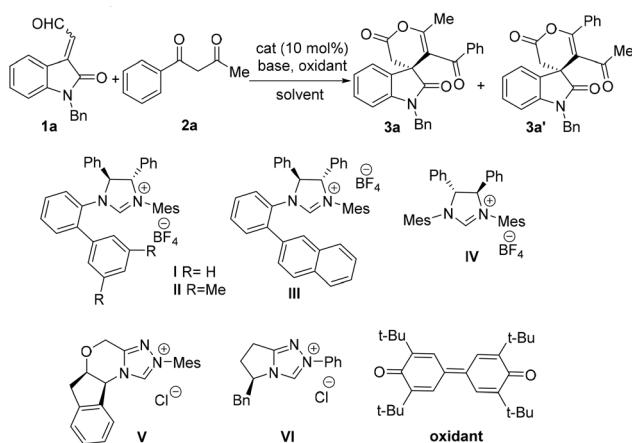


with 1,3-dicarbonyl compound **2a** under the catalysis of C_1 -symmetric diaryl-saturated imidazolium catalysts **I–III**.¹⁸ To our delight, the desired oxidative annulation product **3a** and its regioisomer **3a'** (6 : 1 rr) were obtained in 23% yield and with high enantioselectivity (89% ee) when catalyst **I**/DBU combination were employed under the oxidation of 3,3',5,5'-tetra-*tert*-butyldiphenoquinone in THF (Table 1, entry 1), while catalyst **II** and **III** gave a relatively higher stereoselectivity and regioselectivity (Table 1, entries 2 and 3). As a comparison, the catalytic activity of C_2 -symmetric diaryl-saturated catalyst **IV**, triazolium catalysts **V** and **VI** was further examined, and inferior results with respect to reactivity and selectivity were observed (Table 1, entries 4–6). Then different bases including inorganic and organic bases were varied, and we found that DBU give better

Table 1 Optimization of reaction conditions



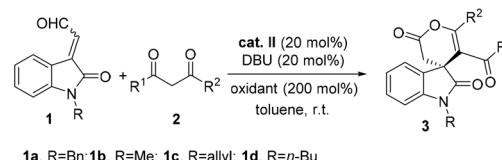
Entry ^a	Cat	Base	Solvent	Time	Yield ^b [%]	Rr ^c	ee ^d [%]
1	I	DBU	THF	1 h	23	6 : 1	89
2	II	DBU	THF	1 h	67	10 : 1	92
3	III	DBU	THF	1 h	65	7 : 1	92
4	IV	DBU	THF	4 h	64	8 : 1	39
5	V	DBU	THF	10 min	53	2 : 1	59
6	VI	DBU	THF	1 h	27	3 : 1	6
7	II	<i>t</i> -BuOK	THF	1 h	62	6 : 1	72
8	II	CsCO ₃	THF	1 h	62	9 : 1	79
9	II	NaOAc	THF	3 h	55	5 : 1	50
10	II	DABCO	THF	10 h	52	5 : 1	89
11	II	DIPEA	THF	24 h	57	5 : 1	84
12	II	NET ₃	THF	24 h	40	6 : 1	77
13	II	DBU	Toluene	1 h	72	12 : 1	91
14	II	DBU	CH ₃ CN	1 h	14	nd	nd
15	II	DBU	Et ₂ O	7 h	40	7 : 1	79
16	II	DBU	Dioxane	24 h	25	4 : 1	93
17	II	DBU	DCM	1 h	49	11 : 1	65
18 ^e	II	DBU	Toluene	1 h	33	9 : 1	91
19 ^f	II	DBU	Toluene	1.5 h	77	12 : 1	94

^a Unless otherwise mentioned, all reactions were performed using **1a** (0.1 mmol) and **2a** (0.2 mmol) in solvent (1.0 mL) in the presence of catalyst (20 mol%), base (20 mol%), and oxidant (0.2 mmol) at room temperature. ^b Combined yields of **3a** and **3a'**. ^c Rr refers to ratio of **3a** and **3a'**, which is determined by ¹H NMR analysis. ^d Determined by HPLC analysis on a chiral column. ^e 10 mol% catalyst was used. ^f 0.1 mmol **1a** and 0.1 mmol **2a** were used.

stereoselectivity and regioselectivity within a shorter time (Table 1, entries 7–12). Solvent screening revealed that toluene gave rise to a better yield (Table 1, entries 13–17). When the catalyst loading was reduced to 10%, the reaction yield was obviously decreased (Table 1, entry 18). For further investigation, we modulated the ratio of **1** and **2** to 1 : 1, and further improvement in yield was observed (Table 1, entry 19).

