RSC Advances



View Article Online

View Journal | View Issue

PAPER

Check for updates

Cite this: RSC Adv., 2018, 8, 40554

Received 11th October 2018 Accepted 19th November 2018

DOI: 10.1039/c8ra08425d

rsc.li/rsc-advances

1 Introduction

Fluorescence imaging plays an important role in bio-life science because of the emission of the probes after excitation can be visualized by the naked eye or at high resolution with optical microscopy.^{1,2} An ideal fluorescent probe for bio-imaging should be bright, biocompatible, water-soluble, and stable against photo bleaching.³ It is well known that traditional organic dyes are favored for routine bio-imaging and especially for applications requiring very accurate quantification.^{4,5} Nevertheless, the problem of hydrophobicity has severely restrained their wide spread applications to biological systems.⁶ For example, hexa-*peri*-hexabenzocoronene (HBC) and its derivatives, as nanographene segments, have attracted considerable attention because of their applications in field-effect transistors and photoconductive devices.^{7–9} However, their

Facile fabrication of water-dispersible nanocomposites based on hexa-*peri*hexabenzocoronene and Fe₃O₄ for dual mode imaging (fluorescent/MR) and drug delivery

Meng-Meng Song, ^b^{ab} Hui-Hui Xiang,^{ac} Meng-Yu Fei,^{ac} Da-Peng Lu,^d Tong-Cui Jiang,^{ab} Yong-Qiang Yu,^c Rui Liu^{*a} and Yu-Xian Shen^{*ab}

The facile fabrication of multifunctional nanocomposites (Fe₃O₄/HBC@F127) consisting of superparamagnetic Fe₃O₄ nanoparticles and fluorescent organic hexa-*peri*-hexabenzocoronene (HBC) molecules incorporated in block copolymer diacylphospholipid–polyethyleneglycol F127 have been demonstrated for dual mode imaging (fluorescent/MR) and drug delivery. The obtained nanocomposites were water-dispersible, stable and biocompatible, as confirmed by dynamic light scattering (DLS) and 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay. Relativity measurements showed a T_2 relaxivity (r_2) of 214.61 mM⁻¹ s⁻¹, which may be used as T_2 -weighted MR imaging agents. *In vitro* imaging studies indicated that the nanocomposites had good MR and fluorescence imaging effects with low cytotoxicity. Besides, the developed nanocomposites significantly inhibited the growth of human hepatoma cells (HepG2). These findings suggested that the facile synthesized multifunctional nanocomposites may be used as a platform for dual mode imaging (both MR and fluorescence) and drug delivery.

applications in biosystems are limited. Multicolor graphene quantum dots (GQDs) with a uniform size of 60 nm diameter and 2–3 nm thickness were synthesized by using a substituted HBC as the carbon source for bio-imaging.¹⁰ Yin *et al.* reported a novel water-soluble HBC derivative with negatively charged side arms that could self-assemble into micro- and nanofibers for bioprobing.¹¹ Considering the planar structure of HBC, it could aggregate and self-assemble into different morphologies in aqueous solution.^{12,13}

Although fluorescence imaging shows high sensitivity at subcellular levels and is suitable for quantifying, it still has limitations in examining deep tissue *in vivo*.¹⁴ Therefore, the combined nanoprobe, for example, a nanoprobe both for fluorescence imaging and MR imaging have great significance due to their complementary capabilities.¹⁵ Among various contrast agents, superparamagnetic Fe₃O₄ nanoparticles have been widely developed as negative contrast agents that can shorten the T_2 relaxation time of water protons, resulting in enhanced imaging contrast and sensitivity.^{16,17}

One important issue related to the applications of HBC molecules and Fe_3O_4 nanoparticles for fluorescence imaging and T_2 weighted MR imaging is how to render the hydrophobic molecules and Fe_3O_4 particles with good water-dispersity, biocompatibility and long blood circulation time. The formation of non-cytotoxic copolymer micelles has been proved to be

^aSchool of Basic Medical Sciences, Anhui Medical University, 81 Meishan Road, Hefei, Anhui, PR China. E-mail: liurui@ahmu.edu.cn; shenyx@ahmu.edu.cn; Tel: +86-551-65161138; +86-551-65113776

^bBiopharmaceutical Research Institute, Anhui Medical University, 81 Meishan Road, Hefei, Anhui, PR China

^cThe First Affiliated Hospital, Anhui Medical University, 218 Jixi Road, Hefei, Anhui, PR China

^dSchool of Pharmacy, Anhui Medical University, 81 Meishan Road, Hefei, Anhui, PR China

Paper

an effective strategy to make the particles meet the above requirements. Polyethylene glycol (PEG) based Pluronic triblock co-polymers are amphiphilic and composed of a hydrophobic central segment of poly(propylene oxide) (PPO) and two hydrophilic segments of poly(ethylene oxide) (PEO), which are widely used as a coating/stabilizing agent for the surface modification of nanoparticles for various biomedical applications.18-21 The amphiphilic block copolymer diacylphospholipid-polyethyleneglycol (F127) can conveniently encapsulate hydrophobic molecules with simple steps and the obtained F127-coated nanocomposites is a kind of ideal drug delivery system with good biocompatibility and prolonged blood circulation time. Hydrophobic drug can also be loaded by physical encapsulation due to its self-assembly characteristic.²²⁻²⁴ Take these factors into consideration, it is believed that the combination of HBC, Fe₃O₄ nanoparticles and polymer could be a new method for the design of multifunctional nanocomposites for dual mode imaging (fluorescent/MR) and drug delivery.

