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Tuning the electronic and magnetic properties of MoS₂ nanotubes with vacancy defects

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By using density functional theory calculations, we evaluated the effects of vacancy defects on the electronic and magnetic properties of MoS₂ nanotubes. While both zigzag and armchair MoS₂ tubes are intrinsically semiconducting, armchair (6, 6) tubes with simple disulfur and mono-molybdenum vacancies, as well as a large vacancy cluster consisting of both Mo and S vacancies (V_{MoS₃}), and zigzag (10, 0) tubes with a mono-molybdenum defect are metallic. In particular, the (6, 6) tube with disulfur and V_{MoS₃} defects is half-metallic, which is promising for applications in spintronic devices. In addition, the (6, 6) tube exhibits an easily tunable magnetic property by introducing vacancies. We found that disulfur, mono-molybdenum, V_{MoS₃} and V_{MoS₆} vacancies are able to cause spin polarization to induce net magnetic moment. This is mainly because the spin states prefer to couple through Mo atoms that are arranged along the zigzag direction (the axial direction of the armchair tube). In contrast, the zigzag (10, 0) tube is relatively hard to tune and is always nonmagnetic, except for the case of V_{MoS₃}. More importantly, atomic- and orbital-projected electron density of states analyses reveal that the net spins are mainly contributed by bare Mo atoms at or near the vacancy edge. For defect-laden MoS₂ tubes that are still semiconducting, the energy gap and effective masses of the charge carriers are highly dependent on tube chirality and defect species. Our present findings highlight the worthwhile semiconducting, metallic, and half-metallic properties of MoS₂ tubes, particularly armchair species, which can be obtained *via* defect engineering; this can find broad applications for the fabrication of nanoelectronic and spintronic devices.

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1. Introduction

Since the first exfoliation by Coleman *et al.* in 2011,¹ molybdenum disulfide (MoS₂) has attracted enormous attention in the recent years. Because of its rich physical and chemical properties, MoS₂ has been increasingly studied for applications in optoelectronics, sensors, field-emitting transistors, and hydrogen evolution catalysis.^{2–8} In addition, MoS₂ has also been utilized in biomedical-related areas, such as biosensors for the detection of protein and DNA,^{9,10} or served as a NIR photo-thermal transducing agent.¹¹ Like graphene, a MoS₂ monolayer can be engineered to be a one-dimensional nanotube (NT),^{12–14} introducing worthwhile electronic and magnetic properties because of its reduced dimensions. However, in contrast to carbon NTs, which can be either metallic or semiconducting depending on their chirality and diameter, MoS₂ NTs have uniform semiconducting features.^{15–17}

It is widely accepted that most nanomaterials, including MoS₂, are rich in defects that can intricately regulate their physical and chemical properties.^{18–21} Numerous experimental and theoretical efforts have demonstrated that point defects on MoS₂ monolayers can induce new physical properties (particularly electron transport)^{19,20,22,23} and chemical activities (acting as the active sites for hydrogen evolution catalysis).^{24–26} Further, it is found that the photoluminescence of MoS₂ can be considerably enhanced by introducing defects, which are fairly essential for bioimaging applications.^{27,28} Despite numerous studies regarding MoS₂ monolayers, little effort has been devoted toward investigating the defects in MoS₂ NTs. This raises an urgent need for a systematic evaluation of the effects of point defects on the electronic and magnetic properties of MoS₂ NTs.

Here, we report a theoretical study of the electronic structures and magnetic properties of MoS₂ NTs with vacancy defects by performing density functional theory (DFT) calculations. While pristine MoS₂ NTs are semiconducting, it is found that (6, 6) MoS₂ NTs with disulfur, mono-molybdenum, or V_{MoS₃} defects, as well as (10, 0) NTs with mono-molybdenum defects, demonstrate metallic conductivity. It is also found that (6, 6) NTs with disulfur and V_{MoS₃} vacancies demonstrate half-metallic conductivity, which can be useful for applications in spintronic devices. Among all the kinds of defects and tubes, net magnetization can be induced for

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armchair MoS₂ NTs with disulfur, mono-molybdenum, V_{MoS₃}, and V_{MoS₆} defects. Zigzag MoS₂ NTs are always nonmagnetic, except for the case of V_{MoS₃}. Generally, the introduction of vacancies tends to increase the effective masses of the charge carriers. The exceptions are holes of zigzag NTs with sulfur defects, where the effective masses are smaller than that of a pristine tube. The worthwhile semiconducting, metallic, and half-metallic properties can find broad applications for the fabrication of nanoelectronic and spintronic devices.

