Synthesis of four-angle star-like CoAl-MMO/BiVO4 p–n heterojunction and its application in photocatalytic desulfurization

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A four-angle star-like Co–Al mixed metal oxide (CoAl-MMO)/BiVO4 heterojunction has been synthesized via a hydrothermal method and following sintering. The CoAl-MMO/BiVO4 is derived from CoAl-LDHs/BiVO4, in which CoAl-LDHs leads to a distribution of amorphous CoAl-MMO. The CoAl-MMO loading on BiVO4 greatly enhances visible light absorption, improves charge separation by band offset charge transfer, and makes flat band potential more negative. The three effects together result in excellent photocatalytic activity. Under visible light irradiation, desulfurization efficiency of thiophene has achieved up to 98.58% on CoAl-MMO/BiVO4 with molar ratio of 0.3 : 5.

Introduction

Organic sulfur from petroleum is a major contributor to environmental pollution. Nowadays, the most widely employed method to remove thiophenic compounds in fuel are hydro-desulfurization (HDS) and adsorption desulfurization. Both of the processes need hydrogen, relatively high-pressure conditions and high energy consumption for deep desulfurization. Therefore, the development of desulfurization process with low energy consumption, mild operating conditions and environmental friendliness has become the research focus.

Photocatalytic desulfurization technology has attracted much attention, because it can provide a cleaner and more environmental friendly way to realize desulfurization. TiO2-based photocatalyst has been widely studied and used in photocatalytic oxidative desulfurization. But since its band gap is 3.2 eV, it can only absorb UV light, which limits its application. Therefore, the development of photocatalysts with visible light response has become a research hotspot. Bismuth vanadate (BiVO4, n-type semiconductor), with a band gap of 2.4 eV, has attracted much attention because it shows activation under visible light irradiation. However, the photocatalytic activity of pure BiVO4 is always low because of the rapid recombination of carriers (electrons and holes). To overcome these drawbacks, strategy of heterogeneous structure construction has been developed to spatially separate photogenerated carriers by band offset. Besides, one widely common strategy is to combine BiVO4 with other noble metal or noble metal oxides, such as Pt, Ag, RuO2. For example, Lin et al. reported a visible-light responsive photocatalyst BiVO4 co-loaded with Pt and RuO2 co-catalysts, which photocalytically oxidized thiophene to SO3 and achieved over 99% of thiophene conversion. Gao et al. synthesized Ag–BiVO4 photocatalysts via hydrothermal method and photocatalytic desulfurization efficiency under visible light irradiation at pH = 7 could be up to 95%. Although the desulfurization efficiency of the above photocatalyst is high, the cost of cocatalysts using noble metal or noble metal oxides is expensive. It is necessary to develop low-cost cocatalysts to combine with BiVO4 to achieve efficient desulfurization.

Herein we report the photocatalytic oxidation of thiophene by Co–Al mixed metal oxide (CoAl-MMO) loaded BiVO4. However, investigation indicates that CoAl-MMO does not act as cocatalyst of BiVO4, but gets three unprecedented effects. The CoAl-MMO loading enhances visible light absorption, improves charge separation by band offset charge transfer, and makes flat band potential more negative. Combination of these effects largely enhances photocatalytic efficiency of thiophene oxidation. The CoAl-MMO derives from a Co-Al-layered double hydroxide (CoAl-LDH) precursor. Since LDH has a uniform distribution of metal cations on the atomic level, sintering leads to CoAl-MMO with a uniform distribution of cobalt and aluminum. It notes that band gap of Al2O3 (ref. 22) is narrow while that of Al2O3 is wide, but neither has been found from our sample by X-ray diffraction (XRD). Besides XRD, samples were characterized by scanning electron microscopy (SEM) and transmission electron microscopy (TEM), ultraviolet-visible diffuse reflectance spectroscopy (UV-vis DRS) and photoluminescence (PL) spectra. The desulfurization activity has been explored under visible-light.

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**Experiment**

**Chemicals**

All the reagents were analytical grade and used without any further purification.

