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Palladium-catalyzed synthesis of quinolines from allyl alcohols and anilines†

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A process for quinoline synthesis through palladium-catalyzed oxidative cyclization of aryl allyl alcohols and anilines is described. This process works in the absence of acid, base and any other additive and has a broad substrate scope, tolerating electron-withdrawing groups such as nitril, trifluoromethyl and so on. A series of quinolines are prepared in satisfactory yields.

Introduction

The quinoline skeleton occurs extensively as an integral part of many natural products, especially in alkaloids, and synthetic pharmacologically active substances due to their biological activities, such as anticancer, antiviral, antibacterial, anti-fungal, anti-inflammatory and antiplatelet aggregation.¹ For example, Camptothecin (**A**) is a cytotoxic quinoline alkaloid clinically used to treat cancer by inhibiting DNA enzyme topoisomerase I; Quinoline compound (**B**) has been developed as a corticotropin releasing factor-1 receptor antagonists for the treatment of depression or other stress-related disorders;² 2-phenylquinolinyl moiety is deemed as the critical pharmacophore of Linsitinib (**C**), a clinical dual inhibitor of insulin-like growth factor-1 receptor (IGF-1R) and insulin receptor (IR) for various cancers treatment (Fig. 1).³

In view of the importance of quinolines, great endeavors have been made to develop the procedures for their synthesis over the past few decades.⁴ Traditionally, the classical strategies, such as Doebner–von Miller reaction, Skraup, reaction, Gould–Jacobs reaction, Combes synthesis, and Friedländer synthesis, employ the condensation reaction of carbonyl

substrates and aniline derivatives under harsh conditions, namely strong acidic or strong basic conditions.⁵ Recently, significant attention has been paid to the development of transition metal-catalyzed protocols for the synthesis of quinoline molecules owing to their advantage of voiding the use of strong acids or strong bases in contrast with traditional methods.⁶ Among them, the condensation of anilines with alcohols has also been developed.⁷ For example, Trillo and Pastor reported that quinolines could be prepared from allyl alcohols and anilines under the catalysis of iron(III)-based Lewis acidic ionic liquid,⁸ and Jiang group disclosed that quinoline derivatives could be obtained from simple aliphatic alcohols and anilines under the Pd(OAc)₂/2,4,6-Collidine/Brønsted acid catalytic system.⁹ More recently, Kapur and his coworkers demonstrated a ruthenium catalyzed [3 + 3] annulation of anilines with allyl alcohols to provide substituted quinolines based on a traceless directing-group strategy.¹⁰ Although these elegant developments could provide a range of alternatives, the development of novel and expeditious methods for the synthesis of quinoline is still in demand. Herein, we report a process for the synthesis of quinoline from allyl alcohols and anilines under the catalysis of Pd(OAc)₂ without any ligand or additive (Scheme 1).

Results and discussion

Our studies began by optimizing the reaction conditions for the synthesis of quinolines, choosing aniline (**1a**) and cinnamic alcohol (**2a**) as model substrates. The results are collected in Table 1. When this reaction was carried out catalyzed by Pd(OAc)₂ in the presence of 20 mol% TsOH in DMSO at 100 °C

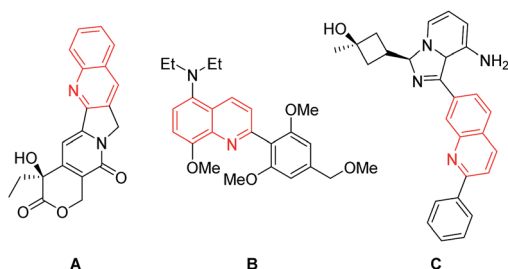
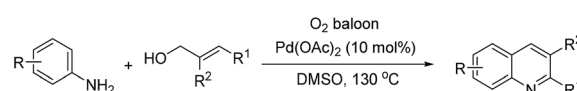


Fig. 1 Examples of bioactive quinolines.

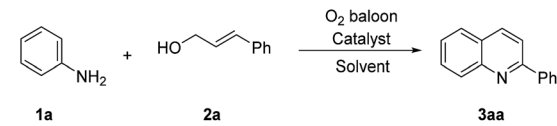
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Scheme 1 Palladium-catalyzed synthesis of quinolines from allyl alcohols and anilines.




