## PCCP



**View Article Online** 

## CORRECTION

Check for updates

**Cite this:** *Phys. Chem. Chem. Phys.,* 2018, **20**, 20733

## Correction: Necessary and sufficient conditions for the successful three-phase photocatalytic reduction of CO<sub>2</sub> by H<sub>2</sub>O over heterogeneous photocatalysts

Kentaro Teramura\*<sup>ab</sup> and Tsunehiro Tanaka\*<sup>ab</sup>

DOI: 10.1039/c8cp91824d

rsc.li/pccp

Correction for 'Necessary and sufficient conditions for the successful three-phase photocatalytic reduction of  $CO_2$  by  $H_2O$  over heterogeneous photocatalysts' by Kentaro Teramura *et al.*, *Phys. Chem. Chem. Phys.*, 2018, **20**, 8423–8431.

The authors would like to make the following corrections to their published article: (1) On page 8424, eqn (5) II O should be shapped to 211 O as shown in the amended equation help

(1) On page 8424, eqn (5),  $H_2O$  should be changed to  $2H_2O$  as shown in the amended equation below:

$$CO_2 + 8H^+ + 8e^- \rightleftharpoons CH_4 + 2H_2O \tag{5}$$

(2) On page 8426, line 2 in the right column, the sentence should be amended as follows:

Very recently, we also discovered that the modification of  $Ga_2O_3$  with Zn species suppresses the evolution of  $H_2$  and does not change the formation rate of CO.

(3) The references in Table 1 on page 8426 should be corrected as shown in the revised Table 1 below:

**Table 1** Summary of photocatalysts tested for the photocatalytic reduction of  $CO_2$  by  $H_2O$  in our research group. Ag co-catalyst loading: 0.25–1.0 wt%; light source: 400 W high-pressure Hg lamp; water volume: 1.0 L;  $CO_2$  flow rate: 30 mL min<sup>-1</sup>; additive: 0.1 M NaHCO<sub>3</sub>

Photocatalyst	Cat. weight/g	Rate of gas evolved/ $\mu$ mol h <sup>-1</sup>						
		$H_2$	$O_2$	CO	Conv. of $CO_2$ (%)	Selec. to CO (%)	$e^{-}/h^{+}$	Ref.
Ag/ZnGa <sub>2</sub> O <sub>4</sub> /Ga <sub>2</sub> O <sub>3</sub>	1.0	16.9	70.1	117.0	0.16	87.4	0.96	33 and 73
Ag/La <sub>2</sub> Ti <sub>2</sub> O <sub>7</sub>	1.0	4.9	5.3	5.2	0.01	51.5	0.95	74
Ag/SrO/Ta <sub>2</sub> O <sub>5</sub>	1.0	3.8	5.1	6.8	0.01	64.2	1.04	75
Ag/ZnGa <sub>2</sub> O <sub>4</sub>	1.0	7.8	78.3	147	0.20	95.0	0.99	37
$Ag/ZnTa_2O_6$	0.5	22.3	20.0	21.9	0.03	49.5	1.11	76
Ag/Sr <sub>2</sub> KTa <sub>5</sub> O <sub>15</sub>	1.0	8.3	34.3	65.5	0.09	88.8	1.08	77
Ag/SrNb <sub>2</sub> O <sub>6</sub>	0.5	1.1	24.8	51.2	0.07	97.9	1.05	46
Ag/Mg-Al LDH/Ga <sub>2</sub> O <sub>3</sub>	1.0	131	167	212	0.29	61.7	1.03	78
Ag/Pr/Ga <sub>2</sub> O <sub>3</sub>	0.5	64.7	150	249	0.34	79.4	1.04	79
Ag/K <sub>2</sub> YTa <sub>5</sub> O <sub>15</sub>	1.0	16.2	43.2	91.9	0.13	85.0	1.25	80
Ag/Sr <sub>1.6</sub> K <sub>0.37</sub> Na <sub>1.43</sub> Ta <sub>5</sub> O <sub>15</sub>	1.0	16.0	53.7	94.6	0.13	85.5	1.03	81
Ag/Yb-Zn/Ga <sub>2</sub> O <sub>3</sub>	0.5	37.6	103	150	0.21	80.0	0.91	82
Ag-Cr/Ga <sub>2</sub> O <sub>3</sub>	0.5	92.9	281	480	0.66	83.8	1.02	83

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>a</sup> Department of Molecular Engineering, Graduate School of Engineering, Kyoto University, Kyotodaigaku Katsura, Nishikyo-ku, Kyoto 615-8510, Japan.

E-mail: teramura@moleng.kyoto-u.ac.jp, tanakat@moleng.kyoto-u.ac.jp

<sup>&</sup>lt;sup>b</sup> Elements Strategy Initiative for Catalysts & Batteries (ESICB), Kyoto University, 1-30 Goryo-Ohara, Nishikyo-ku, Kyoto 615-8245, Japan