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Antimicrobial effect of gallium nitrate against bacteria encountered in burn wound infections

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Recently, biomaterials have been increasingly used for burn wound healing, but burn wound infections and biomaterial-associated infections still need to be addressed. As a novel inorganic antimicrobial, the antimicrobial effect of gallium nitrate was investigated, and the minimum inhibitory concentration (MIC) of gallium nitrate against bacteria that are common in infected burn wounds was determined with a Microbial Viability Assay Kit-WST. The results showed that the MIC of Ga(NO₃)₃ against *E. coli* and *E. faecalis* was 256 µg mL⁻¹, and it was 512 µg mL⁻¹ against *P. aeruginosa*, *K. pneumonia*, *E. cloacae*, *A. baumannii*, *S. maltophilia*, *S. aureus* and *S. epidermidis*. Meanwhile, transmission electron microscopy (TEM) found similar visual evidence of the mechanism by which the gallium ion attacks both Gram-negative and Gram-positive bacteria, which was in agreement with the MIC results. By TEM observation, it was found that detachment of the cell membrane and wall and the appearance of an electron-light region containing condensed substances occurred in both Ga³⁺-treated *E. coli* and Ga³⁺-treated *S. aureus* cells, with smaller morphological changes in Ga³⁺-treated *S. aureus* compared with *E. coli*. This research shows the effective and wide-spectrum antimicrobial properties of gallium nitrate against most bacteria encountered in burn wound infections. Gallium(III) could be a good choice when fighting an infected burn wound, and it is a promising candidate for modifying biomaterials or medical devices to prevent infection in burn wounds.

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1. Introduction

In recent years, due to the development of molecular biology and tissue engineering technology, biomaterials, including engineered skin tissue, have been rapidly developed, making it possible for severe burn patients without enough skin for grafting to still be cured and to have physiological functions re-established.¹ However, it cannot be ignored that biomaterial-associated infections frequently occur with implants and devices, especially those for supporting or restoring normal skin functions.^{2,3} Nevertheless, comparatively few biomaterials have been prepared to efficiently decrease the incidence of biomaterial-associated infections.

At the same time, for burn patients, the wound bed is an ideal medium for the survival and spread of bacteria because it contains rich nutrients due to the leakage of inflammatory mediators and the dissolving of necrotic tissue. What is more serious is that pathogens can easily invade adjacent healthy tissues and spread to the bloodstream in patients with severe

burns who have lost normal immune function, leading to a life-threatening infection, sepsis.⁴ The treatment of wound infection in burn patients has always been one of the most intractable problems, since heavy antibiotic use has greatly increased antibiotic resistance.⁵ Therefore, ideal agents against biomaterial-associated infection and burn wound infection that are effective, can operate against a wide spectrum of pathogens, and are safe are greatly needed, including inorganic antimicrobials.

Inorganic antimicrobials, which provide an environment with a relatively high concentration of metal ions, can change the membrane potential and induce a difference in ion concentrations, thus destroying the molecules' transportation. Meanwhile, the heavy metal ions can inactivate important enzymes by destroying the conformation of the enzyme or damaging/displacing metal ions, which are necessary to sustain the enzymatic activity, resulting in energy metabolism disorder. In addition, intracellular metal ions can combine with nucleic acids, which may interfere with the bacterial proliferation process.^{6–10} For these antibacterial mechanisms, inorganic antimicrobials possess the advantages of effectiveness, the ability to affect a wide spectrum of bacteria, and less vulnerability to acquired resistance.

Recently, several metallic antimicrobial agents, such as silver, copper and zinc, have been introduced and

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investigated.^{11–13} These metallic antimicrobial agents also include gallium.^{14,15} Gallium (Ga, 69.72 g mol^{−1}) is a transition metal element, which has a similar ionic radius to that of iron. Currently, the efficacy and safety of a variety of compounds containing gallium that are widely used to treat cancer and hypercalcemia, have been confirmed.^{16–20} The first investigation of the antimicrobial action of gallium was published in 1931.²¹ It has been reported that gallium (Ga³⁺) exerts a significant inhibitory activity against numerous bacteria including *Staphylococcus aureus*, methicillin-resistant *S. aureus*, *Rhodococcus equi*, *Pseudomonas aeruginosa*, *Acinetobacter baumannii* and *Escherichia coli*.^{22–26} Ga³⁺, in the form of Ga(NO₃)₃ solution (Ganite; Genta), is already FDA approved and given in large doses (i.v.) to treat hypercalcemia of malignancy.¹⁶ Ga³⁺ has an ionic radius identical to Fe³⁺, which is vitally important in many biologic systems as it can disrupt iron-dependent oxidation and reduction. As a result, Fe³⁺ is prevented from combining with proteins and chelating agents, and the bacterial DNA synthesis and various oxidation–reduction reactions are interfered with because of the competitive inhibition by Ga³⁺.²⁷ Due to its antimicrobial ability, Ga³⁺ has been proposed as a topical wound treatment.²⁸

