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High-contrast mechanochromic benzothiadiazole derivatives based on a triphenylamine or a carbazole unit†

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Four triphenylamine or carbazole-based benzothiadiazole fluorescent molecules have been successfully synthesized and characterized. Interestingly, the donor–acceptor (D–A) type luminogens **1**, **2**, **3** and **4** showed different solid-state fluorescence. Furthermore, the four compounds exhibited reversible high-contrast mechanochromism characteristics.

Stimuli-responsive materials receive much attention currently due to their academic importance and potential applications in opto-electronic devices and fluorescent sensors,^{1–7} especially organic smart materials whose solid-state luminescence can be tuned by external stimuli.^{8–13} Mechanochromic fluorescence materials, as a class of smart materials, are also receiving increasing attention.^{14–20} To date, a number of mechanofluorochromic organic molecules have been reported.^{21–24} In contrast, examples of high-contrast mechanochromic luminescence materials are still inadequate. Indeed, many traditional organic materials are aggregation caused quenching (ACQ)-active, and these materials are weakly emissive or nonluminescent in the solid state due to the presence of strong intermolecular electronic interactions in their aggregated state, which promotes the formation of exciplexes and excimers.^{25–27} Obviously, the ACQ effect is unbenevolent to gain high-contrast mechanofluorochromic materials.^{28–30} It is no doubt that mechanochromic molecules with a bright solid-state fluorescence emission are easier to achieve high-contrast mechanofluorochromic phenomenon. Therefore, the corresponding highly emissive smart luminophors have attracted considerable attention.^{31,32}

In general, the emission characteristics of mechanochromic luminescence materials depend strongly on their molecular structures and intermolecular interactions.^{33–35} Therefore, it is an effective method for the realization of mechanofluorochromic materials to change the morphological structures by means of external mechanical stimulus.³⁶

Benzothiadiazole-based derivatives are regarded as attractive candidates for the organic π -conjugated fluorescent dyes owing to their strongly electron-withdrawing feature.^{37–41} Meanwhile,

the benzothiadiazole unit is also advantageous to the construction of donor–acceptor (D–A) type molecules, which have emerged as a significative class of optical materials finding potential value in some areas such as in fluorescent sensors and displays.^{42,43} Motivated by the fact that triphenylamine or carbazole fluorogen has been broadly applied in the field of emissive materials,^{44,45} we attempted to link one triphenylamine or carbazole group to one benzothiadiazole moiety. As a result, we have obtained four D–A type fluorescent molecules on the basis of a combination of the electron-donating triphenylamine or carbazole unit and the electron-accepting benzothiadiazole unit (Fig. 1). Compound **1**, **2**, **3** or **4** contains rotatable aromatic rings, and thus their molecular structures are nonplanar, which is advantageous to the radiative decay in the aggregated state. Indeed, compounds **1**, **2**, **3** and **4** showed bright solid-state fluorescence with different emission colors. In addition, we found that the D–A type luminogens **1**, **2**, **3** and **4** applying the triphenylamine or carbazole moiety as an electron donor and the benzothiadiazole moiety as an electron acceptor exhibited various mechanochromic fluorescence characteristics with good reversibility. Furthermore, luminogen **1** showed mechanofluorochromic behavior involving color change from orange to rare red.

To investigate the solid-state fluorescence behaviors of compounds **1**, **2**, **3** and **4** in detail, the corresponding solid-state

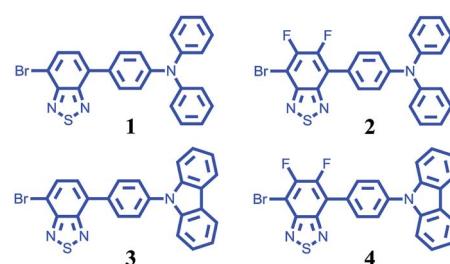


Fig. 1 Molecular structures of the compounds **1**, **2**, **3** and **4**.

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emission spectra were studied initially. As shown in Fig. 2, the fluorescence spectrum of triphenylamine-containing benzothiadiazole derivative **1** exhibited one emission band with the λ_{max} at 575 nm, and the fluorescent molecule exhibited strong orange luminescence with the fluorescence quantum yield (Φ) of 7.13%, and triphenylamine-containing compound **2** exhibited strong yellow luminescence ($\Phi = 7.43\%$) with the λ_{max} at 567 nm. In contrast, the emission spectrum of carbazole-based benzothiadiazole derivative **3** exhibited one emission band with the λ_{max} at 504 nm, and the luminogen exhibited bright green fluorescence with the quantum yield of 16.10%, and carbazole-based compound **4** also exhibited bright green fluorescence ($\Phi = 16.53\%$) with the λ_{max} at 498 nm. Therefore, the photoluminescence (PL) behaviors of compounds **1**, **2**, **3** and **4** could be adjusted *via* introducing various fluorogens containing triphenylamine and carbazole. In addition, the fluorescence lifetimes of **1**, **2**, **3** and **4** were also measured. As shown in Fig. 3, the average lifetime of fluorescent molecule **1** was 0.82 ns, the average lifetime of **2** was 1.56 ns, the average lifetime of **3** was 3.00 ns, and the average lifetime of **4** was 1.41 ns.