2. Computational methods

All the DFT calculations were performed using the Vienna ab initio simulation package (VASP).^{29,30} Projector-augmented wave (PAW)³¹ potentials were used to describe the electron-ion interactions, while the electron exchange–correlation interactions were analyzed using the generalized gradient approximation (GGA)³² in the form of a Perdew–Burke–Ernzerhof (PBE) scheme. A plane-wave cutoff of 500 eV was used for all the calculations. All the atomic positions and lattice vectors were fully optimized by using a conjugate gradient algorithm to obtain the ground-state configuration. Atomic relaxation was performed until the change in total energy was smaller than 0.01 meV and all the Hellmann–Feynman forces on each atom were less than 0.01 eV Å^{−1}, which could guarantee fully relaxed structures. Two MoS₂ NTs, namely, (6, 6) and (10, 0), have been used as the representative models; their electronic and transport properties were studied in our earlier work.¹⁷ The (6, 6) and (10, 0) models comprise 108 and 120 atoms, respectively, and are periodic along the axial direction (*z* direction). These settings result in corresponding lattice lengths of 9.76 and 10.68 Å, which can be regarded as the separation between adjacent defects. A vacuum space of 15 Å was placed in the *x* and *y* directions to avoid mirror interactions. Three experimentally observed point vacancy defects were studied:^{18–21} monosulfur vacancy (V_S), disulfur vacancy (V_{S₂}), and mono-molybdenum vacancy (V_{Mo}). Further, we considered two large vacancy clusters consisting of both Mo and S vacancies, namely, V_{MoS₃} and V_{MoS₆}, which were identified in the transmission electron microscopy (TEM) studies of monolayer MoS₂.¹⁸ Then, *k*-point sampling of 1 × 1 × 7 was used for structure relaxation, while a denser mesh of 1 × 1 × 30 was used to calculate the band structures and electron states. To quantitatively evaluate the vacancy stability, we calculated the vacancy formation energy in the neutral charge state following the standard approach:

$$E_f = E_{\text{defect}} - E_{\text{ref}} + \sum_i n_i \mu_i$$

where E_{defect} and E_{ref} are the total energies of the tube with specific defect and related reference ideal tube, respectively; n_i is the number of removed atoms and μ_i is their chemical potential calculated from the bulk phase.^{33–35}

3. Results and discussion

The structures of (6, 6) and (10, 0) tubes are shown in Fig. 1a and b, respectively. A pristine (6, 6) tube is a semiconductor with an

indirect bandgap of 0.25 eV, while the (10, 0) tube has a direct bandgap of 0.33 eV. The topologies of the five vacancy defects are shown in Fig. 1c (V_S), Fig. 1d (V_{S₂}), Fig. 1e (V_{Mo}), Fig. 1f (V_{MoS₃}), and Fig. 1g (V_{MoS₆}), where the removed atoms are highlighted as dashed-line balls.

3.1 MoS₂ tube with V_S defect

We begin our analysis with a (6, 6) tube with a V_S defect. The presence of a significant number of S vacancies in MoS₂ was directly observed by high-resolution TEM.^{18,19,22} In our calculations, the V_S vacancy has a formation energy of 2.31 eV for the (6, 6) tube and a comparable value of 2.35 eV for the (10, 0) tube. These values are higher than that for the MoS₂ monolayer (1.56 eV) and than those obtained in an earlier computational study.³⁶ The higher formation energy of the defect can be attributed to the curvature of the tube wall. In Fig. 2a and b, we show the plots of the band structures of defect-free and defect-laden (6, 6) tube for comparison. It is found that the (6, 6) tube with V_S remains semiconducting with an indirect bandgap. In addition, the spin-up and spin-down branches are totally degenerated, indicating that the system is nonmagnetic. The bandgap slightly decreases from 0.25 to 0.24 eV. Another significant change is that the valence band (VB) and conduction band (CB) become flat, as shown in Fig. 2b. The flat band usually represents a spatially localized electron state in real space, corresponding to carriers with large effective masses. To quantitatively assess the regulation of defects to charge carriers, we calculated the effective mass by $m^* = \hbar^2 [\partial^2 E(k) / \partial k^2]^{-1}$ at the valence band maximum (VBM, the hole) and conduction band minimum (CBM, the electron). The m^* value of a hole is $-2.16m_e$, which is larger than that of the defect-free (6, 6) tube (only $-0.83m_e$).¹⁷ The m^* value of an electron is $1.16m_e$, which is also larger than that of the defect-free (6, 6) tube (only $0.53m_e$).¹⁷

The isosurfaces of the electron states of the CB and VB are shown in Fig. 2c and d, respectively. Evidently, the CB and VB represent the unsaturated states at Mo atoms around the S vacancy. In addition, it is also found that the electron clouds between Mo 1–2 overlap along the tube's axial direction. In contrast, the cloud at Mo 3 is isolated. First, this should be mainly attributed to the small distance between Mo 1–2 (3.20 Å) as compared to the distance of 3.59 Å for Mo 1–3 or Mo 2–3. In addition, this phenomenon may refer to the fact that an electron coupling is stronger along the zigzag direction (axial direction of the tube) than that along the armchair direction.

For the case of a pristine (10, 0) tube, it is a direct-bandgap semiconductor: both the VBM and CBM are located at the Γ (0, 0, 0) point, as shown in Fig. 2e. However, when one V_S is introduced, the bandgap becomes indirect: the CBM moves to the χ (0, 0, 0.5) point (Fig. 2f). The V_S defect also suppresses the gap, from 0.33 eV for an ideal tube to 0.17 eV for a defect-laden tube. The spin-up and spin-down branches are totally degenerated, indicating no polarization. In Fig. 2g and h, we show the plots of the isosurfaces of the electron states of the VB and CB. It is clear that the VB and CB mainly spread around the Mo-edge atoms near the missing S. The electron clouds are dispersed within Mo 1–2 and Mo 1–3. In contrast, no electron cloud is