The objective of this research was to explore the antimicrobial effect and to determine the minimum inhibitory concentration (MIC) values of Ga(NO₃)₃ solution against bacteria that are common in burn wound infections, by using a Microbial Viability Assay Kit-WST. Meanwhile, transmission electron microscopy (TEM) was chosen with the aim of observing in detail the microscopic morphological and structural changes of bacteria interacting with gallium nitrate.

2. Material and methods

2.1 Chemicals, microbial strains, culture media

Ga(NO₃)₃·xH₂O (crystalline, 99.9% trace metals basis) was bought from Sigma-Aldrich and dissolved in water. Then, inductively coupled plasma mass spectrometry (ICP-MS) was employed to accurately identify the concentration of gallium ions, due to the uncertainty of the water content in the crystalline form.

Nine microbial strains, including Gram-positive: *Staphylococcus aureus* (ATCC25923), *Staphylococcus epidermidis* (FSCC223011), and *Enterococcus faecalis* (FSCC146002), and Gram-negative: *Escherichia coli* (ATCC25922), *Pseudomonas aeruginosa* (ATCC27853), *Klebsiella pneumoniae* (FSCC167002),

Table 1 Optimal cell density for inoculation of different bacteria in WST-8 tests

Strains (Gram-negative)	Density (CFU mL ^{−1})	Strains (Gram-positive)	Density (CFU mL ^{−1})
<i>E. coli</i>	5 × 10 ³	<i>S. aureus</i>	10 ⁵
<i>P. aeruginosa</i>	5 × 10 ⁷	<i>S. epidermidis</i>	10 ⁷
<i>K. pneumoniae</i>	5 × 10 ³	<i>E. faecalis</i>	10 ⁴
<i>E. cloacae</i>	10 ⁴		
<i>A. baumannii</i>	5 × 10 ³		
<i>S. maltophilia</i>	10 ⁷		

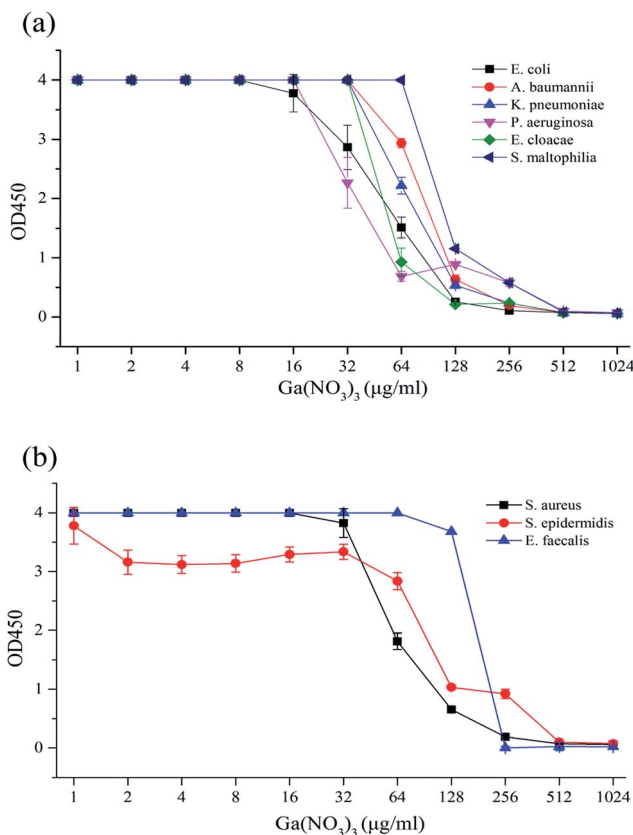


Fig. 1 Antimicrobial susceptibility curves of Ga(NO₃)₃.

