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Efficient organic–inorganic intumescent interfacial flame retardants to prepare flame retarded polypropylene with excellent performance

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In this article, an efficient and simple approach for the preparation of organic–inorganic intumescent interfacial flame retardants, aiming at enhancing the flame-retardant efficiency and interfacial adhesion between matrix and flame retardants was presented. The expandable graphite (EG) was functionalized by using a grafting process containing phosphorous, resulting in the formation of organic–inorganic intumescent flame retardants. Based on the successful grafting reaction, a series of flame-retardant polypropylene (PP) composites with different content of modified EG (MEG) were prepared and evaluated. With the incorporation of 30 wt% of MEG into PP, the satisfactory UL-94 flame retardant grade (V-0) and limiting oxygen index (LOI) as high as 25.3% were obtained. The residues of the PP/MEG composites were significantly increased with PP/EG and PP/EG/DOPO composites. Moreover, the residual char of PP/MEG composites is more compact and integrated. In addition, the formation of organic side chains on the MEG surface by the grafting reaction also contributed to an improvement in the interfacial compatibility, leading to an enhancement in mechanical properties of the composites compared with the PP composites filled with EG. The interfacial grafting flame retardants provided a novel way to prepare organic–inorganic intumescent flame retardants and the as-prepared flame retardants exhibited excellent flame retardant efficiency.

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1. Introduction

Polypropylene (PP) is a polymer material which is widely used in various industries such as in chemical, petroleum and building. However, PP is flammable and will release CO, combustible gas and other toxic gases during combustion which are very harmful to human health and the environment.^{1,2} Furthermore, the flammability of PP is a disadvantage in some applications industries where high flame retardancy is required, for example, in electrical applications, transportation sectors and daily necessities.^{3–7} Therefore, seeking desirable flame retardant for PP is quite necessary. At present, the main method is to mix flame retardant into the polymer directly. Over the past two decades, with the limitation of halogen-containing flame retardants and increasing requirements of high fire security for residential safety and environmental concerns, the main effort on flame retardants of PP has shifted to the development of halogen-free flame retardants. Considering the inherent

flammability of PP and its application requirements, a suitable halogen-free flame retardant should be efficient enough in low additions to enhance flame-retardant performance and simultaneously to maintain the good mechanical and thermal properties of PP.^{8–12}

As a newly emerged intumescent additive, expandable graphite (EG) is one of the halogen-free flame retardants, which has a special graphite intercalation compound where sulfuric acid (H₂SO₄) is intercalated between the carbon layers.^{13–15} EG can exfoliate or expand along the *C* axis of the crystal structure by about 100 times and forms a ‘worm-like’ structure with numerous ordered graphite flakes which can act as a protection layer of the polymer matrix to induce the heat and gas transfer during combustion.^{16,17} However, it is usually observed that the incorporation of EG into polymer matrix not only reduces the compatibility between the matrix and the filler, but also the residual carbon layer is quite loose, which results in less protection to the inner matrix. Therefore, it is proposed that the phosphorus-containing flame retardant which can decompose into phosphoric acid during the high temperature can effectively solve this problem.

Phosphorus-containing flame retardants at high temperature can produce phosphate–carbon complexes which are thick and have high temperature resistance effect. Therefore, it can agglomerate the ‘worm-like’ carbon layer and form a physical

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barrier layer with excellent thermal stability property at high temperature. Phosphorus-containing flame retardants seem to offer a promising halogen-free method for the PP matrix and its composites in the future. Especially, 9,10-dihydro-9-oxa-10-phosphaphenanthrene-10-oxide (DOPO) is one of the most important phosphorus-based flame retardants because of its excellent environmental compatibility. DOPO can act its role not only in the gas phase (*via* gas production), but also in the condensed phase (*via* char formation). With high reactivity, the P–H group of DOPO can react with many functional groups such as epoxides,^{18–21} providing a novel way for preparing the organic–inorganic intumescent interfacial flame-retardants. However, it also has some negative impacts on thermal properties and compatibility between matrix and fillers. In order to graft DOPO onto the EG, the silane coupling agent with epoxide group named γ -glycidyloxypropyltrimethoxysilane is designed as a bridge to combine DOPO and EG. Phosphorus and silicon also have been used to polymers flame retardant.^{22–25} The Si–OH groups produced from siloxane by hydroxylation can react with the hydroxyl groups of EG to form Si–O bonds. The P–H groups of DOPO can react with the epoxides of silane coupling agent to form C–P–O bonds.^{26–28} Herein, we propose a novel resistance mode that can effectively suppresses these problems by grafting DOPO molecular onto the EG using silane coupling agent. The surface modification has been applied to the inorganic to increase the compatibility between the filler and the polymer matrix.

In this work, a novel organic–inorganic interfacial intumescent flame retardant mode (MEG) is proposed by using grafting process containing phosphorous and applied to the EG flame retardant PP composite systems which exhibit excellent flame retardant efficiency. In this mode, the residue char is designed to increase with EG rather than independently produced from the bulk polymer as in the conventional mode. The structure of grafted EG is characterized by Fourier transform infrared (FTIR) and scanning electron microscopy (SEM). The flammability and mechanical properties of PP/MEG, PP/EG, PP/EG/DOPO composites are investigated systematically by thermogravimetry analysis (TGA), limiting oxygen index (LOI), vertical burning test (UL-94), tensile and impact test. X-ray photoelectron spectroscopy is used to explore the char residues of the PP composites.

2. Experimental

2.1. Raw materials

Polypropylene (PP) in this work was a commercial polymer PP-B4808, purchased from Yanshan Petrochemical, China. Its melt flow index (MFI) was 10 g/10 min. Expandable graphite (EG) with particle size of 270 μm was provided by Qingdao Kangboer Graphite Company. Phosphorus-containing flame retardant, 9,10-dihydro-9-oxa-10-phosphaphenanthrene-10-oxide (DOPO) was obtained from Shengshida Tech & Trade Co., Ltd., China. The γ -glycidyloxypropyltrimethoxysilane (silane coupling agent) was purchased from Chengdu Chenbang Chemical Co., Ltd., China. Ethanol and ethylic acid was obtained from Chengdu Kelon Co Ltd., China.