Herein, we developed a novel water-dispersible $Fe_3O_4/HBC@F127$ nanocomposite by simply incorporating two hydrophobic functional particles (HBC molecules and Fe_3O_4 nanoparticles) in a biodegradable polymeric micelle of F127. The formed particle showed good MR and fluorescence imaging effects with low cytotoxicity. It also can be applied as a drug delivery vehicle by loading Doxorubicin (DOX) to inhibit the growth of HepG2 cells. The Fe₃O₄/HBC@F127 multifunctional nanocomposites could be a potential platform for dual mode imaging (fluorescent/MR) and drug delivery.

2 Experimental

2.1 Materials

Pluronic F127 was purchased from Aldrich (St. Louis, USA). DOX·HCl was purchased from Beijing Huafeng United Technology Co. Ltd. (Beijing, China). 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) and 4,6-diamino-2phenyl indole (DAPI) was acquired from Shanghai Sangon Biological Engineering Technology & Services Co., Ltd (China). Dulbecco's Modified Eagle Medium (DMEM), fetal bovine serum (FBS), penicillin, and streptomycin were from Hangzhou Jinuo Biomedical Technology (Hangzhou, China). Other chemicals were purchased from Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China).

2.2 Synthesis of Fe₃O₄/HBC@F127 nanocomposites

Tetramesityl HBC molecules was synthesized according to the reported method.²⁵ Hydrophobic Fe_3O_4 nanoparticles were synthesized as previously reported.²⁶ To synthesize $Fe_3O_4/$ HBC@F127 nanocomposites, F127 (5 mg), HBC (5 mg) and Fe_3O_4 (5 mg) nanoparticles were dissolved in 0.5 mL THF and then added dropwise into 30 mL H₂O and then kept stirring for 30 min. The resulting solution was then dialyzed in pure water for 3 days. Finally, the resulting suspension was filtered through a 0.22 µm membrane before use. DOX (5 mg) was also added in

the first step to obtain DOX loaded ${\rm Fe_3O_4/HBC}$ ${\rm (BF127)}$ nanocomposites.

2.3 Characterizations

Transmission electron micrographs (TEM) were taken by TEM (JEOL JEM-2100F) for particle size determination. Infrared spectra were recorded in the range 4000–400 cm⁻¹ on a Fouriertransform infrared spectrometer (FT-IR, Bomen Hartmann and Braun, MB series). Dynamic light scattering (DLS) measurement was determined by Zetasizer (Nano series, Malvern Instruments). The magnetic measurement was carried out using vibrating sample magnetometer (VSM, MPMS SQUIDM, USA). The absorption and emission spectra was analyzed by using a UV-vis spectroscopy (Specord 205, Jena, Germany) and fluorescence spectroscopy (Cary Eclipse, Varian, USA). The Fe concentration of the particles was determined by inductively coupled plasma analyses (ICP, Agilent 730 ICP-OES, USA). The T_2 relaxation times of the particle suspension with different Fe concentrations in water were measured by a 3.0-T GE Signa HDX MRI scanner (GE, Milwaukee, USA) by using a head coil. T_2 weighted images were obtained from a 4.0 mm-thick section using a 60 mm \times 48 mm field of view (FOV), repetition time (TR) = 3000 ms, echo time (TE) = 40, 60, 80, 100 and 120 ms and reconstructed using a 320 \times 320 image matrix. The relaxivity (r_2) was calculated by a linear fitting of the inverse relaxation time as a function of the Fe concentration.

2.4 Cell culture, cytotoxicity, and cellular uptake

Human hepatoma cells (HepG2) were obtained from American Type Culture Collection (ATCC, Manassas, VA, USA) and cultured in DMEM medium. Each cell culture medium was supplemented with 10% inactivated fetal bovine serum (FBS), 100 mg mL⁻¹ streptomycin, and 100 U mL⁻¹ penicillin at 37 °C under 5% CO₂. For cytotoxicity study, the cells were seeded on a 96-well plate at a density of 1×10^4 cells per well with DMEM medium containing 10% FBS. After 24 h, the culture medium was replaced with 100 µL of medium containing 0–400 µg mL⁻¹ of Fe₃O₄/HBC@F127 nanocomposites. The cytotoxicity was evaluated by determining the cell viability after incubation for 24 h. The number of viable cells was determined by MTT assay according to the reported method.²⁷

For cellular uptake study, the cells (5 × 10⁵) were pre-grown in 6-well culture plates and then Fe₃O₄/HBC@F127 nanocomposites were added at a concentration of 0–40 µg mL⁻¹ in the same medium and then incubated separately for 1 h, 2 h and 24 h. Next, the culture medium was aspirated and the cells were washed three times with 2 mL of PBS containing 2% FBS. The cells were detached by 1× trypsin and centrifuged at 1200 rpm for 5 min. The media was then removed by aspiration. The cells were re-suspended in 2 mL of PBS and 1 × 10⁴ cell accounts were immediately analyzed using a flow cytometer (Beckman Coulter, California, USA). The collected cells were microwave digested and then the cellular uptake Fe was analyzed by ICP.

For cell imaging, the cells (5×10^4) were seeded into a 24-well culture plate containing one glass coverslip per well and

incubated for 24 h. Next, the medium was removed and 0.5 mL of DMEM media containing $Fe_3O_4/HBC@F127$ nanocomposites was added into each well and incubated at 37 °C for 24 h. The coverslips with cells were then placed in empty wells, treated with 1 mL of 4% formaldehyde in PBS, and allowed to sit at room temperature for 30 min. After washing with PBS for three times, the cells were treated with 1 mL of Triton X-100 and incubated for 10 min. Then the cells were washed three times with PBS and then incubated at 37 °C with 0.2 mL of DAPI for 10 min. The cells were analyzed using a confocal laser scanning microscope (CLSM, Zeiss, LSM 880, Germany).

