


Cite this: *Nanoscale Adv.*, 2019, 1, 4359

Ultra-high conductivity of graphene nanoribbons doped with ordered nitrogen†

Xiao-Fei Li,[✉] Wei-Wei Yan,^a Jia-Rui Rao,[✉] Dong-Xue Liu,^a Xiang-Hua Zhang,^{ab} Xinrui Cao^c and Yi Luo^{*d}

Graphene is an attractive candidate for developing high conductivity materials (HCMs) owing to an extraordinary charge mobility. While graphene itself is a semi-metal with an inherently low carrier density, and methods used for increasing carrier density normally also cause a marked decrease in charge mobility. Here, we report that ordered nitrogen doping can induce a pronounced increase in carrier density but does not harm the high charge mobility of graphene nanoribbons (GNRs), giving rise to an unprecedented ultra-high conductivity in the system. Our first-principles calculations for orderly N-doped GNRs (referred to as C₅N-GNRs) show that N-doping causes a significant shift-up of the Fermi level (ΔE_F), resulting in the presence of multiple partially-filled energy bands (PFEDs) that primarily increase the carrier density of system. Notably, the PFEDs are delocalized well with integral and quantized transmissions, suggesting a negligible effect from N-doping on the charge mobility. Moreover, the PFEDs can cross the E_F multiple times as the ribbon widens, causing the conductivity to increase monotonically and reach ultra-high values ($>15G_0$) in sub-5 nm wide ribbons with either armchair or zigzag edges. Furthermore, a simple linear relationship between the doping concentration and the ΔE_F was obtained, which provides a robust means for controlling the conductivity of C₅N-GNRs. Our findings should be useful for understanding the effect of ordered atomic doping on the conductivity of graphene and may open new avenues for realizing graphene-based HCMs.

Received 25th July 2019
Accepted 3rd September 2019

DOI: 10.1039/c9na00458k

rsc.li/nanoscale-advances

Introduction

For many applications in electronics¹ such as field-effect transistors,² nano-electrodes,³ nano-wires,⁴ nano-circuits,⁵ printable conductors,^{6,7} and rechargeable batteries,⁸ high conductivity materials (HCMs) are fundamentally required and play important roles in determining their performance.⁹

Recent advances in synthetic approaches have allowed the use of nanometals to develop HCMs,^{10,11} and a high conductivity up to 105 S cm^{−1} has been achieved in hybrids of silver nanoparticle-nanowires.¹² Conjugated polymers have also been proposed for HCMs owing to their conductive π electrons,¹³ and a high conductivity of 1400 S cm^{−1} has been achieved in recent experiments.¹⁴ Moreover, ionogels,¹⁵ self-healing hydrogels,¹⁶

and metal-organic frameworks (MOFs)¹⁷ have also been proposed for HCMs, and a high conductivity of 40 S cm^{−1} has been obtained in experiments.¹⁸ However, most of the prepared HCMs are composites or hybrids with quite complex structures, making defined structure-conductivity relationships difficult to determine which is disadvantageous for the further development of HCMs.^{19,20}

Graphene has a simple structure and an extraordinary charge mobility,²¹ thus it has long been considered as a very attractive candidate for HCMs,^{22,23} and a high conductivity of 104 S m^{−1} for graphene pastes formed on polyimide substrate was obtained recently.²⁴ Although the structure-conductivity relationship is yet to be defined due to the complex structure of these graphene-based materials, the development of graphene-based HCMs is expected. HCMs require both a high carrier density and a high charge mobility. But these characteristics are like two sides of a coin; very difficult to be obtained in a single material.¹¹ Actually, graphene itself is semi-metallic with an inherently low carrier density.²¹ It is proved that an electric gating effect is a very effective and efficient way to tune the chemical potential and thus the carrier density of graphene. Without using gates, atomic doping is also an efficient approach to tune the chemical potential and thus the carrier density of graphene.²⁵ Among the common atomic boron (B), nitrogen (N), oxygen (O), and sulfur (S) doping approaches,^{26,27}

^aSchool of Optoelectronic Science and Engineering, Institute of Fundamental and Frontier Sciences, University of Electronic Science and Technology of China, Chengdu, Sichuan 610054, China. E-mail: xf.li@uestc.edu.cn

^bDepartment of Electrical and Information Engineering, Hunan Institute of Engineering, Xiangtan, Hunan 411101, China

^cInstitute of Theoretical Physics and Astrophysics, Department of Physics, Xiamen University, Xiamen 361005, China

