# **RSC Advances**



View Article Online

# REVIEW

Check for updates

Cite this: RSC Adv., 2018, 8, 40061

# Applications of Friedel–Crafts reactions in total synthesis of natural products

Majid M. Heravi, 💿 \* Vahideh Zadsirjan, \* Pegah Saedi and Tayebeh Momeni

Over the years, Friedel–Crafts (FC) reactions have been acknowledged as the most useful and powerful synthetic tools for the construction of a special kind of carbon–carbon bond involving an aromatic moiety. Its stoichiometric and, more recently, its catalytic procedures have extensively been studied. This reaction in recent years has frequently been used as a key step (steps) in the total synthesis of natural products and targeted complex bioactive molecules. In this review, we try to underscore the applications of intermolecular and intramolecular FC reactions in the total syntheses of natural products and complex molecules, exhibiting diverse biological properties.

# 1. Introduction

Received 3rd September 2018

Accepted 3rd November 2018

DOI: 10.1039/c8ra07325b

rsc.li/rsc-advances

Historically, the FC reactions are the well-established set of reactions initially discovered by Charles Friedel and James Crafts in 1877. In these reactions, certain substituents are attached on to a suitable aromatic ring.<sup>1</sup> Basically, Friedel-

Department of Chemistry, School of Science, Alzahra University, Vanak, Tehran, Iran. E-mail: mmh1331@yahoo.com; mmheravi@alzahra.ac.ir; z\_zadsirjan@yahoo.com; Fax: +98 2188041344; Tel: +98 9121329147 Crafts reactions imply two main sets of reactions. They are alkylation and acylation reactions. In fact, both types of such reaction proceed *via* the same mechanism, which is the typical reaction of aromatic compounds, electrophilic aromatic substitution.<sup>2-5</sup> FC alkylation comprises the alkylation of an appropriate aromatic ring using an alkyl halide, conventionally in the presence of a strong Lewis acid as the catalyst.<sup>6</sup> Commonly, anhydrous ferric chloride is used as a catalyst, in which the alkyl group initially attaches itself to the former site of the chloride ion.<sup>7</sup> Although this reaction enjoys the



Majid M Heravi was born in 1952 in Mashhad, Iran. He received his BSc degree from the National University of Iran in 1975 and his MSc and PhD degrees from Salford University, England in 1977 and 1980, respectively. He completed his doctoral thesis under the supervision of the late Jim Clarck in Salford University, England. He started his career as a research fellow in Daroupakhsh (a phar-

maceutical company) in 1981 in Tehran, Iran and joined Ferdowsi University of Mashhad, Iran as an assistant professor in 1983, and was promoted to associate professor in 1993 and full professor in 1997 in the aforementioned university. In 1999 he moved to Alzahra University in Tehran, Iran as professor of chemistry, where he still works. He has previously been a visiting professor at UC Riverside, California, USA and Hamburg University, Hamburg, Germany. His research interests focus on heterocyclic chemistry, catalysis, organic methodology and green synthetic organic chemistry.



Vahideh Zadsirjan was born in Shiraz, Iran in 1979. She received her BSc in Pure Chemistry from Kharazmi University in 2002, her MSc degree in 2007, and PhD degree in organic chemistry in 2016 under the Professor supervision of Majid M. Heravi from Alzahra University Tehran, Iran. Her research is focused on heterocyclic chemistry, catalysis, organic methodology and green synthetic organic chemistry.

advantages already known for electrophilic aromatic substitution, it also suffers from the fact that the final product is more nucleophilic than the reactant, thus undesired over-alkylation may take place, giving unwanted side products. In addition, the reaction is limited to using tertiary carbon and secondary alkylating agents, or else the emerging carbocation  $(R^{+})$  may be subjected to a known carbocation rearrangement.7 Indeed, alkylations via FC reaction are not limited to using just alkyl halides. FC reactions can be successfully performed via the generation of any other appropriate intermediate, such as those carbocationic intermediates that are derived from alkenes in the presence of a protic acid, Lewis acid, enones, and epoxides. During the previous years of FC reactions, several other Lewis acids, such as BF<sub>3</sub>, BeCl<sub>2</sub>, TiCl<sub>4</sub>, SbCl<sub>5</sub> or SnCl<sub>4</sub>, have been proven to act as effective catalysts. Besides, strong Brønsted acids such as H<sub>2</sub>SO<sub>4</sub> and HF, or super acids, such as HF · SbF<sub>5</sub> and HSO<sub>3</sub> F·SbF<sub>5</sub>, have also been used successfully and proven to accelerate the FC alkylation reaction. In spite of the great significance of FC alkylation in organic transformation, some major drawbacks and problems still exist, which must be circumvented. For example, the FC alkylation reaction requires stoichiometric or super stoichiometric amounts of a Lewis acid or Brønsted acid. Furthermore, since it needs toxic alkyl halides as a reagent, the formation of large quantities of salts as side products is proven to be inevitable. Thus, development of more eco-friendly strategies and economically feasible chemical processes for such an important reaction is still in great demand. The replacement of the alkyl chlorides by less harmful alkylating agents, for instance alcohols, has undoubtedly been a major development since in this case water is formed as a nontoxic by-product. More importantly, the utilization of activated double bonds and compounds such as styrenes have been found even more advantageous and operative, since no by products are generated at all.

The FC acylation reaction is actually the acylation of certain aromatic compounds. For the FC acylation reaction, acyl chlorides are used as common acylating agents. In this version of the FC reaction, frequently Lewis acid catalysts such as AlCl<sub>3</sub> can be used along with acid anhydrides as a suitable reagent. The reaction conditions for the FC acylation reaction are exactly as same as those for FC alkylation. It is worthwhile to know that the FC acylation reaction shows several advantages over the alkylation variant. Because of the electron-withdrawing nature of the carbonyl motif, the product, which is actually a ketone, is expectedly less reactive than the substrate, thus undesired multiple acylations do not take place. In addition, since the intermediate is not a carbocation, no rearrangement occurs as the generated carbonium ion is stabilized *via* a resonance structure in which the positive charge is located on the electronegative oxygen.

By definition, a natural product is a chemical compound or substance that originates from nature and is produced by a living organism.8 In the more extensive sense, a naturally occurring compound implies any substance produced by a living organism.9 Significantly, the total synthesis of natural products, including semi synthesis, is the state of art in synthetic organic chemistry and nowadays plays a key role in the development of organic chemistry as a whole by providing stimulating and useful and especially biologically active natural products as the target. Generally, the total synthesis of natural products is a non-commercial research activity, aimed at deeper understanding of the synthesis of a desired natural product scaffold and, more importantly, the development of more desirable new synthetic approaches. Natural products are also limited to the purified organic compounds isolated from natural sources that are originated by the pathways of primary or secondary metabolism. Many secondary metabolites are cytotoxic and have been chosen and modified through evolution for use as "chemical warfare" agents toward prey, predators, and competing organisms. The terminology of natural product has also been stretched for profitable purposes and thus refers to cosmetics, dietary supplements, and foods obtained from natural sources with no artificial ingredients as additives.<sup>10-13</sup> Natural products frequently show therapeutic benefit and many of them have been used as folk medicines for treating diseases for centuries and thus have given some knowledge about lead compounds for drug discovery.14 Natural products can be categorized in accordance with their biological function, biosynthetic pathway, or source.

The applications of FC reactions (alkylation and acylation) in organic synthesis have been extensively studied and reviewed previously.<sup>6,15-19</sup> In continuation of our interest in applications

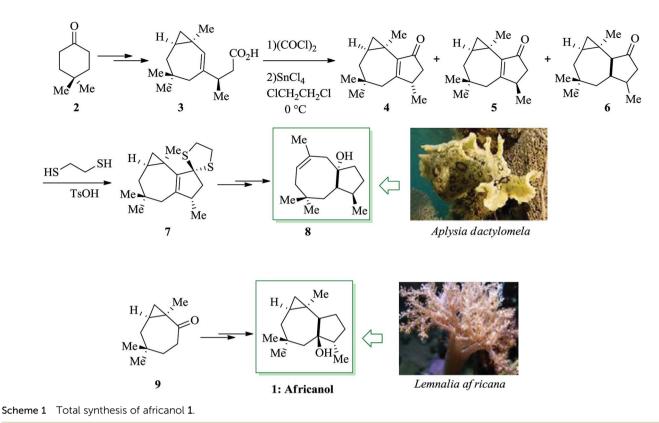


Pegah Saedi was born in Tonekabon, Iran in 1992. She received her BSc degree in Pure Chemistry from Kharazmi University in 2015, and her MSc degree in Organic Chemistry in 2018 under the supervision of Professor Majid M. Heravi from Alzahra University, Tehran, Iran. Her research is focused on heterocyclic chemistry, catalysis, organic methodology and green synthetic organic chemistry.



Tayebeh Momeni was born in 1988 in Karaj, Iran. She received her BSc degree from Imam Khomeini International University in 2010 and her MSc degree from Tehran University under the direction of Professor Mehdi Ghandi in 2012 before moving to Alzahra University, where she became a PhD candidate in Professor Majid. M. Heravi's research group in 2015.

### Review



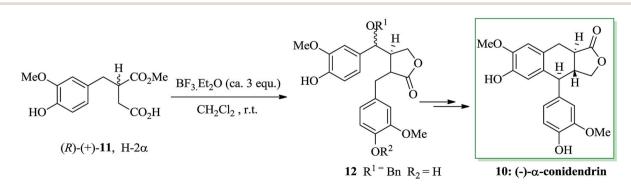
of name reactions<sup>20–35</sup> in the total synthesis of natural products, herein we try to underline the applications of both FC reactions in the total synthesis of naturally occurring compounds, showing diverse biological properties.

# 2. Applications of Friedel–Crafts in total synthesis of natural products

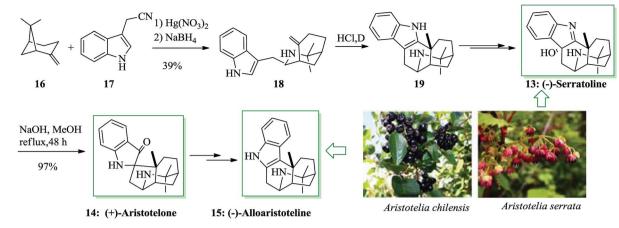
## 2.1. Intramolecular alkylation

Dactylol, **8**, an irregular isoprenoid alcohol, was isolated from the Caribbean sea hare *Aplysia dactylomela* and its structure was identified in 1978 by Schmitz and co-workers.<sup>36,37</sup> The total synthesis of the marine sesquiterpenes africanol **1** and dactylol **8** was achieved from bicyclo[5.1.0]octane precursors. In this route, firstly, compound **2** gave carboxylic acid **3**, which was transformed into compound 7 *via* FC cyclization of its acid chloride and acid-mediated dithioketalization (after several steps). Compound 7 was converted into dactylol 8 after several steps. The configuration of the side chain Me substituent is detected to be proper in relation to dactylol 8 but reverse to that existing in africanol 1. Subsequently, africanol 1 was provided from compound 9 upon several steps (Scheme 1).<sup>38</sup>

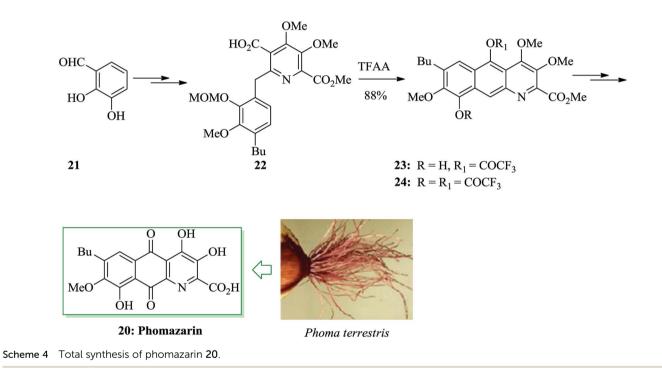
Most lactonic aryltetralin lignans, for example peltatins and podophyllotoxin, contain a similar common structure.  $\alpha$ -Conidendrin was extracted from both wood<sup>39</sup> and waste sulfite liquor.<sup>40</sup> Boissin and co-workers developed the total synthesis of natural (–)-conidendrin **10**. Conidendrin **10** was provided in eight steps starting from the easily accessible and optically active hemisuccinic ester (*R*)-(+)-**11**. The latter gave the monobenzoxy intermediate **12** after several steps, including treatment



(0)



Scheme 3 Total synthesis of natural products (-)-serratoline 13, (+)-aristotelone 14 and (-)-alloaristoteline 15.

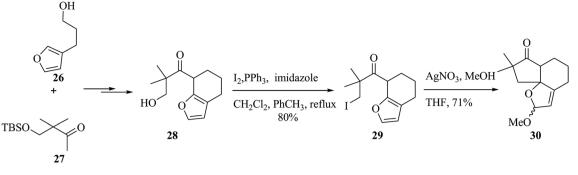


with  $BF_3 \cdot Et_2O$  in dichloromethane at ambient temperature. The benzoxy group of **12** was actually found to be an excellent leaving group during intramolecular FC alkylation. As a result, (-)- $\alpha$ -conidendrin **10** was provided as the desired natural product from **12** in satisfactory yield (Scheme 2).<sup>41</sup>

The Aristotelia alkaloids (–)-serratoline 13, (+)-aristotelone 14 and (–)-alloaristoteline 15 were extracted from Aristotelia chilensis. Serratoline was formerly extracted from Aristotelia serrata, a species native to New Zealand. Heathcock and co-workers reported the total synthesis of (–)-serratoline 13, (+)-aristotelone 14 and (–)-alloaristoteline 15. This approach starts from the reaction between (lS)-(–)- $\beta$ -pinene 16 and 3-indolylacetonitrile 17 to afford (+)-makomakine 18. Next, an intramolecular FC reaction provided (+)-aristotelone 19, which upon two steps afforded alkaloid 13. Subsequently, base-

mediated skeletal rearrangement of 13 provided (+)-aristotelone 14. Finally, alkaloid 14 gave (–)-alloaristoteline 15 upon several steps (Scheme 3).<sup>42</sup>

Phomazarin 20 is the most abundant and extensively studied aza anthraquinone that was extracted from cultures of *Phoma terrestris* Hansen (*Pyrenochaeta terrestris* Hansen) in 1940.<sup>43</sup> Boger and co-workers reported the total synthesis of phomazarin 20. Initially 2,3-dihydroxybenzaldehyde 21 was converted into 22 after several steps. Compound 22 in the presence of TFAA cleanly led to FC closure of the B-ring and simultaneous MOM deprotection, affording a variable and inconsequential mixture of 23 and 24 with no detection of the intermediate bisphenol. Conditions were planned such that only the single mono-(trifluoroacetate) 23 (88%; TFAA) was obtained. Finally,



Scheme 5 The formation of the tricyclic product 30.

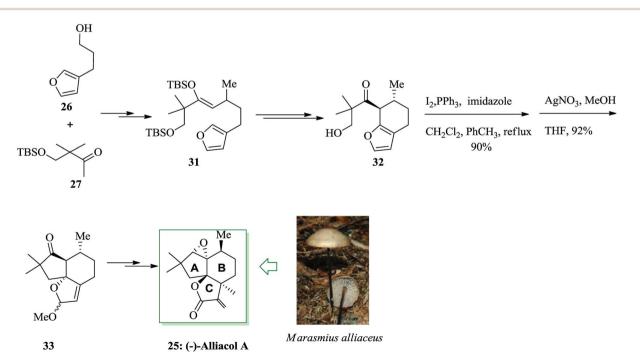
compound 23 afforded phomazarin 20 upon several steps (Scheme 4).44

Alliacol A 25, a sesquiterpene, was extracted in Europe from the culture broth of the fungus *Marasmius alliaceus*.<sup>45</sup> This molecule exhibits antimicrobial properties and prevents DNA synthesis in the ascetic form of Ehrlich carcinoma.<sup>46,47</sup> A tandem anodic coupling-FC alkylation methodology was applied to quickly complete the enantioselective synthesis of alliacol A. In this route, firstly, the reaction between compounds 26 and 27 afforded compound 28 after several steps. Then, it was tried to form the tricyclic unit of the desired naturally occurring compound. For this purpose, the alcohol in 28 was transformed into the relevant iodide, which was applied for either the FC alkylation or radical cyclization method. The FC reaction between iodide and AgNO<sub>3</sub> in MeOH gave a 71% yield of the corresponding tricyclic product 30 (Scheme 5).<sup>48</sup>

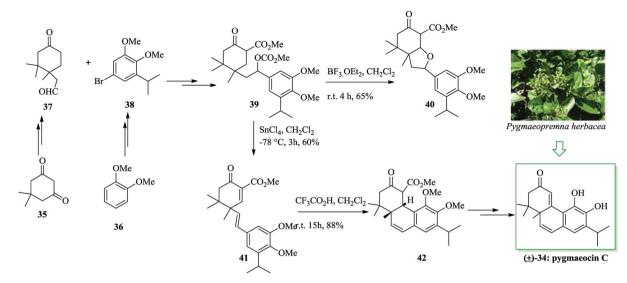
Subsequently, it was attempted to provide substrate **31** containing the B-ring Me group of alliacol A in place. Compounds **26** and **27** gave substrate **31** upon several steps.

This compound provided the bicyclic product **32** upon two steps. Using the bicyclic product, the alcohol **32** was transformed into an iodide and the FC cyclization was finished. A 92% extracted yield of the tricyclic product **33** was provided from the FC reaction. Next, compound **33** was converted into (-)-alliacol A **25** upon several steps (Scheme 6).<sup>48</sup>

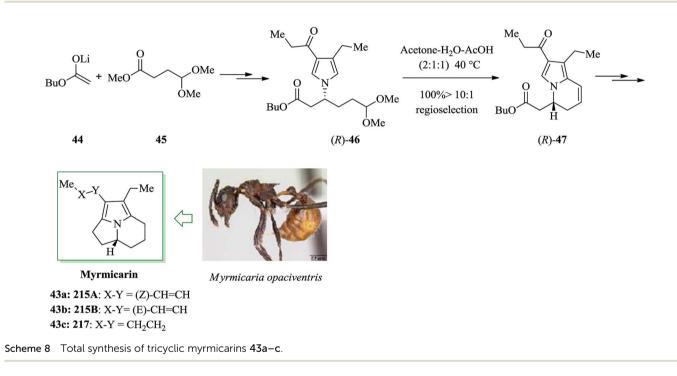
In 1990, Meng and co-workers for the first time reported the isolation of pygmaeocin C 34,<sup>49</sup> a rearranged abietane diterpenoid, from the roots of *Pygmaeopremna herbacea* Moldenke. The initial total synthesis of pygmaeocin C 34, in racemic form, was accomplished in 2005 by Liu and co-workers. Total synthesis of pygmaeocin C 34 was accomplished, starting from dimedone 35 in 14 steps with an overall yield of 4.6%. In this route, an intramolecular FC reaction was considered as a key step. Notably, keto aldehyde 37 was synthesized from dimedone 35 and also compound 38 was synthesized from veratrole 36 *via* several steps. In the following, the reaction of aldehyde 37 and bromide 38 gave the key intermediate 39, upon several steps. It should be mentioned that compound 39 was subjected to



Scheme 6 Total synthesis of (-)-alliacol A 25.



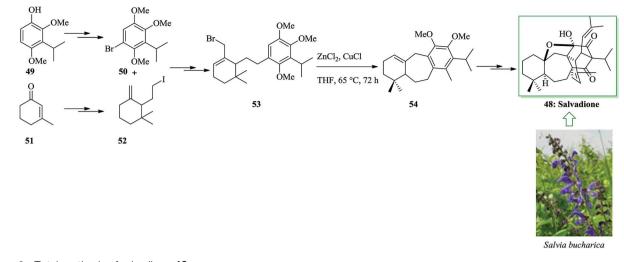
Scheme 7 Total synthesis of pygmaeocin C 34.



a simple intramolecular FC alkylation to give the tricyclic system in the target product. Upon extensive study, it was found that the required cyclization could be obtained from olefin **41** with an extraordinary degree of efficiency. Upon treatment with trifluoroacetic acid (TFA), the latter was submitted to intramolecular FC reaction to afford an 88% yield of the corresponding cyclized product **42** as a result of double bond isomerization with subsequent ring closure. Finally, the tricyclic keto ester **42** was converted into pygmaeocin C **34** upon several steps (Scheme 7).<sup>50</sup>

The myrmicarins, a group of structurally intriguing alkaloids, were initially extracted from the poison gland secretions of the African ant species *Myrmicaria opaciventris*.<sup>51,52</sup> The pyrroloindolizine unit of myrmicarins **43a**, **43b**, and **43c** is a common structural scaffold found in various alkaloids. An asymmetric synthesis of a key dihydroindolizine intermediate 47 for the construction of myrmicarin alkaloids was developed in 2005 by Movassaghi and co-workers. Key conversions in this method contain a stereospecific Pd-mediated N-vinylation, a Cu-mediated asymmetric conjugate reduction, and a regioselective FC reaction. The synthesis of optically potent and isomerically pure samples of (4aR)-myrmicarins **43a**, **43b**, and **43c** and also their respective C4*a*-epimers was achieved using the above-mentioned protocol. Firstly, the lithium enolate **44** and methyl 4-(dimethoxy)-butyrate **45** were reacted to afford compound **46** in several steps. Next, the dihydroindolizine **47** was provided, which is a key intermediate for the construction of the myrmicarin alkaloids. A regioselective FC reaction of the

### Review



Scheme 9 Total synthesis of salvadione 48.

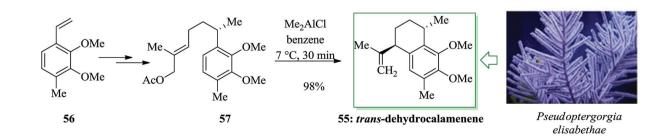
pyrrole ring upon Brønsted acid activation of the dimethoxyacetal **46** and removal of MeOH was achieved to provide the bicyclic vinyl pyrrole **47**. The optimized conditions (acetoneacetic acid-H<sub>2</sub>O, 2 : 1 : 1, 40 °C) were recognized for the quantitative transformation of the β-pyrrolyl ester **46** to the bicyclic vinyl pyrrole to give the corresponding C7*a*-cyclization product **47**. Finally, compound **47** afforded isomerically pure tricyclic myrmicarins **43a–c** after several steps (Scheme 8).<sup>53</sup>

Ahmad and co-workers extracted and identified a remarkable polycyclic triterpene from *Salvia bucharica*, salvadione A **48**.<sup>54</sup> The tricyclic 6–7–6 unit structure of the triterpene salvadione **48** was provided using an effective method starting from the aryl bromide **50** (provided from cyclohexenone **51**) and the alkyl iodide **52** (provided from phenol **49**) containing a methylenecyclohexane group at the terminus. Notably, alkylation of **50** with the iodide **52** afforded the tethered system (allylic bromide) **53**. An intramolecular FC alkylation reaction of allyl bromide **53** provided compound **54** *via* cyclization in the presence of zinc chloride and copper(1) chloride in tetrahydrofuran. The synthesis of tricyclic compound **54** demonstrates a formal total synthesis of salvadione **48**. Finally, the tricycle **54** afforded salvadione **48** upon several steps (Scheme 9).<sup>55</sup>

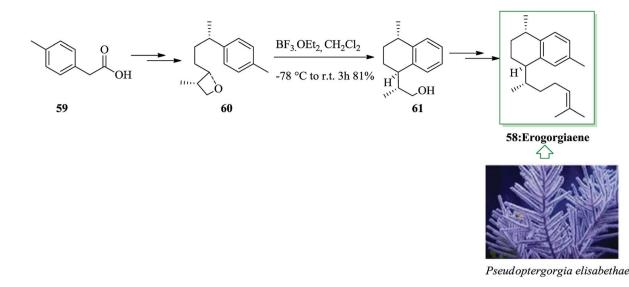
Various aromatic diterpenes with a serrulatane or amphilectane framework have been recognized as bioactive metabolites from marine soft corals, particularly *Pseudopterogorgia elisabethae.*<sup>56</sup> Notable representatives of these compounds are the anti-inflammatory pseudopterosins,<sup>57,58</sup> the cytotoxic and antiviral helioporins,<sup>59</sup> and the pseudopteroxazoles.<sup>60,61</sup> Schmalz *et al.* in 2007 developed an effective and extremely enantioselective synthetic approach to calamenene 55, a *trans*-1,4-difunctionalized tetralin derivative, in 6 linear steps and 57% overall yield initiating from 56.<sup>62</sup> The reaction was initiated from compound 56, which was converted into the corresponding allylic acetate 57 in moderate overall yield. The 1,4-*trans*difunctionalized tetralin framework was selectively provided *via* a FC-type cationic cyclization reaction using Me<sub>2</sub>AlCl as a "proton-scavenging" Lewis acid.<sup>63</sup> The envisioned *trans*-dehydrocalamenene 55 was provided in almost quantitative yield and with very moderate *de* (up to 10 : 1) (Scheme 10).<sup>62</sup>

Erogorgiaene, extracted together with other diterpenes from the West Indian sea whip *Pseudopterogorgia elisabethae*, exhibits favorable anti-mycobacterial properties. A total synthesis of erogorgiaene **58** was developed in 16 steps with an overall yield of 8.2%. The synthesis was based on an extremely diastereoselective intramolecular FC reaction of an oxetane obtained through an enantioselective *syn* aldol coupling. In this route, firstly, easily accessible acid **59** was converted into oxetane **60** in high yield upon several steps. The essential intramolecular FC reaction of oxetane **60** was performed.<sup>64</sup> As expected, the reaction progressed easily, at room temperature, providing a single diastereomer **61**. Finally, alcohol **61** afforded erogorgiaene **58** upon several steps (Scheme **11**).<sup>65</sup>

In 2005, Sattler and co-workers extracted and revealed a group of naturally occurring aromatic  $\beta\text{-}C\text{-}glycoside$ 



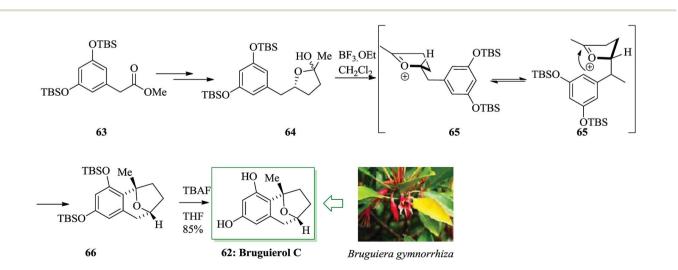
Scheme 10 Total synthesis of trans-dehydrocalamenene 55.



Scheme 11 Total synthesis of erogorgiaene 58

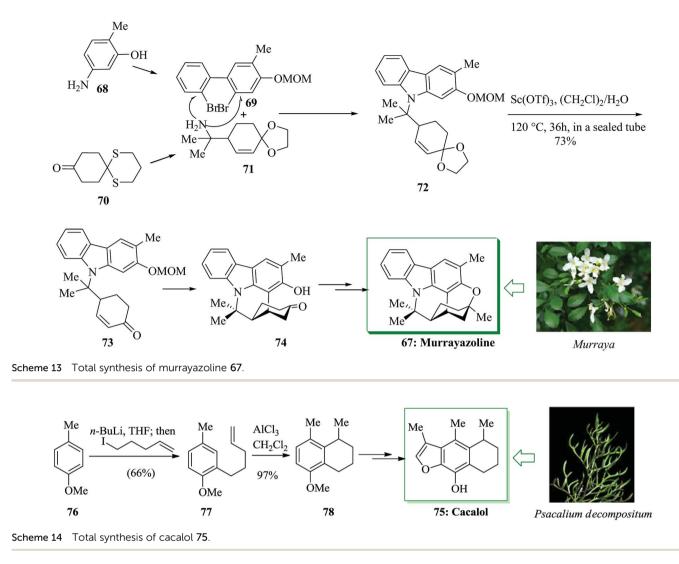
compounds named bruguierols A–C from the stem of the *Bruguiera gymmorrhiza* mangrove tree.<sup>66</sup> In 2007, Jennings and coworkers demonstrated the initial total synthesis of bruguierol C **62** in 7 linear steps from compound **63**.<sup>67</sup> The main step was an intramolecular FC alkylation that eventually supplied the final natural product. For the synthesis of bruguierol C **62**, compound **63** was converted into lactol **64** after several steps. Lastly, an intramolecular trap of the incipient (emerging) oxocarbenium cation using a Marson-type FC alkylation<sup>68</sup> permitted the construction of the masked bruguierol C. The two-step reaction sequence including oxocarbenium construction to provide **65** and also intramolecular FC alkylation afforded the corresponding  $\beta$ -C-glycoside product **66**. Finally, compound **66** afforded the naturally occurring compound **62** in 85% yield (Scheme 12).<sup>67</sup>

Murrayazoline 67, found as mahanimbidine and curryangin, is a carbazole alkaloid extracted as a racemic or an optically active compound from the genus *Murraya*. Murrayazoline and its corresponding carbazole alkaloids were demonstrated to be have antiplatelet aggregation activity.69 A structural explanation investigation demonstrated that murrayazoline is a hexaheterocyclic alkaloid composed of N-functionalized carbazole, cyclohexane and dihydropyran constituents.<sup>70,71</sup> The total synthesis of  $(\pm)$ -murrayazoline 67 was developed in 2008 by Chida and co-workers.72 The characteristic hexa-heterocyclic structure of 67 was generated using a combination of the intramolecular FC-type Michael addition reaction and palladium-mediated carbon-oxygen coupling reactions. Firstly, the two segments 69 and 71 were provided from 5-amino-2methylphenol 68 and 1,5-dithiaspiro-[5,5]undecane-9-one 70,73 respectively, in several steps. Next they afforded the corresponding N-functionalized carbazole 72. The reaction of 72 with scandium(III)triflate in water and dichloroethane induced the deprotection of the ethylene ketal substituent, the intramolecular FC-type Michael addition and the deprotection of the O-MOM group, affording pentacyclic ketone 74 in 73% yield.<sup>74,75</sup>



Scheme 12 Total synthesis of bruguierol C 62.

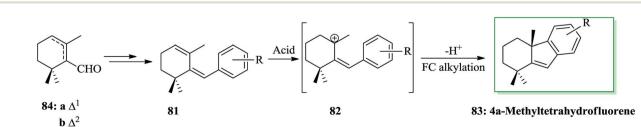
### Review



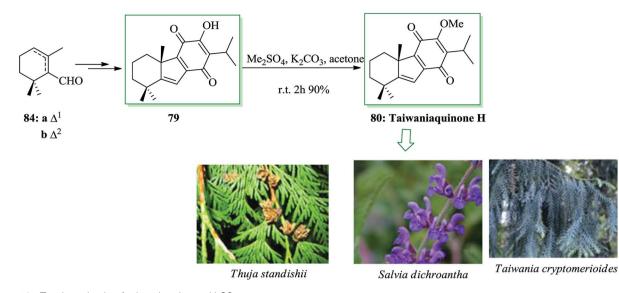
Finally, compound 74 afforded ( $\pm$ )-murrayazoline 67 after several steps (Scheme 13).<sup>72</sup>

Cacalol **75**, a sesquiterpene naturally occurring compound, was extracted from the roots of the shrub *Psacalium decompositum* in northern Mexico,<sup>76,77</sup> that has antihyperglycemic,<sup>78</sup> anti-inflammatory,<sup>79</sup> antimicrobial,<sup>79</sup> and antioxidant<sup>80</sup> properties. A facile synthesis of cacalol **75** was established from 4methylanisole **76** in seven steps and 21–25% overall yield. An intramolecular FC alkylation, Baeyer–Villiger oxidation and alkylation are the key reactions in this approach. The synthesis of cacalol **75** was started with *ortho*-lithiation<sup>81</sup> of 4-methylanisole **76** using *n*-butyllithium in tetrahydrofuran. This anion was alkylated with 5-iodo-1-pentene<sup>82</sup> to afford the alkene 77. Subsequently, this molecule was cyclized *via* an intramolecular FC alkylation using AlCl<sub>3</sub> in dichloromethane to afford the tetralin **78**. Finally, compound **78** afforded cacalol **75** upon several steps (Scheme 14).<sup>83</sup>

Taiwaniaquinoids, a group of tricyclic diterpenoids containing a 4*a*-methyltetra- (and hexa-)hydrofluorene framework have been extracted from various East Asian conifers, for example, *Taiwania cryptomerioides*,<sup>84,85</sup> *Salvia dichroantha*,<sup>86</sup> and *Thuja standishii*.<sup>87</sup> They contain taiwaniaquinone H **80**<sup>85</sup> and dichroanone **79**.<sup>86</sup> A very significant pathway toward taiwaniaquinoids was reported in 2009 by Alvarez-Manzaneda and co-



Scheme 15 Synthesis of 4a-methyltetrahydrofluorene 83 framework.



Scheme 16 Total synthesis of taiwaniaquinone H 80.

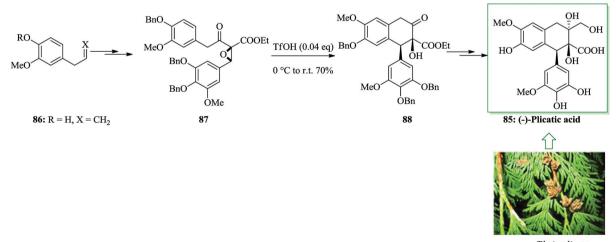
workers. Key steps are the intramolecular FC alkylation reaction of an aryldiene and the degradative oxidation reaction of a methylenedioxy substituent. Using this strategy,  $(\pm)$ -dichroanone 79 (in three steps, 77% overall yield) and  $(\pm)$ -taiwaniaquinone H 80 (in four steps, 70% overall yield) were provided from α-84a or β-cyclocitral 84b. This group designed a probable synthesis of taiwaniaquinoids. The key intermediate would be the aryldiene 81, which resulted in the arylallyl cation 82 under acidic conditions, which, through a quick intramolecular FC alkylation, afforded the 4a-methyltetrahydrofluorene 83 framework (Scheme 15).88

β-Cyclocitral 84a and α-cyclocitral 84b afforded dichroanone 79 upon several steps. Then, the transformation of this quinone into taiwaniaquinone H 80 was achieved (Scheme 16).88

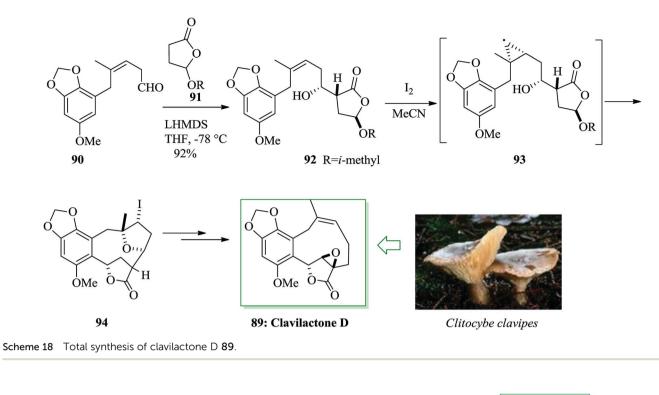
Plicatic acid was identified as the causative agent of occupational asthma.89-91 Plicatic acid was extracted from western red cedar (Thuja plicata) in 1959 by Maclean and co-workers.92 The initial enantioselective total synthesis of (-)-plicatic acid

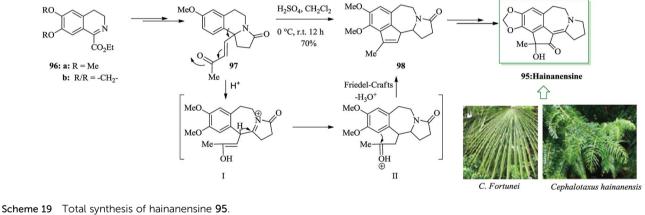
was achieved from eugenol in 12 steps and 14% overall yield. In this approach, a theoretically novel methodology containing an epoxidation-intramolecular enantioselective epoxy-ringopening FC reaction sequence was performed for the asymmetric formation of a structurally complex 2,7'-cyclolignane framework. The synthesis was initiated from eugenol 86, which was converted into epoxide 87 upon several steps. The key intermediate,  $\alpha$ -hydroxy ketone 88, was provided through an intramolecular FC reaction to open the epoxide ring in 86. Noticeably, triflic acid (TfOH) efficiently improved the FC reaction. Finally, compound 88 provided (-)-plicatic acid 85 after several steps (Scheme 17).93

Clavilactone D 89, a tyrosine kinase inhibitor,94 was extracted from cultures of the fungus *Clitocybe clavipes*.<sup>95</sup> A method for the synthesis of clavilactone D was reported in 2009 by Yoshimitsu and co-workers.96 This pathway uses sequential cyclization and FC cyclization to provide a polycyclic lactone fused



Thuja plicata



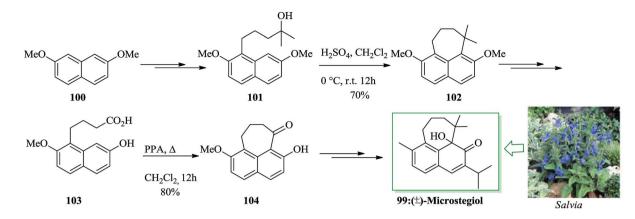


with an aromatic ring. Total synthesis of clavilactone D **89** was initiated from aldehyde **90**. In this route, substrate **92** was synthesized *via* the aldol reaction of chiral lactone **91**<sup>97</sup> with aldehyde **90**.<sup>98</sup> Next, iodo etherification followed *via* FC-type

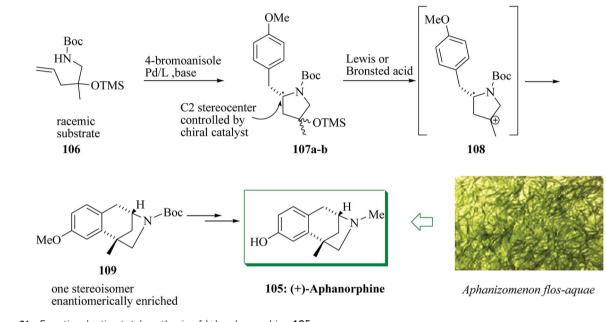
cyclization of alkenyl alcohol **92** with iodine gave the fused polycyclic compound **94**. Finally compound **94** provided clavilactone D **89** upon several steps (Scheme 18).<sup>96</sup>

Hainanensine **95**, racemic, is a structurally distinctive minor alkaloidal constituent identified by Liang and Sun in 1981 from the antileukemia plants *Cephalotaxus hainanensis* and *C. fortunei*.<sup>99</sup> Hainanensine **95** contains marginal antitumor properties,<sup>99</sup> and also Yin *et al.* demonstrated that saturated derivatives and other structural analogues of **95** have demonstrated a series of remarkable biological properties.<sup>100</sup> A simple total synthesis of hainanensine **95** through an efficient acidcatalyzed rearrangement/FC annulation cascade, was developed. Relying on this method, initially, enone **97** was synthesized from 3,4-dihydroisoquinoline **96** after several steps. Upon refluxing enone **97** in formic acid, compound **98** was provided. Seemingly, the construction of rearranged annulation product **98** resulted from more FC-type cyclization-dehydration of the initial ring-expansion intermediates **I** and **II**. Subsequently, compound **98** provided hainanensine **95** upon several steps (Scheme 19).<sup>101</sup>

A rearranged abietane including the cyclohepta[de]naphthalene nucleus, microstegiol **99**,<sup>102-107</sup> was extracted from roots of a wide range of plants of the genus *Salvia*. Microstegiol itself was exhibited to contain antileukemic and modest antibacterial properties.<sup>102,105-107</sup> For the total synthesis of microstegiol **99**, firstly 2,7-dimethoxynaphthalene **100** was converted into compound **101** upon several steps. Exposing tertiary alcohol **101** to sulfuric acid in dichloromethane induced its transformation to 7,7-dimethyltetra-hydrocyclohepta[de]naphthalene **102** (70% yield).<sup>108</sup> Next, compound **102** afforded compound **103** upon



Scheme 20 Total synthesis of  $(\pm)$ -microstegiol 99.

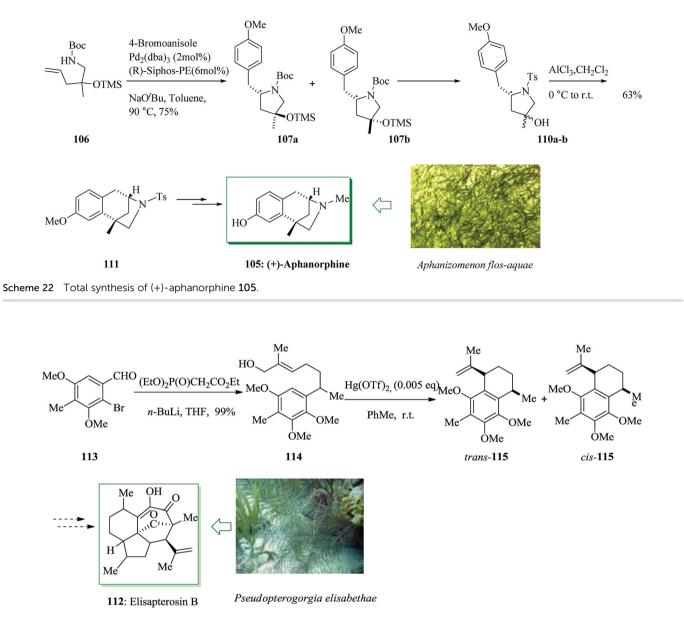


Scheme 21 Enantioselective total synthesis of (+)-aphanorphine 105

several steps. Subsequently, **103** using polyphosphoric acid (PPA) in dichloromethane under reflux conditions was transformed to **104** with 80% yield. Afterwards, compound **104** led to the construction of  $(\pm)$ -microstegiol **99** in 64% yield upon several steps (Scheme 20).<sup>108</sup>

The tricyclic alkaloid aphanorphine was extracted from the blue-green alga *Aphanizomenon flos-aquae* by Clardy and Shimizu in 1988.<sup>109</sup> Aphanorphine shares common structural features with other bioactive benzomorphan alkaloids, for example eptazocine, pentazocine and morphine.<sup>110-114</sup> A short enantioselective synthesis of (+)-aphanorphine **105** was accomplished in 10 steps and 13% overall yield. A racemic  $\gamma$ -aminoalkene derivative was converted into a 1 : 1 mixture of enantiomerically enriched diastereomers employing an enantioselective palladium-mediated carboamination. Next, this mixture was transformed into an enantiomerically enriched masked aphanorphine derivative *via* a FC reaction to make a quaternary all-carbon stereocenter. The catalyst-controlled enantioselective palladium-mediated carboamination reaction of racemic substrate **106** was applied to set the C2 stereocenter of pyrrolidines **107a,b**, thus affording a mixture of enantiomerically enriched diastereomers. Subsequently, this mixture of diastereomers was transformed to **109** in a FC alkylation that proceeded through intermediate carbocation **108**, and formed the all-carbon quaternary stereocenter in an enantio-convergent carbon–carbon bond-forming stage. A two-step sequence of reduction and demethylation was used to transform **109** to aphanorphine (Scheme 21).<sup>115</sup>

The coupling reaction of **106** and 4-bromoanisole was explored for asymmetric alkene carboamination reactions.



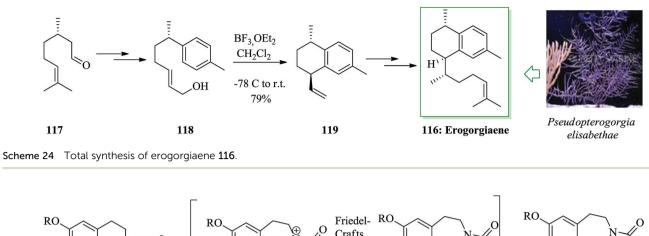
Scheme 23 Total synthesis of elisapterosin B 112.

