

CORRECTION

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 Cite this: *CrystEngComm*, 2019, 21, 7329

Correction: Facile hydrothermal synthesis of hierarchical porous priceite (Ca₄B₁₀O₁₉·7H₂O) microspheres as high-efficiency adsorbents for heavy metal ions removal

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DOI: 10.1039/c9ce90172h

rsc.li/crystengcomm

 Correction for 'Facile hydrothermal synthesis of hierarchical porous priceite (Ca₄B₁₀O₁₉·7H₂O) microspheres as high-efficiency adsorbents for heavy metal ions removal' by Wancheng Zhu *et al.*, *CrystEngComm*, 2019, DOI: 10.1039/c9ce01003c.

The authors regret some citation errors concerning the references in Table 1 of the published paper. Table 1 with the correct references is shown below.

Table 1 Comparison of the adsorption capacity for Pb²⁺ on various adsorbents

Adsorbent	^a S _{BET} (m ² g ⁻¹)	Pb ²⁺ adsorption capacity q _m (mg g ⁻¹)	Ref.
BiOBr microspheres	59.30	6.5	50
Spherical mesoporous silica	—	59.0	51
Organic silica hollow spheres	259.90	75.6	52
Urchin-like α-FeOOH hollow spheres	96.90	80.0	22
Hierarchical porous MgO microrods	50.20	124.4	48
Porous Ca(BO ₂) ₂ microspheres	42.70	140.2	25
Magnetic PNB core-shell microspheres	—	143.0	6
Carnation-like Ca ₄ B ₁₀ O ₁₉ ·7H ₂ O microspheres	32.79	256.4	This work
Fe ₃ O ₄ @SiO ₂ @Zr(OH) _x CSNs	29.00	283.8	53
Hydroxyapatite/Fe ₃ O ₄ microspheres	59.40	440.0	54
Flower-like Ti(HPO ₄) ₂ ·H ₂ O microstructures	122.00	550.0	55

^a S_{BET}: specific surface area.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

