



Cite this: RSC Adv., 2017, 7, 52621

Received 3rd September 2017
Accepted 6th November 2017DOI: 10.1039/c7ra09798k
rsc.li/rsc-advances

A near-infrared phosphorescent iridium(III) complex for imaging of cysteine and homocysteine in living cells and *in vivo*†

Yongquan Wu, Renmiao Wu,^a Huifang Li,^a Hong Zeng,^a Yuanyan Li,^a Qiuhong Wang,^b Mei Shi^{*b} and Xiaolin Fan^{*a}

Based on the cyclization reaction of an aldehyde with an aminoalkyl thiol, a novel NIR-emitting iridium(III) complex (NIR-Ir) showed off-on phosphorescent response toward Cys and Hcy, and was developed to detect Cys/Hcy levels *in vitro* and *in vivo*.

Biological thiols (RSH), such as cysteine (Cys) and homocysteine (Hcy), play very crucial roles in physiological processes to maintain the biological redox homeostasis because of their equilibrium between RSH and RSSR.¹ However, many diseases are related to unusual contents of Cys or Hcy in the human body. Cys deficiency may lead to various symptoms including skin lesions and liver damage.^{2,3} Furthermore, homocysteine is a risk factor for Alzheimer's and cardiovascular diseases.^{4,5} Hence, great attention has been paid to the detection of Hcy and Cys.

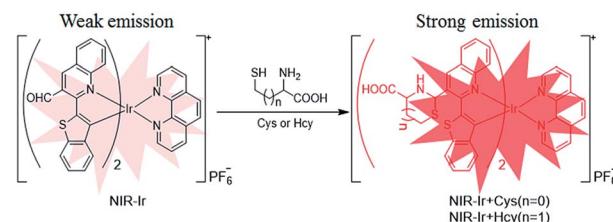
Fluorescent detection has emerged as an attractive tool for the detection of various biological molecules, such as thiols,^{6,7} because of its high sensitivity, selectivity, simplicity and its potential application to not only *in vitro* assays but also *in vivo* imaging studies. For example, Strongin's group has reported a xanthene dye containing aldehyde that afforded thiazolidines or thiazinanes by reaction with Cys or Hcy, resulting in a decrease in fluorescence emission.⁸

Compared to organic molecules, iridium(III) complexes as chemosensors provide several advantages, such as high photoluminescent efficiencies, relatively long lifetimes and significant Stokes shifts.^{9,10} Hence, many good performances were obtained in the assay of cations, anions, biological molecules and as biological labeling reagents.^{11–23} Recently, iridium(III) complex as probe of thiol detection, Che and co-workers developed a FRET-based luminescent iridium(III) complexes for sensing both cysteine and homocysteine.²⁴ Ma's group developed a long lifetime iridium(III) complex chemosensor for the selective detection of cysteine.²⁵ Li, Zhao, Huang, Yi and

co-workers reported a series of iridium(III) complexes for detection and imaging of homocysteine or both homocysteine and cysteine.^{26–34} Additionally, Chao's group reported an azobis(2,20-bipyridine)-bridged dinuclear iridium(III) complex for thiol imaging.³⁵ Other iridium(III) complexes have been designed and synthesized for thiol detection.^{36–39}

In order to deeply understand the roles of Cys/Hcy played in biological system, it is crucial to monitor them not only *in vitro* assays, but also in living cells and living animal. However, most of iridium(III) complexes emit relatively short wavelengths ($\lambda_{\text{em max}} < 650 \text{ nm}$), limiting their use in deep-tissue and living animal imaging because of the shallow penetration depth.¹¹ Compared to visible light, the advantage of NIR light was employed in biological imaging due to minimum photodamage to biological samples, deep tissue penetration, and minimum interference from background autofluorescence in the living systems.^{40–42} Currently, iridium(III) complexes are underdeveloped in the NIR region and few have been used in bioimaging.^{43,44} Therefore, it remains a challenging task to develop iridium(III) complex probes having NIR emission for the detection and *in vivo* imaging of Cys and Hcy.

