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## Correction: Facile hydrothermal synthesis of hierarchical porous priceite (Ca<sub>4</sub>B<sub>10</sub>O<sub>19</sub>·7H<sub>2</sub>O) microspheres as high-efficiency adsorbents for heavy metal ions removal

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 Correction for 'Facile hydrothermal synthesis of hierarchical porous priceite (Ca<sub>4</sub>B<sub>10</sub>O<sub>19</sub>·7H<sub>2</sub>O) microspheres as high-efficiency adsorbents for heavy metal ions removal' by Wancheng Zhu *et al.*, *CrystEngComm*, 2019, DOI: 10.1039/c9ce01003c.

The authors regret some citation errors concerning the references in Table 1 of the published paper. Table 1 with the correct references is shown below.

**Table 1** Comparison of the adsorption capacity for Pb<sup>2+</sup> on various adsorbents

| Adsorbent                                                                                      | <sup>a</sup> S <sub>BET</sub> (m <sup>2</sup> g <sup>-1</sup> ) | Pb <sup>2+</sup> adsorption capacity q <sub>m</sub> (mg g <sup>-1</sup> ) | Ref.      |
|------------------------------------------------------------------------------------------------|-----------------------------------------------------------------|---------------------------------------------------------------------------|-----------|
| BiOBr microspheres                                                                             | 59.30                                                           | 6.5                                                                       | 50        |
| Spherical mesoporous silica                                                                    | —                                                               | 59.0                                                                      | 51        |
| Organic silica hollow spheres                                                                  | 259.90                                                          | 75.6                                                                      | 52        |
| Urchin-like α-FeOOH hollow spheres                                                             | 96.90                                                           | 80.0                                                                      | 22        |
| Hierarchical porous MgO microrods                                                              | 50.20                                                           | 124.4                                                                     | 48        |
| Porous Ca(BO <sub>2</sub> ) <sub>2</sub> microspheres                                          | 42.70                                                           | 140.2                                                                     | 25        |
| Magnetic PNB core-shell microspheres                                                           | —                                                               | 143.0                                                                     | 6         |
| Carnation-like Ca <sub>4</sub> B <sub>10</sub> O <sub>19</sub> ·7H <sub>2</sub> O microspheres | 32.79                                                           | 256.4                                                                     | This work |
| Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub> @Zr(OH) <sub>x</sub> CSNs                     | 29.00                                                           | 283.8                                                                     | 53        |
| Hydroxyapatite/Fe <sub>3</sub> O <sub>4</sub> microspheres                                     | 59.40                                                           | 440.0                                                                     | 54        |
| Flower-like Ti(HPO <sub>4</sub> ) <sub>2</sub> ·H <sub>2</sub> O microstructures               | 122.00                                                          | 550.0                                                                     | 55        |

<sup>a</sup> S<sub>BET</sub>: specific surface area.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

