



Cite this: *RSC Adv.*, 2017, 7, 54934

# Enantioselective synthesis of chiral acylsilanes by copper/HZNU-Phos-catalyzed asymmetric conjugate addition of diethylzinc to $\alpha,\beta$ -unsaturated acylsilanes†

Ji-Yuan Lv,<sup>a</sup> Zheng Xu,<sup>a</sup> Zhan-Jiang Zheng,<sup>a</sup> Li Li,<sup>\*a</sup> Yu-Ming Cui,<sup>a</sup> Jian Cao,<sup>a</sup> Ke-Fang Yang<sup>a</sup> and Li-Wen Xu<sup>id</sup> <sup>\*ab</sup>

Received 30th October 2017  
Accepted 27th November 2017

DOI: 10.1039/c7ra11919d

rsc.li/rsc-advances

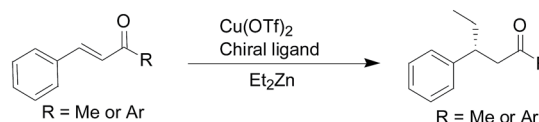
The catalytic asymmetric copper-catalyzed conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes was found to proceed smoothly in moderate to good yields and promising enantioselectivities (up to 85% ee) in the presence of the multifunctional HZNU-Phos with both a phosphine center and BINOL-based diol moiety that played a crucial role in the achievement of the best enantioselectivity for this reaction.

Since Brook reported the synthesis of acylsilanes in 1957,<sup>1</sup> much effort has been devoted to the synthesis and synthetic utility of acylsilanes in organic chemistry because acylsilanes can be considered as unusual carbonyl compounds bonded with a silicon-based bulky group at the carbonyl  $sp^2$  carbon atom.<sup>2</sup> Additionally, acylsilanes are a fascinating class of carbonyl compounds, and as such, growing attention has been paid to the utilization of acylsilanes in a diverse range of transformations. Therefore, it is an increasingly attractive moiety in organic synthesis, as demonstrated by the steric and the electronic effects as well as the unique reactivity pattern of the bulky trisubstituted silyl groups.<sup>3</sup> As a family of functional acylsilanes, the  $\alpha,\beta$ -unsaturated acylsilanes are very attractive building blocks for organic synthesis.<sup>4</sup> While various addition reactions to  $\alpha,\beta$ -unsaturated acylsilanes toward the synthesis of corresponding  $\beta$ -functionalised acylsilanes over past decades,<sup>5</sup> to the best of our knowledge, there is rare examples on the conjugate addition of organometallic reagents to  $\alpha,\beta$ -unsaturated acylsilanes, and asymmetric versions still remain sparse. The chemistry is still remarkably underdeveloped in the case of catalytic asymmetric conjugate addition reaction with  $\alpha,\beta$ -unsaturated acylsilanes. There remains, however, a formidable and exciting challenge associated with the endeavor on the development of highly chemo- and stereo-selective  $\beta$ -functionalization process. It occurred to us that copper catalysis could be used to promote

conjugate addition of organometallic reagents, such as diethylzinc, to  $\alpha,\beta$ -unsaturated acylsilanes. Although enormous efforts have been performed in the asymmetric 1,4-conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated carbonyl compounds,<sup>6</sup> the development of new variants of conjugate addition reaction with excellent enantioselectivity remains a challenge (Fig. 1).

Although recent advances have been made in the field of transformations of acylsilanes, this type of conjugate addition reaction continues to present a formidable challenge because it generally generate mixtures of 1,4-adducts and 1,2-adducts (Fig. 2),<sup>2</sup>  $\alpha$ -hydroxysilanes,<sup>7</sup> silyl enol ethers,<sup>8</sup> silyl ethers or other side-products depending on the Brook rearrangement

(a) Previous work on asymmetric 1,4-addition catalyzed by copper



(b) This work

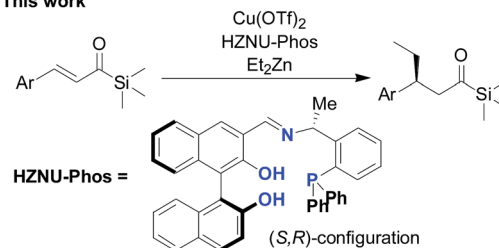


Fig. 1 Copper-catalyzed 1,4-conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated carbonyl compounds: (a) typical examples of asymmetric 1,4-conjugate addition catalysed by copper<sup>6</sup> and (b) copper-catalyzed conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes.

