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Photodissociation dynamics and the dissociation energy of vanadium monoxide, VO, investigated using velocity map imaging†

Alexander S. Gentleman,^{id} Andreas Iskra, Hansjochen Köckert^{id} and Stuart R. Mackenzie^{id}*

Velocity map imaging has been employed to study multi-photon fragmentation of vanadium monoxide (VO) via the C $^4\Sigma^-$ state. The fragmentation dynamics are interpreted in terms of dissociation at the three-photon level, with the first photon weakly resonant with transitions to vibrational energy levels of the C $^4\Sigma^-$ state. The dissociation channels accessed are shown to depend strongly on the vibrational level via which excitation takes place. Analysis of the evolution of the kinetic energy release spectrum with photon energy leads to a refined value for the dissociation energy of ground state VO of $D_0(\text{VO}) = 53\,126 \pm 263 \text{ cm}^{-1}$.

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1. Introduction

Vanadium oxides play important roles in modern industrial processes as catalysts, semiconductors, and in optical devices.^{1,2} Vanadium monoxide (VO) also plays a key role in astrophysical chemistry, with characteristic VO transitions observed in the spectra of both cool (*i.e.* 2500–3000 K) M-class stars and young hot brown dwarfs and used in their characterisation.^{3,4} Furthermore, VO has been postulated to exist in the atmosphere of the large Jupiter exoplanets WASP-121b and WASP-127b discovered by L. Delrez *et al.*⁵ and Lam *et al.*,⁶ respectively.^{7–10}

Largely due to its astrophysical significance, the low-lying electronic states of VO have been characterised extensively. Merer *et al.* has used Fourier transform spectroscopy,^{11–13} intracavity laser-induced fluorescence spectroscopy,¹⁴ and Doppler-limited discharge emission and laser excitation spectra¹⁵ to study the ground and excited states of VO. Many of the same spectral features are clear in the more recent electronic absorption spectra of matrix isolated VO.¹⁶ This group has previously reported resonance enhanced multiphoton ionization (REMPI) spectra of VO in the visible region, characterising transitions from the VO X $^4\Sigma^-$ to the B $^4\Pi$ and C $^4\Sigma^-$ excited states as well as spin-forbidden transitions to the 2 $^2\Pi$ and 3 $^2\Pi$ excited states.¹⁷ VO_x^- ($x = 1-4$) anions have been studied by Wang and co-workers using photo-detachment spectroscopy,¹⁸ with the high-resolution spectrum of VO_2^- studied using the low Photoelectron Velocity Map Imaging (SEVI) technique by Neumark and co-workers.¹⁹ McKemmish *et al.*

have recently reviewed the known spectral lines of VO, as well as the potential energy surfaces of the low-lying states of VO.⁴

The complex electronic structure of VO has attracted considerable interest from theoreticians. In an early computational study involving the use of coupled pair functional method,²⁰ Bauschlicher and Langhoff identified the VO ground state symmetry as X $^4\Sigma^-$,²⁰ thereby confirming the assignment of Carlson and Moser²¹ which preceded high-resolution spectroscopic investigation. The bonding in the ground state of VO comprise a mix of a covalent and ionic contributions leading to an element of $(3d\sigma + 3d\pi)$ triple bond character. The covalent bonding contributions arise from V $(3d) + O(2p)$ interactions, with the lone V 4s electron polarised away from O; the ionic component from electron donation from the O atom $2p\pi$ orbital into the empty $3d\pi$ orbitals on V.²⁰ Low-lying excited states arise predominantly from V 4s excitation into non-bonding π or δ orbitals.

Bauschlicher and Langhoff calculated a dissociation energy of VO, *i.e.*, $D_0(\text{VO})$, of 5.68 eV ($45\,800 \text{ cm}^{-1}$), markedly lower than the experimental values (see Table 1) due to inadequate treatment of the considerable correlation energy.²⁰ A more recent investigation by Miliordos and Mavridis using multi-reference and coupled cluster approaches obtained a value of $D_e = 149.2 \text{ kcal mol}^{-1}$ ($52\,180 \text{ cm}^{-1}$, 6.470 eV),²² which compares better with the experimentally-determined dissociation energies.

Experimentally, Jones and Gole determined the lower-bound D_0 value of VO in their study of the $\text{V} + \text{NO}_2 \rightarrow \text{VO} + \text{NO}$ reaction.²³ Under single-collision conditions, the authors observed chemiluminescence from vibrational levels up to $v' = 11$ of the VO C $^4\Sigma^-$ -X $^4\Sigma^-$ system. With knowledge of $D_0(\text{O-NO})$, the lower-limit value for $D_0(\text{VO})$ of $48\,677 \text{ cm}^{-1}$

Department of Chemistry, Physical and Theoretical Chemistry Laboratory,
South Parks Road, Oxford, OX1 3QZ, UK. E-mail: stuart.mackenzie@chem.ox.ac.uk
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Table 1 Experimental and theoretical literature values of the ground state dissociation energy of VO, $D_0(\text{VO})$

Author	D_0	Method	Year
Jones <i>et al.</i> ²³	$\geq 48\,677\text{ cm}^{-1}$ ($\geq 6.034\text{ eV}$)	Chemiluminescence	1976
Pedley <i>et al.</i> ²⁶	$148.4 \pm 4.5\text{ kcal mol}^{-1}$ $51\,900 \pm 1600\text{ cm}^{-1}$ ($6.435 \pm 0.195\text{ eV}$)	Knudsen cell	1983
Balducci <i>et al.</i> ²⁴	$625.5 \pm 8.5\text{ kJ mol}^{-1}$ $52\,290 \pm 710\text{ cm}^{-1}$ ($6.483 \pm 0.088\text{ eV}$)	Knudsen cell	1983
Bauschlicher <i>et al.</i> ²⁰	$45\,800\text{ cm}^{-1}$ (5.68 eV)	CASSCF	1986
Miliordos <i>et al.</i> ²²	$149.2\text{ kcal mol}^{-1}$ $52\,180\text{ cm}^{-1}$ (6.470 eV)	RCCSD(T) + DKH2/BP	2007
This work	$53\,126 \pm 263\text{ cm}^{-1}$	VMI	2019

