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Printed electronics based on inorganic conductive nanomaterials and their applications in intelligent food packaging

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The diverse demands of consumers for packaging functions and increasingly complex product circulation systems have spurred the development of intelligent food packaging (IFP). Besides the basic functions of traditional food packaging, which include the protection of food, sales promotion and convenient transportation, the developing area of IFP can detect the condition of food (ambient temperature, humidity, corruption degree, etc.) in one whole product cycle, and record and feedback information regarding the quality of the packaged food to form a complete product monitoring system. Recently, cutting-edge printed electronics (PE) technology has opened new opportunities for the realization and expansion of IFP functions. Here we introduce some new printed sensors and radio frequency identification (RFID) tags, which are used to form an IFP system. Since conductive ink is the cornerstone of PE, in this paper, the synthesis and properties of inorganic conductive nanomaterials are also reviewed. The formulation of conductive ink, sintering methods and the flexible substrates used in PE are also discussed. Focusing on the state-of-the-art application of PE in the field of IFP, the purpose of this article is to provide a review of IFP that integrates emerging PE technology with prospective next-generation IFP systems to sense, detect, and record feedback information on products in the supply chain environment.

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1. Introduction

Detecting the freshness and safety of food has become a significant challenge in the global supply chain and with the demand for minimally processed food products. Therefore, there is a continued demand for new analytical technologies that can detect the ambient environment and freshness of packaged food in a more time-effective manner. The continuing evolution of large-area radio frequency identification (RFID) and printed sensor production is spurred by the development of printed electronics (PE). Once we reduce the cost of such packaging, RFID and printed sensors can be applied to food packaging. The synergistic integration of intelligent food packaging (IFP) with PE technologies, which has transitioned from science fiction to an area of focus, satisfies the need for the protection of food from environmental pollution, and plays an active role in the quality and safety of food. It prolongs the shelf life using functions that traditional packaging cannot realize, for example judging the freshness of food, the authenticity of drugs, the environmental factors (temperature, humidity, light etc.) in circulation, and so on.^{1–4} Recent trends in IFP include: (i)

theoretical research on internal environmental factors of packaging that impair the shelf life of food, such as gas composition, temperature, microorganisms, etc.; (ii) materials or strategies that are optimized to detect changes in products and to provide effective information to the packaging users (consumer or production decision-makers); and (iii) challenges such as cost minimization that still lie ahead. This review is an effort to familiarize readers with the great breakthrough of IFP in protecting food quality and safety in the dynamic and diverse circulation environment.

The burgeoning PE technology provides a possibility for the realization of IFP. As an interdisciplinary technology between printing and electronics, PE uses a printing process, such as inkjet printing, screen printing, nanoimprinting or soft lithography, etc. to prepare electronic circuits on a flexible substrate.^{5–11} To date, low-cost has been the main driving force in the development of the electronics. For a cost-effective and high-volume process, a printing method like the roll-to-roll (R2R) manufacturing platform has been used to fabricate PE that is becoming ultra-light and ultra-thin.^{8,12} During the past five decades, microelectronics has dominated electronic technology. Table 1 shows the differences between PE and microelectronics. The complexity and huge investment in hard electronic devices made by traditional microelectronic technology is incompatible with the low-cost, high throughput production which is necessary for practical applications in

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Table 1 A comparison between PE and microelectronics

	PE	Microelectronics
Advantages	Light, can be flexible when it prints on flexible substrates, easily distributed, greatly reduced thickness with invariant integration level, low cost	No onerous requirements of mechanical properties, fast running
Disadvantages	High requirements for mechanical properties and encapsulation	Heavy, hard and dispersed
Manufacturing process	Print, R2R	No mass production, high cost, low efficiency, environmental pollution
Substrates	Flexible substrates (plastic, paper, <i>etc.</i>)	Rigid substrates (silicon substrate, glass substrate)

packaging. Conversely, PE can be applied. For example, RFID has been widely used for electronic labeling in packaging.

Conductive ink is the foundation stone for the fabrication of PE and offers good performance as shown in Fig. 1. In this figure, a complete process is illustrated, from the preparation of conductive ink, to the printing process and applications in IFP. The preparation of high-quality conductive ink is a recent research hotspot. Nano-conductive ink is a functional ink developed rapidly with modern science and technology. It is a kind of multi-component dispersion in solvent, containing mainly resins used as a binder to adhere to the substrates, oils as the carrier or medium of the nanomaterial additives to control the rheological properties of the ink, and nanomaterials as the conductive fillers.^{12–16} In this review, the synthesis and stabilization of two kinds of inorganic conductive nanomaterials (metal nanoparticles (NPs) and carbon nanomaterials), and sintering processes are briefly introduced.

The review will focus on the latest progress in the utilization of PE for the fabrication of various IFP. The main focus of the present study is the utilization of PE to realize new IFP with diversified functions. We will explore the issues around expanding the diversity of functions while reducing the additional cost of packaging. In brief, parallel activities of laboratory research (developing new applications of PE in IFP) and commercialization (reducing cost and increasing production efficiency) are needed to promote the development of new functions and the commercialization of IFP. We hope that PE can be better integrated into IFP to broaden its application prospects and increase market capacity.

2. Printable conductive nanomaterials

2.1 Conductive ink: requirements and conduction mechanism

The formulations and preparation methods of conductive inks are crucial if they are to demonstrate good printability, with the appropriate viscosity and surface tension as well as adhesion to specified substrates.^{13,17,18} However, several key characteristics, such as the viscosity and ink particle size, should be adjustable to fit different printing methods. Table 2 (ref. 19) shows a comparison of different printing methods (piezoelectric inkjet

printing, aerosol jet printing, screen printing, gravure printing, flexographic printing, gravure offset printing and letterpress printing) with respect to several technical parameters (optimum range of viscosity, single printing ink film thickness, pattern resolution and printing speed). It is desirable to prepare nanosheet (NS) and nanoparticle (NP)-based conductive inks with well-enhanced electrical conductivity. There are many factors which can affect the conducting performance of ink, such as its composition and the post-processing methods used. Moreover, the mechanisms, which mainly include percolation, the tunnel effect and the field emission effect, can change with the different ink types and the concentration of conducting fillers. Fig. 2 illustrates the conducting mechanism of conductive ink from the colloid state to solidified film printing on the substrate. The conductive ink remains insulating unless the solvent volatilizes and promotes stable binding between the conductive NPs and the matrix (this process is often denoted as the drying of the ink).

After the drying of the conductive ink on the substrate, the conductive NPs mainly exist in three states: a continuous contact state forming a current path termed as percolation; a partly continuous contact state, in which thermal vibration causes electrons to drift through the gaps between the NPs and create an electronic pathway (namely, the tunnel effect); and a completely discontinuous state in which the particles cannot form an electronic pathway because of large gaps between the NPs.^{19–21}

2.2 Metal NPs

2.2.1 Synthesis of metal NPs. A large number of conductive NPs are a basic prerequisite for conductive ink. Metals with high conductivity are the best candidates for conductive materials. Silver (Ag) ($\rho = 1.586 \Omega \text{ m}$, 20 °C), copper (Cu) ($\rho = 1.678 \Omega \text{ m}$, 20 °C), gold (Au) ($\rho = 2.40 \Omega \text{ m}$, 20 °C), aluminum (Al) ($\rho = 2.46548 \Omega \text{ m}$, 20 °C) and nickel (Ni) ($\rho = 6.84 \Omega \text{ m}$, 20 °C) are mostly used as conductive nanofillers, among which Ag is the most promising both in the laboratory and in practical industrial applications.^{22–24} Currently, metal NPs can mainly be synthesized by two means: top-down methods and bottom-up methods.

