Combining transition metals and transient directing groups for C–H functionalizations

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In the domain of synthetic chemistry, C–H bond activation has always remained in the spotlight for researchers over the last few decades. Although different strategies have been employed to chemically trigger unactivated C–H bonds, transition metal catalyzed directing group (DG) aided C–H bond activation is the most explored pathway of all because of its ability to perform diverse site selective functional metanorphosis. Despite its popularity, tedious synthetic methodology requiring additional steps for the installation and removal of DGs from the target substrate diminishes its efficacy. However, replacement of directing groups by transient directing groups (tDGs) reduces the hurdle to a greater extent without compromising the product yield and selectivity. In this report we have depicted the intense journey of transient directing groups with three (Rh, Ru, and Pd) prevalent second row transition metals.

1 Introduction

Carbon–hydrogen bonds are nature’s most ubiquitous functional unit. Regardless of their easy accessibility, C–H bonds are quite inert due to a very low polarity difference between the two constituent atoms and a high C–H bond dissociation energy. Therefore direct conversion of C–H bonds into C–C or C–heteroatom bonds poses a great challenge to the synthetic community. Therefore, transition metal catalyzed C–H activation has become one of the most exciting topics in the last couple of decades in the domain of organometallic chemistry.

Even with its indisputable evolution, the major complication lies in achieving position-selective C–H functionalization. Irrespective of different chemical environments, all C–H bond energy values more or less coincide in the same range, hence it becomes difficult to perform a regioselective C–H bond transformation. Although a series of different strategies have been employed to address this concern, the use of directing groups (DGs) is still an irreplaceable solution. Directing groups are none other than a variety of σ-coordinating groups that can easily bind to the metal center forming a stable metallacycle which then selectively reaches the proximal C–H bond through its tuned orientation. Employing appropriate directing groups, versatile functionalizations have been achieved at ortho-,meta- and recently at para- positions of different aromatic rings. This strategy is also useful for selective functionalizations...
at various positions (α, β, γ and δ) of different aliphatic systems.16–22 Although enormous progress has been observed in DG assisted C–H bond activation, the major drawback of this approach is the installation of the DG in the substrate and also the removal at the end, and these often demand tedious synthetic procedures. To resolve this problem, two alternative solutions have been applied: (i) use of a transient mediator23–25 in a stoichiometric ratio and (ii) implementation of a traceless directing group.24–27 As well as these two approaches, an alternative way is to develop a functionally capable directing group which can reversibly connect with the substrate as well as to a metal. The comparison between the concepts of covalently linked directing groups and transient directing groups is shown in Fig. 1.

The concept of transient directing groups relies on the reversible binding of an organocatalyst with a substrate containing a particular functional group. The newly generated entity effectively coordinates with the metal center in a monodentate or bidentate fashion to generate the metallacycle which upon reaction with the coupling partner, furnishes the site selective functionalization (Fig. 2). Previously, detailed reviews on transient DG mediated C–H activation have been reported by Besset28 and Ackermann27 independently from two different perspectives. This work reviews the momentous evolution of transition metal catalyzed C–H activation mediated by transient modifiers. In this mini review we have categorized the entire theme into three major parts which are (a) rhodium catalyzed, (b) ruthenium catalyzed, and (c) palladium catalyzed transient DG assisted C–H activation, with detailed mechanisms and substrate scopes as well as their limitations.

2 Transient DG assisted C–H activation with second row transition metals

2.1 Transient DG assisted C–H activation with rhodium

Rhodium is the most exploited transition metal in terms of transient directing group assisted C–H activation. A series of functionalizations have been reported so far which use rhodium based catalysts. The journey started back in 1979 when Suggs reported a detailed study on the hydroacylation of aromatic aldehydes in three steps where Wilkinson’s catalyst [(PPh3)3RhCl] (5 mol%) was used to activate the aldehyde C–H bond.28 Of these three steps, one key step was the formation of aldimines from the corresponding aldehydes and amines via amidine formation. Starting from 2-amino-3-picoline A three different aldimines were prepared, which after reaction with different olefins resulted in the desired hydroacylated products mediated by (iminocarbonyl)rhodium(III) hydrides. However, the protocol was restricted to very limited variants of aromatic aldehydes with no α-hydrogens. Later, in 1997, Jun and coworkers developed a one-step hydroacylation of aldehydes 2 with different terminal olefins 1 through the in situ generation of 3-methyl-2-aminopyridyl aldimines (Fig. 3i).29 The method was not only step-reducing, but a variety of aliphatic aldehydes were also incorporated along with aromatic aldehydes.

