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Enhanced thermoelectric properties in N-type $\text{Mg}_2\text{Si}_{0.4-x}\text{Sn}_{0.6}\text{Sb}_x$ synthesized by alkaline earth metal reduction

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$\text{Mg}_2\text{Si}_{1-x}\text{Sn}_x$ -based compounds have been recognized as promising thermoelectric materials owing to their high figure-of-merit ZT s, abundance of raw constituent elements and nontoxicity. However, further improvement in the thermoelectric performance in this type of material is still constrained by the high thermal conductivity. In this work, we prepared a series of representative $\text{Mg}_2\text{Si}_{0.4-x}\text{Sn}_{0.6}\text{Sb}_x$ ($x = 0, 0.0075, 0.008, 0.009, 0.01, 0.011$) samples *via* the alkaline earth metal reduction method through a combination of ball milling and spark plasma sintering (SPS) processes. The samples featured many dislocations at the grain boundaries and plenty of nanoscale-coherent Mg_2Si – Mg_2Sn spinodal phases; both of which can effectively scatter heat-carrying phonons and have nearly no impact on the carrier transport. Meanwhile, Sb-doping can efficiently optimize the carrier concentration and significantly suppress the bipolar effects. As a result, a maximal ZT of 1.42 at 723 K and engineering (ZT)eng of 0.7 are achieved at the optimal Sb-doping level of $x = 0.01$. This result indicates that the alkaline earth metal reduction method could be an effective route to engineer phonon transport and improve the thermoelectric performance in $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_x$ -based materials.

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Introduction

Thermoelectric materials can directly convert waste heat into electricity, which will improve the ultimate efficiency of thermal energy utilization. Taking the reality of severely increased waste heat energy dissipation into account, thermoelectric materials contribute to the development of green and sustainable energy for society.^{1,2} The performance of thermoelectric materials is usually evaluated by a dimensionless figure-of-merit, $ZT = S^2T/\rho K$, where S , ρ , K and T are the Seebeck coefficient, electrical resistivity, thermal conductivity and absolute temperature, respectively. On the whole, thermoelectric transport consists of two correlative processes: the electrical and thermal transport processes, where electrical transport is generally quantified by the thermoelectric power factor ($\text{PF} = S^2/\rho$), in which S and ρ ($\rho = 1/n\mu$) are counter-related through the carrier concentration n under a single parabolic band (SPB) model for acoustic phonon dominated systems:³

$$S = \frac{8\pi^2 k_B^2}{2eh^2} m^* T \left(\frac{\pi}{3n} \right)^{2/3} \quad (1)$$

Furthermore, the thermal transport process is mostly carried out by acoustic phonons and charge carriers, which are referred to as the phonon part (K_l) and the charge carrier part (K_e). Because K_e and ρ are further coupled through the Wiedemann–Franz law, we can say that $K_e = LT/\rho$, where L is known as the Lorenz number. Thus, the enhancement of ZT values through the independent tunings of thermoelectric parameters S , ρ and K is challenging. However, ideal thermoelectric materials require a synergistically high power factor and low thermal conductivity, and measures should be taken to achieve this goal. Instinctively the fundamental approach to increase electrical transport is to modulate carrier concentrations;^{4–6} other approaches include delicate tuning of the band effective mass and mobility through band engineering.^{7–12} Another strategy is to decrease the relatively independent parameter, lattice thermal conductivity K_l , through increased phonon scatterings by multiple crystal defects, such as point defects, grain boundaries, nanoscale precipitations, dislocations, or by taking advantage of the intrinsically anharmonic phonons for low lattice thermal conductivity, *etc.*^{13–18}

Mg_2Si -based compounds are treated as promising thermoelectric materials because of their nontoxicity and low cost and because of the abundance of the constituent elements. However, the low power factor and high lattice thermal conductivity in pristine Mg_2Si materials inhibit the improvement of thermoelectric performance. Liu *et al.* found that the Mg_2Si – Mg_2Sn solid solutions favor convergence of the

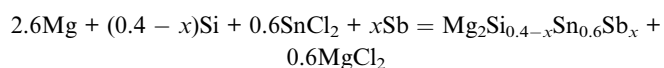
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conduction band, leading to an increase of the density-of-states effective mass, and therefore Seebeck coefficient and ZT .¹⁹ Zhang *et al.* also found that Sb-doping on the Si site can optimize the carrier concentration and effectively tune the band structures, leading to an increased power factor.²⁰ Furthermore, Tang and Jang *et al.* reported that the $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ solid solution doped with Sb could form a coherent interface in nanostructures, thus effectively scattering mid-frequency phonons while having little influence on carrier transport.^{21,22}

The alkaline earth metal reduction method, which was first described in our previous work in half-Heusler materials including both n-type $\text{Hf}_{0.025}\text{Zr}_{0.075}\text{NiSn}_{0.97}\text{Sb}_{0.03}$ and p-type $\text{Nb}_{0.8}\text{Ti}_{0.2}\text{FeSb}$, proves to be effective in inhibiting the thermal transport process and has little influence on electrical transport.²³ Since the high thermal conductivity also constrains the thermoelectric performance of $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_x$ -based materials, we introduce this novel method to synthesize high purity $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ bulk samples, as shown in the following chemical reaction equation:



