Synthesis of propellanes containing a bicyclo[2.2.2]octene unit via the Diels–Alder reaction and ring-closing metathesis as key steps†

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The synthesis of propellanes containing bicyclo[2.2.2]octene via olefin metathesis approach is less explored. Herein, we describe a simple and convenient method to synthesize propellane derivatives containing a bicyclo[2.2.2]octene unit which are structurally similar to 11β-HSD1 inhibitors by sequential usage of the Diels–Alder reaction, C-allylation and ring-closing metathesis (RCM) as the key steps. Additionally, we expanded this approach to an endo-tricyclo[4.2.2.02,5]decene derivative which is a useful monomer for polymer synthesis and we have also synthesized basketene and anthraocene-based propellanes using the same strategy.

Introduction

Propellanes are Y-shaped tricyclic molecules containing a bridgehead C–C bond. They can be classified as carbocyclic and heterocyclic frames and their physical or chemical properties vary with their ring size.1 Heterocyclic propellanes are generally found in nature whereas carbocyclic propellanes are rare and found in secondary metabolites.1–4 Synthesis of the propellane motif is difficult using conventional methods because of their quaternary nature5 and creation of vicinal quaternary carbons is not a trivial task. However, they can be assembled easily using a ring-closing metathesis (RCM) strategy.1–3 The synthesis of propellanes containing a bicyclo[2.2.2]octene system and their metathetic behaviour is worthy of systematic investigation.4

Bicyclo[2.2.2]octene is a rigid and strain free system which is present in diverse and challenging targets varying from biomolecules to natural products.5 Bicyclo[2.2.2]octene fused cyclic amides are potent bioactive molecules. 3,10-Dihydroxydielmentha-5,11-diene-4,9-dione 1m–7 is a diterpenoid isolated from Callitris macleayana heartwood and has a core of bicyclo[2.2.2]octene fused with a 6-membered carbocyclic ring (Fig. 1). Mitindomide 2 (Fig. 1) is a bicyclo[2.2.2]octene analogue which inhibits topoisomerase II and promotes DNA-interstrand cross-linking and thereby increases its anti-neoplastic activity.5a A water soluble derivative 3, which is structurally similar to mitindomide 2 but lacks the cyclobutane ring, exhibits antitumor activity.5e

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It seems that ROMP of an endo-tricyclo[4.2.2.0\(^2\)]deca-3,9-diene system is much faster than that of a norbornene system because of the high reactivity of the cyclobutene moiety present in the endo-tricyclo[4.2.2.0\(^2\)]deca-3,9-diene unit. Recently, Choi and co-workers have proposed a controlled ROMP of terminal alkyne monomer 23 to alkyne copolymer 24 by taking advantage of the strain in the cyclobutene ring of TD (Scheme 2).
Results and discussion

Our synthetic approach to propellane derivatives relies on understanding the metathetic behaviour of the *endo*-tricyclo[4.2.2.0]deca-3,9-diene derivative 29. Its synthesis starts with the preparation of the Diels–Alder (DA) adduct 27, which is easily assembled from commercially available cyclooctatetraene 25 and maleic anhydride 26 using a [4 + 2] cycloaddition reaction. Next, anhydride 27 was treated with aniline in the presence of triethylamine in toluene at 120 °C to give compound 28 in 98% yield. Later, the *endo* adduct 28 was treated with an excess amount of allyl bromide in the presence of 1 M NaHMDS at −78 °C to deliver the diallyl compound 29 in 45% yield (Scheme 3). Compound 29, which contains three distinct types of olefinic moieties, is an interesting substrate for studying metathetic behaviour.

To study the metathesis behaviour of compound 29, we screened various catalysts (Fig. 2) under different reaction conditions (Table 1). Among the various catalysts, only the G-I and G-II catalysts (Table 1, entries 1 and 2) gave a significant yield of the propellane derivative 30. The stereochemistry of compound 30 was further confirmed by single crystal X-ray diffraction studies (Fig. 3).

