




Cite this: *Nanoscale Adv.*, 2019, 1, 1497

A neoteric sandwich-configurational composite film offering synchronous conductive aeolotropy, superparamagnetism and dual-color fluorescence†

Liu Yang, Wensheng Yu, Jiao Tian, Xue Xi, Dan Li, Xiangting Dong, * Qianli Ma,*
Guixia Liu  and Jinxian Wang 

A new type of two-dimensional (2D) sandwich-configurational composite film offering electrically conductive aeolotropism, superparamagnetism and dual-color fluorescence was successfully fabricated *via* electrospinning. The composite film consists of a [polyaniline (PANI)/polymethylmethacrylate (PMMA)]/[Eu(BA)₃phen/PMMA] Janus nanobelt array aeolotropic conductive-fluorescent layer (first layer), a Fe₃O₄/polyvinylpyrrolidone (PVP) superparamagnetic nanofibers layer (second layer) and a Tb(BA)₃phen/polyacrylonitrile (PAN) fluorescent nanofiber layer (third layer), which have been tightly bonded together to form a sandwich-configurational composite film with trifunctionality. Because of the exceptional sandwich-like structure, electrically conductive, superparamagnetic, and fluorescent substances are mutually and efficaciously segregated. Thus, reciprocally pernicious interferences among them can be thoroughly avoided; thus, the sandwich-configurational composite film coinstantaneously possesses superior conductive aeolotropism, fluorescence and magnetism tri-functionality. Thus, the aeolotropic conductive-fluorescent layer and fluorescent layer respectively exhibit excellent red and green fluorescence properties. Further, the conductive aeolotropism and superparamagnetism of the composite film can be severally adjusted *via* regulating the contents of PANI and Fe₃O₄ NPs. Owing to the peculiar nanostructure made of Janus nanobelts in the aeolotropic conductive-fluorescent layer, the conduction ratio reaches 10⁸ times between conductive and insulating directions of the sandwich-configurational composite film. Under the excitation of 290 nm ultraviolet light, prominent red emission at 615 nm can be clearly observed in the aeolotropic conductive-fluorescent layer. Additionally, major green emission at 545 nm can be observed in the fluorescent layer under 314 nm light excitation. Furthermore, due to the exceptional sandwich structure, the properties of each layer of the whole film are relatively independent, the fluorescence intensities of the aeolotropic conductive-fluorescent layer and the fluorescent layer are hardly affected by the magnetic variation of the superparamagnetic layer, and the fluorescence intensity of the fluorescent layer is not influenced by the modulation of the PANI content of the aeolotropic conductive-fluorescent layer. The neoteric sandwich-configurational composite film with concurrent trifunctionality constructed by a facile method has potential applications in many fields. Overall, the academic design and manufacturing means will provide support for the design and construction of new-typed aeolotropic conductive films with multifunctionality.

Received 15th November 2018
Accepted 26th January 2019

DOI: 10.1039/c8na00351c

rsc.li/nanoscale-advances

Introduction

In recent years, aeolotropic conductive films (ACFs), owing to their environmental friendliness and high-density connection in the narrow space, have been widely used in many fields,^{1–3} such as packaging technologies for flat panel displays and high-

tech industry. To the best of our knowledge, the type I ACFs are conductive in the thickness orientation of the film and insulative along the surface.⁴ This type of film has been extensively applied in electronics. Type II ACFs exhibit distinct electrical conductivities in two perpendicular orientations of the surface.^{5,6} Recently, significant attention has been paid to the type II ACFs. Huang *et al.*⁷ prepared aeolotropic conductive polymer films *via* shear flow induction of multi-walled carbon nanotubes to assemble ordered parallel multi-walled carbon nanotubes perpendicular to shear flow. Ma *et al.*⁸ prepared a novel Janus nanobelt array film with strongly aeolotropic conductivity, in which the conductance ratio between two orientations reached up to 10⁸. Type II ACFs only exist in the

Key Laboratory of Applied Chemistry and Nanotechnology at Universities of Jilin Province, Changchun University of Science and Technology, Changchun 130022, China. E-mail: dongxiangting888@163.com; maqianlimail@163.com; Fax: +86 0431 85383815; Tel: +86 0431 85582575

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c8na00351c



exploratory stage in the laboratory prior to large-scale industrial production and utilization. In this case, it is important to design and manufacture new ACFs with superior performance.

The growing demand for multifunctional materials makes it hard for single-function materials to satisfy the demands of modern science and technology. Therefore, more and more researchers focus their attention on multifunctional composite materials.^{9–13} In recent years, magnetic-fluorescence bifunctional nanomaterials have been widely used in drug delivery and biomolecular detection.¹⁴ For example, magnetic-fluorescence bifunctional nanomaterials have important applications in the field of drug delivery, because loaded drugs can be transferred to specific location *via* magnetic navigation. Moreover, real-time location of a drug can be detected by measuring the fluorescence signals emitted by magnetic-fluorescence nanomaterials.¹⁵ Electrical-magnetic bifunctional nanomaterials can promote proliferation and orientation of pre-osteoblasts, which play an important role in bone repair and tissue regeneration. In addition, studies have shown that cell viability under electrical-magnetic stimulation is significantly higher than that of the cell stimulated only by electrical or magnetic stimulation.¹⁶ Consequently, nanomaterials with fluorescent-magnetic-electrical three functions will be more widely used than those with only one or two functions. Conductive polyaniline (PANI),^{17–21} rare earth (RE) fluorescent complexes^{22–25} and Fe₃O₄ magnetic nanoparticles^{26–31} are widely utilized to construct multifunctional materials due to their preminent performance.

Electrospinning has been identified as an outstanding technology to produce nanofibers from polymer solutions or melts.^{32–39} Both type I ACFs and type II ACFs have some limitations due to their monolayer structures. For example, it is difficult to achieve the fluorescence-conduction-superparamagnetism multifunctions without affecting each other. Moreover, they can emit only a single color when exhibiting fluorescence. In order to overcome these shortcomings, in this study we creatively design and construct a sandwich-configurational composite film to simultaneously acquire aeolotropic conduction, superparamagnetism and dual-color fluorescence multifunction. On the basis of type II ACFs, we adopt a layer-by-layer electrospinning technology to fabricate new electrically conductive aeolotropic-superparamagnetism-fluorescence trifunctional [PANI/poly-methylmethacrylate (PMMA)]/[Eu(BA)₃phen/PMMA] Janus nanobelts array/[Fe₃O₄/polyvinylpyrrolidone (PVP)]/[Tb(BA)₃phen/polyacrylonitrile (PAN)] sandwich-configurational composite film (denoted by JAEF/MF/LF sandwich-configurational composite film for short) to achieve the three independent functional partitions in only one sheet of two-dimensional (2D) film. Aeolotropic conductive films concurrently possess superparamagnetic properties and fluorescence, thus, realizing multifunctionality. In this neoteric tri-layered composite film, aeolotropic conductive-fluorescent layer is designed to be the first layer consisting of Janus nanobelts array film to achieve conductive aeolotropism and robust red fluorescence. Janus nanobelt is the basic building unit of the aeolotropic conductive-fluorescent layer with both sides made of PANI/PMMA (conductive side) and Eu(BA)₃-phen/PMMA (insulative-fluorescent side), respectively.

Superparamagnetic layer (second layer) is composed of nanofibers containing PVP and Fe₃O₄ NPs that exhibit superparamagnetic performance. Fluorescent layer (third layer) is made up of nanofibers comprising PAN and Tb(BA)₃phen to ensure splendid green fluorescence. The sandwich-configurational composite film can successfully help realize the effective isolation of PANI and Fe₃O₄ NPs from RE complexes to guarantee strong fluorescence. Therefore, it is anticipated that the neoteric composite film with excellent conductive aeolotropism, fluorescence and superparamagnetic properties will be obtained. In order to highlight the superiorities of JAEF/MF/LF sandwich-configurational composite film, we also prepared four kinds of contrastive samples. On account of its multifunctionality, the sandwich-configurational composite film can be potentially utilized in drug targeting, bone repair, tissue regeneration, electromagnetic interference shielding, microwave absorption, nanodevices construction, *etc.*^{40,41}

Experimental

Chemical reagents

Chemical reagents were described in the ESI section.†

Preparation of OA-modified Fe₃O₄ NPs, PMMA, Tb(BA)₃phen and Eu(BA)₃phen

The synthesis of OA-modified Fe₃O₄ NPs, PMMA, Tb(BA)₃phen and Eu(BA)₃phen was described previously.^{8,42}

Preparation of spinnable liquids

Preparation of spinnable liquids for JAEF/MF/LF sandwich-configurational composite film. Two different types of spinnable liquids were used to prepare the aeolotropic conductive-fluorescent layer (first layer). Spinnable liquid I for the conductive side of Janus nanobelts was prepared as follows. ANI, CSA and PMMA were dissolved in the mixed solvent of CHCl₃ (13.0000 g) and DMF (1.0000 g) under magnetic agitation (denoted by liquid I-a). APS was added to the mixed solvent of CHCl₃ (4.5000 g) and DMF (0.5000 g) (named liquid I-b). Liquid I-a and liquid I-b were placed into an icebox at 0 °C for 20 minutes. Then liquid I-b was added dropwise to liquid I-a in the ice-water bath and the solution was maintained for 3 h. At last, the final substance was kept at 0 °C for 24 h and PANI was obtained *via* polymerization of aniline. For fabricating the insulative-fluorescent side of Janus nanobelts, the spinnable liquid II was prepared as follows. PMMA (1.0000 g) and Eu(BA)₃phen were dissolved in the mixed solvent of CHCl₃ (17.5000 g) and DMF (1.5000 g) under magnetic stirring. In order to determine the optimal ratio of Eu(BA)₃phen to PMMA and obtain the strongest fluorescence intensity, different amounts of Eu(BA)₃phen were added to spinnable liquid II. Spinnable liquid III for making the superparamagnetic layer (second layer) was composed of Fe₃O₄ NPs, PVP and DMF. The as-prepared Fe₃O₄ NPs were dispersed in DMF (4.8000 g) under ultrasonics for 25 min and then PVP (1.0000 g) was added to the mixture. In order to determine the optimal ratio of Tb(BA)₃phen to PAN and acquire the strongest fluorescence intensity,



