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Totally room-temperature solution-processing method for fabricating flexible perovskite solar cells using an $\text{Nb}_2\text{O}_5\text{--TiO}_2$ electron transport layer[†]

Jun Jiang, Shubo Wang, Xuguang Jia, Xiang Fang, Shuai Zhang, Jing Zhang, Wei Liu, Jianning Ding* and Ningyi Yuan *

Flexible perovskite solar cells are new technology-based products developed by the global solar industry and are promising candidates for realizing a flexible and lightweight energy supply system for wearable and portable electronic devices. A critical issue for flexible perovskite solar cells is to achieve high power conversion efficiency (PCE) while using low-temperature solution-based technology for the fabrication of a compact charge collection layer. Herein, we innovatively introduce niobium ethoxide as a precursor additive to TiO_2 NCs, which allows realization of an $\text{Nb}_2\text{O}_5\text{--TiO}_2$ electron transport layer (ETL). The presence of Nb_2O_5 remarkably enhances electron mobility and electrical conductivity of the ETLs. In addition, uniform perovskite films are prepared by an annealing-free solution-based method. The excellent performance of the cell is attributed to its smooth film surface and high electron mobility, and performance is verified by the effective suppressions of charge recombination and time-resolved photoluminescence. PCEs of 15.25% and 13.60% were obtained for rigid substrates (glass/fluorine-doped tin oxide) and an indium tin oxide/PET (poly(ethylene terephthalate)) flexible substrate by using a totally room-temperature solution-processing method, respectively.

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1. Introduction

Perovskite solar cells have developed very rapidly over the last several years and could be in considerable demand for various applications in the future photovoltaic market.^{1–6} Numerous scientific research professionals have accelerated increases in power conversion efficiency (PCE) of perovskite solar cells, and at present, the certified world record for a PCE is 22.1%.⁷ As a light absorption layer, APbX_3 ($\text{A} = \text{CH}_3\text{NH}_3$, $(\text{NH}_2)_2\text{CH}_2$ or Cs , and $\text{X} = \text{I}$, Br , or Cl) is considered as the most promising replacement for silicon solar cells; APbX_3 has outstanding properties such as strong light absorption, weakly bound excitons, long-range charge-carrier diffusion, and apparent tolerance to defects.^{8–10}

A state-of-the-art photovoltaic solar cell (PSC) is prepared in a conventional n-i-p device configuration consisting of an n-type oxide semiconductor capped with a perovskite absorber and a hole-transport layer (HTL). The electron transfer layer (ETL) of

PSCs plays important roles in extracting electrons and blocking holes from perovskite. Metal oxide materials such as TiO_2 , ZnO , and SnO_2 are widely used in most studies aimed at achieving high efficiency because of their environmentally friendly nature, wide band gap, high electron mobility, and good stability.^{11–14} A large number of PSCs based on TiO_2 require a high-temperature sintering process for their crystallization or removal of the dispersion medium.^{8–10,15} However, TiO_2 , ZnO , and SnO_2 ETLs obtained using high-temperature treatment cannot be applied to flexible perovskite solar cells. Recently, methods have been developed for fabricating low-temperature processable TiO_2 ETLs for planar PSCs, such as atomic layer deposition¹⁶ and magnetron sputtering.¹⁷ Compared to vacuum methods, solution-based techniques are generally more cost-effective and scalable and can be used to achieve roll-to-roll process-ability. Therefore, it is imperative to develop effective methods that can restriction migration, improve electron mobility of an ETL, and reduce trap-state density within the perovskite material, thereby eliminating hysteresis and improving the efficiency of PSCs. However, perovskite materials such as $\text{CH}_3\text{NH}_3\text{PbI}_3$ and $\text{CH}(\text{NH}_2)_2\text{PbI}_3$ are mostly used for PSCs, and these materials practically need annealing for 10–60 min to form the black crystalline photoactive layer.^{18–23} Annealing process is not convenient for mass production because it requires additional equipment and increases energy consumption. One method to address this problem is to employ annealing-free processing,¹⁵

School of Materials Science and Engineering, Jiangsu Collaborative Innovation Center of Photovoltaic Science and Engineering, Jiangsu Province Cultivation Base for State Key Laboratory of Photovoltaic Science and Technology, Changzhou University, Changzhou 213164, Jiangsu, China. E-mail: dingjn@cczu.edu.cn; nyuan@cczu.edu.cn

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which can save energy and facilitate industrial production. Herein, we propose a facile strategy to prepare a highly efficient perovskite solar cell *via* room-temperature solution processing. TiO_2 films are fabricated by spin coating a colloidal solution of anatase TiO_2 nanocrystallines (NCs) prepared *via* a low-temperature sol-gel method on a substrate. Considering improved performance of the film, we employed niobium ethoxide as a precursor additive. Nb_2O_5 is considered a better ETL for a perovskite solar cell due to its higher carrier mobility and conduction band edge position. Here, niobium ethoxide is introduced into the TiO_2 dispersion as an aid dispersant. Niobium ethoxide facilitates spontaneous coalescence of the TiO_2 NCs, thereby forming a stable dispersion. The stable dispersion is spin-coated onto the conductive base film without annealing; only ultraviolet (UV) treatment for 15 min yields a uniform and dense $\text{Nb}_2\text{O}_5\text{-TiO}_2$ layer. It is expected that the Nb_2O_5 formed *in situ* would passivate the grain boundary of the TiO_2 NCs, and further form a dense and uniform film. The exceptional performance of the layer is attributed to the excellent optical and electronic properties of the $\text{Nb}_2\text{O}_5\text{-TiO}_2$ material, such as a smooth surface and high electron mobility; these properties make the material a better growth platform for a high-quality perovskite absorber layer. In addition, MAPbI_3 films are deposited as a light absorption layer *via* one-step spin-coating by a simple annealing-free process.