^dHefei National Laboratory for Physical Science at the Microscale, University of Science and Technology of China, Hefei, Anhui 230026, P. R. China. E-mail: yiluo@ustc.edu.cn

† Electronic supplementary information (ESI) available: Fig. S1–S7. See DOI: 10.1039/c9na00458k



N-doping²⁸ is particularly powerful because it introduces impressive additive properties into graphene.^{29–32} While, N atoms usually distribute in the graphene network randomly with various N-species, they can act as scattering centres that lead to conductance fluctuation and substantially restrict the conductivity.^{25,33}

Very recently, orderly N-doped graphene was suggested to grow on Cu(111) *via* self-assembly of C₅NCl₅ precursors.³⁴ The resultant C₅N-graphene exhibits well-defined structures and a high carrier density of 1014 cm⁻². However, the ordered N-doping impact on the carrier mobility of graphene and the conductivity of C₅N-graphene are not yet evaluated; both of which are important for the development of graphene-based HCMs through ordered atomic doping.

For actual applications in nanoelectronics, one needs to reduce the size of graphene into the nanoscale. Sub-10 nm wide graphene nanoribbons (GNRs) are semiconductors, which are suitable for electronic nanodevices.^{35–39} Up to now, both the edge well-defined GNRs^{28,40–44} and orderly N-doped GNRs⁴⁵ are achievable. Thus, obtaining orderly N-doped GNRs (referred to as C₅N-GNRs) *via* advanced fabrication techniques is expected, and the C₅N-GNRs offer a good platform to explore the effect of ordered N-doping on the conductivity.

In this study, we demonstrate that ordered N-doping increases the carrier density by inducing a considerable shift of the Fermi level (E_F) of GNRs, but does not impact significantly on their high mobility. Our first principles results reveal that the conductivity of C₅N-GNRs can increase with increasing width and can reach ultrahigh values ($>15G_0$, $G_0 = \frac{2e^2}{h}$) in sub-5 nm wide ribbons. Moreover, a simple linear relationship between the doping concentration and the ΔE_F is obtained, which provides a robust means for controlling the conductivity of C₅N-GNRs. Our findings are useful in understanding the effect of ordered atomic doping on the conductivity of graphene and will open up new avenues for realizing graphene-based HCMs.

Models and methods

Geometry optimizations and electronic structure calculations were performed using spin-polarized density functional theory (DFT) as implemented in Vienna *Ab initio* Simulation Package (VASP-5.3).⁴⁶ The projector augmented wave (PAW) pseudo-potential was used to describe the core electrons, and a 500 eV cut-off energy was utilized for the valence electrons. The revised Perdew–Burke–Ernzerhof (rPBE) generalized gradient approximation (GGA)⁴⁷ was chosen for the exchange correlation potential. A vacuum layer distance of 10 Å along *x* and *y* directions was used to eliminate interactions between neighbored ribbons. The *k*-mesh of $1 \times 1 \times 11$ and $1 \times 1 \times 23$ were adopted to sample the Brillouin zone (BZ) for geometry optimizations and band structure calculations. The optimization did not stop until the residual force on each atom was smaller than 0.002 eV Å⁻¹.

Electronic transport properties were examined by adopting the nonequilibrium Green's function method in combination with density functional theory (NEGF-DFT), as implemented in

SIESTA 4.2,⁴⁸ by choosing double- ζ plus polarization functions (DZP) basis set for valence electrons, norm-conserving pseudopotentials for core electrons. During the calculations, a 150 Ry cut-off energy was set for real space grids, an energy shift parameter of 0.01 Ry was used to determine the cut-off radii of atomic orbitals, and a *k*-mesh of $1 \times 1 \times 100$ was adopted for achieving the balance between calculated efficiency and accuracy. As an example, the constructed transport device model of the type 1–1 C₅N-AGNR with width $m = 2$ is given in Fig. S1.†

Results and discussion

The edge structures of C₅N-GNRs

Pristine GNRs have two types of edge structures.^{35,36} Armchair-edged GNRs (AGNRs) are nonmagnetic (NM) and can be classified into three families depending on their widths ($W = 3m$, $3m + 1$, and $3m + 2$), whereas zigzag-edged GNRs (ZGNRs) are spin-polarized with tempting antiferromagnetic (AFM) characteristics. Thus, both armchair- and zigzag-edged C₅N-GNRs (C₅N-AGNRs and C₅N-ZGNRs) with different widths are considered in this work.

