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Synthesis and functionalization of 3-bromo-2-(2-chlorovinyl)benzothiophenes as molecular tools†

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An efficient bromocyclization process of *ortho*-substituted arylmethyl sulfide promoted by *N*-methyl-pyrrolidin-2-one hydrotribromide led to the synthesis of 3-bromo-2-(2-(di)chlorovinyl) benzothiophene as a polyhalogenated platform. Various arylations on the C3 atom of such di-substituted benzothiophenes and further functionalizations at the chlorine atoms of the benzothiophenes afforded efficient and rapid access to a small library of stereo-defined 2,3-disubstituted benzothiophenes.

Introduction

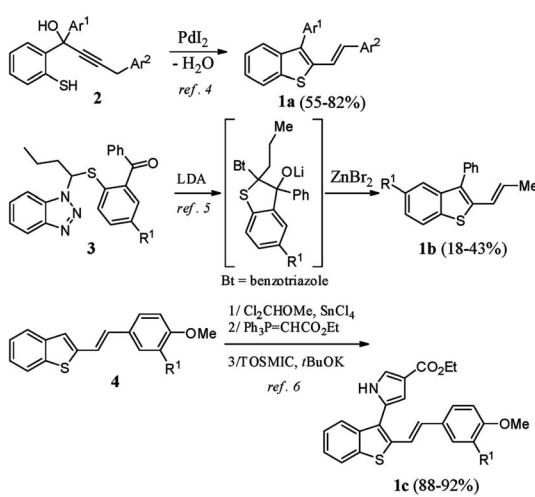
2,3-Disubstituted benzothiophenes have been well studied by the scientific community in past decades, mainly for their numerous biological properties.¹ For example, raloxifene (Evista™), an oral selective estrogen receptor modulator, is prescribed in the prevention and treatment of osteoporosis and is also given for postmenopausal women who are at high risk for invasive breast cancers.² In contrast to the synthesis of 2,3-diarylbenzothiophenes, which is well reported,³ the preparation

of their stereo-defined 2-vinylous analogues **1a–c** is poorly documented (Scheme 1).

To our knowledge, these derivatives having (*E*)-double bonds have been prepared using three different strategies: (i) the Pd-catalyzed heterocyclodehydration of acyclic precursor **2**,⁴ (ii) reaction of **3** with LDA followed by rearrangement in the presence of ZnBr₂,⁵ and (iii) C3-functionalization of 2-substituted benzothiophene **4** through a three-step sequence involving a Rieche formylation, a Wittig reaction, and a pyrrole ring construction under van Leusen reaction conditions.⁶

In a continuation of our work dedicated to the synthesis of functionalized heterocycles,⁷ we described a new method to prepare a variety of stereo-defined polyhalogenated platforms **6** through the *N*-methyl-pyrrolidin-2-one hydrotribromide (MPHT)-promoted bromocyclization of (*Z*)- and (*E*)-chloroenynes **5** and subsequent site-selective Suzuki–Miyaura coupling reactions of **6** to prepare various 2,3-disubstituted benzothiophenes **1** (Scheme 2).

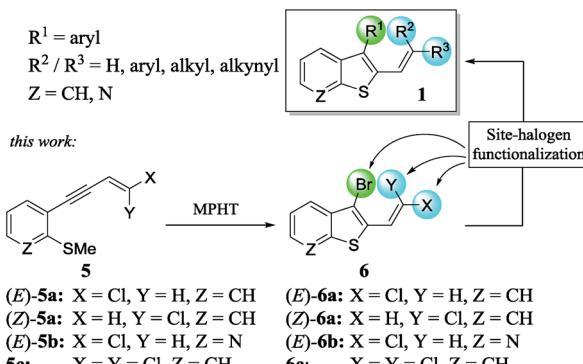
We choose a bromocyclization strategy instead of Larock's iodo heteroannulation⁸ since the site-selective Suzuki–Miyaura coupling of a C–Br vs. C–Cl bond is more challenging from our point of view than the coupling of C–I vs. C–Cl.