Table 1 Constituents of the spinnable liquid I

Liquid I	PANI:PMMA (wt%)	ANI (g)	CSA (g)	APS (g)	PMMA (g)	DMF (g)	CHCl ₃ (g)
L _{I-1}	30%	0.3000	0.7566	0.7636	1.0000	1.5000	17.5000
L _{I-2}	40%	0.4000	0.9685	0.9832	1.0000	1.5000	17.5000
L _{I-3}	50%	0.5000	1.2420	1.2245	1.0000	1.5000	17.5000
L _{I-4}	60%	0.6000	1.5321	1.4830	1.0000	1.5000	17.5000
L _{I-5}	70%	0.7000	1.7596	1.7153	1.0000	1.5000	17.5000

Table 2 Constituents of the spinnable liquid II

Liquid II	Eu(BA) ₃ phen:PMMA (wt%)	Eu(BA) ₃ phen (g)	PMMA (g)	DMF (g)	CHCl ₃ (g)
L _{II-1}	5%	0.0500	1.0000	1.5000	17.5000
L _{II-2}	10%	0.1000	1.0000	1.5000	17.5000
L _{II-3}	15%	0.1500	1.0000	1.5000	17.5000
L _{II-4}	20%	0.2000	1.0000	1.5000	17.5000
L _{II-5}	25%	0.2500	1.0000	1.5000	17.5000

different amounts of Tb(BA)₃phen were added to spinnable liquid IV. Spinnable liquid IV, used for fabricating the fluorescent layer (third layer) was prepared as follows: PAN (1.0000 g) and Tb(BA)₃phen were dissolved in DMF (10.0000 g) under magnetic agitation. The actual constituents of spinnable liquids I, II, III and IV are listed in Tables 1–4.

Preparation of spinnable liquids for contrastive samples.

To emphasize the advantages of JAEF/MF/LF sandwich-configurational composite film, four kinds of contrastive samples were fabricated, including [PANI/PMMA]/[Eu(BA)₃phen/PMMA] Janus nanobelt non-array/[Fe₃O₄/PVP]/[Tb(BA)₃phen/PAN] sandwich-configurational composite film (named JNEF/MF/LF sandwich-configurational composite film for short), [PANI/Eu(BA)₃phen/PMMA] composite nanobelt array/[Fe₃O₄/PVP]/[Tb(BA)₃phen/PAN] sandwich-configurational composite film (named CAEF/MF/LF sandwich-configurational composite film for short), [PANI/Eu(BA)₃phen/PMMA] composite nanobelt non-array/[Fe₃O₄/PVP]/[Tb(BA)₃phen/PAN] sandwich-configurational composite film (denoted by CNEF/MF/LF sandwich-configurational composite film for short) and [PANI/PMMA]/[Eu(BA)₃phen/PMMA] Janus nanobelt array/[Fe₃O₄/Tb(BA)₃phen/PVP/PAN] two-layer composite film (named JAEF/[M–L]F two-layer composite film for short). JNEF/MF/LF sandwich-configurational composite film was fabricated using spinnable liquids L_{I-3}, L_{II-3}, L_{III-2} and L_{IV-3}. The spinnable liquids for the composite nanobelts array layer of the CAEF/MF/LF sandwich-configurational composite film and composite nanobelts non-array layer of the CNEF/MF/LF sandwich-

configurational composite film were fabricated by blending spinnable liquids L_{I-3} and L_{II-3} at a volume ratio of 1 : 1 (named spinnable liquid V); the other two spinnable liquids for superparamagnetic layer and fluorescent layer were spinnable liquids L_{III-2} and L_{IV-3}. The spinnable liquid for superparamagnetic-fluorescent layer of JAEF/[M–L]F two-layer composite film was prepared by mixing equivoluminal spinnable liquids L_{III-2} and L_{IV-3} (named spinnable liquid VI), and spinnable liquids for Janus nanobelts array layer of JAEF/[M–L]F two-layer composite film were spinnable liquids L_{I-3}, L_{II-3}. Table S2† summarizes the constituents of spinnable liquids in detail.

Fabrication of JAEF/MF/LF sandwich-configurational composite film and contrastive samples via electrostatic-spinning

The electrostatic-spinning process and equipment diagrams are shown in Fig. 1. A peculiarly made parallel spinneret was used for fabricating [PANI/PMMA]/[Eu(BA)₃phen/PMMA] Janus nanobelts and aeotropic conductive-fluorescent layer. Two different types of spinnable liquids were firstly utilized to fabricate Janus nanobelts in the aeotropic conductive-fluorescent layer. Spinnable liquids I and II were added to two plastic syringes, respectively. During the electro-spinning process, 2.5 mL of spinnable liquid I and 2.5 mL of spinnable liquid II were consumed, and the two spinnable liquids formed

Table 3 Constituents of the spinnable liquid III

Liquid III	Fe ₃ O ₄ : PVP (mass ratio)		PVP (g)	DMF (g)
	Fe ₃ O ₄ (g)	PVP (g)		
L _{III-1}	0.5 : 1	0.5000	1.0000	4.8000
L _{III-2}	1 : 1	1.0000	1.0000	4.8000
L _{III-3}	2 : 1	2.0000	1.0000	4.8000

Table 4 Constituents of the spinnable liquid IV

Liquid IV	Tb(BA) ₃ phen: PAN (wt%)	Eu(BA) ₃ phen		DMF (g)
		(g)	PAN (g)	
L _{IV-1}	5%	0.0500	1.0000	10.0000
L _{IV-2}	10%	0.1000	1.0000	10.0000
L _{IV-3}	15%	0.1500	1.0000	10.0000
L _{IV-4}	20%	0.2000	1.0000	10.0000
L _{IV-5}	25%	0.2500	1.0000	10.0000



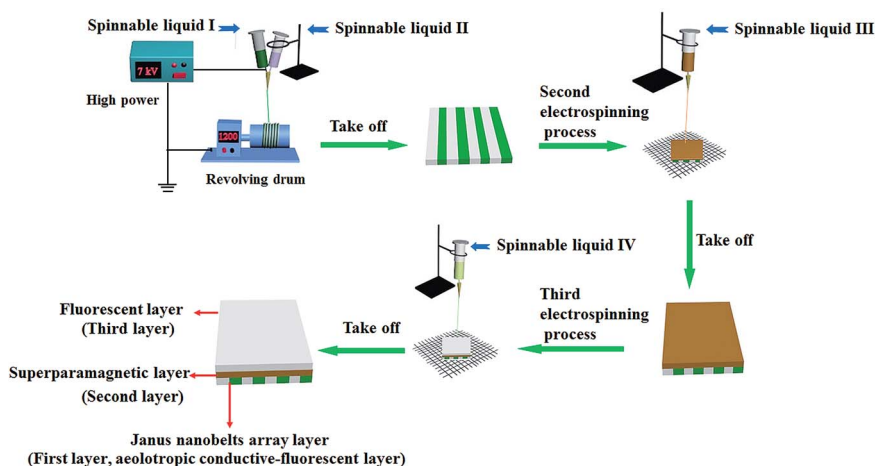


Fig. 1 Device diagram and electro-spinning process for preparation of the sandwich-configurational composite film.

a side-by-side structure at the tip of the nozzle. As a collector, a revolving drum was placed approximately 14 cm from the tip of the plastic nozzle to collect Janus nanobelts. The rotation speed was 1100 rpm and the spinning speed was 0.5 mL per hour. The electro-spinning nozzle and the revolving drum were respectively connected to the positive pole and grounded terminal of a high-voltage direct current power supply, and 7 kV of voltage was applied. Thus, the aeolotropic conductive layer can be obtained. Secondly, in order to obtain the sandwich-configurational composite film, the aeolotropic conductive layer was taken off from the revolving drum and then placed on the wire netting. The wire netting, as the collection device, was placed at the distance of 16 cm from the single spinneret and connected with the grounded terminal of the high voltage power supply. The voltage was kept at 14 kV. In the electro-spinning process, 5 mL of the spinnable liquid III was consumed to construct the superparamagnetic layer. Thirdly, spinnable liquid III was replaced by 5 mL of spinnable liquid IV to continue the electro-spinning process to form a fluorescent layer after the spinnable liquid III was thoroughly removed. At this time, JAEF/MF/LF sandwich-configurational composite film was successfully prepared.

The contrastive samples were fabricated to illustrate the preponderance of the structure of JAEF/MF/LF sandwich-configurational composite film. The electro-spinning process and schematic diagrams of contrastive samples are presented in Table S1.† Moreover, the constituents of spinnable liquids, spinnerets and electro-spinning conditions for fabricating the contrastive samples are also systematically summed up in Table S2.†

Characterization

The characterization methods are described in the ESI† section.