Notably, chlorides and metal substances were used as raw materials to perform ball milling and then the spark plasma sintering (SPS) process. Primarily, the Sb-dopant optimized the carrier concentration and effectively suppressed the bipolar effects in thermal conductivity, raising the PF to a high level. Meanwhile, the dislocations at the grain boundaries and the nanoscale-coherent Mg_2Si – Mg_2Sn spinodal phases in the matrix strongly scattered the transport phonons, leading to reduced lattice thermal conductivity, but only with a trivial impact on electrical transport. As a result, the thermoelectric performance of $\text{Mg}_2\text{Si}_{0.4-x}\text{Sn}_{0.6}\text{Sb}_x$ in the whole temperature range was enhanced, with a maximum ZT of 1.42 at 723 K at the doping level of $x = 0.01$.

It is noted that the thermoelectric performance here is obviously higher than that of the samples synthesized through traditional methods,^{20,22} even though the thermoelectric performance is a little lower than the optimal performance in the report with a maximum ZT of 1.50 at 716 K.²⁴ However, the alkaline earth metal reduction method is less time-consuming and simpler than the synthetic process that combines the molten-salt sealing method with pulsed electric sintering densification.²⁴ This result proved that the alkaline earth metal reduction method with a combination of ball milling and the SPS process is a dominant and simple method to enhance the thermoelectric performance of $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_x$ -based materials.

Experiments

Synthetic procedures

The $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ and five Sb-doped $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ bulk samples were synthesized through a mechanochemical reaction and SPS process. The raw materials including tin(II) chloride anhydrous (SnCl_2 , 98%, powders), magnesium (Mg, 99.8%, turnings),

silicon (Si, 99%, powders) and antimony (Sb, 99.999%, pieces) were weighed according to the stoichiometry of $\text{Mg}_{2.7}\text{Si}_{0.4-x}\text{Sn}_{0.6}\text{Sb}_x$ ($x = 0, 0.0075, 0.008, 0.009, 0.01, 0.011$) in an Ar-filled glove box. The excess Mg was used not only to reduce Sn from SnCl_2 , but also to compensate for evaporation of Mg during the reaction process. All 5 g of materials were then transferred into the stainless ball milling jar to undergo a 6 h high-energy ball milling process (SPEX Sample Prep 800 Mixer/Mill), at between 1425 and 1725 revolutions per minute (rpm). Finally, the obtained precursor powder was loaded into a graphite die (diameter, 12.7 mm) and sintered by SPS at 1023 K for 15 min under a pressure of 50 MPa.

Phase and morphologies analysis

The phase analysis was performed by X-ray diffraction (XRD; Xpert Pro PANalytical XRD) using Cu-K_α radiation ($\lambda = 1.5418 \text{ \AA}$) and operating under 40 kV and 40 mA in a 2θ range between 20° and 80° in the air. The microstructures of the samples were characterized by scanning electron microscopy (SEM; XL30S-FEG) and transmission electron microscopy (TEM; JEM-2100 Plus). The accelerating voltages for the SEM and TEM were 15 kV and 200 kV, respectively.

Thermoelectric properties analysis

The temperature-dependent electrical resistivity and Seebeck coefficient were measured using commercial equipment (LSR-3, Linseis) protected by high purity helium. The temperature-dependent thermal diffusivity (D) was measured by a laser flash system (LFA-1000, Linseis) at the atmosphere of high purity helium, and the total conductivity was calculated by $K_{\text{tot}} = DC_p d$, where C_p is the specific heat capacity estimated by Dulong–Petit law $C_p = 3NR$, N and R are the gas constant and the ratio of atomic number to molar mass, respectively, and d is the density measured by Archimedes drainage method. The room-temperature Hall analysis, including carrier concentration (n) and carrier mobility (μ), was carried out using the van der Pauw method under a magnetic field of 0.5 T.

Results and discussion

Structures and morphologies analysis

Six typical samples were synthesized through this new method, and were characterized by XRD as shown in Fig. 1(a). It can be seen that a single-phase pattern is obtained. The main diffraction peaks of the patterns can be indexed to an anti-fluorite-type structure ($Fm3m$), and the strongest two peak positions of the samples are located between the peaks of pure Mg_2Si and Mg_2Sn , indicating the formation of Mg_2Si – Mg_2Sn solid solution and the successful doping of Sb, which is consistent with previous results.^{20,21,25} As the inset of Fig. 1(b) shows, the crystal structure of $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ includes two types of lattice site, in which 16c is occupied by Mg while Si and Sn co-occupied 8b.²⁶ When Sb partly replaces Si in the $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ matrix, the crystal lattice expands because the atomic radius of Sb is larger than that of Si. The lattice parameters of the six samples are