Herein, a novel NIR-emitting iridium(III) complex $[(\text{bbt-pq-CHO})_2\text{Irphen}]^+\text{PF}_6^-$ (**NIR-Ir**) possessing a Cys/Hcy recognition unit, an aldehyde moiety in the main C^N ligands bbt-pq as a Cys/Hcy chemosensor (Scheme 1). Hcy/Cys can be sensitively



Scheme 1 Chemical structure of NIR-Ir and sensing mechanism of NIR-Ir for Cys and Hcy.

^aSchool of Chemistry and Chemical Engineering, Key Laboratory of Organo-pharmaceutical Chemistry of Jiangxi Province, Gannan Normal University, Ganzhou, 341000, P. R. China. E-mail: faxnl2013@gnnu.cn

^bDepartment of Chemistry, Fudan University, Shanghai 200433, P. R. China. E-mail: shimeif@fudan.edu.cn

† Electronic supplementary information (ESI) available: Complex synthesis and characterization and supplementary data. See DOI: 10.1039/c7ra09798k



and selectively detected using the aldehyde bearing bis-cyclometalated Ir(III) complex as turn-on probe. Therefore, photoluminescence intensity of **NIR-Ir** increased in the presence of Hcy/Cys based on the special cyclization reaction of aldehyde with aminoalkyl thiol groups. Moreover, **NIR-Ir** could be applied for the detection and imaging of Cys/Hcy in living animal due to its NIR emission, indicating its significant potential for biological applications.

The synthesis procedures for ligand L1 and **NIR-Ir** are outlined in Scheme S1 (ESI†). The ligand L1 was easily obtained by the Suzuki coupling reaction of 2-chloroquinoline-3-carbaldehyde with benzo[b]thiophen-2-boronic acid. The experimental details and characterization data for **NIR-Ir** are listed in the ESI.†

The complex **NIR-Ir** displays intense absorption band below 400 nm, which should be attributed to the spin-allowed C^{4N} ligand-centered LC ($\pi-\pi^*$) transitions (Fig. S1, ESI†). The absorption band in the range of 450–600 nm is assigned to the metal-to-C^{4N} ligand charge transfer (MLCT) and C^{4N} ligand to N^{3N} ligand charge transfer (LLCT) transitions.²⁸ The phosphorescent emission range of **NIR-Ir** lay between 610 and 760 nm with maximal wavelength at 680 nm, which belongs to the NIR region (650–900 nm). The photophysical data of **NIR-Ir** were summarized in Table S1 (ESI†). It should be noted that the absorption and emission bands for **NIR-Ir** were extend to 590 nm and 760 nm, respectively, which afford an advantage for *in vivo* imaging.

The detection of Cys and Hcy by **NIR-Ir** was examined using a spectrophotometric method. The variation of **NIR-Ir** absorption spectra in DMSO-HEPES (pH 7.4, 4 : 1 v/v) upon addition of Cys or Hcy was displayed in Fig. 1. As shown in Fig. 1a, with the addition of Cys or Hcy, the absorbance at 338 decreased gradually and the increase at 495 nm. At the same time, the emission intensity at 684 nm of **NIR-Ir** increased significantly with Cys/Hcy concentration (Fig. 1b). According to previous report, the aldehyde group in **NIR-Ir** was a strong electron-withdrawing

group that can quench the luminescence of Ir(III) complex.^{26,27} After Cys or Hcy reacted with aldehyde groups in **NIR-Ir**, the emission intensity of Ir(III) complex enhanced. Moreover, the emission intensities of **NIR-Ir** (5 μ M) with different concentrations of Cys/Hcy show a linear correlation in the range of 10–90 equiv. of Cys/Hcy (Fig. S2, ESI†), and a linear regression equation of $I = 1.37C_{\text{Cys}} + 124.52$ (unit of C is μ M, $R = 0.984$) for Cys and $I = 1.94C_{\text{Hcy}} + 147.66$ (unit of C is μ M, $R = 0.985$) for Hcy is obtained. The detection limit is calculated to be 13.7 μ M for Cys and 9.7 μ M for Hcy ($S/N = 3$), respectively. These facts indicate that the **NIR-Ir** could serve as an indicator of Cys/Hcy.

