



Cite this: *Chem. Commun.*, 2015, 51, 15446

Received 10th July 2015,
Accepted 27th August 2015

DOI: 10.1039/c5cc05740j

www.rsc.org/chemcomm

Synthesis of diversely functionalised 2,2-disubstituted oxetanes: fragment motifs in new chemical space†

Owen A. Davis, Rosemary A. Croft and James A. Bull*

Di-, tri- and tetra-substituted oxetane derivatives with combinations of ester, amide, nitrile, aryl, sulfone and phosphonate substituents are prepared as fragments or building blocks for drug discovery. The synthesis of these novel oxetane functional groups, in new chemical space, is achieved via rhodium-catalysed O–H insertion and C–C bond forming cyclisation.

Recent years has seen considerable enthusiasm for exploring new chemical motifs in medicinal chemistry that enter unexplored chemical and intellectual property (IP) space.^{1,2} Oxetanes motifs have gained significant traction as desirable modules for drug discovery.³ Carreira, Rogers-Evans, Müller and co-workers demonstrated that 3,3-disubstituted oxetanes constitute polar replacements for *gem*-dimethyl groups, and often afford improved solubility and physicochemical properties.⁴ Subsequently, oxetane motifs have been adopted widely in medicinal chemistry programs.^{5,6} 3-Substituted oxetane motifs have been examined in particular, exploiting oxetan-3-one^{3,4} and 3-iodooxetane⁷ as building blocks. New spirocycles,⁸ peptidomimetics⁹ and other oxetane analogues of important compounds have been prepared (*e.g.* Fig. 1).^{6,10}

Examples of 2-substituted oxetanes remain less studied; Pfizer reported a class of 2-oxetanyl containing γ -secretase inhibitors that possessed improved properties relative to tetrahydrofuran and carbocyclic derivatives.⁶ However, oxetanes bearing functional groups on the ring are unexplored, and could provide fascinating new structural units for use by medicinal chemists. The synthesis of functionalised oxetanes continues to be a challenge. Cyclisation by intramolecular Williamson etherification is most common,³ but 2-functionality is often precluded due to unstable oxyanionic intermediates. Also, epoxide-opening and ring-closure methods using sulfoxonium ylides are not tolerant of sensitive electrophilic functionality.¹¹ On the other hand, the photochemical

Paterno–Büchi reaction can introduce functionality from an alkene component, but is often restricted to very highly substituted oxetanes, with undesirable properties for medicinal chemistry.^{12–14}

We have targeted new oxetane derivatives due their small, polar nature and well-defined 3-D shape.¹⁵ We recently reported the preparation of 2-sulfonyl oxetanes, formed through a novel anionic C–C bond forming cyclisation (Scheme 1A).¹⁶ These fragments displayed good pH stability, and stability to liver microsomes.^{16b} We have also established an efficient synthesis of oxetane 2,2-dicarboxylates involving O–H insertion using diazomalonates, and C–C bond forming cyclisation (Scheme 1B).¹⁷

Here we report the synthesis of unusual oxetane motifs bearing medicinally relevant functional groups directly on the

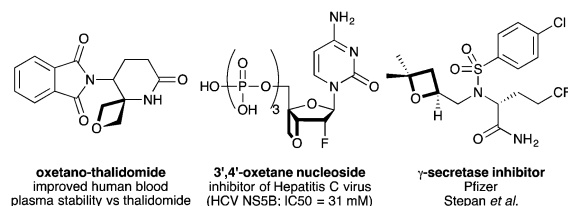
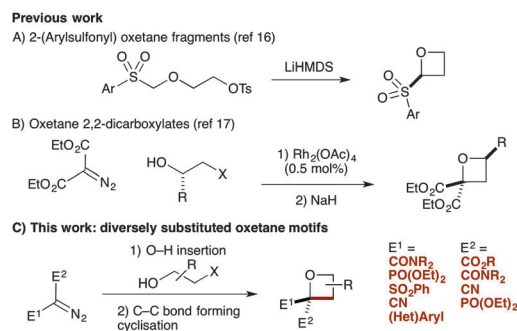


Fig. 1 Biologically active oxetane-containing compounds.