Top-down methods use physical methods (mechanical grinding, laser ablation, physical vapor deposition, *etc.*) to break



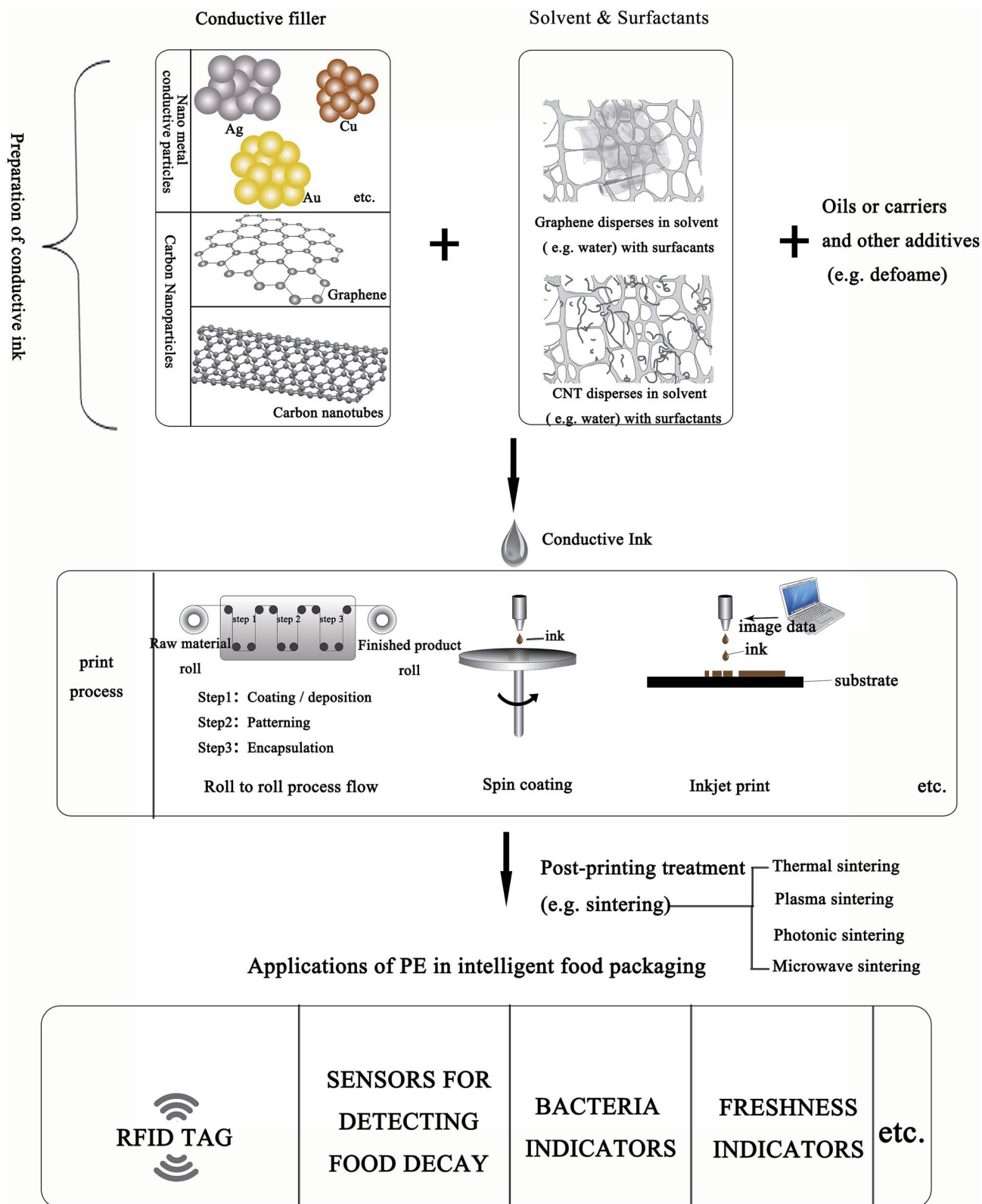


Fig. 1 Summary of the preparation of conductive ink based on inorganic conductive nanomaterials, and the applications of PE in IFP.

bulk metal into nanoscale particles. The most common method among several top-down methods is physical vapor deposition. The metal vapor is prepared by plasma excitation or thermal

heating of the bulk metal, by which the metal wires or metal powders vaporize and condense rapidly through a stream of inert gas like N_2 or Ar in the solution. Wang *et al.*²⁵ have



Table 2 Comparison of different printing technologies

Printing technology	Optimum range of viscosity/cP	Single printing ink film thickness/ μm	Pattern resolution/ μm	Printing speed/ $(\text{m}^2 \text{s}^{-1})$
Piezoelectric inkjet printing	5–20	0.05–1	>20	≤ 0.5
Aerosol jet printing	1–1000	0.1–5	>8	≤ 0.01
Screen printing	500–10 000	≤ 100	>50	0.1–10
Gravure printing	50–500	0.8–8	>20	3–60
Flexographic printing	50–200	0.8–2.5	>50	3–30
Gravure offset printing	>50	0.5–6	>20	0–30
Letterpress printing	50 000–150 000	0.5–1.5	>50	0.5–2

reported the preparation of Ag NPs with sizes between 2 nm and 6 nm by plasma vapor deposition. However, the need for sophisticated equipment and severe operating conditions hamper the commercialization of this high-energy-consumption method.

Conversely, bottom-up routes decompose the precursor in solution or reduce it in the presence of a reducing agent to form a metal ionic liquid which will form unique nanostructures by self-assembly.²⁶ Fig. 3 shows the synthetic process for metal NPs using a chemical reduction method. The bottom-up routes make it possible to prepare a large number of different sizes of metal NPs by adjusting parameters such as the types and dosages of the reducing agent, pH and temperature, to affect the growth of the crystals and finally control the sizes, shapes and distribution properties of the products. The most commonly used reductants include borohydride,²⁷ hydrazine,²⁸ ascorbic acid²⁹ and citrate,³⁰ *etc.* Tippabattini Jayaramudu

*et al.*³¹ have synthesized nano-Ag particles with a uniform average size between 10 and 12 nm using silver nitrate as the precursor and polyethylene glycol as the reducing agent. The advantages of this method which include the production of uniform microstructures and low energy consumption have impelled researchers to put more effort into finding the optimal synthetic parameters. Compared with physical methods, the bottom-up routes are more suitable for the commercial application of IFP.

2.2.2 Stabilization of dispersions. Brownian motion can often cause metal NPs that are dispersed in a solvent to cohere and precipitate during collisions. It is a major challenge in the field of PE to stabilize the metal NPs in the solvent and to prolong the shelf life of the conductive ink. Scientifically, there are two main stabilization methods used to solve this problem, namely electrostatic stabilization and steric stabilization.^{32–34} The most effective way is to add stabilizers into the metal NP

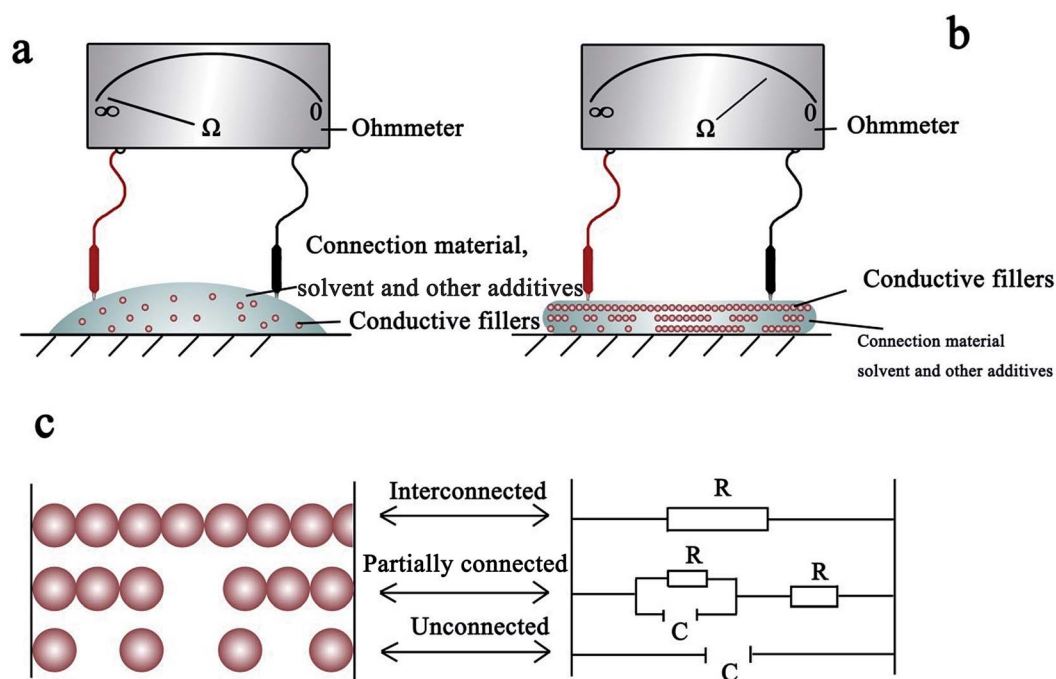


Fig. 2 A schematic diagram of the conduction mechanisms of conductive ink: (a) less conductive particles, where the residual solvent barrier between conductive particles and charge transfer resistance are very large, (b) conductive particles in close contact, where the resistance is lower, and (c) the three connection states of the conductive ink.