2. Experimental section

2.1. Materials

4-*tert*-Butylpyridine (*t*-BP), Li-bis-(trifluoromethanesulfonyl) imide (Li-TFSI), PbI_2 (99.9985%), niobium ethoxide (99.999%), dimethylacetamide (DMAc) and *N*-methyl-2-pyrrolidone (NMP) were ordered from Alfa Aesar; the methylammonium iodide (MAI) (99.5%), and 2,2',7,7'-tetrakis[*N,N*-di(4-methoxyphenyl) amino]-9,9'-spiro-bifluorene (spiro-OMeTAD) (99.8%) were purchased from Xi'an p-OLED Technology Corp. Titanium tetrachloride (TiCl_4 , anhydrous, 99.5%) and chlorobenzene (anhydrous, 99.9%) were purchased from Aldrich (USA). All solvents were used without any further purification.

2.2. Synthesis of TiO_2 NCs and preparation of TiO_2 dispersion with niobium ethoxide

TiO_2 NCs were synthesized by a non-hydrolytic sol-gel reaction according to a modified procedure.²⁴ The resulting precipitates were washed by adding excess ethanol and diethyl ether and purified by centrifugation at 3000 rpm for 5 min. This washing procedure was repeated thrice. To obtain the TiO_2 colloidal solution ($\sim 5 \text{ mg mL}^{-1}$), the washed TiO_2 NCs were dispersed into anhydrous methanol, and ultrasonic treatment was carried out for several hours. To obtain a niobium ethoxide/ TiO_2 mixed precursor suspension for spin coating, the purified TiO_2 NCs (5 mg mL^{-1}) were re-dispersed in ethanol at the desired niobium ethoxide concentration. $\text{Nb}_2\text{O}_5\text{-TiO}_2$ ETLs were fabricated by spin-coating at 3000 rpm under ambient conditions, and the films were free from annealing. The samples were treated again with UV-ozone for 15 min before perovskite deposition.

2.3. Device fabrication

Pre-patterned transparent conducting oxide substrates were sequentially cleaned using ethanol, acetone, isopropanol, and ethanol separately in an ultrasonic bath for 20 min each and then dried under flowing nitrogen. Fluorine-doped tin oxide (FTO) substrates underwent UV-ozone treatment (Model UV-03 UVO₃ cleaner) for 15 min before they were used for spin-coating ETLs.

A highly dispersed solution of TiO_2 NCs and TiO_2 NCs with niobium ethoxide in ethanol were dropped onto substrates and immediately spin-coated at a speed of 3000 rpm for 30 s. The samples again underwent UV-ozone treatment for 15 min before perovskite deposition. Then, the TiO_2 ($\text{Nb}_2\text{O}_5\text{-TiO}_2$)-coated substrates were transferred immediately to a nitrogen-filled glovebox for the deposition of perovskite films.

A MAPbI_3 solution was prepared according to an annealing-free process reported by Fang *et al.*¹⁵ The MAPbI_3 precursor solution (1.2 M) was prepared using a mixed solvent of DMAc and NMP in a volume ratio of 5 : 1. Perovskite films were deposited onto the TiO_2 or $\text{Nb}_2\text{O}_5\text{-TiO}_2$ substrates according to a two-step spin-coating procedure. In the first step, spin coating was carried out at 1000 rpm for 20 s with an acceleration of 200 rpm s^{-1} . In the second step, spin coating was carried out at 5000 rpm for 45 s with an acceleration of 1000 rpm s^{-1} . During the second step, chlorobenzene was dropped onto the spinning substrate at 35 s before the end of the procedure. At the end of the second step, a dark perovskite film was directly formed. A spiro-OMeTAD HTL was prepared according to a process reported by Yang *et al.*²⁵ The HTL was fabricated as follows: a spiro-OMeTAD solution (90 mg mL^{-1}) was dissolved in chlorobenzene using 36 μL 4-*tert*-butylpyridine and 22 μL lithium bis(trifluoromethylsulfonyl)imide (520 mg mL^{-1}) as the dopants in acetonitrile. The spiro-OMeTAD solution was spin-coated onto the perovskite films at 3000 rpm for 30 s. Finally, an 80 nm thick gold coating was deposited using a thermal evaporator.