Results and discussion

Phase analysis

The phase compositions of the Fe_3O_4 NPs and JAEF/MF/LF sandwich-configurational composite film were studied by X-

ray diffractometry (Fig. 2). XRD patterns of Fe_3O_4 NPs are in conformity with the cubic-structured Fe_3O_4 (PDF 89-0691) and characteristic peaks of other impurities do not exist, which indicates that Fe_3O_4 NPs are single and pure phase. When the aeolotropic conductive-fluorescent layer and the fluorescent layer of JAEF/MF/LF sandwich-configurational composite film face the X-ray source, characteristic diffraction peaks of Fe_3O_4 can be observed also. However, the reflection peaks of Fe_3O_4 are weaker because Fe_3O_4 NPs merely disperse throughout the middle layer of JAEF/MF/LF sandwich-configurational composite film and low contents of Fe_3O_4 NPs in the samples. The broad diffraction peaks at *ca.* 22° of the samples are also found, indicating the existence of amorphous substances, such as PMMA, PVP and rare earth complexes.

Morphological and structural analysis

As demonstrated in Fig. 3a, the intersecting surface of JAEF/MF/LF sandwich-configurational composite film can be clearly seen in scanning electron microscope (SEM) images with low magnification. It can be seen that the composite film is made up

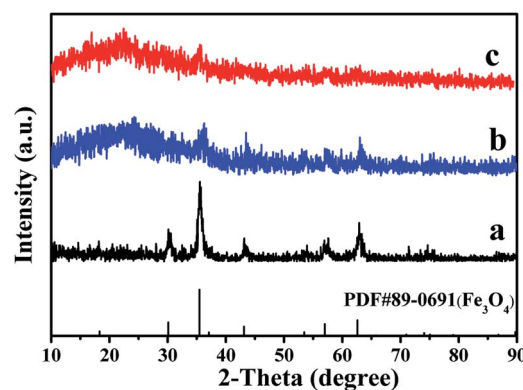


Fig. 2 XRD patterns of (a) Fe_3O_4 NPs, (b) the aeolotropic conductive-fluorescent layer and (c) the fluorescent layer of the JAEF/MF/LF sandwich-configurational composite film ($[L_{I-3}/L_{II-3}]/L_{III-2}/L_{IV-3}$).



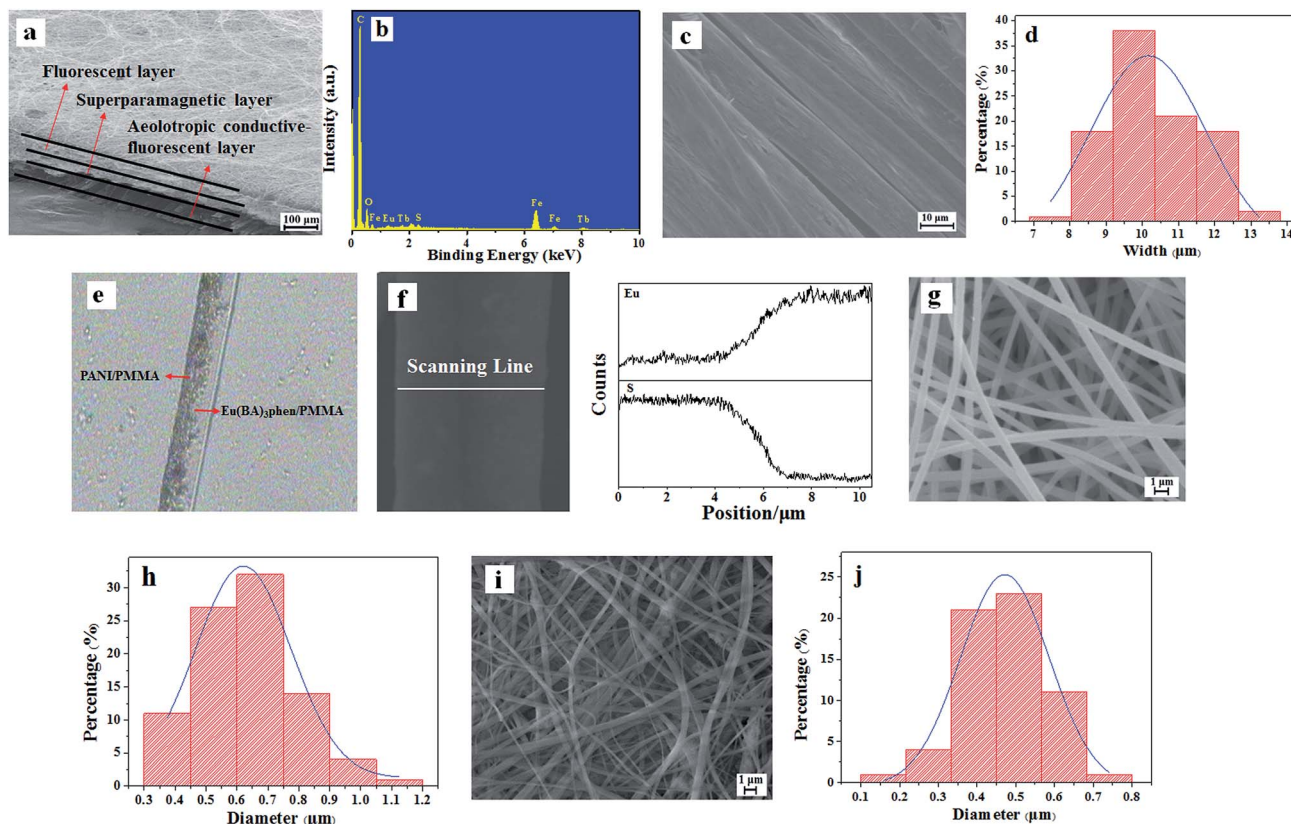


Fig. 3 SEM images of (a) intersecting surface, (c) aeolotropic conductive-fluorescent layer, (g) fluorescent layer and (i) superparamagnetic layer; (b) EDS spectrum of JAEF/MF/LF sandwich-configurational composite film; (d) histogram of width of Janus nanobelts in aeolotropic conductive-fluorescent layer; (e) OM image and (f) EDS line-scan analysis of single Janus nanobelt; (h) diameter distribution of nanofibers in fluorescent layer and (j) superparamagnetic layer.

of three layers. Fig. 3b depicts the energy dispersive spectrum (EDS). The JAEF/MF/LF sandwich-configurational composite film is made of elements Fe, S, Eu, Tb, C, and O. Fig. 3c shows the SEM image of the Janus nanobelts. It is clear that all the width directions of these nanobelts in the aeolotropic conductive layer arrange upwards, almost without exception, due to the fact that this structure helps the nanobelts lay in a state of low energy and more stability onto the revolving drum. The average width of Janus nanobelts is 10.25 μm (Fig. 3d). The internal structure of a Janus nanobelt in the aeolotropic conductive layer can be expressly seen using the optical microscope (OM). As indicated in Fig. 3e, an obvious Janus structure can be observed, and one side of the Janus nanobelt is PANI/PMMA, which is dark-colored. In contrast, the other side is transparent $\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA}$. In order to further examine the Janus structure of the nanobelt, EDS line-scan analysis was carried out (Fig. 3f), where the S and Eu elements delegate CSA doped PANI and $\text{Eu}(\text{BA})_3\text{phen}$, respectively. Elemental S only exists on one side, and elemental Eu merely appears in the other side of Janus nanobelt. Based on the analysis of the above results, we can surely conclude that aeolotropic conductive Janus nanobelts array film (namely the first layer) has been successfully prepared. SEM images (Fig. 3g and i) of the fluorescent layer and the superparamagnetic layer, which are composed of nanofibers, are also provided, respectively. The histograms of the

diameter distribution of the nanofibers in fluorescent layer and superparamagnetic layer are severally demonstrated in Fig. 3h and j, and their average diameters are 630 nm and 480 nm, respectively.

The JAEF/[M-L]F two-layer composite film is also presented. The intersecting surface of the two-layer composite film is shown in Fig. S1a.† Fig. S1b† presents the SEM image of the superparamagnetic-fluorescent layer in the JAEF/[M-L]F two-layer composite film. The average diameter of the nanofibers in superparamagnetic-fluorescent layer is 390 nm (Fig. S1c†). Meanwhile, the first layers of JNEF/MF/LF, CNEF/MF/LF and CAEF/MF/LF sandwich-configurational composite film were examined also. The SEM images (Fig. S1d–f†) demonstrate the arrangement of nanobelts in these contrastive samples. The nanobelts in the JNEF/MF/LF and CNEF/MF/LF sandwich-configurational composite films are out of order (Fig. S1d and e†), but nanobelts in the CAEF/MF/LF sandwich-configurational composite film are ordered (Fig. S1f†).

Physical photos of the sandwich-configurational composite films are shown in Fig. 4. It is obvious that the aeolotropic conductive-fluorescent layer of the sandwich-configurational composite film is blackish green (Fig. 4b) owing to the existence of PANI in the aeolotropic conductive-fluorescent layer, and the fluorescent layer is white (Fig. 4d). It was found that the product (Fig. 4a) has a plain and neoteric three-layered



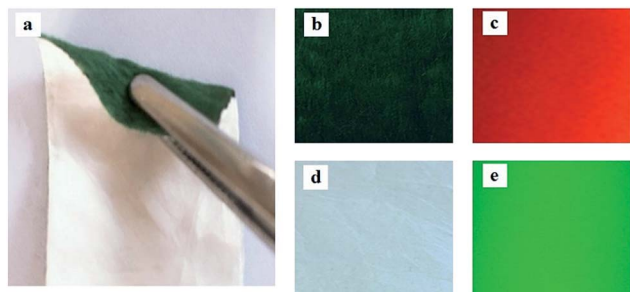


Fig. 4 Images: (a) a folded film, (b) the aeolotropic conductive-fluorescent layer, (c) red fluorescence emission under 290 nm excitation of the aeolotropic conductive-fluorescent layer in darkness (d) the fluorescent layer, and (e) green fluorescence emission under 314 nm excitation of the fluorescent layer in darkness.

structure: different layers show different colors due to the presence of different components. It can be seen in Fig. 4c and e that aeolotropic conductive-fluorescent layer and the fluorescent layer emit red fluorescence under 290 nm UV irradiation and green fluorescence under 314 nm UV illumination in darkness, respectively.

