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Transport in polymer-supported chemically-doped CVD graphene†

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In this study we report on the electron transport in flexible-transparent polymer supported chemically doped chemical vapour deposited (CVD) graphene. We investigate the modified carrier transport following doping with various metal chlorides. An increase in the work function was noted for AuCl₃-, FeCl₃-, IrCl₃-, and RhCl₃-doping, whilst only SnCl₂ doping showed a reduced work function. We attribute this to dopant-mediated charge transfer resulting in the formation of neutral atomic species. The neutral and near-neutral atomic populations produced metallic aggregates, with this agglomeration level critically dependent on the cohesive energy of the metallic component in each dopant. Micron-scale spatial conductivity mapping highlighted the spatially uniform low resistance of AuCl₃-doped graphene. Local conductivity enhancements at grain boundaries and lattice defects within the as-synthesised polycrystalline graphene suggested that the dopant molecules tend to reside at lattice imperfections. Temperature dependent transport studies indicated that the shifted work function improved electrical conductivity due to the increase of barrier transparency between grain boundaries. Variable Range Hopping (VRH) dominated at temperatures <140 K in undoped graphene, whereas combined Nearest Neighbour Hopping (NNH) and diffusive transport appears to play a major role throughout the transport in all doped samples. The findings herein reveal that the underlying extended transport mechanisms associated with chemically doped CVD graphene transferred to polymer supports contrast with the highly localised transport in undoped graphene.

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Introduction

Graphene, a two-dimensional lattice of hexagonally arrayed carbon atoms, has attracted much attention for various commercial applications given its diverse, and largely unique set of electrical, optical and mechanical properties.¹ Though promising and industrially attractive, the high sheet resistance of chemical vapour deposited (CVD) graphene necessitates the development of improved methodologies of enhancing the nascent materials'

electrical conductivity, without which it will remain challenging to integrate into various large area flexible electronics technologies, such as organic light emitting diodes (OLED),^{2–4} capacitive touch screens^{5,6} and photovoltaic cells.^{7–9} Such applications require areal uniformity, mechanical stability, and strong and long-lasting adhesion between the graphene and substrate upon flexing. There is also a pressing need for a means of work function tuning and this, as a result, has been studied for some time. For large area electronics (typically > 10 cm × 10 cm), simultaneous conductance uniformity at the macro and micro scales, is critical. Chemical vapour deposition is one of the few industrially viable methods that allows for such large area, uniform synthesis, along with ambient pressure roll-to-roll processing, of truly monolayer materials of quality that is now comparable to the best exfoliated samples.^{10,11} CVD is gaining ever increasing traction. Coupled to its decreasing cost, CVD is proving to be one of the most attractive approaches for the high yield synthesis of large area graphene materials.¹⁰

To be of use in most electronic applications, the as-synthesised nascent CVD graphene must be transferred to arbitrary insulating substrates from the opaque and conductive catalyst upon which it is grown. Various transfer methods have been developed to maintain the quality of the as-grown graphene during transfer.^{12–14}

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In our earlier work we reported an approach to transfer graphene onto flexible substrates with one of the highest adhesion potentials between the graphene and substrate reported to date.^{15,16} Though such work demonstrated the mechanical robustness of these polymer-supported layered systems, it remains nonetheless critical to elucidate the underlying electron transport in such transferred two dimensional materials, particularly in order to guide us towards further conductivity improvements. The conductivity of pristine monolayer CVD graphene is, at present, too low to replace incumbent transparent conducting oxide electrodes.^{17,18} In pristine graphene, the density of states at, or near to, the Dirac point remains low, dramatically reducing the conductivity. As a result, various doping strategies have been developed as a critical means of increasing conductivity.^{19–22} Spray coating of silver nanowires has been shown to increase the conductivity of the graphene basal plane by ~ 20 times with only a 4.3% decrease in optical transparency.²³ Doping by plasma exposure²⁴ has been shown to accurately control both the work function and electrical conductivity; 10 mTorr Cl_2 plasma at 2 W RF power led to a +0.6 eV work function shift and two-fold increase in conductivity. However, in defective samples, such plasma readily exacerbates, and even nucleates lattice defects and, given the often elevated process temperatures used, this remains a challenging approach for many common polymer substrates. To avoid many of these issues, herein, we report on a room temperature chemical doping strategy based on the use of various metal chlorides in a facile *chemi douche* process. Chemical doping has been shown to decrease the sheet resistance (R_s) of graphene from $\sim 2000 \Omega \text{ sq.}^{-1}$ to $\sim 90 \Omega \text{ sq.}^{-1}$ without significantly compromising the optical transparency (from $\sim 97\%$ to $\sim 80\%$).^{25–27}

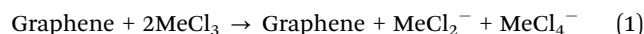
Metal chlorides have been considered as a highly efficient dopants for a wide variety of carbon-based conducting materials.^{22,28–30} The charge transfer rate depends critically on the reaction potential of the selected dopant in relation to the native carbon. The metal chloride chosen for this study where selected due to the high charge transfer propensity and commensurate high reaction potential. In our previous work³¹ we reported on the conductivity and transparency of such systems before and after metal chloride doping, with our findings suggesting that molecular ad/desorption is one of the primary factors contributing to the observed time-dependent shifts in the doping level and, hence, conductivity. In the present report, we extend this earlier work by investigating the detailed temperature dependent electron transport within these doped graphene materials, focussing on the shift in work function and the associated impacts of dopant agglomerate formation.

Experiment

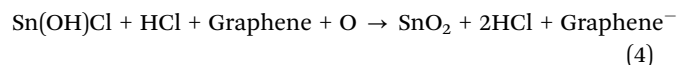
Monolayer graphene was grown by CVD, as reported in ref. 15. The polycrystalline graphene was transferred from the Cu foil catalyst to polyethylene terephthalate (PET) using an Ultraviolet Adhesive (UVA) procedure, as similarly detailed elsewhere.¹⁶ After transfer, the graphene-on-PET was chemically doped with

one of five chloride compounds; AuCl_3 , FeCl_3 , SnCl_2 , IrCl_3 , or RhCl_3 . Dopant solutions were spin-cast onto the transferred graphene samples at 2000 rpm for 1 min after a 30 s spread at 500 rpm. The effects on the conductivity as a function of dopant concentration have been reported elsewhere.³¹ Optical images of typical transferred and doped graphene on PET are shown in the inset of Fig. 1(b) and (c).

