

Cite this: *Chem. Sci.*, 2018, 9, 7138

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Can percolation theory explain the gelation behavior of diblock copolymer worms?†

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It is well known that polymerization-induced self-assembly (PISA) offers an efficient synthetic route for the production of highly anisotropic diblock copolymer worms. When prepared in aqueous media, such worms form thermoresponsive free-standing hydrogels that are (i) readily sterilizable, (ii) can act as a 3D matrix for the culture of normal mammalian cells and (iii) can induce stasis in human stem cell colonies. Herein we critically examine the gelation behavior of two types of diblock copolymer worms in terms of recent advances in percolation theory for rigid rods, which explicitly account for the effect of rod length polydispersity. More specifically, we use small-angle X-ray scattering (SAXS) to determine the weight-average worm contour length, L_w , and the mean worm cross-sectional radius, R . This approach enables a direct comparison to be made between the theoretical critical worm volume fraction, ϕ_c , required for gelation and the experimental values indicated by rheological measurements and tube inversion experiments. Given that these diblock copolymer worms are relatively flexible rather than truly rod-like, reasonable agreement between these two parameters is observed, particularly for shorter, relatively stiff worms. For longer, more flexible worms a proportionality constant of approximately two is required to reconcile theory with experimental values for ϕ_c . These findings are expected to have important implications for the aqueous gelation behavior exhibited by various other anisotropic nanoparticles, such as cellulose nanocrystals and semicrystalline block copolymer rods, and also fibril-forming small molecule (e.g. dipeptide) gelators.

Received 1st June 2018
Accepted 25th July 2018DOI: 10.1039/c8sc02406e
rsc.li/chemical-science

Introduction

It is well known that certain surfactants (or binary mixtures thereof) can form highly anisotropic worms in aqueous solution.^{1–4} These systems have potential applications as thickeners,^{5,6} in drag reduction,⁷ and for enhanced oil recovery.^{4,8} The ‘living’ nature of these self-healing systems has been demonstrated and sophisticated techniques such as contrast variation neutron scattering have been utilized to characterize their structure.^{9,10} Surfactant worms typically exhibit mean contour lengths of the order of 1–10 μm . The concept of worm entanglements as a physical mechanism for gelation has been suggested on the basis of a combination of rheological and theoretical studies.^{10–12}

Diblock copolymer worm gels have been recognized for almost two decades.¹³ Over the last five years or so, the development of polymerization-induced self-assembly (PISA) has enabled the rational, reproducible synthesis of a wide range of diblock copolymer worm gels directly in water, polar solvents (e.g. ethanol) or non-polar solvents (e.g. *n*-alkanes).^{14–22} In particular, diblock copolymer worms prepared *via* dispersion polymerization often exhibit thermoresponsive gelation, undergoing a reversible worm-to-sphere morphological transition either on heating in ethanol or *n*-alkanes^{18,23,24} or on cooling in aqueous solution.^{20,25} In each case, this morphological transition appears to be the result of surface plasticization of the core-forming block, which leads to a subtle change in the packing parameter for the diblock copolymer chains.^{26,27} Typically, the mean worm width is well-defined, is of the order of a few tens of nm and is dictated by the mean degree of polymerization (DP) of the core-forming block. In contrast, the mean worm length is rather ill-defined and is typically of the order of hundreds of nm. Compared to the dimensions reported for surfactant worms, diblock copolymer worms appear to be too short to account for the observed formation of free-standing gels *via* a worm entanglement mechanism.

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† Electronic supplementary information (ESI) available: Full experimental details including synthesis and characterisation of diblock copolymers; GPC chromatograms; SIPLI images; digital images of tube inversion test; calculations for effective worm density, worm cross-sectional radius and volume fraction. See DOI: 10.1039/c8sc02406e



Percolation theory has been used for many years to account for the substantial differences in conductivity thresholds observed for many types of conductive particles dispersed in electrically insulating matrices.^{28–35} Typically, spheres exhibit a percolation threshold volume fraction of around 0.16,^{29,30} whereas highly anisotropic rods (*e.g.* polyaniline needles or carbon nanotubes) form fully-connected conductive networks at significantly lower volume fractions, sometimes below 0.01.^{31–35} Recently, percolation theory has been extended to include polydisperse rods exhibiting a wide range of rod lengths,^{36,37} which is often the case encountered experimentally. More specifically, for cylindrical rods with a high aspect ratio (*i.e.* length/width ratio), Chatterjee³⁶ has used mean field theory to show that the critical volume fraction for the percolation threshold, ϕ_c , can be estimated using eqn (1):