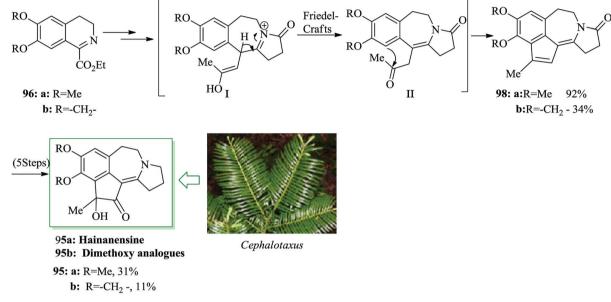
Based on the optimal conditions, this transformation gave a 1 : 1 mixture of diastereomers **107a** and **107b**, which afforded **110a,b**. The enantio-convergent intramolecular FC alkylation reaction of the mixture of diastereomers **110a,b** afforded **111** in 63% yield.<sup>114</sup> Meanwhile, the transformation of **107a,b** to **109** failed. The *N*-tosylated derivative **111** formerly acted as an intermediate in Zhai's syntheses of aphanorphine.<sup>114,116</sup> Afterwards, compound **111** was converted into (+)-aphanorphine **105** upon several steps (Scheme 22).<sup>115</sup>

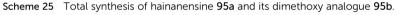
Elisapterosin B **112** was extracted from the gorgonian coral *Pseudopterogorgia elisabethae* and exhibited antitubercular properties, preventing the growth of *Mycobacterium tuberculosis* H37Rv.<sup>117</sup> As part of a method toward the formation of the antitubercular agent elisapterosin B, it was synthesized from two chiral, non-racemic olefinic substrates and their diastereoselective ring closure, achieved in the presence of Hg salts. In this pathway, the synthesis was initiated from aldehyde **113**,<sup>118</sup>

which was converted into **114** upon several steps. The asymmetric construction of **115** was achieved through an intramolecular FC alkylation reaction of **114**. Harmata and coworkers synthesized elisapterosin B **112** by employing benzothiazine chemistry along with  $Hg(OTf)_2$ -mediated diastereoselective intramolecular FC alkylation (Scheme 23).<sup>119</sup>

Erogorgiaene **116**, together with other structurally relevant diterpenes, has been extracted<sup>120</sup> from the West Indian gorgonian octocoral *Pseudopterogorgia elisabethae*. The formal total synthesis of the antitubercular agent erogorgiaene has been accomplished in 12 steps with an overall yield of 14.2%, initiating from (*S*)-(–)-citronellal **117**. This synthetic method includes an enamine-catalyzed 1,4-addition, an aldol condensation, dehydrogenation, a Wittig olefination reaction, intramolecular FC cyclization, TEMPO-BAIB-catalyzed oxidation, and Evans auxiliary-induced diastereoselective methylation. In this route, total synthesis of erogorgiaene **116** was started from







(*S*)-(–)-citronellal **117**, which was converted into allylic alcohol **118** upon several steps. Next, compound **118** was exposed to intramolecular FC cyclization reaction<sup>62,119</sup> with BF<sub>3</sub>·OEt<sub>2</sub> in dry CH<sub>2</sub>Cl<sub>2</sub> to provide bicyclic compound **119** in 79% yield with moderate *de* (8 : 1). Afterwards, erogorgiaene **116** was provided from compound **119** upon several steps (Scheme 24).<sup>121</sup>

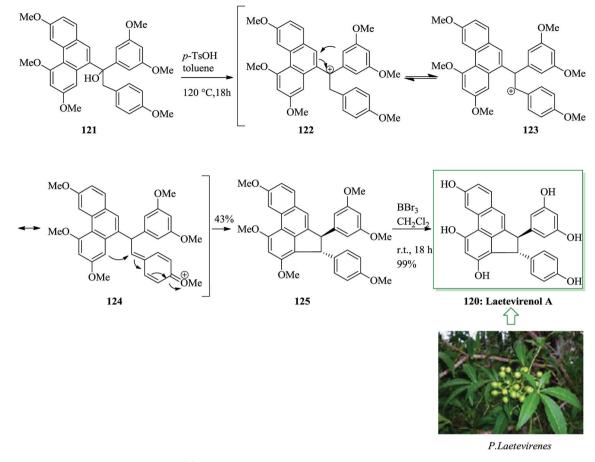
The *Cephalotaxus* genus is a member of the *Cephalotaxaceae* group of conifers. They are a fruitful source of several naturally occurring compounds, especially terpenoids (abietanes, troponoids) and alkaloids (cephalotaxine esters), which often exhibit medicinal activities, particularly in the anticancer area.<sup>122</sup> An uncommon rearrangement-annulation cascade has been demonstrated by Li and co-workers for the total synthesis of hainanensine **95a** and its dimethoxy analogue **95b**.<sup>101</sup> Initially, isoquinolines **96a/96b** provided the intermediate **I** followed by FC cyclization on the residual ketone **II** toward tetracyclic intermediates **98a** and **b**. Five more steps were necessary to attain hainanensine **95a** and its dimethoxy analogues **95b** (Scheme 25).<sup>123</sup>

Laetevirenol A **120**, together with laetevirenol B–E, was extracted from the stems and roots of *Parthenocissus laetevirens* in 2008 by Pan and co-workers.<sup>124</sup> Laetevirenol A exhibits potent

antioxidant properties, probably because of the existence of a phenanthrene scaffold serving as a free radical scavenger.<sup>125-133</sup> The one-pot dehydration/FC alkylation of **121** with *p*-toluenesulfonic acid in toluene provided the corresponding *trans*-cyclized product **125** as a single isomer. Finally, global demethylation of **125** with boron tribromide in dichloromethane afforded laetevirenol A **120** in quantitative yield (Scheme 26).<sup>134</sup>

The *Leucetta* alkaloids, a class of 2-aminoimidazole alkaloids,<sup>135</sup> were extracted from sponges of the *Clathrina* and *Leucetta* groups.<sup>136–138</sup> Structurally, these alkaloids include at least one oxygenated benzyl group. Kealiinines A–C **126a–c**<sup>139</sup> were extracted by the Proksch group in 2004, together with naamidine H and naamine G. It should be mentioned that no biological activity has been stated by Proksch for **126b** and **126c**,<sup>139</sup> but kealiinine A **126a** was found to be potent in the brine shrimp toxicity test.<sup>139</sup> Concise total syntheses of the *Leucetta*obtained alkaloids, kealiinines A–C, were achieved through an intramolecular FC dehydration sequence of a bis benzylic diol. For the synthesis of kealiinine C **126c**, the diol **128c** was synthesized from 1-methyl-4,5-diiodoimidazole **127** upon several steps. Pleasantly, the reaction between unpurified **128c** 

8

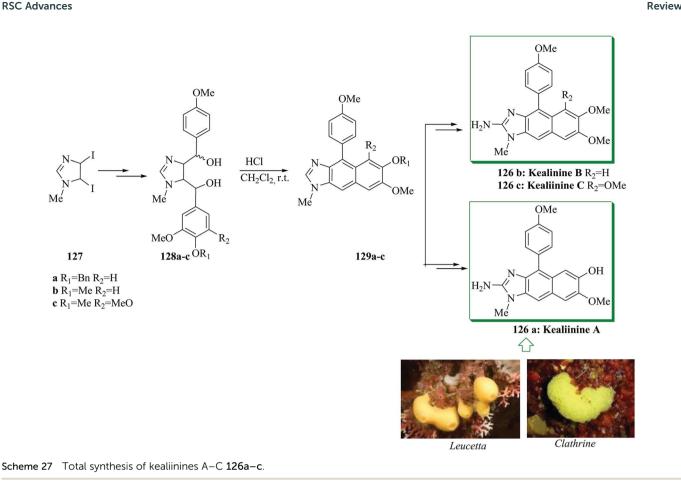


Scheme 26 Total synthesis of laetevirenol A 120.

and hydrochloric acid led to an intramolecular FC cyclizationdehydration sequence to give the naphthimidazole **129c**. Subsequently, compound **129c** provided kealiinine C **126c** upon several steps. Next, it was tried to make the other two group members, kealiinine A **126a** and B **126b**. Their syntheses follow largely the similar methodology, simply substituting the suitable benzaldehyde for **126a** and **126b**, affording the target products after seven steps with 11% overall yield and six steps with 21% overall yield, respectively, starting from compound **127** (Scheme 27).<sup>140</sup>

The dibenzo[a,d]cycloheptene scaffold is a kind of polycyclic framework that occurs broadly in various naturally occurring compounds.<sup>141,142</sup> One of the most direct approaches for the formation of this scaffold is the intramolecular FC acylation reaction.<sup>143,144</sup> Another common synthetic pathway is the equivalent acid-mediated FC alkylation reaction of olefins.<sup>145</sup> The natural diptoindonesin D **130** and pauciflorial F **131** have been extracted from the acetone extract of the tree bark of *Hopea dryobalanoides*<sup>146</sup> and the stem barks of *Vatica pauciflora*<sup>147</sup> respectively. Yang and co-workers established an efficient method for the construction of several dibenzo[a,d] cycloheptenes through FC alkylation reaction. Furthermore, utilizing this approach as the main stage, this group demonstrated short and valuable pathways to form diptoindonesin D **130** and pauciflorial F 131. The formation of diptoindonesin D 130 and pauciflorial F 131 was started with the construction of the biaryl alcohol 134 from the lithiated form of the Me-masked resveratrol bromide 132 and 3,5-dimethoxy benzaldehyde 133.<sup>148</sup> With this key intermediate as the substrate, directly, the polyfunctionalized indene 135 was obtained in the presence of trifluoromethane sulfonic anhydride (Tf<sub>2</sub>O) as a catalyst *via* FC alkylation reaction. Next, oxidation reaction of compound 135 followed *via* deprotection afforded paucifloral F. On the other hand, compound 134 provided ketone 136 with 90% yield.<sup>149</sup> Then, the carbocyclic unit structure of diptoindonesin D 130 was constructed through FC alkylation reaction, providing 137 in 76% yield. Upon oxidation and deprotection of 137, the target natural diptoindonesin D 130 was provided (Scheme 28).<sup>150</sup>

Delle Monache *et al.* described the extraction of a dihydrophenanthrapyrane natural product from the root of *Clusia paralycola*, paralycolin A, that shows cytotoxicity against P388 cells and KB.<sup>151</sup> The structurally relevant natural product cedrelin A **138** was extracted from the bark of *Cedrelinga catenaeformis* Duke by Kakisawa *et al.* in 1991.<sup>152</sup> Paralycolin B has been extracted merely as a methylated derivative **139** upon reaction of a resultant mixture of paralycolin A and paralycolin B with diazomethane. Furthermore, these naturally occurring compounds usually contain an isopropenyl substituent at the



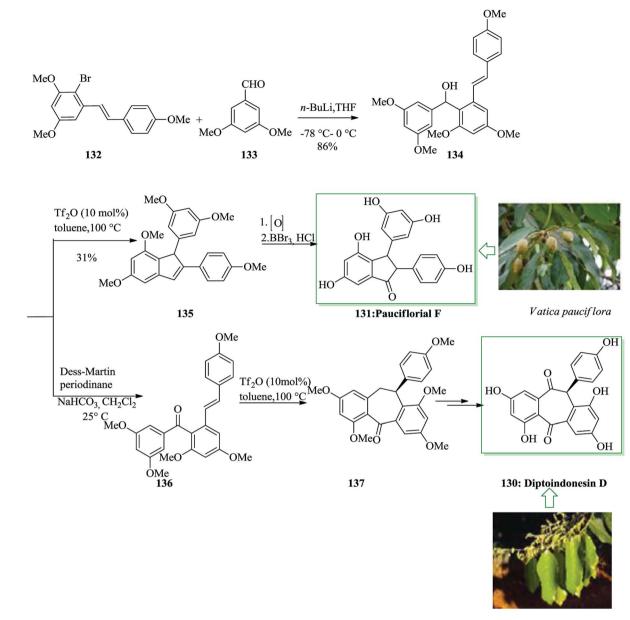
10-position of the 9,10-dihydrophenanthrene framework. Hamada and co-workers in 2013 demonstrated the initial asymmetric total syntheses of cedrelin A and methylated paralycolin B, employing mediated enantioselective intramolecular FC allylic alkylation reaction of phenols as the main step.<sup>153</sup> Total synthesis of cedrelin A 138 was initiated from aldehyde 140,<sup>154</sup> which was transformed into allyl carbonate derivative 141 upon several steps. Next, the mediated enantioselective intramolecular FC allylic alkylation of 141 was examined. With 1.5 equiv. of potassium acetate, the reaction progressed easily to give compound 142 in 98% yield (94% extracted yield) with 66% enantioselectivity. Finally, intermediate 142 has been converted into cedrelin A 138 in 74% yield (12 steps from 140 and 16.5% overall yield) (Scheme 29).153

Subsequently, the asymmetric total synthesis of methylated paralycolin B 139 was accomplished. First, compound 144 was synthesized from 6-bromoveratraldehyde 143 upon several steps. Enantioselective intramolecular FC allylic alkylation reaction of 144 using  $Pd(dba)_2$  and (R,R)-chiral ligand in dichloromethane/methanol mixed solvent afforded 145 in 98% yield with 92% enantioselectivity. Finally, compound 145 provided methylated paralycolin B 139 in 92% yield (10 steps from 143 and 35.2% overall yield) (Scheme 30).154

Members of the genera Ligularia, Euryops, Senecio, and Psacalium of the Asteraceae group are generally dispersed plants rich in furanoeremophilane compounds.155-157 This group of sesquiterpenes includes a linearly fused C6-C6-C4O tricyclic

 $(C_{12})$  scaffold having three methyl groups linked to the 4, 5, and 11 positions. An asymmetric total synthesis of the furanoeremophilane sesquiterpene (+)-9-oxoeuryopsin 146 was reported from 2-methyl-2-cyclohexen-1-one 147 in 7% overall yield. This method was accomplished in seven facile synthetic reactions involving conjugate addition-enolate trapping, imidazoylthiocarbonylation reaction, bisdethiocarbonylation, masked cyanohydrin construction-hydrolysis, dehydration, alkaline hydrolysis, and also FC cyclization. The total synthesis of 146 was initiated from 2-methyl-2-cyclohexen-1-one, which was converted into acid 148 upon several steps. Next, the crude acid 148 was transformed into (+)-9-oxoeuyopsin 146 in 59% overall yield via an intramolecular FC cyclization using tin(IV) chloride as the Lewis acid of the spontaneously provided acid chloride (PCl<sub>5</sub> in benzene at room temperature) (Scheme 31).<sup>158</sup>

Compound 149, an unnamed guinone methide diterpenoid, was initially extracted from the root bark of Bobgunnia madagascariensis.159 A convergent pathway was established to form an antifungal tricyclic o-hydroxy-p-quinone methide diterpenoid 149 and analogues. In 2013, Yang and co-workers finished the total synthesis of the antifungal tricyclic o-hydroxy-p-quinone methide diterpenoid 149. A Stille reaction was applied to introduce the allyl substituent as a protected 2-hydroxyethyl side chain. This group established a BBr3-catalyzed one-pot bisdemethylation/intramolecular FC alkylation reaction to make the tricyclic molecular framework and then completing the total synthesis of 149.160 Compounds 149 and 155 exhibited active



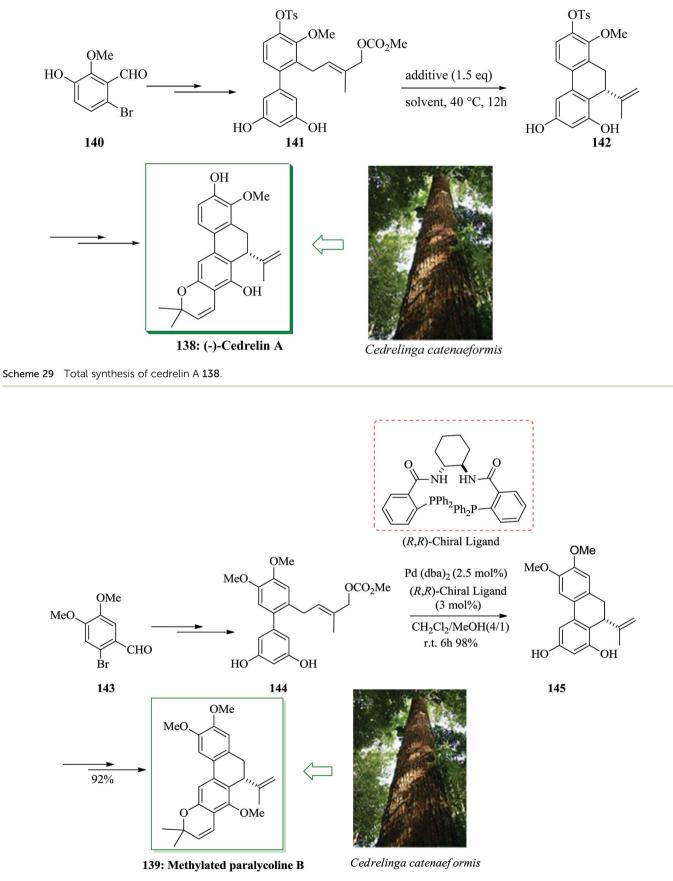
Scheme 28 Total synthesis of diptoindonesin D 130 and pauciflorial F 131.

cytotoxicity against several strains of pathogenic yeasts tested, but etherification of the 2-hydroxyethyl group resulted in significantly attenuated activity. This total synthesis began from commercially available bromobenzene, which in four steps is converted into 3,4-dimethyoxytoluene **150**.<sup>161</sup> The latter after several steps gave the primary alcohol **151**. The application of the BBr<sub>3</sub>-catalyzed one-pot reaction resulted in the completion of the synthesis of **149**. Therefore, reaction of **150** with BBr<sub>3</sub> resulted in bis-demethylation and cyclization in a one-pot manner to generate a tricyclic catechol that, upon oxidation with Ag<sub>2</sub>O, afforded **149**. Selective monoetherification of **149** with dimethoxymethane afforded the analogue **152** (Scheme 32).<sup>160</sup>

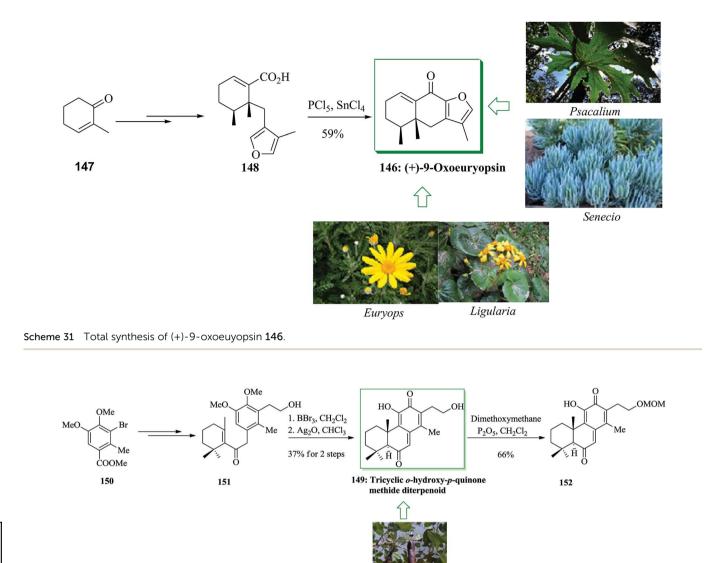
The same group synthesized analogue 155 from carboxylic acid 153.<sup>160</sup> Compound 153 afforded enone 154 upon several

steps. Compound **155** was provided *via* BBr<sub>3</sub>-catalyzed bisdemethylation and intramolecular FC alkylation followed by oxidation of the resulting catechol with Ag<sub>2</sub>O in one-pot fashion. Compounds **149** and **155** were found to be particularly effective against all yeast strains, including *Candida krusei*, *Candida glabrata*, and *Candida parapsilosis* strains (Scheme 33).<sup>161</sup>

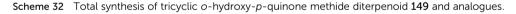
Dimeric epipolythiodiketopiperazine alkaloids are an intriguing group of fungal metabolites notable for their complex molecular scaffolds and biological properties.<sup>162,163</sup> (+)-Bionectins A **156** and C **157**<sup>164</sup> were extracted in 2006 by Zheng and co-workers from fungi of the *Bionectra byssicola* species. (+)-Bionectins A (+)-**156** showed important bacterio-static activity against methicillin-resistant and quinolone-resistant *Staphylococcus aureus* Gram-positive eubacteria. The



Scheme 30 Total synthesis of methylated paralycolin B 139.

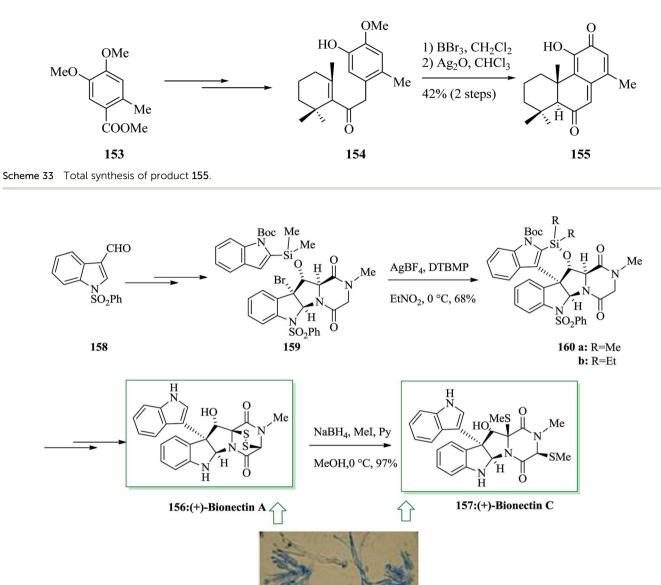


BObgunnia madagascariensis

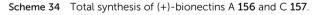


short and effective total synthesis of (+)-bionectins A and C was developed by Movassaghi and co-workers in 2013. This route leading to the total synthesis of these naturally occurring compounds features a new and scalable approach for the synthesis of erythro- $\beta$ -hydroxytryptophan amino acid. It involves an intramolecular FC reaction of a silyl-tethered indole and a new mercaptan reagent for epipolythiodiketopiperazine (ETP) synthesis that can be unraveled under very mild conditions. Based on this approach, firstly indole-3-carboxaldehyde **158** was converted into the corresponding silyl-tethered indole adduct **159** in 74% yield in several steps. Gratifyingly, a silvercatalyzed intramolecular FC reaction progressed easily in nitroethane to give the C3-(3'-indolyl)-silacyclic product **160a** in 68% yield. The structure of a diethyl silyl variant **160b**, provided during optimization studies, was confirmed *via* X-ray analysis. Compound **160a** gave the target natural product (+)-bionectin A **156** in 81% yield after several steps. Reductive methylation of (+)-bionectin A **156** with NaBH<sub>4</sub> and methyl iodide in pyridine and MeOH gave (+)-bionectin C **157** in 97% yield (Scheme 34).<sup>165</sup>

Ammosamides A–C are marine alkaloids that were initially extracted from the marine *Streptomyces* strain CNR-698 by Fenical and co-workers in 2009.<sup>166</sup> Except for ammosamide D, these natural products have a characteristic pyrroloquinoline unit framework, in which the tricyclic structure contains a pyrrole scaffold bridging the C4 and C5 positions of a quinoline ring system.<sup>167,168</sup> Ammosamides A and B exhibit important cytotoxicity against human colon adenocarcinoma HCT-116 cells.<sup>166</sup> A total synthesis of ammosamide B **161**, a member of the pyrroloquinoline alkaloid group extracted from marine *Streptomyces*, was developed. The characteristic unit tricyclic



Bionectra byssicola



structure of **161** was generated through a unique, tandem FC reaction sequence to transform the symmetric tetra-amino functionalized benzene derivative **162** into the tricyclic pyrroloquinoline product **163**. The reaction was initiated with an acid-promoted tandem FC reaction of the diamino and dinitro functionalized benzene derivative **162** to form the tricyclic pyrroloquinoline **163**. The required substrate **162** was easily synthesized from *m*-dichlorobenzene in a two-step sequence.

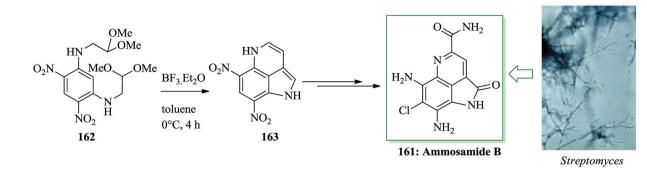
Subsequently, a number of acids were applied to examine the tandem FC reaction and the results demonstrated that the reaction improved in the presence of trifluoroacetic acid (TFA) but does not provide either of the anticipated FC products, only an unidentified oligomeric material. Reactions using the Lewis

acids aluminium chloride and titanium tetrachloride also resulted in the construction of insoluble oligomers and no reaction (**162** was recovered quantitatively) happened once the less reactive Lewis acids zinc chloride, tin tetrachloride and Yb(OTf)<sub>3</sub> were used. The remarkable observation that an indole (side-product) was provided in 52% yield when BF<sub>3</sub>·Et<sub>2</sub>O in dichloromethane was used provided the motivation for a broader investigation of solvent effects on this method. When CH<sub>3</sub>CN or THF was utilized in the presence of BF<sub>3</sub>·Et<sub>2</sub>O as the solvent for the improvement of the reaction, insoluble oligomers were provided. In contrast, tricyclic pyrroloquinoline **163**, *via* tandem FC product, produced the expected product in 35% isolated yield once toluene was applied as the solvent. While This article is licensed under a Creative Commons Attribution-NonCommercial 3.0 Unported Licence.

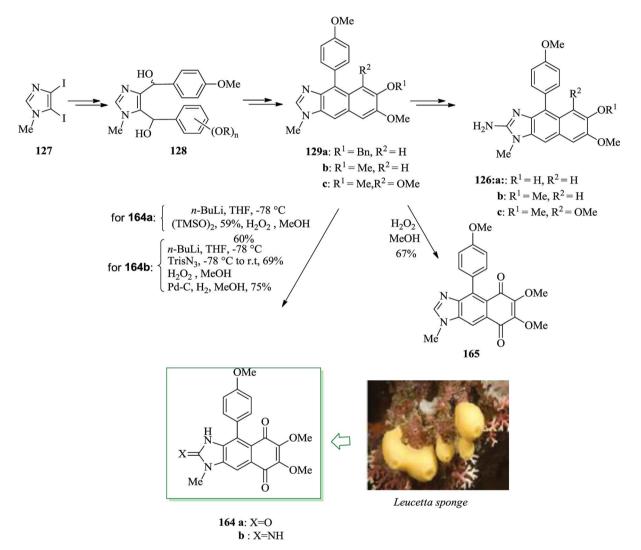
BY-NC

8

Open Access Article. Published on 03 December 2018. Downloaded on 8/26/2025 9:48:33 PM.



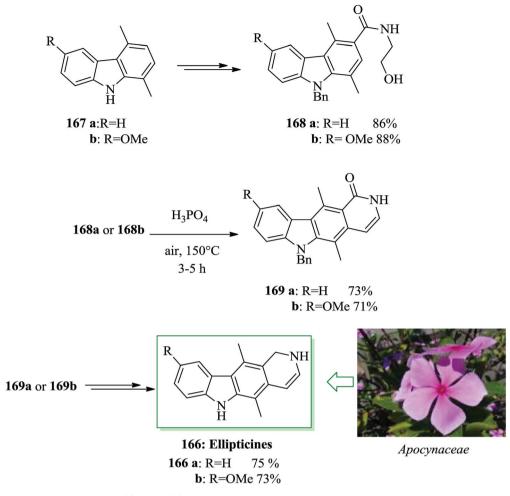




Scheme 36 Total synthesis of kealiiquinone A 164a and the 2-amino congener 164b.

this reaction was low yielding, it was performed efficiently and no significant side products were detected by TLC analysis. Hence, the poor yield of **163** is probably owing to its insolubility during the work-up and purification processes. Consequently, Nagasawa decided to apply the optimal reaction conditions  $(BF_3 \cdot Et_2O \text{ in toluene})$  performing the conversion of **162** to **163** in the synthetic pathway to form ammosamide B **161** and skipped the purification stage. A tandem FC reaction of **162** promoted by  $BF_3 \cdot Et_2O$  in toluene led to the formation of compound **163**. Next, ammosamide B **161** was provided after several steps (Scheme 35).<sup>169</sup>

Marine sponges provide a wide range of structurally unique secondary metabolites with remarkable biological properties.<sup>170,171</sup> Clardy and Scheuer initially extracted kealiiquinone



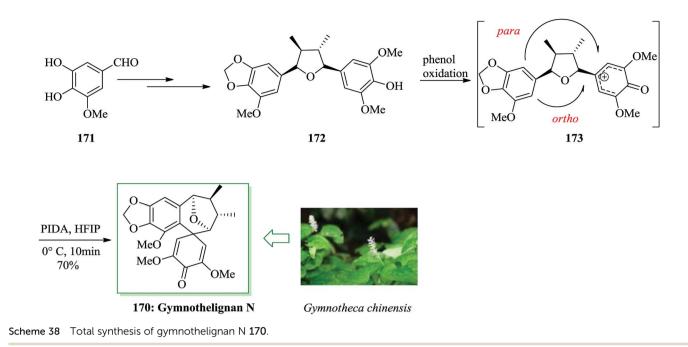
Scheme 37 Total synthesis of ellipticines 166a and 166b.

**164a** from a *Leucetta*-obtained sponge containing a unique imidazobenzoquinone scaffold.<sup>172</sup> Compound **164a** exhibited cytotoxicity, with potentially a unique mechanism of action.<sup>173</sup> Schmitz reported the isolation of 2-deoxy-2-aminokealiiquinone **164b**, but no bioactivity has been reported for this molecule.<sup>174</sup> A brief synthesis for kealiiquinone **164a** and 2-deoxy-2-aminokealiiquinone **164b** was reported in 2013 by Lovely and co-workers. Advanced intermediates including the full naph-thimidazole scaffold are generated *via* FC reaction followed by oxidation. Total synthesis of **164a** and **164b** was started from diiodoimidazole **127**, which was converted into **128** in several steps the latter undergoes sequential FC reaction/dehydration to give compound **129**.<sup>140</sup> Subsequently, compound **129** provided kealiiquinone A **164a** and the 2-amino congener **164b** (Scheme **36**).<sup>175</sup>

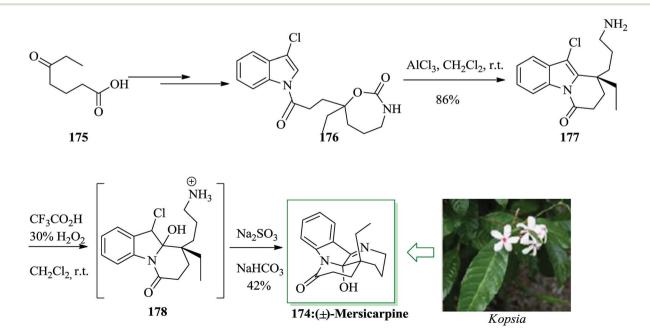
Natural products belonging to the pyrido[4,3-*b*]carbazole group of alkaloids are well represented in the chemical literature.<sup>176-185</sup> Since the first extraction of ellipticine **166a** and 9-methoxyellipticine **166b** in 1959 by Goodwin and co-workers,<sup>177</sup> several other compounds in this group have been extracted from *Apocynaceae* plants.<sup>178</sup> The antineoplastic activity of ellipticine has been developed largely because of the importance of

DNA intercalation and inhibition of topoisomerase II.181 Nagarajan and co-workers in 2014 achieved an convenient synthesis of ellipticine 166a and 9-methoxyellipticine 166b in 7 steps with 23% and 25% overall yields, respectively.186 This synthetic method uses a key phosphoric acid-catalyzed FC cyclodehydration for the generation of the pyridine unit. Total synthesis began from 1,4-dimethylcarbazoles (167a and 167b). In this approach, phosphoric acid-catalyzed FC cyclodehydration was employed as a key step for the construction of pyrido [4,3-b] carbazole alkaloids. This synthesis was started from 1,4-dimethylcarbazoles 167a and 167b and afforded the corresponding amides 168a and 168b after several steps. Amides 168a and 168b, were subjected to the critical FC cyclodehydration with subsequent pyridine ring construction. Compounds 168a and 168b in the presence of phosphoric acid provided dihydropyridocarbazolones 169a and 169b in 73% and 71% yields, respectively. Finally, compounds 169a and 169b were converted into ellipticines 166a and 166b after several steps (Scheme 37).186

In 2012, Xu and co-workers extracted various tetrahydrofuran-type lignans, gymnothelignans A–O, from *Gymnotheca chinensis* Decne, a broadly applied perennial



Chinese herb of Saururacese.187 Among them, gymnothelignan N attracted the interest of the authors owing to its unique 10,11-benzospiro[5.6]dodec-13,15-dien-14-one scaffold across a tetrahydrofuran ring, which is extraordinary in lignans. Total synthesis of gymnothelignan N 170 was achieved in 13 steps from 5-hydroxyvanilin 171 in 6.7% overall yield. In this approach, a syn Evans aldol reaction, an intramolecular hydrogenative dehydration reaction, and a phenol oxidative dearomatization/FC reaction were considered as the key steps. Total synthesis was started from 5-hydroxyvanilin 171, which was converted into compound 172 via several steps. Using the key intermediate 172, the step was performed to generate the critical bioinspired oxidative FC reaction. However, there are many cases of phenol oxidative dearomatization/FC reaction used in natural product synthesis;<sup>188–191</sup> in this substrate, it is considered to be challenging for two aspects. The first one is the desired intramolecular FC reaction, which happens across the tetrahydrofuran ring to generate the seven-membered ring with an all-carbon quaternary center as linkage, while the presumed cation intermediate **173** is constructed *in situ*. The second one is the site selectivity of the FC reaction. The *para* and *ortho* positions of the methoxyl substituent are in competition in this reaction.<sup>190,191</sup> Hence, the usual phenol oxidation hypervalent iodine(m) reagents, for example iodobenzene diacetate (PIDA),

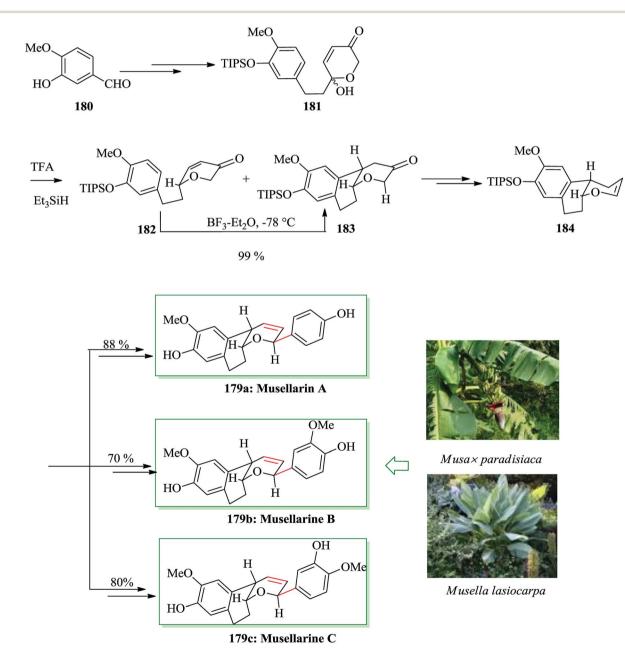


Scheme 39 Total synthesis of mersicarpine 174.

were considered and PIDA in hexafluoroisopropyl alcohol (HFIP) resulted in the FC product in 70% yield. The selectivity of the *ortho/para* ratio was found to be 2.6/1. The *ortho* FC product, so-called gymnothelignan N **170**, could be extracted using careful flash chromatography employing dichloromethane/ methanol as the eluent system. Thus, the total synthesis of gymnothelignan N **170** was achieved upon several steps from 5-hydroxyvanilin **171** in 6.7% overall yield (Scheme 38).<sup>192</sup>

Mersicarpine 174, a structurally fascinating monoterpene indole alkaloid, was extracted from the *Kopsia* species of plants by Kam and co-workers in 2004.<sup>193</sup> This uncommon tetracyclic natural product contains a typical seven-membered cyclic imine and a  $\delta$ -lactam around a completely functionalized hemiaminal stereogenic center. A short total synthesis of mersicarpine 174

was accomplished using a cyclic carbamate for the construction of a tertiary carbocation. The main step includes intramolecular FC alkylation with this carbocation for the formation of a quaternary carbon center and a subsequent oxidation and cyclization cascade for the generation of a seven-membered cyclic imine. This approach permitted for a rapid synthesis of mersicarpine from a simple intermediate employing straightforward chemical operations in a one-pot fashion. The total synthesis of mersicarpine **174** was initiated from compound **175**, which afforded compound **176** upon several steps. The key FC alkylation reaction was employed in the total synthesis of mersicarpine. In this route, compound **176** was reacted with  $AlCl_3$ , a mixture of trifluoroacetic acid and  $H_2O_2$  *via* FC reaction to provide the product **177** in 86% yield. Upon oxidation of **177** 



Scheme 40 Total synthesis of musellarins A–C 179a–c.

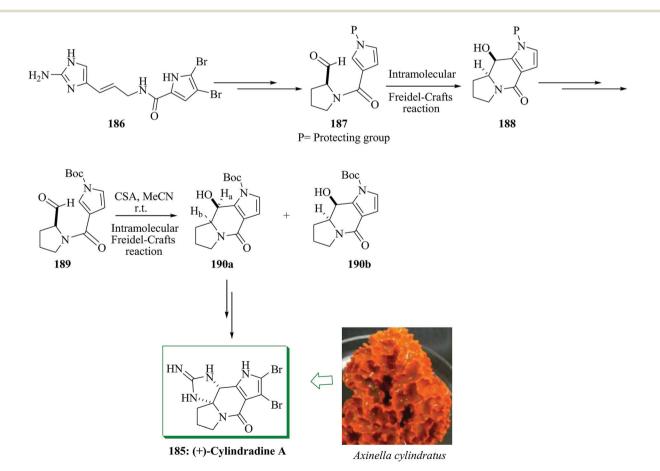
and acidity optimization with Na<sub>2</sub>SO<sub>3</sub>/NaHCO<sub>3</sub>, mersicarpine 174 was provided in 42% yield. As a result, the natural product was obtained in 25% overall yield (Scheme 39).<sup>194</sup>

Naturally occurring diarylheptanoids, a group of secondary plant metabolites, 195,196 exhibit active biological properties, for antioxidant, antibacterial, example anticancer, antiosteoporosis, antifungal and antihepatotoxicity activity.195,196 Among them, musellarins A-C represent an uncommon structural group owing the presence of the rare bicyclic tetrahydropyran scaffold. Musellarin A was extracted in 2002 by Kinghorn<sup>197</sup> and co-workers from hybrid plant fruits of Musa x paradisiaca in Peru.<sup>197,198</sup> In 2014, Tong and co-workers reported the first total syntheses of musellarins A-C in 15-16 steps.<sup>199</sup> The key synthetic features are an Achmatowicz rearrangement, a Kishi reduction, and FC cyclization to make the tricyclic scaffold and Heck coupling reaction of aryldiazonium salts to introduce the aryl substituent into the dihydropyran in a 2,6trans method in the last step of the synthesis. Total syntheses of musellarins A-C were started from isovanillin 180, which was converted into dihydropyranone hemiacetal 181. Kishi reduction of 181 with a combination of trifluoroacetic acid and triethylsilane afforded a 2:1 mixture of dihydropyranone 182 and a surprising FC cyclization adduct 183 in 78% combined yield. In this step the intramolecular FC cyclization reaction of dihydropyranone **182** happened in the presence of  $BF_3 \cdot Et_2O$  to give the tricyclic compound 183. Compound 183, after several steps,

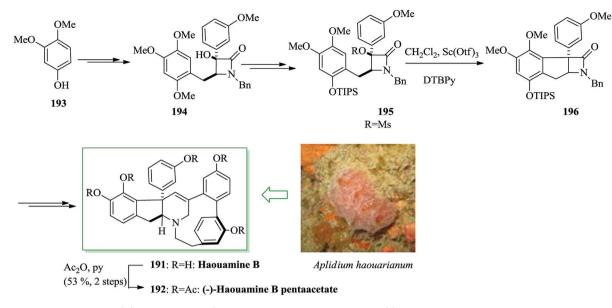
afforded the key intermediate enol ether **184**. Finally, compound **184** was converted into musellarins A–C in 88%, 70% and 80% yields upon several steps (Scheme 40).<sup>199</sup>

Polycyclic pyrrole-imidazole alkaloids (PIAs) are oroidinobtained marine naturally occurring compounds<sup>200,201</sup> with various and complex frameworks. Most of these alkaloids exhibit multiple biological properties, including immunosuppressive, antitumor and adrenoceptoragonistic properties.<sup>202,203</sup> Nagasawa and co-workers in 2014 developed a total synthesis of (+)-cylindradine A through intramolecular FC-type cyclization reaction of pyrrole-aldehyde and oxidative cyclization of tricyclic pyrrolopyrrolidine-guanidine with hypervalent iodine to make the cyclic guanidine structure involving the N,N-aminal scaffold.<sup>204</sup> The total synthesis of cylindradine A was initiated from oroidin 186, which was converted into 187 in several steps. Compound 187 containing aldehyde and 4-carbamoylpyrrole functional substituents gave compound 188 via an intramolecular FC-type reaction. Next, compound 188 was converted into 189 after several steps. The FC adduct 190 was provided in 82% yield as a diastereomeric mixture in a ratio of ca. 7:1 (190a: 190b) via reaction of 189 with camphorsulfonic acid (CSA). Subsequently, (+)-cylindradine A 185 was provided from compound 190a in 58% yield upon several steps (Scheme 41).204

Haouamines were extracted from a tunicate, *Aplidium haouarianum*, on the southern coast of Spain by Zubía and coworkers in 2003.<sup>205</sup> Structurally, haouamines have unique



Scheme 41 Total synthesis of (+)-cylindradine A 185.



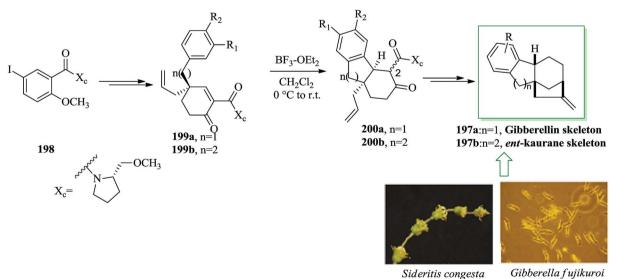
Scheme 42 Total synthesis of (-)-haouamine B 191 and haouamine B pentaacetate 192.

the scandium(III) triflate-catalyzed intramolecular FC alkylation were applied to make the uncommon indane-fused  $\beta$ -lactam 196. In this route, an intramolecular McMurry coupling reaction was considered as the key step. Tokuyama and co-workers initiated the synthesis of (-)-haouamine B from 3,4-dimethoxyphenol 193, which after several steps was transformed into substrate 194. To avoid the unanticipated cyclization, the benzyl group was changed to the more bulky TIPS substituent with the intention of reducing the nucleophilicity of the oxygen atom and a range of Brønsted/Lewis acids were explored. Upon hydrogenolysis of the benzyl ether, introduction of the TIPS substituent followed by mesylation of the tertiary alcohol gave substrate 195 for the FC alkylation. Compound 195 gave indenoβ-lactam 196. The reaction was accomplished using 2,6-di-tertbutylpyridine (DTBPy). The corresponding compound 196 was

provided in 80% yield with the labile TIPS substituent remaining intact. Finally, compound 196 afforded (-)-haouamine B 191 and haouamine B pentaacetate 192 after several steps (Scheme 42).206

Gibberellin<sup>207</sup> and *ent*-kaurane derivatives,<sup>208</sup> are naturally occurring diterpene compounds,<sup>209</sup> having a complex tetracyclic molecular framework presenting a bicyclo[3.2.1]octane system.<sup>210,211</sup> Both gibberellins and ent-kauranes contain remarkable biological properties. Notably, the gibberellins were found to be plant hormones and growth regulators.<sup>212-214</sup> Gibberellins were initially extracted from the fungus Gibberella fujikuroi and ent-kaurane diterpenoids were also extracted from the whole plant Sideritis congesta P. H. Davis & Hub.-Mor.<sup>215</sup>

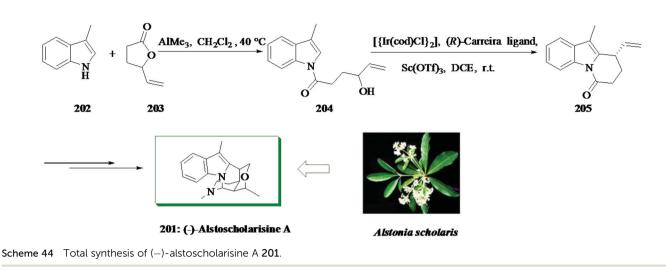
Malachowski and co-workers in 2015 developed an approach for the formation of tricarbocyclic scaffold via an extension of



Scheme 43 Total synthesis of gibberellins 197a or ent-kauranes 197b.

8

#### Review



the asymmetric Birch-Cope sequence along with intramolecular FC alkylation reaction.<sup>216</sup> An asymmetric method for the synthesis of the tetracarbocyclic framework of the gibberellins 197a or the ent-kauranes 197b was demonstrated by the same group.<sup>216</sup> There is a tetracarbocyclic ring system in the unit of both the gibberellins and the ent-kauranes 197b. The asymmetric formation of tetracyclic diterpene frameworks began from the 5-iodo-salicylate derivative 198. Upon several steps, compounds 199a and 199b were produced. Conversion of these products, 199a and 199b, into the tetracarbocyclic scaffold of the gibberellins 197a or ent-kauranes 197b was achieved via FC alkylation in the presence of BF<sub>3</sub>·Et<sub>2</sub>O, which was clearly one of the most charming conjugate additions of an aromatic nucleophile to a cyclohexenone electrophile.217,218 The cyclization happened to generate the cis isomers 200a and 200b, but most of the products were isolated as C-2 epimers, usually as a 1:1 mixture. Finally, the tetracarbocyclic framework of the gibberellins 197a or ent-kauranes 197b was obtained from 200a and 200b upon several steps (Scheme 43).216

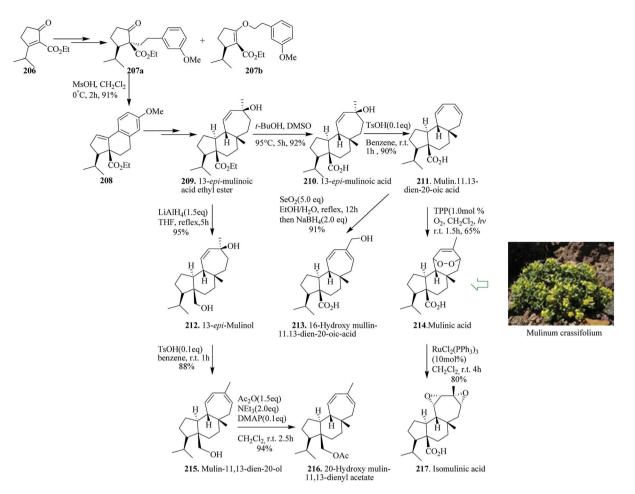
The alstoscholarisines<sup>219</sup> are some fascinating monoterpenoid indole alkaloids having a complex pentacyclic structure extracted from Alstonia scholaris by Luo and co-workers. In 2016, Liang and co-workers reported an asymmetric total synthesis of (-)-alstoscholarisine A 201 in 10 steps.<sup>220</sup> This group demonstrated a short and highly asymmetric total synthesis of (-)-alstoscholarisine A 201, an extracted monoterpenoid indole alkaloid having important bioactivity in promoting adult neuronal stem cells proliferation. A highly asymmetric (99% ee), intramolecular iridium-mediated FC alkylation reaction between indole 204 and a secondary allylic alcohol was used to create the first stereogenic center upon which the other three contiguous chiral centers were easily set through an extremely enantioselective tandem 1,4-addition and aldol reaction. Firstly, an acylation reaction between 3-methylindole 202 and 4-vinylbutyrolactone 203 was achieved using trimethylaluminium,221 providing secondary allylic alcohol 204 in 75% yield. Catalytic enantioselective FC alkylation reaction of 204 with the combination of  $[{\rm Ir(cod)Cl}_2], (R)$ -carreira ligand<sup>222-224</sup> and Lewis acid scandium(III) triflate effectively afforded tricycle 205 in 75% yield with 99% ee. The latter was

converted into (–)-alstoscholarisine A **201** in several steps (Scheme 44).<sup>220</sup>

Terpenoids are structurally intriguing molecules with fascinating biological properties.<sup>225</sup> For instance, mulinane diterpenoids, initially extracted in 1990 from the Chilean shrub Mulinum crassifolium, have been applied as traditional folk medicine for the treatment of illnesses, for example, diabetes and bronchial and intestinal disorders.<sup>226-232</sup> Enantioselective total synthesis of mulinane diterpenoids has been accomplished. In this strategy, an intramolecular FC reaction and Birch reduction were considered as the key steps. Based on this pathway, Xie and co-workers in 2017 reported the divergent asymmetric total synthesis of seven mulinane diterpenoids, namely, 13-epi-mulinolic acid (210, 12 steps, 18% overall yield), mulin-11,13-dien-20-oic acid (211, 13 steps, 17% yield), mulinic acid (214, 14 steps, 11% yield), isomulinic acid (217, 15 steps, 8.6% yield), 16-hydroxy mulin-11,13-dien-20-oic acid (213, 14 steps, 15% yield), mulin-11,13-dien-20-ol (215, 13 steps, 15% yield), and 20-hydroxy mulin-11,13-dienyl acetate (216, 14 steps, 14% yield), along with two analogues, namely, 13-epi-mulinol (212, 12 steps, 17% yield) and 13-epi-mulinolic acid ethyl ester (209, 11 steps, 20% yield), from easily accessible cyclopentenone 206.236

Total synthesis began from 3-isopropyl-2-(ethoxycarbonyl) cyclopentenone 206. Compound 206 gave C-alkylation product 207a in 65% yield, together with a 20% yield of the unanticipated O-alkylation product 207b in several steps. Reaction of 207a with methanesulfonic acid led to an intramolecular FC reaction that afforded tricyclic compound 208 in 91% yields. Subsequently, compound 208 gave 13-epi-mulinolic acid ethyl ester 209 upon several steps. Using compound 209, the same group accomplished the enantioselective total synthesis of mulinane diterpenoids through late-stage functional modification or functionalization. Therefore, the synthesis of seven mulinane diterpenoids from the "biogenic precursor" 209 was initially achieved in several steps. The latter was then converted to various natural products, 13-epi-mulinolic acid 210,233 mulin-11,13-dien-20-oic acid 211,234 mulinic acid 214226 isomulinic acid 217,<sup>226</sup> 16-hydroxymulin-11,13-dien-20-oic acid 213,<sup>235</sup> 13-

Review



Scheme 45 Total synthesis of 13-*epi*-mulinolic acid ethyl ester 209, 13-*epi*-mulinolic acid 210, mulin-11,13-dien-20-oic acid 211, 13-*epi*-mulinol 212, 16-hydroxymulin-11,13-dien-20-oic acid 213, mulinic acid 214 mulin-11,13-dien-20-ol 215, 20-hydroxymulin-11,13-dienyl acetate 216 and isomulinic acid 217.

