Photoluminescence properties of a \( \text{ScBO}_3: \text{Cr}^{3+} \) phosphor and its applications for broadband near-infrared LEDs†

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The rapid extension of solid state lighting technologies offers the possibility to develop broadband near-infrared (NIR) phosphor-converted LEDs (pc-LEDs) as novel NIR light sources. In this paper, a new NIR-emitting phosphor \( \text{ScBO}_3: \text{Cr}^{3+} \) was synthesized by a high temperature solid state reaction method. Phase structure, spectroscopic properties, luminescent lifetime, quantum yield, emitter concentration influences and thermal quenching behavior of \( \text{ScBO}_3: \text{Cr}^{3+} \), as well as its applications for NIR pc-LEDs, were systematically investigated. \( \text{ScBO}_3: \text{Cr}^{3+} \) phosphors exhibit a broad absorption band ranging from 400 to 530 nm, which matches well with the characteristic emission of the blue LED chip. Moreover, \( \text{Cr}^{3+} \) ions occupy the \( \text{Sc}^{3+} \) sites with relatively low crystal field strength in the \( \text{ScBO}_3 \) host, and therefore \( \text{ScBO}_3: \text{Cr}^{3+} \) phosphors show intense broadband emission peaking at \( \sim 800 \) nm upon excitation at \( 460 \) nm, originating from spin-allowed \( ^4T_2 \rightarrow ^4A_2 \) transition of \( \text{Cr}^{3+} \) ions. The optimum \( \text{Cr}^{3+} \) concentration was determined to be \( \sim 2 \) mol\% with a quantum yield of \( \sim 65\% \). A broadband NIR pc-LED prototype device was fabricated by the combination of \( \text{ScBO}_3: \text{Cr}^{3+} \) phosphors and a blue LED chip, which showed a maximum NIR light output power of \( \sim 26 \) mW and a corresponding energy conversion efficiency of \( \sim 7\% \). The results indicate the great potential of \( \text{ScBO}_3: \text{Cr}^{3+} \) phosphors for applications in broadband NIR pc-LEDs.

Introduction

Since the first demonstration of high-brightness blue LEDs and subsequently white pc-LEDs,\(^1,2\) solid state lighting technologies have developed greatly and brought about a revolution in lighting and display industries. White LEDs are emerging as novel lighting sources to replace traditional incandescent and fluorescent lamps in numerous fields, due to their superior advantages including high luminous efficiency, environment friendliness, long lifetime, compactness and good reliability. The widely used method for producing white LED devices is the combination of a blue LED chip and certain phosphor materials with appropriate emission wavelengths. These white pc-LEDs possess simple device structure and low manufacturing cost. Besides white LEDs, the rapid developments of solid state lighting technologies also will provide a promising solution to construct broadband NIR light sources, taking advantage of the phosphor-converted LED technique.\(^3\)

NIR light sources can be widely applied in a variety of fields, such as optical communication, biomedical imaging, spectroscopy, food detection, security surveillance and face (or iris) recognition.\(^8\)–\(^12\) Incandescent bulbs, halogen lamps and ALGaAs-based LEDs are normally used as NIR light sources, however, to some extent suffering from various weaknesses. Incandescent bulbs and halogen lamps can provide continuous light emissions covering from visible to infrared ranges. However, their practical applications are limited by some drawbacks, such as large sizes, short lifetimes, low luminescent efficiency and high working temperature.\(^8\) In contrast, the NIR semiconductor LEDs possess small sizes, long lifetimes and high energy-conversion efficiency. Unfortunately, the LEDs normally show narrow emission bands with the largest full width at half maximum (FWHM) of \( \sim 40 \) nm,\(^5,7\) which are inappropriate for most spectroscopic applications. A possible solution to overcome the limitations of current NIR light sources is to design the NIR-emitting pc-LEDs, which are composed of well-established InGaN blue LED chips and NIR-emitting phosphors. The NIR pc-LEDs can benefit from the outstanding merits of the blue-emitting InGaN LED chips, such as relatively higher thermal stability, continuously increased luminescent efficiency and decreased manufacturing cost. Since the invention of high-brightness blue LEDs in the mid 1990s,\(^2,3\) steady
improvements have been achieved for InGaN-based blue LEDs. According to Haiz's law, the light output per LED package increases by a factor of twenty every decade, and the cost per lumen falls by a factor of ten.\textsuperscript{13} On the other hand, broadband and wavelength-tunable emissions can be expected for NIR pc-LEDs through the rational selection of emitting ions and host materials of NIR-emitting phosphors. Moreover, the continuous emission spectra will be achieved by the appropriate combination between various types of phosphor materials that possess various emission wavelengths upon excitation by the blue light.