As described in Fig. 1(a), the metal chloride (Me_xCl_y) is physically adsorbed, mediating spontaneous charge transfer across well-defined energy levels at the graphene–metal ion interface. In the present system, such an interaction can be expressed as (1)–(3).^{31,32}



The positive Me^{3+} ions in the MeCl_4^- are neutralised following charge donation to the graphene basal plane. Depending on the metal type, the positive reduction potentials of the metal ions result in a reduction in the local electron population in the graphene substrate. Work function therein increased, resulting in p-type doping. Our data suggests that AuCl_3 , FeCl_3 , IrCl_3 and RhCl_3 follow this reaction well, however SnCl_2 differs somewhat.³¹ SnCl_2 reacts with H_2O , producing Sn(OH)Cl and HCl , as;³³



Aqueous SnCl_2 is a reducing agent. Sn^{2+} reacts with bound oxygen species. When the oxygen constituent is removed, the graphene becomes increasingly negatively charged, leading to notable n-type doping.

As seen in Fig. 1(c), upon doping, particles 5–15 μm in diameter form. No such substantial particles were noted prior to doping (Fig. 1(b)). EDX suggests that these particles are very likely dopant agglomerates which form from the incomplete dissolution of the powdered dopant, or as a result of charge transfer induced agglomeration. As described above, the metal chlorides transfer surplus charge to the graphene basal plane leaving charge-neutral metal atoms (Me^0) or, in the somewhat anomalous case of SnCl_2 , SnO_2 . Such neutral metallic constituents tend to aggregate under aqueous conditions and are subsequently fixed upon drying, with their formation dictated by the magnitude of the metal constituents cohesive energy. The total areal coverage and number of agglomerate particles are shown in Fig. 1(d). AuCl_3 -doped graphene shows the largest number of agglomerates, however the size of these agglomerates is smaller by 7.5%/99.2% for $\text{FeCl}_3/\text{IrCl}_3$, respectively. Though the number of agglomerates of the IrCl_3 -doped was smaller than that of the AuCl_3 -doped, the average diameter of the agglomerates was significantly larger (14.4 μm) than that of the AuCl_3 (0.1 μm), with the total agglomerate area being the largest for IrCl_3 -doped samples (9.91%) as a result.

Cohesive energy (CE) is defined as the energy needed to sub-divide a metal into isolated atoms.³⁴ Metals with a low CE



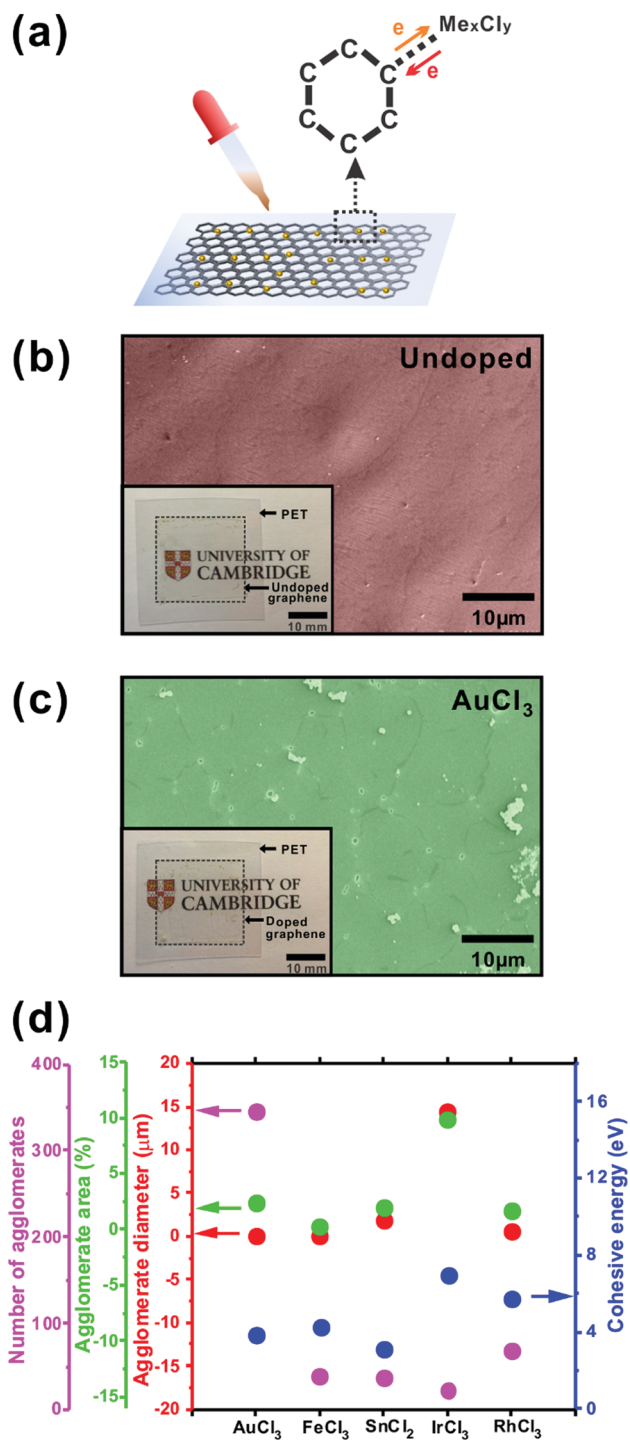


Fig. 1 (a) Doping scheme. SEM micrographs and optical photographs (insets) of (b) undoped and (c) AuCl₃-doped graphene, and (d) comparison of the metal constituent cohesive energy with the number of agglomerates, agglomerate diameter and total agglomerate-covered area.

are readily chemically separated, whilst those of comparatively high CE produce larger particles. The calculated CE of the considered metallic constituents is reported in Fig. 1(d).³⁵ There appears a notable correlation between the magnitude of the CE and the degree of agglomeration of the metallic species. CE seems to play a contributing role in governing

agglomerate formation and thus, concurrently, the resulting optical transparency and sheet resistance. The large CE of Ir (6.94 eV) indicates that Ir has perhaps the greatest likelihood of inducing significant agglomeration, consistent with our experimental findings. Across all the dopants, SnCl₂ was somewhat atypical. The CE_{Sn} = 3.14 eV, which was smaller than that of Au (3.81 eV) and Fe (4.28 eV), however the average diameter of the agglomerates (1.95 μm) was larger than found in both the Au (0.11 μm) and Fe (0.12 μm). The agglomerate covered area of Sn (1.96%) was smaller than that of Au (2.39%), but larger than that of Fe (0.31%), which is largely inconsistent with cohesion-driven agglomeration. Nevertheless, this likely relates to the disparate doping mechanism associated with SnCl₂ compared with the other considered Me_xCl_y. SnO₂ remains after SnCl₂ doping. No Sn neutral species were formed which likely gives rise to the poor correlation between the generation of SnO₂ and CE_{Sn}.

