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A one-pot, multicomponent reaction for the synthesis of novel 2-alkyl substituted 4-aminoimidazo[1,2-*a*][1,3,5]triazines†‡

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A highly selective, one-pot, three-component synthesis of novel 2-alkyl-substituted 4-aminoimidazo[1,2-*a*][1,3,5]triazines has been developed. The scope of the method was explored in two dimensions, varying the structures of trialkyl orthoesters and 2-aminoimidazoles in their reactions with cyanamide. Conveniently performed under microwave irradiation, this method was also proved to be efficient under conventional heating. A library of 24 novel compounds was prepared in high purity using this multicomponent approach. Molecular and crystal structures of representative molecules were studied using X-ray crystallography.

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Introduction

The imidazo[1,2-*a*][1,3,5]triazine ring system is recognised as an important scaffold in medicinal chemistry.² The well-defined targets for imidazo[1,2-*a*][1,3,5]triazines include a variety of enzymes, such as focal adhesion kinase,³ activated Cdc42-associated tyrosine kinase 1,⁴ dipeptidyl peptidase IV,⁵ and phosphodiesterase 5.⁶ This scaffold has been successfully used for the design of ligands for μ -opioid⁷ and A₁ adenosine⁸ receptors. The pharmacological potential of imidazo[1,2-*a*][1,3,5]triazines as anti-cancer,⁹ anti-viral¹⁰ and anti-diabetic¹¹ agents has also been well documented in the literature. The broad spectrum of biological activities of imidazo[1,2-*a*][1,3,5]triazines results in ongoing demand for the development of new practical methods for the synthesis of diversely functionalized imidazo[1,2-*a*][1,3,5]triazines.

Since the first report¹² on the synthesis of the imidazo[1,2-*a*][1,3,5]triazine ring system 50 years ago, a number of methods for the preparation of compounds with this heterocyclic core have been reported. They typically involve triazine ring

construction onto substituted imidazoles¹³ or, more often, the annulation of the imidazole ring onto substituted 1,3,5-triazines.¹⁴ The two most common approaches utilize: (1) reactions of 2-amino-1,3,5-triazines with α -haloaldehydes/ketones^{3,8,12,14a,b} or chalcones,^{14c} and (2) intramolecular cyclisations of 2-amino-1,3,5-triazines substituted at the amino group.^{11,14d-f} Reports on the preparation of imidazo[1,2-*a*][1,3,5]triazines *via* the formation of the 1,3,5-triazine ring onto imidazoles are limited and usually utilize addition of 2-aminoimidazoles to *N*-acyliso(thio)cyanates followed by the intramolecular closure of the 1,3,5-triazine ring.¹³

Multicomponent reactions (MCRs) represent an efficient strategy in modern synthetic chemistry.¹⁵ Minimising the number of synthetic steps in obtaining targeted compounds from available starting reagents is highly desirable in the fields of organic synthesis and drug discovery. Due to their diversity and combinatorial potential, MCRs have become highly appreciated tools for the preparation of libraries of compounds for biological screening.¹⁶ However, issues surrounding chemo- and regio-selectivity of MCRs often pose a challenge and require fine-tuning during method development. The need for rapid and selective construction of biologically active compounds for drug discovery has stimulated intense development in the use of microwave technology. Microwave irradiation has been effectively used as an alternative source of energy, improving efficiency and selectivity of MCRs.^{1,17,18}

We have been exploring the potential for new one-pot, multicomponent, microwave-assisted protocols for the synthesis of azolo[1,3,5]triazines.¹⁸ After the development of the MCRs utilizing aminoazoles, orthoformates and cyanamide, we have been refining the scope of this reaction using a variety of aminoazoles. In particular, the molecular diversity of compounds prepared has been achieved by varying the

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structures of the aminoazole precursors used in this three-component condensation and included libraries of pyrazolo[1,5-*a*][1,3,5]triazines,^{1,18a-c} 1,2,4-triazolo[1,5-*a*][1,3,5]triazines,^{18d,e} and imidazo[1,2-*a*][1,3,5]triazines.^{18f} Herein, we report our attempts to further expand the scope of this MCR by exploring its capacity to accommodate various orthoesters and therefore, introduce a new point of diversity in the reaction products. 2-Aminoimidazoles were selected as aminoazole substrates for the reaction in order to explore the regio- and chemo-selectivity of the process for these challenging substrates bearing endocyclic nitrogen atoms of similar reactivity.

Results & discussion

Synthesis

The starting materials used in our three-component reaction, *viz.* 2-aminoimidazoles (**1**), were prepared using the previously reported method developed by Van der Eycken's group.¹⁹ Optimization of the conditions for our proposed MCR was performed using a model reaction of 2-amino-4-phenylimidazole (**1a**), triethyl orthoacetate, and cyanamide under microwave irradiation (Table 1). The initial attempt to carry out this reaction at 150 °C for 20 min in ethyl acetate led to 4-amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (**5a**) obtained in 42% yield. Despite the moderate yield, the product **5a** was isolated in a chromatography-free process as the sole product. We found that the yield could be improved to 60% by performing the reaction at 160 °C for 35 min. However, despite attempts to further increase the yields by manipulating the reaction conditions, no improvement in the outcome was achieved.

The developed method was tested using two different models of microwave synthesizers: Discover SP (CEM) and Monowave 400 (Anton Paar). The three-component reaction of **1a**, triethyl orthoacetate and cyanamide under the optimized

conditions was conducted in both systems and resulted in the formation of **5a** in 60% and 56% yields, respectively.

Attempts to perform the reaction under reflux in ethyl acetate for 12 h led to the recovery of the starting material **1a** without traces of **5a**. However, an attempt to carry out the MCR of **1a**, triethyl orthoacetate and cyanamide in the Monowave 50 (Anton Paar) reactor using sealed vessels under fast conventional heating imitating the optimised microwave irradiation conditions was successful, resulting in the 54% yield of equally pure **5a**. Therefore, we could exclude any significant contribution of non-thermal microwave effects in promoting this MCR.

In principle, the annulation of the 1,3,5-triazine ring onto the 2-aminoimidazole **1a** in the three-component reaction could proceed with the formation of four regioisomeric structures *viz.* 2-aminoimidazo[1,2-*a*][1,3,5]triazines **2** and **3** as well as 4-aminoimidazo[1,2-*a*][1,3,5]triazines **4** and **5a** (Scheme 1). However, only one product was isolated from the MCR. The structure of this product was assigned on the basis of spectral data and an alternative synthesis of **5a** adopting our previously reported step-wise synthesis of azolo[1,3,5]triazines *via* the 1,3,5-triazine ring annulation onto aminoazoles with a predisposed position of the amino group on the triazine ring.²⁰ This method converted aminoazoles into formamidine intermediates *via* the reaction with orthoformates and morpholine followed by the triazine ring closing reaction of these intermediates with cyanamide. The conversion of 2-amino-4-phenylimidazole (**1a**) to the corresponding acetamidine (**6**) was achieved successfully using the reaction with triethyl orthoacetate and morpholine under microwave irradiation. The subsequent reaction of **6** with cyanamide led to the formation of a product identical to the one obtained from the MCR (Scheme 1). While the step-wise method excluded formation of compounds **2** and **3**, a ring-closing reaction could still afford compounds **4** and/or **5a**. The steric hindrance between the amino group and the phenyl ring in **4** would affect the thermodynamic stability of this potential product, therefore decreasing chances for its formation. Moreover, we could not exclude a rearrangement of **4** into **5a** under the reaction conditions. The rearrangement would proceed *via* intermediate **7** *via* the mechanism similar to the one reported²¹ for the rearrangement of another fused heterocycle with the amino-1,3,5-triazine ring. NOESY experiments performed on the product did not revealed any through-space interactions between the phenyl and methyl groups of the compound thus supporting assignment of structure **5a** as the MCR product. X-ray crystal structure determinations for analogous compounds **5g** and **5p** (*vide infra*) further confirmed the regioselective closure of the triazine ring to the more sterically accessible endocyclic nitrogen of the imidazole.