Scheme 1 Previous syntheses of 3-aryl-2-alkenyl-benzothiophenes **1a–c**.

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Scheme 2 Synthesis of polyhalogenated platforms **6**.

Results and discussion

The required 1,3-chloroenynes (*E*)-5a with an *ortho* nucleophilic methyl sulfide was prepared *via* Pd-catalyzed Sonogashira–Linstrumelle coupling reaction.⁹ We were pleased to observe that in the presence of MPHT, a mild and easy-to-handle brominating agent discovered in our lab,¹⁰ (*E*)-5a¹¹ underwent bromocyclization¹² at rt in CH₂Cl₂ to provide the desired 2,3-disubstituted benzothiophene platform (*E*)-6a in a good (88%) yield¹³ (Scheme 3).

The scope of this bromocyclization was also demonstrated by the synthesis of (*E*)-6b (Scheme 3), which is suitable for the preparation of 7-aza-benzothiophene-containing scaffolds found in drug discovery.¹⁴

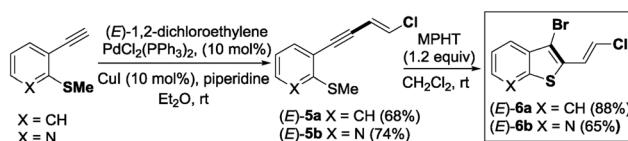
Next, we focus our attention on the identification of efficient experimental conditions for site-selective Suzuki–Miyaura coupling reactions between 3-bromobenzothiophene (*E*)-6a and arylboronic acid 7 (1.3 equiv.) as coupling partners (Table 1). Initially, to compare the reactivity of boronic acids towards a 2-vinylchlorine moiety *vs.* a probably more reactive 3-bromine atom on a benzothiophene scaffold, we tested the conditions used for the coupling of chloroenynes¹⁵ using Pd(PPh₃)₄ (5 mol%) as the catalyst, K₂CO₃ (2 equiv.) as the base, and toluene/MeOH (2 : 1) as the solvent at 90 °C. However, no selectivity was observed, and the expected C-3 monoarylated benzothiophene (*E*)-8a was isolated in 17% yield accompanied with (*E*)-9 (6%) and significant amounts (39%) of the diarylated product (*E*)-1aa (entry 1). This result clearly highlighted that the selective introduction of an aryl substituent at the C-3 position of (*E*)-6a is far from trivial, although the C-Br bond is more reactive than the C-Cl bond. It should be noted that 2,3-disubstituted benzothiophene derivatives (*E*)-8a, (*E*)-9 and (*E*)-1aa can be easily separated by column chromatography on silica gel.

Next, we continued our study by exploring the influence of the palladium- and ligand-controlled site-selective Suzuki–Miyaura cross couplings. No reaction occurred by replacing Pd(PPh₃)₄ with other palladium sources such as PdCl₂(PPh₃)₂, PdCl₂(dppf) and Pd(dba)₂ (entries 2–4), even when increasing the amount of the catalyst from 5 to 20 mol%. By using the N-heterocyclic carbene palladacycle precatalyst [PdCl(dmbo)(IMes)] (Pd–NHC) developed in 2008 by Ying¹⁶ for Heck- and Suzuki-coupling reactions, we were pleased to observe the formation of the desired C-3 monoarylated benzothiophene (*E*)-8a in a moderate but promising yield of 38% after 7 h of reaction (entry 5). In this case, (*E*)-9 was not detected, and a trace amount (3%) of the diarylated product (*E*)-1aa was isolated along with significant amounts (45%) of unreacted (*E*)-6a. Increasing the reaction time from 7 to 24 h (entry 6) improved