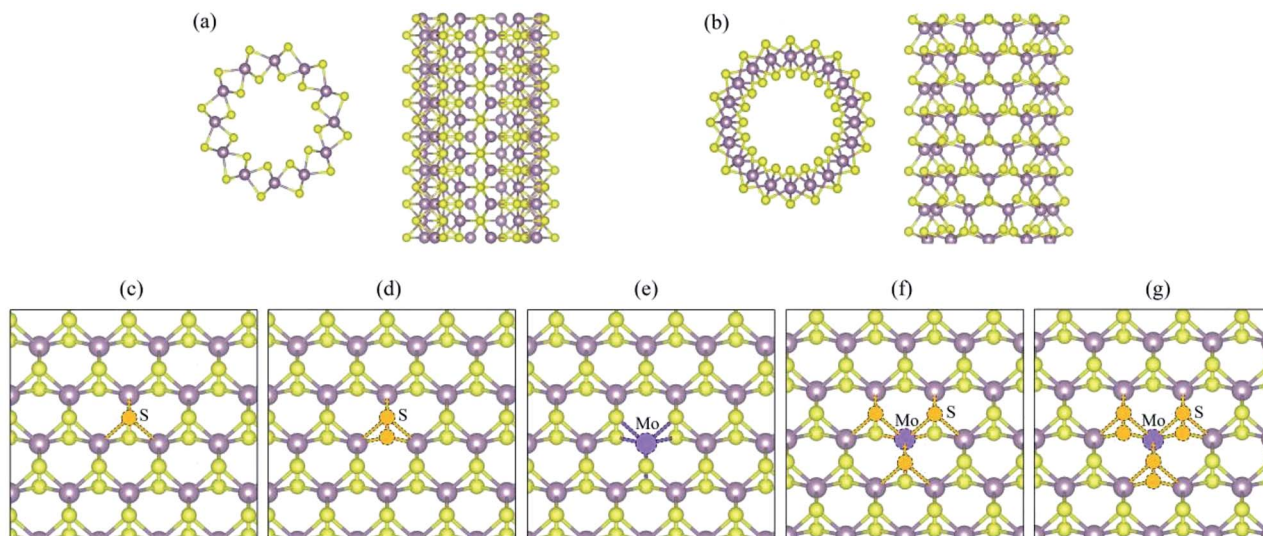


Fig. 1 (a) Top view and side view of a (6, 6) tube; (b) top view and side view of a (10, 0) tube. Topologies of (c) V_S , (d) V_{S_2} , (e) V_{Mo} , (f) V_{MoS_3} , and (g) V_{MoS_6} vacancies. The removed atoms are highlighted as dashed-line balls.

found in Mo 2–3. We mainly attribute this electron cloud pattern to the different distances between the three Mo atoms. In detail, the distance between Mo 1–2 and Mo 1–3 is 3.24 Å;

however, for Mo 2–3, this distance is larger, namely, 3.62 Å. Another feature is that the introduction of the V_S defect changes the m^* values of the hole and electron in an opposite trend. In

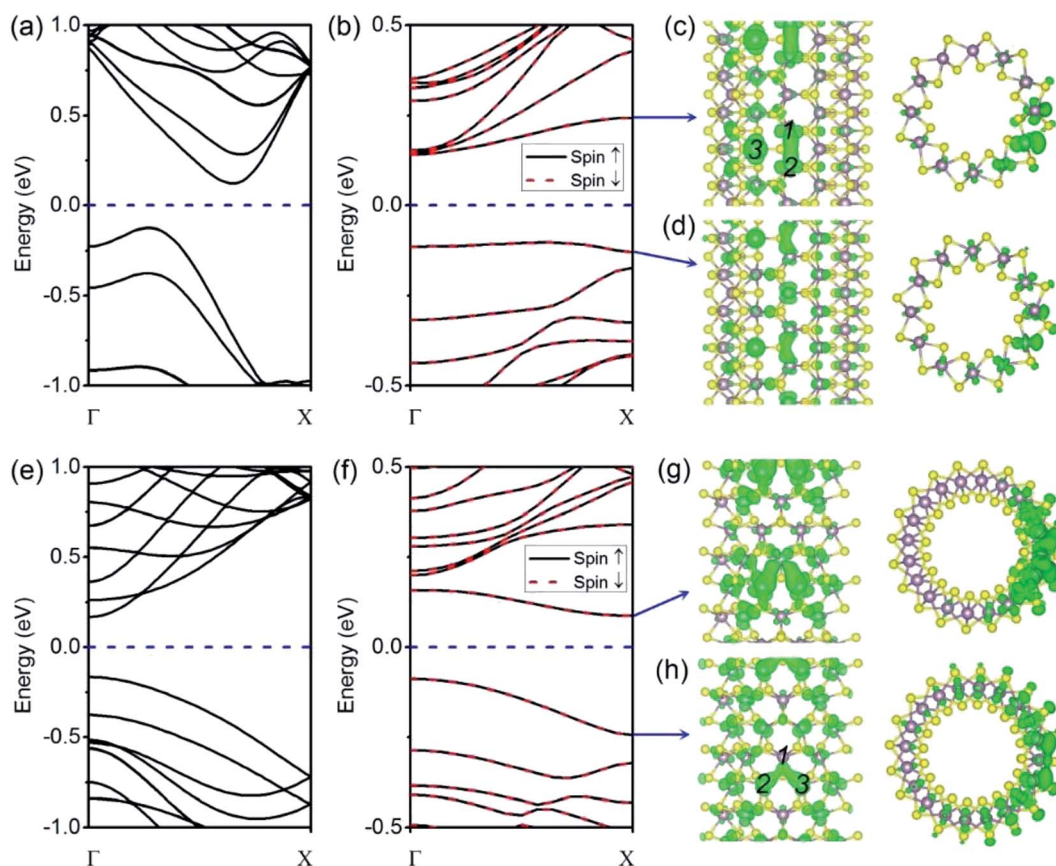


Fig. 2 Band structures of (a) pristine (6, 6) tube and (b) (6, 6) tube with V_S vacancy. Electron isosurfaces of (c) CB and (d) VB. Band structures of (e) pristine (10, 0) tube and (f) (10, 0) tube with V_S vacancy. Electron isosurfaces of (g) CB and (h) VB. The isosurface cutoff is $0.001|e|/\text{\AA}^3$.