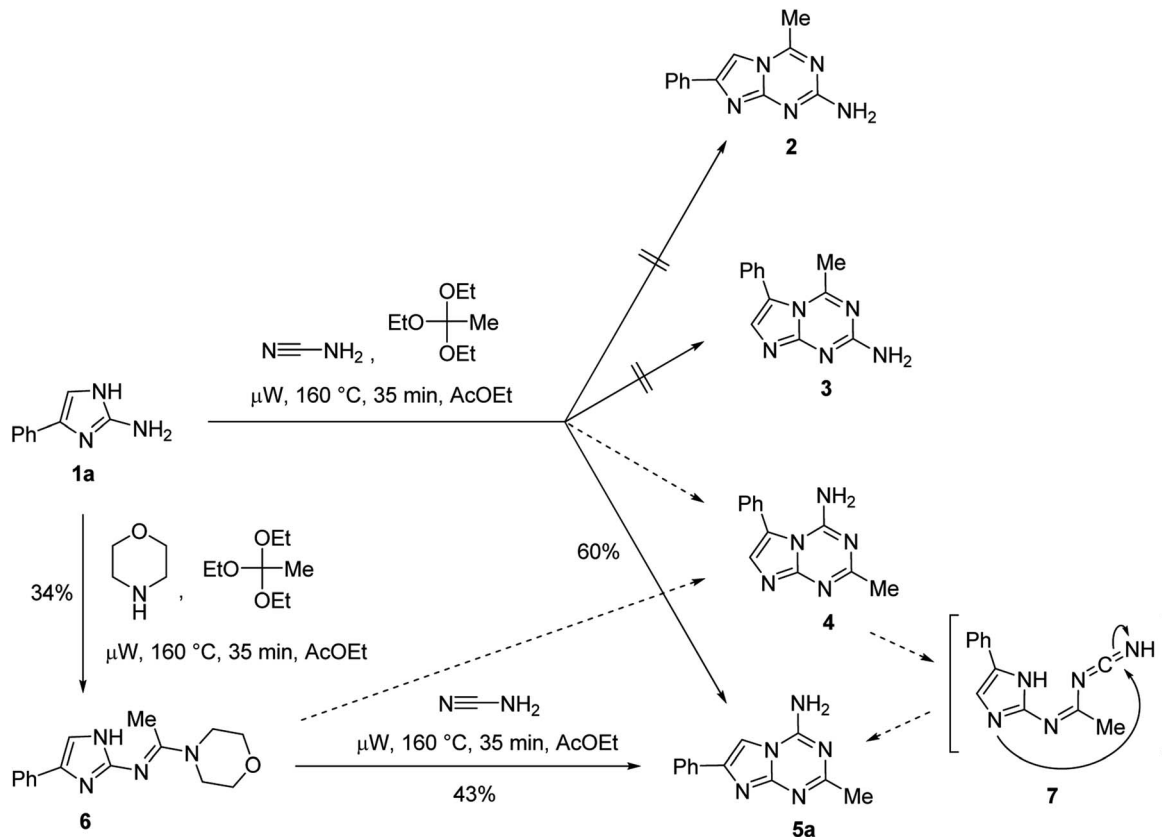
Using the optimized conditions from the model reaction, we explored the scope of our three-component reaction using a variety of trialkyl orthoesters and 4-aryl substituted 2-aminoimidazoles (**1**). The method was successfully applied for the synthesis of a library of 24 new 2-alkyl-4-amino-7-arylimidazo[1,2-*a*][1,3,5]triazines (**5**) (Table 2). The reaction was found to proceed similarly for the full range of tested 2-aminoimidazoles accommodating aryl substituents in the position 4 of the imidazole ring. However, variations in the alkyl orthoester type

Table 1 Optimization of conditions for the model reaction^a

Entry	Temperature (°C)	Time (min)	Isolated yield (%)
1	150	20	42
2	150	30	47
3	160	20	47
4	170	20	41
5	160	30	51
6	160	35	60
7	160	40	47
8 ^b	160	35	45

^a The reaction was performed using a Discover SP CEM microwave synthesizer with 1 mmol of **1a**, 2.5 mmol of triethyl orthoacetate and 2.5 mmol of cyanamide in 2 mL of the ethyl acetate. ^b The reaction was performed using 1 mmol of **1a**, 3 mmol of triethyl orthoacetate and 3 mmol of cyanamide.





Scheme 1 Synthesis of 4-amino-2-methyl-7-phenylimidazo[1,2-a][1,3,5]triazine (5a).

affected product yields more significantly. In general, an increase in length of alkyl chain substituent on compound 5 led to a decrease of isolated yields. The yields of compounds 5a–f and 5g–l obtained from triethyl orthoacetate and triethyl orthopropionate, respectively, were substantially higher than those of 5m–r and 5s–x prepared using triethyl orthobutyrate and trimethyl orthoacetate. The decrease of yields in the homologous series of compounds 5 could be attributed to the relatively lower reactivity of long chain orthoesters used for their preparation. This assumption found a confirmation in the isolation of compounds 5 exclusively from all the reactions. No traces of other isomers 2–4 were detected. Besides products 5, only unreacted 2-aminoimidazoles (1) could be isolated from the reaction mixtures.

X-ray crystallography

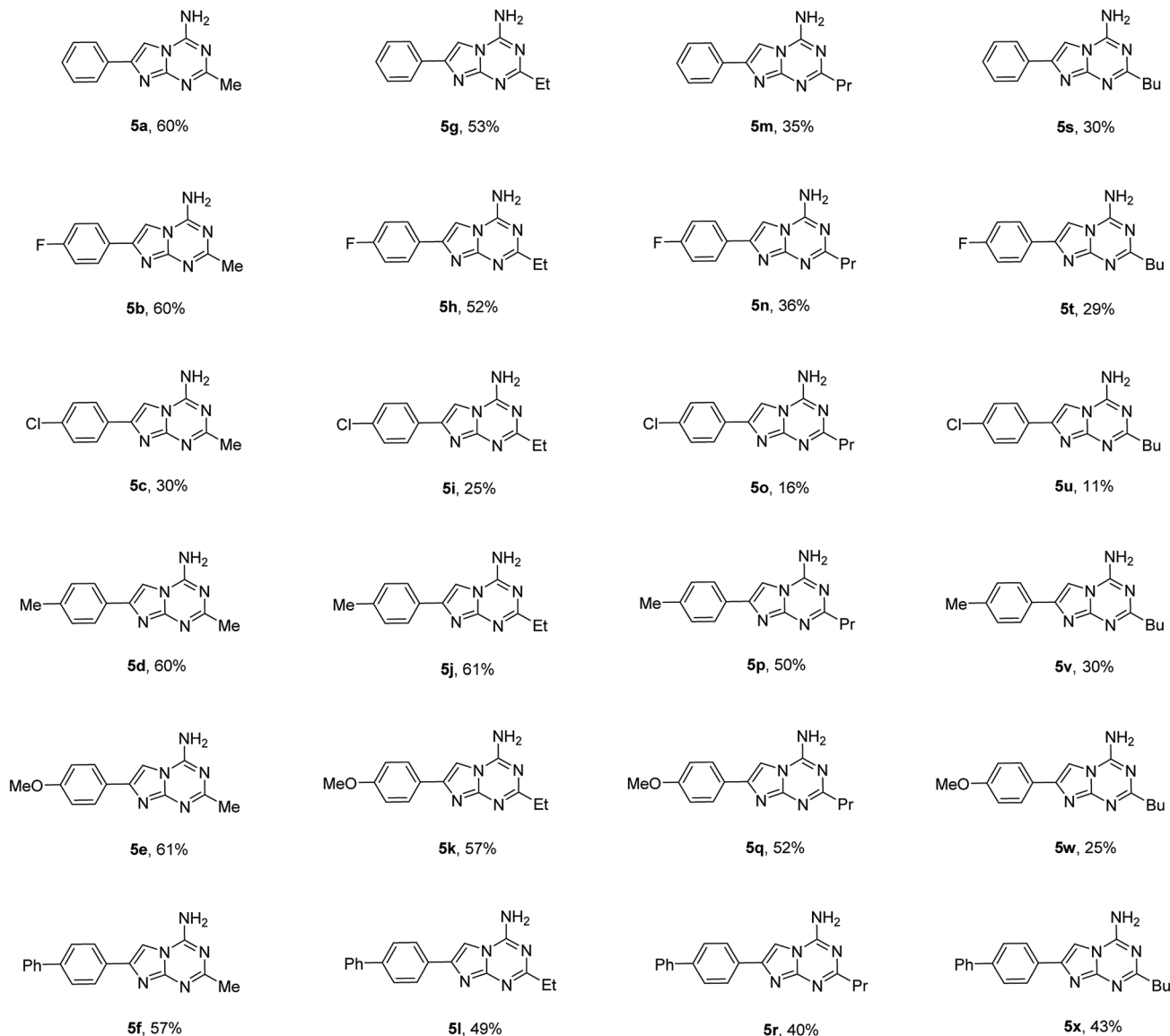
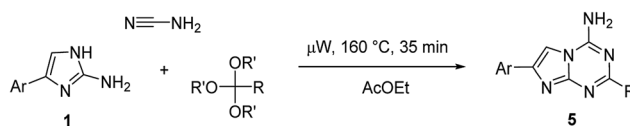
Crystals were obtained for two representative compounds, namely 5g and 5p, and their crystal and molecular structures determined by X-ray crystallography. The molecular structure of 5g is shown in Fig. 1a and comprises an essentially planar nine-membered ring with the r.m.s. deviation of the fitted atoms being 0.0129 Å; the maximum deviations above and below the least-square plane are found for the C2 [0.0215(8) Å] and C9 [0.0232(9) Å] atoms, respectively; the N41 and C71 atoms lie, respectively, 0.0464(13) and 0.0413(13) Å to the same side of the molecule.

Indeed, the entire molecule, with the exception of the C2-bound ethyl group, is essentially co-planar as seen in the dihedral between the imidazo[1,2-a][1,3,5]triazine and phenyl rings of 2.31(5)°. The ethyl group has an +anti-clinal (+ac) disposition with respect to the fused-ring system as seen in the torsion angle of 97.41(12)° for N1–C2–C21–C22.

