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Nanosheets and 2D-nanonetworks by mutually assisted self-assembly of fullerene clusters and DNA three-way junctions†

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Programmable construction of two dimensional (2D) nanoarchitectures using short DNA strands is of utmost interest in the context of DNA nanotechnology. Previously, we have demonstrated fullerene-cluster assisted self-assembly of short oligonucleotide duplexes into micrometer long, semiconducting nanowires. This report demonstrates the construction of micrometer-sized nanosheets and 2D-nanonetworks from the mutual self-assembly of fullerene nanoclusters with three way junction DNA (3WJ-DNA) and 3WJ-DNA with a 12-mer overhang (3WJ-OH), respectively. The interaction of unique sized fullerene clusters prepared from an aniline appended fullerene derivative, F-An, with two 3WJ-DNAs, namely, 3WJ-20 and 3WJ-30, having 20 and 30 nucleobases, respectively at each strand was characterized using UV-visible absorption, circular dichroism and fluorescence techniques. The morphological characterization of nanosheets embedded with F-An clusters was performed *via* AFM, TEM and DLS analyses. The programmability and structural tunability of the resultant nanostructures were further demonstrated using 3WJ-OH containing a cytosine rich, single stranded DNA 12-mer overhang, which forms entangled 2D-nanonetwork structures instead of nanosheets due to the differential interaction of F-An nanoclusters with single and duplex strands of 3WJ-OH. Moreover, the selective modification of the cytosine rich sequence present in 3WJ-OH with silver nanoclusters (AgNCs) resulted in significant enhancement in silver nanocluster fluorescence (~40%) compared to 3WJ-OH/AgNCs owing to the additional stability of AgNCs embedded in 2D nanostructures. This unique strategy of constructing DNA based 2D nanomaterials and their utilization in the integration of functional motifs could find application in the area of DNA nanotechnology and bio-molecular sensing.

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Introduction

The unparalleled precision and sequence programmability of hierarchical DNA nanostructures attracted immense attention due to their accessible integration capability of diverse components with superior functional properties.^{1–5} Due to the structural regularity and precisely defined dimensions, DNA has been extensively used as a scaffold or template for the precise dimension controlled organization of metal nanoparticle arrays,^{6,7} liposomes,⁸ carbon nanotubes,⁹ and proteins¹⁰ *via* various non-covalent strategies. Moreover, the construction of higher order, predefined 1D, 2D and 3D architectures was earlier reported using DNA origami and DNA tiles by

hybridizing a long circular DNA and short stapler strands.^{11–15} In particular, DNA origami has proven useful for immobilization of nanoscale functional objects, such as small molecules,¹⁶ biomolecules,¹⁷ photonic components,¹⁸ and moving DNA machines.¹⁹ For instance, Sleiman and co-workers showed a unique strategy that chemically transfers a discrete pattern of DNA strands from a three-dimensional DNA nanostructure to gold nanoparticles with high site-specific and addressable properties.²⁰ However, addressing the fabrication of higher order DNA nanomaterials non-covalently with the same precision as that of DNA origami without compromising the intrinsic dynamic and functional properties remains as a challenging task.²¹ This could decrease the cost of synthesis and tedious periodic assembly of long multiple DNA components.²²

On the other hand, DNA has been used as a scaffold or template in the bottom-up assembly of metal ions,^{23–25} small molecules,²⁶ oligomers,^{27,28} polymers,^{29,30} and nanoparticles,^{31,32} into various nanostructures. Moreover, the long range ordered assembly of short DNA strands into one dimensional (1D) nanofibers/nanowires was reported using various design strategies such as DNA templated assembly,^{33,34}

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complementary hydrogen bonding,³⁵ DNA condensation^{36,37} and π -chromophore assisted self-assembly.^{38–40} The directional properties and several secondary interactions such as electrostatic, groove binding, intercalation and π - π interactions are the primary reason for the 1D growth of DNA-templated nanostructures. The construction of non-covalently assembled nanosheets and 2D-nanonetworks using short DNA strands with the capability of integrating the functional components of interest is rarely reported. For example, Reji Varghese and co-workers have exquisitely demonstrated the construction of crystalline DNA-decorated nanosheets *via* an amphiphile driven strategy using short DNA strands covalently tethered with hexa-*peri*-benzocoronene⁴¹ and tetraphenylethylene⁴² chromophores. The presence of single stranded DNA (ssDNA) in these nanosheets offers an effective way to immobilize the catalytically active DNA modified gold nanoparticles *via* complementary hybridization. In another example, the Häner group developed chimeric oligomers consisting of a covalently bound short oligonucleotide strand to an array of phosphodiester-linked pyrenes which self-assembled into 1D ribbons and 2D nanosheets depending on the strand composition and conditions driven by the aromatic stacking and hydrophobic interactions.⁴³ They also showed the prospect of decorating the preassembled chimeric DNA conjugates with gold nanoparticle through complementary DNA hybridization.⁴⁴

The development of 2D nanostructures using non-covalently assembled short DNA strands with an opportunity to integrate functional motifs is extremely important. Recently, we have reported the fullerene cluster assisted mutual self-assembly of short duplex DNA strands into micrometer sized, semi-conducting nanowires and chiral templated self-assembly in the presence of CT-DNA using groove binding and π - π interactions.^{45,46} The prospect of fullerene clusters in bringing small oligonucleotide duplexes into nanowires opens up a wide range of opportunities in DNA nanotechnology.^{47–49} The possibilities of extending these assemblies beyond the 1D nanostructure, incorporating other DNA constructs, are immediate challenges which will expand the scope and generality of the scheme. Herein, we report the construction of nanosheets through mutual self-assembly of **F-An** clusters and 'Y' shaped **3WJ-DNA** through multiple non-covalent interactions.

