



# Synthesis of isophthalates from methyl coumalate†

Cite this: *RSC Adv.*, 2017, 7, 56760

George A. Kraus \* and Shuai Wang

Received 30th November 2017  
 Accepted 12th December 2017

DOI: 10.1039/c7ra12935a

[rsc.li/rsc-advances](http://rsc.li/rsc-advances)

Benzoic acid and isophthalic acid are prepared as commodity chemicals by the benzylic oxidation of toluene or xylene using a cobalt or manganese catalyst.<sup>1</sup> Substituted isophthalates are prepared by a variety of methods but benzylic oxidation is a major pathway for 4-phenyl- and 4-methylisophthalic acids.<sup>2</sup> We recently reported the synthesis of *para*-substituted benzoic acids by the reaction of coumalic acid (generated in one step from malic acid) with alkenes.<sup>3</sup> This reaction also proceeds well with methyl coumalate to provide the corresponding esters. The mechanism involves an *in situ* generated bicyclic lactone which loses carbon dioxide and undergoes dehydrogenation (for X = H) or loses an alcohol (for X = ether), providing the aromatic ester, as illustrated in Scheme 1.

With enol ethers the bicyclic lactone from methyl coumalate can be isolated if the reaction temperature is kept below 100 °C. A number of researchers have generated adducts of vinyl ethers with coumalates and have transformed them into substituted cyclohexenes.<sup>4,5</sup> Posner and coworkers recently generated complex cyclitols in high yield from an adduct of a pyrone and a vinyl ether.<sup>6</sup> Okura and coworkers recently reported cycloadditions using pyrone 3,5-dicarboxylates.<sup>7</sup>

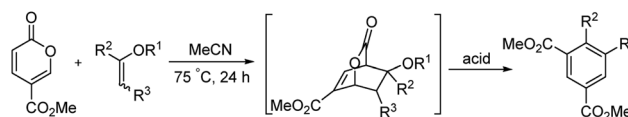
The ready availability of these bicyclic lactones offered the opportunity to open the lactone under acidic or basic conditions to produce isophthalates. This strategy utilizes all of the carbons of methyl coumalate, increasing the atom economy of

the transformation. We report herein that the reaction of the bicyclic lactone intermediates with *para*-toluenesulfonic acid (PTSA) in boiling methanol afforded very good to excellent yields of the diesters of isophthalic acids. This process is shown below in Scheme 2.

Initially, we studied a number of acids and bases to determine the optimal conditions for isophthalates formation. These results are collated in Table 1. Our best conditions involved five mole percent of PTSA in boiling methanol.

Examples are illustrated below in Scheme 3. The yields over the two-step sequence are very good. Because methanol is employed as the solvent, the dimethyl esters predominate.

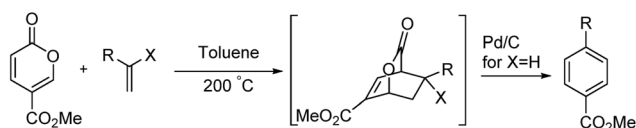
As the results above demonstrate, both cyclic and acyclic vinyl ethers react readily with methyl coumalate. Aromatization



Scheme 2 The reaction with enol ethers to form isophthalates.

Table 1 Optimization of the aromatization reaction

Entry	Cat (5 mol%)	T (°C)	Yield (%)
1	H <sub>2</sub> SO <sub>4</sub>	65	48
2	CF <sub>3</sub> COOH	65	29
3	PTSA	65	82
4	NaOH	65	43
5	K <sub>2</sub> CO <sub>3</sub>	65	57
6	NaOMe	65	65
7	PTSA	45	47

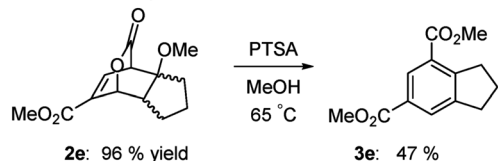
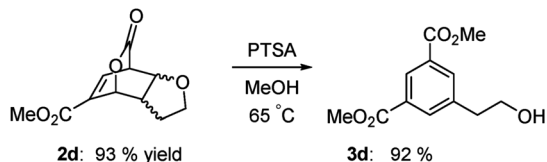
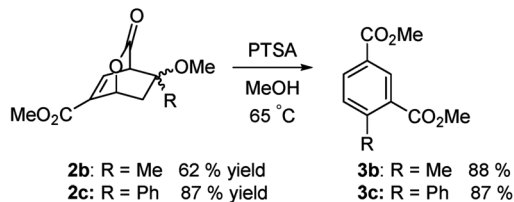
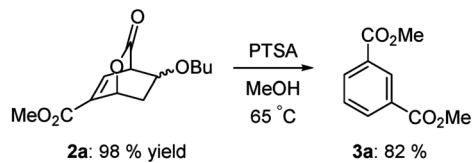


Scheme 1 The reaction of methyl coumalate with alkenes.

