



Cite this: RSC Adv., 2018, 8, 16383

 Received 14th March 2018
 Accepted 25th April 2018

DOI: 10.1039/c8ra02240b

rsc.li/rsc-advances

Lignans and isoflavonoids from the stems of *Pisonia umbellifera*†

 Zhiguo Liu,^a Xilong Zheng,^b Yanan Wang,^c Mengyue Tang,^a Shilin Chen,^a Fangbo Zhang,^a Li Li,^a Cun Zhang^{*a} and Yi Sun^{*a}

Twelve new compounds including four isoflavonolignans (**1a/1b** and **2a/2b**), two neolignans (**3a/3b**), and six isoflavonoids (**5–10**), together with seven known compounds (**4** and **11–16**) were isolated from the stems of *Pisonia umbellifera*. The structures were elucidated on the basis of comprehensive spectroscopic analyses and ECD calculation methods. Compounds **1–3** were present as enantiomers that were successfully separated by chiral HPLC. Compounds **1a/1b** and **2a/2b** are the first examples of isoflavonolignans with a pyranoid ring linking up the isoflavonoid and the monolignol from nature. A putative biosynthetic pathway for the isoflavonolignans was deduced. The anti-inflammatory and cytotoxic activities for all compounds were evaluated.

Introduction

Pisonia umbellifera (J. R. Forst. & G. Forst.) Seem. (Nyctaginaceae), a perennial tree growing in low altitude thickets, is widely distributed throughout Australia, Java, Malaysia, Hawaii, Taiwan and the Hainan province of China. The leaves of *P. umbellifera* are used as a folk medicine to treat ciguatera poisoning and to prevent phytophage nematodes. Previous phytochemical investigations on this plant have resulted in the isolation of triterpenoids, alkaloids, phenolic acids and isoflavonoids, of which some compounds exhibited antitubercular activity against *Mycobacterium tuberculosis*.^{1,2}

Isoflavonolignans are a type of lignoids, which incorporate a C6-aromatic-C3-side chain unit into an isoflavonone nucleus. Isoflavonolignans are derived from two phenylpropanoid units as they are biogenetically related to lignans and neolignans. So far, only a few isoflavonolignans, termed as benzodioxane lignoids have been isolated from Fabaceae family, which structures contain the catechol moieties in flavonoids.^{3–8}

P. umbellifera is a common herb medicine for Li folk in China. In our investigation searching for the novel bioactive constituents from natural resource,^{9,10} we chose the stems of *P. umbellifera* for more detailed study. Twelve new compounds, including four isoflavonolignans (**1a/1b** and **2a/2b**), two neolignans (**3a/3b**), and six isoflavonoids (**5–10**), together with

seven known compounds (**4** and **11–16**) were isolated from its stems (Fig. 1). The absolute configurations of the new isoflavonolignans (**1a/1b** and **2a/2b**) and neolignans (**3a/3b**) were elucidated by comparing their calculated and experimental electronic circular dichroism (ECD) spectra. A putative biosynthetic pathway for the isoflavonolignans was deduced. All the compounds were tested to evaluate their inhibitory effects on NO production in LPS-induced RAW 264.7 macrophages and the cytotoxic activities. Herein, we report the isolation, structure identification, and biological effects of compounds **1–16**.

Results and discussion

Compound **1** was obtained as colorless oil. The molecular formula was determined to be $C_{27}H_{24}O_8$ by HRESIMS at m/z 477.1545 [$M + H$]⁺ (calcd for $C_{27}H_{25}O_8$ 477.1544). The ¹H NMR spectrum (Table 1) of **1** revealed the presence of one enolic hydroxyl group at δ_H 12.54 (5-OH) and one diagnostic proton of isoflavone at δ_H 8.09 (1H, s, H-2). The aromatic region of spectrum displayed the signals of 1,2-disubstituted moieties which were assigned to H-3', H-4', H-5' and H-6', and was consistent with the substitution pattern in ring B of 6,8-dimethylisogenistein.² The 1,3,4-trisubstituted moieties of benzene ring were also observed with the aromatic protons at δ_H 6.95 (1H, d, $J = 7.8$ Hz, H-5''), 6.92 (1H, d, $J = 1.8$ Hz, H-2'') and 6.91 (1H, dd, $J = 7.8$, 1.8 Hz, H-6''). In addition, the aliphatic region of spectrum exhibited an AMXY coupling system with the signals at δ_H 5.02 (1H, d, $J = 9.0$ Hz, H-7''), 3.65 (1H, dd, $J = 10.8$, 4.8 Hz, H-9''), 3.55 (1H, dd, $J = 10.8$, 4.8 Hz, H-9'') and 2.25 (1H, m, H-8''). Detailed analyses of the ¹H and ¹³C NMR spectra indicated that **1** possessed the isoflavonolignan skeleton bearing a phenylpropanoid unit. The linkage of the two parts through a pyranoid ring was

^aInstitute of Chinese Materia Medica, China Academy of Chinese Medical Sciences, Beijing 100700, PR China. E-mail: ysun@icmm.ac.cn; czhang@icmm.ac.cn

^bHainan Branch Institute of Medical Plant Development, Chinese Academy of Medical Sciences, Wanning 571100, PR China

^cInstitute of Materia Medica, Chinese Academy of Medical Sciences & Peking Union Medical College, Beijing 100050, PR China. E-mail: annaleelin@imm.ac.cn

† Electronic supplementary information (ESI) available: 1D and 2D NMR, HR ESI-MS, UV and ECD spectra of **1–3** and **5–10**. See DOI: [10.1039/c8ra02240b](https://doi.org/10.1039/c8ra02240b)



