


 Cite this: *RSC Adv.*, 2018, 8, 20275

DOI: 10.1039/c8ra90046a

[www.rsc.org/advances](http://www.rsc.org/advances)

## Correction: Ascorbic acid/Fe<sup>0</sup> composites as an effective persulfate activator for improving the degradation of rhodamine B

 Xiangyu Wang,<sup>\*a</sup> Yi Du,<sup>a</sup> Huiling Liu<sup>b</sup> and Jun Ma<sup>b</sup>

 Correction for 'Ascorbic acid/Fe<sup>0</sup> composites as an effective persulfate activator for improving the degradation of rhodamine B' by Xiangyu Wang *et al.*, *RSC Adv.*, 2018, 8, 12791–12798.

The authors regret that the unit on the *x*-axis of Fig. 1 was incorrectly written as “% wt” rather than “‰ wt” in the original article. The correct version of Fig. 1 is presented below.

<sup>a</sup>Faculty of Environmental Science and Engineering, Kunming University of Science and Technology, Kunming 650500, P. R. China. E-mail: [imusthlee2014@sina.com](mailto:imusthlee2014@sina.com)

<sup>b</sup>School of Municipal and Environmental Engineering, State Key Laboratory of Urban Water Resources and Environment, Harbin Institute of Technology, Harbin 150090, P. R. China



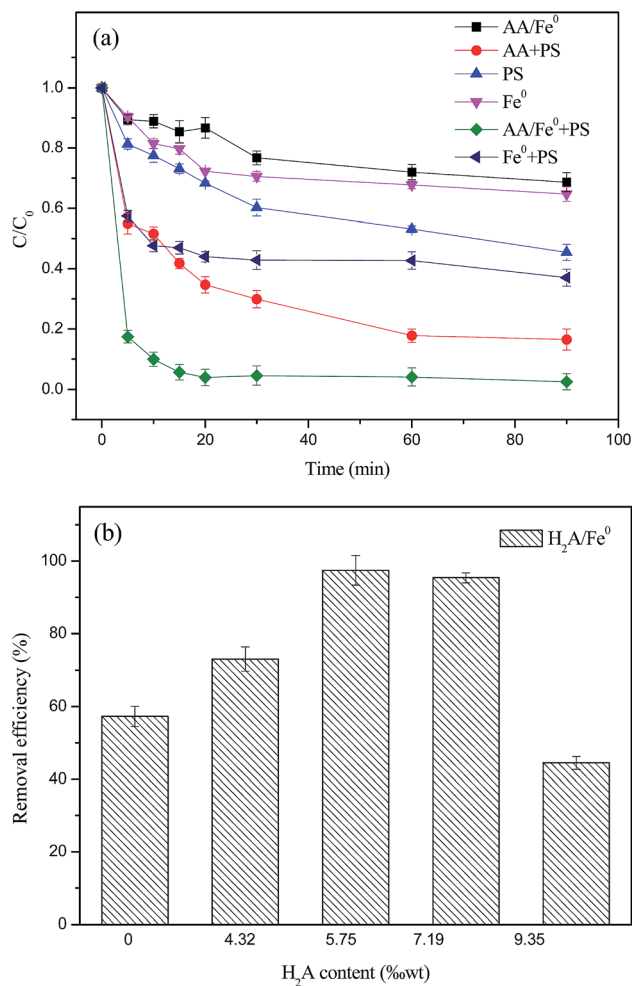


Fig. 1 (a) Comparison of removal efficiency of RhB in different systems ( $C_0 = 50 \text{ mg L}^{-1}$ , PS dosage =  $1.4 \text{ g L}^{-1}$ ,  $\text{Fe}^0$  dosage =  $1 \text{ g L}^{-1}$ ,  $\text{H}_2\text{A}/\text{Fe}^0$  dosage =  $1 \text{ g L}^{-1}$ ,  $\text{H}_2\text{A}$  dosage =  $1.6 \text{ g L}^{-1}$  and  $T = 298 \text{ K}$ ); (b) effect of  $\text{H}_2\text{A}$  concentration on removal efficiency of RhB in the  $\text{H}_2\text{A}/\text{Fe}^0$ -PS system ( $C_0 = 50 \text{ mg L}^{-1}$ ,  $\text{Fe}^0$  dosage =  $0.8 \text{ g L}^{-1}$ ,  $T = 298 \text{ K}$  and the solution volume is  $50 \text{ mL}$ ).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