<sup>a</sup>Key Laboratory of Organosilicon Chemistry and Material Technology of Ministry of Education, Hangzhou Normal University, No 1378, Wenyi West Road, Science Park of HZNU, Hangzhou 311121, P. R. China. E-mail: liwenxu@hznu.edu.cn

<sup>b</sup>Suzhou Research Institute, State Key Laboratory for Oxo Synthesis and Selective Oxidation, Lanzhou Institute of Chemical Physics (LICP), Chinese Academy of Sciences (CAS), P. R. China

† Electronic supplementary information (ESI) available: Experimental procedures, characterization data, crystallographic data in CIF and NMR spectra. See DOI: 10.1039/c7ra11919d



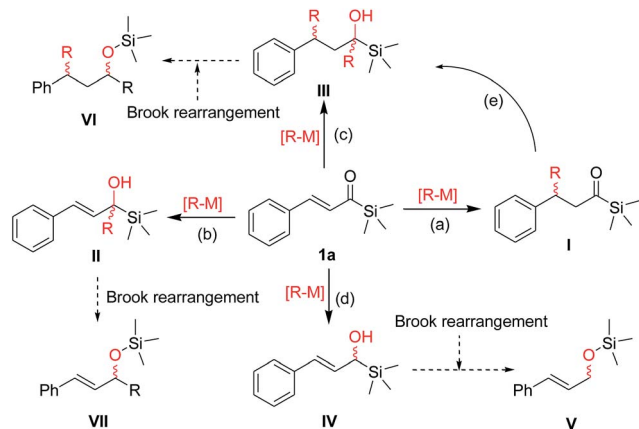


Fig. 2 Possible products from the conjugate addition reaction of organometallic reagents [R-M] to  $\alpha,\beta$ -unsaturated acylsilanes.

reaction (migration of a silyl group from carbon center to an oxygen atom).<sup>9</sup> We recently developed a synthetic methodology for the enantioselective copper-catalyzed conjugate addition of diethylzinc and trapping of the zinc enolate employing a multifunctional N,O,P-ligand (HZNU-Phos) bearing multiple stereogenic centers,<sup>10</sup> which operates with high level of enantioselectivity for chalcone and 4-phenylbutenone and its analogues. The copper/HZNU-Phos catalyst system exhibited the highest catalytic performance to date in term of enantioselectivity (up to >99% ee) and efficiency (TON = 17 600).

Herein, using our multifunctional HZNU-Phos ligand mentioned above, we would like to report a copper-catalyzed conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes,<sup>11</sup> which provides a catalytic method to give various acylsilanes with good enantioselectivity.

Initially, the evaluation of enantioselectivity for the catalytic asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a** were made in the presence of various ligands. As shown in Fig. 3, different types of chiral phosphine ligands and multifunctional heteroatom-containing ligands were used in this work.<sup>12</sup> All these phosphine ligands with different groups can catalyze the conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a** smoothly with good conversions. It can be seen that varied enantioselectivities were achieved from different ligands (Table 1), and the BINOL-derived multifunctional phosphine ligand (**L6**, called HZNU-Phos) showed its great impact on the enantioselective conjugate addition reaction. Interestingly, the ligand **L5** gave no enantioselectivity compared to that with **L6**, which supported the important role of two phenol groups in this reaction. In contrast, the BINAP (2,2'-bis(diphenylphosphino)-1,1'-binaphthyl) and other phosphine ligands evaluated in this work proved substantially no or low enantioselectivity (<67% ee). The BINOL-derived phosphine **L19** that has been used as a highly efficient ligand in copper-catalyzed conjugate addition of  $\text{Et}_2\text{Zn}$ , reported firstly by Endo and Shibata,<sup>13</sup> proved to be poor ligand in this reaction, which gave the desired product **2a** in low enantioselectivity. In addition, there is no activity or enantioselectivity for **L5**, **L9** (Fei-Phos), **L13**, or **L18** (Tao-Phos), albeit