detail, the m^* value of the hole is $-1.37m_e$, which is smaller than that of the defect-free (10, 0) tube ($-1.55m_e$). On the contrary, the m^* value of the electron is $1.68m_e$, which is significantly larger than that of the defect-free (10, 0) tube (only $0.51m_e$).¹⁷

3.2 MoS₂ tube with V_{S₂} defect

As compared to the V_S defect, our results reveal that the V_{S₂} defect induces dramatic changes in the (6, 6) tube. Firstly, the formation energy for a V_{S₂} defect is relatively high, 5.07 eV for the (6, 6) tube and 5.03 eV for the (10, 0) tube. The high formation energy indicates that the removal of disulfur atoms, resulting in three Mo-edge atoms, is difficult. This is consistent with earlier studies in which it was shown that Mo-terminated edges in MoS₂ are highly unstable.³³ From the band structures (Fig. 3a), it is evident that the first change involves the (6, 6) tube bearing a V_{S₂} defect becoming metallic. It is noteworthy that only one band of a spin-down channel crosses the Fermi level. However, the spin-up channel has an indirect bandgap of 0.16 eV. Therefore, the tube is half-metallic: it only allows spin-down transport while blocking spin-up transport. This property reveals the prospective applications of (6, 6) tubes in the design of spintronic devices *via* defect engineering. Secondly, the system is spin-polarized with net magnetic moment of $1.75 \mu_B$. To reveal the origin of spin polarization, the atomic- and orbital-projected electron densities of states (PDOS) were analyzed, and these results are shown in Fig. 3b and c. The net spin was found to be mainly contributed by Mo atoms, with only minor contributions from S atoms. In addition, the spin polarization

mainly happened for the d-orbital electrons of Mo. In addition, the isosurface of net spin (magnetic moment) is shown in Fig. 3d. The shape of the net spin is mainly located on Mo 1 and Mo 2 and has a character of $d_{x^2-y^2}/d_{xy}$ hybrid orbitals. This is consistent with the above PDOS analyses. For a MoS₂ monolayer, the $d_{x^2-y^2}$ and d_{xy} orbitals contribute to the Mo-S bonds becoming energy-degenerated.¹⁷ The net spin near the V_{S₂} defect is mainly contributed by the unsaturated electrons because of the missing S atoms. In addition, among the three Mo atoms at the defect edge, the spin mainly spreads around Mo 1 and Mo 2, which aligns along the tube's axial direction (namely, the zigzag direction). Hence, the spin coupling in MoS₂ NTs is similar to the case of graphene ribbons, where a ferromagnetic coupling can only form in the zigzag-edged graphene ribbons. While for armchair ribbons, the coupling between adjacent spins are weak.³⁷

In contrast to the (6, 6) tube (which becomes half-metallic upon the introduction of a V_{S₂} defect), the (10, 0) tube remains semiconducting. However, the gap becomes indirect because the VBM is located at the Γ point, while the CBM moves to the X point (Fig. 3e). In addition, all the branches of the bands are polarized, although the net spin is zero. Such a spin-polarized band structure exhibits two distinct energy gaps for the two spin components: the gap is 0.23 eV for spin-up, while it is only 0.14 eV for spin-down. In addition, the CB is considerably flat, indicating that the electron carriers are highly localized at the defect site (Fig. 3f) with considerably large m^* . Our results indicate that for electron carriers, the values of m^* are $5.99m_e$ (spin-up) and $4.47m_e$ (spin-down).

