

Cite this: *RSC Adv.*, 2017, 7, 55748

DOI: 10.1039/c7ra90111a

[www.rsc.org/advances](http://www.rsc.org/advances)

## Correction: Removal of Cu(II) from aqueous solution using Fe<sub>3</sub>O<sub>4</sub>–alginate modified biochar microspheres

Changjiang Yu,<sup>ab</sup> Miao Wang,<sup>b</sup> Xinyu Dong,<sup>b</sup> Zaifeng Shi,<sup>b</sup> Xiaopeng Zhang<sup>b</sup> and Qiang Lin<sup>\*ab</sup>Correction for 'Removal of Cu(II) from aqueous solution using Fe<sub>3</sub>O<sub>4</sub>–alginate modified biochar microspheres' by Changjiang Yu *et al.*, *RSC Adv.*, 2017, 7, 53135–53144.

The authors regret that there were some errors in the XRD analysis for SiO<sub>2</sub> in the original article (p. 53142, Section 3.6.4). The two sentences beginning 'The diffraction peaks at  $2\theta = 22.0^\circ$  and  $42.0^\circ$ ...' are corrected as follows:

"The diffraction peaks at  $2\theta = 20.9^\circ$ ,  $40.3^\circ$  and  $42.5^\circ$  are assigned to the (1 0 0), (1 1 1) and (2 0 0) planes, respectively. The angular position of the diffraction line corresponded to the amorphous structure SiO<sub>2</sub>, in accordance with the database of the SiO<sub>2</sub> standard card (JCPDS no. 65-0466)".

The correspondingly updated Fig. 11 is presented below.

In addition, there was an error in the variable name of the right-hand y-axis of Fig. 4. The updated figure, in which 'DTG' has been revised to 'Derivative weight' is presented below.

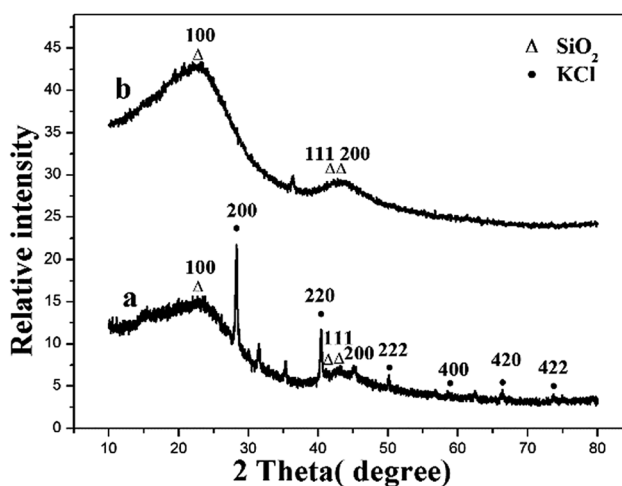


Fig. 11 XRD pattern of biochar before (a) and after (b) Cu(II) adsorption.

<sup>a</sup>Faculty of Environmental Science and Engineering, Kunming University of Science and Technology, No. 68 Wenchang Road, Kunming 650093, China. E-mail: [linqianggroup@163.com](mailto:linqianggroup@163.com)

<sup>b</sup>Key Laboratory of Water Pollution Treatment & Resource Reuse of Hainan Province, College of Chemistry and Chemical Engineering, Hainan Normal University, No. 99 Longkunnan Road, Haikou 571158, China

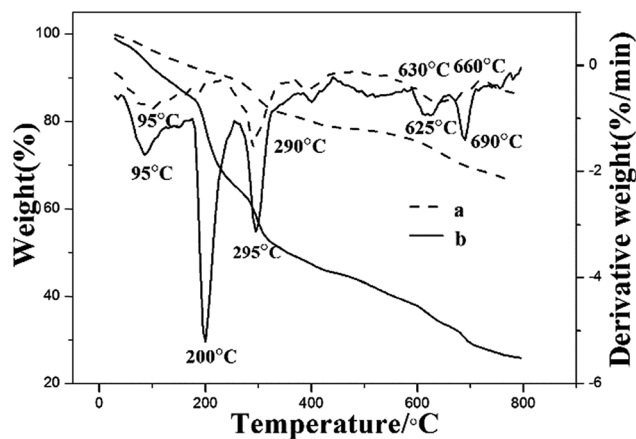


Fig. 4 TG and DTG curves of magnetic microsphere (a) and calcium alginate (b).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

