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Mn₃O₄ nanoparticles cause endoplasmic reticulum stress-dependent toxicity to *Saccharomyces* cerevisiae

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 Mn_3O_4 nanoparticles (NPs) are a significant nanomaterial (NM) due to their excellent physiochemical properties. However, little is known about their biological effects. In this study, we investigated the effect of the synthesized Mn_3O_4 nanoparticles (NPs) (with the size of 10-25 nm) on the important fungus model, Saccharomyces cerevisiae. Growth inhibition assays showed that Mn_3O_4 NPs had dosedependent toxicity to Saccharomyces cerevisiae (IC $_{50}=340$ ppm). The plasma membrane (PM) was not damaged by the NPs, and the addition of ROS scavengers could not attenuate growth inhibition of the NPs to yeast cells, ruling out the contribution of PM damage and oxidative stress to this toxicity. Interestingly, Mn_3O_4 NPs caused HAC1 mRNA splicing and remarkable up-regulation of the unfolded protein response (UPR) genes, indicating that the NPs induce severe endoplasmic reticulum (ER) stress. Moreover, treatment of the NPs severely reduced the activity of both extracellular invertase and surface ferric reductase, which might be attributable to ER stress-related disruption of the secretion pathway. This study uncovers a novel toxicity mechanism of Mn_3O_4 NPs against eukaryotic cells, and provides useful information for assessing the environmental impact of NMs.

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1. Introduction

Oxides of manganese are suitable for super-capacitor applications due to their natural abundance, environmentally friendly nature, and the ability of manganese to exist in various states.¹ The various oxidation states of manganese result in the formation of MnO₂, Mn₂O₃, and Mn₃O₄.¹ Mn₃O₄ nanoparticles (NPs) are one of the most important nanomaterials (NMs) due to their superior electrochemical properties. Besides their application in super-capacitors, they are also used in various fields, such as paints, rubbers, sensors and computer components.² Moreover, Mn₃O₄ NPs have been implemented as catalysts for the oxidation of methane and carbon monoxide (CO)³ and for the abatement of volatile organic compounds and waste gases.⁴ Therefore, the synthesis and application of Mn₃O₄ NPs has become a hot spot in today's research.

With the development of nanotechnology and abundant NMs being incorporated into ecosystems, it is essential and urgent to understand the potential biological impact of NMs.⁵ Because of their small sizes and large surface energy, NMs have much higher biological activity than bulk materials,⁶ they may easily enter into cells through free penetration or receptormediated endocytosis, and actively interact with intracellular components.⁷ These interactions may lead to dysfunction of

Key Laboratory of Molecular Microbiology and Technology, Ministry of Education, College of Life Science, Nankai University, Tianjin, 300071, P. R. China. E-mail: yuqilin033@163.com; nklimingchun@163.com; Tel: +86-022-23508506 protein functions, damage of DNA, interference of signalling pathways and especially accumulation of reactive oxygen species (ROS).^{8,9} Similar to many other NMs, Mn₃O₄ NPs have been demonstrated to be toxic to mammalian cells, which is associated with ROS production. Although the toxicity of Mn₃O₄ NPs has been revealed, the detailed toxicity mechanisms of Mn₃O₄ NPs remain to be investigated.

The ER, as a vital organelle, plays important roles in protein processing and modification, lipid synthesis and signalling communication.10 Owing to its significance in various cellular processes, dysfunction of the ER induces growth inhibition and even apoptosis. ER function may be disturbed by inhibition of protein glycosylation, reduction of disulfide bond formation, calcium depletion from the ER lumen, impairment of protein transport from the ER to the Golgi apparatus, and expression of misfolded proteins.11 Such dysfunction leads to ER stress.12 ER dysfunction contributes to abundant acute and chronic diseases, such as neurodegenerative diseases, diabetes and Alzheimer's disease.13 To alleviate this stress, a conserved signalling pathway, termed the unfolded protein response (UPR), was activated.14 Up to now, little is known about the contribution of ER dysfunction to the toxicity mechanism of nanomaterials.

Saccharomyces cerevisiae is a promising unicellular eukaryotic model organism for evaluating the toxicological effect of nanoparticles, as the cellular structure and functional organization of this fungus share many similarities with those of mammalian organisms.¹⁵ In this study, we for the first time investigated the

potential effect of the synthesized Mn_3O_4 nanoparticles on this representative fungal organism, and explored possible toxicity mechanisms of these nanomaterials in the fungal cells. Interestingly, we found that ER stress, rather than the well-known plasma membrane (PM) damage and ROS accumulation, contributed to the toxicity of Mn_3O_4 nanoparticles.

2. Materials and methods

2.1 Preparation and characterization of Mn₃O₄ NPs

The $\rm Mn_3O_4$ NPs were synthesized according to Liu's method. ¹⁶ The general morphology of the products was characterized by transmission electron microscopy (TEM, Tecnai G2 F-20, FEI, USA). The crystal structure and composition of the products were characterized by X-ray diffraction (XRD, D/max-2500, Japan).