With the optimized reaction conditions in hand, the generality of the reaction was explored (Table 2, entries 1–20). A broad range of differently substituted 1,3-dione bearing electron-donating or electron-withdrawing substituents on *meta*- or *para*-position of aromatic rings were well tolerated, whereas the 1,3-dione with substituents on *ortho*-position of aromatic rings gave complex products as a result of steric hindrance. Although no pronounced electronic effect was observed, we discovered that substrates with the nitro group (**2e**, **2n**) gave lower yields and enantioselectivities (Table 2, entries 5 and 14), while substrates with the methyl group (**2g**, **2j**) gave lower regioselectivities but higher yields (Table 2, entries 7, 8 and 10). Heteroaryl-containing substrate (**2o**) also worked well to give the δ -lactone **3o** with excellent regioselectivity in moderate yield (52%) and high ee (83%) (Table 2, entry 15). Aliphatic 1,3-dione **2p** could participate in the oxidative reaction efficiently, giving

Table 2 Scope of the reaction



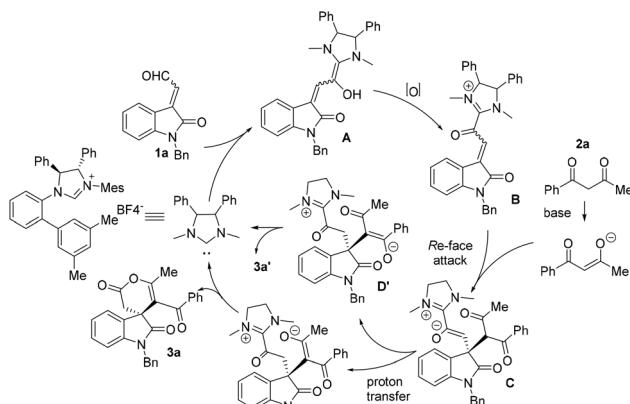
1a, R=Bn; **1b**, R=Me; **1c**, R=allyl; **1d**, R=n-Bu

Entry ^a	R ¹ /R ² (2)	3	Yield ^b [%]	Rr ^c	Ee ^d [%]
1	Ph/Me (2a)	3a	77	12 : 1	94
2	4-BrC ₆ H ₄ /Me (2b)	3b	75	14 : 1	92
3	4-FC ₆ H ₄ /Me (2c)	3c	78	14 : 1	92
4	4-ClC ₆ H ₄ /Me (2d)	3d	69	17 : 1	93
5	4-NO ₂ C ₆ H ₄ /Me (2e)	3e	50	18 : 1	84
6	4-CF ₃ C ₆ H ₄ /Me (2f)	3f	65	14 : 1	90
7	4-MeC ₆ H ₄ /Me (2g)	3g	82	7 : 1	95
8	4- <i>t</i> -BuC ₆ H ₄ /Me (2h)	3h	74	5 : 1	94
9	3-MeOC ₆ H ₄ /Me (2i)	3i	79	12 : 1	94
10	3-MeC ₆ H ₄ /Me (2j)	3j	84	6 : 1	94
11	3-BrC ₆ H ₄ /Me (2k)	3k	69	16 : 1	92
12	3-ClC ₆ H ₄ /Me (2l)	3l	72	16 : 1	92
13	3-FC ₆ H ₄ /Me (2m)	3m	76	17 : 1	93
14	3-NO ₂ C ₆ H ₄ /Me (2n)	3n	45	19 : 1	83
15	2-Thienyl/Me (2o)	3o	52	>25 : 1	83
16	Me/Me (2p)	3p	58	—	90
17 ^e	Ph/Me (2a)	3q	62	>20 : 1	94
18 ^f	Ph/Me (2a)	3r	74	12 : 1	94
19 ^g	Ph/Me (2a)	3s	52	8 : 1	94
20	Ph/Ph (2q)	3t	60	—	96

^a all reactions were performed using **1a** (0.1 mmol) and **2** (0.1 mmol) in toluene (1.0 mL) in the presence of catalyst **B** (20 mol%), DBU (20 mol%), and oxidant (0.2 mmol) at room temperature.