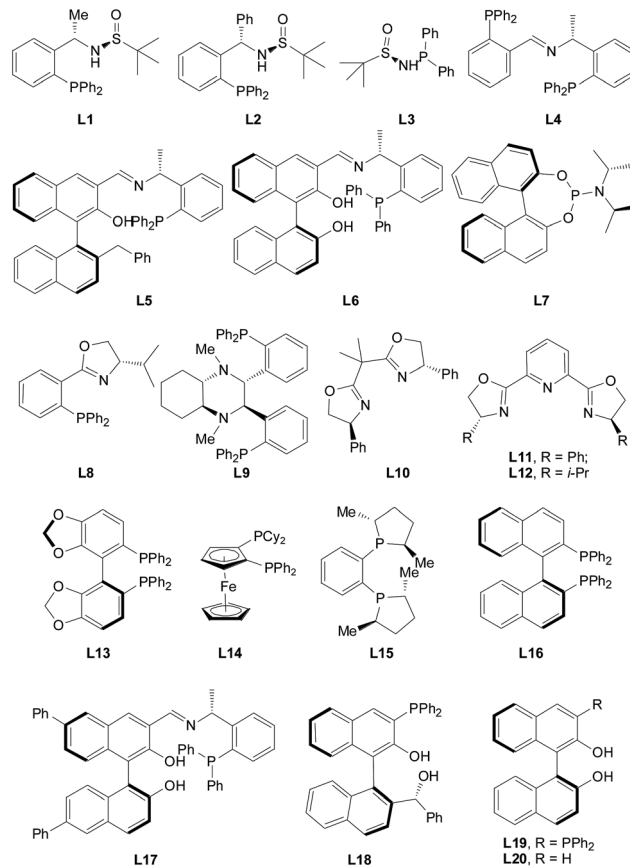


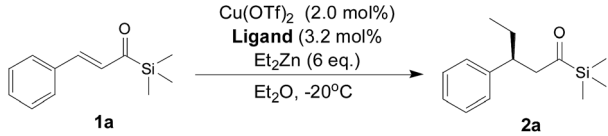
Fig. 3 Representative ligands evaluated in the copper-catalyzed asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a**.

these ligands exhibited excellent enantioselectivity in other organic transformations, which further suggested the crucial role of multifunctional group and multiple stereogenic centers on the (*S,R*)-HZNU-Phos ligands in the copper-catalyzed conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a**. Although the pybox ligands have been used in this reaction, the use of pybox **L11** (2,6-bis((*S*)-4-phenyl-4,5-dihydrooxazol-2-yl)pyridine) and pybox **L12** (2,6-bis((*S*)-4-isopropyl-4,5-dihydrooxazol-2-yl)pyridine) resulted in low to moderate enantioselectivities in this reaction (up to 53% ee). And interestingly, the substituent on the pybox has a large effect on the enantioselectivity. These screenings of ligands appeared to us as that the (*S,R*)-HZNU-Phos ligand (**L6**) was still the best choice for the copper-catalyzed conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a** (Table 1).

After the screening of chiral ligands for catalytic conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a**, it was found that the effect of solvents and additive on reaction enantioselectivity were also important. As shown in Table 2, no enantioselectivity was detected in THF, and the difference in enantioselectivity for other solvents was largely, in which the diethyl ether was further confirmed as the best solvent in this reaction (Entry 4). Notably, we have also found that  $\text{Me}_2\text{Zn}$ ,  $\text{Ph}_2\text{Zn}$ , and  $\text{Et}_3\text{Al}$  were not suitable organometallic nucleophiles



**Table 1** The enantioselectivities of copper-catalyzed asymmetric conjugate addition diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a** in the presence of various ligands (L1–L20)