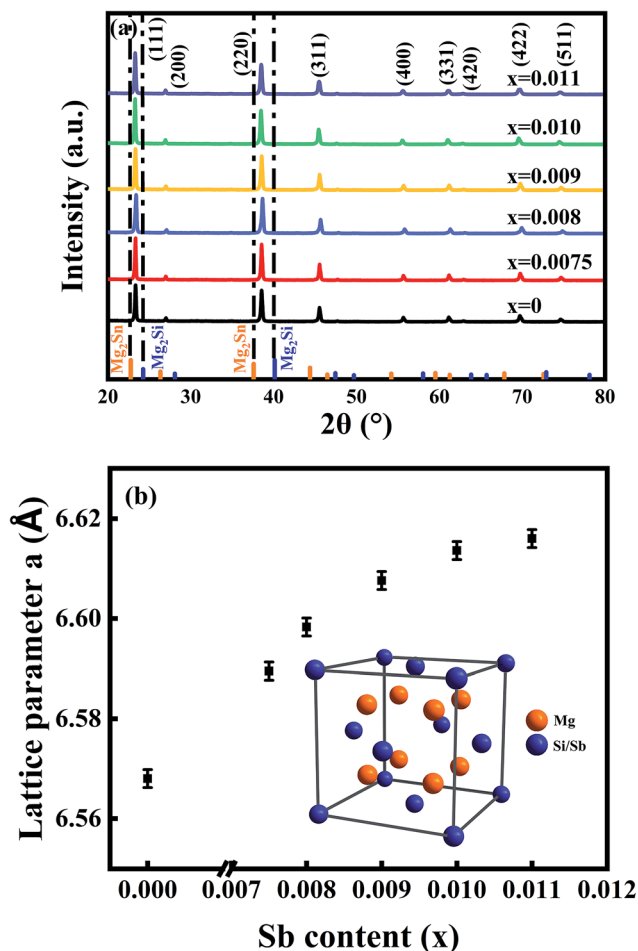


Fig. 1 (a) XRD patterns for all samples and (b) antimony doping dependence of lattice parameters for a series of Sb-doped $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_{0.6}\text{Sb}_x$ ($x = 0, 0.0075, 0.008, 0.009, 0.01, 0.011$) compounds. The inset image shows the crystal structure of $\text{Mg}_2\text{Si/Sn}$.

calculated and displayed as a function of the antimony doping level in Fig. 1(b), which is consistent with previous results.²⁰

Fig. 2(a) presents the morphology and structure of the fractured surface of the $\text{Mg}_2\text{Si}_{0.39}\text{Sn}_{0.6}\text{Sb}_{0.01}$ sample. There are clearly pores along the surface, which can be ascribed to the extrusion of magnesium chloride. However, the density of these samples is still 93–95% of the theoretical density. The grain sizes are around 5–10 μm , and clear edges can be observed at the boundaries, exhibiting good crystallization by using this

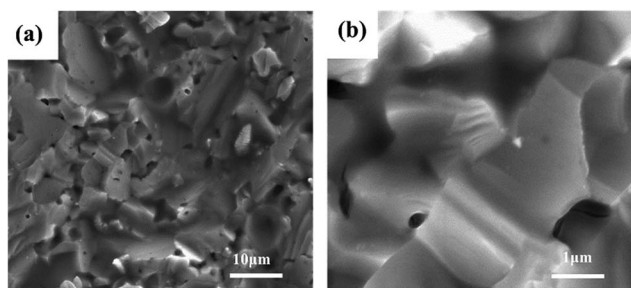


Fig. 2 SEM images of (a) low-magnification and (b) high-magnification of the optimal composite $\text{Mg}_2\text{Si}_{0.39}\text{Sn}_{0.6}\text{Sb}_{0.01}$.

new synthetic method. In Fig. 2(b), the high-magnification SEM image shows that there are no clear inclusions or precipitations between the crystalline boundaries and in the inner grains, indicating the formation of a continuous Mg_2Si – Mg_2Sn solid solution.²²

Fig. 3 shows the TEM images of $\text{Mg}_2\text{Si}_{0.39}\text{Sn}_{0.6}\text{Sb}_{0.01}$. Fig. 3(a) represents the TEM image of the grain boundaries. To identify structural information among the grain boundaries with greater clarity, a magnified TEM image was obtained, as shown in Fig. 3(b). The dislocations at the grain boundaries can clearly be seen, which would scatter the phonons and decrease the thermal conductivity. Furthermore, the dislocations are mainly located at the boundaries, which would be beneficial for the decoupling of electrical transport and phonon transport inside the grains. A similar result can be seen in half-Heusler systems synthesized by the same route.²³

Furthermore, Fig. 3(c) and (d) show the TEM image and the corresponding selected area electron diffraction (SAED) image of the matrix, respectively. According to the SAED pattern, the matrix conforms with the zone axis $[0\ 1\ 1]$ of the $Fm\bar{3}m$

