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Root-like natural fibers in polypropylene prepared *via* directed diffusion and self-assembly driven by hydrogen bonding

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Interfacial properties govern effective transfer of load from a polymer matrix to a reinforced fiber, dictating the mechanical performance of the composite. This paper reports a facile and controlled strategy of preparing root-like natural fibers (NFs) in polypropylene (PP), where the two components are integrated *via* mechanical interlocking provided by the interfacial grown fiber. Specifically, driven by hydrogen bonding between amide groups of the self-assembling molecules (NAs) and abundant hydroxyl groups existing in the NF, the former selectively diffuses and aggregates on the latter and then self-assembles into branched fibers, which can be tailored finely by the annealing temperature and time as well as cooling rate. This study opens up an interfacial manipulation without preliminary surface modification and holds great potential in preparing NF-reinforced polymer composites.

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Introduction

Polymer composites have comprehensive properties superior to the parent polymer and thus attract significant attention in the field of materials science and industrial applications.^{1,2} The inherent incompatibility between most polymers and introduced fillers results in poor interfacial interaction and finally inferior mechanical performances.³ Currently, common ways to improve the interfacial interaction include modifying the filler surface and adding some compatibilizers to minimize the interfacial tension. However, there are still some issues that need to be addressed for polymer composites, such as high consumption of toxic reagents, the soft nature of the low-molecular weight compatibilizer, and so on.^{4,5}

It's evident that interfacial geometry of inorganic filler can influence the interfacial interaction of polymer composites.⁶ When the fillers are mechanically interlocked with polymer matrix, large interfacial friction is generated to promote interfacial load-transferring efficiency and to perfect the resulting composite.^{7–10} Tamrakar achieved the controlled growth of carbon nanotubes at the surface of glass fibers (GF) by means of electrophoretic deposition to significantly enhance the interfacial strength between the fiber and epoxy resin.⁸ Our group also utilized the polarity gaps of the different components to direct interfacial diffusion and recrystallization of some amide-based

self-assembling compounds and prepared pompon-like hydroxyapatite and root-like glass fiber, where the grown fibers at the interface served as interlocks to tightly join the two components.^{11,12} However, the self-assembling molecules generally present high polarity so pristine fillers can't effectively anchor and immobilize the molecules. Accordingly, a polydopamine layer with hydroxyl groups needs to be pre-coated onto the surface for the directed interfacial diffusion.

Benefited from the renewability, abundant availability, lightweight and excellent mechanical properties, natural fibers (NFs) can replace traditional fillers as a green and high-efficient reinforcement for polymer and is becoming a popular choice for preparation of polymer composites.^{13,14} A prominent feature of NFs is rich hydroxyl groups at the surface so they have the potential to function as a preferred acceptor to assemble the amide-based molecules *via* hydrogen bonding.¹⁵ Accordingly, in this paper, an amide-based compound (WBG) which dissolves into polypropylene (PP) melts at high temperature and self-assembles upon cooling,^{16–18} was chosen and the potential interaction with NFs was explored. Then, the interfacial nucleation and growth were investigated and the corresponding regulation mechanism was proposed. Finally, root-like NFs with controlled branched density and dimension were *in situ* built in PP melts.

Experiment

Materials

Isotactic polypropylene (trade name: T30S) with weight average molecular weight of 400 kg mol^{−1} and the molecular weight distribution of 4.6, was provided by Dushanzi Petroleum Chemical Incorporation (Xinjiang, China). Rare earth

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compound (trade name: WBG-II) was supplied by Guangdong Weilinna Incorporation (Guangdong, China). It's a kind of dimetal complex of lanthanum and calcium with dicarboxylic acid and amide-type ligands and the general formula is $\text{Ca}_x\text{La}_{1-x}(\text{LIG1})_m(\text{LIG2})_n$, where x and $1 - x$ are corresponding to the proportion of Ca^{2+} and La^{3+} ion in the complex, and LIG1 and LIG2 represent dicarboxylic acid and amide-type ligand with coordination numbers of m and n .¹⁹ The average size is ~ 20 nm. Cotton fiber was offered by Shanghai Old Cotton Yarn Company (Shanghai, China) and was extracted by dimethylsulfoxide to remove the impurities. The diameter is ~ 15 μm .

Sample preparation

PP samples containing 0.3 wt% of rare earth compounds (NAs) were prepared in a co-rotating twin screw extruder at the temperature of 180 $^\circ\text{C}$ and hot-pressed into ~ 10 μm thin films. Then, a single cotton fiber was embedded between the two PP with NAs films to prepare the composites.