Scheme 1 Synthesis of oxetanes through anionic C–C bond forming cyclisation.

Department of Chemistry, Imperial College London, South Kensington,
London SW7 2AZ, UK. E-mail: j.bull@imperial.ac.uk

† Electronic supplementary information (ESI) available: Experimental procedures, characterisation data, ¹H and ¹³C NMR spectra; details of reaction optimisation and calculated molecular properties. See DOI: 10.1039/c5cc05740j



ring (Scheme 1C). Diversely functionalised oxetanes in new chemical space are prepared in a facile manner from functionalised diazo compounds by O–H insertion and C–C bond forming cyclisation. These simple but highly functionalised motifs provide interesting shape and physicochemical properties for fragments or building blocks for drug discovery that cannot be prepared by other methods.

Targeting 2,2-disubstituted oxetanes, our investigations initially focused on a range of unsymmetrical acceptor–acceptor diazo compounds, with which to examine O–H insertion with 2-bromoethanol (Scheme 1C and Table 1).^{18,19} A series of diazo compounds **1a–g** bearing amide, sulfone, phosphonate, and nitrile groups were prepared from the methylene precursors using TsN₃ or TfN₃.^{18,20,21} We first examined amide–ester diazo **1a** (entry 1). Pleasingly, using catalytic Rh₂(OAc)₄ (0.5 mol%) in benzene at 80 °C successfully mediated O–H insertion of the carbene derived from diazo **1a** into 2-bromoethanol in 75% yield.

Cyclisation was then effected by treatment of **2a** with NaH (1.2 equiv., 0 °C, 1 h, then 25 °C, 1 h) to afford 2-amido-2-ester-oxetane **3a** in excellent yield forming the C(2)–C(3) bond. Using Weinreb amide **1b** (entry 2), a slightly reduced yield was obtained for the O–H insertion (51%), but an excellent yield was obtained on cyclisation of **2b** to oxetane **3b**. Sulfone–ester diazo **1c** smoothly underwent O–H insertion, followed by cyclisation in

just 1 h at 0 °C to provide a new type of sulfonyl oxetane (entry 3).¹⁶ Phosphonate–ester diazo **1d** proceeded similarly, to yield the first example of a 2-oxetanyl phosphonate, **3d**.²² Interestingly, the use of nitrile–ester diazo **1e** led to side reactions under the conditions for O–H insertion, with the carbenoid undergoing competitive insertion into the benzene solvent (1.2:1–2.4:1 O–H insertion:π-insertion). To prevent this unwanted side product, the reaction was performed in CH₂Cl₂ (25 °C, 17 h), and bromide **2e** was isolated in 85% yield (entry 5). Cyclisation at 0 °C afforded nitrile oxetane **3e** after 1 h. The sequence was also successful on a larger scale affording >1.5 g of oxetane **3e** in excellent yield (entry 6). Mixed nitrile–amide and phosphonate–amide diazo compounds were also successful under the same conditions, providing oxetanes **3f** and **3g** respectively (entries 7 and 8). The wide variety of substituted oxetanes that can be readily accessed demonstrates the versatility of this approach.

Substituted bromohydrins were then investigated to form tri- and tetrasubstituted oxetane derivatives (Table 2). The reaction sequence successfully formed oxetanes with 4-alkyl (**5a**) and 4-aryl substituents (**5b–e**) in good yields using similar reaction conditions. The additional substituent had little influence on the O–H insertion step. However, elevated temperature and increased reaction times were required for cyclisation of all substituted examples, requiring 16 h at 25 °C for complete conversion.