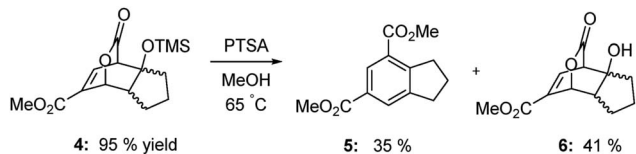
Department of Chemistry, Iowa State University, Ames, IA 50011, USA. E-mail: [gakraus@iastate.edu](mailto:gakraus@iastate.edu)

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c7ra12935a





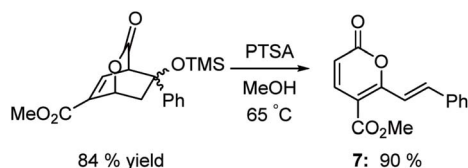
Scheme 3 Isophthalates from lactone intermediates.



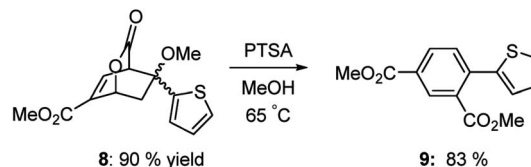
Scheme 4 Isophthalates from silyloxy lactones.

with PTSA in methanol gives good yields of isophthalates. In the case of **3a**, 15% of the product was the butyl methyl isophthalate. The corresponding enol silyl ethers also formed stable adducts with methyl coumalate; however, the adducts do not undergo high-yield aromatization to afford the isophthalates. As shown below in Scheme 4, the adduct from the reaction of methyl coumalate with the enol silyl ether of cyclopentanone (**4**) afforded two main products. Apparently, rapid removal of the silyl ether group in methanol yields an alcohol **6** that undergoes some fragmentation in addition to aromatization.

Support for this hypothesis is shown in Scheme 5. The adduct of the enol silyl ether of acetophenone reacted with PTSA



Scheme 5 Rearrangement of silyloxy lactone.



Scheme 6 Preparation of thiophene substituted isophthalate.

in methanol to afford a product in 90% yield that was clearly not an isophthalate. After examination of the NMR, IR and mass spectrum, we assigned the structure to be pyrone **7** shown in Scheme 5. The mechanism of formation is not clear, but likely involves a retro-aldol reaction and an intramolecular hydride shift followed by loss of water.

Heterocyclic systems also participate effectively in this transformation. Scheme 6 shows the preparation of thiophene substituted isophthalate **9** in very good overall yield.

## Conclusions

The two-step Diels–Alder/aromatization pathway to substituted isophthalates proceeds in high overall yields and is operationally convenient. A variety of alkyl, aryl and heteroaryl isophthalates can be prepared. Adducts of alkyl vinyl ethers afford higher yields of isophthalates than adducts of enol silyl ethers.

## Experimental section

### General procedure for D–A reaction of methyl coumalate with enol ethers

Methyl coumalate (154 mg, 1.0 mmol, 1.0 equiv.), enol ether (3.0 mmol, 3.0 equiv.) and MeCN (1.0 mL) were added to a 15 mL sealed tube. The tube was heated in a 75 °C oil bath and stirred for 24 hours. After the reaction, the solvent was removed under reduced pressure and the residue was purified by silica gel chromatography.

**Methyl-8-butoxy-3-oxo-2-oxabicyclo[2.2.2]oct-5-ene-6-carboxylate (2a).** Colorless oil; 249 mg, yield = 98%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 7.20–7.17 (m, 1H), 5.68–5.65 (m, 1H), 4.09 (m, 1H), 4.06–4.01 (m, 1H), 3.79 (s, 3H), 3.43 (ddd, *J* = 8.3, 5.4, 1.8, 1H), 3.35 (ddd, *J* = 9.0, 5.4, 1.8, 1H), 2.60 (dddd, *J* = 13.3, 7.8, 3.8, 1.5, 1H), 1.59 (d, *J* = 1.9, 1H), 1.46 (pd, *J* = 6.8, 3.8, 2H), 1.28 (tdd, *J* = 14.6, 7.3, 1.6, 2H), 0.87 (td, *J* = 7.3, 1.6, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ = 170.75, 162.64, 138.43, 135.78, 73.41, 71.44, 69.33, 52.30, 47.63, 34.93, 31.64, 19.28, 13.87. HRMS (ESI-QTOF) calcd for [M + H<sup>+</sup>]: 255.1227, found: 255.1226.