Subsequently, the mechanochromic fluorescence characteristics of compounds **1**, **2**, **3** and **4** were investigated. As shown in Fig. 4, the solid sample of luminogen **1** showed a bright orange fluorescence. Interestingly, the orange luminescence was changed to the red luminescence with the λ_{max} at 593 nm upon treating with mechanical force stimulus. Furthermore, the initial orange emission could be restored after treatment of the ground compound **1** with fuming dichloromethane for 1 min. Therefore, **1** showed reversible high-contrast mechanofluorochromic behavior with color change from orange to red, which is a relatively rare color conversion among all mechanochromic fluorescence phenomena.

Similarly, as shown in Fig. 5, compound **2** also showed reversible high-contrast mechanochromic fluorescence behavior. Moreover, the reversible mechanochromic fluorescence of **1** or **2** could be repeated four times between the orange or yellow and red or orange emissions without obvious changes by alternating grinding and dichloromethane treatments. To date, this mechanochromic luminescence conversion of some reported mechanochromism compounds with superior performance is also repeated three or four times,^{46–48} and thus the reversibility of the mechanochromic fluorescence effect of **1** or **2** is good (Fig. 6). On the other hand, as shown in Fig. 7, when

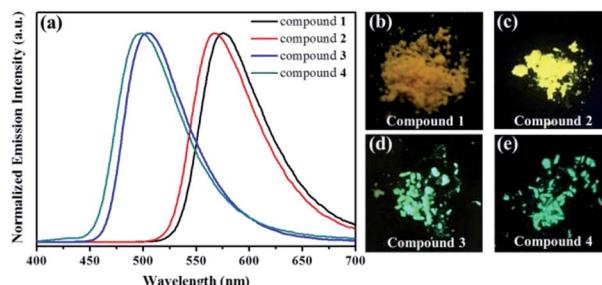


Fig. 2 Solid-state emissive spectra of the compounds **1**, **2**, **3** and **4**, and the related fluorescence images under 365 nm UV light.

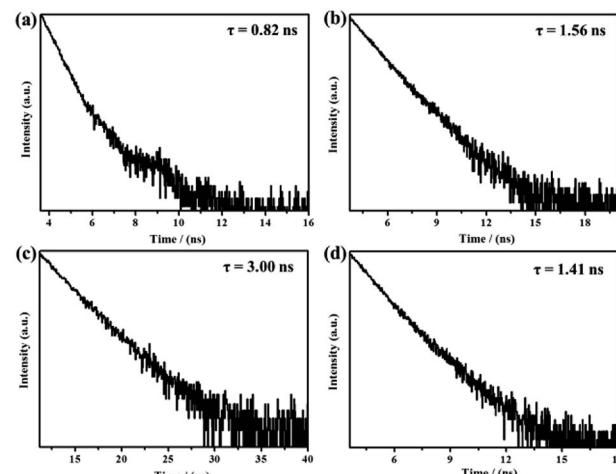


Fig. 3 (a) Time-resolved luminescence (575 nm) of solid sample **1**. Excitation wavelength: 365 nm. (b) Time-resolved luminescence (567 nm) of solid sample **2**. Excitation wavelength: 365 nm. (c) Time-resolved luminescence (504 nm) of solid sample **3**. Excitation wavelength: 365 nm. (d) Time-resolved luminescence (498 nm) of solid sample **4**. Excitation wavelength: 365 nm.

sample **3** were ground in an agate mortar with a pestle, the green emission was changed to the yellow-green fluorescence with the λ_{max} at 533 nm. Moreover, the yellow-green emission could also revert to the original green emission after a 1 min treatment of the ground powder with fuming dichloromethane vapor. Furthermore, as shown in Fig. 8, compound **4** also showed similar mechanochromic fluorescence behavior.