The interactions of **F-An** clusters with two different **3WJ-DNA** namely, **3WJ-20** and **3WJ-30**, were investigated using UV-visible absorption, fluorescence, and circular dichroism (CD) studies and correspondingly, the morphological behaviour was probed through atomic force microscopy (AFM), transmission electron microscopy (TEM) and dynamic light scattering (DLS) techniques. Furthermore, the programmability of the self-assembly processes was explored by using **3WJ-DNA** with a 12-mer cytosine rich overhang (**3WJ-OH**) which formed a 2D-nanonetwork structure with **F-An** clusters. Moreover, the **3WJ-OH/F-An** 2D-nanonetwork was used to selectively stabilize **AgNCs** with enhanced fluorescence properties. To the best of our knowledge, this represents the first example where a non-covalently assembled 2D nanostructure constructed from mutual self-

assembly of short DNA strands and fullerene nanoclusters were used for the stabilization of fluorescent **AgNCs**.

Experimental section

Materials and methods

All the chemicals and reagents were used as received without further purification. Ethidium bromide (**EB**), silver nitrate (AgNO_3 , 99.0%), sodium borohydride (NaBH_4 , 98.0%), sodium phosphate dibasic anhydrous (Na_2HPO_4), sodium phosphate monobasic monohydrate ($\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$) and DMSO solvent (spectroscopic grade, 99.9%) were purchased from Sigma Aldrich. Milli Q deionized water (18.2 M Ω cm) was used throughout the studies. All the experiments were performed in a 10% DMSO-PBS solvent mixture. The detailed synthesis and characterization of the fullerene derivative, **F-An**, was reported previously.⁴⁵ The fullerene derivative, **F-An**, was initially dissolved in DMSO and then diluted with deionized water to obtain appropriate concentrations in a 10% DMSO-PBS solvent mixture.

The UV-visible absorption spectra were recorded on a Shimadzu UV-2600 Spectrophotometer. For UV-visible titrations, the concentration of the fullerene derivative (3 μM) was kept constant and DNA solution (100 μM) in small aliquots was added. The fluorescence emission spectra were recorded on a SPEX FLUOROLOG-3 (FL3-221) spectrofluorimeter. The wavelength of excitation for **EB**-displacement assay was 515 nm. The fluorescence of the DNA/**EB** complex (1 μM /30 μM) was monitored by subsequent addition of **F-An** in small portions. Circular dichroism measurements were performed on a JASCO J-810 spectropolarimeter having a Peltier controlled thermostatic cell holder where the sensitivity, integration time and scan rate were chosen appropriately. All optical measurements were carried out using 1 cm cuvettes at room temperature (25 ± 1 °C) unless otherwise mentioned.

DNA oligonucleotide sequences

Oligonucleotide sequences used in the present work were as follows:

3WJ-20

DNA 1: 5'-GGT GGC GAG AGC GAC GAT CC-3'

DNA 2: 5'-GGA TCG CAG AGT TGA CC-3'

DNA 3: 5'-GGT CAA CTC TTC TCG CCA CC-3'

3WJ-30

DNA 4: 5'-ATT ACG GTG GCG AGA GCG ACG ATC CTG TAT-3'

DNA 5: 5'-ATA CAG GAT CGT CGC AGA GTT GAC CAT GCT-3'

DNA 6: 5'-AGC ATG GTC AAC TCT CGC CAC CGT AAT-3'

3WJ-OH

DNA 7: 5'-GGT GGC GAG AGC GAC GAT CCC CTC CTT CCT CC-3'

DNA 8: 5'-GGA TCG CAG AGT TGA CCC CTC CTT CCT CC-3'

DNA 9: 5'-GGT CAA CTC TTC TCG CCA CCC CTC CTT CCT CC-3'

Phosphoramidites for automated solid phase DNA synthesis were purchased from Glen Research (<http://www.glenresearch.com/index.php>). Other reagents needed for



DNA synthesis were purchased from Sigma Aldrich. Oligonucleotides were synthesized on a H-8 K&A DNA synthesizer by following standard protocols. The synthesized oligonucleotides were purified by LC-6AD Shimadzu Reverse Phase High Performance Liquid Chromatography (RP-HPLC) with an Inertsil® ODS-3, 5 μm , 10 \times 250 mm reversed phase column. The oligonucleotide samples were then desalted using a Sep-Pak C18 cartridge. High resolution mass spectral (HRMS) analysis was performed on a Thermo Scientific Q Exactive Hybrid Quadrupole-Orbitrap electrospray ionization mass spectrometer (ESI-MS) instrument. Measurements were done in ESI mode. The concentrations of the corresponding strands were determined by monitoring UV absorbance at 260 nm. The hybridization of DNA was carried out by annealing (heating at 70 $^{\circ}\text{C}$ for 10 min followed by slow cooling at a rate of 1 $^{\circ}\text{C min}^{-1}$) equal concentrations of single strands (**3WJ-20**: DNA 1, DNA 2 & DNA 3; **3WJ-30**: DNA 4, DNA 5 & DNA 6; **3WJ-OH**: DNA 7, DNA 8 & DNA 9) in 10 mM phosphate buffer containing 2 mM NaCl, pH = 7.4. Melting curves were measured by monitoring the absorbance at 260 nm while the temperature was increased at a rate of 1 $^{\circ}\text{C min}^{-1}$. A stock solution of 100 μM DNA solutions was used for all further studies.

Preparation of DNA stabilized silver nanoclusters

DNA stabilized **AgNCs** were synthesized according to a modified literature procedure.⁵⁰ In a typical experiment, 7.5 μL of **3WJ-OH** (100 μM) was mixed with 5 μL of AgNO_3 (1.35 mM) in 10 mM phosphate buffer and 100 mM NaCl. The mixture was gently stirred for 10 minutes and reduced using a freshly prepared solution of NaBH_4 (2.5 μL , 1.35 mM). The standardized concentrations were chosen to provide a DNA : AgNO_3 : NaBH_4 molar ratio of 1 : 9 : 4.5 which was found to give better fluorescent **AgNCs**. The sample was further vortexed for 2 minutes and allowed to react in the dark at room temperature for 12 hours before spectral measurements. The final concentrations of **3WJ-OH**, AgNO_3 and NaBH_4 were 15 μM , 135 μM and 67.5 μM , respectively (final volume of 50 μL).