The crystallographic asymmetric of 5p comprises two independent molecules, illustrated in Fig. 1b. The molecules present features that differ from those described for 5g. The r.m.s. of the nine atoms of the fused-ring system is 0.0247 Å [0.0103 Å for the second independent molecule] with the N41 and C71 atoms lying 0.0263(15) and 0.0837(14) Å [0.0333(15) and 0.0113(14) Å] to the same side of the plane. A twist in the molecule is evident as seen in the dihedral angle of 18.78(6)° [10.45(3)°] between the rings. The C2-bound *n*-propyl groups have very similar conformations with the C2–C21–C22–C23 torsion angles being –68.78(14) and –72.27(13)°, indicating a –syn-clinal (–sc) disposition in each case. The difference is that in the first independent molecule, the terminal methyl group is orientated away from the amino substituent whereas in the second molecule, it is orientated towards the amino group. An overlay diagram of the three molecules of 5g and 5p is shown in Fig. 1c. The difference in the co-planarity of the aromatic residues of 5g and 5p is clearly evident, as are the different orientations of the *n*-propyl substituents in 5p.

The molecular packing arrangements in the crystals of 5g and 5p feature directional amino–N–H⋯N(ring) hydrogen



Table 2 Three-component synthesis of 2-alkyl-4-amino-7-arylimidazo[1,2-a][1,3,5]triazines (**5**)^a

^a The reaction was performed using a Discover SP CEM microwave synthesizer with 1 mmol of 2-aminoimidazoles (**1**), 2.5 mmol of trialkyl orthoesters and 2.5 mmol of cyanamide at 160 °C for 35 min in 2 mL of ethyl acetate.

bonding. In **5g**, centrosymmetrically-related molecules are connected *via* amino-N–H \cdots N3(triazine) hydrogen bonds to form eight-membered $\{\cdots\text{HNCN}\}_2$ synthons. The second amino-N–H hydrogen atom forms a hydrogen bond to the triazine-N1 atom. The result is a supramolecular layer parallel to (1 0 0) and with a zig-zag topology with the phenyl groups lying to either side, Fig. 2a. The layers stack along the *a*-axis with the primary interactions between them being of the type $\pi\cdots\pi$. The closest

such interactions [3.4699(6) Å] occur between imidazoles, and between phenyl rings [3.8142(7) Å]. A view of the unit cell contents is shown in ESI Fig. S1† and geometric parameters characterizing the intermolecular interactions are given in the figure caption.

A supramolecular layer sustained by amino-N–H \cdots N(ring) hydrogen bonding is also found in the crystal of **5p**. The eight-membered $\{\cdots\text{HNCN}\}_2$ synthons arising from amino-N–H \cdots



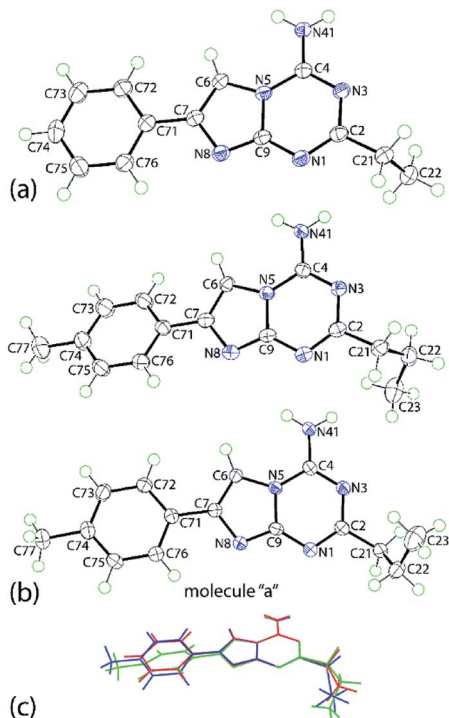


Fig. 1 (a) Molecular structure of **5g** and (b) molecular structures of the two independent molecules comprising the asymmetric unit of **5p**, showing atom labelling scheme and 70% anisotropic displacement parameters. (c) An overlay diagram of **5g** (red image), the first independent molecule of **5p** (green) and the inverted second independent molecule of **5p** (blue). The molecules have been overlapped so the triazine rings are coincident.

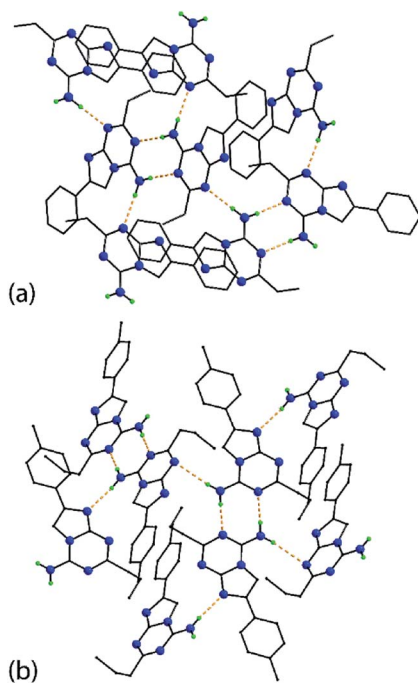


Fig. 2 Supramolecular layers sustained by amino-N-H...N(ring) hydrogen bonding (see ESI Fig. S1†), shown as orange dashed lines, in the crystals of (a) **5g** and (b) **5p**. Non-participating hydrogen atoms have been omitted and the carbon atoms are represented as small spheres.

N3(triazine) hydrogen bonds is formed by each of the independent molecules which self-associate. For the first independent molecule, rather than an amino-N-H...N3(triazine) hydrogen bond as formed by the second independent molecule (and in the crystal of **5g**), an amino-N-H...N8(imidazo) hydrogen bond is formed instead; the latter pairs of interactions involve both independent molecules. As in **5g**, the supramolecular layer, which is approximately parallel to (1 0 1), has a jagged topology with the 4-tolyl rings projecting to either side, Fig. 2b. The connections between layers to consolidate the three-dimensional architecture are of the type $\pi(\text{imidazo}) \cdots \pi(\text{imidazo})$ [3.5695(7) Å] and $\pi(\text{triazine}) \cdots \pi(\text{phenyl})$ [3.8946(7) and 3.9224(7) Å]. A view of the unit cell contents is shown in ESI Fig. S2† with geometric parameters given in the chart caption.

Conclusion

In summary, we have successfully developed a three component reaction for the synthesis of novel 2-alkyl substituted 4-aminoimidazo[1,2-*a*][1,3,5]triazines **5** using easily accessible starting materials and a simple, catalyst- and chromatography-free protocol. The distinct advantage of the proposed approach was found to be high chemo- and regio-selectivity with the exclusive formation of desired products in high purity.

Experimental section

General information

Melting points (uncorrected) were determined on a StuartTM SMP40 automatic melting point apparatus. ¹H and ¹³C NMR spectra were recorded on a Bruker Fourier 300 spectrometer (300 MHz) using DMSO-*d*₆ as a solvent and TMS as an internal reference. Microwave-assisted reactions were carried out in the closed vessel focused single mode using a Discover SP microwave synthesizer (CEM, USA) monitoring reaction temperature by equipped IR sensor. The model reaction was also carried out using Monowave 400 (Anton Paar, Austria) and Monowave 50 (Anton Paar, Austria) reactors.

General method for the synthesis of 4-amino-2-alkyl-7-arylimidazo[1,2-*a*][1,3,5]triazines (5**) under microwave irradiation.** The mixture of 2-aminoimidazoles (**1**, 1 mmol), cyanamide (105 mg, 2.5 mmol) and trialkyl orthoesters (2.5 mmol) in ethyl acetate (2 mL) were irradiated in a 10 mL seamless pressure vial using a microwave system operating at maximal microwave power up to 150 W (Discover SP, CEM) or 850 W (Monowave 400, Anton Paar) at 160 °C for 35 min. After cooling, the precipitate was filtered, washed with ethyl acetate and recrystallised from appropriate solvents.

4-Amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5a**).** Light yellow solid, yield 60%, mp 303–305 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 2.38 (3H, s, CH₃), 7.35 (1H, t, ³*J* = 7.4 Hz, H-4'), 7.47 (2H, t, ³*J* = 7.5 Hz, H-3' and H-5'), 7.85 (2H, d, ³*J* = 7.0 Hz, H-2' and H-6'), 8.15 (1H, s, H-6), 8.35 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 25.3 (CH₃), 101.7 (C-6), 125.2 (C-2' and C-6'), 127.9 (C-4'), 128.7 (C-3' and C-5'), 133.3 (C-1'), 143.2 (C-7), 150.3 & 150.6 (C-4 and C-8a), 164.7 (C-2). Anal. Calcd for



$C_{12}H_{11}N_5$: C, 63.99; H, 4.92; N, 31.09. Found: C, 63.82; H, 5.08; N, 30.91.