the yield of (*E*)-8a from 38 to 85%, but also increased the quantity of the diarylated product (*E*)-1aa from 3% to 10%. Finally, using 10 mol% of this Pd–NHC precatalyst with 2 equiv. of K₂CO₃ in a hot mixture of toluene/MeOH (2 : 1) for 24 h led to the selective C-3 monoarylation of (*E*)-6a, thus providing (*E*)-8a in 90% yield together with 7% of (*E*)-1aa (entry 7). Due to the σ -donation and steric bulk around the metal, this Pd complex with a carbene ligand instead of phosphine ligands facilitates the oxidative addition and the reductive elimination in the palladacycle. Thereby, the selectivity between a C-Br *vs.* a C-Cl bond was increased in the presence of a boronic acid. This result was confirmed by replacing [PdCl(dmbo)(IMes)] by PEPPSITM-IPr precatalyst, and the reaction furnished mainly the mono-coupling product (*E*)-8a (82% yield). The effect of the base was next investigated, and K₃PO₄ gave a similar result to K₂CO₃ (entry 8). All other bases were unsuccessful in achieving efficient coupling reaction, leading to a complex mixture of by-products when using LiOt-Bu (entry 9) or to unchanged starting material (*E*)-6a in the presence of NEt₃ or KOAc (entries 10 and 11). The effect of the solvent was studied, but no improvement was noted when toluene/MeOH was replaced by DMF, THF, MeOH or toluene (entries 12–15). A mixture of toluene associated with MeOH was found to be the best solvent combination, likely for solubility reasons. The conditions used in entry 7 ([PdCl(dmbo)(IMes)] (10 mol%), K₂CO₃ (2 equiv.), and arylboronic acid (1.3 equiv.) in toluene/MeOH (2 : 1) in a sealed tube at 90 °C for 24 h) were then used for other coupling reactions using a variety of boronic acids to demonstrate the versatility and the chemoselectivity of the present protocol (Table 2). As expected, using the experimental conditions depicted in entry 7 of Table 1, a variety of arylboronic acids¹⁷ bearing electron-donating and electron-withdrawing groups were introduced at the C-3 position of (*E*)-6a,b, leading to (*E*)-8a–g in good to excellent yields (59–90%).

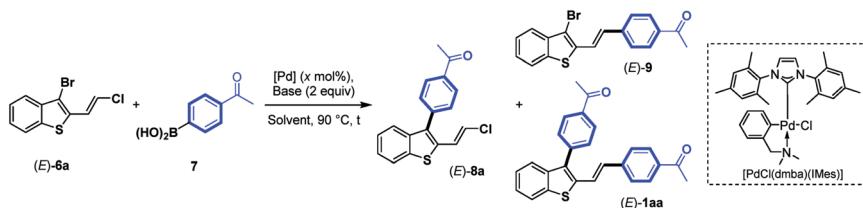
As the next logical extension, Suzuki–Miyaura coupling reactions at the remaining C-Cl bond of benzothiophene compounds (*E*)-8a–d were attempted under the previous conditions reported for the couplings of arylboronic acids with chloroenynes.¹⁵ In the presence of Pd(PPh₃)₄ (5 mol%), K₂CO₃ (2 equiv.), and arylboronic acid (1.2 equiv.) in a hot mixture of toluene/MeOH, we were pleased to observe the successful replacement of the chlorine atom by various aromatic and heteroaromatic rings (Table 3). The reactions proceeded in good yields (75–92%) with electron-poor and electron-rich arylboronic acids used as coupling partners for (*E*)-3-aryl-2-(2-chlorovinyl)benzothiophenes (*E*)-8a–d. It should be noted that using Pd(PPh₃)₄ (10 mol%), K₂CO₃ (2 equiv.) and 4-acetylboronic acid in a slight excess (2.2 equiv.) furnished (*E*)-1aa (see Table 1), in which the boronic acid replaced both the bromine and chlorine atoms of (*E*)-6a (83%).

Because the reaction conditions for the two-step Suzuki–Miyaura couplings are similar (K₂CO₃, toluene/MeOH, 90 °C), we investigated whether the two-step coupling reactions could be carried out in a one-pot fashion directly from (*E*)-6a, avoiding the isolation of the monocoupling products (*E*)-8. Reactions were conducted using K₂CO₃ (2 equiv.) in toluene/MeOH (2 : 1) as the solvent at 90 °C. In the first step, (4-acetylphenyl)boronic



Scheme 3 MPHT promoted the cyclization of 1,3-chloroenyne (*E*)-5a,b.