^b Combined yields of **3** and **3'**. ^c Determined by ¹H NMR analysis.

^d Determined of the major isomer by HPLC analysis on a chiral column. ^e **1b** was used. ^f **1c** was used. ^g **1d** was used.



Scheme 1 Proposed catalytic cycle.

rise to the product in 58% yield and 90% ee within 10 minutes (Table 2, entry 16), while biaryl-substituted substrate **2q** give the corresponding product **3t** in 60% yield and 96% ee over 24 h (Table 2, entry 20). Further investigation revealed that this method was compatible with different *N*-substituted enals, affording the desired δ -lactone **3q-3s** in moderate yield with good regioselectivities and excellent enantioselectivities (Table 2, entries 17–19).

The absolute configuration of the product **3a** was assigned to be *S* by X-ray crystallographic analysis (Cu target, see ESI†).¹⁹ Based on our experimental results and previous literature reports, a proposed catalytic cycle is shown in Scheme 1. The free NHC catalyst, derived from precatalyst by *in situ* deprotonation, could react with isatin-derived unsaturated aldehyde **1a**, giving rise to Breslow intermediate **A**, which could be converted to acyl azolium intermediate **B** through two-electron oxidation and deprotonation. Subsequent Re-face Michael attack of **2a**-derived enolate to species **B** along with the proton transfer sequentially produced intermediates **C** and **D**. Intramolecular cyclization of **D** gave the product **3a** and released the NHC catalyst to next catalytic cycle. Mechanistically, the formation of byproduct **3a'** arose from intramolecular annulation of intermediate **D'**, which was generated from Michael adduct **C** via undesired competing proton transfer step.²⁰

Conclusions

In conclusion, we have developed a C_1 -symmetric NHC catalyzed oxidative spiroannulation of isatin-derived enals. The *in situ* generated isatin-derived α,β -unsaturated acyl azolium species was efficiently trapped by 1,3-dicarbonyl compounds via a Michael addition/cyclization process. A series of synthetically intriguing spirooxindole δ -lactones was readily obtained in moderate to high yields with excellent enantioselectivities under mild conditions. Further application of the C_1 -symmetric NHCs to novel and challenging transformations toward synthesis of synthetically or pharmaceutically valuable molecules are currently underway.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We are grateful to the NSFC (21572087, 21632003), the Key program of Gansu province (17ZD2GC011) and the “111” program from the MOE of P. R. China for financial support.