We find that C₅N-ZGNRs have only one kind of edge structure, while C₅N-AGNRs exhibit three kinds of edge structures, which can be labelled as 1: C–C, 2: C–C', and 3: C–N, respectively, as shown in Fig. 1. Hence, depending on the edge structures, C₅N-AGNRs can be classified into six types named as 1–1, 1–2, 1–3, 2–2, 2–3, and 3–3, respectively.

Ultrahigh conductivity in sub-5 nm wide C₅N-AGNRs

We have calculated the zero-bias transmission $T(E)$ of C₅N-AGNRs at the NM state. The obtained $T(E)$ of type 1–1 C₅N-AGNRs is given in Fig. 2(a), and that of types 1–2, 1–3, 2–2, 2–3, and 3–3 are given in Fig. S3–S7,† respectively.

One can immediately see that there does not exist any transport gaps in each C₅N-AGNRs, suggesting that the AGNRs are uniformly transferred from semiconducting to metallic *via* ordered N-doping. The intensities of $T(E)$ are found to be integer plateaux, just as the AGNRs with fully nitrogenized edges.⁴ It should be noted that previous work has shown that $T(E)$ of the GNRs with disordered N-doping exhibit non-integral intensity

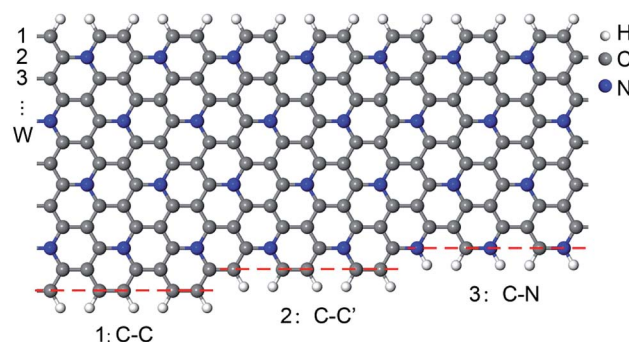


Fig. 1 Schematic presentation of the three kinds of edge structures of C₅N-AGNRs. (1) C–C, double C–C dimer lines, (2) C–C', single C–C dimer line, and (3) C–N, single C–N dimer line.



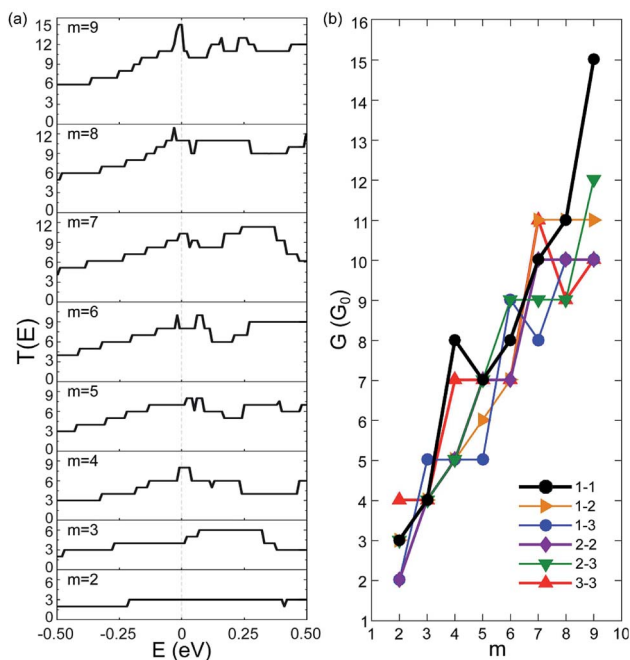


Fig. 2 Zero-bias transmission functions ($T(E)$) and conductivities (G). (a) The $T(E)$ of type 1–1 C_5N -AGNRs with different widths (W), which can be represented by integers m since $W = 3m + 2$, (b) the calculated conductivities of all six types of C_5N -AGNRs as a function of width.

due to the disordered N-dopants acting as scattering centres, which reduced the transmission possibilities.^{25,49} Hence, the ordered N-doping does not introduce any localized states to block transport channels around the E_F , or in other words the high carrier mobility of GNRs is not considerably impacted by the ordered N-doping. Moreover, one can see that the intensity of $T(E)$ near the E_F increases as the ribbon broadens, implying that wider ribbons present more conductive channels. Clearly, the orderly N-doped AGNRs exhibit much better transport capacities than the disorderly N-doped ones.