For oxetanes **5a** and **5b**, a ~1:1 mixture of diastereoisomers was obtained due to the sterically similar ester–sulfone or ester–phosphonate groups.²³ Diazo compounds with a less sterically demanding nitrile group (**1e**, **1h**, **1i**) gave increased diastereoselectivity (up to 3:1 dr). The incorporation of a 3-Ph group gave the 2,2,3-substituted derivative **5f** (dr = 56:44). 2,2,4,4-Tetrasubstituted oxetane **5g** was prepared from the commercially available chlorohydrin (1-chloro-2-methyl-2-propanol)

Table 1 Scope of 2,2-disubstituted oxetanes varying diazo **1a–g**

Entry	E ¹	E ²	Yield 2 ^a (%)	Yield 3 ^b (%)	Oxetane 3
1		CO ₂ Et	a 75	93 ^c	
2		CO ₂ Et	b 51	96 ^c	
3	SO ₂ Ph	CO ₂ Et	c 92	87 ^d	
4	PO(OEt) ₂	CO ₂ Et	d 77 ^e	70 ^d	
5	CN	CO ₂ Bn	e 85 ^f	88 ^d	
6			93 ^{f,g}	93 ^{d,h}	
7		CN	f 67 ⁱ	87 ^d	
8		PO(OEt) ₂	g 84	72 ^d	

^a O–H insertion: **1** (1.1 equiv.), 2-bromoethanol (1.0–2.0 mmol), Rh₂(OAc)₄ (0.5 mol%), PhH, 0.1 M, 80 °C. ^b Cyclisation: **2** (0.3–0.6 mmol), NaH (1.2 equiv.), DMF, 0.025 M. ^c 0 °C for 1 h; then 25 °C for 1 h. ^d 0 °C for 1 h. ^e 1.5 equiv. diazo **1d**. ^f Run in CH₂Cl₂, 0.1 M at 25 °C for 17 h. ^g 10 mmol scale. ^h 8 mmol scale. ⁱ 1.2 equiv. diazo **1f**.

Table 2 Preparation of 2,2,4-, 2,2,3-tri- and 2,2,4,4-tetrasubstituted oxetanes

1c–e,h,i	HO–CH(R)–CH ₂ –Br	4a–h	5a–h		
ⁱ 4a 66%	ⁱ 4b 64%	ⁱ 4c 81%	ⁱ 4d 63%		
ⁱⁱ 5a 92% (dr 55:45) ^{c,g}	ⁱⁱ 5b 71% (dr 54:46) ^g	ⁱⁱ 5c 80% (dr 70:30) ^g	ⁱⁱ 5d 86% (dr 74:26) ^{c,g}		
ⁱ 4e 80%	ⁱ 4f 77%	ⁱ 4g 40%	ⁱ 4h 65%		
ⁱⁱ 5e 43% (dr 70:30) ^g	ⁱⁱ 5f 75% (dr 56:44) ^{c,g}	ⁱⁱ 5g 54%	ⁱⁱ 5h 95% (dr 83:17) ^{c,g}		

Conditions: (i) O–H insertion: **1** (1.1 equiv.), β-bromohydrin (0.5–1.3 mmol), Rh₂(OAc)₄ (0.5 mol%), PhH, 0.1 M, 80 °C. (ii) Cyclisation: **4** (0.3–0.5 mmol), NaH (1.2 equiv.), DMF, 0.025 M, 25 °C, 16 h. ^a dr determined by ¹H NMR of crude reaction mixture. ^b β-Chlorohydrin used. ^c Oxetane isomers separated by silica chromatography. ^d 1.5 equiv. diazo. ^e Run in CH₂Cl₂, 0.1 M at 25 °C for 17 h. ^f 1.2 equiv. diazo. ^g Major diastereoisomer indicated (see ref. 23).



in moderate yields. The synthesis of novel fused bicyclic oxetane scaffold **5h** started by treating *N*-Boc-3-pyrroline with NBS and H₂O to provide 3-bromo-4-hydroxy-pyrrolidine, which gave ether **4h** on O–H insertion. Cyclisation then afforded the bicyclic oxetane **5h** as a 83 : 17 mixture of separable diastereoisomers in 95% combined yield.^{21,23}

Next we examined aryl-ester (donor-acceptor) diazo compounds,²⁴ to generate 2-aryl-oxetane-2-carboxylates (Table 3). Such aryl-substituted heterocycles offer attractive features for fragment screening and are little studied in medicinal chemistry.²⁵ To our delight, the O–H insertion conditions were effective for the phenyl-ester diazo **6a** with an excellent 86% yield using reduced Rh₂(OAc)₄ loading (0.25 mol%, entry 1). The O–H insertion was successful across varied aryl groups (entries 3–8), including the chloropyridyl diazo **6f**. However, using the NaH in DMF conditions for cyclisation gave variable yields and oxetane **8a** and bromide **7a** proved inseparable.