Fluorescence performance

The fluorescence properties of sandwich-configuration composite film were further explored *via* modulating the contents of PANI and Fe_3O_4 NPs. Fig. 5 demonstrates the fluorescence spectra of the aeolotropic conductive-fluorescent layer (first layer) of the sandwich-configuration composite film. As shown in Fig. S2,[†] the optimum excitation (abbreviated as EX) wavelength is 290 nm. It can be found that the emission (abbreviated as EM) intensities of the aeolotropic conductive-fluorescent layer of the film decrease with the increase of PANI content. This phenomenon is caused by the light absorption of dark-colored PANI. The $\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA}$ sides of the Janus nanobelts absorb two kinds of EX light. The first is direct ultraviolet radiation on the $\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA}$ sides, and the second is ultraviolet light reflected or refracted on the PANI/PMMA sides. It is easy to infer that the former plays a major role in stimulating $\text{Eu}(\text{BA})_3\text{phen}$ complexes in the $\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA}$ sides while the latter secondarily contributes to the EX process. As for the latter, since PANI absorbs ultraviolet light (<400 nm), as demonstrated by the UV-Vis

absorption spectrum of PANI in Fig. 5c, more PANI leads to an increase in light absorption and weak reflection or refraction of ultraviolet light to $\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA}$ sides. Therefore, when more PANI is added, the EX light and EM light are partly weakened, resulting in the reduced fluorescence intensity of the aeolotropic conductive-fluorescent layer. On the contrary, the superparamagnetic layer (namely the second layer) has no effect on the fluorescence intensity of the aeolotropic conductive-fluorescent layer (Fig. 5b). This result is caused by the unique structure of the sandwich-configuration composite film. Because the three layers in the sandwich structure are independent of each other and it is difficult for the EX light to pass from the aeolotropic conductive-fluorescent layer to the superparamagnetic layer of the film, the dark-colored Fe_3O_4 NPs in the superparamagnetic layer cannot absorb the EX light and EM light, leading to no change in fluorescence intensity when the amount of Fe_3O_4 NPs increases. This result reveals the structural superiority of the sandwich-configuration composite film.

As shown in Fig. S3,[†] the optimum EX wavelength is 314 nm for the fluorescent layer of the sandwich-configuration composite film. The fluorescence spectra of the fluorescent layer (third layer) of the sandwich-configuration composite film are displayed in Fig. 6. It is satisfactory to find that the fluorescence intensity of the sandwich-configuration composite film remained almost constant when the contents of PANI and Fe_3O_4 NPs were changed. The aeolotropic conductive-fluorescent layer and superparamagnetic layer have little effect on fluorescence properties of the fluorescent layer in the sandwich-configuration composite film, despite the changed contents of PANI in the aeolotropic conductive-fluorescent layer and the presence of Fe_3O_4 NPs in the superparamagnetic layer. The new findings result from the fact that the new-typed sandwich-configuration composite film high effectively separates the fluorescent layer from the aeolotropic conductive-fluorescent layer and the superparamagnetic layer, so that the two layers have almost no detrimental influence on the fluorescent layer.

The fluorescence properties of the first layer of JAEF/MF/LF sandwich-configuration composite film were also compared with those of the first layer of JNEF/MF/LF, CAEF/MF/LF and CNEF/MF/LF sandwich-configuration composite film, as displayed in Fig. 7. Compared with JAEF/MF/LF sandwich-configuration composite film, the EX and EM intensities of

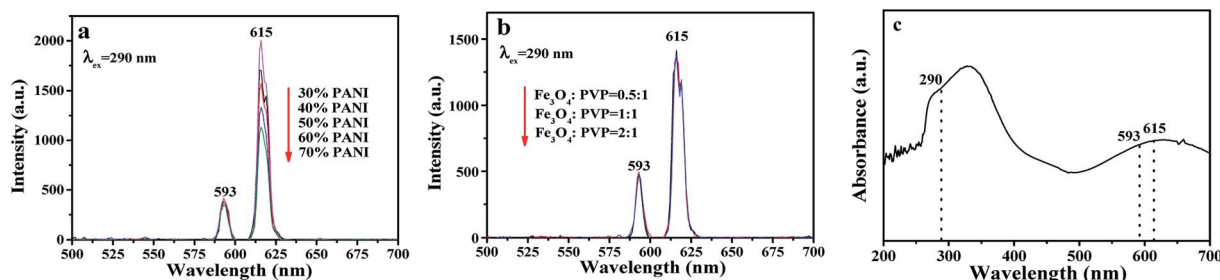


Fig. 5 (a and b) EM spectra of the aeolotropic conductive-fluorescent layer (first layer) doped with different contents of PANI (a) and the superparamagnetic layer (second layer) doped with different ratios of Fe_3O_4 NPs (b); (c) UV-Vis absorbance spectrum of PANI.



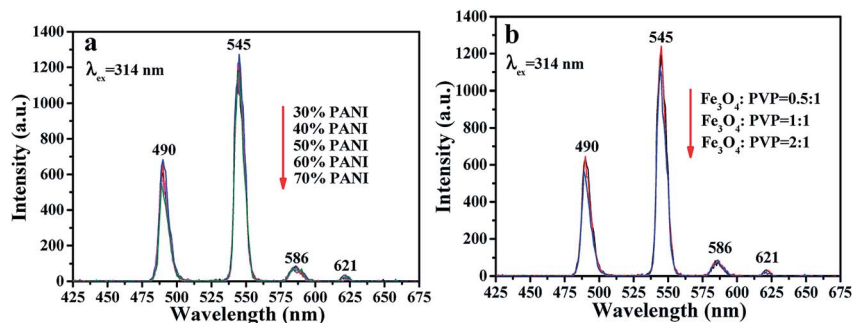


Fig. 6 (a and b) EM spectra of the fluorescent layer doped with different ratios of PANI in the aeolotropic conductive-fluorescent layer (a) and doped with different ratios of Fe_3O_4 NPs in the superparamagnetic layer (b).

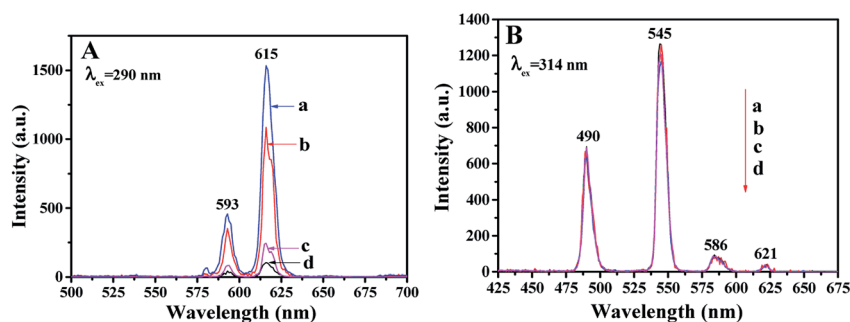


Fig. 7 EM spectra of (A) the first layer and (B) the fluorescent layer of (a) JAEF/MF/LF, (b) JNEF/MF/LF, (c) CAEF/MF/LF and (d) CNEF/MF/LF sandwich-configurational composite film.

the JNEF/MF/LF sandwich-configurational composite film slightly decrease, while the fluorescent intensities of CAEF/MF/LF and CNEF/MF/LF sandwich-configurational composite films are very weak under the same compositions. Fig. 8 is the diagram showing the comparison of fluorescence characteristics of the four samples. As shown in Fig. 8, since the first layer of JAEF/MF/LF sandwich-configurational composite film is composed of highly aligned Janus nanobelts, the surface of the film is densified. The EX light is difficult to pass from the upper nanobelts to the lower ones, so the EM light generated is mainly emitted from the upper Janus nanobelts without reduction. As

for the first layer of JNEF/MF/LF sandwich-configurational composite film demonstrated in Fig. 8, the Janus nanobelts are disordered, resulting in a loose surface of the film. In such case, part of the EX light will reach the lower Janus nanobelts through the voids of the loose upper Janus nanobelts, leading to the consequence that this part of EX light is weakened by the absorption of the upper Janus nanobelts. In the same way, the EM light is also absorbed when passing through the voids in the upper Janus nanobelts. As a result, the film has lower fluorescent intensity than JAEF/MF/LF sandwich-configurational composite film. For the same reason as stated above, the first

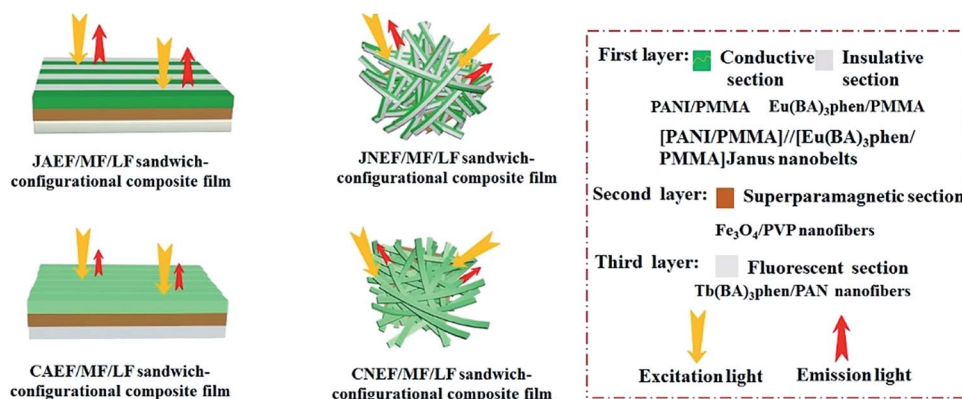


Fig. 8 Schematic of the EX light and EM light in JAEF/MF/LF, JNEF/MF/LF, CAEF/MF/LF and CNEF/MF/LF sandwich-configurational composite film.