We were able to suppress the polymerization mode by carrying out the reaction under an ethylene atmosphere in the presence of the G-I and G-II catalysts. However, with the Hoveyda–Grubbs catalysts (Table 1, entries 3 and 4), the diallyl compound 29 was polymerized along with trace amounts of the propellane derivative 30. We didn’t observe any RRM product under these conditions. The crystal data of compound 30 indicate that due to their orientation, the vinyl groups are not favourably disposed to interact with the allyl groups to deliver the RRM product (Fig. 3).

To understand the metathetic behaviour of the different olefinic moieties present in the bicyclo[2.2.2]octene derivative 29, we are interested in preparing compound 31 with the double bond present in the cyclobutene ring absent. In this regard, compound 28 was subjected to selective hydrogenation with the aid of 10% Pd/C–H₂ to afford the partially saturated compound 31 in 96% yield. Later, it was subjected to allylation under 1 M NaHMDS conditions to generate the diallyl derivative 32 in 53% yield. Then, the diallyl compound 32 was subjected to a metathesis sequence with the G-II catalyst to obtain the propellane derivative 33 in 64% yield (Scheme 4).

We have also studied the metathetic behaviour of the DA adduct 28 using various catalysts under different reaction conditions (Table 2). Among these, the G-I and G-II (Table 2, entries 1 and 2) catalysts gave the cyclobutene ring-opened product 34 in excellent yields. However, we didn’t observe the formation of the completely ring-opened product 35 under these conditions.

To understand the role of the cyclobutane ring in the metathetic behaviour of compound 29, we synthesized the DA adduct 37 from maleic anhydride 26 with 1,3-cyclohexadiene 36 and subsequently 37 was reacted with aniline in the presence of Et₂N to deliver compound 38. The metathetic behaviour of the *endo* adduct 38 was studied with various catalysts under different reaction conditions. Unfortunately, we didn’t obtain the expected ROM product 39. Later, the N-phenyl derivative 38 was treated with allyl bromide in the presence 1 M NaHMDS at −78 °C to afford the diallyl compound 40 in 45% yield. Subsequently, the diallyl derivative 40 was subjected to a metathesis

![Scheme 2](image-url)  
**Scheme 2**  
Polymerization of *endo*-tricyclo[4.2.2.0]deca-3,9-diene (TD).

![Scheme 3](image-url)  
**Scheme 3**  
Synthesis of diallyl *endo*-tricyclo[4.2.2.0]deca-3,9-diene 29.
sequence using the G-II catalyst (5 mol%) under ethylene to produce the propellane derivative 41 in 92% yield (Scheme 5). These results suggest that the double bond present in the cyclobutene of the endo-tricyclo[4.2.2.0]deca-3,9-diene system is highly reactive and the double bond present in the bicyclo[2.2.2]octene system is inert under these conditions. Our attempt to synthesize the exo derivative of 37 and 38 was not successful.

Later, this strategy was extended to design the basketene and anthracene-based propellane derivatives 45 and 50 respectively. Recently, our group reported the synthesis of cage propellanes, however, the basketene-based propellane synthesis is a worthy target for high energy systems. In this context, the DA adduct 27 was subjected to a [2 + 2] photocycloaddition reaction using a 450 W Hg lamp equipped with a Pyrex glass filter for 20 h to produce the cage compound 42, which on treatment with aniline in the presence of trimethylamine in toluene gave the cage compound 43 in 68% yield. Additionally, allylation of the N-phenyl derivative 43 with allyl bromide under 1 M NaHMDS conditions afforded the diallyl compound 44 which underwent RCM with the G-II catalyst at