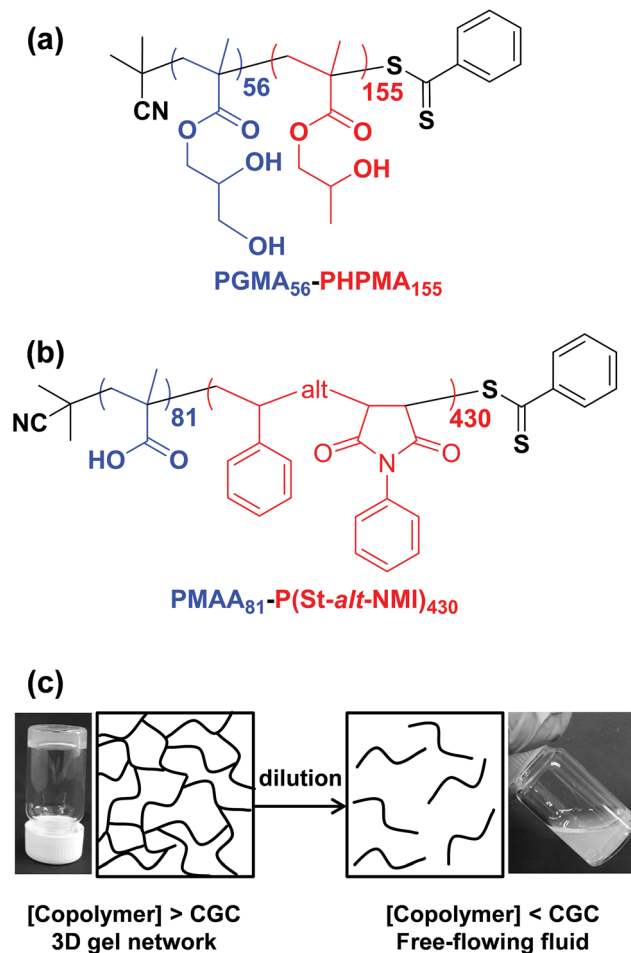
$$\phi_c \approx \left(\frac{R}{L_w}\right) \left(\frac{1 + \frac{\sigma_R^2}{R^2}}{1 + \frac{\sigma_R^2}{4R^2}}\right) \quad (1)$$

where L_w is the weight-average rod length, R is the number-average rod cross-sectional radius and σ_R is the standard deviation of the rod cross-sectional radius. As noted by Chatterjee, for populations of rods with narrow width polydispersities, relatively high aspect ratios, and uncorrelated variations in the widths and lengths, the percolation threshold is governed by the ratio of the number-average radius to the weight-average (rod) length. As noted above, the average worm cross-sectional radius R is well-defined, so σ_R tends to zero. Hence eqn (1) can be simplified to give:

$$\phi_c \approx \frac{R}{L_w} \quad (2)$$

[*N.B.* It can be shown that the approximation made when deriving eqn (2) leads to a small systematic underestimation of ϕ_c (see ESI†)]. Otten and co-workers drew similar conclusions to that of Chatterjee using a somewhat different mathematical approach.³⁷

We postulated that the percolation threshold required for the formation of an extended 3D network of inter-connected electrically conductive rods randomly dispersed in an insulating matrix to produce macroscopic electrical conductivity^{38,39} should be equivalent to that required for formation of a macroscopic physical gel by a colloidal dispersion of rods. Herein, we evaluate to what extent eqn (2) provides a useful description of the gelation behavior observed for two examples of diblock copolymer worms.^{36,37} For this approach to be valid, gelation should occur as a result of multiple inter-worm contacts (see Scheme 1), which would provide an alternative gelation mechanism to the inter-worm entanglements model previously (and correctly) invoked for surfactant worms. The two diblock copolymer systems studied herein were chosen because they represent relatively long, highly flexible worms²⁵ and relatively short, stiff worms, respectively.⁴⁰ Thus they represent two limiting copolymer morphologies for which contrasting experimental data might be anticipated.



Scheme 1 Chemical structures for (a) poly(glycerol mono-methacrylate)-*block*-poly(2-hydroxypropyl methacrylate) (PGMA–PHPMA) diblock copolymers prepared by RAFT aqueous dispersion polymerization and (b) poly(methacrylic acid)-*block*-poly(styrene-*alt*-*N*-phenylmaleimide) copolymers prepared by RAFT dispersion polymerization in a 50/50% w/w ethanol/1,4-dioxane mixture. (c) Schematic cartoon illustrating formation of a continuous 3D network of worms above the critical gelation concentration (CGC) owing to multiple inter-worm contacts. In contrast, these inter-worm contacts are broken on dilution below the CGC, resulting in a free-flowing dispersion rather than a gel.