To further investigate the areal uniformity of the chemical doping, sub-micro-scale scanning spreading resistance microscopy (SSRM)^{36,37} was performed using a conductive cantilever (Applied NanoStructures, Inc., SICONA) atomic force microscope (AFM, Veeco Instruments Inc., Dimension icons). With a spatial resolution of $\Delta x, y \sim 11.7$ nm, 3 μm × 3 μm areas were scanned with the sample edges electrically grounded in all cases. Probing the relationship between surface roughness and spreading resistance upon doping allows for immediate comparisons of the nanoscale and macroscale resistance distributions, the latter of which was measured *via* contact four-point probe mapping. For the undoped graphene, SSRM and surface morphology data appear near coincident (Fig. 2(a) and (c)); valleys of comparable vertical dimensions appear less likely to effectively attach to the graphene during the transfer process with such substrate irregularities prohibiting proximal contact between the substrate and the graphene.³⁸ The graphene in these features is not completely transferred, but rather partially lost; consequently the valleys tend to have higher resistance than the surrounding ridges. These substrate effects have been shown elsewhere to be minimised through the use of suspended graphene.³⁹ Contrasting results were found in some areas of the AuCl₃ doped graphene (black and red arrows, Fig. 2(b) and (d)). Studies elsewhere suggest that such features manifests as highly-resistive line defects. However, in the present case the narrow lines showed a surprising reduction in resistance relative to the nascent surrounding graphene. We find that dopant molecules, which do not noticeably augment the feature topography, adhere to these defect sites which include; abrupt changes in the surface morphology such as folds, grain boundaries, cracks and other lattice non-idealities, with such preferential adsorption reducing the feature resistance to values lower than the surrounding graphene. This grain healing is notable upon Me_xCl_y doping. Chang *et al.* argued that adatoms on graphene have a tendency to dwell on atomic steps or boundaries.⁴⁰ Such molecules are largely beyond the detectable limit of the present AFM morphology study, though nonetheless contribute to a measurable change in the conduction. The dopant distribution is inhomogeneous, with a tendency for accumulation at edges, steps, grain boundaries, or other defects. The conductive edge and grain boundaries may possibly contribute to the charge



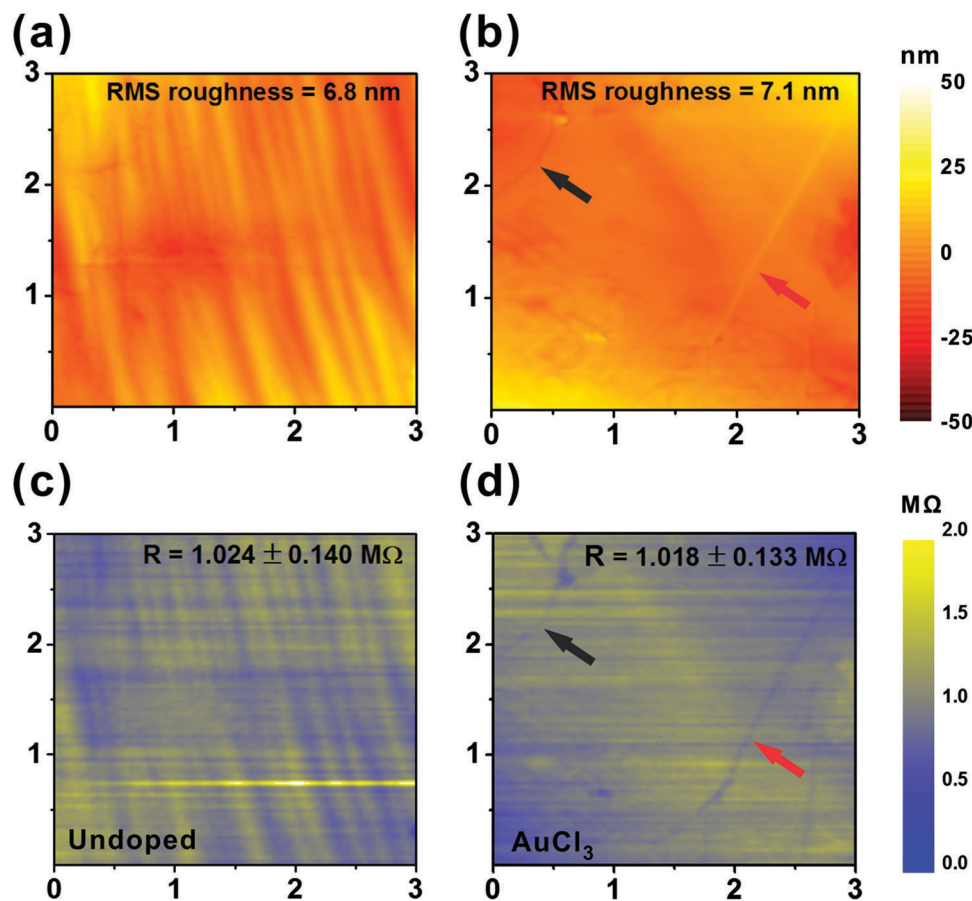


Fig. 2 Atomic Force Microscopic (AFM) maps of (a) undoped, (b) AuCl_3 -doped graphene. Scanning Spreading Resistance Microscopy (SSRM) maps of (c) undoped and (d) AuCl_3 -doped graphene. The black arrow in (b) and in (d) denotes a typical narrow valley and the corresponding low resistance of this feature, respectively.

transport by functioning as a percolation network of sub-nanometre wires, and is the focus of further study.

X-ray Photoemission Spectra (XPS) were acquired using a Mg source (1253 eV) with an energy resolution of 0.092 eV. To measure the change in work function following chemical doping, Ultraviolet Photoemission Spectra (UPS) were measured using a He I lamp (21.2 eV) with an energy resolution of 0.061 eV. For XPS and UPS measurement the as-grown graphene was transferred to polished molybdenum substrates as our earlier efforts indicated that the polymer substrates induced serious charging effects. As shown in Fig. 3(a), the work function of the doped graphene shifted by +0.23 eV (AuCl_3), +0.32 eV (FeCl_3), +0.11 eV (IrCl_3), and +0.13 eV (RhCl_3) from undoped graphene indicating p-doping compared to the undoped scenario, consistent with our Raman results (ESI,† Fig. S4), whereas SnCl_2 -doped graphene exhibited a similar work function to that of the undoped graphene, with only +0.06 eV shift.