2.4. Characterization

Morphologies of the TiO_2 films were characterized by field-emission scanning electron microscopy (FESEM, Zeiss Supra 55). The ζ -potential of the TiO_2 NCs with different concentrations of niobium ethoxide was characterized by using a size analyzer (Zetasizer Nano ZS ZEN3600 instrument, Malvern Instruments) at room temperature with a 633 nm laser. The UV-vis transilluminator spectra of the samples were recorded on a spectrophotometer (UV5800). High-resolution transmission electron microscopy (HR-TEM) was carried out using an electron microscope (JEM-2100, JEOL Ltd., Japan). The root-mean-square (RMS) roughness and topography images of the films were obtained *via* atomic force microscopy (AFM, Veeco Dimension V). The quality and crystalline structure of the samples were confirmed by θ - 2θ X-ray diffraction (XRD) using an X-ray diffractometer (D/max 2500 PC) with a Cu $\text{K}\alpha$ radiation source. A photoluminescence (PL) system (DeltaFlex, Horiba Ltd.) was used to measure the time-resolved PL (TRPL) decay. Photovoltaic performance of the solar cells was measured using



a multisource meter (Model 2400, Keithley, Cleveland, OH, USA) under one sun (AM 1.5G, 100 mW cm⁻²) illumination, which was achieved by using a solar simulator (500 W Xe lamp) (XES-40S1, San-Ei Electric Co., Ltd., Japan) as the light source. The device area of 0.07 cm² was defined by a metal mask. All devices were scanned with a reverse and forward under standard test procedure at a scan rate of 0.2 V s⁻¹. X-Ray photoelectron spectroscopy (XPS) was performed on a photoelectron spectrometer (ESCALAB 250Xi, Thermo Fisher Scientific).

3. Results and discussion

3.1. Characterization of TiO₂ and Nb₂O₅-TiO₂ films

In the spin-coating process, a stably distributed precursor solution of TiO₂ NCs in solution is a prerequisite to form uniform and void-free TiO₂ thin films. The niobium ethoxide-capped TiO₂ NCs have a uniform dispersion in ethanol, are stable for months, and have better anti-settleability properties (Fig. S1†). The ζ -potential is an important factor reflecting colloid stability. The ζ -potential was measured, and the results are shown in Fig. 1. The figure indicates that the ζ -potential value for TiO₂ is 16.3 mV. The ζ -potential increases with the addition of niobium ethoxide. It demonstrated that the niobium ethoxide plays an important role in the dispersion of TiO₂ NCs. Powder XRD measurements of a TiO₂ sample revealed typical diffraction peaks of anatase TiO₂ (Fig. 2a). The diffraction peaks centered at 25.063, 37.682, 47.882, 54.119, 62.418, and 75.029 are assigned to (101), (004), (200), (105), (204), (116), and (215) diffractions, respectively. This result confirms that the as-synthesized TiO₂ NCs are assigned to the anatase crystal structure (PDF no. 21-1272). Scherrer peak width analysis²⁶ revealed that the average size of the nanocrystalline domains is approximately 8.8 nm. To gain further insights into the nanoscale morphology and nanocrystal structures, the morphology of as-synthesized TiO₂ NCs was investigated using the HR-TEM images. The HR-TEM images (Fig. S2a†) revealed that the TiO₂ NCs are around 5–10 nm in diameter, which is consistent with the value obtained by Scherrer peak width

analysis. In addition, the selected area electron diffraction pattern for the TiO₂ NCs (Fig. S2b†) confirms the high crystallinity of the TiO₂ NCs.

Fig. 2b shows the optical transmission spectra of TiO₂ and Nb₂O₅-TiO₂. Both the materials show excellent transmittance in the wavelength range of 400–800 nm. The resultant ETLs coated on the FTO glass substrates shows good optical transparency and higher transmittance than the bare FTO glass, with the transmittance being greater than 80% in the entire visible region. The high transparency of the Nb₂O₅-TiO₂ ETLs is very conducive to the light absorption of the perovskite layer and improves light harvesting. No apparent difference was found between the transmittances of different Nb₂O₅-TiO₂ films deposited on the FTO substrates.

The composition and bonding type of the Nb₂O₅-TiO₂ film were measured using XPS. A typical XPS spectra of Nb₂O₅-TiO₂ is shown in Fig. 3a. Clearly, the O, Ti, and Nb peaks are located at \sim 530.59, \sim 458.99, and \sim 207.51 eV, respectively. The high-resolution Ti 2p (Fig. 3b) spectrum reveals two different peaks located at 459.04 and 464.69 eV, which correspond to Ti 2p^{3/2} and Ti 2p^{1/2}, respectively; accordingly, a spin-orbit coupling of 5.65 eV is obtained, which is the signature of Ti⁴⁺. As shown in the Nb 3d core level spectra (Fig. 3c), Nb 3d^{5/2} and Nb 3d^{3/2} peaks are located at 207.54 and 210.20 eV, respectively, indicating the presence of five-valent niobium in the deposited films.²⁷ The main binding energy of 530.3 eV is attributed to O 1s (Fig. 3d) which indicates the O²⁻ state in TiO₂ and the peak at the higher binding energy of 531.4 eV is attributed to surface oxygen-group absorbance or hydroxyl groups.^{28–31}

