



Cite this: *RSC Adv.*, 2018, 8, 23688

First-principles investigations of manganese oxide (MnO_x) complex-sandwiched bilayer graphene systems

Rafique Muhammad,^{ab} Yong Shuai,^{*a} Ahmed Irfan^b and Tan He-Ping^a

In this paper, we calculate the structural, electronic, magnetic and optical parameters of MnO_x ($x = 0-4$), cluster-sandwiched bilayer graphene (Gr) systems, utilizing first-principles calculations with van der Waals corrections implemented with density functional theory (DFT). Charge transfer is observed from the graphene layers to the MnO_x clusters, thus producing a hole doping phenomenon in the graphene layers. The MnO_x clusters' electronegative nature greatly modifies the electronic structure of bilayer graphene. It is observed that the MnO_x clusters' incorporation in bilayer graphene converts zero band gap semimetal bilayer graphene to a half metallic or dilute magnetic semiconducting material. Interestingly, the Gr/MnO₃/Gr complex structure displays indirect band gap semiconductor behavior for both spin channels, and has a ~ 20 meV band gap value. The band gap during spin up and spin down band channels increases as the size of MnO_x is increased in between the graphene layers. Through spin density diagrams, it is revealed that the MnO_x clusters' incorporation in the graphene layers converts nonmagnetic bilayer graphene to a magnetic substrate. The obtained magnetic moments for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr, sandwiched bilayer graphene systems were found to be 3.53 μ_B , 3.03 μ_B , 2.46 μ_B , 1.03 μ_B and 0.00 μ_B , respectively. Through density of states (DOS) plots, it is inferred that the d orbitals of the Mn atoms are mainly responsible for the generation of magnetic moments in the given bilayer graphene systems. The optical parameters, specifically absorption, reflectivity and refractive coefficients, were obtained for all given systems. The absorption spectrum of bilayer graphene is improved in the visible range when MnO_x clusters are sandwiched between the graphene layers. It is revealed that MnO_x clusters' incorporation in bilayer graphene improves these optical parameters in the low lying energy region. The results obtained during this study provide the basis for future experimental extrapolations to make multilayer graphene systems functional for optoelectronic and spintronic applications.

Received 23rd April 2018
Accepted 6th June 2018

DOI: 10.1039/c8ra03484b

rsc.li/rsc-advances

1. Introduction

Two dimensional (2D) monatomic graphene synthesized in 2004¹ with impressive electrical, electronic and thermal properties,²⁻⁵ has acquired lots of attention from the research community throughout the last decade. The linear dispersion relation near the Fermi energy (E_F) level makes graphene charge carriers act like massless Dirac fermions.⁶ It has been widely established that graphene-based magnetic substrates are more stable and efficient than traditional transition metal based magnets.⁷ Hence, 2D graphene materials have prospective applications for spintronic devices. Magnetism in graphene can be induced through intrinsic defects,⁸ doping nonmetal^{9,10} as well as transition metal impurities.¹¹⁻¹⁴ However, the graphene

charge carriers' mobility is reduced when individual impurities are added in its lattice, subjecting it to stronger interactions between the doped impurities and constituent carbon (C) atoms of graphene.¹⁵ Ideal doping impurities can be those which not only produce stable structures but also maintain the intrinsic behavior of graphene. Superhalogens can be regarded as ideal dopants, since these clusters possess higher electronegativity, with electron affinity values exceeding those of the halogen atoms.^{16,17} Attributed to their strong oxidation nature and high electronegativity,¹⁸ 3d metal tri and tetraoxide superhalogen complex-based doped graphene,^{19,20} h-BN layer²¹ and MoS₂ layer²² structures present higher stability compared to that of individual TM atom doped graphene, h-BN and MoS₂ monolayer structures. Recently, multilayered graphene structures have also attracted lots of interest, due to the possibility of producing a well-defined band gap by breaking the symmetry between constituent graphene layers.^{23,24} Different techniques such as external electrical field applications,^{25,26} and molecular doping in bilayer graphene,^{27,28} have been adopted to produce

^aSchool of Energy Science and Engineering, Harbin Institute of Technology, 92 West Dazhi Street, Harbin 150001, PR China. E-mail: shuaiyong@hit.edu.cn

^bMehran University of Engineering and Technology, SZAB Campus, Khairpur Mirs', Pakistan



a well-defined band gap in the bilayer graphene structures. In addition, some experimental studies have also been carried out on the intercalation of FeCl₃ clusters in graphite and few layer graphene (FLG) systems.^{29–31}

As mentioned earlier, substituting tri and tetraoxide superhalogen clusters in monolayer graphene produces stable structures. Keeping this in mind we extend this concept by incorporating MnO_x clusters in a bilayer graphene system. These MnO_x clusters were sandwiched in between two graphene layers. Recently, X. Zhang *et al.*³² performed first-principles calculations on MnF₃₍₄₎ cluster-sandwiched bilayer graphene systems. The authors reported that these structures display ferromagnetic behavior and that the MnF₃₍₄₎ clusters greatly improve the electronic properties of the bilayer graphene systems without affecting the intrinsic nature of graphene. Since these systems are based on only magnetic superhalogen clusters, in this work we provide a detailed study on individual the Mn atom and different MnO_x cluster (*i.e.* MnO, MnO₂, MnO₃, MnO₄) sandwiched bilayer graphene systems. The MnO_x clusters' substitution in bilayer graphene serves two purposes, (i) these clusters significantly modify the electronic and magnetic parameters, attributed to the hole doping phenomenon induced in both graphene layers by the MnO_x clusters, (ii) weaker interaction between the MnO_x clusters and graphene layers maintain the intrinsic behavior of graphene. These Gr/MnO_x/Gr structures can be regarded as being suitable for spintronic devices such as spin field-effect transistors (FET) or vertical (FET) devices.

Since, MnO_x cluster substitution in bilayer graphene modifies its electronic structure, it also modifies the interband and intraband transitions, which in turn can alter the optical parameters. Hence, in addition to the electronic and magnetic parameters, we also investigated the optical properties – specifically the absorption coefficient, reflectivity and refractive coefficient of MnO_x cluster-incorporated bilayer graphene systems. The optical parameters of monolayer graphene and h-BN systems have been widely investigated.^{33–35} However, no complete account for the optical parameters of bilayer graphene systems is available. Hence, this study provides a substantial contribution towards tuning the optical parameters of bilayer graphene systems through incorporation of foreign clusters.

This work presents the structural, electronic, magnetic and optical parameters of MnO_x cluster-sandwiched bilayer graphene systems through first-principles DFT calculations. This paper is arranged as follows; computational details are produced in Section 2, the obtained results and their general discussion are elaborated in Section 3 and this work is finally concluded in Section 4. It can be presumed that the structural, electronic, magnetic and optical characteristics of such a system has not been defined comprehensively. The results provided in this paper can provide a new approach for tuning the above-mentioned properties, in order to make multilayered graphene systems which are applicable for nanoelectronic and spintronic devices, distinctive to single layered graphene systems.