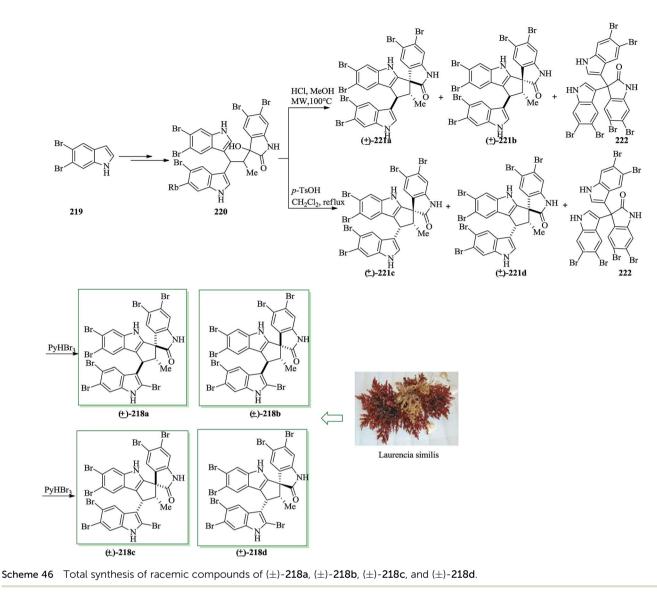
*epi*-mulinol **212**, mulin-11,13-dien-20-ol **215**<sup>229</sup> and 20-hydroxy mulin-11,13-dienyl acetate **216** (Scheme 45).<sup>236</sup>

Similisines A [(+)-218a] and B [(-)-218a], extracted from Laurencia similis in 2013, are the first example of polybrominated spiro-trisindole alkaloids fused to a five-member ring.237 Similisines A and B are trisindole alkaloids that feature a unique polybrominated [5,5] spirooxindole scaffold fused to a fivemembered ring. The first total synthesis of similisines A and B, a pair of unprecedented polybrominated spiro-trisindole enantiomers and all their stereoisomers was achieved in 6 steps from 5,6-dibromoindole 219. In this method, an intramolecular FC cyclization was considered as the key step. Total synthesis of similisines A, B and their stereoisomers started with the construction of trisindole 220, which was obtained from 5,6dibromoindole 219 upon several steps.<sup>238</sup> The five-membered spirocyclic framework of 221 was generated from trisindole 220 via intramolecular FC cyclization. Treatment of 220 with HCl in methanol under microwave irradiation (MWI) provided two racemic isomers  $(\pm)$ -221a (8% yield) and  $(\pm)$ -221b (8% yield), along with the unexpected trisindole 222 in 10% yield. Although the other two corresponding racemic isomers  $(\pm)$ -221c and  $(\pm)$ -221d were detected only in trace amounts, when trisindole 220 was reacted with p-TsOH in DCE under reflux, racemic

isomers (±)-221c and (±)-221d were provided in 7% and 16% yields, respectively, along with trace amounts of racemic (±)-221a and (±)-221b. In this case, the unanticipated trisindole 222 was also provided in 10% yield. Therefore, both the racemic stereo-isomers (±)-221a-(±)-221d could be obtained *via* two abovementioned different conditions. Using the key (±)-221a-(±)-221d, the synthesis of similisines A, B and their stereoisomers was accomplished. Bromination of (±)-221a-(±)-221d with pyridinium tribromide<sup>239,240</sup> afforded the corresponding racemic compounds of (±)-218a (85%), (±)-218b (81%), (±)-218c (85%), and (±)-218d (83%), respectively. On the other hand, (±)-218a was identified to be a racemic mixture of natural similisines A and B (Scheme 46).<sup>237</sup>

## 2.2. Intermolecular alkylations

Brasiliquinones A–C, unique cytotoxic benz(*a*)anthraquinones usually found as angucyclines, were initially extracted from pathogenic species of *Nocardia*. Most of the angucycline antibiotics contain a Me substituent at C-3 while brasiliquinones A–C contain an Et substituent at C-3 providing a unique class of angucyclines. Brasiliquinones B and C are more effective than brasiliquinone A against L I210 tumor cells.<sup>241</sup> Brasiliquinone B

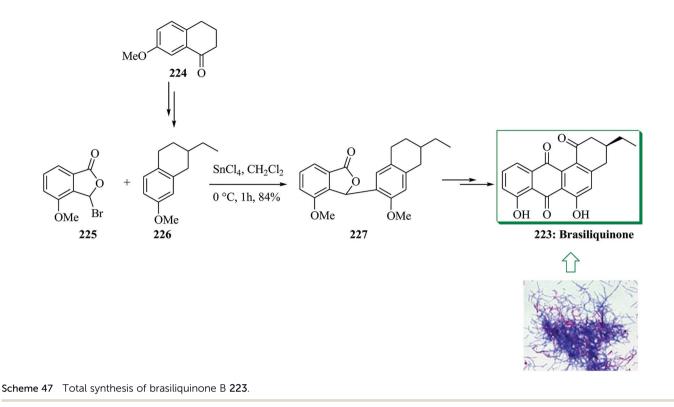


223 has been obtained from 7-methoxy-L-tetralone in several steps making use of an FC alkylation reaction as a main stage. Initially, 7-methoxy-L-tetralone 224 gave the relevant tetralin derivative 226 after several steps. FC alkylation reaction of 226 with 3-bromo-4-methoxyphthalide 225 using SnCl<sub>4</sub> gave the lactone 227 regiospecifically. Finally, compound 227 provided (+)-brasiliquinone B 223 upon several steps (Scheme 47).<sup>242</sup> Deshpande and co-workers in 2002 also used this method for the synthesis of (±)-brasiliquinone B 223.<sup>243</sup>

Methyl 3-(2,4,5-trimethoxyphenyl)propionate **228** has been extracted from the root bark of *Cordia alliodora*, demonstrating excellent larvicidal and antifungal properties in biological tests.<sup>244</sup> Tamariz *et al.* in 2004 developed a one-step synthesis of the antifungal and larvicidal natural product methyl 3-(2,4,5-trimethoxyphenyl)propionate **228** by the reaction between 1,2,4-trimethoxybenzene with **230** under MWI.<sup>245</sup> Even though methyl acrylate **230** failed to afford the desired product **228** under these conditions, the use of *sym*-tetrach loroethane<sup>246</sup> as the solvent at

80 °C for 1 week afforded **228** in 37% yield. The reaction time was shortened to 8 h and the yield improved using MWI (200 W) with the same mixture in a Teflon screw-capped glass tube at 80 °C, affording **228** in 66% yield. The construction of **228**, at the present time, is the shortest synthesis of this biologically potent natural product, since the previously reported synthesis, beginning from  $\beta$ -asarone or asaraldehyde, provided **228** in three steps (Scheme 48).<sup>247</sup>

Podophyllotoxin **231** and its derivatives are significant members of the lignan group of naturally occurring compounds.<sup>248</sup> The biological properties of podophyllotoxin have generated strong attention in the development of synthetic pathways to this natural product.<sup>249,250</sup> Bach and co-workers in 2008 reported a six-step total synthesis of enantiomerically pure (–)-podophyllotoxin starting from the Taniguchi lactone **203** in 35% overall yield, using iron(m)-mediated FC alkylation as a key step.<sup>251</sup> The synthesis was started from the Taniguchi lactone **203**, which is accessible in enantiomerically pure form from 2-

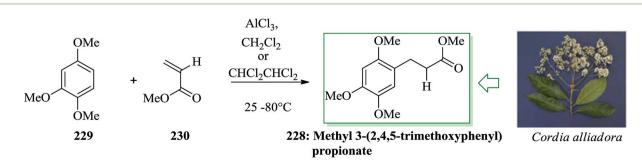


butyne-1,4-diol in two steps with a subsequent conventional resolution<sup>252</sup> or in six steps via an asymmetric iridium-mediated allylation.<sup>253</sup> An aldol reaction with aldehyde 232 gave 233 with high stereoselectivity with respect to the stereogenic center at the  $\alpha$  position of the lactone. Compound 233 was reacted with 234 (X = OH) under the optimized reaction conditions (FeCl<sub>3</sub> in dichloromethane) to afford the FC alkylation reaction product 235 in 99% yield. Finally, alcohol 233 afforded (-)-podophyllotoxin 231 upon several steps. This synthesis demonstrated that a stereogenic center in the  $\beta$  position to an ester or lactone scaffold can be generated diastereoselectively via a Lewis acid-mediated S<sub>N1</sub> reaction if a stereogenic center is already present in the  $\alpha$  position (Scheme 49).<sup>251</sup>

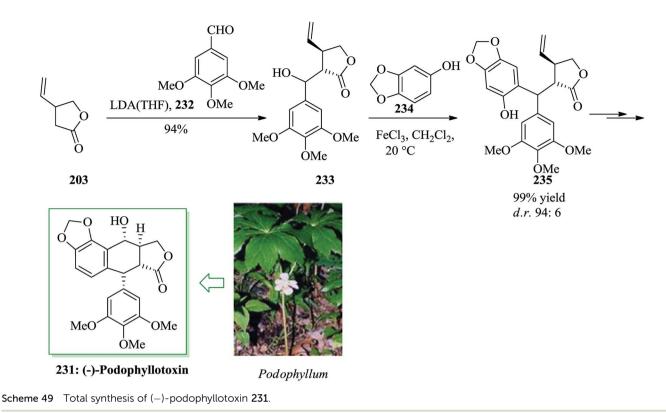
Haplophytine 236 is the major indole alkaloid known in the dried leaves of the Mexican plant Haplophyton cimicidum and was first extracted by Snyder et al. in 1952.254-256 The right-half segment is a hexacyclic aspidosperma alkaloid group, socalled aspidophytine, that is provided by the acidic degradation of (+)-haplophytine.<sup>257</sup> Fukuyama and co-workers in 2009

reported the initial total synthesis of (+)-haplophytine 236 demonstrating the simple assembly of tricyclic ketone 238 via the intramolecular Mannich reaction. FC alkylation reaction and Fischer indole synthesis were considered as the key steps. For the synthesis of (+)-haplophytine 236, initially, tricyclic ketone 238 was synthesized from 237 after several steps. Using the key tricyclic ketone 237, this group attempted the synthesis of the left-hand segment. 7-Benzyloxyindole 239 was converted into iodoindolenine 240 upon several steps. Next, the FC alkylation reaction progressed easily and provided the corresponding coupling product 242 as the major isomer in nearly a 2.4 : 1-2 : 1 diastereoselectivity. In the following, compound 242 afforded 243 after several steps. Finally, hydrazine 243 and ketone 239 provided (+)-haplophytine 236 upon several more steps (Scheme 50).258

Polyphenolic natural products represent a large and growing group of structurally varied compounds exhibiting an extensive series of biological properties.259,260 In 2006, Tan and co-workers revealed the structure and cytotoxic activities against a panel of



Scheme 48 Total synthesis of methyl 3-(2,4,5-trimethoxyphenyl)propionate 228



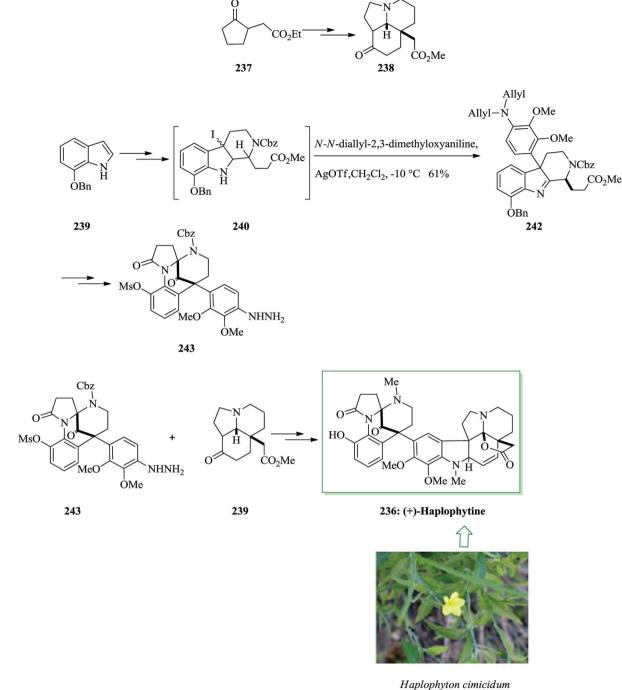
selected cancer cell lines of hopeanol 245, a polyphenol secondary metabolite extracted from the bark of Hopea exalata.261 A subsequent investigation of H. hainanensis led to the isolation of the structurally related hopeahainol A 244. The total synthesis and biological evaluation of the resveratrol-obtained naturally occurring compounds products hopeanol 244 and hopeahainol A 245 in their racemic and antipodal forms were carried out. In this method, an intramolecular FC reaction and a Grob-type fragmentation were considered as the key steps. It was examined for building the quaternary center of hopeahainol A 244 and hopeanol 245 with appropriate appendages. In the first incursion, the synthesis of the simple triaryl methyl ester 250 via an intermolecular FC-type reaction containing tertiary alcohol 247 as the main substrate and phenol 249 as the external nucleophile was attempted. Therefore, dimethyl oxalate 246 provided the tertiary alcohol methyl ester 247 in 66% overall yield. Satisfyingly, treatment of a solution of tertiary alcohol 247 and phenol 249 in dichloromethane in the presence of excess p-TsOH·H<sub>2</sub>O resulted in the construction of the desired product 250 in high yield (97%), probably via the intermediacy of carbocation 248 (Scheme 51a). Then, the feasibility of synthesizing the more relevant and advanced model system 257 was attempted. In this case, the corresponding tertiary alcohol methyl ester substrate 253 was synthesized from methyl glyoxalate derivative 251 by addition of Grignard reagent 252 (81% yield). The latter underwent a smooth FC-type reaction with resorcinol 255 providing the desired tetracyclic model system 257 (90% yield), probably via transient intermediate carbocation 254 and diphenolic methyl ester 256. The latter apparently undergoes spontaneous lactonization under the optimized reaction conditions (Scheme 51b).<sup>141</sup>

The successful formation of model system 257 (comprising four of the six ring systems of hopeahainol A) provided motivation for the next logical step, which was to attempt to use an established approach to assemble the entire carbon framework of hopeahainol A 244. Using protected nucleophilic partner 258 in the FC reaction with tertiary alcohol methyl ester 253 gave regioisomeric product 260 in 80% yield (Scheme 52).<sup>141</sup>

Next, a simple model study directed toward the formation of model system 263 was designed. Thus, methyl glyoxalate 251 provided tertiary alcohol 261 upon several steps. Treatment of the latter with  $BF_3 \cdot Et_2O$  in dichloromethane led to the construction of the corresponding FC reaction product 263 in 86% yield. It is suggested that compound 263 probably formed *via* generation of intermediate 262 (Scheme 53).<sup>141</sup>

Then, the next phase of the campaign was provided by the initial formation of the diphenolic  $\gamma$ -lactone **268** and/or **269** *via* hydroxyester **265**. The aldehyde **264** provided diphenolic hydroxyester **265** upon several steps. Treatment of **265** ( $\sim 1 : 1$  *dr*) with BF<sub>3</sub>·Et<sub>2</sub>O in dichloromethane at room temperature provided diastereomeric products **268** and **269** (86% yield,  $\sim 1 : 1.3 \ dr$ ) *via* FC reaction. Compounds **268** and **269** were probably formed *via* diastereomeric transition states **266** and **267**, respectively (Scheme 54).<sup>141</sup>

Compounds 268 and 269 provided epoxide 270 in several steps. Intramolecular FC reaction of epoxide 270 ( $\sim 1 : 1 dr$ ) resulted in the formation of two diastereomeric cyclized products 274 (major) and 273 (minor) in  $\sim 2 : 1$  ratio in the presence of tin(v) chloride in dichloromethane in 62% yield. Finally, compounds 272 and 273 provided (–)-hopeahainol A 244 in



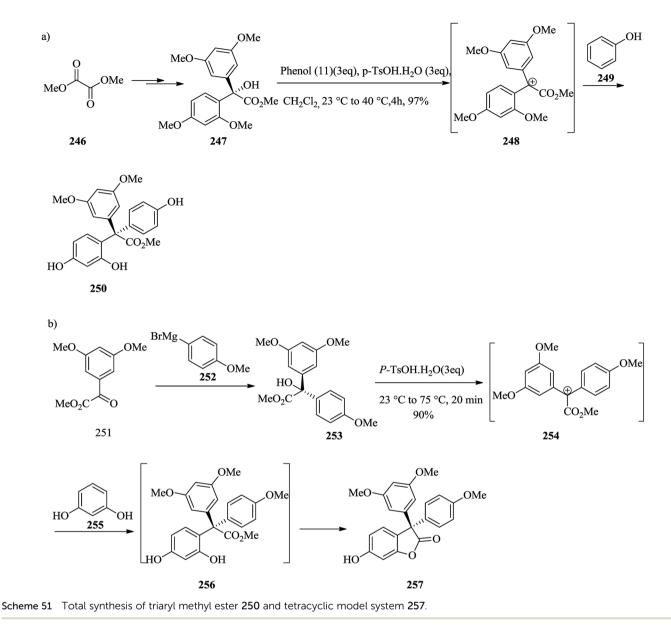
Scheme 50 Total synthesis of (+)-haplophytine 236.

indpropriyton cumera

84% overall yield after several steps. Subsequently, (-)-hopeahainol A 244 was transformed into (-)-hopeanol (-)-245 via several steps (Scheme 55).<sup>141</sup>

Dictyodendrins were initially extracted in 2003 by Fusetani and Matsunaga from the marine sponge *Dictyodendrilla verongiformis* collected from Nagashima Island in Southern Japan.<sup>262</sup> These compounds contain the highly functionalized pyrrolo[2,3-*c*]carbazole moiety. Tokuyama and co-workers in 2010 reported the first total synthesis of dictyodendrin A 275 and B 276 using an unprecedented one-pot benzyne-catalyzed indoline construction/cross-coupling sequence, employing transmetalation to Cu species.<sup>263</sup> In this route, for the synthesis of **275**, initially, *para*-nitrophenol **277** afforded **278** after several steps. The latter initially underwent FC alkylation reaction with **279** in the presence of AgOTf. Then upon pinacolborylation at the bromo group of **278**, the azidephenyl group was introduced *via* Suzuki–Miyaura coupling to afford **281**. Finally, azide **281** was converted into dictyodendrin A **275** in 8.2% overall yield over 21 steps, starting from *para*-nitrophenol **277** (Scheme 56).<sup>263</sup>

Review

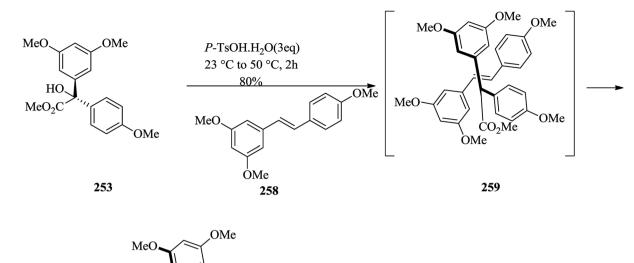


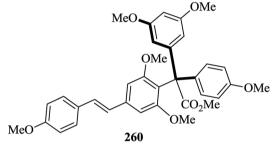
Subsequently, total synthesis of dictyodendrin B 276 was successfully achieved in 12% overall yield starting from *para*nitrophenol 277. By using bromoindole 278, regioselective FC acylation occurred in the presence of zinc chloride<sup>264</sup> to afford 283, which was converted into dictyodendrin B 276 in seven steps (Scheme 57).<sup>263</sup>

(–)-Ardeemin **284a** and (–)-acetylardeemin **284b** were extracted in 1993 from the fungus *Aspergillus fischerii*.<sup>265</sup> (–)-Acetylardeemin **284b** is one of the most active preventers of multidrug resistance.<sup>266,267</sup> Total synthesis of multidrug-resistant inhibitors (–)-ardeemin **284a**, acetylardeemin **284b**, and (–)-formylardeemin **285** was accomplished, starting from bromopyrroloinoline **286** in 10 steps by Song and co-workers in 2012. The key step includes direct alkylation of **286** with prenyl tributylstannane **287** to provide **288** through a silver-promoted enantioselective FC reaction. Significant installation of the isoprenyl substituent provided a good overall yield for this FC

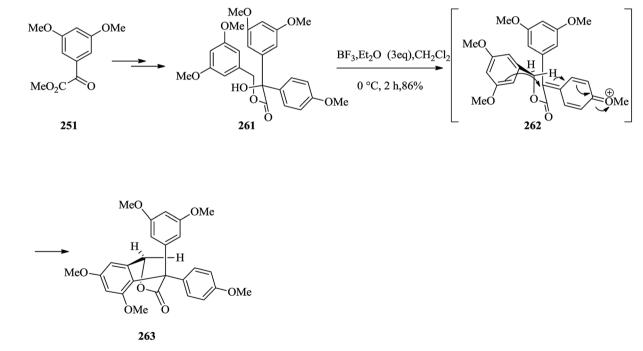
reaction. Firstly, direct isoprenylation of bromopyrroloindoline **286** with **287** gave **288** through the FC reaction in the presence of silver perchlorate, **287**, and caesium carbonate. Based on optimal reaction conditions, the main intermediate **288** was generated from **286** in 91% yield in a single step, leading to the total synthesis of **284** and **285**. Next, compound **288** gave (–)-formylardeemin **285** upon several steps. Deformylation of **285** with aqueous sodium hydroxide in methanol afforded (–)-ardeemin **284a** in 85% yield. (–)-Acetylardeemin **284b** was obtained from **284a** employing an already reported method (Scheme **58**).<sup>266</sup>

Epipolythiodiketopiperazine alkaloids, a structurally varied group of secondary fungal metabolites, exhibit a broad range of biological properties, including antiviral, antifungal, antibiotic, and cytotoxic activities.<sup>268,269</sup> It is noteworthy that these mycotoxins were characterized by identification of a bridged polysulfide linkage across the cyclic dipeptide substructure.<sup>270–276</sup>



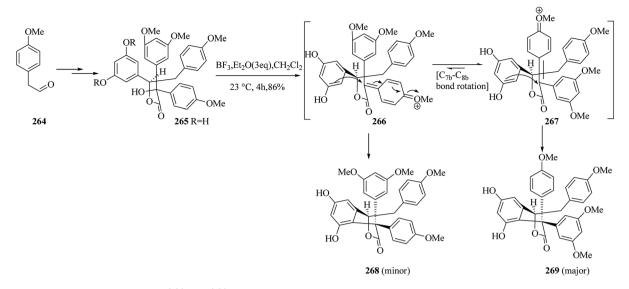


Scheme 52 The formation of compound 260.





(+)-Gliocladin B **289**,<sup>277,278</sup> a new epidithiodiketopiperazine, and (+)-gliocladin C **290**, an atypical non-thiolated triketopiperazine, were extracted by Usami and co-workers in 2004 from a strain of *Gliocladium roseum* OUPS-N132. (+)-Gliocladins show remarkable cytotoxicity against the murine P388 lymphocytic leukemia cellline.<sup>277</sup> A short and asymmetric total synthesis of (+)-gliocladin B and C was developed in 2012 by Movassaghi and co-workers. The unified synthesis of (+)-gliocladins B **289** and C **290** was started from the bromocyclization of diketopiperazine (-)-**291**<sup>276,279</sup> in just three steps starting from *N*-Boc-L-tryptophan and sarcosine methyl ester. Compound **291** reacted with bromine to give *endo*-tetracyclic bromide (+)-**292** in 75% yield



Scheme 54 The formation of products 268 and 269.

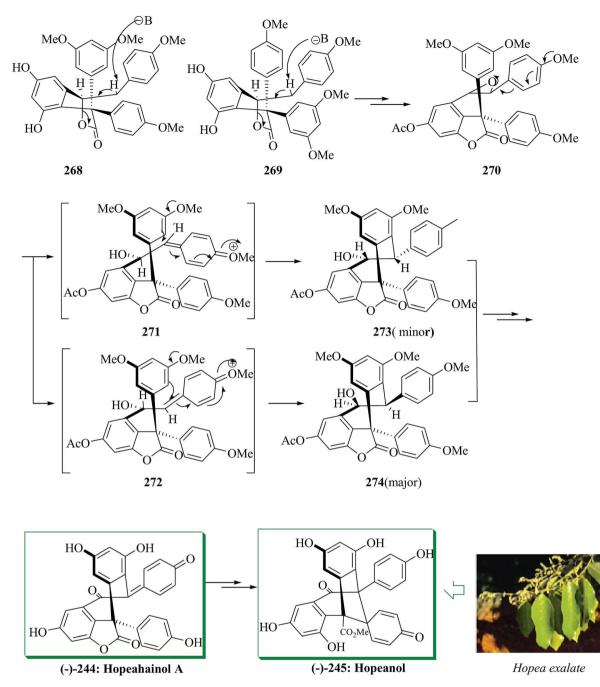
(*endo*-diastereomer). Significantly, coupling of bromide (+)-292 with indole 293 promoted by AgBF<sub>4</sub> in nitroethane proceeded smoothly to give the corresponding C3-(3'-indolyl)hexahy-dropyrroloindole (+)-294 in 83% yield. 5-Bromo-1-triisopropylsilylindole 293 was shown to be a significant nucleophile for the corresponding regio- and stereo-selective FC-type reaction. Finally, (+)-294 provided (+)-gliocladin B 289 and C 290 after several steps (Scheme 59).<sup>280</sup>

Licorice is a popular food additive applied as a sweetener in chewing gums, tobaccos, and candies, and is also one of the most broadly applied medicinal herbs having anticancer<sup>281,282</sup> antiparasitic,283 antibacterial,284 superoxide-scavenging285 and antioxidant properties.<sup>286</sup> Seven chalcones, licochalcones A-E and G and the closely related compound echinatin, have been extracted and identified from the root of Glycyrrhiza inflata (licorice),<sup>287,288</sup> except for the artificial synthetic licochalcone F.<sup>289</sup> Among these reported chalcones, licochalcone C shows higher cytotoxicity compared to the analogous licochalcone B.281 In addition, licochalcone C exhibits remarkable inhibitory activity against the PTP1B enzyme.<sup>290</sup> Wang and co-workers reported a concise, four-step synthesis of licochalcone C 295a and its regioisomer, tentatively called licochalcone H 295b, via acidcatalyzed Claisen-Schmidt condensation reaction as a main step. Initially, a key intermediate, 2,4-dihydroxy-3-(3-methylbut-2-en-1-yl)benzaldehyde 298a, was synthesized from the reaction between 2,4-dihydroxybenzaldehyde 296 and 3hydroxy-3-methyl-1-butene 297. Lewis acid-promoted FC alkylation of 297 in dioxane using  $BF_3 \cdot OEt_2$  as a catalyst gave 298. In this approach, the electron-rich aromatic ring attacked the electrophile in the presence of the Lewis acid catalyst,  $BF_3 \cdot OEt_2$ , and 3-hydroxy-3-methyl-1-butene 297, followed by aromatization of the benzene ring to provide the desired compound 298a in 18% yield based on recovery of 2,4-dihydroxybenzaldehyde (70%). The other regioisomer 298b was provided as a major product in 38% yield, probably because of the desirable alkylation of the sterically less demanding C-5 in the FC reaction.

Finally, compounds **298a** and **298b** gave **295a** and **295b** in 52 and 67% yield, respectively (Scheme 60).<sup>291</sup>

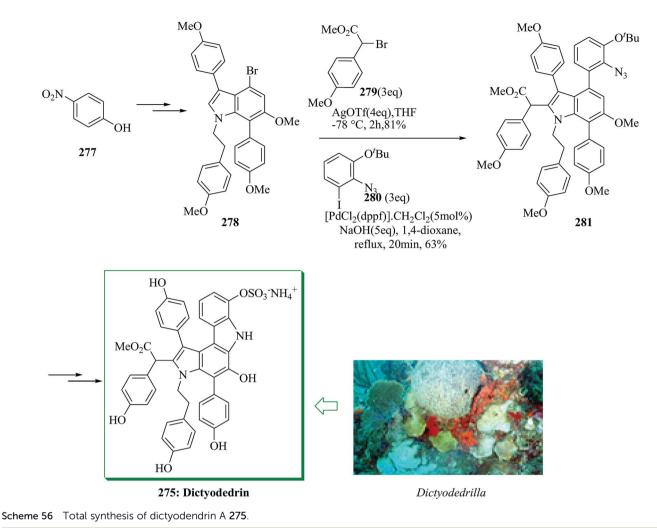
 $(\pm)$ -Yuehchukene 299, a unique dimeric indole alkaloid, was initially extracted as a racemate from the roots of Murrava paniculata (L.) Jack and other related Murraya species.<sup>292,293</sup> Yuehchukene exhibits remarkable anti-implantation property in hamsters and rats. A concise total synthesis of yuehchukene was successfully accomplished through organocatalytic FC alkylation reaction of indole to a sterically encumbered *a*-alkyl enal with excellent ees (up to 96%) as the vital step. It should be mentioned that a racemization method during the subsequent cyclization steps of the conjugate adduct to yuehchukene has been detected. In this route, at the beginning of exploring feasibility, a FC conjugate addition of dienecarbaldehyde 300 and indole 301 was successfully performed to provide the conjugate addition adduct 304. The best result was obtained in the reaction between two equivalents of 300 and indole 301 in the presence of trifluoroacetic acid in dichloromethane at room temperature, which gave 303 in 22% yield. Various acid additives have been examined with the catalyst 302294 in this reaction, and the best result was obtained under the optimized reaction conditions, affording a 26% yield of the desired product 304 (trans/cis = 73: 27) with 60% ee of the trans adduct. The absolute configuration of this FC alkylation adduct has been identified to be (1S, 2R). Next, yuehchukene 299 was provided through alkylation-cyclization of adduct 304 (93% ee) using indole 301 in 36% yield (Scheme 61).295

Several natural and synthetic heterocyclic quinones have shown significant biological properties, including antiprotozoan, antitumoral, and antibiotic properties.<sup>296,297</sup> Many of these compounds contain antineoplastic chemotherapeutic activities.<sup>298</sup> Quinones, a group of molecules known in various drugs, were applied in the clinical therapy of solid tumors.<sup>299</sup> Ellipticine was extracted from the leaves of *Ochrosia elliptica* Labill.<sup>300</sup> Ellipticine quinone **305**<sup>301,302</sup> is an essential synthetic intermediate in the Gribble synthesis of ellipticines, which



Scheme 55 Total synthesis of (–)-hopeahainol A 244 and (–)-hopeanol 245.

display antitumor activity.<sup>303</sup> A straightforward pathway for the synthesis of biologically potent ellipticine quinones was developed in 2014 by Nagarajan and co-workers. The key steps included are FC hydroxyalkylation with subsequent oxidation followed by directed *ortho*-lithiation of indole-2-carboxylate esters with functionalized quinoline and pyridine carboxaldehydes. For the synthesis of ellipticine quinone **305**, firstly, compound **306** and pyridine-3-carboxaldehyde **307** were reacted to afford **308** as acid under FC hydroxyalkylation reaction. The acid **308** was directly transformed into ellipticine quinone **305** using LiTMP in 43% yield (Scheme 62).<sup>304</sup> Studying the extract of leaves of *P. adunctum* by Orjala and coworkers<sup>305</sup> resulted in the extraction of natural adunctin E.<sup>305</sup> A wide range of naturally occurring compounds, namely methyllinderatin **310** and linderol A **312**, were extracted from the fresh bark of *Lindera umbellata* by Ichino<sup>306</sup> and Mimaki and coworkers,<sup>307</sup> respectively. In 2007, Portet and co-workers<sup>308</sup> extracted hostmanin A **309** together with methylinderatin **310**, adunctin E **311** and relevant naturally occurring compounds from the leaves of *Piper hostmannianum*. Methyllinderatin **310** exhibited antibacterial effects toward *M. luteus* and also exhibited potent antiplasmodial activity against chloroquine sensitive and resistant strains of *Plasmodium falciparum*.



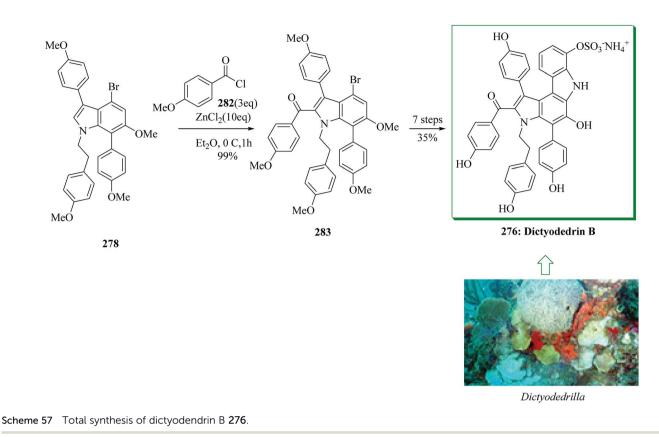
(-)-Linderol A 312 inhibited melanin biosynthesis of cultured B-16 melanoma cells without providing any cytotoxicity in the cultured cells.<sup>307</sup> Adunctin E 311 and linderol A 312 have four contiguous stereocenters. Dethe et al. in 2015 reported<sup>309</sup> protecting group free,<sup>310, 311</sup> short and asymmetric total synthesis of (+)-hostmanin A 309, (+)-methyllinderatin 310, (+)-linderol A 312 and adunctin E 311. A one-step method was established for the asymmetric synthesis of hexahydrodibenzofuran derivatives employing a modified FC reaction. The established approach was used for the synthesis of a wide range of naturally occurring compounds involving (+)-hostmanin A, (+)-methyllinderatin, and (-)-linderol A. For the synthesis of these natural products, the diastereoselective modified FC reaction was used,<sup>312, 313</sup> which included two needed constituents, dihydrochalcone derivative 314 and alcohol 315. The desired chalcone derivative 314 was provided from acetophenone derivative 313 upon several steps. Using the required segments, the stage was set to examine the key FC coupling reaction of a mixture of 314 and 315 using  $BF_3 \cdot OEt_2$ , which resulted in rapid construction of the extremely regio- and diastereo-selective coupling product methyllinderatin 310 in 91% yield with >20:1 diastereomeric ratio (only major diastereomer is depicted in Scheme 63). Moreover, one-step synthesis of (+)-hostmanin A 309 has been

accomplished. Therefore, the reaction of dihydrochalcone derivative **314** and alcohol **315** using *p*-toluenesulfonic acid under reflux in toluene provided (+)-hostmanin A **309** as a single diastereomer with 79% yield, whose spectral data and optical rotation were identical to those reported for the natural product.<sup>308</sup> Then, the total synthesis of (–)-linderol A **312** and adunctin E **311** was explored, which were finally obtained from methyllinderatin **310** upon several steps through various pathways.<sup>309</sup>

# 2.3. Intramolecular acylation

The benzo[*c*]phenanthridine alkaloid nitidine was first extracted from the root bark and root wood of *Zanthoxylum nitidum*, a woody climber that grows in most areas of Hong Kong.<sup>314</sup> A unique method for the total synthesis of benzophenanthridine alkaloids is presented in the context of the construction of the antileukemic agent nitidine chloride **316**. 4,5-Dimethoxyhomophthalic anhydride **317** was reacted with a solution of 3,4-methylenedioxybenzylidenemethylamine **318**, which after other steps gave compound **319**. Intramolecular FC acylation reaction of **319** with poly(phosphoric acid) afforded the tetracyclic ketone **320**. Next, compound **320** gave nitidine, separated

Review



as the chloride **316** in 47% yield, along with a 24% yield of the hydrogenolysis product **321** (Scheme 64).<sup>315</sup>

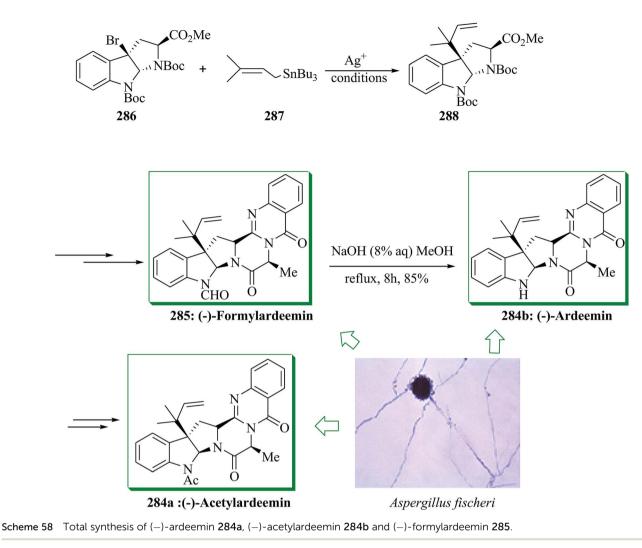
Anthraquinones as naturally occurring compounds extracted from both plants<sup>316</sup> and insects<sup>317</sup> have stirred up the interest of synthetic organic chemists as significant synthetic targets. Erythroglaucin 322a has been extracted from species of Cortinarius while catenarin 322b was first extracted from cultures of Helminthosporium catenarium. The regiospecific, concise, and effective synthesis of the natural anthraquinones erythroglaucin 322a and catenarin 322b was successfully achieved. The synthesis of erythroglaucin 322a and catenarin 322b was achieved through generation of ortho-metalated N,N-diethylbenzamide as an intermediate. Firstly, 3,5-dimethoxy-N,Ndiethylbenzamide 323 afforded a high yield of the o-benzylbenzoic acid 324 upon several steps, which under mild FC cyclization in the presence of trifluoroacetic anhydride<sup>318,319</sup> afforded the anthracenol 325. NMR and IR spectral analysis disclosed evidence for a concentration-dependent anthracenol 325a and anthracenone 325b equilibrium. Finally, compound 325 gave catenarin 322a and erythroglaucin 322b in 22% and 29% overall yields, respectively (Scheme 65).320

Interleukin-8 (IL-8), a chemoattractant for neutrophils, is provided by macrophages and endothelial cells.<sup>321</sup> IL-8 is implicated in a broad series of acute and chronic inflammatory disorders.<sup>322</sup> Frondosins A–D were recently extracted from the sponge *Dysidea frondosa*. In these compounds, the unifying architectural theme is the existence of bicyclo[5.4.0]undecane ring systems in the context of permuted linkages to several hydroquinone-based scaffolds.<sup>323</sup> Frondosin B **326**<sup>324</sup> includes a benzofuran ring system fused to a nor-sesquiterpenoid (14carbon) scaffold. A concise synthesis of ( $\pm$ )-frondosin B **326**, an interleukin-8 receptor antagonist, was accomplished from 5methoxysalicylaldehyde **327**. The seven-membered ring in ketone **329**, the common intermediate for both syntheses, was made *via* FC reaction. This method was initiated from 5methoxysalicylaldehyde **327**, which provided carboxylic acid **328** upon several steps. Fortunately, the critical FC reaction could be performed. Therefore, reaction of **328** with oxalyl chloride and reaction of the resultant acid chloride with SnCl<sub>4</sub> afforded the corresponding ketone **329** in moderate yield.<sup>325</sup> Finally, compound **329** was converted into ( $\pm$ )-frondosin B **326** after several steps (Scheme 66).<sup>326</sup>

Various sesquiterpene metabolites have been identified from species of gorgonians.<sup>327</sup> These compounds show remarkable biological properties, including antifungal, cytotoxicity and immunostimulatory activity.<sup>328</sup> Echinofuran **330**,<sup>329</sup> a type of furanosesquiterpenoid tetrahydrolinderazulene, was extracted from the gorgonian *Echinogorgia praelonga* in 1992.<sup>330</sup> The initial total synthesis of ( $\pm$ )-echinofuran **330** was achieved in 12 steps with overall yield of 0.55%. The synthesis of **330** was accomplished using 3-methyl-4-(trimethylsilyl)furan **331** as a precursor. A Suzuki coupling reaction and a Lewis acid-catalyzed FC cyclization reaction were the main steps in the formation of the desired ring system **333** (Scheme 67).

For the synthesis of  $(\pm)$ -echinofuran **330**, next, benzyl ether **335** was obtained starting from 1,3-cyclopentanedione **334**, after several steps. Saponification of **335** with sodium hydroxide was followed instantaneously, without more purification, by a FC

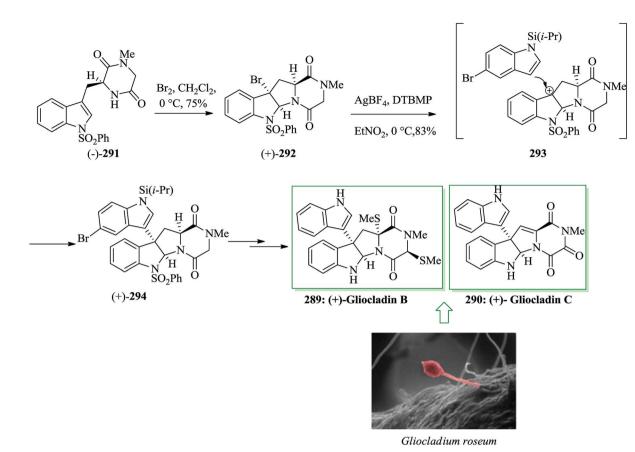
### Review



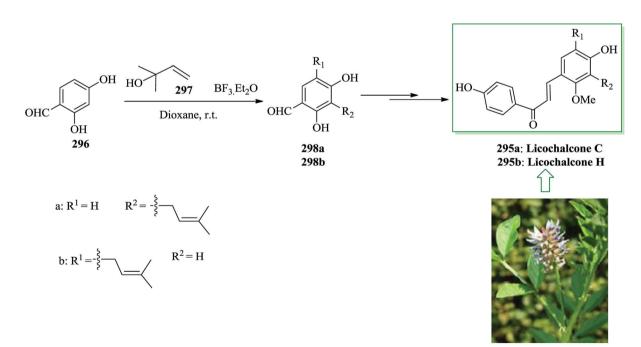
cyclization reaction, providing **336**. Finally, compound **336** was converted into  $(\pm)$ -echinofuran **330** upon several steps (Scheme 68).<sup>331</sup>

Preussomerins A-F were the first members of this group to be reported<sup>332</sup> and preussomerins G-L are the most recent additions.333 Structurally, all the preussomerins contain two naphthalene nucleus linked by three oxygen atoms, providing a bis-spiroacetal system. This significant head-to-tail trioxabicyclo[3.3.1]nonane core is a unique natural product unit. The synthesis of newly extracted members of the preussomerin group, preussomerins K and L (4% overall yields in both cases), was demonstrated by Taylor et al.<sup>334</sup> The main stages are the functionalization of a 2-arylacetal anion, one-pot FC cyclizationdeprotection and reductive opening of epoxides. The total synthesis was commenced from methoxyphenol 338, which afforded the diacid 339 in several steps. The next step of the synthesis was FC cyclization reaction of the diacid 339. Upon activating of the latter as the acid chloride by a homogenous solution of aluminium chloride in nitromethane, the corresponding FC adduct 340 was obtained. However, if the reaction was permitted to stand for longer time in the presence of excess aluminium chloride, two new compounds, identified as 341a and 341b, started forming, demonstrating that upon a quick cyclization, the remaining Lewis acid gradually mediated the removal of the phenolic methoxy groups. Lastly, compound **341b** gave preussomerin L **336** and preussomerin K **337** through a different multi-step synthesis (Scheme 69).<sup>334</sup>

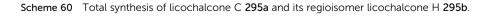
Preussomerin F was extracted from the coprophilous fungus Preussia isomera by Gloer and co-workers in 1990.335 Preussomerin K was also extracted independently by Isaka et al. from a lichenicolous fungus Microsphaeropsis sp. BCC 3050.336 All of the preussomerins exhibit pronounced antifungal activity as well as preventing Gram-positive bacteria. All the members of the preussomerin group are characterized by the same key structural aspect, containing the naphthalene units linked by three oxygen atoms, providing a bis-spiroacetal functionality. Taylor and co-workers in 2004 developed and reported the first total synthesis of the racemic natural products preussomerin F (3% overall yield, 13 steps) and preussomerins K and L (4% overall yields in both cases, 11 and 10 steps, respectively) from cheap and simply accessible initiating precursors via the usage of a straightforward, non-biomimetic methodology including the functionalization of 2-arylacetal anions, simultaneous onepot FC cyclization-deprotection and substrate-regiocontrolled hydrogenation as the key steps. The total synthesis of the racemic naturally occurring compounds preussomerins F, K



Scheme 59 Total synthesis of (+)-gliocladin B 289 and C 290.



Glycyrrhiza inf lata

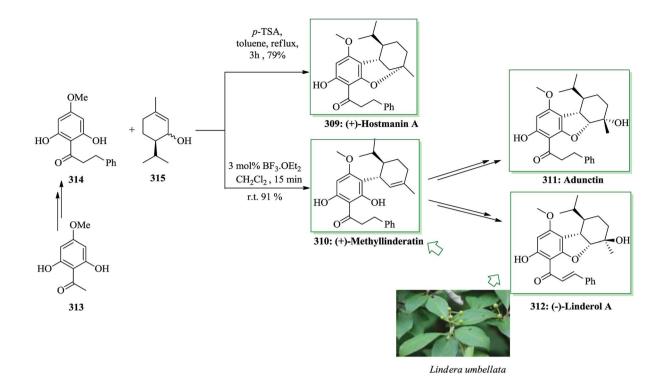


Method A: cat-302 (20mol%), 301 (S)-303 (40mil%), CHCl3/EtOAc(1:1) BA(20 mol%) 25 °C, 15days Method A: cat-IV (20mol%), OHC CH<sub>2</sub>Cl<sub>2</sub> (0.1 mol) OHC (S)-A11(20mil%), CHCl<sub>3</sub>/EtOAc(1:1) 300 301 304: MethodA: 25% yield, 94:6 20 °C, 15days Method B: 11% yield, 96:4 OMe ЮH 302 (S)-303 Murrava paniculata 299: (±)- Yuehchukene Scheme 61 Total synthesis of yuehchukene 299  $R_2$ CHO AlCl<sub>3</sub>,dry CH<sub>2</sub>Cl<sub>2</sub>  $CO_2R_1$ LiTMP, THF -78 °C, 3h 0 °C to r.t. 3h 307 306 308 C 43%-72% 78 %  $R_1 = {}^tBu$  $R_1 = R_2 = H$  $R_2 = H$ Ochrosia elliptica 305: Ellipticine quinone

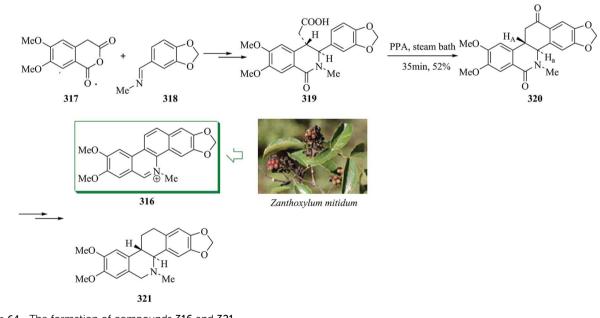


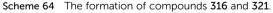
and L started from 4-methoxyphenol, which was transformed into the diacid **339** in several steps. Then the reaction of diacid **339** with a homogenous solution of aluminium chloride in nitromethane at ambient temperature provided the corresponding FC adduct **340** in satisfactory yield (75%). However, if the reaction was permitted to stand for a longer time in the presence of an excess of aluminium chloride, two new compounds, characterized as **341a** and **341b**, started to appear, demonstrating that after a fast cyclization, the lasting Lewis acid gradually removed the phenolic methoxy substituents. Finally, compound **341b** gave preussomerins F, K and L upon several steps (Scheme 70).<sup>337</sup>

The search for phytoalexins in *Musa* over the years led to the isolation and structural identification of 9- and 4-phenylphenalenones and related compounds, for example dimeric phenylphenalenones, phenylnaphthalic anhydrides, perinaphthenones, and an oxabenzochrysenone.<sup>338</sup> 4-Methoxy-1*H*-phenalen-1-one (4-methoxyperinaphthenone) **343**, a subunit



Scheme 63 Total synthesis of (+)-hostmanin A 309, (+)-methyllinderatin 310, (+)-linderol A 312 and adunctin E 311.

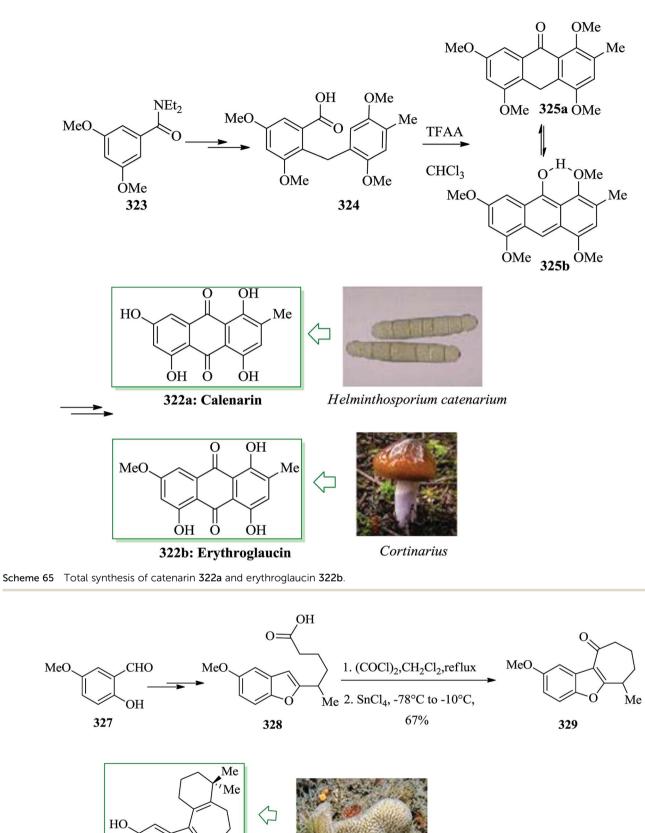




known in some *Musa* phytoalexins and relevant naturally occurring compounds from *Haemodoraceae*, was obtained *via* a Heck–Fujiwara coupling reaction and a FC acylation as the carbon–carbon bond-forming reactions. It was actually synthesized from 2-methoxynaphthalene in five steps with an overall yield of 36%. The formation of 4-methoxy-1*H*-phenalen-1-one (4-methoxyperinaphthenone, **343**) and other perinaphthenones alike was achieved *via* the cyclization of *b*-1-naphthylpropanoic acids provided by the malonic ester synthesis,

employing 4-methoxynaphthalenes **344** as the starting material.<sup>339</sup> Based on this method, 4-methoxynaphthalene gave **345** upon several steps. Compound **345** was cyclized employing FC reaction conditions and gave 4-methoxy-1*H*-phenalen-1-one (4methoxyperinaphthenone) **343** as the sole product in 48% yield (Scheme 71).<sup>340</sup>

Xestodecalactone C 346 was extracted from the fungus *Penicillium* cf. *montanense*, which in turn was extracted from *Xestospongia exigua*. This molecule is structurally related to

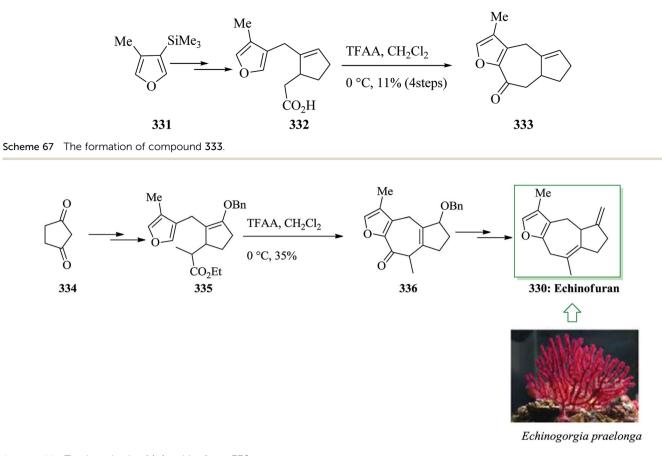


Dysidea frondosa

Scheme 66 Total synthesis of  $(\pm)$ -frondosin B 326.