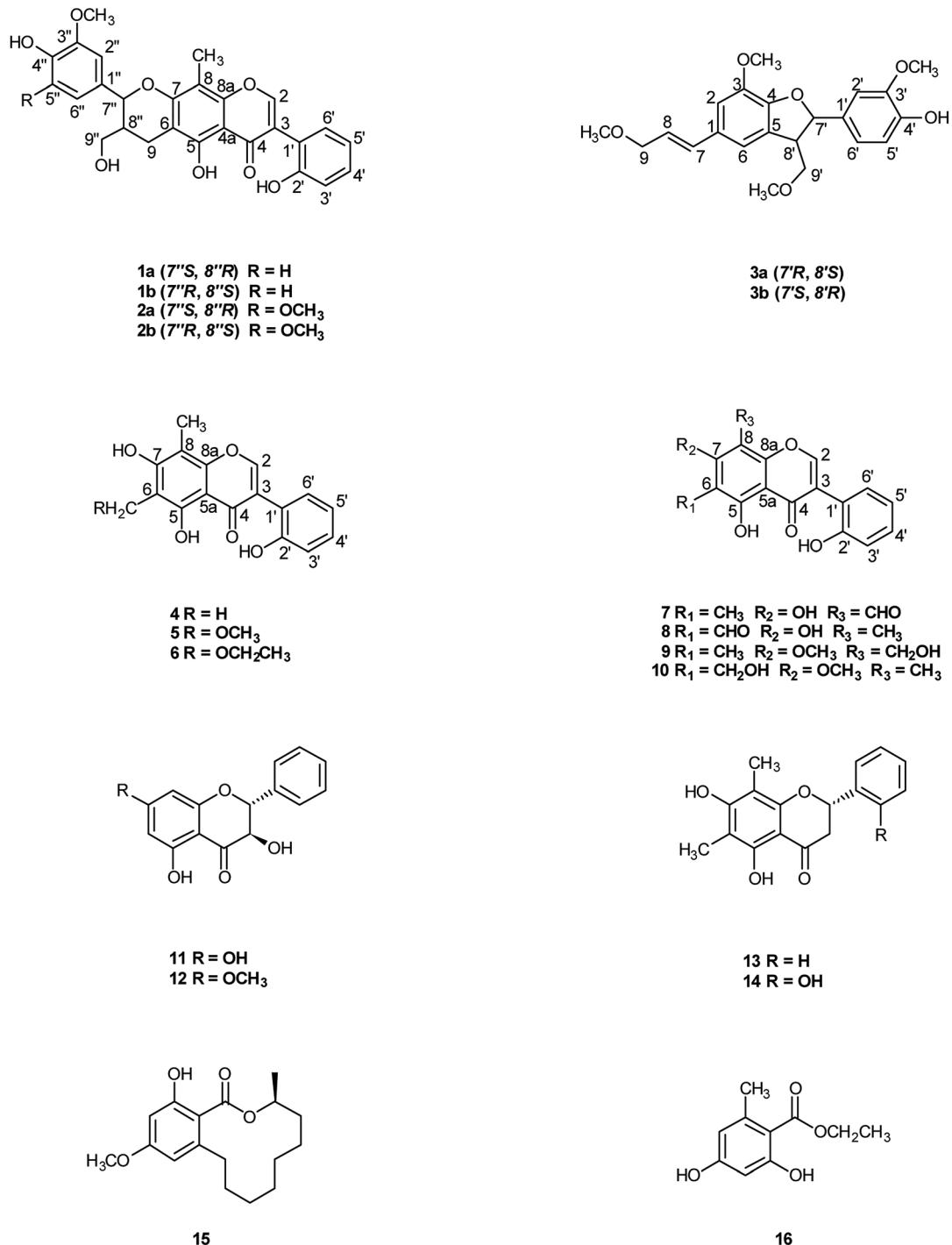


Fig. 1 Structures of compounds 1–16.

confirmed by the 2D NMR spectra. Based on the HMBC correlations (Fig. 2a) from CH₂-9 to C-7'' (δ_{C} 81.0) and C-9'' (δ_{C} 63.4), and from H-7'' to C-9 (δ_C 21.8) and C-9'' (δ_C 63.4), and the key ¹H-¹H COSY correlations from H-8'' to H-7'', CH₂-9'' and CH₂-9, respectively. The coupling constant ($J = 9.0$ Hz) between H-7'' and H-8'' in ¹H NMR spectrum and the NOESY correlations (Fig. 2b) of H-8''/H-6'' and H-7''/CH₂-9'' suggested a *trans* configuration of the chiral centers on the pyranoid ring.

The absolute configuration of **1** was determined by comparing its experiment and quantum chemical calculations of the electronic circular dichroism spectra. Notably, the specific rotation of **1** displayed value of zero (*c* 0.1, MeOH). Thus, chiral-phase HPLC analysis and resolution of **1a** and **1b** were carried out on a Daicel Chiralpak AS-H chiral column using *n*-hexane/isopropanol (7 : 3) as mobile phase (Fig. 3a). The absolute configurations of **1a** and **1b** were determined by ECD

Table 1 ^1H (600 MHz) and ^{13}C NMR (150 MHz) NMR data for compounds 1–3

Position	1^a		2^b		3^b	
	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}
1						132.3
2	8.09, s	156.0	8.19, s	157.2	6.98, br s	112.2
3		122.9		122.0		145.6
4		182.5		182.8		149.4
4a		104.8		106.0		
5		157.2		158.1		130.4
6		106.5		107.0	6.98, br s	116.6
7		159.7		160.1	5.59, d (16.2)	134.3
8		103.2		103.8	6.19, m	124.3
8a		153.3		154.7		
9	2.74, dd (17.4, 10.2) 2.98, dd (17.4, 5.4)	21.8	2.73, dd (16.8, 10.8) 2.99, dd (16.8, 4.8)	22.6	4.08, dd (6.0, 1.2)	74.3
8-CH ₃	2.22, s	7.7	2.23, s	7.6		
3-OCH ₃					3.90, s	56.8
9-OCH ₃					3.39, s	58.0
1'		120.4		119.8		134.3
2'		156.2		156.8	6.96, d (1.8)	110.6
3'	7.10, dd (7.8, 1.2)	119.8	6.94, dd (7.8, 1.2)	117.2		149.1
4'	7.36, ddd (7.8, 7.8, 1.8)	130.8	7.29, ddd (7.8, 7.8, 1.8)	132.7		147.7
5'	7.00, ddd (7.8, 7.8, 1.2)	121.3	6.92, ddd (7.8, 7.8, 1.2)	120.7	6.79, d (7.8)	116.2
6'	7.19, dd (7.8, 1.8)	129.9	7.26, dd (7.8, 1.8)	130.9	6.84, dd (7.8, 1.8)	119.9
7'					5.50, d (6.6)	89.7
8'					3.63, m	52.7
9'					3.65, m	75.7
					3.72, m	
3'-OCH ₃					3.84, s	56.4
9'-OCH ₃					3.41, s	59.3
1''		131.1		131.4		
2''	6.92, d (1.8)	109.1	6.73, s	105.3		
3''		147.0		149.4		
4''		146.1		136.7		
5''	6.95, d (7.8)	114.6		149.4		
6''	6.91, dd (7.8, 1.8)	120.0	6.73, s	105.3		
7''	5.02, d (9.0)	81.0	5.02, d (8.4)	82.0		
8''	2.25, m	39.7	2.26, m	40.8		
9''	3.55, dd (10.8, 4.8) 3.65, dd (10.8, 4.8)	63.4	3.43, dd (10.8, 4.8) 3.55, dd (10.8, 4.8)	63.4		
3''-OCH ₃	3.92, s	56.2	3.87, s	56.8		
5''-OCH ₃			3.87, s	56.8		
5-OH	12.54, s					

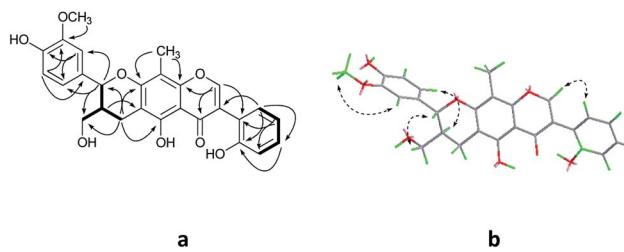
^a Recorded in CDCl₃. ^b Recorded in methanol-d₄.