Characterization

Fourier transform infrared spectroscopy (FTIR). NF and NA powders were mixed at the weight ratio of 1 : 1 and the FTIR spectra were recorded from 4000 to 600 cm^{-1} on a Nicolet 20SXB spectrometer (Thermo Fisher Scientific Inc., USA). As a control, pristine NF and NA powders were also measured.

Polarized light microscope (PLM) observation. The assembly that single fiber was sandwiched between the two NAs-containing PP film, was placed under PLM (Leica DM2500P) connected with hot stage (Linkam THMS600, Linkam Scientific Instruments Ltd., UK), heated to a given temperature and kept for a certain time, followed by cooling down at a constant rate to the preset temperatures. The PP/NF interfacial structures were recorded.

Scanning electron microscope (SEM) investigation. ~ 5 g NAs were dissolved into 100 ml 1,2-dichlorobenzene. Then, NFs were put into the NA solution for 0.5 h at 80 $^\circ\text{C}$. The surface morphology of the treated NFs was investigated by a FEI Inspect F-SEM instrument with an acceleration voltage of 20 kV.

Results and discussion

Hydroxyl groups enriched at the surface of NF can provide the possibility for the formation of hydrogen bonds with amide group carried by NAs. The underlying interaction between NA and NF was evaluated by FTIR spectra (Fig. 1a). NF exhibited the broad peaks at 3432 cm^{-1} due to the presence of $-\text{OH}$ groups; the sharp absorption at 3317 cm^{-1} with a shoulder peak appeared in the NAs, where the former was corresponding to self-association $-\text{NH}$ groups and the latter came from free ones.²⁰ One can observe that in the case of NF/NA mixture, the shoulder representing free $-\text{NH}$ groups took the distinct shift from 3458 cm^{-1} to 3421 cm^{-1} , accompanied by a blue shift of $-\text{C}=\text{O}$ group situated at 1718 cm^{-1} . These features confirmed that there was hydrogen bonding between amide groups of the NAs and abundant hydroxyl groups existing in the NF. This was also verified by the coating of NF by some NAs *via* dip-coating experiment, as shown in Fig. 1b. Similar phenomena were also investigated by previous researchers.²¹ As melting PP is nonpolar, the existence of NF-NA hydrogen bonding guided the directed diffusion of NA onto the surface of NF and triggered the subsequent epitaxial crystallization upon cooling, into root-like fiber with laterally grown NA fiber. The corresponding mechanism on the formation of root-like NF in PP matrix is proposed, as illustrated in Fig. 1c.

For convenient observation, a single NF was embedded in two PP films with 0.3 wt% NAs and heated to 250 $^\circ\text{C}$, where NAs completely dissolved into PP melt.¹² One can observe that upon cooling, the dissolved NAs gradually separated out from PP melts to crystallize epitaxially on the NF surface. As shown in Fig. 2b–e, in the initial stage, only few short NAs fibers of 13 μm emerged on the surface of NF; with the decreasing temperature, the density and dimension of the interracially-grown NAs fibers increased. When the temperature reached 156 $^\circ\text{C}$, the NAs fibers grew to 90 μm and completely covered the NF. Moreover, ascribing to the highly nucleating efficiency of NA on PP,²² the branched NA fibers held rich heterogeneous nuclei for PP and induced the lateral growth of PP crystals, into transcrystalline layers featuring the excellent interfacial interaction (Fig. 2f). Analogous to tree root, the branched NA fiber can interlock NF