layer of CAEF/MF/LF sandwich-configurational composite film has stronger fluorescent intensity than that of the first layer of CNEF/MF/LF sandwich-configurational composite film. As observed in Fig. 7A, the fluorescent intensity of the first layer of JAEF/MF/LF sandwich-configurational composite film and JNEF/MF/LF sandwich-configurational composite film are significantly higher than those of CAEF/MF/LF sandwich-configurational composite film and CNEF/MF/LF sandwich-configurational composite film. As for JAEF/MF/LF sandwich-configurational composite film and JNEF/MF/LF sandwich-configurational composite film, the constituent units are Janus nanobelts that effectively isolate $\text{Eu}(\text{BA})_3\text{phen}$ complexes from PANI, resulting in greatly reduced absorption of EX light and EM light by PANI. Composite nanobelts are building units of CAEF/MF/LF sandwich-configurational composite film and CNEF/MF/LF sandwich-configurational composite film, in which the uniform distribution of the $\text{Eu}(\text{BA})_3\text{phen}$ and PANI leads to the consequence that the dark-colored PANI strongly absorbs and weakens the EX light and the EM light. From the above analysis, it is found that JAEF/MF/LF sandwich-configurational composite film has the highest fluorescent intensity compared with the three contrastive samples. Hence, we can reasonably conclude that the structures of nanobelts and films significantly affect the fluorescent intensity, and Janus nanobelt array composite film has better performance than other contrastive samples. We also explored the EM spectra of the fluorescent layer of the JAEF/MF/LF, JNEF/MF/LF, CAEF/MF/LF and CNEF/MF/LF sandwich-configurational composite film. It can be seen from Fig. 7B that the fluorescent property of the fluorescent layer is not affected by the change of the first layer. It can be concluded that three layers in the sandwich structure are independent of each other, and the EX light is hard to pass through the fluorescent layer to the superparamagnetic layer and first layer of the film, leading to the result that the fluorescent layer is not affected by other layers.

Besides, to further demonstrate the advantages of the unique structure of the JAEF/MF/LF sandwich-configurational composite film, fluorescence studies of the JAEF/MF/LF sandwich-configurational composite film and JAEF/[M-L]F two-layer composite film were performed. Compared with the fluorescent layer of the JAEF/MF/LF sandwich-configurational composite film, JAEF/[M-L]F two-layer composite film (superparamagnetic-fluorescent layer) exhibits extremely weak emissions of Tb^{3+} . Fig. 9 shows the emission spectra of the JAEF/MF/LF sandwich-configurational composite film (fluorescent layer) and JAEF/[M-L]F two-layer composite film (superparamagnetic-fluorescent layer). In order to explain these results, in Fig. 10 we present the schematic diagrams of the fluorescent properties of JAEF/MF/LF sandwich-configurational composite film and JAEF/[M-L]F two-layer composite film. $\text{Tb}(\text{BA})_3\text{phen}$ of the JAEF/[M-L]F two-layer composite film directly contacts the dark-colored Fe_3O_4 NPs which strongly absorb EX and EM light, resulting in a serious decrease in fluorescent intensity. On the other hand, many researches have shown that elemental Fe can cause fluorescence quenching of RE fluorescent compounds, which also leads to a decrease of fluorescence intensity.⁴³ On the contrary, the fluorescent layer is

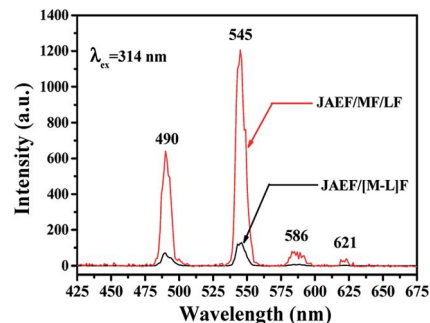


Fig. 9 Emission spectra of the fluorescent layer in JAEF/MF/LF sandwich-configurational composite film and the superparamagnetic-fluorescent layer in JAEF/[M-L]F two-layer composite film.

separated from the superparamagnetic layer and the aeolotropic conductive-fluorescent layer in the JAEF/MF/LF sandwich-configurational composite film. The fluorescent intensity of the fluorescent layer is almost unaffected by the other two layers. It can be seen from the above analysis that the JAEF/MF/LF sandwich-configurational composite film possesses better fluorescence properties than the JAEF/[M-L]F two-layer composite film, strongly implying that the sandwich-configurational composite film is better than the two-layer composite film for constructing peculiar materials with multi-functionalities.

Electrical conduction analysis

The electrical properties of the layers containing PANI of the JAEF/MF/LF sandwich-configurational composite film and four contrastive samples were researched. For all specimens, the

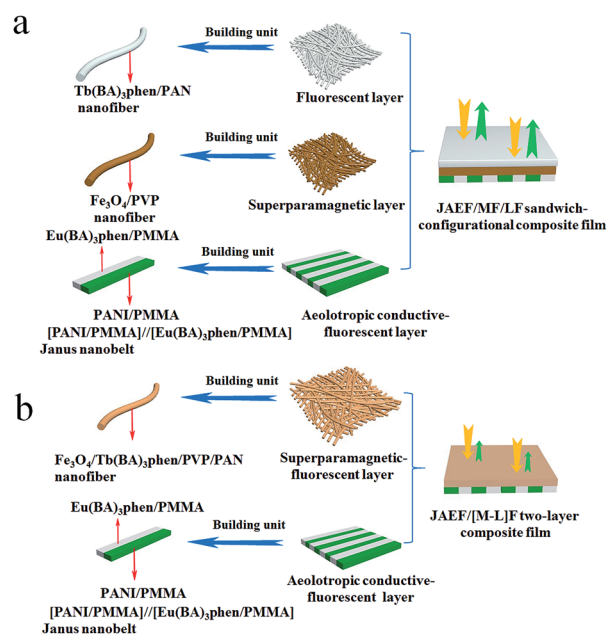


Fig. 10 Schematic of the EX light and EM light of (a) the fluorescent layer in JAEF/MF/LF sandwich-configurational composite film and (b) the superparamagnetic-fluorescent layer in JAEF/[M-L]F two-layer composite film.



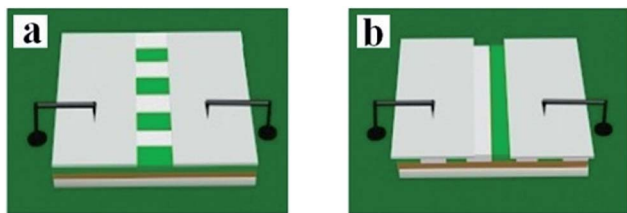


Fig. 11 Schematic for the conductance test of samples: (a) parallel orientation test and (b) perpendicular orientation test of the nanobelts in the JAEF/MF/LF sandwich-configurational composite film.

mass percentages of $\text{Eu}(\text{BA})_3\text{phen}$ to PMMA, $\text{Tb}(\text{BA})_3\text{phen}$ to PAN and the mass ratio of Fe_3O_4 to PVP were fixed at 15%, 15% and 1 : 1, respectively. Moreover, the mass percentages of PANI to PMMA were varied from 30% to 70%. In the two orthogonal orientations on the horizontal surface of the film, the length orientation (conductive orientation) and width orientation (insulative orientation) of the nanobelts were labeled as A and B, respectively. The test methods for the films are shown in Fig. 11. Two pieces of soldering tin ($1 \times 0.45 \text{ cm}^2$) as electrodes were attached to the sample (cut into $1 \text{ cm} \times 1 \text{ cm}$) *via* conductive adhesive. The distance between two pieces of soldering tin was 0.1 cm. The electrical conductances of JAEF/MF/LF sandwich-configurational composite film (aeolotropic conductive-fluorescent layer) and four contrastive samples are listed in Table 5. It is reported that the charge transport capacity provided by connective conductive network is a key factor for the conduction of PANI.⁴⁴ For the JAEF/MF/LF sandwich-configurational composite film, it can be seen from Table 5

that the higher is the content of PANI in the Janus nanobelts in aeolotropic conductive-fluorescent layer, the higher electrical conductance of A orientation can be obtained, while the conductance changes within B orientation is very small. Accordingly, by regulating the ratio of PANI to PMMA, the conductance of the product can be adjusted so that the aeolotropic ability also can be tailored. The diagrammatic sketch of the conductive mechanism of JAEF/MF/LF sandwich-configurational composite film with various amounts of PANI in the Janus nanobelts array layer (aeolotropic conductive-fluorescent layer) is displayed in Fig. 12. Electrons can move unhindered along the length orientation of the Janus nanobelts when current is generated in the film. As the PANI content increases, the increase in conductance is due to the formation of a more continuous PANI network structure which results in a more effective charge transport. Moreover, with an increase of PANI content, the green color of the conductive side of Janus nanobelts becomes deeper. The conductance along with the width orientation still retains *ca.* -10 orders of magnitude, regardless of how much the PANI content increases. As seen in Table 5, for the JAEF/MF/LF sandwich-configurational composite film, with the change of Fe_3O_4 NPs content in the superparamagnetic layer, the conductance of the film does not change significantly. Fig. 13 displays the diagrammatic sketch of the conduction mechanism of JAEF/MF/LF sandwich-configurational composite film with various ratios of Fe_3O_4 . Owing to the same content of PANI and the same structure in the aeolotropic conductive-fluorescent layer, they have similar conductance and the same color in the aeolotropic conductive-fluorescent layer of the film. For the fluorescent layer, the conductance value cannot be acquired (lower than the