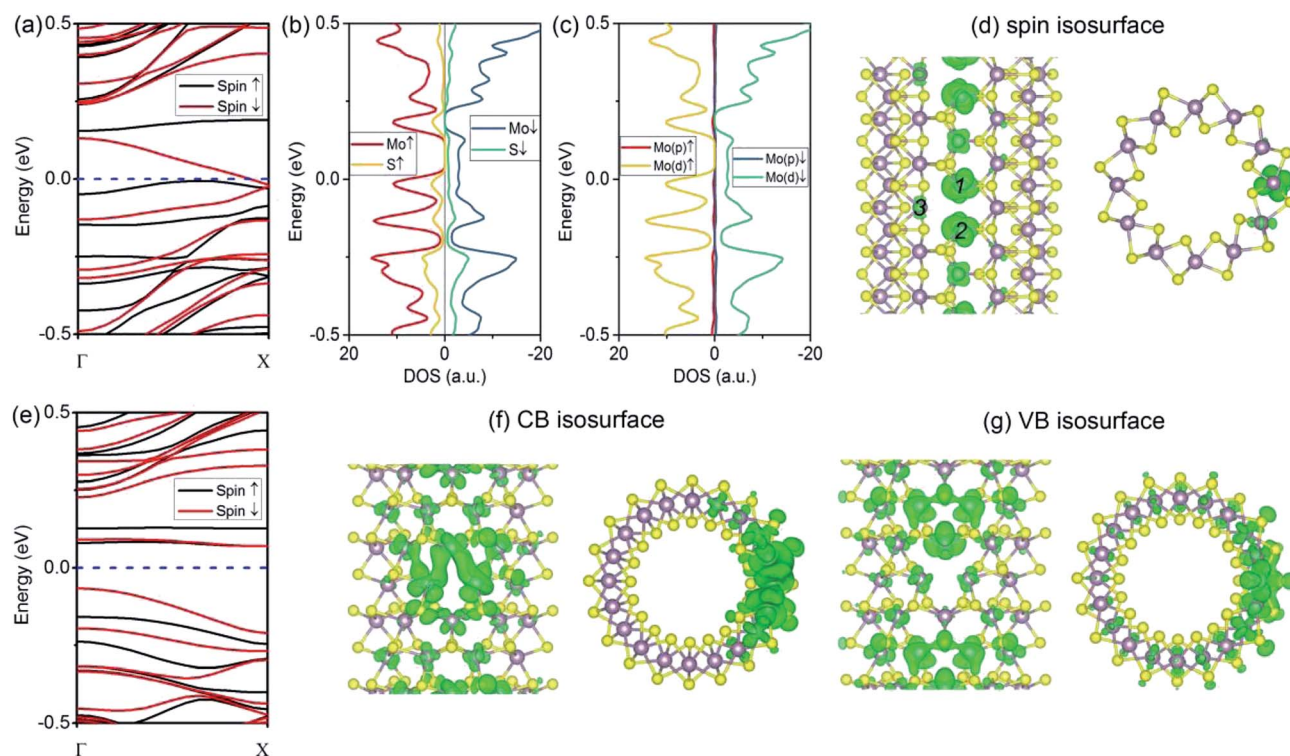


Fig. 3 (a) Band structures of a (6, 6) tube with a V_{S₂} defect; (b) atomic-projected DOS; (c) Mo-orbital-projected DOS; (d) spin isosurface (cutoff = $0.003|e|/\text{\AA}^3$); (e) band structures of a (10, 0) tube with a V_{S₂} defect; (f) and (g) charge isosurfaces of CB and VB (cutoff = $0.001|e|/\text{\AA}^3$).



These values are considerably larger than that of an ideal (10, 0) tube ($0.51m_e$). Hence, the CB is better recognized as an "impurity state." Similarly, the VB of spin-up is also considerably flat at the VBM, resulting in large m^* values for holes ($-4.65m_e$) (as compared to $-1.55m_e$ for an ideal tube). In Fig. 3f and g, we show the plots of the isosurfaces corresponding to the VB and CB. It is clear that the electron states are localized. The only exception is that the VB of spin-down is more dispersed in energy, representing holes of $-1.32m_e$.

3.3 MoS₂ tube with V_{Mo} defect

The introduction of a V_{Mo} defect corresponds to the formation energy of 2.25 eV for (6, 6) tubes and 1.89 eV for (10, 0) tubes. This value is comparable to that of a V_S vacancy. The removal of a single Mo atom leaves S-edge atoms, which are predicted to be stable from an earlier study.³³ The V_{Mo} defect also induced significant changes to the (6, 6) tube. As depicted by the spin-polarized band structures (Fig. 4a), the system becomes metallic, with one band representing spin-up and one band representing spin-down crossing the Fermi level. Therefore, in contrast to the (6, 6) tube bearing a V_{S₂} defect, the tube with the V_{Mo} defect has no spin selectivity. As the electron occupations are different for the two spin components, the system has a net magnetic moment of $1.10 \mu_B$, which is smaller than that of V_{S₂}. From the PDOS analyses (Fig. 4b and c), the net spin was mainly attributed to the d-orbital electrons from the Mo atoms. This phenomenon is consistent with the case of (6, 6) tubes with V_{S₂} defects. The spin isosurface is shown in Fig. 4d, from which it is

evident that spin polarization takes place not at the bare S atoms at the V_{Mo} edge, but mainly at the Mo atoms around the defect.

Finally, we assess the effects of V_{Mo} defects on (10, 0) tubes. From the band structures shown in Fig. 4e, it is clear that all the bands that represent spin-up and spin-down are degenerated; therefore, the system is nonpolarized. This is similar to the cases of the (6, 6) and (10, 0) tubes with V_S defects. The most significant change is that there are two bands at the Fermi level, contributing to metallic conductivity. From the isosurfaces of the electron states representing these two bands (Fig. 4f and g), it can be seen that the conducting states are mainly contributed by the Mo atoms that are spread along the tube's axial direction.