Enterobacter cloacae (FSCC145003), *Acinetobacter baumannii* (ATCC19606), and *Stenotrophomonas maltophilia* (ATCC51331) were selected. Culture media were as follows: tryptic soy broth (TSB), Mueller–Hinton (MH) broth, and brain heart infusion (BHI) broth. All above were bought from GUANGDONG HUANKAI MICROBIAL SCI. & TECH. CO., LTD.

2.2 Inductively coupled plasma mass spectrometry (ICP-MS)

Determination of the gallium ion concentration was performed on a NexION 300X apparatus (Perkin-Elmer, Waltham, MA). Operating conditions of 1300 W Rf power, 13 L min^{−1} plasma flow, 0.82 L min^{−1} nebulizer flow and peak jump mode were used. The calibration solution was made from a dilution of a standard stock solution (10 mg L^{−1}) with 2% nitric acid. The analytical process was carried out in accordance with the order of blank, calibration solution and sample. Both ⁶⁹Ga and ⁷¹Ga

Table 2 The MIC of Ga(NO₃)₃ against bacteria strains

Strains (Gram-negative)	MIC (μg mL ^{−1})	Strains (Gram-positive)	MIC (μg mL ^{−1})
<i>E. coli</i>	256	<i>S. aureus</i>	512
<i>P. aeruginosa</i>	512	<i>S. epidermidis</i>	512
<i>K. pneumoniae</i>	512	<i>E. faecalis</i>	256
<i>E. cloacae</i>	512		
<i>A. baumannii</i>	512		
<i>S. maltophilia</i>	512		



isotopes were used in order to control for possible interference and matrix effects.²⁹

2.3 Susceptibility testing

The Microbial Viability Assay Kit-WST, purchased from Dojindo Laboratories (Kumamoto, Japan), offers a method of microbial metabolism detection by colorimetry. According to the manufacturer's technical manual, WST-8, employed as a colorimetric indicator, is directly proportional to the number of living microorganisms.

Since the species and the metabolic activity have effects on the sensitivity (O.D. value), previous work has been done to optimize the number of cells and colouring reaction time for each test. After cultivation overnight, a bacterial suspension of 10^7 CFU mL⁻¹ using McFarland standards was prepared, further diluted with sterile saline to provide various densities, and then inoculated into a 96-well plate. After incubation for 6 h at 35 °C, 10 µL of the colouring reagent was added to each well, followed by another 2–3 h incubation. The O.D. value was measured at 450 nm with a microplate reader (Bio-Rad Laboratories, USA), and then the cell density corresponding to the O.D. value within a range of 2–4 was selected.

Bacteria were incubated at 37 °C in broth at 150 rpm overnight and then diluted to the proper density according to McFarland standards. After 90 µL MH broth was added to each well of columns 1 to 10 of the 96-well microplate, Ga(NO₃)₃ solution that had been prepared was added to the first well, followed by a serial two-fold dilution from columns 1 to 10. After that, another 90 µL MH broth was added to each well of columns 1 to 10, making the final volume 180 µL and dilutions of 1024 to 2 µg mL⁻¹. Columns 11 and 12 were the positive and negative controls, respectively. A microbial suspension (10 µL) was inoculated into each well of columns 1 to 11, and the microplate was incubated for 6 h at 35 °C. Then, 10 µL colouring reagent (comprising nine parts WST solution and one part electron mediator reagent for Gram-negative bacteria, or nine parts WST solution and one part eight-fold diluted electron mediator reagent for Gram-positive bacteria) was added to each well of the microplate, with another incubation for 2–3 h at 35 °C. Finally, the microplate was read at 450 nm with a microplate reader (Bio-Rad Laboratories,

USA), and the negative control was read as the blank value. The MIC value was defined as the lowest concentration of gallium nitrate solution where the absorbance change was no more than 0.05 compared to the blank value.^{30–32} The results are reported as the mean of the experiment that was conducted in triplicate.