(6.034 eV) was determined. Balducci *et al.* used a combination of Knudsen cell and mass spectrometry to investigate the temperature dependence of the $\text{VO} + \text{Eu} \rightleftharpoons \text{EuO} + \text{V}$ gas-phase equilibrium.²⁴ From the heat of reaction and the well-known $D_0(\text{EuO})$, they determined a dissociation energy, $D_0(\text{VO})$ of $625.5 \pm 8.5\text{ kJ mol}^{-1}$ ($52\,290 \pm 710\text{ cm}^{-1}$, $6.483 \pm 0.088\text{ eV}$). Finally, Pedley and Marshall used the available thermochemical data from older Knudsen cell experiments to determine the value of $D_0(\text{VO})$ as $148.4 \pm 4.5\text{ kcal mol}^{-1}$ ($51\,900 \pm 1600\text{ cm}^{-1}$, $6.435 \pm 0.195\text{ eV}$).^{25,26}

Of all of the geometric and electronic properties of VO that have been elucidated by previous experimental and theoretical work, the experimental $D_0(\text{VO})$ values suffer from large experimental uncertainties limited by either/both the resolution of the technique at the time, or/and the precision of the least well-determined value in a thermodynamic cycle. We have developed a velocity map imaging (VMI) spectrometer coupled with a laser ablation source capable of investigating the photodissociation dynamics of the small metal oxides. By using VMI to measure the photofragment kinetic energy release (KER) spectra, dissociation energies can be determined directly by extrapolation of the observed KER as a function of photoexcitation energy. In this way we have determined improved ground state dissociation energies for systems including: (i) Au_2 ,²⁷ (ii) Au-RG ($\text{RG} = \text{Ar, Kr, Xe}$),^{28,29} (iii) Ag_2 , AgO , and Ag-RG ($\text{RG} = \text{Ar, Kr, Xe}$),³⁰ (iv) $\text{Li}(\text{NH}_3)_4$,³¹ (v) Cu_2 and CuO ,³² and (vi) CrO and MoO .³³

Here, we report the results of a VMI investigation into the multiphoton dissociation dynamics of VO *via* vibrational energy levels of the well-characterised $\text{C } ^4\Sigma^-$ state. The rich dissociation dynamics reflect the complex electronic structure of VO and, unlike in other systems we have studied, the dissociation channels accessed are strongly dependent on the $\text{C } ^4\Sigma^-$ state vibrational level *via* which excitation takes place. Extrapolation of the observed total kinetic energy release as a function of total photoexcitation energy yields an improved ground state dissociation energy.

II. Experimental

The VMI instrument used in these studies is a modified version of that described in our previous studies, the new variant being capable of studying both neutral and charged species. In previous

work, the VMI lens assembly comprised a conical extractor lens assembly with a resistively-coupled electrode chain, as designed by Wrede *et al.*³⁴ For the work presented here, the entire electrode chain was grounded. The new instrument was operated in neutral-dissociation mode, with the method used to generate and detect VO in the gas-phase identical to that used in prior work.¹⁷ In brief, laser ablation at 532 nm (2nd harmonic) of a rotating disc target of pure vanadium (Goodfellow, 99.98%) generates V- and VO-containing species which are entrained in a pulse of helium delivered *via* a solenoid valve (Parker-Hannifin, Series 9) from a stagnation pressure of 3 bar.

Photoexcitation of VO in the region $17\,000\text{--}23\,500\text{ cm}^{-1}$ was provided by the focussed output of an optical parametric oscillator (OPO, Continuum Panther) pumped by the 355 nm (3rd harmonic) output of a Nd:YAG laser (Continuum Surelite). The OPO pulse both photodissociates VO and photoionizes the resultant V fragments within a VMI lens assembly. Extraction of the ion sphere occurs collinear to the direction of the molecular beam and images are recorded at a 10 Hz repetition rate for *ca.* 100k–500k ion counts using a commercial MCP/scintillator (P45)/camera assembly (Photek) employing on the fly centroid averaging. Reconstruction of the central slice of the ion sphere was performed using the Polar Onion Peeling (POP) algorithm developed by Verlet and co-workers.³⁵

In the visible region probed here, the only two species observed in the molecular beam were V and VO. The REMPI spectrum of VO recorded (Fig. 1) is identical to that reported previously by our group with no new bands observed. All images presented here are of V fragments arising from photodissociation events induced when the visible output from the OPO is in the vicinity of $\text{C } ^4\Sigma^- \text{--X } ^4\Sigma^- (\nu', 0)$ transitions in VO. A magnification factor of 1.055 is used throughout this work, which is consistent both with simulations of ion trajectories with our electrode arrangement^{36,37} and our previous experiments.^{30,33}

III. Results and discussion

A. $(1 + n)$ REMPI spectrum of VO

In order to confirm the presence of VO in the molecular beam, the $(1 + n)$ REMPI spectrum was recorded in the region of the $\text{C } ^4\Sigma^- (\nu')\text{--X } ^4\Sigma^- (\nu'' = 0)$ transitions (Fig. 1a). The VO^+ REMPI



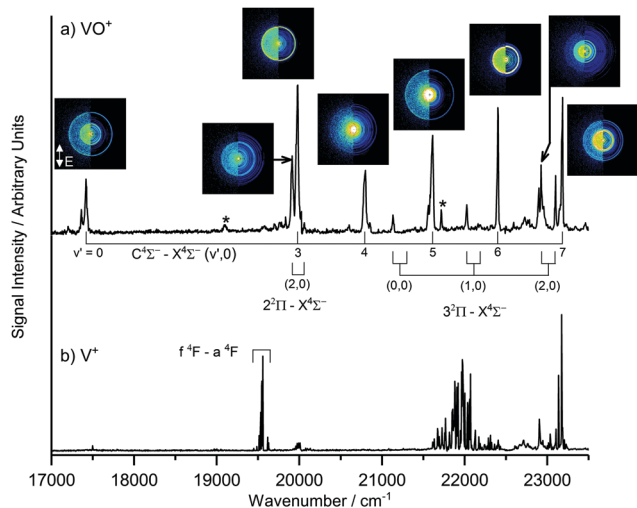


Fig. 1 (a) One colour ($1 + n$) REMPI spectrum of VO recorded in the range $17\,000\text{--}23\,500\text{ cm}^{-1}$. (b) The corresponding signal observed in the V^+ channel. Band assignments are from Hopkins *et al.* [ref. 16] and asterisks (*) mark unassigned bands. Velocity map images of V^+ photofragments recorded at energies corresponding to various vibronic transitions in (a) are included and illustrate the rich dynamical behaviour across this spectral region. The polarisation of the photoexcitation laser is indicated in the $v' = 0$ image (*i.e.* vertically-polarised in the plane of the detector).

spectrum reproduces all transitions identified previously.¹⁷ The corresponding spectrum recorded simultaneously in the V^+ ion channel is shown in Fig. 1b for comparison. V^+ is detected following: (i) direct ionization of neutral V atoms in the molecular beam (*e.g.*, the $2 + 1$ ionization *via* the f^4F state observed near $19\,550\text{ cm}^{-1}$), or (ii) photodissociation of V-containing molecules/clusters coupled with V ionization. Only in the latter case do the velocity map images show evidence of the characteristic rings associated with V-atom recoil from a co-fragment in a dissociation process.