**Preparation of BiVO₄ and CoAl-MMO/BiVO₄ samples**

In a typical synthesis, under stirring conditions, 0.17 mmol P123 (analysis) and 5 mmol Bi(NO₃)₃·5H₂O were dissolved in a mixed solution of 5 mL HNO₃ solution (3 mol L⁻¹) and 20 mL ethylene glycol solution to obtain solution A. 5 mmol NH₄VO₃ was dissolved in 20 mL deionized water with 70 °C to obtain solution B. Solution A and B at room temperature were stirred for 30 min each, then solution A was added to B dropwise. Then pH value was adjusted with NH₄H₂O (14 wt%) and was stirred for 60 min to obtain BiVO₄ precursor. A certain amount of Co(NO₃)₂·6H₂O and Al(NO₃)₃·9H₂O was dissolved in deionized water (Co²⁺/Al³⁺ molar ratio = 3 : 1), then a solution containing 1.0 mol L⁻¹ NaOH and 0.30 mol L⁻¹ Na₂CO₃ was added into it. Stirring for 60 min, Co–Al layered double hydroxide (CoAl-LDHs) precursor was obtained. CoAl-LDHs precursor was added to BiVO₄ precursor, pH value was adjusted to 10 with NaOH (1.0 mol L⁻¹) solution and stirred for 60 another min. The mixed solution was transferred to a 100 mL Teflon lined stainless steel autoclave and hydrothermal treatment was conducted at 180 °C for 12 h. After that, the resulting CoAl-LDHs/BiVO₄ catalysts were washed with deionized water and absolute ethanol, dried at 80 °C for 3 h. Molar ratio of CoAl-LDHs/BiVO₄ for 0 : 5 (BiVO₄), 0.1 : 5, 0.3 : 5, 0.5 : 5 were controlled. Final products were obtained by sintering in a muffle furnace at a ramp of 1 °C min⁻¹ from 30 °C to 400 °C and maintaining for 4 h. The final products were noted as CoAl-MMO/BiVO₄.

**Characterizations**

X-ray powder diffraction (XRD) measurements were performed on a Bruker D8 Advance diffractometer operated at 40 kV and 40 mA at the scanning range from 2 to 70 degree with Cu-Kα radiation (λ = 0.15406 nm). The particle morphologies of the products were observed by scanning electron microscopy (SEM) (Hitachi SU 8010) and by transmission electron microscopy (TEM) (JEM 2200FS). A Cary 5000 UV-vis spectrometer (Agilent Technologies) was used to obtain the reflectance spectra of the samples over a range of 400–800 nm. Electrochemical analysis was carried out with a standard three-electrode system with a Pt plate as the counter electrode, Hg/HgCl₂ (saturated with KCl) as a reference electrode, and ITO glass coated with the sample was used as the working electrode. A 0.5 M Na₂SO₄ solution was used as electrolyte. A 500 W Xe arc lamp (CHF-XM35-500W) was utilized as light source. Transient short-circuit photocurrent measurements and Mott–Schottky experiments with amplitude of 50 mV and a frequency of 1000 Hz were taken on a CHI660E workstation.

**Photocatalytic oxidative desulfurization tests**

The photocatalytic oxidative desulfurization tests were carried out in a quartz tube reactor with a water condenser at atmospheric pressure and room temperature, using air and H₂O₂ (30%) as oxidants. The model oil, with sulfur content of 200 ppm is prepared by dissolving thiophene into n-octane. 50 mg photocatalyst and 50 mL model oil were added to the reactor. The suspension was stirred in dark for 30 min to obtain an adsorption–desorption equilibrium between the CoAl/BiVO₄ photocatalyst and model oil. Then, 0.128 mL H₂O₂ was added and the suspension was irradiated by a 500 W Xe arc lamp (CHF-XM35-500W), the airflow velocity was of 5 mL min⁻¹. Desulfurized oil was collected periodically and extracted with acetonitrile. Sulfur content was determined by a TSN-5000 series fluorescence nitrogen/sulfur analyzer (Jiangfen Electroanalytical Instrument Co., Ltd., China).

**Results and discussion**

Fig. 1 shows XRD patterns of BiVO₄ and CoAl-MMO/BiVO₄ samples. The diffraction peaks at 2θ of 18.6°, 28.9°, 30.5°, 34.4°, 35.3°, 39.4°, 42.3°, 46.1°, 46.6°, 47.3°, 53.3°, 58.3°, and 59.9° can be observed, which index to monoclinic BiVO₄ (JCPDS no. 14-0688) and correspond to the crystalline planes of (101), (013), (112), (200), (020), (211), (105), (123), (204), (024), (301), (303) and (224). It is clearly observed from the graph that as CoAl-MMO loading ratio increases, the intensity of CoAl-MMO/BiVO₄ samples becomes weaker; and no other peaks can be found in XRD patterns of CoAl-MMO/BiVO₄ catalysts, which may due to the amorphous characteristics of CoAl-MMO.

