Uptake and distribution characteristics of the novel fungicide pyraoxystrobin in cucumber plants

Xunyue Liu,a † Xia Chen,b c Xingcheng Dingc and Huiming Wu a

The uptake and distribution characteristics of a novel fungicide, pyraoxystrobin, labeled with 14C on its pyrazol ring, were investigated in cucumber (Cucumis sativus) seedlings. Foliar applied pyraoxystrobin rapidly penetrated the treated leaf and reached a maximum uptake of 68% after 5 d. The translocation of absorbed 14C in cucumber seedlings was both acropetal and basipetal. However, over 74% of the absorbed 14C-pyraoxystrobin remained in the treated leaves. The order of its distribution in the plant was as follows: treated leaf > stalk above the treated leaf > leaves above the treated leaf > stalk below the treated leaf > leaves below the treated leaf > cotyledon > root. Seedlings grown in soils containing bound residues (BR) of pyraoxystrobin revealed that the BRs were not easily absorbed or translocated by the plant. Soil type had a large effect on root uptake, with the highest uptake among the three tested soils from red clay.

1. Introduction

Strobilurins are an important class of agricultural fungicidal compound. They are modeled after natural compounds isolated from several basidiomycete species that inhabit decaying plant materials in woodland soils in many countries.1,2 These compounds exhibit fungicidal activities by binding to the coenzyme Qo site of cytochrome b, which blocks the electron transfer between cytochrome b and cytochrome c1.3–5 In 2005, more than 10 strobilurin fungicides were commercially available; however, significant increases in resistance were observed in a range of important plant pathogens against the strobilurin class of fungicides after a short period of field applications.6,7 Thus, developing new strobilurin analogues to address the increasing demand of resistant pathogens has attracted the attention of agricultural chemists.

SYP-3343, ((E)-2-(2-((3-(4-chlorophenyl)-1-methyl-1H-pyrazole-5-yloxy)methyl)phenyl)-3-methoxyacrylate), is a novel pyraoxystrobin fungicide that was independently developed by the ShenYang Research Institute of Chemical Industry in China. It is highly effective and shows broad-spectrum activity against Pseudoperonospora cubensis, Blumeria graminis, Erysiphe cichoracearum, Plasmopara viticola and Magnaporthe grisea.8–10 It has obtained temporary registration status in China for cucumber (Cucumis sativus) downy mildew at an application rate of 80–100 g active ingredient per hectare.11 Some pesticides that are applied to the soil, along with their degradation products, are often easily bound to organic compounds and mineral constituents of the soils through physical and/or chemical interactions, forming the so-called bound residues (BRs).12,13 Soil BRs can be released again through the soil microorganism activities and physicochemical processes, and the released compounds may be available for uptake by plants and biotic communities.14 Based on the potentially hazardous impact of released BRs from many pollutants, such as atrazine, fenitrothion and tetrabromobisphenol A,15–17 it is necessary to gather more information regarding the bioactivity of BRs in the biota.

SYP-3343 is a novel fungicide against Pseudoperonospora cubensis for cucumber, which is an important vegetable for human consumption. Safety assessments must consider the absorption, translocation, release and uptake of BRs by cucumber plants. The environmental behavior of SYP-3343 in soils has been addressed;15,16 however, the absorption, translocation, and released BRs of SYP-3343 in plants remain unknown. The objectives of the present study were to evaluate the absorption and translocation of SYP-3343 from (1) foliar applications to cucumber plants and (2) cucumber seedlings grown in soils containing BRs, using the 14C isotopic tracer method under simulated field conditions.