**Methyl-8-methoxy-8-methyl-3-oxo-2-oxabicyclo[2.2.2]oct-5-ene-6-carboxylate (2b).** Colorless oil; 140 mg, yield = 62%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 7.22 (dd, *J* = 6.2, 2.1, 1H), 5.60 (m, 1H), 3.81 (d, *J* = 6.1, 1H), 3.76 (s, 3H), 3.10 (s, 3H), 2.15 (dd, *J* = 14.0, 3.6, 1H), 1.84 (dd, *J* = 14.2, 1.8, 1H), 1.42 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ = 171.19, 162.70, 139.57, 135.00, 75.66, 73.58, 52.95, 52.21, 50.19, 40.90, 23.97. HRMS (ESI-QTOF) calcd for [M + H<sup>+</sup>]: 227.0914, found: 227.0911.

**Methyl-8-methoxy-3-oxo-8-phenyl-2-oxabicyclo[2.2.2]oct-5-ene-6-carboxylate (2c).** White solid; 250 mg, yield = 87%. <sup>1</sup>H NMR



(400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.45–7.30 (m, 3H), 7.24 (m, 2H), 7.16 (m, 1H), 5.78 (m, 1H), 4.24 (d,  $J$  = 6.5, 1H), 3.76 (s, 3H), 3.03 (s, 3H), 2.77 (dd,  $J$  = 14.2, 4.1, 1H), 2.31 (d,  $J$  = 14.3, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 170.22, 162.52, 140.17, 138.09, 137.68, 129.02, 128.62, 126.90, 81.19, 73.82, 52.43, 52.20, 51.44, 39.70. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 289.1071, found: 289.1076.

**Methyl-8-oxo-2,3,3a,4,7,7a-hexahydro-4,7-(epoxymethano)benzo furan-5-carboxylate (2d).** White solid; 208 mg, yield = 93%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.21 (dd,  $J$  = 6.7, 2.0, 1H), 5.52 (m, 1H), 4.30 (dd,  $J$  = 8.7, 3.5, 1H), 4.11 (tdd,  $J$  = 8.6, 4.1, 1.9, 1H), 3.92 (ddt,  $J$  = 7.1, 3.6, 1.7, 1H), 3.87–3.80 (m, 1H), 3.79 (s, 3H), 2.66 (dddd,  $J$  = 9.3, 7.9, 6.3, 1.6, 1H), 2.23–2.00 (m, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 169.98, 162.52, 139.14, 138.27, 77.90, 76.50, 71.24, 52.47, 48.80, 43.39, 27.44. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 225.0757, found: 225.0758.

**Methyl-7a-methoxy-8-oxo-2,3,3a,4,7,7a-hexahydro-1H-4,7-(epoxy methano)indene-5-carboxylate (2e).** White solid; 242 mg, yield = 96%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.23 (m, 1H), 5.46 (m, 1H), 3.99 (d,  $J$  = 6.1, 1H), 3.79 (s, 3H), 3.17 (s, 3H), 2.16 (td,  $J$  = 7.7, 6.7, 1.7, 1H), 2.05 (dq,  $J$  = 9.3, 4.7, 3.5, 2H), 1.79 (dtd,  $J$  = 11.7, 6.2, 3.5, 2H), 1.72–1.60 (m, 1H), 1.53 (ddd,  $J$  = 13.8, 8.8, 6.8, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 171.31, 162.93, 139.56, 136.62, 88.78, 77.21, 52.33, 52.16, 51.28, 50.90, 33.05, 27.66, 27.20. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 253.1071, found: 253.1068.