The concentration of DNA-**AgNCs** was referenced to the concentration of DNA unless otherwise mentioned. In the case of **3WJ-OH/F-An** stabilized **AgNC** preparation, a similar procedure was followed in which **3WJ-OH** was replaced with **3WJ-OH/F-An** (1 : 1, 15 μM each). Similarly, the control experiments were also carried out in the absence of a DNA template and in the absence of an overhang (using **3WJ-20**). A solution of 20 μL DNA-**AgNCs** (15 μM) diluted to 1 mL using 10 mM phosphate buffer and 100 mM NaCl was used for all further photophysical and morphological studies.

AFM/TEM sample preparations

Samples for AFM analysis were prepared by drop-casting 10 μL solution of **3WJ-20/F-An**, **3WJ-30/F-An** and **3WJ-OH/F-An** (1 : 1, 3 μM each) onto a freshly cleaved mica sheet under a humidity controlled atmosphere (<40% relative humidity) and were dried under air. A BRUKER MULTIMODE AFM operating in a tapping mode regime was used to record AFM images under ambient conditions. Micro-fabricated TiN cantilever tips (NT-MDT-NSG

series) with a resonance frequency of 299 kHz and a spring constant of 20 to 80 Nm^{-1} were used. AFM section analysis was done offline. TEM samples were prepared by drop casting 10 μL solution of the sample on top of a carbon-coated copper grid (400 mesh). The samples were allowed to adsorb on the grid overnight under ambient conditions and Millipore water (2 \times 10 μL) was added to remove buffer salts. TEM analysis was performed with a JEOL-JEM0310 high-resolution transmission electron microscope with an accelerating voltage of 100 kV and a beam current of 65 A. The samples were imaged with a Hamamatsu ORCA CCD camera. The DLS experiments were performed in quartz cuvettes using the same solutions which are used for microscopic studies. The hydrodynamic diameters of the nanoclusters were determined using a Malvern Zeta Nano-ZS system.

Results and discussion

Design and synthesis

The detailed synthesis of the aniline appended C_{60} -fullerene derivative, **F-An**, with a triethyleneglycol spacer (Fig. 1a) and the characterization of unique sized nanoclusters (3–5 nm) it forms in 10% DMSO-PBS were reported previously.⁴⁵ Both fullerene and substituted aniline groups are known to have groove binding and intercalative interactions with DNA.^{51,52} The goal of our approach is to construct DNA based 2D nanostructures from mutual self-assembly of short DNA strands and **F-An** nanoclusters. In this context, we have designed three 'Y' shaped **3WJ-DNA** structures having variable size and shape in order to investigate their interaction and self-assembly in the presence of **F-An** nanoclusters. All the oligonucleotide sequences used in the present work (Fig. 1b–d) were synthesized through solid phase DNA synthesis, purified *via* HPLC and characterized by ESI-MS analysis. Three different **3WJ-DNA** namely **3WJ-20** (DNA 1, DNA 2 and DNA 3), **3WJ-30** (DNA 4, DNA 5 and DNA 6) and

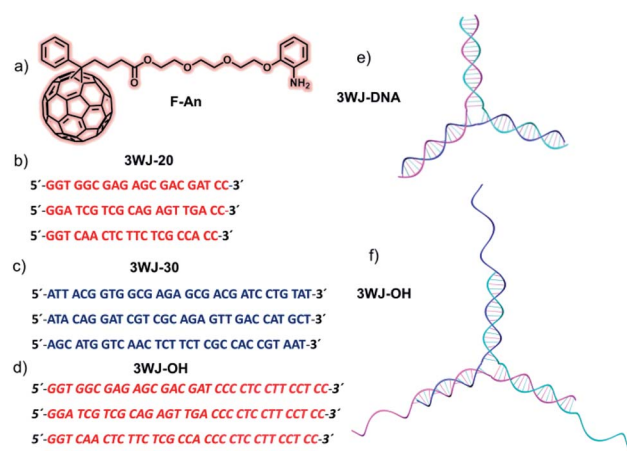


Fig. 1 (a) Chemical structure of the fullerene derivative, **F-An**, and oligonucleotide sequences used for the construction of (b) **3WJ-DNA** with 20 nucleobases (**3WJ-20**) and (c) 30 nucleobases (**3WJ-30**) and (d) **3WJ-overhang DNA** (**3WJ-OH**). General pictorial representation of (e) **3WJ-DNA** and (f) corresponding **3WJ-OH**.



3WJ-OH (DNA 7, DNA 8 and DNA 9) were assembled *via* complementary hybridization of respective short DNA strands following a modified literature report.⁵³ Here, **3WJ-20** and **3WJ-30** DNA consist of 20 and 30 nucleobases which form 'Y' shaped **3WJ-DNA** structures (Fig. 1e), while **3WJ-OH** is similar to **3WJ-20** with an extended, cytosine rich, ssDNA 12-mer sequence as an overhang and hence consists of both dsDNA and ssDNA within the 3WJ structure (Fig. 1f). The formation of 3WJ structures was confirmed by thermal denaturation studies.⁵⁴ All the interaction studies and morphological characterization were carried out using 10% DMSO-PBS (containing 10 mM phosphate buffer and 2 mM NaCl, pH = 7.4) unless otherwise mentioned.