2. Materials and methods

2.1 Materials

14C labeled SYP-3343 was synthesized in our laboratory with the help of the ShenYang Research Institute of Chemical Industry in China,17 and the chemical structure is shown in Fig. 1. The specific activity was 4.53 × 108 Bq g−1. The radiochemical purity
and chemical purity were both greater than 98%. The scintillation-grade cocktail ingredients 2,5-diphenyloxazole (POPO) and 1,4-di-[2’-(5’-phenyloxazolyl)]-benzene (POPOP) were purchased from Acros Organics (Geel, Belgium). Dimethyldibenzene, dichloromethane, acetone, methanol and ethyl acetate were analytical grade. The methanol, acetonitrile and acetic acid used for the chromatographic analyses were of chromatographic grade.

2.2 Leaf treatment experiment

Cucumber seeds were soaked in 30 °C water for 6 h and then germinated in Petri dishes containing filter papers moistened with distilled water that were incubated in an artificial weather box (28 °C; humidity, 80%, light/dark, 10 h/14 h). The seedlings were sown into flowerpots (15 cm × 11 cm × 9.5 cm) containing soil (pH 6.95, 20.98% silt, 36.32% sand and 42.70% clay contents) and then cultivated under greenhouse conditions (28/16 °C, day/night; humidity, 80%; light/dark, 10 h/14 h) until they had sprouted two leaves. Plant containers were watered manually as needed during the whole treatment period.

The treatment and analysis methods were modified based on the procedures used by Lycan et al. Three-leaf stage cucumber seedlings were selected for treatment. The second fully expanded leaf of selected plants was treated with 20 μL of 5 μg mL⁻¹ 14C-SYP-3343 solution containing 0.25% of the surfactant Tween-80 and the radioactive concentration was 83 kBq mL⁻¹. Prior to treatment, the soil surface of the flowerpots was covered with absorbent papers to prevent the solution from contaminating the soil during the treatment process. After treatment, the seedlings were cultivated under the previous conditions until sampling. Cucumber seedlings were harvested in batches of three plants at 0.25, 0.5, 1, 3, 5, 8 and 14 d after treatment (DAT). The treated leaves were excised from the plants and eluted with methanol for 30 s at a flow rate of 5.0 mL min⁻¹ using a Waters 600E pump (Waters Co., Massachusetts, USA) to remove the unabsorbed SYP-3343. The leachate was directly collected in 20 mL scintillation vials and mixed with 10 mL of scintillation cocktail A (5 g POPO, 0.5 g POPPOP, 650 mL of dimethyldibenzene and 350 mL of glycol-ether). The 14C radioactivity was measured using a liquid scintillation counter (LSC; Wallac 1414, Turku, Finland). Before formal experiments began, 14C SYP-3343 was applied to leaves and after 30 s sampled to evaluate the efficiency of our leaf-wash method.²⁹ The method removed ~98.5 ± 2.5% of the unabsorbed 14C-SYP-3343. The treated plants were divided into seven parts (the roots were washed in advance with tap water to remove soil): treated leaf (TL), leaves above the treated leaf (LATL), stalk above the treated leaf (SATL), leaves below the treated leaf (LBTL), stalk below the treated leaf (SBTL), cotyledon and roots. After drying to constant weights in an oven at 80 °C, all plant parts were combusted using a biological oxidizer (OX-600 Biological Oxidizer, R. J. Harvey Instrument Co., NJ, USA). The duration of combustion was 4.0 min, and the combustion temperature was maintained at 900 °C. Released 14CO₂ from the sample combustion, was absorbed in 20 mL of scintillation cocktail B (5 g POPO, 0.5 g POPPOP, 600 mL dimethyldibenzene, 225 mL glycol-ether and 175 mL ethanolamine). The radioactivity was determined using an LSC to quantify the absorbed 14C-SYP-3343 in plant tissues. The recovery efficiency was greater than 90% as determined by combusting the 14C standard compound. Foliar absorption was calculated by dividing the amount of 14C recovered from the analyzed plant part by the applied 14C activity. Absorption of 14C in cucumber plants was expressed as the percentage of activity in a given tissue over the total absorbed radioactivity. Additionally, two seedlings, which had been subjected to 4.22 (±0.01) × 10³ Bq of applied 14C-SYP-3343 were sampled at 8 DAT. The roots were washed with tap water, and the TL was washed with methanol. The whole plants were oven-dried at 80 °C to constant weights. Then, they were wrapped in a layer of preservative film and placed in separate cassettes for two weeks exposure. An imaging analyzer (FUJIFILM BAS-1800, Tokyo, Japan) was used to observe its transport in cucumber seedlings. Non-treated plants were used as controls.