Table 5 Conductance of the JAEF/MF/LF sandwich-configurational composite film with different amounts of PANI and various ratios of Fe_3O_4 to PVP and the four contrastive samples

Samples	Conductance (S)		A/B ratios	Aeolotropism degree
	A	B		
JAEF/MF/LF sandwich-configurational composite film [L_{I-1}/L_{II-3}]/ L_{III-2}/L_{IV-3} (30% PANI)	7.82×10^{-6}	2.63×10^{-10}	2.97×10^4	Medium
JAEF/MF/LF sandwich-configurational composite film [L_{I-2}/S_{II-3}]/ L_{III-2}/L_{IV-3} (40% PANI)	6.43×10^{-5}	2.58×10^{-10}	2.49×10^5	Strong
JAEF/MF/LF sandwich-configurational composite film [L_{I-3}/L_{II-3}]/ L_{III-2}/L_{IV-3} (50% PANI)	6.85×10^{-4}	3.24×10^{-10}	2.11×10^6	Strong
JAEF/MF/LF sandwich-configurational composite film [L_{I-4}/L_{II-3}]/ L_{III-2}/S_{IV-3} (60% PANI)	8.62×10^{-3}	2.75×10^{-10}	3.13×10^7	Very strong
JAEF/MF/LF sandwich-configurational composite film [L_{I-5}/L_{II-3}]/ L_{III-2}/S_{IV-3} (70% PANI)	7.62×10^{-2}	3.36×10^{-10}	2.27×10^8	Very strong
JAEF/MF/LF sandwich-configurational composite film [L_{I-3}/L_{II-3}]/ L_{III-1}/L_{IV-3} (Fe_3O_4 : PVP = 0.5 : 1)	6.34×10^{-4}	2.89×10^{-10}	2.19×10^6	Strong
JAEF/MF/LF sandwich-configurational composite film [L_{I-3}/L_{II-3}]/ L_{III-3}/S_{IV-3} (Fe_3O_4 : PVP = 2 : 1)	5.89×10^{-4}	2.64×10^{-10}	2.23×10^6	Strong
JNEF/MF/LF sandwich-configurational composite film ($[L_{I-3}/L_{II-3}]/L_{III-2}/L_{IV-3}$)	3.05×10^{-4}	2.96×10^{-4}	1.03	None
CAEF/MF/LF sandwich-configurational composite film ($L_{VI}/L_{III-2}/L_{IV-3}$)	2.58×10^{-7}	8.31×10^{-10}	310	Weak
CNEF/MF/LF sandwich-configurational composite film ($L_{VI}/L_{III-2}/L_{IV-3}$)	2.68×10^{-7}	2.92×10^{-7}	0.92	None
JAEF/[M-L]F two-layer composite film ($[L_{I-3}/L_{II-3}]/L_{VI}$)	5.73×10^{-4}	3.05×10^{-10}	1.88×10^6	Strong



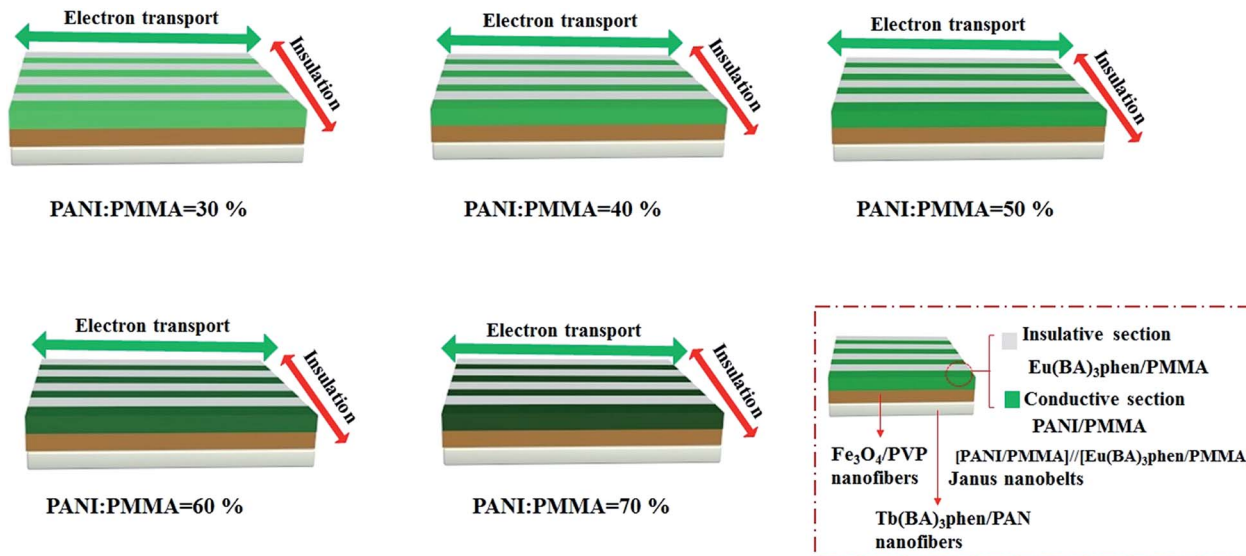


Fig. 12 Schematic of conductive mechanism and color variation with different amounts of PANI in the aeolotropic conductive-fluorescent layer of the JAEF/MF/LF sandwich-configurational composite film.

detection limit), which means that the fluorescent layer is insulated. Hence, the peculiarity of each layer is mutually independent, which is ascribed to the exceptional sandwich structure of the film.

It is also shown in Table 5 that the JAEF/MF/LF sandwich-configurational composite film and the JAEF/[M-L]F two-layer composite film possess almost the same electrical properties, but the conductance of their conductive orientation is higher than that of JNEF/MF/LF, CAEF/MF/LF and CNEF/MF/LF sandwich-configurational composite film with the same ingredients. The conduction mechanism diagrams of the JAEF/MF/LF sandwich-configurational composite film and four contrastive samples are shown in Fig. 14. In the Janus nanobelts array layer (first layer) of the JAEF/MF/LF sandwich-configurational composite film, the Janus nanobelts are arrayed in the same orientation, causing the flow orientation of current to stay fixed. On the contrary, because Janus nanobelts in the JNEF/MF/LF sandwich-configurational composite film are not ordered, the orientation of current flow is disordered. Under these conditions, all orientations of the film possess similar electrical conduction, resulting in a decrease in conductance in a specific orientation. CAEF/MF/LF and the CNEF/MF/LF sandwich-

configurational composite films exhibit observably lower conductance owing to the presence of an insulating material, $(\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA})$, in the PANI/ $(\text{Eu}(\text{BA})_3\text{phen}/\text{PMMA})$ composite nanobelts, leading to blocking the construction of a continuous conductive network. The JAEF/[M-L]F two-layer composite film has almost the same electrical properties as the JAEF/MF/LF sandwich-configurational composite film because insulating materials such as $\text{Tb}(\text{BA})_3\text{phen}$, Fe_3O_4 , PVP and PAN are dispersed in the superparamagnetic-fluorescent layer. The conductance ratios (namely A/B values) between the two orientations for different samples are also listed in Table 5 in detail. The conductance ratios between A and B are almost the same when the PANI content is fixed. When the content of Fe_3O_4 varies, the conductance does not change significantly. Nevertheless, with the increase of PANI contents, the A/B ratios markedly increase and the level of conductive aeolotropism also increases. In other words, tunable conductive aeolotropism can be realized. Most important of all, the A/B ratios reach up to *ca.* 10^8 times when the mass percentage of PANI is settled at 70%. As far as we know, this is the highest reported value for the nanobelts array films.

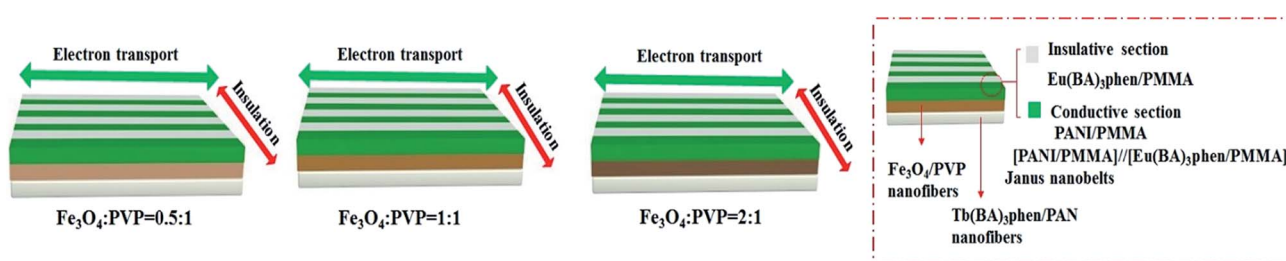


Fig. 13 Schematic of conductive mechanism with various ratios of Fe_3O_4 to PVP in the superparamagnetic layer of the JAEF/MF/LF sandwich-configurational composite film.