The Voigt-convolution fitted⁴¹ $\text{C}1\text{s}$ peak of the undoped and the Me_xCl_y doped graphene are shown in Fig. 3(b)–(g). Two principle carbon bonding components; $\text{C}-\text{C}$ sp^2 at 284.6 eV and $\text{C}-\text{C}$ sp^3 at 285.8 eV are shown, alongside the oxygen containing functional groups $\text{C}-\text{O}$ (hydroxyl) and $\text{C}=\text{O}$ (carboxyl) at 286.0 eV and 287 eV, respectively). In all doped cases, the sp^3 peak does

not significantly increase, indicating there is little defect generation induced by the doping procedure. Our Raman studies independently confirm this. Conversely, however; the $\text{C}-\text{O}$ peak cross-section increases when doped with AuCl_3 (286.4 eV), FeCl_3 (287.2 eV), SnCl_2 (287.6 eV), and IrCl_3 (287.5 eV), even though there is no observable $\text{C}-\text{O}$ peak in the vicinity of 286 eV in the undoped and RhCl_3 -doped graphene. The highest $\text{C}-\text{O}$ peak occurred upon SnCl_2 -doping. It is possible that residual SnO_2 products deleteriously bond to the graphene basal plane. Charge transfer can be examined *via* the $\text{C}1\text{s}$ sp^2 peak shift, attributable to an work function modification. As described in Fig. 3(h), following chemical doping, the undoped graphene peak (284.28 eV) shifted to lower values for AuCl_3 (284.19 eV), FeCl_3 (284.19 eV), IrCl_3 (284.22 eV), and RhCl_3 (284.11 eV) indicating p-doping, whilst it moved to higher binding energy in SnCl_2 -doped graphene (284.34 eV) indicating n-doped characteristics compared to the undoped graphene. There was no additional peak generated from the metal chloride–carbon bonding which is a significant benefit compared to doping mediated by substitutional⁴² or thin film deposition.⁴³ Chemical doping of polymer supported graphene under the present framework appears to be a promising approach to increase the conductivity and therein afford a means of controlling work function without a severely affecting the atomic structure of the as-grown graphene.



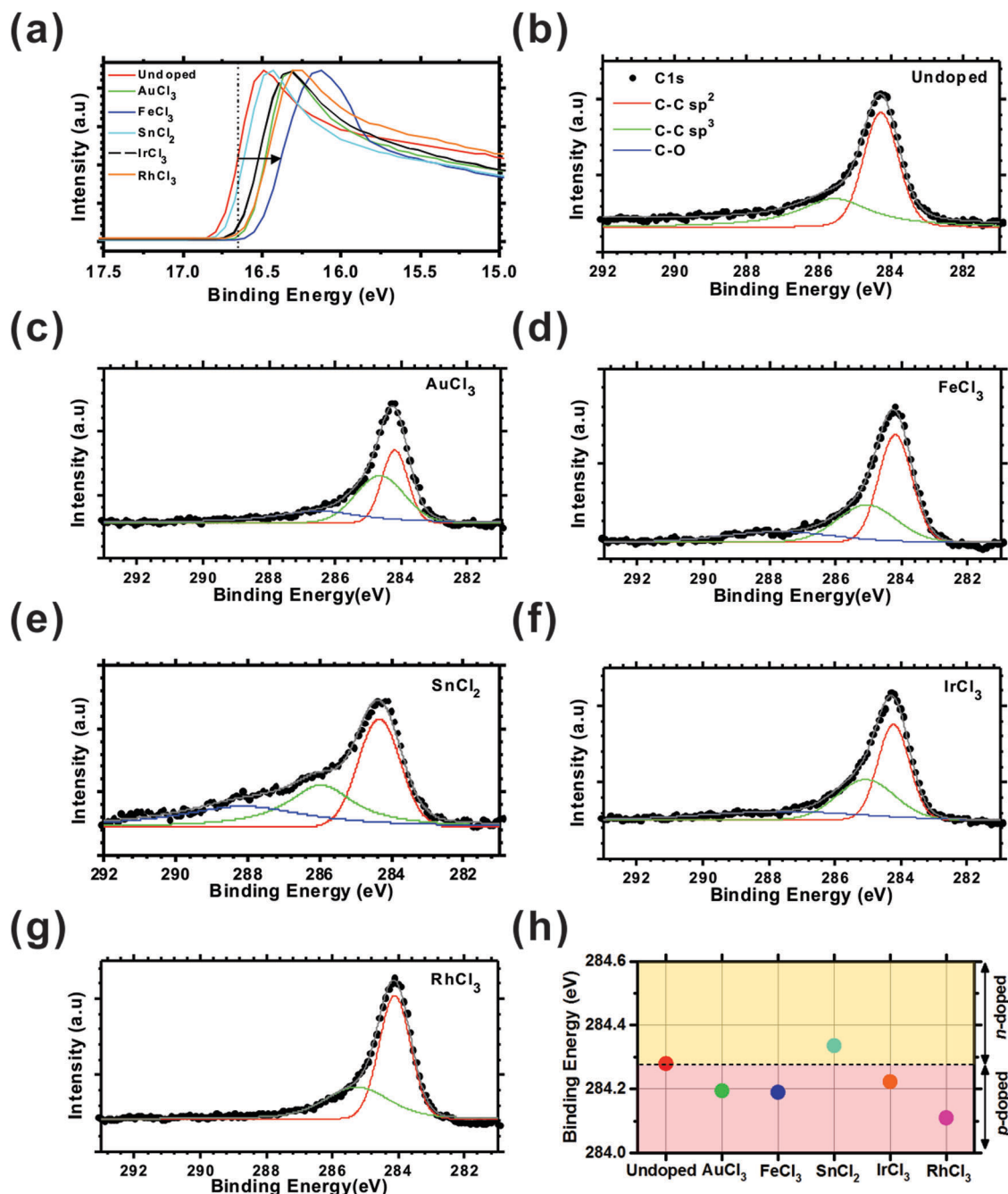


Fig. 3 (a) Secondary UPS cut-off and XPS C1s peak for (b) undoped, (c) AuCl_3 -, (d) FeCl_3 -, (e) SnCl_2 -, (f) IrCl_3 -, and (g) RhCl_3 -doped graphene. (h) C-C sp^2 peak position of the undoped and doped graphene.

UPS and XPS collectively suggest that charge transfer underpins the present chemical doping framework. However, direct evidence on the amount of charge transfer can be gleaned by Hall measurements (Fig. 4). Van der Pauw geometries were fabricated and samples measured at room temperature under a 2 kG field.

The R_s of the graphene decreased from $2148 \, \Omega \, \text{cm}^{-2}$ (undoped graphene) to $1026 \, \Omega \, \text{cm}^{-2}$ (AuCl_3), $853 \, \Omega \, \text{cm}^{-2}$ (FeCl_3), $1521 \, \Omega \, \text{cm}^{-2}$ (SnCl_2), and $1109 \, \Omega \, \text{cm}^{-2}$ (RhCl_3). The R_s reduction is consistent with our earlier macro-scale four-point probe measurements.³¹ Only IrCl_3 -doped graphene exhibited a higher

R_s ($68 \, \text{M}\Omega \, \text{cm}^{-2}$) than the undoped samples (Fig. 4(a)), which differs from other published works.^{22,28,29} This may be associated with the topological insulating behaviour of Ir on graphene^{44–46} though further investigations are ongoing to explore this.