To illustrate the transport capacities, we plotted the zero-bias conductivity G of C_5N -AGNRs in Fig. 2(b). One can see that each C_5N -AGNR possesses an integral multiple of quantized conductivity. And as a whole, the conductivity exhibits a linear growing trend with the ribbon width. It reaches a very high value ($10G_0$) when $m = 9$. Especially for type 1–1, an ultrahigh conductivity of $15G_0$ was achieved. Such high conductivities have never been reported in other N-doped GNRs.^{4,25,49,50} Considering that type 1–1 C_5N -AGNRs belongs to the family of $W = 3m + 2$, the ribbon with $m = 9$ is only 3.58 nm in width, which is particularly suitable for nanoelectronics. Moreover, the linear growing trend implies that ultrahigh conductivities exist uniformly in all types of C_5N -AGNRs once their widths reach the sub-5 nm level.

To gain insight into the uniformly ultrahigh conductivity, we calculated the electronic structure of the six types of C_5N -AGNRs, and the result of type 1–1 is given in Fig. 3. It is evident that each ribbon is metallic with some well-delocalized partially-filled energy bands (PFEBS) crossing the E_F . Meanwhile, the typical band splitting around the

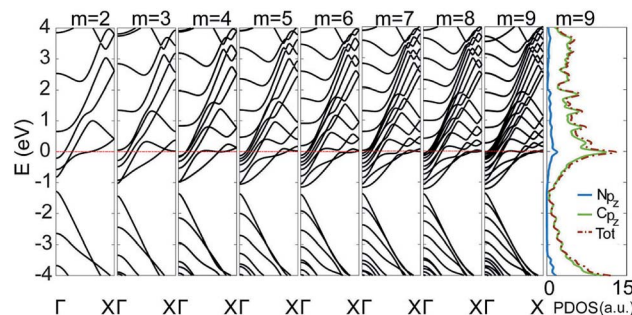


Fig. 3 The band structures of type 1–1 C_5N -AGNRs with different widths of $m = 2$ to 9, plus the partial density of states (PDOSs) of the ribbon with $m = 9$.

Dirac point of pristine AGNRs is retained, and the center of the split gap is located below the E_F by ~ 1 eV. It means that the E_F of pristine AGNRs has been shifted up into the conduction band (CB) region by the N-doping. The number of PFEBS increased with ribbon widening. For example, the number is 3 for $m = 2$, and it reaches 12 for $m = 9$. The 12 PFEBS run across the E_F 15 times in total, some of them are folded and cross the E_F more than once, which directly results in the ultrahigh conductivity of $15G_0$ as the carrier mobility is not affected by the N-doping.

Previous experiments have shown that N-dopants in graphene have an effective interaction force ranging up to 1.1 nm.³⁰ The distance of neighboring N-dopants in a C_5N -AGNR is only 4.26 Å, which ensures a strong interaction among N-dopants. This leads to the formation of well-delocalized PFEBS, just like the AGNRs with totally nitrogenized edges.⁴ Since a wider AGNR usually has more sub-bands in the conduction band (CB) region, a wider C_5N -AGNR provides a higher density of states near the Fermi level, *via* increasing the number of PFEBS and/or enhancing the folding of PFEBS. In general, the fact that ordered N-doping induces well-delocalized PFEBS and high density of states around the Fermi level is the source of the ultrahigh conductivity in sub-5 nm wide C_5N -AGNRs.

To further understand the origins of PFEBS, we have added the plot of partial density of states (PDOSs) for type 1–1 C_5N -AGNR with $m = 9$ in Fig. 3. A sharp DOS peak with an intensity of $13 \mu A$ presents right at the E_F . A previous study has shown an intensity of $3 \mu A$ in C_5N -graphene,³⁴ which is already much higher than that of graphene and GNRs. Of note, is that C_5N -graphene exhibits only 2 PFEBS, but the narrow C_5N -AGNR with $m = 2$ already has 3 PFEBS. This means that the high conductivity of C_5N -graphene is inherited and carried forward in multiples in C_5N -AGNRs, suggesting an advantage of C_5N -AGNRs for HCMs. Moreover, it shows that the N p_z orbitals distribute in a wide energy range, and the peak positions are consistent with that of the C p_z orbitals. This implies that the redundant electrons of N atoms are fully integrated into the ribbon *via* coupling with C p_z orbitals, resulting in the presence of multiple PFEBS in C_5N -AGNRs. Since a well-delocalized PFEBS can contribute a unit conductivity, the C_5N -AGNRs of multiple well-delocalized PFEBS naturally own high conductivities.