Fig. 4a and b show the top-view SEM images of the TiO₂ and Nb₂O₅-TiO₂ films. The Nb₂O₅-TiO₂ film is uniform and dense and shows no apparent pinholes, indicating the high quality of the film. The surface morphology of the Nb₂O₅-TiO₂ film is not like to that of the pristine TiO₂ film. The results of simultaneous energy-dispersive spectroscopy (EDS) of the Nb₂O₅-TiO₂ film are shown in Fig. S3.† When excess niobium ethoxide (30% and 40%) is added, the surface morphology indicates the formation of a porous surface with a large number of pinholes, directly causing a deterioration in the quality of the thin film (Fig. S4†). Fig. 4c and d shows the AFM height images of the TiO₂ and Nb₂O₅-TiO₂ films. The RMS roughness decreased from 13.8 to 10.5 nm because of the introduction of Nb₂O₅, indicating that the Nb₂O₅-TiO₂ films have flat surfaces. A smooth surface is essential for growing high-quality perovskite films, reducing surface defect trap interfaces with ETLs and HTLs, and enhancing charge extraction at the interface between the ETL and the perovskite layer.^{32–34}

To investigate the effect of introducing Nb₂O₅ on the electrical properties, the electron mobility was studied by the space charge limited current (SCLC) method using an electron-only device structure. The sample structure used for this measurement was FTO/PCBM/TiO₂ (15% Nb₂O₅-TiO₂)/PCBM/Ag. The details are shown in the ESI.† Fig. S5† shows the current density–voltage (*J*–*V*) curves for the TiO₂ and Nb₂O₅-TiO₂ films fitted using the Mott–Gurney law.^{34,35} It's apparent that the electron mobility of the ETL film increases considerably because of the introduction of Nb₂O₅. The electron mobility

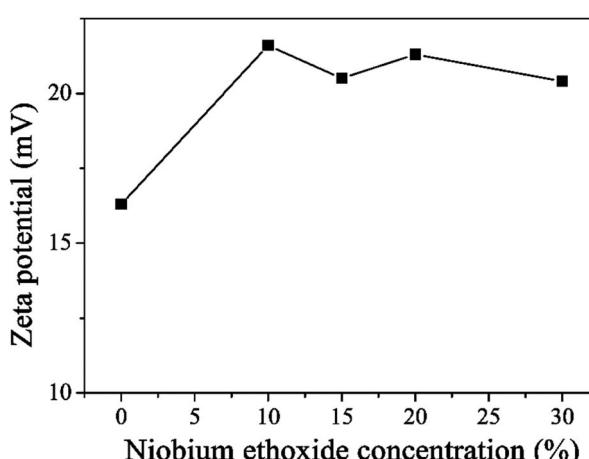


Fig. 1 The zeta potential of TiO₂ NCs with different concentration of niobium ethoxide.



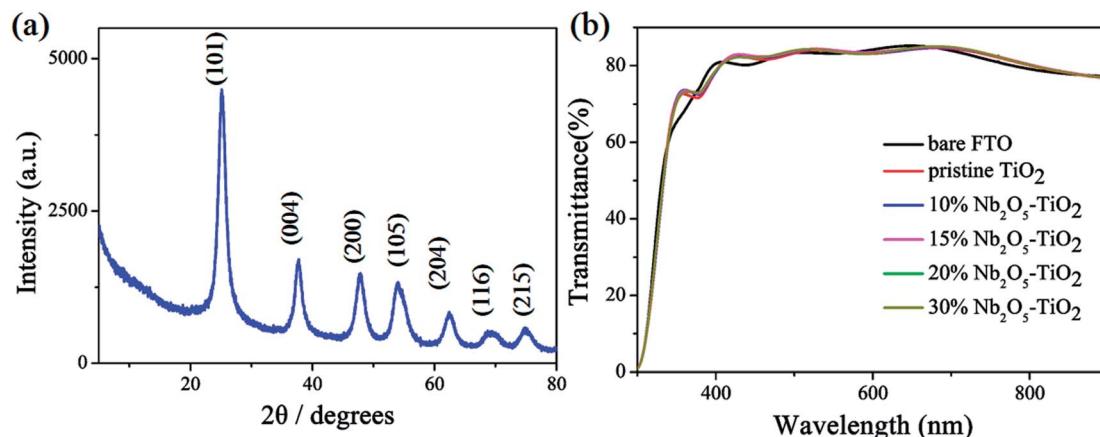


Fig. 2 (a) Powder X-ray diffraction pattern of TiO_2 NCs. (b) Transmittance spectra of bare FTO, pristine TiO_2 , and Nb_2O_5 - TiO_2 with different niobium ethoxide contents.

increases from 7.09×10^{-4} to $1.14 \times 10^{-3} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, which is very close to previous reports that TiO_2 modified craft with annealing process.³⁶

3.2. Photovoltaic performance

The above measurements show that the quality of the TiO_2 NCs film improves considerably because of the introduction of Nb_2O_5 ; *e.g.*, the film has a smoother surface and enhanced

electron mobility. Next, PSCs were designed and fabricated using the TiO_2 and Nb_2O_5 - TiO_2 ETLs. Fig. S6[†] shows the detailed device structure, in which FTO is employed as the anode, the TiO_2 or Nb_2O_5 - TiO_2 film as the ETL, MAPbI_3 as the absorber layer, spiro-OMeTAD as the HTL, and a gold layer as the cathode. A cross-sectional SEM image of the completed device architecture is shown in Fig. S7.[†] The smooth morphology of the Nb_2O_5 - TiO_2 films is beneficial for forming