The carrier density of undoped graphene was $2.0 \times 10^{12} \, \text{cm}^{-2}$, which is comparable to published values (9.3×10^{11} – $3.4 \times 10^{12} \, \text{cm}^{-2}$),^{43,47–50} whereas the carrier density of the doped graphene increased by roughly an order of magnitude ($8.5 \times 10^{12} \, \text{cm}^{-2}$ (SnCl_2)– $2.8 \times 10^{13} \, \text{cm}^{-2}$ (AuCl_3)). Only IrCl_3 doped graphene showed



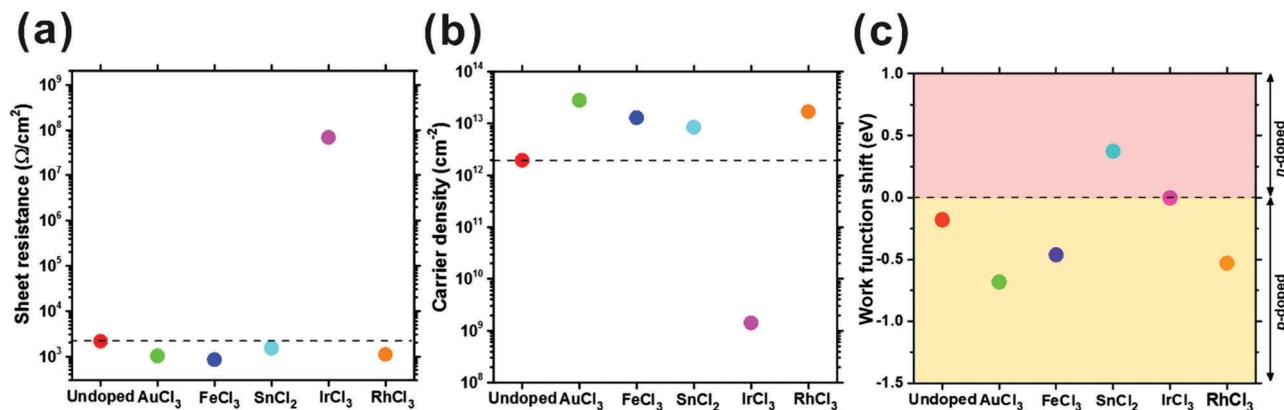


Fig. 4 Transport in undoped and Me_xCl_y-doped graphene: (a) sheet resistance (R_s), (b) carrier density, and (c) work function shift calculated with empirical carrier density values.

a lower carrier density ($1.4 \times 10^9 \text{ cm}^{-2}$). The Hall carrier mobilities for undoped and doped graphene can be obtained by⁵¹

$$\mu = \frac{\sigma V_{\text{H}} d}{IB} \quad (5)$$

where V_{H} , I , d , and B are applied Hall voltage, current, graphene thickness and applied magnetic field (2 kG), respectively. σ is the conductivity which is obtained from the measured resistivity. The calculated mobility of the undoped graphene was $1483 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, which is in the range of reported values for CVD graphene ($100\text{--}3300 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$).^{39,52} Doped graphenes showed one or half order lower mobility ($162 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ (SnCl₂)– $562 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ (FeCl₃)) than the undoped, but are close to the reported values ($200\text{--}450 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$) for nitrogen-doped CVD graphene.⁵³ This is probably due to the carrier scattering associated with various non-idealities in the transport system, such as residual charged impurities. Chen *et al.* demonstrated that graphene with charged impurities can show metallic resistance at low temperature ($< 100 \text{ K}$) where the resistance tended to increase at high temperatures ($> 100 \text{ K}$) with increasing carrier impurity-mediated scattering.⁴⁹ The work function shift of the undoped graphene was negatively shifted by 0.18 eV, indicating slight p-doping, possibly due to unavoidable doping from ambient oxygen.⁵⁴ Chemically doped graphene clearly showed a non-negligible work function shift (-0.5 eV (AuCl₃) to $+0.55 \text{ eV}$ (SnCl₂)), as shown in Fig. 4(c). The work function depends on the net charge carrier density.⁵⁵ Surface dipole formation may adjust the surface potential and hence, the work function.⁵⁵ Our Hall measurements demonstrate that the present chemical doping methodology can successfully tune the work function of graphene. Successful p-type tuning will enhance the luminous efficiency when used as an anode for advanced OLED applications or solar cells, whilst low work function tuning may be beneficial for optimised electron emission applications.

Kelvin Probe Force Microscopy (KPFM, Bruker Co.) was used to independently corroborate the measured work function shifts upon chemical doping, as well as concurrently exploring any microscale spatial variation. The surface potential of graphene was scanned over $20 \mu\text{m} \times 20 \mu\text{m}$ and the work function

(ϕ_{Graphene}) extracted (Fig. 5(a) and (b)). Undoped graphene had a work function of 4.72 eV, which was slightly higher than previously reported (4.66 eV).^{56,57} This slight increase is likely due to unintentional doping of the graphene following interaction with ambient atmosphere.^{54,58,59} The work function of the SnCl₂-doped graphene was shifted by -0.13 eV from the undoped graphene, whilst the remainder of the doped graphenes showed positive shifts in their work functions ranging from 0.07 to 0.31 eV. The areal distribution of the work function appears especially uniform with a standard deviation per unit area ranging from only 0.04 eV (RhCl₃) to 0.21 eV (FeCl₃).

In Fig. 5(c) we compare the work function shift for all doped samples obtained by KPFM, UPS, Hall measurement and Density Function Theory (DFT). DFT calculations were performed using Material Studio 7.0 and DMol³, with calculations based on a $3 \text{ nm} \times 3 \text{ nm}$ basal plane. The generalized gradient approximation with Perdew–Burke–Ernzerhof functional (GGA-PBE) was applied for the exchange–correlation function. The trend in the work function shift from the DFT models is in broad agreement with our KPFM results. However, FeCl₃, SnCl₂ and RhCl₃ all showed a $\sim 0.3 \text{ eV}$ higher work function compared to KPFM, UPS and Hall measurements. The difference is likely attributable to non-idealities associated with the graphene, its synthesis and doping; including, but not limited to, unintentional doping of the graphene upon exposure to ambient conditions and the catalyst etchant, in addition to the proximal substrate and UV adhesive. Nevertheless, the observed trends are consistent with our other empirical findings.

Achieving a high degree of charge transfer is central to increasing the doping level. This depends on two properties; the Gibbs free energy and the reduction potential of the dopants. The Gibbs free energy (ΔG) is defined as the thermodynamic potential used to calculate the maximum or reversible work that may be performed by a thermodynamic system at a constant temperature.⁶⁰ Charge transfer is endothermic ($\Delta G < 0$); it readily accelerates the reaction when ΔG is increasingly negative, as illustrated Fig. 5(d). The metal ions (Me^{3+}) in the dopants are increasingly likely to achieve neutrality (Me^0) with larger $|\Delta G|$ by positive charge transfer to the graphene, resulting in a larger