4-Amino-2-methyl-7-(4-fluorophenyl)imidazo[1,2-a][1,3,5]triazine (5b). White solid, yield 60%, mp > 350 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 2.38 (3H, s, CH₃), 7.30 (2H, dd, $^3J_{HF} = 8.9$ Hz, $^3J_{HH} = 8.9$ Hz, H-3' and H-5'), 7.87 (2H, dd, $^4J_{HF} = 5.5$ Hz, $^3J_{HH} = 8.9$ Hz, H-2' and H-6'), 8.11 (1H, s, H-6), 8.37 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 25.3 (CH₃), 101.5 (C-6), 115.7 (d, $^2J_{CF} = 21.7$ Hz, C-3' and C-5'), 127.2 (d, $^3J_{CF} = 8.3$ Hz, C-2' and C-6'), 129.8 (d, $^4J_{CF} = 2.9$ Hz, C-1'), 142.3 (C-7), 150.4 & 150.6 (C-4 and C-8a), 161.9 (d, $^1J_{CF} = 244.7$ Hz, C-4'), 164.8 (C-2). Anal. Calcd for $C_{12}H_{10}FN_5$: C, 59.25; H, 4.14; N, 28.79. Found: C, 59.02; H, 4.30; N, 28.63.

4-Amino-2-methyl-7-(4-chlorophenyl)imidazo[1,2-a][1,3,5]triazine (5c). Brown solid, yield 30%, mp > 350 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 2.37 (3H, s, CH₃), 7.53 (2H, d, $^3J = 8.6$ Hz, H-3' and H-5'), 7.9 (2H, d, $^3J = 8.58$ Hz, H-2' and H-6'), 8.16 (1H, s, H-6), 8.38 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 25.3 (CH₃), 102.1 (C-6), 126.9 (C-2' and C-6'), 128.8 (C-3' and C-5'), 132.1 (C-1'), 132.3 (C-4'), 142.1 (C-7), 150.6 & 150.4 (C-4 and C-8a), 164.9 (C-2). Anal. Calcd for $C_{12}H_{10}ClN_5$: C, 55.50; H, 3.88; N, 26.97. Found: 55.42; H, 4.01; N, 26.75.

4-Amino-2-methyl-7-(4-methylphenyl)imidazo[1,2-a][1,3,5]triazine (5d). White solid, yield 60%, mp > 350 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 2.34 (3H, s, CH₃), 2.37 (3H, s, CH₃), 7.27 (2H, d, $^3J = 7.9$ Hz, H-3' and H-5'), 7.73 (2H, d, $^3J = 8.1$ Hz, H-2' and H-6'), 8.09 (1H, s, H-6), 8.30 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 20.8 (CH₃), 25.3 (CH₃), 101.1 (C-6), 125.2 (C-2' and C-6'), 129.3 (C-3' and C-5'), 130.5 (C-1'), 137.3 (C-4'), 143.4 (C-7), 150.2 & 150.5 (C-4 and C-8a), 164.5 (C-2). Anal. Calcd for $C_{13}H_{13}N_5$: C, 65.25; H, 5.48; N, 29.27. Found: 65.18; H, 5.60; N, 29.18.

4-Amino-2-methyl-7-(4-methoxyphenyl)imidazo[1,2-a][1,3,5]triazine (5e). Light yellow solid, yield 61%, mp 338–339 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 2.37 (3H, s, CH₃), 3.80 (3H, s, OCH₃), 7.03 (2H, d, $^3J = 8.9$ Hz, H-3' and H-5'), 7.77 (2H, d, $^3J = 8.9$ Hz, H-2' and H-6'), 8.02 (1H, s, H-6), 8.29 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 25.3 (CH₃), 55.1 (OCH₃), 100.4 (C-6), 114.2 (C-3' and C-5'), 125.8 (C-1'), 126.6 (C-2' and C-6'), 143.3 (C-7), 150.3 & 150.5 (C-4 and C-8a), 159.2 (C-4'), 164.4 (C-2). Anal. Calcd for $C_{13}H_{13}N_5O$: C, 61.17; H, 5.13; N, 27.43. Found: C, 61.05; H, 5.23; N, 27.32.

4-Amino-2-methyl-7-(4-biphenyl)imidazo[1,2-a][1,3,5]triazine (5f). Light brown solid, yield 57%, mp > 350 °C (DMF). 1H NMR (300 MHz, DMSO- d_6): δ 2.38 (3H, s, CH₃), 7.38 (1H, t, $^3J = 7.3$ Hz, H-4''), 7.49 (2H, t, $^3J = 7.5$ Hz, H-3'' and H-5''), 7.73 (2H, d, $^3J = 7.1$ Hz, H-2'' and H-6''), 7.79 (2H, d, $^3J = 8.4$ Hz, H-3' and H-5'), 7.94 (2H, d, $^3J = 8.4$ Hz, H-2' and H-6'), 8.20 (1H, s, H-6), 8.36 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 25.3 (CH₃), 101.8 (C-6), 125.8 (C-3' and C-5'), 126.4 (C-3'' and C-5''), 127.0 (C-2' and C-6'), 127.4 (C-4''), 128.9 (C-2'' and C-6''), 132.4 (C-1'), 139.5 (C-4' and C-1''), 142.9 (C-7), 150.4 & 150.6 (C-4 and C-8a), 164.7 (C-2). Anal. Calcd for $C_{18}H_{15}N_5$: C, 71.74; H, 5.02; N, 23.24. Found: C, 71.68; H, 5.11; N, 23.10.

4-Amino-2-ethyl-7-phenylimidazo[1,2-a][1,3,5]triazine (5g). Light yellow solid, yield 53%, mp 299–301 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 1.25 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.66 (2H, q, $^3J = 7.5$ Hz, CH₂), 7.35 (1H, t, $^3J = 7.4$ Hz, H-4'), 7.47 (2H, t, $^3J =$

7.5 Hz, H-3' and H-5'), 7.86 (2H, d, $^3J = 7.0$ Hz, H-2' and H-6'), 8.16 (1H, s, H-6), 8.36 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 31.5 (CH₂), 101.7 (C-6), 125.2 (C-2' and C-6'), 127.9 (C-4'), 128.7 (C-3' and C-5'), 133.3 (C-1'), 143.3 (C-7), 150.4 & 150.8 (C-4 and C-8a), 168.6 (C-2). Anal. Calcd for $C_{13}H_{13}N_5$: C, 65.25; H, 5.48; N, 29.27. Found: C, 65.22; H, 5.52; N, 29.12.

4-Amino-2-ethyl-7-(4-fluorophenyl)imidazo[1,2-a][1,3,5]triazine (5h). White solid, yield 52%, mp 339–340 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 1.25 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.65 (2H, q, $^3J = 7.5$ Hz, CH₂), 7.31 (2H, dd, $^3J_{HF} = 8.9$ Hz, $^3J_{HH} = 8.9$ Hz, H-3' and H-5'), 7.88 (2H, dd, $^4J_{HF} = 5.5$ Hz, $^3J_{HH} = 8.9$ Hz, H-2' and H-6'), 8.12 (1H, s, H-6), 8.35 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 31.5 (CH₂), 101.5 (C-6), 115.7 (d, $^2J_{CF} = 21.8$ Hz, C-3' and C-5'), 127.2 (d, $^3J_{CF} = 8.3$ Hz, C-2' and C-6'), 129.9 (d, $^4J_{CF} = 3.0$ Hz, C-1'), 142.4 (C-7), 150.4 & 150.8 (C-4 and C-8a), 161.9 (d, $^1J_{CF} = 245.1$ Hz, C-4'), 168.7 (C-2). Anal. Calcd for $C_{13}H_{12}FN_5$: C, 60.69; H, 4.70; N, 27.22. Found: C, 60.62; H, 4.83; N, 27.11.

4-Amino-2-ethyl-7-(4-chlorophenyl)imidazo[1,2-a][1,3,5]triazine (5i). Brown solid, yield 25%, mp 340–342 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 1.24 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.65 (2H, q, $^3J = 7.5$ Hz, CH₂), 7.53 (2H, d, $^3J = 8.7$ Hz, H-3' and H-5'), 7.85 (2H, d, $^3J = 8.7$ Hz, H-2' and H-6'), 8.17 (1H, s, H-6), 8.36 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 31.5 (CH₂), 102.1 (C-6), 126.9 (C-2' and C-6'), 128.8 (C-3' and C-5'), 132.2 (C-1'), 132.3 (C-4'), 142.1 (C-7), 150.5 & 150.8 (C-4 and C-8a), 168.8 (C-2). Anal. Calcd for $C_{13}H_{12}ClN_5$: C, 57.04; H, 4.42; N, 25.59. Found: C, 56.97; H, 4.50; N, 25.49.