High selectivity is very important for an excellent chemosensor. Hence, studies of the selective response of **NIR-Ir** have been carried out with other elementary amino acids. The ratio of photoluminescent intensity at 684 nm (F/F_0) was selected to indicate the changes with or without addition of various amino acids. Amino acids such as Ala, Arg, Asp, Gln, Glu, Gly, GSH, His, Ile, Leu, Lys, Met, Phe, Pro, Ser, Thr, Trp, Tyr, and Val, did not produce detectable spectral changes (Fig. 2, black bars), indicating that the formation of thiazolidine and thiazinane was a key for the selective recognition of Cys and Hcy. Moreover, the competition experiments were performed to evaluate the specificity of **NIR-Ir** for Cys/Hcy over a complex mixture of amino acids. After the addition of 80 equiv. of Cys/Hcy to the above mixture, the phosphorescence intensity significantly increased (Fig. 2, blue and red bars). This competition experiment demonstrated that other elementary amino acids do not interfere with **NIR-Ir** phosphorescent detection of Cys/Hcy.

To seek further detailed information on the binding mechanism, the ¹H NMR interaction of **NIR-Ir** with Hcy/Cys was investigated. When excess Hcy or Cys was added to a solution of **NIR-Ir** in DMSO-d₆, the aldehyde resonance (10.91 ppm) of **NIR-Ir** disappeared, which resulting from thiazinane (or thiazolidine) was formed by the interaction of aldehyde with Hcy (or

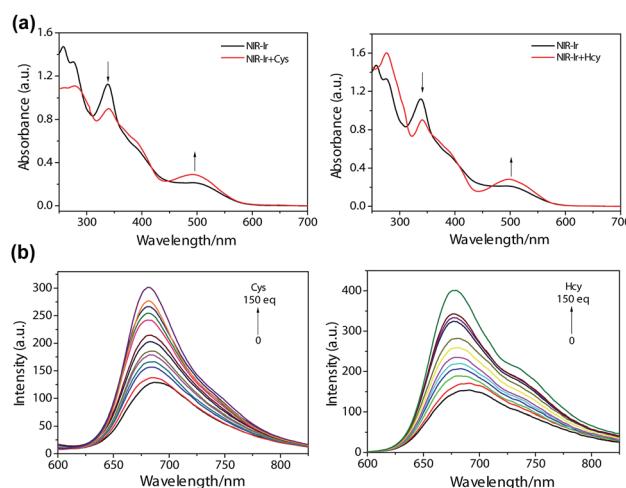


Fig. 1 Changes in absorption (a) and emission (b) spectra of **NIR-Ir** in DMSO-HEPES (4 : 1 v/v) upon addition of various amounts of Cys or Hcy.

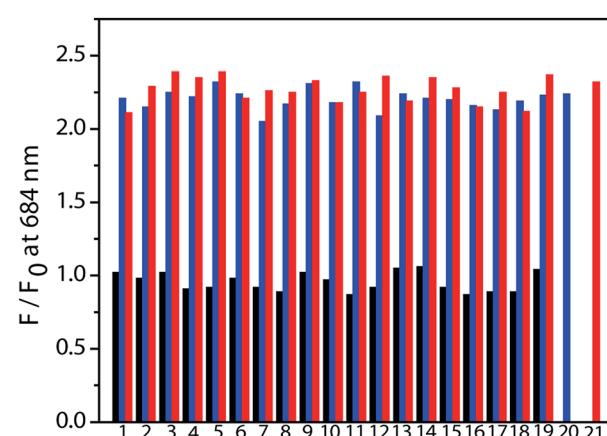


Fig. 2 Luminescent responses of **NIR-Ir** (10 μ M) in DMSO-HEPES (4 : 1 v/v) to various amino acids (80 equiv.) at 37 °C. Bars represent the ratio F/F_0 of luminescence intensity at 670 nm. Black bars represent the addition of 80-fold various amino acids to a 10 μ M solution of **NIR-Ir**. Blue and red bars represent the addition of Cys and Hcy (80 equiv.) to the above solution, respectively. (1) Gly; (2) Glu; (3) Gln; (4) GSH; (5) Met; (6) Arg; (7) Tyr; (8) Ala; (9) Lys; (10) Leu; (11) Pro; (12) Try; (13) Ser; (14) Thr; (15) Asp; (16) Asn; (17) Val; (18) Ile; (19) His; (20) Cys; (21) Hcy.