Table 1 Optimizing the reaction conditions for palladium-catalyzed synthesis of quinolines^a


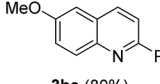
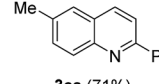
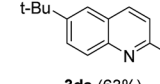
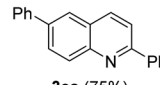
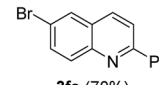
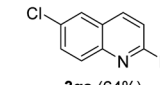
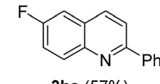
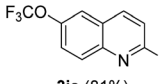
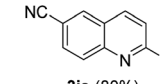
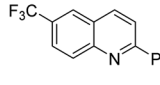
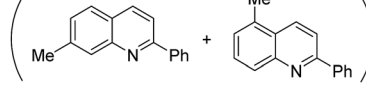
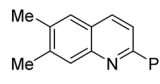
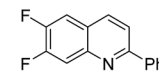
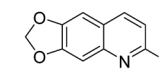
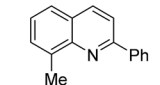
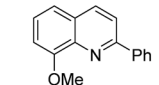
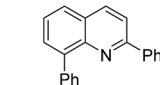
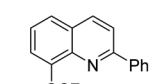
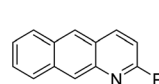
Entry	Catalyst	Solvent	Temperature	Yield ^b (%)
1 ^c	Pd(OAc) ₂	DMSO	100	79
2	Pd(OAc) ₂	DMSO	100	75
3	Pd(TFA) ₂	DMSO	100	16
4	PdCl ₂	DMSO	100	8
5	Pd ₂ (dba) ₃	DMSO	100	Trace
6	Pd(CH ₃ CN) ₂ Cl ₂	DMSO	100	5
7	Pd(PPh ₃) ₄ Cl ₂	DMSO	100	Trace
8	Pd(PPh ₃) ₄	DMSO	100	Trace
9	Pd(OAc) ₂	Dioxane	100	26
10	Pd(OAc) ₂	Toluene	100	Trace
11	Pd(OAc) ₂	DMF	100	Trace
12	Pd(OAc) ₂	NMP	100	Trace
13	Pd(OAc) ₂	DMAC	100	Trace
14	Pd(OAc) ₂	DMSO	130	90 (85) ^d
15	Pd(OAc) ₂	DMSO	150	82

^a All reactions were performed with **1a** (0.5 mmol), **2a** (0.5 mmol), catalyst (10 mol%), in the indicated solvent (2.0 mL) under 1 atm oxygen at selected temperature for 12 h. ^b GC yield based on **2a**. ^c 20 mol% TsOH was used. ^d Isolated yield.

for 12 h, the expected quinoline **3aa** was obtained in 79% GC yield (entry 1). Unexpectedly, the reaction worked well in absence of acid (entry 2). Our experimental results showed that Pd(TFA)₂ and PdCl₂ could not drive this reaction efficiently (entries 3–4). We then examined other palladium catalysts containing special ligands, such as Pd₂(dba)₃, Pd(CH₃CN)₂Cl₂, Pd(PPh₃)₄Cl₂, Pd(PPh₃)₄, and found that they could not promote the transformation preferably either (entries 5–8). The effect of solvents was next investigated. Using dioxane as the reaction medium provided the target product **3aa** in 26% yield (entry 9). However, trace amount of product **3aa** was detected when the reaction was performed in toluene or amide solvents such as DMF, NMP, and DMAC (entries 10–13). The results of testing reaction temperature revealed that increasing the temperature to 130 °C could improve the yield of quinoline product to 90% while higher temperature gave rise to a lower yield (entries 14–15). On the basis of the experimental results mentioned above, the optimal reaction conditions are listed in entry 14.