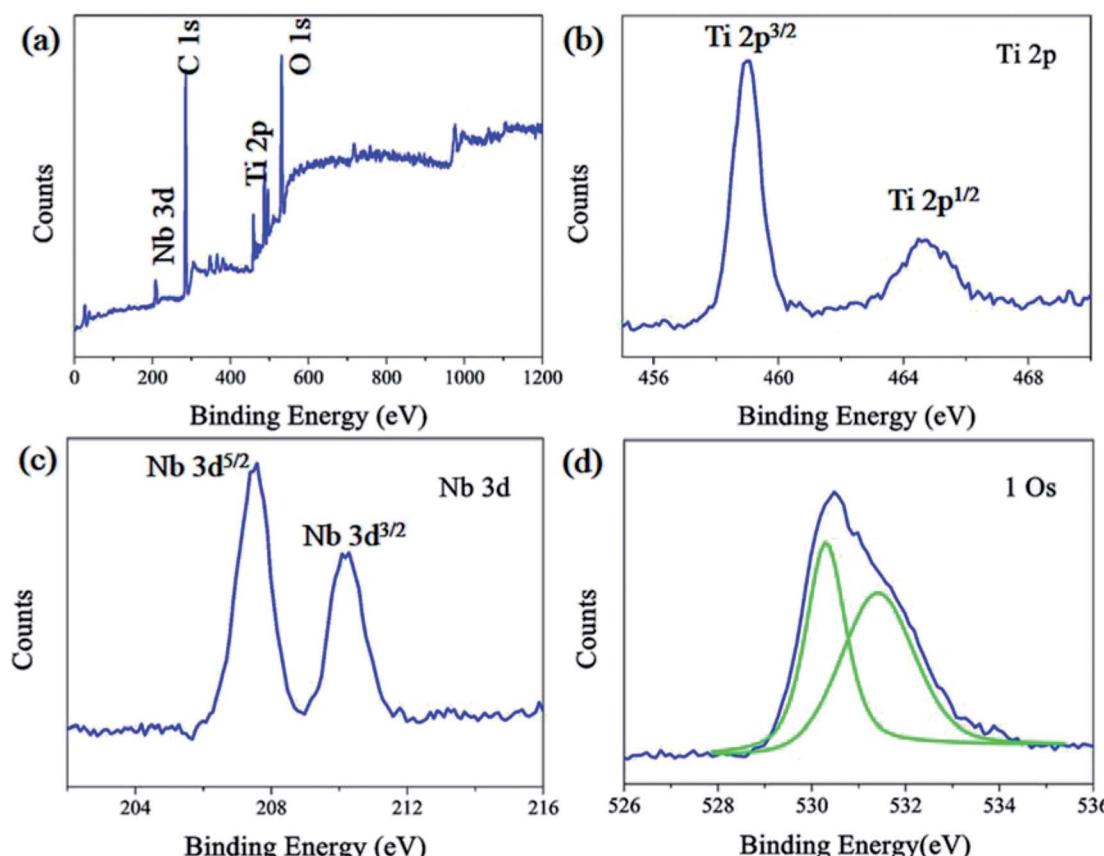


Fig. 3 (a) Typical XPS survey of Nb_2O_5 - TiO_2 film. High-resolution XPS (b) Ti 2p, (c) Nb 3d, and (d) O 1s peaks of the Nb_2O_5 - TiO_2 film deposited on a glass substrate.



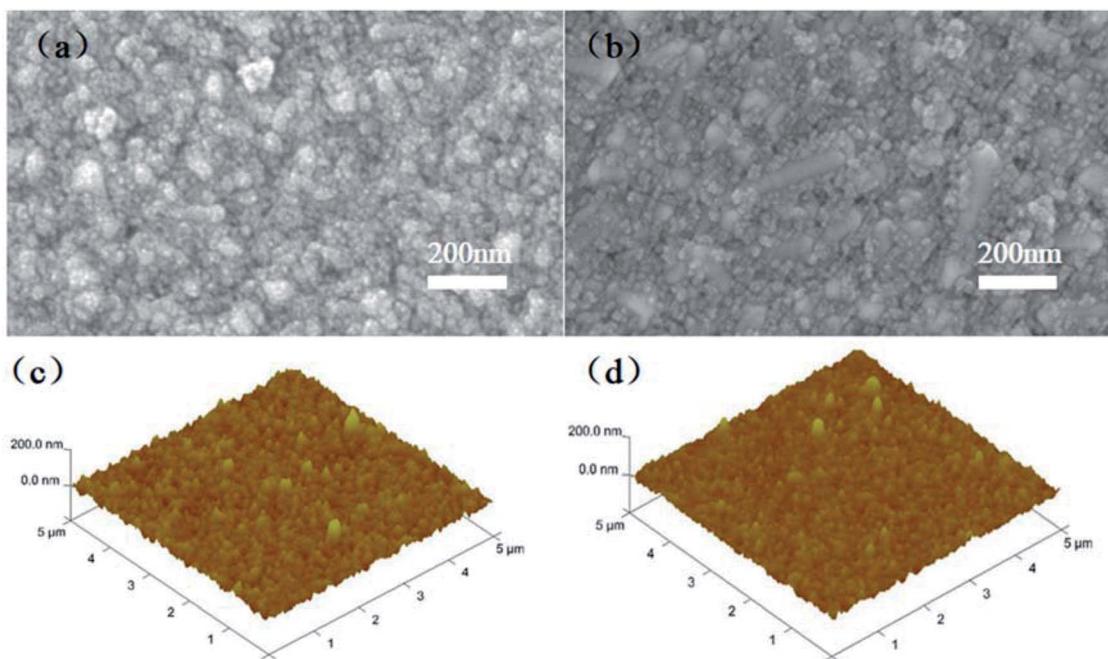


Fig. 4 Top-view SEM images of (a) TiO_2 and (b) Nb_2O_5 – TiO_2 films coated on FTO substrates. AFM height images of the (c) TiO_2 and (d) Nb_2O_5 – TiO_2 films.