4-Amino-2-ethyl-7-(4-methylphenyl)imidazo[1,2-a][1,3,5]triazine (5j). Light yellow solid, yield 61%, mp > 350 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 1.25 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.34 (3H, s, CH₃), 2.65 (2H, q, $^3J = 7.5$ Hz, CH₂), 7.27 (2H, d, $^3J = 8.0$ Hz, H-3 and H-5), 7.75 (2H, d, $^3J = 8.1$ Hz, H-2' and H-6'), 8.10 (1H, s, H-6), 8.31 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 20.8 (CH₃), 31.5 (CH₂), 101.2 (C-6), 125.2 (C-2' and C-6'), 129.3 (C-3' and C-5'), 130.6 (C-1'), 137.3 (C-4'), 143.5 (C-7), 150.3 & 150.7 (C-4 and C-8a), 168.5 (C-2). Anal. Calcd for $C_{14}H_{15}N_5$: C, 66.38; H, 5.97; N, 27.65. Found: C, 66.33; H, 6.08; N, 27.58.

4-Amino-2-ethyl-7-(4-methoxyphenyl)imidazo[1,2-a][1,3,5]triazine (5k). White solid, yield 57%, mp 328–330 °C (EtOH). 1H NMR (300 MHz, DMSO- d_6): δ 1.24 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.64 (2H, q, $^3J = 7.6$ Hz, CH₂), 3.80 (3H, s, OCH₃), 7.04 (2H, d, $^3J = 8.9$ Hz, H-3' and H-5'), 7.78 (2H, d, $^3J = 8.9$ Hz, H-2' and H-6'), 8.02 (1H, s, H-6), 8.29 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 31.4 (CH₂), 100.4 (C-6), 114.2 (C-2' and C-6'), 125.9 (C-4'), 126.6 (C-3' and C-5'), 143.4 (C-7), 150.3 & 150.6 (C-4 and C-8a), 159.2 (C-1'), 168.3 (C-2). Anal. Calcd for $C_{14}H_{15}N_5O$: C, 62.44; H, 5.61; N, 26.01. Found: C, 62.38; H, 5.74; N, 25.90.

4-Amino-2-ethyl-7-(4-biphenyl)imidazo[1,2-a][1,3,5]triazine (5l). White solid, yield 49%, mp 348–350 °C (DMF). 1H NMR (300 MHz, DMSO- d_6): δ 1.26 (3H, t, $^3J = 7.6$ Hz, CH₃), 2.66 (2H, q, $^3J = 7.6$ Hz, CH₂), 7.38 (1H, t, $^3J = 7.3$ Hz, H-4''), 7.49 (2H, t, $^3J = 7.5$ Hz, H-3'' and H-5''), 7.73 (2H, d, $^3J = 7.2$ Hz, H-2'' and H-6''), 7.79 (2H, d, $^3J = 8.4$ Hz, H-3' and H-5'), 7.95 (2H, d, $^3J = 8.3$ Hz, H-2' and H-6'), 8.21 (1H, s, H-6), 8.36 (2H, br s, NH₂). ^{13}C NMR (75 MHz, DMSO- d_6): δ 11.7 (CH₃), 31.5 (CH₂), 101.9 (C-6), 125.8 (C-3' and C-5'), 126.4 (C-3'' and C-5''), 127.0 (C-2' and C-6'), 127.4



(C-4''), 128.9 (C-2'' and C-6''), 132.4 (C-1'), 139.5 & 139.6 (C-4' and C-1''), 143.0 (C-7), 150.5 & 150.8 (C-4 and C-8a), 168.7 (C-2). Anal. Calcd for C₁₉H₁₇N₅: C, 72.36; H, 5.43; N, 22.21. Found: C, 72.31; H, 5.52; N, 22.14.

4-Amino-2-propyl-7-phenylimidazo[1,2-a][1,3,5]triazine (5m). Light yellow solid, yield 35%, mp 272–274 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.94 (3H, t, ³J = 7.4 Hz, CH₃), 1.77 (2H, m, ³J = 7.4 Hz, CH₂), 2.60 (2H, t, ³J = 7.4 Hz, CH₂), 7.35 (1H, t, ³J = 7.3 Hz, H-4'), 7.47 (2H, t, ³J = 7.5 Hz, H-3' and H-5'), 7.85 (2H, d, ³J = 7.1 Hz, H-2' and H-6'), 8.15 (1H, s, H-6), 8.35 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.6 (CH₃), 20.4 (CH₂), 40.2 (CH₂), 101.7 (C-6), 125.2 (C-2' and C-6'), 127.9 (C-4'), 128.7 (C-3' and C-5'), 133.3 (C-1'), 143.3 (C-7), 150.3 & 150.8 (C-4 and C-8a), 167.6 (C-2). Anal. Calcd for C₁₄H₁₅N₅: C, 66.38; H, 5.97; N, 27.65. Found: C, 66.31; H, 6.02; N, 27.60.

4-Amino-2-propyl-7-(4-fluorophenyl)imidazo[1,2-a][1,3,5]triazine (5n). White solid, yield 36%, mp 290–292 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.94 (3H, t, ³J = 7.4 Hz, CH₃), 1.76 (2H, m, ³J = 7.4 Hz, CH₂), 2.60 (2H, t, ³J = 7.4 Hz, CH₂), 7.31 (2H, dd, ³J_{HF} = 8.9 Hz, ³J_{HH} = 8.9 Hz, H-3' and H-5'), 7.87 (2H, dd, ⁴J_{HF} = 5.5 Hz, ³J_{HH} = 8.9 Hz, H-2' and H-6'), 8.11 (1H, s, H-6), 8.34 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.6 (CH₃), 20.4 (CH₂), 40.2 (CH₂), 101.5 (C-6), 115.7 (d, ²J_{CF} = 21.7 Hz, C-3' and C-5'), 127.2 (d, ³J_{CF} = 8.3 Hz, C-2' and C-6'), 129.9 (d, ⁴J_{CF} = 3.5 Hz, C-1'), 142.4 (C-7), 150.4 & 150.8 (C-4 and C-8a), 161.9 (d, ¹J_{CF} = 244.5 Hz, C-4'), 167.7 (C-2). Anal. Calcd for C₁₄H₁₄FN₅: C, 61.98; H, 5.20; N, 25.81. Found: C, 61.92; H, 5.29; N, 25.72.

4-Amino-2-propyl-7-(4-chlorophenyl)imidazo[1,2-a][1,3,5]triazine (5o). Light brown solid, yield 16%, mp 294–295 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.94 (3H, t, ³J = 7.4 Hz, CH₃), 1.76 (2H, m, ³J = 7.4 Hz, CH₂), 2.60 (2H, t, ³J = 7.4 Hz, CH₂), 7.53 (2H, d, ³J = 8.7 Hz, H-3' and H-5'), 7.85 (2H, d, ³J = 8.9 Hz, H-2' and H-6'), 8.17 (1H, s, H-6), 8.36 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.6 (CH₃), 20.4 (CH₂), 40.2 (CH₂), 102.1 (C-6), 126.9 (C-2' and C-6'), 128.8 (C-3' and C-5'), 132.2 (C-1'), 132.3 (C-4'), 142.1 (C-7), 150.4 & 150.8 (C-4 and C-8a), 167.8 (C-2). Anal. Calcd for C₁₄H₁₄ClN₅: C, 58.44; H, 4.90; N, 24.34. Found: C, 58.37; H, 5.03; N, 24.22.

4-Amino-2-propyl-7-(4-methylphenyl)imidazo[1,2-a][1,3,5]triazine (5p). Light yellow solid, yield 50%, mp 305–307 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.94 (3H, t, ³J = 7.4 Hz, CH₃), 1.77 (2H, m, ³J = 7.4 Hz, CH₂), 2.34 (3H, s, CH₃), 2.60 (2H, t, ³J = 7.4 Hz, CH₂), 7.27 (2H, d, ³J = 7.3 Hz, H-3' and H-5'), 7.75 (2H, d, ³J = 8.0 Hz, H-2' and H-6'), 8.10 (1H, s, H-6), 8.30 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.6 (CH₃), 20.4 (CH₂), 20.8 (CH₃), 40.2 (CH₂), 101.2 (C-6), 125.2 (C-3' and C-5'), 129.3 (C-2' and C-6'), 130.6 (C-1'), 137.3 (C-4'), 143.5 (C-7), 150.3 & 150.7 (C-4 and C-8a), 167.5 (C-2). Anal. Calcd for C₁₅H₁₇N₅: C, 67.39; H, 6.41; N, 26.20. Found: C, 67.32; H, 6.53; N, 26.06.

4-Amino-2-propyl-7-(4-methoxyphenyl)imidazo[1,2-a][1,3,5]triazine (5q). White solid, yield 52%, mp 295–297 °C (MeOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.94 (3H, t, ³J = 7.4 Hz, CH₃), 1.76 (2H, m, ³J = 7.4 Hz, CH₂), 2.59 (2H, t, ³J = 7.4 Hz, CH₂), 3.80 (3H, s, OCH₃), 7.04 (2H, d, ³J = 8.9 Hz, H-3' and H-5'), 7.78 (2H, d, ³J = 8.9 Hz, H-2' and H-6'), 8.03 (1H, s, H-6), 8.30 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.6 (CH₃), 20.4 (CH₂), 40.2 (CH₂), 55.1 (OCH₃), 100.4 (C-6), 114.2 (C-3' and C-5'), 125.9 (C-1'), 126.6 (C-2' and C-6'), 143.4

(C-7), 150.3 & 150.6 (C-4 and C-8a), 159.2 (C-4'), 167.3 (C-2). Anal. Calcd for C₁₅H₁₇N₅O: C, 63.59; H, 6.05; N, 24.72. Found: C, 63.52; H, 6.20; N, 24.59.