The outline of the reaction mechanism was proposed as follows: (1) generation of aldimine a from the corresponding aldehyde 2 and 2-amino-3-picoline A which acts as a co-catalyst, (2) formation of (iminocarbonyl)rhodium(III) hydride b by C–H activation, (3) olefin coordination at the Rh center forming intermediate c, and (4) β-migratory insertion to generate an (iminocarbonyl)rhodium(III) alkyl complex d followed by (5) reductive elimination to form the linear ketone 3 with regeneration of the catalyst again (Fig. 3ii). The reaction was further modified through the use of a catalytic amount of benzoic acid along with aniline as an additive where they both triggered the aldimine formation in the initial step (Fig. 3iii).30 Later, in 2002 the same analogy was applied by Jun et al. to obtain the ortho-alkylated aromatic ketones starting from a masked ketone i.e. ketimine 4 with a variety of unactivated olefins.31 The protocol covered

![Fig. 1 Directing groups vs. transient directing groups.](image1)

![Fig. 2 Transient directing group mediated C–H activation.](image2)
Jun's catalytic system was further revised with an additional phosphine tether which overall offered bidentate coordination to the metal center and subsequently promoted the C–H activation step to provide intermolecular hydroacylated products (Table 2).33

In the acylated products, different functional groups like ester 3a, internal alkene 3b, alcohol 3c, and acid 3d were well-tolerated. After the successful intermolecular hydroacylation, the authors intended to extend the scope of this work further to achieve intramolecular hydroacylation, for which they chose o-vinylbenzaldehydes 7a–d in the presence of 5 mol% of [Rh(COD)]BF₄ catalyst and 5 mol% of ligand B (Table 3). Different functional groups including carboxylic ester 8a, free phenol 8b, nitro group 8c, and chloride 8d were well-suited to the reaction conditions, earning excellent yields.

Influenced by all of these developments, Dong and co-workers developed an α-alkylation of 1,2-diketones with different types of aromatic as well as aliphatic olefins in the presence of stoichiometric amounts of 2-aminopyridine C (Table 4).

The reaction continued through an enamine mediated C–H activation. A range of electron rich as well as electron deficient olefin coupling sources were equally fruitful. Regardless of

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**Table 1** Ortho-alkylation of masked ketones with activated olefins

<table>
<thead>
<tr>
<th><strong>5</strong></th>
<th>R= OMe</th>
<th>[Rh(PPPh₃)₃Cl] (5 mol%)</th>
<th>Toluene, 150 °C, 2 h</th>
<th><strong>6</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>5a</strong></td>
<td>R= OEt</td>
<td><strong>6a</strong></td>
<td>93%</td>
<td></td>
</tr>
<tr>
<td><strong>5b</strong></td>
<td>R= NMe₂</td>
<td><strong>6b</strong></td>
<td>75%</td>
<td></td>
</tr>
<tr>
<td><strong>5c</strong></td>
<td>R= SO₃Ph</td>
<td><strong>6c</strong></td>
<td>43%</td>
<td></td>
</tr>
<tr>
<td><strong>5d</strong></td>
<td>R= CN</td>
<td><strong>6d</strong></td>
<td>32%</td>
<td></td>
</tr>
</tbody>
</table>

**Table 2** Intermolecular hydroacylation with an *in situ* generated bidentate tDG

<table>
<thead>
<tr>
<th><strong>1a</strong></th>
<th>RCHO</th>
<th><strong>3a-e</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>1a</strong></td>
<td>R= Ph</td>
<td><strong>3a</strong> 73%</td>
</tr>
<tr>
<td><strong>1b</strong></td>
<td>R= Me</td>
<td><strong>3b</strong> 57%</td>
</tr>
<tr>
<td><strong>1c</strong></td>
<td>R= OMe</td>
<td><strong>3c</strong> 75%</td>
</tr>
<tr>
<td><strong>1d</strong></td>
<td>R= NMe₂</td>
<td><strong>3d</strong> 51%</td>
</tr>
</tbody>
</table>

**Table 3** Intramolecular hydroacylation with an *in situ* generated bidentate transient DG

<table>
<thead>
<tr>
<th><strong>7a-d</strong></th>
<th>B</th>
<th><strong>8a-d</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>7a</strong></td>
<td>R= m-COOMe, 96%</td>
<td><strong>8a</strong></td>
</tr>
<tr>
<td><strong>7b</strong></td>
<td>R= m-CN, 95%</td>
<td><strong>8b</strong></td>
</tr>
<tr>
<td><strong>7c</strong></td>
<td>R= p-N0₂, 93%</td>
<td><strong>8c</strong></td>
</tr>
<tr>
<td><strong>7d</strong></td>
<td>R= o-Cl, 91%</td>
<td><strong>8d</strong></td>
</tr>
</tbody>
</table>