326: Frondosin B

Me



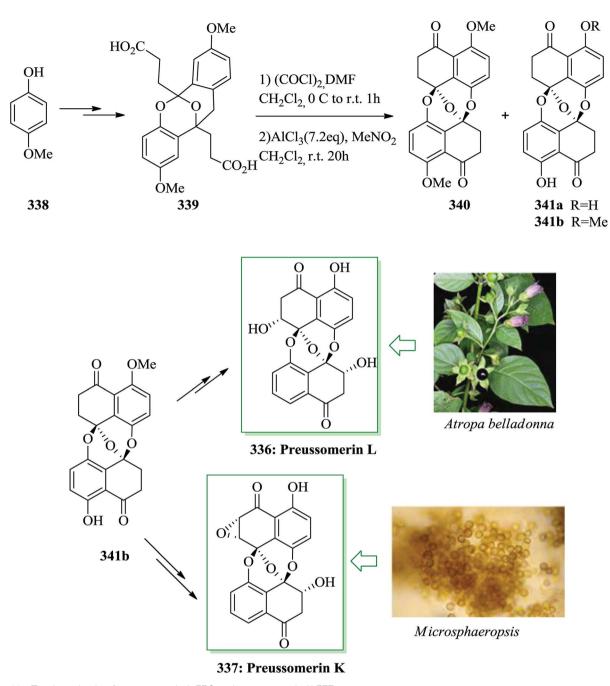
Scheme 68 Total synthesis of  $(\pm)$ -echinofuran 330.

a wide range of compounds extracted from terrestrial fungi, involving sporostatin 347. Sporostatin (M5032, 347), extracted from the fungus Sporormiella sp., is an inhibitor of cyclic adenosine 3',5'-monophosphate phosphodiesterase.341 Total synthesis of xestodecalactone C and epi-sporostatin was developed in 10 steps using Prins cyclizations, Mitsunobu reaction and intramolecular FC acylation reaction. This method is convergent and extremely enantioselective. The synthesis of xestodecalactone 346 and epi-sporostatin 347 was started with chiral homoallyl alcohol 349. The starting material 349 was synthesized in two steps from S-(–)-benzyl glycidyl ether 348. Next, compound 349 was converted into acid 350 after several steps. The corresponding macrolide 351 was provided in 41% yield via intramolecular FC reaction of the carboxylic acid 350 in the presence of a mixture of trifluoroacetic acid and trifluoroacetic acid anhydride.342,343 Demethylation reaction of 351 with freshly synthesized  $AlI_3$  afforded the target molecule  $346^{344,345}$  in 96% yield, whereas the same reaction at ambient temperature provided 347 in 94% yield.342,343 Finally, compound 351 afforded 346 and 347 in 96% and 94% yield, respectively (Scheme 72).346

The aaptamines, first found in nature by Nakamura and coworkers in 1982,<sup>347</sup> are marine alkaloids that include a benzo[*de*] [1,6]naphthyridine scaffold.<sup>348,349</sup> The aaptamine group involves aaptamine **352**, first extracted from an Okinawan specimen of *Aaptos aaptos*. The aaptamines exhibit various remarkable biological properties. Owing to their antagonistic effects on  $\beta$ - adrenergic receptors, a cardiac property was found for 352.<sup>350,351</sup> A formal total synthesis of the marine alkaloid aaptamine 352 has fruitfully been accomplished *via* the development of a new methodology to provide 2,3,3*a*,4,5,6-hexahydroaaptamine 356. This alternative access to 356 has been achieved in eight steps and in 11% overall yield, with quinolin-4-one 355 as a key intermediate. The synthesis was started from market-accessible 2,3-dimethoxybenzoic acid 353, which after several steps afforded acid 354. Then, acid 354 was exposed to FC-type acylation reaction with excess PPE in toluene to obtain 355 in 95% yield. Finally, after several steps, intermediate 355 afforded 2,3,3*a*,4,5,6-hexahydroaaptamine 356, which is an intermediate for the synthesis of aaptamine 352 (Scheme 73).<sup>352</sup>

Sporostatin (M5032, **34**7), extracted from the fungus of *Sporormiella* sp., is an inhibitor of cyclic adenosine 3',5'-monophosphate phosphodiesterase (cAMP-PDE),<sup>353</sup> composed of a ten-membered macrolide derivative with a 1,3-di-hydroxybenzene ring. A simple and effective total synthesis of sporostatin was achieved in five steps beginning from (*S*)-propylene oxide **357**. In this method, esterification, cross-metathesis, and intramolecular FC reaction were known as the key steps. Total synthesis of sporostatin (M5032, **347**) was started from (*S*)-propylene oxide **357**, which afforded the relevant  $\alpha$ , $\beta$ -unsaturated carboxylic acid **358** after several steps. Then, the macrolide **359** was obtained in 41% yield when the carboxylic acid **358** was exposed to an intramolecular FC

Review

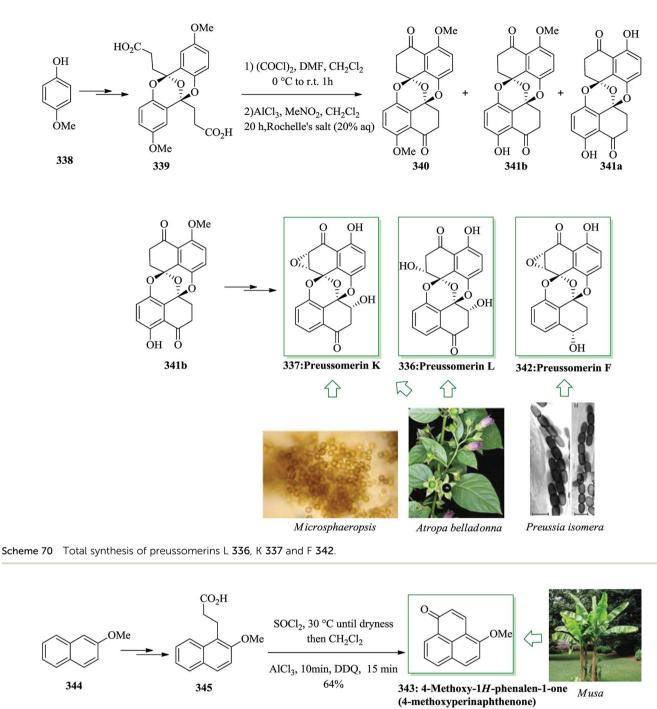


Scheme 69 Total synthesis of preussomerin L 336 and preussomerin K 337

reaction with a mixture of trifluoroacetic acid and trifluoroacetic anhydride.<sup>344,354–356</sup> Deprotection of the MeO substituents of **359** employing freshly synthesized AlI<sub>3</sub> (ref. 357–359) provided the final product **347** in 94% yield (Scheme 74).<sup>360</sup>

In 2002, 10-membered macrolides fused to the 1,3-dihydroxybenzene ring, for example xestodecalactones A, B, and C **346**, were extracted from the fungus *Penicillium* cf. *mantanense* obtained from the marine sponge *Xestospongia exigua*.<sup>358</sup> Xestodecalactones A–C exhibited antifungal and antibacterial properties.<sup>361</sup> A simple and extremely effective enantioselective total synthesis of xestodecalactone C **346**, a polyketide natural product, was accomplished using Keck's asymmetric allylation reaction, an iodine-induced electrophilic cyclization, and an intramolecular FC acylation as key steps. The synthesis of xestodecalactone C **346** was initiated from propane-1,3-diol, which was protected with *p*-methoxybenzyl (PMB) bromide to provide the relevant propan-1-ol **360**. Next, compound **360** gave the acid **361** in several steps. The corresponding macrolide **362** was obtained in 40% yield *via* an intramolecular FC acylation reaction of the carboxylic acid **361** using CF<sub>3</sub>COOH/(CF<sub>3</sub>CO)<sub>2</sub>O<sup>346</sup> Finally, compound **362** was transformed into the desired natural product **346** after several steps (Scheme 75).<sup>362</sup>

The curvularins are octaketides composed of a 12-membered macrolide framework fused to a 1,3-dihydroxybenzene scaffold. (11 $\beta$ )-11-Methoxycurvularin 363 is a member of the curvularin group extracted from the mycelium of the hybrid strain ME 005

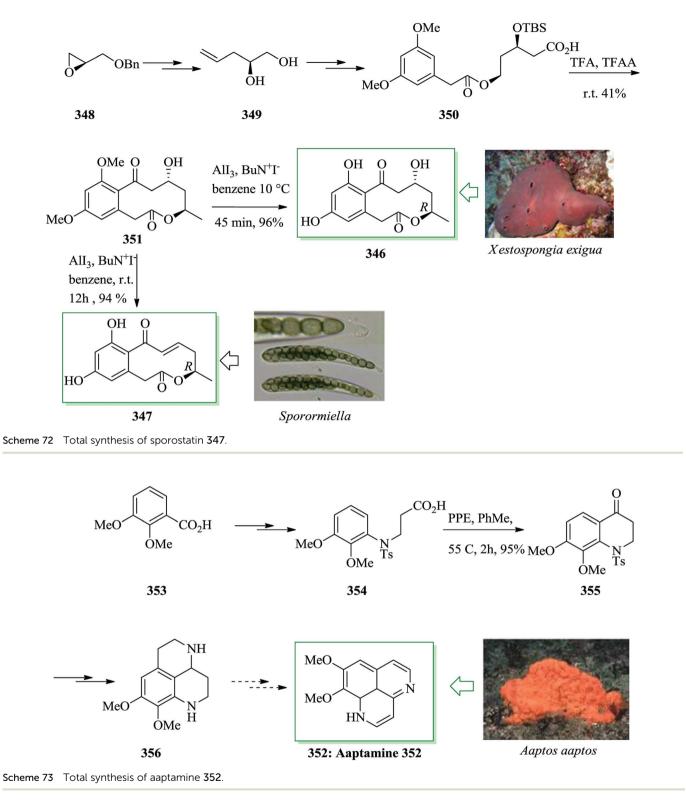


Scheme 71 Total synthesis of 4-methoxy-1H-phenalen-1-one (4-methoxyperinaphthenone) 343.

obtained from *Penicillium citreoviride* 4692 and 6200 (ref. 363) that demonstrated significant cytotoxicity toward a panel of four human-cancer cell lines.<sup>364</sup> A simple and extremely effective enantioselective total synthesis of  $(11\beta)$ -11-methoxycurvularin 363 was accomplished in 2010 by Venkateswarlu and coworkers.<sup>365</sup> The synthesis was started with a Cu-catalyzed regioselective ring opening of (2S)-2-methyloxirane 357, a Keck enantioselective allylation and intramolecular FC acylation as key steps. In this strategy, oxirane 357 was used as the starting material, which was converted into compound 364

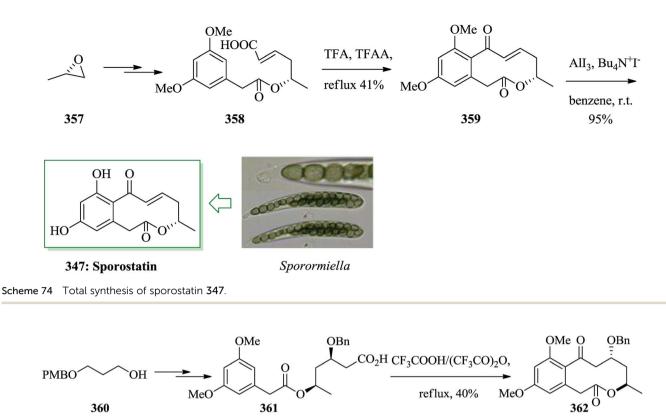
upon several steps. The corresponding macrolide **365** was obtained in 41% yield *via* intramolecular FC reaction of the carboxylic acid **364** in the presence of  $CF_3COOH/(CF_3CO)_2O$ . Next, demethylation of **365** using freshly synthesized AlI<sub>3</sub> afforded the natural product **363** in 96% yield (Scheme 76).<sup>365</sup>

Dipterocarpaceae is a rich source of a wide range of biologically potent oligostilbenoids.<sup>366,367</sup> Not surprisingly, many remarkable pharmacological functions of this group were also identified, including antifungal, anti-inflammatory, antibacterial, and anticancer properties.<sup>368–371</sup> Isolation of a novel



oligostilbenoid containing strong cytotoxicity, diptoindonesin G, from the tree bark of *Hopea mengarawan* was demonstrated by Syah *et al.*<sup>372</sup> This natural product exhibits active immunosuppressive properties.<sup>373</sup> A very significant total synthesis of diptoindonesin G **366** was developed using a domino dehydrative cyclization/intramolecular FC acylation/regioselective demethylation reaction of aryloxyketone **371** in the presence

of boron trichloride, in which the tetracyclic 6*H*-anthra[1,9-*bc*] furan-6-one framework was generated through the 3-arylbenzofuran in a one-pot fashion. Accordingly, a coupling reaction of **367** and **368** took place, which was followed by dehydrative cyclization and subsequent direct arylation at the C2 position of the benzofuran scaffold to afford the key intermediate **369**. The latter was then transformed into compound **370** upon



Xastospongia exigua

361

OH

ö

Scheme 75 Total synthesis of xestodecalactone C 346

HO

OH 0

360

treatment with NaOH and then TFAA by intramolecular FC acylation reaction. Upon treatment with BBr3 the latter was converted into G 366 in 93% yield (Scheme 77).374

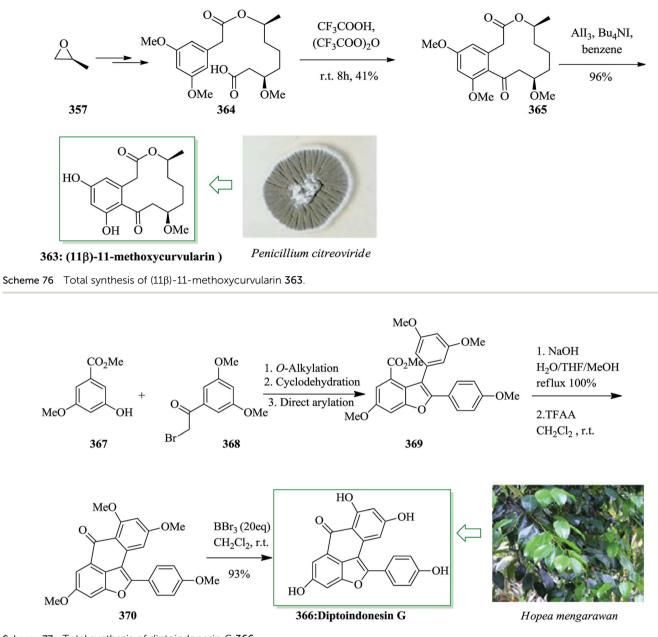
346: Xestodecalactone C

The same strategy was applied for the synthesis of diptoindonesin G 366 using compound 371 instead of 370 as the precursor. Compound 371 was treated with BCl<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> to give the demethylated compound 372 in 95% yield.375-378 Finally, the latter was transformed into the desired natural product diptoindonesin G 366 upon several steps (Scheme 78).374

Kinamycin antibiotics379,380 have been extracted from of Streptomyces murayamaensis culture broth<sup>381,382</sup> with antitumor and antiviral properties.<sup>382</sup> The structure was at first identified as N-cyanobenzo[b]carbazoloquinone. The structure of so-called 'prekinamycin' extracted by Gould was revised to, temporarily, diazobenzo[b]fluorene 373.383 This compound with the structure 373 was extracted as a metabolite provided by S. murayamensis mutant MC2 (ref. 384) and finally named as prekinamycin.385 Total synthesis of prekinamycin 373 was accomplished through Suzuki coupling reaction of naphthaleneboronic acid and a bromobenzene derivative,

intramolecular FC reaction of 2-(naphthalen-2-yl)benzoic acid, and diazotization in ten steps starting from 3,5-dimethylphenol 374, in 7% yield. Initially, 3,5-dimethylphenol 374 was converted into carboxylic acid 375 upon several steps. The intramolecular FC reaction of 375 with (CF<sub>3</sub>CO)<sub>2</sub>O<sup>386</sup> or POCl<sub>3</sub>/K<sub>2</sub>CO<sub>3</sub> (ref. 387) in a one-pot fashion resulted in a complex mixture. Transformation of acid 375 to acid chloride 376, followed by intramolecular FC reaction in the presence of tin(IV) chloride, afforded the key MOM-deprotected benzo[b]fluorenone 377 in merely 4% yield. Finally, compound 377 was transformed into the desired natural product prekinamycin 373 in 47% yield (Scheme 79).388

Diazobenzofluorene is a naturally occurring compound, classified as a group of structurally complex molecules bearing a tetracyclic (ABCD) scaffold along with a diazo group. Some epoxykinamycins contain FL-120B and the closely related FL-120B' 378.389 Monomeric diazobenzofluorenes were found to exhibit antitumor activity. Enantioselective synthesis of FL-120B' was accomplished employing Sharpless asymmetric epoxidation, Stille cross-coupling, and intramolecular FC



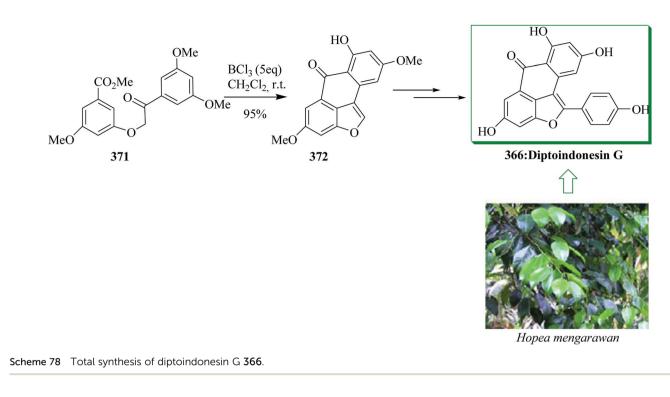
Scheme 77 Total synthesis of diptoindonesin G 366.

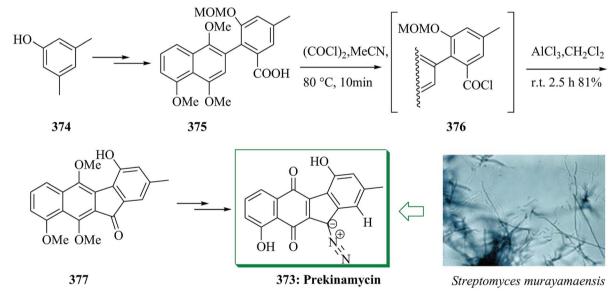
reactions as key steps. The synthesis of FL-120B' represents the first total synthesis of an epoxide-comprising, diazobenzo-fluorene natural product. Firstly, quinone **379** gave carboxylic acid **380** upon several steps. Then, carboxylic acid **380** in the presence of trifluoroacetic anhydride (TFAA) underwent an intramolecular FC acylation, affording the corresponding ketone **381**. Finally, ketone **381** was transformed into FL-120B' **378** after several steps (Scheme 80).<sup>390</sup>

The trikentrins **382** and **384** were extracted from the marine sponge *Trikentrion flabelliforme* by Capon<sup>391</sup> and exhibited antibacterial properties. In a concise total synthesis of (+)-*cis*-trikentrin A, initially 4-ethyl-7-vinylindole underwent nickel(II)-mediated enantioselective hydrovinylation to afford the desired 2-but-3-enyl derivative **384a** in excellent yield (99%) and high *ee* (96%). Compound **384a** was converted into the corresponding

carboxylic acid **385a** in several steps. The latter was subjected to an intramolecular FC reaction and a Wittig reaction affording the *exo*-methylene compound **387a** in about 70% yield from the acid **385a**. Subsequently, compound **387a** was converted into the natural product (+)-*cis*-trikentrin **382** after several steps. Noticeably, synthesis of a more complex natural product, (+)-*cis*trikentrin B, demonstrates the application of the hydrovinylation product, **384b**, and a late-step functionalization of a 6,8-dimethylcyclopent[g]indole nucleus. Employing the aforementioned protocol, **384b** was transformed into (+)-*cis*trikentrin-B **383** after several steps (Scheme 81).<sup>392</sup>

Amaryllidaceae isocarbostyril alkaloids have been attractive synthetic targets for organic synthetic chemists during the last two decades.<sup>393,394</sup> The isolation of *trans*-dihydronarciclasine **388** from the Chinese medical plant *Zephyranthes candida* was

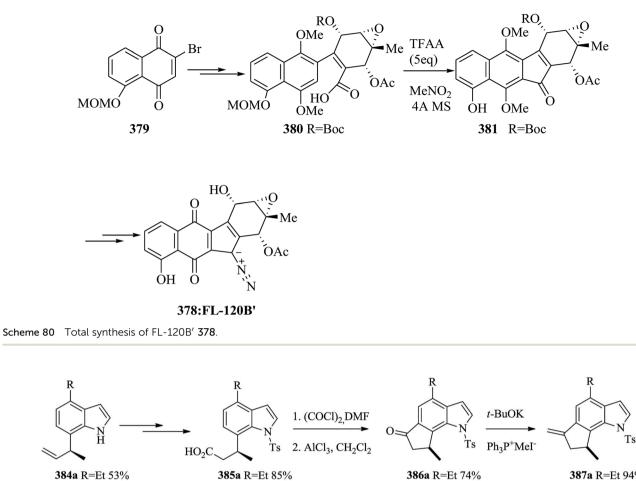




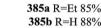
Scheme 79 Total synthesis of prekinamycin 373.

demonstrated in 1990.<sup>395</sup> An extremely enantioselective and effective total synthesis of *trans*-dihydronarciclasine **388** in an entirely substrate-controlled strategy from easily accessible chiral starting materials was developed in which the desired target **388** was obtained in 16% overall yield. The total synthesis of *trans*-dihydronarciclasine **388** was commenced by preparing triacetate **390** from the enantiomerically enriched allylic alcohol **389** *via* several steps. Using 1.1 equivalents of Tf<sub>2</sub>O and 1.5 equivalents of 2-chloropyridine, and heating at 35 °C, the product **392** was obtained with very high regioselectivity (12.5 : 1) and in 76% yield. After two steps, the desired target (+)-**388** was obtained (Scheme 82).<sup>396</sup>

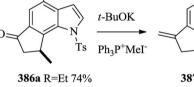
Colchicine,<sup>397,398</sup> extracted from *Colchicum autumnale*, is a cytostatic drug that strongly binds to tubulin, the main constitutive protein of microtubules.<sup>399</sup> Colchicine and its structural analogues, for example allocolchicine<sup>400</sup> and combretastatin A-4,<sup>401</sup> symbolize promising lead structures for the development of new anticancer agents addressing the colchicine binding site of tubulin.<sup>402,403</sup> and 5,6,7-trimethoxy-4-(1methyl-1*H*-indol-6-yl)-2*H*-chromen-2-one,<sup>404</sup> respectively that were known to prevent microtubule assembly *in vitro* at nanomolar concentrations. Schmalz and co-workers in 2012 provided a unique group of indole-comprising allocolchicine analogues. Using 3,4,5-trimethoxy-phenylpropionic acid **394**,



384b R=H 83%

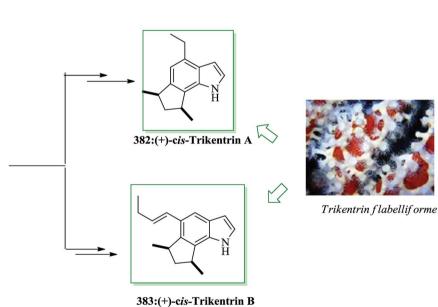






386b R=H 68%

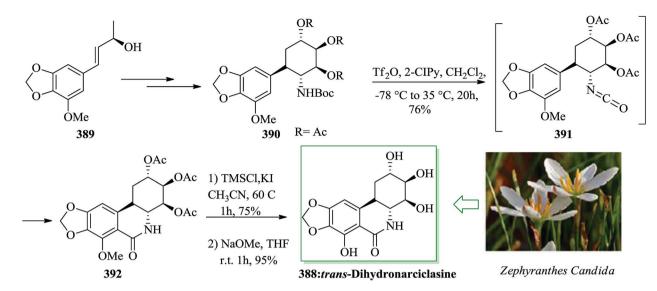
387a R=Et 94% 387b R=H 95%



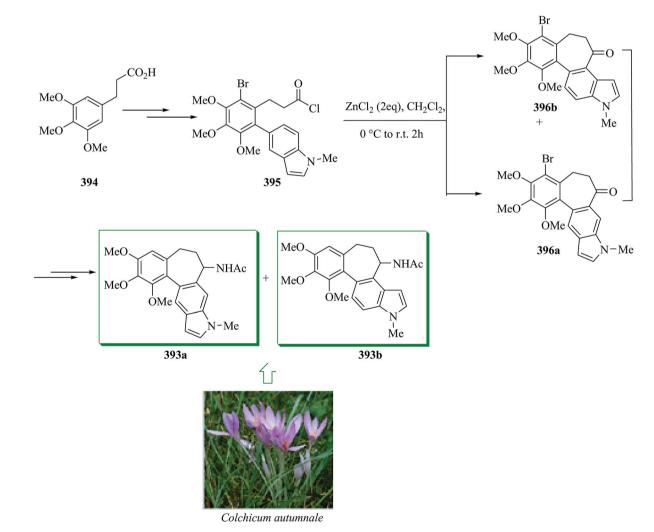
Scheme 81 Total synthesis of (+)-cis-trikentrin 382 and (+)-cis-trikentrin-B 383.

the target compounds rac-393a and rac-393b were obtained in 11 steps with overall yields of 14% and 3%, respectively, through a Suzuki cross-coupling reaction and a FC cyclization reaction as the key carbon-carbon bond-forming reactions. Firstly, 3,4,5-trimethoxyphenylpropionic acid 394 afforded a clear solution including acid chloride 395 upon several steps.

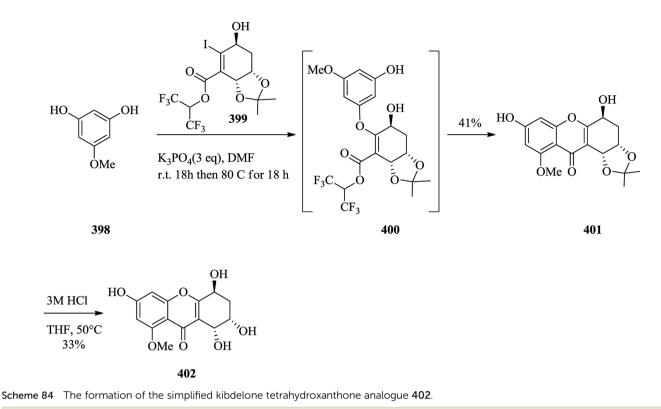
3



Scheme 82 Total synthesis of trans-dihydronarciclasine 388.



Scheme 83 Total synthesis of allocolchicinoids rac-393a and rac-393b.

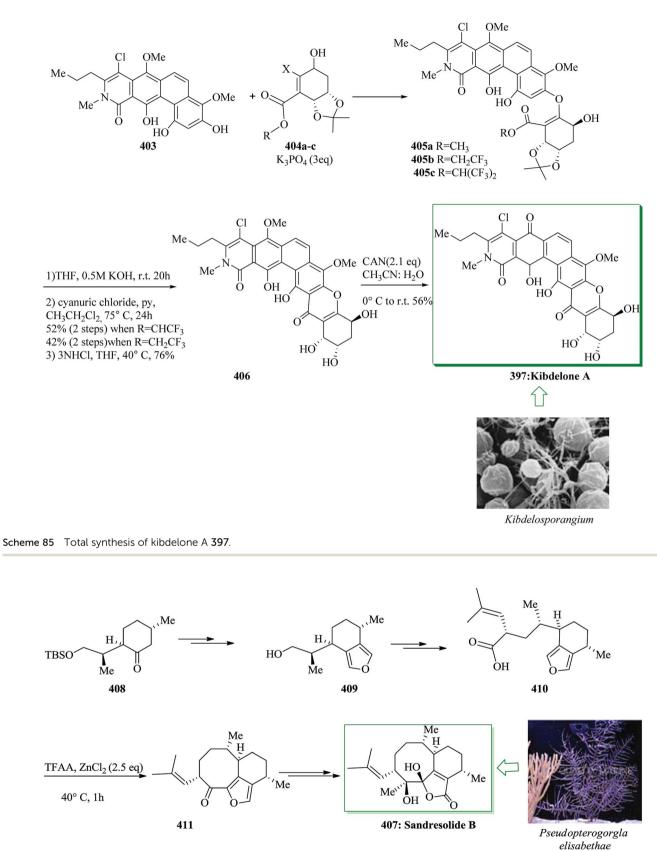


In a first attempt to induce the projected FC cyclization, ZnCl<sub>2</sub> was added to a crude solution of **395** in dichloromethane at ambient temperature. The corresponding ketones **396a** and **396b** were obtained in merely 5% yield (as a 4 : 1 mixture). Finally, compounds **396a** and **396b** provided the indole-obtained allocolchicinoids *rac*-**393a** and *rac*-**393b**, respectively, after several steps (Scheme 83).<sup>405</sup>

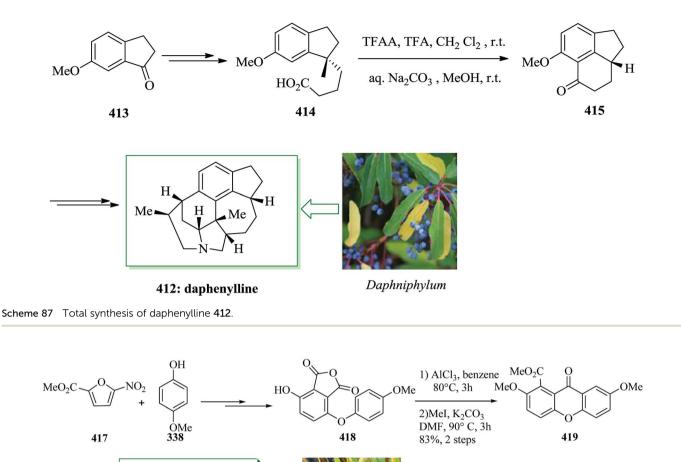
The kibdelones are a unique group of bioactive heterocyclic polyketides provided by a rare soil actinomycete, Kibdelosporangium sp. (MST-108465). Significantly, kibdelones A-C were found to be active at low nanomolar concentrations against tumor cell lines.<sup>406</sup> The total synthesis of kibdelone A 397 was achieved through In(m)-mediated anylation of a heterocyclic quinone monoketal and iodine-catalyzed oxidative photochemical electrocyclization. A one-pot oxa-Michael/FC method was tried using monomethoxy phloroglucinol 398407 to produce the first simplified analogues of kibdelones. Under the optimal conditions, using the extremely activated HFIP ester 399 resulted in the finding that tetrahydroxanthone construction based on basic conditions was possible in a one-pot method, avoiding extraction of the acid- and heat-sensitive vinylogous carbonate 400. Furthermore, the oxa-Michael reaction progressed in a siteselective manner at ambient temperature; however, the FC cyclization reaction needed heating to 60 °C using K<sub>3</sub>PO<sub>4</sub> to provide tetrahydroxanthone 401 in 41% yield. The usage of base was needed for the FC cyclization reaction by the isolation of vinylogous carbonate 400 and its subsequent thermolysis in dimethylformamide (60 °C), which was shown to be unsuccessful. Final deprotection of 401 gave the simplified kibdelone tetrahydroxanthone analogue 402 (Scheme 84).408

Then, total synthesis of the triol unit of kibdelone A **397** was examined. The reaction of **403** and **404a-c** gave the corresponding products **405a-c**. Additional FC cyclization was explored employing a one- or two-pot method with the naturally occurring compounds **405a-c** based on thermal, Lewis acid-mediated, and *N*-methylimidazole-improved conditions; however, in these cases merely initiating material, aromatized F-ring, or retro-Michael products were detected. Cyclization reaction was accomplished employing a two-step sequence through saponification/cyanuric chloride activation of vinylogous carbonates **405b** and **405c** in 52% and 42% yield over two steps, respectively. Lastly, deprotection of the acetonide scaffold of the F-ring and oxidation reaction of the B-ring employing ceric ammonium nitrate (CAN) in H<sub>2</sub>O/acetonitrile gave kibdelone A **397** in moderate yield (Scheme 85).<sup>408</sup>

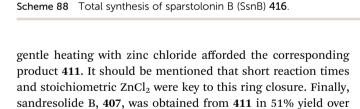
The Caribbean octocoral *Pseudopterogorgia elisabethae* is a chemically prolific species that has attracted the attention of scientists.<sup>409</sup> Sandresolide B **407**<sup>410-412</sup> was first reported in 1999.<sup>413</sup> The total synthesis of the diterpenoid sandresolide B from a common furan framework was achieved and reported in 2013 by Trauner and co-workers. Key stages contain Pdcatalyzed carbonylation, lanthanide mediated ring closure, Myers alkylation reaction, intramolecular FC acylation, photooxygenation, and a Kornblum–DeLaMare rearrangement.<sup>414</sup> Total synthesis of **407** was started from ketone **408**, which was converted into compound **409** upon several steps. Next, compound **409** afforded acid **410** upon several steps. Using the key acid **410**, the seven-membered ring of the sandersolides was provided through an intramolecular FC acylation reaction. Activation of the acid with trifluoroacetic anhydride followed by



Scheme 86 Total synthesis of sandresolide B 407.



Spaganium stoloniferum



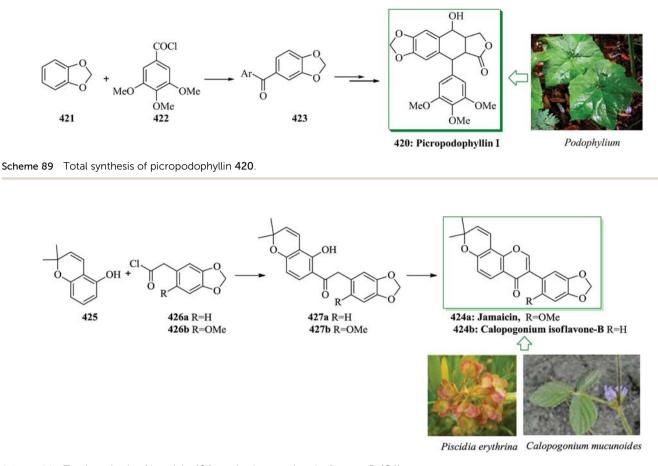
416:SsnB

HO

two seps (Scheme 86).414 Daphenylline 412, extracted from the fruit of Daphniphyllum longeracemosum by Hao and co-workers in 2009, is the first member of the Daphniphyllum alkaloids that includes a benzene ring in the unit structure.415 Daphenylline contains a rearranged 22-norcalyciphylline A type framework,416 the highly fused hexacyclic system of which has generated significant interest in the chemical community. Total synthesis of (-)-daphenylline, a hexacyclic Daphniphyllum alkaloid, was accomplished in 2016 by Fukuyama and co-workers.417 In this approach, an asymmetric Negishi coupling reaction, an intramolecular FC reaction, Sonogashira coupling and Claisen rearrangement are considered as key steps. Total synthesis of (-)-daphenylline 412 was initiated from 6-methoxyindan-1-one 413, which was converted into carboxylic acid 414 after several steps. Next, an intramolecular FC reaction of 414 was achieved by reaction with

trifluoroacetic anhydride (TFAA) and trifluoroacetic acid (TFA) to provide cyclic ketone **415**. Finally, daphenylline **412** was obtained after several steps (Scheme 87).<sup>417</sup>

In 2010, sparstolonin B (SsnB) 416 was first extracted from the Chinese herb Spaganium stoloniferum by Liang et al.<sup>418</sup> In 2015, Sun and co-workers reported that sparstolonin B can prevent high-fat-diet-induced obesity in rats, and also prevents LPS-induced production of MCP-1, interleukin-6, and tumor necrosis factor-a in 3T3-L1 adipocytes.419 Shibata and coworkers in 2017 developed a unique method for the syntheses of sparstolonin B (SsnB) 416 with an overall yield of 38% over 10 steps from the affordable, market-accessible initiating precursor methyl 5-nitro-2-furoate. In this method, Diels-Alder reaction and intramolecular FC reaction were found as the key steps. Total synthesis of 416 was started with aromatic substitution of methyl 5-nitro-2-furoate with sodium 4-methoxybenzenolate as a nucleophile to afford the cyclic anhydride 419 in 95% yield. The intramolecular FC reaction of anhydride 419 in benzene afforded a mixture of the corresponding ester 420 and a side-product 420b, provided via elimination of an O-



Scheme 90 Total synthesis of jamaicin 424a and calopogonium isoflavone-B 424b.

of sparstolonin B (SsnB) **416** with an overall yield of 38% over 10 steps from the affordable, market-accessible initiating precursor methyl 5-nitro-2-furoate. In this method, Diels–Alder reaction and intramolecular FC reaction were found as the key steps. Total synthesis of **416** was started with aromatic substitution of methyl 5-nitro-2-furoate with sodium 4-methoxybenzenolate as a nucleophile to afford the cyclic anhydride **419** in 95% yield. The intramolecular FC reaction of anhydride **419** in benzene afforded a mixture of the corresponding ester **420** and a side-product **420b**, provided *via* elimination of an *O*-methyl substituent.<sup>420</sup> Without more purification, reaction of the product mixture with iodomethane afforded intermediate **420** in 83% yield over two steps. Lastly, compound **420** gave sparstolonin B (SsnB) **416** upon several steps (Scheme 88).<sup>421</sup>

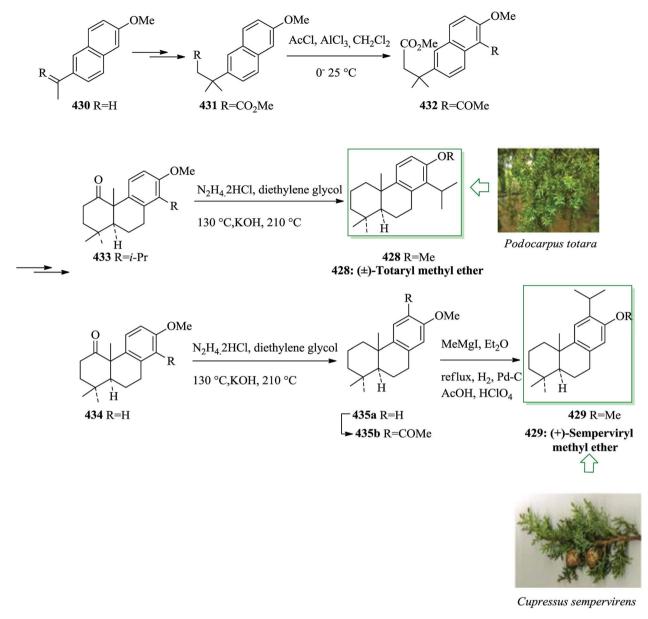
## 2.4. Intermolecular acylation

The lignan lactone picropodophyllin **420** has been obtained from various species of *Podophyllum*.<sup>422,423</sup> Gensler and coworkers demonstrated total synthesis of picropodopyllin. Initially, 3, 4-methylenedioxy-3',4',5'-trimethoxybenzophenone **422** was synthesized from the FC reaction between methylenedioxybenzene **421** and trimethoxybenzoyl chloride **422**. Subsequently, compound **423** provided picropodophyllin **420** after several steps (Scheme 89).<sup>424</sup>

Isoflavones<sup>425</sup> are the most abundant subset of the flavonoid group of compounds, which also includes pterocarpans, rotenoids,426 and coumestans.427 Structurally, isoflavones are highly functionalized and oxygenated derivatives of 3-phenylchromans. Jamaicin 424a has been extracted from Piscidia erythrina bark root.428 Calopogonium isoflavone-B 424b was extracted from ether extracts of Calopogonium mucunoides seeds. The FC acylation reaction has been examined on various acid-sensitive substrates. Based on the proper conditions of changing Lewis acids, solvents, and reaction temperatures, the acylation indeed occurred, therefore obviating the necessity for functional group protection-deprotection sequences. By the use of these methods, the naturally occurring isoflavones jamaicin 424a and calopogonium isoflavone-B 424b have been obtained. Jamaicin 424a and calopogonium isoflavone-B 424b have been obtained from the deoxybenzoin systems 427b and 427a, respectively, by any of a wide range of excellent formylation methodology.429,430 These crucial and sensitive deoxybenzoins were synthesized directly via the FC acylation reaction between 2,2-dimethyl-5hydroxychrom-3-ene 425 and a homologated piperonal-426a or sesamol-426b obtained acid halide (Scheme 90).431

The tricyclic diterpenes sempervirol and totarol incorporate a *trans*-fused octahydrophenanthrene core as the basic carbocyclic scaffold. Totarol includes a modified abietane framework and has been extracted as a major constituent of the heartwood

#### Review

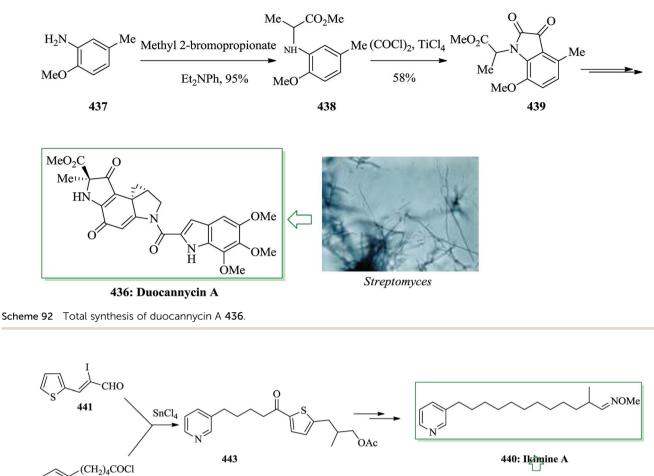


Scheme 91 Total synthesis of  $(\pm)$ -totaryl methyl ether 428 and (+)-semperviryl methyl ether 429.

methoxynaphthalene afforded<sup>435</sup> a mixture of 6-acyl and 1-acyl derivatives. Compound **432** provided the *trans*-fused ketones **434** and **433** *via* several steps. Huang–Minlon reduction<sup>436</sup> of **433** gave (+)-totaryl methyl ether **428** in 84% yield. An analogous reduction of the ketone **434** provided octahydrophenanthrene **435a**. FC acylation reaction between **435a** and AcCl using tin(rv) chloride gave the methyl ketone **435b** as the only product in 80% yield. Reaction of **435b** with MeMgI and catalytic hydrogenolysis of the resulting carbinol in acetic acid with a few drops of perchloric acid gave (+)-semperviryl methyl ether **429** in 80% yield (Scheme 91).<sup>434</sup>

Duocannycin A **436**<sup>437-439</sup> extracted from *Streptomyces* sp. was found to be an enormously active antitumor antibiotic that is effective against several strains of experimental cancer cell lines. Total synthesis of **436** was accomplished using methoxycarbonylation of the C<sub>4</sub>-position of the 5-aminoindoline by way of the isatin and subsequent Dieckmann cyclization to the methyl 2-methylindoxyl-2-carboxylate as key steps. Total synthesis of **436** was started from 6-methoxy-3-methylaniline **437**. Next, one carbon unit was introduced effectively using isatin derivative **439** obtainable *via* the FC reaction of *N*-alkyl derivative **438** with oxalyl chloride. Finally, compound **439** was converted into *dl*-**436** after several steps (Scheme 92).<sup>440</sup>

Ikimine A has been extracted from *Niphates* sp.<sup>441</sup> The cytostatic ikimine A **440** was synthesized *via* acylation of 2-methyl-3-(2-thiophene)-L-propylacetate **441**, reductive de-sulfurization, and functional group transformation to methyl branched racemic and nonracemic chiral 3-alkylpyridines. SnCl<sub>4</sub> mediated the acylation reaction between 2-methyl-3-(2-thiophene)-Lpropylacetate **441** and the acid chloride **442** to afford the ketone





Scheme 93 Ikimine A (α-methyl-3-pyridinedodecanal-O-methyloxime) 440.

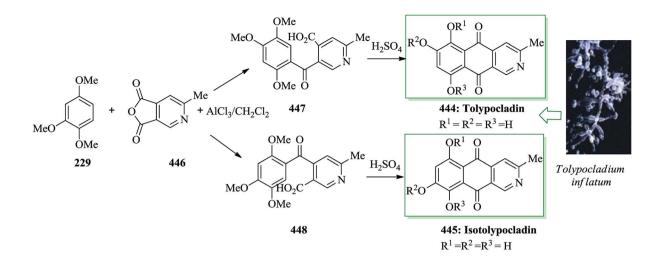
**443.** Reaction with TiCl<sub>4</sub> or H<sub>3</sub>PO<sub>4</sub> as FC catalysts or acylation reaction with *in situ* made acyl trifluoroacetates using phosphoric acid<sup>442</sup> afford lower yields. Lastly, compound **443** afforded ikimine A ( $\alpha$ -methyl-3-pyridinedodecanal-*O*-methyloxime) **440** *via* several steps (Scheme 93).<sup>443</sup>

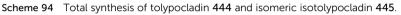
+HCl

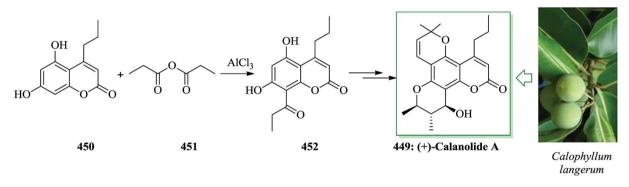
442

Tolypocladin **444** has been extracted from *Tolypocladium inflatum* DSM 915 and identified as 3-methyl-5,6(or 7),8-trihydroxy-2-aza-anthraquinone-(9,10).<sup>444</sup> Total synthesis of the microbial metabolite tolypocladin **444** and isomeric isotolypocladin **445** was developed through FC acylation reaction *via* the condensation of 2-aza-anthraquinone-(9,10) ring. The total synthesis was commenced by the reaction of 1,2,4trimethoxy-benzene **229** and 2-methyl-pyridine 4,5-dicarboxylic acid anhydride **446** based on FC condition reactions. The resultant key products **447** and **448** can be separated though recrystallization. Therefore, cyclization and demethylation of the key products **447** and **448** using sulfuric acid afforded **444** (Scheme 94).<sup>445</sup> (+)-Calanolide A,<sup>446</sup> initially extracted from the dried fruits and twigs of *Calophyllum lanigerum* var. *Austrocoriaceum*, is a non-nucleoside reverse transcriptase inhibitor currently in clinical trials as an anti-HIV agent. Zhang and co-workers in 2000 used an investigation method to perform the FC acylation reaction that was applied in the total synthesis of (+)-calanolide A **449**. A FC acylation reaction was applied to form 5,7dihydroxy-4-propyl-8-propionylcoumarin **452** as a key intermediate for the formation of (+)-calanolide A. This reaction was performed by reacting 5,7-dihydroxycoumarin **450** with 1 equiv. of propionic anhydride aluminium chloride under reflux in dichloroethane (DCE). The optimal reaction conditions used 7 equiv. of aluminium chloride in 1.0 mL of nitromethane, with a predicted excellent yield (97%) (Scheme 95).<sup>447</sup>

Kaempferol **453**<sup>448,449</sup> and other flavonoids have been found to be potential antitumor agents and human immunodeficiency virus (HIV) type 1 integrase inhibitors.<sup>450</sup> Kaempferol and its 3-*O*-glycoside derivatives were identified in several plants,







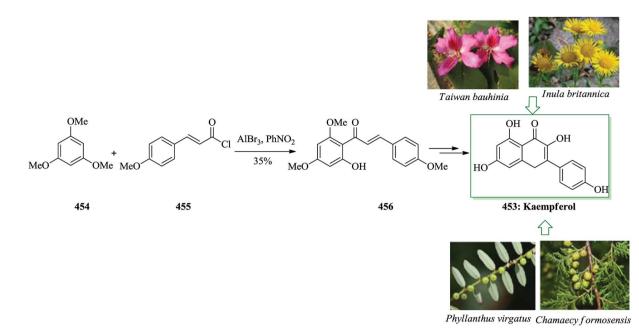
Scheme 95 Total synthesis of (+)-calanolide A 449.

vegetables, fruits, and beverages, for example *Taiwan Bauhinia*,<sup>451</sup> *Chamaecyparis formosensis*,<sup>452</sup> *Inula britannica*,<sup>453</sup> French beans,<sup>454</sup> asparagus,<sup>455</sup> blueberries,<sup>456</sup> grape fruit juice,<sup>457</sup> teas,<sup>458</sup> and honey.<sup>459</sup> Kaempferol **453** was obtained in seven steps with 47% overall yield from 1,3,5-trimethoxybenzene **454**. Firstly, the FC reaction of trimethoxy benzene **454** and 4-methoxycinnamic acid chloride **455** with AlBr<sub>3</sub> as a catalyst provided *o*-hydroxychalcone **456** in 35% yield after several steps. Finally, compound **456** provided the corresponding product **453** upon several steps (Scheme 96).<sup>460</sup>

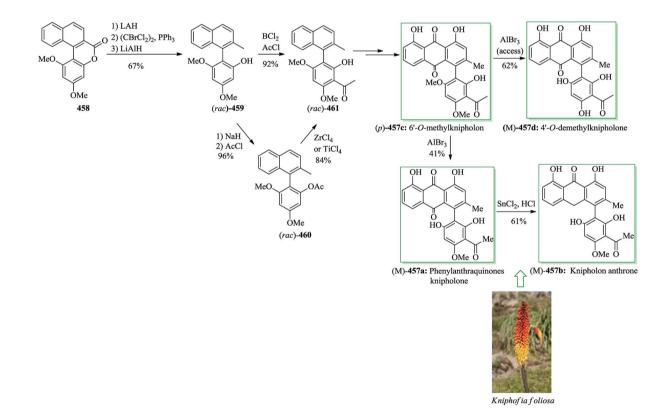
Several naturally occurring phenylanthraquinones, namely knipholone **457a**, knipholone anthrone **457b**, 6'-O-methylknipholone **457c** and 4'-O-demethylknipholone **457d**, demonstrated moderate to high antiplasmodial activity *in vitro* against *Plasmodium falciparum*,<sup>461</sup> the carrier of the most lethal malaria tropica. M-Knipholone **457a** has been extracted from *Kniphofia foliosa*.<sup>462</sup> Bringmann *et al.* in 2002 demonstrated the initial significant synthetic method toward natural knipholone-type phenylanthraquinones. Firstly, lactone **458** afforded the racemic biaryl **459**.<sup>463</sup> With this substrate, the introduction of the *C*-acetyl substituent succeeded both directly, *via* FC acetylation reaction, and in a two-step method, *via* an *ortho*-selective Fries reaction<sup>464</sup> on ester **461**, and afforded the *C*-acetyl compound **460** in moderate yields, without any removal of the methyl ether. Therefore, compound **460** afforded 6'-*O*-methylknipholone **457c**, the first natural phenylanthraquinone. Selective mono-*O*-demethylation of **457c** at C-6' employing aluminium bromide afforded the target molecule knipholone **457a**, reduction of which afforded its anthrone **457b**. The fully *O*-demethylated derivative **457d** was obtained by employing aluminium bromide in excess (Scheme 97).<sup>463</sup>

The polycitones A 462 and B 463, polybrominated pyrrole alkaloids, were extracted by Kashman et al.465 from marine ascidians of the genus Polycitor. Polycitone A is an active inhibitor of retroviral reverse transcriptases and cellular DNA polymerases.<sup>466</sup> Polycitone B 463 was obtained in four steps from pyrrole dicarboxylic acid 464, involving FC reaction of the desired acid chloride with anisole. The transformation of 462 into polycitone A 463 was accomplished in two steps through Mitsunobu alkylation of the pyrrolic NH group. Polycitone A was obtained in 18% overall yield and gave the probability of changing the groups on the pyrrole ring. This synthetic method was started from dicarboxylic acid 464, which reacted with oxalyl chloride followed by elimination of the solvent and rigorous drying, which afforded the crude acid chloride 465. Compound 465 was treated with anisole under FC reaction conditions to provide the diketone 466 in 66% yield. In the following, compound 466 afforded polycitone B 462 after

(0)



Scheme 96 Total synthesis of kaempferol 453.

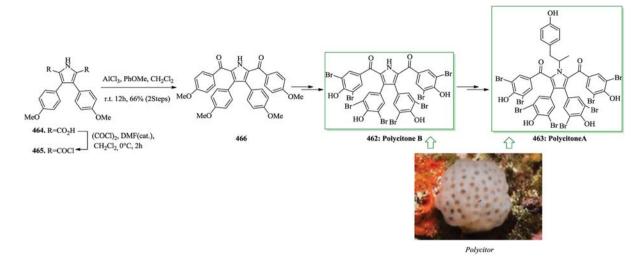


Scheme 97 Total synthesis of phenylanthraquinones knipholone 457a, knipholone anthrone 457b, 6'-O-methylknipholone 457c and 4'-O-demethylknipholone 457d.