Notes and references

- For selected reviews, see: (a) C. Marti and E. M. Carreira, *Eur. J. Org. Chem.*, 2003, 2209; (b) C. V. Galliford and K. A. Scheidt, *Angew. Chem., Int. Ed.*, 2007, **46**, 8748; (c) R. Dalpozzo, G. Bartoli and G. Bencivenni, *Chem. Soc. Rev.*, 2012, **41**, 7247; (d) N. R. Ball-Jones, J. J. Badillo and A. K. Franz, *Org. Biomol. Chem.*, 2012, **10**, 5165.
- For selected examples, see: (a) X. H. Chen, Q. Wei, S. W. Luo, H. Xiao and L. Z. Gong, *J. Am. Chem. Soc.*, 2009, **131**, 13819; (b) A. P. Antonchick, C. Gerding-Reimers, M. Catarinella, M. Schürmann, H. Preut, S. Ziegler, D. Rauh and H. Waldmann, *Nat. Chem.*, 2010, **2**, 735; (c) B. Tan, N. R. Candeias and C. F. Barbas III, *Nat. Chem.*, 2011, **3**, 473; (d) Y. Liu, M. Nappi, E. Arceo, S. Vera and P. Melchiorre, *J. Am. Chem. Soc.*, 2011, **133**, 15212; (e) G. Zhan, M. L. Shi, Q. He, W. J. Lin, Q. Ouyang, W. Du and Y. C. Chen, *Angew. Chem., Int. Ed.*, 2016, **55**, 2147; (f) X. Han, W. L. Chan, W. Yao, Y. Wang and Y. Lu, *Angew. Chem., Int. Ed.*, 2016, **55**, 6492.
- (a) Y. G. Chen, J. C. Wu, G. Y. Chen, C. R. Han and X. P. Song, *Chem. Biodiversity*, 2011, **8**, 1958; (b) S. S. Ma, W. L. Mei, Z. K. Guo, S. B. Liu, Y. X. Zhao, D. L. Yang, Y. B. Zeng, B. Jiang and H. F. Dai, *Org. Lett.*, 2013, **15**, 1492; (c) J. Z. Huang, C. L. Zhang, Y. F. Zhu, L. L. Li, D. F. Chen, Z. Y. Han and L. Z. Gong, *Chem.-Eur. J.*, 2015, **21**, 8389.
- (a) D. Du, Z. Y. Hu, J. L. Jin, Y. Y. Lu, W. F. Tang, B. Wang and T. Lu, *Org. Lett.*, 2012, **14**, 1274; (b) S. Zhao, J. B. Lin, Y. Y. Zhao, Y. M. Liang and P. F. Xu, *Org. Lett.*, 2014, **16**, 1802.
- For selected reviews, see: (a) D. Enders, O. Niemeier and A. Henseler, *Chem. Rev.*, 2007, **107**, 5606; (b) V. Nair, S. Vellalath and B. P. Babu, *Chem. Soc. Rev.*, 2008, **37**, 2691; (c) A. Grossmann and D. Enders, *Angew. Chem., Int. Ed.*, 2012, **51**, 314; (d) J. Izquierdo, G. E. Hutson, D. T. Cohen and K. A. Scheidt, *Angew. Chem., Int. Ed.*, 2012, **51**, 11686; (e) X. Bugaut and F. Glorius, *Chem. Soc. Rev.*, 2012, **41**, 3511; (f) S. J. Ryan, L. Candish and D. W. Lupton, *Chem. Soc. Rev.*, 2013, **42**, 4906; (g) M. N. Hopkinson, C. Richter, M. Schedler and F. Glorius, *Nature*, 2014, **510**, 485; (h) J. Mahatthananchai and J. W. Bode, *Acc. Chem. Res.*, 2014, **47**, 696; (i) D. M. Flanigan, F. Romanov-Michailidis, N. A. White and T. Rovis, *Chem. Rev.*, 2015, **115**, 9307; (j) Y. Wang, D. Wei and W. Zhang, *ChemCatChem*, 2018, **10**, 338; (k) X. Y. Chen, Q. Liu, P. Chauhan and D. Enders, *Angew. Chem., Int. Ed.*, 2018, **57**, 3862.
- (a) T. Ukai, S. Tanaka and S. Dokawa, *J. Pharm. Soc. Jpn.*, 1943, **63**, 296; (b) D. Enders, K. Breuer, J. Rumsink and



J. H. Teles, *Helv. Chim. Acta*, 1996, **79**, 1899; (c) D. Enders and U. Kallfass, *Angew. Chem., Int. Ed.*, 2002, **41**, 1743; (d) M. S. Kerr, J. R. Alaniz and T. Rovis, *J. Am. Chem. Soc.*, 2002, **124**, 10298; (e) Y. Hachisu, J. W. Bode and K. Suzuki, *J. Am. Chem. Soc.*, 2003, **125**, 8432; (f) H. Y. Zhao, F. W. Foss and R. Breslow, *J. Am. Chem. Soc.*, 2008, **130**, 12590; (g) Q. Liu, S. Perreault and T. Rovis, *J. Am. Chem. Soc.*, 2008, **130**, 14066; (h) S. P. Lathrop and T. Rovis, *J. Am. Chem. Soc.*, 2009, **131**, 13628; (i) A. T. Biju, N. E. Wurz and F. Glorius, *J. Am. Chem. Soc.*, 2010, **132**, 5970; (j) T. Jousseaume, N. E. Wurz and F. Glorius, *Angew. Chem., Int. Ed.*, 2011, **50**, 1410; (k) X. Fang, X. Chen, H. Lv and Y. R. Chi, *Angew. Chem., Int. Ed.*, 2011, **50**, 11782; (l) C. A. Rose, S. Gundala, C. L. Fagan, J. F. Franz, S. J. Connolly and K. Zeitler, *Chem. Sci.*, 2012, **3**, 735; (m) M. Schedler, D. S. Wang and F. Glorius, *Angew. Chem., Int. Ed.*, 2013, **52**, 2585.