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Fig. 3 Intermolecular hydroacylation (i) by Jun and co-workers in 1997 (ii) further modification of Jun's work in 2000 (iii) proposed mechanistic outline for intermolecular hydroacylation.
several difficulties with the ketimine formation, in 2014, Mo and Dong reported α-alkylation of cyclic ketone derivatives with a modified rhodium(II) complex [Rh(coe)Cl]$_2$ using simple ethylene gas where 7-azaindoline acted as an effective transient directing group (Fig. 4). The transformation was highly regioselective for the less crowded 5-position of the cyclic ketone, overriding the electronic bias of the substituents present on the ketones. In particular, this work excluded the possibility of forming other side products. The use of the Rh(I) dimer complex and ligand D was found to be essential in this case. The enamine formation in the initial step was promoted by 10 mol% TsOH·H$_2$O which again supported the proof-of-concept. To accelerate the oxidative addition of enamine C–H bonds, a sterically hindered electron rich NHC ligand (IMes) was employed (5 mol%). A plausible mechanistic outline for this reaction is shown in Fig. 5. At the very beginning, an in situ enamine f was formed, followed by oxidative addition of the Rh complex, generating g. In the next step, olefin 13 coordinated with the metal to form complex h, which after reductive elimination generated i. Hydrolysis of this intermediate accomplished the desired ketones 14.

Apart from imine based transient directing groups, Bedford reported Rh(I) catalyzed ortho-arylation of phenols via a phosphinite based transient directing group (Fig. 6). Previous methods for the preparation of ortho-arylated phenols were limited to Suzuki or Stille coupling reactions which demand stoichiometric quantities of coupling partners like boronic acid derivatives or organotin compounds. On the other hand, the incorporation of aryloxide groups into phosphite or phosphinite systems assured the formation of thermodynamically and kinetically stable low-strain five membered metallacycles for the ortho-C–H activation of phenols. The authors envisaged two intertwined catalytic pathways involving directed ortho-C–H arylation of phenol assimilated with a PR$_2$(OAr) organocatalyst, followed by transesterification of the phosphinite ligand, to release substituted phenol (Fig. 6). Bedford’s elegant work was inspired by Lewis’s work in 1985 where they elegantly demonstrated ruthenium catalyzed ortho-ethylation and deuteration of phenols involving in situ reversible transesterification of triarylphosphite ligands with phenol. In 2008, Breit et al. used the same phosphinites as temporary directing groups to execute hydroformylation of homoallylic alcohols 17 to furnish γ-lactones and γ-lactols with highly branched regioisomers.

Table 4 α-Alkylation of diketones using 2-aminopyridine as a transient modifier

![Table 4](image)

Fig. 4 α-Alkylation of cyclic ketones with the 7-azaindoline tDG precursor.

![Fig. 4](image)

Fig. 5 Proposed mechanism for α-alkylation of cyclic ketones using 7-azaindoline as a tDG.

![Fig. 5](image)

Fig. 6 Rh-catalyzed ortho-arylation of phenols using a phosphite tDG.
Early methodologies were selective for the linear hydroformylation of alkenes whereas the use of phosphine or phosphite ligands revised the regiochemistry in favour of branched isomers (Fig. 7). Indeed, both E and Z alkenes provided the same selective branched lactols 18 under the optimized reaction conditions. Interestingly, Tan and co-workers also achieved selective branched hydroformylation of homoallylic alcohols through an amino phosphate based ligand.\(^{38}\) In 2012, Douglas excellently exploited a similar approach for the design of seven membered cyclic benzoketones via intramolecular hydroacylation.\(^{39}\) The report was influenced by Dong’s work in 2009, where the presence of a heteroatom in the ketone ring was essential (Fig. 8).\(^{40}\) The use of a catalytic amount of ligand E accelerated the rate of hydroacylation by reducing decarbonylation of the ketones. The same Rh(i) catalyst was then implemented by Dong and co-workers to furnish \(\alpha\)-alkylated ketones, where 2-aminopyridine acted as the key component for the intermediate imine formation. A catalytic quantity of 5-methyl-2-benzoic acid played a crucial role in improving enamine formation.\(^{41}\) In addition, \(\alpha\)-alkenylation of cyclic ketones 23 was achieved with internal alkynes 24 using the same 7-azaindoline D as for the transient modifier. The methodology enabled the production of both the \(\alpha,\beta\)- as well as the \(\beta,\gamma\)-enones in moderate to excellent yields (Fig. 9).\(^{42}\) Thereafter, in 2015, Rovis and Piou successfully implemented the \(\text{syn}\)-carboamination of doubly substituted alkenes in the presence of a Rh(III) catalyst with enoxypthalimides as the amine source (Fig. 10).\(^{43}\) Very recently, Jiao and co-workers demonstrated an imine based transient DG assisted amidation of benzaldehyde derivatives (Fig. 11).\(^{44}\) They used 3-phenyl-1,4,2-dioxazol-5-one as the potential aminating source and 4-trifluoromethyl aniline F proved its efficacy as a catalytic modifier.