2.3 Root uptake from soils containing bound residues

Root uptake from soils containing bound residues was conducted as Han et al. reported. Brief, three BR-containing soils named (S1, S2 and S3) (physico-chemical property shown in Table 1) were got from our previous study. The soils were mixed with each corresponding fresh soil to adjust the specific activity to 20.7 Bq g⁻¹. 130 g of mixed soils were placed in a 200 mL plastic pot for the cultivation of cucumber plants, and

<table>
<thead>
<tr>
<th>No.</th>
<th>Soil type</th>
<th>pH</th>
<th>OM (%)</th>
<th>CEC (meq. 100 g⁻¹)</th>
<th>Silt (%)</th>
<th>Sand (%)</th>
<th>Clay (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>S1</td>
<td>Coastal soil</td>
<td>8.23</td>
<td>1.23</td>
<td>11.77</td>
<td>31.32</td>
<td>26.32</td>
<td>42.36</td>
</tr>
<tr>
<td>S2</td>
<td>Loamy soil</td>
<td>6.95</td>
<td>2.11</td>
<td>10.65</td>
<td>36.32</td>
<td>20.98</td>
<td>42.70</td>
</tr>
<tr>
<td>S3</td>
<td>Red soil</td>
<td>5.53</td>
<td>2.65</td>
<td>8.93</td>
<td>33.25</td>
<td>21.01</td>
<td>45.75</td>
</tr>
</tbody>
</table>

a Organic matter. b Cation exchange capacity.
distilled water was added to adjust the soil moisture to 60% of the water-holding capacity. Five germinated cucumber seeds were sown into each pot, and three replicates were performed for each treatment. Seeds were cultivated in the artificial climate incubator (28/16 °C, day/night; humidity, 80%; light/dark, 10 h/14 h) and irrigated every 2–3 d. The seedlings were sampled after 14 and 21 d. Each seedling was divided into shoots and roots, and the roots were rinsed with tap water. All plant parts were dried to a constant weight at 80 °C and combusted in the biological oxidizer. The released 14CO2 was absorbed in 20 mL liquid scintillation cocktail B, and the radioactivity was measured using an ultra-low liquid scintillation spectrometer (ULLSS; Quantulus 1220, Wallac) to estimate the amounts of BR that were absorbed by the cucumber plant at different times. The recovery efficiency of the above combustion procedure was greater than 95%.

2.4 Statistical analysis

The 14C radioactivity was calculated by deducting the background (20 dpm for ULLSS and 50 dpm for LSC). Each treatment was replicated three times except the autoradiography which was done two times. All of the data were analyzed using Origin 7.5 (Microcal Software, Northampton, MA, USA). The standard deviations (mean ± standard deviation, n = 3) presented in the text were calculated from the repeated measurements. Statistical analysis by one-way analysis of variance (ANOVA) was performed with SPSS 20.0 (IBM Corporation, Armonk, NY).