Cys) (Fig. S3, ESI[†]).²⁷ Meanwhile, calculations based on density functional theory (DFT)⁴⁵ for **NIR-Ir**, **NIR-Ir + Hcy** and **NIR-Ir + Cys** were done to further understand the selectivity of **NIR-Ir** for detecting Hcy/Cys. Electron distributions of the frontier molecular orbitals (MOs) for these Ir(III) complexes are shown in Fig. S4 (ESI[†]). Energies for these frontier MOs as well as their corresponding HOMO–LUMO gaps are listed in Table S2,[†] TD-DFT obtained vertical absorption and emission energies are collected in Table S3.[†] TD-NTO analyses for hole-particle wave function pairs of natural transition orbital for the S_1 and T_1 states are displayed in Fig. S5 and S6 (ESI[†]), respectively. For **NIR-Ir**, HOMO wave function is mainly distributed on the phenyl part of the cyclometalated ligands and the Ir center. The LUMO is localized on the N³⁺N ligand. Similar distributions of the frontier MOs are observed for **NIR-Ir + Cys** and **NIR-Ir + Hcy**. Calculated HOMO–LUMO energy gaps of **NIR-Ir**, **NIR-Ir + Cys**, and **NIR-Ir + Hcy** are 3.00, 3.19 and 3.04 eV, respectively. It means ligand modification has an effect on the frontier MOs. As a result, the vertical $S_0 \rightarrow S_1$ absorption energies for these Ir(III) complexes are also influenced, because the $S_0 \rightarrow S_1$ absorption is dominated by the HOMO to LUMO transition as obtained from further NTO analysis based on the optimized complexes in the ground state (Fig. S5, ESI[†]). For the aim to get more information for emission, we also did NTO analyses for the hole-particle wave function pairs based on the optimized complexes in the triplet state. As shown in Fig. S6 (ESI[†]), it proved that, for these Ir(III) complexes, emission is mainly derived from the $[\pi C^3N-\pi N^3N^*]^3LLCT$ and $[\pi\pi(\text{Ir})-\pi N^3N^*]^3MLCT$ states.

A cellular level fluorescence probe should bear low cytotoxicity and the ability to enter the cell at working concentrations. Herein, the MTT assay was carried out (Fig. S7, ESI[†]). After 24 h incubation with **NIR-Ir**, even at concentrations below 25 μM , which was five-fold the working concentration, the cell viabilities were above 80% signify that the probe with low cytotoxicity.

The ability of **NIR-Ir** to monitor Cys/Hcy within living cells was investigated by confocal fluorescence microscopy. The optical windows at the NIR channel 650–750 nm was chosen as signal output for detection. The reaction was allowed to proceed at 37 °C for 30 min to obtain the adequate interaction between **NIR-Ir** and intracellular Cys/Hcy. As shown in Fig. 3a, MCF-7 cells incubated with 5 μM **NIR-Ir** for 30 min at 37 °C showed marked intracellular luminescence. By contrast, when MCF-7 cells were pretreated with 200 μM *N*-ethylmaleimide (NEM, as a thio-reactive compound) at 37 °C for 1 h, and then incubated with complex **NIR-Ir** for 30 min at 37 °C, MCF-7 cells showed very weak phosphorescence emission (Fig. 3b). Moreover, with the incubation time of **NIR-Ir** increased to 2 h after pre-incubation with *N*-ethylmaleimide, the luminescence intensity increased (Fig. 3c), indicating the appearance of new Cys/Hcy in living MCF-7 cells and further reaction of **NIR-Ir** with the new Hcy/Cys, and **NIR-Ir** can detect the changes of Cys/Hcy level in the living cells. Furthermore, the photobleaching property when using **NIR-Ir** as an imaging probe was investigated at different excitation wavelengths. Under continuous irradiation at 543 or 405 nm with high power density for 10 min, the photoluminescence intensity of **NIR-Ir** was maintained at >90%