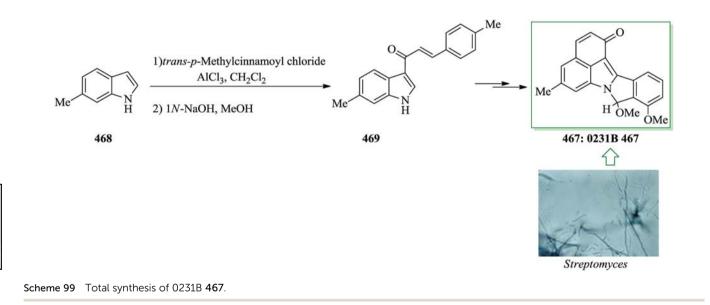
several steps and then polycitone B **462** was transformed into polycitone A **463** after several steps (Scheme 98).<sup>356</sup>

The unique inhibitors of  $3\alpha$ -hydroxysteroid dehydrogenase, 0231A and 0231B, have been extracted from a fermentation

broth of *Streptomyces* sp. HKI0231.<sup>467</sup> These compounds are promising as lead structures for anti-inflammatory agents. 0231A and 0231B have a unique benz[c,d]indol-3(1*H*)-one scaffold in their molecules. Nakatsuka and co-workers in 2003



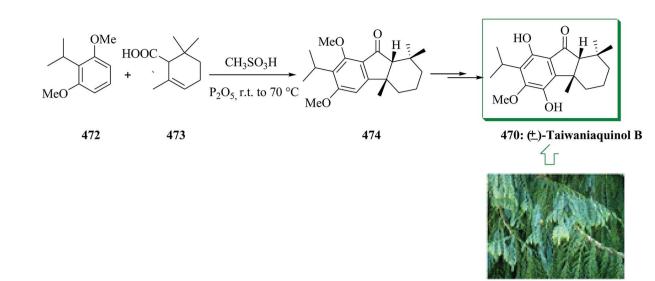
Scheme 98 Total synthesis of polycitone B 462 and polycitone A 463.



conditions to provide the diketone **466** in 66% yield. In the following, compound **466** afforded polycitone B **462** after several steps and then polycitone B **462** was transformed into polycitone A **463** after several steps (Scheme 98).<sup>356</sup>

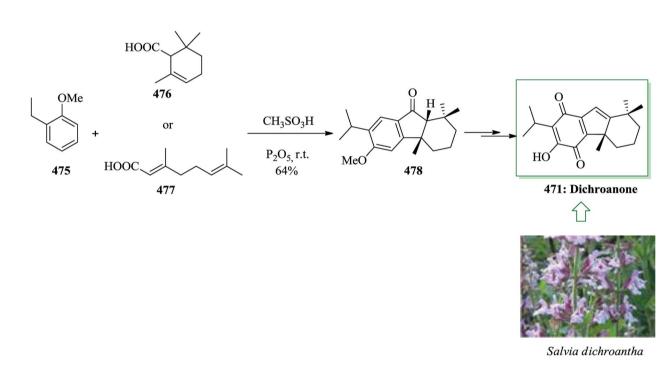
The unique inhibitors of  $3\alpha$ -hydroxysteroid dehydrogenase, 0231A and 0231B, have been extracted from a fermentation broth of *Streptomyces* sp. HKI0231.<sup>467</sup> These compounds are promising as lead structures for anti-inflammatory agents. 0231A and 0231B have a unique benz[c,d]indol-3(1H)-one scaffold in their molecules. Nakatsuka and co-workers in 2003 reported the synthesis of 0231B **467** in 10 steps (8.1% overall yield) from 6-methylindole **468**.<sup>468</sup> Upon masking the nitrogen with a pivaloyl group, the introduction of a *trans-p*-methylcinnamoyl group was achieved at the 3-position *via* FC acylation,<sup>469</sup> and subsequent de-protection of the pivaloyl group gave the *trans-p*- methylcinnamoyl derivative **469** in an 87% overall yield from **468**. Finally, compound **469** was converted into 0231B **467** upon several steps (Scheme 99).<sup>468</sup>

Various diterpenoids have the 4a-methyltetra- (and hexa-) hydrofluorene framework, for example, taiwaniaquinols A, B, and D and taiwaniaquinone D and H, as well as diverse structurally relevant diterpenoids. Dichroanone, dichroanal B, and standishinal were extracted from *Taiwania cryptomerioides*, *Salvia dichroantha* and *Thuja standishii*.<sup>84,86,470</sup> A significant acidpromoted domino FC acylation/alkylation reaction was known as the key step. Significantly, the formal total syntheses of diterpenoids ( $\pm$ )-taiwaniaquinol B **470** and ( $\pm$ )-dichroanone **471** were performed. Total synthesis of ( $\pm$ )-taiwaniaquinol B **470** was accomplished in 2008 by She and co-workers.<sup>471</sup> Total synthesis of **470** was accomplished from the reaction of 1,3-



Taiwania cryptomerioides

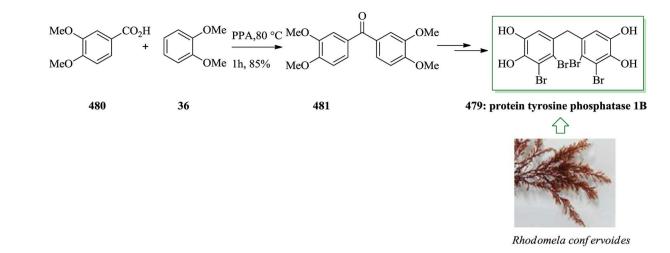
### Scheme 100 Total synthesis of $(\pm)$ -taiwaniaquinol B 470.



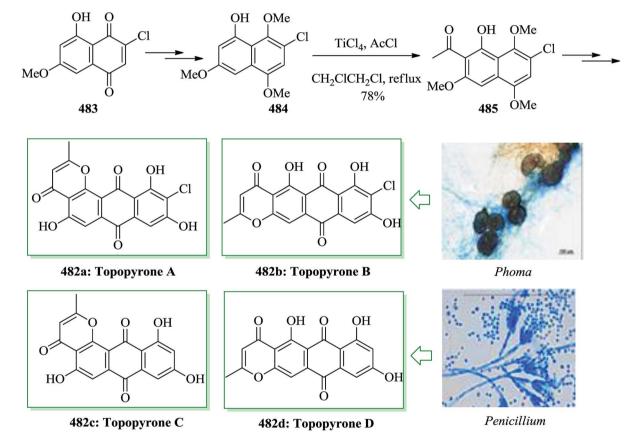


major isomer **478** was obtained by employing the domino FC acylation/alkylation reaction of compound **475** with geranic acid **477**, giving a 64% yield after purification by column chromatography. Finally, compound **478** afforded ( $\pm$ )-dichroanone **471** in several steps and its spectra were found to be identical with those formerly reported (Scheme 101).<sup>86,471,472</sup>

Protein tyrosine phosphatase 1B (PTP1B) has aroused intensive research interest because of its involvement in the insulin signaling cascade as a major negative regulator.<sup>473</sup> Bioassay-guided separation of the ethanol extract using a wide range of chromatographic techniques resulted in a series of bromophenols. One of them, named bis-(2,3-dibromo-4,5dihydroxyphenyl)-methane, exhibited surprising inhibitory activity against PTP1B. Bis-(2,3-dibromo-4,5-dihydroxyphenyl)methane **479**, which was extracted from red algae *Rhodomela confervoides*, was demonstrated as a natural bromophenol with important inhibition against PTP1B. Total synthesis of compound **479** was achieved with an overall yield of 24%. Initially, 3,4-dimethoxybenzoic acid **481** on reaction with 1,2dimethoxybenzene **36** using polyphosphoric acid at 80 °C for one hour easily gave **481** in 85% yield.<sup>474</sup> Finally, compound **481** 





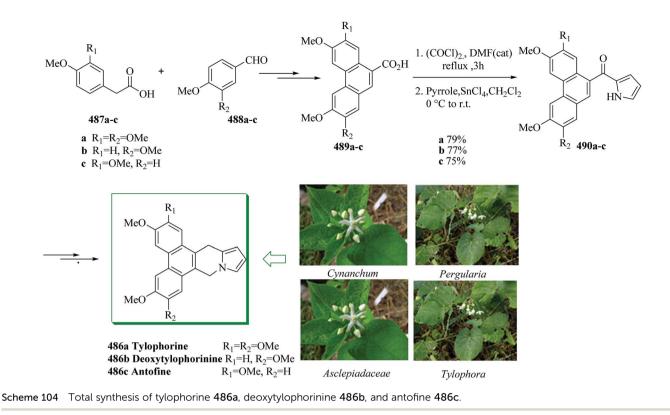


Scheme 103 Total synthesis of topopyrone A 482a, B 482b, C 482c and D 482d

afforded compound **479** after several steps in 24% overall yield (Scheme 102).<sup>475</sup>

The topopyrones, a group of planar anthraquinone polyphenols, were identified by the orientation of the pyrone ring appended on one side as well as the presence or lack of a chlorine at C7. Topopyrones A **482a** and B **482b** contain a chloro group at C7, *i.e.*, that part of the molecule distant from the pyrone ring, while topopyrones C **482c** and D **482d** are unfunctionalized at C7. The topopyrones were extracted from

the culture broths of two fungi, *Phoma* sp. BAUA2861 and *Penicillium* sp. BAUA4206. Particularly, topopyrones A, B, C, and D selectively inhibited the growth of yeast expressing human topoisomerase I. The topopyrones exhibit a unique group of extremely cytotoxic topoisomerase I poisons. Efficient total synthesis of all four naturally occurring members of this group was achieved in 2008 by Hecht *et al.*<sup>476</sup> Main steps are Diels-Alder reaction and a titanium-catalyzed *ortho*-directed FC acylation. For the synthesis of topopyrone A **482a**, B **482b**, C **482c** 

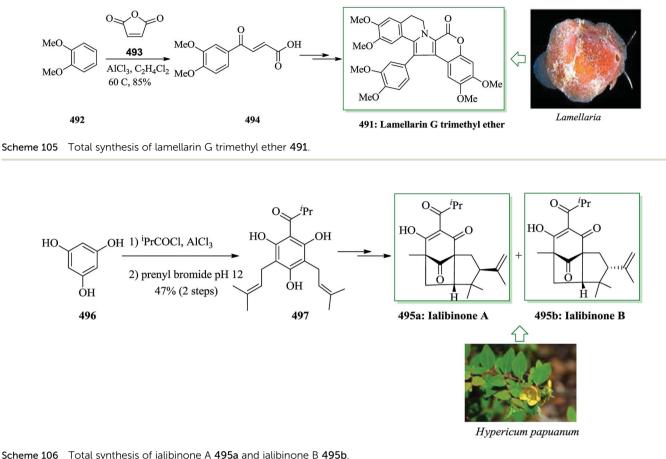


and D **482d**, initially quinone **483**<sup>477</sup> was converted into phenol **484** in several steps. At this point  $\text{TiCl}_4$  was selected to perform regiocontrolled FC acylations. Actually, it was known that reaction of **484** in 1,2-dichloroethane with titanium tetrachloride and AcCl under reflux introduced the needed acyl side chain selectively at the *ortho* position giving **485** in 78% yield. Finally, compound **485** gave topopyrone A **482a**, B **482b**, C **482c** and D **482d** after several steps (*via* different routes) (Scheme 103).<sup>476</sup>

Phenanthroindolizidine alkaloids, extracted from Pergularia, Cynanchum, Tylophora, and some genera of the Asclepiadaceae group, show significant biological, pharmacological,478,479 and antitumor properties.480-483 (-)-Antofine 486c, extracted from Cynanchum komarovii, exhibited high antiviral activity against the tobacco mosaic virus (TMV).484,485 Wang et al. in 2008 reported the total synthesis of tylophorine 486a, deoxytylophorinine 486b, and antofine 486c starting from pyrrole in 48%, 44%, and 46% overall yields, respectively.486 In this route for the synthesis of phenanthroindolizidine alkaloids 486a, 486b, and **486c**, the reaction of benzeneacetic acid **487a-c** and aromatic aldehyde 488a-c afforded phenanthrene-9-carboxylic acid 489ac via several steps. Then, the intermolecular FC acylation of the corresponding acyl chloride (synthesized by chlorination of acid 489a with oxalyl chloride) and pyrrole was catalyzed by tin(IV) chloride to form pyrrolidine ring system 490a under mild conditions in 79% yield. After three steps, the racemic tylophorine 486a was provided. Thus, the shortest synthetic pathway to a large-scale construction of tylophorine 486a was accomplished starting from pyrrole under mild conditions without any protecting group in 48% overall yield. This group

demonstrated the construction of deoxytylophorinine 486b and antofine 486c. Deoxytylophorinine 486b and antofine 486c were obtained in 44% and 46% overall yields, respectively, in six steps including a Perkin condensation reaction, intramolecular oxidative coupling reaction of acids, chlorination of acids 489b,c, and subsequent intermolecular FC reactions with pyrrole in one pot, deketonization of 2-acylpyrroles 490b,c, catalytic hydrogenation of 2-alkylpyrroles, and Pictet-Spengler cyclomethylenation. The versatility and flexibility of this approach were exhibited by the large-scale construction of three representative phenanthroindolizidine alkaloids, tylophorine antofine 486a, deoxytylophorinine 486b, and 486c (Scheme 104).486

Lamellarin G, a marine natural product, includes a 5-oxa-6baza dibenzo[a,i]fluoren-6-one framework. Lamellarins were extracted from the prosobranch mollusk Lamellaria sp. and the ascidians.487 Some of these lamellarins488,489 show active biological properties,490 for example cytotoxicity to a series of cancer cell lines, cell division inhibition, and immunomodulation. A modular synthesis of the lamellarin G trimethyl ether was established using various reactions in sequence, including FC acylation, esterification, haloarylation, and oxidative cyclization. This method provided the complete structure of lamellarin G trimethyl ether in four steps with 44% overall yield. Total synthesis of lamellarin G trimethyl ether was accomplished in 2009 by Yadav et al.491 In this approach, 4-(3,4-dimethoxyphenyl)-4-oxo-but-2-enoic acid 494 was obtained from veratrole 492 and maleic anhydride via FC reaction.492 Subsequently, compound 494 afforded the desired target product lamellarin G trimethyl ether 491 upon several steps (Scheme 105).491



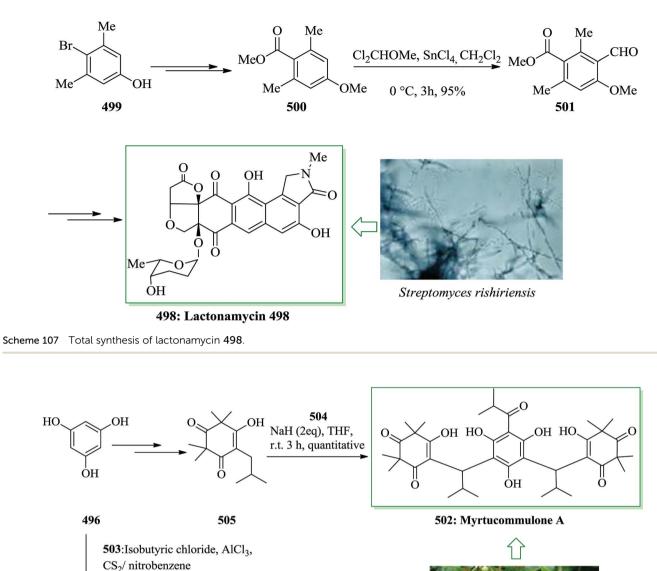
The epimeric ialibinones A and B 495, uncommon types of PPAP, were extracted from the leaves of Hypericum papuanum by Sticher et al. in 2000.493 They show antibacterial activity against cytotoxic against KB cell lines.493,494 The tricyclic natural products ialibinone A and ialibinone B were synthesized as a 41:59 mixture in four steps starting from phloroglucinol. The synthetic sequence included acylation reaction of phloroglucinol through FC reaction, double prenylation, dearomatizing methylation, and oxidative free radical cyclization. Initially, the bis-prenylated acylphloroglucinol 497 was synthesized in two steps from phloroglucinol 496 via FC acylation reaction. Next, compound 497 provided the target natural products, ialibinone A 495a and ialibinone B 495b, as an inseparable mixture in 35% overall yield (Scheme 106).495

Lactonamycin 498, extracted from a culture broth of Streptomyces rishiriensis MJ773-88K4, exhibited active antimicrobial properties against Gram-positive bacteria and vancomycinresistant Enterococcus (VRE) as well as antitumor properties.496,497 Structurally, compound 498 contains a hexacyclic system and a glycosidic bond at the *t*-alcohol.<sup>497</sup> The first total synthesis of lactonamycin 498 was accomplished in 2010 by Tatsuta et al.498 The synthesis includes sequential intramolecular conjugate addition reaction of alcohols to the acetylenic ester, asymmetric glycosylation of the tertiary alcohol, and Michael-Dieckmann type cyclization with the thioester. The

synthesis of lactonamycin 498 was started from 4-bromo-3,5dimethylphenol 499, that after two steps gave ester 500.499 Then, FC-type formylation reaction<sup>500</sup> of **500**, gave aldehyde **501**. Upon several steps, compound 501 gave lactonamycin 498 (Scheme 107).498

Myrtucommulone A 502 was developed in 1974 by Kashman et al. as a substance known in the common myrtle Myrtus communis L.<sup>501</sup> Jauch et al. in 2010 reported synthesis of myrtucommulone A 502 from isobutyryl phloroglucinol 504, isobutyraldehyde, and syncarpic acid in one step. Isobutyryl phloroglucinol 504 is easily accessible via FC acylation of phloroglucinol 496 in 70-80% yield.502,503 Compound 496 after three steps gave syncarpic acid.504,505 Compound 503 is also accessible from 496 upon several steps. In this part, two acidcatalyzed FC alkylation reactions occurred consecutively in a one-pot reaction. Running the FC alkylation under basic conditions needed removal of the acid and extraction of 505 as a crude product prior to reaction with 504, which had been deprotonated with two equivalents of sodium hydride in tetrahydrofuran. Finally, the synthesis of 502 was completed in three hours at ambient temperature in quantitative yield (Scheme 108).506

The tylophora alkaloids have obtained synthetic and medicinal attention owing to their panoply of biological properties, such as antibacterial, anticancer, antifungal, antiviral,



Scheme 108 Total synthesis of myrtucommulone A 502.

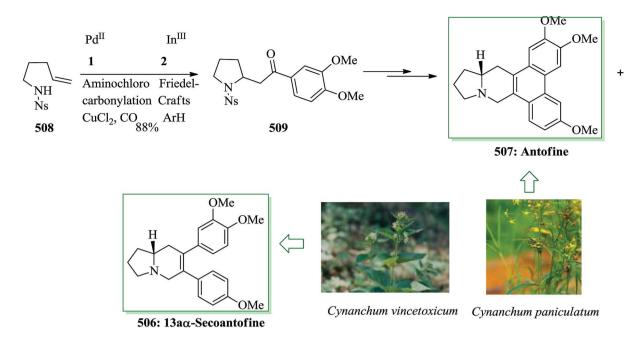
OH

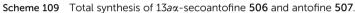
ÓН 504 reflux, 70%

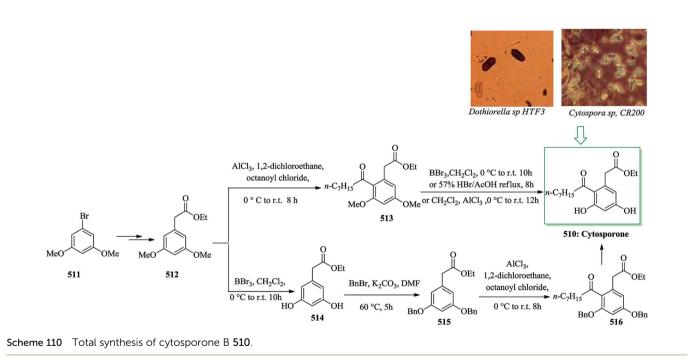
and anti-inflammatory properties.<sup>507</sup> Antofine and  $13a\alpha$ secoantofine were extracted from the root of *Cynanchum paniculatum* Kitagawa (*Asclepiadaceae*)<sup>508</sup> and from aerial parts of *Cynanchum vincetoxicum*,<sup>509</sup> respectively.<sup>510</sup> A fast synthetic method to the tylophora alkaloids antofine and  $13a\alpha$ -secoantofine was achieved. The key step in this synthetic methodology was the multicatalytic aminochlorocarbonylation/FC acylation reaction of a protected 4-pentenyl amine to give a  $\beta$ -pyrrolidinyl ketone intermediate. Thus, after exposing the easily accessible nosyl protected alkenyl amine **508** and veratrole to amino-chlorocarbonylation, it was gratifying to find that the  $\beta$ -pyrrolidinyl ketone **509** could be isolated in high yield. Finally, compound **509** afforded 13*a* $\alpha$ -secoantofine and antofine in 19% and 48% overall yield, respectively (Scheme 109).<sup>511</sup>

Mytus communis

HO

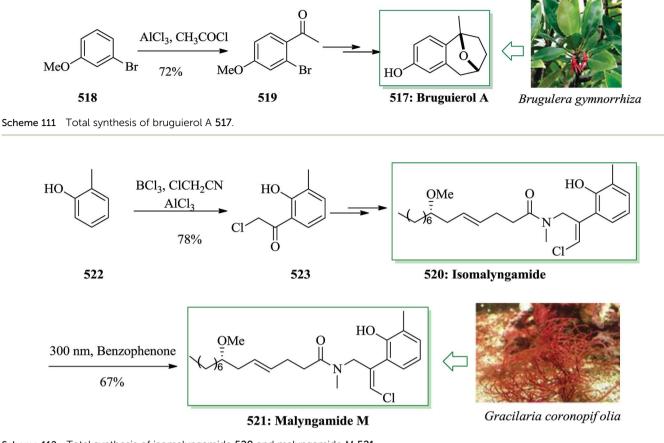






Cytosporone B **510**, extracted from *Cytospora* sp. CR200 and *Dothiorella* sp. HTF3,<sup>512</sup> exhibits antibacterial activity.<sup>512-515</sup> The total synthesis of cytosporone B was achieved from 1-bromo-3,5-dimethoxybenzene. The key steps are sequential Grignard reaction and Lemieux-van Rudloff oxidation followed by a deprotection of the methyl aromatic ether to phenol and subsequent FC acylation. In this route, total synthesis of cytosporone B was started from 1-bromo-3,5-dimethoxybenzene **511**. Next, it was converted into the key intermediate **512** *via* 

three steps. Then, with the key intermediate **512** obtained by oxidation and esterification, the sequential FC acylation and deprotection were accomplished subsequently. The FC reaction progressed with 86% yield of the purified product. The ether in **512** was deprotected to phenol **514** using BBr<sub>3</sub>, followed by etherification with benzyl bromide to provide the intermediate **515** (56% yield, two steps). The hydroxyl protection step was followed by an attempt at acylation of this protected phenol with octanoyl chloride catalyzed by aluminium chloride in 83%



Scheme 112 Total synthesis of isomalyngamide 520 and malyngamide M 521

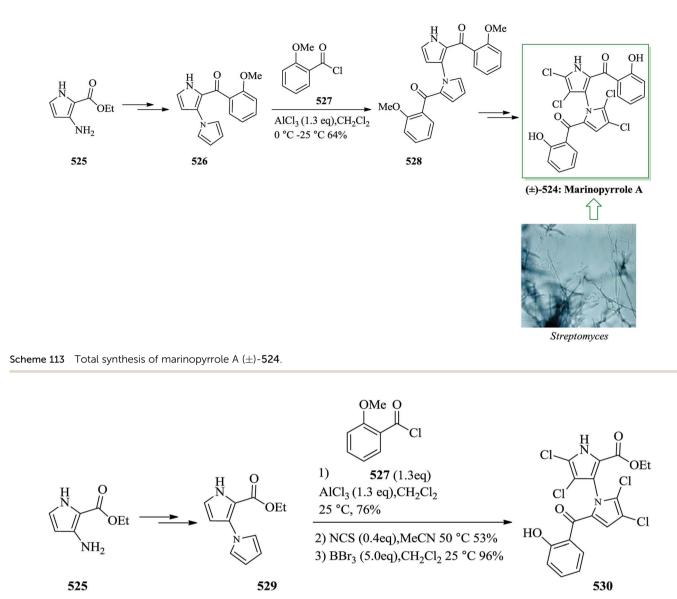
yield. Finally, the debenzylation of ethyl 2-(3,5-bis(benzyloxy)-2-octanoylphenyl)acetate **516** under  $H_2$  progressed very easily (95% yield) and very pure cytosporone B **510** was obtained (Scheme 110).<sup>512</sup>

Bruguierol A **517** was extracted and identified by Sattler *et al.* from the stem of *Bruguiera gymnorrhiza* tree in 2005.<sup>66</sup> This natural product has a unique structure characterized by a 2,3benzofused 8-oxabicyclo[3.2.1]octane core. Total synthesis of  $(\pm)$ -bruguierol A **517** was achieved in 10 steps and with an overall 16.8% yield. The embedded unique 8-oxabicyclo[3.2.1] octane unit framework in this natural product was generated through a novel Sc(OTf)<sub>3</sub>-catalyzed intramolecular [3 + 2] cycloaddition of cyclopropane. The synthesis of bruguierol A **517** was achieved *via* FC acylation reaction of 3-bromoanisole **518** to give compound **519** in 72% yield.<sup>516,517</sup> Finally,  $(\pm)$ -bruguierol A **517** was synthesized after several steps from compound **519** (Scheme 111).

The malyngamides, usual metabolites of *Lyngbya majuscula*, are N-functionalized amides of long chain fatty acids. A subgroup of malyngamides contains a communal and remarkable terminal vinyl chloride functionality, for example malyngamides A, B, M, and isomalyngamides A, B, M. Malyngamide M **521** was extracted from the Hawaiian red alga *Gracilaria coronopifolia*.<sup>518</sup> It was the first example of a natural aromatized malyngamide that possesses a special terminal vinyl chloride substructure. The first concise and significant asymmetric

synthesis of malyngamide M 521 was achieved in nine steps from o-cresol in 12% overall yield. The key steps included the Wittig reaction, FC acetylation, amidation, and isomerization reaction. The isomalyngamide M 520 was also obtained in 2010 Cao et al.<sup>519</sup> In this pathway, total synthesis of bv malyngamide M 521 was started from o-cresol 522. FC acetylation reaction of 522 with chloroacetonitrile using boron trichloride and aluminium trichloride afforded phenol 523 in 78% yield.520 Next, after several steps, compound 523 gave isomalyngamide 520 in 83% yield. Finally, compound 520 was transformed into a mixture of 521/Z-521 (2.5 : 1) when exposed to UV light using benzophenone in dichloromethane for 8 h at room temperature.521 Fortunately, the two isomers could be simply separated by flash chromatography over silica gel, and pure 521 was provided in 67% yield (Scheme 112).519

Marinopyrrole A **524** is an alkaloid endowed with promising antibiotic properties against methicillin-resistant *Staphylococcus aureus* (MRSA)<sup>523</sup> that was extracted from an obligate marine *Streptomyces*. This structurally uncommon molecule exists at room temperature as enantiopure M-(–)-atropisomers. Nicolaou *et al.* developed total synthesis of the antibiotic marinopyrrole A **524** in five steps and 16% overall yield from aminopyrrole **525**.<sup>524</sup> The synthesis of marinopyrrole (–)-**524** (natural) and (+)-**524** (unnatural) was started from aminopyrrole **525**, which was converted into tricycle **526** upon several steps. FC arylation reaction of **526** and acid chloride **527**, catalyzed by



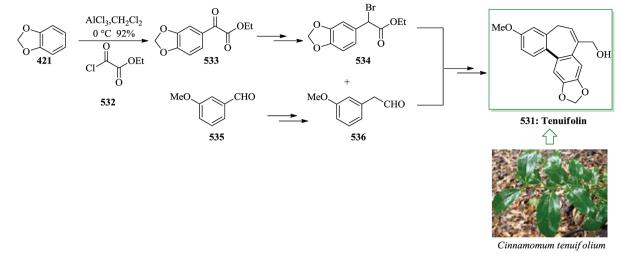
Scheme 114 Total synthesis of mono-arylated marinopyrrole 530.

aluminium chloride in dichloromethane, resulted in the marinopyrrole unit structure **528** in 64% yield. Finally, compound **528** afforded racemic marinopyrrole A [ $(\pm)$ -**524**] *via* several steps (Scheme 113).<sup>524</sup>

In a similar method, mono-arylated marinopyrrole **530** was synthesized from bis-pyrrole **529** (obtained from compound **525**), *via* a three-step sequence containing a FC C-arylation with acid chloride **527**, *tetra*-chlorination with NCS in acetonitrile, and demethylation with BBr<sub>3</sub> in dichloromethane to provide the target product **530** in 39% overall yield (Scheme 114).<sup>525</sup>

Biaryl compounds, containing a 6,7,6-ring system, include a wide range of naturally occurring compounds and synthetic pharmaceuticals, for example alkaloids as well as cyclolignans. Several of them showed activity against various cancer cell lines. Lin *et al.* extracted a unique sesquiterpenoid, tenuifolin **531**, from the stems of *Cinnamomum tenuifolium*.<sup>526</sup> In the structure of **531** exists a similar moiety to allocolchicine and a preliminary bioassay indicated that **531** only exhibited weak antiproliferative activity against tumor cell line DU145.<sup>526</sup> The first total synthesis of a sesquiterpenoid, tenuifolin **531**, was accomplished in seven linear steps from market-accessible benzodioxole **421**. A FC acylation reaction of **421** using ethyl 2-chloro-2-oxoacetate **532** gave compound **533** in 92% yield. Next, compound **533** was converted into bromoester **534** *via* several steps. Lastly, bromoester **534** and aldehyde **536** afforded tenuifolin **531** in excellent yield (91%) (Scheme 115).<sup>527</sup>

Dictyodendrins were extracted in 2003 by Fusetani and Matsunaga from the extracts of a Japanese marine invertebrate, the marine sponge *Dictyodendrilla verongiformis*, which was from Nagashima Island in Kagoshima, Japan. These compounds were the first marine alkaloids that contain inhibitory activity against telomerase.<sup>528</sup> A highly effective total synthesis of dictyodendrins A–E was accomplished. The synthesis demonstrates a novel benzyne-catalyzed one-pot indoline construction/cross-coupling sequence. Firstly, 2,6dibromo-phenylethylamine **539** was transformed into the



Scheme 115 Total synthesis of tenuifolin 531.

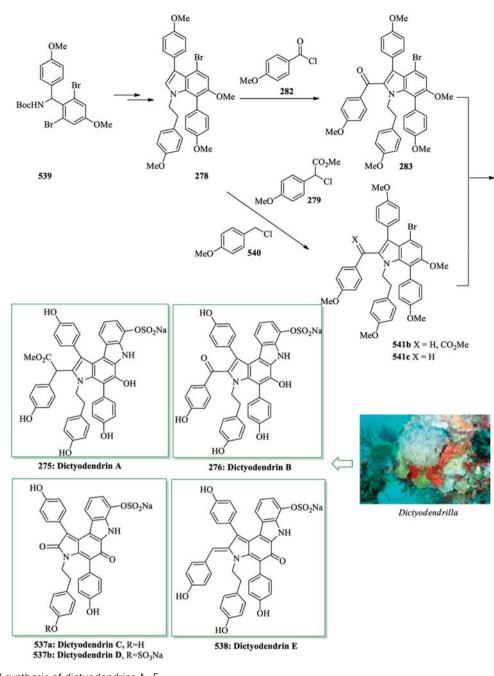
pivotal indole 278 via removal of the Boc group, DDQ oxidation to the indole,<sup>529</sup> and attachment of apara anisylethyl group onto the nitrogen atom. Then, the stage was set for the installation of subunits on the indole 2-position for the synthesis of dictyodendrin A 275, B 276, and E 538. 2-Acylindole derivative 283 was provided in almost quantitative yield through FC acylation reaction<sup>264</sup> with para-methoxybenzoyl chloride 282 and zinc chloride as a promoter (Table 1, entry 1). In contrast, installation of the para-anisyl-acetate group was needed for synthesis of dictyo-dendrin A 275 (Table 1, entry 2). It was found that silver triflate530 was effective and the reaction progressed to afford the corresponding product 541 in high yield (Table 1, entry 3). To provide the intermediate to dictyodendrin E 538, it was necessary to introduce the para-methoxybenzyl group onto the indole 2-position. Unfortunately, reaction with ZnCl<sub>2</sub> or silver triflate did not afford the corresponding product at all. It was assumed that the difficulty with para-methoxybenzylation was because of the high reactivity of para-methoxybenzyl chloride 540 and the presence of electron-rich aromatic rings in the substrate, which induce nonselective benzylation. Moreover, compound 278 afforded intermediates 541 and 542. Finally, compound 541 provided the dictyodendrins upon several steps (Scheme 116).531

Entry	RX	Lewis acid	Product	Yield [%]
1 <sup><i>a</i></sup>	202	7-0	282	00
-	282	$ZnCl_2$	283	99
$2^a$ $3^b$	279	$ZnCl_2$	541	
	279	AgOTf	541	81
$4^a$	540	$ZnCl_2$	542	Trace <sup>c</sup>
$5^b$	$540^{d}$	AgOTf	542	_

<sup>a</sup> Reaction condition: ZnCl<sub>2</sub> (10 equiv.), RX (2 equiv.), Et<sub>2</sub>O, 0 °C.
 <sup>b</sup> Reaction condition: AgOTf (4 equiv.), RX (3 equiv.), THF, -78 °C.
 <sup>c</sup> Complex mixture. <sup>d</sup> 540 two equivalents. Reaction temperature: -78 to 0 °C.

Miles et al. extracted cadinane sesquiterpene lactones heritol 544<sup>532</sup> and heritonin 543<sup>533</sup> from the sap of the mangrove plant Heritiera littoralis in the Philippines, which were demonstrated to possess ichthyotoxicity in parts per million quantities to Tilapia nilotica fingerlings. Chavan et al. reported the highly diastereoselective total synthesis of racemic heritonin 543 and heritol 544 via intramolecular cyclization on a preprovided sensitive butenolide functionality from market-purchasable initiating precursors in eight and nine purification operation in 43% and 33% overall yield, respectively. Synthesis of heritol 544 and heritonin 543 was started with FC acylation reaction of o-cresol methyl ether 545 and allylacetyl chloride in CH<sub>2</sub>Cl<sub>2</sub> at ambient temperature to provide the desired allylic keto compound 546 in 95% yield. Next, compound 546 was converted into acid 547 via several steps. Then, compound 547 provided the key intermediate tetralone 548 through intramolecular FC acylation in 95% yield (over two steps). Compound 548 afforded compound 549 via several steps. The latter was converted into heritonin 543 in 96% yield and lastly compound 543 gave heritol 544 in 80% yield (Scheme 117).534

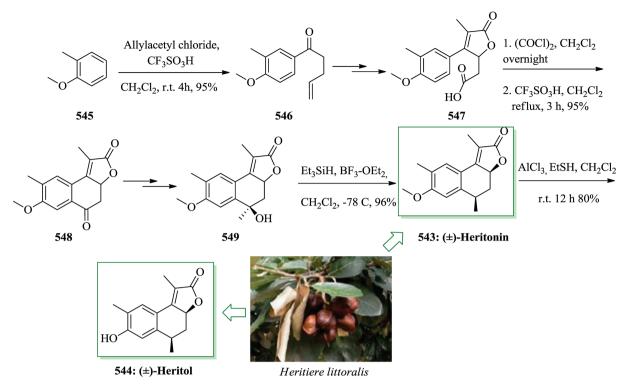
Lysidicins A-C were extracted from Lisidicie rhodostegia Hance (Fabaceae) by Yu et al. in 2006535 and the isolation of lysidicins D-H was reported by the same group in 2007536 and 2010.537 Meanwhile, among the lysidicin group, lysidicin A 553 has the most unique and complicated structure in which two acetals form a spiro[furan-furofuran] ring system. Watanabe et al. in 2012 reported the initial total synthesis of  $(\pm)$ -lysidicin A significantly through single and cascade Claisen rearrangements and FC acylation reaction with AgOTf.538 In this method, the overall yield was 3.5% in 15 steps from diol 551. The total synthesis was started from diol 551, which was converted into benzyl ether 552 after several steps. Next, FC acylation reaction of 552 gave 553a and 553b in 27% and 35% yield (in two steps), respectively. Both 550a and 550b are regioisomeric compounds of lysidicin A; they were separately exposed to acid-mediated isomerization of the acetalic spiro[furan-furofuran] ring

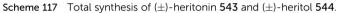


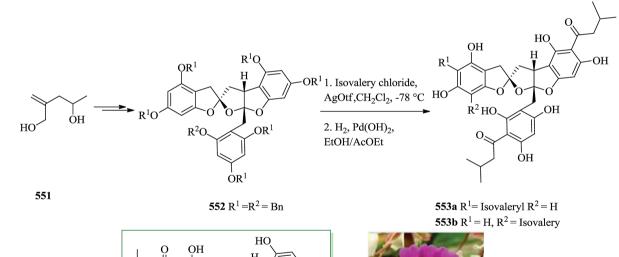
Scheme 116 Total synthesis of dictyodendrins A–E.

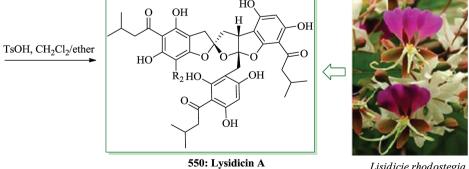
*et al.* in 2012 reported the initial total synthesis of  $(\pm)$ -lysidicin A significantly through single and cascade Claisen rearrangements and FC acylation reaction with AgOTf.<sup>538</sup> In this method, the overall yield was 3.5% in 15 steps from diol **551**. The total synthesis was started from diol **551**, which was converted into benzyl ether **552** after several steps. Next, FC acylation reaction of **552** gave **553a** and **553b** in 27% and 35% yield (in two steps), respectively. Both **550a** and **550b** are regioisomeric compounds of lysidicin A; they were separately exposed to acid-mediated isomerization of the acetalic spiro[furan-furofuran] ring system and finally total synthesis of ( $\pm$ )-lysidicin A **550** was achieved (Scheme 118).<sup>538</sup>

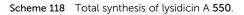
The cultivation of *Streptomyces armeniacus* under specific conditions resulted in the discovery of the armeniaspiroles, a unique group of compounds biosynthetically related to streptopyrroles.<sup>539</sup> Armeniaspiroles, a unique group of natural products extracted from *Streptomyces armeniacus*, are identified by a novel spiro[4.4]non-8-ene moiety. Several derivatives of armeniaspiroles were obtained *via* halogenation, alkylation, addition/removal or reduction. A total synthesis of the 5-chloro analogue of armeniaspirole A was achieved in a linear six-step sequence. 5-Chloro-armeniaspirole A shows moderate activity against a series of multidrug-resistant, Gram-positive bacterial pathogens. A biomimetic synthesis of the armeniaspirole unit was envisaged starting from **556**. The pyrrolo-phenone



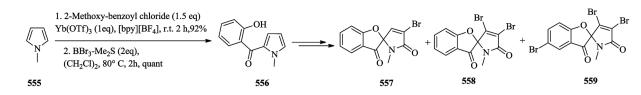


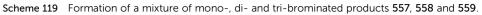


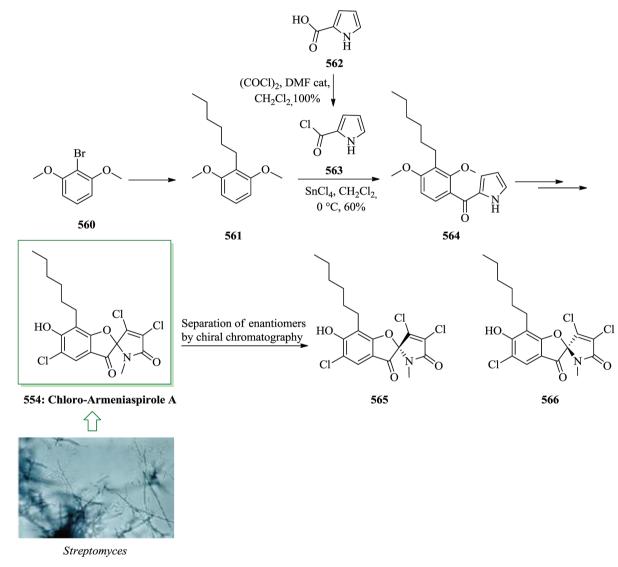




Lisidicie rhodostegia



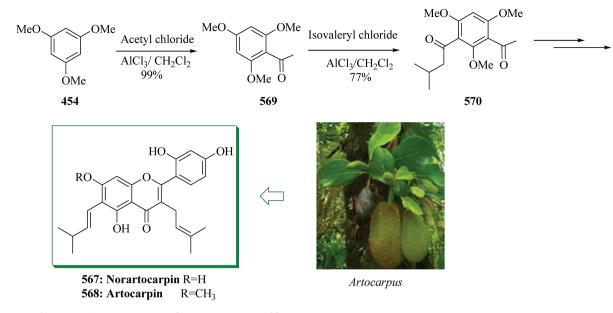




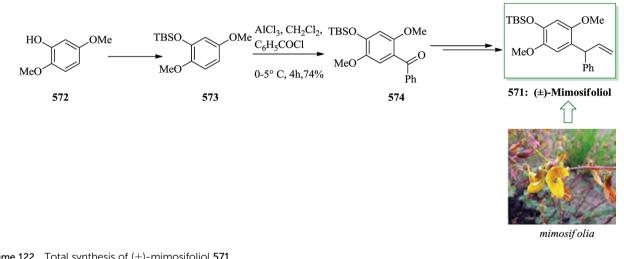
Scheme 120 Total synthesis of 5-chloro-armeniaspirol 554.

compound **564** in a satisfactory 60% yield. Compound **564** gave compound **554** in 72% yields after several steps. The two enantiomers of racemic 5-chloro-armeniaspirole A **554** were finally separated by chiral chromatography to afford enantiopure 5-chloro-armeniaspirole A **565** (*R*-configuration) and **566** (*S*-configuration).<sup>540</sup>

Norartocarpin 567 and artocarpin 568 are natural isoprenylated flavonoids isolated from the genus *Artocarpus*.<sup>541-543</sup> They have a wide range of remarkable biological properties, including inhibitor effects on melanin biosynthesis and  $5\alpha$ -reductase, as well as antibacterial and cytotoxic activity.<sup>544</sup> The total syntheses of norartocarpin and artocarpin, two biologically remarkable natural flavonoids with two regioisomeric isoprenyl side chains, were accomplished for the first time through a linear reaction sequence of 9 and 12 steps with overall yields of 14% and 3.5%, respectively, starting from 1,3,5-trimethoxybenzene. Starting from 1,3,5-trimethoxybenzene **454**, twice sequential FC acylation of **454** in CH<sub>2</sub>Cl<sub>2</sub> with aluminium



Scheme 121 Synthesis of norartocarpin 567 and artocarpin 568



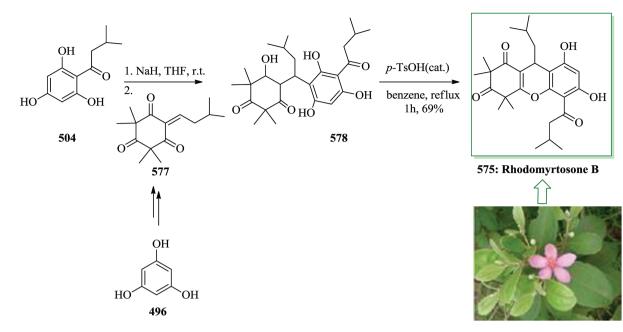
Scheme 122 Total synthesis of  $(\pm)$ -mimosifoliol 571.

chloride using acetyl chloride and 3-methylbutanoyl chloride as the acylated reagents, respectively, gave the intermediate 570 in moderate yields (75% for two steps). In the course of FC acylation, the corresponding demethylation simultaneously happened, which may be related to the effect of two carbonyl groups at ortho positions. On the other hand, if the two FC reactions were accomplished in a reversed sequence, 2,4,6-trimethoxyacetylbenzene 569 rather than the desired target molecule 570 was obtained as the major product, *i.e.*, the isopentanoyl group was removed through reverse FC reaction during the second step. Finally, compounds 567 and 568 were obtained from 570 via several steps (Scheme 121).545

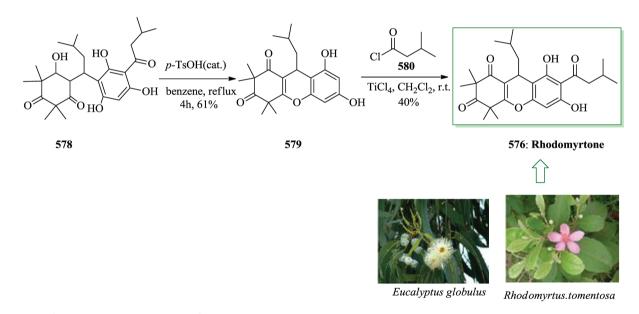
(+)-R-Mimosifoliol 571, a neoflavonoid isolated from the rootwood of Aeschynomene mimosifolia, demonstrated weak activity in a DNA-strand scission assay.546 Racemic (±)-mimosifoliol was obtained in 5 steps through an o-quinone methide

intermediate and subsequently (+)-mimosifoliol was obtained through cycloaddition of the same intermediate with a chiral enol ether.547 A two-step method was developed to vinylate the diarylmethanes at the bridging  $CH_2$  with lateral lithiation and formylation followed by a Wittig reaction. This strategy was used in the racemic synthesis of the natural product mimosifoliol. Completion of the total synthesis of  $(\pm)$ -mimosifoliol needed a protected hydroxyl substituent. For this purpose, compound 572 was protected with a tert-butyldimethylsilyl group to afford 573. The latter was acylated to provide ketone 574. Careful control of the temperature of the FC reaction was needed to prevent accidental deprotection. Finally, compound 574 afforded  $(\pm)$ -mimosifoliol 571 after several steps (Scheme 122).548

The flowering plant Rhodomyrtus tomentosa (Aiton) Hassk. of the group Myrtaceae is applied in Thailand to treat a wide range



Scheme 123 Total synthesis of rhodomyrtosone B 575.