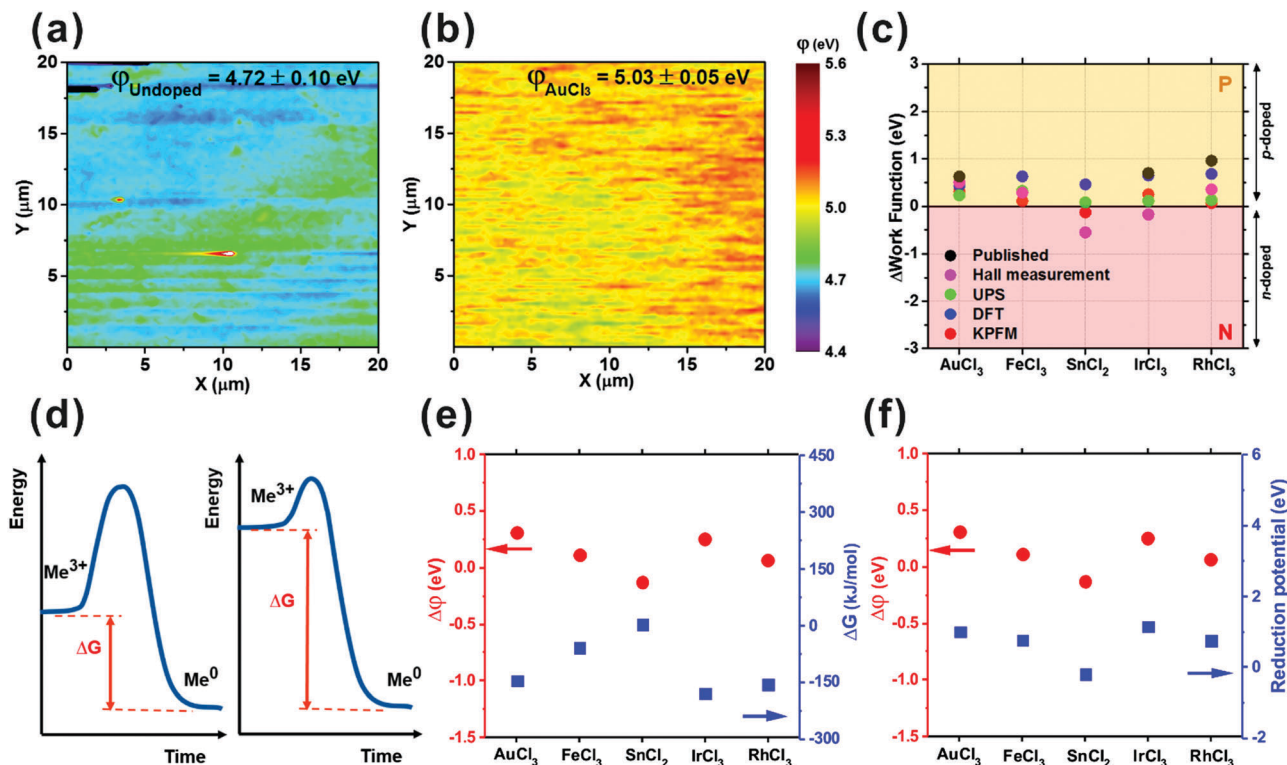


Fig. 5 KPFM maps of (a) undoped and (b) AuCl_3 -doped graphene, (c) comparison of measured work function and published data,^{21,22,28,56} (d) an example plot of the charge transfer reaction of Me_xCl_y -doped CVD graphene and comparison of work function shift values measured by KPDM with (e) Gibb's free energy and (f) reduction potential.

work function shift. The Gibb's free energy values of the constituent metal atoms for each dopant are compared with the empirically assessed work function values (KPDM) in Fig. 5(e).⁶¹ A mirrored trend in the work function values in relation to ΔG is evident. A large work function shift is observed in AuCl_3 (0.31 eV) and IrCl_3 -doped graphene (0.25 eV) with both having large, negative ΔG (AuCl_3 : $-145.7 \text{ kJ mol}^{-1}$ and IrCl_3 : $-179.9 \text{ kJ mol}^{-1}$).⁶¹

Another influential parameter in driving the rate and direction of charge transfer in the present 2D system is the reduction potential.¹¹ As seen in Fig. 5(f), the work function of the doped graphene shows a similar trend to that of the reduction potential of the metallic ions. The reduction reaction mediates p-type doping in graphene, leaving bound Me^0 . If the reduction potential is high, the rate of charge transfer increases resulting in a large increase in the work function. For n-type doping from SnCl_2 , the reduction potential during oxidation by Sn^{2+} is -0.19 V ⁶¹ indicating that it has a tendency to donate electrons to graphene, whilst the other dopants tend to accept electrons with the positive potentials (AuCl_3 : 1.00 V, FeCl_3 : 0.77 V, IrCl_3 : 1.16 V, and RhCl_3 : 0.76 V).⁶¹ We thus anticipate that the graphene doped with AuCl_3 , FeCl_3 , IrCl_3 and RhCl_3 show nominally p-type behaviour, whereas graphene doped with SnCl_2 would exhibit nominally n-type behaviour, which our empirical findings corroborate.

Variable Range Hopping (VRH) has been widely implicated as a dominant transport mechanism in such 1D and 2D ensemble nanomaterials.^{62,63} VRH is conduction by phonon-assisted

quantum tunnelling between disordered localised states near E_F , where the resistance (R) can be expressed as a function of temperature, T as:^{63,64}

$$R(T)^{-1} = R_0^{-1} \exp \left[- \left(\frac{T_0}{T} \right)^{\frac{1}{p}} \right] \quad (6)$$

where T_0 is the tunnelling parameter and $p = (d + 1)$, where d is the dimensionality of hopping. Tunnelling transport is characterised by Mott's 3D VRH, Mott's 2D VRH and Efros/Shklovskii (ES) VRH with $\exp(-T^{1/4})$, $\exp(-T^{1/3})$ and $\exp(-T^{1/2})$ dependence, respectively.⁶⁵ At low temperatures, the density of states in a partially ordered system is not constant and is largely affected by Coulombic interactions between hopping charge carriers. Thus, ES VRH can be observed only at low temperatures ($< 10 \text{ K}$) where the thermal potential is insufficient to overcome such transport phenomena.⁶⁶

VRH has been considered elsewhere to explain transport in undoped graphene with Fermi level (E_F) at the Dirac point.⁶³ However, work function shift achieved by chemical doping means that E_F can clearly be adjusted by chemical doping to provide otherwise different transport regimes, which needs another transport model rather than VRH. To investigate conduction in our doped graphene, the current-voltage (I - V) characteristics were measured as a function of temperature from 77 K to room temperature (297 K) using a Keithley 4200SCS SMU. Cr/Au (10/80 nm) contacts were thermally evaporated and defined



via shadow masking of circular transmission lines (Fig. 6(a)). A range of channel lengths was considered ($L = 50\text{--}1000\ \mu\text{m}$), granting us simultaneous empirical access to the contact resistance. The differential resistance, $R = (dV/dI)_{V=0}$ was calculated for all temperatures. Undoped graphene shows an exponentially decreasing resistivity with temperature, over the measured temperature range, but other doped graphene shows increasing resistivity as the temperature rose above $\sim 110\ \text{K}$, as shown in Fig. 6(c); an effect that is not well captured using standard VRH-like models.