Results and discussion

Initially, we sought literature data for colloidal dispersions of rigid rods to support our hypothesis. Recently, Nordenström *et al.* reported an interesting study of the aqueous gelation behavior of various cellulosic nanorods of varying dimensions and surface charge.⁴¹ More specifically, a series of six cellulose nanorods were prepared with varying length/diameter ratios (or aspect ratios) and their (de)gelation behavior was characterized using dynamic light scattering (DLS). The critical volume fraction for gelation was shown to be inversely proportional to the aspect ratio. Furthermore, it was postulated that gelation was simply a result of multiple contacts with neighboring nanorods, which arrests their translational diffusion in solution. However, no specific link was made to the recent mathematical advances



developed to describe the percolation behavior of polydisperse rods.^{36,37} Of the six types of cellulose nanorods reported by Nordenström *et al.*,⁴¹ the most relevant to the present study of neutral worms is that with the lowest surface charge, which had a mean length of 520 nm (determined by DLS) and a mean radius of 3.35 nm (measured by AFM studies). Using eqn (2), we calculate the theoretical percolation volume fraction, ϕ_c , for such cellulosic nanorods to be 0.0064, which is in reasonably good agreement with the experimental ϕ_c of 0.0073 reported by Nordenström and co-workers.⁴¹ Thus our hypothesis of physical equivalence between the respective critical percolation thresholds required for solid-state electrical conductivity and physical gelation appears to have some merit.

The poly(glycerol monomethacrylate)₅₆-poly(2-hydroxypropyl methacrylate)₁₅₅ [PGMA₅₆-PHPMA₁₅₅] and poly(methacrylic acid)₈₁-poly(styrene-*alt*-*N*-phenylmaleimide)₄₃₀ [PMAA₈₁-P(St-*alt*-NMI)₄₃₀] worm gels evaluated in this study were prepared using PISA as described by Blanazs *et al.*²² and Yang and co-workers⁴⁰ respectively (see Scheme 1 for the relevant chemical structures). More specifically, the highly flexible PGMA₅₆-PHPMA₁₅₅ worms were synthesized *via* reversible addition-fragmentation chain transfer (RAFT) aqueous dispersion polymerization of 2-hydroxypropyl methacrylate (HPMA), and are clearly highly anisotropic as judged by transmission electron microscopy (TEM, see Fig. 1a). In contrast, the relatively short, stiff PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms were prepared by RAFT dispersion alternating copolymerization of styrene with *N*-phenylmaleimide using a 1 : 1 ethanol/1,4-dioxane mixture. These latter worms are much less anisotropic (see Fig. 1b). In both cases, the diblock copolymer chains possess relatively narrow molecular weight distributions as determined by gel permeation chromatography (GPC) and comparison to their respective macro-CTAs indicates high blocking efficiencies (see Fig. S1†). The mean aspect ratio (*i.e.* length/width ratio) for each type of worm can be determined using small-angle X-ray scattering (SAXS), as described below.⁴² TEM analysis confirms that the worm cross-sectional radius is well-defined in both cases. More specifically, the mean core radius, r_c , for PGMA₅₆-PHPMA₁₅₅ and PMAA₈₁-P(St-*alt*-NMI)₄₃₀ is estimated to be 11.1 ± 1.3 and 19.2 ± 2.1 nm, respectively.

In contrast, the worm contour length, L_w , is clearly rather ill-defined. This is because such worms are formed *via* stochastic

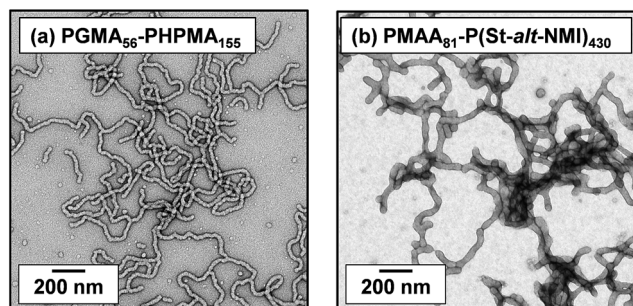


Fig. 1 TEM images obtained for dry dispersions of (a) PGMA₅₆-PHPMA₁₅₅ and (b) PMAA₈₁-P(St-*alt*-NMI)₄₃₀ diblock copolymer worms prepared by PISA.

1D fusion of multiple spheres during PISA.⁴⁴ In principle, SAXS is a powerful technique for characterizing block copolymer nano-objects, not least because X-ray scattering is averaged over many millions of particles and hence much more statistically robust than TEM studies.⁴⁵ Accordingly, SAXS patterns were recorded for the two worm dispersions at 1.0% w/w copolymer concentration, see Fig. 2. Both SAXS patterns exhibit a gradient of approximately -1 at low q , which is indicative of highly anisotropic rods (or worms). Fitting such patterns using an established worm model⁴⁶ provides detailed and robust structural information, including the weight-average L_w and cross-sectional worm radius R . For example, SAXS indicates an L_w of approximately 1100 nm for PGMA₅₆-PHPMA₁₅₅ worms, with a corresponding core radius, r_c , of 8.5 ± 0.9 nm (see Fig. 2a). However, the highly hydrated stabilizer chains also contribute to the overall effective worm dimensions. Given the mean DP of the PGMA₅₆ chains, the thickness of this additional stabilizer layer is estimated to be 3.6 nm by SAXS analysis. Thus the