2.4 Transmission electron microscope (TEM)

Representatives of Gram-negative and Gram-positive strains, *E. coli* and *S. aureus*, were cultivated in TSB broth (99 mL) at 150 rpm at 37 °C for 16 h. Then, 1 mL of Ga(NO₃)₃ solution was added to the media, making the final concentration 32 µg mL⁻¹ for both strains of bacteria. Cultivation was continued for 4 to 10 h, and a culture with no Ga(NO₃)₃ served as a blank control. Five millilitres of culture media was collected and centrifuged, washed with PBS, and then fixed with 2.5% glutaraldehyde. After graded alcohol dehydration and Epon embedding, the sample was cut with an ultramicrotome and then stained with uranyl acetate and lead citrate. Then, the ultrathin section sample was ready for TEM observation.³³

3. Results

3.1 Optimal cell density for inoculation of different bacteria in WST-8 tests

The sensitivity of WST-8 is related to the species of bacteria or the metabolic activity, so experiments have been done to optimize the number of cells for each species. The results are shown in Table 1.

3.2 Determination of the MIC of Ga(NO₃)₃ against bacteria strains

A few insoluble flocs existed in higher concentrations of medium-diluted Ga(NO₃)₃ solution, so 10 µL of 0.2% EDTA solution was added to each well before measurement to eliminate the interference with absorbance and the microbial viability assay. The O.D. value obtained was in accordance with the ordinate, the abscissa concentration of the Ga(NO₃)₃ solution for a susceptibility curve for analysis of the results (Fig. 1).

According to Fig. 1(a), which shows strains of Gram-negative bacteria, the susceptibility curve was stable at the beginning

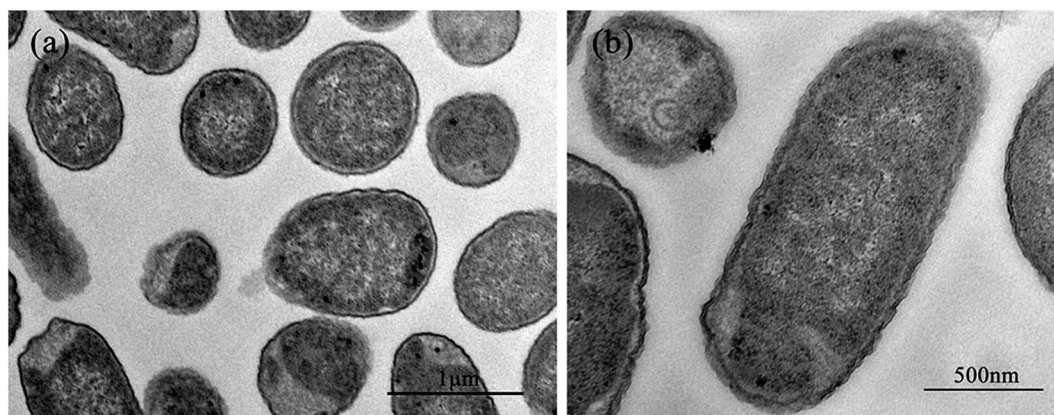


Fig. 2 Microstructure of normal *E. coli* cells.



and began to decline at $16\text{--}64\ \mu\text{g mL}^{-1}$ with the rising $\text{Ga}(\text{NO}_3)_3$ concentration, which prompted the inhibition of bacterial growth. As the concentration constantly rose, the curve reached its lowest point close to the blank value at $256\text{--}512\ \mu\text{g mL}^{-1}$, which revealed the fact that bacterial growth was completely inhibited. Based on the definition of the MIC in WST-8 tests,^{30–32} the concentration of $\text{Ga}(\text{NO}_3)_3$ at this point was read as the MIC against the experimental strain. Identical phenomena were found in Fig. 1(b), which shows strains of Gram-positive bacteria.

The MIC of $\text{Ga}(\text{NO}_3)_3$ was $256\ \mu\text{g mL}^{-1}$ against *E. coli* and *E. faecalis* and $512\ \mu\text{g mL}^{-1}$ against *P. aeruginosa*, *K. pneumonia*, *E.*

cloacae, *A. baumannii*, *S. aureus*, *S. epidermidis* and *S. maltophilia*, as shown in Table 2.

3.3 Microscopic appearance of *E. coli* after Ga^{3+} treatment

Fig. 2 shows the normal microstructure of *E. coli* cells. Electron density uniformity was found in the cells with smooth borders, while DNA molecules, which appear as electron-light areas in TEM, were distributed randomly in the cells.