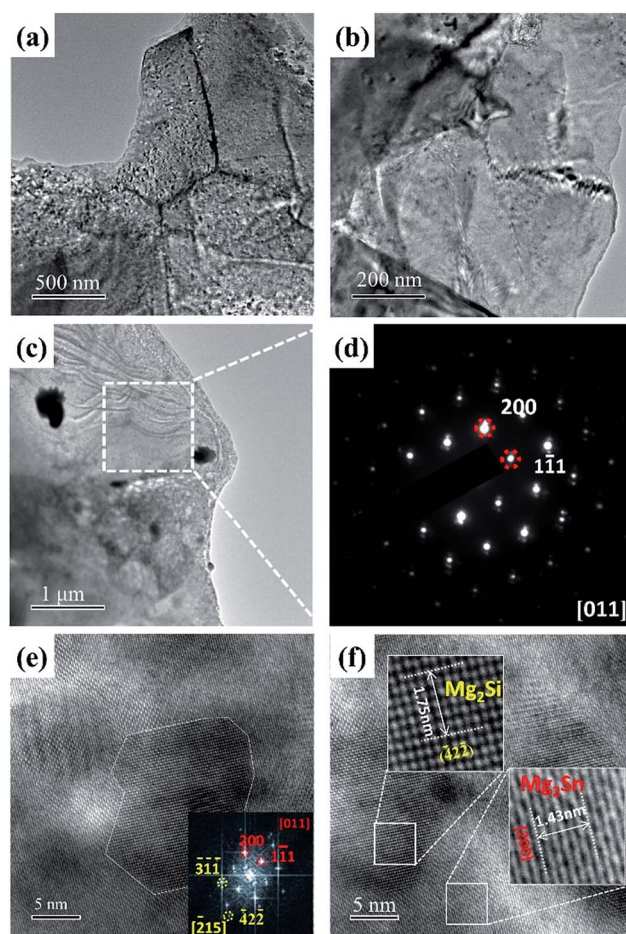


Fig. 3 (a) TEM image of the grain boundary. (b) TEM image of the dislocations in the matrix. (c) TEM image of the matrix. (d) Selected area electron diffraction image of the matrix. (e) High resolution TEM image of the selected (inset shows the fast Fourier transform pattern of the matrix). (f) High resolution TEM image of coherent matrix with the optimal doping level $x = 0.01$.



structure. The Fast Fourier Transform pattern of the matrix in Fig. 3(e) shows that there are two sets of diffraction pattern, which indicates that the matrix is composed of different structures. The red points are assigned to the zone axis $[0\ 1\ 1]$ of Mg_2Sn , and the yellow points are assigned to the zone axis $[2\ 1\ 5]$ of Mg_2Si . In order to obtain more information about the matrix, a high resolution TEM image of the matrix is shown in Fig. 3(f). From the high resolution TEM image, it is clear that the matrix is composed of many spinodal phases, consistent with the result of fast Fourier transform patterns. Furthermore, the inter-planar distances of the planes in the selected areas correspond to $(2\ 0\ 0)$ and $(4\ 2\ 2)$ of Mg_2Sn and Mg_2Si , respectively. It is generally believed that the coherent spinodal structure in the matrix would not have a significant impact on the electrical transports. However, the spinodal structures might result in strong phonon scattering owing to mass fluctuation and lattice dislocations. Thus, the coherent spinodal structures would facilitate the thermoelectric transport in these materials by decoupling the thermal and charge carrier transports, leading to enhanced ZT .

Thermal conductivity analysis

Fig. 4(a) shows the temperature dependence of total thermal conductivity (K_{tot}) for all samples. As the doping level increases, the total thermal conductivity increases; while the total thermal conductivity decreases as the temperature increases. As shown in Fig. 4(a), the thermal conductivity of the undoped sample reaches a minimum and starts to increase at 423 K, indicating the occurrence of a bipolar effect. The bipolar thermal conductivity has a contribution from the minority of carriers in the materials, which cannot be ignored when the temperature is sufficiently high.²⁷ Thus, there would be a significant increase of thermal conductivity in the undoped sample. However, as for the Sb-doped samples, the thermal conductivity was continuously decreasing, and the bipolar effect was substantially suppressed.

The total thermal conductivity can be divided into three parts: lattice thermal conductivity (K_l), carrier thermal

conductivity (K_e) and bipolar thermal conductivity (K_b). The carrier thermal conductivity is calculated through the Wiedemann–Franz law as mentioned above, in which the Lorenz number can be estimated according to SPB model as follows:²⁸

$$L = \frac{k_B^2}{e^2} \frac{(\lambda + 1)(\lambda + 3)F_\lambda(\eta)F_{\lambda+2}(\eta) - (\lambda + 2)^2 F_{\lambda+1}(\eta)^2}{(\lambda + 1)^2 F_\lambda(\eta)^2} \quad (2)$$

Here, λ is the scattering factor, which is equal to 0 for the acoustic phonon scattering dominated system.²⁹ However, it is complicated to calculate the Lorenz number through calculation of the Fermi integral. By using the simplified equation proposed by Hyun-Sik Kim *et al.*:³⁰

$$L = 1.5 + \exp\left(-\frac{|S|}{116}\right) \quad (3)$$

where L and S are on the basis of $10^{-8} \text{ W } \Omega^{-1} \text{ K}^{-2}$ and $\mu \Omega \text{ K}^{-1}$, respectively. The accuracy of Lorenz number is nearly 95% when the system follows the SPB model under the assumption of acoustic phonon scattering dominating.³⁰