2.2 Preparation of Mn₃O₄ NP solutions

 $\rm Mn_3O_4$ NPs stock solution was prepared in YPD medium (yeast extract 1%, peptone 2%, glucose 2%) or SC-sucrose medium (yeast nitrogen source 0.67%, sucrose 2%, amino acid mixture 0.2%) with an initial concentration of 10, 000 ppm. Then the stock solution was sonicated for 30 min (AS3120, Auto science, China) and a series of 2-fold dilutions were prepared in the corresponding medium.

2.3 Yeast strains and growth conditions

Usually, the wild-type strain InvSc1 (Invitrogen, USA) was used in each experiment. In order to detect the damage to the endoplasmic reticulum of the Mn_3O_4 NPs, the ER stress reporting strain WT+pJC104 which contains the plasmid PFKS2-LacZ and the hac1 Δ mutant to detect its sensitivity to Mn_3O_4 NPs.

Growth inhibition was assessed by hemocytometer counting. Overnight-cultured yeast cells were suspended in fresh YPD medium to an optical density at 600 nm (OD₆₀₀) of 0.2. 1 mL of cell suspension was mixed with 1 mL of diluted Mn₃O₄ NPs suspension in glass tubes, and the final concentrations of the NPs were 0, 100, 400, 800 ppm. After cultured with shaking at 140 rpm and 30 $^{\circ}$ C for 12 h, the cells in each tube were counted using a hemocytometer. The percent of growth was calculated as the number of cells in each group divided by the number of cells in the control ×100. Since the IC₅₀ of the Mn₃O₄ NPs is 340 ppm (as described in the Results), we used this concentration in the following experiments to explore the toxicity mechanisms of the NPs. To evaluate the effect of ROS scavengers on cell growth under Mn₃O₄ NP treatment, 10 mM N-acetyl cysteine (NAC), 10 mM thiourea or 10 mM vitamin C (VC) were added into the co-incubation system.

2.4 Mn²⁺ dissolution assays

To assess $\rm Mn^{2+}$ dissolution from $\rm Mn_3O_4$ NPs, the nanoparticles were suspended in yeast culturing supernatant with the concentrations of 400 ppm. The mixtures were subsequently cultured at 30 °C with shaking at 140 rpm for 12 h. Then the cells were harvested, washed with deionized water several times and suspended. The suspensions were repeatedly frozen and

thawed, and then were vortexed violently for 20 minutes to thoroughly release the intracellular ${\rm Mn}^{2^+}$. After centrifugation at 12 000 rpm for 10 min to pellet ${\rm Mn}_3{\rm O}_4$ NPs, the supernatants were digested by 10% (v/v) HNO₃, and ${\rm Mn}^{2^+}$ contents were then determined using inductively coupled plasma (ICP-AES, Thermo Elemental, USA).

2.5 PM damage assays

Propidium iodide (PI) staining was used to determine the effect of Mn_3O_4 NPs on PM damage. Yeast cells were co-cultured with Mn_3O_4 NPs for 12 h, harvested, suspended in PBS buffer, and stained with 10 μg mL $^{-1}$ of PI for 5 min. The cells were then harvested and observed by a fluorescence microscope (BX-51, Olympus, Japan) with the RFP filter set. The percent of PI-positive (PM damaged) cells was calculated as the number of PI-positive cells divided by the number of total cells $\times 100$. At least 40 fields were counted.

2.6 ROS detection

Cells were treated with Mn_3O_4 NPs shaking at 140 rpm for 12 h. The cells were then washed and suspended in PBS. 500 mL of suspensions were stained with 10 µg mL $^{-1}$ of dichlorofluorescein diacetate (DCFH-DA, dissolved in ethanol, Sigma, USA) at 30 °C for 30 min. The cells were harvested, washed twice with PBS buffer, and then observed by a fluorescence microscope (BX-51, Olympus, Japan) with the RFP filter set. The percent of ROS-positive cells was calculated as the number of ROS-positive cells divided by the number of total cells $\times 100$. At least 40 fields were counted.

2.7 ER damage detection and β -galactosidase assays

The ER stress reporting strain WT+pJC104 were treated with Mn₃O₄ NPs as described above. Cells were then harvested and suspended in 1 mL working Z buffer (60 mM Na₂HPO₄, 40 mM NaH₂PO₄, 10 mM KCl, 1 mM MgSO₄, 0.027% (v/v) β-mercaptoethanol, pH 7.0). 150 μL of suspensions were permeabilized with 20 μL chloroform and 50 μL SDS (0.1%, m/v) at 30 °C for 5 min, mixed with 500 μL *O*-nitrophenyl-*b*-D-galactopyranoside (ONPG, 5000 ppm, BBI, USA), and incubated at 30 °C for certain time (*T*). Reactions were stopped by addition of 500 mL Na₂CO₃ (1 M) when the mixtures turned yellow. The cell number of the suspensions was also detected by hemocytometers. Suspensions were centrifuged at 12 000 rpm for 10 min, and the optical density of the supernatants at 420 nm (OD₄₂₀) was determined. Miller units of activity were calculated as $(OD_{420} \times 1000)/(cell number \times T \times 10^{-7})$.

2.8 HAC1 mRNA splicing

A pair of primers for detection of *HAC1* mRNA splicing were designed on the outside of the intron, HAC1-5rt (CCGTAGA CAACAACAATTTG) and HAC1-3rt (CATGAAGTGATGAAGA AATC). *HAC1* products were obtained by PCR using the cDNA template. The size of the intron-containing *HAC1* fragment was 433 bp and when the UPR pathway was activated, the size of the

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spliced HAC1 fragment was 181 bp. The PCR products were separated by 2% agarose gel electrophoresis.