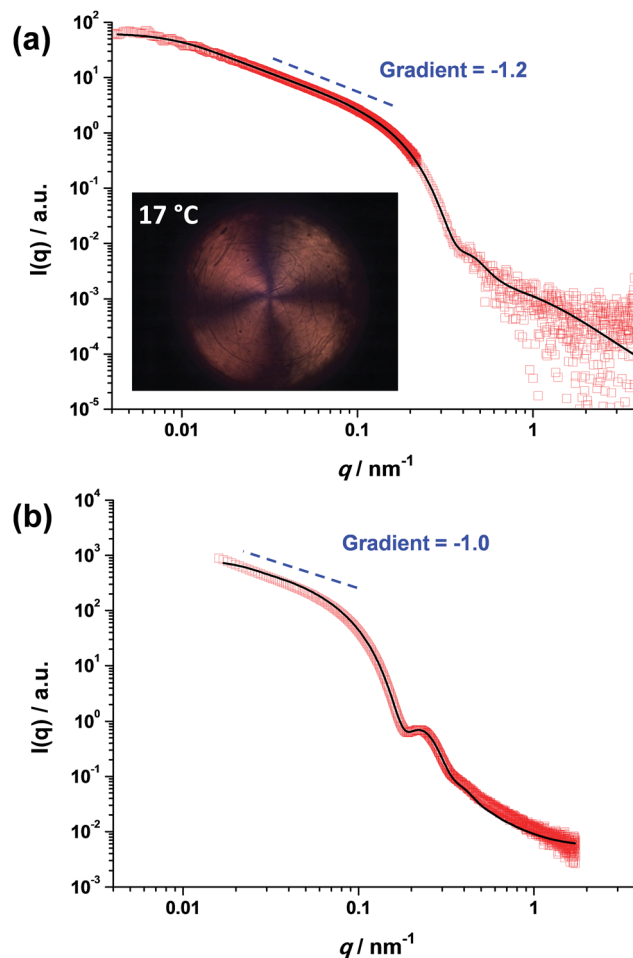


Fig. 2 SAXS patterns recorded for 1.0% w/w dispersions of (a) PGMA₅₆-PHPMA₁₅₅ worms at 18 °C and (b) PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms at 20 °C. Inset in (a) shows a shear-induced polarized light image of the corresponding 5.0% w/w worm dispersion obtained at a maximum shear rate of 20 s⁻¹. The Maltese cross observed is the distinctive signature for birefringence, indicating *in situ* worm alignment.⁴³



effective worm cross-sectional radius, R , for these 'hairy' PGMA₅₆-PHPMA₁₅₅ worms is calculated to be 12.1 ± 0.9 nm (see ESI† for calculation details). In contrast, SAXS analysis of the PMAA₈₁-P(St-*alt*-NMI)₄₃₀ diblock copolymer worms suggests an L_w of approximately 296 nm (see Fig. 2b). These latter worms have an r_c value of 20.0 ± 2.7 nm and a stabilizer thickness of 6.6 nm, giving an overall R value of 26.6 ± 3.0 nm. Hence the mean aspect ratios (or L_w/R values) for the PGMA₅₆-PHPMA₁₅₅ and PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms are 89 and 11, respectively. These strikingly different aspect ratios are useful in the context of the present study because they enable a more rigorous test of the percolation theory recently developed for polydisperse rods.^{36,37} Thus, according to eqn (2), the critical percolation volume fraction, ϕ_c , required to form a 3D gel network comprising PGMA₅₆-PHPMA₁₅₅ worms is expected to be significantly lower than that required for gelation when using the PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms.

It is well-known that semi-concentrated dispersions of such PGMA-PPHMA worm gels exhibit thermoresponsive behavior, with degelation occurring on cooling below the critical gelation temperature (CGT) as a result of a worm-to-sphere transition.^{20,25} When applying percolation theory to such thermosensitive systems, it is important to determine the characteristic temperature that corresponds to long, linear worms (as opposed to branched worms or worm clusters). This is readily achieved using shear-induced polarized light imaging (SIPLI), as recently reported by Mykhaylyk and co-workers.⁴³ Briefly, an aqueous worm dispersion is subjected to applied shear using an optorheometer, which enables simultaneous interrogation of the sample using polarized light. The appearance of a distinctive Maltese cross motif indicates shear-induced alignment of the highly anisotropic worms. If such experiments are performed as a function of temperature, the temperature at which the brightest Maltese cross is observed corresponds to the formation of the most linear (*i.e.* longest) worms. Such measurements are shown in the inset of Fig. 2a and S2† and indicate an optimum temperature of 17 °C, which is very close to that at which the SAXS studies were performed (18 °C). It is noteworthy that the PMAA₈₁-P(St-*alt*-NMI)₄₃₀ diblock copolymer worms do not exhibit such thermoresponsive behavior, so the temperature at which SAXS analysis is conducted is not particularly important in this case.

Utilizing the structural information provided by SAXS in combination with eqn (2), the theoretical critical volume fraction (ϕ_c) required for the percolation threshold (and hence macroscopic gelation) is predicted to be 0.011 ± 0.001 and 0.090 ± 0.009 for the PGMA₅₆-PHPMA₁₅₅ and PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms, respectively. This approximate eight-fold difference simply reflects the substantial difference in aspect ratio for these two types of worms.