Micromorphological changes were found in *E. coli* cells after Ga^{3+} treatment (Fig. 3), and Fig. 3(a) gives an overview of these changes. The cell membrane was found to be rough, and shrinkage and detachment of the cell membrane and wall were

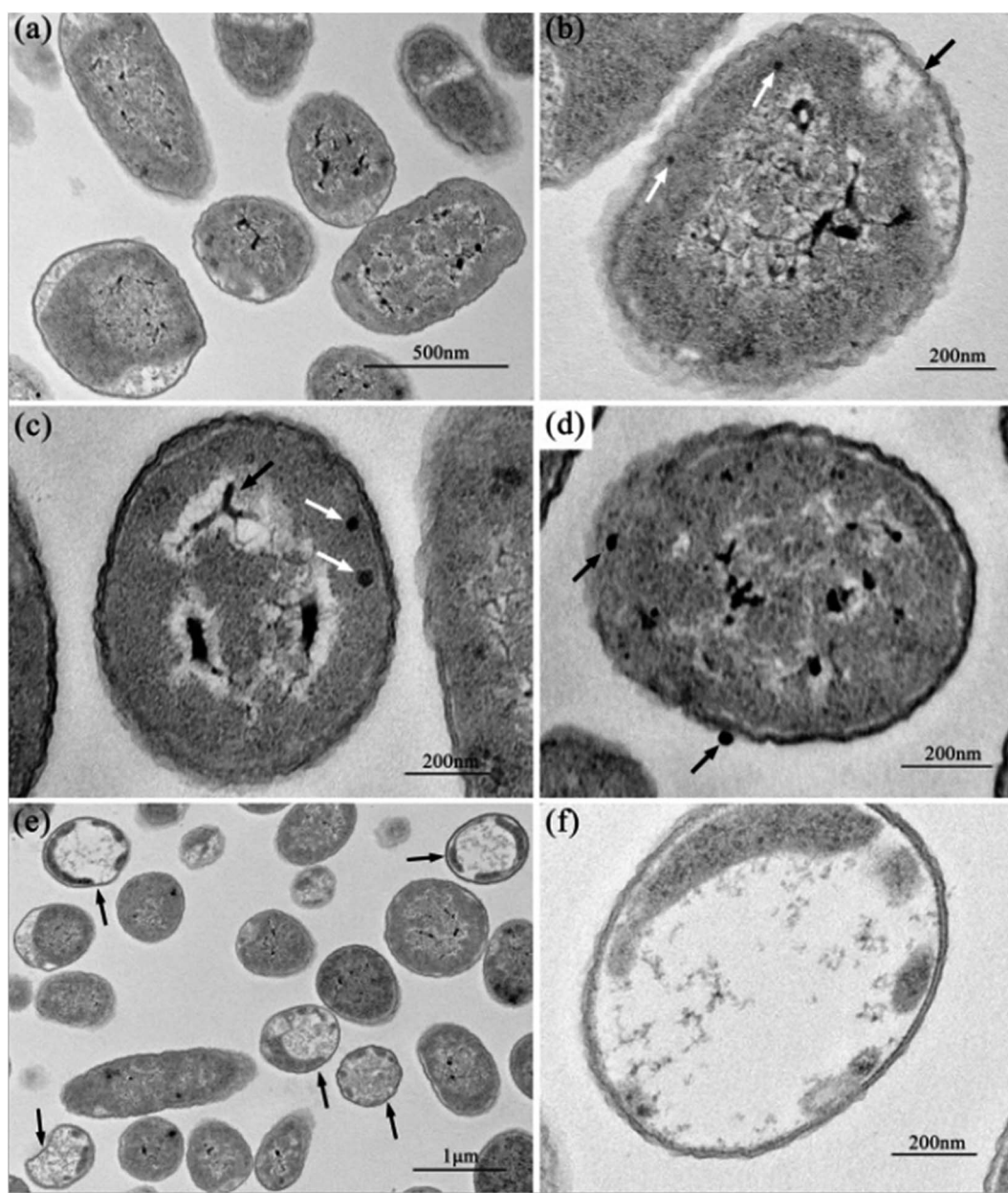


Fig. 3 Microstructure of Ga^{3+} -treated *E. coli* cells. (a) An overview. (b) A large gap between the cell membrane and cell wall (black arrow) and intracellular high-density electronic granules (white arrow). (c) Strip-type electron-dense substances inside the electron-light region (black arrow), and intracellular high electron density granules (white arrow). (d) High electron density granules adhere to the cell wall. (e) and (f) Cells composed of a large electron-light region with substances sporadically distributed.