<table>
<thead>
<tr>
<th>Entry</th>
<th>Catalyst (mol%)</th>
<th>Solvent</th>
<th>Temp (°C)/time (h)</th>
<th>Yield (%)</th>
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<tbody>
<tr>
<td>1</td>
<td>G-I (5 mol%)</td>
<td>CH₂Cl₂</td>
<td>rt/6 h</td>
<td>65</td>
</tr>
<tr>
<td>2</td>
<td>G-II (5 mol%)</td>
<td>CH₂Cl₂</td>
<td>rt/6 h</td>
<td>58</td>
</tr>
<tr>
<td>3</td>
<td>GH-I (5 mol%)</td>
<td>CH₂Cl₂</td>
<td>rt/5 h</td>
<td>Trace</td>
</tr>
<tr>
<td>4</td>
<td>GH-II (5 mol%)</td>
<td>CH₂Cl₂</td>
<td>rt/3 h</td>
<td>Trace</td>
</tr>
<tr>
<td>5</td>
<td>G-I (10 mol%)</td>
<td>Toluene</td>
<td>Reflux/3 h</td>
<td>10</td>
</tr>
<tr>
<td>6</td>
<td>G-II (10 mol%)</td>
<td>CH₂Cl₂</td>
<td>rt/5 h</td>
<td>35</td>
</tr>
<tr>
<td>7</td>
<td>G-I (5 mol%) + Ti(OPrᵢ)₄ (catalytic)</td>
<td>CH₂Cl₂</td>
<td>rt/4 h</td>
<td>32</td>
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The cubane bond angle of 90° was confirmed on the basis of 1H NMR and 13C NMR spectral parameters and corresponding anhydride. Anthracene derived propellane was further supported by the HRMS data. Along similar lines, bicyclo[2.2.2]octene propellane derivatives were found to have larger bond angles compared to other derivatives due to distortion upon the [2 + 2] cyclo-addition reaction of 30. The gas-phase calculations were carried out using the Gaussian 09 program.

We have also carried out geometry optimization studies to illustrate the spatial demands and rigidity of these propellanes. Gas-phase calculations were carried out for 30, 33, 41, and 45 at the M062X/6-31G** level of theory. All calculations were carried out using the Gaussian 09 program. As shown by the optimized geometry study, the C1–C2–C7 and C1–C2–C7 of the bicyclo[2.2.2]octene propellane derivatives 30, 33, and 41 are similar. However, the propellane derivative 45 was found to have larger bond angles compared to other derivatives due to distortion upon the [2 + 2] cyclo-addition reaction of 27. The C2–C3–C10 of the basketene derivative 45 is 89.56° which is close to the cubane bond angle of 90° (Fig. 4). Because of the rigidity of the propellane containing bicyclo[2.2.2]octene derivatives and their structural similarity to the reported compounds 5–9, our methodology is useful for synthesizing biologically active derivatives such as these.

**Experimental section**

All commercially available reagents were used without further purification and the reactions involving air-sensitive catalysts or reagents were performed in degassed solvents. Moisture-sensitive materials were transferred using the syringe-septum technique and the reactions were maintained under a nitrogen atmosphere. Analytical thin-layer chromatography (TLC) was performed on (7.5 × 2.5 cm) glass plates coated with the Acme silica gel GF 254 (containing 13% calcium sulfate as a binder) using a suitable mixture of EtOAc and petroleum ether for development. Column chromatography was performed using Acme silica gel (100–200 mesh) with an appropriate mixture of EtOAc and petroleum ether. Coupling constants (J) are given in hertz (Hz) and chemical shifts are denoted in parts per million (ppm) downfield from the internal standard, tetramethylsilane (TMS). The abbreviations, s, d, t, q, m, dd, brs, td, tt and dt refer to singlet, doublet, triplet, quartet, multiplet, doublet of doublets, broad singlet, triplet of doublet, triplet of triplets and doublet of triplets, respectively. Infrared (IR) spectra were recorded on a Nicolet Impact-400 FT-IR spectrometer. Nuclear Magnetic Resonance (NMR) spectra were generally recorded on a Bruker (AvanceTM 400 or AvanceTM III 500) spectrometer operating at 400 or 500 MHz for 1H and 100.6 or 125.7 MHz for 13C nuclei. The high-resolution mass spectrometric (HRMS) measurements were carried out using a Bruker (Maxis Impact) or Micromass Q-ToF spectrometer.

The general procedure for the endo-anhydride synthesis

**Compound 27.** Off-white solid (190 mg starting from 150 mg of 26, 68%), Rf = 0.56 (10% EtOAc/hexane). 1H and 13C NMR spectra of compound 27 were matched with those of the reported literature.