For MR imaging, after the co-incubation with Fe₃O₄/HBC@F127 nanocomposites, the cells were washed thoroughly, then dispersed and suspended by 2 mL 0.5% agarose gel in a 5 mL EP tube. MR imaging was performed on a 3.0-T GE Signa HDX MRI scanner (GE, Milwaukee, WI, USA) using a T_2 -weighted Fast Spin Echo sequence (TR = 3000 ms, TE = 100 ms, slice thickness of 4 mm).

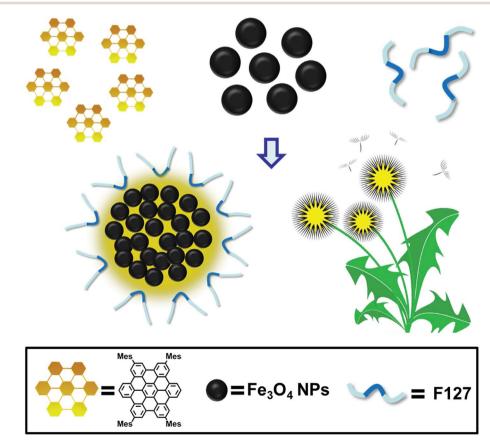
3 Results

3.1 Fabrication and characterizations of Fe₃O₄/HBC@F127 nanocomposites

The schematic illustration of the simply formation of $Fe_3O_4/HBC@F127$ nanocomposites was shown in Scheme 1. Hydrophobic oleic acid-coated Fe_3O_4 nanoparticles and HBC molecules were dissolved in THF with F127, and then the mixed

solution was added into deionized H₂O to form the dandelionlike micelles. After completely dialysis, the micelles containing Fe₃O₄ nanoparticles and HBC molecules were formed, which abbreviated as Fe₃O₄/HBC@F127. Pluronic F127 is selected to encapsulate HBC and Fe₃O₄ nanoparticles due to its amphiphilic nature and excellent biocompatibility as a copolymer consisting poly(ethylene oxide)–poly(propylene oxide)–poly(ethylene oxide) blocks, PEO100–PPO65–PEO100. During the formation of nanocomposites, the hydrophobic PPO is more likely wrapped inside the particles with hydrophobic HBC and Fe₃O₄ to prevent aggregation, protein adsorption, and recognition by the reticuloendothelial system (RES).²⁸

The morphology of Fe₃O₄/HBC@F127 nanocomposites was studied by TEM. Fe₃O₄ nanoparticles were obtained as colloidal particles dispersed in common nonpolar or weakly polar solvents, such as hexane, tetrahydrofuran, chloroform, *etc.* A representative TEM image shows that Fe₃O₄ nanoparticles modified by oleic acid in hexane are highly monodisperse, with an average diameter of about 10 nm (Fig. 1A). After entrapped with HBC in block copolymer F127, some hydrophobic Fe₃O₄ nanoparticles are aggregated in the inner space of the F127 micelle due to the hydrophobic interaction (Fig. 1B). Simultaneously, HBC has a strong tendency to aggregate through π - π stacking when entrapped into block copolymer F127 due to its planar core. The fabricated Fe₃O₄/HBC@F127 nanocomposites have a nearly spherical morphology. It can be seen clearly that



Scheme 1 Schematic illustration of the formation of dandelion-like Fe₃O₄/HBC@F127 nanocomposites.

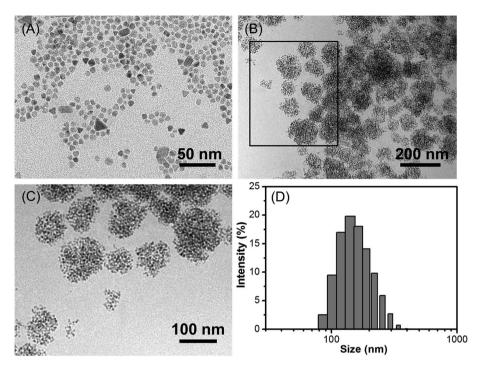


Fig. 1 Morphology characterizations of Fe₃O₄/HBC@F127 nanocomposites. (A) TEM micrographs of Fe₃O₄ nanoparticles, (B) Fe₃O₄/HBC@F127 nanocomposites, (C) a magnified view of the rectangular area of (B) and (D) size distribution of Fe₃O₄/HBC@F127 nanocomposites by DLS.

each particles are made of many Fe₃O₄ nanoparticles in the magnified TEM image (Fig. 1C). The mean hydrodynamic diameter was in the range of 100-300 nm measured in deionized H2O. Besides, Fe3O4/HBC@F127 nanocomposites are stable in deionized water for a period of 3 months without aggregation, which is desirable for application in biomedicine.

Fourier-transform infrared (FT-IR) spectroscopy was performed to investigate the effective compound of HBC molecules, Fe₃O₄ nanoparticles and F127. As shown in Fig. 2, the absorption band at around 597 cm⁻¹ of Fe₃O₄/HBC@F127 nanocomposites corresponding to the Fe-O bond in Fe₃O₄ conforms the presence of Fe₃O₄ nanoparticles.²⁹ Even though the absorption bands of HBC at 875, 850, 828, 766 and 750 cm^{-1}

are suppressed in Fe₃O₄/HBC@F127 nanocomposites (circled in Fig. 2) due to the strong background of Fe_3O_4 nanoparticles, they still can be clearly figured out in the magnified spectrum. Moreover, the characteristic absorption bands of F127 at 2886.4 cm⁻¹ (C–H asymmetric stretching) and 1111.4 cm⁻¹ (C– O-C stretching) also appear in the FT-IR spectrum, which indicate the effective coating onto the surface of Fe₃O₄ nanoparticles.³⁰ The characteristic band of F127 at 2882 cm⁻¹ (C-H asymmetric stretching) is overlapped with the peaks at 2915 cm⁻¹ and 2855 cm⁻¹ in FT-IR spectrum of $Fe_3O_4/$ HBC@F127 nanocomposites, which is hard to identify.