2.2 Transient DG assisted C–H activation with ruthenium
A notable advancement in this domain was achieved by Lewis and co-workers. They reported Ru-catalysed \(\text{ortho}\)-deuteration of phenol using a phosphite transient DG under a D\(_2\) atmosphere. The authors discovered that the presence of a catalytic amount of KOPh can speed up the transesterification process between phenol and triphenylphosphites via an active metal-lacycle G (Fig. 12).\(^{45}\) Ruthenium(II) catalyzed \(\text{ortho}\)-arylation of benzaldehydes between imines (masked benzaldehydes) 34 and aryl bromides 35 was also reported by Inoue et al. (Fig. 13).\(^{46}\) Recently, Rasheed and Zhang synthesized \(\text{ortho}\)-(sulfonylamino)benzaldehydes in the presence of 3 mol% [Ru(\(p\)-cymene)Cl\(_2\)] as the catalyst using organic azides as the source of \(N\)-functionalities in conjunction with a catalytic quantity of aniline H (Fig. 14).\(^{47}\) 5 mol% AgSbF\(_6\) was found to be the best chloride ion scavenger in this case. With the optimized reaction conditions in
hand, the authors explored the generality of this reaction. Both electron donating groups and electron withdrawing groups were well-tolerated. The catalytic cycle for this reaction included the generation of a cationic ruthenium species \( \text{Ru}^2 \) ion exchange with AgSbF\(_6\) (Fig. 15). The active catalyst \([\text{Ru}(\text{p-cymene})](\text{SbF}_6)_{2}\) was thus generated. Eventually, the ruthenium catalyst was bound to the \textit{in situ} formed imine \( j \) which then underwent C–H activation to form complex \( k \). Furthermore, the N center of tosyl azide was ligated to the metal center in an oxidatively additive manner to produce \( l \). The exclusion of nitrogen and the concomitant rearrangement led to the complex \( m \) which upon treatment with acid afforded the \textit{ortho}-aminated product \( 39 \).

2.3 Transient DG assisted C–H activation with palladium

Although Rh-catalyzed transient directing group assisted C–H functionalizations have been well-explored, no such reports with palladium metal catalysts were found in the literature until the work of Yu and co-workers. They disclosed the arylation of C(sp\(^3\))–H bonds of \textit{ortho}-alkyl benzaldehydes in 2016 (Fig. 16).\(^{48}\) However, the traditionally-used iminopyridine or iminooxazoline directing groups failed to show adequate directing abilities in this case due to their strong coordination with the Pd metal center, thus disfavouring the targeted C–H bond cleavage. However, a catalytic amount of amino acids solved the problem in the presence of a mixture of protic solvents. A proportionate (3 : 1) mixture of 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) and acetic acid (AcOH) improved the product formation in the case of the aryl system while a combination of AcOH and water in a ratio of 9 : 1 enhanced the scope for \( \beta \)-arylation of aliphatic ketones.

This functionalization was further extended for the synthesis of fluorenones by the Sorensen group using low-cost anthranilic acid as a transient DG.\(^ {49}\) In addition to benzaldehydes, \textit{ortho}-arylation of aromatic ketones \( \textit{via} \) a glycine transient DG was also reported by Jin \textit{et al.} in 2017.\(^ {50}\) More recently, Yu \textit{et al.} revealed a series of Pd-catalyzed functionalizations at the \textit{ortho}-position of the benzaldehydes using a bidentate amino acid based transient DG, followed by an Ir(III) catalyzed C–H amidation with monodentate aniline \( K \) as an effective transient directing group (Fig. 17).\(^ {51}\) A library of functionalized compounds was synthesized with a wide substrate scope and remarkable applications. In a similar way, Hu,\(^ {52}\) Yu,\(^ {53}\) Bull,\(^ {54}\) and Li and Ge\(^ {55}\) also achieved \( \beta \)-arylation of unactivated C(sp\(^3\))–H bonds where...
acetohydrazide and β-alanine turned out to be some of the most efficient active transient modifiers (Fig. 18).