Fig. 2 The key HMBC (H→C) and COSY (—) (a), and NOESY (b) correlations of 1.

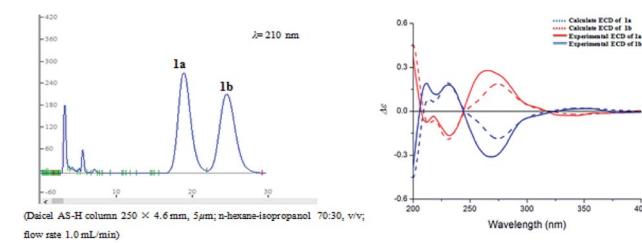


Fig. 3 (A) Chiral-phase HPLC analytical chromatogram of 1. (B) Comparison of calculated and experimental ECD spectra of 1.



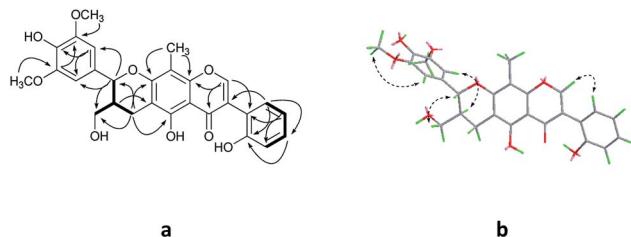


Fig. 4 The key HMBC ($\text{H} \rightarrow \text{C}$) and COSY (—) (a), and NOESY (b) correlations of 2.

calculations. The experimental ECD curves of **1a** and **1b** were similar to the computed curves of **1a** ($7''S$ and $8''R$) and **1b** ($7''R$ and $8''S$), respectively (Fig. 3b). Therefore, the structures of **1a** and **1b** were finally established as ($7''S,8''R$)-5-hydroxy-7''-(4''-hydroxy-3''-methoxyphenyl)-8''-(hydroxymethyl)-3-(2'-hydroxyphenyl)-8-methyl-7'',8''-dihydropyrano[3,2-g]chromen-4(9H)-one and ($7''R,8''S$)-5-hydroxy-7''-(4''-hydroxy-3''-methoxyphenyl)-8''-(hydroxymethyl)-3-(2'-hydroxyphenyl)-8-methyl-7'',8''-dihydropyrano[3,2-g]chromen-4(9H)-one, respectively. They are named as pisonone A and pisonone B. As far as we know, these two compounds are the first examples of isoflavanolignans featuring a pyranoid ring in nature.

Compound 2 was isolated as yellow oil, and had a molecular formula of $\text{C}_{28}\text{H}_{26}\text{O}_6$ as determined by HRESIMS ion at m/z 507.1649 [$\text{M} + \text{H}$]⁺. The ¹H and ¹³C NMR data (Table 1) showed similar signals to those of **1**, and indicated that the only difference was the presence of another methoxy group in **2**. This substitution was placed at C-5'' by the correlations (Fig. 4a) between the OCH_3 (δ_{H} 3.87, s) and C-5'' (δ_{C} 149.4) in the HMBC spectrum. The relative configurations of C-7'' and C-8'' were deduced from analysis of the coupling constant ($J = 8.4$ Hz) and the key NOESY correlations (Fig. 4b) of H-8''/H-6'' and H-7''/CH₂-9'', and suggested a *trans* relationship between them. For the same reason as **1**, the racemate of **2a** and **2b** was obtained by the chiral-phase separation. The calculated ECD spectra were in good agreement with the experimental spectra (Fig. 5), confirming that compounds **2a** and **2b** possessed the ($7''S$ and $8''R$) and ($7''R$ and $8''S$) configurations, respectively. Thus, the planar structure of **2** were elucidated as 5-hydroxy-7''-(4''-hydroxy-3'',5''-dimethoxyphenyl)-8''-(hydroxymethyl)-3-(2'-hydroxyphenyl)-8-methyl-7'',8''-dihydropyrano[3,2-g]chromen-4(9H)-one. Compounds **2a** and **2b** are named as pisonone C and pisonone D, respectively.

Compound 3 was acquired as colorless oil. On the basis of its HRESIMS at m/z 409.1621 [$\text{M} + \text{Na}$]⁺, its molecular formula was

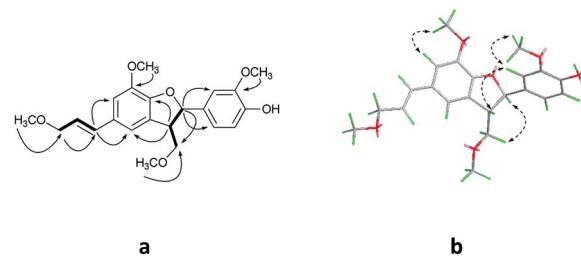


Fig. 6 The key HMBC ($\text{H} \rightarrow \text{C}$) and COSY (—) (a), and NOESY (b) correlations of 3.

determined as $\text{C}_{22}\text{H}_{26}\text{O}_6$. The ¹H NMR spectrum (Table 1) showed two groups of aromatic protons at δ_{H} 6.98 (2H, br s, H-2 and H-6), 6.96 (1H, d, $J = 1.8$ Hz, H-2'), 6.84 (1H, dd, $J = 7.8$, 1.8 Hz, H-6') and 6.79 (1H, d, $J = 7.8$ Hz, H-5') corresponding to 1,3,4,5-tetrasubstituted and 1,3,4-trisubstituted phenyl moieties. In addition, two *trans* olefinic protons at δ_{H} 5.59 (1H, d, $J = 16.2$ Hz, H-7) and 6.19 (1H, m, H-8), one oxymethine protons at δ_{H} 5.50 (1H, d, $J = 6.6$ Hz, H-7'), two oxymethylene protons at δ_{H} 4.08 (2H, dd, $J = 6.0$, 1.2 Hz, H-9), 3.72 (1H, m, H-9') and 3.65 (1H, m, H-9') were observed. The ¹³C NMR and HSQC spectra revealed the presence of 22 carbon resonances, including 12 phenyl carbons, 2 olefinic carbon, 4 methoxy carbons, and 4 aliphatic carbons. Comparison of the NMR data with those of the dehydrodiconiferyl alcohol,¹¹ they had a similar planar structure except for the presence of a methoxy group in **3**. The methoxy group was determined at C-9' by the HMBC correlations (Fig. 6a). According to the $J_{7',8'}$ value (6.6 Hz) and key NOESY correlations (Fig. 6b) of H-8'/H-2' and H-7'/CH₂-9', the relative configuration of C-7' and C-8' was suggested to be *trans*. The racemate of **3a** and **3b** was also separated by chiral HPLC, and the absolute configuration of their furan rings was identified by comparing of their experimental and calculated ECD data (Fig. 7). Thus, compounds **3a** and **3b** were determined as shown and named as ($7''R,8''S$)-9'-methoxy-dehydrodiconiferyl alcohol and ($7''S,8''R$)-9'-methoxy-dehydrodiconiferyl alcohol.

