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## Correction: Facile hydrothermal synthesis of hierarchical porous priceite ( $\text{Ca}_4\text{B}_{10}\text{O}_{19}\cdot7\text{H}_2\text{O}$ ) microspheres as high-efficiency adsorbents for heavy metal ions removal

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Correction for ‘Facile hydrothermal synthesis of hierarchical porous priceite ( $\text{Ca}_4\text{B}_{10}\text{O}_{19}\cdot7\text{H}_2\text{O}$ ) microspheres as high-efficiency adsorbents for heavy metal ions removal’ by Wancheng Zhu *et al.*, *CrystEngComm*, 2019, DOI: 10.1039/c9ce01003c.

The authors regret some citation errors concerning the references in Table 1 of the published paper. Table 1 with the correct references is shown below.

**Table 1** Comparison of the adsorption capacity for  $\text{Pb}^{2+}$  on various adsorbents

Adsorbent	$^a S_{\text{BET}}$ ( $\text{m}^2 \text{g}^{-1}$ )	$\text{Pb}^{2+}$ adsorption capacity $q_m$ ( $\text{mg g}^{-1}$ )	Ref.
BiOBr microspheres	59.30	6.5	50
Spherical mesoporous silica	—	59.0	51
Organic silica hollow spheres	259.90	75.6	52
Urchin-like $\alpha$ -FeOOH hollow spheres	96.90	80.0	22
Hierarchical porous MgO microrods	50.20	124.4	48
Porous $\text{Ca}(\text{BO}_2)_2$ microspheres	42.70	140.2	25
Magnetic PNB core–shell microspheres	—	143.0	6
Carnation-like $\text{Ca}_4\text{B}_{10}\text{O}_{19}\cdot7\text{H}_2\text{O}$ microspheres	32.79	256.4	This work
$\text{Fe}_3\text{O}_4@\text{SiO}_2@\text{Zr}(\text{OH})_x$ CSNs	29.00	283.8	53
Hydroxyapatite/ $\text{Fe}_3\text{O}_4$ microspheres	59.40	440.0	54
Flower-like $\text{Ti}(\text{HPO}_4)_2\cdot\text{H}_2\text{O}$ microstructures	122.00	550.0	55

<sup>a</sup>  $S_{\text{BET}}$ : specific surface area.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