Experimental ϕ_c values can be estimated from tube inversion tests, which were performed at ambient temperature (17–18 °C) for varying copolymer volume fractions (see Fig. S3† and related calculations). These observations indicated ϕ_c values of approximately 0.025 ± 0.002 and 0.121 ± 0.004 for the PGMA₅₆-PHPMA₁₅₅ and PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms, respectively. Very similar ϕ_c values (0.025 ± 0.002 and 0.113 ± 0.004) were

obtained from oscillatory rheology, see Fig. 3. In this case, degelation is indicated by the point of intersection of the storage modulus (G') and the loss modulus (G'') curves, and these latter experiments are considered more reliable.

In polymer physics, the worm-like chain model is used to describe the behavior of semi-flexible polymers.⁴⁷ For such worm-like chains, the Kuhn length is equal to twice the persistence length, where the latter parameter quantifies the chain stiffness. In principle, the behavior of the long, flexible diblock copolymer worms described in our study is analogous to that of an individual polymer chain.⁴⁸ The Kuhn lengths derived from SAXS studies of both types of worms are included in Table 1. The Kuhn length for the short, stiff worms is simply equal to the weight-average worm contour length (L_w). In contrast, the Kuhn length for the long, flexible worms is much lower than L_w , which implies significant flexibility.

Clearly, there is some discrepancy between the theoretical and experimental ϕ_c values summarized in Table 1. However, percolation theory is derived assuming rigid rods, whereas the PGMA₅₆-PHPMA₁₅₅ worms clearly exhibit significant flexibility (see TEM images in Fig. 1). This necessarily reduces the effective weight-average worm length L_w , which in turn leads to a higher ϕ_c value. Given this important caveat, the fair agreement

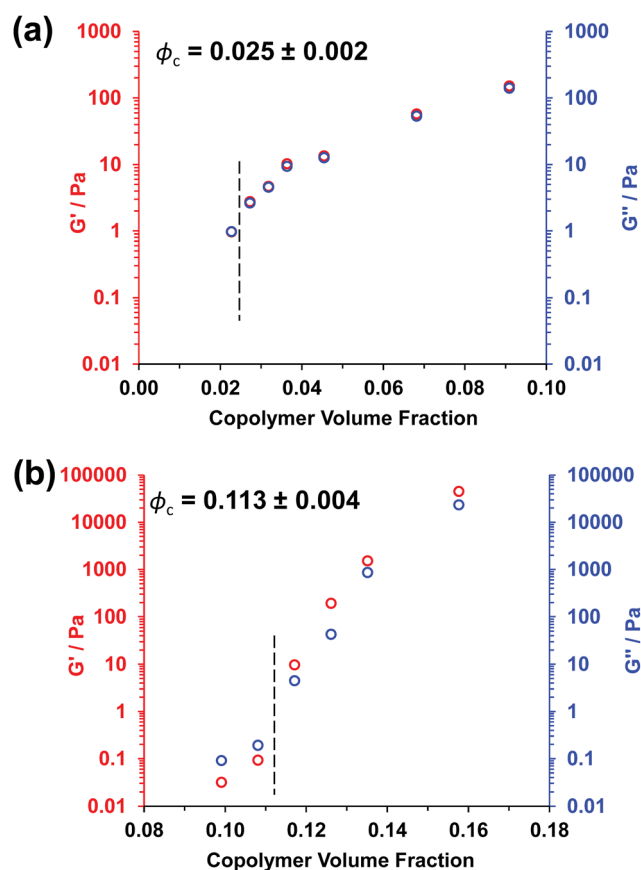


Fig. 3 Oscillatory rheology data obtained for (a) PGMA₅₆-PHPMA₁₅₅ worms at 17 °C and (b) PMAA₈₁-P(St-*alt*-NMI)₄₃₀ worms at 20 °C at varying copolymer volume fractions. The critical gelation volume fraction (or ϕ_c) is determined by the point of intersection of the loss modulus (G'') and storage modulus (G') curves.



Table 1 Summary of weight-average worm length, cylindrical cross-sectional radius, aspect ratio, Kuhn length, theoretical and experimental volume fractions calculated for relatively flexible PGMA₅₆–PHPMA₁₅₅ worms and relatively stiff PMAA₈₁–P(St-*alt*-NMI)₄₃₀ worms

Anisotropic nanoparticle type	Weight-average worm length, L_w /nm	Worm cross-sectional radius, R /nm	Aspect ratio (L_w/R)	Kuhn length/nm	ϕ_c , theory ^a	ϕ_c , experiment ^b
PGMA ₅₆ –PHPMA ₁₅₅ worms	1100	12.3	89	300	0.011	0.025
PMAA ₈₁ –P(St- <i>alt</i> -NMI) ₄₃₀ worms	296	26.6	11	296	0.090	0.113

^a Calculated using eqn (2). ^b Determined by oscillatory rheology.

observed between the experimental and theoretical ϕ_c values supports our hypothesis that such worm gels form a 3D network simply *via* multiple contacts between neighbouring worms. In contrast, the PMAA₈₁–P(St-*alt*-NMI)₄₃₀ worms are much stiffer (the glass transition temperature for the core-forming P(St-*alt*-NMI)₄₃₀ block is around 208 °C.⁴⁰ Thus, better agreement between experimental and theoretical ϕ_c values is expected, and indeed observed.