The optical properties of HBC in THF and Fe₃O₄/HBC@F127 nanocomposites dispersed in deionized H2O were investigated

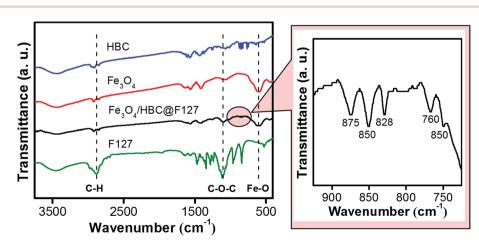


Fig. 2 FT-IR spectra of HBC molecules, Fe₃O₄ nanoparticles, Fe₃O₄/HBC@F127 nanocomposites and F127.

by UV-vis spectrometer and fluorescence spectrometer. The absorption and emission intensity are both normalized in Fig. 3. Both the absorption and emission spectra of Fe₃O₄/HBC@F127 nanocomposites in deionized H₂O are similar with those of HBC in THF. The UV-vis absorption spectra showed a strong absorption at around 360 nm, which was selected as the excitation wavelength (Fig. 3A). The emission spectrum of Fe₃O₄/HBC@F127 nanocomposites showed multiple emission bands with maximum peaks at 468, 506 and 586 nm (Fig. 3B). As shown in the inset in Fig. 3B, the Fe₃O₄/HBC@F127 nanocomposites also present an intense greenish photoluminescence under UV lamp irradiation.

The magnetic property of Fe₃O₄ nanoparticles and Fe₃O₄/HBC@F127 nanocomposites was evaluated by field-dependent magnetization measurements at 300 K (Fig. 4). The lack of hysteresis indicates the superparamagnetic nature of the Fe₃O₄ nanoparticles. The saturated magnetization of Fe₃O₄ nanoparticles was 35.57 emu g⁻¹ when the applied magnetic field reaches 20 kOe. The saturated magnetization decreased slightly (34.67 emu g⁻¹) when Fe₃O₄ nanoparticles formed Fe₃O₄/HBC@F127 nanocomposites, which was attributed to the

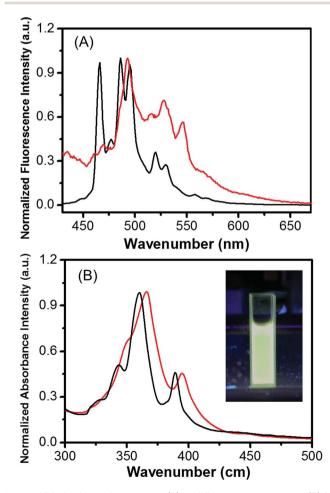


Fig. 3 UV-vis absorption spectra (A) and fluorescence spectra (B) for HBC in THF and Fe_3O_4 /HBC@F127 nanocomposites in deionized H₂O (black line-HBC in THF, red line-Fe₃O₄/HBC@F127 nanocomposites in deionized H₂O).

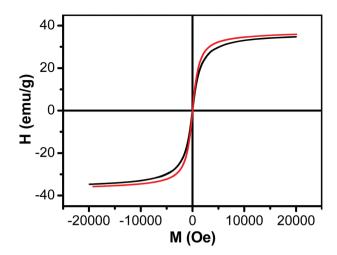


Fig. 4 Magnetization curve as a function of field for Fe_3O_4 nanoparticles (red line) and $Fe_3O_4/HBC@F127$ nanocomposites (black line) at 300 K.

decrease of Fe₃O₄ content of the nanocomposites. The data around zero field showed the coercivity of Fe₃O₄/HBC@F127 nanocomposites was 26 Oe, which demonstrated that the Fe₃O₄/HBC@F127 nanocomposites were superparamagnetic without hysteresis.

3.2 MR relaxivity measurements

The MR imaging properties of Fe₃O₄/HBC@F127 nanocomposites were evaluated by measuring the transverse relaxation time T_2 on a 3.0 T scanner. Their efficiency as contrast agents is determined by calculating the transverse relaxation rate from a linear fit of the inverse relaxation times as a function of the iron concentration. The transverse relaxation time T_2 of water was reduced by Fe₃O₄/HBC@F127 nanocomposites relative to the control PBS. As shown in Fig. 5A, we observed that the T_2 -weighted MR signal intensity decreased with the increase of Fe₃O₄/HBC@F127 nanocomposites concentration (measured in μ M Fe) at different TE values (TR = 3000 ms). As expected, the relaxation rate r_2 is equal to $1/T_2$, which is linearly proportional to Fe concentration (Fig. 5B). T_2 relaxivities r_2 for our Fe₃O₄/ HBC@F127 nanocomposites is 214.61 mM⁻¹ s⁻¹, which is higher than that of Feridex ($r_2 = 108.2 \text{ mM}^{-1} \text{ s}^{-1}$).³¹

3.3 In vitro imaging studies

The encapsulated HBC molecules enabled confocal microscopic tracking of the nanocomposites after internalized by cells. In order to observe the cell imaging effect, HepG2 cells were treated with Fe₃O₄/HBC@F127 nanocomposites at a concentration of 10 μ g mL⁻¹ for 24 h. The confocal microscopic images show that large amount of green particles are dispersed in the cytoplasm, which proves the cellular uptake of the nanocomposites (Fig. 6).