3.4 MoS₂ tube with V_{MoS₃} defect

Next, we consider the vacancy type with a relatively large size, *i.e.*, V_{MoS₃}. From our calculations, the introduction of a V_{MoS₃} vacancy requires a formation energy of 5.21 eV for the (6, 6) tube and 5.33 eV for the (10, 0) tube. These values are comparable to that of the V_{S₂} vacancy in MoS₂ tubes, as discussed above. In fact, the two species of defects, V_{S₂} and V_{MoS₃}, have been directly observed experimentally in the MoS₂ monolayer. Therefore, they can exist in the MoS₂ tube. For the electronic properties, we firstly considered the case of the (6, 6) tube. From the band structures shown in Fig. 5a, it is evident that the first change involves the (6, 6) tube bearing V_{MoS₃} becoming metallic. In addition, similar to the case of the (6, 6) tube with V_{S₂} (which is conducting only for spin-down), the bands representing the

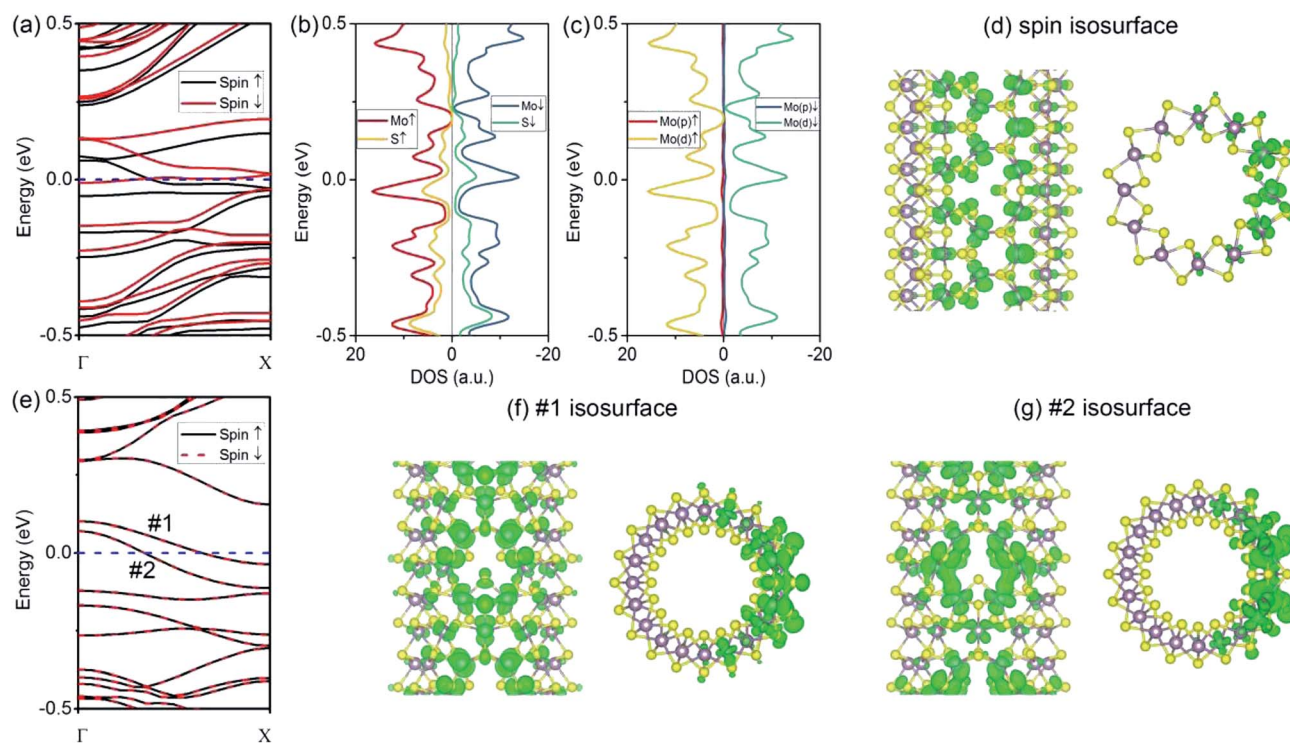


Fig. 4 (a) Band structures of a (6, 6) tube with a V_{Mo} defect; (b) atomic-projected DOS; (c) Mo-orbital-projected DOS; (d) spin isosurface (cutoff = $0.001|e|/\text{\AA}^3$); (e) band structures of a (10, 0) tube with a V_{Mo} defect; (f and g) charge isosurfaces of two bands at the Fermi level (cutoff = $0.001|e|/\text{\AA}^3$).



spin-down channel cross the Fermi level. However, the bandgap for the spin-up bands is 0.12 eV. Therefore, the (6, 6) tube with V_{MoS_3} is conducting for the spin-down channel, while the spin-up channel is blocked. This phenomenon is similar to the case of the (6, 6) tube with a V_{S_2} defect. As the bands near the Fermi level are severely modulated by the V_{MoS_3} defect, the net magnetic moment of each vacancy is as large as $4 \mu_{\text{B}}$. From the PDOS analyses shown in Fig. 5b and c, it is clear that the net spin was mainly contributed by the d-orbital electrons from the Mo atoms. Spatially, the net spin is located at the Mo atoms near the vacancy edge, as shown in Fig. 5d.

For the (10, 0) tube, it is still semiconducting, which is similar to the case of a pristine (10, 0) tube. As shown in Fig. 5e, there are still worthwhile spintronics that may exist for the (10, 0) tube. In detail, it is a direct-bandgap semiconductor for the spin-down channel, while it is indirect for the spin-up channel. Secondly, the system is also spin-polarized with integer net magnetic moment of $2 \mu_{\text{B}}$. From the PDOS analyses shown in Fig. 5f and g, the net spin was mainly attributed to the d-orbital electrons from the Mo atoms. From the spatial distributions shown in Fig. 5h, it is clear that all of them are localized at the Mo atoms even if the S atoms at the vacancy edges have dangling bonds. By comparing the spin distributions in the (6, 6) tube with those in the (10, 0) tube, it is found that the net spins on the Mo atoms arranged along the zigzag direction (the (6, 6) tube's axial direction) can form a strong ferromagnetic coupling, as shown in Fig. 5d, which is absent in the case of the (10, 0) tube (Fig. 5h).