2.2. Preparation of grafted expandable graphite (MEG)

Modification of EG particles was performed according to the following steps. All materials were dried for 6 h to eliminate possible absorbed water on the surface of the particles. A calculated amount of DOPO and silane coupling agent were mixed in the three-necked flask and heated for 4 h at 160 °C to obtain the DOPO–silane coupling agent. Then a mixed solution A was prepared with moderate amount of deionized water and EG, and the pH was adjusted to 4 by adding ethylic acid. Then solution B which composed of DOPO–silane coupling agent and ethanol was added in the solution A and stirred for 4 h at 60 °C. Finally, the obtained mixed solution was repeatedly washed with ethanol aqueous solution to remove the excess reactants from the surface. The grafted EG product was called MEG. Schemes 1 and 2 illustrate the grafted synthesis of MEG intumescent flame retardants.

2.3. Preparation of the composites

PP pellets, EGs (MEG or EG), and DOPO (only for bulk mode formulation) were premixed in a mixer and then the mixture was extruded in a twin-screw extruder (TSE-20A1600-4-40, made in Nanjing, China) with a rotation rate of 150 rpm. The temperatures from hopper to die were 170, 180, 190, 205, 210, 210, and 205 °C respectively. The extruded granulates were cooled in a water bath and cut into pellets, then dried in drier at 60 °C for 6 h. The dried granulates were injected into standard test bars at 200, 210, and 205 °C from hopper to die in an injection molding machine (EM80-V, Chen De Plastics Machinery).

In addition, there is about 200 g EG used for grafting in this experiment and 216 g product obtains finally. Based on the increased mass of product, a 7.4% grafting ratio for DOPO on the surface of EG is calculated. Therefore, the DOPO/EG (2.22/30) = 7.4% in the PP/EG/DOPO composites is reasonable. The formulations of the flame-retardant PP composites were presented in Table 1.

2.4. Measurements and characterization

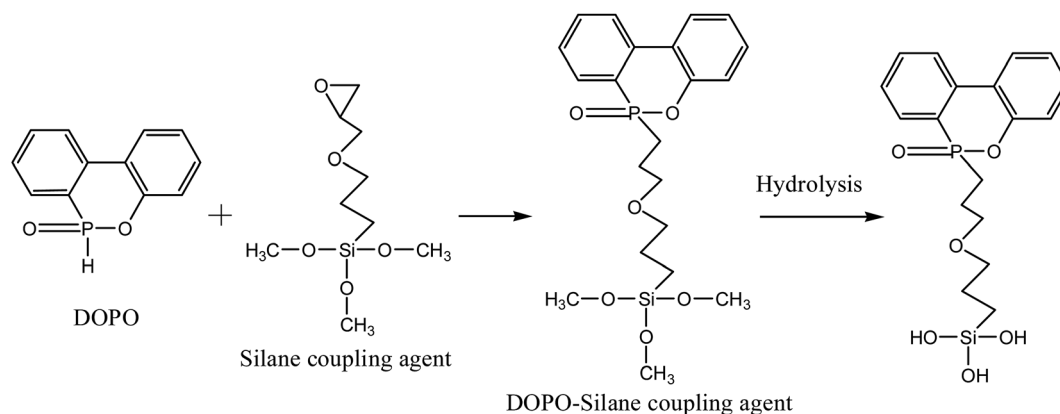
Fourier transform infrared spectra (FTIR) of EG, MEG and DOPO were recorded on a Nicolet 560 spectrophotometer. The powders of materials were mixed with KBr powders, and then the mixture was compressed into plates for FTIR spectra analysis.

The LOI values were surveyed using a JF-4 type instrument (manufactured by Jiangning Analysis Instrument Factory, Nanjing China) on sheets 120 \times 6.5 \times 3 mm³ according to the standard oxygen index test (ASTM D 2863-77).

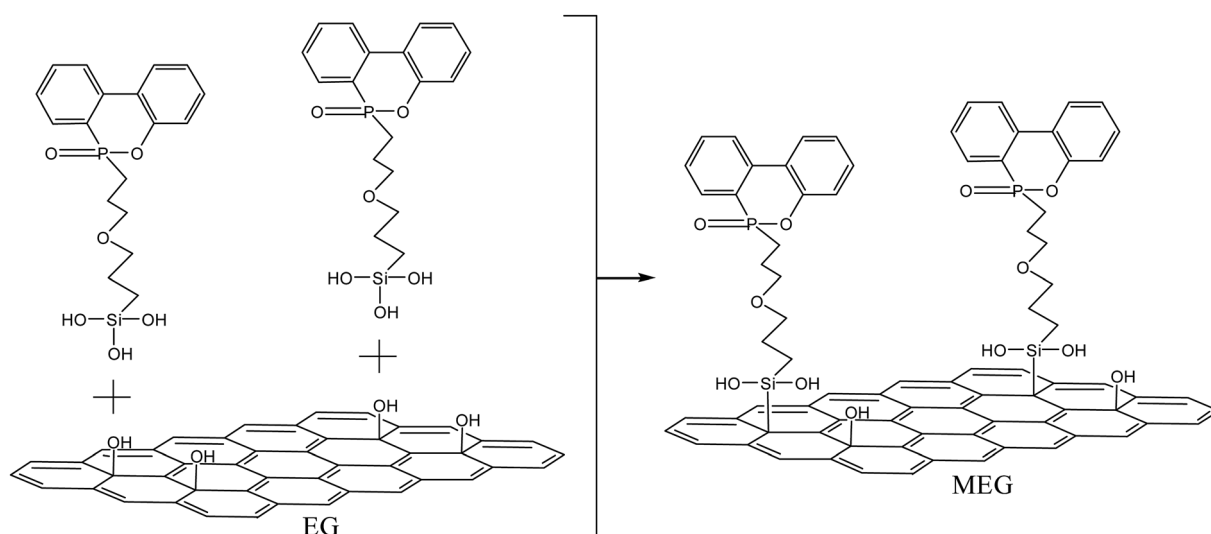
The Underwriters Laboratories-94 (UL-94) vertical burning test was carried out on a CZF-1 type instrument (Nanjing Jiangning Analytical Instrument Factory, China), with the bar dimensions of 127 \times 12.7 \times 2.7 mm³ according to ASTM D635-77.

Thermal gravimetric analysis (TGA) was performed on a NETZSCH TG 209F1 thermogravimetric analyzer at a heating rate of 10 °C min^{−1} under nitrogen with a flow rate of 60 mL





Scheme 1 Reaction scheme of DOPO and silane coupling agent.



Scheme 2 Reaction schemes of DOPO-silane coupling agent and EG.

min^{-1} and the scan range was from 30 to 700 °C. About 8–10 mg of the sample was used in this test.