The scope of the protocol was further evaluated after the optimal reaction conditions were established. Cinnamic alcohol **2a** was reacted with a series of substituted anilines under the best reaction conditions, and the results are listed in Table 2. *para*-substituted anilines were firstly probed. Generally, moderate to excellent yields could be achieved from both electron-donating groups and electron-withdrawing groups substituted anilines (**3ba–3ka**). For example, 4-methoxyaniline and 4-(trifluoromethyl)aniline afforded the corresponding quinolines in 80% and 84% yield, respectively (**3ba**, **3ka**). *meta*-substituted anilines were then explored. When 3-methylaniline

Table 2 Scope of anilines for Pd(OAc)₂-catalyzed synthesis of quinolines^{a,b}


		
3ba (80%)	3ca (71%)	3da (63%)
		
3ea (75%)	3fa (70%)	3ga (64%)
		
3ha (57%)	3ia (81%)	3ja (80%)
		
3ka (84%)	3la : 3la' = 6:1 (68%)	
		
3ma (75%)	3na (82%)	3oa (62%)
		
3pa (69%)	3qa (64%)	3ra (74%)
		
3sa (78%)	3ta (76%)	

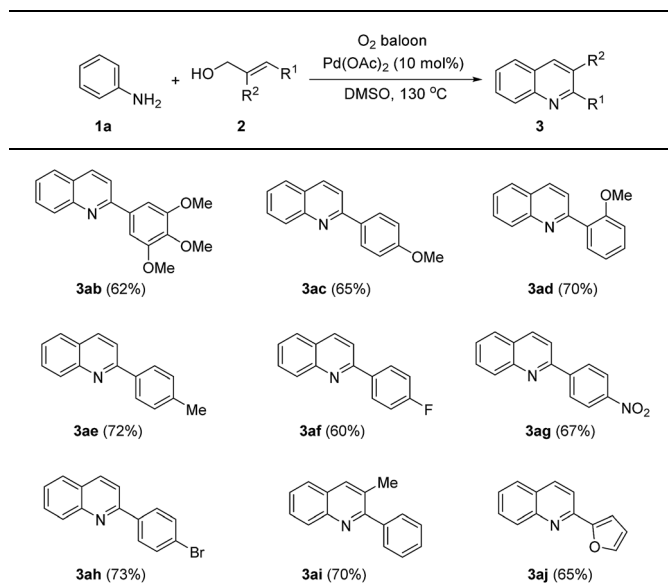
^a Reactions were performed with **1a** (0.5 mmol), **2** (0.5 mmol), Pd(OAc)₂ (10 mol%), in DMSO (2 mL) at 130 °C for 12 h. ^b Isolated yield.

was subjected to the optimized conditions, 7-methyl substituted product **3la** and 5-methyl substituted product **3la'** were obtained in the ratio of 5 : 1, which suggested that steric hindrance of the substituents had significant influence on the regioselectivity of the oxidative cyclization reaction, and 3,4-disubstituted anilines furnished 6,7-disubstituted quinoline products selectively (**3ma–3oa**). *ortho*-substituted anilines were finally tested. As shown in Table 2, all *ortho*-substituted anilines went through the transformation smoothly, producing the desired quinoline compounds in good yields (**3pa–3sa**). Moreover, beta-naphthylamine was also found to be a good partner of cinnamic alcohol under the standard conditions and gave 2-phenylbenzo[*g*]quinoline in 76% yield (**3ta**).

A variety of allyl alcohols were treated with aniline **1a**. And the results are summarized in Table 3. We found that cinnamic alcohols, whose phenyl rings were substituted by electron-donating groups, reacted smoothly with aniline to provide



Table 3 Scope of allyl alcohols for Pd(OAc)₂-catalyzed synthesis of quinolines^{a,b}

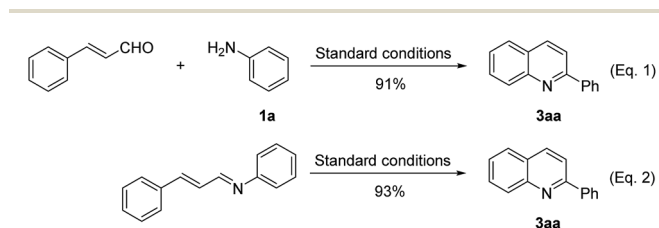


^a Reactions were performed with **1** (0.5 mmol), **2a** (0.5 mmol), Pd(OAc)₂ (10 mol%), in DMSO (2 mL) at 130 °C for 12 h. ^b Isolated yield.

target quinolines in moderate to good yields (**3ab–3ae**). Likewise, moderate to good yields could be achieved from cinnamic alcohols whose phenyl rings were substituted by F and NO₂ (**3ag–3ah**). It is noteworthy that this procedure for the preparation of quinoline derivatives tolerates aryl bromide which allows further functionalization (**3ah**). To our delight, 2,3-disubstituted quinoline could be obtained from α -methyl cinnamic alcohol (**3ai**). Heteroaryl allyl alcohol was finally assessed and 2-furyl quinoline was successfully isolated in a yield of 65% (**3aj**).