2.9 Quantitative real-time PCR (qRT-PCR)

The treated cells were harvested, and total RNA was extracted from the cells as previously described. An oligo (dT)-primer RT reagent Kit (Promega, Madson, USA) was used for reverse transcriptional synthesis of cDNA, and a Trans Start Green qPCR Supermix Kit (TRANSGEN, Beijing, China) was used for qRT-PCR. Transcription levels of the genes involved in the UPR genes, including PMT4 (encoding protein mannosyl transferase), YSY6 (encoding a homolog of mammalian RAMP4 protein involved in secretion), GAA1 (encoding a subunit of the GPI: protein transamidase complex), ERD2 (encoding an HDEL receptor),17 INO1 (encoding protein inositol-3-phosphate synthase) and SAC6 (encoding protein fimbrin), were normalized against the levels of ACT1 in different samples. The primers used for the reactions are listed in Table 1. The $2^{-\Delta\Delta C_t}$ method was used to calculate the relative expression compared to the control.

2.10 Invertase and ferric reductases assay

Invertase activity was estimated by an invertase analysis kit (Suzhou Keming Biotechnology Co., Ltd). The activity of cellsurface ferric reductases was detected using the bathophenanthroline disulfonic acid (BPS) method.18

Table 1 Primers used in qRT-PCR

Primer	Sequence (5′–3′)
ACT1-5rt	TGTCTTCCCATCTATCGTCG
ACT1-3rt	CGACGATAGATGGGAAGACA
PMT4-5rt	GTTGCTGAACATTGGCTCT
PMT4-3rt	AAGGAACCATCATAGCCACA
INO1-5rt	GACTTTGTCGTCTCTGGTTG
INO1-3rt	AAGAGGCTTCACCAAGGAC
SAC6-5rt	AGCAAGCAAACTGGAACAG
SAC6-3rt	GTGGTTTGTAATGGTGCCTT
GAA1-5rt	TCGGCACATTGGCATTTCCT
GAA1-3rt	GCTCTATCTCATCTTTCGGC
ERD2-5rt	AGAACCAACACCATTGCG
ERD2-3rt	ACACTCTCCAACCATACAGA
YSY6-5rt	ACAGACACCAAGACAAAGAC
YSY6-3rt	CCACCTACGAGAAGAAACAG
FKS2-5rt	CGATTGATTGGGTGAGACG
FKS2-3rt	AAATCTTTGTGTCGCCTTCC
PRM5-5rt	TCCACACACATACCCAGT
PRM5-3rt	TTTCTGTTGGATTCGGATGC
CHS1-5rt	CAGACCCGTAAACGATGGA
CHS1-3rt	CGATAGAAGACACCGAGGAT
CHS3-5rt	CGATTGTGGCTTTCCTGAC
CHS3-3rt	CTTTACCAGCATCTGACCAG
DFG5-5rt	AGGTTGGTTGGCAATGGTA
DFG5-3rt	CCGCAATCTGGAACAAACA
CRH1-5rt	AACCAACTACAATGACGCTC
CRH1-3rt	AACTGCGAAATCTTCTTGGG
ECM33-5rt	GTTGGTGGTGGTTTCATCAT
ECM33-3rt	GCACCACCTCTAACAGACT
SUC2-5rt	ATTGCCTGGGCTTCAAACTG
SUC2-3rt	GACCAGGACCAGCATTACT

2.11 Determination of residual sucrose content in culture medium

Cells were treated with or without 400 ppm Mn₃O₄ NPs, shaking at 140 rpm for 12 h. The cultures were then centrifuged, and then 0.1 mL of the supernatant were mixed with 0.1 mL of 30% KOH, incubated for 10 min at 100 °C. After cooled to the room temperature, 3 mL of the anthrone reagent was added into the mixture, co-incubated at 40 °C for 10-15 min. The optical density at 620 nm (OD₆₂₀) was detected. Based on the standard curve, the contents of residual sucrose in the culture medium were calculated.

2.12 Statistical analysis

Each text was performed with three replicates, and the values represent the means \pm standard deviation (SD) of three experiments. Significant difference between the treatments was determined using the t-test (P < 0.05). All statistical analyses were performed by Statistical Packages for the Social Sciences (SPSS, Version 20).

3. Results and discussion

Characterization of the synthesized Mn₃O₄ NPs

Transmission electron microscope (TEM) images and XRD data of the synthesized Mn₃O₄ nanoparticles are shown in Fig. 1. The samples displayed nanoparticle structure with cube-like morphology and with the dimension of 10-25 nm (Fig. 1a). XRD analysis revealed that the obtained nanoparticles were indexed to the pure Mn₃O₄ structures (Fig. 1b).

3.2 Mn₃O₄ NPs show toxicity to S. cerevisiae

As demonstrated in Fig. 2, the NPs had inhibitory effect on yeast growth, and 340 ppm leads to approximate 50% decrease of the growth biomass ($IC_{50} = 340$ ppm). With increase of concentration of Mn₃O₄ NPs, the growth inhibition effect was enhanced. In the following experiments, we used 400 ppm of the NPs for exploring their toxicity mechanisms.