Table 3 Scope of 2-aryl-oxetane 2-carboxylates

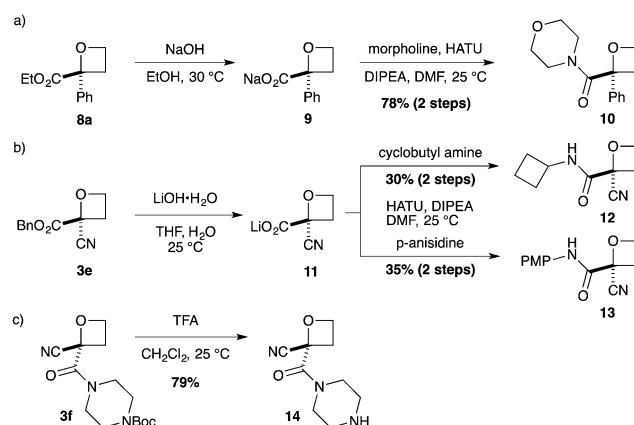
Entry	Ar	E	Yield 7 ^a (%)	Yield 8 ^b (%)	Oxetane 8
1	Ph	CO ₂ Et	a 86	93	
2			80 ^c	76 ^d	
3	Ph	CO ₂ iPr	b 87 ^e	72	
4	3-ClPh	CO ₂ Et	c 69	97	
5	4-CF ₃ Ph	CO ₂ Et	d 64	77	
6	4-MeOPh	CO ₂ Et	e 86	22	
7				41 ^f	
8		CO ₂ Et	f 54	76	
9	3-ClPh		g 73	90	

^a O–H insertion: **6** (1.1–1.5 equiv.), 2-bromoethanol (0.4–2.0 mmol), Rh₂(OAc)₄ (0.25 mol%), PhH, 0.1 M, 80 °C. ^b Cyclisation: **7** (0.3–0.5 mmol), LiHMDS (1.4 equiv.), THF, 0.025 M, 0 °C, 1 h. ^c Reaction on 7.5 mmol scale. ^d Reaction on 5.0 mmol scale. ^e 0.5 mol% Rh₂(OAc)₄. ^f 2.0 equiv. LiHMDS.

Alternative bases and conditions were examined,²¹ and the use of LiHMDS (1.4 equiv.) in THF was found to give quantitative conversion to oxetane **8a** with a 93% isolated yield (entry 1). This reaction sequence was similarly successful on a larger scale (entry 2). The modified cyclisation conditions worked well for substituted aryl derivatives including halogenated (entries 4 and 5), and pyridyl derivatives (entry 8). However, the electron rich *p*-methoxy phenyl substrate (entry 6) led to low conversion and isolated yield. Using a larger excess of LiHMDS (2 equiv.) gave an improved yield of 41% (entry 7). The cyclisation conditions were also effective for aryl-amide ether **7g** (entry 9). This approach provided direct access to compounds that conform well to fragment guidelines for fragment screening.²¹

As a more divergent approach to similar compounds, hydrolysis of ester oxetane **8a** was examined (Scheme 2a). Using NaOH afforded sodium salt **9** as an attractive oxetane containing building block. Amide coupling with morpholine, directly from the salt,²⁶ gave amide **10** in a good yield over two steps. Finally, we prepared additional 2-nitrile oxetane derivatives. The nitrile group is receiving considerable attention in drug discovery,^{27,28} being metabolically robust and a strong hydrogen bond acceptor. Therefore, we considered that the combination with an oxetane motif could present a fascinating structure for medicinal chemistry.^{13,29} From nitrile-oxetane benzyl ester **3e**, ester hydrolysis was best achieved using LiOH to afford lithium salt **11** (Scheme 2b). Amide bond formation gave nitrile amide oxetanes **12** and **13** in 30–35% yields over 2 steps. Importantly, treatment of oxetane **3f** with TFA promoted Boc deprotection, **14**, without other degradation, providing encouraging preliminary information on acid stability (Scheme 2c; 10 equiv. TFA, 22 h).