**Compound 37.** Off-white solid (242 mg starting from 150 mg of 36, 74%), Rf = 0.48 (10% EtOAc/hexane). 1H and 13C NMR spectra of compound 37 were matched with those of the reported literature.

**Compound 42.** White solid (180 mg starting from 345 mg of 27, 52%), Rf = 0.56 (10% EtOAc/hexane). 1H and 13C NMR spectra of compound 42 were matched with those of the reported literature.

**Compound 47.** White solid (650 mg starting from 490 mg of 46, 86%), Rf = 0.62 (10% EtOAc/hexane). 1H and 13C NMR spectra of compound 47 were matched with those of the reported literature.

The general procedure for the endo-N-phenyl adduct synthesis

Aniline (1 mmol, 1 equiv.) and trimethylamine (1 mL) were added to anhydride (1 mmol, 1 equiv.) in toluene in a sealed tube. The resulting mixture was heated at 120 °C for 12 h. After the depletion of anhydride, the extra solvent was distilled off.
The residue was washed with diethyl ether (2 × 30 mL) and dried under high vacuum to yield a pure N-phenyl adduct.

**Compound 28.** Yield 98% (310 mg, starting from 240 mg of 27), white solid. $^1$H and $^{13}$C NMR spectra of compound 28 were matched with those of the reported literature.\textsuperscript{10a}

**Compound 38.** White solid (465 mg starting from 480 mg of 37, 68%), $R_f = 0.52$ (10% EtOAc/hexane). $^1$H and $^{13}$C NMR spectra of compound 38 were matched with those of the reported literature.

**Compound 43.** White solid (65 mg starting from 100 mg of 42, 48%), $R_f = 0.52$ (10% EtOAc/hexane), mp: 165–167°C. IR (neat): $\nu_{\text{max}}$ C=O stretch at 1705 cm$^{-1}$. $^1$H NMR (CDCl$_3$, 400 MHz): $\delta = 7.49$–7.43 (m, 2H), 7.41–7.35 (m, 1H), 7.25–7.21 (m, 2H), 3.38–3.32 (m, 4H), 3.29–3.25 (m, 2H), 3.13–3.09 (m, 2H), 2.99 (t, 2H, $J = 1.9$ Hz). $^{13}$C NMR (CDCl$_3$, 100 MHz): $\delta = 178.50$, 132.19, 129.13, 128.68, 126.65, 43.90, 38.95, 36.75, 32.79.

HRMS (ESI, Q-ToF) $m/z$: found [M + H]$^+$ = 278.1189, calculated = 278.1181 for C$_{18}$H$_{15}$NO$_2$.

**Compound 48.** Pale yellow solid (295 mg starting from 300 mg of 47, 78%), $R_f = 0.55$ (10% EtOAc/hexane). $^1$H and $^{13}$C NMR spectra of compound 48 were matched with those of the reported literature.\textsuperscript{19b}

**Synthesis of compound 31**

To compound 28 (320 mg, 1.15 mmol) in EtOAc, 10% Pd/C was added at room temperature, the reaction mixture was stirred at room temperature until completion. The reaction mixture was passed through a celite column and washed with EtOAc several times to yield pure compound 31. White solid (310 mg, 96%), $R_f = 0.68$ (5% EtOAc/hexane). IR (neat): $\nu_{\text{max}}$ C=O stretch at 1704 cm$^{-1}$. $^1$H NMR (CDCl$_3$, 400 MHz): $\delta = 7.44$ (t, 2H, $J = 7.5$ Hz), 7.36 (t, 1H, $J = 7.5$ Hz), 7.18 (d, 2H, $J = 8.6$ Hz), 6.43 (t, 2H, $J = 3.5$ Hz), 3.23 (brs, 2H), 2.83 (br, s, 2H), 2.62–2.57 (m, 2H), 2.10–2.02 (m, 2H). $^{13}$C NMR (CDCl$_3$, 100 MHz): $\delta = 178.1$, 132.1, 132.1, 129.13, 128.7, 126.6, 45.0, 37.0, 36.5, 22.3.

