CORRECTION

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Correction: A toxicology-informed, safer by design approach for the fabrication of transparent electrodes based on silver nanowires

Djadidi Toybou,^{ab} Caroline Celle,^a Catherine Aude-Garcia,^c Thierry Rabilloud^{*c} and Jean-Pierre Simonato^{*a}

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Correction for 'A toxicology-informed, safer by design approach for the fabrication of transparent electrodes based on silver nanowires' by Djadidi Toybou *et al.*, *Environ. Sci.: Nano*, 2019, DOI: 10.1039/c8en00890f.

In the Fig. 2 image (parts a and b), Fig. 2 caption and the sentence starting "Thus, in our goal to fabricate..." in the "Results and discussion" section, there was an error in the units of the KBr concentration. The value "350 mmol L⁻¹" in these locations should read "350 μmol L⁻¹". The corrected image, caption and sentence are shown below.

The corrected sentence is: "Thus, in our goal to fabricate "on demand" AgNW size standards with defined lengths and diameters, we carried out experiments by setting the KBr amount at 350 μmol L⁻¹ to control the diameter, and we introduced NaCl in various amounts in order to modulate the length".

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

^a Univ. Grenoble Alpes, LITEN DTNM, CEA, F-38054 Grenoble, France. E-mail: jean-pierre.simonato@cea.fr

^b Univ. Grenoble Alpes, ISTERRE, CNRS, F-38000, Grenoble, France

^c Univ. Grenoble Alpes, Chemistry and Biology of Metals, CEA BIG, CNRS UMR 5249, F-38054 Grenoble, France. E-mail: thierry.rabilloud@cea.fr

Fig. 2 Tuning of the length of thin AgNWs by NaCl concentration and reaction temperature. (a) Length as a function of the chloride concentration, with a constant KBr concentration of $350 \mu\text{mol L}^{-1}$. Increasing the quantity of chloride decreases the mean length of AgNWs. (b) The diameter of AgNWs remains constant around 40 nm while the length decreases from $28 \mu\text{m}$ to $3 \mu\text{m}$. The KBr concentration is set at $350 \mu\text{mol L}^{-1}$. (c) The length remains almost constant while the reaction temperature is modified. (d) Diameter tuning by reaction temperature modification, where increasing the temperature induces a diameter decrease.