Scheme 124 Synthesis of rhodomyrtone 576

of ailments. Rhodomyrtone **576** was isolated from the ethanol extract. The same compound was extracted from the bark of small twigs of *Eucalyptus globulus* Labill, which is a member of the *Myrtaceae* group.<sup>549,550</sup> This acylphloroglucinol exhibits significant antibacterial activity. Additional acylphloroglucinols could be isolated from the leaves of *R. tomentosa*, which contain the isomer rhodomyrtosone B **575**.<sup>551</sup> In 2013, Maier *et al.* developed a concise synthesis of the isomeric acylphloroglucinols rhodomyrtosone B **575** and rhodomyrtone **576**.<sup>552</sup> In this pathway, for the synthesis of **575**, phloroglucinol **496** was converted into the cross-conjugated enedione **577** *via* several steps. Since enone **577** is prone to isomerization,<sup>553</sup> it was

instantaneously treated with the enolate made *via* the reaction of acylphlorglucinol **504** with sodium hydride. In this way, a reasonable yield of the key intermediate **578** could be obtained. Refluxing a benzene solution of hydroxy ketone **578** in the presence of catalytic quantities of *p*-toluenesulfonic acid resulted in rhodomyrtosone B **575** in 69% yield (Scheme 123).<sup>552</sup>

The other isomer, the antibiotic rhodomyrtone **576**, was obtained from **578** through a sequence of acid-catalyzed cyclization, retro FC reaction, and reacylation (Scheme 124).<sup>552</sup>

8,5'-Neolignans having an 8-aryl-2,3-dihydrobenzofuran framework are the most plentiful natural products known in various groups of plants. These dihydrobenzofuran neolignans

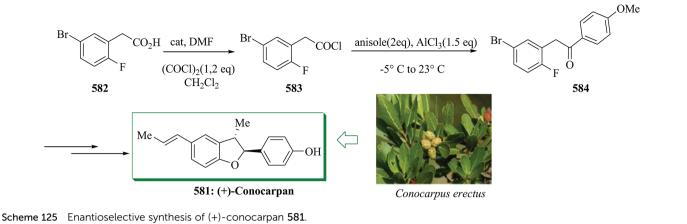


exhibit a wide range of biological activities, including cytotoxic, antiviral, and antifungal.<sup>554,555</sup> (+)-Conocarpan 581, isolated from the wood of Conocarpus erectus by Hayashi and Thomson in 1975,556 showed a range of biological activities, including insecticidal, antifungal, anti-inflammatory, and antitrypanosomal.<sup>557–559</sup> The enantioselective synthesis of natural (+)-conocarpan was reported in 2013 by Chen et al.560 The highlights of the synthesis are the asymmetric hydrogenation of prochiral ketones and intramolecular ring closure. Total synthesis of (+)-conocarpan 581 was started from 2'-fluoro-5'bromophenyl acetic acid 582. The acid was transformed into acid chloride 583 followed by FC reaction with a slight excess of anisole to provide ketone 584 in 92% yields over two steps. Subsequently, compound 584 provided (+)-conocarpan 581 in 84% yield (Scheme 125).560

Marmycin A, a unique angucycline analogue, was isolated by Fenical *et al.* in 2007 from the culture broth of a marine sediment-obtained actinomycete related to the genus *Streptomyces.* FC acylation and Dess–Martin oxidation were considered as the key steps in the synthesis of marmycin A **585**. In this route, firstly anthracen-1-amine **589** and glycal **588** gave the C1' *N*-glycosidation product **590**. Next, *N*-trifluoroacetylation of **590** using TFAA/triethylamine followed by O-deprotection with ammonium resulted in a FC acylation product **591b** in 70% overall yield, other than the desired *N*-Boc product **591a**. Finally, compound **591b** provided compound **585** in several steps (Scheme 126).<sup>561</sup>

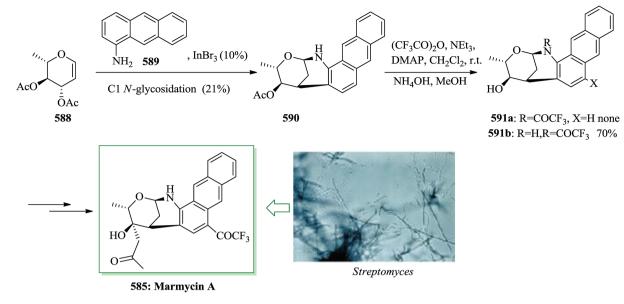
To confirm the FC acylation using  $(CF_3CO)_2O$  and the nucleophilic addition occurring on the carbonyl group, diastereomeric *N*-glycosidation products **592** and **593** were reacted with  $(CF_3CO)_2O$ /triethylamine/DMAP and then ammonium to afford FC acylation products **594** and **595**, respectively, in 70% overall yield. Quaternary alcohols **586** and **587** were obtained from **594** and **595**, respectively, in 55% overall yield after several steps (Scheme 127).<sup>561</sup>

Marinamide **596** and its methyl ester **602** are two isoquinolinone alkaloids that were isolated from the metabolite of mixed fermentation of two mangrove endophytic fungi (strains no. 1924 and 3893) from the South China Sea. These two compounds were identified as 4-(2-pyrrolyl)-1-isoquinolone-3carboxylic acid **596** and methyl 4-(2-pyrrolyl)-1-isoquinolone-3-

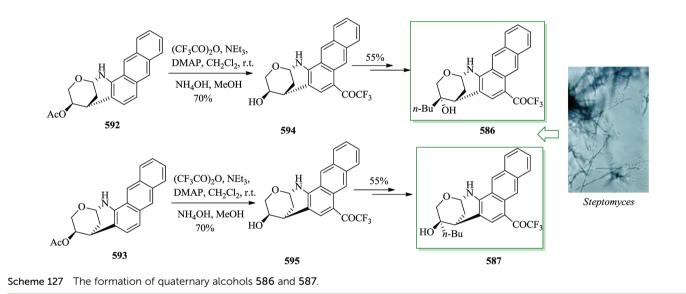
carboxylate 602, respectively. Compounds 596 and 602 exhibited significant antibacterial and562 antitumor properties. The first total synthesis of isoquinolinone alkaloid marinamide 596 and its methyl ester 602 was developed in 2013 by Ji et al. 563 The key steps included a regioselective FC reaction of 1-benzyl-1Hpyrrole to generate the intermediate 601. The synthesis of intermediate 601 is the key step for the formation of marinamide and its methyl ester, which included a regioselective FC acylation of 1-benzyl-1H-pyrrole to generate a 2-acylpyrrole derivative. Phthalic anhydride 597 afforded the acyl chloride 598 after several steps. Compound 598 reacted with 599, which was obtained by reacting pyrrole 600 with benzyl bromide in anhydrous dimethyl sulfoxide, via the catalysis of zinc powder in toluene at ambient temperature to effectively give the key intermediate 601 in 68% yield.564 Compound 601 gave the methyl ester of marinamide 602.565 Finally, hydrolysis of 602 with hydrochloric acid gave marinamide 596 (Scheme 128).563

The naphtho[2,3-c]furandiones (iso furanonaphthoquinones), a relatively small group of secondary metabolites with various biological properties, have been isolated from fungal, botanical, bacterial and insect sources. Monosporascone 603 was isolated from the fungus Gelasinospora pseudoreticulata.<sup>566</sup> Monosporascone is the only known isofuranonaphthoquinone with oxygenation at the 5 and 7 positions. The first total synthesis of the natural isofuranonaphthoquinone monosporascone 603 was exhibited in 2014 by Piggott et al.567 The five-step synthesis established involves a silver acetylide-acid chloride coupling, domino Diels-Alder-retro-Diels-Alder reaction, and an intramolecular FC acylation reaction. Total synthesis of the natural product monosporascone 603 was accomplished in 57% yield overall. The synthesis of monosporascone 603 was started from the reaction of silver acetylide 605<sup>568</sup> and acid chloride 604,<sup>569</sup> which after four steps gave acid chloride 606. Reaction of acid chloride 606 with five equivalents of aluminium chloride,<sup>570</sup> with an extended reaction period to permit selective demethylation of the perimethoxy group, gave monosporascone 603 in satisfactory yield (Scheme 129).571

In 2006, Merck researchers reported the discovery of a unique and active antibiotic, (–)-platensimycin **607**, isolated from *Streptomyces platensis* strain MA7327, which originated



Scheme 126 Total synthesis of marmycin A 585

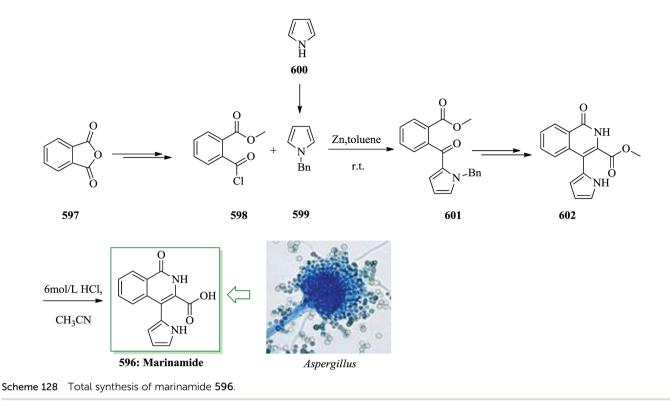


from South Africa.<sup>572</sup> Lear et al. developed its total synthesis via a four-step construction of the aromatic amine segment and an improved stereocontrolled assembly of the ketolide segment, (-)-platensic acid. Key synthetic advances contain a modified Lieben haloform reaction, a sterically controlled chemo- and diastereo-selective organocatalytic conjugate reduction and a bismuth(m)-mediated FC cyclization. The longest linear sequence is 21 steps with an overall yield of 3.8% from eugenol. In this approach, firstly, compound 608 was converted into lactol 609 via several steps. A practical limitation of this method is the excessive requirement for very toxic and ecologically harmful tin(IV)chloride to drive the FC arylation. Hence, Bi(OTf)<sub>3</sub> was found to be the most superior in reactivity and gave 610 in 94% yield. Subsequently, compound 610 was converted into the tetracyclic dienone 613 after several steps. Then, the Lewis acid-catalyzed cyclization of the cis-tosyl lactol 611, which

could be accessed directly from 608, was explored. The reaction of 611 with tin(*w*) chloride needed a greater excess (8–10 equiv.) to achieve a high-yielding FC transformation. Inspired by Bartoli, Sambri, and co-workers,573 they eventually turned to lithium perchlorate as a cocatalyst to Bi(OTf)<sub>3</sub> to help drive the FC cyclization of 611. The best catalytic combination of 5 mol% Bi(OTf)<sub>3</sub> with 3 equivalents of lithium perchlorate eventually provided 612 in 94% yield. Subsequently, compound 612 was converted into 613 via several steps and then compound 613 was transformed into 614 via several more steps (Scheme 130).574

A unique synthesis of the aromatic section of platensimycin **607** was accomplished. This synthesis was started with FC acylation of **615** with acetic acid in hot PPA to provide the acetophenone **616**. Next, compound **616** was converted into compound **617** *via* several steps (Scheme 131).<sup>574</sup>

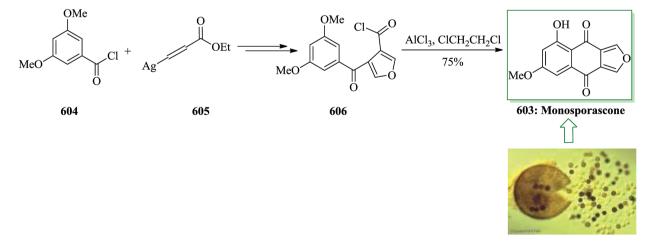
8



Finally, platensic acid **614** and the aniline core **617** afforded (–)-platensimycin **607** in 60% yield upon several steps (Scheme 132).<sup>574</sup>

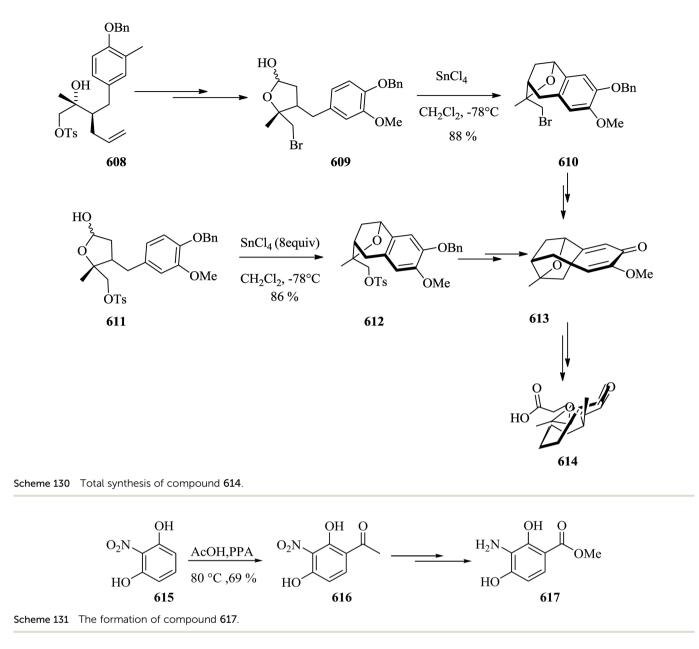
The natural product 2,4,6-trihydroxy-3-geranylacetophenone (tHGA) isolated from the medicinal plant *Melicope ptelefolia* was demonstrated to show lipoxygenase (LOX) inhibitory activity. It is known that LOX plays a significant role in inflammatory response as it catalyzes the oxidation of unsaturated fatty acids.<sup>575,576</sup> Shaari *et al.* in 2014 reported synthesis of tHGA analogues through simple FC acylation and alkylation reactions.<sup>577</sup> In this method, they reported the formation of 3-geranyl-1-(2'-methylpropanoyl)phloroglucinol **618**, a natural product known in *Hypericum empetrifolium*. Direct FC acylation of phloroglucinol **496** was achieved with isobutyryl chloride as the acylating agent using anhydrous AlCl<sub>3</sub> to provide compound **619**.<sup>578</sup> Subsequently, the geranyl scaffold was introduced through electrophilic substitution of geranyl bromide with anhydrous  $K_2CO_3$  as the base in dry MeOH under reflux to afford the FC alkylation product **618** in satisfactory yield (19.5%) (Scheme 133).<sup>577</sup>

The merochlorins, a structurally uncommon group of four chlorinated meroterpenoids, have potent antimicrobial activity and were isolated from a marine strain of *Streptomyces* bacteria.<sup>579</sup> Merochlorin A **620** contains four contiguous stereocenters embedded in a bicyclo[3.2.1]octanone scaffold. The complex, polycyclic structures of merochlorins A, obtained from



Monosporascus cannonballus

Scheme 129 Total synthesis of monosporascone 603.

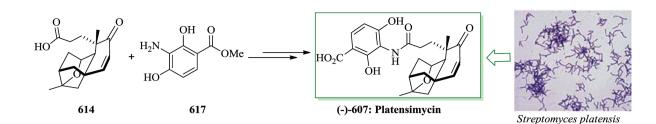


a unique terpene side chain, have aroused the interest of chemists.  $^{\tt 580, 581}$ 

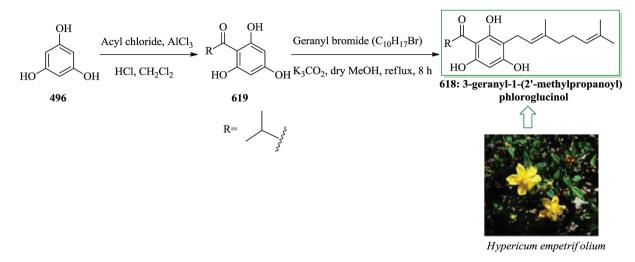
Total synthesis of merochlorin A was exhibited alongside the biosynthetic studies of George *et al.* in 2015.<sup>582</sup> Synthesis of merochlorin A **620** was started with a FC reaction of methyl 3,5-dimethoxyphenylacetate **621** and chloroacetyl chloride to afford ketone **623** in 56% yield. Finally, merochlorin A **620** was obtained from compound **623** in 42% yield *via* several steps (Scheme 134).<sup>582</sup>

Curvulone B **624**, a *cis*-di functionalized tetrahydropyran natural product, is one of a wide range of relevant compounds isolated from the marine fungus *Curvularia* sp., which was isolated from the marine alga *Gracilaria folifera*.<sup>583</sup> A total synthesis of curvulone B was achieved using a FC reaction and a highly *cis*-selective intramolecular oxa-Michael addition. This synthetic method was accomplished in ten steps and 39% overall yield. Total synthesis of Curvulone B **624** was started from 3,5-dihydroxybenzoic acid **625**. Next, compound **625** afforded ester **626** upon several steps. The FC acylation reaction of ester **626** was performed with whole regioselectivity with a 2-chlorobenzyl protecting group to afford ketone **628** in 89% yields. Finally, ketone **628** afforded the target natural product **624** *via* several steps (Scheme 135).<sup>584</sup>

Triumphalone **629** and isotriumphalone **630**, highly oxidized monomeric phloroglucinols<sup>585,586</sup> with two stereogenic centers, were extracted from *Melaleuca triumphalis* by Brophy *et al.*<sup>587</sup> The first total synthesis of  $(\pm)$ -triumphalone was accomplished in 8 steps from phloroglucinol. Synthetic triumphalone was converted into  $(\pm)$ -isotriumphalone in one step. In this route, firstly phloroglucinol **496** was converted into **632** through the FC acylation reaction in 82% yield.<sup>588</sup> Subsequently,  $(\pm)$ -triumphalone **629** was provided from compound **632** via several steps and also compound **629** was transformed to  $(\pm)$ -isotriumphalone **630** in one step. An spontaneous change of **629** to



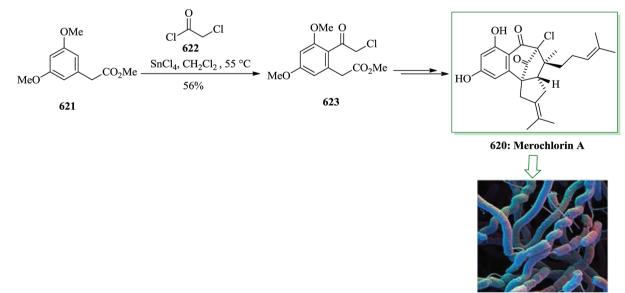
Scheme 132 Total synthesis of (-)-platensimycin 607.



Scheme 133 Total synthesis of 3-geranyl-1-(2'-methylpropanoyl)phloroglucinol 618.

630 was detected. The structure change gradually occurred to afford a 3:7 mixture of 629 and 630 upon standing for 24 months (Scheme 136).589

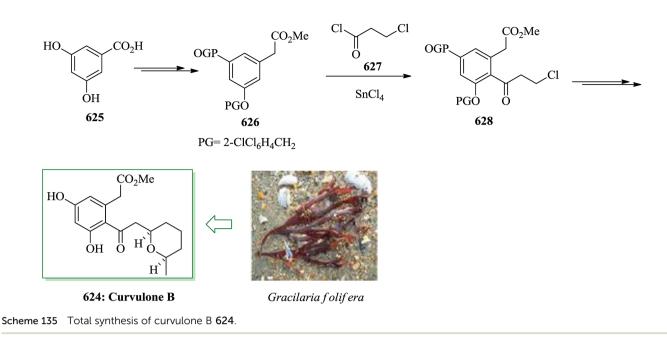
Flavonoids, a group of polyphenolic compounds that are abundant in plants,<sup>590</sup> include dietary components of potential importance to health.591 In addition, flavanoids are emerging as a potentially significant unique group of pharmaceutical lead substrates.590 Among the several classes of flavonoids, dihydrochalcones exhibit a series of biologically remarkable activiincluding antioxidant,<sup>592,593</sup> antiinflammatory,594 ties, antileishmanial,595 antidiabetic,596 anticancer597 and molluscidal<sup>598</sup> properties.<sup>599</sup> Peng et al. described the isolation and



Streptomyces bacteria

Scheme 134 Total synthesis of merochlorin A 620.

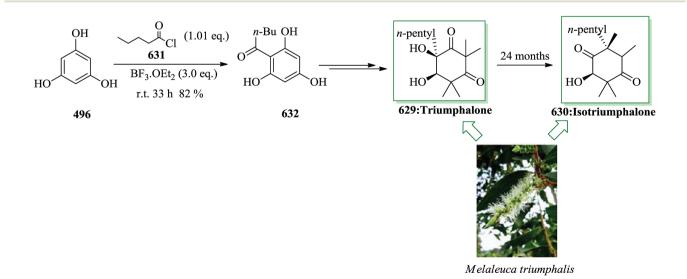
0N-78



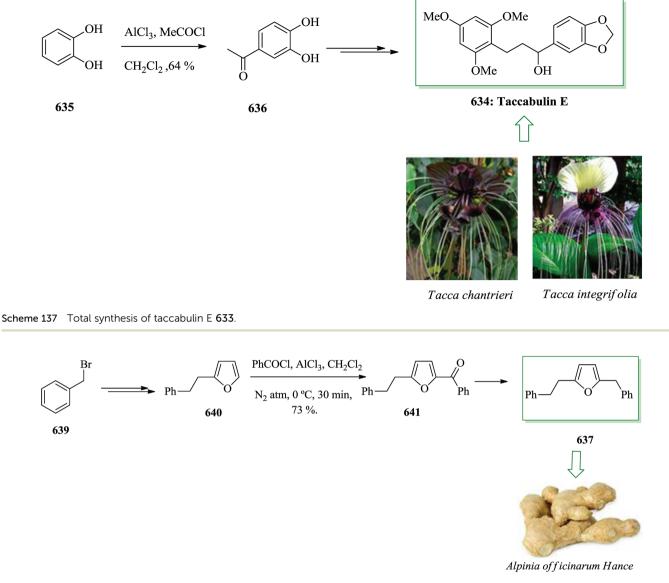
identification of a significant dihydrochalcone, named taccabulin E **633**, from extracts of the plant species *Tacca chantrieri* and *Tacca integrifolia*.<sup>600</sup> Spring *et al.* in 2015 reported the concise and divergent total synthesis of dihydrochalcone **633**,<sup>601</sup> which was started from catechol **634**. Next, FC acylation reaction of **634** permitted access to **635** in moderate yield.<sup>602,603</sup> Finally, compound **635** was converted into taccabulin E **633** in 94% yield upon several steps (Scheme 137).<sup>601</sup>

Diarylheptanoids, with two aryl groups at C(1) and C(7) of a  $C_7$  chain, are a significant group of natural products isolated from nature.<sup>604–608</sup> Generally, diarylheptanoids are categorized into three groups comprising linear diarylheptanoids, macrocyclic biarylheptanoids, and macrocyclic diaryl ether heptanoids. Sun *et al.*<sup>609</sup> isolated two furan-cyclized diarylheptanoids, **637** and **638**, from rhizomes of *Alpinia officinarum Hance* (Zingiberaceae) as minor constituents and they were demonstrated to have satisfactory cytotoxicity against the IMR-32 human neuroblastoma cell line. Seçen *et al.* in 2015 reported a synthetic strategy for the formation of two natural diarylheptanoids, 2-benzyl-5-(2-phenylethyl)furan **637** and 2methoxy-4-{[5-(2-phenyl-ethyl)furan-2-yl]methyl}phenol **638**.<sup>610</sup> Total synthesis of the natural product **637** was started from benzyl bromide **639**, which afforded 2-(2-phenylethyl)furan **640** *via* several steps. FC acylation of compound **640** with benzoyl chloride using aluminium chloride afforded ketone **641** in 73% yield. Finally, the reduction of **641** afforded **637** in 81% yield (Scheme 138).<sup>610</sup>

Then, this group used 2-(2-phenylethyl)furan **640** as the initiating precursor to form natural product **638**. In this route, compound **640** was reacted with 4-(acetyloxy)-3-methoxybenzoyl chloride **642** *via* FC reaction to afford ketone **643** in a satisfactory yield (59%). Subsequently, natural product **642** was



Scheme 136 Total synthesis of triumphalone 629 and isotriumphalone 630.

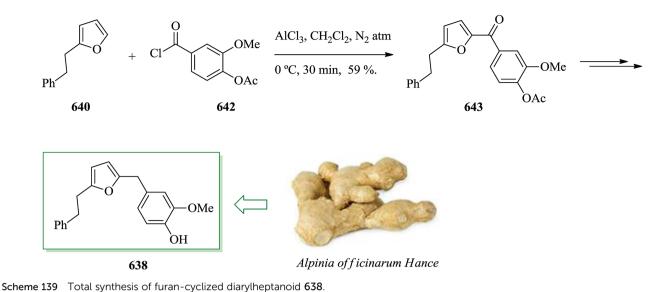


Scheme 138 Total synthesis of furan-cyclized diarylheptanoid 637.

synthesized from compound 643 upon several steps (83%) (Scheme 139).<sup>610</sup>

Stigmatellin A and B, two unique antibiotics, were isolated from the gliding bacterium *Stigmatella aurantiaca*.<sup>611</sup> They are powerful inhibitors of electron transport in chloroplasts and mitochondria. The absolute configuration of Stigmatellin A was confirmed as (*S*,*S*,*S*,*S*) by chemical correlation.<sup>612</sup> A significant and enantioselective method for the formal total synthesis of stigmatellin A was developed in 2017 by Yadav *et al*.<sup>613</sup> The key steps included in this synthesis are desymmetrization, FC acylation reaction, regioselective demethylation, Baker–Venkataraman rearrangement and Grubbs cross-metathesis. Total synthesis of stigmatellin A was started from tetramethoxy benzene **645**. The FC acylation reaction of tetramethoxy benzene **645** with propanoyl chloride using aluminium chloride gave the corresponding ketone **646**.<sup>614</sup> Finally, compound **646** gave stigmatellin A **644** after several steps (Scheme 140).<sup>613</sup>

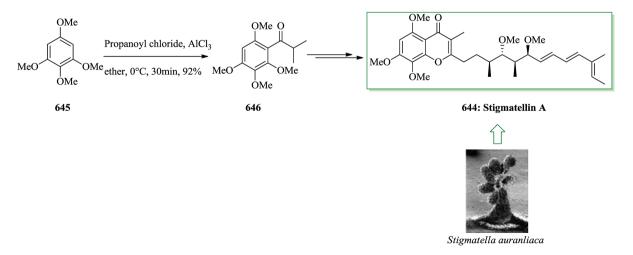
γ-Lycorane is an alkaloid isolated from the plants of the Amaryllidacae group.615 Various methods for the synthesis of Ylycorane 647 have been reported, despite, as several authors have pointed out, its apparent lack of useful pharmacological activities.616,617 A total synthesis of Y-lycorane was accomplished using N-tosylpyrrole as a key framework. The synthesis uses both an intermolecular and an intramolecular FC reaction, and also an entirely diastereoselective hydrogenation of a late-step pyrrole intermediate. For the synthesis of  $\gamma$ -lycorane 647, Ntosyl pyrrole 648 was subjected to an FC acylation reaction with succinic anhydride. The resulting acid 650 afforded the corresponding alcohol 651 via several steps. The reaction between alcohol 651 and amberlyst-15 easily afforded the corresponding FC product 652 in 82% yield by construction of the novel C-C bond at the  $\alpha$ -position. Finally, compound 652 gave  $\gamma$ -lycorane 647 via several steps (Scheme 141).618



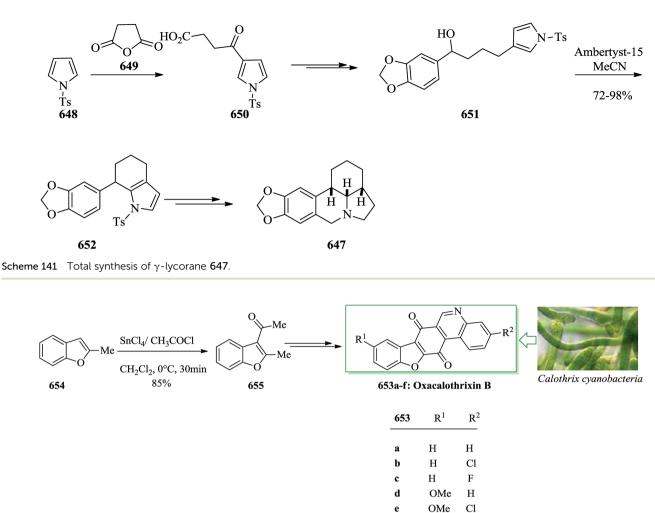
Calothrixin A and B are unique indolo[3,2-*j*]phenanthridine alkaloids that were isolated from *Calothrix* cyanobacteria in 1999.<sup>619</sup> These natural products show significant biological activities, for example antimalarial and anticancer, and they inhibit bacterial RNA polymerase.<sup>620</sup> The total synthesis of oxacalothrixin B, an isostere of the biologically significant carbazoloquinone alkaloid calothrixin B, was accomplished from 2-acetyl-3-methylbenzofuran. In this route, initially FC acylation reaction of 2-methylbenzofuran **654**<sup>621</sup> using CH<sub>3</sub>COCl with tin(IV) chloride in dry dichloromethane at 0 °C gave 3-acetyl-2methylbenzo[*b*]furan **655** in moderate yield. Next, compound **655** afforded oxacalothrixin B **653a–f** in moderate yields after several steps (Scheme 142).<sup>622</sup>

Coumarins constitute a significant group of heterocyclic compounds that are known as benzo- $\alpha$ -pyrones, wherein a pyran ring is fused with a benzene ring. Two unique gerany-lated coumarins, mammeasins C and D (**656** and **657**), were isolated together with 20 other coumarins.<sup>623</sup> Compounds **656** and **657** are rare coumarins, wherein a dioxaphenalene type scaffold is generated by attaching a pyran ring to the coumarin

unit. Both these compounds exhibit active aromatase inhibitory activity comparable to that of aminoglutethimide, which was applied as a reference standard.<sup>623</sup> The first total synthesis of the geranylated pyranocoumarins, mameasins C 656 and D 657, aromatase inhibitors isolated from the flowers of Mammea siamensis, was achieved in five steps, starting from phloroglucinol 496. Based on this method, total synthesis of 656 and 657 was started from 496, in which 496 was exposed to FC acylation with crotonyl chloride. This reaction gave the corresponding chromanone 5,7-dihydroxy-2-methylchroman-4-one 659, albeit in a low yield of 31%. Next compound 659 provided pyranocoumarin 660 in 69% yield. FC acylation reaction of 660 with butyryl chloride or isobutyryl chloride was applied for the acylation of phenolic compounds,506 and two regioisomers (663/ 664) in 4:1 and 5:1 ratio were provided. Finally, compounds 663a and 663b afforded the corresponding 656 and 657 in 23% and 15% yield after 2 steps, respectively. In addition, examining of the intermediates obtained in the synthetic route to 656 and 657 revealed that de-geranylated pyranocoumarins 663 and 664



Scheme 140 Total synthesis of stigmatellin A 644.



Scheme 142 Total synthesis of oxacalothrixin B 653a-f.

exhibit superior aromatase inhibitory activity as compared to the natural products **656** and **657** (Scheme 143).<sup>624</sup>

Isoflavone C-glycosides, in which the sugar scaffold is attached by a carbon-carbon bond directly to the isoflavone ring, are not simply hydrolyzed in acidic gastric juices, unlike Oglycosides and aglycone. These glycosides exhibit several biological properties, including radioprotective,625 anti-myocardial ischemic,626 mitogenic and colony-stimulating,627 and antidiabetic628 activities. Among these compounds, puerarin, which is known mainly in Pueraria radix, exhibits a strong antimyocardial ischemic influence.629 Two isoflavone C-glycosides (6-tert-butyl puerarin and 6-tert-butyl-4'-methoxypuerarin) were obtained using FC acetylation reaction and Vilsmeier-Haack cyclization in five steps with overall yields of 14.6% and 14.2%, respectively. Initially, C-glucosyl acetophenone 666 was converted into 2 *C*-β-D-glucopyranoside **667** after several steps. The key intermediate deoxybenzoin 669 was obtained from FC acetylation reaction of 667 using anhydrous AlCl<sub>3</sub> with a 63.6% yield.630,631 Lastly, compound 669 gave 6-tert-butyl-4'-

methoxypuerarin **665a** and 6-*tert*-butyl puerarin **665b** with 95.5% and 98.6% yields, <sup>632,633</sup> respectively (Scheme 144).<sup>634</sup>

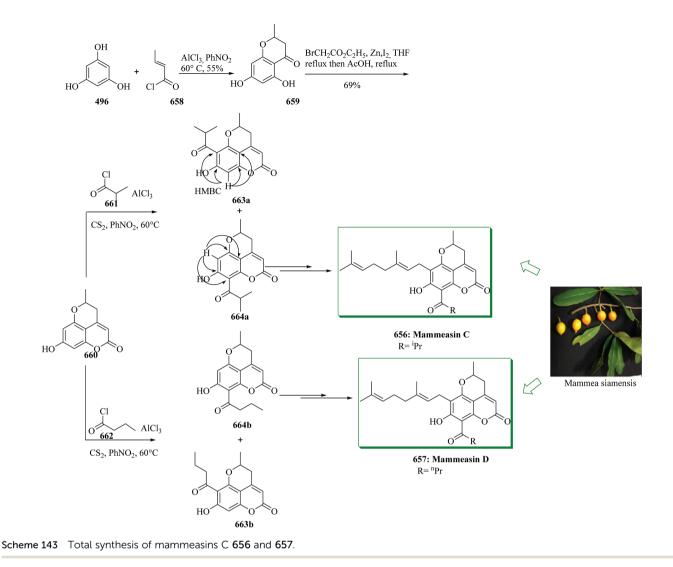
# 2.5. Miscellaneous

f

OMe

F

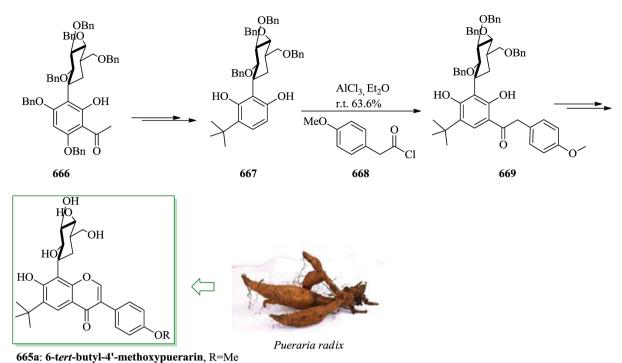
With the increase in the advanced age population, neurodegenerative disorders, for example Alzheimer's and Parkinson's disease, are emerging as a major social issue, therefore resulting in great demand for new therapeutic drugs to prevent these diseases.635 Significant efforts at searching for small-moleculebased natural products with neurotrophic activities resulted in the isolation of (-)-talaumidin 670 from Brazilian Aristolochia arcuata Masters.636 Talaumidin and its analogues show important neurite outgrowth-promoting and neuroprotective activities in the primary cultured rat corticaland additionally in the hippocampal neurons.636 The initial asymmetric total synthesis of neurotrophic (-)-talaumidin 670 was developed in 16 steps from 4-benzyloxy-3-methoxybenzaldehyde in ca. 10.7% overall yield. In this method, the key steps are Evans asymmetric anti-aldol reaction, hydroboration/oxidation and epimerization and FC arylation reaction. Total synthesis of



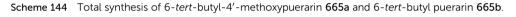
(–)-talaumidin **670** was started from the reaction between 4benzyloxy-3-meth-oxybenzaldehyde **671** and (*S*)-4-benzyl-3propionyl-2-oxazolidinone **672** to provide five-membered acetal **673** *via* several steps. Then, FC-type arylation reaction of **673** with 1,2-methylenedioxybenzene **421** using tin(rv) chloride in dichloromethane afforded only the corresponding (5*S*)-**675** in 89% yield along with 2% of talaumidin **670**. Finally, debenzylation of **675** with Pd(OH)<sub>2</sub> in ethanol provided (–)-(2*S*,3*S*,4*S*,5*S*)-**670** in 77% yield (Scheme 145).<sup>637</sup>

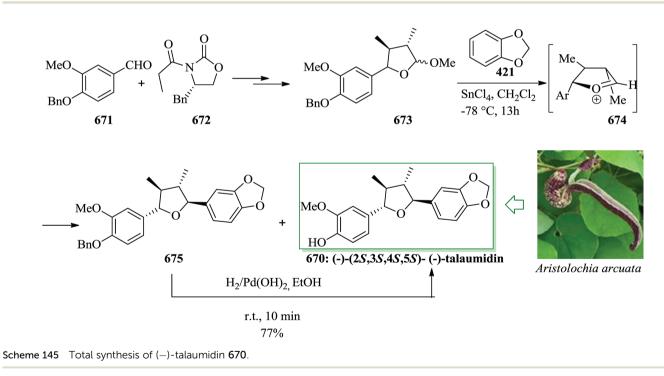
An oxidative FC reaction containing various aromatic compounds catalyzed by a hypervalent iodine reagent was accomplished by employing polyfunctionalized phenols. This reaction provided rapid access to extremely functionalized compounds, including a dienone, a quaternary carbon center, and an aromatic ring. The product's framework is found in various naturally occurring compounds. Based on this method, total synthesis of compounds belonging to the Amaryllidaceae alkaloids group, for example *O*-methyljoubertiamine, mesembrine, and its natural derivative the dihydro-*O*-methylsceletenone, was accomplished in eight/nine steps. The synthetic pathway to these molecules features a significant transformation on the basis of a Fukuyama and Michael-retro-Michael tandem method. The synthesis of **679** was started from 2,4,6-trimethylphenol **676**, which afforded compound **677** *via* several steps. Subsequently, reaction of compound **678a** or **678b** with TBAF leads in satisfactory yield to the bicyclic compound **679** with complete regioselectivity regarding the alkene, including the silyl substituent, the latter being subsequently removed under the reaction conditions. This oxidative FC reaction permits quick access to substituted synthons, including a quaternary carbon center, a dienone, and an aromatic scaffold, and various functionalities can be present on the side chain. Such an intermediate could provide some opportunity for total synthesis of different naturally occurring compounds. The most well-known belong to the group of *Amaryllidaceae* alkaloids (Scheme 146).<sup>638</sup>

Based on this method, two alkaloids of this group were synthesized. Mesembrine **680a** and *O*-methyljoubertiamine **681** are alkaloids present in *Sceletium tortuosum*. They were demonstrated to be very active serotonin reuptake inhibitors, potent at very low doses. The 4,5-dihydro-4'-O-methyl-sceletenone **680b** is a simpler natural derivative of mesembrine



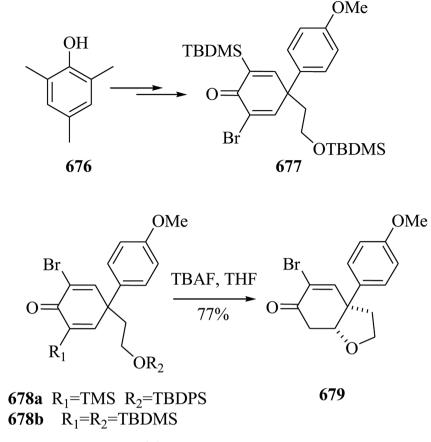
665b: 6-*tert*-butyl puerarin, R=H





extracted from *Aptenia cordifolia*. Starting from 2-(4-hydroxyphenyl)ethanol **682**, total synthesis of mesembrine and 4,5dihydro-4'-*O*-methylsceletenone was achieved *via* several steps in 86% yield. Furthermore, the transformation of 4,5-dihydro-4'-*O*-methylsceletenone, **680b**, into *O*-methyljoubertiamine **681** was accomplished by reaction with iodomethane (Scheme 147).<sup>638</sup> (2)-Talaumidin **683** and (2)-galbelgin **684**, naturally occurring lignans, were isolated from *Aristolochia arcuata*<sup>636</sup> and *Piper futokadsura*. <sup>639</sup> They are 2,5-diaryl-3,4-dimethyl-tetrahydrofuran lignans. <sup>640</sup> Among them, (2)-talaumidin **683** demonstrated important neurotrophic activity in the primary culture of rat cortical neurons and could serve as a promising lead compound for the treatment of neurodegenerative disorders, for example

8



Scheme 146 The formation of the bicyclic compound 679.

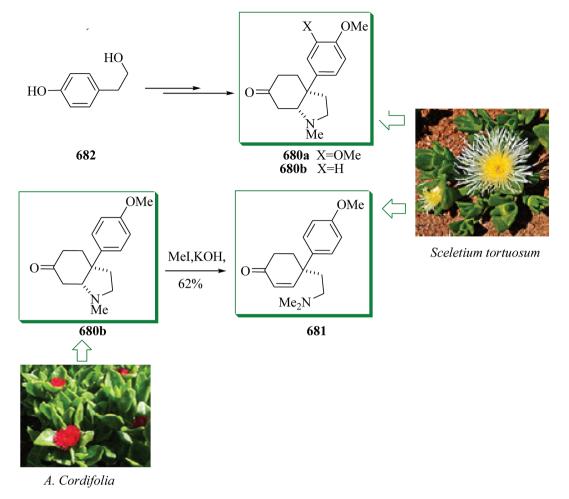
Alzheimer's and Parkinson's disease,641 whereas (2)-galbelgin 684 exhibits anti-HBV activity.<sup>642</sup> (2)-Talaumidin 683 and (2)galbelgin 684 were obtained starting from 4-pentenoic acid with an overall yield of about 17.8 and 16.9%, respectively. The key steps contain an Evans asymmetrical anti-aldol reaction, TBS protection, hydroboration, oxidation, and FC arylation. Firstly, 4-pentenoic was converted into 686 via several steps. Next, the methyl acetal 686 was transformed into 2,5-diaryl-3,4dimethyltetrahydrofuran via FC-type arylation reaction using tin(IV) chloride in dichloromethane.<sup>637,643</sup> The FC-type arylation reaction was performed with the epimerization at the C2 position of 686 to give 2,3-trans-3,4-trans-4,5-trans-tetrahydrofuran 687a and 687b as a single diastereomer. Lastly, removal of the Bn protecting group in 687a afforded (2)-talaumidin 683 (90%). Furthermore, the total synthesis of (2)-galbelgin 684 was achieved in two steps from the intermediate 687b (Scheme 148).644

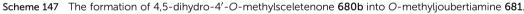
Extracts from the plants *Pycnanthus angolensis* and *Holostylis reniformis* were applied to treat malaria throughout Africa and Brazil, respectively.<sup>645</sup> Extracted lignans from these and other plants,<sup>250</sup> such as (–)-8'*-epi*-aristoligone **689**, have demonstrated promising antiplasmodial activity against a chloroquine-resistant strain of *Plasmodium falciparum*.<sup>250</sup> The development of a unique one-pot oxidative [3,3] rearrangement/FC arylation reaction permitted the fast and stereocontrolled synthesis of several tetralone- and naphthyl-type lignan natural products with antimalarial activity. For the synthesis of **689** and **688**,

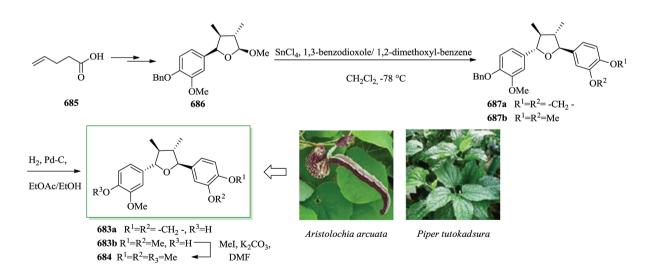
initially hydrazine **691** serves as a linchpin for the reaction of aryl aldehyde **690** with arene **693** *via* initial construction of hydrazone **692**, which is followed by a hypervalent-iodide oxidative [3,3] rearrangement/FC arylation in one-pot fashion to provide benzhydryl derivative **694**. Next, *via* several steps, the one-pot oxidative [3,3] rearrangement/FC arylation of **695** with 1,2-dimethoxybenzene **36** progressed with complete chirality transfer to provide benzhydryl **696** in 77% yield. Subsequently, compound **696** was transformed into (-)-8'*-epi*-aristoligone **689** in 28% yield and (-)-cyclogalgravin **688** in 73% yield *via* several steps (Scheme 149).<sup>646</sup>

Based on this method, the total synthesis of **697** and **698** was achieved as follows. Construction of hydrazone **699** from piperonal followed *via* oxidative [3,3] rearrangement and FC arylation reaction with 1,2-dimethoxybenzene **36** resulted in the formation of diastereomeric benzhydryl **700** (80% yield, 5 : 1 d.r.). Next, the final two natural products, (-)-8'*-epi*-aristotetralone **698** and (-)-pycnanthulignene B **697**, were synthesized in 43% and 70% yield, respectively, from compound **700** *via* several steps (Scheme 150).<sup>646</sup>

Santalin Y occurs in "red sandalwood", the hardwood of *Pterocarpus santalinus*. The yellow pigment santalin Y was isolated by Nohara *et al.* in 1995 as a minor component.<sup>647</sup> A biomimetic total synthesis of santalin Y, a structurally complex but racemic natural product, was developed in 2015 by Strych *et al.* Santalin Y was obtained *via* seven steps (longest linear





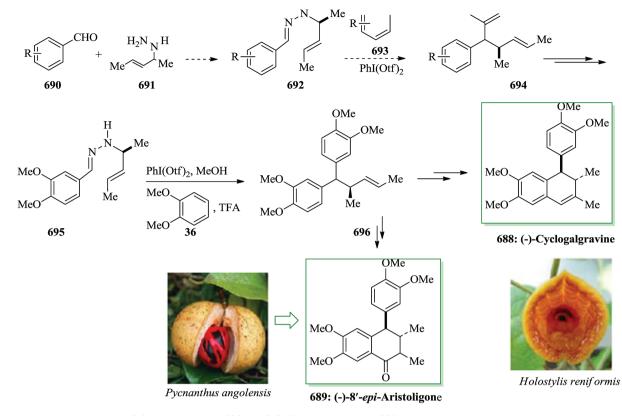


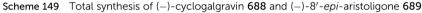
Scheme 148 Total synthesis of (2)-talaumidin 683 and (2)-galbelgin 684.

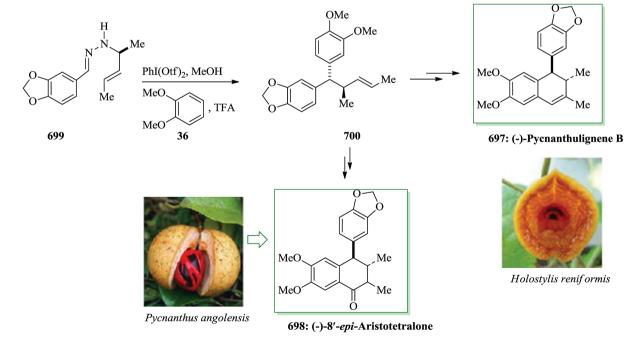
sequence) and in 8% overall yield. The key steps were a (3 + 2) cycloaddition reaction and an intramolecular FC reaction. The total synthesis of santalin Y was started from isoflavylium ion **702**, which was converted into a tetrahydrofuran intermediate

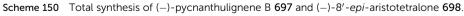
**703** *via* several steps. Next, a subsequent intramolecular FC cyclization provided santalin Y (Scheme 151).<sup>648</sup>

Epipolythiodiketopiperazine alkaloids represent a structurally complex and biologically potent group of secondary fungal



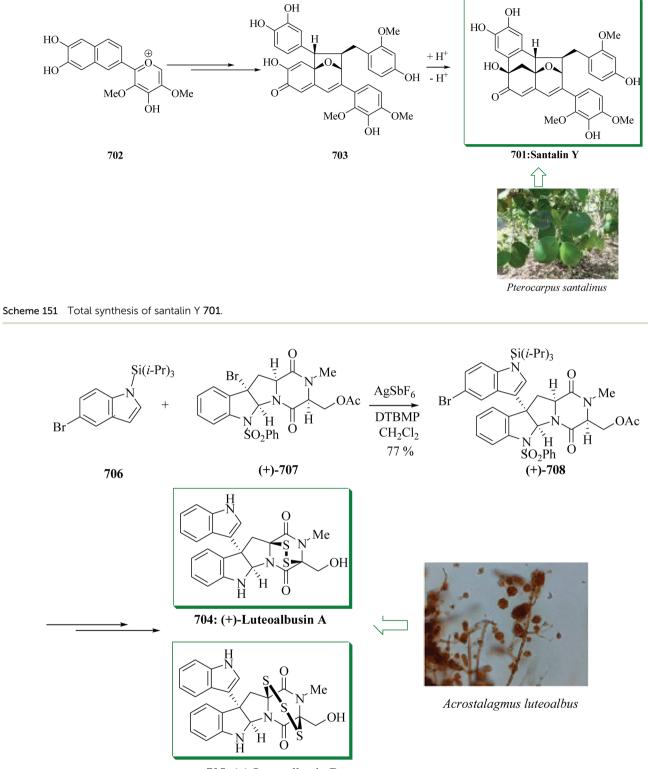






metabolites.<sup>162,268,649</sup> Several members of this group of natural products share a cyclo-tryptophan unit and an eponymous epipolythiodiketopiperazine (ETP) substructure. Among them, (+)-luteoalbusin A **704** and (+)-luteoalbusin B **705** were isolated

from the marine fungi *Acrostalagmus luteoalbus* SCSIO F457. The first total synthesis of (+)-luteoalbusins A and B was developed in 2015 by Movassaghi *et al.*<sup>650</sup> The total synthesis of alkaloids (+)-**704** and (+)-**705** was started with the Ag-catalyzed

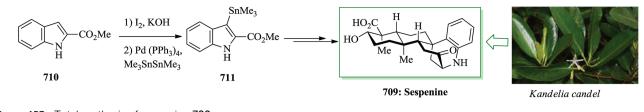


705: (+)-Luteoalbusin B

Scheme 152 Total synthesis of (+)-luteoalbusin A 704 and (+)-luteoalbusin B 705.

FC arylation of diketopiperazine (+)-**707**. Reaction of a solution of C3-bromo diketopiperazine (+)-**707** in  $CH_2Cl_2$  with indoles **706** and silver(1) hexafluoroantimonate using 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) gave the corresponding C3-

indolylhexacycle (+)-708 in 77% yield. Lastly, compound 708 produced (+)-luteoalbusins A 704 and B 705 via several steps (Scheme 152).<sup>650</sup>



Scheme 153 Total synthesis of sespenine 709.

In 2011, Ding et al. identified sespenine 709, which is a structurally uncommon polycyclic molecule isolated from Streptomyces sp. HKI0595.651 Li et al. in 2016 reported a ten-step (the longest linear sequence) synthesis of this molecule from market-accessible precursors.652 Sharpless enantioselective epoxidation, Stille-Miyata coupling reaction and Prins cyclization/FC/retro FC were considered as the key steps. Total synthesis of sespenine 709 was started from methyl indole-2carboxylate 710. Initially, the feasibility of FC-type allylation at the C3 position of 710 was investigated.<sup>653,654</sup> In this pathway, stannane 711 was synthesized from 710 through a two-step sequence described by Routier et al.655 Iodination based on basic conditions gave the desired C3 iodide, which was transformed to 711 via a methodology-mediated stannylation [Pd(PPh<sub>3</sub>)<sub>4</sub>, Me<sub>3</sub>SnSnMe<sub>3</sub>]. Afterwards, sespenine 709 was obtained after several steps (Scheme 153).652

# 3. Conclusion

In conclusion, in this review, we tried to underscore the significance and importance of the FC reaction as an old reaction with a new and interesting perspective owing to its applications in the important and new field of total synthesis of naturally occurring compounds. We showed how this old reaction can play an important and key role in the total synthesis of natural products proven to exhibit diverse biological activities. Indeed, nowadays, the FC reaction is considered as one of the most significant basic reaction classifications in the total synthesis of the most important class of natural products. The important compounds from biological points of views are totally synthesized via a pathway including at least one step involving this old but useful FC reaction as a key step. As a result, the FC reaction as an old but extremely useful reaction is frequently applied in the total synthesis of natural products, particularly those with various biological activities.

# Conflicts of interest

There are no conflicts to declare.

# References

- 1 C. Friedel and J. M. Crafts, *Compt. Rend.*, 1877, **84**, 1392–1395.
- 2 C. C. Price, Org. React., 1962, 3, 1.
- 3 J. Groves, Chem. Soc. Rev., 1972, 1, 73-97.
- 4 S. C. Eyley and F. Ince, Synlett, 1991, 4, 721-723.
- 5 H. Heaney, Compr. Org. Synth., 1991, 2, 733-752.

- 6 M. Rueping and B. J. Nachtsheim, *Beilstein J. Org. Chem.*, 2010, **6**, 1–31.
- 7 M. B. Smith and J. March, *March's advanced organic chemistry: reactions, mechanisms, and structure*, John Wiley & Sons, 2007.
- 8 J. Gershenzon and N. Dudareva, *Nat. Chem. Biol.*, 2007, 3, 408–414.
- 9 G. Samuelson, Drugs of natural origin: a textbook of pharmacognosy, Taylor & Francis Group, 1999.
- 10 J. R. Hanson, *Natural products: the secondary metabolites*, Royal Society of Chemistry, 2003.
- 11 D. Williams and T. Lemke, *Foye's Principles of Medicinal Chemistry*, Lippincott Williams Wilkins, Philadelphia, 5th edn, 2002, vol. 1, p. 25, ISBN 0-683-30737.
- 12 R. A. Maplestone, M. J. Stone and D. H. Williams, *Gene*, 1992, **115**, 151–157.
- 13 P. Hunter, EMBO Rep., 2008, 9, 838-840.
- 14 J. W.-H. Li and J. C. Vederas, Science, 2009, 325, 161-165.
- 15 N. Calloway, Chem. Rev., 1935, 17, 327-392.
- 16 G. A. Olah and G. S. Prakash, *Across Conventional Lines:* Selected Papers of George A. Olah, World scientific, 2003.
- 17 R. M. Roberts and A. A. Khalaf, *Friedel–Crafts alkylation chemistry: a century of discovery*, Marcel Dekker Inc, 1984.
- 18 T. B. Poulsen and K. A. Jørgensen, Chem. Rev., 2008, 108, 2903–2915.
- 19 N. A. Paras and D. W. MacMillan, J. Am. Chem. Soc., 2001, 123, 4370–4371.
- 20 M. M. Heravi and E. Hashemi, *Tetrahedron*, 2012, **45**, 9145–9178.
- 21 M. M. Heravi, E. Hashemi and F. Azimian, *Tetrahedron*, 2014, 1, 7–21.
- 22 M. M. Heravi and V. Zadsirjan, *Tetrahedron: Asymmetry*, 2013, 24, 1149–1188.
- 23 M. M. Heravi, E. Hashemi and N. Nazari, *Mol. Diversity*, 2014, **18**, 441–472.
- 24 M. M. Heravi and V. F. Vavsari, *RSC Adv.*, 2015, 5, 50890–50912.
- 25 M. M. Heravi, E. Hashemi and N. Ghobadi, *Curr. Org. Chem.*, 2013, **17**, 2192–2224.
- 26 M. M. Heravi, T. B. Lashaki and N. Poorahmad, *Tetrahedron: Asymmetry*, 2015, **26**, 405–495.
- 27 M. M. Heravi, V. Zadsirjan and Z. Bozorgpour Savadjani, *Curr. Org. Chem.*, 2014, **18**, 2857–2891.
- 28 M. M. Heravi and N. Nazari, *Curr. Org. Chem.*, 2015, **19**, 2358–2408.
- 29 M. M. Heravi, V. Zadsirjan and B. Farajpour, *RSC Adv.*, 2016,
   6, 30498–30551.