The surface morphologies of the EG, MEG, burned EG and burned MEG were observed on a scanning electron microscope (SEM) (Model JSM-7500F, Japan) with a conductive gold coating and with an acceleration voltage of 5.0 kV.

X-ray photoelectron spectroscopy (XPS) of PP/EG30 and PP/MEG30 were recorded to determine the elements in the char residue on a XSAM 800 machine.

The tensile strength of composites was measured at ambient temperature using a tensile tester AGS-J (Autograph SHIMADZU) according to GB 1040-92 (crosshead speed 100 mm min^{-1}). The Notched Izod impact tests were measured at ambient temperature on a pendulum impact tester (Model JBS-300B; Shandong Drick Instruments, China) according to the ASTM D3420 standard. The results were the average values of at least five specimens.

3. Results and discussion

3.1. The synthesis reaction and characterization of MEG

The mechanism of grafting synthesis for MEG are illustrated in Schemes 1 and 2, which contains two steps. One is the P-H groups of DOPO can react with the epoxides of silane coupling agent to form C-P-O bonds and the hydrolysis reaction of DOPO-silane coupling agent, another is grafting DOPO-silane coupling agent on the surface of EG. The Si-OH groups produced from siloxane by hydroxylation can react with the hydroxyl groups of EG to form Si-O bonds. Therefore, the

Table 1 Formulations of pure PP and its flame-retardant composites

Samples	PP (wt%)	MEG (wt%)	EG (wt%)	DOPO (wt%)
PP	100	0	0	0
PP/MEG10	90	10	0	0
PP/MEG20	80	20	0	0
PP/MEG30	70	30	0	0
PP/EG30	70	0	30	0
PP/EG/DOPO	70	0	27.78	2.22



grafting reaction of DOPO onto the surface of EG is easy to be successfully realized.

Fig. 1 shows the FTIR spectra of DOPO, EG and MEG. The characteristic bands of EG and MEG at 1637, 2854 and 2923 cm^{-1} are attributed to the C=C stretching vibrations and C-H stretching vibrations in the aliphatic hydrocarbons, respectively. In the spectrum of EG, the strong peaks at 3432 and 1128 cm^{-1} are assigned to the stretching vibrations of -OH and C-O groups, respectively. This indicates that EG has -OH groups that provide the ability to synthesize the product of hydrolysis reaction of DOPO and silane coupling agent. In the DOPO's spectrum, the band at 2437 cm^{-1} is attributed to P-H stretching. After grafting onto EG, this peak is disappeared, and a new peak at 754 cm^{-1} attributed to the P-C stretching vibration is appeared which suggests the occurrence of the reaction between P-H and epoxide groups. That of MEG has another new absorption peak at 1581 cm^{-1} compared with the spectrum of EG, attributing to the benzene ring vibration of DOPO. This can confirm that DOPO is successfully grafted onto the surface of EG. Additionally, the 3432 cm^{-1} of MEG shows a relatively reduction in the intensity, which confirms the reaction between DOPO and EG. Assignments of the peaks in FTIR spectrum are presented in Table 2.

SEM analysis is used to investigate the surface morphology to confirm the formation of MEG in Fig. 2. The surface of EG is very smooth and the graphite sheet layer is clearly observed in Fig. 2a. As can be seen, the rough MEG surface seen in Fig. 2b indicates the continuous covering on the surface. This result confirms the grafting reaction is happened. Because of DOPO and silane coupling agent are easily dissolved in ethanol aqueous solution, so the excess reactants can easily be washed out by enough ethanol aqueous solution.

The TGA curves of EG and MEG in N_2 atmosphere are presented in Fig. 3. Correspondingly, the temperatures, at which the relative mass loss is 5% ($T_{5\%}$), 30% ($T_{30\%}$) and 50% ($T_{50\%}$), respectively, are obtained from TGA curves and listed in Table 3.

Table 2 Assignments of the peaks in FTIR spectra shown in Fig. 1

Wavenumbers (cm^{-1})	Assignment
3432	Stretching vibration of O-H groups
2923, 2854	C-H stretching vibration in -CH ₃ or -CH ₂ -
2437	Stretching vibration of P-H bond
1637	Stretching vibration of C=C bond
1581	Benzene ring
754	Stretching vibration of P-C bond

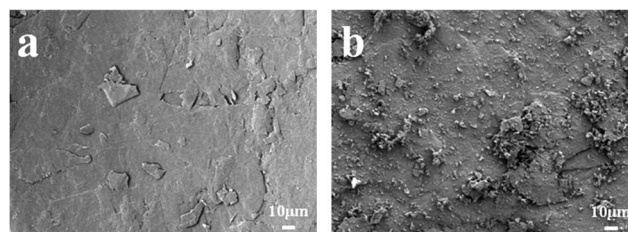


Fig. 2 The surface morphology of (a) EG and (b) MEG.

The $T_{5\%}$ and $T_{30\%}$ of EG are 222.2 °C and 266.8 °C respectively. However, the MEG exhibits a lower thermal stability at temperature ranging from 150 °C to 300 °C due to the low thermal stability of DOPO and silane coupling agent which will decompose to phosphate and SiO_2 , but a higher thermal stability at temperature above 300 °C compared with the EG. The charred residue of MEG is 55.6% at 600 °C and is much higher than that of EG. It is the thermal decomposition of DOPO that promotes EG to form a compact char layer under high temperature. Therefore, the MEG achieves better thermal stability performance than EG at high temperature.

3.2. The flame-retardant performances of the composites

Limiting oxygen index (LOI) and UL-94 vertical burning tests have been widely used to evaluate the flammability of polymers

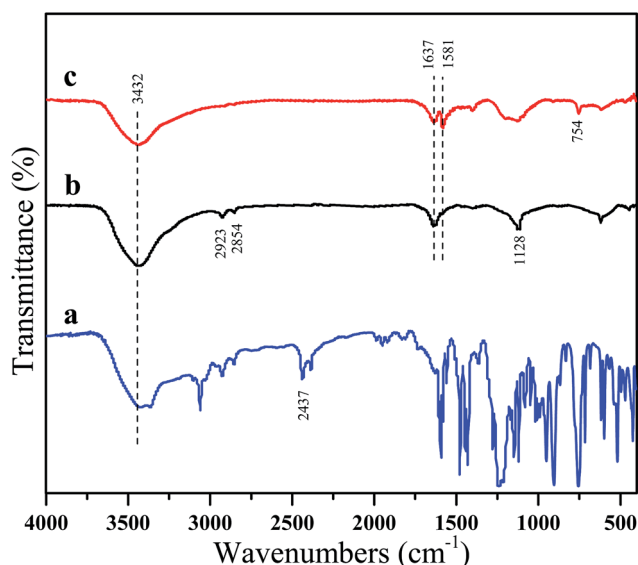


Fig. 1 FTIR spectra of (a) DOPO, (b) EG and (c) MEG.