7 (a) M. He, J. R. Struble and J. W. Bode, *J. Am. Chem. Soc.*, 2006, **128**, 8418; (b) M. He, G. J. Ue and J. W. Bode, *J. Am. Chem. Soc.*, 2006, **128**, 15088; (c) J. Kaeobamrung, M. C. Kozlowski and J. W. Bode, *Proc. Natl. Acad. Sci. U. S. A.*, 2010, **107**, 20661; (d) L. Hao, Y. Du, H. Lv, X. Chen, H. Jiang, Y. Shao and Y. R. Chi, *Org. Lett.*, 2012, **14**, 2154; (e) L. Wang, Q. Ni, M. Blgmel, T. Shu, G. Raabe and D. Enders, *Chem.-Eur. J.*, 2015, **21**, 8033.

8 (a) C. Burstein and F. Glorius, *Angew. Chem., Int. Ed.*, 2004, **43**, 6205; (b) S. S. Sohn, E. L. Rosen and J. W. Bode, *J. Am. Chem. Soc.*, 2004, **126**, 14370; (c) C. Guo, M. Schedler, C. G. Daniliuc and F. Glorius, *Angew. Chem., Int. Ed.*, 2014, **53**, 10232; (d) C. Guo, B. Sahoo, C. G. Daniliuc and F. Glorius, *J. Am. Chem. Soc.*, 2014, **136**, 17402; (e) C. Guo, M. Fleige, D. Janssen-Müller, C. G. Daniliuc and F. Glorius, *Nat. Chem.*, 2015, **7**, 842; (f) L. Wang, S. Li, M. Blümel, R. Puttreddy, A. Peuronen, K. Rissanen and D. Ender, *Angew. Chem., Int. Ed.*, 2017, **56**, 8516.

9 (a) S. J. Ryan, L. Candish and D. W. Lupton, *J. Am. Chem. Soc.*, 2009, **131**, 14176; (b) F. G. Sun, L. H. Sun and S. Ye, *Adv. Synth. Catal.*, 2011, **353**, 3134; (c) J. Cheng, Z. Huang and Y. R. Chi, *Angew. Chem., Int. Ed.*, 2013, **52**, 8592.

10 For reviews on oxidative NHC catalysis, see: (a) C. E. I. Knapke, A. Imami and A. J. Wangelin, *ChemCatChem*, 2012, **4**, 937; (b) S. D. Sarkar, A. Biswas, R. C. Samanta and A. Studer, *Chem.-Eur. J.*, 2013, **19**, 4664.

11 (a) J. Guin, S. De Sarkar, S. Grimme and A. Studer, *Angew. Chem., Int. Ed.*, 2008, **47**, 8727; (b) B. E. Maki, A. Chan, E. M. Phillips and K. A. Scheidt, *Org. Lett.*, 2007, **9**, 371; (c) S. D. Sarkar, S. Grimme and A. Studer, *J. Am. Chem. Soc.*, 2010, **132**, 1190; (d) S. D. Sarkar and A. Studer, *Org. Lett.*, 2010, **12**, 1992; (e) X. Zhao, K. E. Ruhl and T. Rovis, *Angew. Chem., Int. Ed.*, 2012, **51**, 12330; (f) J. Mo, X. Chen and Y. R. Chi, *J. Am. Chem. Soc.*, 2012, **134**, 8810; (g) E. G. Delany, C. L. Fagan, S. Gundala, A. Mari, T. Broja, K. Zeitler and S. J. Connolly, *Chem. Commun.*, 2013, **49**, 6510; (h) J. Mo, L. Shen and Y. R. Chi, *Angew. Chem., Int. Ed.*, 2013, **52**, 8588; (i) H. Inoue and K. Higashiura, *J. Chem. Soc., Chem. Commun.*, 1980, 549.