To design NIR-emitting phosphors excited by the blue light, the emitting ions should be first of all chosen reasonably. The utilization of 4f–4f transitions of trivalent lanthanide ions is hampered by their spectrally narrow and low intensity absorption.\textsuperscript{4} For divalent or trivalent lanthanide ions (e.g., \textit{Eu}\textsuperscript{2+} and \textit{Ce}\textsuperscript{3+}) featuring 4f–5d optical transitions, it is difficult to achieve the light emissions in the NIR spectral range due to the large energy separation between 4f and 5d levels. Recently, Bi-doped crystals and glasses have attracted intense attention due to their super broadband emissions from visible to NIR.\textsuperscript{14–16} However, the luminescent behaviour of Bi-active centre has not been well understood, and it is still a big challenge to improve its NIR emission efficiency.\textsuperscript{6} Transition metal ions with incomplete 3d shells (3d\textsuperscript{n}, n < 10) have a number of low-lying energy states, between which radiative transitions in the NIR spectral region may occur. Moreover, because they are strongly coupled to the coordination ligands in hosts, optical transitions of transition metal ions are significantly affected by the crystal field symmetry and strength. In general, transition metal ions show relatively strong and broad excitation bands, as well as wavelength-tunable broadband emissions. As typical transition metal ions, \textit{Cr}\textsuperscript{3+} ions in the material ruby (\textit{Al}_{2}\textit{O}_{3}:\textit{Cr}\textsuperscript{3+}) have been thoroughly investigated by spectroscopists for over a hundred years.\textsuperscript{17} Laser-related spectroscopy of \textit{Cr}\textsuperscript{3+} ions in a variety of crystals has also been extensively studied for the developments of tunable solid state lasers.\textsuperscript{18} Recently, NIR long-persistent luminescence of \textit{Cr}\textsuperscript{3+}-activated gallate phosphors has gained considerable attention due to their potential applications in optical information storage, night-vision surveillance and bio-imaging.\textsuperscript{19–22} \textit{Cr}\textsuperscript{3+} ions exhibit strong absorption band in the visible spectral range and can be effectively excited by the blue light. In addition, in contrast to sharp emission lines arising from the \textit{2E} → \textit{4A}_2 transition of \textit{Cr}\textsuperscript{3+} ions in strong-field sites (e.g., \textit{Al}_{2}\textit{O}_{3}:\textit{Cr}\textsuperscript{3+}), a tunable broad emission band arising from the \textit{4T}_2 → \textit{4A}_2 transition can be observed when \textit{Cr}\textsuperscript{3+} ions occupy lattice sites with relatively weak crystal field.\textsuperscript{23} Thees optical characteristics imply that \textit{Cr}\textsuperscript{3+} ions are an ideal choice of luminescence centres for the development of novel NIR-emitting phosphors.

In this paper, NIR-emitting \textit{Sc}BO\textsubscript{3}:\textit{Cr}\textsuperscript{3+} phosphors were synthesized via a solid state reaction method. \textit{Sc}_{2}O\textsubscript{3} (99.99%), \textit{H}_3BO\textsubscript{3} (99.95%) and \textit{Cr}_2O\textsubscript{3} (99.95%) were used as starting materials. Stoichiometric amounts of raw materials were weighed and thoroughly mixed, except that 25% excess of \textit{H}_3BO\textsubscript{3} was added to compensate its evaporation loss in course of high temperature sintering. Subsequently, the powder mixture was transferred into an alumina crucible and calcined in a muffle furnace at 1300 °C for 10 h. Finally, the as-prepared products were naturally cooled down to room temperature and ground to fine powders for subsequent characterization.

**Experimental section**

**Phosphor synthesis**

\textit{Sc}_{1–x}\textit{Cr}_x\textit{BO}_3 (x = 0.005–0.1) phosphors were synthesized via a solid state reaction method. \textit{Sc}_{2}O\textsubscript{3} (99.99%), \textit{H}_3BO\textsubscript{3} (99.95%) and \textit{Cr}_2O\textsubscript{3} (99.95%) were used as starting materials. Stoichiometric amounts of raw materials were weighed and thoroughly mixed, except that 25% excess of \textit{H}_3BO\textsubscript{3} was added to compensate its evaporation loss in course of high temperature sintering. Subsequently, the powder mixture was transferred into an alumina crucible and calcined in a muffle furnace at 1300 °C for 10 h. Finally, the as-prepared products were naturally cooled down to room temperature and ground to fine powders for subsequent characterization.