- Open Access Article. Published on 03 December 2018. Downloaded on 8/26/2025 9:48:33 PM. EV-NO This article is licensed under a Creative Commons Attribution-NonCommercial 3.0 Unported Licence.
- 30 M. M. Heravi, T. Ahmadi, M. Ghavidel, B. Heidari and H. Hamidi, *RSC Adv.*, 2015, 5, 101999–102075.
- 31 M. M. Heravi, V. Zadsirjan, H. Hamidi and P. H. T. Amiri, *RSC Adv.*, 2017, 7, 24470–24521.
- 32 M. M. Heravi, V. Zadsirjan, M. Esfandyari and T. B. Lashaki, *Tetrahedron: Asymmetry*, 2017, **28**, 987–1043.
- 33 M. M. Heravi, T. B. Lashaki, B. Fattahi and V. Zadsirjan, *RSC Adv.*, 2018, **8**, 6634–6659.
- 34 M. M. Heravi, S. Rohani, V. Zadsirjan and N. Zahedi, RSC Adv., 2017, 7, 52852–52887.
- 35 A. T. K. Koshvandi, M. M. Heravi and T. Momeni, *Appl.* Organomet. Chem., 2018, **32**.
- 36 R. C. Gadwood, R. M. Lett and J. E. Wissinger, *J. Am. Chem. Soc.*, 1986, **108**, 6343–6350.
- 37 F. J. Schmitz, K. H. Hollenbrek and D. J. Vanderah, *Tetrahedron*, 1978, **34**, 2719–2722.
- 38 L. A. Paquette and W. H. Ham, *J. Am. Chem. Soc.*, 1987, **109**, 3025–3036.
- 39 F. Brauns, J. Org. Chem., 1945, 10, 216-218.
- 40 I. A. Pearl, J. Org. Chem., 1945, 10, 219-221.
- 41 P. Boissin, R. Dhal and E. Brown, *Tetrahedron Lett.*, 1989, 30, 4371–4374.
- 42 D. Stoermer and C. H. Heathcock, *J. Org. Chem.*, 1993, 58, 564–568.
- 43 F. Kögl, G. Van Wessem and O. Elsbach, *Recl. Trav. Chim. Pays-Bas*, 1945, **64**, 23–29.
- 44 D. L. Boger, J. Hong, M. Hikota and M. Ishida, J. Am. Chem. Soc., 1999, 121, 2471–2477.
- 45 T. J. King, I. W. Farrell, T. G. Halsall and V. Thaller, J. Chem. Soc., Chem. Commun., 1977, 727–728.
- 46 T. Anke, W. Watson, B. Giannetti and W. Steglich, J. Antibiot., 1981, 34, 1271–1277.
- 47 H. Hoffmann and J. Rabe, *Angew. Chem., Int. Ed. Engl.*, 1985, 24, 94–110.
- 48 J. Mihelcic and K. D. Moeller, *J. Am. Chem. Soc.*, 2004, **126**, 9106–9111.
- 49 Q. Meng and M. Hesse, Helv. Chim. Acta, 1990, 73, 455-459.
- 50 C.-L. Chin, D. D.-P. Tran, K.-S. Shia and H.-J. Liu, *Synlett*, Georg Thieme Verlag, 2005.
- 51 F. Schröder, S. Franke, W. Francke, H. Baumann, M. Kaib,
  J. M. Pasteels and D. Daloze, *Tetrahedron*, 1996, 52, 13539–13546.
- 52 F. Schröder, V. Sinnwell, H. Baumann, M. Kaib and W. Francke, *Angew. Chem., Int. Ed.*, 1997, **36**, 77–80.
- 53 M. Movassaghi and A. E. Ondrus, *Org. Lett.*, 2005, 7, 4423–4426.
- 54 V. U. Ahmad, M. Zahid, M. S. Ali, Z. Ali, A. Jassbi, M. Abbas, J. Clardy, E. Lobkovsky, R. Tareen and M. Z. Iqbal, *J. Org. Chem.*, 1999, **64**, 8465–8467.
- 55 M. E. Maier and A. Bayer, *Eur. J. Org. Chem.*, 2006, **2006**, 4034-4043.
- 56 T. J. Heckrodt and J. Mulzer, in *Natural products synthesis II*, Springer, 2005, pp. 1–41.
- 57 A. Majdalani and H.-G. Schmalz, Synlett, 1997, 1997, 1303– 1305.
- 58 P. J. Kocienski, A. Pontiroli and L. Qun, *J. Chem. Soc., Perkin Trans.* 1, 2001, 2356–2366.

- 59 S. E. Lazerwith, T. W. Johnson and E. Corey, *Org. Lett.*, 2000, 2, 2389–2392.
- 60 M. Harmata and X. Hong, Org. Lett., 2005, 7, 3581-3583.
- 61 R. G. Kerr, A. C. Kohl and T. A. Ferns, J. Ind. Microbiol. Biotechnol., 2006, 33, 532–538.
- 62 S. Werle, T. Fey, J. M. Neudörfl and H.-G. Schmalz, *Org. Lett.*, 2007, 9, 3555–3558.
- 63 B. B. Snider, D. J. Rodini, M. Karras, T. C. Kirk, E. A. Deutsch, R. Cordova and R. T. Price, *Tetrahedron*, 1981, 37, 3927–3934.
- 64 F. Muehlthau and T. Bach, *Synthesis*, 2005, 2005, 3428–3436.
- 65 J. Yadav, A. Basak and P. Srihari, *Tetrahedron Lett.*, 2007, **48**, 2841–2843.
- 66 L. Han, X. Huang, I. Sattler, U. Moellmann, H. Fu, W. Lin and S. Grabley, *Planta Med.*, 2005, **71**, 160–164.
- 67 D. M. Solorio and M. P. Jennings, J. Org. Chem., 2007, 72, 6621–6623.
- 68 C. M. Marson, J. Campbell, M. B. Hursthouse and K. A. Malik, *Angew. Chem., Int. Ed.*, 1998, 37, 1122–1124.
- 69 T. S. Wu, Y. Y. Chan, M. J. Liou, F. W. Lin, L. S. Shi and K. T. Chen, *Phytother. Res.*, 1998, **12**, S80–S82.
- 70 T.-S. Wu, M.-L. Wang and P.-L. Wu, *Phytochemistry*, 1996, 43, 785–789.
- 71 H. Furukawa, T. Wu, T. Ohta and C. Kuoh, *Chem. Pharm. Bull.*, 1985, **33**, 4132–4138.
- 72 A. Ueno, T. Kitawaki and N. Chida, *Org. Lett.*, 2008, **10**, 1999–2002.
- 73 H. Rupp, W. Schwarz and H. Musso, *Eur. J. Inorg. Chem.*, 1983, **116**, 2554–2563.
- 74 W. Zhuang, T. Hansen and K. A. Jørgensen, *Chem. Commun.*, 2001, 347–348.
- 75 S. Yamazaki and Y. Iwata, J. Org. Chem., 2006, 71, 739-743.
- 76 J. Romo and P. Joseph-Nathan, *Tetrahedron*, 1964, 20, 2331– 2337.
- 77 H. Kakisawa, Y. Inouye and J. Romo, *Tetrahedron Lett.*, 1969, **10**, 1929–1932.
- 78 W. D. Inman, J. Luo, S. D. Jolad, S. R. King and R. Cooper, J. Nat. Prod., 1999, 62, 1088–1092.
- 79 M. Jimenez-Estrada, R. R. Chilpa, T. R. Apan, F. Lledias,
  W. Hansberg, D. Arrieta and F. A. Aguilar, *J. Ethnopharmacol.*, 2006, 105, 34–38.
- 80 K. Shindo, M. Kimura and M. Iga, *Biosci., Biotechnol., Biochem.*, 2004, 68, 1393–1394.
- 81 V. Snieckus, Chem. Rev., 1990, 90, 879-933.
- 82 A. Padwa and N. Kamigata, J. Am. Chem. Soc., 1977, 99, 1871–1880.
- 83 B. L. Kedrowski and R. W. Hoppe, J. Org. Chem., 2008, 73, 5177–5179.
- 84 W.-H. Lin, J.-M. Fang and Y.-S. Cheng, *Phytochemistry*, 1995, 40, 871–873.
- 85 C.-I. Chang, J.-Y. Chang, C.-C. Kuo, W.-Y. Pan and Y.-H. Kuo, *Planta Med.*, 2005, **71**, 72–76.
- 86 K. Kawazoe, M. Yamamoto, Y. Takaishi, G. Honda, T. Fujita, E. Sezik and E. Yesilada, *Phytochemistry*, 1999, 50, 493–497.
- 87 H. Ohtsu, M. Iwamoto, H. Ohishi, S. Matsunaga and R. Tanaka, *Tetrahedron Lett.*, 1999, 40, 6419–6422.

- 88 E. Alvarez-Manzaneda, R. Chahboun, E. Cabrera, E. Alvarez, R. Alvarez-Manzaneda, R. Meneses, H. Es-Samti and A. Fernandez, *J. Org. Chem.*, 2009, 74, 3384–3388.
- 89 A. Cartier, H. Chan, J.-L. Malo, L. Pineau, K. Tse and M. Chan-Yeung, J. Allergy Clin. Immunol., 1986, 77, 639–645.
- 90 D. N. Weissman and D. M. Lewis, Occup. Med., 2000, 15, 385-398.
- 91 M. Chan-Yeung, H. Chan, K. Tse, H. Salari and S. Lam, J. Allergy Clin. Immunol., 1989, 84, 762–768.
- 92 J. Gardner, G. Barton and H. Maclean, *Can. J. Chem.*, 1959, 37, 1703–1709.
- 93 B.-F. Sun, R. Hong, Y.-B. Kang and L. Deng, J. Am. Chem. Soc., 2009, 131, 10384–10385.
- 94 G. Cassinelli, C. Lanzi, T. Pensa, R. A. Gambetta, G. Nasini,
  G. Cuccuru, M. Cassinis, G. Pratesi, D. Polizzi and
  M. Tortoreto, *Biochem. Pharmacol.*, 2000, 59, 1539–1547.
- 95 L. Merlini, G. Nasini, L. Scaglioni, G. Cassinelli and C. Lanzi, *Phytochemistry*, 2000, 53, 1039–1041.
- 96 T. Yoshimitsu, S. Nojima, M. Hashimoto, K. Tsukamoto and T. Tanaka, *Synthesis*, 2009, **2009**, 2963–2969.
- 97 O. M. Moradei and L. A. Paquette, Org. Synth., 2003, 66-74.
- 98 M. E. Layton, C. A. Morales and M. D. Shair, J. Am. Chem. Soc., 2002, 124, 773–775.
- 99 N. Sun and X. Liang, Acta Chim. Sin., 1981, 16, 24.
- 100 D. Yin, C. Xu, Y. Gao, D. Liu, S. Wen and J. Guo, *Acta Pharm. Sin.*, 1992, 27, 824–829.
- 101 Z.-W. Zhang and W.-D. Z. Li, *Org. Lett.*, 2010, **12**, 1649–1651.
- 102 A. Ulubelen, G. Topcu, N. Tan, L.-J. Lin and G. A. Cordell, *Phytochemistry*, 1992, **31**, 2419–2421.
- 103 A. Ulubelen, U. Sönmez and G. Topcu, *Phytochemistry*, 1997, 44, 1297–1299.
- 104 A. Ulubelen, G. Topçu, H.-B. Chai and J. M. Pezzuto, *Pharm. Biol*, 1999, **37**, 148–151.
- 105 A. Ulubelen, S. Öksüz, U. Kolak, C. Bozok-Johansson, C. Çelik and W. Voelter, *Planta Med.*, 2000, **66**, 458–462.
- 106 G. Topcu, E. N. Altiner, S. Gozcu, B. Halfon, Z. Aydogmus, J. Pezzuto, B.-N. Zhou and D. G. Kingston, *Planta Med.*, 2003, **69**, 464–467.
- 107 A. Kabouche, N. Boutaghane, Z. Kabouche, E. Seguin, F. Tillequin and K. Benlabed, *Fitoterapia*, 2005, 76, 450– 452.
- 108 R. A. Taj and J. R. Green, J. Org. Chem., 2010, 75, 8258-8270.
- 109 N. Gulavita, A. Hori, Y. Shimizu, P. Laszlo and J. Clardy, *Tetrahedron Lett.*, 1988, **29**, 4381–4384.
- 110 M. Medjahdi, J. C. González-Gómez, F. Foubelo and M. Yus, *Eur. J. Org. Chem.*, 2011, 2011, 2230–2234.
- 111 M. Li, P. Zhou and H. F. Roth, Synthesis, 2007, 2007, 55-60.
- 112 J. F. Bower, P. Szeto and T. Gallagher, *Chem. Commun.*, 2005, 5793–5795.
- 113 Z. Ma and H. Zhai, Synlett, 2007, 2007, 0161-0163.
- 114 Z. Ma, H. Hu, W. Xiong and H. Zhai, *Tetrahedron*, 2007, **63**, 7523–7531.
- 115 D. N. Mai, B. R. Rosen and J. P. Wolfe, *Org. Lett.*, 2011, **13**, 2932–2935.

- 116 P. A. Donets, J. L. Goeman, J. Van der Eycken, K. Robeyns, L. Van Meervelt and E. V. Van der Eycken, *Eur. J. Org. Chem.*, 2009, **2009**, 793–796.
- 117 A. D. Rodríguez, C. Ramírez, I. I. Rodríguez and C. L. Barnes, *J. Org. Chem.*, 2000, **65**, 1390–1398.
- 118 M. Harmata, W. Ying and C. L. Barnes, *Tetrahedron Lett.*, 2009, **50**, 2326–2328.
- 119 W. Ying, C. L. Barnes and M. Harmata, *Tetrahedron Lett.*, 2011, **52**, 177–180.
- 120 A. D. Rodríguez and C. Ramírez, *J. Nat. Prod.*, 2001, **64**, 100–102.
- 121 J. S. Yadav, B. Thirupathaiah and A. Al Khazim Al Ghamdi, *Eur. J. Org. Chem.*, 2012, 2012, 2072–2076.
- 122 L. Fu, N. Li and R. Mill, *Flora of China*, 1999, vol. 4, pp. 85– 88.
- 123 H. Abdelkafi and B. Nay, Nat. Prod. Rep., 2012, 29, 845-869.
- 124 S. He, B. Wu, Y. Pan and L. Jiang, *J. Org. Chem.*, 2008, 73, 5233–5241.
- 125 L. Li, G. E. Henry and N. P. Seeram, *J. Agric. Food Chem.*, 2009, **57**, 7282–7287.
- 126 T. Ito, Y. Akao, H. Yi, K. Ohguchi, K. Matsumoto, T. Tanaka, M. Iinuma and Y. Nozawa, *Carcinogenesis*, 2003, 24, 1489– 1497.
- 127 K.-S. Huang, M. Lin, L.-N. Yu and M. Kong, *Tetrahedron*, 2000, **56**, 1321–1329.
- 128 H. J. Kim, E. J. Chang, S. H. Cho, S. K. Chung, H. D. Park and S. W. Choi, *Biosci., Biotechnol., Biochem.*, 2002, 66, 1990–1993.
- 129 J.-R. Dai, Y. F. Hallock, J. H. Cardellina and M. R. Boyd, *J. Nat. Prod.*, 1998, **61**, 351–353.
- 130 A. E. Bala, A. Kollmann, P. H. Ducrot, A. Majira, L. Kerhoas, R. Delorme and J. Einhorn, *Pestic. Sci.*, 1999, 55, 206–208.
- 131 S. Sotheeswaran and V. Pasupathy, *Phytochemistry*, 1993, 32, 1083–1092.
- 132 C. W. Choi, Y. H. Choi, M.-R. Cha, Y. S. Kim, G. H. Yon, K. S. Hong, W.-K. Park and S. Y. Ryu, *Planta Med.*, 2011, 77, 374–376.
- 133 M. H. Jang, X. L. Piao, J. M. Kim, S. W. Kwon and J. H. Park, *Phytother. Res.*, 2008, **22**, 544–549.
- 134 Y. L. Choi, B. T. Kim and J.-N. Heo, *J. Org. Chem.*, 2012, 77, 8762–8767.
- 135 Z. Jin, Nat. Prod. Rep., 2011, 28, 1143-1191.
- 136 P. B. Koswatta and C. J. Lovely, *Nat. Prod. Rep.*, 2011, 28, 511–528.
- 137 J. Sullivan, R. Giles and R. Looper, *Curr. Bioact. Compd.*, 2009, **5**, 39–78.
- 138 M. Roué, E. Quévrain, I. Domart-Coulon and M.-L. Bourguet-Kondracki, *Nat. Prod. Rep.*, 2012, 29, 739– 751.
- 139 W. Hassan, R. Edrada, R. Ebel, V. Wray, A. Berg, R. van Soest, S. Wiryowidagdo and P. Proksch, *J. Nat. Prod.*, 2004, 67, 817–822.
- 140 J. Das, P. B. Koswatta, J. D. Jones, M. Yousufuddin and C. J. Lovely, *Org. Lett.*, 2012, **14**, 6210–6213.
- 141 K. Nicolaou, Q. Kang, T. R. Wu, C. S. Lim and D. Y.-K. Chen, *J. Am. Chem. Soc.*, 2010, **132**, 7540–7548.

- 142 K. e. C. Nicolaou, T. R. Wu, Q. Kang and D. Y. K. Chen, *Angew. Chem.*, 2009, **121**, 3492–3495.
- 143 A. Dorn, V. Schattel and S. Laufer, *Bioorg. Med. Chem. Lett.*, 2010, **20**, 3074–3077.
- 144 S. A. Laufer, G. M. Ahrens, S. C. Karcher, J. S. Hering and R. Niess, *J. Med. Chem.*, 2006, **49**, 7912–7915.
- 145 Y.-P. Xue and W.-D. Z. Li, J. Org. Chem., 2011, 76, 57-64.
- 146 E. H. Hakim, L. D. Juliawaty, Y. M. Syah, L. bin Din, E. L. Ghisalberti, J. Latip, I. M. Said and S. A. Achmad, Z. Naturforsch., C: J. Biosci., 2005, 60, 723–727.
- 147 T. Ito, T. Tanaka, M. Iinuma, K.-i. Nakaya, Y. Takahashi, R. Sawa, J. Murata and D. Darnaedi, *J. Nat. Prod.*, 2004, 67, 932–937.
- 148 L. Botella and C. Nájera, Tetrahedron, 2004, 60, 5563-5570.
- 149 P. S. Baran and N. Z. Burns, J. Am. Chem. Soc., 2006, **128**, 3908–3909.
- 150 G.-W. Wang, H.-L. Wang, D. A. Capretto, Q. Han, R.-B. Hu and S.-D. Yang, *Tetrahedron*, 2012, **68**, 5216–5222.
- 151 F. Delle Monache, G. Delle Monache, J. F. Cavalcanti and R. M. Pinheiro, *Tetrahedron Lett.*, 1987, **28**, 563–566.
- 152 K. Ezaki, M. Satake, T. Kusumi and H. Kakisawa, *Tetrahedron Lett.*, 1991, **32**, 2793–2796.
- 153 Y. Suzuki, N. Matsuo, T. Nemoto and Y. Hamada, *Tetrahedron*, 2013, **69**, 5913–5919.
- 154 R. Bandichhor, A. N. Lowell and M. C. Kozlowski, J. Org. Chem., 2011, 76, 6475–6487.
- 155 A. Pinder, in *Fortschritte der Chemie Organischer Naturstoffe/ Progress in the Chemistry of Organic Natural Products*, Springer, 1977, pp. 81–186.
- 156 A. Romo de Vivar, A.-L. Pérez-Castorena, A. Arciniegas and J. L. Villaseñor, *J. Mex. Chem. Soc.*, 2007, **51**, 160–172.
- 157 Y. Saito, M. Hattori, Y. Iwamoto, Y. Takashima, K. Mihara, Y. Sasaki, M. Fujiwara, M. Sakaoku, A. Shimizu and X. Chao, *Tetrahedron*, 2011, 67, 2220–2231.
- 158 A. L. Silva, R. A. Toscano and L. A. Maldonado, J. Org. Chem., 2013, 78, 5282–5292.
- 159 F. Schaller, L. Rahalison, N. Islam, O. Potterat, K. Hostettmann, H. Stoeckli-Evans and S. Mavi, *Helv. Chim. Acta*, 2000, 83, 407–413.
- 160 J. Huang, D. Foyle, X. Lin and J. Yang, *J. Org. Chem.*, 2013, **78**, 9166–9173.
- 161 J. K. Kim, Y. H. Kim, H. T. Nam, B. T. Kim and J.-N. Heo, *Org. Lett.*, 2008, **10**, 3543–3546.
- 162 D. M. Gardiner, P. Waring and B. J. Howlett, *Microbiology*, 2005, **151**, 1021–1032.
- 163 N. Boyer, K. C. Morrison, J. Kim, P. J. Hergenrother and M. Movassaghi, *Chem. Sci.*, 2013, 4, 1646–1657.
- 164 C.-J. Zheng, C.-J. Kim, K. S. Bae, Y.-H. Kim and W.-G. Kim, *J. Nat. Prod.*, 2006, **69**, 1816–1819.
- 165 A. Coste, J. Kim, T. C. Adams and M. Movassaghi, *Chem. Sci.*, 2013, 4, 3191–3197.
- 166 C. C. Hughes, J. B. MacMillan, S. P. Gaudêncio, P. R. Jensen and W. Fenical, *Angew. Chem., Int. Ed.*, 2009, **48**, 725–727.
- 167 A. Miyanaga, J. E. Janso, L. McDonald, M. He, H. Liu, L. Barbieri, A. S. Eustáquio, E. N. Fielding, G. T. Carter and P. R. Jensen, *J. Am. Chem. Soc.*, 2011, 133, 13311–13313.

- 168 Y. Aotani, H. Nagata and M. Yoshida, *J. Antibiot.*, 1997, **50**, 543–545.
- 169 Y. Takayama, T. Yamada, S. Tatekabe and K. Nagasawa, *Chem. Commun.*, 2013, **49**, 6519–6521.
- 170 J. W. Blunt, B. R. Copp, R. A. Keyzers, M. H. G. Munro and M. R. Prinsep, *Nat. Prod. Rep.*, 2012, **29**, 144–222.
- 171 J. C. Morris and A. J. Phillips, *Nat. Prod. Rep.*, 2011, **28**, 269–289.
- 172 R. K. Akee, T. R. Carroll, W. Y. Yoshida, P. J. Scheuer, T. J. Stout and J. Clardy, *J. Org. Chem.*, 1990, 55, 1944–1946.
- 173 S. Nakamura, N. Tsuno, M. Yamashita, I. Kawasaki, S. Ohta and Y. Ohishi, *J. Chem. Soc., Perkin Trans.* 1, 2001, 1, 429– 436.
- 174 X. Fu, J. R. Barnes, T. Do and F. J. Schmitz, *J. Nat. Prod.*, 1997, **60**, 497–498.
- 175 J. Das, A. Bhan, S. S. Mandal and C. J. Lovely, *Bioorg. Med. Chem. Lett.*, 2013, 23, 6183–6187.
- 176 M. Alvarez and J. A. Joule, *Tetrahedron Lett.*, 2001, 42, 15–317.
- 177 S. Goodwin, A. Smith and E. Horning, *J. Am. Chem. Soc.*, 1959, **81**, 1903–1908.
- 178 R. Woodward, G. Iacobucci and I. Hochstein, *J. Am. Chem. Soc.*, 1959, **81**, 4434–4435.
- 179 P. Juret, A. Tanguy, A. Girard, J. Le Talaer, J. Abbatucci, N. Dat-Xuong, J. Le Pecq and C. Paolett, *Eur. J. Cancer*, 1978, 14, 205–206.
- 180 H.-J. Knölker and K. R. Reddy, *Chem. Rev.*, 2002, **102**, 4303–4428.
- 181 M. Stiborová, C. A. Bieler, M. Wiessler and E. Frei, *Biochem. Pharmacol.*, 2001, 62, 1675–1684.
- 182 C. M. Miller and F. O. McCarthy, *RSC Adv.*, 2012, **2**, 8883–8918.
- 183 D. Mal, B. K. Senapati and P. Pahari, *Tetrahedron*, 2007, 63, 3768–3781.
- 184 H. Y. Lee, G. S. Chen, C. S. Chen and J. W. Chern, J. *Heterocycl. Chem.*, 2010, 47, 454–458.
- 185 M. Dračínský, J. Sejbal, B. Rygerová and M. Stiborová, *Tetrahedron Lett.*, 2007, **48**, 6893–6895.
- 186 N. Ramkumar, M. S. Raghavendra and R. Nagarajan, *Synlett*, 2014, **25**, 2791–2793.
- 187 D. He, L. Ding, H. Xu, X. Lei, H. Xiao and Y. Zhou, J. Org. Chem., 2012, 77, 8435–8443.
- 188 L. H. Mejorado and T. R. Pettus, J. Am. Chem. Soc., 2006, 128, 15625–15631.
- 189 K. e. C. Nicolaou, D. J. Edmonds, A. Li and G. S. Tria, *Angew. Chem.*, 2007, **119**, 4016–4019.
- 190 Y. Kita, M. Arisawa, M. Gyoten, M. Nakajima, R. Hamada, H. Tohma and T. Takada, *J. Org. Chem.*, 1998, **63**, 6625– 6633.
- 191 H. Tohma, H. Morioka, S. Takizawa, M. Arisawa and Y. Kita, *Tetrahedron*, 2001, **57**, 345–352.
- 192 H. Li, Y. Zhang, X. Xie, H. Ma, C. Zhao, G. Zhao and X. She, *Org. Lett.*, 2014, **16**, 4440–4443.
- 193 T.-S. Kam, G. Subramaniam, K.-H. Lim and Y.-M. Choo, *Tetrahedron Lett.*, 2004, **45**, 5995–5998.
- 194 Z. Lv, Z. Li and G. Liang, Org. Lett., 2014, 16, 1653-1655.
- 195 A. l. Biechy and S. Z. Zard, Org. Lett., 2009, 11, 2800-2803.

- 196 J. Zhu, G. Islas-Gonzalez and M. Bois-Choussy, Org. Prep. Proced. Int., 2000, 32, 505-546.
- 197 D. S. Jang, E. J. Park, M. E. Hawthorne, J. S. Vigo, J. G. Graham, F. Cabieses, B. D. Santarsiero, A. D. Mesecar, H. H. Fong, R. G. Mehta, J. M. Pezzuto and A. D. Kinghorn, J. Agric. Food Chem., 2002, 50, 6330-6334.
- 198 L.-B. Dong, J. He, X.-Y. Li, X.-D. Wu, X. Deng, G. Xu, L.-Y. Peng, Y. Zhao, Y. Li and X. Gong, Nat. Prod. Bioprospect., 2011, 1, 41-47.
- 199 Z. Li, T.-F. Leung and R. Tong, Chem. Commun., 2014, 50, 10990-10993.
- 200 A. Al Mourabit and P. Potier, Eur. J. Org. Chem., 2001, 2001, 237-243.
- 201 H. Hoffmann and T. Lindel, Synthesis, 2003, 2003, 1753-1783.
- 202 T. Imaoka, M. Iwata, T. Akimoto and K. Nagasawa, Nat. Prod. Bioprospect., 2013, 8, 961-964.
- 203 N. M. Hewlett and J. J. Tepe, Org. Lett., 2011, 13, 4550-4553.
- 204 M. Iwata, K. Kanoh, T. Imaoka and K. Nagasawa, Chem. Commun., 2014, 50, 6991-6994.
- 205 L. Garrido, E. Zubía, M. J. Ortega and J. Salvá, J. Org. Chem., 2003, 68, 293-299.
- 206 Y. Momoi, K. i. Okuyama, H. Toya, K. Sugimoto, K. Okano and H. Tokuyama, Angew. Chem., 2014, 126, 13431-13435.
- 207 L. N. Mander, Nat. Prod. Rep., 2003, 20, 49-69.
- 208 P. A. García, A. B. De Oliveira and R. Batista, Molecules, 2007, 12, 455-483.
- 209 J. R. Hanson, Nat. Prod. Rep., 2009, 26, 1156-1171.
- 210 M. Presset, Y. Coquerel and J. Rodriguez, Chem. Rev., 2012, 113, 525-595.
- 211 L. Zhu, S.-H. Huang, J. Yu and R. Hong, Tetrahedron Lett., 2015, 56, 23-31.
- 212 E. H. Colebrook, S. G. Thomas, A. L. Phillips and P. Hedden, J. Exp. Biol., 2014, 217, 67-75.
- 213 J.-M. Davière and P. Achard, Development, 2013, 140, 1147-1151.
- 214 P. J. Davies, The plant hormones: their nature, occurrence, and functions, in Plant hormones, Springer, Dordrecht, 2010, pp. 1-15.
- 215 G. Topçu, A. Ertaş, M. Öztürk, D. Dinçel, T. Kılıç and B. Halfon, Phytochem. Lett., 2011, 4, 436-439.
- 216 S. J. Burke, W. P. Malachowski, S. K. Mehta and R. Appenteng, Org. Biomol. Chem., 2015, 13, 2726-2744.
- 217 D. F. Taber and R. B. Sheth, J. Org. Chem., 2008, 73, 8030-8032.
- 218 H.-J. Liu and D. D.-P. Tran, Tetrahedron Lett., 1999, 40, 3827-3830.
- 219 X.-W. Yang, C.-P. Yang, L.-P. Jiang, X.-J. Qin, Y.-P. Liu, Q.-S. Shen, Y.-B. Chen and X.-D. Luo, Org. Lett., 2014, 16, 5808-5811.
- 220 X. Liang, S.-Z. Jiang, K. Wei and Y.-R. Yang, J. Am. Chem. Soc., 2016, 138, 2560-2562.
- 221 C. Beemelmanns, V. Blot, S. Gross, D. Lentz and H. U. Reissig, Eur. J. Org. Chem., 2010, 2010, 2716-2732.
- 222 M. A. Schafroth, D. Sarlah, S. Krautwald and E. M. Carreira, J. Am. Chem. Soc., 2012, 134, 20276-20278.

- 223 J. Y. Hamilton, D. Sarlah and E. M. Carreira, J. Am. Chem. Soc., 2013, 135, 994-997.
- 224 O. F. Jeker, A. G. Kravina and E. M. Carreira, Angew. Chem., 2013, 125, 12388-12391.
- 225 T. J. Maimone and P. S. Baran, Nat. Chem. Biol., 2007, 3, 396-407.
- 226 L. A. Loyola, G. Morales, B. Rodriguez, J. Jiménez-Barbero, M. C. d. l. Torre, A. Perales and M. R. Torres, Tetrahedron, 1990, 46, 5413-5420.
- 227 L. A. Loyola, G. Morales, M. C. De La Torre, S. Pedreros and B. Rodríguez, Phytochemistry, 1990, 29, 3950-3951.
- 228 C. Areche, B. Sepulveda, A. S. Martin, O. Garcia-Beltran, M. Simirgiotis and A. Canete, Org. Biomol. Chem., 2014, 12, 6406-6413.
- 229 C. Areche, F. Rojas-Alvarez, C. Campos-Briones, C. Lima, E. G. Pérez and B. Sepúlveda, J. Pharm. Pharmacol., 2013, 65, 1231-1238.
- 230 G. A. Wächter, S. G. Franzblau, G. Montenegro, E. Suarez, R. H. Fortunato, E. Saavedra and B. N. Timmermann, J. Nat. Prod., 1998, 61, 965-968.
- 231 L. A. Loyola, J. Bórquez, G. Morales, A. San-Martín, J. Darias, N. Flores and A. Giménez, Phytochemistry, 2004, 65, 1931-1935.
- 232 G. A. Wächter, G. Matooq, J. J. Hoffmann, W. M. Maiese, M. P. Singh, G. Montenegro and B. N. Timmermann, J. Nat. Prod., 1999, 62, 1319-1321.
- 233 L. A. Loyola, J. Bórquez, G. Morales and A. San-Martín, Phytochemistry, 2000, 53, 961-963.
- 234 L. A. Loyola, J. Bórquez, G. Morales and A. S. Martin, Phytochemistry, 1997, 44, 649-651.
- 235 C. Areche, L. A. Loyola, J. Borquez, J. Rovirosa and A. San-Martin, Magn. Reson. Chem., 2008, 46, 765-768.
- 236 Y. T. Liu, L. P. Li, J. H. Xie and Q. L. Zhou, Angew. Chem., Int. Ed., 2017, 56, 12708-12711.
- 237 W.-S. Sun, S. Su, R.-X. Zhu, G.-Z. Tu, W. Cheng, H. Liang, X.-Y. Guo, Y.-Y. Zhao and Q.-Y. Zhang, Tetrahedron Lett., 2013, 54, 3617-3620.
- 238 B. P. Bandgar and K. A. Shaikh, Tetrahedron Lett., 2003, 44, 1959-1961.
- 239 S. Luo, C. A. Zificsak and R. P. Hsung, Org. Lett., 2003, 5, 4709-4712.
- 240 P. Feng, Y. Fan, F. Xue, W. Liu, S. Li and Y. Shi, Org. Lett., 2011, 13, 5827-5829.
- 241 M. Tsuda, H. Sato, Y. Tanaka, K. Yazawa, Y. Mikami, T. Sasaki and J. i. Kobayashi, J. Chem. Soc., Perkin Trans. 1, 1996, 1773-1775.
- 242 M. L. Patil, H. B. Borate, D. E. Ponde, B. M. Bhawal and V. H. Deshpande, Tetrahedron Lett., 1999, 40, 4437-4438.
- 243 M. L. Patil, H. B. Borate, D. E. Ponde and V. H. Deshpande, Tetrahedron, 2002, 58, 6615-6620.
- 244 J.-R. Ioset, A. Marston, M. P. Gupta and K. Hostettmann, J. Nat. Prod., 2000, 63, 424-426.
- 245 R. Aguilar, A. Benavides and J. Tamariz, Synth. Commun., 2004, 34, 2719-2735.
- 246 A. Chatterjee and B. Hazra, Tetrahedron, 1977, 33, 1983-1987.

This article is licensed under a Creative Commons Attribution-NonCommercial 3.0 Unported Licence. Open Access Article. Published on 03 December 2018. Downloaded on 8/26/2025 9:48:33 PM.

- 247 A. K. Sinha, R. Dogra and B. P. Joshi, *Indian J. Chem., Sect. B:* Org. Chem. Incl. Med. Chem., 2002, **41B**, 635–638.
- 248 M. Gordaliza, P. A. García, J. M. Miguel del Corral, M. A. Castro and M. A. Gómez-Zurita, *Toxicon*, 2004, 44, 441–459.
- 249 R. S. Ward, Phytochem. Rev., 2003, 2, 391-400.
- 250 R. S. Ward, Nat. Prod. Rep., 1999, 16, 75-96.
- 251 D. Stadler and T. Bach, Angew. Chem., Int. Ed., 2008, 47, 7557–7559.
- 252 K. Kondo and F. Mori, Chem. Lett., 1974, 3, 741-742.
- 253 C. Gnamm, S. Foerster, N. Miller, K. Broedner and G. Helmchen, *Synlett*, 2007, **2007**, 0790–0794.
- 254 J. Saxton, in *The Alkaloids: Chemistry and Physiology*, Elsevier, 1965, vol. 8, pp. 673–678.
- 255 E. F. Rogers, H. R. Snyder and R. F. Fischer, *J. Am. Chem. Soc.*, 1952, 74, 1987–1989.
- 256 H. R. Snyder, R. F. Fischer, J. F. Walker, H. E. Els and G. A. Nussberger, *J. Am. Chem. Soc.*, 1954, **76**, 2819–2825.
- 257 P. Yates, F. N. MacLachlan, I. D. Rae, M. Rosenberger,
  A. G. Szabo, C. R. Willis, M. P. Cava, M. Behforouz,
  M. V. Lakshmikantham and W. Zeiger, *J. Am. Chem. Soc.*,
  1973, 95, 7842–7850.
- 258 H. Ueda, H. Satoh, K. Matsumoto, K. Sugimoto, T. Fukuyama and H. Tokuyama, *Angew. Chem., Int. Ed.*, 2009, 48, 7600–7603.
- 259 C. S. Yang, J. D. Lambert and S. Sang, Arch. Toxicol., 2009, 83, 11–21.
- 260 L. Bonfili, V. Cecarini, M. Amici, M. Cuccioloni, M. Angeletti, J. N. Keller and A. M. Eleuteri, *FEBS J.*, 2008, 275, 5512–5526.
- 261 H. M. Ge, C. Xu, X. T. Wang, B. Huang and R. X. Tan, *Eur. J.* Org. Chem., 2006, **2006**, 5551–5554.
- 262 K. Warabi, S. Matsunaga, R. W. M. van Soest and N. Fusetani, *J. Org. Chem.*, 2003, **68**, 2765–2770.
- 263 K. Okano, H. Fujiwara, T. Noji, T. Fukuyama and H. Tokuyama, *Angew. Chem.*, 2010, **122**, 6061–6065.
- 264 E. Duval and G. D. Cuny, *Tetrahedron Lett.*, 2004, 45, 5411–5413.
- 265 J. P. Karwowski, M. Jackson, R. R. Rasmussen, P. E. Humphrey, J. B. Poddig, W. L. Kohl, M. H. Scherr, S. Kadam and J. B. Mcalpine, *J. Antibiot.*, 1993, 46, 374–379.
- 266 K. M. Depew, S. P. Marsden, D. Zatorska, A. Zatorski,
  W. G. Bornmann and S. J. Danishefsky, *J. Am. Chem. Soc.*, 1999, 121, 11953–11963.
- 267 S. Takiguchi, T. Iizuka, Y.-s. Kumakura, K. Murasaki, N. Ban, K. Higuchi and T. Kawasaki, *J. Org. Chem.*, 2010, 75, 1126–1131.
- 268 U. Anthoni, C. Christophersen and P. H. Nielsen, Alkaloids: Chem. Biol. Perspect., 1999, 13, 163–236.
- 269 P. Waring, R. D. Eichner and A. Müllbacher, *Med. Chem. Res.*, 1988, **8**, 499–524.
- 270 T. Rezanka, M. Sobotka, J. Spízek and K. Sigler, *Anti-Infect.* Agents Med. Chem., 2006, 5, 187–224.
- 271 C. L. Chai and P. Waring, Redox Rep., 2000, 5, 257-264.
- 272 P. Trown, Biochem. Biophys. Res. Commun., 1968, 33, 402-407.
- 273 T. Hino and T. Sato, Tetrahedron Lett., 1971, 12, 3127-3129.

- 274 Y. Kishi, T. Fukuyama and S. Nakatsuka, *J. Am. Chem. Soc.*, 1973, **95**, 6492–6493.
- 275 Y. Kishi, S. Nakatsuka, T. Fukuyama and M. Havel, J. Am. Chem. Soc., 1973, **95**, 6493–6495.
- 276 J. Kim, J. A. Ashenhurst and M. Movassaghi, *Science*, 2009, 324, 238–241.
- 277 Y. Usami, J. Yamaguchi and A. Numata, *Heterocycles*, 2004, 63, 1123–1129.
- 278 B. V. Bertinetti, M. A. Rodriguez, A. M. Godeas and G. M. Cabrera, *J. Antibiot.*, 2010, **63**, 681.
- 279 M. Movassaghi, M. A. Schmidt and J. A. Ashenhurst, *Angew. Chem.*, 2008, **120**, 1507–1509.
- 280 N. Boyer and M. Movassaghi, *Chem. Sci.*, 2012, **3**, 1798–1803.
- 281 E. J. Park, H. R. Park, J. S. Lee and J. Kim, *Planta Med.*, 1998, 64, 464–466.
- 282 G. Yoon, B. Y. Kang and S. H. Cheon, *Arch. Pharmacal Res.*, 2007, **30**, 313–316.
- 283 S. F. Nielsen, M. Chen, T. G. Theander, A. Kharazmi and S. B. Christensen, *Bioorg. Med. Chem. Lett.*, 1995, 5, 449– 452.
- 284 H. Haraguchi, K. Tanimoto, Y. Tamura, K. Mizutani and T. Kinoshita, *Phytochemistry*, 1998, **48**, 125–129.
- 285 H. Haraguchi, H. Ishikawa, K. Mizutani, Y. Tamura and T. Kinoshita, *Bioorg. Med. Chem.*, 1998, **6**, 339–347.
- 286 T. Nomura and T. Fukai, in *Fortschritte der Chemie* organischer Naturstoffe/Progress in the Chemistry of Organic Natural Products, Springer, 1998, pp. 1–140.
- 287 T. Saitoh and S. Shibata, *Tetrahedron Lett.*, 1975, **16**, 4461–4462.
- 288 G. Yoon, Y. Do Jung and S. H. Cheon, *Chem. Pharm. Bull.*, 2005, **53**, 694–695.
- 289 Y. Na, J.-H. Cha, H.-G. Yoon and Y. Kwon, *Chem. Pharm.* Bull., 2009, 57, 607–609.
- 290 G. Yoon, W. Lee, S.-N. Kim and S. H. Cheon, *Bioorg. Med. Chem. Lett.*, 2009, **19**, 5155–5157.
- 291 Z. Wang, Y. Cao, S. Paudel, G. Yoon and S. H. Cheon, *Arch. Pharmacal Res.*, 2013, **36**, 1432–1436.
- 292 Y.-C. Kong, K.-F. Cheng, R. C. Cambie and P. G. Waterman, *J. Chem. Soc., Chem. Commun.*, 1985, 47–48.
- 293 Y.-C. Kong, K.-F. Cheng, K.-H. Ng, P. P.-H. But, S.-X. Yu, H.-T. Chang, R. C. Cambie, T. Kinoshita, W.-s. Kan and P. G. Waterman, *Biochem. Syst. Ecol.*, 1986, 14, 491–497.
- 294 C. Cassani, R. Martín-Rapún, E. Arceo, F. Bravo and P. Melchiorre, *Nat. Protoc.*, 2013, **8**, 325.
- 295 N. S. Dange, B.-C. Hong and G.-H. Lee, *RSC Adv.*, 2014, 4, 59706–59715.
- 296 Y. S. Tsizin, Chem. Heterocycl. Compd., 1978, 14, 925-940.
- 297 P. D. Bass, D. A. Gubler, T. C. Judd and R. M. Williams, *Chem. Rev.*, 2013, **113**, 6816–6863.
- 298 I. Kock, D. Heber, M. Weide, U. Wolschendorf and B. Clement, J. Med. Chem., 2005, 48, 2772–2777.
- 299 L. Z. Benet, D. Kroetz, L. Sheiner, J. Hardman and L. Limbird, *Goodman and Gilman's the pharmacological basis of therapeutics*, 1996, pp. 3–27.
- 300 A. Clarysse, A. Brugarolas, P. Siegenthaler, R. Abele, F. Cavalli, R. De Jager, G. Renard, M. Rozencweig,

H. H. Hansen and E. E. C. T. Group, *Eur. J. Cancer Clin. Oncol.*, 1984, **20**, 243–247.

- 301 G. W. Gribble, M. G. Saulnier, M. P. Sibi and J. A. Obaza-Nutaitis, J. Org. Chem., 1984, 49, 4518–4523.
- 302 M.-L. Bennasar, T. Roca and F. Ferrando, *J. Org. Chem.*, 2005, **70**, 9077–9080.
- 303 P. H. Bernardo, C. L. Chai, G. A. Heath, P. J. Mahon, G. D. Smith, P. Waring and B. A. Wilkes, *J. Med. Chem.*, 2004, 47, 4958–4963.
- 304 N. Ramkumar and R. Nagarajan, J. Org. Chem., 2014, 79, 736–741.
- 305 J. Orjala, A. D. Wright, C. A. Erdelmeier, O. Sticher and T. Rali, *Helv. Chim. Acta*, 1993, **76**, 1481–1488.
- 306 K. Ichino, Phytochemistry, 1989, 28, 955-956.
- 307 Y. Mimaki, A. Kameyama, Y. Sashida, Y. Miyata and A. Fujii, *Chem. Pharm. Bull.*, 1995, **43**, 893–895.
- 308 B. Portet, N. Fabre, V. Roumy, H. Gornitzka, G. Bourdy, S. Chevalley, M. Sauvain, A. Valentin and C. Moulis, *Phytochemistry*, 2007, 68, 1312–1320.
- 309 D. H. Dethe and B. D. Dherange, *J. Org. Chem.*, 2015, **80**, 4526–4531.
- 310 I. S. Young and P. S. Baran, Nat. Chem., 2009, 1, 193.
- 311 E. Roulland, Angew. Chem., Int. Ed., 2011, 50, 1226-1227.
- 312 C. C. Price and M. Lund, J. Am. Chem. Soc., 1940, 62, 3105–3107.
- 313 F. Mühlthau, D. Stadler, A. Goeppert, G. A. Olah, G. S. Prakash and T. Bach, *J. Am. Chem. Soc.*, 2006, **128**, 9668–9675.
- 314 H. R. Arthur, W. Hui and Y. Ng, J. Chem. Soc., 1959, 1840– 1845.
- 315 M. Cushman and L. Cheng, J. Org. Chem., 1978, 43, 286-288.
- 316 R. H. Thomson, *Naturally Occurring Quinones*, Academic Press, 2nd edn, 1971.
- 317 K. Brown, Chem. Soc. Rev., 1975, 4, 263-288.
- 318 R. Ferrier and J. Tedder, J. Chem. Soc., 1957, 1435-1437.
- 319 K. Kim, M. Spatz and F. Johnson, *Tetrahedron Lett.*, 1979, 20, 331–334.
- 320 S. De Silva, M. Watanabe and V. Snieckus, *J. Org. Chem.*, 1979, **44**, 4802–4808.
- 321 R. C. Hoch, I. U. Schraufstätter and C. G. Cochrane, J. Lab. Clin. Med., 1996, 128, 134–145.
- 322 M. Seitz, B. Dewald, N. Gerber and M. Baggiolini, J. Clin. Invest., 1991, 87, 463–469.
- 323 R. J. Capon, in *Studies in Natural Products Chemistry*, Elsevier, 1995, vol. 15, pp. 289–326.
- 324 A. D. Patil, A. J. Freyer, L. Killmer, P. Offen, B. Carte, A. J. Jurewicz and R. K. Johnson, *Tetrahedron*, 1997, 53, 5047–5060.
- 325 G. Bhide, N. Tikotkar and B. Tilak, *Tetrahedron*, 1960, **10**, 223–229.
- 326 M. Inoue, M. W. Carson, A. J. Frontier and S. J. Danishefsky, J. Am. Chem. Soc., 2001, **123**, 1878–1889.
- 327 M. K. Li and P. J. Scheuer, *Tetrahedron Lett.*, 1984, 25, 587–590.
- 328 S. Sakemi and T. Higa, Experientia, 1987, 43, 624-625.

- 329 H. Fukui, N. Yoshikawa and M. Tabata, *Phytochemistry*, 1984, 23, 301-305.
- 330 J.-i. Tanaka, H. Miki and T. Higa, *J. Nat. Prod.*, 1992, 55, 1522–1524.
- 331 H.-K. Yim, Y. Liao and H. N. Wong, *Tetrahedron*, 2003, **59**, 1877–1884.
- 332 H. A. Weber and J. B. Gloer, *J. Org. Chem.*, 1991, **56**, 4355–4360.
- 333 K. Krohn, U. Flörke, M. John, N. Root, K. Steingröver, H.-J. Aust, S. Draeger, B. Schulz, S. Antus and M. Simonyi, *Tetrahedron*, 2001, 57, 4343–4348.
- 334 E. Quesada, M. Stockley and R. J. Taylor, *Tetrahedron Lett.*, 2004, **45**, 4877–4881.
- 335 H. A. Weber, N. C. Baenziger and J. B. Gloer, J. Am. Chem. Soc., 1990, 112, 6718–6719.
- 336 P. Seephonkai, M. Isaka, P. Kittakoop,
  P. Palittapongarnpim, S. Kamchonwongpaisan,
  M. Tanticharoen and Y. Thebtaranonth, *Planta Med.*, 2002, 68, 45–48.
- 337 E. Quesada, M. Stockley, J. P. Ragot, M. E. Prime, A. C. Whitwood and R. J. Taylor, *Org. Biomol. Chem.*, 2004, 2, 2483–2495.
- 338 T. Kamo, N. Hirai, K. Iwami, D. Fujioka and H. Ohigashi, *Tetrahedron*, 2001, 57, 7649–7656.
- 339 R. Cooke, B. Johnson and W. Segal, *Aust. J. Chem.*, 1958, **11**, 230–235.
- 340 J. Nanclares, J. Gil, J. Saez, B. Schneider and F. Otálvaro, *Tetrahedron Lett.*, 2008, **49**, 3844–3847.
- 341 F. Sponga, L. Cavaletti, A. Lazzarini, A. Borghi, I. Ciciliato, D. Losi and F. Marinelli, *J. Biotechnol.*, 1999, **70**, 65–69.
- 342 Q. Liang, J. Zhang, W. Quan, Y. Sun, X. She and X. Pan, *J. Org. Chem.*, 2007, 72, 2694–2697.
- 343 D. Konwar, R. C. Boruah and J. S. Sandhu, *Tetrahedron Lett.*, 1990, **31**, 1063–1064.
- 344 G. Bringmann, G. Lang, M. Michel and M. Heubes, *Tetrahedron Lett.*, 2004, **45**, 2829–2831.
- 345 R. Kasar, R. Khan, V. Deshpande and N. Ayyangar, *Tetrahedron Lett.*, 1991, **32**, 1599–1600.
- 346 J. Yadav, N. Thrimurtulu, K. U. Gayathri, B. S. Reddy and A. Prasad, *Tetrahedron Lett.*, 2008, **49**, 6617–6620.
- 347 H. Nakamura, J. i. Kobayashi, Y. Ohizumi and Y. Hirata, *Tetrahedron Lett.*, 1982, 23, 5555–5558.
- 348 S. Aoki, H. Wei, K. Matsui, R. Rachmat and M. Kobayashi, *Bioorg. Med. Chem.*, 2003, **11**, 1969–1973.
- 349 B. F. Bowden, B. J. McCool and R. H. Willis, *J. Org. Chem.*, 2004, **69**, 7791–7793.
- 350 H. Nakamura, J. i. Kobayashi, Y. Ohizumi and Y. Hirata, J. Chem. Soc., Perkin Trans. 1, 1987, 173–176.
- 351 Y. Ohizumi, A. Kajiwara, H. Nakamura and J. i. Kobayashi, J. Pharm. Pharmacol., 1984, **36**, 785–786.
- 352 E. L. Larghi, B. V. Obrist and T. S. Kaufman, *Tetrahedron*, 2008, **64**, 5236–5245.
- 353 Y. Murakami, A. Ishii, S. Mizuno, S. Yaginuma and Y. Uehara, *Anticancer Res.*, 1999, **19**, 4145–4149.
- 354 T. Yoshino, F. Ng and S. J. Danishefsky, *J. Am. Chem. Soc.*, 2006, **128**, 14185–14191.