Compound 5 was obtained as yellow needles. It showed a quasi-molecular ion peak at m/z 329.1020 [$\text{M} + \text{H}$]⁺, corresponding to the molecular formula of $\text{C}_{18}\text{H}_{16}\text{O}_6$. The 1D NMR spectra (Table 2) displayed similar substitution pattern as **4**, the major difference between the two compounds was the methyl etherification at C-9 in **5**. The above conclusion was confirmed by the HMBC correlations of OCH_3 with C-9 (δ_{C} 67.8). Thus, the structure of **5** was assigned as 5,7,2'-trihydroxy-6-methylene-8-methyl-9-methoxyisoflavone, and named as pisonone E.

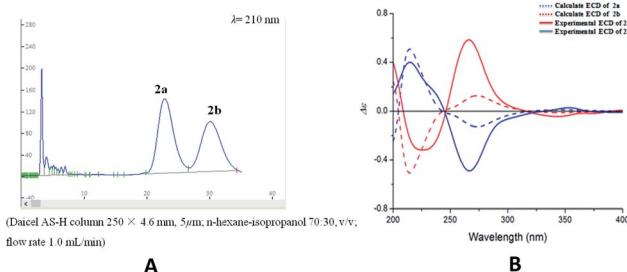


Fig. 5 (A) Chiral-phase HPLC analytical chromatogram of 2. (B) Comparison of calculated and experimental ECD spectra of 2.

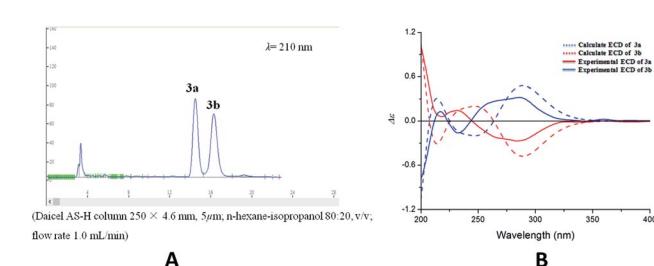


Fig. 7 (A) Chiral-phase HPLC analytical chromatogram of 3. (B) Comparison of calculated and experimental ECD spectra of 3.



Table 2 ^1H (600 MHz, CDCl_3) and ^{13}C (150 MHz, CDCl_3) NMR data for compounds 5–7

Position	5		6		7	
	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}
2	8.09, s	156.0	8.08, s	155.9	8.12, s	154.8
3		123.1		123.0		125.2
4		182.5		182.4		181.3
4a		104.6		104.5		104.5
5		156.5		156.3		165.4
6		105.6		105.9		109.9
7		162.4		162.5		168.2
8		103.5		103.4		102.5
8a		154.9		154.8		158.2
9	4.90, s	67.8	4.92, s	65.9		
6-CH ₃					2.16, s	6.7
8-CHO					10.37, s	189.8
8-CH ₃	2.25, s	7.4	2.24, s	7.3		
9-OCH ₃	3.53, s	58.9				
9-OCH ₂ CH ₃			3.70, dd (14.4, 7.2)	67.2		
9-OCH ₂ CH ₃			1.33, t (7.2)	15.1		
1'		120.2		120.2		119.0
2'		156.2		156.1		155.9
3'	7.10, dd (7.8, 1.2)	119.8	7.10, dd (7.8, 1.2)	119.8	7.11, dd (7.8, 1.2)	119.9
4'	7.36,ddd (7.8, 7.8, 1.8)	130.8	7.36,ddd (7.8, 7.8, 1.8)	130.8	7.40,ddd (7.8, 7.8, 1.8)	131.4
5'	7.00,ddd (7.8, 7.8, 1.2)	121.3	7.00,ddd (7.8, 7.8, 1.2)	121.3	7.04,ddd (7.8, 7.8, 1.2)	121.6
6'	7.18, dd (7.8, 1.8)	129.9	7.17, dd (7.8, 1.8)	129.9	7.19, dd (7.8, 1.8)	130.2
5-OH	12.59, s		12.58, s			
7-OH	9.33, br s		9.62, br s			
2'-OH	8.22, br s		8.25, br s			

Compound **6** was isolated as a yellow powder with the molecular formula of $\text{C}_{19}\text{H}_{18}\text{O}_6$ (HRESIMS and NMR data). It has more 14 mass units than that of compound **5**. The ^1H and ^{13}C NMR data of **6** (Table 2) were similar to those of **5**, indicating that the only difference between the two structures was the presence of an additional oxymethylene group in **6**. The oxymethylene group was assigned to C-9 by the HMBC correlations from OCH_2CH_3 -9 to C-9 (δ_{C} 65.9) and OCH_2CH_3 -9 (δ_{C} 15.1), as well as the ^1H – ^1H COSY correlations from OCH_2CH_3 -9 (δ_{H} 3.70) to OCH_2CH_3 -9 (δ_{H} 1.33). Thus, the structure of **6** was defined as 5,7,2'-trihydroxy-6-methylene-8-methyl-9-ethoxyisoflavone, and named as pisonone F.

Compounds **7** and **8** were obtained as colorless oil, and determined to have the same molecular formula $\text{C}_{17}\text{H}_{12}\text{O}_6$ by ^{13}C NMR and HRESIMS data. Extensive analysis of 1D and 2D NMR data (Tables 2 and 3) indicated that their structures were similar to that of **4**, and the major differences were the presence of methyl groups at C-6 or C-8, replaced by an aldehyde group. The position of the aldehyde groups were located at C-8 in **7** and at C-6 in **8**, respectively, which were elucidated by the HMBC correlations from CHO-8 to C-7 (δ_{C} 168.2) and C-8a (δ_{C} 158.2) in **7**, as well as the HMBC correlations from CHO-6 to C-5 (δ_{C} 166.0) and C-7 (δ_{C} 166.2) in **8**. Thus, the structures of **7** and **8** were established as 5,7,2'-trihydroxy-6-methyl-8-aldehydeisoflavone and 5,7,2'-trihydroxy-6-aldehyde-8-methylisoflavone, and named as pisonone G and pisonone H, respectively.