**Phosphor characterization**

The phase structure of phosphors was identified by powder X-ray diffraction (XRD) measurements on a D8 Discover diffractometer (Bruker) with Cu K\textit{α} radiation (\textit{λ} = 1.5406 Å). Rietveld structure refinements were conducted using the general structure analysis system (GSAS) program.\textsuperscript{26} The particle morphology was observed on a Sirion field-emission scanning electron microscope (FE-SEM, FEI). Diffuse reflection spectrum measurements were performed on a Cary 5000 UV-vis-NIR spectrophotometer (Varian) equipped with an internal diffuse reflectance accessory. The PL excitation spectra were measured on an F-7000 fluorescence spectrophotometer (Hitachi). The PL emission spectra were obtained on a Maya2000 portable spectrometer (Ocean Optics) using a blue LED (\textit{λ}_\text{em} = 450 nm) as the excitation source. Temperature-dependent (30–250 °C) emission spectra were measured with the assistance of a self-designed heating system. The luminescent decay curves were measured by a FluoroMax-4 fluorescence spectrometer (Horiba Jobin Yvon). The quantum yield (QY) was measured by an integrating sphere equipped with a Maya2000 spectrometer and a blue LED, and the spectrometer was carefully calibrated by a standard tungsten lamp. The quantum yield is defined as the percentage of the number of emitted photons to that of absorbed photons, which can be calculated using the following equation:\textsuperscript{27}

\[ \eta = \frac{L_{\text{sample}}}{E_{\text{reference}} - E_{\text{sample}}} \times 100\% \]  

where \( \eta \) is the quantum yield, \( L_{\text{sample}} \) is the integrated emission intensity of the sample, whereas \( E_{\text{reference}} \) and \( E_{\text{sample}} \) represent the integrated intensities of the excitation light with and without the sample in the integrating sphere, respectively.
NIR pc-LED fabrication and performance measurement

NIR pc-LEDs were fabricated by applying a mixture of ScBO$_3$:Cr$^{3+}$ phosphors and transparent silicon resin on an InGaN blue LED chip ($\lambda_{em} = 455$ nm). Photographs of the lighted pc-LED were taken by a cellphone’s camera (Samsung S7), which used a Si-based photodetector covering the spectral range up to ~1100 nm. When taking the picture, an 800 nm long-pass filter was used to remove all the visible light. For electroluminescence (EL) measurements, the packaged LED device was operated at a DC forward bias of ~3.0 V with various injection currents, and the EL spectral distribution was recorded using an integrating sphere and a corrected Maya2000 spectrometer. The NIR light output power of the pc-LED was measured by a FieldMate light power meter (Coherent) equipped a PM10 thermopile sensor, where a 600 nm long-pass filter was used to remove the residual blue light from the LED chip. The energy conversion efficiency was defined as the ratio of the NIR light output power to the input electrical power.

Results and discussion

Structural analysis

Rietveld structure refinements of as-prepared ScBO$_3$ and ScBO$_3$:0.02Cr$^{3+}$ samples were performed to identify their phase structure. The crystallographic data (ICSD# 79-0097) reported by Keszler et al. was adopted as the initial model.\textsuperscript{28} Fig. 1 shows the refined XRD patterns of ScBO$_3$ and ScBO$_3$:0.02Cr$^{3+}$, and the final refinement parameters are listed in Table 1. The refined profiles confirm the phase purity without unidentified diffraction peaks from impurity, regardless of Cr$^{3+}$ doping. Both ScBO$_3$ and ScBO$_3$:Cr$^{3+}$ are found to crystallize as a rhombohedral structure with a space group of R3c (no. 167). Fig. 2 displays the crystal cell of ScBO$_3$ and the coordination environment of [ScO$_6$] group. In ScBO$_3$, Sc$^{3+}$ ions are coordinated by six oxygen atoms and B$^{3+}$ ions are surrounded by three oxygen atoms. ScBO$_3$ exhibits a layered calcite-like structure, which is composed of [ScO$_6$] octahedron and trigonal planar [BO$_3$] group. Structural connectivity between the [ScO$_6$] and [BO$_3$] group occurs only via corner-sharing.