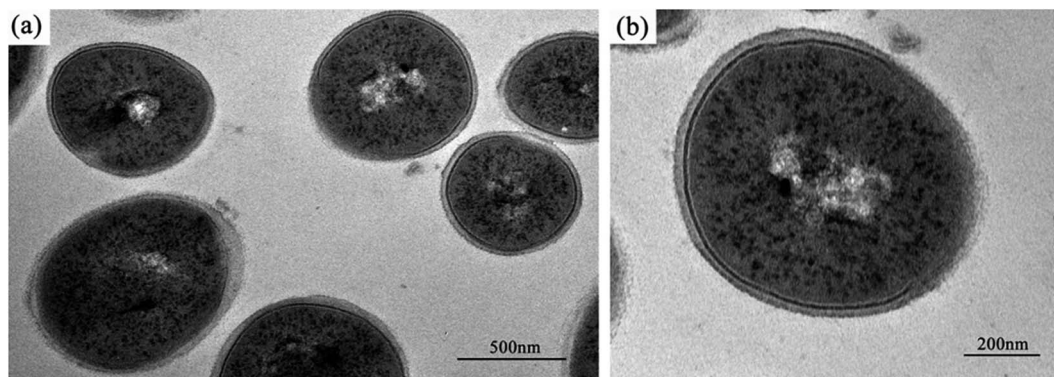


Fig. 4 Microstructure of normal *S. aureus* cells.

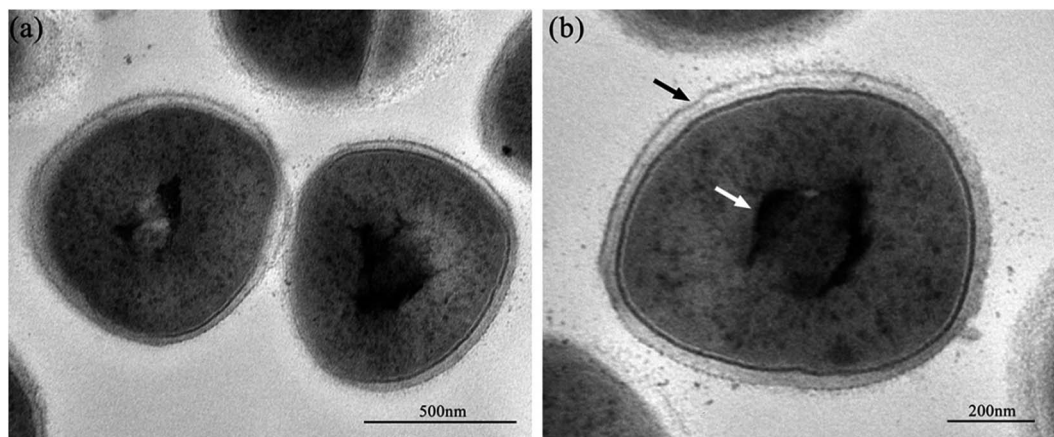


Fig. 5 Microstructure of Ga^{3+} -treated *S. aureus* cells. (a) Intracellular electron-dense substances. (b) Cell membrane and wall was parted (black arrow), and intracellular condensed substances were present (white arrow).

also found in Fig. 3(b) (black arrow), while some part of the cell wall appeared indistinct in the TEM image, which may indicate damage of cell wall (Fig. 3(b) and (d)). Electron-light areas gathered in the centre of the treated cells, and some strip-type electron-dense substances were obvious in the electron-light region in Fig. 3(c) (black arrow). There were some high electron density granules adhering to the cell wall (Fig. 3(d)) or in the cytoplasm (Fig. 3(b) and (c), white arrow), which may be an indication of Ga^{3+} combining with some kind of molecule on the cell wall and/or invading into the cytoplasm.

Meanwhile, a particular appearance was found in some TEM images (Fig. 3(e) and (f)). Some cells, which appeared as lower electron densities in the picture, were composed of a large electron-light region with substances sporadically distributed, and some had indistinct images of cell walls. However, internal structures or substances were difficult to find in these cells. These cells were similar to capsules containing broken segments, which may be the final phase of the treated cells.