To illustrate the richness of the dissociation dynamics observed in this region, Fig. 1a also shows velocity map images of the V^+ ion recorded in resonance with known $C^4\Sigma^- - X^4\Sigma^- (v',0)$ and $2^2\Pi - X^4\Sigma^- (v',0)$ transitions in VO. It is clear by inspection that the images display dramatic changes throughout this region.

The V^+ signals observed at *ca.* $22\,000\text{ cm}^{-1}$ cannot be assigned to any known atomic transitions in V. This signal could originate from metastable V^* in the molecular beam or else the fragmentation of larger V-containing species. In most cases though, sharp rings in the V^+ images arise only in the vicinity of vibronic transitions in VO confirming this as the parent molecule.

B. V^+ photofragment images *via* several VO $C^4\Sigma^- (v')$ levels

Velocity map images recorded in the V^+ channel while resonant with $C^4\Sigma^- - X^4\Sigma^- (v',0)$ transitions in VO are shown in Fig. 2 along with the extracted V^+ kinetic energy release (KER) spectra of each following reconstruction of the central slice of the ion sphere using the POP algorithm.³⁵ The images, and KER spectra, show clear differences depending on the vibrational state accessed in the $C^4\Sigma^-$ excited state. This contrasts with our previous observations involving the two-colour spectroscopy of

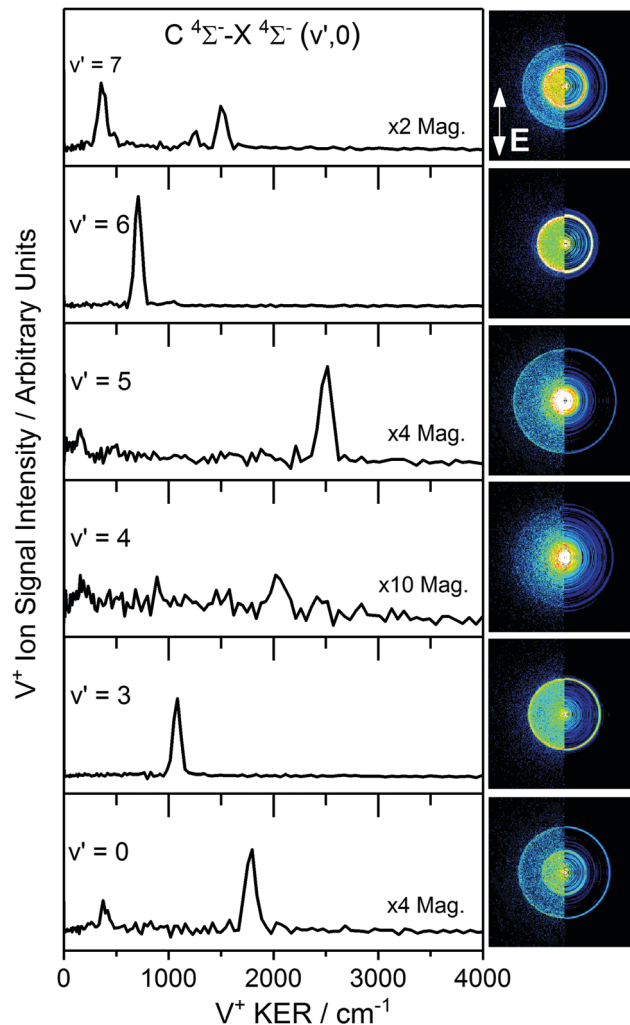


Fig. 2 Vanadium ion velocity map images and corresponding V^* kinetic energy release (KER) spectra recorded at various band positions of the $C^4\Sigma^- - X^4\Sigma^- (v',0)$ transition in the VO ($1 + n$) REMPI spectrum. The left-half of each image displays the raw collected image, with the right-half displaying the reconstructed central slice using the polar ion peeling algorithm. The polarisation of the photoexcitation laser is indicated in the $v' = 7$ image (*i.e.* vertically-polarised to the plane of the detector).

$\text{CuO} \rightarrow \text{Cu}^* 2D_{3/2} + \text{O}^* 1D$ following $\text{CuO } F^2\Pi (v') \leftarrow X^2\Pi (v'' = 0)$ excitation³² in which a single ring was observed corresponding to the same dissociation channel across the range of images.

Some V^+ images in Fig. 2 show a single strong, isotropic ring (*via* $C^4\Sigma^- v' = 3, 6$) whilst others (*via* $C^4\Sigma^- v' = 0, 5, 7$) clearly show multiple features. The image recorded *via* $C^4\Sigma^- v' = 4$ is significantly weaker than those recorded from other vibrational states and shows only weak features above the noise level. We proceed by considering the dissociation dynamics observed in individual images before assembling all information to determine dissociation channels common to different images and extrapolate to determine a best estimate of the dissociation energy.

C. VO dissociation *via* the $C^4\Sigma^- (v = 3)$ level

One challenge encountered in studying dissociation of neutral precursors generated by laser ablation is that of identifying



unambiguously the parent molecule. Where dissociation to the same channel occurs over a wide enough region of photon energy, consideration of the fragment kinetic energy release as a function of photon energy permits determination of the mass of the co-fragment. Fig. 3 shows V^+ velocity map images recorded in the vicinity of the $C^4\Sigma^- - X^4\Sigma^- (3,0)$ transition of VO along with the extracted V^+ KER spectra. A small but clear increase in the V^+ KER with photoexcitation energy is observed as additional energy above the dissociation threshold is released as kinetic energy of the photofragments. By recording 19 such images in the spectral region $19\,910\text{--}20\,075\text{ cm}^{-1}$ (Fig. 3b), the variation of V^+ KER with photon energy confirmed VO dissociation (*i.e.*, O atom co-fragment) at the three-photon level, consistent with the observation of VO REMPI spectrum (see ESI† for details). In principle, the data would also be consistent with photofragmentation of V_3 at the single-photon level but there is no evidence of V_3 in the molecular beam despite the ionisation energy for V_3 ($44\,336\text{ cm}^{-1}$)³⁸ being

lower than that of V ($54\,411.7\text{ cm}^{-1}$).³⁹ However, the most compelling evidence for VO dissociation comes from the fact that rings in the V^+ images are only present at the energies that correspond to known vibronic transitions in VO.