For the ScBO$_3$ host, the lattice parameters are calculated to be $a = b = 4.7516(5)$ Å, $c = 15.2835(7)$ Å, $V = 298.84(1)$ Å$^3$, and $Z = 6$. For ScBO$_3$:0.02Cr$^{3+}$, the smaller lattice parameters are determined: $a = b = 4.7481(9)$ Å, $c = 15.2641(1)$ Å, $V = 298.03(1)$ Å$^3$. The effective ionic radius of Cr$^{3+}$ is 0.615 Å when the coordination number (CN) is equal to 6, which is close to that of Sc$^{3+}$ (0.745 Å) and far larger than that of B$^{3+}$ (0.27 Å). Therefore, Cr$^{3+}$ ions prefer to occupy the Sc$^{3+}$ sites with octahedral coordination in ScBO$_3$:Cr$^{3+}$ phosphors. Due to the smaller ionic radius of Cr$^{3+}$, the calculated cell parameters and Sc–O bond length of ScBO$_3$:0.02Cr$^{3+}$ are slightly smaller than that of pure ScBO$_3$ (Table 1).

XRD patterns of ScBO$_3$:Cr$^{3+}$ phosphors with various Cr$^{3+}$ concentrations are shown in Fig. 3a. All the diffraction peaks of ScBO$_3$:xCr$^{3+}$ ($x = 0.01$–0.08) phosphors agree well with those of a rhombohedral ScBO$_3$ structure (JCPDS Card no. 79-0097). No impurity phase can be recognized with the increase of Cr$^{3+}$ contents.

Noting that the XRD peaks of ScBO$_3$:xCr$^{3+}$ phosphors shift to higher diffraction angle with increasing the Cr$^{3+}$ concentration

| Table 1 Rietveld structure refinements and lattice parameters for ScBO$_3$ and ScBO$_3$:0.02Cr$^{3+}$ |
|-----------------------------------------------|------------------|------------------|
| Space group | $R3c$ | Rhombohedral |
| Symmetry | | |
| Cell parameters | | |
| $a = b$ (Å) | 4.7516(5) | 4.7481(9) |
| $c$ (Å) | 15.2835(7) | 15.2641(1) |
| $\alpha = \beta = \gamma$ (°) | 90°, 120° | 90°, 120° |
| $Z$, volume (Å$^3$) | 6, 298.84(1) | 6, 298.03(1) |
| $R_p$ (%), $R_{wp}$ (%) | 6.43%, 8.57% | 8.45%, 11.51% |
| $\chi^2$ | 3.163 | 6.672 |
| Bond distance and angle | | |
| B–O (Å) | 1.37624(1) | 1.3916E(16) |
| Sc–O (Å) | 2.12198(2) | 2.1124(9) |
| $\angle$OBO (°) | 120 | 120 |
| $\angle$OScO (°) | 87.686(1), 180.000(0), 87.483(25), 180.000(0), 92.314(1) | 92.517(25) |
Fig. 2 (a) The crystal structure representation of ScBO₃ and (b) the coordination environment of [ScO₆] octahedron.

(3b), indicative of the decrease of lattice interplanar spacing due to the smaller ionic radius of Cr³⁺. The SEM image (Fig. 3c) shows that the obtained ScBO₃:Cr³⁺ phosphor is mainly composed of a large number of spherical particles with an average size of 1–3 μm, except that a few large-sized particles (>5 μm) are also observed.

Spectroscopic properties

When Cr³⁺ (3d³) ions occupy lattice sites with octahedral coordination, their energy level distribution can be illustrated by the Tanabe–Sugano diagram (Fig. 4a). Except for ⁵E and ⁴T₁ levels, most of Cr³⁺ crystal field levels (e.g., ⁴T₂, ⁴T₁, ²A₁) show strong dependences on Dq/B values, where Dq and B are the crystal field strength and Racah parameters, respectively. Optical transitions of Cr³⁺ ions can be understood in more detail with the aid of their configuration coordinate diagrams (Fig. 4b).