3. Results and discussion

3.1 Absorption and translocation in cucumber seedlings after leaf treatment

The 14C radioactivity recovered in TL, LATL, SATL, LBTL, SBTL, cotyledon and root were added together to calculate the total absorption amount in cucumber seedlings. The absorption of 14C-SYP-3343 in cucumber seedlings is shown in Fig. 2. At 0.25 DAT, 20% of the applied SYP-3343 was absorbed. This increased to 68% at 5 DAT. However, it then remained at this level until 14 DAT (Fig. 2). Pesticides transfer from leaves into plants owing to diffusion through leaf cuticles and the transpiration of the symplasm. The hydrophilic–lipophilic balance is necessary for germicidal activity, membrane permeability and conductivity.22 As reported by Liu et al.,23 the octanol/H2O

Table 2 Translocation and distribution of 14C SYP-3343 in cucumber seedling after leaf treatment (% of applied)*

<table>
<thead>
<tr>
<th>Days after treatment (DAT)</th>
<th>0.25</th>
<th>0.5</th>
<th>1</th>
<th>3</th>
<th>5</th>
<th>8</th>
<th>14</th>
</tr>
</thead>
<tbody>
<tr>
<td>TL</td>
<td>74.06 ± 0.09a</td>
<td>74.28 ± 0.13a</td>
<td>74.74 ± 0.12a</td>
<td>75.76 ± 0.07a</td>
<td>75.98 ± 0.18a</td>
<td>76.09 ± 0.11a</td>
<td>76.15 ± 0.12a</td>
</tr>
<tr>
<td>LATL</td>
<td>5.10 ± 0.14a</td>
<td>5.72 ± 0.18b</td>
<td>5.85 ± 0.13b</td>
<td>5.89 ± 0.21b</td>
<td>6.54 ± 0.12c</td>
<td>6.64 ± 0.24c</td>
<td>7.23 ± 0.22d</td>
</tr>
<tr>
<td>SATL</td>
<td>6.68 ± 0.22a</td>
<td>6.73 ± 0.12ab</td>
<td>6.81 ± 0.16ab</td>
<td>6.95 ± 0.18b</td>
<td>7.34 ± 0.13c</td>
<td>7.53 ± 0.24c</td>
<td>7.37 ± 0.17c</td>
</tr>
<tr>
<td>LBTL</td>
<td>4.20 ± 0.18a</td>
<td>4.09 ± 0.27a</td>
<td>4.03 ± 0.18ab</td>
<td>3.87 ± 0.11bc</td>
<td>3.65 ± 0.22c</td>
<td>3.69 ± 0.19c</td>
<td>3.63 ± 0.13c</td>
</tr>
<tr>
<td>SBTL</td>
<td>6.62 ± 0.26a</td>
<td>5.66 ± 0.21b</td>
<td>5.53 ± 0.17bc</td>
<td>5.42 ± 0.11c</td>
<td>4.43 ± 0.21d</td>
<td>4.32 ± 0.24d</td>
<td>4.29 ± 0.22d</td>
</tr>
<tr>
<td>Cotyledon</td>
<td>0.88 ± 0.03a</td>
<td>0.77 ± 0.02b</td>
<td>0.76 ± 0.03b</td>
<td>0.65 ± 0.01c</td>
<td>0.66 ± 0.01c</td>
<td>0.66 ± 0.02c</td>
<td>0.6 ± 0.01d</td>
</tr>
<tr>
<td>Root</td>
<td>0.75 ± 0.02a</td>
<td>0.73 ± 0.01a</td>
<td>0.62 ± 0.01b</td>
<td>0.63 ± 0.01b</td>
<td>0.62 ± 0.01b</td>
<td>0.61 ± 0.01b</td>
<td>0.58 ± 0.01c</td>
</tr>
</tbody>
</table>

* Data are the mean of three replications ± standard error of the mean. Means of plant parts within a row by the same letter are not significantly different according to ANOVA (P < 0.05).
partition coefficient \( \log P \) is 3.93 that means \(^{14}\text{C}\)-SYP-3343 is a liposoluble pesticide which can permeate membrane epidermal cells and be absorbed by cucumber seedlings. Bromilow et al. reported that the compound transport characters were affect by its \( \log P \), when \( \log P \) is near 4, the compound will be difficult to migrate neither up nor down.\(^{23}\) Our findings here was assistant with this finding.