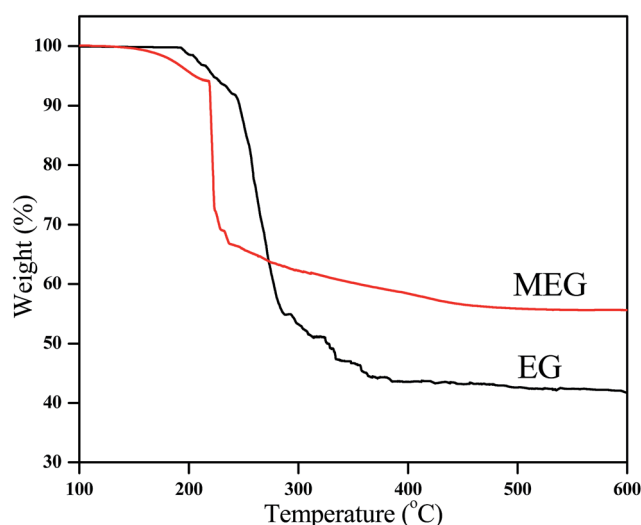


Fig. 3 TGA curves of EG and MEG in N_2 atmosphere.



Table 3 TGA data of PP and its flame-retardant composites^a

Samples	$T_{5\%}$ (°C)	$T_{30\%}$ (°C)	$T_{50\%}$ (°C)	T_p (°C)	Residues at 600 °C (wt%)
EG	222.2	266.8	—	—	41.8
MEG	204.9	227.6	—	—	55.6
PP	371.1	428.9	444.0	454.2	0.3
PP/MEG30	338.4	433.7	457.2	464.7	25.8
PP/EG30	367.4	465.2	475.9	476.6	24.9
PP/EG/DOPO	356.0	454.4	469.2	472.4	24.5

^a $T_{5\%}$, $T_{30\%}$, and $T_{50\%}$, onset decomposition temperature of 5%, 30%, and 50% weight loss, respectively; T_p , peak temperature of DTG.

and their composites. The flame retardant performances of PP composites filled with EG, MEG and DOPO separately are characterized and shown in Table 4. It is observed that pure PP exhibits a LOI value of 18.1% and is not classified in the UL-94 tests. This means pure PP is highly flammable. The addition of inorganic–organic intumescent flame retardants has great effect on the flammability of materials. With the addition of MEG into PP, the LOI value of PP/MEG composites increases gradually. When 10 wt% of MEG is added, the LOI value rises to 21.9%, however, this sample is failed in the UL-94 test. As the content of MEG increases to 30 wt%, the LOI value is improved to 25.3% from 18.1% for pure PP and the satisfied flame retardant property (UL-94 V-0) is obtained.

Unfortunately, EG and EG/DOPO have slight effects on the LOI value and UL-94 rating. As listed in Table 4, the LOI values of PP/EG30 and PP/EG/DOPO are only 23.2 and 24.2%, respectively. In addition, PP/EG30 fails in the UL-94 test, however, PP/EG/DOPO only achieves the UL-94 V-1 rating. The flammability is a little better than the PP/EG30, showing that DOPO and EG have a synergistic effect in the flame-retardant PP, but cannot improve the flame retardancy notably. Although EG is known to be an efficient flame-retardant additive that performs both as a carbon source and as a gas source.

On the other hand, the PP/MEG30 system exhibits outstanding flame-retardant properties and the UL-94 result reaches V-0 rating. The results indicate that the inorganic–organic interfacial intumescent flame retardants play great roles in the fire resistances. It is the DOPO–silane coupling agent system that can catalyze polymer matrix to form a dense carbon layer when the MEG particles are heated and the formed compact layer can effectively block the outside heat and oxygen to protect the internal polymer matrix. However, the DOPO

particles in PP/EG/DOPO is randomly dispersed in EG and PP matrix. The phosphate–carbon complexes cannot achieve synergistic flame retardant effect with EG at high temperature. The surprising improvement in the flame retardancy with only 2.22% DOPO–silane coupling agent concentrated on the interface implies that the flame-retardant method extremely efficient.

The digital photographs of the burned splines and SEM morphologies of pure PP and PP composites after combustion are shown in Fig. 4 and 5, respectively. It is clearly seen that the pure PP burns sharply with dripping continuously, which is very dangerous and easy to cause “secondary damage”, and little char residue is found on the surface of PP sample (Fig. 4a). Some intumescent char residues are observed covered on the surface of PP/MEG composites, and the content of char residues increase with increasing the MEG content, as shown in Fig. 4b and c. But the quantity and quality of char layers are not good



Fig. 4 Char morphologies of pure PP and the PP composites after UL-94 test: (a) PP, (b) PP/MEG10, (c) PP/MEG20, (d) PP/MEG30, (e) PP/EG30, (f) PP/EG/DOPO.

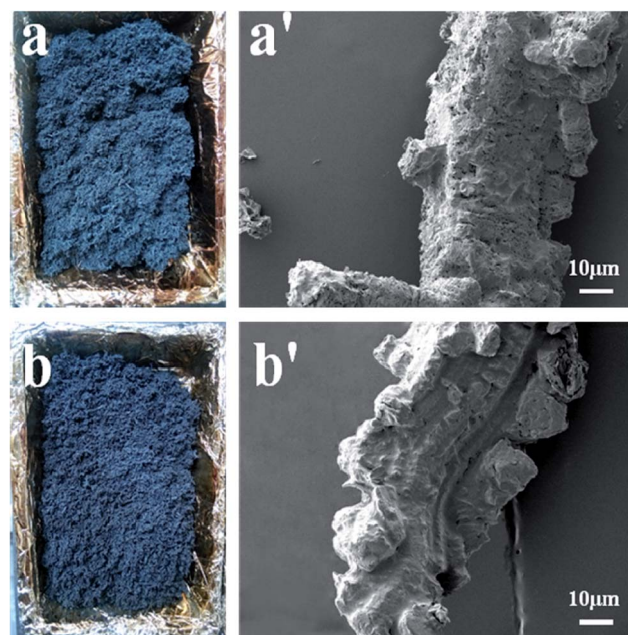


Fig. 5 The digital photographs and SEM photographs of residue surface of the flame-retardant PP composites after burning: (a, a') PP/EG30 and (b, b') PP/MEG30.