Fig. 5a shows the diffuse reflection spectra of ScBO₃ and ScBO₃:xCr³⁺ samples. Two broad absorption bands centred at ~460 and 645 nm were detected after the incorporation of Cr³⁺ ions into the ScBO₃ host, which are originated from the ⁴A₂ → ⁴T₂ and ⁴A₂ → ⁴T₁ transitions, respectively. The emission band shapes of Cr³⁺-doped compounds are normally characterized by two broad absorption bands in the visible spectral range, arising from the spin-allowed ⁴A₂ → ⁴T₂ and ⁴A₂ → ⁴T₁ transitions, respectively. The emission band shapes of Cr³⁺-doped compounds are determined by the host crystal field strength, depending on the fact whether the ⁵E or ⁴T₂ level is lowest. As shown in Fig. 4a, ⁵E and ⁴T₂ levels cross at Dq/B = 2.3. When Cr³⁺ ions occupy high-field sites (Dq/B > 2.3), ⁵E will be the emitting state and the Cr³⁺ emission spectra is characterized by sharp emission lines (⁵E → ⁴A₂). When Cr³⁺ ions occupy low-field sites (Dq/B < 2.3), the ⁴T₂ → ⁴A₂ transition will dominate the Cr³⁺ luminescence featuring a very broad emission band (Fig. 4b).

Fig. 5b represents the excitation and emission spectra of ScBO₃:0.02Cr³⁺ phosphors. Upon excitation at 450 nm, the PL spectrum displays a broad emission band extending from 700 to 950 nm, with a maximum at ~800 nm and a FWHM value of ~120 nm. The broad emission band should be attributed to the spin-allowed transitions of Cr³⁺ ions in an octahedral crystal field (Dq: the octahedral crystal field parameter, B: Racah parameter).
The $Dq/B$ value is about 2.15, which is lower than 2.3, indicating that Cr$^{3+}$ ions occupy lattice sites in ScBO$_3$ with lower crystal field strength (Fig. 4a). Based on structural analysis, Cr$^{3+}$ ions preferentially occupy the Sc$^{3+}$ sites in ScBO$_3$:Cr$^{3+}$ phosphors. The weak crystal field should be related to longer Sc$-$O bond distance in ScBO$_3$, because the crystal field strength is inversely proportional to the distance between the central ions and coordination ions.\(^{31}\)

**Cr$^{3+}$ concentration effects**

Fig. 6a shows the PL emission spectra of ScBO$_3$:xCr$^{3+}$ phosphors with various Cr$^{3+}$ contents. The spectral profile changes little with the increase of Cr$^{3+}$ concentrations. The peak wavelengths of broad emission bands of ScBO$_3$:xCr$^{3+}$ phosphors exhibit a slight red-shift with increasing $x$ values (Fig. S2†), despite the fact that the Sc$-$O distance decreases at higher Cr$^{3+}$ contents. We believe the re-absorption between activator ions play an important role in ScBO$_3$:xCr$^{3+}$ phosphor with larger $x$ values. As shown in Fig. 5b, the long-wavelength excitation band of ScBO$_3$:Cr$^{3+}$ slightly overlaps with the emission band, and the re-absorption between Cr$^{3+}$ ions will reduce the emission at the blue wing and result in the red-shift of emission band.

The emission intensities of ScBO$_3$:xCr$^{3+}$ phosphors increase with increasing $x$ values and reach a maximum at a critical concentration of $x = 0.02$. A further increase in the Cr$^{3+}$ concentration results in the decrease of the emission intensity due to the concentration quenching effect (Fig. 6b). Nonradiative

\[ E(4T_2) = 10Dq \]

\[ \frac{Dq}{B} = \frac{15(\Delta E/Dq - 8)}{(\Delta E/Dq)^2 - 10(\Delta E/Dq)} \]

where $\Delta E$ is the energy separation between $4T_2$ and $4T_1$ states. The value of $\Delta E$ is determined by the direct comparison between the spectral positions of $^4A_2 \rightarrow 4T_2$ and $^4A_2 \rightarrow 4T_1$ excitation bands. The energy value ($E$) of the $^4T_2$ level is determined from the excitation peak of the $^4A_2 \rightarrow 4T_2$ transition, at the same time taking into account the influences of the Stokes shift.\(^{31}\) Herein, the difference in force constants between the $^4A_2$ and $^4T_2$ (or $^4T_1$) parabolas is approximately ignored (Fig. 4b). The values of $Dq$ and $B$ are calculated to be $\sim$1400 and 651 cm$^{-1}$, respectively.