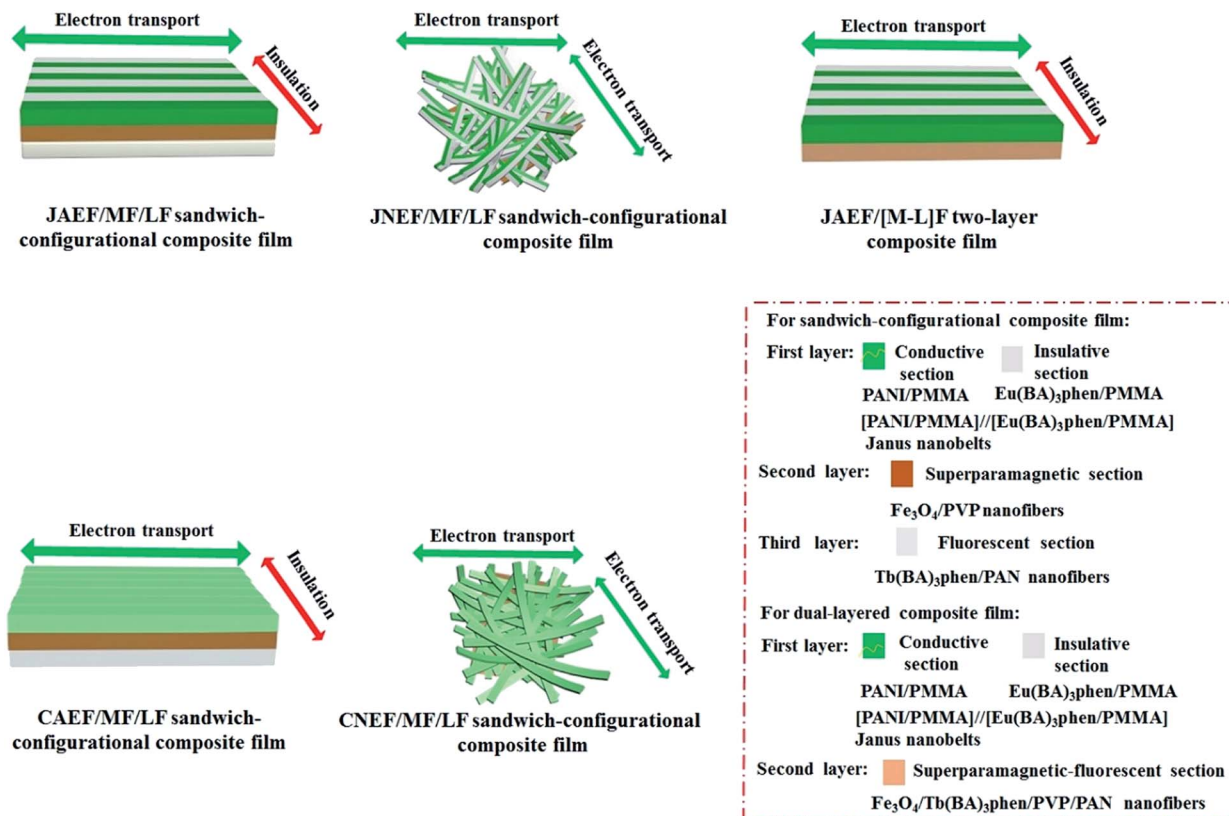


Fig. 14 Conductive schematic diagrams of the JAEF/MF/LF sandwich-configurational composite film along with four contrastive samples.

In order to find the outstanding aeolotropic conduction of the JAEF/MF/LF sandwich-configurational composite film, conductance values of contrastive samples were also measured, using the same methods (depicted in Fig. 11). The current direction of the JAEF/MF/LF sandwich-configurational composite film is settled such that the direction parallel to the length of the nanobelts is conductive and the vertical orientation is insulative; therefore, the film exhibits high conductance ratio. That is to say, this type of film displays strong aeolotropism ($[L_{I-3}/L_{II-3}]/L_{III-2}/L_{IV-3}$, 50% PANI, $A/B \approx 10^6$). The JNEF/MF/LF sandwich-configurational composite film exhibits no electrically conductive aeolotropism, as seen in

Table 5, because the film has electrical conduction in all orientations. The conductance ratio of A/B (two perpendicular orientations) is *ca.* 1. Although the CAEF/MF/LF sandwich-configurational composite film possesses aeolotropism, the aeolotropism is relatively weak ($A/B \approx 310$). The interfaces between the nanobelts can serve as the insulative media, but the insulation is not strong enough, leading to weak aeolotropism. For the CNEF/MF/LF sandwich-configurational composite film, both high conductance and aeolotropism ($A/B \approx 0.92$) are not acquired. The above results fully demonstrate that the JAEF/MF/LF sandwich-configurational composite film possesses the highest degree of aeolotropism compared with the three sandwich-configurational contrastive samples.

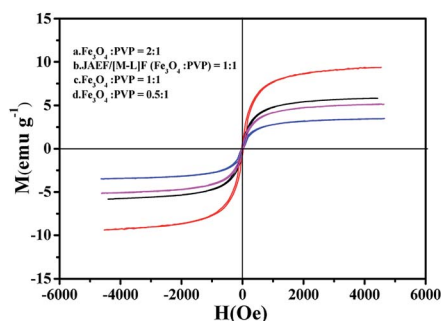


Fig. 15 Hysteresis loops of JAEF/MF/LF sandwich-configurational composite film of different Fe₃O₄ NPs mass ratios and JAEF/[M-L]F two-layer composite film.

Table 6 Saturation magnetization of JAEF/MF/LF sandwich-configurational composite film of different Fe₃O₄ NPs mass ratios and JAEF/[M-L]F two-layer composite film

Samples (emu g ⁻¹)	Saturation magnetization (Ms)
JAEF/MF/LF [L _{I-3} //L _{II-3}]/L _{III-3} /S _{IV-3} (Fe ₃ O ₄ : PVP = 2 : 1)	9.3
JAEF/MF/LF [L _{I-3} //L _{II-3}]/L _{III-2} /S _{IV-3} (Fe ₃ O ₄ : PVP = 1 : 1)	5.1
JAEF/MF/LF [L _{I-3} //L _{II-3}]/L _{III-1} /S _{IV-3} (Fe ₃ O ₄ : PVP = 0.5 : 1)	3.4
JAEF/[M-L]F (Fe ₃ O ₄ : PVP = 1 : 1)	5.8



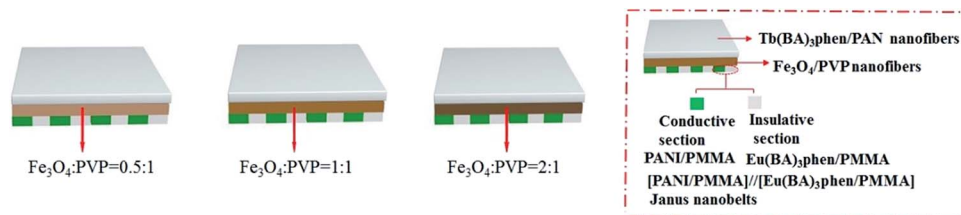


Fig. 16 Schematic of the JAEF/MF/LF sandwich-configurational composite film with various mass ratios of Fe_3O_4 NPs to PVP.

Superparamagnetic properties

The hysteresis loops of the JAEF/MF/LF sandwich-configurational composite film containing different mass ratios of Fe_3O_4 NPs and the JAEF/[M-L]F two-layer composite film are demonstrated in Fig. 15, and Table 6 summarizes their saturation magnetizations.

With the increase of Fe_3O_4 NPs content, the saturation magnetization of the JAEF/MF/LF sandwich-configurational composite film increases from 3.4 to 9.3 emu g^{-1} , indicating that the composite film possesses tunable superparamagnetism. The superparamagnetic performance of the JAEF/MF/LF sandwich-configurational composite film $\{[L_{I-3}/L_{II-3}]/L_{III-2}/L_{IV-3}\}$ is similar to that of the JAEF/[M-L]F two-layer composite film due to the fact that they have the same quantity of Fe_3O_4 NPs in theory. The diagrammatic sketch of the JAEF/MF/LF sandwich-configurational composite film, containing different mass ratios of Fe_3O_4 NPs to PVP, is revealed in Fig. 16. The color of the superparamagnetic layer becomes darker with the increase of Fe_3O_4 NPs content. This is because Fe_3O_4 NPs are black, and the presence of more Fe_3O_4 NPs contributes to the darker color of the superparamagnetic layer in the JAEF/MF/LF sandwich-configurational composite film.

Conclusions

A brand-new sandwich-configurational composite film has been successfully fabricated for the first time using a facile three-step electrospinning technology. The aeolotropic conductive-fluorescent layer (first layer) of the sandwich-configurational composite film is composed of orderly arranged Janus nanobelts which consist of a conductive side and an insulating side, making the film possess electrically conductive aeolotropism. The degree of conduction aeolotropism on the aeolotropic conductive-fluorescent layer can be adjusted by regulating the content of PANI, and the conductivity of conductive orientation is about 10^8 orders of magnitude higher than that of the insulating orientation. Disordered nanofibers severally form superparamagnetic and fluorescent layers containing Fe_3O_4 NPs and $\text{Eu}(\text{BA})_3\text{phen}$, achieving excellent superparamagnetic and fluorescent properties. The superparamagnetic properties of the composite film are tailored by varying the content of Fe_3O_4 NPs. Moreover, it is very satisfying to find that the fluorescent layer exhibits excellent fluorescence properties which are not influenced by Fe_3O_4 NPs and PANI. Microscopic partition and macroscopic partition simultaneously exist in the 2D sandwich-configurational composite film, and a highly efficient

combination of microscopic partition with macroscopic partition is actualized in the novel film. Microscopic partition, actualized in the Janus nanobelt and used as a building unit, ensures that the film possesses high conductive aeolotropism, while the macroscopic partition, realized in the whole 2D film, guarantees no mutual interference among electricity, superparamagnetism and fluorescence. Macroscopically, electrically conductive aeolotropism, superparamagnetism and fluorescence are highly integrated, meanwhile, the three functions are also mutually independent in the novel 2D film. Namely, the three functions are highly integrated as well as absolutely mutually independent. The neoteric sandwich-configurational composite film also can be popularized to assemble various three functions to realize multi-functionalization without mutual pernicious interferences among single functions. This neoteric sandwich-configurational composite film has a great potential for nanodevice construction, microwave absorption and other fields. Even more importantly, the design idea and preparative means presented in this work are of great significance towards new-typed aeolotropic conductive films with multifunctionality.