F-An nanocluster/3WJ-DNA interactions

The interactions of **F-An** clusters with **3WJ-20** and **3WJ-30** were studied using UV-visible absorption spectroscopy. Upon sequential addition of **3WJ-20** or **3WJ-30** to 10% DMSO-PBS containing **F-An** clusters, the absorption spectra showed a gradual increase at 260 nm with a concomitant decrease in the corresponding **F-An** absorption at 336 nm (Fig. S1a and S2a†). The changes in the absorption at 260 nm include contributions from both decrease in the nucleobase absorption in the presence of **F-An** and enhancement due to the increase in **3WJ-DNA** concentration. These opposing trends result in an apparent increase in the absorption, but with a clear decrease compared to a blank **3WJ-DNA** titration experiment without the **F-An** nanoclusters (Fig. S3a†). Both **3WJ-20** and **3WJ-30** exhibited 26% and 28% hypochromicity at 336 nm, respectively, with an isosbestic point at 300 nm (inset of Fig. S1a and S2a†). The intrinsic binding constant was calculated through a half reciprocal plot and was found to be $5.46 \times 10^5 \text{ M}^{-1}$ (Fig. S1b and S2b†). The control titration between **F-An** clusters and DNA 1 single strands under similar experimental conditions did not show any hypochromicity at 336 nm (Fig. S3b†). These results clearly indicate the interaction between **F-An** clusters and **3WJ-DNA**, through intercalative or groove binding modes. Ethidium bromide (**EB**) based fluorescence displacement assay was used to get more insight into the mode of binding. The titration of **3WJ-20** or **3WJ-30** with **EB** provided highly fluorescent complexes (Fig. S4†) due to the intercalative interaction of **EB** with the duplex. The subsequent titration of **F-An** clusters with the **3WJ-30/EB** complex showed negligible changes in the fluorescence intensity even up to 30 μM **F-An** revealing negligible displacement of the intercalated **EB** chromophores (Fig. S5†). These results indicate the inability of **F-An** nanoclusters to competitively displace the intercalated **EB** chromophores and hence rule out any significant intercalative mode of binding for the **F-An** nanoclusters. Furthermore, thermal denaturation studies of **3WJ-30** in the presence of **F-An** clusters (1 : 1, 3 μM each; $T_m = 48^\circ\text{C}$) showed a destabilization of $\sim 7^\circ\text{C}$ compared to **3WJ-30** alone (3 μM ; $T_m = 55^\circ\text{C}$, Fig. S6b†), which is indicative of possible groove interactions. Negligible CD changes for **3WJ-DNA** in the presence of 1 : 1 **F-An** clusters also rule out the intercalative mode of binding (Fig. S7†). These results unambiguously confirm the interaction of **F-An** clusters with **3WJ-DNA** through groove binding interactions of the fullerene and aniline moieties.^{45,46,52,53}

The morphological studies accompanied by the interaction of **F-An** clusters with **3WJ-DNA** were performed using AFM, TEM and DLS techniques. AFM studies of both **3WJ-30/F-An** (1 : 1) and **3WJ-20/F-An** (1 : 1) solutions drop cast on mica revealed the formation of micrometer sized, few layered nanosheets with a height ranging from 2–3 nm indicating a 2D nanoassembly as envisaged (Fig. 2a and b). The sheet-like morphology is a consequence of the three-way branched DNA geometry with a propensity of nanostructure growth in 2D directions which is in contrast to the 1D self-assembly of **F-An** clusters in the presence of unbranched dsDNA.⁴⁵

High resolution TEM analysis of drop cast solutions of 1 : 1 **3WJ-30/F-An** also exhibited micrometer sized nanosheets, confirming the observations made during the AFM analyses (Fig. 2c). Furthermore, the embedded **F-An** clusters with an average size of $\sim 3.2 \text{ nm}$ were clearly visible in the high resolution TEM image (Fig. 2d), indicating the role of these nanoclusters in the observed nanosheet morphology. Under identical conditions, the AFM and TEM analyses of **F-An** alone showed 3–5 nm clusters as reported earlier (Fig. S8†).⁴⁵ EDAX measurements validated the signatures of phosphorus and nitrogen atoms throughout the nanosheets confirming the presence of **3WJ-DNA** (Fig. S9†). Both **3WJ-30** and **3WJ-OH** in 10% DMSO-PBS did not show any DLS number average size distribution, indicating a completely dissolved monomeric state under the experimental conditions (Fig. S10a†), whereas **F-An** alone showed a hydrodynamic radius of 10 nm. **3WJ-30** and **3WJ-OH** in the presence of **F-An** showed observable changes in the DLS compared to bare **3WJ-DNA** and **F-An** with a smooth correlogram (Fig. S10b†) signifying the self-assembly of **3WJ-DNA** and **F-An** nanoclusters. A schematic representation of the proposed self-assembly of **3WJ-DNA** and **F-An** clusters is given in Fig. 2e.

F-An nanocluster/3WJ-OH DNA interactions

To further explore the application of these nanosheets, we replaced **3WJ-DNA** by 3WJ-overhang DNA (**3WJ-OH**, Fig. 1d and f) with two clear objectives: (i) to get more insight into the resultant nanostructure due to the differential interaction of **F-An** clusters with single stranded and double stranded DNA and (ii) application of these nanostructures in the construction of DNA overhang stabilized fluorescent AgNCs. Initially, the morphological behaviour of the **3WJ-OH/F-An** (1 : 1) complex was probed using AFM and TEM analyses. AFM studies exhibited a micrometer sized entangled 2D-nanonetwork structure with a height of about 2–3 nm representing the distinct interaction of **F-An** clusters with **3WJ-OH** compared to **3WJ-DNA** (Fig. 3a and b). This could be attributed to the existence of a single stranded DNA overhang in **3WJ-OH** which perturbs the continuous 2D growth of the nanostructure unlike that in **3WJ-DNA**. TEM analysis of the same solution also revealed the 2D-nanonetwork structure, in agreement with the result obtained from AFM studies (Fig. 3c and d). EDAX analysis further confirmed the presence of **3WJ-OH** DNA as observed in the case of nanosheets (Fig. S11†). The self-assembly process between **3WJ-OH** and **F-An** clusters is pictorially illustrated in Fig. 3e.