3.3 Both PM damage and oxidative stress do not contribute to the toxicity of Mn₃O₄ NPs

PM damage and oxidative stress are two well-known mechanisms by which NPs cause cytotoxicity.9 However, PI staining, which indicates PM damage, showed that only <1% cells were damaged by Mn₃O₄ NPs, even the concentrations reached to 800 ppm (Fig. 3). Hence, PM damage is not involved in the toxicity mechanism of Mn₃O₄ NPs.

To evaluate whether ROS production and consequent oxidative stress are associated with the toxicity of Mn₃O₄ NPs, we measured ROS production in the Mn₃O₄ NP-treated yeast cells by DCFH-DA staining. Fluorescence microscopy demonstrated that the number of ROS-accumulated cells increased with the increase of the concentration of Mn₃O₄ NPs (Fig. 4a). The number of ROS-accumulated cells treated with 400 ppm Mn₃O₄ NPs is 6 to 7 times as much as control (Fig. 4a). Therefore, Mn₃O₄ NPs may lead to intracellular ROS accumulation.

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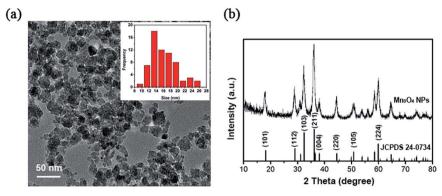


Fig. 1 Characterization of the synthesized Mn₃O₄ NPs. (a) Transmission electron microscopy (TEM). (b) XRD patterns.

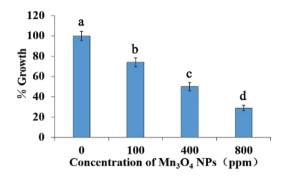


Fig. 2 Growth inhibition of the synthesized $\rm Mn_3O_4$ NPs to *S. cerevisiae*. The yeast cells were co-incubated with different concentrations of $\rm Mn_3O_4$ NPs for 12 h and counted. The percent of growth in each group was calculated as the cell number of group divided by the control group $\times 100$. Error bars indicate the standard deviations (n=3). Identical letters indicate no significant difference between the treatments (P < 0.05).

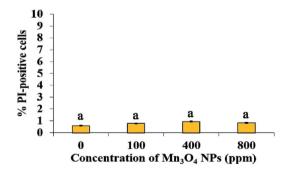


Fig. 3 Mn_3O_4 NPs did not cause obvious PM damage. The treated cells were stained with PI and observed by fluorescence microscopy. The percent of PI-positive cells was calculated. Error bars indicate the standard deviations (n=3). Identical letters indicate no significant difference between the treatments (P < 0.05).

However, the addition of ROS scavengers, including NAC, thiourea and VC, could not restore the growth of yeast cells under $\rm Mn_3O_4$ NP treatment (Fig. 4b). These results indicated that ROS accumulation caused by $\rm Mn_3O_4$ NPs was not involved in the toxicity mechanism of these NPs.

3.4 Mn²⁺ dissolution only partially contribute to the toxicity

Ion dissolution is an important factor that leads to the toxicity of NMs. ¹⁹ The released ions released from NPs may result in lysosomal and mitochondrial damage and ultimately cell death. ²⁰ We further determined Mn²⁺ dissolution from the synthesized Mn₃O₄ NPs. ICP assays showed that Mn²⁺ dissolution was enhanced with the increase of Mn₃O₄ NP concentrations. Approximately 1.699 ppm of Mn²⁺ dissolved from 400 ppm of the Mn₃O₄ NPs and absorbed by cells (Fig. 5a), revealing that only 0.42% of manganese released from the materials at this concentration enters the cell for toxicity.

To investigate whether Mn^{2+} dissolution is associated with the toxicity of the NPs, we then tested the effect of the dissolved Mn^{2+} (1.699 ppm) on yeast growth. After 12 h of incubation, only 20% growth was inhibited by the dissolved Mn^{2+} , while 50% growth was inhibited by Mn_3O_4 NPs (Fig. 5b). Therefore, Mn^{2+} dissolution only partially contributed to the toxicity of Mn_3O_4 NPs.

3.5 Mn₃O₄ NPs cause severe ER stress

The ER plays important roles in abundant intracellular processes, such as protein synthesis, folding, modification, and transport. The protein-folding capacity of the ER is adaptable: when the capacity of the ER is exceeded and, as a result, unfolded proteins accumulate in the ER, an intracellular signalling pathway, the unfolded protein response (UPR), is induced.²¹ ER stress may result in dysfunction of the ER, leading to growth inhibition and apoptosis.^{22,23} We evaluated ER stress using the reporting strain WT+pJC104.²⁴ β-Galactosidase assays showed that Mn₃O₄ NPs remarkably up-regulated the expression of the UPR promoter-governed *LacZ* gene, indicating that ER stress was exerted by Mn₃O₄ NPs (Fig. 6a).