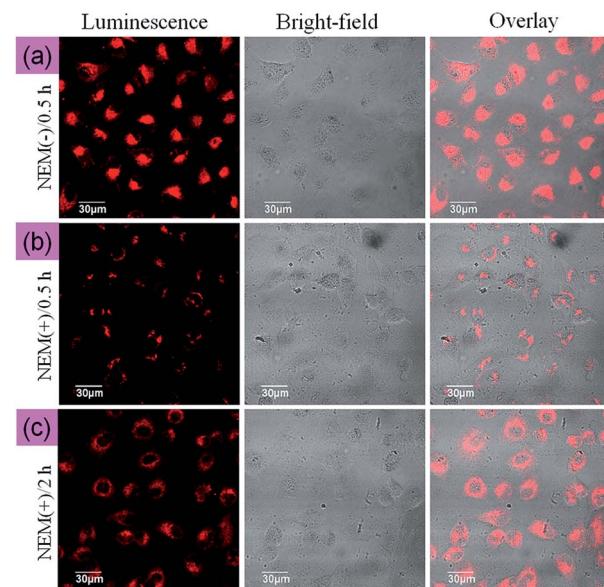


Fig. 3 Luminescence images of **NIR-Ir**-incubated MCF-7 cells. (a) MCF-7 cells incubated with 5 μM **NIR-Ir** for 30 min. (b) MCF-7 cells incubated with 200 μM *N*-ethylmaleimide for 1 h and then further incubated with 5 μM **NIR-Ir** for 30 min. (c) MCF-7 cells incubated with 200 μM *N*-ethylmaleimide for 1 h and then further incubated with 5 μM **NIR-Ir** for 2 h. Emission was collected by an NIR channel at 650–750 nm, excitation at $\lambda_{\text{ex}} = 543$ nm.

and ~75% of its initial value for 543 nm and 405 nm, respectively (Fig. S8, ESI[†]), indicating that excitation at long wavelength benefits the photo-stability of **NIR-Ir** in imaging.

Being encouraged by successful utilization of **NIR-Ir** in imaging Cys/Hcy in living cells, we evaluated the suitability of **NIR-Ir** for monitoring Cys in living mice. **NIR-Ir** (20 μM , 100 μL) and Cys (200 μM , 100 μL) were subcutaneously injected into a living mouse in order. As illustrated in Fig. 4a, after being

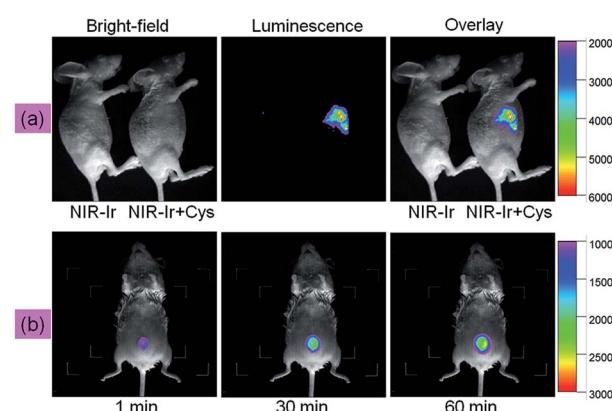


Fig. 4 (a) *In vivo* NIR images of **NIR-Ir**-pretreated living mice injected with 100 μL normal saline (left mice) or 200 μM Cys solution (right mice), the concentration of **NIR-Ir** is 20 μM . (b) Images of a shaved living mice that is given a subcutaneous injection of **NIR-Ir** (20 μM , 100 μL) and Cys (200 μM , 100 μL). Images were taken after incubation for 1, 30, and 60 min, respectively. The luminescence emission were collected at 650–750 nm as detection signals, upon irradiation at 532 nm laser.