2. Computational details

First-principles calculations based on DFT were utilized to investigate the structural, electronic, magnetic and optical parameters of MnO_x cluster-sandwiched bilayer graphene systems using the generalized-gradient approximation (GGA) implemented in the Vienna Ab-initio simulation package (VASP).^{36,37} VASP adopts projector augmented wave (PAW) potentials³⁸ with the Perdew–Burke–Ernzerhof (PBE) exchange–correlation functional³⁹. All the calculations were performed in spin-polarized mode. The Grimme (DFT-D2)⁴⁰ method was employed for van der Waals (vdW) corrections due to its appropriate account of long-range vdW interactions.^{41–43} A 500 eV cutoff energy was applied for the wave function to be expanded in plane waves. A 4 × 4 bilayer graphene supercell was adopted in which a vacuum layer thickness of 20 Å in the Z-direction was added to remove the interference between neighboring layers. A 9 × 9 × 1 *T*-centered *k*-point mesh for Brillouin zone (BZ) sampling was employed to achieve good convergence parameters. Hellmann–Feynman forces of less than 0.01 eV Å⁻¹ and a total change in energy of less than 10⁻⁶ eV were selected for the geometry optimization process. The Gaussian smearing method was utilized to avoid the problems related to partial occupancies. To remove the inaccuracies generated from electrostatic potential, atomic forces and the total energy caused by periodic boundary conditions, the dipole correction technique was also adopted.⁴⁴

3. Results and discussion

3.1 Structural and magnetic properties of MnO_x cluster-sandwiched bilayer graphene

Bilayer graphene structures were incorporated with an individual Mn atom and different MnO_x (*i.e.* MnO, MnO₂, MnO₃ and MnO₄) clusters, as described in Fig. 1(a)–(e), respectively. In this section, the structural and magnetic parameters of these systems were investigated using the first-principles method.

The bond distances of the C–C atoms, Mn–O atoms and the distance of the MnO_x clusters from both graphene layers is also shown in Fig. 1(a)–(e), respectively. After performing the geometry relaxation process on these systems, it is established that the C atoms lying above and below the MnO_x clusters drift away from the graphene 2D plane, and thus a change in the C–C atoms' bond lengths occurs. After incorporation of the MnO_x clusters in bilayer graphene, the distance between the graphene layers also increases. It is interesting to observe that the 2D planar structure of both graphene layers is maintained, which suggests that the intrinsic behavior of the graphene layer will remain unaltered. From the structures given in Fig. 1(a)–(e), it is observed that the sandwiched Mn atom of the MnO_x cluster tries to maintain its position just below the C atoms of the graphene layer. Moreover, these MnO_x clusters try to reduce their distance from a single graphene layer (either the upper or lower layer) as evident by the bond lengths of Mn–C atoms described in Fig. 1(a)–(e), respectively.^{32,45}

The total magnetization of MnO_x cluster-sandwiched bilayer graphene supercells, magnetic moments of individual the Mn



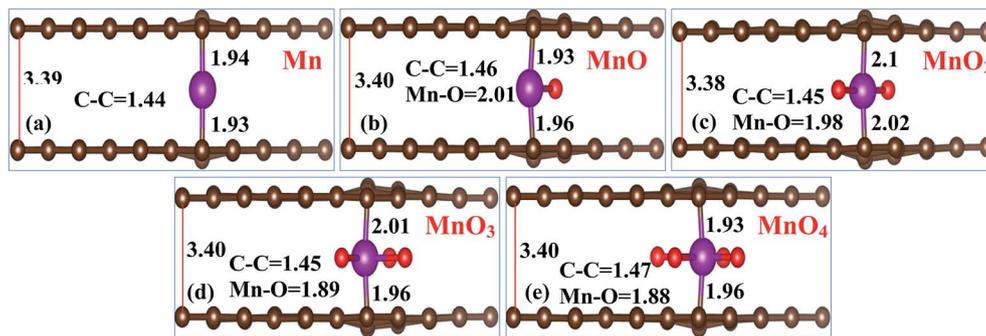


Fig. 1 Side view of the atomic structures of MnO_x cluster-sandwiched bilayer graphene systems (where, $x = 0-4$). The small black and red balls represent the C and O atoms respectively, and the big ball represents the Mn atom.

atom and the average equatorial bond lengths of the Mn–C, Mn–O and C–C atoms for all the MnO_x cluster-sandwiched bilayer graphene systems are presented in Table 1. After MnO_x cluster incorporation in bilayer graphene, the Mn–C bond lengths were in between 1.93 and 2.1 Å, the C–C bond lengths were in the range of 1.44–1.47 Å and the obtained variation for the C–C bond lengths was in between 0.2 and 0.5 Å. The obtained total magnetic moments of different Gr/ MnO_x /Gr systems listed in Table 1, clearly indicate that MnO_x cluster incorporation into bilayer graphene significantly improves its ferromagnetic behavior as demonstrated by the larger magnetic moment values for each given system. Only the MnO_4 cluster-sandwiched bilayer graphene carries a 0.00 μ_B magnetic moment. This non-magnetic behavior of the Gr/ MnO_4 /Gr system, can be attributed to quenching of the magnetic moment of the Mn atom by the four O atoms present around the Mn atom. The negative magnetic moment of the O atoms nullifies the magnetic moments of the Mn atoms thus producing a 0.00 μ_B net magnetic moment for Gr/ MnO_4 /Gr system.^{20,46}

In order to determine whether the given Gr/ MnO_x /Gr systems are thermodynamically stable enough or not, we calculated the binding energy for each system using the following expression³² and the obtained binding energies are presented in Table 1,

$$E_b = E_{(\text{gp}+\text{MnO}_x)} - (2E_{(\text{gp})} + E_{\text{MnO}_x}) \quad (1)$$

where the terms $E_{(\text{gp})}$, E_{MnO_x} and $E_{(\text{gp}+\text{MnO}_x)}$ represent the total energy of pure monolayer graphene, the total energy of the

Table 1 Total magnetization of the supercell (μ_{tot} , in μ_B), magnetic moments of the individual Mn atom, the binding energies E_b (eV) and bond distances of the Mn–C ($d_{\text{Mn-C}}$, in Å), C–C ($d_{\text{C-C}}$, in Å) and Mn–O ($d_{\text{Mn-O}}$, in Å) atoms for all MnO_x cluster-sandwiched bilayer graphene systems

Impurity	μ_{tot} in μ_B	μ_{Mn} in μ_B	$d_{\text{Mn-C}}$ (Å)	$d_{\text{C-C}}$ (Å)	$d_{\text{Mn-O}}$ (Å)	E_b (eV)
Mn	3.53	3.054	1.94	1.44	—	2.52
MnO	3.03	3.026	1.96	1.46	2.01	2.91
MnO ₂	2.46	2.5	2.1	1.45	1.98	3.08
MnO ₃	1.03	1.5	2.01	1.45	1.89	3.85
MnO ₄	0.00	0.00	1.96	1.47	1.88	4.31

MnO_x clusters and the total energy of the MnO_x cluster-sandwiched bilayer graphene systems, respectively. The obtained binding energies for different cluster-sandwiched bilayer graphene systems presented in Table 1 clearly indicate that superhalogen cluster incorporation in bilayer graphene is thermodynamically favorable as most of the systems contain positive binding energies. These predictions are consistent with previous reports.^{32,47–49}

The obtained total magnetic moments μ_{tot} and individual Mn atom magnetic moments μ_{Mn} provided in Table 1, clearly demonstrate that incorporation of the MnO_x cluster in bilayer graphene is a suitable way to make it a magnetic material. Only the MnO_4 cluster-sandwiched graphene system displays non-magnetic behavior. To further support our predictions of the magnetic properties of MnO_x cluster-sandwiched bilayer graphene systems, we elaborated on the magnetic coupling nature of the Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr and Gr/MnO₃/Gr bilayer graphene systems and the resulting diagrams are presented in Fig. 2(a)–(d), respectively. The Gr/ MnO_4 /Gr system was omitted due to its 0.00 μ_B magnetic moment. Fig. 2(a)–(d) describes the top and side views of the spin densities ($\rho \uparrow - \rho \downarrow$) of the different MnO_x cluster-sandwiched bilayer graphene systems.