4-Amino-2-propyl-7-(4-biphenyl)imidazo[1,2-a][1,3,5]triazine (5r). Light brown solid, yield 40%, mp 337–338 °C (DMF). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.95 (3H, t, ³J = 7.4 Hz, CH₃), 1.78 (2H, m, ³J = 7.4 Hz, CH₂), 2.62 (2H, t, ³J = 7.4 Hz, CH₂), 7.38 (1H, t, ³J = 7.3 Hz, H-4''), 7.49 (2H, t, ³J = 7.5 Hz, H-3'' and H-5''), 7.73 (2H, d, ³J = 7.2 Hz, H-2'' and H-6''), 7.79 (2H, d, ³J = 8.4 Hz, H-3' and H-5'), 7.95 (2H, d, ³J = 8.3 Hz, H-2' and H-6'), 8.21 (1H, s, H-6), 8.36 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.7 (CH₃), 20.4 (CH₂), 40.2 (CH₂), 101.9 (C-6), 125.8 (C-3' and C-5'), 126.4 (C-3'' and C-5''), 127.0 (C-2' and C-6'), 127.4 (C-4''), 128.9 (C-2'' and C-6''), 132.4 (C-1'), 139.5 & 139.6 (C-4' and C-1''), 143.0 (C-7), 150.4 & 150.8 (C-4 and C-8a), 167.7 (C-2). Anal. Calcd for C₂₀H₁₉N₅: C, 72.92; H, 5.81; N, 21.26. Found: C, 72.88; H, 5.94; N, 21.13.

4-Amino-2-butyl-7-phenylimidazo[1,2-a][1,3,5]triazine (5s). Light yellow solid, yield 30%, mp 267–269 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.92 (3H, t, ³J = 7.3 Hz, CH₃), 1.36 (2H, m, ³J = 7.4 Hz, CH₂), 1.72 (2H, m, ³J = 7.5 Hz, CH₂), 2.62 (2H, t, ³J = 7.5 Hz, CH₂), 7.35 (1H, t, ³J = 7.3 Hz, H-4'), 7.47 (2H, t, ³J = 7.5 Hz, H-3' and H-5'), 7.85 (2H, d, ³J = 7.0 Hz, H-2' and H-6'), 8.15 (1H, s, H-6), 8.35 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.7 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 101.7 (C-6), 125.2 (C-2' and C-6'), 127.9 (C-4'), 128.7 (C-3' and C-5'), 133.3 (C-1'), 143.3 (C-7), 150.3 & 150.8 (C-4 and C-8a), 167.8 (C-2). Anal. Calcd for C₁₅H₁₇N₅: C, 67.39; H, 6.41; N, 26.20. Found: C, 67.32; H, 6.55; N, 26.04.

4-Amino-2-butyl-7-(4-fluorophenyl)imidazo[1,2-a][1,3,5]triazine (5t). White solid, yield 29%, mp 296–298 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.92 (3H, t, ³J = 7.3 Hz, CH₃), 1.36 (2H, m, ³J = 7.4 Hz, CH₂), 1.72 (2H, m, ³J = 7.5 Hz, CH₂), 2.62 (2H, t, ³J = 7.5 Hz, CH₂), 7.31 (2H, dd, ³J_{HF} = 8.9 Hz, ³J_{HH} = 8.9 Hz, H-3' and H-5'), 7.88 (2H, dd, ⁴J_{HF} = 5.6 Hz, ³J_{HH} = 8.9 Hz, H-2' and H-6'), 8.12 (1H, s, H-6), 8.34 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.8 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 101.5 (C-6), 115.7 (d, ²J_{CF} = 21.7 Hz, C-3' and C-5'), 127.2 (d, ³J_{CF} = 8.3 Hz, C-2' and C-6'), 129.9 (d, ⁴J_{CF} = 3.0 Hz, C-1'), 142.4 (C-7), 150.4 & 150.8 (C-4 and C-8a), 161.9 (d, ¹J_{CF} = 244.8 Hz, C-4'), 167.9 (C-2). Anal. Calcd for C₁₅H₁₆FN₅: C, 63.14; H, 5.65; N, 24.55. Found: C, 63.05; H, 5.77; N, 24.42.

4-Amino-2-butyl-7-(4-chlorophenyl)imidazo[1,2-a][1,3,5]triazine (5u). Light brown solid, yield 11%, mp 287–288 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.91 (3H, t, ³J = 7.3 Hz, CH₃), 1.35 (2H, m, ³J = 7.4 Hz, CH₂), 1.72 (2H, m, ³J = 7.5 Hz, CH₂), 2.62 (2H, t, ³J = 7.5 Hz, CH₂), 7.53 (2H, d, ³J = 8.6 Hz, H-3' and H-5'), 7.84 (2H, d, ³J = 8.6 Hz, H-2' and H-6'), 8.16 (1H, s, H-6), 8.36 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.7 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 102.1 (C-6), 126.9 (C-2' and C-6'), 128.8 (C-3' and C-5'), 132.2 & 132.33 (C-1' and C-4'), 142.1 (C-7), 150.4 & 150.8 (C-4 and C-8a), 168.1 (C-2). Anal. Calcd for C₁₅H₁₆ClN₅: C, 59.70; H, 5.34; N, 23.21. Found: C, 59.63; H, 5.41; N, 23.15.

4-Amino-2-butyl-7-(4-methylphenyl)imidazo[1,2-a][1,3,5]triazine (5v). Light yellow solid, yield 25%, mp 287–289 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.92 (3H, t, ³J = 7.3 Hz, CH₃), 1.36 (2H, m, ³J = 7.4 Hz, CH₂), 1.72 (2H, m, ³J = 7.5 Hz, CH₂), 2.34



(3H, s, CH₃), 2.62 (2H, t, ³J = 7.5 Hz, CH₂), 7.27 (2H, d, ³J = 7.9 Hz, H-3' and H-5'), 7.74 (2H, d, ³J = 8.1 Hz, H-2' and H-6'), 8.09 (1H, s, H-6), 8.30 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.7 (CH₃), 20.8 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 101.2 (C-6), 125.2 (C-2' and C-6'), 129.3 (C-3' and C-5'), 130.5 (C-1'), 137.3 (C-4'), 143.5 (C-7), 150.3 & 150.7 (C-4 and C-8a), 167.7 (C-2). Anal. Calcd for C₁₆H₁₉N₅: C, 68.30; H, 6.81; N, 24.89. Found: C, 68.24; H, 6.90; N, 24.81.

4-Amino-2-butyl-7-(4-methoxyphenyl)imidazo[1,2-*a*][1,3,5]triazine (5w). Light yellow solid, yield 25%, mp 279–281 °C (MeOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.92 (3H, t, ³J = 7.3 Hz, CH₃), 1.36 (2H, m, ³J = 7.4 Hz, CH₂), 1.72 (2H, m, ³J = 7.5 Hz, CH₂), 2.62 (2H, t, ³J = 7.5 Hz, CH₂), 3.81 (3H, s, OCH₃), 7.04 (2H, d, ³J = 8.9 Hz, H-3' and H-5'), 7.78 (2H, d, ³J = 8.8 Hz, H-2' and H-6'), 8.03 (1H, s, H-6), 8.30 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.8 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 55.1 (OCH₃), 100.4 (C-6), 114.2 (C-3' and C-5'), 125.9 (C-1'), 126.6 (C-2' and C-6'), 143.4 (C-7), 150.3 & 150.6 (C-4 and C-8a), 159.2 (C-4'), 167.6 (C-2). Anal. Calcd for C₁₆H₁₉N₅O: C, 64.63; H, 6.44; N, 23.55. Found: C, 64.55; H, 6.54; N, 23.43.

4-Amino-2-butyl-7-(4-biphenyl)imidazo[1,2-*a*][1,3,5]triazine (5x). Light brown solid, yield 43%, mp 327–329 °C (DMF). ¹H NMR (300 MHz, DMSO-*d*₆): δ 0.92 (3H, t, ³J = 7.3 Hz, CH₃), 1.37 (2H, m, ³J = 7.4 Hz, CH₂), 1.73 (2H, m, ³J = 7.5 Hz, CH₂), 2.63 (2H, t, ³J = 7.5 Hz, CH₂), 7.38 (1H, t, ³J = 7.3 Hz, H-4''), 7.49 (2H, t, ³J = 7.5 Hz, H-3'' and H-5''), 7.73 (2H, d, ³J = 7.5 Hz, H-2'' and H-6''), 7.79 (2H, d, ³J = 8.4 Hz, H-3' and H-5'), 7.94 (2H, d, ³J = 8.4 Hz, H-2' and H-6'), 8.20 (1H, s, H-6), 8.36 (2H, br s, NH₂). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 13.7 (CH₃), 21.7 (CH₂), 29.3 (CH₂), 37.9 (CH₂), 101.8 (C-6), 125.8 (C-3' and C-5'), 126.4 (C-3'' and C-5''), 127.0 (C-2' and C-6'), 127.4 (C-4''), 128.9 (C-2'' and C-6''), 132.4 (C-1'), 139.5 (C-4' and C-1''), 143.0 (C-7), 150.4 & 150.7 (C-4 and C-8a), 167.9 (C-2). Anal. Calcd for C₂₁H₂₁N₅: C, 73.44; H, 6.16; N, 20.39. Found: C, 73.38; H, 6.26; N, 20.31.