Having established oxygen as the co-fragment, the V^+ KER spectra can be recast as total kinetic energy release (TKER) spectra and used to determine the dissociation energy of VO. First it is necessary to assign the fragmentation channels observed. As discussed earlier, the most recent experimental value for the ground state dissociation energy of VO is that of Balducci *et al.*, *i.e.* $D_0(\text{VO}) = 625.5 \pm 8.5\text{ kJ mol}^{-1}$ ($52\,345 \pm 726\text{ cm}^{-1}$).²⁴ This $D_0(\text{VO})$ value was used to initially identify the exit channel accessed *via* the photofragmentation of VO in the vicinity of the $C^4\Sigma^- - X^4\Sigma^- (3,0)$ transition.

A plot of TKER *versus* $3h\nu$ yields a straight line

$$\text{TKER} = 3h\nu - (D_0(\text{VO}) + E_V + E_O), \quad (1)$$

from which a value of $55\,560 \pm 170\text{ cm}^{-1}$ is obtained for $(D_0(\text{VO}) + E_V + E_O)$ by extrapolation to $\text{TKER} = 0$ (see ESI†). Assuming the dissociation energy of $52\,350\text{ cm}^{-1}$ Balducci *et al.* generates $E_V + E_O \approx 3\,200 \pm 900\text{ cm}^{-1}$. Given the sparsity of V and O atom terms in the low energy region, the only channel consistent with this measurement is $V(a^6D) + O(^3P)$ at $2100\text{--}2600\text{ cm}^{-1}$ depending on spin-orbit levels.^{40,41} The next excited electronic terms would be $V(^4D)$ and/or $O(^1D)$ some 6200 cm^{-1} or $15\,970\text{ cm}^{-1}$ higher in wavenumber, respectively.

With some confidence in this assignment of fragment terms generated by fragmentation *via* the $C^4\Sigma^- (\nu = 3)$ level we can reverse the argument and determine a more refined value for the dissociation energy from our measurement. With a weighted averages of the $V(^6D)$ and $O(^3P)$ spin-orbit levels and within the assumption of dissociation at the three-photon level we calculate an initial estimate of the ground state dissociation energy $D_0(\text{VO})$ of $53\,200 \pm 300\text{ cm}^{-1}$.

D. Images recorded in the vicinity of the VO $C^4\Sigma^- - X^4\Sigma^- (7,0)$ transition

The V^+ velocity map image recorded *via* the $C^4\Sigma^- (\nu = 7)$ level, is shown in Fig. 2. Multiple rings/dissociation channels are observed in the image which persist following excitation throughout the spectral region around the VO $C^4\Sigma^- - X^4\Sigma^- (7,0)$ transition.

Fig. 4 shows the TKER peak positions as a function of three-photon photoexcitation wavenumber for images recorded in the vicinity of the $C^4\Sigma^- - X^4\Sigma^- (7,0)$ band ($23\,100\text{--}23\,250\text{ cm}^{-1}$) as well as around the $3^2\Pi - X^4\Sigma^- (2,0)$ band ($22\,800\text{--}22\,950\text{ cm}^{-1}$). At this total energy many dissociation channels are open and the TKER spectra observed compare well with the production of V electronic states between 8400 and $15\,600\text{ cm}^{-1}$ above the ground term as listed in Table 2.

In all cases the O atoms is produced in the 3P ground state and, in the absence of spin-orbit level resolution, the lines shown in Fig. 4 assume weighted mean wavenumbers for each atomic term.

Fig. 5 illustrates the assigned exit channels accessed upon three-photon dissociation of VO *via* resonance with the $C^4\Sigma^-$

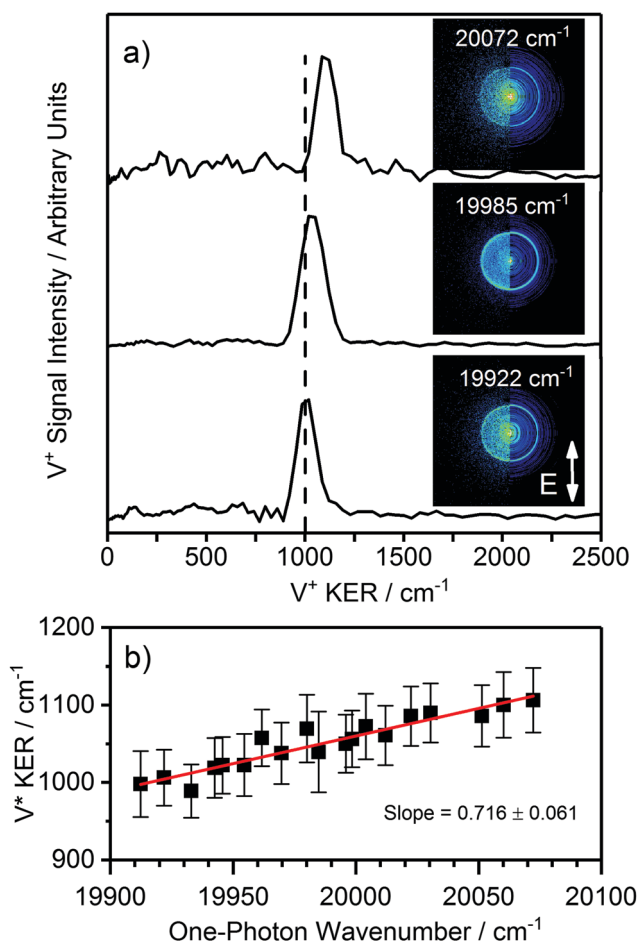


Fig. 3 (a) V^+ kinetic energy release (KER) spectra channel recorded at various photon energies in the vicinity of the $C^4\Sigma^- - X^4\Sigma^- (3,0)$ transition of VO. The KER (and hence the size of the ring in the image) increases with increasing photoexcitation energy. (b) Plotting the variation in V^* KER with photon energy allows a determination of (i) the number of photons involved in the dissociation process and (ii) the mass of the co-fragment (see ESI† for more details).