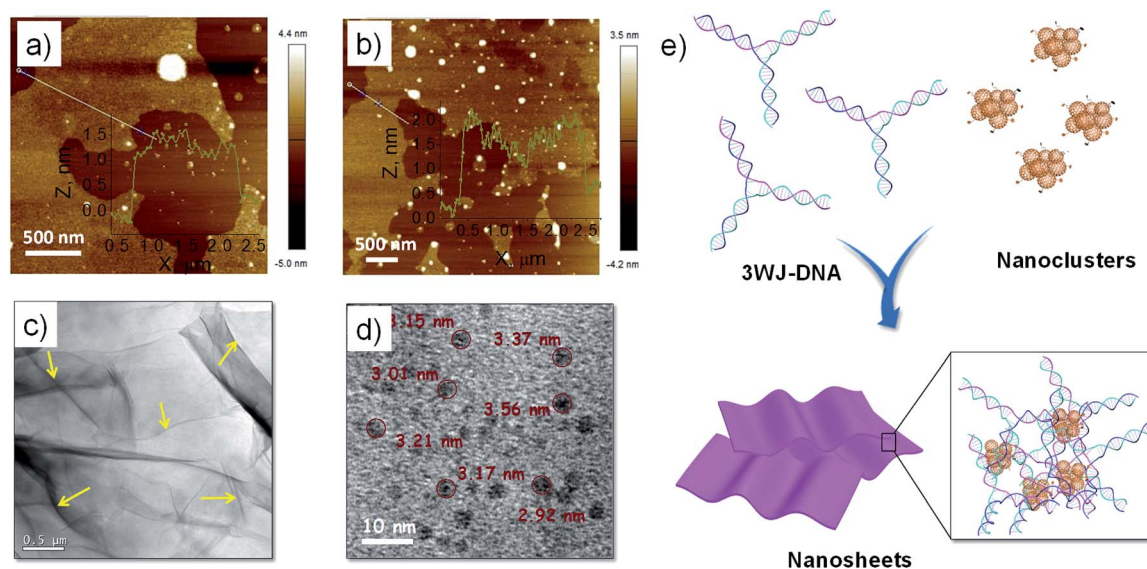


Fig. 2 AFM height image of (a) 3WJ-30/F-An and (b) 3WJ-20/F-An (1 : 1, 3 μ M each) showing micrometer sized nanosheets. Inset of a and b shows the corresponding height images. (c) Corresponding TEM image of 3WJ-30/F-An showing few layered nanosheets (indicated with yellow arrow marks). (d) Zoomed portion of the TEM image in c revealing the \sim 3 nm sized F-An clusters embedded in the nanosheets. (e) Schematic illustration of mutually assisted self-assembly between 3WJ-DNA and F-An clusters into nanosheets.

3WJ-OH/F-An stabilized silver nanoclusters

Ever since the first report by Dickson and co-workers,⁵⁴ DNA-stabilized fluorescent AgNCs have attracted widespread interest due to their many advantages including easy synthesis and sequence dependent emission properties.^{55–57} Though DNA-stabilized AgNCs have shown great promise for information storage,⁵⁸ bio-labelling⁵⁹ and sensing,^{60–62} the synthesis of stable AgNCs with enhanced fluorescence remains challenging. The

cytosine rich overhang sequence present in 3WJ-OH was chosen specifically to obtain fluorescence turn on emission in the presence of templated AgNCs.⁶³ The detailed description about the synthesis of DNA-stabilized AgNCs is provided in the Experimental section.

The emission of AgNCs stabilized with 3WJ-OH and 3WJ-OH/F-An 2D-nanonetworks was monitored under similar conditions. The 3WJ-OH/AgNC complex showed characteristic