In yeast, the expression of UPR target genes is controlled by the UPR-specific transcription factor Hac1p.²⁴ The splicing of mRNA by Ire1p is a key step in the activation of the classical UPR pathway.²⁵ The expression of Hac1p is posttranscriptionally regulated. Dimer Irelp has endonuclease activity, splicing *HAC1* mRNA introns. Haclp is a transcription factory, entering into the nucleus to activate the expression of UPR genes.²⁶ Here, we analysed *HAC1* mRNA splicing in the experimental group treated with 400 ppm Mn₃O₄ NPs and control group. The results

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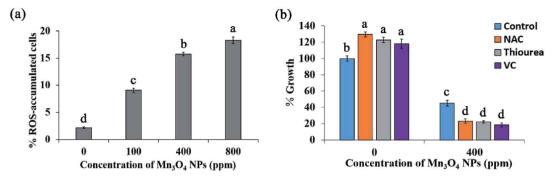


Fig. 4 Mn_3O_4 NPs caused ROS accumulation that is not involved in the toxicity of the NPs. (a) ROS accumulation. The treated yeast cells were stained with DCFH-DA, and observed by fluorescence microscopy. The cells displaying whole-cell green fluorescence were ROS-positive cells. The percent of ROS-accumulated cells was then calculated. (b) Addition of ROS scavengers could not restore the growth of NP-treated cells. The yeast cells were treated with 400 ppm Mn_3O_4 NPs or not (control), 10 mM N-acetyl cysteine and 400 ppm Mn_3O_4 NPs plus 10 mM NAC, 10 mM thiourea and 400 ppm Mn_3O_4 NPs plus 10 mM thiourea, 10 mM VC and 400 ppm Mn_3O_4 NPs plus 10 mM VC (VC) for 12 h, and then the cells were counted. The percent of growth in each group was calculated as the cell number of group divided by the control group $\times 100$. The error bars indicate the standard deviations (n = 3). Identical letters indicate no significant difference between the treatments (P < 0.05).

showed that the size of the control group was 433 bp, with no splicing occurred. In contrast, under the treatment of NPs, HAC1 mRNA was significant spliced. This indicated the activation of ER stress with higher activity in 400 ppm NP treated strains (Fig. 6b). Moreover, we used the hac1 Δ mutant to detect its sensitivity to Mn₃O₄ NPs. As shown in Fig. 6c, the mutant exhibited higher sensitivity to Mn₃O₄ NPs than the wild-type strain, suggesting the important role of UPR in tolerance to Mn₃O₄ NP-caused ER stress.

To confirm Mn₃O₄ NPs caused ER stress, we further detected expression of several UPR genes, including *PMT4*, *INO1*, *SAC6*, *YSY6*, *GAA1* and *ERD2*. qRT-PCR analysis revealed that these UPR genes were remarkably up-regulated by Mn₃O₄ NP treatment (Fig. 6d). These results indicated that Mn₃O₄ NPs caused severe ER stress and consequent activation of the UPR pathway.

3.6 Mn₃O₄ NPs attenuate protein secretion

The yeast cells secrete abundant proteins from the ER to the cell surface, such as the invertase and ferric reductase, maintaining the regular absorption of essential nutrients.²⁷ Since the ER

governs protein secretion,28 we supposed that Mn3O4 NPs may have an impact on protein secretion by causing ER stress. To verify this, we first tested the activity of invertase with treatment of Mn₃O₄ NPs in the medium with sucrose as the sole carbon source. The yeast invertase, which is encoded by the SUC2 gene, is translated in the ER and then secreted from this organelle to the extracellular environment, functioning in utilization of extracellular sucrose.29 qRT-PCR showed that the NP-treated cells had decreased expression of SUC2 (Fig. 7a), suggesting that the NPs rendered reduction of cellular total invertase activity. However, the treated cells showed significant higher intracellular invertase activity (Fig. 7b). In addition, NP treatment led to significant decrease of sucrose uptake into the yeast cells (Fig. 7c), confirming that the NPs reduced the activity of extracellular invertase. These results indicated that the NPs severely reduced secretion of intracellular invertase.

Ferric reductase is also expressed in the ER and secreted to the cell surface for mediating absorption of extracellular ferric ions. BPS staining revealed that the NP-treated cells had significant decreased activity of cell-surface ferric reductase

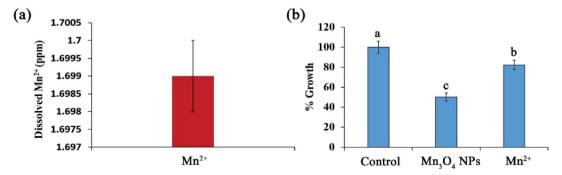


Fig. 5 Dissolution of Mn^{2+} only partially contribute to the toxicity of Mn_3O_4 NPs. (a) Dissolution of Mn^{2+} . The yeast suspensions were mixed with 400 ppm Mn_3O_4 NPs, co-incubated for 12 h, and the supernatant was used for detection of dissolved Mn^{2+} . (b) Growth inhibition of dissolved Mn^{2+} to yeast growth. The yeast cells were co-incubated with 400 ppm Mn_3O_4 NPs or 1.699 ppm Mn^{2+} (dissolved concentration of Mn^{2+}) for 12 h and counted. The percent of growth in each group was calculated. The error bars indicate standard deviations, n=3. Identical letters indicate no significant differences among treatments (P < 0.05).