The spin density diagrams provided in Fig. 2(a)–(d) suggest that, the sandwiched MnO_x clusters induce magnetic moments on the C atoms of the graphene layer lying above or below the impurity clusters. A parallel spin polarization direction is achieved between the MnO_x clusters and C atoms of bilayer graphene. Moreover, the unfilled d-orbitals of Mn atoms carry clockwise spin direction as shown by the yellow isosurface ($0.002 \text{ e } \text{Å}^{-3}$), for all the given MnO_x clusters except for the MnO_3 cluster. The direction of spin polarization between Mn and O atoms was antiparallel for the MnO and MnO_2 clusters, while it was parallel for the MnO_3 cluster as shown in Fig. 2(b)–(d). From the spin density diagrams it can be further presumed that despite having weaker interactions between the MnO_x clusters and graphene layers, the distortion in the spin density observed in both graphene layers was not localized at the periphery of the MnO_x clusters. Rather, the spin density was distributed on both the graphene layers.^{32,48,50}

The charge density differences and charge transfer phenomena for all MnO_x cluster-sandwiched bilayer graphene systems were investigated through Bader analysis,^{51,52} in order



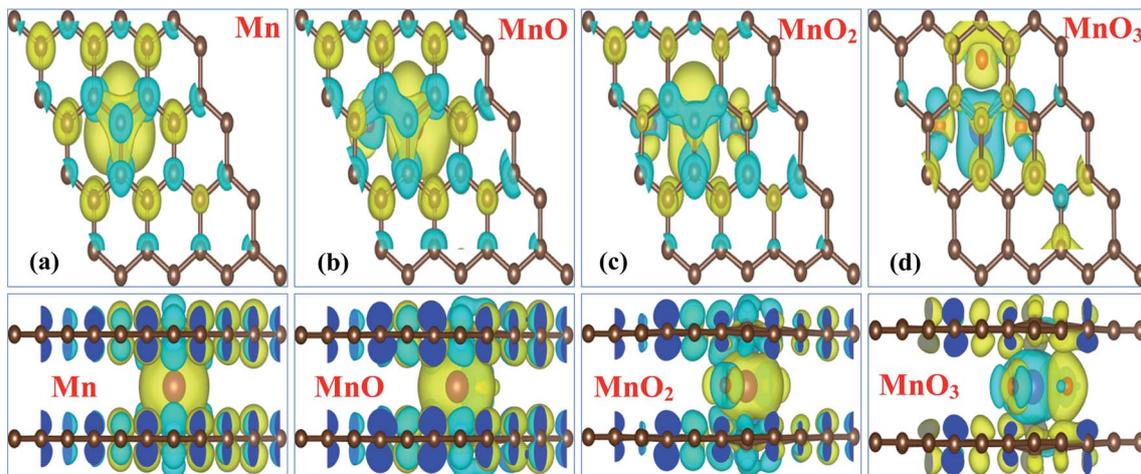


Fig. 2 Top and side views of the spin density diagrams for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr and Gr/MnO₃/Gr cluster-sandwiched bilayer graphene systems. Positive and negative spin densities are described by yellow and cyan isosurfaces ($0.002 \text{ e } \text{Å}^{-3}$), respectively. The blue color appearing at the edges is generated due to interference of the neighboring graphene layers.

to further understand the behaviors of MnO_x clusters incorporated in bilayer graphene systems. The charge density difference can be described by $\Delta\rho = \rho_{\text{MnO}_x\text{-BL/graphene}} - \rho_{\text{BL/graphene}} - \rho_{\text{MnO}_x}$. Where $\rho_{\text{MnO}_x\text{-BL/graphene}}$, $\rho_{\text{BL/graphene}}$ and ρ_{MnO_x} represent the charge density of the MnO_x cluster-sandwiched bilayer graphene systems, the charge density of the graphene layers and the charge density of the MnO_x clusters respectively. The

resulting charge density difference diagrams for all MnO_x cluster-sandwiched bilayer graphene systems are shown in Fig. 3(a)–(e). Electron gain and loss configurations are presented by the yellow ($0.0003 \text{ e } \text{Å}^{-3}$) and cyan ($0.0003 \text{ e } \text{Å}^{-3}$) isosurfaces, respectively. The charge density difference diagrams of Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr cluster-sandwiched bilayer graphene systems

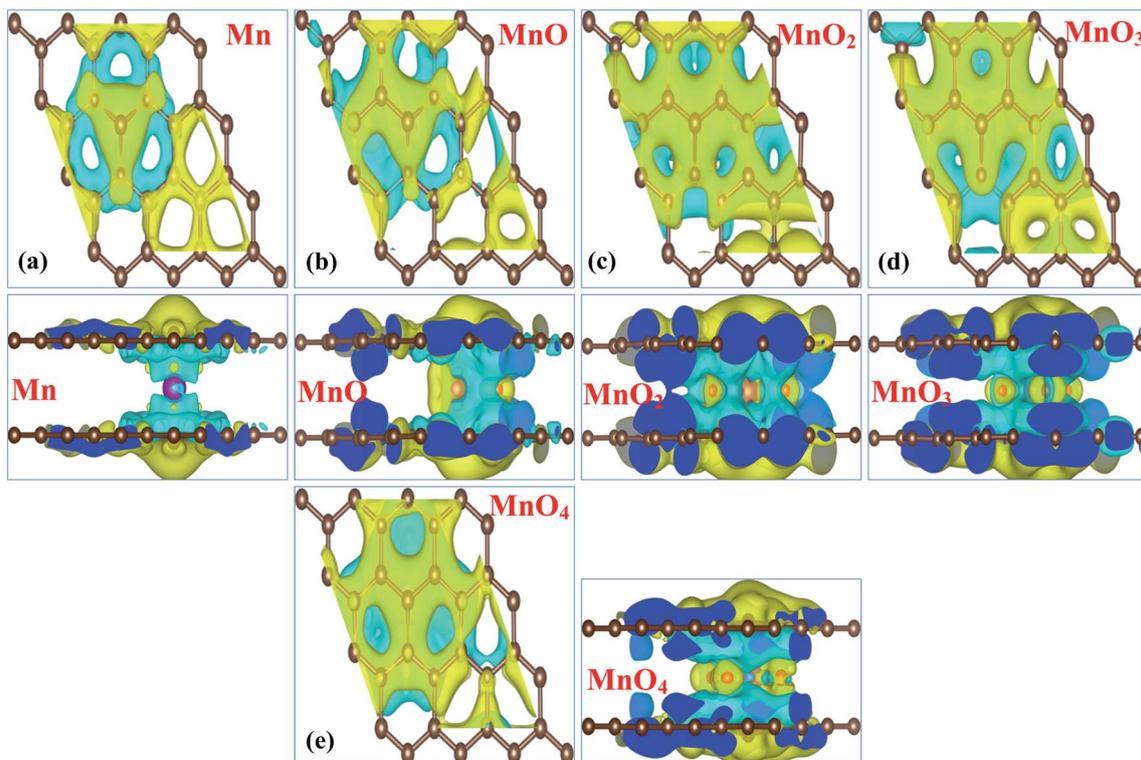


Fig. 3 Top and side views of charge density difference diagrams for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr cluster-sandwiched bilayer graphene systems. Electron gain and electrons loss configurations are described by yellow and cyan isosurfaces ($0.0003 \text{ e } \text{Å}^{-3}$), respectively. The blue color appearing at the edges is generated due to interference of the neighboring graphene layers.