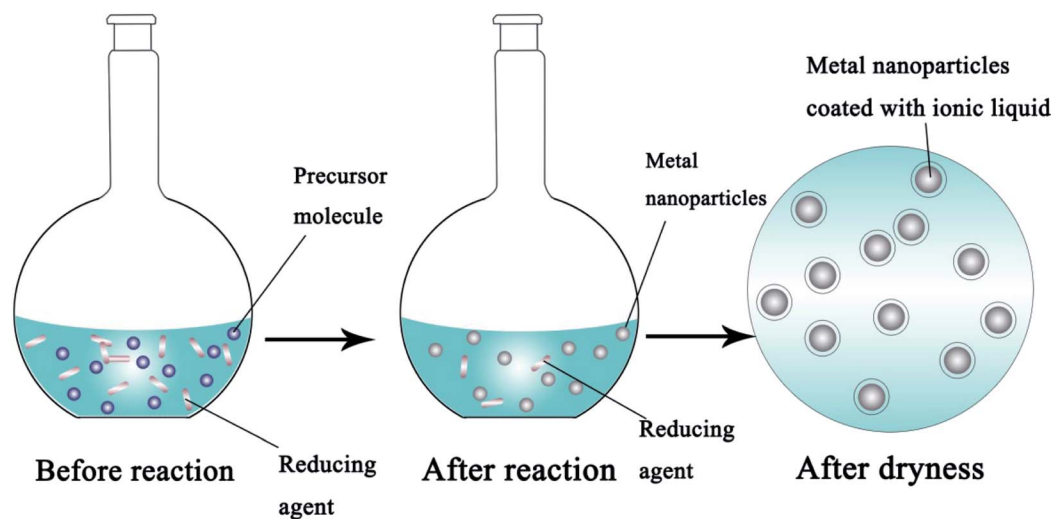


Fig. 3 Schematic diagram of the "bottom up" route of metal NP synthesis by a chemical reduction method.

preparation process while the embryo of metal is still growing. The addition into the solution of ionic-free amphiphilic polymers containing both a hydrophobic portion and a hydrophilic one can form stable conductive inks. Moreover, it enables a higher concentration of metal NPs to disperse effectively in the solution. Polyvinylpyrrolidone (PVP),³⁵ sodium carboxymethyl cellulose,³⁶ naphthalene sulfonic acid, formaldehyde condensate, polyacrylate *etc.* are the most commonly reported stabilizers.³⁷ For example, Mihwa Seo *et al.*³⁸ have studied how the types of organic stabilizer (in neodecanoic acid-stabilized Ag NPs (Ag-ND), hexyl amine-stabilized Ag particles (Ag-HA) and polyvinylpyrrolidone-stabilized Ag particles (Ag-PVP)) influence the ink's resistivity. They concluded that Ag-ND has the lowest resistivity ($3.5 \mu\Omega \text{ cm}$) compared to $15 \mu\Omega \text{ cm}$ for Ag-PVP, and 4.0 and $5.7 \mu\Omega \text{ cm}$ for Ag-HA1, and Ag-HA2 (larger sized NPs). Fig. 4 shows the SEM images of the four nano Ag films made using different stabilizers at different sintering temperatures.

2.2.3 Stabilization against oxidation. As a low-priced metal, Cu is an alternative to the noble metals such as Ag and Au which have low-melting-temperatures together with high conductivity (10^{-3} to $10^{-2} \Omega \text{ cm}$). However, Cu is prone to oxidation which causes a deterioration in resistivity and a higher sintering temperature.²² Accordingly, concerted efforts are being made to search for effective methods to stabilize Cu against oxidation. One of the most common ways is to form a protective coat around Cu NPs in solution to protect them from being oxidized; another method is to add antioxidants such as ascorbic acid. Cheng *et al.*³⁹ presented a method of synthesizing Cu NPs with an average size of 140 nm which remained in a stable state for 3 months without oxidation at room temperature. They used PVP as a capping agent and ascorbic acid as an antioxidant. Nevertheless, the aforementioned approaches can only retard the oxidation of metal NPs. More effective ways to prevent from oxidation need to be developed. Recently, some research groups have synthesized a kind of antioxidative Cu NP with a dense shell. Yim *et al.*⁴⁰ fabricated hybrid copper-silver-graphene (CSG) nanoplatelet conductive inks which showed the desired

oxidation resistance even at high temperatures. This CSG film displayed an electrical resistance of a few ohms at 190°C . Pajor-Świerzy *et al.*⁴¹ recently presented a kind of conductive ink based on Cu and Ag core-shell particles. It was air stable with a lowest resistivity of $6.9 \pm 0.7 \mu\Omega \text{ cm}$ that corresponds to 25% of the conductivity of bulk Cu.

2.2.4 Sintering and resistivity. A sintering process is often needed to obtain high electrical conductivity that parallels that of the bulk metal. Thermal sintering with temperatures ranging from 150°C to 300°C is a widely used method. Fig. 4 shows the effects of different sintering temperatures on the microstructures of NPs. The result shows that the density increases rapidly with rising temperature from 100 – 220°C . Table 3 shows the sintering temperature and the conductivity of various metallic nano-conductive inks for PE. Photonic sintering^{42,43} (which is a high-temperature thermal processing of thin films using a broad-spectrum pulse of light from a flash lamp), plasma sintering⁴⁴ (which needs sophisticated equipment and compatibility with plastic substrates) and microwave sintering⁴⁵ (which has a small penetration depth) are also commonly reported methods. However, the biggest challenge arises where patterns need to be printed on flexible substrates such as plastic film that cannot withstand severe conditions. Hirotaka Koga *et al.*⁴⁶ developed a rapid low-temperature sintering method to fabricate Ag conductive ink for PE. The as-printed films were sintered at 75°C and yielded high conductivities of $2.74 \times 10^6 \text{ S m}^{-1}$. In addition, the optimal sintering temperature strongly depends on the ligands. For example, the optimal sintering temperatures for Ag NPs are 140°C (ref. 47) and 450 K (ref. 48) respectively for PVP and poly(diallyldimethylammonium chloride) ligands. Anni⁴⁹ used xylene as the dispersant to prepare Au NP conductive ink with the best sintering temperature of 180°C , and Yun⁵⁰ prepared Ag inks with an optimal sintering temperature of 70°C using a hydroxyethyl cellulose adhesive agent. Ligands can make conductive NPs more uniform and stable in the dispersion system, but the presence of ligands prevents the transfer of charge between conductive particles



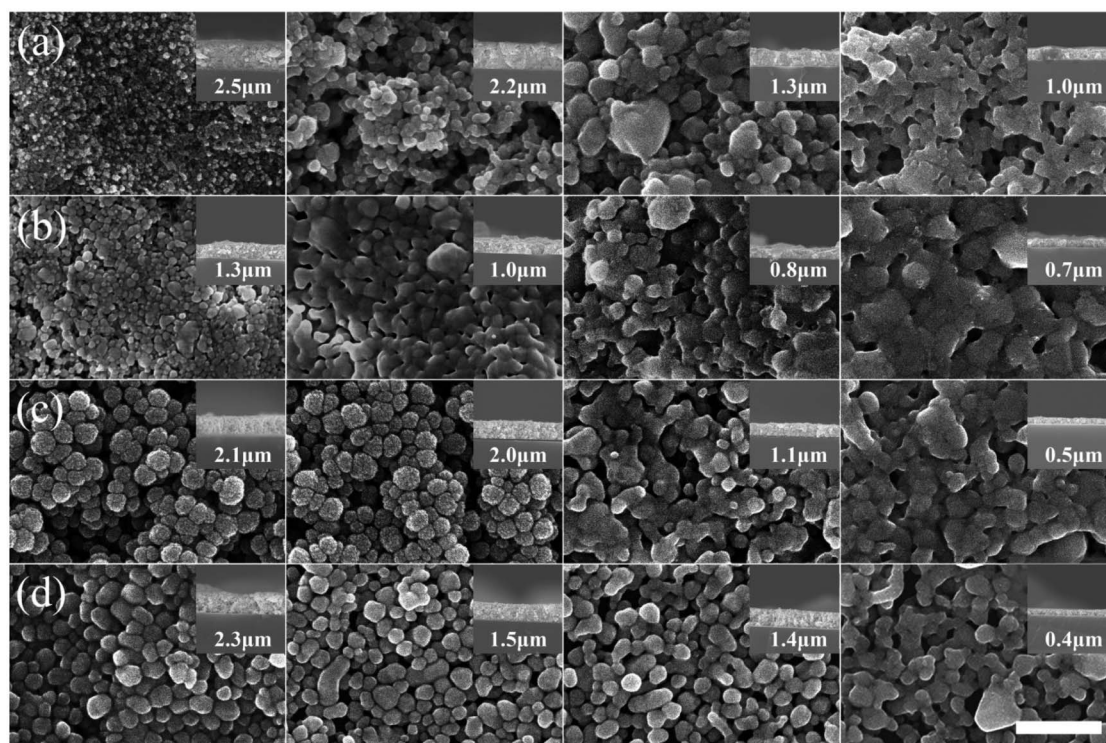


Fig. 4 SEM images (top and side views) of nano Ag films: (a) Ag-ND, (b) Ag-HA1, (c) Ag-HA2, and (d) Ag-PVP sintered at various temperatures (left to right: 100 °C, 140 °C, 180 °C, and 220 °C); scale bar = 500 nm [reprinted with permission from ref. 38, M. Seo, J. S. Kim, J. G. Lee, S. B. Kim and S. M. Koo, The effect of silver particle size and organic stabilizers on the conductivity of silver particulate films in thermal sintering processes, *Thin Solid Films*, 2016, **616**, 366–374. Copyright© Elsevier].

and increases the resistance of printed circuits, so different ligands have different sintering temperatures.