The Lorenz number L and electrical thermal conductivity K_e are calculated and shown in Fig. 4(b) and (c), respectively. Fig. 4(d) shows the sum of the lattice thermal conductivities and bipolar thermal conductivities of all samples. As for the Sb-doped samples, the lattice thermal conductivity decreases as the Sb-doping level increases, which can be explained by the enhancement of scattering effects due to dislocations and the nanoscale spinodal structure, as shown in Fig. 3, and it approximately follows a T^{-1} law that indicates the Umklapp process dominated phonon transport.³¹ However, as for the undoped sample, the sudden increase of $K_l + K_b$ from 423 K indicates the occurrence of a bipolar effect and the minority carrier plays a significant role in the thermal transport.⁴

Electrical transport analysis

Room-temperature Hall measurements were performed, as shown in Fig. 5(a). The carrier concentration increases with the Sb-doping level, which is ascribed to the donor effect by Sb^{5+} atom substitution on Si^{4+} sites in the host matrix. At the same time, the carrier mobility decreases slightly, which is due to the scattering of the carrier by point defects and other nanoscale structures.

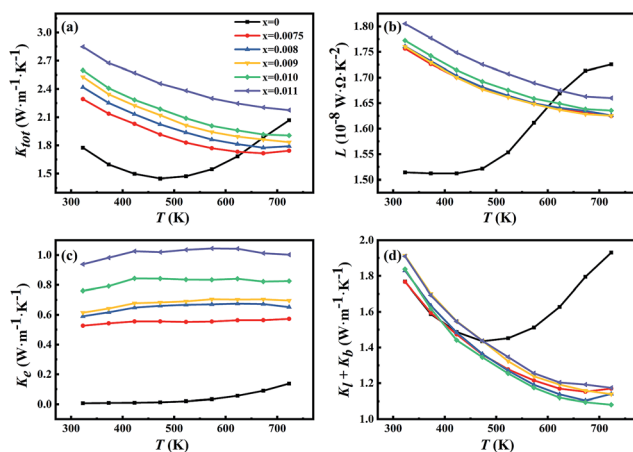


Fig. 4 Temperature dependence of (a) total thermal conductivity, (b) Lorenz number, (c) carrier thermal conductivity and (d) lattice and bipolar conductivity for all samples.

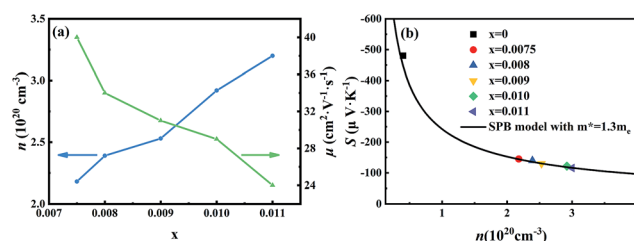


Fig. 5 (a) Room-temperature Hall carrier concentration and carrier mobility of a series of Sb-doped $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_{0.6}\text{Sb}_x$ ($x = 0.0075, 0.008, 0.009, 0.01, 0.011$) compounds. (b) Pisarenko relation plot (black solid line) based on the assumption of a SPB mode with $m^* = 1.3m_e$ at room temperature compared with the scatter plot of the experimental data of all samples.



The plot of the room-temperature Seebeck coefficient as a function of carrier concentration is shown in Fig. 5(b), compared with the Pisarenko line under the assumption of a simple parabolic band model:²⁸

$$S = \frac{k_B}{e} \left\{ \eta - \frac{(\lambda + 2)F_{\lambda+1}(\eta)}{(\lambda + 1)F_{\lambda}(\eta)} \right\} \quad (4)$$

$$n = \frac{4\pi(2m_b^*k_B T)^{3/2}}{h^3} F_{1/2}(\eta) \quad (5)$$

Here, k_B , e and h are the Boltzmann constant, electron charge and Planck constant, respectively. m^* is the effective mass which approaches $1.3m_e$ according to the previous results.²⁰ η is the reduced Fermi level defined as $\eta = (E_F - E_C)/(k_B T)$ where E_F is the chemical potential and E_C is the energy of the conduct band top. Finally, $F_{\lambda}(\eta)$ is the Fermi integral described as follows:

$$F_{\lambda}(\eta) = \int_0^{\infty} \frac{\xi^{\lambda} d\xi}{\exp(\xi - \eta) + 1} \quad (6)$$

As Fig. 5(b) shows, the experimental data match well with the Pisarenko curve at room temperature, revealing that the SPB model still appropriately describes the electronic transport of those $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ -based samples at room temperature.