incubated for a certain time, luminescence intensity of right mouse increased over time, which can be contributed to the interaction of **NIR-Ir** with Cys. However, in the left control, no obvious fluorescence was observed under the same experimental condition. Furthermore, the emission intensity changes of **NIR-Ir** for Cys sensing with different time were investigated. A shaved living mice was preinjected subcutaneously with Cys (200 μ M, 100 μ L), and then treated with **NIR-Ir** (20 μ M, 100 μ L) *via* subcutaneous injection. As shown in Fig. 4b, the mice exhibited weakened emission signals within 1 min incubation, whereas the mice showed significantly stronger emission intensity with increasing the incubation time for 60 min. The above results confirmed that the NIR probe **NIR-Ir** possessing the unique ability to monitor and image Cys in complicated living organisms.

In conclusion, a novel NIR-emitting iridium(III) complex **NIR-Ir** was designed and employed it as a switch-on chemosensor for detection and imaging of Cys/Hcy. **NIR-Ir** displays distinguishable advantages: compared with present iridium(III) complexes, it is a good candidate for detecting Cys/Hcy *in vitro* and *in vivo* (Table S4†). To the best of our knowledge, this is the first time that a NIR-emitting iridium(III) switch-on probe was used for imaging of Cys/Hcy in small animal (mice). Moreover, in order to eliminate auto-fluorescent from organism, **NIR-Ir** displays a large Stokes shift (>250 nm) and a relatively long emission lifetime that allows it to be distinguished from background auto-fluorescent *via* the use of time-resolved photoluminescence technique (TRPT) or fluorescence lifetime imaging microscopy (FLIM). The present results offer valuable information and create a new platform for designing NIR-emitting probe based on phosphorescent metal complexes for *in vivo* imaging.

Ethical statement

“Animal procedures were in agreement with the guidelines of the Institutional Animal Care and Use Committee of Gannan Normal University and performed in accordance with the institutional guidelines for animal handling. All of the animal experiments were approved by the Science and Technology Committee of Jiangxi Province (China)”.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by Natural Science Foundation of China (21501031, 51478123, 21463003), the Key Technology R&D Program of Jiangxi Province (20143ACG70019), and Natural Science Foundation of Jiangxi Province of China (20151BAB213001).