Table 4 Formulations and flammability of pure PP and its flame-retardant composites

Samples	UL-94	LOI (%)
PP	NR	18.1
PP/MEG10	NR	21.9
PP/MEG20	NR	23.1
PP/MEG30	V-0	25.3
PP/EG30	NR	23.2
PP/EG/DOPO	V-1	24.2



enough to act as an effective protective shield for underlying polymer matrix, so the PP/MEG10 and PP/MEG20 samples cannot obtain good results in LOI and UL-94 tests. While for PP/MEG30, an expanding and compact char layer is clearly seen. And this sample quickly extinguishes off the fire. It is observed from Fig. 5b and b' that the charred residue of PP/MEG30 is more compact, continuous, and integrated. The char layer can act as an efficient and good barrier to prevent the combustible gas and heat flow from transferring into the polymer materials. The main reason is that the thick phosphate-carbon complexes and silica produced by the decomposition of DOPO and silane coupling agent can effectively agglomerate the carbon layer at high temperature. Therefore, the good synergistic effect between organic (DOPO) and inorganic (EG) intumescent flame retardants. However, compared with MEG, when the same content of EG is added into PP (Fig. 4e), it can be found that PP/EG30 sample shows a high flame rate as the bars burned out without extinguishment after ignition. In addition, the char layer after burning is loose, discontinuous and brittle, and there are many holes on the surface of the char layer, as shown in Fig. 5a and a'. This will promote the flammable gas and heat flow transfer into underlying materials. While for PP/EG/DOPO, the morphology of charred residue shown in Fig. 4f shows that the vertical flammability is violent compared to PP/MEG sample. This indicates that there is not enough fire retardant present to overcome the wick effect and enhance the flame retardancy remarkably.

3.3. Thermal stability of the composites

The TGA and DTG curves of pure PP, PP/MEG30, PP/EG30 and PP/EG/DOPO composites as a functional temperature at the heating of $10\text{ }^{\circ}\text{C min}^{-1}$ in nitrogen are presented in Fig. 6 and 7, respectively, and the detailed data are listed in Table 3. In addition, the T_p is defined as the temperature which maximum weight loss rate occurs, corresponding to the peak temperature of DTG. It is clearly found that the thermal degradation of pure PP in N_2 atmosphere is characterized by one-step with $T_{5\%}$ and T_p at $371.1\text{ }^{\circ}\text{C}$ and $454.2\text{ }^{\circ}\text{C}$, respectively. And the resulting charred residue at $600\text{ }^{\circ}\text{C}$ is about 0.3 wt%.

While for PP/MEG30, PP/EG30 and PP/EG/DOPO composites, the two decomposition steps of weight loss are exhibited: the first step takes place from around 220 to $380\text{ }^{\circ}\text{C}$; the second one, from 380 to $500\text{ }^{\circ}\text{C}$, may be attributed to the degradation of residual materials. The flame-retardant PP composites present a lower weight loss rate and higher charred residues at $600\text{ }^{\circ}\text{C}$ than pure PP. Compared with pure PP, it is observed that the $T_{5\%}$ of PP/MEG30, PP/EG30 and PP/EG/DOPO are reduced by $32.7\text{ }^{\circ}\text{C}$, $3.7\text{ }^{\circ}\text{C}$ and $15.1\text{ }^{\circ}\text{C}$, respectively. This is attributed to the thermal stability of DOPO and silane coupling agent is lower than EG, and the thermal stability of EG is lower than PP. PP/MEG30 contains DOPO, silane coupling agent and EG, so it has the lowest $T_{5\%}$ temperature. PP/EG30 contains only EG, so it has the highest $T_{5\%}$ temperature. However, PP/MEG30 exhibits an enhanced thermal stability at temperature ranging from 400 to $600\text{ }^{\circ}\text{C}$. PP/MEG30 has the highest charred residue which is 25.8 wt% at $600\text{ }^{\circ}\text{C}$ compared to PP/EG and PP/EG/DOPO. The

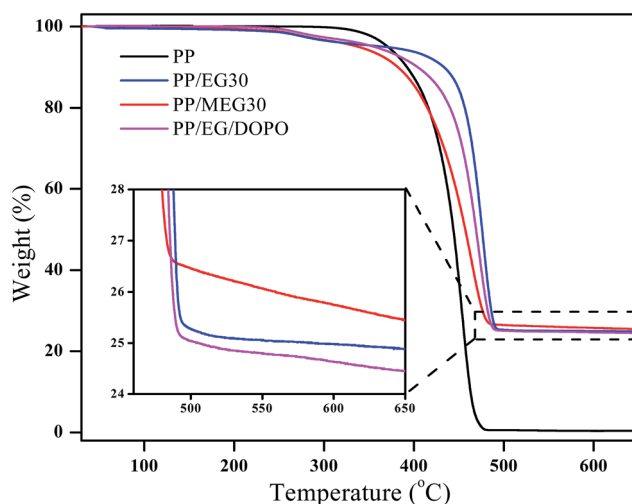


Fig. 6 TGA curves of PP, PP/EG30, PP/MEG30, and PP/EG/DOPO composites in N_2 atmosphere.

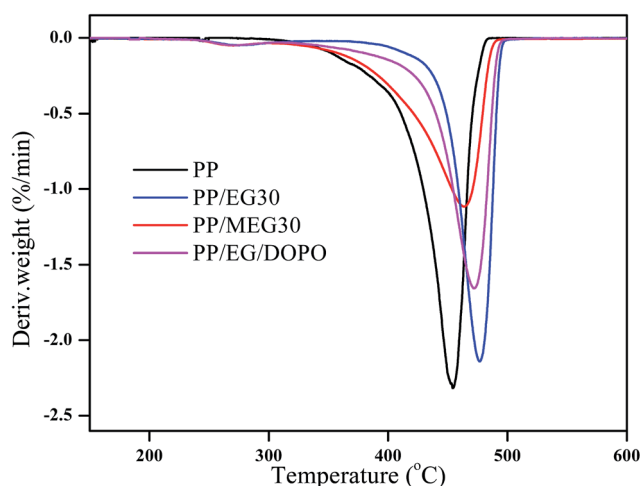


Fig. 7 DTG curves of PP, PP/EG30, PP/MEG30, and PP/EG/DOPO composites in N_2 atmosphere.

higher charred residue produced, the higher barrier properties of heat and gas transfer between PP matrix and combustion zone obtained.