2.3 Carbon NSs and NPs

Because of their excellent chemical, electrical, mechanical, and optical properties as well as their abundance, graphene and carbon nanotubes (CNTs) are emerging as promising candidates for metal NPs.^{6,9,12,13,18,55–57} Since 2004, more and more of the literature has focused on the preparation of graphene and CNTs. Where and how we can apply graphene in new fields and for new functions has become a wide-spread challenge.

2.3.1 Graphene. Graphene is 2D structured material with a single atomic layer of sp^2 hybridized carbon atoms.^{58–61} There are several methods used to synthesize graphene: mechanical

exfoliation, liquid exfoliation (LE), microfluidization, solid exfoliation, oxidation–exfoliation–reduction and chemical vapor deposition (CVD), *etc.*^{22,62}

2.3.2 Synthesis of graphene. Initially, graphene was prepared by mechanical exfoliation from graphite.^{9,63–65} Due to the weak interlayer attractions in graphite, graphene can be easily exfoliated with simple equipment.^{6,66} However there are several disadvantages of this method: (i) the degree of exfoliation must be controlled carefully; and (ii) it has a low production efficiency which makes it unsuitable for large scale commercial manufacture.

Oxidation–exfoliation–reduction is the most commonly reported method used to prepare graphene. As shown in Fig. 5, firstly, strong oxidizers⁶⁵ are used, such as fuming nitric acid,^{67,68} potassium perchlorate⁶⁹ or potassium permanganate⁷⁰ to oxidize the graphite. Oxygen-containing groups such as carboxyl and hydroxyl are formed, causing an increase in the interlayer distance of graphite from 3.35 Å to 7–10 Å.^{71–74} The as-prepared graphene oxide (GO) can then easily be exfoliated by ultrasonic treatment, and finally graphene can be prepared by reduction. For example, Hong *et al.*⁷⁴ prepared graphene ink with the Hummers' method using hydrazine as a reducing agent under anoxic conditions. Then, they mixed the as-prepared graphene with cellulose acetate, acetone, *n*-butyl pyridinium hexafluorophosphate and cyclohexanone, and sonicated for a given amount of time. They fabricated electrodes for Cd^{2+} and Pd^{2+} determination using the as-prepared graphene

Table 3 Sintering temperatures and conductivities of various metal NP inks for PE

Conductive particles	Sintering temperature	Conductivity	Ref
Ag NPs	300 °C	3.2 $\mu\Omega$ cm	47
Cu ion	90 °C	$2.3 \times 10^{-6} \Omega$ cm	51
Cu NPs	150 °C	3.23 $\mu\Omega$ cm	44
Au NPs	180 °C	$1.6 \times 10^{-7} \Omega$ m	49
Zn NPs	25 °C	$3 \times 10^5 S m^{-1}$	52
Ni NPs	150 °C	76.34 $\mu\Omega$ cm	53
Cu/Ag NPs	100 °C	5.12 $\mu\Omega$ cm	54



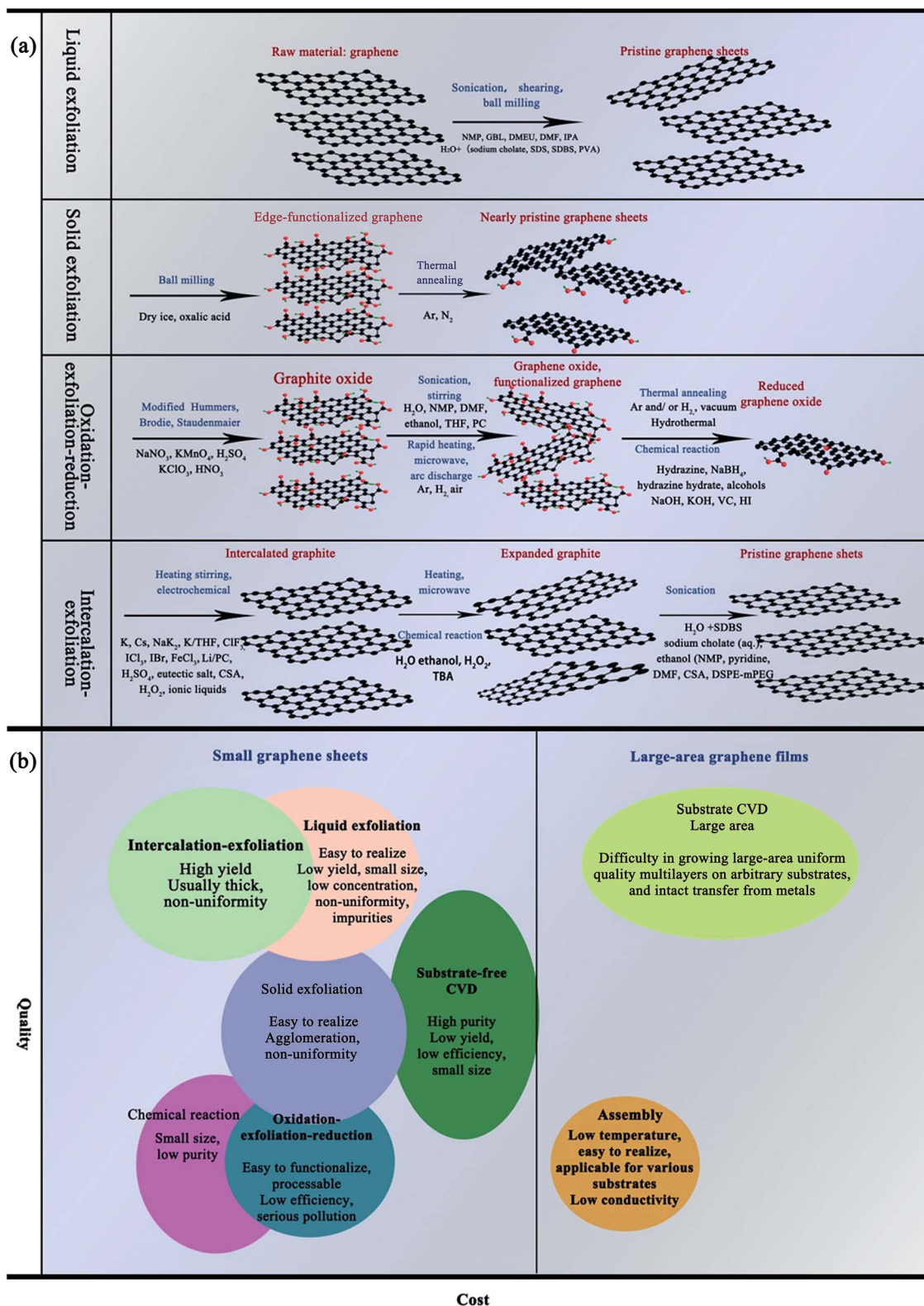


Fig. 5 (a) Schematic diagram of the synthesis of graphene from graphite. (b) Comparison of the quality and cost of graphene products manufactured by different methods [reprinted with permission from ref. 75, W. Ren and H. M. Cheng, The global growth of graphene, *Nat Nanotechnol.* 2014, 9, 726–730. Copyright© Springer Nature].



ink by screen printing and obtained an excellent transfer resistance ($430\ \Omega$). Fig. 5 shows several methods used for synthesizing graphene from graphite, and compares the quality and cost of the graphene products manufactured by the different methods.

LE is another method used to produce graphene. The process involves ultrasonic treatment of the graphite dispersed in solvent followed by centrifugation to obtain graphene.⁷⁶ Arhin *et al.*⁷⁷ prepared inkjet-printed graphene ink using graphene produced by ultrasonic-assisted liquid phase exfoliation. They ultrasonicated 100 mg graphite flakes with 1.5 mg PVP in 10 mL isopropanol for 12 h at $15\ ^\circ\text{C}$ and then centrifuged at 4000 rpm for 1 h. As shown in Fig. 6, graphene was obtained with an average flake thickness of $5.9 \pm 0.2\ \text{nm}$ and with 56% of flakes being 4 nm thick. Recently, microfluidization of graphite (which is similar to LE) has been developed by stripping graphite in aqueous solutions with high shear rates. This is a simple and scalable production route for conductive inks for large-area printing in flexible electronics.