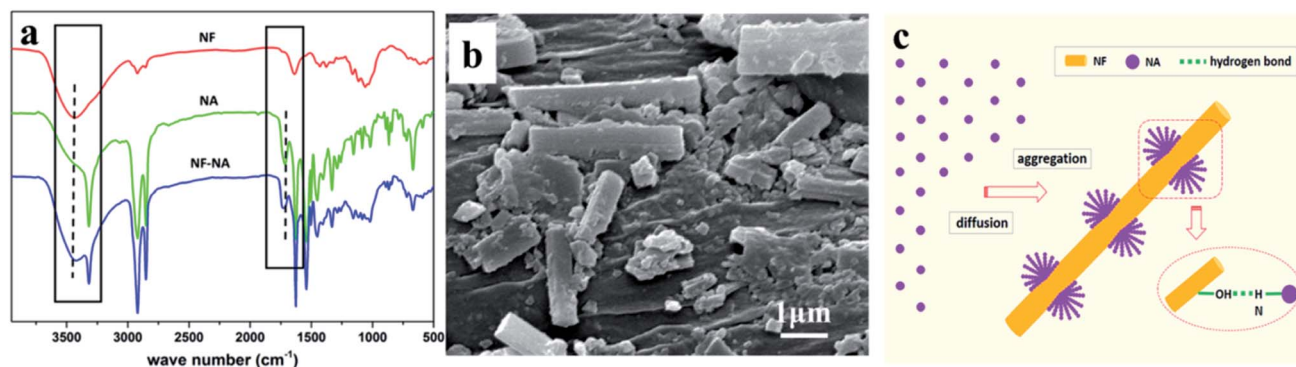


Fig. 1 FTIR spectra of NF, NA and NF-NA powders (a), SEM photo of NF dip-coated by NAs (b) and schematic illustration on the formation of root-like NF in PP matrix (c).



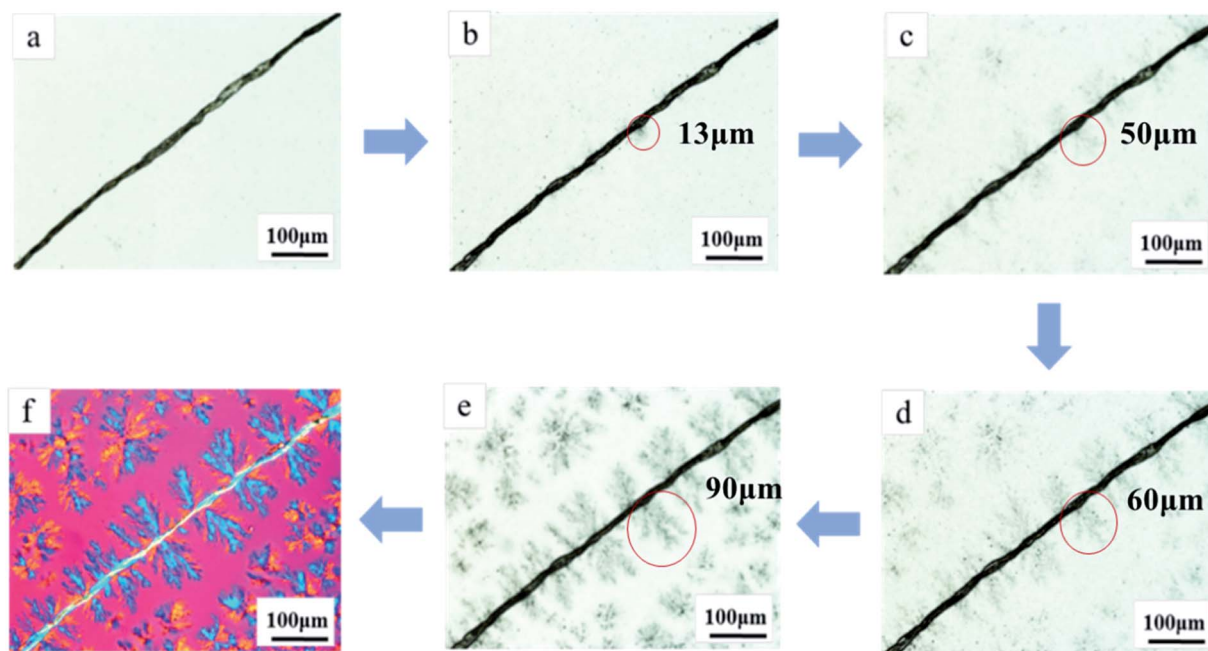


Fig. 2 PLM photos for NF/PP containing 0.3 wt% NA annealed at 250 °C for 2 min during the cooling process at a rate of 30 °C min⁻¹: (a) 250 °C; (b) 170 °C; (c) 162 °C; (d) 156 °C; (e) 140 °C; (f) the corresponding photo of (e) taken in polarized mode.

with surrounding PP matrix to significantly enhance the interfacial friction, resulting in the difficulty of NF pulling-out from polymer matrix under mechanical load. Accordingly, the load can transfer effectively from PP matrix to the NF through the interfacially-grown NA fiber, which in turn endowed the strong interfacial interaction of the NF-PP interface.^{9,10} Different from the previous chemical modification methods, this study provided a new idea to modify the interface between the fillers and polymer matrix *via* interfacial interlocking; moreover, compared to the glass fiber or hydroxyapatite featuring inert activity to NAs, NFs carry massive active hydroxyl groups to interact with amide groups of NA so the interfacial fibrous

crystals can be directly generated without preliminary surface modification.