3.4 Microscopic appearance of *S. aureus* after Ga^{3+} treatment

The normal microstructure of *S. aureus* cells is shown in Fig. 4. It was found that *S. aureus* cells have a comparatively thicker cell

wall and an apparent nucleoid containing DNA molecules, compared with *E. coli* cells.

Except for the smooth borders that were maintained after Ga^{3+} treatment in *S. aureus* cells, similar micromorphological changes occurred compared with *E. coli* cells (Fig. 5), such as the detachment of cell membrane and wall (Fig. 5(b), black arrow) and the appearance of condensed substances (Fig. 5(b), white arrow). In addition to these similarities, a few differences were also noted. On microscopic observation, the size of the electron-light region in *S. aureus* cell was smaller, and the density was much higher. In addition, the cell structure remained integral, and no broken segments had been found. This observation may suggest that *S. aureus* has a much stronger anti-gallium defence system.

4. Discussion

Wound infection, which may cause delays in re-epithelialization and poor wound healing, is one of the most common and intractable complications in burn patients. Even worse is that the pathogens in the infected wounds of patients with severe burns may cause bacteraemia, sepsis, multiple organ dysfunction syndrome (MODS), or even death.³⁴ According to recent



microbiological monitoring data,^{35,36} the detection rate of Gram-negative bacteria is much higher in burn wound infections, including *P. aeruginosa*, *E. coli*, *A. baumannii*, *K. pneumonia*, *E. cloacae* and *S. maltophilia*, followed by Gram-positive bacteria, including *S. aureus*, *S. epidermidis* and *E. faecalis*. To face the challenge of increasing drug resistance, novel antimicrobial agents are greatly needed, and the use of gallium may represent a brand new treatment against bacterial infection.

Pathogens are exposed to iron limitation during infection, and must compete for Fe^{3+} retained by chelating proteins,³⁷ so the competition for Fe becomes critical in the struggle between pathogen and host. Since Ga^{3+} and Fe^{3+} show very similar nuclear radii, the chemical behaviour of Ga^{3+} closely resembles that of Fe^{3+} .¹⁸ It is well known that Fe^{3+} is vitally important in many biologic systems and in the pathogen infection process. Therefore, the similarities in nuclear radius and coordination chemistry between Ga^{3+} and Fe^{3+} is what we can exploit to disturb iron metabolism to achieve the goal of antibacterial function. Our experiments observed the antibacterial activity of gallium nitrate ($\text{Ga}(\text{NO}_3)_3$) solution against bacteria common in burn wound infections, and the MIC determined in our study was $512 \mu\text{g mL}^{-1}$ for most strains of bacteria, with the exception of a MIC of $256 \mu\text{g mL}^{-1}$ against *E. coli* and *E. faecalis*. The results showed that gallium nitrate was a broad-spectrum antimicrobial agent against Gram-positive and Gram-negative bacteria. Furthermore, the MIC value exhibited that gallium nitrate had almost the same effect against both types of bacteria, which was strongly related to its antibacterial mechanism. If Ga^{3+} can affect most types of bacteria in burn wound infections, it may be an outstanding choice when we are faced with an infected wound.

In addition, biomaterial implants and devices have become one of the most important therapeutic methods currently when we try to repair a wound including burns. However, the deficiency in this approach or even life-threatening problem we must pay attention to is biomaterial-associated infection. Microbial contamination and infection can adversely affect all types of biomaterial implants and devices.^{2,3} Several methods have been explored and clinically studied in recent years to prevent implant infections in these contexts. For instance, antimicrobial agent coating strategies were extensively investigated. Kassetty³⁸ coated a dermal substitute with host defence peptides, and its antibacterial effects were evaluated. Chen³⁹ reported the *in vivo* antibacterial activity of a titanium substrate coated with melamine, a novel antimicrobial peptide that is being broadly studied.⁴⁰ Biocatalytic polymer-based antimicrobial coatings were also introduced and evaluated by Dave.⁴¹ Moreover, several ions, including silver,^{42,43} zinc,⁴⁴ and copper,⁴⁵ were exploited for the antibacterial modification of medical devices. In addition, other antibacterial strategies such as the use of non-adhesive⁴⁶ or photocatalytic materials⁴⁷ were also discussed.