Besides, CVD is a method commonly used in industry to produce high-quality graphene sheets with large areas. Osikoya *et al.* fabricated a CVD graphene sheet grown from acetylene as the carbon source, with a large available surface area, while the sheet resistance was as low as $125\ \Omega\ \text{sq}^{-1}$.^{78–80} However, the biggest challenges for this technique are the high cost and complex processes involved.

2.3.3 CNTs. CNTs as another nanomaterial have been utilized in fabricating conductive inks because of their extraordinary electrical, optical, and mechanical properties.^{81–83} He and Tjong⁸ prepared a kind of conductive ink containing 0.1 wt% multi-walled CNTs (MWNTs) and 0.2 wt% GO. The conductive ink was utilized in the scalable production of conductive films with a sheet resistance of $380\ \Omega\ \text{sq}^{-1}$ and 85% transparency. Shin *et al.*⁸³ developed a kind of CNT-based conductive ink that could be utilized in the fabrication of 2D flexible electronics. They used DNA as the surfactant to disperse the CNTs in methacryloyl and hyaluronic acid to fabricate eco-friendly conductive ink. Flexible electronics using this kind of ink displayed a 25% variation in the measured resistance even with a 180° folding angle. Liao *et al.*¹³ reported a highly conductive carbon-based aqueous ink for electroluminescent devices, printed capacitive sensors and flexible wearable electronics.

2.3.4 Stabilization of dispersed carbon NS and NP-based conductive inks. Although promising for application, the poor dispersibility of graphene and CNTs is a stumbling block in the field of conductive inks and hinders the realization of high conductivities. Theoretically, graphene and CNTs are apt to aggregate and precipitate irreversibly in solvent due to van der Waals forces and π - π stacking between the graphene lamellae. The methods used to promote the dispersion of graphene can

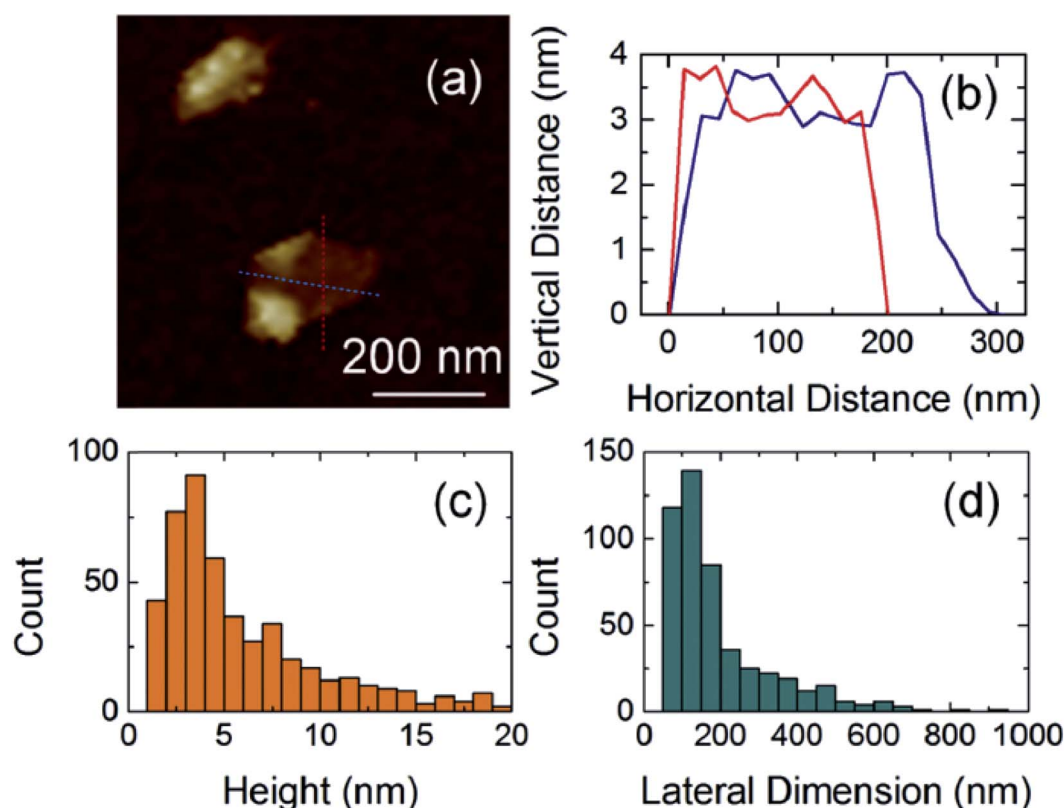


Fig. 6 Atomic force microscopy characterization of dispersed graphene flakes: (a) image and (b) section of a typical graphene flake. Distributions of (c) flake thickness and (d) flake lateral dimensions [reprinted with permission from ref. 77, D. Dodoo-Arhin, R. C. T. Howe, G. Hu, Y. Zhang, P. Hiralal, A. Bello, G. Amaratunga and T. Hasan, Inkjet-printed graphene electrodes for dye-sensitized solar cells, *Carbon*, 2016, **105**, 33–41. Copyright© Elsevier].



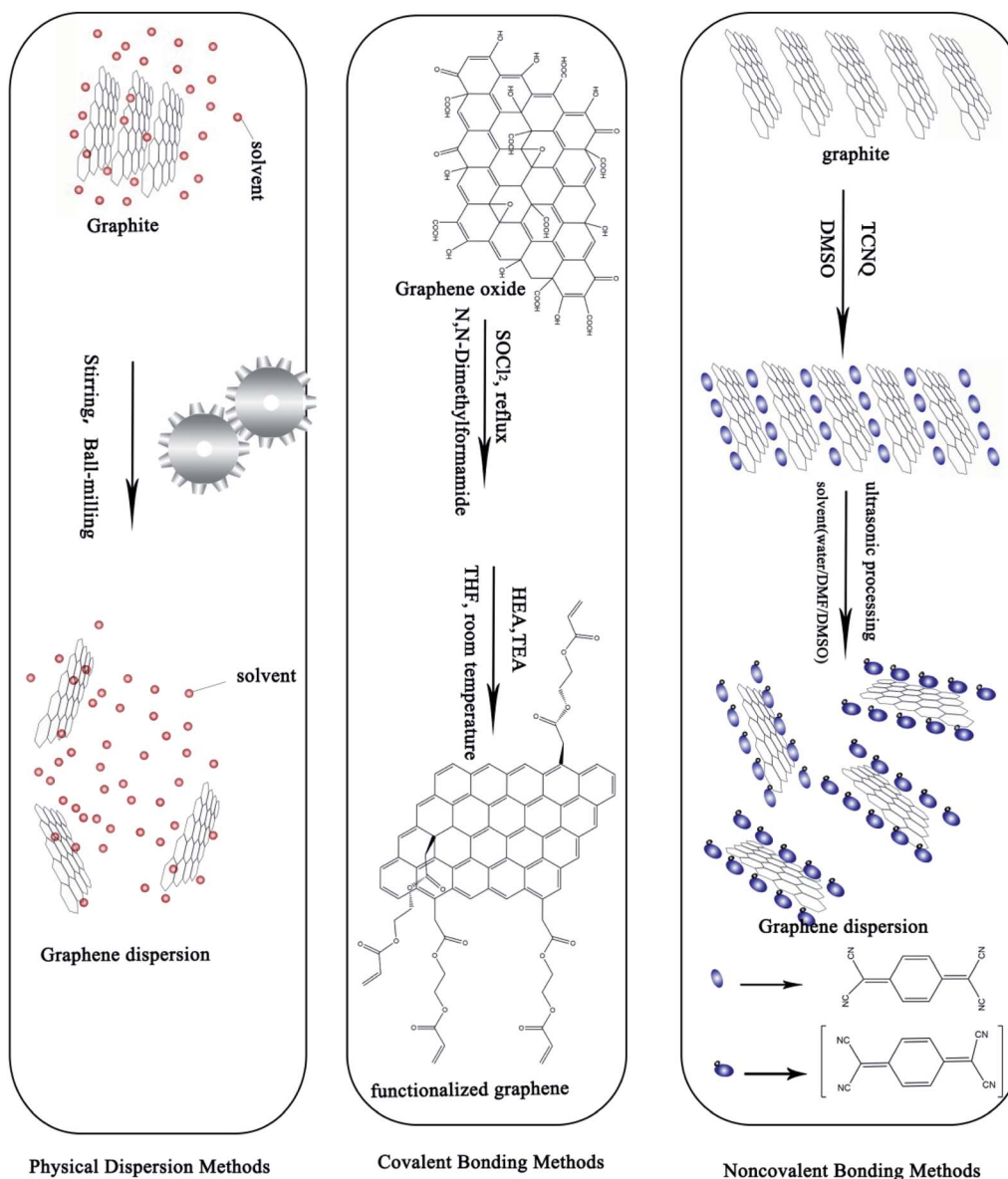


Fig. 7 Schematic diagram of the dispersion methods used for graphene.

be divided into three categories, namely physical dispersion methods, noncovalent bonding methods and covalent bonding methods (Fig. 7).