The morphology of samples has been characterized by SEM and TEM (Fig. 2). The morphology of BiVO₄ shows four angles with star-like and the particle size is uniform. After loaded with CoAl-MMO, the morphology becomes more regular. Low molar loading ratio makes the surface of CoAl-MMO/BiVO₄ particles look smooth, as molar loading ratio reaches 0.3 : 5, the surface become rough. And under magnified TEM many small particles on the surface can be observed (Fig. 2(d)). As red dashed line and red arrows point, CoAl-MMO nanoparticles are amorphous and dispersed on BiVO₄ surface. As shown in inset graph of Fig. 2(b) and (c), the thickness of CoAl-MMO/BiVO₄ with molar ratio 0.1 : 5 and 0.3 : 5 are about 750 nm and 1 μm, respectively, meaning that with the loading ratio increasing, more CoAl-
MMO formed on BiVO₄ surface and the particles become thicker. The particle size rises also, which may be due to CoAl-LDH playing a role of template.

Fig. 3(A) shows UV-vis diffuse reflectance spectra of BiVO₄ and CoAl-MMO/BiVO₄. BiVO₄ shows an absorption region between 200–530 nm, which covers both UV and partial visible light region. Interestingly, after loaded with CoAl-MMO the absorption region extends to 700 nm, which is throughout the UV to visible light region. For crystalline semiconductor, the optical absorption near the band edge follows the formula:

\[
\alpha hν = A(hν - E_g)^{n/2}
\]

\(\alpha, h, E_g\) and \(A\) refer to coefficient, light frequency, band gap and a constant (\(A = 1\)), respectively. \(n\) depends on the characteristics of the transition in a semiconductor, for direct transition, \(n = 1\), for indirect transition \(n = 4\). For BiVO₄, \(n = 1\). The plots of \((\alpha hν)^2\) versus photon energy \((hν)\) is shown in the inset graph of Fig. 3(A), and the band gap of BiVO₄ and CoAl-MMO/BiVO₄ can be obtained by extrapolating the curve to \(\alpha = 0\). As a result, the band gaps CoAl-MMO/BiVO₄ with loading ratio of 0 : 5, 0.1 : 5, 0.3 : 5 and 0.5 : 5 obtained are 2.4 eV, 2.35 eV, 2.06 eV and 2.08 eV, respectively. Compared to BiVO₄, CoAl-MMO/BiVO₄ samples exhibit stronger absorption in the visible light range and narrower band gap. BiVO₄ is a n-type semiconductor while CoAl₂O₄ p-type (although not detected by XRD but may exist as amorphous), a p–n heterojunction should have been formed at the interface of CoAl₂O₄ and BiVO₄ particles, which extends the absorption range and makes the band gap narrower. For sample with molar loading ratio of 0.3 : 5, these small particles on the surface leads to expanding of contact area and stable heterojunction structure. However, for sample 0.5 : 5, the absorption capacity become lower and the band gap slightly increases. It may be due to that some growing amount of CoAl-MMO particles are not well-loaded on the surface of BiVO₄, but just scatter around, which makes the heterojunction structure unstable and restrain the interface interaction between CoAl-MMO and BiVO₄.

Photoluminescence (PL) spectra of semiconductor materials derive from recombination of photo-induced charge carriers. Higher PL intensity means higher recombination rate of
carriers (electrons and holes) and photocatalytic activity becomes correspondingly lower.\textsuperscript{29-31} Fig. 3(B) shows PL spectra of BiVO\(_4\) and CoAl-MMO/BiVO\(_4\) samples with excitation wavelength of 320 nm. The PL emission wavelength of all samples is centered at 423 nm, different from 590 nm reported by R. Tang \textit{et al.} on BiVO\(_4\) nanosheets and its complex with graphene.\textsuperscript{8} BiVO\(_4\) alone has the highest PL intensity, and the PL intensity decreases by different extent with CoAl-MMO loading amount, the least one obtained on sample ratio of 0.3 : 5. It's caused by the electron transfer from CoAl\(_2\)O\(_4\) conduction band (CB) to BiVO\(_4\) CB, and meanwhile holes transfer from BiVO\(_4\) valence band (VB) to CoAl\(_2\)O\(_4\) VB under the potential of band energy difference, namely, band offset.\textsuperscript{29,32} The above migration of photogenerated carriers makes electrons and holes spatial separation, and tremendous reduces their recombination probability. Accordingly, photogenerated carriers have longer life time to take part in photocatalytic reactions, and the photocatalytic activity of CoAl-MMO/BiVO\(_4\) would be improved upon BiVO\(_4\). As a photocatalytic activity test, transient photocurrent responses of the composite electrodes with on-off cycles of intermittent visible-light irradiation are studied, as shown in Fig. 3(C). The anodic photocurrent in Na\(_2\)SO\(_4\) solution represents photocatalytic water oxidation efficiency of generated holes on surface of sample electrode. Among the catalysts, BiVO\(_4\) shows the lowest photocurrent, meaning that the least amount of holes participates in water oxidation, caused by the worst separation efficiency between carriers.\textsuperscript{33} With the molar ratio being 0.3 : 5, CoAl-MMO/BiVO\(_4\) generates the highest photocurrent density.