The participation of MEG improves the charred residue remarkably and has an obvious effect on the thermal degradation behaviors of PP composites. The TGA curves of PP composites suggest a two-stage decomposition process. In the first step, it is seen from Fig. 6 that the initial decomposition temperature of the PP composites is lower than that of pure PP due to the low thermal stability particles of DOPO and silane coupling agent which decomposes to form SiO_2 and viscous phosphate-carbon complexes. At second step, with increasing temperatures, the phosphate-carbon complexes begin to dehydrate the polymer matrix and the agglomerate carbon layer to form a physical barrier layer. Then the decomposition of PP matrix becomes relatively slower due to the continuous and dense carbon layer. According to the DTG curves shown in



Fig. 7, it can be clearly seen that the temperature of the maximum decomposition rate of PP/MEG30 is the lowest. These results indicate that the addition of MEG has less influence on initial decomposition temperature, but greater influence on the thermal stabilities of PP composites at high temperature.

3.4. Thermo-oxidative degradation of the composites

Dynamic FTIR is employed to evaluate the solid pyrolysis products of PP composites. The FTIR spectra of PP/EG30 and PP/MEG30 with increasing the pyrolysis temperatures are shown in Fig. 8 and 9, respectively, and the assignments of FTIR spectra are collected in Table 5. It is obviously seen from Fig. 8 that the peaks at 2960, 2920, 2840, 1640, 1460, 1380 and 997 cm^{-1} , assigning to the CH_2 and CH_3 vibration and the C-C stretching vibration of aromatic ring, are the characteristic absorptions of pure PP.²⁹ The band at 3440 cm^{-1} decreased at the temperature of 350 °C, and this can be explained by the release of water. The relative intensities of other characteristic peaks do not change below 350 °C. While the temperature rises to 400 °C, it's found that the absorption peaks at 2960, 2920, 2840, 1640, 1460, 1380 and 997 cm^{-1} disappear, indicating that the main decomposition happens at this stage. This is consistent with the TGA results.

While for PP/MEG30, the peaks at 2960, 2920, 2840, 1640, 1460, 1380 and 997 cm^{-1} are the characteristic absorptions of PP, as shown in Fig. 9. The FTIR spectra show very similar features compared with that of PP/EG30. When the temperature rises to 350 °C, it is found that the intensities of most of peaks decrease sharply, indicating that the main decomposition of PP/MEG30 happens at this stage. The presence of MEG catalyzes the degradation of PP matrix, which corresponds well with TGA results. When the temperature is over 350 °C, two peaks at 1260 and 1120 cm^{-1} appear, while those are not visible in the FTIR spectra of PP/EG30. The peaks can be ascribed to the stretching vibration of $\text{P}=\text{O}$ and $\text{P}-\text{O}-\text{P}$ in phosphate-carbon complexes.³⁰

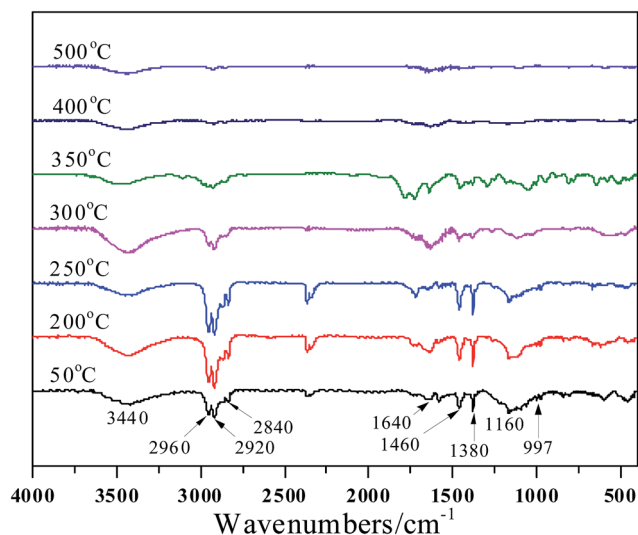


Fig. 8 The dynamic FTIR spectra at different pyrolysis temperatures of PP/EG30.

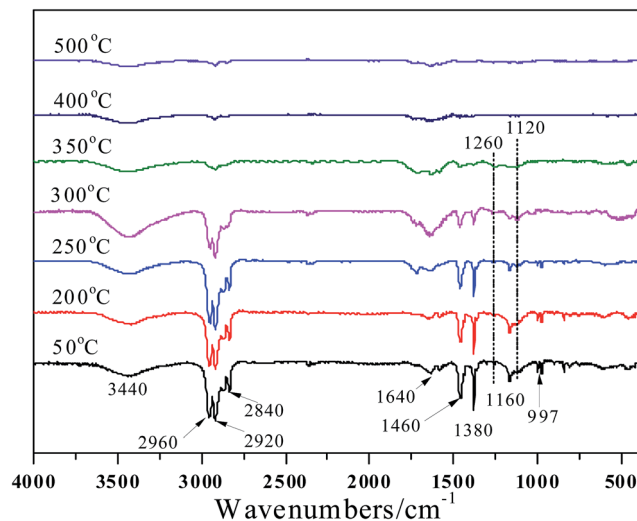


Fig. 9 The dynamic FTIR spectra at different pyrolysis temperatures of PP/MEG30.

Table 5 Assignment of FTIR spectra of the PP/EG30 and PP/MEG30 samples

Wavenumbers (cm^{-1})	Assignment
3440	Stretching vibration of O-H groups
2960, 2920, 2840	C-H stretching vibration in $-\text{CH}_3$ or $-\text{CH}_2-$
1640	C-C stretching vibration of aromatic ring
1460	Scissor vibration of $-\text{CH}_2$
1380	Symmetric vibration of $-\text{CH}_3$
1260	Stretching vibration of $\text{P}=\text{O}$ bond
1160	C-O stretching vibration
1120	Stretching vibration of $-\text{P}-\text{O}-\text{P}-\text{O}$
997	Vibration of CH_2-CH (CH_3)

The presence of polyphosphate catalyzes the formation of the compactable char, which can delay the thermal degradation during the combustion. Therefore, the better flame-retardant performances can be achieved for PP/MEG30. At higher temperature (>400 °C), most of peaks are disappeared and only exist the residual char.