Synthesis of 4-amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5a) using Monowave 400 (Anton Paar) microwave synthesizer. The mixture of 2-amino-4-phenyl-1*H*-imidazole (**1a**, 159 mg, 1 mmol), cyanamide (105 mg, 2.5 mmol) and triethyl orthoacetate (0.46 mL, 2.5 mmol) in ethyl acetate (2 mL) were irradiated in a 10 mL seamless pressure vial using a microwave system operating at maximal microwave power up to 850 W (Monowave 400, Anton Paar) at 160 °C for 35 min. After cooling, the precipitate was filtered and washed with ethyl acetate to give **5a** in the 56% yield.

Synthesis of 4-amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5a) via heating in a sealed vessel. The mixture of 2-amino-4-phenyl-1*H*-imidazole (**1a**, 159 mg, 1 mmol), cyanamide (105 mg, 2.5 mmol) and triethyl orthoacetate (0.46 mL, 2.5 mmol) in ethyl acetate (2 mL) were heated in a 10 mL seamless pressure vial in an enclosed system (Monowave 50, Anton Paar) at 160 °C for 35 min. After cooling, the precipitate was filtered to obtain a compound, which was identical to **5a** prepared under microwave irradiation. Yield 54%.

Step-wise approach for the synthesis of 4-amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5a)

***N,N'*-Morpholino-*N*-[3(5)-phenylimidazo-5(3)yl]acetamide (6).** A mixture of 2-amino-1*H*-imidazole **1a** (159 mg, 1 mmol) with

triethyl orthoacetate (0.46 mL, 2.5 mmol) and morpholine (0.22 mL, 2.5 mmol) in ethyl acetate (2 mL) were irradiated in a 10 mL seamless pressure vial using the Discover SP (CEM) microwave system operating at maximal microwave power up to 150 W at 160 °C for 35 min. After cooling, the precipitate was filtered and recrystallised from EtOH to give pure formamidine **6**. Light yellow solid, yield 34%, mp 217–219 °C (EtOH). ¹H NMR (300 MHz, DMSO-*d*₆): δ 2.28 (3H, s, CH₃), 3.55–3.56 (4H, m, (CH₂)₂), 3.62–3.65 (4H, m, (CH₂)₂), 7.10 (1H, t, ³J = 7.3 Hz, H-4'), 7.22 (1H, s, CH), 7.29 (2H, t, ³J = 7.7 Hz, H-3' and H-5'), 7.66 (2H, d, ³J = 6.9 Hz, H-2' and H-6'), 11.08 & 11.44 (1H, br s, NH). Anal. Calcd for C₁₅H₁₈N₄O: C, 66.64; H, 6.71; N, 20.73. Found: C, 66.46; H, 6.94; N, 20.55.

4-Amino-2-methyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5a). A mixture of **6** (270 mg, 1 mmol) with cyanamide (105 mg, 2.5 mmol) in ethyl acetate (2 mL) were irradiated in a 10 mL seamless pressure vial using the Discover SP (CEM) microwave system operating at maximal microwave power up to 150 W at 160 °C for 35 min. After cooling, the precipitate was filtered to obtain a compound, which was identical to **5a** prepared using the multicomponent reaction. Yield 43%.

X-Ray crystallographic analysis. Intensity data for **5g** and **5p** were measured at 100 K on an Rigaku/Oxford Diffraction XtaLAB Synergy diffractometer (Dualflex, AtlasS2) fitted with CuK α radiation ($\lambda = 1.54178 \text{ \AA}$) so that $\theta_{\max} = 67.1^\circ$. Data reduction and Gaussian absorption corrections were by standard methods.²² The structures were solved by direct-methods²³ and refined on F^2 (ref. 24) with anisotropic displacement parameters, C-bound H atoms in the riding model approximation and N-bound H atoms refined with N–H = $0.88 \pm 0.01 \text{ \AA}$. A weighting scheme of the form $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$ where $P = (F_o^2 + 2F_c^2)/3$ was introduced in each case. The molecular structure diagrams showing 70% probability displacement ellipsoids were generated by ORTEP for Windows²⁵ and the packing diagrams with DIAMOND.²⁶ Additional data analysis was made with PLATON.²⁷

Crystal data for 4-amino-2-ethyl-7-phenylimidazo[1,2-*a*][1,3,5]triazine (5g). $M = 239.28$, orthorhombic, $Pccn$, $a = 16.6173(2)$, $b = 11.17070(10)$, $c = 12.4074(2) \text{ \AA}$, $V = 2303.15(5) \text{ \AA}^3$, $Z = 8$, $D_x = 1.380 \text{ g cm}^{-3}$, $F(000) = 1008$, $\mu = 0.709 \text{ mm}^{-1}$, no. reflns meas. = 14 762, no. unique reflns = 2060 ($R_{\text{int}} = 0.026$), no. reflns with $I \geq 2\sigma(I) = 1935$, no. parameters = 170, $R(\text{obs. data}) = 0.032$, $wR2(\text{all data}) = 0.085$. CCDC deposition number: 1840247.

Crystal data for 4-amino-2-propyl-7-(4-methoxyphenyl)imidazo[1,2-*a*][1,3,5]triazine (5p). $M = 267.33$, triclinic, $P\bar{1}$, $a = 10.7534(2)$, $b = 11.8995(2)$, $c = 12.1124(3) \text{ \AA}$, $\alpha = 71.918(2)$, $\beta = 68.962(2)$, $\gamma = 78.712(2)^\circ$, $V = 1368.93(5) \text{ \AA}^3$, $Z = 4$, $D_x = 1.297 \text{ g cm}^{-3}$, $F(000) = 568$, $\mu = 0.650 \text{ mm}^{-1}$, no. reflns meas. = 32 816, no. unique reflns = 4893 ($R_{\text{int}} = 0.029$), no. reflns with $I \geq 2\sigma(I) = 4530$, no. parameters = 377, $R(\text{obs. data}) = 0.034$, $wR2(\text{all data}) = 0.092$. CCDC deposition number: 1840248.

Conflicts of interest

There are no conflicts to declare.