The formation process of root-like NFs involved the directed diffusion of the NA molecules at the NF surface and the subsequent epitaxial growth. Therefore, fine control of the interfacial grown NA fibers can be achieved by adjusting the diffusion and growth processes. The molecular diffusion is a kinetic process, depending on the annealing temperature and time. As shown in Fig. 3a–c, with the increasing annealing temperature, NAs displayed stronger mobility and the interfacial diffusion became more remarkable, leading to denser and longer branched NA fibers. Further, at the prolonged annealing time, NAs had higher

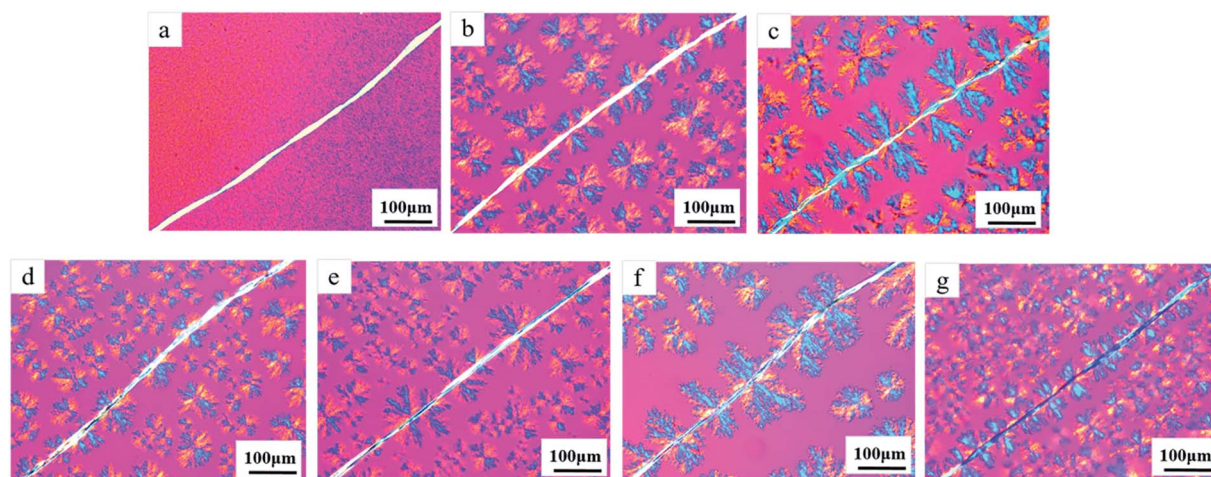


Fig. 3 PLM photos for single NF/PP with 0.3% NA annealed for 2 min at different temperatures followed by cooled at the rate of 30 °C min⁻¹: 200 °C (a), 230 °C (b) and 250 °C (c); cooled at the rate of 30 °C min⁻¹ annealed at 230 °C for different annealing times: 0.5 min (d) and 1 min (e); annealed for 2 min at 230 °C followed by cooled at different cooling rate: 10 °C min⁻¹ (f) and 50 °C min⁻¹ (g).



probability to diffuse from PP bulk matrix to the NF surface and thus root-like NF with larger and denser NA fibers was generated. Additionally, the growth of branched NA fiber can be regulated by controlling the cooling rate. Slow cooling can facilitate the perfect growth of NA fibers. As shown in Fig. 3f and g, when the cooling rate was reduced from 50 to 10 °C min⁻¹, larger NA fibers were generated at the surface of NF.

Conclusions

The hydrogen bonding between amide groups of NAs and abundant hydroxyl groups existing in the NF fibers can direct the interfacial diffusion of the former and induce the lateral growth at the NF surface, into root-like NFs with mechanically interlocking with PP matrix, which is favorable to the load transferring from polymer matrix to the reinforced NF. Moreover, fine control of the interfacial grown NA fibers is achieved by adjusting the diffusion and growth processes, which are in turn determined by the annealing temperature, annealing time and cooling rate. Compared to conventional chemical modification, the functional interfacial geometry is *in situ* formed in PP melts without any complicated chemical pre-treatment, which holds great potential in the NF-reinforced polymer composites.

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