3.5 MoS_2 tube with V_{MoS_6} defect

Finally, we analyze the V_{MoS_6} defect, which has the largest size. Initially, we analyzed the formation of such a defect. From our calculations, the formation energy of a V_{MoS_6} defect is rather high: 17.31 eV for the (6, 6) tube and 16.63 eV for the (10, 0) tube. For comparison, the corresponding value in a MoS_2 monolayer is estimated to be 11.0 eV.³⁶ Although the formation energy is high, such a vacancy is still found in the experiment,¹⁸ but the concentration might be low. For the (6, 6) tube with a V_{MoS_6} defect, the band structure (Fig. 6a) indicates that it is semiconducting. The two spin branches near the Fermi level are rather flat, indicating spatially localized states. Among all the situations in the present study, the V_{MoS_6} defect in the (6, 6) tube induces the most significant spin polarization with net magnetic moment of $6 \mu_{\text{B}}$. From the atomic- and orbital-projected DOS data (Fig. 6b and c), the net spin is mainly attributed to the d-orbital of Mo atoms, which is similar to the other spin-polarized states in the aforementioned systems. The spin isosurface shown in Fig. 6d also points to the fact that the spins are highly localized at the Mo atoms at the defect edge. By labeling the four atoms in Fig. 6d that bear a net spin (atom 4 is antiferromagnetically coupled), it is also worth noticing that the adjacent spins can form a strong coupling along the zigzag direction (tube's axis). In contrast, the (10, 0) tube with a V_{MoS_6} defect is nonmagnetic, although it is also rich in Mo-edge atoms (Fig. 6e). The CB and VB are relatively flat,

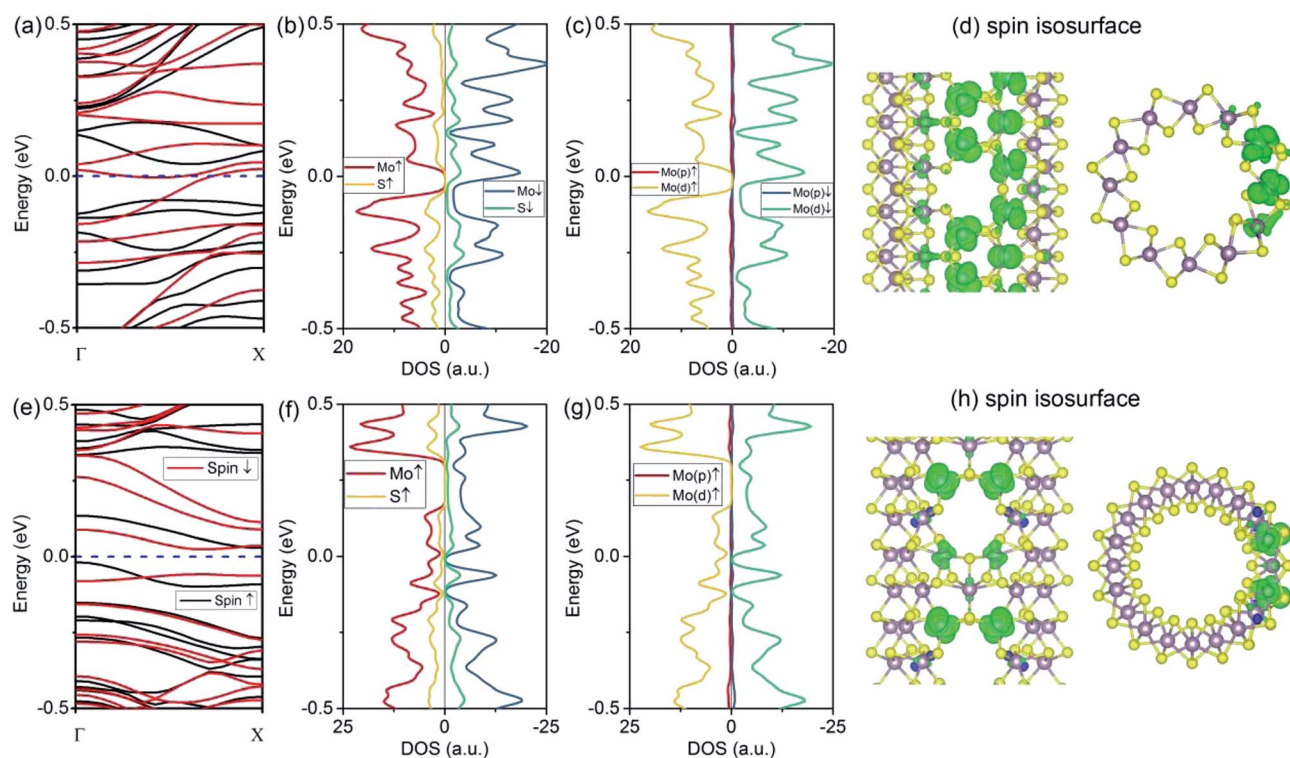


Fig. 5 (a) Band structures of a (6, 6) tube with a V_{MoS_3} defect; (b) atomic-projected DOS; (c) Mo-orbital-projected DOS; (d) spin isosurface (cutoff = 0.003 $|e|/\text{\AA}^3$); (e) band structures of a (10, 0) tube with a V_{MoS_3} defect; (f) atomic-projected DOS; (g) Mo-orbital-projected DOS; (h) spin isosurface (cutoff = 0.003 $|e|/\text{\AA}^3$).