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Notes and references

- 1 F. P. L. Lim, N. R. Halcovitch, E. R. T. Tiekink and A. V. Dolzhenko, *Tetrahedron*, 2018, **74**, 1868–1879.
- 2 F. P. L. Lim and A. V. Dolzhenko, *Eur. J. Med. Chem.*, 2014, **85**, 371–390.
- 3 P. Dao, N. Smith, C. Tomkiewicz-Raulet, E. Yen-Pon, M. Camacho-Artacho, D. Lietha, J.-P. Herbeuval, X. Coumoul, C. Garbay and H. Chen, *J. Med. Chem.*, 2015, **58**, 237–251.
- 4 X. Jiao, D. J. Kopecky, J. Liu, J. Liu, J. C. Jaen, M. G. Cardozo, R. Sharma, N. Walker, H. Wesche, S. Li, E. Farrelly, S.-H. Xiao, Z. Wang and F. Kayser, *Bioorg. Med. Chem. Lett.*, 2012, **22**, 6212–6217.
- 5 J. Feng, S. L. Gwaltney, J. A. Stafford, M. B. Wallace and Z. Zhang, *US Pat.*, 20060135767, 2006.
- 6 H. Haning, U. Niewoehner, T. Schenke, T. Lampe, A. Hillisch and E. Bischoff, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 3900–3907.
- 7 (a) M. Rzadkowska, E. Szacon, D. Matosiuk, E. Kedzierska and S. Fidecka, *Acta Pol. Pharm.*, 2012, **69**, 1270–1275; (b) D. Matosiuk, S. Fidecka, L. Antkiewicz-Michaluk, J. Lipkowski, I. Dybala and A. E. Koziol, *Eur. J. Med. Chem.*, 2002, **37**, 761–772.
- 8 E. Novellino, E. Abignente, B. Cosimelli, G. Greco, M. Iadanza, S. Laneri, A. Lavecchia, M. G. Rimoli, F. Da Settimo, G. Primofiore, D. Tuscano, L. Trincavelli and C. Martini, *J. Med. Chem.*, 2002, **45**, 5030–5036.
- 9 (a) F. Saczewski, M. Maruszak and P. J. Bednarski, *Arch. Pharm.*, 2008, **341**, 121–125; (b) I. Lakomska, B. Golankiewicz, J. Wietrzyk, M. Pelczynska, A. Nasulewicz, A. Opolski, J. Sitkowski, L. Kozerski and E. Szlyk, *Inorg. Chim. Acta*, 2005, **358**, 1911–1917.
- 10 (a) D. Dukhan, F. Leroy, J. Peyronnet, E. Bosc, D. Chaves, M. Durka, R. Storer, P. La Colla, F. Seela and G. Gosselin, *Nucleosides, Nucleotides Nucleic Acids*, 2005, **24**, 671–674; (b) J. O. Ojwang, S. Ali, D. F. Smee, J. D. Morrey, C. D. Shimasaki and R. W. Sidwell, *Antiviral Res.*, 2005, **68**, 49–55.
- 11 P. Cui, T. L. Macdonald, M. Chen and J. L. Nadler, *Bioorg. Med. Chem. Lett.*, 2006, **16**, 3401–3405.
- 12 J. Kobe, B. Stanovnik and M. Tisler, *Chem. Commun.*, 1968, 1456.
- 13 (a) Z. Wang, H. K. Huynh, B. Han, R. Krishnamurthy and A. Eschenmoser, *Org. Lett.*, 2003, **5**, 2067–2070; (b) P. Rao and S. A. Benner, *J. Org. Chem.*, 2001, **66**, 5012–5015; (c) E. J. Prisbe, J. P. H. Verheyden and J. G. Moffatt, *J. Org. Chem.*, 1978, **43**, 4784–4794; (d) J. J. Voegel, M. M. Altorfer and S. A. Benner, *Helv. Chim. Acta*, 1993, **76**, 2061–2069.
- 14 (a) P. Dao, C. Garbay and H. Chen, *Tetrahedron*, 2013, **69**, 3867–3871; (b) T. Hanami, H. Oda, A. Nakamura, H. Urata, M. Itoh and Y. Hayashizaki, *Tetrahedron Lett.*, 2007, **48**, 3801–3803; (c) J. J. Li, C. Song, D.-M. Cui and C. Zhang, *Org. Biomol. Chem.*, 2017, **15**, 5564–5570; (d) L. B. Krasnova, J. E. Hein and V. V. Fokin, *J. Org. Chem.*, 2010, **75**, 8662–8665; (e) B. Han, B. Jaun, R. Krishnamurthy and A. Eschenmoser, *Org. Lett.*, 2004, **6**, 3691–3694; (f) Z. Wang, H. K. Huynh, B. Han, R. Krishnamurthy and A. Eschenmoser, *Org. Lett.*, 2003, **5**, 2067–2070.
- 15 (a) J. F. Allochio Filho, B. C. Lemos, A. S. de Souza, S. Pinheiro and S. J. Greco, *Tetrahedron*, 2017, **73**, 6977–7004; (b) S. Maddila, S. B. Jonnalagadda, K. K. Gangu and S. N. Maddila, *Curr. Org. Synth.*, 2017, **14**, 634–653; (c) J.-P. Wan, L. Gan and Y. Liu, *Org. Biomol. Chem.*, 2017, **15**, 9031–9043; (d) I. A. Ibarra, A. Islas-Jacome and E. Gonzalez-Zamora, *Org. Biomol. Chem.*, 2018, **16**, 1402–1418; (e) P. Patil, B. Mishra, G. Sheombarsing, K. Kurpiewska, J. Kalinowska-Tluscik and A. Doemling, *ACS Comb. Sci.*, 2018, **20**, 70–74; (f) E. A. Muravyova, S. M. Desenko, R. V. Rudenko, S. V. Shishkina, O. V. Shishkin, Y. V. Sen'ko, E. V. Vashchenko and V. A. Chebanov, *Tetrahedron*, 2011, **67**, 9389–9400; (g) P. V. Santhini, R. A. Krishnan, S. A. Babu, B. S. Simethy, G. Das, V. K. Praveen, S. Varughese and J. John, *J. Org. Chem.*, 2017, **82**, 10537–10548.
- 16 (a) R. Pathoor and D. Bahulayan, *New J. Chem.*, 2018, **42**, 6810–6816; (b) E. Ezzatzadeh and Z. Hossaini, *Nat. Prod. Res.*, 2018, DOI: 10.1080/14786419.2018.1428598; (c) P. Sathiyachandran, P. Manogaran, V. N. Nesterov, V. V. Padma and K. J. Rajendra Prasad, *Eur. J. Med. Chem.*, 2018, **150**, 851–863; (d) P. R. Mali, P. K. Shirsat, N. Khomane, L. Nayak, J. B. Nanubolu and H. M. Meshram, *ACS Comb. Sci.*, 2017, **19**, 633–639.
- 17 (a) A. J. Ansari, S. Sharma, R. S. Pathare, K. Gopal, D. M. Sawant and R. T. Pardasani, *ChemistrySelect*, 2016, **1**, 1016–1021; (b) A. S. Medvedeva, M. M. Demina, T. V. Kon'kova, T. L. H. Nguyen, A. V. Afonin and I. A. Ushakov, *Tetrahedron*, 2017, **73**, 3979–3985; (c) N. M. Drosos, C. Kakoulidou, M. Raftopoulou, J. Stephanidou-Stephanatou, C. A. Tsoleridis and A. G. Hatzidimitriou, *Tetrahedron*, 2017, **73**, 1–7; (d) F. B. Souza, D. C. Pimenta, H. A. Stefani and A. Helio, *Tetrahedron Lett.*, 2016, **57**, 1592–1596; (e) P. Vincetti, A. Brianza, N. Scalacci, G. Costantino, D. Castagnolo and M. Radi, *Tetrahedron Lett.*, 2016, **57**, 1464–1467; (f) M. Veeranarayana Reddy, B. Siva Kumar, K. T. Lim, B. G. Cho and Y. T. Jeong, *Tetrahedron Lett.*, 2016, **57**, 476–478; (g) A. Kumbhar, S. Jadhav, R. Shejwal, G. Rashinkar and R. Salunkhe, *RSC Adv.*, 2016, **6**, 19612–19619; (h) R. Mishra and L. H. Choudhury, *RSC Adv.*, 2016, **6**, 24464–24469; (i) B. Jiang, T.-S. Zhang, R. Fu, W.-J. Hao, S.-L. Wang and S.-J. Tu, *Tetrahedron*, 2016, **72**, 5652–5658; (j) M. Kumar, T. Kaur, V. K. Gupta and A. Sharma, *RSC Adv.*, 2015, **5**, 17087–17095; (k) P. Wadhwa, T. Kaur and A. Sharma, *RSC Adv.*, 2015, **5**, 44353–44360.



- 18 (a) F. P. L. Lim, G. Luna and A. V. Dolzhenko, *Tetrahedron Lett.*, 2014, **55**, 5159–5163; (b) F. P. L. Lim, G. Luna and A. V. Dolzhenko, *Tetrahedron Lett.*, 2015, **56**, 7016–7019; (c) F. P. L. Lim, G. Luna and A. V. Dolzhenko, *Tetrahedron Lett.*, 2015, **56**, 521–524; (d) A. V. Dolzhenko, S. A. Kalinina and D. V. Kalinin, *RSC Adv.*, 2013, **3**, 15850–15855; (e) S. A. Kalinina, D. V. Kalinin and A. V. Dolzhenko, *Tetrahedron Lett.*, 2013, **54**, 5537–5540; (f) F. P. L. Lim, S. T. Low, E. L. K. Ho, N. R. Halcovitch, E. R. T. Tiekink and A. V. Dolzhenko, *RSC Adv.*, 2017, **7**, 51062–51068.
- 19 (a) D. S. Ermolatev, E. P. Svidritsky, E. V. Babaev and E. Van der Eycken, *Tetrahedron Lett.*, 2009, **50**, 5218–5220; (b) D. S. Ermolat'ev and E. V. Van der Eycken, *J. Org. Chem.*, 2008, **73**, 6691–6697; (c) D. S. Ermolat'ev, B. Savaliya, A. Shah and E. Van der Eycken, *Mol. Diversity*, 2011, **15**, 491–496.
- 20 D. V. Kalinin, S. A. Kalinina and A. V. Dolzhenko, *Heterocycles*, 2013, **87**, 147–154.
- 21 Y. S. Agasimundin, F. T. Oakes and N. J. Leonard, *J. Org. Chem.*, 1984, **50**, 2474–2480.
- 22 Rigaku Oxford Diffraction, *CrysAlis PRO*, Yarnton, Oxfordshire, England, 2017.
- 23 G. M. Sheldrick, *Acta Crystallogr., Sect. A: Found. Crystallogr.*, 2008, **64**, 112–122.
- 24 L. J. Farrugia, *J. Appl. Crystallogr.*, 2012, **45**, 849–854.
- 25 G. M. Sheldrick, *Acta Crystallogr., Sect. C: Struct. Chem.*, 2015, **71**, 3–8.
- 26 K. Brandenburg, *DIAMOND, Crystal Impact GbR*, Bonn, Germany, 2006.
- 27 A. L. Spek, *Acta Crystallogr., Sect. D: Biol. Crystallogr.*, 2009, **65**, 148–155.