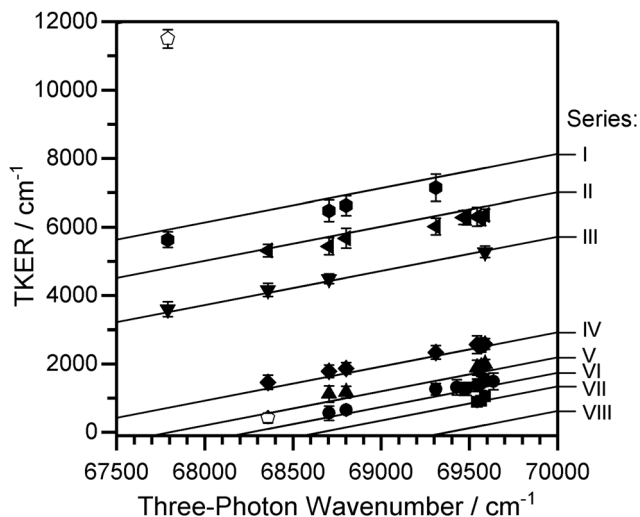


Fig. 4 Peak positions in the TKER spectra plotted as a function of three-photon energy. Uncertainties represent the FWHM of the TKER peak in consideration. The solid lines represent the simulated TKERs based on the $D_0(\text{VO})$ value determined from V^+ velocity map images recorded in the $\nu' = 3$ region, and the identified quantum states of the vanadium and oxygen photofragments (Series I–VII, see Table 2 and text for details). For completion, Series VIII represents the TKER expected for the $\text{V}^4\text{F} + \text{O}^1\text{D}$ channel. The open data points could not be assigned unambiguously to any series.

Table 2 Assigned photofragments resulting from the photodissociation of VO in the vicinity of $\text{C}^4\Sigma^- (\nu' = 3, 7) - \text{X}^4\Sigma^- (\nu'' = 0)$ transitions together with $D_0(\text{VO})$ determined from each series. The weighted-average of the $D_0(\text{VO})$ values is evaluated from the contributions from the data collected around $\nu' = 3$, and from Series I–VII in Fig. 4

	Channel	$D_0(\text{VO})/\text{cm}^{-1}$
$\nu' = 3$	$\text{V}(3d^4(^5\text{D})4s; a^6\text{D}) + \text{O}(^3\text{P})$	$53\,190 \pm 242$
$\nu' = 7$	Series I $\text{V}(3d^4(^3\text{D})4s; a^4\text{D}) + \text{O}(^3\text{P})$	$53\,507 \pm 356$
	Series II $\text{V}(3d^34s^2; a^4\text{P}) + \text{O}(^3\text{P})$	$53\,419 \pm 293$
	Series III $\text{V}(3d^34s^2; a^2\text{G}) + \text{O}(^3\text{P})$	$53\,136 \pm 262$
	Series IV $\text{V}(3d^34s^2; a^2\text{P}) + \text{O}(^3\text{P})$	$53\,075 \pm 230$
	Series V $\text{V}(3d^34s^2; a^2\text{D}_2) + \text{O}(^3\text{P})$	$53\,007 \pm 225$
	Series VI $\text{V}(3d^3(^3\text{H})4s; a^4\text{H}) + \text{O}(^3\text{P})$	$53\,066 \pm 234$
	Series VII $\text{V}(3d^4(^3\text{P}_2)4s; b^4\text{P}) + \text{O}(^3\text{P})$	$53\,143 \pm 329$
	Weighted-average:	$53\,190 \pm 261$

($\nu' = 3, 7$) levels. All observed dissociation channels lie above the VO ionization energy and thus neutral dissociation could, in principle, compete with VO^+ photofragmentation. However, fitting assuming VO^+ photodissociation led to less satisfactory assignments. Similarly, dissociation producing O^1D was considered (Fig. 5 – Series VIII) but lies too high in energy.

With knowledge of the product quantum state distributions of photofragments together with the measured TKER allows for a more refined determination of the parent molecule ground state dissociation energy. For the photodissociation channels identified in Fig. 4, fixing the gradient at 1 (for dissociation at the 3-photon level) and performing linear-fits yields $D_0(\text{VO})$ values given in Table 2. The table also includes an overall weighted-average for $D_0(\text{VO})$ calculated from values determined from each individual contribution in Series I–VII ($53\,190 \pm 261 \text{ cm}^{-1}$), and the

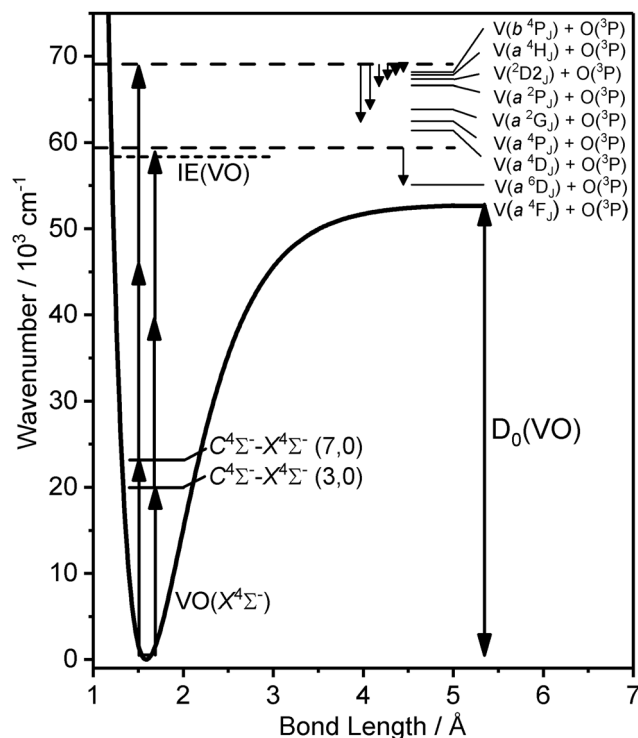


Fig. 5 Potential energy curve of the ground $\text{X}^4\Sigma^-$ state of VO shown together with various dissociation thresholds accessed by the three-photon dissociation of VO via resonance with vibronic levels of the excited $\text{C}^4\Sigma^-$ state (energies of $\nu' = 3$ and 7 are shown as examples).

value determined from the V^+ velocity map images recorded via the $\text{C}^4\Sigma^- \nu' = 3$ vibronic level (Section IIIB). The $D_0(\text{VO})$ values determined from Series I–VII are in good agreement with each other, and with that determined from the data recorded in the vicinity of the $\nu' = 3$ vibronic transition further supporting the assignments of the identified exit channels.

The weighted-average value for the VO dissociation energy based on Series I–VII can be compared with available literature values listed in Table 1. This value was determined by calculating the weighted arithmetic mean of all $D_0(\text{VO})$ values determined from each individual series, with each value being weighted relative to the number of data points comprising each individual series (with the error calculated from the square-root of the sum of weighted variance of the error from each series). Our value agrees to within mutual uncertainties with the Knudsen cell measurements but lies towards the upper end of the range. It should be noted that there is an intrinsic uncertainty in the measurements performed with Knudsen cell mass spectrometry, which can only determine dissociation energies relative to that of a reference sample for strongly-bound systems such as VO.