To explore the feasibility of the $Fe_3O_4/HBC@F127$ nanocomposites used as contrast agents for T_2 -weighed MR imaging of cancer cells, HepG2 cells were incubated with $Fe_3O_4/$ HBC@F127 nanocomposites at various iron concentrations for

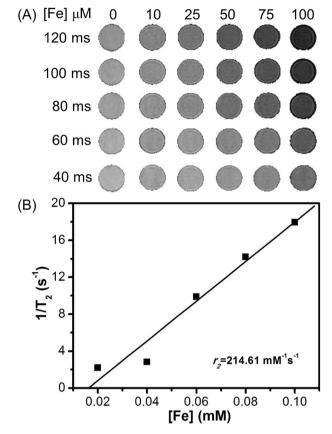


Fig. 5 Magnetic resonance imaging properties of $Fe_3O_4/HBC@F127$ nanocomposites: (A) T_2 -weighted MR images of $Fe_3O_4/HBC@F127$ nanocomposites in PBS solution at 3.0 T MRI system at various iron concentrations for different values of echo time (TE, TR = 3000 ms). (B) T_2 relaxivity plot of aqueous suspension of $Fe_3O_4/HBC@F127$ nanoparticles vs. iron concentration (data as values obtained from curve fitting).

24 h and then MR imaging investigation was performed. It can be seen from the T_2 -weighted MR phantom images that cells incubated with Fe₃O₄/HBC@F127 nanocomposites show a significantly darker signal compared with that in control group (Fig. 7A), which indicates the negative contrast enhancement becomes stronger as Fe concentration increases. The quantitative analysis of the MR signal intensity was also carried by plotting the MR signal intensity of the cells as function of Fe concentration (Fig. 7B). The MR signal intensity of cells treated by Fe₃O₄/HBC@F127 nanocomposites is much lower than that of the control cells treated with PBS, which suggests that the nanocomposites can specifically affect the MR signal.

3.4 Effects of DOX loaded Fe₃O₄/HBC@F127on cell viability

Cytotoxicity studies were carried out to determine the cell viabilities of $Fe_3O_4/HBC@F127$ nanocomposites and DOX loaded $Fe_3O_4/HBC@F127$ nanocomposites in HepG2 cells. MTT results shows that the developed $Fe_3O_4/HBC@F127$ nanocomposites are non-cytotoxic in the given concentration range (Fig. 8A). The cell viability is found no significant difference between $Fe_3O_4/HBC@F127$ nanocomposites treated group and

control group even when the concentration of Fe₃O₄/ HBC@F127 is high up to 400 μ g mL⁻¹. However, the cell viability exhibited significant change when exposed to DOX loaded Fe₃O₄/HBC@F127 nanocomposites. DOX loaded Fe₃O₄/ HBC@F127 nanocomposites exhibits the typical concentrationdependent antiproliferative effect on HepG2 cells. There are significant differences between DOX loaded Fe₃O₄/HBC@F127 treated group and control group when the concentration of $Fe_3O_4/HBC@F127$ nanocomposites is larger than 200 µg mL⁻¹ (the DOX loading amount is approximately 20 μ g mg⁻¹ Fe₃O₄/ HBC@F127 nanocomposites). The toxicity of DOX loaded Fe₃O₄/HBC@F127 nanocomposites is further visualized directly by cellular imaging (Fig. 8B-D), which confirms that higher percentage of cell death as compared to the control. Cells treated with an equivalent amount of control nanocomposites (without drug) does not show any morphology change, as the cell growth is almost identical as the control cells.

3.5 Cellular uptake studies

The amount of the cellular uptake of iron was quantitatively measured by using ICP. The cellular Fe uptake displays both concentration-dependent and time-dependent manner for HepG2 cells. As shown in Fig. 9A, when HepG2 cells are incubated with Fe₃O₄/HBC@F127 nanoparticles at a Fe concentration of 0.2 mM, the amount of the cellular iron increases in the first several hours and then nearly reaches a saturation point, which also suggests that the uptake of Fe₃O₄/HBC@F127 nanocomposites occurs within 6 hours. As shown in Fig. 9B, when HepG2 cell are incubated with Fe₃O₄/HBC@F127 nanocomposites for 24 h, the amount of cellular Fe grows in the tested concentration from 0 to 0.8 mM, and higher Fe concentration leads to an increased Fe uptake in the cells. Due to the fluorescence signal of HBC in Fe₃O₄/HBC@F127 nanocomposites, flow cytometry analysis was used to investigate the cellular uptake of Fe3O4/HBC@F127 nanocomposites in HepG2 cells (Fig. 9C and D). Compared with the control group, cells incubated with Fe₃O₄/HBC@F127 nanocomposites show a fluorescent shift. When the concentration increases to 0.4 mM, the fluorescent intensity does not change significantly (Fig. 9C), which is different from ICP analysis results. This can be ascribed to that the excessive nanoparticles are not able to be washed thoroughly in the cell collection process for ICP analysis as the concentration increases. The fluorescent intensity of cells incubated with Fe₃O₄/HBC@F127 nanocomposites for 2 h shows only a slight shift compared with control group, suggesting a weak cellular uptake efficacy. While the incubation time was prolonged to 24 h, the increased fluorescent intensity showed a better cellular uptake efficacy, which confirmed the time dependent uptake of Fe₃O₄/HBC@F127 nanoparticles by HepG2 cells (Fig. 9D).