In summary, we have developed a versatile strategy to prepare diversely functionalised oxetane derivatives, involving O–H insertion and C–C bond forming cyclisation. A wide variety of functional groups have been introduced on the oxetane ring, accessing new chemical space. We believe these oxetane motifs will provide interesting new structural elements for medicinal chemistry programs as well as synthesis. Efforts towards enantioselective synthesis and studies on the stability and medicinal chemistry properties of the new oxetane derivatives will be reported in due course. We gratefully acknowledge EPSRC



Scheme 2 Divergent synthesis of oxetane derivatives.



(CAF, EP/J001538/1; Impact Acceleration Account, EP/K503733/1), and Imperial College London. We thank EPSRC National Mass Spectrometry Facility, Swansea.

Notes and references

- (a) J.-L. Reymond, R. van Deursen, L. C. Blum and L. Ruddigkeit, *Med. Chem. Commun.*, 2010, **1**, 30; (b) R. J. Hall, P. N. Mortenson and C. W. Murray, *Prog. Biophys. Mol. Biol.*, 2014, **116**, 82; (c) D. H. Drewry and R. Macarron, *Curr. Opin. Chem. Biol.*, 2010, **14**, 289; (d) C. M. Marson, *Chem. Soc. Rev.*, 2011, **40**, 5514; (e) W. R. Pitt, D. M. Parry, B. G. Perry and C. R. Groom, *J. Med. Chem.*, 2009, **52**, 2952; (f) F. W. Goldberg, J. G. Kettle, T. Kogej, M. W. D. Perry and N. P. Tomkinson, *Drug Discovery Today*, 2015, **20**, 11.
- For examples: (a) J. A. Burkhard and E. M. Carreira, *Org. Lett.*, 2008, **10**, 3525; (b) J. A. Burkhard, C. Guérot, H. Knust, M. Rogers-Evans and E. M. Carreira, *Org. Lett.*, 2010, **12**, 1944; (c) J. A. Burkhard, B. Wagner, H. Fischer, F. Schuler, K. Müller and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2010, **49**, 3524; (d) D. B. Li, M. Rogers-Evans and E. M. Carreira, *Org. Lett.*, 2011, **13**, 6134; (e) D. Foley, R. Doveston, I. Churcher, A. Nelson and S. P. Marsden, *Chem. Commun.*, 2015, **51**, 11174.
- J. A. Burkhard, G. Wuitschik, M. Rogers-Evans, K. Müller and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2010, **49**, 9052.
- (a) G. Wuitschik, M. Rogers-Evans, K. Müller, H. Fischer, B. Wagner, F. Schuler, L. Polonchuk and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2006, **45**, 7736; (b) G. Wuitschik, E. M. Carreira, B. Wagner, H. Fischer, I. Parrilla, F. Schuler, M. Rogers-Evans and K. Müller, *J. Med. Chem.*, 2010, **53**, 3227.
- (a) J. Du, B.-K. Chun, R. T. Mosley, S. Bansal, H. Bao, C. Espiritu, A. M. Lam, E. Murakami, C. Niu, H. M. Micolochick Steuer, P. A. Furman and M. J. Sofia, *J. Med. Chem.*, 2014, **57**, 1826; (b) E. M. Skoda, J. R. Sacher, M. Z. Kazancioglu, J. Saha and P. Wipf, *ACS Med. Chem. Lett.*, 2014, **5**, 900; (c) J. E. Dowling, M. Alimzhanov, L. Bao, M. H. Block, C. Chuaqui, E. L. Cooke, C. R. Denz, A. Hird, S. Huang, N. A. Larsen, B. Peng, T. W. Pontz, C. Rivard-Costa, J. C. Saeh, K. Thakur, Q. Ye, T. Zhang and P. D. Lyne, *ACS Med. Chem. Lett.*, 2013, **4**, 800.