presented in Fig. 3(a)–(e) describe a similar nature of charge difference between the graphene layers and MnO_x clusters. A charge depletion pattern (*i.e.* the cyan isosurface $0.0003 \text{ e } \text{\AA}^{-3}$) is observed around the MnO_x clusters, while individual the Mn and O atoms are filled with yellow isosurface ($0.0003 \text{ e } \text{\AA}^{-3}$), suggesting electron gain behavior of the impurity clusters. This partial distribution of the cyan and yellow isosurfaces in between bilayer graphene, as visible in Fig. 3(a)–(e), suggests that the interaction between graphene layers and the MnO_x clusters is covalent in nature. Moreover, it can be suggested that the direction of charge transfer is from the graphene layers to the MnO_x clusters. As the bilayer graphene transfers its charge carriers to the impurity clusters, it can also be presumed that the hole doping process is initiated in both the graphene layers. These results are consistent with previously reported results.^{21,53–55}

3.2 Electronic structures of MnO_x cluster-sandwiched bilayer graphene systems

Here, we investigate the electronic structures of MnO_x cluster-sandwiched bilayer graphene systems through first-principles calculations. Spin polarized band structures and projected DOS plots are determined for all MnO_x cluster-sandwiched bilayer graphene systems. A thirty K -points grid is adopted along the path Γ – M – K – Γ in the IBZ in order to obtain the band

structure diagrams with fine quality. $9 \times 9 \times 1$ Γ -centered BZ sampling is applied for calculating the spin polarized projected (PDOS) plots for all the Gr/ MnO_x /Gr sandwiched bilayer graphene systems. A Gaussian width of 0.02 eV was utilized for smearing the energy eigenvalues.

Spin polarized electronic structures of MnO_x cluster-sandwiched bilayer graphene systems are provided in Fig. 4(a)–(e). For comparison, the electronic structure of pure bilayer graphene is also shown in Fig. 4, named ‘pure bilayer (BL) graphene’. The well recognized quadratic dispersion of energy near the high symmetric K -points and the straddling of valence and conduction bands at the Fermi energy level (E_F) is visible in the electronic structure of bilayer graphene shown in Fig. 4. The electronic structure of pure bilayer graphene calculated during this work is in consensus with previous studies.^{56–58} Significant change in the electronic structure of bilayer graphene is predicted after MnO_x cluster incorporation. The Dirac cone appearing at the high symmetric K -point shifts down to the valence band by ~ 0.9 eV and ~ 1 eV during spin up and spin down channels, respectively for the Gr/Mn/Gr sandwiched bilayer graphene systems as shown in Fig. 4(a). For the Gr/MnO/Gr system, the Dirac cone also creeps down to the valence band by ~ 0.5 eV and ~ 0.8 eV during spin up and spin down channels, respectively. During this, a band gap of ~ 0.02 eV and ~ 0.04 eV appears at the high symmetric K -point during spin up and spin down channels, respectively as is evident in Fig. 4(b). For the Gr/

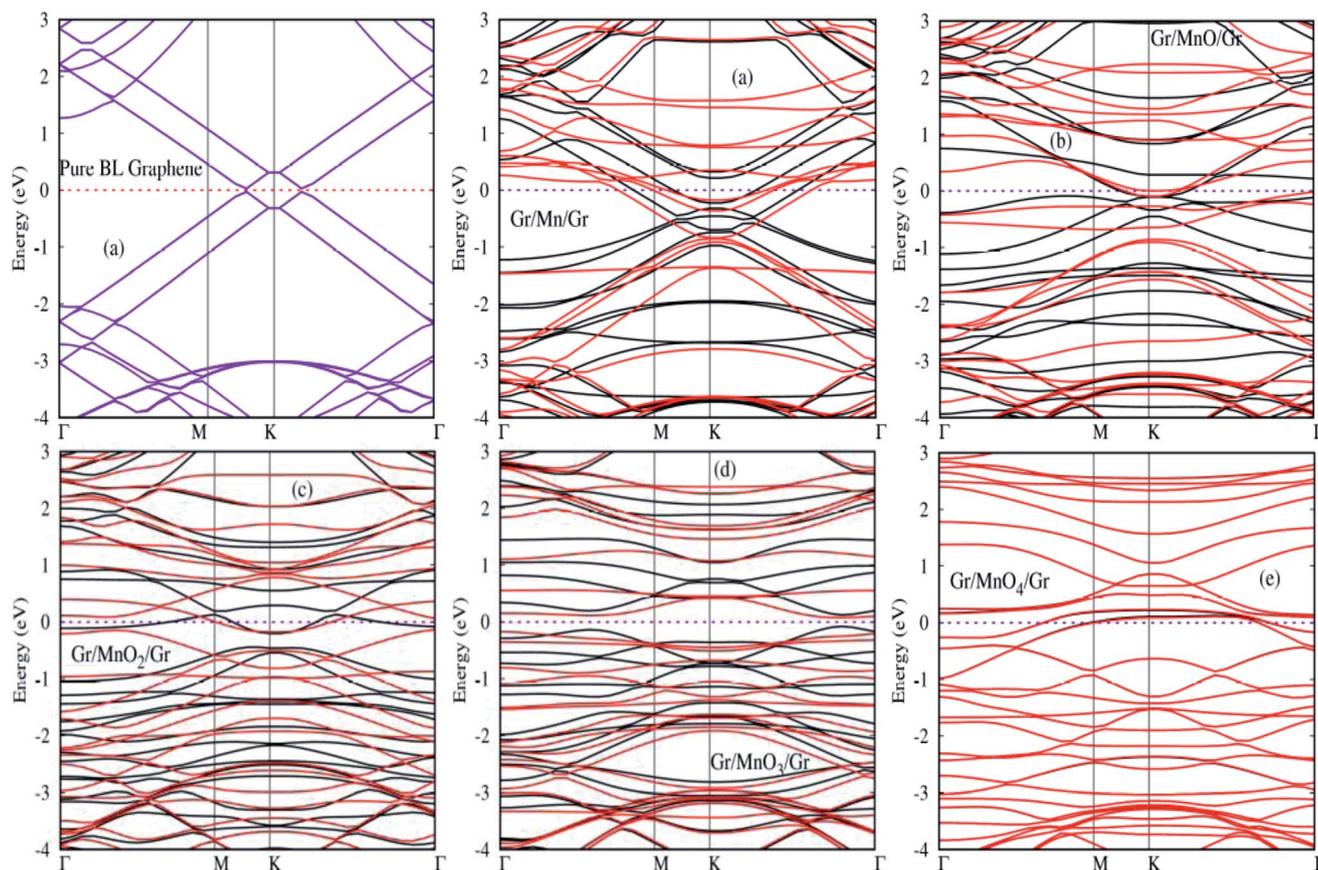


Fig. 4 Spin polarized band structures for MnO_x cluster-sandwiched bilayer graphene (4×4) supercell structures.