From Table 1, the theoretical ϕ_c for stiff worms is approximately eight times greater than that for flexible worms. In contrast, the corresponding experimental ϕ_c ratio is approximately four. This suggests that eqn (2) can be rewritten as:

$$\phi_c \approx k \frac{R}{L_w} \quad (3)$$

where the proportionality constant k varies by at least a factor of two depending on the degree of worm flexibility. It is perhaps also worth emphasizing here that, given the worm dimensions indicated by SAXS studies, the ‘worm entanglements’ mechanism invoked to account for the gelation of surfactant worms does not appear to account satisfactorily for the physical gelation observed for these much less anisotropic diblock copolymer worms.

Conclusions

In summary, recent advances in percolation theory for poly-disperse rods provide an improved understanding of the gelation behavior exhibited by diblock copolymer worms. Combined with experimental data, this suggests that a 3D gel network forms primarily *via* multiple contacts between neighbouring worms, rather than as a result of worm entanglements. In view of the growing number of studies utilizing worm-based hydrogels for various biomedical applications,^{25,49–58} this new physical insight is likely to be important for the design of next-generation diblock copolymer worm gels, as well as the growing literature on block copolymer rods.^{58–62} Indeed, it seems likely that our findings are also relevant to the growing literature on supramolecular gels composed of amphiphilic small molecules^{63–71} as well as hydrogels based on cellulose and silica nanorods.^{41,72–77}

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

S. P. A. is the recipient of a five-year ERC Advanced Investigator grant (PISA 320372). EPSRC is thanked for a Platform grant (EP/J007846/1) and a Particle Technology Fellowship grant (EP/R003009/1). The Leverhulme Trust is thanked for post-doctoral funding of M. J. D. (RPG-2016-330).

Notes and references

- M. E. Cates and S. J. Candau, *J. Phys.: Condens. Matter*, 1990, **2**, 6869–6892.
- F. C. MacKintosh, S. A. Safran and P. A. Pincus, *Europhys. Lett.*, 1990, **12**, 697–702.
- G. Porte, J. Appell and Y. Poggi, *J. Phys. Chem.*, 1980, **84**, 3105–3110.
- Z. Chu, C. A. Dreiss and Y. Feng, *Chem. Soc. Rev.*, 2013, **42**, 7174–7203.
- T. S. Davies, A. M. Ketner and S. R. Raghavan, *J. Am. Chem. Soc.*, 2006, **128**, 6669–6675.
- B. Chase, W. Chmiliowski, R. Marcinew, C. Mitchell, Y. Dang, K. Krauss, E. Nelson, T. Lantz, C. Parham and J. Plummer, *Oilfield Rev.*, 1997, **9**, 20–33.
- J. L. Zakin, B. Lu and H.-W. Bewersdorff, *Rev. Chem. Eng.*, 1998, **14**, 253–320.
- J. Yang, *Curr. Opin. Colloid Interface Sci.*, 2002, **7**, 276–281.
- N. A. Spensley, M. E. Cates and T. C. B. McLeish, *Phys. Rev. Lett.*, 1993, **71**, 939–942.
- C. A. Dreiss, *Soft Matter*, 2007, **3**, 956–970.
- H. Rehage and H. Hoffmann, *J. Phys. Chem.*, 1988, **92**, 4712–4719.
- M. E. Cates, *Macromolecules*, 1987, **20**, 2289–2296.
- Y.-Y. Won, H. T. Davis and F. S. Bates, *Science*, 1999, **283**, 960–963.
- X. Zhang, S. Boissé, W. Zhang, P. Beaunier, F. D’Agosto, J. Rieger and B. Charleux, *Macromolecules*, 2011, **44**, 4149–4158.
- Y. W. Pei and A. B. Lowe, *Polym. Chem.*, 2014, **5**, 2342–2351.
- Y. W. Pei, N. C. Dharsana, J. A. Van Hensbergen, R. P. Burford, P. J. Roth and A. B. Lowe, *Soft Matter*, 2014, **10**, 5787–5796.
- M. Semsarilar, E. R. Jones, A. Blanazs and S. P. Armes, *Adv. Mater.*, 2012, **24**, 3378–3382.
- L. A. Fielding, J. A. Lane, M. J. Derry, O. O. Mykhaylyk and S. P. Armes, *J. Am. Chem. Soc.*, 2014, **136**, 5790–5798.