Physical methods encompass ultrasonic treatment, stirring, and ball-milling. Baomin Wang *et al.*⁸⁴ prepared a graphene nanoplatelet suspension by ultrasonic treatment. They used a minimum sonication time to effectively disperse the 0.1 g L^{-1} concentration of bundled graphene nanoplatelets. Weifeng Zhao *et al.*⁸⁵ used ball-milling to exfoliate graphene in a variety of organic solvents such as N,N -dimethylformamide, at a concentration of up to 0.08 mg mL^{-1} , achieving a yield higher than 32.0 wt%. With low cost, simple operation and the possibility of application to large scale production, physical methods are a popular choice among many researchers. However, physical methods cause damage to the structure of graphene and produce a low dispersion rate which is unsatisfactory.

Noncovalent bonding methods based on π - π interactions and surfactant modification can form a stable dispersion of graphene. π - π interactions make use of π bonds in aromatic molecules (polycyclic aromatic hydrocarbons (PAHs), peptides, *etc.*) to stabilize the graphene dispersion. Jun Wang *et al.*⁷⁷ prepared a kind of graphene dispersion using PAHs. They revealed that after loading with PAHs, the conformation and aggregation of the graphene and GO nanosheets dramatically changed. Surfactant addition is the most common way to prepare stable graphene dispersions. To prepare stable conductive ink various surfactants such as PAH and sodium dodecyl sulfate (SDS)⁸⁶ are needed together with a solvent like water or N -methylpyrrolidone, *etc.* Song *et al.*⁸⁷ prepared a kind of graphene ink powder containing graphene and excess ethyl cellulose (2 : 3, w/w) dispersed in terpineol with 20% w/v of solid in the solvent. They used this graphene





Table 4 Components and conductivity properties of several carbon NP and NS inks

Ink type	R_s (Ω sq ⁻¹)	Film thickness or transparency	GNP or CNT content/wt%	Substrate	Printing method	Post-printing treatment	Ref.
LE graphene in aqueous media using a water soluble cellulose stabilizer mixed with 2, 3-butanediol	300 Ω sq ⁻¹	2.2 μ m for 20 prints	8.5 mg mL	Glass and polyethylene terephthalate (PET) substrate	Ink-jet printing	Dried at 100 °C in a vacuum oven for 1 h	93
Graphene electrochemically exfoliated in (NH ₄) ₂ SO ₄ electrolyte and dispersed in DMF	1.81 k Ω sq ⁻¹	16.0 nm with 80% transparency	0.1 mg mL ⁻¹	PET	Dry transfer	Low temperature annealing (<i>i.e.</i> , 300 °C)	94
Graphene electrochemically exfoliated in (NH ₄) ₂ SO ₄ electrolyte and dispersed in ethanol and water	5–10 k Ω sq ⁻¹	15–20 nm	—	SiO ₂ /Si substrate	Ink-jet printing	Annealing at 350 °C	56
LE graphene in isooctane using a diblock polymer containing 11 cholesterol units (CEM) and 7 ethyl-hexyl acrylate units (EHA) as dispersant	80 Ω sq ⁻¹	6–7 μ m	4 mg mL	Glass substrates and paper	Direct writing by a ballpoint pen loaded with graphene ink	Instantly dry	95
Intercalated EG dispersed in a copolymer of PVP and polyvinyl acetate (PVAc) (for simplicity termed the binder) in isopropanol	30 Ω sq ⁻¹	25 μ m	53 mg mL	PET	Screen printing	Dried at 100 °C for 5 min	96
Single-walled CNTs dispersed in aqueous solution dissolved with sodium dodecylsulphonate	80 Ω sq ⁻¹	27 μ m and 80% transmittance	0.1 wt%	PET	Rod-coating	No post-printing treatment required	97
Few-walled CNTs in pure isopropanol without surfactant	4.75 Ω sq ⁻¹	1 μ m	2–2.5 mg mL ⁻¹	Heated glass substrate	Spray-coated	Annealed at a temperature of 120 °C for 1 h	98
MWCNTs and Ag NPs in water with the assistance of sodium dodecylbenzenesulfonate (SDBS)	300 Ω sq ⁻¹	50 prints	5 mg mL ⁻¹ MWCNTs and 4 mg mL ⁻¹ Ag	Paper	Ink-jet printing	No post-printing treatment required	99
Urea functionalized SWNTs dispersed in water without surfactant	Range from 15 to 1700 Ω sq ⁻¹	Transmittance range of 2–80% at 550 nm	—	Glass	Rod-coating	No post-printing treatment required	100

ink to print conductive graphene patterns on plastic with high-resolution (3.2 μm line width and 1 μm spacing) using a transfer printing method. In addition, carbon NSs and NPs will not melt during heat treatment, in contrast to metal NPs, but the dispersants existing in the carbon NSs and NPs need heat treatment for removal. Secor *et al.*⁸⁸ heat treated graphene ink at 250 $^{\circ}\text{C}$ for 30 minutes to achieve a resistance of 4 $\text{m}\Omega\text{ cm}$. Lee *et al.*⁸⁹ reported that sintering printed graphene NSs (GNSS) at 400 $^{\circ}\text{C}$ could effectively lead to the elimination of excess SDS on the GNSS and increase the C=C content. However, these organic solvents are toxic. Developing non-toxic stabilizing agents and surfactants or modified graphene or CNTs with excellent dispersivity in water has become the subject of many studies in the literature. Michel *et al.*⁹⁰ synthesized a kind of surfactant-free graphene ink in a mixture of terpeneol and cyclohexanone. Furthermore, the conductive ink had an excellent electrical conductivity with a resistance of only 1.1 $\text{m}\Omega\text{ m}$ compared with 7.1 $\text{m}\Omega\text{ m}$ for the surfactant-assisted ink (*N*-methyl-2-pyrrolidone and ethyl cellulose). This formulation is very eco-friendly and has excellent conductivity because of the biomaterials, terpeneol and cyclohexanone. Koutsoukis *et al.*⁹¹ recently developed a highly conductive formulation using pristine graphene and dihydroxyphenyl-functionalized MWNTs mixed with another hydrophilic polymer which displayed desirable conductivity ($R_s = 22.6\ \Omega\ \text{sq}^{-1}$, $\sigma = 4971\ \text{S m}^{-1}$).

Covalent bonding methods are most commonly used to functionalize graphene and CNTs. Attachment of a large number of hydroxyl, epoxy, carboxyl, or carbonyl groups to the surface of GO significantly improves the solubility and dispersion stability of the graphene in most common organic solvents.

2.3.5 Electrical conductivity of carbonaceous nanomaterials. It is widely believed that the low electrical conductivity of carbonaceous nanomaterials constrains their application. Commonly, charge transport mechanisms are mainly divided into two categories, namely particle contact and the tunnel effect. Key factors that determine the integrity and degree of compactness of a conducting circuit are the number of contacting particles, and the distance between the conductive particles. The realization of high conductivity conductive inks is closely related to the content, shape, size and polarity of the conductive fillers. Generally conductive materials that are small in size but large in specific surface area are the best candidate for conductive fillers. Therefore, when CNTs with slender fibrous structures and nanoflake graphene materials are chosen as conductive materials, good dispersibility and stability are crucial to the conductivity. At present, many reports are focusing on improving the conductivity of carbonaceous nanomaterials. Table 4 shows the components, conductivity properties and printing methods of several carbon NP inks. Overgaard *et al.*⁹² demonstrated a method for highly conductive semitransparent graphene. Circuits using screen-printing had a low sheet resistance of 327 $\Omega\ \text{sq}^{-1}$ for thin semitransparent layers and 37% transmittance. Some researchers have combined carbonaceous nanomaterials with other conductive materials to get higher conductivity. Koutsoukis *et al.*⁹¹

synthesized a kind of highly conductive water-based polymer/graphene nanomaterial. A conductive ink with a 22.6 $\Omega\ \text{sq}^{-1}$ sheet resistivity and 6459 S m^{-1} conductivity was obtained.