**Methyl-8-oxo-7a-((trimethylsilyloxy)-2,3,3a,4,7,7a-hexahydro-1H-4,7-(epoxymethano)indene-5-carboxylate (4).** White solid; 294 mg, yield = 95%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.22 (dd,  $J$  = 5.0, 0.9, 1H), 5.43–5.41 (m, 1H), 3.80 (s, 3H), 3.78 (d,  $J$  = 6.3, 1H), 2.17 (dd,  $J$  = 8.8, 6.6, 1H), 2.11–2.01 (m, 1H), 1.98–1.88 (m, 1H), 1.86–1.68 (m, 3H), 1.61 (m, 1H), 0.09 (s, 9H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 171.43, 163.04, 140.94, 136.12, 85.74, 77.44, 55.75, 53.24, 52.34, 39.70, 27.95, 27.05, 2.05. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 311.1309, found: 311.1316.

**Methyl-3-oxo-8-phenyl-8-((trimethylsilyloxy)-2-oxabicyclo[2.2.2]oct-5-ene-6-carboxylate.** White solid; 290 mg, yield = 84%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.50–7.43 (m, 2H), 7.35 (m, 3H), 7.31–7.25 (m, 1H), 5.80 (m, 1H), 3.96 (dd,  $J$  = 6.4, 0.7, 1H), 3.84 (s, 3H), 3.02 (d,  $J$  = 3.8, 1H), 2.33–2.19 (m, 1H), –0.08 (s, 9H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 170.34, 162.93, 144.06, 141.38, 135.43, 128.64, 128.28, 125.82, 77.37, 73.57, 57.73, 52.41, 43.30, 1.64. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 347.1309, found: 347.1312.

**Methyl 8-methoxy-3-oxo-8-(thiophen-2-yl)-2-oxabicyclo[2.2.2]oct-5-ene-6-carboxylate (8).** Oil; 263 mg, yield = 90%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.30 (dd,  $J$  = 5.1, 1.2, 1H), 7.09 (dd,  $J$  = 6.6, 2.3, 1H), 6.91 (dd,  $J$  = 5.1, 3.6, 1H), 6.86 (dd,  $J$  = 3.6, 1.2, 1H), 5.80–5.68 (m, 1H), 4.18 (d,  $J$  = 6.6, 1H), 3.73 (s, 3H), 3.07 (s, 3H), 2.72 (dd,  $J$  = 14.2, 4.0, 1H), 2.27 (dd,  $J$  = 14.2, 1.6, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 169.43, 162.36, 144.83, 138.15, 137.14, 126.92, 126.79, 126.41, 78.79, 73.59, 53.99, 52.40, 51.40, 39.36. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 295.0635, found: 295.0637.

### General procedure for the formation of isophthalates

A solution of 0.2 mmol of bicyclic adduct, one crystal of PTSA and 1.0 mL of methanol was boiled for 18 hours. The solvent

was removed under reduced pressure and the residue was purified by silica gel chromatography.

**Dimethyl isophthalate (3a).** White solid; 37.5 mg, yield = 82%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.68 (q,  $J$  = 1.6, 1H), 8.22 (dt,  $J$  = 7.8, 1.5, 2H), 7.54–7.47 (m, 1H), 3.94 (s, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 166.38, 133.94, 130.84, 130.70, 128.76, 52.52. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 195.0652, found: 195.0650.

**Dimethyl 4-methylisophthalate (3b).** White solid; 36.2 mg, yield = 88%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.55 (d,  $J$  = 1.9, 1H), 8.02 (dd,  $J$  = 8.1, 1.8, 1H), 7.30 (d,  $J$  = 8.0, 1H), 3.91 (s, 3H), 3.90 (s, 3H), 2.64 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 167.26, 166.42, 145.68, 132.73, 132.07, 131.99, 129.84, 128.06, 52.31, 52.15, 22.00. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 209.0808, found: 209.0808.

**Dimethyl [1,1'-biphenyl]-2,4-dicarboxylate (3c).** White solid; 46.9 mg, yield = 87%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.49 (d,  $J$  = 1.8, 1H), 8.17 (dd,  $J$  = 8.0, 1.8, 1H), 7.46 (d,  $J$  = 8.0, 1H), 7.42–7.36 (m, 3H), 7.34–7.25 (m, 2H), 3.95 (s, 3H), 3.67 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 168.29, 166.16, 146.84, 140.27, 132.12, 131.18, 131.07, 129.14, 128.27, 128.22, 127.96, 52.46, 52.29. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 271.0965, found: 271.0967.