Fig. 6 shows the temperature-dependent electrical resistivity, Seebeck coefficient, power factor and ZT for all samples and related ref. 20 and 22. Among them, Fig. 6(a) indicates that the electrical resistivity decreases sharply with the Sb-doping in $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$. Fig. 6(b) represents the magnified electrical resistivity of five Sb-doped samples, which clearly shows that the resistivity decreases with increases in Sb-doping level and increase of carrier concentration, as shown in the room-temperature Hall measurements. Furthermore, the electrical resistivity of five Sb-doped samples increases with temperature, indicating the characteristics of heavily doped semiconductor behavior. Clearly, it is proven that by using the new simple synthetic route, Sb can be effectively doped into $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ and so increase the carrier concentration, leading to transformation from non-generated semiconductor behaviors to generated semiconductor behaviors in $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$.⁴

In addition, the resistivity of the undoped sample decreases from 423 K, ascribed to the intrinsic excitation and the contribution of minority carrier to the carrier transport.³² However, no obvious decrease in the electrical resistivity of Sb-doped samples in comparison with the undoped sample was observed, indicating that adding Sb could significantly increase the density of the major carriers. Fig. 6(c) and (d) show that the Seebeck coefficient decreases as the Sb-doping level increases, which originates from the increase of carrier concentration according to eqn (1). As for the undoped sample, there is a clear decrease from 423 K, which is due to the occurrence of bipolar effects. Because of the opposite electric charge between minority and majority carriers, the Seebeck coefficient decreases clearly when the bipolar effect occurs. Further, the sudden decrease of Seebeck coefficient is consistent with that in electrical resistivity and the increase of thermal conductivity.²⁷ Fig. 6(e) presents the temperature dependence of the power

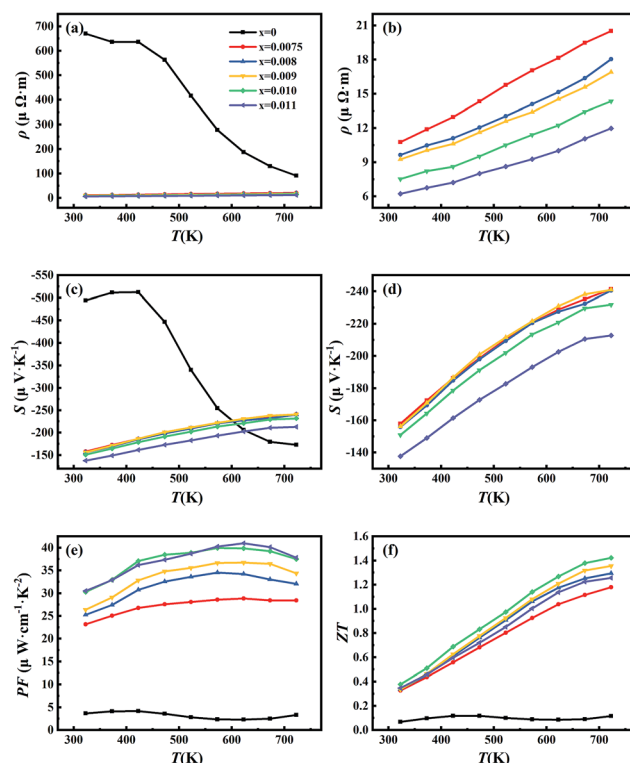


Fig. 6 Temperature dependence of (a) electrical resistivity, (c) Seebeck coefficient, (e) power factor and (f) ZT for all samples. (b) and (d) are the magnified electrical resistivity and Seebeck coefficient for all Sb-doped samples, respectively.

factor for all samples. It indicates that PF increases with doping level. When the doping level reaches above 0.01, the PF hits the maximum value from the whole temperature range. The increase of PF with Sb-doping level could be explained by two reasons. On one hand, the optimization of carrier concentration through Sb-doping approaches the maximal PF . Furthermore, the optimized carrier concentration by adding Sb effectively suppress the bipolar effects. Overall, the dimensionless figure-of-merit for all samples is calculated and shown in Fig. 6(f). It is shown that with increasing Sb-doping level, the ZT increases and reaches a maximum of 1.42 at 723 K at a doping level $x = 0.01$. The high ZT in the work originates for two reasons: on one hand, Sb-doping optimizes the carrier concentration and suppresses the bipolar effects, leading to optimized thermoelectric power factor. On the other hand, the new synthetic approach allows the formation of a number of dislocations at the grain boundaries and plenty of coherent spinodal structures in the matrix of the $\text{Mg}_2\text{Si}_{0.4}\text{Sn}_{0.6}$ samples, which would effectively scatter the heat-carrying phonons, while having little impact on the charge carrier transport. Thus, the decoupling of electrical and phonon transports is realized in Sb-doped $\text{Mg}_2\text{Si}_{0.4-x}\text{Sn}_{0.6}\text{Sb}_x$ with enhanced thermoelectric performance.

In addition, Fig. 7(a) shows that the samples synthesized through the alkaline earth metal reduction method are higher than those of Sb-doped $\text{Mg}_2\text{Si}_{1-x}\text{Sn}_x$ based composites synthesized through traditional methods from the literature,^{20,22,33,34} particularly at high temperatures. Furthermore, the synthesis



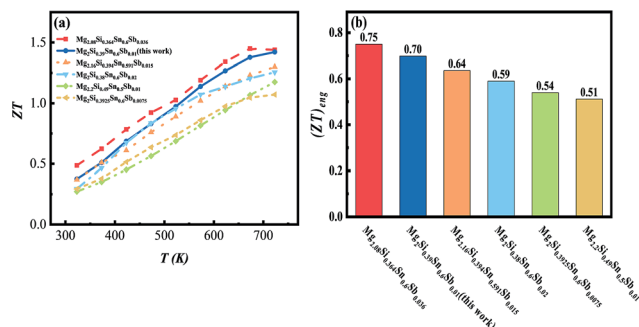


Fig. 7 (a) Temperature-dependent ZTs and (b) engineering ZTs of optimal composite Mg₂Si_{0.39}Sn_{0.6}Sb_{0.01} and the literature.