**Table 2** The metathetic behaviour of compound 28 with different catalysts

<table>
<thead>
<tr>
<th>Entry</th>
<th>Catalyst</th>
<th>Time</th>
<th>Product 34, yield (%)</th>
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<tr>
<td>1</td>
<td>G-I</td>
<td>6 h</td>
<td>83</td>
</tr>
<tr>
<td>2</td>
<td>G-II</td>
<td>6 h</td>
<td>85</td>
</tr>
<tr>
<td>3</td>
<td>GH-I</td>
<td>5 h</td>
<td>71</td>
</tr>
<tr>
<td>4</td>
<td>GH-II</td>
<td>12 h</td>
<td>68</td>
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<td>5</td>
<td>M71 SIMes</td>
<td>12 h</td>
<td>65</td>
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<td>6</td>
<td>M71 1\textsuperscript{st} gen</td>
<td>12 h</td>
<td>69</td>
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<tr>
<td>7</td>
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<td>12 h</td>
<td>58</td>
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<tr>
<td>8</td>
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<td>12 h</td>
<td>60</td>
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<tr>
<td>9</td>
<td>M73 SIMes</td>
<td>12 h</td>
<td>63</td>
</tr>
<tr>
<td>10</td>
<td>Ru-I</td>
<td>5 h</td>
<td>75</td>
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</table>

**Scheme 5** Synthesis of the bicyclo[2.2.2]octene propellane 41.
HRMS (ESI, Q-ToF) \( m/z \): found \([M + Na]^+ = 302.1151\), calculated = 302.1146 for \( C_{18}H_{17}NO_2 \).

The general procedure for allylation using NaHMDS

NaHMDS (4.33 mmol, 8 eq.) and allyl bromide (4.3 mmol, 8 eq.) were added to the \( \text{N} \)-phenyl adduct compound (0.55 mmol, 1 eq.) in dry THF at \(-78^\circ \text{C}\) under a nitrogen atmosphere. The resulting mixture was brought to room temperature in 12 h and stirred at rt for 1 h. After the completion of the reaction, THF was removed under reduced pressure and the residue was partitioned between ethyl acetate and water. The organic layer was concentrated and purified by column chromatography to yield the allylated compound.

**Compound 29.** Off-white solid (41 mg starting from 70 mg of 28, 45%), \( R_t = 0.75 \) (5% EtOAc/hexane), mp: 156.5–157.2 \( ^\circ \text{C} \). IR (neat): \( \nu_{\text{max}} \, \text{C=O} \) stretch at 1764, 1701 \( \text{cm}^{-1} \). \( ^1\text{H} \) NMR (CDCl\(_3\), 400 MHz): \( \delta = 7.47-7.40 \) (m, 3H), 7.40–7.32 (m, 1H), 7.22–7.12 (m, 2H), 6.25–6.13 (m, 2H), 6.03 (dd, \( J = 2.5 \text{ Hz}, 4.5 \text{ Hz}, 2 \text{ Hz} \)), 5.82 (s, 2H), 5.27–5.17 (m, 4H), 3.13–3.03 (m, 4H), 2.71–2.50 (m, 5H). \( ^{13}\text{C} \) NMR (CDCl\(_3\), 100 MHz): \( \delta = 180.2, 137.6, 133.9, 132.1, 130.2, 129.2, 128.7, 126.7, 118.8, 51.8, 42.2, 38.9, 35.5. \) HRMS (ESI, Q-ToF) \( m/z \): found [M + Na\(^+\)] = 380.1621, calculated = 380.1619 for \( C_{24}H_{23}NO_2 \).