MnO₂/Gr system, the Dirac cone shifts into the valence band by ~ 0.2 eV for both spin up and spin down channels and a band gap of ~ 0.03 eV is produced at the high symmetric *K*-point for both channels as shown in Fig. 4(c). Interestingly, the Gr/MnO₃/Gr bilayer graphene system displays indirect band gap semiconductor behavior for both spin up and spin down band channels. The band gap achieved during spin up and spin down band channels for this system was found to be ~ 1.1 eV and ~ 0.6 eV as shown in Fig. 4(d). As listed in Table 1, the Gr/MnO₄/Gr bilayer system possesses a $0.00 \mu_B$ magnetic moment; hence its bands appear at the same energy level without having any polarization. From Fig. 4(e), it can be maintained that Gr/MnO₄/Gr exhibits half metallic properties as some impurity states are observed at the (E_F) level. The obtained band gap values for different Gr/MnO_{*x*}/Gr bilayer graphene systems have to be considered carefully, since the DFT method slightly underestimates the electronic band gap. Results provided in Fig. 4 are consistent with previous reports.^{32,57,58}

We also investigated the spin polarized total and projected PDOS plots for all the Gr/MnO_{*x*}/Gr cluster-sandwiched bilayer graphene systems, in order to determine the origins of the magnetic moments and the electronic hybridization between the MnO_{*x*} clusters and graphene layers. The TDOS and PDOS on the d orbitals of the Mn atom, the p orbitals of the O atoms and the orbitals of the C atoms of the graphene layers were

calculated and are presented in Fig. 5(a)–(e), respectively. In these plots, the E_F level is described by a solid grey line drawn at 0 eV energy in the respective PDOS plots. It is assumed that some surface states will be induced by the MnO_{*x*} clusters in between the valence and conduction bands, as can be observed in Fig. 5(a)–(e). When covalent bonding occurs between the graphene layers and MnO_{*x*} clusters, it produces a robust hybridization which, in turn, induces some surface states at the E_F level. As the Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr cluster-sandwiched bilayer graphene systems carried $3.53 \mu_B$, $3.03 \mu_B$, $2.46 \mu_B$, $1.03 \mu_B$ and $0.00 \mu_B$ magnetic moments, respectively, the C and O atom p orbitals and Mn atom d orbitals of the Gr/MnO_{*x*}/Gr structures which contain finite magnetic moments, displayed spin polarization. Only the Gr/MnO₄/Gr cluster-sandwiched bilayer graphene system does not contain any spin polarization in its orbitals since it possesses a 0.00 magnetic moment. The above statements can be verified through the PDOS plots presented in Fig. 5(a)–(e). The PDOS plots of Fig. 5(a)–(e) support the prediction that the magnetic moments of the MnO_{*x*} cluster-sandwiched bilayer graphene systems are induced by five different orbitals (d_{xy} , d_{yz} , d_{z^2} , d_{xz} and $d_{x^2-y^2}$) of the Mn atom. For all the MnO_{*x*} cluster-sandwiched bilayer graphene systems, the d_{z^2} , d_{xy} and $d_{x^2-y^2}$ orbitals of the Mn atoms do not hybridize with the O atom p orbitals, while the d_{xy} and d_{yz} orbitals of the Mn atom produce soft hybridization with the O atom p

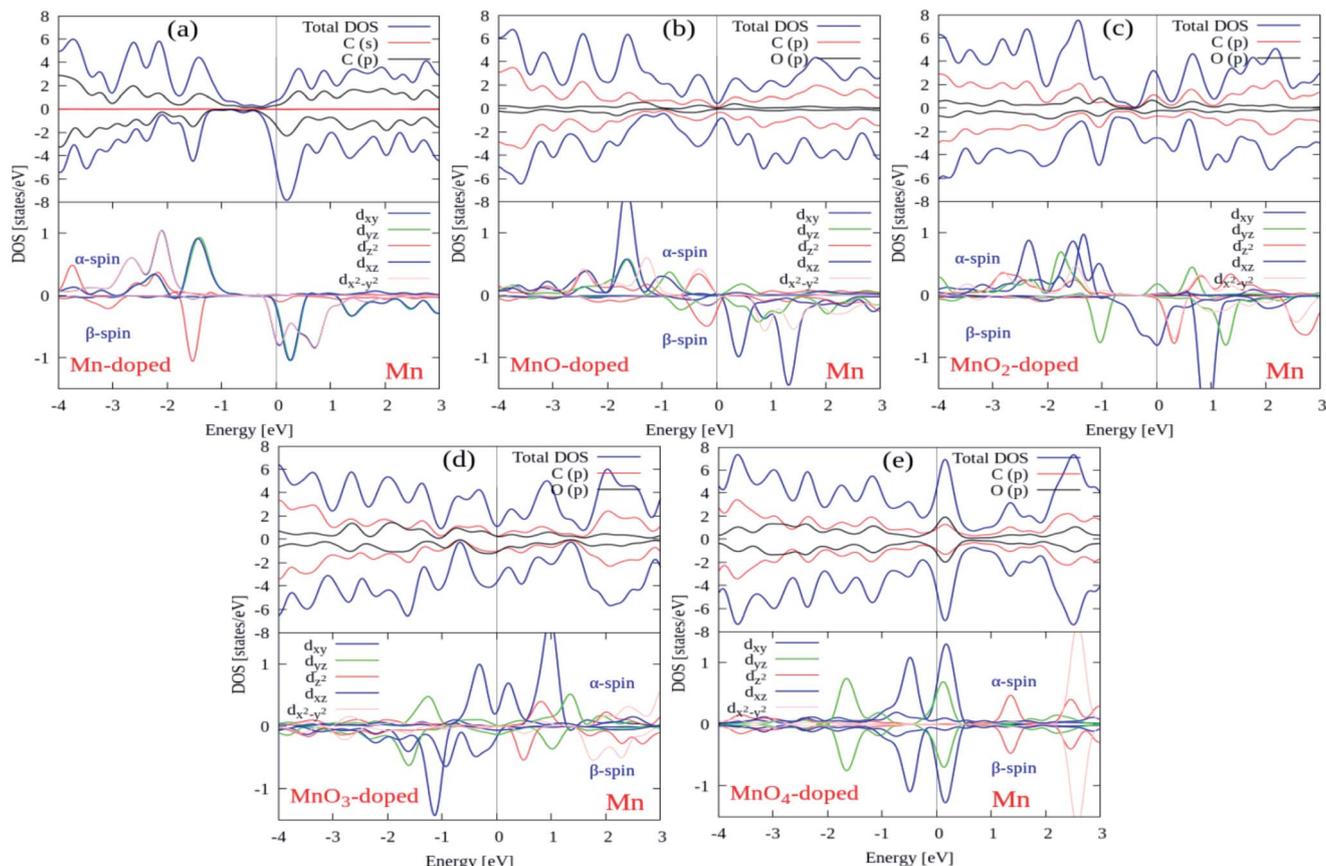


Fig. 5 Total and projected DOS plots for Gr/MnO_{*x*}/Gr cluster-sandwiched bilayer graphene systems. The solid grey line appearing at 0 eV energy level depicts the Fermi energy level.



orbitals as described in Fig. 5(a)–(e). These plots further support the predictions that the opening of the band gap at a high symmetric K -point occurs at different energy levels during both the spin up and spin down band channels as is evident in Fig. 4(a)–(e) as well. From the electronic structure and PDOS plots it can be concluded that 2D magnetic substrates can be synthesized by incorporating superhalogen MnO_x clusters in bilayer graphene systems.