Compounds **9** and **10** were isolated as colorless oil. Its molecular formula was $\text{C}_{18}\text{H}_{16}\text{O}_6$ as deduced by their HRESIMS and NMR data.

Comparison of their ^1H and ^{13}C NMR data (Table 3) suggested that they displayed similar structure to that of **4**, the differences between them were the presence of a hydroxymethyl group and a methoxy group in **9** and **10**. The HMBC correlations from CH_2OH -8 to C-7 (δ_{C} 164.6) and C-8a (δ_{C} 153.5) in **9** and the correlations from CH_2OH -6 to C-5 (δ_{C} 158.4) and C-7 (δ_{C} 163.7) in **10** indicated that the hydroxymethyl groups were placed at C-8 in **9** and at C-6 in **10**, respectively. The methoxy groups in both **9** and **10** were located at C-7 by the relevant HMBC and NOESY experiments. Thus, the structures of **9** and **10** were deduced to be 5,2'-dihydroxy-6-methyl-7-methoxy-8-hydroxymethylisoflavone and 5,2'-dihydroxy-6-hydroxymethyl-7-methoxy-8-methylisoflavone, and named as pisonone I and pisonone J.

The seven known compounds were identified as 6,8-dimethylisogenistein (**4**),² pinobanksin (**11**),¹² alpinone (**12**),¹³ desmethoxymatteucinol (**13**),¹⁴ matteucin (**14**),¹⁵ ozoroalide (**15**),¹⁶ and ethyl orsellinate (**16**)¹⁷ by comparing their spectroscopic data with the literature values.

The analysis of the biosynthetic pathway of the isoflavanolignans (**1a/1b**, **2a/2b**) revealed that the structures were formed via phenylpropanoid pathway transforming phenylalanine into 4-coumaroyl-CoA, which extended the isoflavone and monolignol pathways. The condensation of three malonyl-CoAs and one 4-coumaroyl-CoA produced the isoflavone fragment **A**, which was methylated on the C-6 and C-8 of the B-ring. The methyl group at C-6 was then oxidized to yield fragment **B**, which was the precursor of the isoflavanoid part for compounds **1** and **2**. On the other hand, coniferyl alcohol (fragment **C**)



Table 3 ^1H (600 MHz, CDCl_3) and ^{13}C (150 MHz, CDCl_3) NMR data for compounds 8–10

Position	8		9		10	
	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}
2	8.09, s	156.2	8.15, s	156.2	8.17, s	156.6
3		123.6		123.7		123.6
4		182.7		182.8		183.0
4 α		104.0		107.6		107.9
5		166.0		160.2		158.4
6		106.7		115.6		118.3
7		166.2		164.6		163.7
8		103.7		112.5		110.0
8 α		159.2		153.5		154.9
6-CHO	10.42, s	193.0				
6-CH ₂ OH					4.84, s	62.6
6-CH ₃			2.26, s	8.7		
7-OCH ₃			3.95, s	54.2	3.92, s	55.0
8-CH ₂ OH			4.93, s	62.5		
8-CH ₃	2.23, s	6.7			2.34, s	8.5
1'		119.2		119.8		119.7
2'		155.9		156.1		155.9
3'	7.11, dd (7.8, 1.2)	119.8	7.11, dd (7.8, 1.2)	119.8	7.11, dd (7.8, 1.2)	119.8
4'	7.39, ddd (7.8, 7.8, 1.8)	131.3	7.38, ddd (7.8, 7.8, 1.8)	131.0	7.38, ddd (7.8, 7.8, 1.8)	131.0
5'	7.03, ddd (7.8, 7.8, 1.2)	121.6	7.02, ddd (7.8, 7.8, 1.2)	121.5	7.02, ddd (7.8, 7.8, 1.2)	121.5
6'	7.19, dd (7.8, 1.8)	130.0	7.17, dd (7.8, 1.8)	130.1	7.18, dd (7.8, 1.8)	130.0
5-OH			12.62, s		12.58, s	

formed the lignin moiety. Finally, the biosynthesis of isoflavanolignans (**1a/b** and **2a/b**) occurred by the elimination of water and cyclization between flavonoid precursor and coniferyl alcohol (Fig. 8).

All compounds were evaluated for their *in vitro* inhibitory effects on LPS-induced NO production in macrophages (Table 4). Among them, compounds **13** and **15** exhibited moderate inhibitory activities with IC_{50} values of 33.15 and 11.30 μM , respectively.

The cytotoxic activities of compounds **1–16** were evaluated against human chronic myelogenous leukemia (K562) and human lung carcinoma (A549) cell lines that present adherent and non-adherent cancer cell type models (Table 5). Compound **5** exhibited the strongest activity against K562 cell line among all the isolates, whereas some of the compounds showed moderate activities against both cell lines. Compounds **1–3** were proved to be inactive in both assays.

Conclusions

Pisonia umbellifera is a folk medicine in Hainan province of China. In our investigation searching for novel bioactive constituents from natural source, twelve new compounds along with seven known ones were isolated from *P. umbellifera*. Compounds **1** and **2** are the first examples of isoflavanolignans featuring a pyranoid ring in nature, which enriches the structural types of this lignans. The anti-inflammatory and cytotoxic activities of all compounds were evaluated. Compounds **13** and **15** exhibited moderate *in vitro* anti-inflammatory activities, and compound **5** exhibited the strongest activity against K562 cell line among all the isolates.

Experimental

General experimental procedures

Optical rotations were measured with a Perkin-Elmer 241 polarimeter. The UV spectra were conducted on a Shimadzu UV-2201 spectrometer. ECD spectra were recorded on a JASCO J-815 spectrometer. HRESIMS were recorded on Agilent Accurate-Mass-Q-TOF MS 6520 system equipped with electrospray ionization (ESI) source. NMR spectra were recorded on a Bruker AV-600 NMR spectrometer using TMS as an internal standard. The chromatographic silica gel (200–300 mesh) and polyamide (100–200 mesh) were purchased from Qingdao Marine Chemical Factory (Qingdao, China) and ODS (50 μm) was produced by YMC Co. LTD., Kyoto, Japan. Sephadex LH-20 was purchased from GE Healthcare. RP-HPLC separations were conducted using a Shimadzu LC-20AT liquid chromatograph with a YMC-PACK ODS-A column (250 \times 10 mm, 5 μm) and Knauer K-2600 UV detector. Chiral separation was conducted in a Daicel AS-H column (250 \times 4.6 mm, 5 μm) from Daicel Chemical Industries, Ltd. (Japan). TLC spots were visualized under UV light and by dipping into 10% H_2SO_4 in EtOH followed by heating. RAW 264.7 cells, human chronic myelogenous leukemia (K562) and human lung carcinoma (A549) cell lines were obtained from National infrastructure of cell line resource.