**Compound 32.** White solid (101 mg starting from 150 mg of 31, 53%), \( R_t = 0.56 \) (5% EtOAc/hexane), mp: 129.3–133.5 \( ^\circ \text{C} \). IR (neat); \( \nu_{\text{max}} \, \text{C=O} \) stretch at 1742, 1705 \( \text{cm}^{-1} \). \( ^1\text{H} \) NMR (CDCl\(_3\), 500 MHz): \( \delta = 7.48-7.40 \) (m, 2H), 7.38–7.33 (m, 1H), 7.20–7.16 (m, 2H), 6.47 (dd, \( J = 3.32, 1.20 \)), 6.20–6.07 (m, 2H), 5.19 (q, \( J = 1.44 \text{ Hz} \)), 5.13 (brs, 1H), 3.01–2.97 (m, 2H), 2.82–2.75 (m, 2H), 2.57 (td, \( J = 1.45 \text{ Hz} \)), 2.53 (td, 1H, \( J = 5.81 \text{ Hz}, 1.33 \text{ Hz} \)), 2.42 (dd, 2H, \( J = 14.6 \text{ Hz}, 8.3 \text{ Hz} \)), 2.06–2.02 (m, 2H), 1.48–1.43 (m, 2H) ppm. \( ^{13}\text{C} \) NMR (CDCl\(_3\), 125 MHz): \( \delta = 180.3, 134.0, 133.7, 129.1, 128.6, 126.7, 50.9, 41.7, 35.6, 31.0, 22.0 \) ppm. HRMS (ESI, Q-ToF) \( m/z \): found [M + H\(^+\)] = 360.1958, calculated = 360.1953 for \( C_{24}H_{26}NO_2 \).

**Compound 40.** White solid (60 mg starting from 100 mg of 38, 45%), \( R_t = 0.8 \) (5% EtOAc/hexane), mp: 116.1–118.4 \( ^\circ \text{C} \). IR (neat): \( \nu_{\text{max}} \, \text{C=O} \) stretch at 1705 \( \text{cm}^{-1} \). \( ^1\text{H} \) NMR (CDCl\(_3\), 400 MHz): \( \delta = 7.47-7.40 \) (m, 2H), 7.40–7.32 (m, 1H), 7.22–7.09 (m, 2H), 6.36–6.27 (m, 2H), 6.23–6.07 (m, 2H), 5.27–5.10 (m, 4H), 2.97 (brs, 2H), 2.72–2.46 (m, 4H), 1.88–1.76 (m, 2H), 1.55–1.27 (m, 2H). \( ^{13}\text{C} \) NMR (CDCl\(_3\), 100 MHz): \( \delta = 180.5, 134.2, 132.2, 129.2, 128.7, 126.7, 118.7, 51.5, 37.5, 35.9, 18.4. \) HRMS (ESI, Q-ToF) \( m/z \): found [M + Na\(^+\)] = 356.1624, calculated = 356.1626 for \( C_{22}H_{23}NO_2 \).

**Compound 44.** Colourless liquid (25 mg starting from 40 mg of 43, 50%), \( R_t = 0.74 \) (5% EtOAc/hexane). IR (neat): \( \nu_{\text{max}} \, \text{C=O} \) stretch at 1694, 1657 \( \text{cm}^{-1} \). \( ^1\text{H} \) NMR (CDCl\(_3\), 400 MHz): \( \delta = 7.47-7.40 \) (m, 2H), 7.40–7.32 (m, 1H), 7.22–7.09 (m, 2H), 6.36–6.27 (m, 2H), 6.23–6.07 (m, 2H), 5.27–5.10 (m, 4H), 2.97 (brs, 2H), 2.72–2.46 (m, 4H), 1.88–1.76 (m, 2H), 1.55–1.27 (m, 2H). \( ^{13}\text{C} \) NMR (CDCl\(_3\), 100 MHz): \( \delta = 180.5, 134.2, 132.2, 129.2, 128.7, 126.7, 118.7, 51.5, 37.5, 35.9, 18.4. \) HRMS (ESI, Q-ToF) \( m/z \): found [M + Na\(^+\)] = 356.1624, calculated = 356.1626 for \( C_{22}H_{23}NO_2 \).