As can be seen in Fig. 9, the reversibility of the mechanofluorochromic behavior of compound **3** or **4** is also excellent. Next, the powder X-ray diffraction (XRD) patterns were studied in order to ensure the morphological characteristics. As can be seen in Fig. 10, the XRD patterns of compound **1** or **2** exhibited a number of sharp reflection peaks, suggesting that the unground compound **1** or **2** was crystalline in nature. However, the ground powder sample became amorphous, with a lack of sharp diffraction peaks. Therefore, the change in fluorescence of compound **1** or **2** could be attributed to the conversion from a crystalline state to an amorphous state. On the other hand,

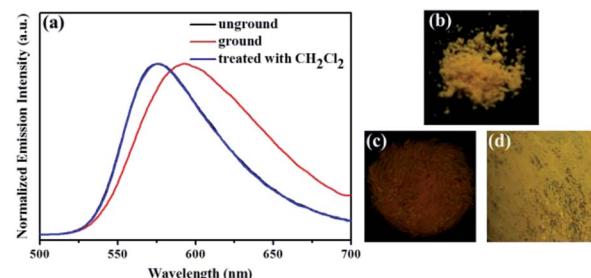


Fig. 4 (a) PL spectra of solid sample **1** at different conditions. Excitation wavelength: 365 nm. (b) Fluorescence image of the unground sample **1** under 365 nm UV light. (c) Fluorescence image of the ground sample **1** under 365 nm UV light. (d) Fluorescence image of the ground sample **1** after treatment with dichloromethane under 365 nm UV light.



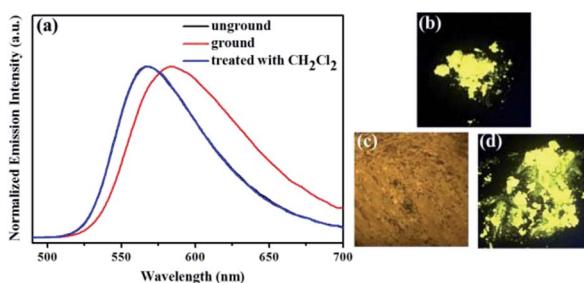


Fig. 5 (a) PL spectra of solid sample 2 at different conditions. Excitation wavelength: 365 nm. (b) Fluorescence image of the unground sample 2 under 365 nm UV light. (c) Fluorescence image of the ground sample 2 under 365 nm UV light. (d) Fluorescence image of the ground sample 2 after treatment with dichloromethane under 365 nm UV light.

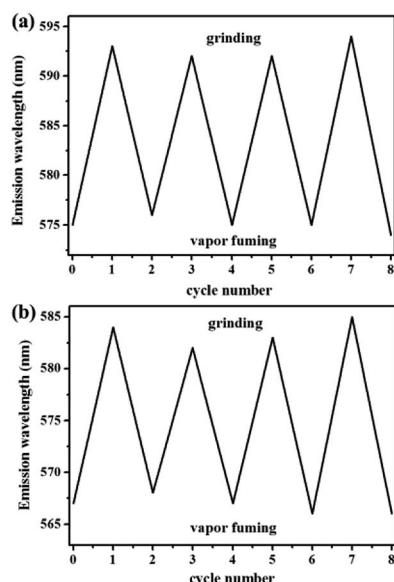


Fig. 6 (a) Repetitive experiment of mechanofluorochromic effect for compound 1. (b) Repetitive experiment of mechanofluorochromic effect for compound 2.

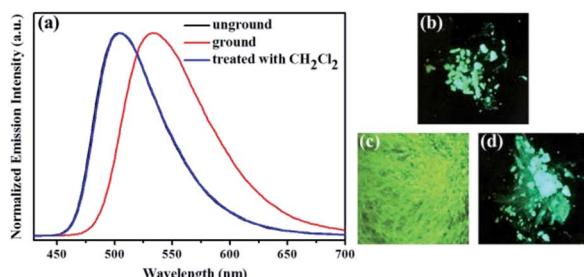


Fig. 7 (a) PL spectra of solid sample 3 at different conditions. Excitation wavelength: 365 nm. (b) Fluorescence image of the unground sample 3 under 365 nm UV light. (c) Fluorescence image of the ground sample 3 under 365 nm UV light. (d) Fluorescence image of the ground sample 3 after treatment with dichloromethane under 365 nm UV light.

when the ground sample was exposed to dichloromethane vapor for 1 min, the sharp and intense peaks reappeared, indicative of the recovery of the crystalline nature. As presented

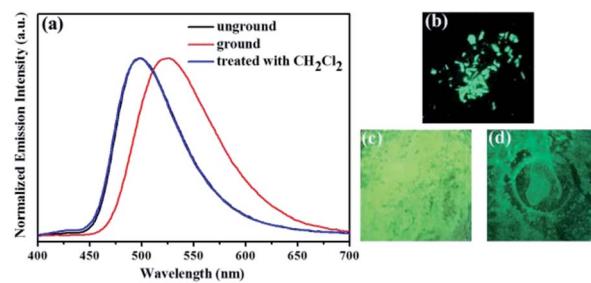


Fig. 8 (a) PL spectra of solid sample 4 at different conditions. Excitation wavelength: 365 nm. (b) Fluorescence image of the unground sample 4 under 365 nm UV light. (c) Fluorescence image of the ground sample 4 under 365 nm UV light. (d) Fluorescence image of the ground sample 4 after treatment with dichloromethane under 365 nm UV light.