Table 1 Optimization of the site-selective coupling reaction between (E)-6a and arylboronic acid 7^a

Entry	[Pd]	x	Base	Time (h)	Solvent	Yield ^b of 8a (%)	Yield ^b of 9 (%)	Yield ^b of 1aa (%)
1	Pd(PPh ₃) ₄	5	K ₂ CO ₃	7	Toluene/MeOH (2 : 1)	17	6	39
2	PdCl ₂ (PPh ₃) ₂	5	K ₂ CO ₃	7	Toluene/MeOH (2 : 1)	0	0	0
3	PdCl ₂ (dpdf)	5	K ₂ CO ₃	7	Toluene/MeOH (2 : 1)	0 ^c	0	0
4	Pd(db ₂) ₂	5	K ₂ CO ₃	7	Toluene/MeOH (2 : 1)	0 ^c	0	0
5	[PdCl(dbma)(IMes)]	5	K ₂ CO ₃	7	Toluene/MeOH (2 : 1)	38 ^d	0	3
6	[PdCl(dbma)(IMes)]	5	K ₂ CO ₃	24	Toluene/MeOH (2 : 1)	85	0	10
7	[PdCl(dbma)(IMes)] ^f	10	K ₂ CO ₃	24	Toluene/MeOH (2 : 1)	90	0	7
8	[PdCl(dbma)(IMes)]	10	K ₃ PO ₄	24	Toluene/MeOH (2 : 1)	85	0	8
9	[PdCl(dbma)(IMes)]	10	LiOt-Bu	24	Toluene/MeOH (2 : 1)	— ^e	—	—
10	[PdCl(dbma)(IMes)]	10	NEt ₃	24	Toluene/MeOH (2 : 1)	0 ^c	0	0
11	[PdCl(dbma)(IMes)]	10	KOAc	24	Toluene/MeOH (2 : 1)	0 ^c	0	0
12	[PdCl(dbma)(IMes)]	10	K ₂ CO ₃	24	DMF	0 ^c	0	0
13	[PdCl(dbma)(IMes)]	10	K ₂ CO ₃	24	THF	0 ^c	0	0
14	[PdCl(dbma)(IMes)]	10	K ₂ CO ₃	24	MeOH	72	0	17
15	[PdCl(dbma)(IMes)]	10	K ₂ CO ₃	24	Toluene	21	0	4

^a Conditions: (E)-6a (1 mmol), 7 (1.3 mmol), [Pd] (0.05 mmol or 0.1 mmol), base (2 equiv.) and solvent (18 mL) were heated in a sealed tube at 90 °C for time indicated in the table under argon atmosphere. ^b Yield of isolated product. ^c (E)-6a was recovered unchanged. ^d 45% of (E)-6a were recovered. ^e A complex mixture of unidentified products was obtained. ^f Replacing [PdCl(dbma)(IMes)] by Pd-PEPPSI™-IPr furnished 8a in a slightly lower yield of 82%.

acid (1.3 equiv.) reacted with (E)-6a. When consumption of the substrate was complete according to TLC, Pd(PPh₃)₄ (5 mol%) and 4-fluoroboronic acid were added to the reaction mixture. Accordingly, we were pleased to isolate (E)-1ab containing two different aryl groups in a good overall yield of 68% (Scheme 4). One can also note that the one-pot synthesis of (E)-1ak (65%) was successfully achieved from (E)-6a by inverting the addition

order of boronic acids (first 4-fluoroboronic acid followed by 4-acetylboronic acid).