Plant material

The plant material was collected in Hainan province, China, and identified by Professor Xilong Zheng, Hainan Branch Institute of Medical Plant Development, Chinese Academy of Medical Sciences, Wanning, China. The voucher specimen (GM-20170121) was deposited in the Institute of Chinese Materia



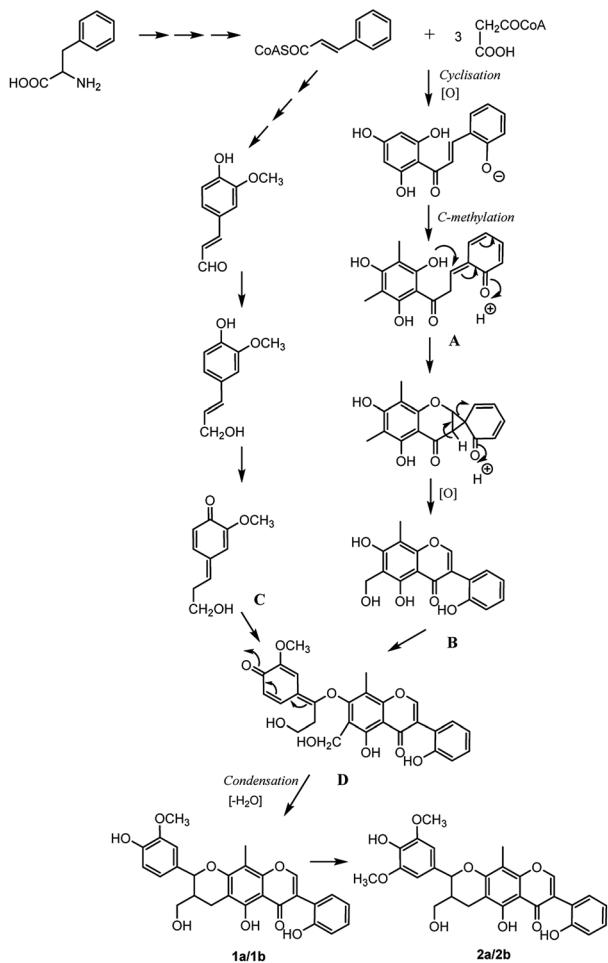


Fig. 8 Possible biosynthetic pathway of compounds **1a/1b** and **2a/2b** from *P. umbellifera*.

Medica, China Academy of Chinese Medical Sciences, Beijing, China.

Extraction and isolation

The dried stems of *P. umbellifera* (5 kg) were extracted with 95% EtOH (3×20 L) three times (3 h, 2 h, 1 h) under reflux. The EtOH extracts were concentrated to afford a crude extract (500 g), which was suspended in H₂O and partitioned successively with petroleum ether, CH₂Cl₂ and EtOAc. The CH₂Cl₂ extract (35 g) was subjected to column chromatography (CC) on silica gel using a gradient system of increasing polarity of petroleum ether-acetone (100 : 0–100, v/v) to give 10 fractions (Fr. A–Fr. J). Fr. D (3.2 g) was chromatographed over RP-C₁₈ silica gel eluted with MeOH–H₂O (from 60 to 80%) to give three fractions Fr. D1–Fr. D3. Fr. D1 (200 mg) was subjected to Sephadex LH-20 CC eluted with MeOH to give two subfractions (Fr. D1A–Fr. D1B). Fr. D1A was repeatedly recrystallized in petroleum ether-acetone (1 : 1) to give **3** (2.6 mg) and **15** (2.1 mg). Fr. D1B was further purified by a semi-preparative HPLC (CH₃CN–H₂O, 75 : 25) to afford **5** (3.4 mg), **6** (8.9 mg) and **13** (2.9 mg). Fr. F (2.0 g) was divided into subfractions Fr. F1 and Fr. F2 by reversed-phase ODS CC eluted with MeOH–H₂O (from 50 to 80%).

Table 4 Inhibitory effect of compounds **1–16** on LPS-induced NO production in macrophages^a

Compounds	IC ₅₀ ± SD ^b (μM)	Compounds	IC ₅₀ ± SD (μM)
1a	>100	8	>100
1b	>100	9	>100
2a	>100	10	>100
2b	>100	11	53.91 ± 0.45
3a	>100	12	84.32 ± 0.72
3b	>100	13	33.15 ± 1.18
4	73.90 ± 0.24	14	>100
5	73.13 ± 0.66	15	11.30 ± 0.87
6	>100	16	52.09 ± 1.21
7	>100	Resveratrol	40.20 ± 0.94

^a Resveratrol was used as positive control. Results are presented as the means ± SD ($n = 3$). ^b Concentration necessary for 50% inhibition (IC₅₀).

(2.4 mg) and **12** (3.7 mg) were obtained from Fr. F1 using Sephadex LH-20 column (MeOH). Fr. F2 was separated by semi-preparative HPLC on an ODS column (CH₃CN–H₂O, 55 : 45) to obtain compounds **9** (2.6 mg) and **10** (3.0 mg). Fr. H (2.5 g) was purified on a polyamide column, eluted with a gradient of MeOH–H₂O from 2 : 8 to 9 : 1, to yield four subfractions (Fr. H1–Fr. H4). Fr. H1 (320 mg) was subjected to a silica gel column eluted with petroleum ether–EtOAc (from 10 : 1 to 2 : 1), followed by recrystallization from MeOH to give **4** (4.1 mg) and **11** (3.5 mg). **1** (2.3 mg) and **14** (3.2 mg) was isolated from Fr. H2 by a Sephadex LH-20 column chromatography (CH₂Cl₂–MeOH, 1 : 1). Fr. H4 was further purified by using semi-preparative HPLC (CH₃CN–H₂O, 65 : 15) to yield **7** (2.6 mg), **8** (2.4 mg) and **16** (4.6 mg).