3. Applications in intelligent food packaging

As shown in Fig. 8, IFP technologies can mainly be divided into two categories, one which encompasses diagnostic technologies (time-temperature integrators/indicators, gas indicators, freshness indicators, bacteria indicators, *etc.*) and the other which encompasses communicating technologies (rapid frequency identification (RFID) tags, electronic article surveillance (EAS) tags, electro-magnetic identification (EMID) tags, *etc.*).¹⁰¹

3.1 Diagnostic technologies in IFP using PE

In the diagnostic technologies, printed sensors are usually used as indicators to detect the inner environment of the food packaging.^{1–4,93} Printed sensors offer potential to improve efficiency without expensive equipment, and have been studied in lots of research. However, there are still some obstacles to be overcome: the sensors must be lightweight and flexible, production costs need to be decreased, sensitivity needs to be increased and the sensors must be eco-friendly.¹⁰²

Flexible printed sensors for food monitoring are complex integrated systems, which are typically composed of: (1) conducting electrodes based on conductive ink, (2) flexible/stretchable substrates, such as PI, PET, PEN, and (3) sensing materials. Fig. 9 shows the diagram of a flexible gas sensor in an IFP system. The sensing materials can monitor the environment inside of the package. The chemical signal can be changed into an electrical one by an electrode printed with effective

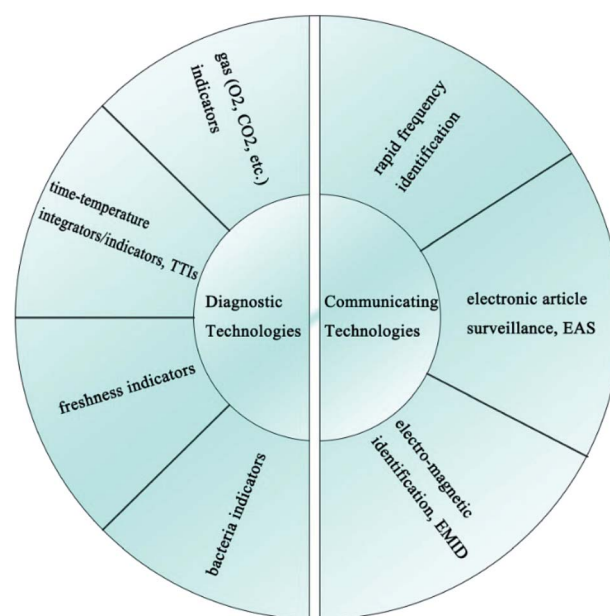


Fig. 8 The main categories of IFP technologies.



conductive ink. The following section presents several flexible sensors that can be applied to IFP.

3.1.1 Gas sensors. A reduction or increase in water content, oxidation reactions and aerobic microorganisms accelerate food decay.^{103–105} The content of the air (two or three constituents out of carbon dioxide (CO₂), oxygen (O₂) and nitrogen (N₂)) and the humidity in the packaging should be adjusted according to the food being packaged and the requirements for freshness. CO₂ is a kind of gas bacteriostat—a low concentration of CO₂ promotes microbiological reproduction, and a high concentration of CO₂ can inhibit the growth and reproduction of most aerobic bacteria that cause food decay.¹⁰⁶ The concentration of O₂ should be kept low in food packaging to reduce both the respiration of fresh fruits and vegetables and the oxidation of food, although a high O₂ content retains the color of fresh meat. An inert gas like N₂ which does not react with food can reduce the oxidation rate of fats and aromatics in food. Table 5 (ref. 107) shows the recommended content of gases in the packaging of various foods.

Gas sensors are usually fabricated so that chemical reactions or changes in acidity lead to physical changes, for example, a change in resistance.¹⁰⁸ They are crucial for controlling and recording changes in gas content and contribute to the extension of product shelf life. For example, Vargas-Sansalvador *et al.*¹⁰⁹ reported the development of a water-based ionic liquid CO₂ sensor which can be printed on food packaging. The

Table 5 Recommendation content of gases in the packaging of various foods

Type of food	O ₂ (%)	CO ₂ (%)	N ₂ (%)
Red meat	60–85	15–40	—
Cooked meat/bacon	—	20–35	65–80
Poultry	—	25	75
Lean fish	—	40	30
Oily fish	—	60	40
Salmon	—	60	20
Hard cheese	—	100	—
Soft cheese	—	30	70
Bread	—	60–70	30–40
Milk free cake	—	60	40
Dairy cake	—	—	100
Fresh spaghetti	—	—	100
Fruit and vegetables	3–5	3–5	85–95

sensor had an excellent optical performance and was especially sensitive to low concentrations of CO₂. Lang and Jedermann¹¹⁰ demonstrated the detection of gas content in packaging for various gases such as O₂, CO₂, alcohols and so on. S. Santoro *et al.*¹¹¹ prepared an O₂ sensor based on the immobilization of tris-1,10-phenanthroline ruthenium which was printed onto a food packaging material to monitor oxygen permeation. This kind of sensor could also be used as a leakage indicator.

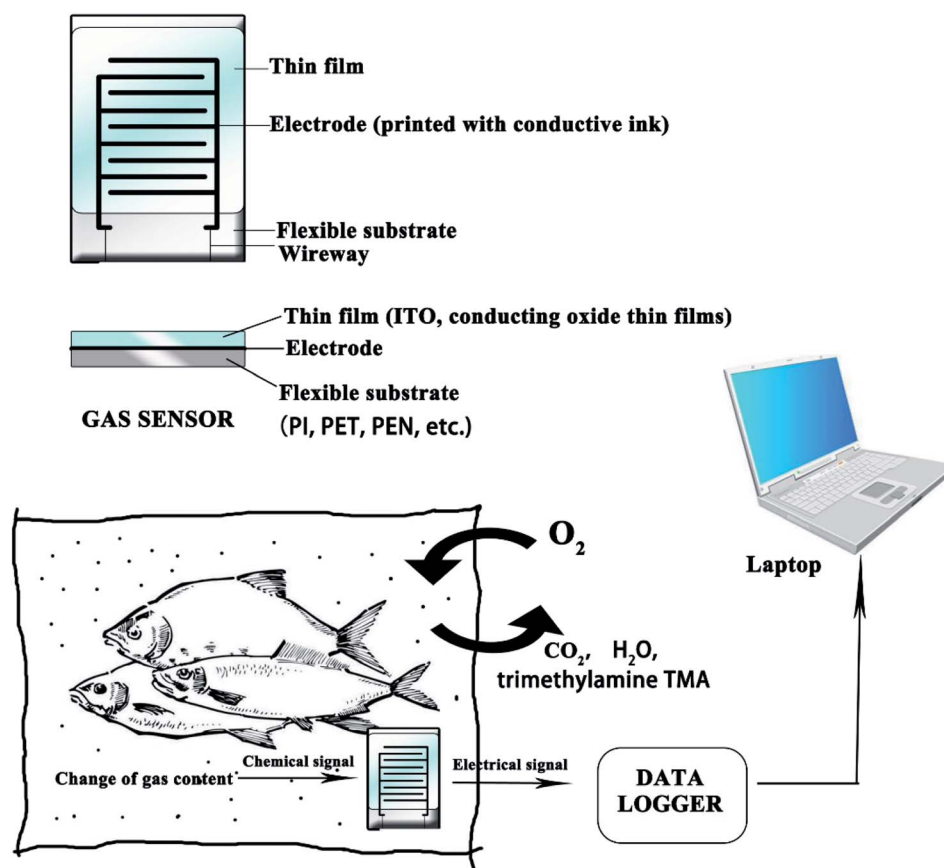


Fig. 9 A diagram of an IFP system based on gas sensors.



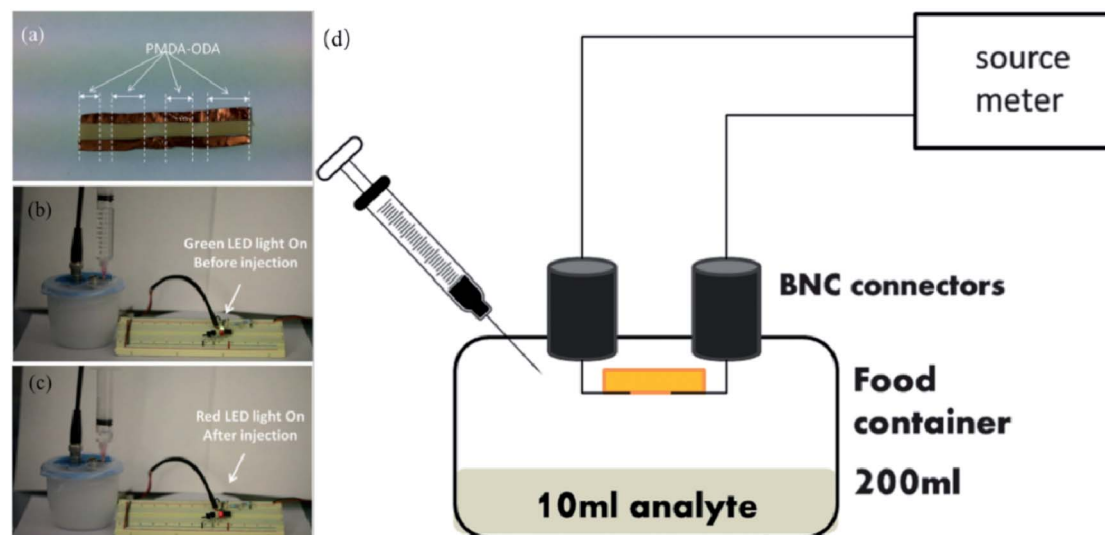


Fig. 10 An NH_3 sensor based on pyromellitic dianhydride-4,4-oxydianiline. (a) The darker parts of cotton is the cured polymer. (b) Before injection of NH_3 solution the green led is on. (c) After the injection of NH_3 solution the red led turns on. (d) Schematic representation of the experimenting set-up [adapted from ref. 112, E. L. Papadopoulou, D. Morselli, M. Prato, A. Barcellona, A. Athanassiou and I. S. Bayer, An efficient pure polyimide ammonia sensor, *Journal of Materials Chemistry C*, 2016, 4, 7790–7797. Copyright© Royal Society of Chemistry].