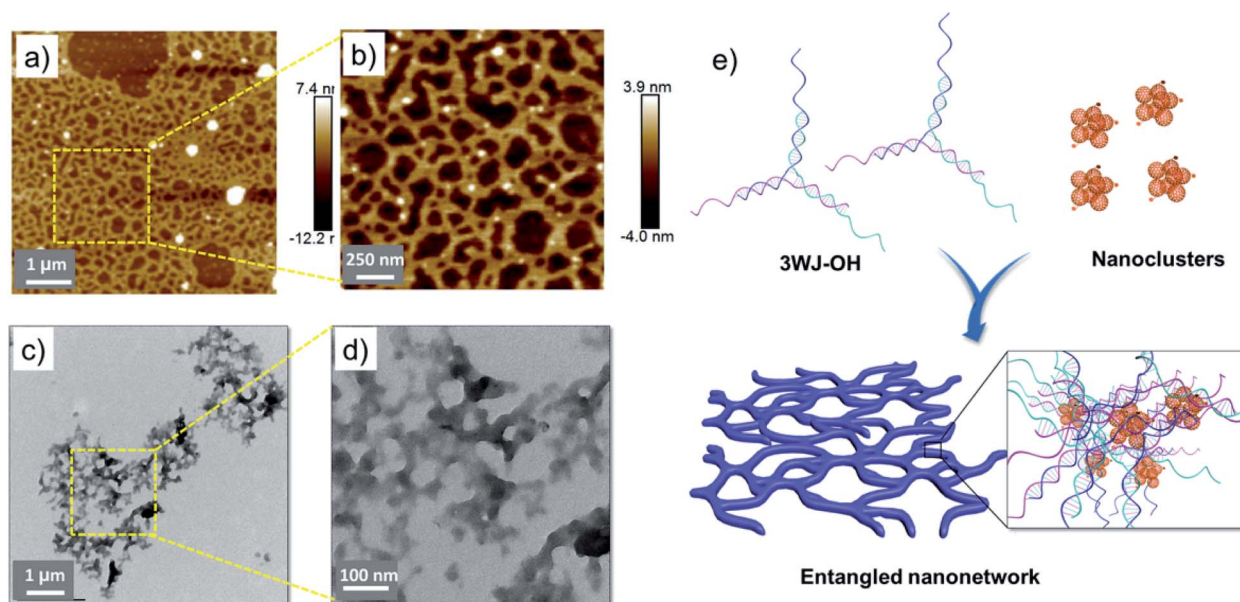


Fig. 3 AFM height image of (a) 3WJ-OH/F-An and (b) zoomed AFM image of the selected portion showing the micrometer sized entangled 2D-nanonetwork. (c) Corresponding TEM image of 3WJ-OH/F-An and (d) zoomed TEM image of the selected portion. (e) Schematic illustration of mutually assisted self-assembly of 3WJ-OH and F-An clusters into an entangled 2D-nanonetwork structure.



fluorescence emission maximum at 632 nm ($\lambda_{\text{exc}} = 560$ nm, Fig. 4a(ii)), which corresponds to the cytosine rich overhang stabilized AgNCs as reported earlier.⁶³ Furthermore, 3WJ-OH/F-An 2D-nanonetwork stabilized AgNCs showed pronounced emission intensity under similar excitation conditions compared to 3WJ-OH/AgNCs (Fig. 4a(iii)), reporting ~40% fluorescence enhancement relative to the bare 3WJ-OH/AgNCs (inset, Fig. 4a). Both 3WJ-20 and 3WJ-30 without the cytosine rich overhang essential for the formation of the oligonucleotide stabilized AgNCs showed negligible formation of fluorescent AgNCs under similar conditions. The control samples 3WJ-30/AgNPs, F-An alone and AgNPs without DNA did not show any fluorescence which supports the requirement of a DNA-overhang for the fluorescent AgNC formation and stabilization (Fig. 4a(i)). The AgNC fluorescence enhancement and stabilization by various 3WJ-DNAs are schematically represented in Fig. 4b–d.

Further analysis of 3WJ-OH/AgNCs using HR-TEM showed the presence of 3–5 nm sized silver nanoclusters (Fig. S12a†) along with larger nanoparticles. The signature of silver atoms from the obtained 3WJ-OH stabilized nanoclusters was evidenced from EDAX analysis (Fig. S13a†). TEM analyses of 3WJ-OH/F-An stabilized AgNCs clearly indicated the formation of nanosheet structures unlike previously observed 3WJ-OH/F-An 2D-nanonetworks (Fig. S12b and S12c†). This structural realignment of the nanonetwork into rigid nanosheets might be due to the stabilization of AgNCs on the ssDNA overhang present in the 3WJ-OH/F-An 2D-nanonetwork. The presence of silver atoms throughout the nanosheets was further proved through EDAX measurement (Fig. S13b†). These morphological and fluorescence results demonstrate the structural

programmability of 2D nanoarchitectures constructed from 3WJ-DNA/F-An and their application as a platform for the immobilization of AgNCs.

Conclusions

In summary, we have presented a new strategy for hierarchical self-assembly of unmodified branched DNA and fullerene nanoclusters into 2D nanomaterials. The interactions between unique sized F-An clusters and 3WJ-DNA used in this study lead to the formation of micrometer sized nanosheets. Furthermore, we have demonstrated the structural tunability and generality of this approach using 3WJ-OH, a modified 3WJ-DNA with a cytosine rich single strand overhang, which upon interaction with F-An clusters formed extended entangled 2D-nanonetwork structures. The application of these extended nanostructures was presented by comparing the fluorescence of AgNCs stabilized in the presence of 3WJ-OH/F-An, 3WJ-OH and 3WJ-DNA. 3WJ-OH/F-An stabilized AgNCs exhibited ~40% enhanced emission relative to bare 3WJ-OH/AgNCs. Thus, the mutual self-assembly strategies demonstrated here could be useful for self-assembly of branched DNA into 2D nanomaterials with possible post-functionalization and hence open up a new avenue in the field of DNA nanotechnology.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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References

- 1 F. A. Aldaye, A. L. Palmer and H. F. Sleiman, *Science*, 2008, **321**, 1795–1799.
- 2 K. M. M. Carneiro, G. D. Hamblin, K. D. Hänni, J. Fakhoury, M. K. Nayak, G. Rizis, C. K. McLaughlin, H. S. Bazzi and H. F. Sleiman, *Chem. Sci.*, 2012, **3**, 1980–1986.
- 3 S. K. Albert, H. V. P. Thelu, M. Golla, N. Krishnan, S. Chaudhary and R. Varghese, *Angew. Chem., Int. Ed.*, 2014, **53**, 8352–8357.
- 4 H. V. P. Thelu, S. K. Albert, M. Golla, N. Krishnan, D. Ram, S. M. Srinivasula and R. Varghese, *Nanoscale*, 2018, **10**, 222–230.
- 5 B. Chakraborty, N. Jonoska and N. C. Seeman, *Chem. Sci.*, 2012, **3**, 168–176.
- 6 J. Zheng, P. E. Constantinou, C. Micheel, A. P. Alivisatos, R. A. Kiehl and N. C. Seeman, *Nano Lett.*, 2006, **6**, 1502–1504.