4 Discussions

Multifunctional nanostructures have attracted wide attention, such as the combination of magnetic nanoparticles and fluorescence nanoparticles or molecules yields a bi-functional

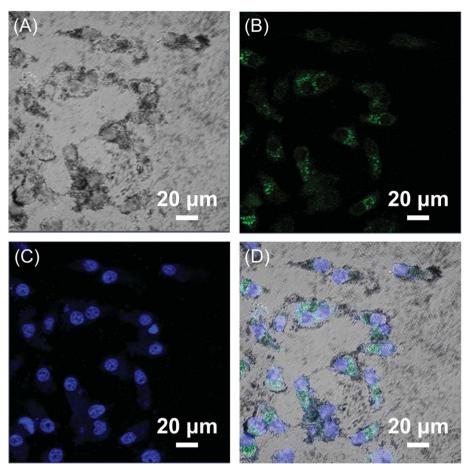


Fig. 6 CLSM images of HepG2 cells exposed to 10 μ g mL⁻¹ Fe₃O₄/HBC@F127 nanocomposites for 24 h. (A) Bright field, (B) Fe₃O₄/HBC@F127 nanocomposites, (C) DAPI, and (D) merged image of (A)–(C).

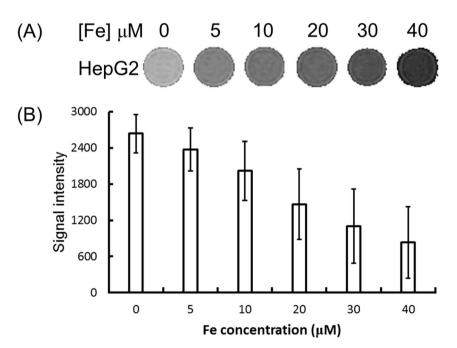


Fig. 7 T_2 -weighted MR images. HepG2 cells were treated with Fe₃O₄/HBC@F127 nanocomposites at various Fe concentrations for 24 h. The images were collected by MR (TR = 3000 ms, TE = 100 ms) (A) and the signal intensity was analyzed (B).

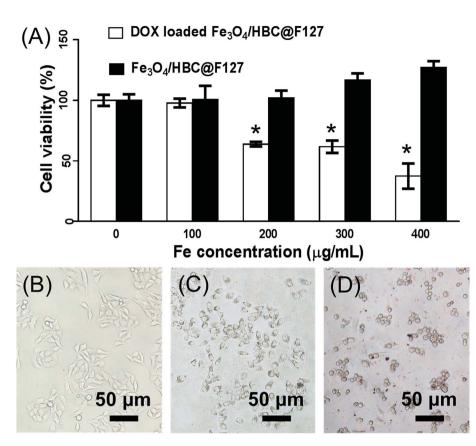


Fig. 8 Effects of DOX loaded Fe₃O₄/HBC@F127 on cell viability. (A) The cytotoxic effects of Fe₃O₄/HBC@F127 and DOX loaded Fe₃O₄/HBC@F127 on HepG2 cells were detected by MTT assay. *p < 0.05 compared with control group (B) control cells treated with DMEM. (C) Cells treated with Fe₃O₄/HBC@F127 nanoparticles. (D) Cells treated with DOX loaded Fe₃O₄/HBC@F127 nanocomposites at Fe concentration of 100 µg mL⁻¹. The data were presented as mean \pm SD, n = 4.

nanocomposites for fluorescence imaging and MR imaging.32-34 Although multifunctional nanocarriers combined with imaging and therapy has reported in many literatures, this work also has advantages. Compared with the previous publications, this work developed a trifunctional nanocomposites not only for cancer diagnosis via MR/fluorescence dual modal imaging but also can load drug molecules for cancer therapy. Besides, we prepared this multifunctional nanocomposites by using a facile and mild method without complex chemical reaction. Hydrophobic HBC molecules and hydrophobic Fe₃O₄ nanoparticles were encapsulated by using biocompatible copolymer F127 due to the hydrophobic interaction and π - π stacking interaction. The significance of this combination also lies in broadening the application of hydrophobic HBC molecules in biomedicine. On one hand, this combination increases the water dispersibility of HBC molecules via F127 micellar solubilization, which is a prerequisite condition for biological application; on the other hand this combination reduces the toxicity of organic HBC molecules. As is well-known, toxicity is an important issue for biomedical imaging and drug delivery applications. It has been reported that the surface modification of the nanoparticles had a significant influence on the cytotoxicity.35 HBC molecules aggregated and then located in the inner space of F127 micelles with a compatible shell layer, which reduce its toxicity obviously. After self-assembly, HBC aggregates also exhibited

similar brilliant fluorescent properties, HepG2 cells incubated with Fe $_3O_4$ /HBC@F127 nanocomposites exhibited strong intracellular green fluorescence signal.

Supermagnetic Fe₃O₄ nanoparticles is a kind of ideal MR imaging contrast agents due to their high transverse relaxation profiles.³⁶ Usually, a high r₂ relaxivity is helpful for MR imaging. Here, the ultra-small Fe₃O₄ nanoparticles forms aggregates in the formation of Fe₃O₄/HBC@F127 nanocomposites and therefore the nanocomposites have a high r_2 value of 214.61 mM⁻¹ s⁻¹. Compared with FDA approved Fe₃O₄ nanoparticlesbased T_2 contrast agent Feridex ($r_2 = 108.2 \text{ mM}^{-1} \text{ s}^{-1}$),³¹ this r_2 value is high enough to yield strong negative contrast signal. The MR signal of the tumor cell samples are also efficiently decreased after incubation with Fe3O4/HBC@F127 nanocomposites, which indicated that the magnetic property of the composites are not affected by the cellular physiological environment and therefore can be used as contrast agents for MR imaging. However, it has been suggested that the potential of $Fe_3O_4/HBC@F127$ nanocomposites as an T_2 contrast agent should be assessed through in vivo experiments.

To sum up, $Fe_3O_4/HBC@F127$ nanocomposites show good fluorescence/MR imaging ability and cell inhibition effect, which ensure its potential application for the diagnosis and treatment of cancer disease.