3.3 Optical properties of MnO_x cluster-sandwiched bilayer graphene

Finally, we determine the effects of MnO_x cluster incorporation on the optical characteristics of bilayer graphene through the Random Phase Approximation (RPA)⁵⁹ approach implemented in DFT. Local field effects are not included, while interband transitions are considered in this technique, hence some inaccuracies can be observed in the dielectric constant parameter in the lower energy range. For the optical parameters calculations, we adopted $11 \times 11 \times 1$ Γ -centered BZ sampling, thus our obtained optical parameters contain some distortions in their profiles. If the number of K -points is increased, the profile can easily be smoothed, but a lack of available computational power limited our K -point mesh to smaller values.

Optical parameters, specifically absorption coefficient ' α ', reflectivity ' R ', refraction index ' n ' and extinction coefficient ' k ' were calculated during this study. These parameters can easily be extracted from the dielectric constant values. The summation of real and imaginary parts *i.e.* $\epsilon = \epsilon' + i\epsilon''$ together provides the dielectric constant parameter. The imaginary dielectric tensor can be determined through the summation of empty states as provided in the following expression,

$$\epsilon''_{\alpha\beta}(\omega) = \frac{4\pi^2 e^2}{\Omega} \lim_{q \rightarrow 0} \frac{1}{q^2} \sum_{c,v,k} 2\omega_k \delta(E_{ck} - E_{vk} - \omega) \times \langle u_{ck+e_\alpha q} | u_{vk} \rangle \langle u_{ck+e_\beta q} | u_{vk} \rangle \quad (2)$$

here, the α and β indices are the Cartesian components, e_α and e_β represent the unit vectors along the three directions, c and v correspond to the conduction and valence bands, respectively. The respective energy of the conduction and valence bands are

denoted by E_{ck} and E_{vk} and the u_{ck} term depicts the cell periodic part of the orbital at the k -point K .

Using the Kramers–Kronig transformation, the real dielectric tensor can be obtained as

$$\epsilon'_{\alpha\beta}(\omega) = 1 + \frac{2}{\pi} P \int_0^\infty \frac{\epsilon''_{\alpha\beta}(\omega') \omega'}{\omega'^2 - \omega^2 + i\eta} d\omega' \quad (3)$$

here, P defines the principle value. The aforementioned technique is extensively elaborated on in ref. 59.

Through these dielectric tensor values, the absorption coefficient ' α ', reflectivity ' R ', refraction index ' n ' and extinction coefficient ' k ' can easily be determined. This is explained in detail in ref. 33.

The refractive index ' n ' and extinction coefficient ' k ' plots for the pure and MnO_x cluster-sandwiched bilayer graphene systems are presented in Fig. 6(a) and (b). These plots are generated in a single plot to demonstrate the comparison between the pure and MnO_x cluster-sandwiched bilayer graphene systems. The static refraction ' n ' and extinction ' k ' (*i.e.* the ' n ' and ' k ' values observed at 0 eV energy) values were found to be 2.7 and 0 for pure bilayer graphene. When MnO_x clusters were sandwiched in bilayer graphene, the static refraction values were increased but the static extinction coefficient value remained same, as shown in Fig. 6(a) and (b). The static refractive index values for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr cluster-sandwiched bilayer graphene systems were found to be 4.8, 4.75, 5.01, 6.5 and 3.6 respectively, as shown in Fig. 6(a).

Furthermore, the first extinction peak intensity for pure bilayer graphene is observed at 2 eV with a peak intensity of 2.4 as shown in Fig. 6(b), but this peak is shifted to lower energy values after MnO_x cluster incorporation in bilayer graphene. This indicates a red shift in the extinction coefficient of bilayer graphene. The value of first extinction peak maintains its intensity in the range of 1.8–2.8 and in energy interval of 0–1.7 eV, depending upon the type of MnO_x cluster-sandwiched in between bilayer graphene as evident in Fig. 6(b).^{60,61} Since the extinction coefficient is also linked to the absorption coefficient parameter, the red shift produced in the extinction coefficient will reflect a red shift in the absorption coefficient as well, and this is also demonstrated in the upcoming section.

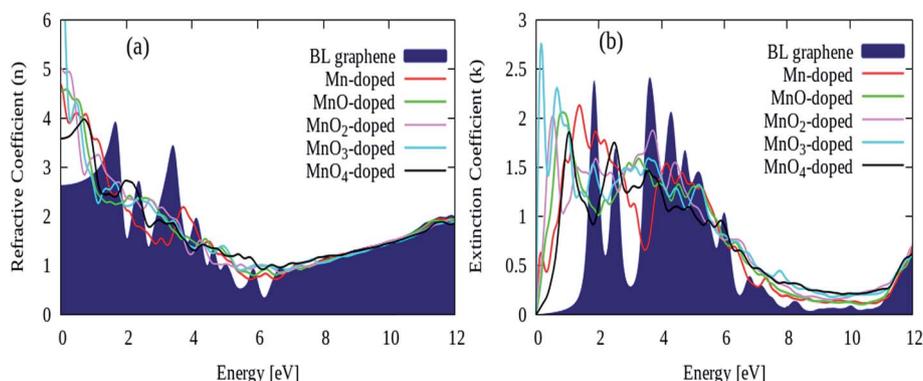


Fig. 6 (a) Refractive index ' n ' and (b) extinction coefficient ' k ' values of pure and different Gr/ MnO_x /Gr cluster-sandwiched bilayer graphene systems.



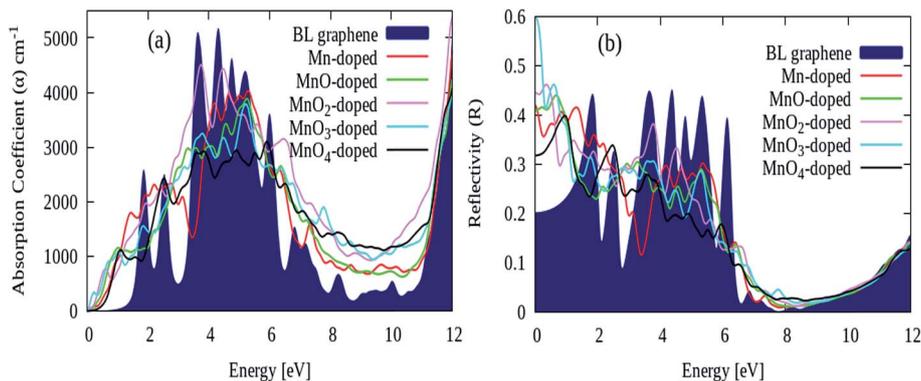


Fig. 7 (a) Absorption coefficient ' α ' and (b) reflectivity ' R ' parameters of pure and different Gr/MnO_x/Gr cluster-sandwiched bilayer graphene systems.