Ultrahigh conductivity in sub-5 nm wide C₅N-ZGNRs

Fig. 4(a) and (b) show the zero-bias conductivities and band structures of C₅N-ZGNRs with different widths (*W*), respectively. One can see that the conductivities of C₅N-ZGNRs also show a growing trend with width broadening, and an ultrahigh conductivity of 15*G*₀ is also achieved for the ribbon width *W* = 23 (about 4.90 nm in width). It is remarkable that there exists only 2 PFEBS in the ribbon with *W* = 6, 7, 8, and 9, just like in the C₅N-graphene, while the number of times the two PFEBS cross the *E*_F increases from 1 to 6 as the ribbon widens, due to the shift-down of the sub-bands near the *E*_F which leads to the enhanced folding feature of the PFEBS. Even in the wide C₅N-ZGNR with *W* = 23, only 6 PFEBS are present, but the folding feature makes them cross the *E*_F 15 times in total, giving rise to the high conductivity. Hence, it is safe to say that ultrahigh conductivity is also achievable in sub-5 nm wide C₅N-ZGNRs.

Linear relationship between ΔE_F and doping concentration

We have plotted ΔE_F ($\Delta E_F \equiv E_{F(C_5N-GNR)} - E_{F(GNR)}$) as the function of the doping concentration (*c*) for all considered GNRs in Fig. 5. From the inset, one can see that the type 1–3 and type 2–2 C₅N-AGNRs and the C₅N-ZGNRs possess a constant doping concentration of 16.7%, just as in the C₅N-graphene. While in the other four types of C₅N-AGNR, the doping concentration either increases or decreases exponentially with increasing width, and can ultimately converge to the value of 16.7%.

Overall, a simple relationship between ΔE_F and doping concentration (*c*) can be seen, which can be described by a linear fitting of $\Delta E_F = 9.23c - 0.24$. It shows that the doping concentration plays a decisive role in determining the ΔE_F . Significant deviations appear only for narrow ribbons,

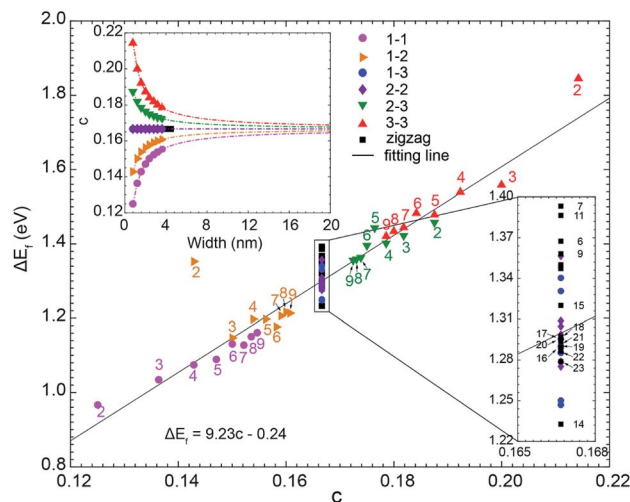


Fig. 5 The scatter plot of the shift of Fermi level (ΔE_F) as a function of doping concentration (*c*). Inset shows the relationship between doping concentration (*c*) and the ribbon width.

especially for the ribbons with *m* = 2, due to strong quantum confinement which only exists in narrow ribbons.

Conclusions

In conclusion, we have systematically studied the electronic structure and electron conductivity of orderly nitrogen-doped graphene nanoribbons (C₅N-GNRs) from first principles calculations. By carefully evaluating the edge type (armchair and zigzag), edge structure (single C–N, single C–C, and double C–C), ribbon width, doping concentration, and quantum confinement effect, a simple linear relationship between the shift-up of the Fermi level and the doping concentration is obtained. It is found that the ordered N-doping increases the carrier density by shifting the Fermi level to the CB region but does not considerably impact the high carrier mobility of the system due to the nonexistence of localized states near the Fermi level. Hence, ultrahigh conductivities (>15*G*₀) can be achieved in sub-5-nanometer-wide C₅N-GNRs, which calls for an increased effort in using ordered atomic doping for band structure engineering of carbon nanomaterials to utilize high conductivity materials at the nanoscale.