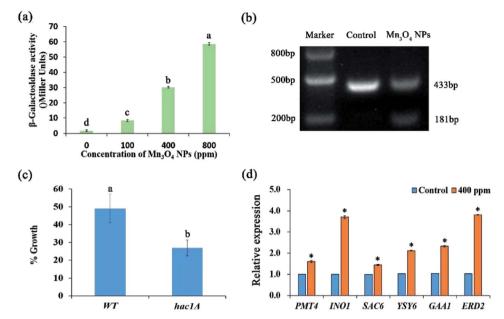


Fig. 6 Mn_3O_4 NPs remarkably activate the UPR pathway. (a) Expression of the UPR promoter-governed expression of β -galactosidase. The yeast cells containing the UPR reporting plasmid pJC104 were treated with different concentrations of Mn_3O_4 NPs, and β -galactosidase activity was determined. (b) The level of HACI mRNA splicing in the strain. The RT-PCR product was 433 bp, indicating that the mRNA was not spliced. The product is 188 bp, indicating that the mRNA is spliced. (c) The yeast cells were co-incubated with different concentrations of Mn_3O_4 NPs for 12 h and counted. The percent of growth in each group was calculated as the cell number of group divided by the control group $\times 100$. (d) The InvSc1 cells were treated with 400 ppm Mn₃O₄ NPs, and total RNA and cDNA were prepared. Transcription levels of the tested genes were normalized against the levels of ACT1 in different samples. The error bars indicate standard deviations, n = 3. Identical letters indicate no significant differences among treatments (P < 0.05). * in (d) indicates significant difference between the treatment and the control (P < 0.05).

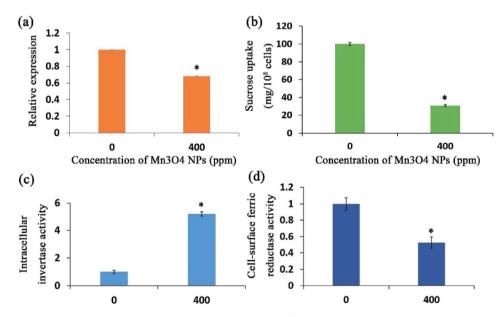


Fig. 7 Mn_3O_4 NPs attenuated secretion of both invertase and ferric reductase. (a) Expression of SUC2. The treated cells were harvested, and total RNA and cDNA were prepared. The transcription levels of SUC2 were detected by RT-PCR. (b) Intracellular invertase activity. The yeast cells were treated with 400 ppm Mn₃O₄ NPs for 12 h, and the intracellular invertase activity was detected by an invertase analysis kit. (c) Sucrose utilization. The yeast cells were treated with 400 ppm Mn_3O_4 NPs for 12 h, and then the culture was centrifuged, obtaining the supernatant to determine the sucrose content. The decreased sucrose content in the medium was calculated. (d) Cell-surface ferric reductase activity. The yeast cells were treated by the NPs and then harvested for assays. The error bars indicate standard deviations, n = 3. *Significant difference between the treatment and the control (P < 0.05)

(Fig. 7d). Hence, the NP treatment also severely reduced secretion of ferric reductase to the cell surface. Taken together, the Mn₃O₄ NPs attenuated protein secretion in the yeast cells.

4. Discussion

Because of the widespread applications of nanomaterials, their toxicity to the living beings, especially those exposed to emissions, is of great concern. However, as one of the main decomposers, the fungal population is usually neglected in evaluating the toxicity of nanoparticles. Fungi are the principal decomposers in the ecosystem, and play an essential role in maintaining ecological cycles and balances. In this study, we investigate the possible toxicity of Mn₃O₄ NPs, one kind of popular nanomaterials, on the model fungus *Saccharomyces cerevisiae*, and investigated the toxic mechanisms. Growth inhibition assays showed that the synthesized Mn₃O₄ NPs have inhibitory effect on yeast cells. Therefore, the potential hazardous effect of Mn₃O₄ NPs to the fungal populations should be considered when they were released into the environment.

Some studies have shown that the toxicity of Mn_3O_4 NPs is related to ROS accumulation, mitochondrial damage, and ion dissolution. In many types of organisms, the mitochondria are one of the main sources of ROS, and ROS scavengers can remove free radicals reduce oxidative stress and hence attenuate the toxicity of the NPs. However, in this study, although Mn_3O_4 NPs caused ROS accumulation, addition of ROS scavengers, including NAC, thiourea and VC, did not restore the yeast growth under the treatment of Mn_3O_4 NPs. Therefore, we supposed that ROS is the by-product of Mn_3O_4 NPs, but is not the cause of the toxicity of the NPs. In the same of the toxicity of the NPs. In the same of the same of the toxicity of the NPs. In the same of the sa

The ER, which functions in protein folding and assembly, lipid biosynthesis, vesicular traffic, and cellular calcium storage, is sensitive to alterations of homeostasis. Proper ER function is essential for cell survival, and perturbation of its function induces cellular damage and results in apoptosis.22 ER function can be disturbed by inhibition of protein glycosylation, a reduction in disulfide bond formation, calcium depletion from the ER lumen, impairment of protein transport from the ER to the Golgi apparatus, and the expression of misfolded proteins. Such dysfunction causes proteotoxicity in the ER, collectively termed ER stress.32 In our study, we found that the yeast strain which contain LacZ reporter plasmid pJC104 expressed high levels of β-galactosidase activity, HAC1 mRNA splicing significantly and abundant UPR genes were also upregulated by Mn₃O₄ NPs. These results indicated that the Mn₃O₄ NPs will cause ER stress and consequent activation of the UPR pathway (Fig. 8). Since the ER function is essential for cell growth, the NP-caused ER stress may lead to decreased cell growth.