Fig. 7(a) demonstrates the absorption coefficient ' α ' parameter of pure and different MnO_x cluster-sandwiched bilayer graphene systems. The pure bilayer graphene absorption coefficient, shown in navy, depicts a value of zero in between the 0 and 1 eV energy interval as shown in Fig. 7(a). The absorption coefficient was described in the energy range of a 0–12 eV energy interval, in order to determine the modification in the absorption spectrum in the visible range. The first main absorption peak induced by low energy $\pi \rightarrow \pi^*$ transitions is observed at ~ 4 eV energy with ~ 5000 peak intensity. After MnO_x cluster incorporation, the absorption spectrum of bilayer graphene starts to increase in value from 0 eV energy, indicating a rise in the absorption coefficient in the lower energy range. As observed with the extinction coefficient, a red shift in the absorption coefficient quantity is also observed, as shown in Fig. 7(a). Moreover, the first minimum peak observed at ~ 4 eV energy reduces in intensity when MnO_x clusters are sandwiched in bilayer graphene. It is worth noting that the absorption coefficient quantity also increases in value in the 6–11 eV energy range after the incorporation of MnO_x clusters in bilayer graphene as shown in Fig. 7(a). It can be concluded that the incorporation of MnO_x clusters significantly improves the absorption coefficient of bilayer graphene in the visible range of the spectrum.^{62,63}

Similarly, the reflectivity parameters of pure and different MnO_x cluster-sandwiched bilayer graphene systems are presented in Fig. 7(b). The static reflectivity of pure bilayer graphene is found to be 0.2, while MnO_x cluster incorporation increases this parameter to 0.41, 0.4, 0.44, 0.6 and 0.31 for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr cluster-sandwiched bilayer graphene systems, respectively. Overall, it is observed that the static reflectivity is improved while the reflectivity parameter decreases in value over a higher energy range as demonstrated in Fig. 7(b).^{64,65}

4. Conclusion

Different Gr/MnO_x/Gr cluster-sandwiched bilayer graphene systems were examined by first-principles DFT calculations. The high electronegative nature of the MnO_x clusters supports the

stability of bilayer graphene as shown by the calculated binding energies. Spin densities of Gr/MnO_x/Gr cluster-sandwiched systems suggest that the electron rich nature of MnO_x clusters induces positive spin electrons into the graphene layers and significant magnetic behavior can be expected from these complex structures. Even with weaker electronic interactions between C and MnO_x clusters, stronger magnetic coupling can be achieved for the MnO_x sandwiched systems. The obtained magnetic moments for Gr/Mn/Gr, Gr/MnO/Gr, Gr/MnO₂/Gr, Gr/MnO₃/Gr and Gr/MnO₄/Gr complexes were found to be 3.53 μ_B , 3.03 μ_B , 2.46 μ_B , 1.03 μ_B and 0.00 μ_B respectively. The direction of charge transfer is from bilayer graphene to the MnO_x clusters, thus inducing a hole doping process in bilayer graphene.

Through spin polarized electronic structures, it is revealed that the MnO_x clusters' incorporation greatly modifies the band structure of graphene; firstly by moving the Dirac cone up/down the E_F level along with inducing a finite band gap at the high symmetric K -point. Interestingly, Gr/MnO₃/Gr sandwiched bilayer graphene depicts an indirect band gap semiconducting property with a band gap of ~ 1.1 eV and ~ 0.6 eV during spin up and spin down band channels, respectively. The obtained PDOS plots further reveal that the sandwiched systems with finite magnetic moments display orbital polarization, along with a finite band gap at different energy levels during spin up and spin down bands for given systems. Moreover, it is predicted that five different orbitals (*i.e.* d_{xy} , d_{yz} , d_{z^2} , d_{xz} and $d_{x^2-y^2}$) of the Mn atom are mainly responsible for the induction of magnetic moments in bilayer graphene. This spin-polarized electronic behavior of the Gr/MnO_x/Gr sandwiched systems suggests that the given systems can be utilized for spintronic device applications.

The optical parameters of MnO_x cluster-sandwiched bilayer graphene systems reveal that, the static ' n ' and ' R ' parameters of MnO_x sandwiched systems attain larger values in comparison to the static ' n ' and ' R ' parameters of pure bilayer graphene. Similarly, the ' α ' and ' k ' parameters of the MnO_x sandwiched systems demonstrate a red shift and these two parameters are improved in the visible range. A combination of the electronic, magnetic and optical parameters of MnO_x cluster-sandwiched bilayer graphene systems revealed that these Gr/MnO_x/Gr



cluster-sandwiched structures display significant potential for applications in novel spintronic and optoelectronic device applications, which are distinctive to pure bilayer graphene structures.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (No. 51522601, 51421063) and the Higher Education Commission, Pakistan under SRGP (No. 21-1778/SRGP/R&D/HEC/2017). In addition, we would like to acknowledge the support that NVIDIA provided us through the GPU Grant Program.