- 19 L. A. Fielding, M. J. Derry, V. Ladmiraal, J. Rosselgong, A. M. Rodrigues, L. P. D. Ratcliffe, S. Sugihara and S. P. Armes, *Chem. Sci.*, 2013, **4**, 2081–2087.
- 20 R. Verber, A. Blanazs and S. P. Armes, *Soft Matter*, 2012, **8**, 9915–9922.
- 21 S. Boisse, J. Rieger, K. Belal, A. Di-Cicco, P. Beaunier, M.-H. Li and B. Charleux, *Chem. Commun.*, 2010, **46**, 1950–1952.
- 22 A. Blanazs, A. J. Ryan and S. P. Armes, *Macromolecules*, 2012, **45**, 5099–5107.
- 23 Y. Pei, O. R. Sugita, L. Thurairajah and A. B. Lowe, *RSC Adv.*, 2015, **5**, 17636–17646.
- 24 Y. Pei, L. Thurairajah, O. R. Sugita and A. B. Lowe, *Macromolecules*, 2015, **48**, 236–244.
- 25 A. Blanazs, R. Verber, O. O. Mykhaylyk, A. J. Ryan, J. Z. Heath, C. W. I. Douglas and S. P. Armes, *J. Am. Chem. Soc.*, 2012, **134**, 9741–9748.
- 26 J. N. Israelachvili, D. J. Mitchell and B. W. Ninham, *J. Chem. Soc., Faraday Trans. 2*, 1976, **72**, 1525–1568.
- 27 A. Blanazs, S. P. Armes and A. J. Ryan, *Macromol. Rapid Commun.*, 2009, **30**, 267–277.
- 28 F. Lux, *J. Mater. Sci.*, 1993, **28**, 285–301.
- 29 S. Hotta, S. D. D. V. Rughooputh and A. J. Heeger, *Synth. Met.*, 1987, **22**, 79–87.
- 30 B. D. Malhotra, S. Ghosh and R. Chandra, *J. Appl. Polym. Sci.*, 1990, **40**, 1049–1052.
- 31 E. C. Cooper and B. Vincent, *J. Phys. D: Appl. Phys.*, 1989, **22**, 1580–1585.
- 32 P. Banerjee and B. M. Mandal, *Synth. Met.*, 1995, **74**, 257–261.
- 33 C. Y. Yang, Y. Cao, P. Smith and A. J. Heeger, *Synth. Met.*, 1993, **53**, 293–301.
- 34 J. Sandler, M. S. P. Shaffer, T. Prasse, W. Bauhofer, K. Schulte and A. H. Windle, *Polymer*, 1999, **40**, 5967–5971.
- 35 J. K. W. Sandler, J. E. Kirk, I. A. Kinloch, M. S. P. Shaffer and A. H. Windle, *Polymer*, 2003, **44**, 5893–5899.
- 36 A. P. Chatterjee, *J. Chem. Phys.*, 2010, **132**, 224905.
- 37 R. H. J. Otten and d. S. P. van, *J. Chem. Phys.*, 2011, **134**, 094902.
- 38 T. Komori and K. Makishima, *Text. Res. J.*, 1977, **47**, 13–17.
- 39 V. Kumar and A. Rawal, *Polymer*, 2016, **97**, 295–299.
- 40 P. Yang, L. P. D. Ratcliffe and S. P. Armes, *Macromolecules*, 2013, **46**, 8545–8556.
- 41 M. Nordenström, A. Fall, G. Nyström and L. Wågberg, *Langmuir*, 2017, **33**, 9772–9780.
- 42 Strictly speaking, SAXS reports the volume-average dimensions of nano-objects. Hence the volume-average worm length L_v is directly related to the weight-average worm length L_w by the worm density. Moreover, given that the worm cross-sectional radius R is narrowly distributed, the volume-average (and weight-average) radius lies close to the number-average value.
- 43 O. O. Mykhaylyk, N. J. Warren, A. J. Parnell, G. Pfeifer and J. Laeuger, *J. Polym. Sci., Part B: Polym. Phys.*, 2016, **54**, 2151–2170.
- 44 N. J. Warren and S. P. Armes, *J. Am. Chem. Soc.*, 2014, **136**, 10174–10185.
- 45 H. Schnablegger and Y. Singh, *The SAXS Guide*, Anton Paar GmbH, Austria, 2011.
- 46 J. S. Pedersen, *J. Appl. Crystallogr.*, 2000, **33**, 637–640.
- 47 M. Doi and S. F. Edwards, *The Theory of Polymer Dynamics*, Clarendon Press, Oxford, 1988.
- 48 W. J. Zhang, B. Charleux and P. Cassagnau, *Soft Matter*, 2013, **9**, 2197–2205.
- 49 K. A. Simon, N. J. Warren, B. Mosadegh, M. R. Mohammady, G. M. Whitesides and S. P. Armes, *Biomacromolecules*, 2015, **16**, 3952–3958.
- 50 I. Canton, N. J. Warren, A. Chahal, K. Amps, A. Wood, R. Weightman, E. Wang, H. Moore and S. P. Armes, *ACS Cent. Sci.*, 2016, **2**, 65–74.
- 51 D. E. Mitchell, J. R. Lovett, S. P. Armes and M. I. Gibson, *Angew. Chem., Int. Ed.*, 2016, **55**, 2801–2804.
- 52 Y. Li, N. Khuu, A. Gevorkian, S. Sarjinsky, H. Therien-Aubin, Y. Wang, S. Cho and E. Kumacheva, *Angew. Chem.*, 2017, **56**, 6083–6087.
- 53 S. Cho, Y. Li, M. Seo and E. Kumacheva, *Angew. Chem., Int. Ed.*, 2016, **55**, 14014–14018.
- 54 D. Velasco, M. Chau, H. Therien-Aubin, A. Kumachev, E. Tumarkin, Z. Jia, G. C. Walker, M. J. Monteiro and E. Kumacheva, *Soft Matter*, 2013, **9**, 2380–2383.
- 55 H. Thérien-Aubin, Y. Wang, K. Nothdurft, E. Prince, S. Cho and E. Kumacheva, *Biomacromolecules*, 2016, **17**, 3244–3251.
- 56 K. Younghoon, D. Paul, A. C. David and E. D. Dennis, *Nanotechnology*, 2005, **16**, 484–491.
- 57 P. Dalhaimer, F. S. Bates and D. E. Discher, *Macromolecules*, 2003, **36**, 6873–6877.
- 58 P. H. J. Kouwer, M. Koepf, V. A. A. Le Sage, M. Jaspers, A. M. van Buul, Z. H. Eksteen-Akeroyd, T. Woltinge, E. Schwartz, H. J. Kitto, R. Hoogenboom, S. J. Picken, R. J. M. Nolte, E. Mendes and A. E. Rowan, *Nature*, 2013, **493**, 651–655.
- 59 J. B. Gilroy, T. Gadt, G. R. Whittell, L. Chabanne, J. M. Mitchels, R. M. Richardson, M. A. Winnik and I. Manners, *Nat. Chem.*, 2010, **2**, 566–570.
- 60 J. A. Massey, K. Temple, L. Cao, Y. Rharbi, J. Raez, M. A. Winnik and I. Manners, *J. Am. Chem. Soc.*, 2000, **122**, 11577–11584.
- 61 S. K. Patra, R. Ahmed, G. R. Whittell, D. J. Lunn, E. L. Dunphy, M. A. Winnik and I. Manners, *J. Am. Chem. Soc.*, 2011, **133**, 8842–8845.
- 62 P. A. Rugar, L. Chabanne, M. A. Winnik and I. Manners, *Science*, 2012, **337**, 559–562.
- 63 L. E. Buerkle and S. J. Rowan, *Chem. Soc. Rev.*, 2012, **41**, 6089–6102.
- 64 D. J. Adams and P. D. Topham, *Soft Matter*, 2010, **6**, 3707–3721.
- 65 R. J. Wojtecki, M. A. Meador and S. J. Rowan, *Nat. Mater.*, 2011, **10**, 14–27.
- 66 B. O. Okesola and D. K. Smith, *Chem. Soc. Rev.*, 2016, **45**, 4226–4251.
- 67 A. R. Hirst, I. A. Coates, T. R. Boucheteau, J. F. Miravet, B. Escuder, V. Castelletto, I. W. Hamley and D. K. Smith, *J. Am. Chem. Soc.*, 2008, **130**, 9113–9121.