Conflicts of interest

There are no conflicts of interest to declare.

Acknowledgements

This study was financially supported by National Natural Science Foundation of China (51573023, 50972020), Natural Science Foundation of Jilin Province (20170101101JC, 20180520011JH), Industrial Technology Research and Development Project of Jilin Province Development and Reform Commission (2017C052-4), Science and Technology Research Planning Project of the Education Department of Jilin Province during the 13th five-year plan period (JJKH20170608KJ, JJKH20181122KJ), and Innovative Foundation (XJLJG-2017-04) and Youth Foundation (XQNJJ-2016-01, XQNJJ-2017-17) of Changchun University of Science and Technology.

Notes and references

- 1 S. W. Li, Y. Y. Feng, P. Long, C. Q. Qin and W. Feng, *J. Mater. Chem.*, 2017, 5, 5068–5075.
- 2 S. Fujii, Y. Suzuki, J. Kawamata and R. Tsunashima, *J. Mater. Chem. C*, 2015, 3, 7153–7158.



- 3 H. Kang, D. Kim and S. Baik, *Phys. Chem. Chem. Phys.*, 2014, **16**, 18759–18764.
- 4 Y. W. Chen, Y. H. Guo, S. Batra, E. Wang, Y. P. Wang, X. Q. Liu, Y. M. Wang and M. Cakmak, *Nanoscale*, 2015, **7**, 14636–14642.
- 5 C. Mao, J. R. Huang, Y. T. Zhu, W. Jiang, Q. X. Tang and X. J. Ma, *J. Phys. Chem. Lett.*, 2013, **4**, 43–47.
- 6 Q. H. Tang, Y. F. Wang, M. Zou and W. J. Guo, *Nanoscale*, 2018, **10**, 4344–4353.
- 7 J. R. Huang, Y. T. Zhu, W. Jiang, J. H. Yin, Q. X. Tang and X. D. Yang, *ACS Appl. Mater. Interfaces*, 2014, **6**, 1754–1758.
- 8 Q. L. Ma, J. X. Wang, X. T. Dong, W. S. Yu and G. X. Liu, *Adv. Funct. Mater.*, 2015, **25**, 2436–2443.
- 9 Y. T. Geng, P. Zhang, Q. T. Wang, Y. X. Liu and K. Pan, *J. Mater. Chem. A*, 2017, **5**, 5390–5396.
- 10 J. Y. Liu, Q. L. Ma, J. Tian, X. Xi, D. Li, X. T. Dong, W. S. Yu, X. L. Wang, J. X. Wang and G. X. Liu, *RSC Adv.*, 2018, **8**, 22887–22896.
- 11 Q. L. Ma, W. S. Yu, X. T. Dong, J. X. Wang and G. X. Liu, *Nanoscale*, 2014, **6**, 2945–2952.
- 12 Z. J. Wang, Q. L. Ma, X. T. Dong, D. Li, X. Xi, W. S. Yu, J. X. Wang and G. X. Liu, *ACS Appl. Mater. Interfaces*, 2016, **8**, 26226–26234.
- 13 Y. W. Liu, Q. L. Ma, M. Yang, X. T. Dong, Y. Yang, J. X. Wang, W. S. Yu and G. X. Liu, *Chem. Eng. J.*, 2016, **284**, 831–840.
- 14 C. Y. Wen, H. Y. Xie, Z. L. Zhang, L. L. Wu, J. Hu, M. Tang, M. Wu and D. W. Pang, *Nanoscale*, 2016, **8**, 12406–12429.
- 15 S. L. Gai, P. P. Yang, C. X. Li, W. X. Wang, Y. L. Dai, N. Niu and J. Lin, *Adv. Funct. Mater.*, 2010, **20**, 1166–1172.
- 16 P. Li, S. P. Zhang, K. Li, J. X. Wang, M. L. Liu, X. N. Gu and Y. B. Fan, *J. Mater. Chem. A*, 2018, **6**, 4952–4962.
- 17 C. C. Liu, X. H. Jiang, Y. Y. Zhao, W. W. Jiang, Z. M. Zhang and L. M. Yu, *Electrochim. Acta*, 2017, **231**, 53–60.
- 18 K. L. Zhou, H. Wang, J. T. Jiu, J. B. Liu, H. Yan and K. Suganuma, *Chem. Eng. J.*, 2018, **345**, 290–299.
- 19 X. B. Liu, Z. P. Wu and Y. H. Yin, *Chem. Eng. J.*, 2017, **323**, 330–339.
- 20 T. K. Gupta, B. P. Singh, R. B. Mathur and S. R. Dhakate, *Nanoscale*, 2013, **6**, 842–851.
- 21 X. Xi, W. S. Yu, Q. L. Ma, D. Li, X. T. Dong, J. X. Wang and G. X. Liu, *RSC Adv.*, 2018, **8**, 31608–31617.
- 22 Q. H. Li, S. L. Zhou, S. W. Wang, X. C. Zhu, L. J. Zhang, Z. J. Feng, L. P. Guo, F. H. Wang and Y. Wei, *Dalton Trans.*, 2013, **42**, 2861–2869.
- 23 B. Yan, L. M. Zhao, X. M. Wang and Y. Zhao, *RSC Adv.*, 2011, **1**, 1064–1071.
- 24 D. Liu and Z. G. Wang, *RSC Adv.*, 2012, **2**, 10085–10090.
- 25 L. B. Fan, Q. L. Ma, J. Tian, D. Li, X. Xi, X. T. Dong, W. S. Yu, J. X. Wang and G. X. Liu, *J. Mater. Sci.*, 2018, **53**, 2290–2302.
- 26 R. X. Gao, X. R. Mu, J. J. Zhang and Y. H. Tang, *J. Mater. Chem. A*, 2014, **2**, 783–792.
- 27 J. C. Feng, S. R. Yu, J. Li, T. Mo and P. Li, *Chem. Eng. J.*, 2016, **286**, 216–222.
- 28 S. M. Zhu, J. C. Fu, H. L. Li, L. Zhu, Y. Hu, W. X. Xia, X. X. Zhang, Y. Peng and J. L. Zhang, *ACS Nano*, 2018, **12**, 3442–3448.
- 29 F. Chen, C. Li, Y. J. Zhu, X. Y. Zhao, B. Q. Lu and J. Wu, *Biomater. Sci.*, 2013, **1**, 1074–1081.
- 30 C. Luo, X. X. Wang, J. Q. Wang and K. Pan, *Compos. Sci. Technol.*, 2016, **133**, 97–103.
- 31 X. Xi, J. X. Wang, X. T. Dong, Q. L. Ma, W. S. Yu and G. X. Liu, *Chem. Eng. J.*, 2014, **254**, 259–267.
- 32 N. Lv, Z. G. Wang, W. Z. Bi, G. M. Li, J. L. Zhang and J. Z. Ni, *J. Mater. Chem. A*, 2016, **4**, 4402–4409.
- 33 N. Lv, J. L. Zhang, G. M. Li, X. Wang and J. Z. Ni, *J. Phys. Chem. C*, 2017, **121**, 11926–11931.
- 34 P. Zhang, C. L. Shao, X. H. Li, M. Y. Zhang, X. Zhang, C. Y. Su, N. Lu, K. X. Wang and Y. C. Liu, *Phys. Chem. Chem. Phys.*, 2013, **25**, 10453–10458.
- 35 Y. C. Chou, C. L. Shao, X. H. Li, C. Y. Su, H. C. Xu, M. Y. Zhang, P. Zhang, X. Zhang and Y. C. Liu, *Appl. Surf. Sci.*, 2013, **285**, 509–516.
- 36 H. Ji, R. Zhao, N. Zhang, C. X. Jin, X. F. Lu and C. Wang, *NPG Asia Mater.*, 2018, **10**, 749–760.
- 37 M. Q. Chi, S. H. Chen, M. X. Zhong, C. Wang and X. F. Lu, *Chem. Commun.*, 2018, **54**, 5827–5830.
- 38 X. Xi, Q. L. Ma, X. T. Dong, D. Li, W. S. Yu, J. X. Wang and G. X. Liu, *ChemPlusChem*, 2018, **83**, 108–116.
- 39 Q. L. Ma, J. X. Wang, X. T. Dong, W. S. Yu, G. X. Liu and J. Xu, *J. Mater. Chem.*, 2012, **22**, 14438–14442.
- 40 Z. J. Li, Z. L. Hou, W. L. Song, X. D. Liu, W. Q. Cao, X. H. Shao and M. S. Cao, *Nanoscale*, 2016, **8**, 10415–10424.
- 41 L. L. Gao, X. Chen, H. Gao and S. B. Zhang, *Mater. Sci. Eng., A*, 2010, **527**, 5115–5121.
- 42 J. Tian, Q. L. Ma, X. T. Dong, M. Yang, Y. Yang, J. X. Wang, W. S. Yu and G. X. Liu, *J. Mater. Sci.: Mater. Electron.*, 2015, **26**, 8413–8420.
- 43 X. Shen and B. Yan, *RSC Adv.*, 2014, **5**, 6752–6757.
- 44 X. B. Li, Q. L. Ma, J. Tian, X. Xi, D. Li, X. T. Dong, W. S. Yu, X. L. Wang, J. X. Wang and G. X. Liu, *Nanoscale*, 2017, **9**, 18918–18930.