Scheme 6 Synthesis of the basketene-based and anthracene derived propellane derivatives 45 and 50, respectively.
stretch at 1708 cm$.^\text{1\text{H NMR (CDCl}_3, 500 MHz):} \delta = 7.47-7.43$ (m, 2H), 7.39–7.37 (m, 1H), 7.20–7.18 (m, 2H), 6.17–6.06 (m, 2H), 5.18–5.12 (m, 4H), 3.28 (sex, 4H, $J = 3.0$ Hz), 3.27–3.23 (m, 2H), 3.15–3.12 (m, 4H), 2.65 (dd further split into triplet, 2H, $J = 1.28$ Hz, 6.24 Hz, 14.52 Hz), 2.43 (dd, 2H, $J = 8.04$ Hz, 14.52 Hz).

$^{13}$C NMR (CDCl$_3$, 125 MHz): $\delta = 180.0, 134.3, 129.2, 128.6, 126.7, 118.4, 46.5, 43.0, 39.2, 37.1, 37.0, 36.5$. HRMS (ESI, Q-ToF) $m/z$: found [M + Na]$^+$ = 380.1621, calculated = 380.1619 for C$_{24}$H$_{23}$NO$_2$.

**Compound 49.** White solid (67 mg starting from 100 mg of 48, 55%), $R_f = 0.75$ (5% EtOAc/hexane), mp: 182.8–186.1 °C. $^1$H NMR (CDCl$_3$, 400 MHz): $\delta = 7.50-7.13$ (m, 13H), 6.53–6.40 (m, 2H), 6.34–6.17 (m, 2H), 5.33–5.00 (m, 4H), 4.65 (s, 2H), 2.42 (dd, 2H, $J = 5.6$ Hz, 14.4 Hz), 2.14 (dd, 2H, $J = 8.5$ Hz, 14.6 Hz).

$^{13}$C NMR (CDCl$_3$, 100 MHz): $\delta = 178.5, 140.0, 139.4, 133.7, 131.6, 129.1, 128.8, 127.1, 126.8, 126.7, 126.5, 125.3, 119.4, 55.7, 51.6, 37.3$. HRMS (ESI, Q-ToF) $m/z$: found [M + Na]$^+$ = 454.1777, calculated = 454.1783 for C$_{30}$H$_{25}$NO$_2$.

**Fig. 4** The optimized geometries of 30, 33, 41 and 45 at the M06-2X/6-31G** level of theory. C = grey, H = white, N = blue and O = red.

### The general procedure for metathesis

The compound (0.12 mmol) was dissolved in dry dichloromethane or toluene and degassed with nitrogen gas followed by ethylene gas. The catalyst was added to the reaction under an ethylene atmosphere and the reaction was carried out under the conditions given above. After the removal of the solvent, the residue was purified using column chromatography to obtain the corresponding compound.

**Compound 30.** White solid (51 mg starting from 70 mg of 29, 68%), $R_f = 0.35$ (5% EtOAc/hexane), mp: 177.6–181.5 °C. IR (neat): $v_{\text{max}}$ C=O stretch at 1755, 1705 cm$^{-1}$. $^1$H NMR (CDCl$_3$, 500 MHz): $\delta = 7.49-7.40$ (m, 2H), 7.40–7.33 (m, 1H), 7.20–7.11 (m, 2H), 6.46–6.35 (m, 2H), 6.10–6.00 (m, 2H), 5.66–5.51 (m, 2H), 5.12–4.97 (m, 4H), 3.09–2.99 (m, 2H), 2.88 (brs, 2H), 2.78–2.65 (m, 2H), 2.40 (dd, 2H, $J = 14.6$ Hz).

$^{13}$C NMR (CDCl$_3$, 125 MHz): $\delta = 181.3, 139.5, 133.9, 132.3, 129.2, 128.7, 126.5,$
115.6, 52.5, 43.0, 42.4, 28.3. HRMS (ESI, Q-ToF) m/z: found [M + Na]+ = 352.1317, calculated = 352.1312 for C28H18NO2.