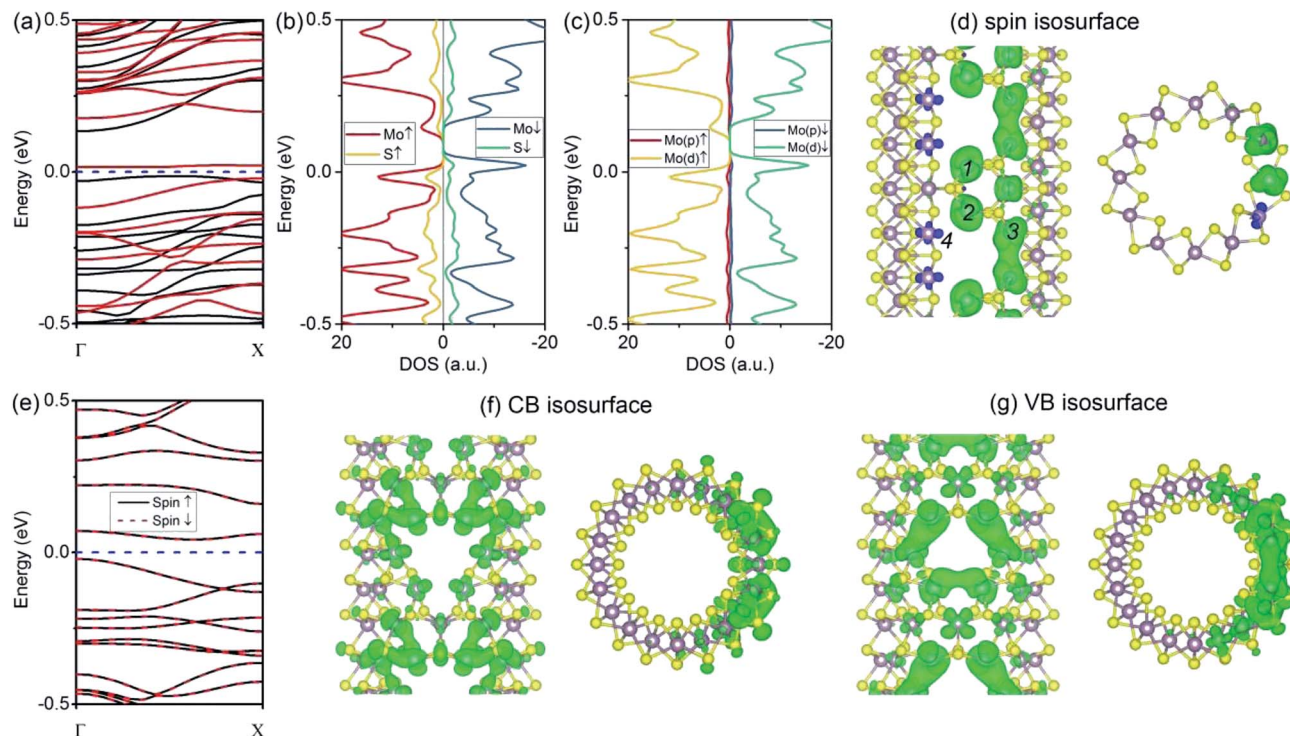


Fig. 6 (a) Band structures of a (6, 6) tube with a V_{MoS_6} defect; (b) atomic-projected DOS; (c) Mo-orbital-projected DOS; (d) spin isosurface (cutoff = $0.005|e|/\text{\AA}^3$); (e) band structures of a (10, 0) tube with a V_{MoS_6} defect; (f and g) charge isosurfaces of the CB and VB (cutoff = $0.002|e|/\text{\AA}^3$).

representing the local states, as demonstrated in Fig. 6f and g. From the abovementioned results, it is evident that spin polarization in the (10, 0) tube can only happen by introducing V_{MoS_3} defects.

4. Conclusions

To summarize, we have systematically investigated the electronic and magnetic properties of MoS_2 NTs with vacancy defects. We found that the (6, 6) tube with $V_{S_2}/V_{Mo}/V_{MoS_3}$ defects and the (10, 0) tube with V_{Mo} vacancy become metallic. It is interesting that the (6, 6) tube with V_{S_2}/V_{MoS_3} vacancies has a half-metallic character, which only allows the transport of electrons with only one spin direction, but exhibits an indirect bandgap of 0.16 eV for the opposite spin direction. In contrast, the (6, 6) tube with V_S/V_{MoS_6} defects and (10, 0) tube with $V_S/V_{S_2}/V_{MoS_3}/V_{MoS_6}$ defects remain semiconducting. With regard to the magnetic properties, it is found that the (6, 6) tube with $V_{S_2}/V_{Mo}/V_{MoS_3}/V_{MoS_6}$ defects can bear net magnetic moment. For the (10, 0) tube, magnetization was only observed for V_{MoS_3} defects. Therefore, our results demonstrate that defect engineering is an effective way to control the electronic and magnetic properties of MoS_2 NTs. By controlling the types of vacancies, metallic and half-metallic conductivities can be realized. We hope that our present findings can stimulate related experimental studies in these areas and assist formulating the ultimate functions of MoS_2 materials for future nanoelectronic and spintronic devices.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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