Papadopoulou¹¹² obtained a highly sensitive ammonia (NH_3) gas sensor for food packaging technology based on pyromellitic dianhydride-4,4-oxydianiline, as shown in Fig. 10. When an NH_3 -releasing solution is injected into the package, the signal indicator of the sensor changes from blue to red. As the freshness of meat is closely related to the change of NH_3 content in the packaging, this sensor would provide a technology for a real-time display of freshness of packaged meat.

3.1.2 Sensors for detecting food decay. Food quality changes can be broken down into two types: microbial metabolites such as toxins, gases, smells and mucus can lead to pH

changes that cause food decay; and fat and pigment oxidation can have a negative effect on food flavor.^{113–115} Food decay is detected by the reaction of the microbial metabolites (organic acids, ethanol, volatile nitrogen compounds and biogenic amines) with components contained in chemical sensors.^{116–118} For example, a titania nanotubular sensor was successfully fabricated on a printed circuit board for the detection of food decay products including dimethyl sulfide from decayed eggs and ethyl acetate from decayed tomatoes. Schaudé *et al.*¹¹⁹ fabricated a kind of colorimetric sensor for NH_3 and biogenic amines in packaging. They selected a pH sensitive dye as the indicator which is induced to change color from green to red when it is exposed to amines during food spoilage. Alreshaid *et al.*¹²⁰ reviewed the gas sensors which can detect certain gases to determine the freshness of packaged foods. Matindoust *et al.*¹²¹ also developed an NH_3 gas sensor printed on flexible polyaniline films for protein-rich foods. Fig. 11 shows this IFP system for detecting NH_3 gas. The voltage of the electrodes through the film is directly proportional to the time and the gas concentration; chemical signals are changed into electrical ones.

Recently, biosensors have also been used to indicate freshness.¹²² Biosensors are made up of biological components (enzymes,¹²³ antibodies¹²⁴ and cells) and physical components that can decode the biological signal and change it to a physical one.¹²⁵ Yang *et al.* presented some different CNT-based biosensors.¹²⁶ For example, enzyme CNT-based biosensors were mentioned in their review. The large specific surface area of CNTs promotes enzyme conjugation so that such sensors can detect glucose and protein by bioreaction. A kind of highly efficient biosensor for glucose detection based on ZnO/ZnS core/shell nanotube arrays (CSNAs) was fabricated by Tarish *et al.*¹²⁷ They demonstrated a method in which their

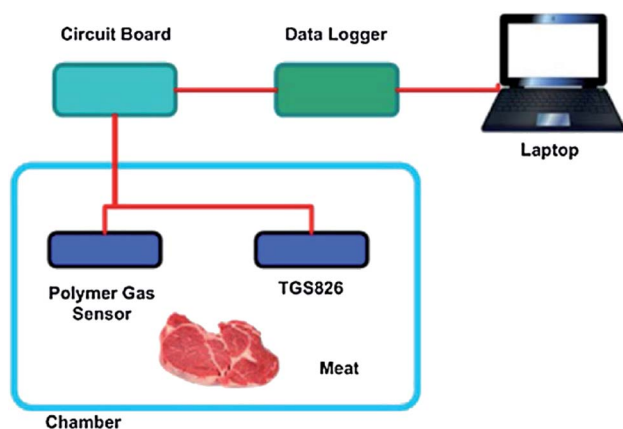


Fig. 11 A schematic representation of an IFP system for monitoring NH_3 gas in a chamber [reprinted with permission from ref. 121, S. Matindoust, A. Farzi, M. Baghaei Nejad, M. H. Shahrokh Abadi, Z. Zou and L.-R. Zheng, Ammonia gas sensor based on flexible polyaniline films for rapid detection of spoilage in protein-rich foods, *Journal of Materials Science: Materials in Electronics*, 2017, 28, 7760–7768. Copyright© Springer Nature].



ZnO/ZnS CSNA electrode gave a higher sensing performance than other reported sensors.

3.2 Communicating technologies in IFP using PEs

RFID technologies are unlike sensors and indicators and fall into another category. RFID tags contain transponders and antennas, satisfying the need of consumers for transparency in the supply chain and providing necessary and effective data for the production and sales decision-makers.^{4,101,128,129} Direct ink-jet printing of antennas on plastic and paper substrates with the use of conductive NP inks is a promising approach to the production of low-cost RFID tags. RFID tags have broad application prospects for the following reasons: (i) they offer a large storage capacity (one to thousands of bytes); (ii) their readable distance through a scanner has a wide range (from a few centimeters to about two hundred meters); (iii) they can be operated without contact, and no particular angle is needed to read information; (iv) they are readable in diverse conditions (even in the dark or under water); and (v) they are recyclable.^{129,130} However, the biggest disadvantage of RFID is its cost, which is a stumbling block to wide-spread commercialization. Using RFID tags in IFP, the link of the supply chain at which the food begins to corrupt can be identified and RFID can speed customer checkout by recognition technology. Recently, many companies have started to use this technology to build a bridge between supermarkets, suppliers and consumers. Suppliers label their goods with RFID tags and once the goods reach the store center of the supermarket, the product labels are immediately scanned and the ID numbers are stored. When customers buy commodities, the ID numbers are deleted from the list of shelf products.

Recently more intelligent RFID tags have been developed, such as sensor-enabled RFID tags to provide information on humidity and temperature. Borgese *et al.*¹³¹ fabricated a novel chipless RFID humidity sensor by inkjet printing. Variations in the relative humidity level from 50% to 90% gave a frequency shift of up to 270 MHz. It was based on a finite artificial impedance surface. Lorite *et al.*¹²⁹ presented an RFID-assisted critical temperature indicator for supply chain monitoring using R2R printing. When the critical temperature that was detrimental to food storage was reached, the resistance rapidly increased (from 1 k Ω to 8 k Ω).

4. Outlook

PE offers wide application prospects in IFP as discussed. However, there are still some challenges and problems to be solved. The following goals remain: (i) decreasing the cost; (ii) simplifying production processes and increasing throughput; (iii) further diversifying applications in IFP; and (iv) developing eco-friendly raw materials with better properties.

Conductive ink plays a crucial role in PE. Optimizing the properties of conductive ink is still a driving force in many literature reports. Nowadays, high-cost Ag ink is still universal in commercially available ink. Therefore, many researchers are focusing on developing low-cost alternative Cu, Al, and Ni

nanomaterials with better properties. However, these types of nanomaterials are easily oxidized, causing deterioration of conductivity so researchers are searching for effective methods to stabilize these nanomaterials against oxidation. The addition of stabilizers is one common approach being considered; a second recent approach is to hinder oxidation of the NPs by, for example, synthesizing NPs with a dense shell that gives them antioxidant properties. Sintering is a key process for improving conductivity. Researchers are developing new sintering methods that avoid destruction of plastic or flexible substrates. For example, low temperature sintering has been studied recently. Low-cost carbonaceous materials are always considered as good candidates for metal-based conductive inks. However, their lower conductivity compared to that of metal NPs is a drawback. Some researchers still consider carbonaceous materials to offer the most prospects because many literature reports describe effective methods to improve their conductivity.

It should be noted that PE integrated with IFP has recently led to more effective functions to protect products. Various sensors for monitoring the changes in food and the inner environment of food packaging have been fabricated. RFID tags satisfy the present market, offering supply chain transparency and providing more information to retailers and manufacturers. However, the cost of RFID is indeed a big obstacle for further application. Developing lower-cost RFID tags with long service lives compatible with the shelf lives of products is still a burning issue. The application of RFID tags could be expanded for example through sensor-enabled RFID tags.

One more important application of PE is in flexible batteries and IFP displays. Nowadays, small volume batteries printed on flexible substrates as capacitors for high energy storage density together with flexible displays have been used to diversify packaging functions.

In this review, we have discussed issues from conductive ink synthesis to PE application to IFP, with the help of some reported articles. We trust that this review may help readers to better understand the preparation of conductive inks and the application of PE to IFP.

Conflicts of interest

The authors declare that they have no conflicts of interest.

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