E. VO photodissociation at other photon energies

Using the weighted-average $D_0(\text{VO})$ value determined above ($53\,190 \pm 261 \text{ cm}^{-1}$, Table 2), it is possible to simulate TKER spectra for comparison with the experimental spectra in order to identify the dissociation channels accessed throughout this spectral region.



Fig. 6 shows the experimental TKER spectra extracted from images recorded at photoexcitation energies corresponding to peak band positions of the VO $C^4\Sigma^-(v' = 4-7)-X^4\Sigma^-(v'' = 0)$ transitions, together with the corresponding simulated TKER spectra. In each case the features observed in the experimental spectrum compare well with those in the simulated TKER spectrum assuming production of the V fragments in a range of excited electronic terms.

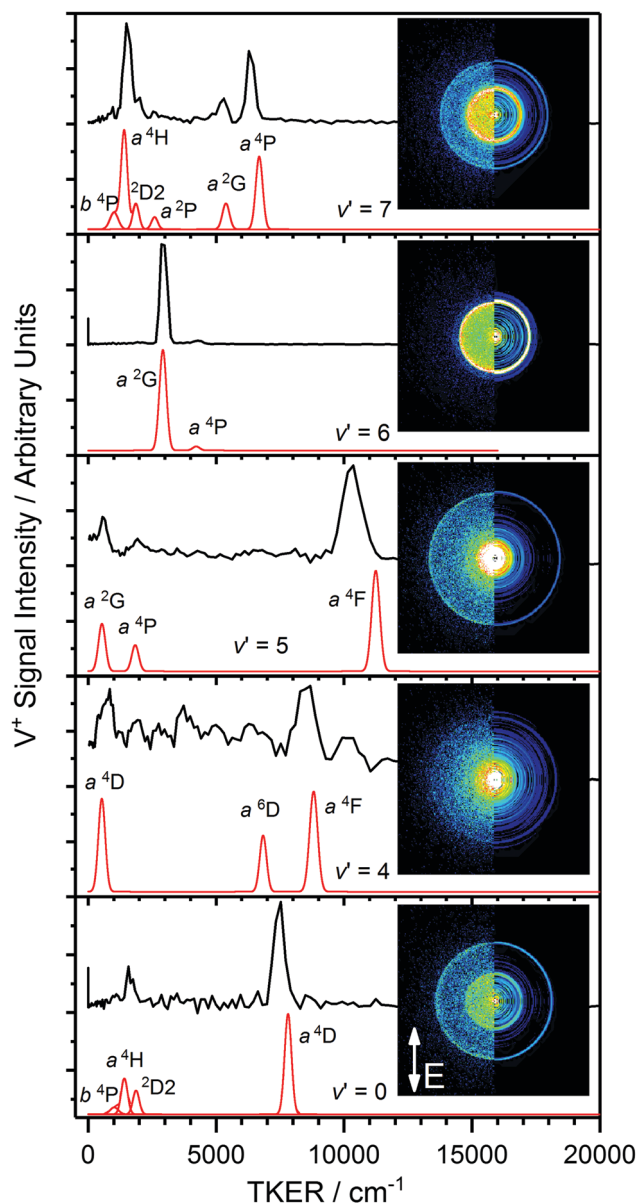


Fig. 6 TKER spectra recorded at the peak band positions of the $C^4\Sigma^-(v' = 4-7)-X^4\Sigma^-(v'' = 0)$ transitions. Black: experimental TKER spectra. Red: simulated TKER spectra for different product quantum states of the vanadium atom photofragment. A $D_0(\text{VO}) = 53190 \pm 261 \text{ cm}^{-1}$ value was used, assuming photodissociation at the three-photon level. FWHMs include uncertainties in the $D_0(\text{VO})$ value, in addition to the weighted averages of V and O atom spin orbit levels. Relative amplitudes of the simulated features are fitted to the experimental data. The oxygen atom co-fragment is assumed to be formed in the ground electronic state (*i.e.*, ^3P). Inset: V^+ velocity map images recorded at the peak vibronic band position.

The cleanest velocity map image is that recorded *via* $C^4\Sigma^-(v' = 6)$ with one strong ring readily assigned to the $V^*(a^2G) + O^3P$ channel and a much weaker outer ring corresponding to $V(a^4P) + O^3P$. The same two channels are visible as low TKER inner rings in the image recorded *via* $C^4\Sigma^-(v' = 5)$ this time in 2:1 intensity ratio, together with a third, outer ring assigned to $V(a^4F) + O^3P$ (*i.e.*, ground state fragments). The a^2G and a^4P terms of V^* are observed a third time in the image *via* $C^4\Sigma^-(v' = 7)$ but in this case in an intensity ratio *ca.* 1:3. Several additional low TKER dissociation channels are observed in this image, most significantly the $V^* a^4H + O^3P$ channel.

As remarked upon earlier, the image recorded *via* $C^4\Sigma^-(v' = 4)$ has anomalously poor signal-to-noise reflecting low ion counts/inefficient photofragmentation at this particular energy. Nevertheless, weak features are observed which can be tentatively assigned to the $O^3P + V^* a^4F$, a^6D and a a^4D channels.

If our working value for the dissociation energy is correct, 1 + 2 excitation *via* the C-X origin band is insufficient to reach the dissociation threshold. Hence the velocity map image recorded *via* $C^4\Sigma^-(v' = 0)$ must represent a 1 + 3, four-photon excitation (to $\sim 69675 \text{ cm}^{-1}$). This lies energetically close to the three-photon energy *via* the $v' = 7$ transition (69590 cm^{-1}) and indeed some of the same channels are observed in the same low TKER ($< 3000 \text{ cm}^{-1}$).

At this stage, the trends in the dissociation channels accessed *via* various $C^4\Sigma^-(v)$ levels become clear and, as shown in Fig. 7, for some features (notably the $V^* a^4P + O^3P$ and $V^* a^2G + O^3P$ channels) it is possible to plot the variation in TKER as a function of total excitation wavenumber across different excitation regions. This, in turn, allows extrapolation to