Entry	Ligand	Solvent	T (°C)	Yield <sup>a</sup> (%)	ee <sup>b</sup> (%)
1	L1	Et <sub>2</sub> O	-20 °C	36	-57
2	L2	Et <sub>2</sub> O	-20 °C	48	-20
3	L3	Et <sub>2</sub> O	-20 °C	20	-9
4	L4	Et <sub>2</sub> O	-20 °C	25	-11
5	L5	Et <sub>2</sub> O	-20 °C	79	0
6	L6	Et <sub>2</sub> O	-20 °C	80	85
7	L7	Et <sub>2</sub> O	-20 °C	33	15
8	L8	Et <sub>2</sub> O	-20 °C	28	11
9	L9	Et <sub>2</sub> O	-20 °C	50	0
10	L10	Et <sub>2</sub> O	-20 °C	65	-28
11	L11	Et <sub>2</sub> O	-20 °C	77	-9
12	L12	Et <sub>2</sub> O	-20 °C	40	-53
13	L13	Et <sub>2</sub> O	-20 °C	35	0
14	L14	Et <sub>2</sub> O	-20 °C	56	-20
15	L15	Et <sub>2</sub> O	-20 °C	75	-67
16	L16	Et <sub>2</sub> O	-20 °C	78	-60
17	L17	Et <sub>2</sub> O	-20 °C	37	-25
18	L18	Et <sub>2</sub> O	-20 °C	59	0
19	L19	Et <sub>2</sub> O	-20 °C	65	-11
20	L20	Et <sub>2</sub> O	-20 °C	27	-9

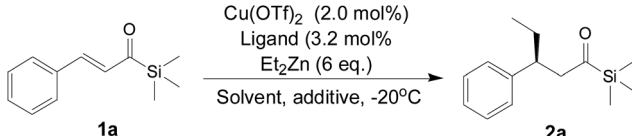
<sup>a</sup> Reaction conditions:  $\alpha,\beta$ -unsaturated acylsilane **1a** (0.5 mmol), Cu(OTf)<sub>2</sub> (2 mol%), chiral ligand (3.2 mol%). The conversions were detected by GC-MS. <sup>b</sup> The enantiomeric excess (ee) was determined by chiral HPLC analysis.

in this copper-catalyzed conjugate addition reaction of **1a** (see Table S1 of ESI<sup>†</sup>).

In addition, the desired product was obtained with almost the same level of enantioselectivity when general inorganic bases, such as KHF<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, KHSO<sub>4</sub>, or KH<sub>2</sub>PO<sub>4</sub>, were used as additive in this reaction (Table 2, Entries 5–8). However, further improvement with chiral cinchona alkaloid, such as cinchonidine, and with Lewis acid (ZnCl<sub>2</sub>), were proved to be unsuccessful because of inferior enantioselectivity in these cases (40% ee and 11% ee respectively, entries 9 and 10). Overall, the adduct **2a** could be obtained in 85% ee when the reaction run in Et<sub>2</sub>O in the presence of Cu(OTf)<sub>2</sub>/HZNU-Phos (**L6**) at -20 °C. Therefore, these reaction conditions would be the most suitable for the catalytic asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane at present.

Having identified an efficient chiral phosphine ligand with suitable reaction conditions, we continued to evaluate the substrate scope of the catalytic asymmetric copper-catalyzed asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes. As shown in Table 3, a variety of aromatic groups on  $\alpha,\beta$ -unsaturated acylsilanes were examined for this reaction, and the corresponding products **2** were obtained in moderate to good yields with promising enantioselectivities

**Table 2** Screening of reaction conditions: the enantioselectivities of copper-catalyzed asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilane **1a**<sup>a</sup>



Entry	Solvent	Additive	Time (h)	Yield	ee <sup>b</sup> (%)
1	THF	—	9	75	0
2	DCM	—	9	83	71
3	Toluene	—	9	85	69
4	Et <sub>2</sub> O	—	9	80	85
5	Et <sub>2</sub> O	KHF <sub>2</sub>	5	65	76
6	Et <sub>2</sub> O	K <sub>2</sub> CO <sub>3</sub>	5	67	76
7	Et <sub>2</sub> O	KHSO <sub>4</sub>	5	75	78
8	Et <sub>2</sub> O	KH <sub>2</sub> PO <sub>4</sub>	5	68	76
9	Et <sub>2</sub> O	Cinchonidine	5	55	40
10	Et <sub>2</sub> O	ZnCl <sub>2</sub>	5	23	11