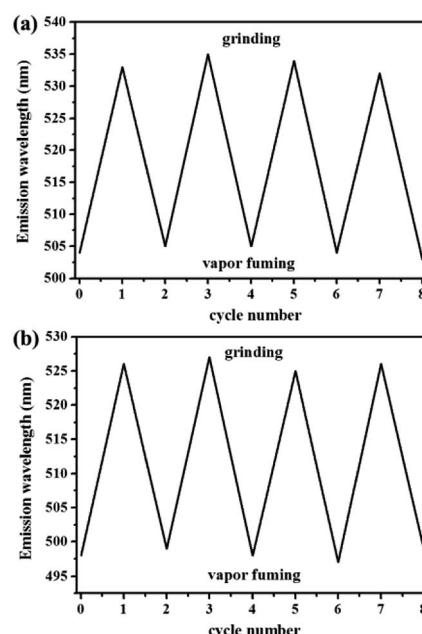


Fig. 9 (a) Repetitive experiment of mechanofluorochromic effect for compound 3. (b) Repetitive experiment of mechanofluorochromic effect for compound 4.

in Fig. 11, the structural transition of the powder sample of compound 3 or 4 was similar to that of 1 or 2. Based on the above mentioned analysis, the powder XRD results demonstrated that the interesting mechanochromic fluorescence characteristics of compounds 1, 2, 3 and 4 were ascribed to the switchable morphology transition between the crystalline state and the amorphous state.

In conclusion, in this work, four triphenylamine or carbazole-based benzothiadiazole fluorescent molecules were successfully synthesized. The compounds 1, 2, 3 and 4 belonged to the highly solid-state emissive donor-acceptor (D-A) type luminescent molecules. It is noteworthy that the four D-A type luminogens exhibited high-contrast mechanofluorochromic characteristics. Furthermore, the reversibility of their mechanochromic phenomena is good. The results of powder XRD

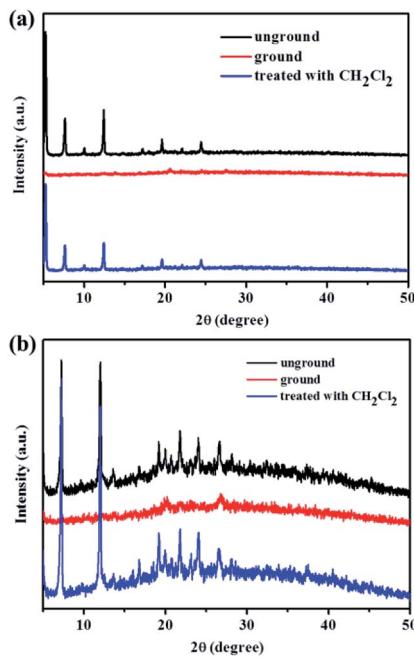


Fig. 10 (a) Powder XRD patterns of compound 1 in different solid states. (b) Powder XRD patterns of compound 2 in different solid states.

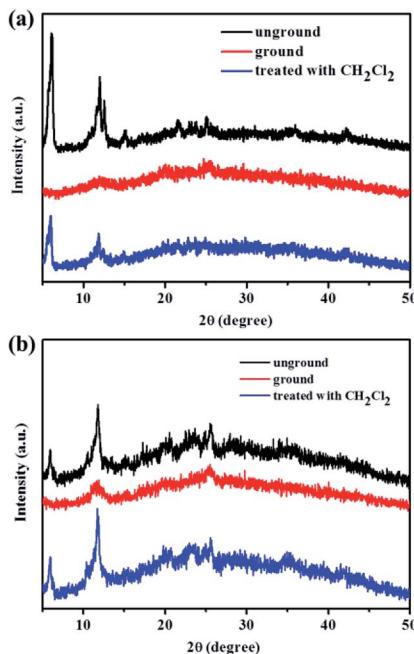


Fig. 11 (a) Powder XRD patterns of compound 3 in different solid states. (b) Powder XRD patterns of compound 4 in different solid states.

experiments confirmed that this switchable morphology transformation is responsible for the reversible mechanochromic fluorescence characteristics of **1**, **2**, **3** and **4**. This work is valuable for designing high-contrast mechanochromic materials involving red light-emitting feature.

Conflicts of interest

There are no conflicts to declare.

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