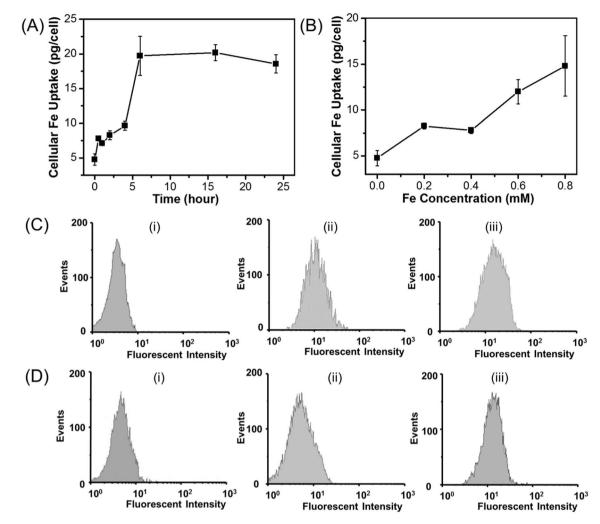


Fig. 9 Cellular uptake of $Fe_3O_4/HBC@F127$ nanocomposites. (A and B) Fe uptake analysis by ICP at different concentrations and different time points. (C) Flow cytometry histograms of HepG2 cells incubated with $Fe_3O_4/HBC@F127$ nanocomposites for 24 h. (i) Control, (ii) 0.4 mM and (iii) 0.8 mM. (D) Flow cytometry histograms of HepG2 cells incubated with $Fe_3O_4/HBC@F127$ nanocomposites at a concentration of 0.2 mM: (i) control, (ii) 2 h and (iii) 24 h.

5 Conclusions

Water-dispersible $Fe_3O_4/HBC@F127$ nanocomposites were developed as multifunctional MR/fluorescence imaging agents for cancer cells. *In vitro* tests indicated that the MR/ fluorescence imaging could be realized with high sensitivity by the application of $Fe_3O_4/HBC@F127$ nanocomposites. Besides, $Fe_3O_4/HBC@F127$ nanocomposites could also delivery anticancer drugs to inhibit the proliferation of HepG2 cells. The fabricated $Fe_3O_4/HBC@F127$ nanocomposites with good biocompatibility are hopeful to be developed as a new multifunctional fluorescence/MR imaging agent for the diagnosis of cancer and a delivery vehicle for the treatment of cancer.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (51302004 to Meng-Meng Song, and 81673438 to Yu-Xian Shen), Anhui Provincial Natural Science Foundation of China (1308085QH141 to Meng-Meng Song) and Scientific Research Foundation of the Institute for Translational Medicine of Anhui Province (SRFITMAP, 2017zhyx34 to Yu-Xian Shen). This work was also supported by University Natural Science Research Project of Anhui Province (KJ2016A334 to Rui Liu).

References

- 1 B. Pelaz, C. H. Alexiou, R. A. Alvarez-Puebla, F. Alves, A. M. Andrews, S. Ashraf, L. P. Balogh, L. Ballerini,
 - A. Bestetti, C. Brendel, S. Bosi, M. Carril, W. C. W. Chan,
- C. Y. Chen, X. D. Chen, X. Y. Chen, Z. Cheng, D. X. Cui,
- J. Z. Du, C. Dullin, A. Escudero, N. Feliu, M. Y. Gao,

M. George, Y. Gogotsi, A. Grunweller, Z. W. Gu, N. J. Halas, N. Hampp, R. K. Hartmann, M. C. Hersam, P. Hunziker, J. Jian, X. Y. Jiang, P. Jungebluth, P. Kadhiresan, K. Kataoka, A. Khademhosseini, J. Kopecek, N. A. Kotov, H. F. Krug, D. S. Lee, C. M. Lehr, K. W. Leong, X. J. Liang, M. L. Lim, L. M. Liz-Marzan, X. M. Ma, P. Macchiarini, H. Meng, H. Mohwald, P. Mulvaney, A. E. Nel, S. M. Nie, Ρ. Nordlander, T. Okano, J. Oliveira, T. H. Park, M. Penner, M. Prato, V. Puntes, V. M. Rotello, R. A. Samarakoon, R. E. Schaak, Y. Q. Shen, S. Sjoqvist, A. G. Skirtach, M. G. Soliman, M. M. Stevens, H. W. Sung, B. Z. Tang, R. Tietze, B. N. Udugama, J. S. VanEpps, T. Weil, P. S. Weiss, I. Willner, Y. Z. Wu, L. L. Yang, Z. Yue, Q. Zhang, Q. Zhang, X. E. Zhang, Y. L. Zhao, X. Zhou and W. J. Parak, ACS Nano, 2017, 11, 2313-2381.