process is less time-consuming and simpler compared with the process of the optimal report.²⁴

Since the temperature-dependent ZTs of the thermoelectric materials varies in a wide temperature range, the engineering dimensionless figure-of-merit ((ZT)_{eng}) proposed by Kim *et al.* was used to describe the average performance in the whole temperature, which is calculated based on the following equation:³⁵

$$ZT_{\text{eng}} = \frac{\left(\int_{T_c}^{T_h} S(T) dT \right)^2}{\int_{T_c}^{T_h} \rho(T) dT \int_{T_c}^{T_h} \kappa(T) dT} \Delta T \quad (7)$$

where T_h and T_c , and ΔT are the hot-side temperature, cold-side temperature and temperature difference between both sides, respectively. As shown in Fig. 7(b), the (ZT)_{eng} of the optimal composite Mg₂Si_{0.39}Sn_{0.6}Sb_{0.01} and the literature values at a T_c of 323 K and a $\Delta T = 400$ K are displayed. Apparently, the maximal (ZT)_{eng} reaches 0.70 for the composition of Mg₂Si_{0.39}Sn_{0.6}Sb_{0.01} in this work, which is a little lower than the optimal report, while it is obviously higher than those Sb-doped Mg₂Si_{1-x}Sn_x-based composites synthesized through traditional methods. Thus, the alkaline earth metal reduction method is a timesaving and simple route to enhance the thermoelectric performance of Mg₂Si_{1-x}Sn_x-based materials.

Conclusion

In this work, a series of Sb-doped n-type Mg₂Si_{0.4-x}Sn_{0.6}Sb_x ($x = 0, 0.0075, 0.008, 0.009, 0.01, 0.011$) bulk materials have been prepared using an alkaline earth metal reduction method through a combination of ball milling and an SPS process. The samples feature many dislocations at the grain boundaries, and plenty of nanoscale-coherent Mg₂Si-Mg₂Sn spinodal phases, both of which can effectively scatter heat-carrying phonons and do little to impact carrier transport. Meanwhile, the Sb-dopant can efficiently optimize the carrier concentration and significantly suppress the bipolar effects. As a result, a maximal ZT of 1.42 at 723 K and a maximal (ZT)_{eng} of 0.7 with $T_c = 323$ K and $T_h = 723$ K are obtained at the optimal Sb-doping level of $x = 0.01$, which is a competitive synthesis process compared with previous reports.^{20,22,24,33,34} This result proved that the thermoelectric performance of Mg₂Si_{0.4}Sb_{0.6} could be enhanced by

defect engineering through an alkaline earth metal reduction method combined with the SPS process.

Conflicts of interest

The authors have no conflicts to declare.

Acknowledgements

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Notes and references

- 1 J. He and T. M. Tritt, *Science*, 2017, **357**, eaak9997.
- 2 J. Yang, L. Xi, W. Qiu, L. Wu, X. Shi, L. Chen, J. Yang, W. Zhang, C. Uher and D. J. Singh, *npj Comput. Mater.*, 2016, **2**, 15015.
- 3 M. Jonson and G. D. Mahan, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1980, **21**, 4223–4229.
- 4 J. Shuai, J. Mao, S. Song, Q. Zhu, J. Sun, Y. Wang, R. He, J. Zhou, G. Chen, D. J. Singh and Z. Ren, *Energy Environ. Sci.*, 2017, **10**, 799–807.
- 5 J. Mao, J. Shuai, S. Song, Y. Wu, R. Dally, J. Zhou, Z. Liu, J. Sun, Q. Zhang, C. Dela Cruz, S. Wilson, Y. Pei, D. J. Singh, G. Chen, C. W. Chu and Z. Ren, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, 10548–10553.
- 6 D. Li, H. Zhao, S. Li, B. Wei, J. Shuai, C. Shi, X. Xi, P. Sun, S. Meng, L. Gu, Z. Ren and X. Chen, *Adv. Funct. Mater.*, 2015, **25**, 6478–6488.
- 7 Y. Pei, H. Wang and G. J. Snyder, *Adv. Mater.*, 2012, **24**, 6125–6135.
- 8 G. Tan, L. D. Zhao and M. G. Kanatzidis, *Chem. Rev.*, 2016, **116**, 12123–12149.
- 9 C. Fu, T. Zhu, Y. Liu, H. Xie and X. Zhao, *Energy Environ. Sci.*, 2015, **8**, 216–220.
- 10 C. Fu, S. Bai, Y. Liu, Y. Tang, L. Chen, X. Zhao and T. Zhu, *Nat. Commun.*, 2015, **6**, 8144.
- 11 T. Zou, X. Qin, Y. Zhang, X. Li, Z. Zeng, D. Li, J. Zhang, H. Xin, W. Xie and A. Weidenkaff, *Sci. Rep.*, 2015, **5**, 17803.
- 12 S. Li, Y. Liu, F. Liu, D. He, J. He, J. Luo, Y. Xiao and F. Pan, *Nano Energy*, 2018, **49**, 257–266.
- 13 K. Biswas, J. He, I. D. Blum, C. I. Wu, T. P. Hogan, D. N. Seidman, V. P. Dravid and M. G. Kanatzidis, *Nature*, 2012, **489**, 414–418.
- 14 J. He, S. N. Girard, M. G. Kanatzidis and V. P. Dravid, *Adv. Funct. Mater.*, 2010, **20**, 764–772.
- 15 Y. Yu, D.-S. He, S. Zhang, O. Cojocaru-Mirédin, T. Schwarz, A. Stoffers, X.-Y. Wang, S. Zheng, B. Zhu, C. Scheu, D. Wu, J.-Q. He, M. Wuttig, Z.-Y. Huang and F.-Q. Zu, *Nano Energy*, 2017, **37**, 203–213.
- 16 B. Poudel, Q. Hao, Y. Ma, Y. Lan, A. Minnich, B. Yu, X. Yan, D. Wang, A. Muto, D. Vashaee, X. Chen, J. Liu,



