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## Correction: Water oxidation by a copper(II) complex: new findings, questions, challenges and a new hypothesis

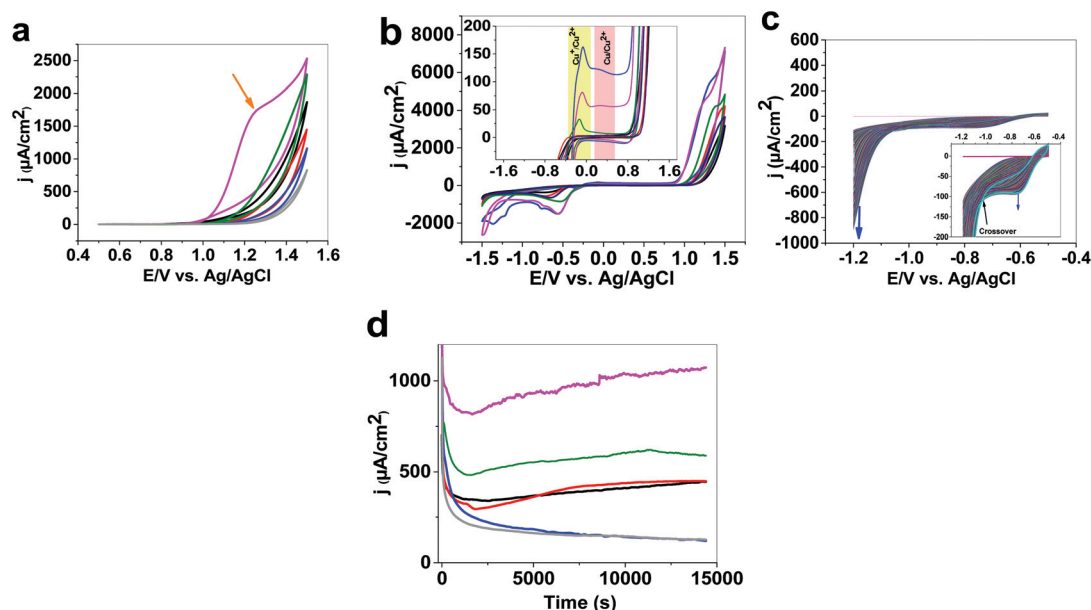
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Correction for 'Water oxidation by a copper(II) complex: new findings, questions, challenges and a new hypothesis' by Mohammad Mahdi Najafpour *et al.*, *Dalton Trans.*, 2018, 47, 9021–9029.

The authors would like to correct part b of Fig. 1, which, in the published manuscript, is the same as part a. The correct Fig. 1 is shown below:



**Fig. 1** CVs (scan rate: 50.0 mV s<sup>-1</sup>; a, b) for 1 (red), 2 (blue), 3 (olive), 4 (gray), copper(II) salt (pink), and bare FTO electrode (black as a reference) in the buffer solution. The continuous CV for 2 at (–1.2)–(–0.5) V (c). Amperometry for 1 (red), 2 (blue), 3 (olive), 4 (gray), copper(II) salt (pink) and bare FTO electrode (black as a reference) in the buffer solution (d). The orange arrows indicate the peak attributed to uncomplexed Cu(II) species. The conditions: in the phosphate buffer solution (0.25 M, pH = 11.0) at room temperature. The electrochemical studies were carried out with a conventional three-electrode setup, in which FTO, Ag/AgCl/KCl<sub>sat</sub> and platinum foil served as working, reference and auxiliary electrodes, respectively. The concentrations of copper(II) compounds were 0.25 mM.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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