In light of recent developments, one can believe that the synthesis of our proposed C₅N-GNRs is feasible. As an example, Cui, *et al.*³⁴ has proposed a kinetic pathway toward C₅N-graphene on Cu(111) using C₅NCl₅ precursors. Chen *et al.*⁴³ have found that oriented GNRs can grow in hexagonal boron nitride (h-BN) trenches. Moreover, the GNR with ordered N-doping at the edges is achievable *via* bottom-up techniques. Once C₅N-graphene is realized in experiments, C₅N-GNRs can also be obtained from C₅N-graphene *via* modern lithographically patterned technologies, such as producing GNRs from graphene sheets.

Conflicts of interest

There are no conflicts to declare.

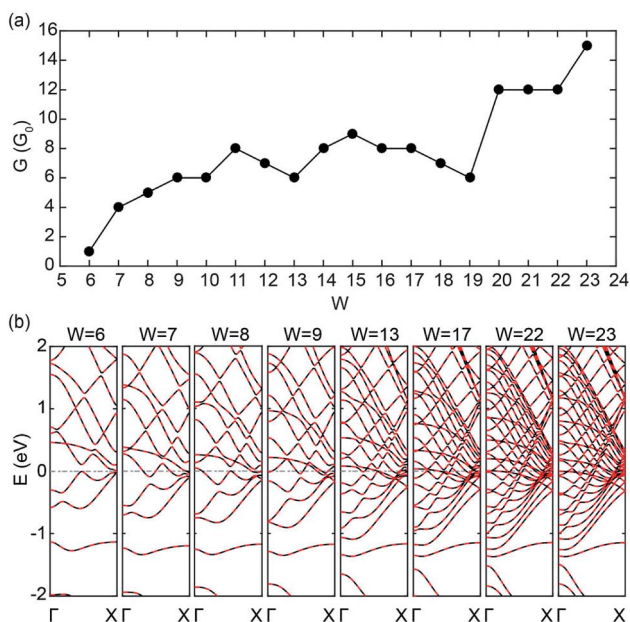


Fig. 4 (a) Zero-bias conductivities and (b) band structure of C₅N-ZGNRs with different widths (*W*) at the AFM ground state.



Acknowledgements

This work was supported by the National Key Research and Development Program of China (2017YFA0303500), National Natural Science Foundation of China (21643011, 61701175, 21633007, 21790350), Hunan Provincial Natural Science Foundation of China (Grant No. 27JJ3050), the Fundamental Research Foundations for the Central Universities, Sichuan Provincial Postdoctoral Science Foundation of China, and Anhui Initiative in Quantum Information Technologies.