- 355 F. Bracher and B. Schulte, *Eur. J. Org. Chem.*, 1997, **1997**, 1979–1982.
- 356 A. T. Kreipl, C. Reid and W. Steglich, *Org. Lett.*, 2002, 4, 3287–3288.
- 357 S. Lai, Y. Shizuri, S. Yamamura, K. Kawai, Y. Terada and H. Furukawa, *Tetrahedron Lett.*, 1989, **30**, 2241–2244.
- 358 R. A. Edrada, M. Heubes, G. Brauers, V. Wray, A. Berg, U. Gräfe, M. Wohlfarth, J. Mühlbacher, K. Schaumann, G. Bringmann and P. Proksch, *J. Nat. Prod.*, 2002, 65, 1598–1604.
- 359 M. V. R. Reddy, A. J. Yucel and P. V. Ramachandran, *J. Org. Chem.*, 2001, **66**, 2512–2514.
- 360 J. Yadav, N. Thrimurtulu, K. U. Gayathri, B. S. Reddy and A. Prasad, *Synlett*, 2009, **2009**, 790–792.
- 361 V. Bernan, M. Greenstein and W. Maiese, in *Advances in applied microbiology*, Elsevier, 1997, vol. 43, pp. 57–90.
- 362 K. Rajesh, V. Suresh, J. J. P. Selvam, C. Bhujanga Rao and Y. Venkateswarlu, *Helv. Chim. Acta*, 2009, **92**, 1866–1872.
- 363 S. Lai, Y. Shizuri, S. Yamamura, K. Kawai and H. Furukawa, *Bull. Chem. Soc. Jpn.*, 1991, **64**, 1048–1050.
- 364 J. He, E. K. Wijeratne, B. P. Bashyal, J. Zhan, C. J. Seliga, M. X. Liu, E. E. Pierson, L. S. Pierson, H. D. VanEtten and A. L. Gunatilaka, *J. Nat. Prod.*, 2004, 67, 1985–1991.
- 365 K. Rajesh, V. Suresh, J. J. P. Selvam, D. C. Babu and Y. Venkateswarlu, *Helv. Chim. Acta*, 2010, **93**, 147–152.
- 366 Y. Takaya and M. Niwa, *Trends Heterocycl. Chem.*, 2001, 7, 41–54.
- 367 T. Shen, X.-N. Wang and H.-X. Lou, *Nat. Prod. Rep.*, 2009, 26, 916–935.
- 368 U. Samaraweera, S. Sotheeswaran and M. U. S. Sultanbawa, *Phytochemistry*, 1982, **21**, 2585–2587.
- 369 H. Kurihara, J. Kawabata, S. Ichikawa, M. Mishima and J. Mizutani, *Phytochemistry*, 1991, **30**, 649–653.
- 370 M. Ohyama, T. Tanaka, T. Ito, M. Iinuma, K. F. Bastow and K.-H. Lee, *Bioorg. Med. Chem. Lett.*, 1999, **9**, 3057–3060.
- 371 K.-S. Huang, M. Lin and G.-F. Cheng, *Phytochemistry*, 2001, 58, 357–362.
- 372 L. D. Juliawaty, E. H. Hakim, S. A. Achmad, Y. M. Syah, J. Latip and I. M. Said, *Nat. Prod. Commun.*, 2009, 4, 947– 950.
- 373 H. M. Ge, W. H. Yang, Y. Shen, N. Jiang, Z. K. Guo, Q. Luo, Q. Xu, J. Ma and R. X. Tan, *Chem.-Eur. J.*, 2010, **16**, 6338– 6345.
- 374 K. Kim and I. Kim, Org. Lett., 2010, 12, 5314-5317.
- 375 T.-L. Ho, Tandem organic reactions, John Wiley & Sons, 1992.
- 376 L. F. Tietze, G. Brasche and K. Gericke, *Domino reactions in organic synthesis*, John Wiley & Sons, 2006.
- 377 L. F. Tietze and U. Beifuss, Angew. Chem., Int. Ed. Engl., 1993, 32, 131–163.
- 378 I. Kim, S. G. Kim, J. Choi and G. H. Lee, *Tetrahedron*, 2008, 64, 664–671.
- 379 S. J. Gould, Chem. Rev., 1997, 97, 2499-2510.
- 380 J. Marco-Contelles and M. T. Molina, *Curr. Org. Chem.*, 2003, 7, 1433–1442.
- 381 S. Ito, T. Matsuya, S. Omura, M. Otani, A. Nakagawa, H. Takeshima, Y. Iwai, M. Ohtani and T. Hata, *J. Antibiot.*, 1970, 23, 315–317.

- 382 T. Hata, S. Omura, Y. Iwai, A. Nakagawa, M. Otani, S. Ito and T. Matsuya, *J. Antibiot.*, 1971, **24**, 353–359.
- 383 S. J. Gould, N. Tamayo, C. R. Melville and M. C. Cone, J. Am. Chem. Soc., 1994, 116, 2207–2208.
- 384 S. J. Gould, J. Chen, M. C. Cone, M. P. Gore, C. R. Melville and N. Tamayo, *J. Org. Chem.*, 1996, **61**, 5720–5721.
- 385 P. J. Proteau, Y. Li, J. Chen, R. T. Williamson, S. J. Gould, R. S. Laufer and G. I. Dmitrienko, *J. Am. Chem. Soc.*, 2000, 122, 8325–8326.
- 386 X. Lei and J. A. Porco, *J. Am. Chem. Soc.*, 2006, **128**, 14790–14791.
- 387 H. Ishii, I.-S. Chen, S. Ueki, T. Masuda, K. Morita and T. Ishikawa, *J. Chem. Soc., Perkin Trans.* 1, 1987, 2415–2420.
- 388 S. Kimura, S. Kobayashi, T. Kumamoto, A. Akagi, N. Sato and T. Ishikawa, *Helv. Chim. Acta*, 2011, **94**, 578–591.
- 389 H.-C. Lin, S.-C. Chang, N.-L. Wang and L.-R. Ceng, J. Antibiot., 1994, 47, 675–680.
- 390 S. S. Scully and J. A. Porco Jr, *Angew. Chem., Int. Ed.*, 2011, **50**, 9722–9726.
- 391 R. J. Capon, J. K. Macleod and P. J. Scammells, *Tetrahedron*, 1986, 42, 6545–6550.
- 392 W. Liu, H. J. Lim and T. RajanBabu, J. Am. Chem. Soc., 2012, 134, 5496–5499.
- 393 Z. Jin, Nat. Prod. Rep., 2011, 28, 1126-1142.
- 394 A. Kornienko and A. Evidente, *Chem. Rev.*, 2008, **108**, 1982– 2014.
- 395 G. R. Pettit, G. M. Cragg, S. B. Singh, J. A. Duke and D. L. Doubek, *J. Nat. Prod.*, 1990, 53, 176–178.
- 396 S. Hwang, D. Kim and S. Kim, *Chem.–Eur. J.*, 2012, **18**, 9977–9982.
- 397 H. G. Capraro, A. Brossi, Tropolonic colchicum alkaloids, in *The Alkaloids: Chemistry and Pharmacology*, Academic Press, 1984, vol. 23, pp. 1–70.
- 398 T. Graening and H. G. Schmalz, *Angew. Chem.*, 2004, **116**, 3292–3318.
- 399 K. W. Wood, W. D. Cornwell and J. R. Jackson, *Curr. Opin. Pharmacol.*, 2001, **1**, 370–377.
- 400 O. Boyé and A. Brossi, in *The Alkaloids: Chemistry and Pharmacology*, Elsevier, 1992, vol. 41, pp. 125–176.
- 401 G. C. Tron, T. Pirali, G. Sorba, F. Pagliai, S. Busacca and A. A. Genazzani, *J. Med. Chem.*, 2006, **49**, 3033–3044.
- 402 C. Rappl, P. Barbier, V. Bourgarel-Rey, C. Grégoire, R. Gilli, M. Carre, S. Combes, J.-P. Finet and V. Peyrot, *Biochemistry*, 2006, 45, 9210–9218.
- 403 D. G. I. Kingston, J. Nat. Prod., 2009, 72, 507-515.
- 404 O. G. Ganina, E. Daras, V. Bourgarel-Rey, V. Peyrot, A. N. Andresyuk, J.-P. Finet, A. Y. Fedorov, I. P. Beletskaya and S. Combes, *Bioorg. Med. Chem.*, 2008, **16**, 8806–8812.
- 405 N. Sitnikov, J. Velder, L. Abodo, N. Cuvelier, J. Neudörfl,
  A. Prokop, G. Krause, A. Y. Fedorov and H. G. Schmalz, *Chem.-Eur. J.*, 2012, 18, 12096–12102.
- 406 R. Ratnayake, E. Lacey, S. Tennant, J. H. Gill and R. J. Capon, *Chem.-Eur. J.*, 2007, **13**, 1610–1619.
- 407 K. Oda, N. Nishizono and Y. Tamai, *Heterocycles*, 2005, **65**, 1985–1988.
- 408 D. K. Winter, M. A. Endoma-Arias, T. Hudlicky, J. A. Beutler and J. A. Porco Jr, *J. Org. Chem.*, 2013, **78**, 7617–7626.

- 410 C. C. Hughes, A. K. Miller and D. Trauner, *Org. Lett.*, 2005, 7, 3425–3428.
- 411 A. K. Miller, C. C. Hughes, J. J. Kennedy-Smith, S. N. Gradl and D. Trauner, *J. Am. Chem. Soc.*, 2006, **128**, 17057–17062.
- 412 P. A. Roethle, P. T. Hernandez and D. Trauner, *Org. Lett.*, 2006, **8**, 5901–5904.
- 413 A. D. Rodríguez, C. Ramírez and I. I. Rodríguez, *Tetrahedron Lett.*, 1999, **40**, 7627–7631.
- 414 I. T. Chen, I. Baitinger, L. Schreyer and D. Trauner, *Org. Lett.*, 2013, **16**, 166–169.
- 415 Q. Zhang, Y.-T. Di, C.-S. Li, X. Fang, C.-J. Tan, Z. Zhang, Y. Zhang, H.-P. He, S.-L. Li and X.-J. Hao, *Org. Lett.*, 2009, 11, 2357–2359.
- 416 H. Morita and J. i. Kobayashi, Org. Lett., 2003, 5, 2895–2898.
- 417 R. Yamada, Y. Adachi, S. Yokoshima and T. Fukuyama, *Angew. Chem.*, 2016, **128**, 6171–6174.
- 418 Q. Liang, Q. Wu, J. Jiang, J. a. Duan, C. Wang, M. D. Smith,
  H. Lu, Q. Wang, P. Nagarkatti and D. Fan, *J. Biol. Chem.*,
  2011, 286, 26470–26479.
- 419 M. Wang, L. Xiu, J. Diao, L. Wei and J. Sun, *Eur. J. Pharmacol.*, 2015, **769**, 79–85.
- 420 A. Oleinik and E. Adamskaya, *Chem. Heterocycl. Compd.*, 1983, **19**, 1221–1224.
- 421 S. Nishigaki, Y. Shibata and K. Tanaka, J. Org. Chem., 2017, 82, 11117–11125.
- 422 A. Stoll, J. Renz and A. Von Wartburg, *Helv. Chim. Acta*, 1954, **37**, 1747–1762.
- 423 A. Von Wartburg, E. Angliker and J. Renz, *Helv. Chim. Acta*, 1957, **40**, 1331–1357.
- 424 W. J. Gensler, C. M. Samour, S. Y. Wang and F. Johnson, *J. Am. Chem. Soc.*, 1960, **82**, 1714–1727.
- 425 P. M. Dewick, *The Flavonoids: Advances in Research*, ed. J. B. Harborne and T. J. Mabry, Chapman and Hall, London, 1982, p. 535.
- 426 A. Jain, A. Kumar and A. Kohli, *J. Sci. Ind. Res.*, 1978, 37, 606–621.
- 427 M. Darbarwar, V. Sundaramurthy and N. S. Rao, *J. Sci. Ind. Res.*, 1976, **35**, 297–312.
- 428 O. Stamm, H. Schmid and J. Büchi, *Helv. Chim. Acta*, 1958, **41**, 2006–2021.
- 429 A. Pelter and S. Foot, Synthesis, 1976, 1976, 326.
- 430 R. J. Bass, J. Chem. Soc., Chem. Commun., 1976, 78-79.
- 431 P. F. Schuda and W. A. Price, *J. Org. Chem.*, 1987, **52**, 1972–1979.
- 432 T. Easterfield and J. McDowell, *Trans. R. Soc. N. Z.*, 1915, **48**, 518–520.
- 433 L. Mangoni and R. Caputo, *Tetrahedron Lett.*, 1967, **8**, 673–675.
- 434 S. Das, S. Bhattachryya and D. Mukherjee, *Tetrahedron*, 1992, **48**, 9101–9110.
- 435 C. Giordano, M. Villa and R. Annunziata, *Synth. Commun.*, 1990, **20**, 383–392.
- 436 H. Minlon, J. Am. Chem. Soc., 1946, 68, 2487-2488.

- 437 M. Ichimura, K. Muroi, K. Asano, I. Kawamoto, F. Tomita, M. Morimoto and H. Nakano, J. Antibiot., 1988, 41, 1285– 1288.
- 438 T. Yasuzawa, T. Iida, K. i. Muroi, M. Ichimura, K. Takahashi and H. Sano, *Chem. Pharm. Bull.*, 1988, **36**, 3728–3731.
- 439 I. Takahashi, K.-I. Takahashi, M. Ichimura, M. Morimoto,K. Asano, I. Kawamoto, F. Tomita and H. Nakano, *J. Antibiot.*, 1988, 41, 1915–1917.
- 440 Y. Fukuda, Y. Itoh, K. Nakatani and T. Shiro, *Tetrahedron*, 1994, **50**, 2793–2808.
- 441 J. i. Kobayashi, C.-m. Zeng, M. Ishibashi, H. Shigemori, T. Sasaki and Y. Mikami, J. Chem. Soc., Perkin Trans. 1, 1992, 1291–1294.
- 442 C. Galli, Synthesis, 1979, 1979, 303-304.
- 443 F. Bracher and T. Papke, Nat. Prod. Lett., 1994, 4, 223-226.
- 444 U. Gräfe, W. Ihn, D. Tresselt, N. Miosga, U. Kaden, B. Schlegel, E.-J. Bormann, P. Sedmera and J. Novák, *Biol. Met.*, 1990, 3, 39–44.
- 445 W. Werner, U. Gräfe, W. Ihn, D. Tresselt, S. Winter and E. Paulus, *Tetrahedron*, 1997, **53**, 109–118.
- 446 Y. Kashman, K. R. Gustafson, R. Fuller, J. McMahon, M. Currens, J. R. Buckheit, S. Hughes, G. Cragg and M. Boyd, *J. Med. Chem.*, 1992, 35, 2735–2743.
- 447 J. Zhang, E. W. Kirchhoff, D. E. Zembower, N. Jimenez, P. Sen, Z.-Q. Xu and M. T. Flavin, *Org. Process Res. Dev.*, 2000, 4, 577–580.
- 448 N. Chikao, E. Nobuyasu, T. Shinkichi, M. Akihisa, K. Koji and F. Masako, *Biol. Chem.*, 1987, **51**, 139.
- 449 M. Joyeux, A. Lobstein, R. Anton and F. Mortier, *Planta Med.*, 1995, **61**, 126–129.
- 450 M. R. Fesen, K. W. Kohn, F. Leteurtre and Y. Pommier, *Proc. Natl. Acad. Sci. U. S. A.*, 1993, **90**, 2399–2403.
- 451 Y. H. Kuo and M. H. Yeh, J. Chin. Chem. Soc., 1997, 44, 379– 383.
- 452 T.-C. Lin, J.-M. Fang and Y.-S. Cheng, *Phytochemistry*, 1999, **51**, 793–801.
- 453 E. J. Park, Y. Kim and J. Kim, J. Nat. Prod., 2000, 63, 34-36.
- 454 J. Hempel and H. Böhm, *J. Agric. Food Chem.*, 1996, 44, 2114–2116.
- 455 T. Kartnig, A. Gruber and J. Stachel, *Planta Med.*, 1985, **51**, 288.
- 456 F. Kader, B. Rovel, M. Girardin and M. Metche, *Food Chem.*, 1996, **55**, 35–40.
- 457 A. Freiburghaus, H. Ha, J. Chen, P. Leuenberger and F. Follath, *Eur. J. Clin. Pharmacol.*, 1995, **48**, 367–371.
- 458 M. G. Hertog, P. C. Hollman and B. Van de Putte, *J. Agric. Food Chem.*, 1993, **41**, 1242–1246.
- 459 F. Ferreres, F. A. Tomáas-Barberáan, M. I. Gil and F. Tomáas-Lorente, *J. Sci. Food Agric.*, 1991, **56**, 49–56.
- 460 Y. J. Lee and T. D. Wu, J. Chin. Chem. Soc., 2001, 48, 201–206.
- 461 G. Bringmann, D. Menche, M. Bezabih, B. M. Abegaz and R. Kaminsky, *Planta Med.*, 1999, 65, 757–758.
- 462 E. Dagne and W. Steglich, *Phytochemistry*, 1984, **23**, 1729–1731.

- 463 G. Bringmann, D. Menche, J. Kraus, J. Mühlbacher,
  K. Peters, E.-M. Peters, R. Brun, M. Bezabih and
  B. M. Abegaz, *J. Org. Chem.*, 2002, 67, 5595–5610.
- 464 D. C. Harrowven and R. F. Dainty, *Tetrahedron Lett.*, 1996, 37, 7659–7660.
- 465 A. Rudi, T. Evan, M. Aknin and Y. Kashman, *J. Nat. Prod.*, 2000, **63**, 832–833.
- 466 S. Loya, A. Rudi, Y. Kashman and A. Hizi, *Biochem. J.*, 1999, 344, 85.
- 467 P. Kleinwaechter, B. Schlegel, I. Groth, A. Haertl and U. Graefe, *J. Antibiot.*, 2001, **54**, 510–512.
- 468 T. Komoda, Y. Shinoda and S.-i. Nakatsuka, *Biosci., Biotechnol., Biochem.*, 2003, **67**, 659–662.
- 469 S.-i. Nakatsuka, K. Teranishi and T. Goto, *Tetrahedron Lett.*, 1994, **35**, 2699–2700.
- 470 W.-H. Lin, J.-M. Fang and Y.-S. Cheng, *Phytochemistry*, 1996, 42, 1657–1663.
- 471 S. Tang, Y. Xu, J. He, Y. He, J. Zheng, X. Pan and X. She, *Org. Lett.*, 2008, **10**, 1855–1858.
- 472 R. M. McFadden and B. M. Stoltz, *J. Am. Chem. Soc.*, 2006, **128**, 7738–7739.
- 473 A. R. Saltiel and C. R. Kahn, Nature, 2001, 414, 799.
- 474 M. Harig, B. Neumann, H. G. Stammler and D. Kuck, *Eur. J.* Org. Chem., 2004, **2004**, 2381–2397.
- 475 J. Li, S. J. Guo, H. Su, L. J. Han and D. Y. Shi, *Chin. Chem. Lett.*, 2008, **19**, 1290–1292.
- 476 M. A. Elban and S. M. Hecht, *J. Org. Chem.*, 2008, 73, 785–793.
- 477 J. Savard and P. Brassard, Tetrahedron, 1984, 40, 3455-3464.
- 478 J. P. Michael, Nat. Prod. Rep., 2001, 18, 520-542.
- 479 E. Gellert, J. Nat. Prod., 1982, 45, 50-73.
- 480 A. G. Damu, P.-C. Kuo, L.-S. Shi, C.-Y. Li, C.-S. Kuoh, P.-L. Wu and T.-S. Wu, *J. Nat. Prod.*, 2005, **68**, 1071–1075.
- 481 P.-L. Wu, K. Rao, C.-H. Su, C.-S. Kuoh and T.-S. Wu, *Heterocycles*, 2002, **57**, 2401–2408.
- 482 R. S. Gupta and L. Siminovitch, *Biochemistry*, 1977, 16, 3209–3214.
- 483 E. Gellert and R. Rudzats, J. Med. Chem., 1964, 7, 361-362.
- 484 T.-y. An, R.-q. Huang, Z. Yang, D.-k. Zhang, G.-r. Li, Y.-C. Yao and J. Gao, *Phytochemistry*, 2001, **58**, 1267–1269.
- 485 Z.-Q. Huang, Y.-X. Liu, Z.-J. Fan, Q.-M. Wang, G.-R. Li, Y.-C. Yao, X.-S. Yu and R.-Q. Huang, *Fine Chem. Intermed.*, 2007, **37**, 20–24.
- 486 K.-L. Wang, M.-Y. Lü, Q.-M. Wang and R.-Q. Huang, *Tetrahedron*, 2008, **64**, 7504–7510.
- 487 B. S. Davidson, Chem. Rev., 1993, 93, 1771-1791.
- 488 R. J. Andersen, D. J. Faulkner, C. H. He, G. D. Van Duyne and J. Clardy, *J. Am. Chem. Soc.*, 1985, **107**, 5492–5495.
- 489 R. A. Davis, A. R. Carroll, G. K. Pierens and R. J. Quinn, J. Nat. Prod., 1999, 62, 419–424.
- 490 A. Carroll, B. Bowden and J. Coll, *Aust. J. Chem.*, 1993, **46**, 489–501.
- 491 J. S. Yadav, K. U. Gayathri, B. V. S. Reddy and A. R. Prasad, *Synlett*, 2009, **2009**, 43–46.
- 492 G. Baddeley, S. Makar and M. Ivinson, *J. Chem. Soc.*, 1953, 3969–3971.

- 493 K. Winkelmann, J. Heilmann, O. Zerbe, T. Rali and O. Sticher, *J. Nat. Prod.*, 2000, **63**, 104–108.
- 494 K. Winkelmann, J. Heilmann, O. Zerbe, T. Rali and O. Sticher, *Helv. Chim. Acta*, 2001, **84**, 3380–3392.
- 495 N. S. Simpkins and M. D. Weller, *Tetrahedron Lett.*, 2010, 51, 4823–4826.
- 496 N. Matsumoto, T. Tsuchida, M. Maruyama, R. Sawa,
  N. Kinoshita, Y. Homma, Y. Takahashi, H. Iinuma,
  H. Naganawa and T. Sawa, *J. Antibiot.*, 1996, 49, 953–954.
- 497 N. Matsumoto, T. Tsuchida, M. Maruyama, N. Kinoshita, Y. Homma, H. Iinuma, T. Sawa, M. Hamada, T. Takeuchi and N. Heida, *J. Antibiot.*, 1999, **52**, 269–275.
- 498 K. Tatsuta, H. Tanaka, H. Tsukagoshi, T. Kashima and S. Hosokawa, *Tetrahedron Lett.*, 2010, **51**, 5546–5549.
- 499 M. E. Jung and J. A. Hagenah, J. Org. Chem., 1987, 52, 1889– 1902.
- 500 A. D. Patten, N. N. Huy and S. J. Danishefsky, *J. Org. Chem.*, 1988, **53**, 1003–1007.
- 501 Y. Kashman, A. Rotstein and A. Lifshitz, *Tetrahedron*, 1974, 30, 991–997.
- 502 R. Mani and K. Venkataraman, Curr. Sci., 1954, 23, 220-221.
- 503 N. Gokan, H. Kikuchi, K. Nakamura, Y. Oshima, K. Hosaka and Y. Kubohara, *Biochem. Pharmacol.*, 2005, **70**, 676–685.
- 504 W. Crow, T. Osawa, K. Platz and D. Sutherland, *Aust. J. Chem.*, 1976, **29**, 2525–2531.
- 505 M. Bolte, W. Crow and S. Yoshida, *Aust. J. Chem.*, 1982, 35, 1411–1419.
- 506 H. Müller, M. Paul, D. Hartmann, V. Huch, D. Blaesius, A. Koeberle, O. Werz and J. Jauch, *Angew. Chem., Int. Ed.*, 2010, **49**, 2045–2049.
- 507 J. P. Michael, Nat. Prod. Rep., 2005, 22, 603-626.
- 508 S. K. Lee, K.-A. Nam and Y.-H. Heo, *Planta Med.*, 2003, **69**, 21–25.
- 509 D. Stærk, J. Christensen, E. Lemmich, J. Ø. Duus, C. E. Olsen and J. W. Jaroszewski, *J. Nat. Prod.*, 2000, 63, 1584–1586.
- 510 D. Stærk, A. K. Lykkeberg, J. Christensen, B. A. Budnik, F. Abe and J. W. Jaroszewski, *J. Nat. Prod.*, 2002, **65**, 1299– 1302.
- 511 L. M. Ambrosini, T. A. Cernak and T. H. Lambert, *Tetrahedron*, 2010, **66**, 4882–4887.
- 512 S. F. Brady, M. M. Wagenaar, M. P. Singh, J. E. Janso and J. Clardy, *Org. Lett.*, 2000, 2, 4043–4046.
- 513 J. D. Hall, N. W. Duncan-Gould, N. A. Siddiqi, J. N. Kelly, L. A. Hoeferlin, S. J. Morrison and J. K. Wyatt, *Bioorg. Med. Chem.*, 2005, **13**, 1409–1413.
- 514 T. Ohzeki and K. Mori, *Biosci., Biotechnol., Biochem.*, 2003, 67, 2584–2590.
- 515 Y. Zhan, X. Du, H. Chen, J. Liu, B. Zhao, D. Huang, G. Li, Q. Xu, M. Zhang and B. C. Weimer, *Nat. Chem. Biol.*, 2008, 4, 548.
- 516 X. H. Cheng and C. J. Fu, *Chin. J. Chem.*, 2007, **25**, 1762–1765.
- 517 F. Wang, Y. J. Zhang, H. Wei, J. Zhang and W. Zhang, *Tetrahedron Lett.*, 2007, **48**, 4083–4086.
- 518 Y. Kan, T. Fujita, H. Nagai, B. Sakamoto and Y. Hokama, *J. Nat. Prod.*, 1998, **61**, 152–155.

# View Article Online RSC Advances

- 519 J. Chen, Z.-F. Shi, L. Zhou, A.-L. Xie and X.-P. Cao, *Tetrahedron*, 2010, **66**, 3499–3507.
- 520 T. Toyoda, K. Sasakura and T. Sugasawa, J. Org. Chem., 1981, 46, 189–191.
- 521 L. Xu, J. Jin, M. Lal, P. Daublain and M. Newcomb, Org. Lett., 2007, 9, 1837–1840.
- 522 R. Ciochina and R. B. Grossman, *Chem. Rev.*, 2006, **106**, 3963–3986.
- 523 C. C. Hughes, A. Prieto-Davo, P. R. Jensen and W. Fenical, *Org. Lett.*, 2008, **10**, 629–631.
- 524 K. Nicolaou, N. L. Simmons, J. S. Chen, N. M. Haste and V. Nizet, *Tetrahedron Lett.*, 2011, **52**, 2041–2043.
- 525 R. H. Furneaux and P. C. Tyler, J. Org. Chem., 1999, 64, 8411-8412.
- 526 R.-J. Lin, M.-J. Cheng, J.-C. Huang, W.-L. Lo, Y.-T. Yeh, C.-M. Yen, C.-M. Lu and C.-Y. Chen, *J. Nat. Prod.*, 2009, 72, 1816–1824.
- 527 C. Tang, Z. Li, Y. Wang, J. Xu, L. Kong, H. Yao and X. Wu, *Tetrahedron Lett.*, 2011, **52**, 3275–3278.
- 528 K. Shin-ya, K. Wierzba, K.-i. Matsuo, T. Ohtani, Y. Yamada, K. Furihata, Y. Hayakawa and H. Seto, *J. Am. Chem. Soc.*, 2001, **123**, 1262–1263.
- 529 N. S. Girgis, H. B. Cottam and R. K. Robins, *J. Heterocycl. Chem.*, 1988, **25**, 361–366.
- 530 T. Sasaki, A. Nakanishi and M. Ohno, J. Org. Chem., 1982, 47, 3219–3224.
- 531 H. Tokuyama, K. Okano, H. Fujiwara, T. Noji and T. Fukuyama, *Chem.–Asian J.*, 2011, **6**, 560–572.
- 532 D. H. Miles, D. S. Lho, A. A. De la Cruz, E. D. Gomez, J. A. Weeks and J. L. Atwood, *J. Org. Chem.*, 1987, 52, 2930–2932.
- 533 D. H. Miles, A.-M. Ly, V. Chittawong, A. A. de la Cruz and
  E. D. Gomez, *J. Nat. Prod.*, 1989, 52, 896–898.
- 534 S. P. Chavan, S. Garai and U. R. Kalkote, *Tetrahedron*, 2012, 68, 8509–8514.
- 535 Y.-C. Hu, X.-F. Wu, S. Gao, S.-S. Yu, Y. Liu, J. Qu, J. Liu and Y.-B. Liu, *Org. Lett.*, 2006, **8**, 2269–2272.
- 536 X.-F. Wu, Y.-C. Hu, S. Gao, S.-S. Yu, Y.-H. Pei, W.-Z. Tang and X.-Z. Huang, *J. Asian Nat. Prod. Res.*, 2007, **9**, 471–477.
- 537 X.-F. Wu, Y.-C. Hu, S.-S. Yu, N. Jiang, J. Ma, R.-X. Tan, Y. Li, H.-N. Lv, J. Liu and S.-G. Ma, *Org. Lett.*, 2010, **12**, 2390–2393.
- 538 Y. Ogura, K. Ishigami and H. Watanabe, *Tetrahedron*, 2012, 68, 1723–1728.
- 539 C. Dufour, J. Wink, M. Kurz, H. Kogler, H. Olivan, S. Sablé,
  W. Heyse, M. Gerlitz, L. Toti and A. Nußer, *Chem.-Eur. J.*, 2012, 18, 16123–16128.
- 540 C. Couturier, A. Bauer, A. Rey, C. Schroif-Dufour and M. Broenstrup, *Bioorg. Med. Chem. Lett.*, 2012, **22**, 6292– 6296.
- 541 T. Zhao, G. R. Yan, S. L. Pan, H. Y. Wang and A. J. Hou, *Chem. Biodiversity*, 2009, **6**, 2209–2216.
- 542 E. T. Arung, K. Shimizu and R. Kondo, *Chem. Biodiversity*, 2007, 4, 2166–2171.
- 543 A.-R. Han, Y.-J. Kang, T. Windono, S. K. Lee and E.-K. Seo, *J. Nat. Prod.*, 2006, **69**, 719–721.
- 544 E. T. Arung, K. Shimizu and R. Kondo, *Planta Med.*, 2006, 72, 847–850.

- 545 W.-J. Zhang, J.-F. Wu, P.-F. Zhou, Y. Wang and A.-J. Hou, *Tetrahedron*, 2013, **69**, 5850–5858.
- 546 F. Fullas, L. J. Kornberg, M. C. Wani, M. E. Wall, N. R. Farnsworth, T. E. Chagwedera and A. D. Kinghorn, *J. Nat. Prod.*, 1996, **59**, 190–192.
- 547 C. Selenski and T. R. Pettus, *J. Org. Chem.*, 2004, **69**, 9196–9203.
- 548 I. Slabu, S. B. Rossington, P. M. Killoran, N. Hirst and J. A. Wilkinson, *Tetrahedron Lett.*, 2013, 54, 1489–1490.
- 549 G. A. Mohamed and S. R. Ibrahim, *ARKIVOC*, 2007, **15**, 281–291.
- 550 M. I. Choudhary, N. Khan, M. Ahmad, S. Yousuf, H.-K. Fun, S. Soomro, M. Asif, M. A. Mesaik and F. Shaheen, *Org. Lett.*, 2013, **15**, 1862–1865.
- 551 A. Hiranrat and W. Mahabusarakam, *Tetrahedron*, 2008, **64**, 11193–11197.
- 552 M. Morkunas, L. Dube and M. E. Maier, *Tetrahedron*, 2013, 69, 8559–8563.
- 553 P. Kalsi, J. Singh, W. Crow and B. Chhabra, *Phytochemistry*, 1987, **26**, 3367–3369.
- 554 H. Yeo, J. H. Lee and J. Kim, *Arch. Pharmacal Res.*, 1999, 22, 306.
- 555 Y. Tezuka, M. Terazono, T. Kusumoto, I. Tomoco, Y. Hatanaka, S. Kadota, M. Hattori and O. Namba, *Helv. Chim. Acta*, 2000, **83**, 29.
- 556 T. Hayashi and R. H. Thomson, *Phytochemistry*, 1975, 14, 1085–1087.
- 557 S. Apers, A. Vlietinck and L. Pieters, *Phytochem. Rev.*, 2003, 2, 201–217.
- 558 P. S. Luize, T. Ueda-Nakamura, B. P. Dias Filho, D. A. G. Cortez and C. V. Nakamura, *Biol. Pharm. Bull.*, 2006, 29, 2126–2130.
- 559 L. Baumgartner, S. Sosa, A. G. Atanasov, A. Bodensieck, N. Fakhrudin, J. Bauer, G. D. Favero, C. Ponti, E. H. Heiss and S. Schwaiger, *J. Nat. Prod.*, 2011, 74, 1779–1786.
- 560 C.-y. Chen and M. Weisel, Synlett, 2013, 24, 189-192.
- 561 X. Zhang, X. Jiang, C. Ding, Q. Yao and A. Zhang, Org. Biomol. Chem., 2013, 11, 1383–1389.
- 562 F. Zhu and Y. Lin, Chin. Sci. Bull., 2006, 51, 1426.
- 563 C.-L. Feng, S.-G. Zhang, J.-Q. Chen, J. Cai and M. Ji, *Chin. Chem. Lett.*, 2013, 24, 767–769.
- 564 J. S. Yadav, B. V. S. Reddy, G. Kondaji, R. Srinivasa Rao and S. Praveen Kumar, *Tetrahedron Lett.*, 2002, **43**, 8133–8135.
- 565 A. Merz, R. Schropp and E. Dötterl, *Synthesis*, 1995, **1995**, 795–800.
- 566 R. D. Stipanovic, J. Zhang, B. D. Bruton and M. H. Wheeler, *J. Agric. Food Chem.*, 2004, **52**, 4109–4112.
- 567 K. A. Punch and M. J. Piggott, *Org. Biomol. Chem.*, 2014, **12**, 2801–2810.
- 568 B. J. Albert and K. Koide, J. Org. Chem., 2008, 73, 1093–1098.
- 569 G. Pickaert, M. Cesario and R. Ziessel, J. Org. Chem., 2004, 69, 5335–5341.
- 570 M. J. Piggott and D. Wege, Aust. J. Chem., 2003, 56, 691-702.
- 571 H. Fujimoto, H. Okuyama, Y. Motohashi, E. Yoshida and M. Yamazaki, *JSM Mycotoxins*, 1995, **1995**, 61–66.

- 572 J. Wang, S. M. Soisson, K. Young, W. Shoop, S. Kodali, A. Galgoci, R. Painter, G. Parthasarathy, Y. S. Tang and R. Cummings, *Nature*, 2006, 441, 358.
- 573 G. Bartoli, M. Locatelli, P. Melchiorre and L. Sambri, *Eur. J. Org. Chem.*, 2007, **200**7, 2037–2049.
- 574 S. T. C. Eey and M. J. Lear, *Chem.–Eur. J.*, 2014, **20**, 11556–11573.
- 575 S. Suryati, Phytochemicals and Anti-inflammatory Activity of Melicope ptelefolia Champ Ex, Benth Masters's Thesis, Universiti Putra Malaysia, Malaysia, 2005.
- 576 K. Shaari, S. Safri, F. Abas, N. H. Lajis and D. Israf, *Nat. Prod. Res.*, 2006, **20**, 415–419.
- 577 C. H. Ng, K. Rullah, M. F. F. M. Aluwi, F. Abas, K. W. Lam, I. S. Ismail, R. Narayanaswamy, F. Jamaludin and K. Shaari, *Molecules*, 2014, **19**, 11645–11659.
- 578 W. Reininger and A. Hartl, *US Pat.*, 4053517, U.S. Patent and Trademark Office, Washington, DC, 1977.
- 579 L. Kaysser, P. Bernhardt, S.-J. Nam, S. Loesgen, J. G. Ruby,
   P. Skewes-Cox, P. R. Jensen, W. Fenical and B. S. Moore,
   *J. Am. Chem. Soc.*, 2012, 134, 11988–11991.
- 580 H. P. Pepper and J. H. George, *Angew. Chem., Int. Ed.*, 2013, 52, 12170–12173.
- 581 R. Meier, S. Strych and D. Trauner, *Org. Lett.*, 2014, **16**, 2634–2637.
- 582 H. P. Pepper and J. H. George, Synlett, 2015, 26, 2485–2490.
- 583 J. Dai, K. Krohn, U. Flörke, G. Pescitelli, G. Kerti, T. Papp, K. E. Kövér, A. C. Bényei, S. Draeger and B. Schulz, *Eur. J.* Org. Chem., 2010, **2010**, 6928–6937.
- 584 R. W. Bates, K. Wang, G. Zhou and D. Z. Kang, Synlett, 2015, 26, 751–754.
- 585 I. P. Singh and S. B. Bharate, *Nat. Prod. Rep.*, 2006, **23**, 558–591.
- 586 I. P. Singh, J. Sidana, S. B. Bharate and W. J. Foley, *Nat. Prod. Rep.*, 2010, **27**, 393–416.
- 587 J. J. Brophy, D. C. Craig, R. J. Goldsack, C. J. Fookes,
  D. N. Leach and P. G. Waterman, *Phytochemistry*, 2006,
  67, 2085–2089.
- 588 G. Wei and B. Yu, Eur. J. Org. Chem., 2008, 2008, 3156-3163.
- 589 E. Nishimura, Y. Ohfune and T. Shinada, *Tetrahedron Lett.*, 2015, **56**, 539–541.
- 590 B. Wu, W. Zhang, Z. Li, L. Gu, X. Wang and P. G. Wang, *J. Org. Chem.*, 2011, **76**, 2265–2268.
- 591 A. Briot, C. Baehr, R. Brouillard, A. Wagner and C. Mioskowski, *J. Org. Chem.*, 2004, **69**, 1374–1377.
- 592 D. H. Silva, Y. Zhang, L. A. Santos, V. S. Bolzani and M. G. Nair, J. Agric. Food Chem., 2007, 55, 2569–2574.
- 593 P. W. Snijman, E. Joubert, D. Ferreira, X.-C. Li, Y. Ding, I. R. Green and W. C. Gelderblom, *J. Agric. Food Chem.*, 2009, 57, 6678–6684.
- 594 J. Somsrisa, P. Meepowpan, S. Krachodnok, H. Thaisuchat, S. Punyanitya, N. Nantasaen and W. Pompimon, *Molecules*, 2013, **18**, 6898–6907.
- 595 A. Hermoso, I. A. Jiménez, Z. A. Mamani, I. L. Bazzocchi, J. E. Piñero, A. G. Ravelo and B. Valladares, *Bioorg. Med. Chem.*, 2003, **11**, 3975–3980.
- 596 D. Ahmed, V. Kumar, M. Sharma and A. Verma, *BMC Complementary Altern. Med.*, 2014, 14, 155.

- 597 J. Zhang and Y. Qin, 2013 21st International Conference on Geoinformatics, 2013.
- 598 E. Lemmich, C. O. Adewunmi, P. Furu, A. Kristensen, L. Larsen and C. E. Olsen, *Phytochemistry*, 1996, 42, 1011– 1013.
- 599 E. Szliszka, Z. P. Czuba, B. Mazur, A. Paradysz and W. Krol, *Molecules*, 2010, **15**, 5336–5353.
- 600 J. Peng, A. L. Risinger, C. Da, G. A. Fest, G. E. Kellogg and S. L. Mooberry, *J. Nat. Prod.*, 2013, **76**, 2189–2194.
- 601 T. H. Sum, T. J. Sum, J. E. Stokes, W. R. Galloway and D. R. Spring, *Tetrahedron*, 2015, **71**, 4557–4564.
- 602 A. T. Tran, N. P. West, W. J. Britton and R. J. Payne, *ChemMedChem*, 2012, 7, 1031–1043.
- 603 D. Maes, M. E. Riveiro, C. Shayo, C. Davio, S. Debenedetti and N. De Kimpe, *Tetrahedron*, 2008, **64**, 4438–4443.
- 604 G. Keserü and M. Nógrádi, in *Studies in Natural Products Chemistry*, Elsevier, 1995, vol. 17, pp. 357–394.
- 605 P. Ruedi and M. Juch, Curr. Org. Chem., 1999, 3, 623-646.
- 606 C. Per, U. P. Claeson, P. Tuchinda and V. Reutrakul, in Studies in Natural Products Chemistry, Elsevier, 2002, vol. 26, pp. 881–908.
- 607 H. Lv and G. She, Nat. Prod. Commun., 2010, 5, 1687-1708.
- 608 H. Lv and G. She, Rec. Nat. Prod., 2012, 6, 321-333.
- 609 Y. Sun, K. Tabata, H. Matsubara, S. Kitanaka, T. Suzuki and K. Yasukawa, *Planta Med.*, 2008, 74, 427–431.
- 610 H. Seçinti and H. Seçen, *Helv. Chim. Acta*, 2015, **98**, 938-944.
- 611 G. Höfle, B. Kunze, C. Zorzin and H. Reichenbach, *Eur. J. Org. Chem.*, 1984, **1984**, 1883–1904.
- 612 D. Enders and S. Osborne, J. Chem. Soc., Chem. Commun., 1993, 424–426.
- 613 J. Yadav, G. Revathi and B. S. Reddy, *Tetrahedron Lett.*, 2017, 58, 3943–3946.
- 614 T. Horie, H. Tominaga, Y. Kawamura, T. Hada, N. Ueda,
  Y. Amano and S. Yamamoto, *J. Med. Chem.*, 1991, 34, 2169–2176.
- 615 B. D. Chapsal and I. Ojima, Org. Lett., 2006, 8, 1395-1398.
- 616 R. J. Huntley and R. L. Funk, *Tetrahedron Lett.*, 2011, 52, 6671–6674.
- 617 K. Tomooka, M. Suzuki, K. Uehara, M. Shimada and T. Akiyama, *Synlett*, 2008, **2008**, 2518–2522.
- 618 B. N. Do Doan, X. Y. Tan, C. M. Ang and R. W. Bates, *Synthesis*, 2017, **49**, 4711–4716.
- 619 R. W. Rickards, J. M. Rothschild, A. C. Willis, N. M. de Chazal, J. Kirk, K. Kirk, K. J. Saliba and G. D. Smith, *Tetrahedron*, 1999, 55, 13513–13520.
- 620 N. T. Doan, P. R. Stewart and G. D. Smith, *FEMS Microbiol. Lett.*, 2001, **196**, 135–139.
- 621 R. AI and N. A. Bumagin, J. Organomet. Chem., 1998, 560, 163–167.
- 622 B. M. Ramalingam and A. K. Mohanakrishnan, *Tetrahedron Lett.*, 2017, **58**, 2919–2922.
- 623 K. Ninomiya, K. Shibatani, M. Sueyoshi, S. Chaipech,Y. Pongpiriyadacha, T. Hayakawa, O. Muraoka andT. Morikawa, *Chem. Pharm. Bull.*, 2016, 64, 880–885.

- 624 G. Tanabe, N. Tsutsui, K. Shibatani, S. Marumoto, F. Ishikawa, K. Ninomiya, O. Muraoka and T. Morikawa, *Tetrahedron*, 2017, **73**, 4481–4486.
- 625 H. H. Kinfe, H. S. Long, M. A. Stander and B. E. Van Wyk, *S. Afr. J. Bot.*, 2015, **100**, 75–79.
- 626 Z. Dong, J. Guo, X. Xing, X. Zhang, Y. Du and Q. Lu, *Biomed. Pharmacother.*, 2017, **89**, 297–304.
- 627 K. Kawaguchi, S. de Mello Alves, T. Watanabe, S. Kikuchi,M. Satake and Y. Kumazawa, *Planta Med.*, 1998, 64, 653–655.
- 628 Y. Tao, Y. Jiang, W. Li and B. Cai, *Anal. Methods*, 2016, 8, 4211-4219.
- 629 S. Liu, C. Zhang, Q. Shi, G. Li, M. Song, Y. Gao, C. Xu, H. Xu,
  B. Fan, S. Yu, C. Zheng, Q. Zhu, B. Wu, L. Peng, H. Xiong,
  Q. Wu and S. Liang, *Brain Res. Bull.*, 2014, **101**, 57–63.
- 630 Y. Zou, S. Zhang, G. Wang, X. Wen, S. Liu, T. Peng, Y. Gao and L. Wang, *J. Carbohydr. Chem.*, 2016, **35**, 387–395.
- 631 J. F. W. McOmie and S. A. Saleh, *Tetrahedron*, 1973, **29**, 4003–4005.
- 632 N. Öztaşkın, Y. Çetinkaya, P. Taslimi, S. Göksu and İ. Gülçin, *Bioorg. Chem.*, 2015, **60**, 49–57.
- 633 D. Shi, J. Li, B. Jiang, S. Guo, H. Su and T. Wang, *Bioorg. Med. Chem. Lett.*, 2012, 22, 2827–2832.
- 634 Y. Zou, S. Zhang, X. Wen, S. Liu, T. Peng, Y. Gao and L. Wang, *Tetrahedron Lett.*, 2017, **58**, 2835–2837.
- 635 M. P. Mattson, Nat. Rev. Mol. Cell Biol., 2000, 1, 120.
- 636 H. Zhai, M. Nakatsukasa, Y. Mitsumoto and Y. Fukuyama, *Planta Med.*, 2004, **70**, 598–602.
- 637 T. Esumi, D. Hojyo, H. Zhai and Y. Fukuyama, *Tetrahedron Lett.*, 2006, **47**, 3979–3983.
- 638 K. C. Guerard, C. Sabot, L. Racicot and S. Canesi, *J. Org. Chem.*, 2009, **74**, 2039–2045.
- 639 T. Konishi, T. Konoshima, A. Daikonya and S. Kitanaka, *Biol. Pharm. Bull.*, 2005, **53**, 121–124.
- 640 S. Hanessian and G. J. Reddy, *Synlett*, 2007, **2007**, 0475–0479.

- 641 H. Zhai, T. Inoue, M. Moriyama, T. Esumi, Y. Mitsumoto and Y. Fukuyama, *Biol. Pharm. Bull.*, 2005, **28**, 289–293.
- 642 R.-L. Huang, C.-F. Chen, H.-Y. Feng, L.-C. Lin and C.-J. Chou, *J. Chin. Clin. Med.*, 2001, **12**, 179–192.
- 643 S. Aketani, K. Tanaka, K. Yamamoto, A. Ishihama, H. Cao, A. Tengeiji, S. Hiraoka, M. Shiro and M. Shionoya, *J. Med. Chem.*, 2002, 45, 5594–5603.
- 644 P. Xue, L.-P. Wang, X.-Z. Jiao, Y.-J. Jiang, Q. Xiao, Z.-G. Luo, P. Xie and X.-T. Liang, *J. Asian Nat. Prod. Res.*, 2009, **11**, 281– 287.
- 645 V. F. de Andrade-Neto, T. da Silva, L. M. X. Lopes, V. E. do Rosário, F. de Pilla Varotti and A. U. Krettli, *Antimicrob. Agents Chemother.*, 2007, **51**, 2346–2350.
- 646 J. C. Reddel, K. E. Lutz, A. B. Diagne and R. J. Thomson, *Angew. Chem.*, 2014, **126**, 1419–1422.
- 647 J. Kinjo, H. Uemura, T. Nohara, M. Yamashita, N. Marubayashi and K. Yoshihira, *Tetrahedron Lett.*, 1995, 36, 5599–5602.
- 648 S. Strych, G. Journot, R. P. Pemberton, S. C. Wang,
  D. J. Tantillo and D. Trauner, *Angew. Chem., Int. Ed.*, 2015, 54, 5079–5083.
- 649 N. J. Patron, R. F. Waller, A. J. Cozijnsen, D. C. Straney,
  D. M. Gardiner, W. C. Nierman and B. J. Howlett, *BMC Evol. Biol.*, 2007, 7, 174.
- 650 T. C. Adams, J. N. Payette, J. H. Cheah and M. Movassaghi, Org. Lett., 2015, 17, 4268–4271.
- 651 L. Ding, A. Maier, H.-H. Fiebig, W.-H. Lin and C. Hertweck, Org. Biomol. Chem., 2011, 9, 4029–4031.
- 652 Y. Sun, Z. Meng, P. Chen, D. Zhang, M. Baunach, C. Hertweck and A. Li, *Org. Chem. Front.*, 2016, **3**, 368–374.
- 653 Z. Liu, L. Liu, Z. Shafiq, D. Wang and Y.-J. Chen, *Lett. Org. Chem.*, 2007, 4, 256–260.
- 654 X. Zhu and A. Ganesan, J. Org. Chem., 2002, 67, 2705-2708.
- 655 A. Bourderioux, A. Ouach, V. Beneteau, J.-Y. Mérour and S. Routier, *Synthesis*, 2010, 2010, 783–790.