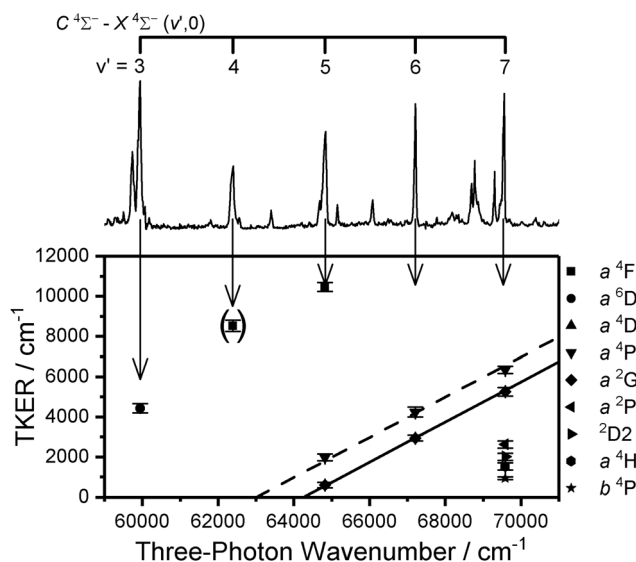


Fig. 7 Bottom: Peak positions in the TKER spectra extracted from V^+ images recorded at energies corresponding to the $C^4\Sigma^-(v' = 3-7)-X^4\Sigma^-(v'' = 0)$ vibronic transitions of VO plotted as a function of three-photon energy. Top: The (1 + n) REMPI spectrum of VO included for visual comparison. The error bars represent the FWHM of the peak in consideration. The dashed and solid lines represent the simulated TKER for the a^4P and a^2G series, respectively. The TKER peak corresponding to the energy of the $v' = 4$ vibronic band is shown in parentheses to indicate the poor quality of the V^+ image recorded at this band position.



much lower TKER (in the case of the ^2G state to within 500 cm^{-1} of threshold), providing a yet further improved value for the ground-state dissociation energy of $53\,126 \pm 263\text{ cm}^{-1}$ (from the $\text{V}^* \ ^2\text{G}$ series). The largest contribution to the uncertainty comes from the (spectrally unresolved) distribution of spin-orbit levels in both fragments for which, in the absence of any better information, a statistical distribution has been assumed. This value is consistent with those in Table 2 from extrapolation in other regions but, in our view, represents the most reliable value for $D_0(\text{VO})$ determined from this study.

F. Bond dissociation energy of VO^+

With an improved $D_0(\text{VO})$ value derived above, a simple thermodynamic cycle permits the determination of the bond energy for the corresponding cation, VO^+ . $D_0(\text{VO})$ and $D_0(\text{VO}^+)$ are linked by the respective ionisation energies of V and VO which are well-known:

$$D_0(\text{VO}) - D_0(\text{VO}^+) = \text{IE}(\text{VO}) - \text{IE}(\text{V}).$$

The most accurate $\text{IE}(\text{V})$ and $\text{IE}(\text{VO})$ values measured to date are $\text{IE}(\text{V}) = 54411.67 \pm 0.17\text{ cm}^{-1}$ from James *et al.*³⁹ and $\text{IE}(\text{VO}) = 58380.0 \pm 0.7\text{ cm}^{-1}$ recently measured by Luo *et al.*⁴² Together with our $D_0(\text{VO})$ value, a $D_0(\text{VO}^+)$ value of $49\,158 \pm 263\text{ cm}^{-1}$ ($6.095 \pm 0.033\text{ eV}$) is obtained. This value lies towards the upper end of the range of previous experimental measures of this value but agrees with most within mutual uncertainties (see Table S2 in ESI†).

IV. Conclusions

Photofragment velocity map images have been recorded over the visible range in the vicinity of the $\text{C } ^4\Sigma^- - \text{X } ^4\Sigma^- (v',0)$ vibronic transitions of VO. Analysis of the variation in kinetic energy release with photon energy permits a refined value for the dissociation energy of $D_0(\text{VO}) = 53\,126 \pm 263\text{ cm}^{-1}$ to be determined. This value, a comparatively direct measure of D_0 , is consistent with all images recorded in this study and compares well with values determined *via* other experimental techniques as well as those calculated theoretically.

This VMI investigation reveals a strong dependence of the dissociation channels accessed on the intermediate vibronic state of VO accessed *via* which the excitation takes place. Although the spectroscopy of VO is well-understood for the ground and low-lying electronic states, higher-lying states remain poorly characterised. Additional spectroscopic and theoretical investigations would be required to fully understand the potential energy surfaces accessed and crossings explored in the photodissociation process. In the spectral region explored here, evidence is found for production of the V atom photofragment in all electronic states between the ground a ^4F electronic state, and the eighth excited electronic state, b ^4P . We find no evidence for the production of the oxygen co-fragment in anything other than the ground ^3P term.

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Conflicts of interest

There are no conflicts to declare.

References

- M. F. Becker, A. B. Buckman, R. M. Walser, T. Lepine, P. Georges and A. Brun, *Appl. Phys. Lett.*, 1994, **65**, 1507–1509.
- R. Grabowski, S. Pietrzyk, J. Sloczynski, F. Genser, K. Wcislo and B. Grzybowska-Swierkosz, *Appl. Catal., A*, 2002, **232**, 277–288.
- N. M. White and R. F. Wing, *Astrophys. J.*, 1978, **222**, 209–219.
- L. K. McKemmish, S. N. Yurchenko and J. Tennyson, *Mon. Not. R. Astron. Soc.*, 2016, **463**, 771–793.
- L. Delrez, A. Santerne, J. M. Almenara, D. R. Anderson, A. Collier-Cameron, R. F. Diaz, M. Gillon, C. Hellier, E. Jehin, M. Lendl, P. F. L. Maxted, M. Neveu-Van Malle, F. Pepe, D. Pollacco, D. Queloz, D. Segransan, B. Smalley, A. M. S. Smith, A. H. M. J. Triaud, S. Udry, V. Van Grootel and R. G. West, *Mon. Not. R. Astron. Soc.*, 2016, **458**, 4025–4043.
- K. W. F. Lam, F. Faedi, D. J. A. Brown, D. R. Anderson, L. Delrez, M. Gillon, G. Hebrard, M. Lendl, L. Mancini, J. Southworth, B. Smalley, A. H. M. Triaud, O. D. Turner, K. L. Hay, D. J. Armstrong, S. C. C. Barros, A. S. Bonomo, F. Bouchy, P. Boumis, A. C. Cameron, A. P. Doyle, C. Hellier, T. Henning, E. Jehin, G. King, J. Kirk, T. Loudon, P. F. L. Maxted, J. J. McCormac, H. P. Osborn, E. Palle, F. Pepe, D. Pollacco, J. Prieto-Arranz, D. Queloz, J. Rey, D. Segransan, S. Udry, S. Walker, R. G. West and P. J. Wheatley, *Astron. Astrophys.*, 2017, **599**, A3.
- I. Hubeny, A. Burrows and D. Sudarsky, *Astrophys. J.*, 2003, **594**, 1011–1018.
- J. J. Fortney, K. Lodders, M. S. Marley and R. S. Freedman, *Astrophys. J.*, 2008, **678**, 1419–1435.
- T. M. Evans, D. K. Sing, H. R. Wakeford, N. Nikolov, G. E. Ballester, B. Drummond, T. Kataria, N. P. Gibson, D. S. Amundsen and J. Spake, *Astrophys. J., Lett.*, 2016, **822**, L4.
- T. M. Evans, D. K. Sing, T. Kataria, J. G. Oyal, N. Nikolov, H. R. Wakeford, D. Deming, M. S. Marley, D. S. Amundsen, G. E. Ballester, J. K. Barstow, L. Ben-Jaffel, V. Bourrier, L. A. Buchhave, O. Cohen, D. Ehrenreich, A. G. Munoz, G. W. Henry, H. Knutson, P. Lavvas, A. L. des Etangs, N. K. Lewis, M. Lopez-Morales, A. M. Mandell, J. Sanz-Forcada, P. Tremblin and R. Lupu, *Nature*, 2017, **548**, 58–61.
- A. S. C. Cheung, A. W. Taylor and A. J. Merer, *J. Mol. Spectrosc.*, 1982, **92**, 391–409.
- A. J. Merer, G. Huang, A. S. C. Cheung and A. W. Taylor, *J. Mol. Spectrosc.*, 1987, **125**, 465–503.
- R. S. Ram, P. F. Bernath, S. P. Davis and A. J. Merer, *J. Mol. Spectrosc.*, 2002, **211**, 279–283.
- G. I. Huang, A. J. Merer and D. J. Clouthier, *J. Mol. Spectrosc.*, 1992, **153**, 32–40.
- A. S. C. Cheung, P. G. Hajigeorgiou, G. Huang, S. Z. Huang and A. J. Merer, *J. Mol. Spectrosc.*, 1994, **163**, 443–458.