![Image](https://example.com/figure5.png)  
**Fig. 5** (a) Diffuse reflection spectra of ScBO$_3$ and ScBO$_3$:xCr$^{3+}$; (b) excitation and emission spectra of ScBO$_3$:0.02Cr$^{3+}$ phosphors.
energy migration among luminescence centres can occur via exchange interaction or multipole–multipole interaction. In this case, the critical energy transfer distance \( R_c \) can be approximately estimated by the following equation proposed by Blasse: \(^\text{31}\)

\[
R_c \approx 2 \left( \frac{3V}{4\pi \chi c N} \right)^{1/3}
\]

(4)

where \( V \) is the unit cell volume, \( \chi \) is the critical activator concentration and \( N \) is the number of the total activator sites per unit cell. For ScBO\(_3\):0.02Cr\(^{3+}\), \( V = 298.03 \text{ Å}^3 \), \( \chi = 0.02 \), \( N = 6 \), and therefore the \( R_c \) value is determined to be \( \sim 16.8 \) Å. The exchange interaction is a short-range effect and usually takes effect only when the distance between activators is shorter than 5 Å. Therefore, nonradiative electric multipolar interaction is the preferred energy transfer mechanism between Cr\(^{3+}\) ions in ScBO\(_3\). In this case, the luminescent intensity per activator can be expressed by the following equation: \(^\text{34}\)

\[
\frac{I}{x} = \frac{k}{1 + \beta(x)^{\theta/3}}
\]

(5)

where \( I \) is the PL intensity, \( k \) and \( \beta \) are constants depending on the interaction type and the host lattice, and \( x \) represents the Cr\(^{3+}\) doping concentration. According to the research work of Van Uitert, \(^\text{34}\) \( \theta = 3 \) corresponds to the energy transfer among the nearest-neighbor ions, whereas \( \theta = 6, 8, \) and 10 correspond to the dipole–dipole, dipole–quadrupole, and quadrupole–quadrupole interactions, respectively. As shown in Fig. 6c, the correlation between \( \log(I/x) \) and \( \log(x) \) can be fitted linearly and the slope \( (\theta/3) \) is determined to be \( \sim 1.06 \). The calculated \( \theta \) value is close to 3, implying that the quenching tends to be proportional to the activator concentration and the concentration quenching mechanism of ScBO\(_3\):xCr\(^{3+}\) phosphors can be attributed to the nonradiative energy transfer among the nearest-neighbor ions. \(^\text{34–36}\)

Fig. 7 shows the luminescent decay curves of ScBO\(_3\):xCr\(^{3+}\) phosphors with various \( x \) values. All the decay curves can be well fitted by a single exponential function (Fig. S3†). The luminescent lifetime \( (\tau) \) are determined to approximately 105, 94, 83, 67 and 41 \( \mu s \) for ScBO\(_3\):xCr\(^{3+}\) where \( x = 0.01, 0.02, 0.03, 0.05, 0.08 \). The decrease of the Cr\(^{3+}\) lifetime is attributed to the increased luminescent quenching effect at higher Cr\(^{3+}\) concentrations. The well fitting results of luminescent decay curves by a single exponential function also indicate that the Cr\(^{3+}\) ions occupy only single lattice site in the ScBO\(_3\) host.

**Temperature-dependent PL properties**

Luminescent thermal stability is another important factor of the phosphors for pc-LED applications. It is well known that the emission intensity of luminescent materials usually decreases in different degrees at elevated temperatures due to increased nonradiative transitions. The working temperature of LED devices can be higher than 100 °C, \(^\text{37}\) and therefore the phosphor materials should maintain their luminescent intensity at higher temperatures for pc-LED applications. Fig. 8a represents the PL spectra of ScBO\(_3\):0.02Cr\(^{3+}\) at various temperatures (30–250 °C), where a thermal quenching behaviour can be observed. With the temperature increasing from 30 to 150 °C, the emission intensity of ScBO\(_3\):0.02Cr\(^{3+}\) decreases by approximately 51% from the initial intensity (Fig. 8b). NIR-emitting ScBO\(_3\):Cr\(^{3+}\) phosphors show lower thermal stability than the commercial

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**Fig. 7** Luminescent decay curves of ScBO\(_3\):xCr\(^{3+}\) phosphors with various Cr\(^{3+}\) concentrations \( (\lambda_{ex} = 450 \text{ nm}) \).