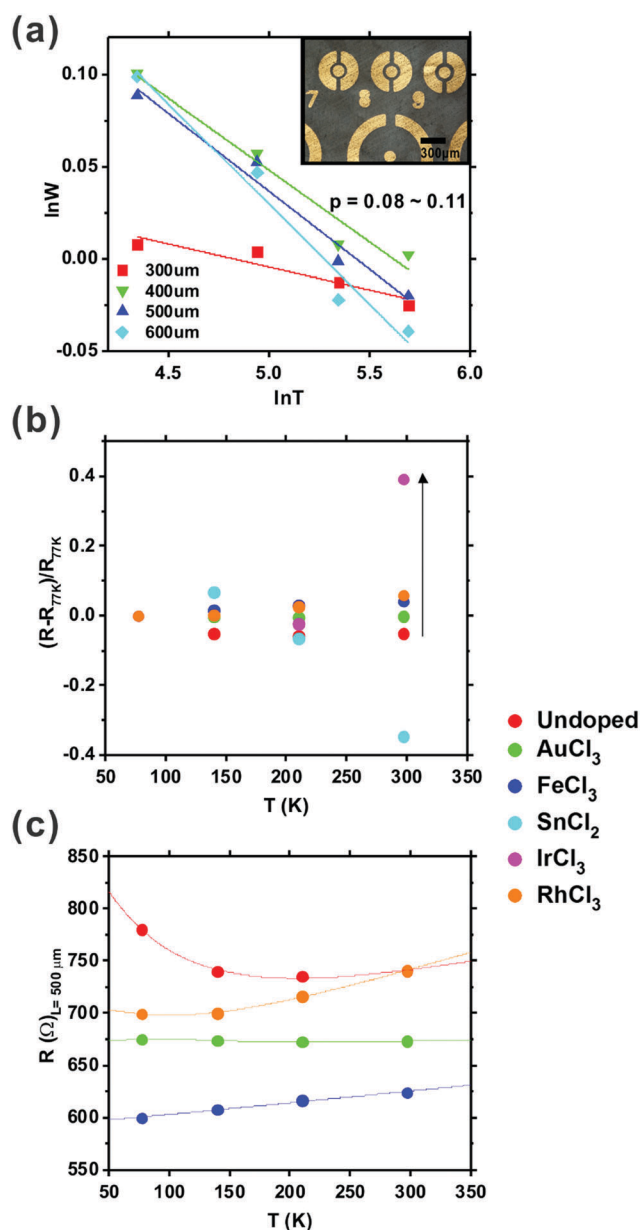


Fig. 6 (a) A logarithmic derivative of the resistance calculated from the current–voltage (I – V) measured at various temperatures. Inset: Microscopic image of the Cr/Au circular transmission lines on the PET-supported CVD graphene. (b) The normalised resistance change with increasing temperature, and (c) example differential resistance calculated from the measured I – V ($L = 300\ \mu\text{m}$).

The Zhabrodsii's Reduced Activation Method (RAM)⁶⁷ was used in order to determine the functional form of the T -dependent resistivity of VRH conduction. In this way the reduced activation energy ($W = T^{-1} \partial \ln R / \partial T^{-1}$) can be expressed as $W(T) = A + x(T_0/T)^p$ by substituting the VRH model, $R(T)^{-1} = R_0^{-1} \exp[-(T/T_0)]^{\frac{1}{p}}$, we have

$$\ln W(T) \approx \ln(x(T_0)^p) - p \ln(T) \quad (7)$$

Using eqn (7) we find the resistance of the AuCl_3 -doped graphene (Fig. 6(a)). From the fitting, the extracted exponent, p , was $0.08\text{--}0.11$ which is related to neither Mott VRH nor ES VRH. Such exponent values lie outside of the VRH range ($0.25\text{--}0.33$) suggesting that doping forces the transport to deviate from conventional VRH transport in such 2D systems. To recognise the T -dependent R trend modified by chemical doping, the change in resistance from the initial resistance at $77\ \text{K}$ ($R - R_{77\text{K}}$) was calculated (Fig. 6(b)). The undoped graphene shows a decreasing R ($R - R_{77\text{K}} < 0$), but the R of the doped graphene increases ($R - R_{77\text{K}} > 0$) with increasing T . Chemical doping reduces the resistance but the resistance increases slightly due to the increasing T . The doped graphene has a larger R increase than the undoped graphene (at room temperature), as highlighted in Fig. 6(b). Evidently an augmented model is required to better explain such results.

Park *et al.* suggested a three-termed transport model for the temperature-dependent conduction in polycrystalline graphene.⁶⁷ Adding Thermal Activation (TA) and Nearest Neighbour Hopping (NNH) to the existing VRH model, this composite modelled can be expressed as;⁶⁷

$$R(T)^{-1} = R_{\text{TA}}^{-1} \exp\left(-\frac{E_a}{k_B T}\right) + R_{\text{NNH}}^{-1} \exp\left(-\frac{E_n}{k_B T}\right) + R_{\text{VRH}}^{-1} \exp\left[-\left(\frac{T_0}{T}\right)^{\frac{1}{p}}\right] \quad (8)$$

where E_a is the thermal activation energy, k_B is the Boltzmann constant, E_n is the NNH activation energy, and R_{TA} , R_{NNH} , and R_{VRH} are the TA, NNH, and VRH constants, respectively. Here the grain boundaries function as metallic sub-nano wires with tunnelling between neighbouring metallic wires dominated by NNH. Charge puddling can also help charge carriers tunnel to neighbouring puddles, in a NNH process, along trapped states.⁶⁸ The thermally activated carriers can hop readily *via* the assistance of metallic wires or linear puddling. Similarly, in our study, such conductive line defects were observed by SSRM. Our findings suggest that edge-mediated conduction in doped graphene can be enhanced by preferential aggregation of dopant molecules. The Park model matched our findings well for $T < 140\ \text{K}$ ($R^2 = 0.94$), but the increasing resistance for $T > 140\ \text{K}$ did not match the model as the NNH component shows only an exponential decay. In such a model, a term is lacking to explain the increasing trend observed in our experimental data.