- (a) A. F. Stepan, K. Karki, W. S. McDonald, P. H. Dorff, J. K. Dutra, K. J. DiRico, A. Won, C. Subramanyam, I. V. Efremov, C. J. O'Donnell, C. E. Nolan, S. L. Becker, L. R. Pustilnik, B. Sneed, H. Sun, Y. Lu, A. E. Robshaw, D. Riddell, T. J. O'Sullivan, E. Sibley, S. Capetta, K. Atchison, A. J. Hallgren, E. Miller, A. Wood and R. S. Obach, *J. Med. Chem.*, 2011, **54**, 7772; (b) A. F. Stepan, G. W. Kauffman, C. E. Keefer, P. R. Verhoest and M. Edwards, *J. Med. Chem.*, 2013, **56**, 6985.
- (a) M. A. J. Duncton, M. A. Estiarte, R. J. Johnson, M. Cox, D. J. R. O'Mahony, W. T. Edwards and M. G. Kelly, *J. Org. Chem.*, 2009, **74**, 6354; (b) M. A. J. Duncton, M. A. Estiarte, D. Tan, C. Kaub, D. J. R. O'Mahony, R. J. Johnson, M. Cox, W. T. Edwards, M. Wan, J. Kincaid and M. G. Kelly, *Org. Lett.*, 2008, **10**, 3259.
- (a) J. A. Burkhard, C. Guérot, H. Knust and E. M. Carreira, *Org. Lett.*, 2012, **14**, 66; (b) G. Wuitschik, M. Rogers-Evans, A. Buckl, M. Bernasconi, M. Märki, T. Godel, H. Fischer, B. Wagner, I. Parrilla, F. Schuler, J. Schneider, A. Alker, W. B. Schweizer, K. Müller and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2008, **47**, 4512.
- (a) N. H. Powell, G. J. Clarkson, R. Notman, P. Raubo, N. G. Martin and M. Shipman, *Chem. Commun.*, 2014, **50**, 8797; (b) M. McLaughlin, R. Yazaki, T. C. Fessard and E. M. Carreira, *Org. Lett.*, 2014, **16**, 4070.
- For oxetane-thalidomide, see: (a) J. A. Burkhard, G. Wuitschik, J.-M. Plancher, M. Rogers-Evans and E. M. Carreira, *Org. Lett.*, 2013, **15**, 4312; (b) for nucleoside analogue: W. Chang, J. Du, S. Rachakonda, B. S. Ross, S. Convers-Reignier, W. T. Yau, J.-F. Pons, E. Murakami, H. Bao, H. M. Steuer, P. A. Furman, M. J. Otto and M. J. Sofia, *Bioorg. Med. Chem. Lett.*, 2010, **20**, 4539.
- (a) T. Sone, G. Lu, S. Matsunaga and M. Shibasaki, *Angew. Chem., Int. Ed.*, 2009, **48**, 1677; (b) E. D. Butova, A. V. Barabash, A. A. Petrova, C. M. Kleiner, P. R. Schreiner and A. A. Fokin, *J. Org. Chem.*, 2010, **75**, 6229.
- (a) M. Abe, *J. Chin. Chem. Soc.*, 2008, **55**, 479; (b) F. Vogt, K. Jödicke, J. Schröder and T. Bach, *Synthesis*, 2009, 4268.
- For Paterno-Büchi reactions with 1,2-dicyanoethylene, forming nitrile oxetanes, see: (a) W. S. Chung, N. J. Turro, S. Srivastava and W. J. Le Noble, *J. Org. Chem.*, 1991, **56**, 5020; (b) C. W. Funke and H. Cerfontain, *J. Chem. Soc., Perkin Trans. 1*, 1976, 1902; (c) J. A. Barltrop and H. A. J. Carless, *J. Am. Chem. Soc.*, 1972, **94**, 1951.
- For 2,2-disubstituted oxetanes by lithiation of phenyl oxetane, see: D. I. Coppi, A. Salomone, F. M. Perna and V. Capriati, *Chem. Commun.*, 2011, **47**, 9918.