One of the most important function of the ER is regulation of protein secretion. We here selected two well-known secreted proteins, invertase and ferric reductase to intestate the effect of Mn_3O_4 NPs on protein secretion. The extracellular invertase is responsible for the extracellular hydrolysis of sucrose³³ and permits yeasts to use sucrose as a carbon and energy source. We first tested the expression of the gene and found that the

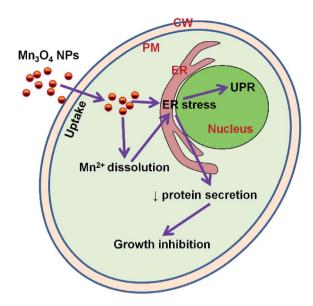


Fig. 8 Model figure illustrating the toxicity mechanism of Mn_3O_4 NPs to yeast cells. ER, endoplasmic reticulum; PM, plasma membrane; CW, cell wall; UPR, unfolded protein response.

expression of the gene was down-regulated. Then we examined the intracellular invertase activity. As expected, the intracellular enzyme activity increased significantly under Mn₃O₄ NP treatment. As the ER function is impaired, invertase secretion process is blocked, leading to invertase accumulation in the cells, followed by reduced absorption of sucrose and decreased cell growth. Ferric reductases is mainly related to the absorption of iron, Fe³⁺ can be reduced to cells can be used to facilitate the form of Fe²⁺ absorption of cells. ³⁴ In yeast cells, ferric reductases is mainly transported to the cell membrane governed by the ER. Decreased secretion of this enzyme may reduce uptake of extracellular Fe³⁺, which may also affect cell growth. Therefore, it can be speculated that Mn₃O₄ NPs cause ER stress leading to decreased protein secretion that is required for cell growth, and consequently result in decreased cell growth (Fig. 8).

5. Conclusions

This study demonstrates that Mn₃O₄ NPs have toxicity to the fungal model organism, *Saccharomyces cerevisiae*. This toxicity is not attributed to PM damage and oxidative stress, but is dependent on ER stress. This study reveals a novel toxicity mechanism of NMs in eukaryotic cells.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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References

- 1 S. Nagamuthu, S. Vijayakumar and G. Muralidharan, *Energy Fuels*, 2013, 27, 3508.
- 2 Z.-Y. Tian, P. Mountapmbeme Kouotou, N. Bahlawane and P. H. Tchoua Ngamou, *J. Phys. Chem. C*, 2013, **117**, 6218.
- 3 E. Stobbe, B. De Boer and J. Geus, Catal. Today, 1999, 47, 161.
- 4 M. Baldi, E. Finocchio, F. Milella and G. Busca, *Appl. Catal., B*, 1998, **16**, 43; M. Baldi, E. Finocchio, C. Pistarino and G. Busca, *Appl. Catal., A*, 1998, **173**, 61.
- 5 R. D. Handy, F. Von der Kammer, J. R. Lead, M. Hassellöv, R. Owen and M. Crane, *Ecotoxicology*, 2008, 17, 287; Y. Bai, Y. Zhang, J. Zhang, Q. Mu, W. Zhang, E. R. Butch, S. E. Snyder and B. Yan, *Nat. Nanotechnol.*, 2010, 5, 683.
- 6 L.-C. Cheng, X. Jiang, J. Wang, C. Chen and R.-S. Liu, Nanoscale, 2013, 5, 3547.
- 7 L. Zhang, F. Gu, J. Chan, A. Wang, R. Langer and O. Farokhzad, *Clin. Pharmacol. Ther. Ser.*, 2008, 83, 761.
- 8 M. S. Wason, J. Colon, S. Das, S. Seal, J. Turkson, J. Zhao and C. H. Baker, *Nanomedicine*, 2013, **9**, 558; E. Sanfins, J. Dairou, F. Rodrigues-Lima and J.-M. Dupret, *Nanoparticle-protein interactions: from crucial plasma proteins to key enzymes, presented at Journal of Physics: Conference Series*, 2011.
- 9 F. Marano, S. Hussain, F. Rodrigues-Lima, A. Baeza-Squiban and S. Boland, *Arch. Toxicol.*, 2011, **85**, 733.
- 10 L. Walter and G. Hajnóczky, J. Bioenerg. Biomembr., 2005, 37, 191.
- 11 L. Dara, C. Ji and N. Kaplowitz, *Hepatology*, 2011, 53, 1752.
- R. J. Kaufman, D. Scheuner, M. Schröder, X. Shen, K. Lee,
 C. Y. Liu and S. M. Arnold, *Nat. Rev. Mol. Cell Biol.*, 2002,
 3, 411; S. Oyadomari and M. Mori, *Cell Death Differ.*, 2004,
 11, 381.
- 13 M. Schröder and R. J. Kaufman, *Mutat. Res., Fundam. Mol. Mech. Mutagen.*, 2005, **569**, 29.
- 14 H. P. Harding, M. Calfon, F. Urano, I. Novoa and D. Ron, *Annu. Rev. Cell Dev. Biol.*, 2002, **18**, 575.
- K. Kasemets, S. Suppi, K. Künnis-Beres and A. Kahru, *Chem. Res. Toxicol.*, 2013, 26, 356; T. Nomura, J. Miyazaki,
 A. Miyamoto, Y. Kuriyama, H. Tokumoto and Y. Konishi,
 Environ. Sci. Technol., 2013, 47, 3417.
- 16 M. Liu, L. Piao, L. Zhao, S. Ju, Z. Yan, T. He, C. Zhou and W. Wang, *Chem. Commun.*, 2010, 46, 1664.
- 17 T. T. Wimalasena, B. Enjalbert, T. Guillemette, A. Plumridge, S. Budge, Z. Yin, A. J. Brown and D. B. Archer, *Fungal Genet. Biol.*, 2008, 45, 1235; Q. Yu, B. Zhang, J. Li, B. Zhang, H. Wang and M. Li, *Free Radicals Biol. Med.*, 2016, 99, 572.