The conduction models outlined above focus on undoped graphene with E_F at the Dirac point. There was no consideration for highly doped graphene with high carrier density showing

metallic-like behaviour, as our empirical findings suggest. For such quasi-metallic behaviour, resistance may arise from electron-phonon scattering or electron-defect scattering, especially at elevated temperatures. We propose a model that explains the electrical conduction of doped graphene for the entirety of the temperature range considered. By introducing an increasing resistance term at high temperatures, associated with phonon coupling effects, we find a temperature dependency of the form;

$$R(T)^{-1} = R_{TA}^{-1} \exp\left(-\frac{E_a}{k_B T}\right) + R_{NNH}^{-1} \exp\left(-\frac{E_n}{k_B T}\right) + R_{VRH}^{-1} \exp\left[-\left(\frac{T_v}{T}\right)^{\frac{1}{3}}\right] + R_M^{-1} [1 + \alpha(T - T_0)] \quad (9)$$

where E_n is the NNH activation parameter, T_v is a tunnelling parameter, α is the temperature coefficient, T_0 is the reference temperature at which the resistance linearly increases with increasing T , and, R_M is the resistance at T_0 .⁶⁹ The R vs. T curves of undoped and doped graphenes were successfully fitted with the suggested model across the entire temperature range considered, as illustrated (solid lines) in Fig. 6(c). For $T > 140$ K, the curves from the doped graphene samples are well matched ($R^2 = 0.97$ (AuCl₃) – 0.99 (RhCl₃)) with the proposed model with higher α (1.1×10^{-3} (AuCl₃)– 9.4×10^{-3} (RhCl₃)) than the undoped (7.1×10^{-4}) indicating that doped graphene has a large number of charge carriers experiencing notable scattering, indicative of diffusive transport in a principally metallic system. Diffusive transport is known to be readily affected by the local scattering of carriers. The scattering of carriers by the substrate- and interface-mediated phonons might well contribute to the increase in R at high temperatures, though additional studies are required to further explore this. The charged impurities

are also the cause of carrier scattering as discussed above.⁴⁹ The external phonon contribution and impurities are strongly dependent on charge carrier density.⁷⁰ Transport in chemically doped graphene at high temperatures appears to be principally diffusive, and is readily influenced by both lattice and substrate scattering.

As described in Fig. 7, there are various potential barriers within the transferred graphene, though such said barriers are commonly low and easily overcome. Grain boundaries and defect states manifest other transport barriers which are increasingly opaque to transmission *via* hopping. For E_F at the Dirac point in pristine, theoretically undoped graphene, the electron's become increasingly localised with high barriers at defects. It has been shown elsewhere that the grain boundary barrier height is inversely proportional to the doping level;⁷¹ the barrier height is lowered by the increasing doping level. Conversely, n-doping shifts E_F to higher potentials therein reducing the effective barrier height; the wave function is extended and electrons move increasingly freely, as described in Fig. 7(b) and (c). Although the electron population may be localised between any given defect there remains a finite probability that tunnelling may occur *via* NNH. The barriers for holes are in the opposite direction upon p-doping and a similar transport mechanism can be applied.

At high temperatures, charge carriers in the polymer-supported doped graphene appear to move with little barrier interference, enabling a low macroscale resistance, but a large number of extended carriers experience scattering from the substrate or other carriers. Upon doping, the possibility of carrier scattering is increased due to the increased number of carriers. VRH is dominant in pristine graphene, whilst doped graphene lies on the border between NNH and diffusive transport. Practically, measuring the T -dependent R of our undoped graphene we note a combination of VRH and NNH. The findings

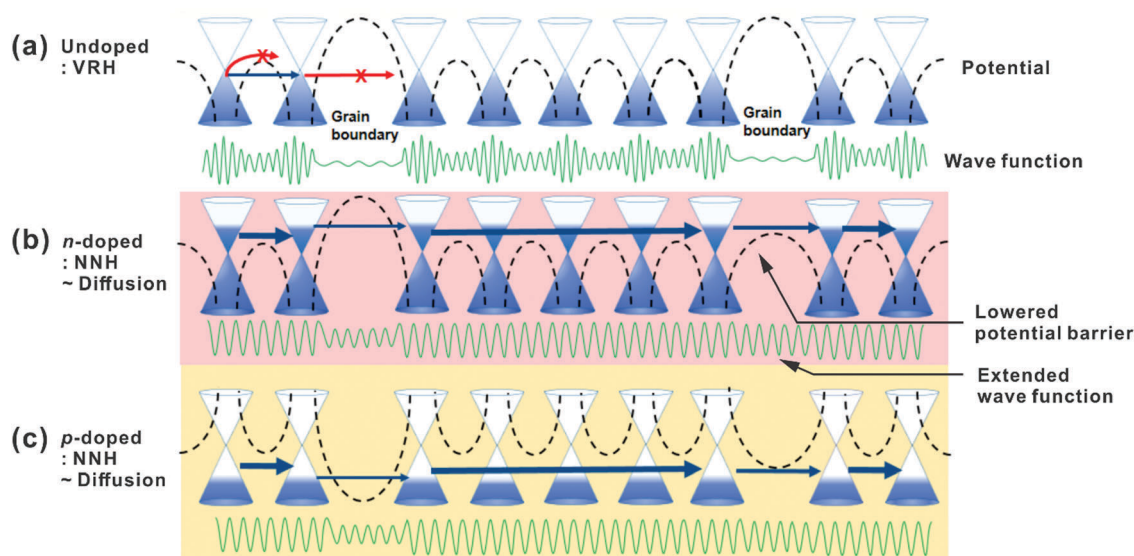


Fig. 7 Transport in (a) pristine graphene, (b) n-doped, (c) p-doped graphene as a function of lattice and substrate induced potential barriers, such as grain boundaries.



clarify that the electron transport of graphene can be extended rather than localised by chemical doping even at low temperatures, whilst scattering at high temperatures limits the absolute reduction in resistance. The elucidated transport mechanism is crucial for the control of work function and resistance of graphene by chemical doping, which underpins future graphene-based transparent flexible electronics devices.

Conclusion

Here we suggest a possible chemical doping scheme to augment charge transport in chemically-vapour deposited (CVD) graphene transferred to optically transparent and mechanically flexible polymer substrates. By conducting broad metrology studies, charge transport was studied before and after metal chloride doping with AuCl_3 , FeCl_3 , SnCl_2 , IrCl_3 and RhCl_3 . Efficient charge transfer mediated by our doping scheme allows for accurate concurrent control over the resistance ($3.5 \text{ k}\Omega$ to $0.93 \text{ }\Omega$) and work function (-0.13 eV (SnCl_2) \sim 0.31 eV (AuCl_3)). The agglomerates observed on graphene are generated from the neutral atoms whose propensity towards aggregation is principally dictated by their specific cohesive energy. SSRM mapping highlighted that dopant species tended to accrue at atomic facets and boundaries resulting in dominant NNH and diffusive transport which we independently verified through transport studies. Charge transport was investigated at various temperatures with Mott 2D VRH and NNH dominant in undoped graphene whilst NNH and diffusive transport were shown to play a major role in the doped graphenes with such contact interface nano-engineering which makes properly aligned work functions and improving electrical conductance between two inhomogeneous material and the demonstrable control over the work function by chemical doping. The presented work function tuning suggests that graphene is an ideal material for applications where work function optimisation is functionally essential in a new generation of transparent flexible electronics – such as in electron field emission applications, carrier injection layers in OLED and solar cells – as well as highlighting one possible means of advanced contact interface engineering.

Conflicts of interest

There are no conflicts to declare.

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