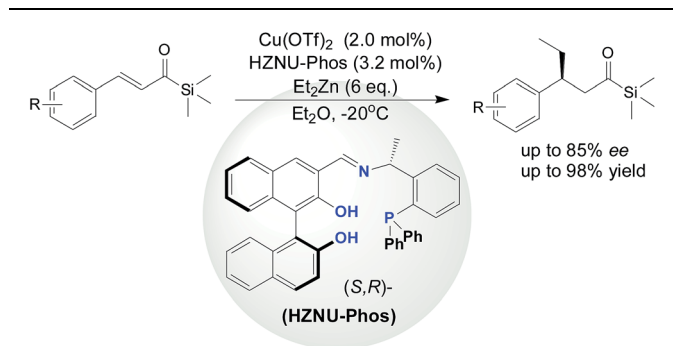
<sup>a</sup> Reaction conditions: **1a** (0.5 mmol), Cu(OTf)<sub>2</sub> (2 mol%), (*S,R*)-HZNU-Phos (3.2 mol%). The conversions were completed (>99%) that detected by GC-MS. <sup>b</sup> The enantiomeric excess (ee) was determined by chiral HPLC analysis.

(Table 3). Most of  $\alpha,\beta$ -unsaturated acylsilanes resulted in moderate to good ee values. However, the substituted groups on the *ortho*-position of phenyl ring that derived from aromatic aldehydes were proved to a disadvantageous factor in the enantioselective conjugate addition cycloaddition. For example, the variation of phenyl of substrate **1a** to *ortho*-halide substituted phenyl derivatives were generally found to be underwent conjugate addition in low enantioselectivities but with good isolated yields (Entries 7–10 of Table 3). Furthermore, the use of *meta*-Cl substituted  $\alpha,\beta$ -unsaturated acylsilane instead of *ortho*-Cl substituted  $\alpha,\beta$ -unsaturated acylsilane also underwent selective conjugate addition with good enantioselectivity and excellent yields (Entry 11, 98% yield and 68% ee). Unfortunately, the X-ray data of product **3** are not available at present. Notably, in the past years, electronic circular dichroism (ECD) has been proved to be a reliable and alternative option to determine absolute configurations of enantioenriched molecules.<sup>14</sup> In order to determine the absolute configuration of the chiral acylsilane product **3**, we compared the calculated CD spectrum and experimental CD spectrum of the acylsilane compound **3d** (see ESI, Fig. S1–S5<sup>†</sup>), in which the (*R*)-configuration of the stereogenic sp<sup>3</sup> carbon center on chiral acylsilane product **3d** is proposed to be more possibly than that of acylsilane product **3d** with (*S*)-configuration based on the calculated results (Fig. S4<sup>†</sup>) as well as the experimental CD spectra of chiral 4-phenyl-hexan-2-one (*S*-configuration, Fig. S1<sup>†</sup>).<sup>10</sup>

In summary, we have investigated the catalytic asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes. This conjugate addition reaction was performed smoothly in moderate to good yields and promising enantioselectivities (up to 85% ee). The preliminary study suggested



**Table 3** Substrate scope of copper-catalyzed asymmetric conjugate addition of diethylzinc to  $\alpha,\beta$ -unsaturated acylsilanes<sup>a</sup>



Entry	R	Yield <sup>b</sup> (%)	ee <sup>c</sup> (%)
1	H	<b>3a</b> : 56	85
2	<i>p</i> -Me	<b>3b</b> : 40	80
3	<i>p</i> -Et	<b>3c</b> : 65	72
4	<i>p</i> -Cl	<b>3d</b> : 60	66
5	<i>p</i> -CF <sub>3</sub>	<b>3e</b> : 60	52
6	<i>p</i> -Br	<b>3f</b> : 60	72
7	<i>o</i> -F	<b>3g</b> : 32	26
8	<i>o</i> -Cl	<b>3h</b> : 68	9
9	<i>o</i> -Br	<b>3i</b> : 70	34
10	<i>o</i> -CF <sub>3</sub>	<b>3j</b> : 40	35
11	<i>m</i> -Cl	<b>3k</b> : 98	68

<sup>a</sup> Reaction conditions:  $\alpha,\beta$ -unsaturated acylsilane **1** (0.5 mmol), Cu(OTf)<sub>2</sub> (2 mol%), chiral ligand is (*S,R*)-HZNU-Phos (3.2 mol%).