References

- 1 K. S. Novoselov, A. K. Geim, S. V. Morozov, D. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *Science*, 2004, **306**, 666–669.
- 2 A. K. Geim and K. S. Novoselov, *Nat. Mater.*, 2007, **6**, 183–191.
- 3 F. Guinea, N. Peres, K. Novoselov, A. Geim and A. C. Neto, *Rev. Mod. Phys.*, 2009, **81**, 109–162.
- 4 Z. Jiang, Y. Zhang, Y.-W. Tan, H. Stormer and P. Kim, *Solid State Commun.*, 2007, **143**, 14–19.
- 5 S. D. Sarma, S. Adam, E. Hwang and E. Rossi, *Rev. Mod. Phys.*, 2011, **83**, 407.
- 6 K. Novoselov, A. K. Geim, S. Morozov, D. Jiang, M. Katsnelson, I. Grigorieva, S. Dubonos and A. Firsov, *nature*, 2005, **438**, 197–200.
- 7 T. Alonso-Lanza, A. Ayuela and F. Aguilera-Granja, arXiv preprint arXiv:1606.00165, 2016.
- 8 X. Cui, R. Zheng, Z. Liu, L. Li, B. Delley, C. Stampfl and S. P. Ringer, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 125410.
- 9 D. Soriano, N. Leconte, P. Ordejón, J.-C. Charlier, J.-J. Palacios and S. Roche, *Phys. Rev. Lett.*, 2011, **107**, 016602.
- 10 K. M. McCreary, A. G. Swartz, W. Han, J. Fabian and R. K. Kawakami, *Phys. Rev. Lett.*, 2012, **109**, 186604.
- 11 K. T. Chan, J. Neaton and M. L. Cohen, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **77**, 235430.
- 12 Z. Qiao, S. A. Yang, W. Feng, W.-K. Tse, J. Ding, Y. Yao, J. Wang and Q. Niu, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **82**, 161414.
- 13 J. Ding, Z. Qiao, W. Feng, Y. Yao and Q. Niu, arXiv preprint arXiv:1108.6235, 2011.
- 14 H. Zhang, C. Lazo, S. Blügel, S. Heinze and Y. Mokrousov, *Phys. Rev. Lett.*, 2012, **108**, 056802.
- 15 A. Krasheninnikov, P. Lehtinen, A. S. Foster, P. Pyykkö and R. M. Nieminen, *Phys. Rev. Lett.*, 2009, **102**, 126807.
- 16 G. Gutsev and A. Boldyrev, *Adv. Chem. Phys.*, 1985, **61**, 169–221.
- 17 G. Gutsev and A. Boldyrev, *Chem. Phys.*, 1981, **56**, 277–283.
- 18 K. Pradhan, G. L. Gutsev, C. A. Weatherford and P. Jena, *J. Chem. Phys.*, 2011, **134**, 144305.
- 19 M. Rafique, Y. Shuai, H.-P. Tan and H. Muhammad, *Appl. Surf. Sci.*, 2017, **399**, 20–31.
- 20 M. Rafique, Y. Shuai, H.-P. Tan and H. Muhammad, *Appl. Surf. Sci.*, 2017, **408**, 21–33.
- 21 J. Meng, D. Li, Y. Niu, H. Zhao, C. Liang and Z. He, *Phys. Lett. A*, 2016, **380**, 2300–2306.
- 22 D. Li, Y. Niu, H. Zhao, C. Liang and Z. He, *Phys. Lett. A*, 2014, **378**, 1651–1656.
- 23 E. V. Castro, K. Novoselov, S. Morozov, N. Peres, J. L. Dos Santos, J. Nilsson, F. Guinea, A. Geim and A. C. Neto, *Phys. Rev. Lett.*, 2007, **99**, 216802.
- 24 H. Min, B. Sahu, S. K. Banerjee and A. MacDonald, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **75**, 155115.
- 25 Y. Zhang, T.-T. Tang, C. Girit, Z. Hao, M. C. Martin, A. Zettl, M. F. Crommie, Y. R. Shen and F. Wang, *Nature*, 2009, **459**, 820–823.
- 26 H. Miyazaki, K. Tsukagoshi, A. Kanda, M. Otani and S. Okada, *Nano Lett.*, 2010, **10**, 3888–3892.
- 27 J. Park, S. B. Jo, Y. J. Yu, Y. Kim, J. W. Yang, W. H. Lee, H. H. Kim, B. H. Hong, P. Kim and K. Cho, *Adv. Mater.*, 2012, **24**, 407–411.
- 28 D. L. Duong, S. M. Lee, S. H. Chae, Q. H. Ta, S. Y. Lee, G. H. Han, J. J. Bae and Y. H. Lee, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 205413.
- 29 N. Kim, K. S. Kim, N. Jung, L. Brus and P. Kim, *Nano Lett.*, 2011, **11**, 860–865.
- 30 M. Dresselhaus and G. Dresselhaus, *Adv. Phys.*, 1981, **30**, 139–326.
- 31 K. Ohhashi and I. Tsujikawa, *J. Phys. Soc. Jpn.*, 1974, **36**, 422–430.
- 32 X. Zhang, D. Li, J. Meng, R. Yan, Y. Niu, H. Zhao, C. Liang and Z. He, *Comput. Mater. Sci.*, 2016, **124**, 316–322.
- 33 A. Marinopoulos, L. Reining, A. Rubio and V. Olevano, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2004, **69**, 245419.
- 34 O. Sedelnikova, L. Bulusheva and A. Okotrub, *J. Chem. Phys.*, 2011, **134**, 244707.
- 35 T. Eberlein, U. Bangert, R. Nair, R. Jones, M. Gass, A. Bleloch, K. Novoselov, A. Geim and P. Briddon, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **77**, 233406.
- 36 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758.
- 37 G. Kresse and J. Furthmüller, *Comput. Mater. Sci.*, 1996, **6**, 15–50.
- 38 P. E. Blöchl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953.
- 39 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 40 S. Grimme, *J. Comput. Chem.*, 2006, **27**, 1787–1799.
- 41 S. Grimme, C. Mück-Lichtenfeld and J. Antony, *J. Phys. Chem. C*, 2007, **111**, 11199–11207.
- 42 J. Antony and S. Grimme, *Phys. Chem. Chem. Phys.*, 2008, **10**, 2722–2729.
- 43 N. Kharche and S. K. Nayak, *Nano Lett.*, 2011, **11**, 5274–5278.
- 44 G. Makov and M. Payne, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1995, **51**, 4014.



- 45 T. Rasheed, S. A. Siddiqui and N. Bouarissa, *J. Fluorine Chem.*, 2013, **146**, 59–65.
- 46 M. Rafique, Y. Shuai, M. Xu, G. Zhang and Y. Guo, *Phys. E*, 2017, **93**, 26–38.
- 47 K. S. Kim, A. L. Walter, L. Moreschini, T. Seyller, K. Horn, E. Rotenberg and A. Bostwick, *Nat. Mater.*, 2013, **12**, 887.
- 48 J. Lin, W. Fang, W. Zhou, A. R. Lupini, J. C. Idrobo, J. Kong, S. J. Pennycook and S. T. Pantelides, *Nano Lett.*, 2013, **13**, 3262–3268.
- 49 J. Han, D. Kang and J. Dai, *RSC Adv.*, 2018, **8**, 19732–19738.
- 50 S. D. Chakarova-Käck, E. Schröder, B. I. Lundqvist and D. C. Langreth, *Phys. Rev. Lett.*, 2006, **96**, 146107.
- 51 W. Tang, E. Sanville and G. Henkelman, *J. Phys.: Condens. Matter*, 2009, **21**, 084204.
- 52 G. Henkelman, A. Arnaldsson and H. Jónsson, *Comput. Mater. Sci.*, 2006, **36**, 354–360.
- 53 M. Wu, C. Cao and J. Jiang, *New J. Phys.*, 2010, **12**, 063020.
- 54 D. Li, C. Wang, Y. Niu, H. Zhao and C. Liang, *Chem. Phys. Lett.*, 2014, **601**, 16–20.
- 55 Z. Chen, I. Santoso, R. Wang, L. F. Xie, H. Y. Mao, H. Huang, Y. Z. Wang, X. Y. Gao, Z. K. Chen and D. Ma, *Appl. Phys. Lett.*, 2010, **96**, 213104.
- 56 S.-O. Guillaume, B. Zheng, J.-C. Charlier and L. Henrard, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 035444.
- 57 S. Latil and L. Henrard, *Phys. Rev. Lett.*, 2006, **97**, 036803.
- 58 S. Latil, V. Meunier and L. Henrard, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **76**, 201402.
- 59 M. Gajdoš, K. Hummer, G. Kresse, J. Furthmüller and F. Bechstedt, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2006, **73**, 045112.
- 60 S. Chopra, *ChemPhysChem*, 2015, **16**, 1948–1953.
- 61 C. Shen, C. Tseng, C. Lin, L. Li and H. L. Liu, *Thin Solid Films*, 2014, **571**, 675–679.
- 62 K. F. Mak, L. Ju, F. Wang and T. F. Heinz, *Solid State Commun.*, 2012, **152**, 1341–1349.
- 63 C. Yelgel, Doctoral degree thesis, University of Exeter, 2013.
- 64 F. Nelson, V. Kamineni, T. Zhang, E. Comfort, J. Lee and A. Diebold, *Appl. Phys. Lett.*, 2010, **97**, 253110.
- 65 L. Yang, J. Deslippe, C.-H. Park, M. L. Cohen and S. G. Louie, *Phys. Rev. Lett.*, 2009, **103**, 186802.