Pisonones A (1a) and B (1b). Colorless oil; $[\alpha]_D^{25}$ (undetected); UV (MeOH) λ_{\max} (log ϵ) 237 (0.36) nm; 271 (0.12) nm; ECD (MeOH) λ_{\max} ($\Delta\epsilon$) **1a**, 232 (−0.17) and 266 (+0.28) nm; **1b**, 231 (+0.18) and 268 (−0.31) nm; ¹H NMR (600 MHz, CDCl₃) and ¹³C NMR (150 MHz, CDCl₃) data, see Table 1; HRESIMS *m/z* 477.1545 [M + H]⁺ (calcd for C₂₇H₂₅O₈, 477.1544).

Pisonones C (2a) and D (2b). Yellow oil; $[\alpha]_D^{25}$ (undetected); UV (MeOH) λ_{\max} (log ϵ) 239 (0.47) nm; 270 (0.11) nm; ECD (MeOH) λ_{\max} ($\Delta\epsilon$) **2a**, 231 (−0.33) and 266 (+0.64) nm; **2b**, 212 (+0.48) and 267 (−0.53) nm; ¹H NMR (600 MHz, MeOD) and ¹³C NMR (150 MHz, MeOD) data, see Table 1; HRESIMS *m/z* 507.1649 [M + H]⁺ (calcd for C₂₈H₂₇O₉, 507.1650).

9'-Methoxy-dehydrodiconiferyl alcohol (3a and 3b). Colorless oil; $[\alpha]_D^{25}$ (undetected); UV (MeOH) λ_{\max} (log ϵ) 240 (0.21) nm; 280 (0.48) nm; ECD (MeOH) λ_{\max} ($\Delta\epsilon$) **3a**, 232 (+0.14) and 284 (−0.27) nm; **3b**, 233 (−0.20) and 287 (+0.29) nm; ¹H NMR (600 MHz, MeOD) and ¹³C NMR (150 MHz, MeOD) data, see Table 1; HRESIMS *m/z* 409.1621 [M + Na]⁺ (calcd for C₂₂H₂₆NaO₆, 409.1622).

Pisonone E (5). Yellow needles; UV (MeOH) λ_{\max} (log ϵ) 222 (0.75) nm; 267 (0.15) nm; ¹H NMR (600 MHz, CDCl₃) and ¹³C NMR (150 MHz, CDCl₃) data, see Table 2; HRESIMS *m/z* 329.1020 [M + H]⁺ (calcd for C₁₈H₁₇O₆, 329.1020).

Table 5 Cytotoxicity of compounds **1–16** (IC_{50} in μM)^a on A549 and K562 cells

Compounds	A549	K562	Compounds	A549	K562
1a	–	–	10	–	–
2a	–	–	11	36.8	18.4
3a	–	–	12	35.0	17.5
4	33.6	16.8	13	–	17.6
5	13.6	6.7	14	–	16.7
6	13.1	14.6	15	34.2	7.5
7	–	35.3	16	–	11.2
8	–	–	Adriamycin	2.9	2.0
9	–	–			

^a Presents a pair of enantiomers; “–” presents IC_{50} values $> 50 \mu\text{M}$.

Pisonone F (6). Yellow powder; UV (MeOH) λ_{max} ($\log \epsilon$) 232 (0.11) nm; 268 (0.22) nm; ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) data, see Table 2; HRESIMS m/z 365.0995 [$\text{M} + \text{Na}$]⁺ (calcd for $\text{C}_{19}\text{H}_{18}\text{NaO}_6$, 365.0996).

Pisonone G (7). Colorless oil; UV (MeOH) λ_{max} ($\log \epsilon$) 241 (0.42) nm; 274 (0.81) nm; ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) data, see Table 2; HRESIMS m/z 313.0707 [$\text{M} + \text{H}$]⁺ (calcd for $\text{C}_{17}\text{H}_{13}\text{O}_6$, 313.0707).

Pisonone H (8). Colorless oil; UV (MeOH) λ_{max} ($\log \epsilon$) 241 (0.56) nm; 287 (0.45) nm; ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) data, see Table 3; HRESIMS m/z 313.0707 [$\text{M} + \text{H}$]⁺ (calcd for $\text{C}_{17}\text{H}_{13}\text{O}_6$, 313.0707).

Pisonone I (9). Colorless oil; UV (MeOH) λ_{max} ($\log \epsilon$) 240 (0.68) nm; 263 (0.21) nm; ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) data, see Table 3; HRESIMS m/z 329.1020 [$\text{M} + \text{H}$]⁺ (calcd for $\text{C}_{18}\text{H}_{17}\text{O}_6$, 329.1020).

Pisonone J (10). Colorless oil; UV (MeOH) λ_{max} ($\log \epsilon$) 234 (0.77) nm; 263 (0.17) nm; ^1H NMR (600 MHz, CDCl_3) and ^{13}C NMR (150 MHz, CDCl_3) data, see Table 3; HRESIMS m/z 329.1020 [$\text{M} + \text{H}$]⁺ (calcd for $\text{C}_{18}\text{H}_{17}\text{O}_6$, 329.1020).

ECD computation section

The absolute configurations shown as **1a/2a/3a** were used as the input configuration of theoretical calculations. Firstly, a systematic conformational analysis was performed to find all possible conformers within a 3 kcal mol⁻¹ energy window in the MMFF94 force field. For **3a**, the 3-methoxypropen-1-yl group was replaced by vinyl group to reduce the computation. The obtained conformers were further optimized and identified as the stable conformers at the B3LYP/6-31G (d) level by the Gaussian 09 program.¹⁸ Main conformers (Boltzmann distribution $> 1\%$) of each compound were chosen for ECD calculations. All quantum computations are performed on an IBM cluster machine located at the High Performance Computing Center of Peking Union Medical College. The lowest 120 electronic excitations were calculated and energies, oscillator strengths, and rotational strengths (velocity) of each electronic excitation were obtained. ECD spectra were then simulated with a half-bandwidth of 0.30 or 0.40 eV. By comparison of the

calculated and experimental ECD spectra, their configurations were established.

Chiral-phase separation

Compounds **1** and **2** were both separated by a chiral-phase chromatographic AS-H column (4.6 \times 250 mm, 5 μm), using *n*-hexane/isopropanol (7 : 3) as mobile phase to yield **1a** ($t_{\text{R}} = 18.9$ min, 0.4 mg), **1b** ($t_{\text{R}} = 24.5$ min, 0.5 mg), **2a** ($t_{\text{R}} = 22.8$ min, 0.5 mg) and **2b** ($t_{\text{R}} = 30.1$ min, 0.6 mg), respectively. The enantiomers of **3a** ($t_{\text{R}} = 14.5$ min, 0.5 mg) and **3b** ($t_{\text{R}} = 16.3$ min, 0.5 mg) were also obtained by chiral HPLC column eluted with *n*-hexane/isopropanol (8 : 2).

