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CORRECTION

View Article Online



Cite this: Phys. Chem. Chem. Phys., 2018, 20, 3847

Correction: High-temperature X-ray diffraction and thermal expansion of nanocrystalline and coarse-crystalline acanthite α -Ag₂S and argentite β-Ag₂S

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DOI: 10.1039/c8cp90018c

rsc.li/pccp

Correction for 'High-temperature X-ray diffraction and thermal expansion of nanocrystalline and coarse-crystalline acanthite α -Ag₂S and argentite β -Ag₂S' by S. I. Sadovnikov et al., Phys. Chem. Chem. Phys., 2016. 18. 4617-4626.

The authors wish to draw the readers' attention to their previous related study, published in *Physics of the Solid State*, which should have been cited in this *Physical Chemistry Chemical Physics* paper.

The study published in this Physical Chemistry Chemical Physics paper contains new experimental X-ray diffraction data, differential thermal and thermogravimetric analysis (DTA-DTG) results and data on the acanthite-argentite phase transformation enthalpy. This Physical Chemistry Chemical Physics paper was accepted before the publication of ref. 1 but published after ref. 1. Therefore ref. 1 should have been cited in this *Physical Chemistry Chemical Physics* paper.

The authors regret not giving the correct attribution for Fig. 4, 6, 7, 8 and 9 in the paper, which were reproduced for the readers' information. The figures are reproduced below with the correct copyright permission text.

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Correction

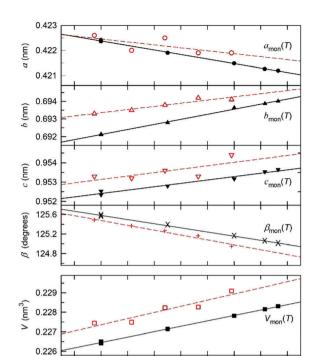


Fig. 4 The effect of temperature T on the unit cell parameters a, b, c, β , and volume V, and on the volumetric thermal expansion coefficient β_V of coarse- and nanocrystalline acanthite. The approximation of the experimental data by the solid line and the closed symbols (●), (▲), (▼), (★), (★), and (♦) corresponds to coarse-crystalline acanthite and the approximation by the dotted line and the open symbols (\bigcirc) , (\triangle) , (∇) , (+), (\square) , and (\lozenge) corresponds to nanocrystalline acanthite. Reproduced from ref. 1 with permission from Springer.

340 360 T (K) $\beta_V(T)$

75

65 60 55

280

300 320

Ax106 (K1) 70 **PCCP**

(200) 623 K β -Ag₂S (space group $Im\overline{3}m$) 593 K 563 K 533 K 503 K 473 K 463 K 453 K 443 K 36.4 36.6 36.8 37.0 Θ 2θ (220)623 K 593 K Counts (arb. units) 563 K 533 K 503 K 500 K 473 K 463 K 453 K

Fig. 6 Evolution of XRD patterns of coarse-crystalline argentite β -Ag₂S in the temperature range of 446–623 K. The inset shows a systematic displacement of the (200) diffraction reflection of bcc argentite with increase of measuring temperature. Reproduced from ref. 1 with permission from Springer.

 2θ (degrees)

40

50

20

30

443 K

60

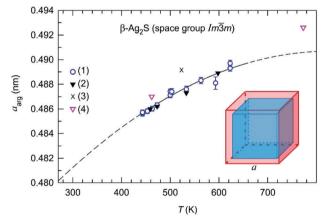


Fig. 7 Dependence of the lattice constant a_{arg} of argentite β -Ag₂S on the temperature T: (1) data of present work; (2), (3), and (4) data, ^{22,24,27} respectively. The approximations of measured lattice constant a_{arg} by the function (10) in the temperature range of 440–660 K is shown by solid lines. Reproduced with some changes from ref. 1 with permission from Springer.

Correction

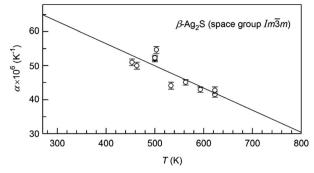


Fig. 8 Temperature dependence of linear thermal expansion coefficient α_{arg} of argentite β -Ag₂S and its approximation by the function (12). Reproduced from ref. 1 with permission from Springer.

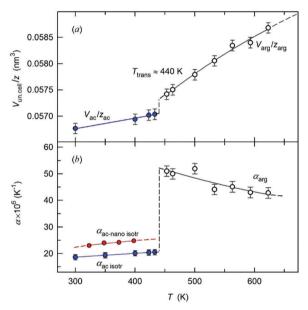


Fig. 9 The temperature dependencies of reduced volume $V_{un,cell}/Z$ (a) and isotropic linear thermal expansion coefficient α (b) of silver sulfide in the of range 300-623 K. At \sim 440 K, there take place jumps of the reduced volume and the thermal expansion coefficient α attributed to the first-order acanthite—argentite phase transformation. Isotropic linear thermal expansion coefficient $\alpha_{ac-nanoisotr}$ of nanocrystalline acanthite α -Ag₂S is larger than $\alpha_{ac\,isotr}$ of coarse-crystalline acanthite. Reproduced with changes from ref. 1 with permission from Springer.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

References

1 A. I. Gusev, S. I. Sadovnikov, A. V. Chukin and A. A. Rempel, Phys. Solid State, 2016, 58, 251-257.