In order to gain some insight into the mechanism of the oxidative annulation reaction, some control experiments were carried out. When cinnamaldehyde reacted with aniline **1a** under the optimal reaction conditions, quinoline **3aa** was detected by in 91% yield (eqn (1)). When *N*-cinnamylideneaniline was subjected to the standard reaction conditions, product **3aa** was given in 93% GC yield (eqn (2)), which revealed that this transformation occurred *via* aldimine as the key intermediate (Scheme 2).

On the basis of the above mentioned results and relevant reports in the literature,^{11,6a} a reasonable reaction pathway for palladium-catalyzed quinoline synthesis from allyl alcohols



Scheme 2 Control experiments.

with anilines is illustrated in Scheme 3. Firstly, cinnamic alcohol **2a** is oxidized by dioxygen under the catalysis of palladium to produce α,β -unsaturated aldehyde **A**, which goes through a condensation with aniline to give imine **B**. Dimerization then takes place to generate diazetidine intermediates **C** with the help of palladium. Its C–N bond cleaves and the following irreversible cyclization forms intermediates **E**, which isomerizes readily into **F** driven by the charge, followed by an intramolecular nucleophilic attack along with the elimination of 1 mol of imine, to provide 2-phenyl-1,2-dihydroquinolin **G**. Palladium-catalyzed dehydro-aromatization subsequently occurs to yield the final product.

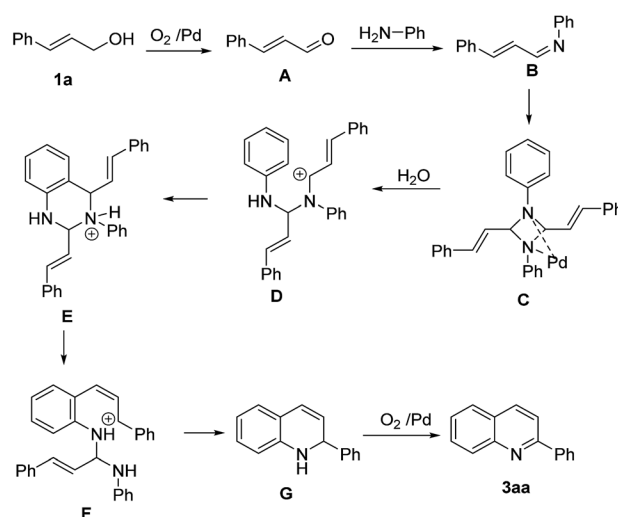
Experimental

General information

All reagents were of analytical grade and obtained from commercial suppliers and used without further purification. ¹H and ¹³C NMR spectra were recorded on a Bruker DRX-400 spectrometer; CDCl₃ was used as solvent and TMS as an internal standard.

Typical procedure for palladium-catalyzed synthesis of quinolines from allyl alcohols and anilines

Allyl alcohol (0.5 mmol), aniline (0.5 mmol), Pd(OAc)₂ (11.2 mg, 10 mol%) and DMSO (2 mL) were added into a test tube attached to an oxygen balloon (1 atm). The system was stirred magnetically and heated at 130 °C with an oil bath for 12 h. And then the reaction was quenched by the addition of 10 mL water. The aqueous solution was extracted with ethyl acetate (3 × 10 mL) and the combined extract was dried with anhydrous MgSO₄. The solvent was vacuumed and the crude product was isolated by TLC (solvent: hexane/ethyl acetate) to give the pure quinoline product.



Scheme 3 Possible reaction pathway for palladium-catalyzed synthesis of quinolines from allyl alcohols and anilines.



Conclusions

In conclusion, we have developed a process for quinine synthesis through palladium-catalyzed oxidative cyclization of aryl allyl alcohols and anilines. This methodology would offer an alternative approach to biologically important quinine derivatives from commercially available starting materials. More importantly, this process works without acids, bases or any other additives.

Acknowledgements

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