Notes and references

1 Z. A. Wood, E. Schröder, J. Robin Harris and L. B. Poole, *Trends Biochem. Sci.*, 2003, **28**, 32–40.

- 2 Y. Zhou and J. Yoon, *Chem. Soc. Rev.*, 2012, **41**, 52–67.
- 3 X. Chen, Y. Zhou, X. Peng and J. Yoon, *Chem. Soc. Rev.*, 2010, **39**, 2120–2135.
- 4 H. Refsum, P. M. Ueland, O. Nygård and S. E. Vollset, *Annu. Rev. Med.*, 1998, **49**, 31–62.
- 5 S. Seshadri, A. Beiser, J. Selhub, P. F. Jacques, I. H. Rosenberg, R. B. D'Agostino, P. W. F. Wilson and P. A. Wolf, *N. Engl. J. Med.*, 2002, **346**, 476–483.
- 6 Z. Guo, S. Nam, S. Park and J. Yoon, *Chem. Sci.*, 2012, **3**, 2760–2765.
- 7 Q. Gao, W. Zhang, B. Song, R. Zhang, W. Guo and J. Yuan, *Anal. Chem.*, 2017, **89**, 4517–4524.
- 8 O. Rusin, N. N. S. Luce, R. A. Agbaria, J. O. Escobedo, S. Jiang, I. M. Warner, F. B. Dawan, K. Lian and R. M. Strongin, *J. Am. Chem. Soc.*, 2004, **126**, 438–439.
- 9 (a) Q. Zhao, F. Li and C. Huang, *Chem. Soc. Rev.*, 2010, **39**, 3007–3030; (b) H. Shi, H. Sun, H. Yang, S. Liu, G. Jenkins, W. Feng, F. Li, Q. Zhao, B. Liu and W. Huang, *Adv. Funct. Mater.*, 2013, **23**, 3268–3276; (c) H. Shi, X. Ma, Q. Zhao, B. Liu, Q. Qu, Z. An, Y. Zhao and W. Huang, *Adv. Funct. Mater.*, 2014, **24**, 4823–4830; (d) S. J. Liu, Y. L. Zhang, H. Liang, Z. J. Chen, Z. Y. Liu and Q. Zhao, *Opt. Express*, 2016, **24**, 15757; (e) S. J. Liu, N. Zhou, Z. J. Chen, H. J. Wei, Y. N. Zhu, S. Guo and Q. Zhao, *Opt. Lett.*, 2017, **42**, 13.
- 10 D.-L. Ma, V. P.-Y. Ma, D. S.-H. Chan, K.-H. Leung, H.-Z. He and C.-H. Leung, *Coord. Chem. Rev.*, 2012, **256**, 3087–3113.
- 11 Q. Zhao, C. Huang and F. Li, *Chem. Soc. Rev.*, 2011, **40**, 2508–2524.
- 12 D.-L. Ma, S. Lin, W. Wang, C. Yang and C.-H. Leung, *Chem. Sci.*, 2017, **8**, 878–889.
- 13 Y. Q. Wu, H. Jing, Z. S. Dong, Q. Zhao, H. Z. Wu and F. Y. Li, *Inorg. Chem.*, 2011, **50**, 7412–7420.
- 14 Y. Wu, M. Shi, L. Zhao, W. Feng, F. Li and C. Huang, *Biomaterials*, 2014, **35**, 5830–5839.
- 15 L. Murphy, A. Congreve, L.-O. Palsson and J. A. G. Williams, *Chem. Commun.*, 2010, **46**, 8743–8745.
- 16 W. Wang, Z. Mao, M. Wang, L.-J. Liu, D. W. J. Kwong, C.-H. Leung and D.-L. Ma, *Chem. Commun.*, 2016, **52**, 3611–3614.
- 17 K. K.-S. Tso, K.-K. Leung, H.-W. Liu and K. K.-W. Lo, *Chem. Commun.*, 2016, **52**, 4557–4560.
- 18 J. Liu, C. Jin, B. Yuan, X. Liu, Y. Chen, L. Ji and H. Chao, *Chem. Commun.*, 2017, **53**, 2052–2055.
- 19 J. Yellol, S. A. Perez, G. Yellol, J. Zajac, A. Donaire, G. Vigueras, V. Novohradsky, C. Janiak, V. Brabec and J. Ruiz, *Chem. Commun.*, 2016, **52**, 14165–14168.
- 20 F. Zhang, X. Liang, W. Zhang, Y.-L. Wang, H. Wang, Y. H. Mohammed, B. Song, R. Zhang and J. Yuan, *Biosens. Bioelectron.*, 2017, **87**, 1005–1011.
- 21 W. Zhang, F. Zhang, Y.-L. Wang, B. Song, R. Zhang and J. Yuan, *Inorg. Chem.*, 2017, **56**, 1309–1318.
- 22 Y. Wu, G. Zeng, N. Lvye, W. Wu, T. Jiang, R. Wu, W. Guo, X. Li and X. Fan, *J. Mater. Chem. B*, 2017, **5**, 4973–4980.
- 23 T. S.-M. Tang, H.-W. Liu and K. K.-W. Lo, *Chem. Commun.*, 2017, **53**, 3299–3302.
- 24 H.-Y. Shiu, M.-K. Wong and C.-M. Che, *Chem. Commun.*, 2011, **47**, 4367–4369.