However, it should be noted that attempts to achieve one-pot coupling in the presence of only one catalyst [PdCl(dmba)(IMes)] (10–20 mol%) furnished (E)-1ab in low yield

Table 2 Suzuki coupling reactions of (E)-6a,b with a variety of arylboronic acids

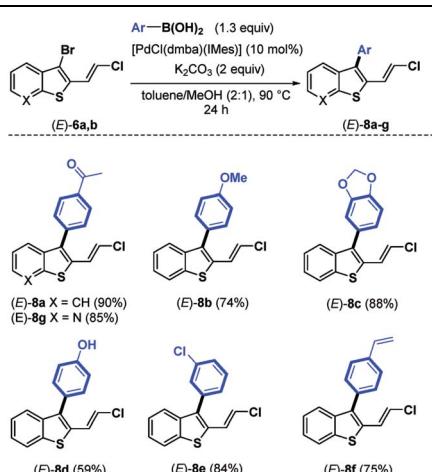
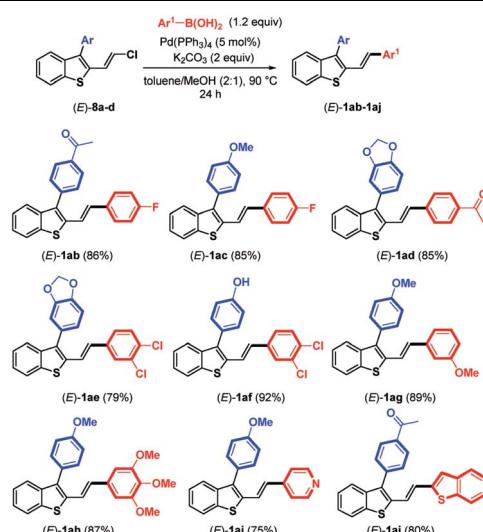
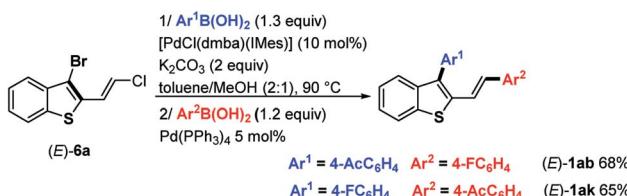
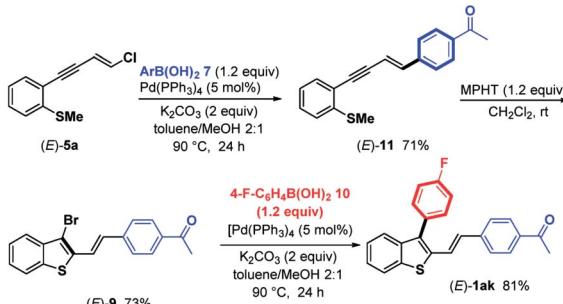


Table 3 Suzuki coupling reactions of (E)-8a-d with a variety of arylboronic acids

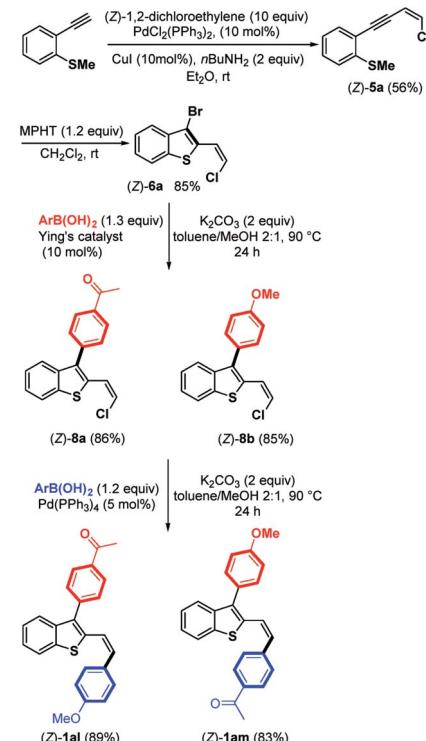


Scheme 4 One-pot synthesis of $(E\text{-}1ab)$ and $(E\text{-}1ak)$ from $(E\text{-}6a)$.Scheme 5 Synthesis of $(E\text{-}1ak)$.