highly crystalline and compact perovskite films. Fig. S8[†] shows the top-view SEM images of the perovskite film, which exhibits smooth surfaces, big crystalline size, and good coverage. Table S1[†] lists the device performance optimized as a function of the Nb_2O_5 content; the key J – V parameters are summarized in ESI.[†] It appears that the PCE increased from 13.47% to 15.25%. It is apparent that at the optimum Nb_2O_5 content, the device performance reaches over 15%. The corresponding photovoltaic parameters for the champion cells are summarized in Table 1. The PSCs based on TiO_2 ETLs have a short-circuit current density (J_{sc}) of 19.55 mA cm^{-2} , an open circuit voltage (V_{oc}) of 0.99 V, a fill factor (FF) of 0.698, and a PCE of 13.47%. Compared to the TiO_2 ETL-based PSC, the Nb_2O_5 – TiO_2 ETL-based PSC has enhanced parameter values: J_{sc} is 20.49 mA cm^{-2} , V_{oc} is 1.04 V, FF is 0.716, and the PCE is 15.25%, which is the best efficiency achieved in this study. Compared to the TiO_2 ETL-based device, all the key J – V parameters of the Nb_2O_5 – TiO_2 ETL-based device are considerably better. Fig. 5a shows the J – V curves for the champion devices based on both TiO_2 and Nb_2O_5 – TiO_2 ETLs measured in the reverse and forward scan directions. Compared to the control device, the photovoltaic performance of these champion devices is considerably better; the larger J_{sc} and FF

are attributed to the improved electron mobility and better hole blocking effect of the Nb_2O_5 – TiO_2 ETL, and the high V_{oc} may be due to the reduced charge recombination and improved electron extraction.^{37–39} Fig. 5b shows the incident photon-to-current efficiency spectra for various ETLs. The integrated current density value for the pristine TiO_2 -based cell is 18.92 mA cm^{-2} , and it increases to 19.50 mA cm^{-2} for the Nb_2O_5 – TiO_2 -based device; that value is in good agreement with the J – V measurement value. Performance statistics for 30 individual cells with TiO_2 and Nb_2O_5 – TiO_2 ETLs are shown in Fig. 6. Clearly, PCEs show a narrower distribution with a smaller standard deviation for the Nb_2O_5 – TiO_2 -based cells, indicating good reproducibility.

3.3. Recombination

To gain insight into the electron extraction and transport mechanism, the steady-state PL and TRPL were measured for the perovskite absorber layer deposited on both the ETL-based substrates. Fig. 7a shows that the spectral peak for the FTO/perovskite sample at 766 nm apparently has the highest PL intensity, indicating serious recombination occurring in the sample. The FTO/ Nb_2O_5 – TiO_2 /perovskite sample has the lowest PL intensity, even lower than that of the FTO/ TiO_2 /perovskite sample. Interestingly, compared to the PL peak for the control samples, the peak for the FTO/ Nb_2O_5 – TiO_2 /perovskite film exhibits a considerable blue-shift.⁴⁰ This suggests that Nb_2O_5 – TiO_2 substrate is favorable to obtain higher-quality perovskite films than the control group. Perovskite with higher crystallinity and considerably less trap density than the corresponding bandgap will decrease (*i.e.*, the conduction band minimum will move down).⁴¹ Fig. 7b shows the time-resolved

Table 1 Key parameters of champion PSCs based on pristine TiO_2 and Nb_2O_5 – TiO_2 ETLs

	Scan direction	V_{oc} (V)	J_{sc} (mA cm^{-2})	FF	PCE (%)
Nb_2O_5 – TiO_2	Reverse	1.04	20.49	0.716	15.25%
	Forward	1.02	20.47	0.678	14.17%
TiO_2	Reverse	0.99	19.55	0.698	13.47%
	Forward	0.93	19.76	0.664	12.17%



