## **Environmental** Science Nano



## CORRECTION

**View Article Online** 



Cite this: Environ. Sci.: Nano, 2019, **6**, 695

## Correction: A toxicology-informed, safer by design approach for the fabrication of transparent electrodes based on silver nanowires

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DOI: 10.1039/c9en90006c

rsc li/es-nano

Correction for 'A toxicology-informed, safer by design approach for the fabrication of transparent electrodes based on silver nanowires' by Djadidi Toybou et al., Environ. Sci.: Nano, 2019, DOI: 10.1039/ c8en00890f.

In the Fig. 2 image (parts a and b), Fig. 2 caption and the sentence starting "Thus, in our goal to fabricate..." in the "Results and discussion" section, there was an error in the units of the KBr concentration. The value "350 mmol L<sup>-1</sup>" in these locations should read "350 µmol L<sup>-1</sup>". The corrected image, caption and sentence are shown below.

The corrected sentence is: "Thus, in our goal to fabricate "on demand" AgNW size standards with defined lengths and diameters, we carried out experiments by setting the KBr amount at 350 umol L-1 to control the diameter, and we introduced NaCl in various amounts in order to modulate the length".

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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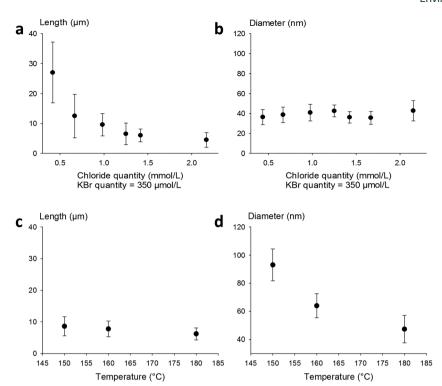


Fig. 2 Tuning of the length of thin AgNWs by NaCl concentration and reaction temperature. (a) Length as a function of the chloride concentration, with a constant KBr concentration of 350 µmol L<sup>-1</sup>. Increasing the quantity of chloride decreases the mean length of AgNWs. (b) The diameter of AgNWs remains constant around 40 nm while the length decreases from 28  $\mu m$  to 3  $\mu m$ . The KBr concentration is set at 350  $\mu m$ ol L $^{-1}$ . (c) The length remains almost constant while the reaction temperature is modified. (d) Diameter tuning by reaction temperature modification, where increasing the temperature induces a diameter decrease.