- 2 P. N. Hedde and G. U. Nienhaus, *Biophys. Rev.*, 2010, 2, 147–158.
- 3 L. Shang and G. U. Nienhaus, Biophys. Rev., 2012, 4, 313-322.
- 4 M. Fernandez-Suarez and A. Y. Ting, *Nat. Rev. Mol. Cell Biol.*, 2008, **9**, 929–943.
- 5 B. N. Giepmans, S. R. Adams, M. H. Ellisman and R. Y. Tsien, *Science*, 2006, **312**, 217–224.
- 6 U. Resch-Genger, M. Grabolle, S. Cavaliere-Jaricot, R. Nitschke and T. Nann, *Nat. Methods*, 2008, 5, 763–775.
- 7 W. H. Wong, T. Khoury, D. Vak, C. Yan, D. J. Jones,
 M. J. Crossley and A. B. Holmes, *J. Mater. Chem.*, 2010, 20, 7005–7014.
- 8 W. Pisula, A. Menon, M. Stepputat, I. Lieberwirth, U. Kolb,
 A. Tracz, H. Sirringhaus, T. Pakula and K. Müllen, *Adv. Mater.*, 2005, 17, 684–688.
- 9 H. H. Dam, K. Sun, E. Hanssen, J. M. White, T. Marszalek,
 W. Pisula, J. Czolk, J. Ludwig, A. Colsmann, M. Pfaff,
 D. Gerthsen, W. W. H. Wong and D. J. Jones, ACS Appl. Mater. Interfaces, 2014, 6, 8824–8835.
- 10 R. Liu, D. Wu, X. Feng and K. Müllen, J. Am. Chem. Soc., 2011, 133, 15221–15223.
- 11 M. Yin, J. Shen, W. Pisula, M. Liang, L. Zhi and K. Müllen, *J. Am. Chem. Soc.*, 2009, **131**, 14618–14619.
- 12 X. L. Feng, W. Pisula, T. Kudernac, D. Q. Wu, L. J. Zhi, S. De Feyter and K. Müllen, *J. Am. Chem. Soc.*, 2009, **131**, 4439– 4448.
- 13 J. S. Wu, A. Fechtenkotter, J. Gauss, M. D. Watson, M. Kastler, C. Fechtenkotter, M. Wagner and K. Müllen, J. Am. Chem. Soc., 2004, 126, 11311–11321.
- 14 J. H. Shen, Y. F. Li, Y. H. Zhu, X. L. Yang, X. Z. Yao, J. Li, G. J. Huang and C. Z. Li, *J. Mater. Chem. B*, 2015, 3, 2873– 2882.
- 15 L. Jiang, Q. Zhou, K. Mu, H. Xie, Y. Zhu, W. Zhu, Y. Zhao, H. Xu and X. Yang, *Biomaterials*, 2013, **34**, 7418–7428.

- 16 Y. W. Jun, J. T. Jang and J. Cheon, *Adv. Exp. Med. Biol.*, 2007, 620, 85–106.
- 17 L. H. Reddy, J. L. Arias, J. Nicolas and P. Couvreur, *Chem. Rev.*, 2012, **112**, 5818–5878.
- 18 J. J. Lin, J. S. Chen, S. J. Huang, J. H. Ko, Y. M. Wang, T. L. Chen and L. F. Wang, *Biomaterials*, 2009, **30**, 5114–5124.
- K. C. Barick, E. Rathee, S. L. Gawali, A. Sarkar, A. Kunwar,
 K. I. Priyadarsini and P. A. Hassan, *RSC Adv.*, 2016, 6, 98674–98681.
- 20 A. V. Kabanov, E. V. Batrakova and V. Y. Alakhov, J. Controlled Release, 2002, 82, 189–212.
- 21 S. S. Kulthe, N. N. Inamdar, Y. M. Choudhari, S. M. Shirolikar, L. C. Borde and V. K. Mourya, *Colloids Surf.*, B, 2011, 88, 691–696.
- 22 E. V. Batrakova and A. V. Kabanov, *J. Controlled Release*, 2008, **130**, 98–106.
- 23 W. D. Ma, H. Xu, C. Wang, S. F. Nie and W. S. Pan, *Int. J. Pharm.*, 2008, **350**, 247–256.
- 24 P. K. Singh, V. K. Pawar, A. K. Jaiswal, Y. Singh, C. H. Srikanth, M. Chaurasia, H. K. Bora, K. Raval, J. G. Meher, J. R. Gayen, A. Dube and M. K. Chourasia, *Int. J. Biol. Macromol.*, 2017, **105**, 1220–1231.
- 25 R. Yamaguchi, S. Hiroto and H. Shinokubo, *Org. Lett.*, 2012, 14, 2472–2475.
- 26 X. Wang, J. Zhuang, Q. Peng and Y. Li, *Nature*, 2005, **437**, 121–124.
- 27 M. M. Song, H. L. Xu, J. X. Liang, H. H. Xiang, R. Liu and Y. X. Shen, *Mater. Sci. Eng.*, C, 2017, 77, 904–911.
- 28 M. Garinot, V. Fievez, V. Pourcelle, F. Stoffelbach, A. des Rieux, L. Plapied, I. Theate, H. Freichels, C. Jerome, J. Marchand-Brynaert, Y. J. Schneider and V. Preat, *J. Controlled Release*, 2007, **120**, 195–204.
- 29 R. D. Waldron, Phys. Rev., 1955, 99, 1727-1735.
- 30 B. Feng, R. Y. Hong, L. S. Wang, L. Guo, H. Z. Li, J. Ding,
 Y. Zheng and D. G. Wei, *Colloids Surf.*, A, 2008, 328, 52–59.
- 31 Y. Ryu, Z. Jin, J. J. Lee, S. H. Noh, T. H. Shin, S. M. Jo, J. Choi, H. Park, J. Cheon and H. S. Kim, *Angew. Chem.*, 2015, 54, 923–926.
- 32 K. Li, Z. P. Zhang, M. Luo, X. Yu, Y. Han, H. P. Wei, Z. Q. Cui and X. E. Zhang, *Nanoscale*, 2012, 4, 188–193.
- 33 R. Zhou, S. Sun, C. Li, L. Wu, X. Hou and P. Wu, ACS Appl. Mater. Interfaces, 2018, 10, 34060–34067.
- 34 C. Zhang, S. S. Moonshi, W. Wang, H. T. Ta, Y. Han, F. Y. Han, H. Peng, P. Kral, B. E. Rolfe, J. J. Gooding, K. Gaus and A. K. Whittaker, ACS Nano, 2018, 12, 9162–9176.
- 35 N. Lewinski, V. Colvin and R. Drezek, Small, 2008, 4, 26-49.
- 36 Y. Hu, S. Mignani, J. P. Majoral, M. Shen and X. Shi, *Chem. Soc. Rev.*, 2018, 47, 1874–1900.