(28%) along with $(E\text{-}8a)$ (42%) after 48 h of reaction, clearly indicating that the Pd–NHC precatalyst was not efficient to introduce an aryl substituent on the C–Cl bond of $(E\text{-}8a)$.

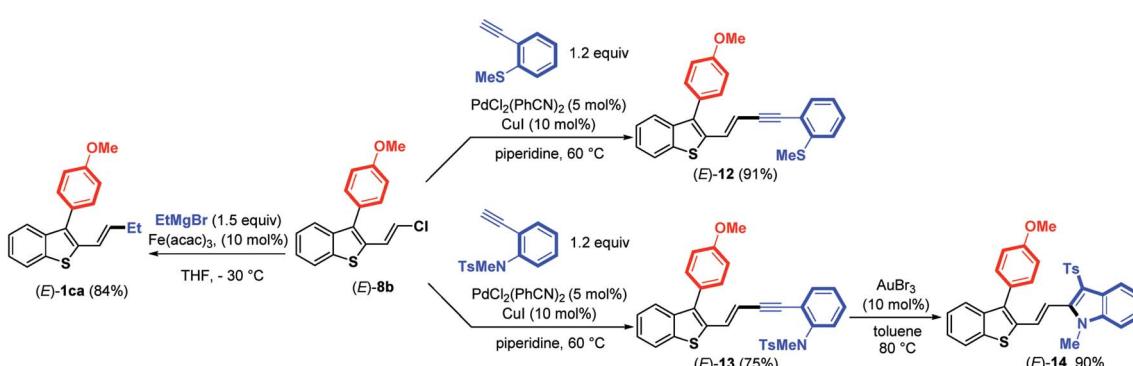
2,3-Disubstituted benzothiophenes $(E\text{-}1ak)$ were also accessed by inverting the MPHT-bromo-cyclization process followed by the arylation of the resulting 3-bromoposition of the thiophene backbone (Scheme 5).

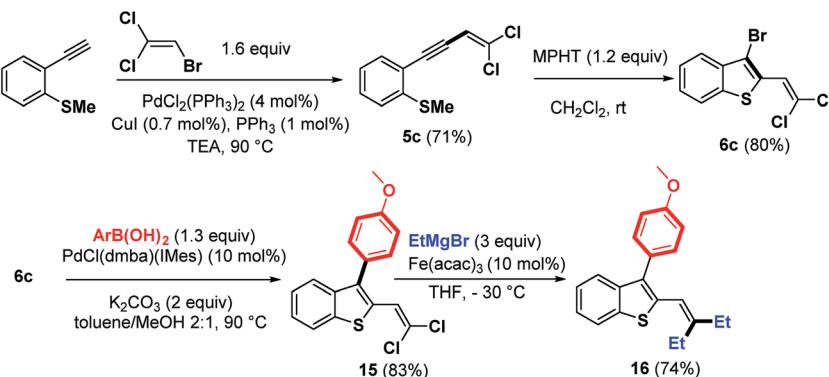
In detail, chloroynne $(E\text{-}5a)$ was initially coupled with (4-acetylphenyl)boronic acid **7** in the presence of $\text{Pd}(\text{PPh}_3)_4$ (5 mol%), K_2CO_3 (2 equiv.) in toluene/MeOH (2 : 1) at 90 °C to give the expected 1,4-diarylenyne $(E\text{-}11)$ in 71% yield.¹⁵ $(E\text{-}11)$ then undergoes MPHT-promoted bromo-cyclization to furnish the 3-bromo-benzothiophene $(E\text{-}9)$ (73%). Further Suzuki–Miyaura coupling between $(E\text{-}9)$ and (4-fluoro-phenyl)boronic acid **10** in the presence of $\text{Pd}(\text{PPh}_3)_4$ (5 mol%) led to $(E\text{-}1ak}$ in a good (81%) yield. The comparison of the two strategies, bromocyclization/Suzuki/Suzuki to prepare $(E\text{-}1ab\text{--aj})$ and Suzuki/bromocyclization/Suzuki to obtain $(E\text{-}1ak}$, shows that they are equivalent in terms of overall yield and ease of implementation.