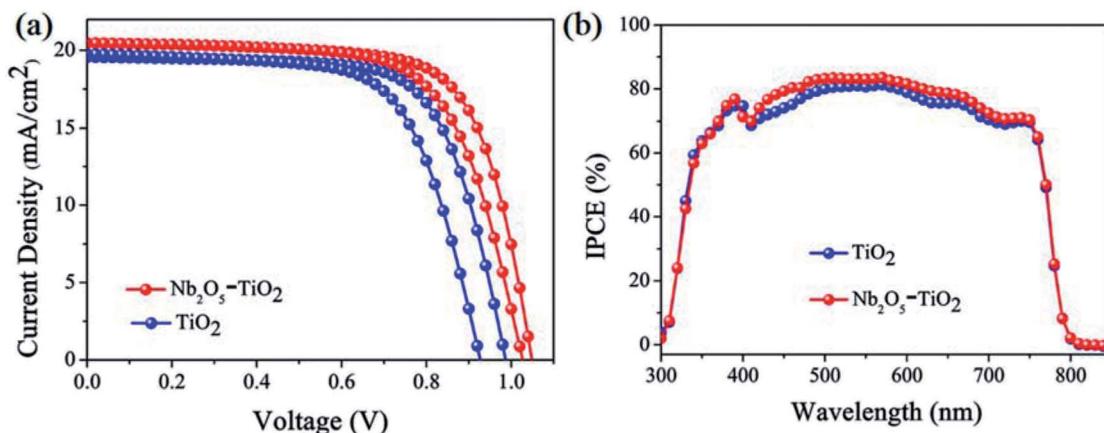


Fig. 5 (a) J – V curves and (b) IPCE spectra of the best-performing PSCs made with TiO₂ and Nb₂O₅–TiO₂ ETLs on FTO glass.

photoluminescence (TRPL) decay curves obtained to analyze the PL lifetimes of the perovskite films prepared on both the ETL-based substrates. PL decay transients and the corresponding PL lifetimes were obtained by performing fitting using a bi-exponential decay function, $f(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2)$, where τ_1 is the fast transient component representing the surface properties and τ_2 is the slow component resulting from the bulk properties.^{42,43} The fitted parameters for the glass/perovskite, glass/TiO₂/perovskite, and glass/Nb₂O₅–TiO₂/perovskite samples are summarized in Table S2.† For the glass/CH₃NH₃PbI₃ sample, τ_1 is 1.05 ns (1.89%) and τ_2 is 429.50 ns (98.11%), with an amplitude average lifetime of 49.40 ns. For the TiO₂/CH₃NH₃PbI₃ sample, τ_1 is 1.36 ns (2.93%) and τ_2 is 255.33 ns (97.07%), with an amplitude average lifetime of 39.48 ns. For the Nb₂O₅–TiO₂/CH₃NH₃PbI₃ sample, τ_1 is 0.92 ns (11.68%) and τ_2 is 223.58 ns (88.32%), with an average lifetime was 7.60 ns. Therefore, it is apparent that the photon-induced electrons transfer from perovskite to the Nb₂O₅–TiO₂ ETL is faster and more effective than that from perovskite to the TiO₂ ETL. Thus, the PL characterizations confirm that the large enhancement in the photovoltaic performances of the Nb₂O₅–TiO₂ ETL can be attributed to the enhanced charge extraction.

3.4. Flexible perovskite solar cells

The totally room-temperature UV process for efficient ETLs is very suitable for fabricating high-performance flexible PSCs. The excellent performance of uniform and dense Nb₂O₅–TiO₂ ETLs prompted us to fabricate flexible perovskite solar cells using flexible substrates. Nb₂O₅–TiO₂ ETLs were successfully fabricated on indium tin oxide (ITO)/PET substrates, and perovskite, spiro-OMeTAD, and a gold electrode were sequentially deposited using the same methods as those used for the FTO glass-based devices. Fig. 8a shows the J – V curves for the perovskite solar cells fabricated using flexible ITO/PET substrates, and the inset illustrates a photograph of a flexible perovskite solar cell fabricated using flexible ITO/PET substrates. For the flexible device, J_{sc} is 20.04 mA cm^{−2}, V_{oc} is 0.99 V, and the FF is 0.69, giving the best PCE of 13.60%. The PCE is lower than that of the rigid device because of the decreased J_{sc} , V_{oc} and FF; the decrease is probably caused by the higher series resistance and lower transmittance of the ITO/PET substrate in the short wavelength spectrum.¹⁷ Fig. 8b shows the J – V curves for the flexible device after it is recovered from the given bending radius. The key J – V parameters of the devices are summarized in Table S3.† After the device is bent with $R = 10$, 5,

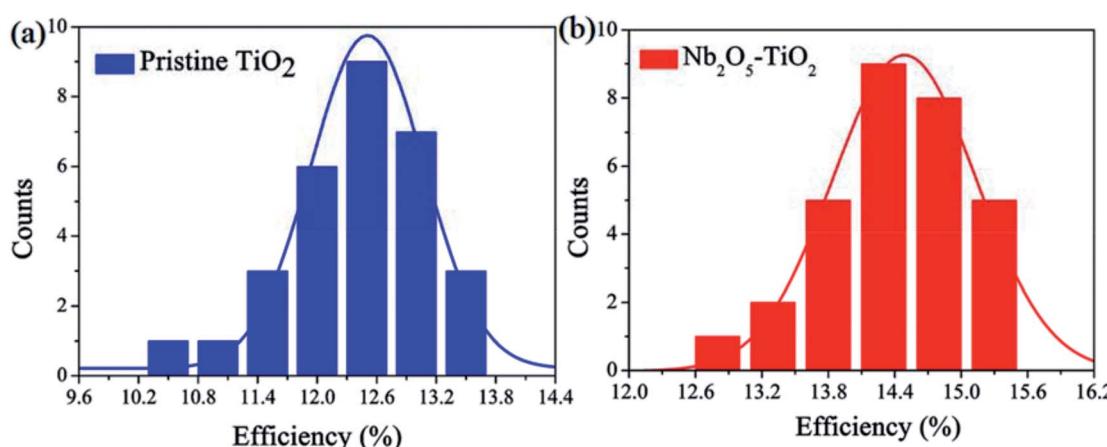


Fig. 6 PCE distribution histogram of PSCs based on TiO₂ (a) and Nb₂O₅–TiO₂ (b) ETLs.