**Compound 33.** White solid (42 mg starting from 70 mg of 32, 64%), Rf = 0.38 (5% EtOAc/hexane), mp: 115.2–118.3 °C. IR (neat): v_max C=O stretch at 1706 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ = 7.45–7.38 (m, 2H), 7.36 (tt, 2H, J = 7.2 Hz, 1.4 Hz), 7.16–7.12 (m, 4H), 6.39–6.29 (m, 2H), 6.06–5.92 (m, 2H), 2.89–2.75 (m, 2H), 2.68–2.55 (m, 2H), 2.25 (d, J = 14.6 Hz, 2H), 1.91–1.80 (m, 2H), 1.40–1.29 (m, 2H). ¹³C NMR (CDCl₃, 100 MHz): δ = 177.5, 139.3, 131.9, 131.8, 129.2, 128.7, 126.5, 115.5, 46.8, 44.2, 38.3. HRMS (ESI, Q-ToF) m/z: found [M + Na]+ = 352.1317, calculated = 352.1312 for C28H18NO2.

**Compound 40.** White solid (36 mg starting from 36 mg of 40, 92%), Rf = 0.55 (5% EtOAc/hexane), mp: 150–152 °C. IR (neat): v_max C=O stretch at 1703 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ = 7.46–7.29 (m, 3H), 7.17–7.07 (m, 2H), 6.39–6.29 (m, 2H), 6.06–5.92 (m, 2H), 2.86–2.75 (m, 2H), 2.68–2.55 (m, 2H), 2.25 (d, J = 14.6 Hz, 2H), 1.91–1.80 (m, 2H), 1.40–1.29 (m, 2H). ¹³C NMR (CDCl₃, 100 MHz): δ = 181.9, 131.5, 129.1, 129.1, 128.6, 126.6, 52.2, 36.6, 28.2, 19.5. HRMS (ESI, Q-ToF) m/z: found [M + Na]+ = 328.1306, calculated = 328.1313 for C20H19NO2.

**Compound 45.** White solid (7.1 mg starting from 10 mg of 44, 65%), Rf = 0.58 (5% EtOAc/hexane), mp: 150–152 °C. IR (neat): v_max C=O stretch at 1702 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ = 7.46–7.29 (m, 3H), 7.17–7.07 (m, 2H), 6.39–6.29 (m, 2H), 6.06–5.92 (m, 2H), 2.86–2.75 (m, 2H), 2.68–2.55 (m, 2H), 2.25 (d, J = 14.6 Hz, 2H), 1.91–1.80 (m, 2H), 1.40–1.29 (m, 2H). ¹³C NMR (CDCl₃, 100 MHz): δ = 181.9, 131.5, 129.1, 129.1, 128.6, 126.6, 52.2, 36.6, 28.2, 19.5. HRMS (ESI, Q-ToF) m/z: found [M + Na]+ = 328.1306, calculated = 328.1313 for C20H19NO2.

**Conclusion**

We have developed a simple synthetic strategy to synthesize propellane containing bicyclo[2.2.2]octene derivatives which are structurally similar to 11β-HSD1 inhibitors. We have also designed interesting bicyclo[2.2.2]octene derivatives which contain a variety of unsaturated moieties to study their metathetic behaviour. The allylic groups present in compounds 40 and 49 undergo RCM to generate the corresponding propellane derivatives 41 and 50, respectively. Surprisingly, compound 40 didn’t undergo RRM due to the extra stability of the bicyclo[2.2.2]octene moiety towards metathesis. Similarly, the allylic groups present in compound 32 underwent RCM to generate the propellane derivative 33. However, compound 29 underwent RCM and ROM to generate the propellane derivative 30. We have extended this methodology to the cage compound 42 to produce the basketene-based propellane derivative 45. As shown by the geometry optimization studies, these propellane derivatives are rigid and the bond angles of the propellane derivatives 30, 33, and 41 are similar, whereas the basketene derivative 45 has larger values for its bond angles. These propellane derivatives will find useful applications in medicinal and polymer chemistry. Since cage molecules and propellanes have various applications as high energy materials, our results may have further implications in this area.

**Conflicts of interest**

There are no conflicts of interest.

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**References**