Scheme 7 Bromocyclization of $(Z\text{-}5a)$ into $(Z\text{-}6a)$ and subsequent Suzuki–Miyaura coupling reactions.

Next, the reactivity of 3-aryl-2-(2-chlorovinyl)benzothiophene $(E\text{-}8b)$ in Pd-catalyzed couplings was examined to demonstrate the usefulness of such molecular tools. Gratifyingly, the use of $\text{PdCl}_2(\text{PhCN})_2$ (5 mol%) and CuI (10 mol%) as the catalysts in piperidine at 60 °C (ref. 18) allowed the coupling to proceed efficiently between $(E\text{-}8b)$ and *ortho*-substituted arylalkynes to provide (E) -enynes **12** and **13** in good yields (Scheme 6).

Taking advantage of the structure of enyne $(E\text{-}13)$ having an *ortho*-*N*-tosyl-*N*-methylaniline function, we achieved a gold-catalyzed cyclization¹⁹ to give, after tosyl migration, 2-alkenyl-3-sulfonylindole $(E\text{-}14)$ in 90% yield. We further demonstrated that the C–Cl bond in $(E\text{-}8b)$ can react with a Grignard reagent (EtMgBr) in the presence of a catalytic amount of $\text{Fe}(\text{acac})_3$ (ref. 20) to introduce an alkyl substituent, thus providing 2-

Scheme 6 Coupling reactions of vinylchloride $(E\text{-}8b)$ with terminal *ortho*-substituted arylalkynes and EtMgBr .



Scheme 8 Synthesis of substituted-3-bromobenzothiophenes 6c and its functionalization.

benzothiophene (*E*)-1ca in a good (84%) yield with no trace of (*E*)-double bond isomerization. Altogether, these results clearly highlight that it is possible to create various C–C bonds through the coupling of alkyl Grignard reagents (Csp^3), arylboronic acids (Csp^2) or terminal alkynes (Csp) with the C–Cl bond of benzothiophenes (*E*)-8, demonstrating the synthetic potential of these molecular tools.

Due to the biological interest in benzothiophenes containing a (*Z*)-double bond²¹ at the C-2 position as tubulin polymerization inhibitors, we investigated the preparation of (*Z*)-6a and explored its coupling to provide stereoselectively (*Z*)-2,3-disubstituted benzothiophene derivatives (Scheme 7). Thus, by replacing piperidine with *n*BuNH₂ as the base,⁹ we were able to synthesize the expected chloroenyne (*Z*)-5a (56%) through the Sonogashira–Linstrumelle coupling of (2-ethynylphenyl)(methyl)sulfane with (*Z*)-1,2-dichloroethylene in the presence of catalytic amounts of $PdCl_2(PPh_3)_2$ and CuI.

Interestingly, the reaction conditions used for the synthesis of (*E*)-6a (MPHT, CH_2Cl_2) also allowed us to selectively prepare (*Z*)-6a (85%) through the bromo-cyclization of (*Z*)-5a (Scheme 7).

Further Suzuki–Miyaura monoarylation reactions of (*Z*)-6a with electron-poor phenylboronic acid 7 or the electron-rich (4-methoxyphenyl)boronic acid in the presence of Pd–NHC¹⁶ led to the expected coupling products (*Z*)-8a and (*Z*)-8b in good yields of 86% and 85%, respectively. Further arylation of the (*Z*)- Csp^2 –Cl bond can be accomplished by arylboronic acids in the presence of $Pd(PPh_3)_4$ as a palladium source. Push-pull

benzothiophene derivatives (*Z*)-1al and (*Z*)-1am were synthesized in excellent yields from (*Z*)-8a and (*Z*)-8b, respectively, with no trace of (*Z*)-double bond isomerization (Scheme 7).