- M. S. Dresselhaus, G. Chen and Z. Ren, *Science*, 2008, **320**, 634.
- 17 H.-S. Kim, S. I. Kim, K. H. Lee, S. W. Kim and G. J. Snyder, *Phys. Status Solidi B*, 2017, **254**, 1600103.
- 18 Z. Wang, G. Wang, R. Wang, X. Zhou, Z. Chen, C. Yin, M. Tang, Q. Hu, J. Tang and R. Ang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 22401–22407.
- 19 W. Liu, X. Tan, K. Yin, H. Liu, X. Tang, J. Shi, Q. Zhang and C. Uher, *Phys. Rev. Lett.*, 2012, **108**, 166601.
- 20 Q. Zhang, J. He, T. J. Zhu, S. N. Zhang, X. B. Zhao and T. M. Tritt, *Appl. Phys. Lett.*, 2008, **93**, 102109.
- 21 J. Jang, B. Ryu, S.-J. Joo, B.-S. Kim, B.-K. Min, H.-W. Lee, S.-D. Park, H. S. Lee and J. E. Lee, *J. Alloys Compd.*, 2018, **739**, 129–138.
- 22 W. Liu, X. Tang, H. Li, K. Yin, J. Sharp, X. Zhou and C. Uher, *J. Mater. Chem.*, 2012, **22**, 13653.
- 23 H. Zhao, B. Cao, S. Li, N. Liu, J. Shen, S. Li, J. Jian, L. Gu, Y. Pei, G. J. Snyder, Z. Ren and X. Chen, *Adv. Energy Mater.*, 2017, **7**, 1700446.
- 24 G. Peng, I. Berkun, D. R. Schmidt, F. M. Luzenski, X. U. Lu and P. B. Sarac, *J. Electron. Mater.*, 2014, **43**, 1790–1803.
- 25 W. Liu, X. Tang and J. Sharp, *J. Phys. D: Appl. Phys.*, 2010, **43**, 085406.
- 26 D. Boudemagh, D. Fruchart, R. Haettel, E. K. Hlil, A. Lacoste, L. Ortega, N. E. Skryabina, J. Toboła and P. Wolfers, *Solid State Phenom.*, 2011, **170**, 253–258.
- 27 J. J. Gong, A. J. Hong, J. Shuai, L. Li, Z. B. Yan, Z. F. Ren and J. M. Liu, *Phys. Chem. Chem. Phys.*, 2016, **18**, 16566–16574.
- 28 A. F. May, E. S. Toberer, A. Saramat and G. J. Snyder, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **80**, 125205.
- 29 X. Shi, Y. Pei, G. J. Snyder and L. Chen, *Energy Environ. Sci.*, 2011, **4**, 4086.
- 30 H.-S. Kim, Z. M. Gibbs, Y. Tang, H. Wang and G. J. Snyder, *APL Mater.*, 2015, **3**, 041506.
- 31 W. Hao, C. Jin, L. Tianqi, Z. Kunjie, L. Shan, L. Jun and Z. Huaizhou, *Chin. Phys. B*, 2018, **27**, 047212.
- 32 F. Hao, P. Qiu, Y. Tang, S. Bai, T. Xing, H.-S. Chu, Q. Zhang, P. Lu, T. Zhang, D. Ren, J. Chen, X. Shi and L. Chen, *Energy Environ. Sci.*, 2016, **9**, 3120–3127.
- 33 W. Liu, X. Tang, H. Li, J. Sharp, X. Zhou and C. Uher, *Chem. Mater.*, 2011, **23**, 5256–5263.
- 34 X. Zhang, H. Liu, S. Li, F. Zhang, Q. Lu and J. Zhang, *Mater. Lett.*, 2014, **123**, 31–34.
- 35 H. S. Kim, W. Liu, G. Chen, C. W. Chu and Z. Ren, *Proc. Natl. Acad. Sci. U. S. A.*, 2015, **112**, 8205–8210.