- 16 O. Hubner, J. Hornung and H. J. Himmel, *J. Chem. Phys.*, 2015, **143**, 024309.
- 17 W. S. Hopkins, S. M. Hamilton and S. R. Mackenzie, *J. Chem. Phys.*, 2009, **130**, 144308.
- 18 H. B. Wu and L. S. Wang, *J. Chem. Phys.*, 1998, **108**, 5310–5318.
- 19 J. B. Kim, M. L. Weichman and D. M. Neumark, *J. Chem. Phys.*, 2014, **140**, 034307.
- 20 C. W. Bauschlicher and S. R. Langhoff, *J. Chem. Phys.*, 1986, **85**, 5936–5942.
- 21 K. D. Carlson and C. Moser, *J. Chem. Phys.*, 1966, **44**, 3259–3265.
- 22 E. Miliordos and A. Mavridis, *J. Phys. Chem. A*, 2007, **111**, 1953–1965.
- 23 R. W. Jones and J. L. Gole, *J. Chem. Phys.*, 1976, **65**, 3800–3802.
- 24 G. Balducci, G. Gigli and M. Guido, *J. Chem. Phys.*, 1983, **79**, 5616–5622.
- 25 P. Coppens, S. Smoes and J. Drowart, *Trans. Faraday Soc.*, 1967, **63**, 2140–2148.
- 26 J. B. Pedley and E. M. Marshall, *J. Phys. Chem. Ref. Data*, 1983, **12**, 967–1031.
- 27 W. S. Hopkins, S. M. Hamilton, P. D. McNaughtner and S. R. Mackenzie, *Chem. Phys. Lett.*, 2009, **483**, 10–15.
- 28 W. S. Hopkins, A. P. Woodham, R. J. Plowright, T. G. Wright and S. R. Mackenzie, *J. Chem. Phys.*, 2010, **132**, 214303.
- 29 W. S. Hopkins, A. P. Woodham, R. J. Plowright, T. G. Wright and S. R. Mackenzie, *J. Chem. Phys.*, 2011, **134**, 094311.
- 30 G. A. Cooper, A. Kartouzian, A. S. Gentleman, A. Iskra, R. van Wijk and S. R. Mackenzie, *J. Chem. Phys.*, 2015, **143**, 124302.
- 31 W. S. Hopkins, A. P. Woodham, N. M. Tonge, A. M. Ellis and S. R. Mackenzie, *J. Phys. Chem. Lett.*, 2011, **2**, 257–261.
- 32 I. S. Parry, A. C. Hermes, A. Kartouzian and S. R. Mackenzie, *Phys. Chem. Chem. Phys.*, 2014, **16**, 458–466.
- 33 G. A. Cooper, A. S. Gentleman, A. Iskra and S. R. Mackenzie, *J. Chem. Phys.*, 2017, **147**, 013921.
- 34 E. Wrede, S. Laubach, S. Schulenburg, A. Brown, E. R. Wouters, A. J. Orr-Ewing and M. N. R. Ashfold, *J. Chem. Phys.*, 2001, **114**, 2629–2646.
- 35 G. M. Roberts, J. L. Nixon, J. Lecointre, E. Wrede and J. R. R. Verlet, *Rev. Sci. Instrum.*, 2009, **80**, 053104.
- 36 SIMION 8.0, Scientific Instrument Services Inc., USA.
- 37 D. A. Dahl, *Int. J. Mass Spectrom.*, 2000, **200**, 3–25.
- 38 M. S. Ford and S. R. Mackenzie, *J. Chem. Phys.*, 2005, **123**, 084308.
- 39 A. M. James, P. Kowalczyk, E. Langlois, M. D. Campbell, A. Ogawa and B. Simard, *J. Chem. Phys.*, 1994, **101**, 4485–4495.
- 40 The resolution of the VMI spectrometer is too low to resolve individual spin-orbit components of the electronic states of vanadium and oxygen. Instead, the weighted arithmetic mean energy of the spin-orbit components for a particular electronic state with the corresponding weight standard deviation is considered. The J level energies for V ($3d^4(^5D)4s$); a $^6D_J = 1/2-9/2$ are: 2112, 2153, 2220, 2311, 2425 cm^{-1} . This gives a weighted mean energy of V ($a\ ^6D$) = $2293 \pm 128\ \text{cm}^{-1}$. Similarly for O ($^3P_J = 2-0$): 0, 158, and 227 cm^{-1} . Hence, O ($^3P_J = 2-0$) = $78 \pm 119\ \text{cm}^{-1}$.
- 41 A. Kramida, Y. Ralchenko and J. Reader and the NIST ASD Team, with all atomic energies quoted from the NIST Atomic Spectra Database (version 5.6.1) [Online], National Institute of Standards and Technology, Gaithersburg, MD, 2018.
- 42 Z. H. Luo, Y. C. Chang, H. Huang and C. Y. Ng, *J. Phys. Chem. A*, 2015, **119**, 11162–11169.