As seen in Table 2, the TL contained a significantly greater amount of \(^{14}\text{C}\)-SYP-3343 (\( >74.06\% \)) compared with other parts, which indicated that most of the absorbed \(^{14}\text{C}\)-SYP-3343 remained in the TL and could not be easily transported. The rest of the SYP-3343 was transferred to the acropetal (following the xylem transpiration stream from lower to higher in the plant) and basipetal (following the photosynthetic transportation in phloem from higher to lower in the plant). However, the acropetal translocation was greater than that of the basipetal, especially by 3 DAT. The LATL (7.23%) and SATL (7.37%) values after 14 d were approximately two times greater than those of LBTL (3.63%) and SBTL (4.29%). The \(^{14}\text{C}\) content in root was always less than that in cotyledon within the experimental time.

In the autoradiographs of cucumber seedlings taken at 8 DAT with \(^{14}\text{C}\)-SYP-3343, which appears as a bright color, a deeper red indicates that a high \(^{14}\text{C}\) concentration existed (Fig. 3). The \(^{14}\text{C}\) concentrations in the different cucumber plant parts followed the order TL > SATL > LATL > SBTL > LBTL > cotyledon > root. Thus, the absorption and transportation of \(^{14}\text{C}\)-SYP-3343 was confirmed in cucumber plants. The axillary bud appeared to have a high \(^{14}\text{C}\) concentration, indicating that SYP-3343 may be transported to cucumber fruit. Further experiments are needed to confirm this finding.

### 3.2 Root absorption and translocation in cucumber seedlings grown in soils containing bound residues of \(^{14}\text{C}\)-SYP-3343

To determine the potential release of soil bound \(^{14}\text{C}\) SYP-3343 residues and their absorption and translocation in cucumber seedlings grown in the BR-containing soil, cucumber seedlings were divided into two parts: one part contained the stem and leaf, and the other contained the remaining root. Table 3 shows the percentage of \(^{14}\text{C}\) radioactivity in each plant part at 14 and 21 d after the cucumber seedlings had been growing in \(^{14}\text{C}\)-SYP-3343 BR-containing soil. The \(^{14}\text{C}\) could be released from BRs into the root and then transported up to the stems and leaves. The amount of \(^{14}\text{C}\) in the stem and leaf section of the plant increased with treatment time. There was a greater \(^{14}\text{C}\)-SYP-3343 distribution in shoots than in roots at both 14 and 21 DAT in the three soil types (Table 3). The BRs of \(^{14}\text{C}\)-labeled prometryn, malathion, parathion and carbofuran were more likely to be transported into shoots than that of SYP-3343.\(^{24–27}\) However, the uptake percentage for cucumber was very limited, and the maximum amount detected was only 0.21% at 21 DAT. The interaction between soil and pesticide was complicated. Many factors, such as soil physicochemical properties, microbial community structure and enzymatic activities, play roles in the release of BRs.\(^{28}\) During the cultivation period, different soil conditions may have contributed to the different BR release rates. Much greater amounts of the released \(^{14}\text{C}\) in red soil (S3) were absorbed into cucumber seedlings than in coastal soil (S1) and loamy soil (S2).

### 4. Conclusions

The leaf absorption of \(^{14}\text{C}\) SYP-3343 into cucumber seedlings was rapid, and the total absorption reached a maximum value of 68% by 5 DAT. Translocation occurred from the TL to all cucumber seedling parts, and the accumulation was greater in the acropetal than in the basipetal. Foliar applications may affect the cucumber fruit within 5 d. Soil properties can affect the release of the BRs, but less than 0.21% of the BRs can be taken up through seedling roots. This suggested that the BRs of SYP-3343 in the soil could not be easily transferred into cucumber fruit. More studies should be conducted to determine if SYP-3343 poses a potential risk to the environment and/or human health.

### Conflicts of interest

There are no conflicts to declare.

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### References