Aside from the qualitative analysis by FTIR, quantitative analysis is performed to determine the residual char of PP/EG30 and PP/MEG30 composites (heated in muffle furnace for 20 min at 600 °C) by using XPS to measure variation of the characteristic element concentrations on the surface of charred residues. Fig. 10 and Table 6 show the XPS spectra and data for PP/EG30 and PP/MEG30 samples, respectively. For C1s spectrum of PP/EG30 has four peaks:^{31,32} the peak at around 284.62 eV is attributed to C-H and C-C in aromatic species; the peak at around 286.05 eV is assigned to C-O (hydroxyl group); the peak at around 287.51 eV is corresponded to carbonyl, and the peak at around 290.29 eV is corresponded to carboxyl groups. O1s spectrum has the peaks around 533.75 and 533.40 eV are assign to $-\text{O}-$ in C-O-C or C-OH group, the peak at around 531.95 eV is attributed to $\text{C}=\text{O}$. In addition, for the PP/MEG30 sample, O1s spectrum has another peak around 532.70 eV. It is reported that



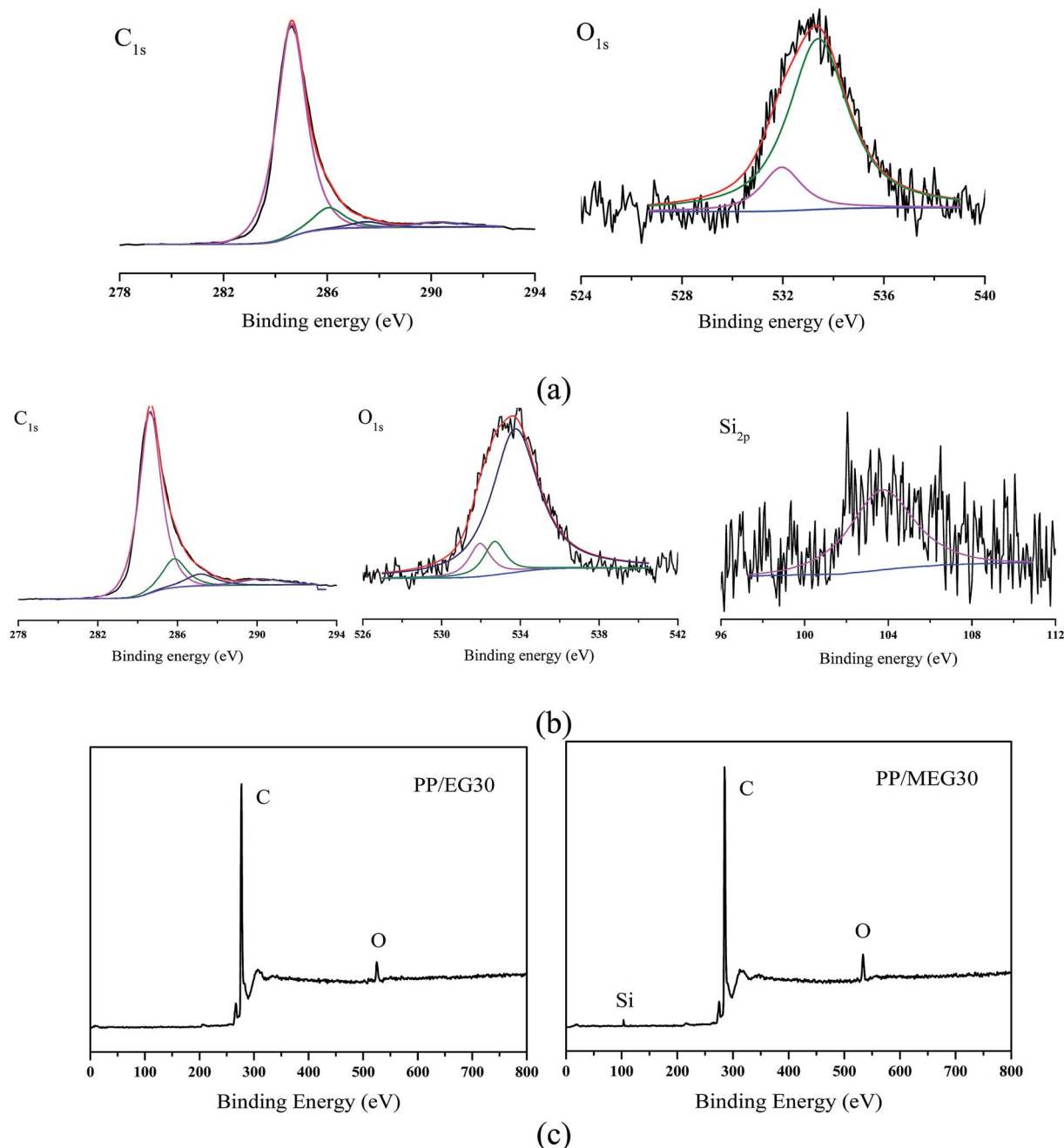


Fig. 10 XPS analysis of PP/EG30 and PP/MEG30: (a) C_{1s} , O_{1s} spectra of PP/EG30, (b) C_{1s} , O_{1s} and Si_{2p} spectra of PP/MEG30, (c) wide scanning spectra of PP/EG30 and PP/MEG30.

the peak at 532.70 eV can be attributed to the Si–O group. For the Si spectrum, the peak at 103.71 eV can be assigned to the Si–O group formed by decomposition products of silane coupling agent at high temperature.

It can be found from Table 6 that the addition of MEG influences the flame retardancy and the thermal degradation of the composites due to the difference in element concentrations. The atom percent of carbon in the residues char of PP/EG30 and PP/MEG30 are 96.02% and 93.96%, respectively. And the atom percent of oxygen and silicon in the residues char of PP/MEG30 are higher than that in the char residues char of PP/EG30. The

presence of Si element promotes the synergistic flame retardancy for organic–inorganic intumescent flame retardants in the flame-retardant PP composites. These mean that the incorporation of DOPO–silane coupling agent into PP can promote the formation of compact residues char and enhance the thermal oxidative resistance of the PP matrix.