<sup>b</sup> Isolated yield. <sup>c</sup> The enantiomeric excess (ee) was determined by HPLC analysis on a chiral stationary phase (see ESI for details).

that the multifunctional HZNU-Phos play a crucial role in this asymmetric conjugate addition reaction, in which both phosphine center and BINOL-based diol moiety were indispensable functional groups to the achievement of the best enantioselectivity for this reaction at present, in which the experimental results further indicate the powerful potential of HZNU-Phos in the asymmetric catalysis. Further studies on the investigation of catalytic performance of HZNU-Phos in catalytic asymmetric transition metal-catalyzed organic transformations are currently underway and will be reported in due future.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

This Project was supported by the National Natural Science Foundation of China (No. 21472031, 21703051, and 21773051), Zhejiang Provincial Natural Science Foundation of China (LZ18B020001, LY16E030009, LY17B030005, and LY17E030003), and Science and Technology Department of Zhejiang Province (2015C31138). The authors also thank Dr Z. R. Qu, Dr C. Q. Sheng, Dr K. Z. Jiang, and Dr Q. H. Pan (all at HZNU) for their technical and analytical support.

## Notes and references

- 1 A. G. Brook, *J. Am. Chem. Soc.*, 1957, **79**, 4373.
- 2 H. J. Zhang, D. L. Priebbenow and C. Bolm, *Chem. Soc. Rev.*, 2013, **42**, 8540.
- 3 (a) P. C. Bulman-Page, M. J. McKenzie, S. S. Klair and S. Rosenthal, in *The Chemistry of Organosilicon Compounds, Part 2*, ed. S. Patai and Z. Rappoport, Wiley, Chichester, UK, 1998, vol. 2, p. 1599; (b) P. C. Bulman-Page, M. J. McKenzie, S. S. Klair and S. Rosenthal, *Patai's Chemistry of Functional Groups*, Wiley, 2009, DOI: 10.1002/9780470682531.pat0189; (c) P. C. Bulman-Page and M. J. McKenzie, in *Science of Synthesis*, ed. I. Fleming, Thieme, Stuttgart, 2001, vol. 4, p. 513; (d) H. Qi and D. P. Curran, in *Comprehensive Organic Functional Group Transformations*, ed. A. R. Katrinsky, O. Meth-Cohn, C. W. Rees and C. J. Moody, Pergamon, Oxford, UK, 1995, p. 409; (e) M. A. Brook, *Silicon in Organic, Organometallic, and Polymer Chemistry*, John Wiley & Sons, New York, 2000.
- 4 (a) H. J. Reich, M. J. Kelly, R. E. Olson and R. C. Holtan, *Tetrahedron*, 1983, **39**, 949; (b) H. J. Reich and M. J. Kelly, *J. Am. Chem. Soc.*, 1982, **104**, 1119.
- 5 (a) R. L. Danheiser and D. M. Fink, *Tetrahedron Lett.*, 1985, **26**, 2509; (b) R. L. Danheiser and D. M. Fink, *Tetrahedron Lett.*, 1985, **26**, 2513; (c) A. Ricci, A. Degl'Innocenti, G. Borselli and G. Reginato, *Tetrahedron Lett.*, 1987, **28**, 4093; (d) K. Narasaka, H. Kusama and Y. Hayashi, *Tetrahedron*, 1992, **48**, 2059.
- 6 For recent examples, see: (a) N. Mistry and S. P. Fletcher, *Adv. Synth. Catal.*, 2016, **358**, 2489; (b) T. T. Yang, Y. L. Zhang, P. Cao, M. Wang, L. Li, D. Li and J. Liao, *Tetrahedron*, 2016, **72**, 2707; (c) L. Han, Y. Lei, P. Xing, X. L. Zhao and B. Jiang, *J. Org. Chem.*, 2015, **80**, 3752; (d) B. H. Lipshutz, S. L. Huang, W. W. Y. Leong, G. F. Zhong and N. A. Isley, *J. Am. Chem. Soc.*, 2012, **134**, 19985; (e) K. Dohi, J. Kondo, H. Yamada, R. Arakawa and S. Sakaguchi, *Eur. J. Org. Chem.*, 2012, 7143; (f) M. Magrez, J. Wencel-Delord, A. Alexakis, C. Crevisy and M. Mauduit, *Org. Lett.*, 2012, **14**, 3576; (g) L. Palais, L. Babel, A. Quintard, S. Belot and A. Alexakis, *Org. Lett.*, 2010, **12**, 1988; (h) M. Welker, S. Woodward, L. F. Veiros and M. J. Calhorda, *Chem.-Eur. J.*, 2010, **16**, 5620; (i) M. Sada, T. Furuyama, S. Komagawa, M. Uchiyama and S. Matsubara, *Chem.-Eur. J.*, 2010, **16**, 10474; (j) M. Sada and S. Matsubara, *J. Am. Chem. Soc.*, 2010, **132**, 432; (k) J. Wencel-Delord, A. Alexakis, C. Crevisy and M. Mauduit, *Org. Lett.*, 2010, **12**, 4335; (l) A. Quintard and A. Alexakis, *Adv. Synth. Catal.*, 2010, **352**, 1856; For representative reviews, see: (m) A. Alexakis, J. E. Bäckvall, N. Krause, O. Pàmies and M. Diéguez, *Chem. Rev.*, 2008, **108**, 2796.
- 7 (a) P. F. Cirillo and J. S. Panek, *J. Org. Chem.*, 1990, **55**, 6071; (b) B. F. Bonini, M. Comes-Franchini, M. Fochi, G. Mazzanti, C. Nanni and A. Ricci, *Tetrahedron Lett.*, 1998, **39**, 6737; (c) E. P. Lodge and C. H. Heathcock, *J. Am. Chem. Soc.*, 1987, **109**, 3353.