**Fig. 8** (a) Temperature-dependent emission spectra, (b) temperature-dependent normalized integrated intensities and (c) activation energy plots using the modified Arrhenius equation for ScBO\(_3\):0.02Cr\(^{3+}\) phosphors \( (\lambda_{ex} = 450 \text{ nm}) \).
visible-emitting phosphors, such as Y3Al5O12:Ce3+ and CaAlSiN3:Eu2+, for which larger than 80% of the emission intensity at room temperature can be retained at 150 °C.36,39 The thermal quenching of ScBO3:Cr3+ can be caused via two possible routes:31 (1) multi-phonon emission directly across the energy gap between the excited state 4T2 and the ground state 4A2; (2) nonradiative transition through the crossing of 4T2 and 4A2 parabolas (Fig. 4b). The increased thermal quenching of ScBO3:Cr3+ is probably related to the narrow energy gap between 4T2 and 4A2 levels, as well as the larger Stokes shift of the phosphor.

The temperature-dependence of the emission intensity can be described by a modified Arrhenius equation:30

$$I(T) = \frac{I_0}{1 + A \exp(-E_a/kT)}$$

(6)

where $I_0$ is the initial intensity, $I(T)$ is the intensity at a given temperature, $A$ is a constant, $k$ is the Boltzmann constant and $E_a$ is the activation energy for the thermal quenching. Fig. 8c depicts a plot of ln($I/I_0 - 1$) versus $1/T$, and the activation energy ($E_a$) is determined to be ~0.36 eV for ScBO3:Cr3+ phosphors through the best fit of the Arrhenius equation.

**Fabrication and performance of NIR pc-LED**

In order to demonstrate the applicability of ScBO3:Cr3+ phosphors, a NIR pc-LED prototype was fabricated by the combination of a 455 nm InGaN LED chip and the ScBO3:0.02Cr3+ phosphor. Photographs of the as-fabricated pc-LED device and the lighted one are shown in the insets of Fig. 9a. Bight NIR emission could be photographed by a cellphone’s camera after the visible light was removed by an 800 nm long-pass filter. Fig. 9a shows a downshift on the emission that called here electroluminescence (EL) spectrum, even if the is no direct electron-excitation of the as-fabricated pc-LED at various drive currents. The blue emission band at ~455 nm comes from the LED chip and the broad emission band between 700–950 nm is attributed to ScBO3:Cr3+ phosphors. With the increase in the drive current at a forward bias of ~3 V, no obvious changes in the EL profile can be found, except for the continuous increase of the luminescence intensity.

Fig. 9b represents the dependences of the NIR light output on the injection current of the pc-LED device. A continuous increase in the NIR light output power is found with increasing the injection current. A maximum output power of ~26 mW is achieved at a drive current of 120 mA. The energy conversion efficiency is calculated to be approximately 7%, much larger than the reported values of NIR pc-LEDs based on trivalent lanthanide doped glass phosphors.44 Noting that the energy conversion efficiency of the LED chip used here is about 20% and the blue light from the LED chip still remains for the NIR pc-LED device. The energy conversion efficiency can be further improved by engineering the package structure and utilizing more efficient blue LED chips. These results demonstrate the great potential of ScBO3:Cr3+ phosphors as an alternative NIR component for applications in broadband NIR pc-LEDs.

**Conclusions**

ScBO3:Cr3+ phosphors were synthesized via a solid state reaction method and investigated for their structural and luminescent characteristics. The phase structure of as-prepared phosphors was confirmed by the Rietveld analysis. In the ScBO3 host, Cr3+ ions preferentially occupied Sc3+ sites with low-field octahedral coordination. Therefore, a broadband emission ($\lambda_{max} = 800$ nm) in the NIR spectral range between 700–950 nm was observed for ScBO3:Cr3+ upon excitation by the blue light, which was attributed to $^4T_2 \rightarrow ^4A_2$ transitions of Cr3+ ions. The luminescent quenching was proportional to the Cr3+ concentration in Sc1-xCrxBO3 phosphors, and the maximum emission intensity was found at $x = 0.02$ with a quantum yield of ~65%. The thermal quenching properties of ScBO3:Cr3+ phosphors were also inspected and the activation energy was determined to be ~0.36 eV. By integrating ScBO3:Cr3+ phosphors on the blue LED chip, a broadband NIR pc-LED was obtained with a maximum NIR light output of ~26 mW and the corresponding energy conversion efficiency of ~7%. The results suggest that ScBO3:Cr3+ phosphors can potentially serve as conversion phosphors for broadband NIR pc-LED devices.

**Conflicts of interest**

There are no conflicts to declare.
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Notes and references