It is worth mentioning that these studies involved arylation of both methyl as well as methylene C(sp3)–H bonds of the aliphatic carbonyl system. Even though transient DG assisted C–H activation is widespread using Rh and Pd metal catalysts, in 2016, Shi et al. first described a transient auxiliary mediated C–H amidation of aryl aldehydes using tosyl azide as the amidating source. Initially they started with simple aldehyde 45 as their model substrate which failed to give the desired aminated product. Then, they replaced the aldehyde with imine 46 which successfully led to the product in 91% yield. Realizing the importance of the iminyl group, the authors explored the use of aldehydes directly as the substrate, with 3-trifluoromethylamine in a catalytic amount to carry out the reaction via intermediate imine formation which furnished ortho-amination of benzaldehydes as expected, in excellent yields (Fig. 18ii). Until 2016, all in situ generated imine based transient templates were utilized to functionalize different aromatic and aliphatic aldehydes or ketone substrates where a catalytic quantity of amine derivatives was always added to the reaction mixture to form the intermediate imine DG. In 2016, Dong and co-workers first thought to switch ligand i.e. they chose free amines as their substrates and added quinoline-8-carbaldehyde in trace amounts to obtain the imine DG. The exo DG enabled efficacious arylation at the γ-position of the free amines (Fig. 19). Later, Ge et al. also discovered γ-C–H arylation of free amines with different aryl iodides as arylation partners. The use of glyoxylic acid as the temporary directing group furnished diverse arylated free amines. A similar strategy was implemented thereafer by the Yu group in 2016 with Boc protected amines to accomplish γ-C(sp3)–H arylation of free primary amines using 2-hydroxy-nicotinaldehyde as the transient DG. This strategy involved commercially obtainable tert-leucine as the catalytic chiral auxiliary which enabled the formation of chiral biaryls with outstanding enantioselectivity (95–99% ee) (Fig. 20).

Notably, in 2018, Sorensen and co-workers developed reaction conditions for Pd-catalyzed ortho-methylation and fluorination of benzaldehydes utilizing potassium methyltrifluoroborate as the methyl source or 1-fluoro-2,4,6-trimethylpyridinium triflate as
the effective fluorinating agent in the presence of orthanilic acid M (45 mol%) or aniline-2,4-disulfonic acid N (50 mol%) respectively (Fig. 21).11

For the first time, orthanilic acid has been used as a potential transient DG for C–H functionalization. Interestingly, employment of one extra sulfonic acid group in N facilitated the C–F reductive elimination step by increasing the electrophilicity of the Pd(n) center.

This increased electrophilicity then attenuated the polarity difference of the Pd(iv)–F bond to execute a facile reductive elimination. Notably, the reaction has covered a broad substrate scope irrespective of electron rich and electron poor systems. For further mechanistic interest, the authors were able to trap the C–H activated metallacycle in crystal form.

3 Conclusions

In conclusion, transition metal catalyzed functionalization of C–H bonds heralds the commencement of a new horizon for the synthesis of potential complex molecular scaffolds. However, transient directing group mediated C–H activation is one of the recent advances in this field which can equally assure site-selective functional transformation of C–H bonds whilst overcoming the technical drawbacks of directed C–H activation. This strategy has already been recognized as a significant tool for different functional alterations of C–H bonds which include arylation, amination, halogenation, alkylation, and many more. Despite its hassle-free synthetic pathway, the protocol is solely restricted to different aromatic and aliphatic aldehydes, ketones, amines and phenol derivatives. This strategy is not only functional group specific but it is also restricted to proximal C–H bonds of the respective functional groups. Systematic design of transient DGs can further extend the scope of distal C–H activation. On top of that, the strategy mainly relies on 4d transition metals like rhodium, palladium, ruthenium and iridium. Therefore experiments with 3d transition metals are highly anticipated due to their abundance as well as their eco-friendly nature. Although the approach is yet not generalized for all possible functional transformations, further remodelling will definitely shape the field of C–H activation.

Conflicts of interest

There are no conflicts to declare.

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References