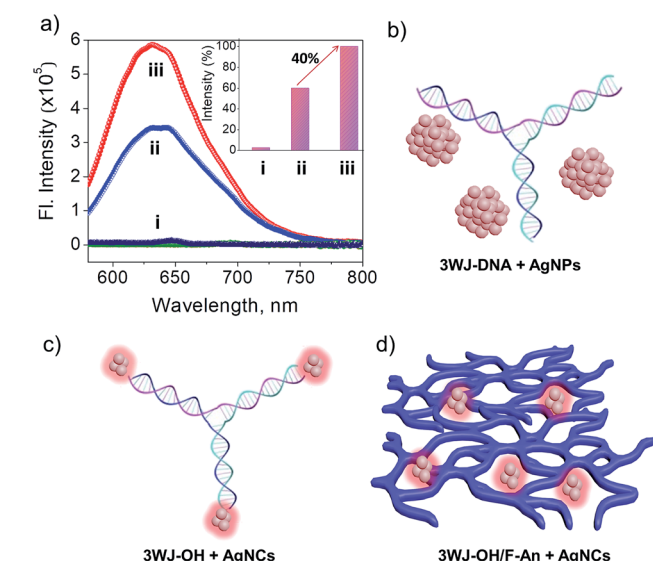


Fig. 4 (a) The fluorescence emission spectra of AgNCs in the presence of different DNA templates, (i) 3WJ-30, F-An alone, and AgNCs without DNA; (ii) 3WJ-OH and (iii) 3WJ-OH/F-An ($\lambda_{\text{exc}} = 560$ nm). The corresponding relative fluorescence intensity (inset). The pictorial representation of (b) 3WJ-30 + AgNPs, (c) 3WJ-OH/AgNCs and (d) 3WJ-OH/F-An 2D-nanonetwork stabilized AgNCs.