- For interest in less-planar compounds in drug discovery: (a) F. Lovering, J. Bikker and C. Humblet, *J. Med. Chem.*, 2009, **52**, 6752; (b) M. Ishikawa and Y. Hashimoto, *J. Med. Chem.*, 2011, **54**, 1539; (c) F. Lovering, *Med. Chem. Commun.*, 2013, **4**, 515; (d) A. Nadin, C. Hattotuwigama and I. Churcher, *Angew. Chem., Int. Ed.*, 2012, **51**, 1114.
- (a) K. F. Morgan, I. A. Hollingsworth and J. A. Bull, *Chem. Commun.*, 2014, **50**, 5203; (b) K. F. Morgan, I. A. Hollingsworth and J. A. Bull, *Org. Biomol. Chem.*, 2015, **13**, 5265.
- (a) O. A. Davis and J. A. Bull, *Angew. Chem., Int. Ed.*, 2014, **53**, 14230; (b) O. A. Davis and J. A. Bull, *Synlett*, 2015, 1283.
- For extensive studies on O-H insertion with varied diazo compounds, see: (a) G. G. Cox, D. J. Miller, C. J. Moody, E.-R. H. B. Sie and J. J. Kulagowski, *Tetrahedron*, 1994, **50**, 3195; (b) C. J. Moody, E. R. H. B. Sie and J. J. Kulagowski, *Tetrahedron*, 1992, **48**, 3991; (c) for a review: D. Gillingham and N. Fei, *Chem. Soc. Rev.*, 2013, **42**, 4918.
- For an O-H insertion on bromoethanol, see: U. Trstenjak, J. Ilaš and D. Kikelj, *Tetrahedron Lett.*, 2013, **54**, 3341.
- (a) J. C. Lee and J. Y. Yuk, *Synth. Commun.*, 1995, **25**, 1511; (b) D. Marcoux, S. Azzi and A. B. Charette, *J. Am. Chem. Soc.*, 2009, **131**, 6970; (c) R. P. Wurz, W. Lin and A. B. Charette, *Tetrahedron Lett.*, 2003, **44**, 8845; (d) D. Marcoux and A. B. Charette, *Angew. Chem., Int. Ed.*, 2008, **47**, 10155.
- See ESI† for further details.
- For an example of a highly substituted 2-phosponate-oxetan-3-one: T. Maegawa, K. Otake, K. Hirokawa, A. Goto and H. Fujioka, *Org. Lett.*, 2012, **14**, 4798.
- Major diastereoisomers shown determined by nOe studies (**5a**, **5d**, **5f**, **5h**) or by analogy. See ESI† for further details.
- Aryl ester diazo compounds prepared using pABSA reagent. (a) H. M. L. Davies, W. R. Cantrell Jr., K. R. Romines and J. S. Baum, *Org. Synth.*, 1992, **70**, 93; (b) H. M. L. Davies, M. V. A. Grazini and E. Aouad, *Org. Lett.*, 2001, **3**, 1475.
- For an isolated report of related structure, see: E. K. Bayburt, B. Clapham, P. B. Cox, J. F. Daanen, M. J. Dart, G. A. Gfesser, A. Gomtsyan, M. E. Kort, P. R. Kym, R. G. Schmidt and E. A. Voight, *Novel trpv3 modulators*, *US Pat.*, US2013/0131036 A1, 2013.
- J. D. Goodreid, P. A. Duspara, C. Bosch and R. A. Batey, *J. Org. Chem.*, 2014, **79**, 943.
- For a review of nitrile containing pharmaceuticals, see: F. F. Fleming, L. Yao, P. C. Ravikumar, L. Funk and B. C. Shook, *J. Med. Chem.*, 2010, **53**, 7902.
- For anastrozole: (a) M. Milani, G. Jha and D. A. Potter, *Clin. Med.: Ther.*, 2009, **1**, 141; (b) T. E. Hughes, M. D. Mone, M. E. Russell, S. C. Weldon and E. B. Villhauer, *Biochemistry*, 1999, **38**, 11597.
- For other nitrile oxetanes: (a) P. Sulmon, N. De Kimpe, N. Schamp, B. Tinant and J.-P. Declercq, *Tetrahedron*, 1988, **44**, 3653; (b) T. S. Cantrell, *J. Chem. Soc., Chem. Commun.*, 1975, 637.