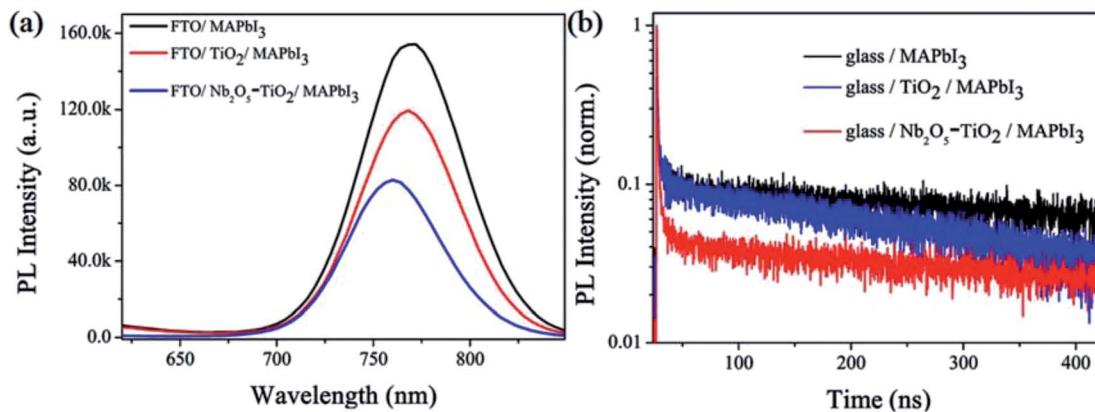


Fig. 7 (a) Steady-state photoluminescence (PL) spectra and (b) normalized time-resolved photoluminescence (TRPL) spectra of pristine perovskite and perovskite layers grown on TiO₂ and Nb₂O₅-TiO₂ ETLs.

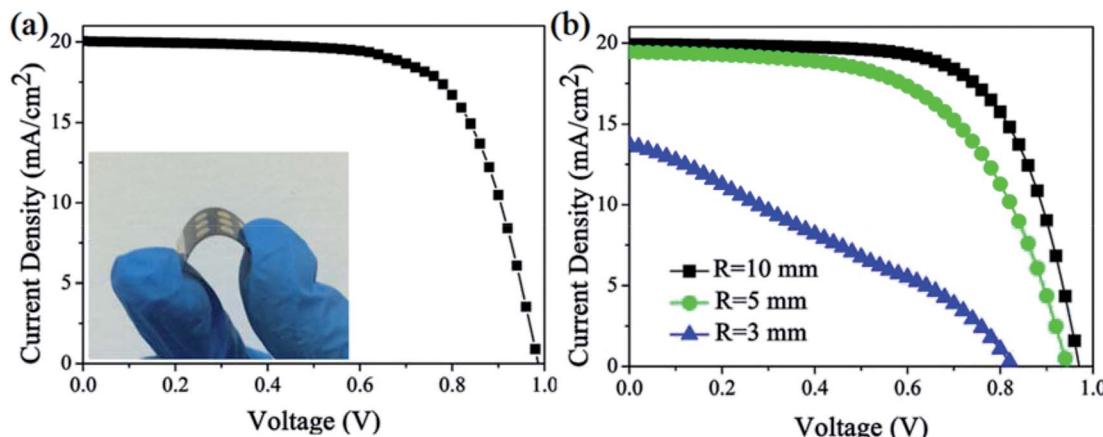


Fig. 8 (a) The J - V curves of best PCE of the perovskite solar cells device using flexible PET/ITO substrates. The inset shows the flexible PSC based on Nb₂O₅-TiO₂ ETLs coated on ITO/PET substrate. (b) J - V curves of the flexible PSCs obtained after recovery from bending at various radius (R = 10, 5, and 3 mm).

and 3 mm, the PCE values degenerate to 13.03%, 10.70%, and 3.39%, respectively. As shown in Fig. 8b, the performance of the flexible device does not show serious degradation when the bending radius is 10 mm. When the bending radius is 3 mm, the brittle ITO breaks⁴⁴ and the PCE is greatly reduced simultaneously, indicating that the flexible devices show good mechanical stability and the Nb₂O₅-TiO₂ ETLs is a promising electron transport material.

4. Conclusion

We have demonstrated that Nb₂O₅-TiO₂ is an excellent ETL material for perovskite solar cells, with the champion cell showing a considerably higher PCE (15.25%) than that of devices based on a pristine TiO₂ ETL and that of a rigid substrate (13.47%). In the proposed process, niobium ethoxide facilitates the spontaneous coalescence of the TiO₂ NCs, thereby forming Nb₂O₅-TiO₂ ETL. Our results suggest that low-temperature solution-processed Nb₂O₅-TiO₂ could be a good ETL candidate for producing efficient perovskite solar cells. Our facile strategy is highly suitable for fabricating high-

performance flexible PSCs because it does not need a high-temperature process and can easily modify the ETL *via* the direct addition of a reagent. This approach will pave the way for further advances in flexible PSCs and is feasible for large scale roll-to-roll processing.

Conflicts of interest

There are no conflicts to declare.

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