- 7 J. Sharma, R. Chhabra, Y. Liu, Y. Ke and H. Yan, *Angew. Chem., Int. Ed.*, 2006, **118**, 744–749.
- 8 N. Dave and J. Liu, *ACS Nano*, 2011, **5**, 1304–1312.
- 9 A. H. Bae, T. Hatano, N. Nakashima, H. Murakami and S. Shinkai, *Org. Biomol. Chem.*, 2004, **2**, 1139–1144.
- 10 M. Slim, N. Durisic, P. Grutter and H. F. Sleiman, *ChemBioChem*, 2007, **8**, 804–812.
- 11 T. Tørring, N. V. Voigt, J. Nangreave, H. Yan and K. V. Gothelf, *Chem. Soc. Rev.*, 2011, **40**, 5636–5646.
- 12 Z. Zhang, J. Song, F. Besenbacher, M. Dong and K. V. Gothelf, *Angew. Chem., Int. Ed.*, 2013, **125**, 9389–9393.
- 13 J. F. Rahbani, J. C. C. Hsu, P. Chidchob and H. F. Sleiman, *Nanoscale*, 2018, **10**, 13994–13999.
- 14 Y. Ke, G. Bellot, N. V. Voigt, E. Fradkov and W. M. Shih, *Chem. Sci.*, 2012, **3**, 2587–2597.
- 15 M. R. Jones, N. C. Seeman and C. A. Mirkin, *Science*, 2015, **347**, 1260901.
- 16 N. V. Voigt, T. Tørring, A. Rotaru, M. F. Jacobsen, J. B. Ravnsbæk, R. Subramani, W. Mamdouh, J. Kjems, A. Mokhir, F. Besenbacher and K. V. Gothelf, *Nat. Nanotechnol.*, 2010, **5**, 200.
- 17 M. Endo and H. Sugiyama, *Acc. Chem. Res.*, 2014, **47**, 1645–1653.
- 18 B. Ding, Z. Deng, H. Yan, S. Cabrini, R. N. Zuckermann and J. Bokor, *J. Am. Chem. Soc.*, 2010, **132**, 3248–3249.
- 19 N. D. Derr, B. S. Goodman, R. Jungmann, A. E. Leschziner, W. M. Shih and S. L. Reck-Peterson, *Science*, 2012, **338**, 662–665.
- 20 T. G. W. Edwardson, K. L. Lau, D. Bousmail, C. J. Serpell and H. F. Sleiman, *Nat. Chem.*, 2016, **8**, 162–170.
- 21 S. Sun, H. Yao, F. Zhang and J. Zhu, *Chem. Sci.*, 2015, **6**, 930–934.
- 22 A. V. Pinheiro, D. Han, W. M. Shih and H. Yan, *Nat. Nanotechnol.*, 2011, **6**, 763–772.
- 23 H. D. A. Mohamed, S. M. D. Watson, B. R. Horrocks and A. Houlton, *J. Mater. Chem. C*, 2015, **3**, 438–446.
- 24 J. Pate, F. Zamora, S. M. D. Watson, N. G. Wright, B. R. Horrocks and A. Houlton, *J. Mater. Chem. C*, 2014, **2**, 9265–9273.
- 25 S. M. D. Watson, A. R. Pike, J. Pate, A. Houlton and B. R. Horrocks, *Nanoscale*, 2014, **6**, 4027–4037.
- 26 V. Malinovskii, D. Wenger and R. Häner, *Chem. Soc. Rev.*, 2010, **39**, 410–422.
- 27 B. Datta and G. Schuster, *J. Am. Chem. Soc.*, 2008, **130**, 2965–2973.
- 28 B. Datta, G. B. Schuster, A. McCook, S. C. Harvey and K. Zakrzewska, *J. Am. Chem. Soc.*, 2006, **128**, 14428–14429.
- 29 L. Dong, T. Hollis, S. Fishwick, B. A. Connolly, N. G. Wright, B. R. Horrocks and A. Houlton, *Chem.–Eur. J.*, 2007, **13**, 822–828.
- 30 S. M. D. Watson, M. A. Galindo, B. R. Horrocks and A. Houlton, *J. Am. Chem. Soc.*, 2014, **136**, 6649–6655.
- 31 S. Saha, D. Bhatia and Y. Krishnan, *Small*, 2010, **6**, 1288–1292.
- 32 Q. Song, Y. Shi, D. He, S. Xu and J. Ouyang, *Chem.–Eur. J.*, 2015, **21**, 2417–2422.
- 33 P. G. A. Janssen, J. Vandenbergh, J. L. J. van Dongen, E. W. Meijer and A. P. H. J. Schenning, *J. Am. Chem. Soc.*, 2007, **129**, 6078–6079.
- 34 K. C. Hannah and B. A. Armitage, *Acc. Chem. Res.*, 2004, **37**, 845–853.
- 35 N. Avakyan, A. A. Greschner, F. Aldaye, C. J. Serpell, V. Toader, A. Petitjean and H. F. Sleiman, *Nat. Chem.*, 2016, **8**, 368–376.
- 36 T. Zhou, A. Llizo, C. Wang, G. Xu and Y. Yang, *Nanoscale*, 2013, **5**, 8288–8306.
- 37 G. Pardatscher, D. Bracha, S. S. Daube, O. Vonshak, F. C. Simmel and R. H. Bar-Ziv, *Nat. Nanotechnol.*, 2016, **11**, 1076–1081.
- 38 K. M. M. Carneiro, F. A. Aldaye and H. F. Sleiman, *J. Am. Chem. Soc.*, 2010, **132**, 679–685.
- 39 S. K. Albert, M. Golla, H. V. P. Thelu, N. Krishnan and R. Varghese, *Chem.–Eur. J.*, 2017, **23**, 8348–8352.
- 40 Y. Vyborna, M. Vybornyi, A. V. Rudnev and R. Häner, *Angew. Chem., Int. Ed.*, 2015, **54**, 7934–7938.
- 41 S. K. Albert, I. Sivakumar, M. Golla, H. V. P. Thelu, N. Krishnan, K. L. Joseph Libin, Ashish and R. Varghese, *J. Am. Chem. Soc.*, 2017, **139**, 17799–17802.
- 42 N. Krishnan, M. Golla, H. V. P. Thelu, S. K. Albert, S. Atchimnaidu, D. Perumal and R. Varghese, *Nanoscale*, 2018, **10**, 17174–17181.
- 43 Y. Vyborna, M. Vybornyi and R. Häner, *J. Am. Chem. Soc.*, 2015, **137**, 14051–14054.
- 44 Y. Vyborna, M. Vybornyi and R. Häner, *Chem. Commun.*, 2017, **53**, 5179–5181.
- 45 S. K. Vittala, S. K. Saraswathi and J. Joseph, *Chem.–Eur. J.*, 2017, **23**, 15759–15765.
- 46 S. K. Vittala and J. Joseph, *Faraday Discuss.*, 2018, **207**, 459–469.
- 47 M. G. Dirk, M. I. Beatriz, A. M. Carmen, W. Mateusz and M. Nazario, *Chem. Soc. Rev.*, 2009, **38**, 1587–1597.
- 48 S. S. Babu, H. Möhwald and T. Nakanishi, *Chem. Soc. Rev.*, 2010, **39**, 4021–4035.
- 49 G. I. Livshits, A. Stern, D. Rotem, N. Borovok, G. Eidelstein, A. Migliore, E. Penzo, S. J. Wind, R. Di Felice, S. S. Skourtis, J. C. Cuevas, L. Gurevich, A. B. Kotlyar and D. Porath, *Nat. Nanotechnol.*, 2014, **9**, 1040–1046.
- 50 C. I. Richards, S. Choi, J.-C. Hsiang, Y. Antoku, T. Vosch, A. Bongiorno, Y.-L. Tzeng and R. M. Dickson, *J. Am. Chem. Soc.*, 2008, **130**, 5038–5039.
- 51 A. M. Cassell, W. A. Scrivens and J. M. Tour, *Angew. Chem., Int. Ed.*, 1998, **37**, 1528–1531.
- 52 R. Nagarajan, W. Liu, J. Kumar, S. K. Tripathy, F. F. Bruno and L. A. Samuelson, *Macromolecules*, 2001, **34**, 3921–3927.
- 53 A. L. Benvin, Y. Creeger, G. W. Fisher, B. Ballou, A. S. Waggoner and B. A. Armitage, *J. Am. Chem. Soc.*, 2007, **129**, 2025–2034.
- 54 J. T. Petty, J. Zheng, N. V. Hud and R. M. Dickson, *J. Am. Chem. Soc.*, 2004, **126**, 5207–5212.
- 55 C. M. Ritchie, K. R. Johnsen, J. R. Kiser, Y. Antoku, R. M. Dickson and J. T. Petty, *J. Phys. Chem. C*, 2007, **111**, 175–181.



- 56 J. Sharma, H. C. Yeh, H. Yoo, J. H. Werner and J. S. Martinez, *Chem. Commun.*, 2010, **46**, 3280–3282.
- 57 D. Schultz and E. G. Gwinn, *Chem. Commun.*, 2012, **48**, 5748–5750.
- 58 T. Li, L. Zhang, J. Ai, S. Dong and E. Wang, *ACS Nano*, 2011, **5**, 6334–6338.
- 59 J. Yu, S. Choi and R. M. Dickson, *Angew. Chem., Int. Ed.*, 2009, **48**, 318–320.
- 60 E. Sharon, N. Enkin, H. B. Albada and I. Willner, *Chem. Commun.*, 2015, **51**, 1100–1103.
- 61 J. Sharma, H.-C. Yeh, H. Yoo, J. H. Werner and J. S. Martinez, *Chem. Commun.*, 2011, **47**, 2294–2296.
- 62 J. Chen, X. Ji, P. Tinnefeld and Z. He, *ACS Appl. Mater. Interfaces*, 2016, **8**, 1786–1794.
- 63 C. I. Richards, S. Choi, J.-C. Hsiang, Y. Antoku, T. Vosch, A. Bongiorno, Y.-L. Tzeng and R. M. Dickson, *J. Am. Chem. Soc.*, 2008, **130**, 5038–5039.