Notes and references

- 1 L. Lu, Y. Shen, X. Chen, L. Qian and K. Lu, *Science*, 2004, **304**, 422–426.
- 2 Y. Liang, X. Liang, Z. Zhang, W. Li, X. Huo and L. Peng, *Nanoscale*, 2015, **7**, 10954–10962.
- 3 T. M. Higgins and J. N. Coleman, *ACS Appl. Mater. Interfaces*, 2015, **7**, 16495–16506.
- 4 L. Liu, X.-F. Li, Q. Yan, Q.-K. Li, X.-H. Zhang, M. Deng, Q. Qiu and Y. Luo, *Phys. Chem. Chem. Phys.*, 2017, **19**, 44–48.
- 5 M. Park, J. Im, M. Shin, Y. Min, J. Park, H. Cho, S. Park, M.-B. Shim, S. Jeon, D.-Y. Chung, J. Bae, J. Park, U. Jeong and K. Kim, *Nat. Nanotechnol.*, 2012, **7**, 803.
- 6 N. Matsuhisa, M. Kaltenbrunner, T. Yokota, H. Jinno, K. Kuribara, T. Sekitani and T. Someya, *Nat. Commun.*, 2015, **6**, 7461.
- 7 F. Hoeng, A. Denneulin and J. Bras, *Nanoscale*, 2016, **8**, 13131–13154.
- 8 H. Kim, D.-H. Seo, S.-W. Kim, J. Kim and K. Kang, *Carbon*, 2011, **49**, 326–332.
- 9 J.-H. Choi, H. Wang, S. J. Oh, T. Paik, P. Sung, J. Sung, X. Ye, T. Zhao, B. T. Diroll, C. B. Murray and C. R. Kagan, *Science*, 2016, **352**, 205–208.
- 10 S. Jung, S. J. Chun, J. T. Han, J. S. Woo, C.-H. Shon and G.-W. Lee, *Nanoscale*, 2016, **8**, 5343–5349.
- 11 C. Zhang, Z. Ni, J. Zhang, X. Yuan, Y. Liu, Y. Zou, Z. Liao, Y. Du, A. Narayan, H. Zhang, T. Gu, X. Zhu, L. Pi, S. Sanvito, X. Han, J. Zou, Y. Shi, X. Wan, S. Y. Savrasov and F. Xiu, *Nat. Mater.*, 2019, **18**, 482.
- 12 F. Xu and Y. Zhu, *Adv. Mater.*, 2012, **24**, 5117–5122.
- 13 X. Huang, P. Sheng, Z. Tu, F. Zhang, J. Wang, H. Geng, Y. Zou, C.-a. Di, Y. Yi, Y. Sun, W. Xu and D. Zhu, *Nat. Commun.*, 2015, **6**, 7408.
- 14 Z. Li, G. Ma, R. Ge, F. Qin, X. Dong, W. Meng, T. Liu, J. Tong, F. Jiang and Y. Zhou, *Angew. Chem., Int. Ed.*, 2016, **55**, 979–982.
- 15 Y. Ding, J. Zhang, L. Chang, X. Zhang, H. Liu and L. Jiang, *Adv. Mater.*, 2017, **29**, 1704253.
- 16 Z. Deng, Y. Guo, X. Zhao, P. X. Ma and B. Guo, *Chem. Mater.*, 2018, **30**, 1729–1742.
- 17 T. N. Tu, N. Q. Phan, T. T. Vu, H. L. Nguyen, K. E. Cordova and H. Furukawa, *J. Mater. Chem. A*, 2016, **4**, 3638–3641.
- 18 D. Sheberla, L. Sun, M. A. Blood-Forsythe, S. Er, C. R. Wade, C. K. Brozek, A. Aspuru-Guzik and M. Dincă, *J. Am. Chem. Soc.*, 2014, **136**, 8859–8862.
- 19 Y. Chang, A. D. Mohanty, S. B. Smedley, K. Abu-Hakme, Y. H. Lee, J. E. Morgan, M. A. Hickner, S. S. Jang, C. Y. Ryu and C. Bae, *Macromolecules*, 2015, **48**, 7117–7126.
- 20 G. Manthilake, N. Bolfan-Casanova, D. Novella, M. Mookherjee and D. Andraut, *Sci. Adv.*, 2016, **2**, e1501631.
- 21 A. K. Geim, *Science*, 2009, **324**, 1530–1534.
- 22 L. Zhang, F. Zhang, X. Yang, G. Long, Y. Wu, T. Zhang, K. Leng, Y. Huang, Y. Ma, A. Yu and Y. Chen, *Sci. Rep.*, 2013, **3**, 1408.
- 23 M. A. Worsley, P. J. Pauzauskie, T. Y. Olson, J. Biener, J. H. Satcher Jr and T. F. Baumann, *J. Am. Chem. Soc.*, 2010, **132**, 14067–14069.
- 24 M. Lin, Y. Gai, D. Xiao, H. Tan and Y. Zhao, *Chem. Phys. Lett.*, 2018, **713**, 98–104.
- 25 B. Biel, X. Blase, F. Triozon and S. Roche, *Phys. Rev. Lett.*, 2009, **102**, 096803.
- 26 X. Wang, G. Sun, P. Routh, D.-H. Kim, W. Huang and P. Chen, *Chem. Soc. Rev.*, 2014, **43**, 7067–7098.
- 27 W.-W. Yan, X.-F. Li, X.-H. Zhang, X. Cao and M. Deng, *J. Mater. Chem. C*, 2019, **7**, 6241–6245.
- 28 X. Wang, X. Li, L. Zhang, Y. Yoon, P. K. Weber, H. Wang, J. Guo and H. Dai, *Science*, 2009, **324**, 768–771.
- 29 Y. Wang, Y. Shao, D. W. Matson, J. Li and Y. Lin, *ACS Nano*, 2010, **4**, 1790–1798.
- 30 R. Lv, Q. Li, A. R. Botello-Mendez, T. Hayashi, B. Wang, A. Berkdemir, Q. Hao, A. L. Elias, R. Cruz-Silva, H. R. Gutierrez, Y. A. Kim, H. Muramatsu, J. Zhu, M. Endo, H. Terrones, J.-C. Charlier, M. Pan and M. Terrones, *Sci. Rep.*, 2012, **2**, 586.
- 31 X.-F. Li, Q.-K. Li, J. Cheng, L. Liu, Q. Yan, Y. Wu, X.-H. Zhang, Z.-Y. Wang, Q. Qiu and Y. Luo, *J. Am. Chem. Soc.*, 2016, **138**, 8706–8709.
- 32 Q.-K. Li, X.-F. Li, G. Zhang and J. Jiang, *J. Am. Chem. Soc.*, 2018, **140**, 15149–15152.
- 33 T. Schiros, D. Nordlund, L. Palova, D. Prezzi, L. Zhao, K. S. Kim, U. Wurstbauer, C. Gutierrez, D. Delongchamp and C. Jaye, *Nano Lett.*, 2012, **12**, 4025–4031.
- 34 P. Cui, J.-H. Choi, C. Zeng, Z. Li, J. Yang and Z. Zhang, *J. Am. Chem. Soc.*, 2017, **139**, 7196–7202.
- 35 M. Y. Han, B. Özyilmaz, Y. Zhang and P. Kim, *Phys. Rev. Lett.*, 2007, **98**, 206805.
- 36 Y.-W. Son, M. L. Cohen and S. G. Louie, *Phys. Rev. Lett.*, 2006, **97**, 216803.
- 37 X. Wang, Y. Ouyang, X. Li, H. Wang, J. Guo and H. Dai, *Phys. Rev. Lett.*, 2008, **100**, 206803.
- 38 J. Cai, C. A. Pignedoli, L. Talirz, P. Ruffieux, H. Söde, L. Liang, V. Meunier, R. Berger, R. Li, X. Feng, K. Mullen and R. Fasel, *Nat. Nanotechnol.*, 2014, **9**, 896–900.
- 39 N. Richter, Z. Chen, A. Tries, T. Precht, A. Narita, K. Müllen, K. Asadi, M. Bonn and M. Kläui, arXiv preprint arXiv:1806.00962, 2018.
- 40 P. Ruffieux, S. Wang, B. Yang, C. Sánchez-Sánchez, J. Liu, T. Dienel, L. Talirz, P. Shinde, C. A. Pignedoli, D. Passerone, T. Dumslaff, X. Feng, K. Mullen and R. Fasel, *Nature*, 2016, **531**, 489–492.
- 41 A. Narita, X. Feng, Y. Hernandez, S. A. Jensen, M. Bonn, H. Yang, I. A. Verzhbitskiy, C. Casiraghi, M. R. Hansen,