- 18 M. Carlson, B. C. Osmond and D. Botstein, *Genetics*, 1981, 98, 25.
- 19 S. J. Klaine, P. J. Alvarez, G. E. Batley, T. F. Fernandes, R. D. Handy, D. Y. Lyon, S. Mahendra, M. J. McLaughlin and J. R. Lead, *Environ. Toxicol. Chem.*, 2008, 27, 1825.
- 20 M. Pandurangan, G. Enkhtaivan, B. Venkitasamy, B. Mistry, R. Noorzai, B. Y. Jin and D. H. Kim, *Biol. Trace Elem. Res.*, 2016, 170, 309.
- 21 K. J. Travers, C. K. Patil, L. Wodicka, D. J. Lockhart, J. S. Weissman and P. Walter, *Cell*, 2000, **101**, 249.
- 22 R. Chen, L. Huo, X. Shi, R. Bai, Z. Zhang, Y. Zhao, Y. Chang and C. Chen, *ACS Nano*, 2014, **8**, 2562.
- 23 O. V. Tsyusko, J. M. Unrine, D. Spurgeon, E. Blalock, D. Starnes, M. Tseng, G. Joice and P. M. Bertsch, *Environ. Sci. Technol.*, 2012, 46, 4115.
- 24 J. S. Cox and P. Walter, Cell, 1996, 87, 391.
- 25 T. Kawahara, H. Yanagi, T. Yura and K. Mori, *Mol. Biol. Cell*, 1997, **8**, 1845.
- 26 U. Rüegsegger, J. H. Leber and P. Walter, Cell, 2001, 107, 103.
- 27 B. Esmon, P. Novick and R. Schekman, Cell, 1981, 25, 451;
 N. Xu, K. Qian, Y. Dong, Y. Chen, Q. Yu, B. Zhang, L. Xing and M. Li, Res. Microbiol., 2014, 165, 252; E. F. Bailão, P. S. Lima, M. G. Silvabailão, A. M. Bailão, G. R. Fernandes, D. J. Kosman and C. M. Soares, Front Microbiol., 2015, 6, 821; T. Roitsch, M. E. Balibrea, M. Hofmann, R. Proels and A. K. Sinha, J. Exp. Bot., 2003, 54, 513.
- 28 Y. Ma and L. M. Hendershot, *J. Chem. Neuroanat.*, 2004, 28, 51.
- 29 Y. Li, W. J. Fu, N. N. Liu, M. J. Tan, G. L. Liu and Z. M. Chi, *J. Mol. Catal. B: Enzym.*, 2015, **111**, 71.
- 30 G. Applerot, A. Lipovsky, R. Dror, N. Perkas, Y. Nitzan, R. Lubart and A. Gedanken, *Adv. Funct. Mater.*, 2009, 19, 842; J. Meng, Y. Ji, J. Liu, X. Cheng, H. Guo, W. Zhang, X. Wu and H. Xu, *Nanotoxicology*, 2014, 8, 686.
- 31 A. Deniaud, E. Maillier, D. Poncet, G. Kroemer, C. Lemaire and C. Brenner, *Oncogene*, 2008, 27, 285.
- 32 R. J. Kaufman, D. Scheuner, M. Schröder, X. Shen, K. Lee, C. Y. Liu and S. M. Arnold, *Nat. Rev. Mol. Cell Biol.*, 2002, 3, 411; K. Mori, *Cell*, 2000, 101, 451.
- 33 L. F. G. De and A. Sols, Biochim. Biophys. Acta, 1962, 56, 49.
- 34 Q. Yu, Y. Dong, N. Xu, K. Qian, Y. Chen, B. Zhang, L. Xing and M. Li, FEMS Yeast Res., 2014, 14, 1037; A. T. Mckie, G. O. Latunde-Dada, S. Miret, J. A. Mcgregor, G. J. Anderson, C. D. Vulpe, J. M. Wrigglesworth and R. J. Simpson, Biochem. Soc. Trans., 2002, 30, 722.