NO production inhibition bioassay

The nitrite concentration in the medium was measured as an indicator of NO production according to the Griess reaction. Briefly, RAW 264.7 cells were seeded into 96-well plates at density of 1×10^5 cell per well and stimulated with 1 $\mu\text{g mL}^{-1}$ of LPS in the presence or absence of test compounds. After incubation at 37 °C for 24 h, 100 μL of cell-free supernatant was mixed with 100 μL of Griess reagent containing equal volumes of 2% (w/v) sulfanilamide in 5% (w/v) phosphoric acid and 0.2% (w/v) of *N*-(1-naphthyl) ethylenediamine solution to determine nitrite production. Absorbance was measured in microplate reader at 540 nm against a calibration curve with NaNO_2 standards. Experiments were performed in triplicate, and data were expressed as the mean \pm SD of three independent experiments.^{19,20}

Assay for cytotoxic activity

The cytotoxic activities of compounds **1–16** against human lung adenocarcinoma strain A-549 were determined using the 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-*H*-tetrazolium bromide (MTT) assay. K562 cell viability was assessed by trypan blue exclusion test. Both A-549 and K562 cells were maintained in DMEM medium containing 10% Fetal Bovine Serum (FBS) and 0.4% penicillin-streptomycin solution (100 \times) at 37 °C under 5% CO_2 . When A549 cells showed logarithmic growth, diluted them to a concentration of 1×10^4 cells per mL, the diluted cell suspensions (200 μL , containing 2000 cells per well) were placed into 96-well microtiter plates and incubated at 37 °C for 24 h, at a 5% CO_2 atmosphere. K562 was prepared at a concentration of 1×10^4 cells per mL directly, and diluted to 96-well microtiter plates. After 72 h incubation, every well was added with the MTT or MTS solution and maintained for another 4 h. Then the cells were fixed and stained with 150 μL DMSO for 10 min. The absorption was measured at 570 nm in an Ultra Microplate Reader (Elx 808, BIO-TEX Instruments, Inc.).²¹

Conflicts of interest

There are no conflicts to declare.



Acknowledgements

This work was supported by the National Natural Science Foundation of China (No. 81502968), National Major New Drug Creation Project of China (No. 2014ZX09509001-001), and CAMS Innovation Fund for Medical Sciences (CIFMS) 2016-12M-3-010.

Notes and references

- 1 C. Lavaud, S. Beauviere, G. Massio, L. L. Men-Olivier and G. Bourdy, *Phytochemistry*, 1996, **43**, 189–194.
- 2 H. T. Kuo, C. F. Peng, H. Y. Huang, C. H. Lin, I. S. Chen and I. L. Tsai, *Planta Med.*, 2011, **77**, 736–741.
- 3 S. Bezuidenhout Catherine, C. B. Bzuidenhout Barend, E. Brandt Vincent and D. Ferreira, *J. Chem. Soc., Perkin Trans. 1*, 1988, 1237.
- 4 D. S. Jang, E. J. Park, M. E. Hawthorne, J. S. Vigo, J. G. Graham, F. Cabieses, B. D. Santarsiero, A. D. Mesecar, H. H. S. Fong, R. G. Mehta, J. M. Pezzuto and A. D. Kinghorn, *J. Nat. Prod.*, 2003, **66**, 583–587.
- 5 K. Ma, T. Ishikawa, H. Seki, K. Furihata, H. Ueki and S. Narimatsu, *Heterocycles*, 2005, **65**, 893–900.
- 6 Z. Zhang, G. M. Liu, Y. H. Wang, C. S. Li, Y. L. Ren, H. Xiao, X. J. Hao and H. P. He, *Chin. Chem. Lett.*, 2007, **18**, 297–299.
- 7 R. J. Yang, Y. S. Lan, Z. J. Huang, C. L. Shao, H. Liang, Z. F. Chen and J. Li, *Rec. Nat. Prod.*, 2012, **6**, 212–217.
- 8 G. P. Oliveira, T. M. G. Silva, C. A. Camara, A. L. B. D. Santana, M. S. A. Moreira and T. M. S. Silva, *Phytochem. Lett.*, 2017, **22**, 61–70.
- 9 Z. G. Liu, Z. L. Li, J. Bai, D. L. Meng, N. Li, Y. H. Pei, F. Zhao and H. M. Hua, *J. Nat. Prod.*, 2014, **77**, 792–799.
- 10 Z. G. Liu, Z. L. Li, J. Bai, D. L. Meng, N. Li, Y. H. Pei, F. Zhao and H. M. Hua, *Bioorg. Med. Chem. Lett.*, 2016, **26**, 1–5.
- 11 Y. B. Liu, X. R. Cheng, J. J. Qin, S. K. Yan, H. Z. Jin and W. D. Zhang, *Chin. J. Nat. Med.*, 2011, **9**, 115–119.
- 12 G. Julien, R. Tristan, C. Stephanie, P. P. Antonio, D. Marie-Cecile, C. C. Marie-France and M. Jean-Michel, *J. Agric. Food Chem.*, 2017, **65**, 8884–8891.
- 13 H. Fujinori, T. Satoshi and M. Junya, *Phytochemistry*, 1991, **30**, 2197–2198.
- 14 C. F. Massaro, M. Katouli, T. Grkovic, H. Vu, R. J. Quinn, T. A. Heard, C. Carvalho, M. Manley-Harris, H. M. Wallace and P. Brooks, *Fitoterapia*, 2014, **95**, 247–257.
- 15 W. S. Feng, X. W. Cao, X. K. Zheng and H. X. Kuang, *Acta Pharm. Sin.*, 2005, **40**, 443–446.
- 16 P. J. M. Abreu and Y. H. Liu, *Fitoterapia*, 2007, **78**, 388–389.
- 17 T. I. B. Lopes, R. G. Coelho, N. C. Yoshida and N. K. Honda, *Chem. Pharm. Bull.*, 2008, **56**, 1551–1554.
- 18 M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven Jr, J. A. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, *Gaussian 09, Rev. C 01*, Gaussian, Inc., Wallingford CT, 2009.
- 19 J. Li, F. Zhao, M. Z. Li, L. X. Chen and F. Qiu, *J. Nat. Prod.*, 2010, **73**, 1667–1671.
- 20 V. M. Dirsch, H. Stuppner and A. M. Vollmar, *Planta Med.*, 1998, **64**, 423–426.
- 21 W. Ebrahim, A. Aly, V. Wray, A. Mádi, M. H. Teiten, F. Gaascht, B. Orlikova, M. U. Kassack, W. H. Lin, M. Diederich, T. Kurtán, A. Debbab and P. Proksch, *J. Med. Chem.*, 2013, **56**, 2991–2999.