The scope of this strategy involving bromocyclization followed by Pd-catalyzed coupling reactions was also extended to the synthesis of polyhalogenated platform 6c, suitable for the elaboration of benzothiophenes containing a trisubstituted double bond at the C-2 position (Scheme 8).

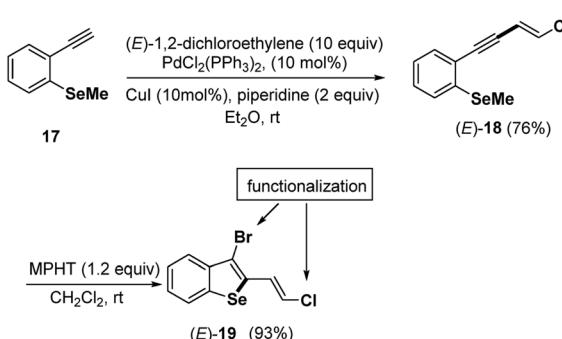
As expected, the bromocyclization process of 5c promoted by MPHT furnished 6c in 80% yield. An illustration of its synthetic potential is shown in Scheme 8. Trihalogenated benzothiophene 6c was first monoarylated at the C-3 position using Pd–NHC precatalyst as a palladium source to give 15 (83%). This product was then coupled with EtMgBr (3 equiv.) in the presence of a catalytic amount of Fe(acac)₃ (ref. 20) at $-30\text{ }^\circ\text{C}$ to give benzothiophene 16 in 74% yield.

Finally, under the mild conditions (MPHT 1.2 equiv., CH_2Cl_2 , rt) described above for the bromocyclization of chloroenynes 5 having an *ortho*-methyl sulfide group on the aromatic moiety, we successfully transformed the *ortho*-SeMe-aryl chloroenyne (*E*)-18 into a 2,3-disubstituted benzoselenophene 19 (93%), which is possibly useful for further chemical transformations (Scheme 9).

Efforts are in progress in our lab to develop satisfactory coupling conditions at both the bromine and chlorine atoms of (*E*)-19 and (*Z*)-19 benzoselenophenes.

Conclusion

In summary, we have demonstrated that MPHT is a mild and efficient brominating agent useful for the room-temperature bromocyclization of *ortho*-alkynylaryl methyl sulfides 5. The resulting 3-bromo-2-(2-(di)chlorovinyl)benzothiophenes 6 may serve as di- or tri-halogenated benzothiophene platforms useful for chemoselective and successive coupling reactions (Suzuki, Sonogashira, *etc.*) leading to rapid and convergent access to a series of 2,3-disubstituted benzothiophenes. We have also demonstrated that it is possible to access to these benzothiophene targets using a complementary strategy involving the arylation of stereo-defined chloroenynes 5a followed by bromocyclization and a second C-3 functionalization on the resulting 3-

Scheme 9 Synthesis of (*E*)-19.

bromo-benzothiophene (arylation, alkynylation, or alkylation). Finally, we have shown that this bromocyclization process is also efficient for arylselenomethyl ether **18**, which was transformed into the novel and potentially functionalizable 2-chlorovinyl-3-bromobenzo selenophene platform **19**. We believe that these novel methodologies will find broad applications in synthetic organic chemistry and in pharmaceutical sciences.

Conflicts of interest

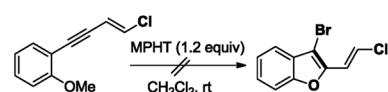
There are no conflicts to declare.

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12 It is also possible to prepare a 3-iodo-2-(chlorovinyl) benzothiophene using I_2 and chloroynene (*E*)-**5a** (82%).

13 When NBS (1.2 equiv.) was used in place of MPHT, (*E*)-**6a** was obtained in a poor yield of 20%. Using Br_2 , the reaction furnished (*E*)-**6a** in 80%. Solid MPHT is more convenient to handle than Br_2 particularly when small quantities were used (1.2 equiv., 61 μ L).

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