25 Z. Mao, M. Wang, J. Liu, L.-J. Liu, S. M.-Y. Lee, C.-H. Leung and D.-L. Ma, *Chem. Commun.*, 2016, **52**, 4450–4453.

26 H. Chen, Q. Zhao, Y. Wu, F. Li, H. Yang, T. Yi and C. Huang, *Inorg. Chem.*, 2007, **46**, 11075–11081.

27 L. Xiong, Q. Zhao, H. Chen, Y. Wu, Z. Dong, Z. Zhou and F. Li, *Inorg. Chem.*, 2010, **49**, 6402–6408.

28 Y. Ma, S. Liu, H. Yang, Y. Wu, C. Yang, X. Liu, Q. Zhao, H. Wu, J. Liang, F. Li and W. Huang, *J. Mater. Chem.*, 2011, **21**, 18974–18982.

29 X. Cao, Y. Wu, K. Liu, X. Yu, B. Wu, H. Wu, Z. Gong and T. Yi, *J. Mater. Chem.*, 2012, **22**, 2650–2657.

30 X. Liu, N. Xi, S. Liu, Y. Ma, H. Yang, H. Li, J. He, Q. Zhao, F. Li and W. Huang, *J. Mater. Chem.*, 2012, **22**, 7894–7901.

31 S. Liu, W. Qiao, G. Cao, Y. Chen, Y. Ma, Y. Huang, X. Liu, W. Xu, Q. Zhao and W. Huang, *Macromol. Rapid Commun.*, 2013, **34**, 81–86.

32 Y. Ma, S. Liu, H. Yang, Y. Wu, H. Sun, J. Wang, Q. Zhao, F. Li and W. Huang, *J. Mater. Chem. B*, 2013, **1**, 319–329.

33 Y. Tang, H.-R. Yang, H.-B. Sun, S.-J. Liu, J.-X. Wang, Q. Zhao, X.-M. Liu, W.-J. Xu, S.-B. Li and W. Huang, *Chem.-Eur. J.*, 2013, **19**, 1311–1319.

34 (a) W. Xu, X. Zhao, W. Lv, H. Yang, S. Liu, H. Liang, Z. Tu, H. Xu, W. Qiao, Q. Zhao and W. Huang, *Adv. Healthcare Mater.*, 2014, **3**, 658–669; (b) S. J. Liu, A. Q. Xu, Z. J. Chen, Y. Ma, H. R. Yang, Z. J. Shi and Q. Zhao, *Opt. Express*, 2016, **24**, 28247.

35 G. Li, Y. Chen, J. Wu, L. Ji and H. Chao, *Chem. Commun.*, 2013, **49**, 2040–2042.

36 N. Zhao, Y.-H. Wu, L.-X. Shi, Q.-P. Lin and Z.-N. Chen, *Dalton Trans.*, 2010, **39**, 8288–8295.

37 K. Huang, I. W. Bulik and A. A. Marti, *Chem. Commun.*, 2012, **48**, 11760–11762.

38 K. Chen and M. Schmittel, *Analyst*, 2013, **138**, 6742–6745.

39 H. Gao, Z. Li, Y. Zhao, H. Qi and C. Zhang, *Sens. Actuators, B*, 2017, **245**, 853–859.

40 H. Chen, B. Dong, Y. Tang and W. Lin, *Acc. Chem. Res.*, 2017, **50**, 1410–1422.

41 F. Kong, Z. Liang, D. Luan, X. Liu, K. Xu and B. Tang, *Anal. Chem.*, 2016, **88**, 6450–6456.

42 J.-Y. Xie, C.-Y. Li, Y.-F. Li, J. Fei, F. Xu, J. Ou-Yang and J. Liu, *Anal. Chem.*, 2016, **88**, 9746–9752.

43 J. H. Palmer, T. Brock-Nannestad, A. Mahammed, A. C. Durrell, D. VanderVelde, S. Virgil, Z. Gross and H. B. Gray, *Angew. Chem., Int. Ed.*, 2011, **50**, 9433–9436.

44 G. Zhang, H. Zhang, Y. Gao, R. Tao, L. Xin, J. Yi, F. Li, W. Liu and J. Qiao, *Organometallics*, 2014, **33**, 61–68.

45 A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648–5652.