12 For selected examples, see: (a) S. D. Sarkar and A. Studer, *Angew. Chem., Int. Ed.*, 2010, **49**, 9266; (b) A. Biswas, S. D. Sarkar, R. Fröhlich and A. Studer, *Org. Lett.*, 2011, **13**, 4966; (c) Z. Q. Rong, M. Q. Jia and S. L. You, *Org. Lett.*, 2011, **13**, 4080; (d) Z. Q. Zhu, X. L. Zheng, N. F. Jiang, X. L. Wan and J. C. Xiao, *Chem. Commun.*, 2011, **47**, 8670; (e) A. G. Kravina, J. Mahatthananchai and J. W. Bode, *Angew. Chem., Int. Ed.*, 2012, **51**, 9433; (f) S. Bera, R. C. Samanta, C. G. Daniliuc and A. Studer, *Angew. Chem., Int. Ed.*, 2014, **53**, 9622; (g) X. X. Wu, B. Liu, Y. X. Zhang, M. Jeret, H. L. Wang, P. C. Zheng, S. Yang, B. A. Song and Y. R. Chi, *Angew. Chem., Int. Ed.*, 2016, **55**, 12280.

13 For one report concerning trapping isatin-derived unsaturated acyl azolium by performed enamine, see: (a) D. Xie, L. Yang, Y. Lin, Z. Zhang, D. Chen, X. Zeng and G. Zhong, *Org. Lett.*, 2015, **17**, 2318, For a racemic example, see: (b) L. L. Zhao, X. S. Li, L. L. Cao, R. Zhang, X. Q. Shi and J. Qi, *Chem. Commun.*, 2017, **53**, 5985.

14 H. Lu, J. Y. Liu, C. G. Li, J. B. Lin, Y. M. Liang and P. F. Xu, *Chem. Commun.*, 2015, **51**, 4473.

15 (a) Y. Wang, H. Lu and P. F. Xu, *Acc. Chem. Res.*, 2015, **48**, 1832; (b) T. P. Gao, J. B. Lin, X. Q. Hu and P. F. Xu, *Chem. Commun.*, 2014, **50**, 8934; (c) L. Tian, X. Q. Hu, Y. H. Li and P. F. Xu, *Chem. Commun.*, 2013, **49**, 7213; (d) Y. Y. Zhao, S. Zhao, J. K. Xie, X. Q. Hu and P. F. Xu, *J. Org. Chem.*, 2016, **81**, 10532.

16 (a) R. Liu and J. Zhang, *Org. Lett.*, 2013, **15**, 2266; (b) R. Liu and J. Zhang, *Chem.-Eur. J.*, 2013, **19**, 7319; (c) T. Mukaiyama, K. Ogata, I. Sato and Y. Hayashi, *Chem.-Eur. J.*, 2014, **20**, 13583.

17 For a racemic synthesis of this scaffold using isatin-derived 2-bromoenals, see: (a) J. Xu, W. Zhang, Y. Liu, S. Zhu, M. Liu, X. Hua, S. Chen, T. Lu and D. Du, *RSC Adv.*, 2016, **6**, 18601, For a catalytic asymmetric synthesis using isatin-derived unsaturated acids with moderate ees, see: (b) W. Zhang, J. Xu, J. Cao, C. Fang, J. Zhu, T. Lu and D. Du, *Tetrahedron*, 2017, **73**, 3249.

18 (a) K. P. Jang, G. E. Hutson, R. C. Johnston, E. O. McCusker, P. H. Y. Cheong and K. A. Scheidt, *J. Am. Chem. Soc.*, 2014, **136**, 76; (b) A. Lee and K. A. Scheidt, *Angew. Chem., Int. Ed.*, 2014, **53**, 7594.

19 CCDC 1010784 (3a).

20 With regard to the mechanism, the hemiacetal formation-Coates-Claisen rearrangement pathway proposed by Bode *et al.* cannot be ruled out at current stage, see: (a) J. Mahatthananchai, P. Zheng and Jeffrey W. Bode, *Angew. Chem., Int. Ed.*, 2011, **50**, 1673; (b) J. Mahatthananchai, J. Kaeobamrung and J. W. Bode, *ACS Catal.*, 2012, **2**, 494; (c) E. Lyngvi, J. W. Bode and F. Schoenebeck, *Chem. Sci.*, 2012, **3**, 2346.