- 68 A. R. Hirst, B. Escuder, J. F. Miravet and D. K. Smith, *Angew. Chem., Int. Ed.*, 2008, **47**, 8002–8018.
- 69 M. Zelzer and R. V. Ulijn, *Chem. Soc. Rev.*, 2010, **39**, 3351–3357.
- 70 S. Fleming and R. V. Ulijn, *Chem. Soc. Rev.*, 2014, **43**, 8150–8177.
- 71 V. Jayawarna, M. Ali, T. A. Jowitt, A. F. Miller, A. Saiani, J. E. Gough and R. V. Ulijn, *Adv. Mater.*, 2006, **18**, 611–614.
- 72 K. R. Peddireddy, I. Capron, T. Nicolai and L. Benyahia, *Biomacromolecules*, 2016, **17**, 3298–3304.
- 73 M. C. Li, Q. L. Wu, K. L. Song, S. Lee, Y. Qing and Y. Q. Wu, *ACS Sustainable Chem. Eng.*, 2015, **3**, 821–832.
- 74 A. E. Way, L. Hsu, K. Shanmuganathan, C. Weder and S. J. Rowan, *ACS Macro Lett.*, 2012, **1**, 1001–1006.
- 75 T. Saito, T. Uematsu, S. Kimura, T. Enomae and A. Isogai, *Soft Matter*, 2011, **7**, 8804–8809.
- 76 C. Zhou, Q. Wu, Y. Yue and Q. Zhang, *J. Colloid Interface Sci.*, 2011, **353**, 116–123.
- 77 R. P. Murphy, K. Hong and N. J. Wagner, *Langmuir*, 2016, **32**, 8424–8435.