**Dimethyl 5-(2-hydroxyethyl)isophthalate (3d).** White solid; 43.4 mg, yield = 92%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.50 (q,  $J$  = 1.5, 1H), 8.08 (t,  $J$  = 1.3, 2H), 3.91 (s, 6H), 3.90 (t,  $J$  = 6.5, 2H), 2.95 (t,  $J$  = 6.5, 2H), 1.93 (s, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 166.41, 139.95, 134.50, 130.78, 128.93, 63.15, 52.49, 38.74. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 239.0914, found: 239.0912.

**Dimethyl 2,3-dihydro-1H-indene-4,6-dicarboxylate (3e).** White solid; 21.7 mg, yield = 47%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.52–8.37 (m, 1H), 8.03 (t,  $J$  = 1.4, 1H), 3.91 (s, 3H), 3.91 (s, 3H), 3.31 (t,  $J$  = 7.6, 2H), 2.96 (t,  $J$  = 7.6, 2H), 2.12 (p,  $J$  = 7.4, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 166.93, 166.82, 152.32, 146.65, 129.89, 129.26, 128.66, 126.62, 52.37, 52.14, 34.20, 32.43, 25.08. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 235.0965, found: 235.0965.

**Methyl (E)-2-oxo-6-styryl-2H-pyran-5-carboxylate (7).** Yellow solid; 46.0 mg, yield = 90%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 7.88–7.84 (m, 2H), 7.59 (d,  $J$  = 7.3, 1H), 7.48 (d,  $J$  = 15.8, 1H), 7.47 (m, 3H), 7.08 (d,  $J$  = 15.8, 1H), 6.79 (d,  $J$  = 7.2, 1H), 3.80 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 167.84, 161.55, 159.80, 144.94, 138.51, 131.65, 130.77, 129.22, 126.01, 121.88, 119.89, 101.79, 51.93. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 257.0808, found: 257.0810.

**Dimethyl 4-(thiophen-2-yl)isophthalate (9).** White solid; 43.6 mg, yield = 83%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 8.36 (t,  $J$  = 2.3, 1H), 8.22–8.01 (m, 1H), 7.56 (dd,  $J$  = 8.1, 2.3, 1H), 7.46–7.34 (m, 1H), 7.20–6.92 (m, 2H), 3.94 (s, 3H), 3.78 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  = 168.52, 166.01, 140.87, 138.43, 131.92, 131.84, 131.31, 130.87, 129.37, 127.67, 127.26, 127.14, 52.67, 52.58. HRMS (ESI-QTOF) calcd for [M + H]<sup>+</sup>: 277.0529, found: 277.0526.

### Conflicts of interest

This work was funded through a subcontract from a STTR to SusTerea (NSF IIP-1622799). GK is a founder of SusTerea.



## Acknowledgements

We thank the NSF Engineering Center for Biorenewable Chemicals (CBiRC) which was awarded NSF grant EEC-0813570 for partial support of this research. We also thank NSF STTR Phase I grant (IIP-1622799) for partial support of this research.

## References

- 1 R. J. Sheehan, *Terephthalic Acid, Dimethyl Terephthalate, and Isophthalic Acid*, *Ullmann's Encyclopedia of Industrial Chemistry*, 2000.
- 2 G. N. Koshel, S. G. Koshel, M. V. Postnova, N. V. Lebedeva, E. A. Kuznetsova, M. S. Belysheva and T. A. Yun'kova, *Kinet. Catal.*, 2004, **45**, 821–825.
- 3 (a) G. A. Kraus, S. Riley and T. Cordes, *Green Chem.*, 2011, **13**, 2734–2736; (b) J. J. Lee, G. R. Pollock, D. Mitchell, L. Kasuga and G. A. Kraus, *RSC Adv.*, 2014, **4**, 45657–45664.
- 4 I. E. Markó and G. R. Evans, *Tetrahedron Lett.*, 1993, **34**, 7309–7312.
- 5 (a) T. Shimo, F. Muraoka and K. Somekawa, *Nippon Kagaku Kaishi*, 1989, 1765–1771; (b) T. Shimo, T. Iwakiri and K. Somekawa, *J. Heterocycl. Chem.*, 1992, **29**, 199–201.
- 6 R. D. Slack, M. A. Siegler and G. H. Posner, *Tetrahedron Lett.*, 2013, **54**, 6267–6270.
- 7 O. Keisho, R. Tamura, K. Shigehara, E. Masai, M. Nakamura, Y. Otsuka, Y. Katayama and Y. Nakao, *Chem. Lett.*, 2014, **43**, 1349–1351.