3.5. Flame retardant mechanism analysis

The surface flame retardant distribution and flame resistance modes of PP/MEG30 are illustrated in Fig. 11. It can be



Table 6 XPS results for the surface of the PP/EG30 and PP/MEG30 samples

Elements	PP/EG30			PP/MEG30		
	Content (%)	Binding energy (eV)	Peak attribution	Content (%)	Binding energy (eV)	Peak attribution
C	96.02	284.62	C-C, C-H	93.96	284.63	C-C, C-H
		286.05	C-O		285.84	C-O
		287.51	C=O		287.14	C=O
		290.29	-COOH		290.20	-COOH
O	3.98	531.95	C=O	5.67	531.95	C=O
		533.40	-O-		532.70	Si-O
		—	—		533.75	-O-
Si	0.00	—	—	0.37	103.71	Si-O

speculated that the lamella of EG, the particles of DOPO and silane coupling agent are dispersed in the PP matrix. After ignition, PP matrix releases flammable gas because of the thermal decomposition of the composites. For one thing, EG expands and forms worm-like structure, and then acts as a physical barrier for inhibiting the heat and gas transfer. For another, the thermal decomposition of DOPO-silane coupling agent produces SiO_2 and polyphosphoric acid, which promotes the degradation of PP matrix to form compact char layer on the surface of PP. Accordingly, it is only if a great amount of continuous and dense chars that obtain from physical barrier of EG and catalyze charring formation of DOPO. As the organic-inorganic intumescent interfacial behaviors are controlled by dynamics or thermos-dynamics, the wick effect of EG will be greatly weakened. This will improve the fire resistance of the composites.

3.6. Mechanical properties of the composites

Incorporation of the organic phase and inorganic phase with poor compatibility generally results in unsatisfactory interfacial combination. Compatibility always has a great influence on the mechanical performances of polymer matrix composites. Table 7 provides the mechanical properties of pure PP and flame-retardant PP composites. As listed in Table 7, the tensile strength, elongation at break, and notched impact strength of pure PP are 31.0 MPa, 446.3%, and 7.2 kJ m^{-2} , respectively.

Compared with pure PP, the addition of inorganic particles (EG or MEG) results in the deterioration of mechanical properties for flame retardant PP composites. In addition, the mechanical performances listed in Table 7 also strongly indicate the difference in the interfacial compatibility among the systems. The tensile strength, elongation at break and impact strength of PP/MEG30 are 27.5 MPa, 9.1%, and 5.3 kJ m^{-2} , which increases about 7.8%, 18.2%, and 8.2%, respectively as compared to those of PP/EG. This result can be explained as the excellent compatibility between MEG and PP matrix. Obviously, the excellent mechanical performances greatly enhance the commercial value and extend the application fields for the flame retarded PP composites.

To observe the dispersion, interfacial adhesion and reinforcing mechanism of PP/EG30 and PP/MEG30 composites, the surface morphology is analyzed by SEM and the results are shown in Fig. 12. For the PP/EG30, some big holes and obvious gaps in the PP matrix can be seen between the EG and polymer

Table 7 The mechanical properties of PP and its flame-retardant composites

Samples	Tensile strength (MPa)	Elongation at break (%)	Notched impact strength (kJ m^{-2})
PP	31.0	446.3	7.2
PP/MEG10	28.3	59.7	6.0
PP/MEG20	27.8	25.0	5.9
PP/MEG30	27.5	9.1	5.3
PP/EG30	25.5	7.7	4.9
PP/EG/DOPO	24.6	6.3	4.6

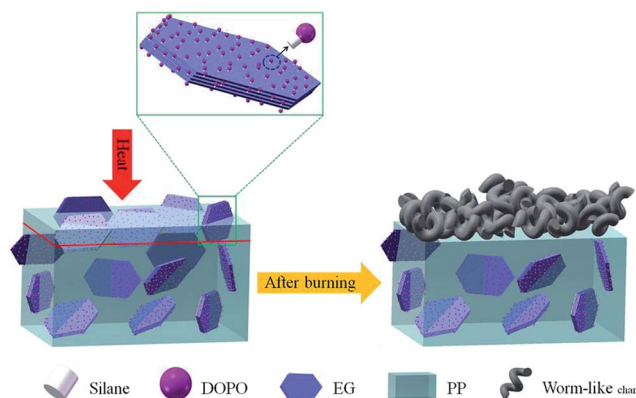


Fig. 11 The flame retardant distribution and flame resistance modes.

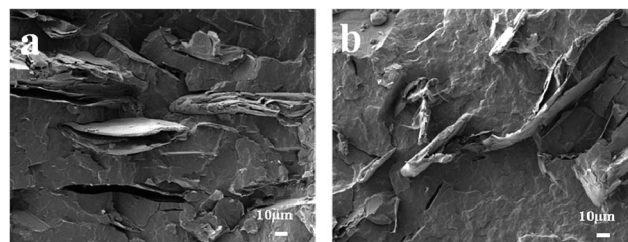


Fig. 12 The interfacial morphology of (a) PP/EG30 and (b) PP/MEG30.



matrix, which indicates poor interfacial combinations. However, for the PP/MEG30 system, after being grafted with DOPO and silane coupling agent, the MEG has well dispersion in the PP matrix and almost no obvious interfaces are observed between MEG and the matrix. As expected, the DOPO–silane coupling agent structure can enhance the dispersion of MEG and the compatibility between the PP matrix and MEG because of the produced DOPO–silane coupling agent side chains with –OH, –O–, –P–O– groups.

4. Conclusions

A novel and efficient flame-retardant additive, MEG, which is constructed by DOPO, silane coupling agent and EG, is synthesized and characterized based on our newly proposed novel organic–inorganic intumescent interfacial flame resistance mode. The thermal behaviors and flame retardances of the flamer retarded PP composites are investigated and found to be strongly influence by the structural features of the additives used. The MEG particles showed outstanding flame-retardant efficiency on the PP/MEG30 composites because of the excellent thermal stability and high residual char yields. This observation is confirmed by the results of the LOI flammability, UL-94, TGA, the dynamic FTIR and XPS test. The wick effect of EG is efficiency eliminated due to the interfacial effect between organic and inorganic additives. Besides, the satisfactory flame retardant grade (V-0) is achieved when approximate 30 wt% of intumescent interfacial flame retardant MEG is incorporated into the PP matrix. Both flame-retardant MEG works in the gas phase and in the condensed phase. In addition, the organic side chains on the surface of EG improve the compatibility between MEG and the PP matrix, which results in the enhancement of the mechanical properties of the PP composites.

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