Meanwhile, many studies about medical devices with Ga^{3+} have been performed. Herron⁴⁸ modified the surface of a biosynthetic dressing using Ga^{3+} -loaded polyelectrolyte multilayers (PEMs), and confirmed its efficacy in controlling biofilm formation by *P. aeruginosa*. Mouriño¹⁴ coated 45S5

Bioglass®-based scaffolds with gallium-releasing 3D alginate, which empowered the composite scaffold ability of an anti-bacterial effect and improved mechanical properties. Zhu⁴⁹ has reported that PVC plates coated with gallium nitrate could reduce bacterial cell aggregation on the PVC surface and inhibit biofilm formation associated with *P. aeruginosa* and *S. pyogenes*. As our study has determined the minimum inhibitory concentration of gallium nitrate against most pathogens in wound infections, it is worth performing further studies about how to modify medical devices or biomaterials with Ga^{3+} , for example the method of sustained-release gelatine microspheres or microfilms, to prevent infection in burn wound.

The TEM scanning images showed us that similar phenomena occurred in Ga^{3+} -treated *E. coli* and *S. aureus* cells, typical Gram-negative and Gram-positive bacterium. Those phenomena were the detachment of the cell membrane and cell wall and the appearance of an electron-light region containing condensed substances. These results suggested that the gallium ions inhibited both Gram-negative and Gram-positive bacteria, which was consistent with our MIC results.

The data show that Ga exerts its deleterious effects in two ways. First, gallium inactivates proteins by displacing Fe from its cognitive protein, and second, gallium causes the release of free Fe into the microenvironment, producing oxidative radicals that can induce cell death.⁵⁰ According to the reported article,³³ which used a similar TEM method, we can suppose the condensed substances to be DNA molecules. DNA replication is influenced when DNA molecules are in a condensed form.³³ Therefore, it may be visual evidence of the dysfunction or damage of DNA molecules caused by gallium ion's interference. According to the reports, heavy metals can react with proteins and inactivate them.¹⁰ Since gallium (Ga^{3+}), a transition metal element, can combine with proteins, especially some significant enzymes, by replacing iron ions, we propose the cell wall-adherent or intracellular electron-dense granules seen in Fig. 3(b)–(d) should be a combination of gallium ions and proteins. On the other hand, internal structures or substances were absent, and only few broken segments can be seen in the same treated *E. coli* cells in Fig. 3(e) and (f), which is interpreted as the disruption caused by the reaction of gallium ions with internal proteins. These segments may be more visual evidence of gallium ions' antibacterial functions.

Although less significant morphological changes occurred and the cell structure remained integral after Ga^{3+} treatment in *S. aureus* cells compared with *E. coli* cells, the MIC results showed us the same antibacterial consequence of Ga^{3+} against both Gram-negative and Gram-positive bacteria. Considering this, despite images showing unbroken structures after treatment of *S. aureus* cells, the dysfunction or damage of DNA molecules and internal proteins had already occurred.

The Microbial Viability Assay Kit-WST is a brand new detection method of microbial metabolism by colorimetry. WST-8 provided in the kit produces a water-soluble formazan dye, the amount of which is proportional to the number of living microorganisms, by a reduction reaction of intracellular dehydrogenase through an electronic mediator. The WST-8 colorimetric method has been applied to susceptibility testing



of various kinds of bacteria, and its advantages compared to the broth microdilution methods have been confirmed.³² It should be noted that, the accuracy of the results could be affected by the inoculated cell density according to different species, previous work should be done to decide the optimal cell densities for each test. Besides this, as a simple, rapid and effective MIC detection method, it is worthy of promotion.

5. Conclusion

Our study shows that gallium nitrate, as an efficient, inorganic antimicrobial that can fight most bacteria encountered in burn wound infections and against which there is little resistance, could be a good choice for clinicians faced with an infected burn wound. Gallium nitrate may also be a potential tool for biomaterial or medical devices to prevent infection in burn wounds as well. However, further studies are needed on the biocidal effect of gallium nitrate on fungi, which is also common in infected burn wounds, especially in patients who have used antibiotics intensively. Meanwhile, methods of effectively modifying biomaterials or medical devices with gallium ions still need further exploration.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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