- 8 A. Tsubouchi, N. Sasaki, S. Enatsu and T. Takeda, *Tetrahedron Lett.*, 2013, **54**, 1264.
- 9 (a) A. G. Brook, *Acc. Chem. Res.*, 1974, **7**, 77; (b) E. J. Corey and S. Lin, *J. Am. Chem. Soc.*, 1996, **118**, 8765; (c) R. Unger, T. Cohen and I. Marek, *Eur. J. Org. Chem.*, 2009, 1749; (d) R. Unger, F. Weisser, N. Chinkov, A. Stanger, T. Cohen and I. Marek, *Org. Lett.*, 2009, **11**, 1853; (e) R. Unger, T. Cohen and I. Marek, *Tetrahedron*, 2010, **66**, 4874; (f) R. B. Lettan II, T. E. Reynolds, C. V. Galliford and K. A. Scheidt, *J. Am. Chem. Soc.*, 2006, **128**, 15566; (g) R. B. Lettan II, C. V. Galliford, C. C. Woodward and K. A. Scheidt, *J. Am. Chem. Soc.*, 2009, **131**, 8805; (h) M. Honda, T. Nakajima, M. Okada, K. Yamaguchi, M. Suda, K.-K. Kunimoto and M. Segi, *Tetrahedron Lett.*, 2011, **52**, 3740; (i) B. Liu and C. D. Lu, *J. Org. Chem.*, 2011, **76**, 4205; (j) C. Wang, Z. Gan, J. Lu, X. Wu and Z. Song, *Tetrahedron Lett.*, 2011, **52**, 2462.
- 10 F. Ye, Z. J. Zheng, W. H. Deng, L. S. Zheng, Y. Deng, C. G. Xia and L. W. Xu, *Chem.-Asian J.*, 2013, **8**, 2242.
- 11 A. Nikolaev and A. Orellana, *Org. Lett.*, 2015, **17**, 5796.
- 12 Z. Xu and L. W. Xu, *Chem. Rec.*, 2015, **15**, 925.
- 13 K. Endo, M. Ogawa and T. Shibata, *Angew. Chem., Int. Ed.*, 2010, **49**, 2410.
- 14 For recent reviews, see: (a) G. Pescitelli, L. D. Bari and N. Berova, *Chem. Soc. Rev.*, 2011, **40**, 4603; (b) D. Slade, D. Ferreira and J. P. J. Marais, *Phytochemistry*, 2005, **66**, 2177.