- A. H. Koch, G. Fytas, O. Ivasenko, B. Li, K. S. Mali, T. Balandina, S. Mahesh, S. D. Feyter and K. Mullen, *Nat. Chem.*, 2014, **6**, 126.
- 42 C. Ma, Z. Xiao, H. Zhang, L. Liang, J. Huang, W. Lu, B. G. Sumpter, K. Hong, J. Bernholc and A.-P. Li, *Nat. Commun.*, 2017, **8**, 14815.
- 43 L. Chen, L. He, H. S. Wang, H. Wang, S. Tang, C. Cong, H. Xie, L. Li, H. Xia, T. Li, *et al.*, *Nat. Commun.*, 2017, **8**, 14703.
- 44 Z. Chen, H. I. Wang, J. Teyssandier, K. S. Mali, T. Dumsloff, I. Ivanov, W. Zhang, P. Ruffieux, R. Fasel, H. J. Rader, D. Turchinovich, S. D. Feyter, X. Feng, M. Klau, A. Narita, M. Bonn and K. Mullen, *J. Am. Chem. Soc.*, 2017, **139**, 3635–3638.
- 45 I.-Y. Jeon, H.-J. Choi, M. J. Ju, I. T. Choi, K. Lim, J. Ko, H. K. Kim, J. C. Kim, J.-J. Lee, D. Shin, S.-M. Jung, J.-M. Seo, M.-J. Kim, N. Park, L. Dai and J.-B. Baek, *Sci. Rep.*, 2013, **3**, 2260.
- 46 G. Kresse and J. Furthmüller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 11169.
- 47 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 48 J. M. Soler, E. Artacho, J. D. Gale, A. García, J. Junquera, P. Ordejón and D. Sánchez-Portal, *J. Phys.: Condens. Matter*, 2002, **14**, 2745.
- 49 H. Ren, Q.-X. Li, Y. Luo and J. Yang, *Appl. Phys. Lett.*, 2009, **94**, 173110.
- 50 X.-F. Li, K.-Y. Lian, Q. Qiu and Y. Luo, *Nanoscale*, 2015, **7**, 4156–4162.