To investigate the electronic effect, Mott–Schottky (MS) measurement\textsuperscript{34,35} has been performed on BiVO\(_4\) and CoAl-MMO/BiVO\(_4\) series, as shown in Fig. 4(A). For CoAl-MMO/BiVO\(_4\) with molar loading ratio of 0 : 5, 0.1 : 5, 0.3 : 5 and 0.5 : 5, the \(V_{\text{fb}}\) are \(-0.54\) V, \(-0.61\) V, \(-0.72\) V and \(-0.64\) V vs. SCE (equivalent to \(-0.06\) V, \(-0.13\) V, \(-0.24\) V and \(-0.16\) V vs. NHE at PH = 0), respectively. It is well known that CB potential \((E_{\text{CB}})\) of a n-type semiconductor is \(0.2\) V more negative than \(V_{\text{fb}}\) and is dependent on carrier concentration and electron effective mass. It can be seen that \(E_{\text{CB}}\) of the samples are more negative than the standard redox potential of \(O_2/\text{O}_2^-\) (0.28 V vs. NHE). It indicates the photogenerated electrons could react readily with adsorbed \(O_2\) to produce ‘\(O_2^-\)’ (ref. 36), and ‘\(O_2^-\)’ is active free radical in desulfurization reactions. Photocatalytic desulfurization activity has been investigated via thiophene decomposition, the result shown in Fig. 4(B). The most negative \(V_{\text{fb}}\) of \(-0.24\) V corresponds to the highest desulfurization efficiency, is obtained with CoAl-MMO/BiVO\(_4\) ratio of 0.3 : 5.

In summary, there are three significant merits of CoAl-MMO loading on BiVO\(_4\) via sintering mixture of CoAl-LDH and BiVO\(_4\). Firstly, it exhibits red shift of the whole band edge upon CoAl-MMO loading. That widens and enhances visible light absorption and utilization. Secondly, nanosized n–p heterojunction has been formed between CoAl-MMO particles and BiVO\(_4\). It endows carriers’ spatial separation, which has been proved by PL intensity decrease. Thirdly, more negative \(V_{\text{fb}}\) has been obtained (dashed lines in Scheme 1), which benefits formation of ‘\(O_2^-\)’. Interestingly, the above three effects changes along with CoAl-MMO loading amount, and 0.3 : 5 is the best in each aspect. Thus, based on the three merits, highly efficient thiophene desulfurization has been realized on CoAl-MMO/BiVO\(_4\). Schematic description of the mechanism for thiophene desulfurization on CoAl-MMO/BiVO\(_4\) has been shown in Scheme 1. Along with CoAl-MMO loading amount modifying, band gap
energy of CoAl-MMO/BiVO₄ becomes smaller and more visible light has been absorbed to excite carriers. 3 colored lines above $E_{CB}$ in Scheme 1 indicate that $E_{CB}$ moves upward, corresponding to band gap shortening, caused by changing loading amount of CoAl-MMO. Due to the band offset of heterojunction, photo-generated holes and electrons transfer to CoAl-MMO and BiVO₄ respectively (black curved arrows in Scheme 1). Electron on BiVO₄ reacts with absorbed O₂ to ’O₂’. On the other hand, hole on CoAl-MMO reacts with OH– forming ‘OH, meanwhile some hole is readily captured by thiophene to forming radical cation $\text{C}_4\text{H}_4\text{S}^+$ (blue curved arrows in Scheme 1). Interaction of the active oxygen species (’O₂– and ’OH) with the sulfur radical cations $\text{C}_4\text{H}_4\text{S}^+$) induces a series of oxidation reactions, at last, thiophene has been almost completely oxidized to SO₂, CO₂, and H₂O. As a result of the profound effects of CoAl-MMO loading and optimization, the desulfurization efficiency reaches up to 98.58% on 0.3 : 5 CoAl-MMO/BiVO₄.

Conclusions

Four-angle star-like CoAl-MMO/BiVO₄ photocatalysts has been synthesized by a hydrothermal method and later sintering. CoAl-MMO loading on BiVO₄ dispersed as amorphous particles on its surface, brings advantages to photocatalytic performance from three aspects, such as visible light enhancement, heterojunctions for carrier spatial separation, and making $V_h$ more negative as well. Therefore, photocatalytic desulfurization efficiency by CoAl-MMO/BiVO₄ has been improved largely, compared to BiVO₄ under visible light irradiation, from under 75% to over 97%. After optimizing the loading amount, 98.58% conversion of thiophene has been achieved with molar loading ratio of 0.3 : 5. This work has demonstrated that CoAl-MMO is not only cost-effective, but also plays a significant role in enhancing the photocatalytic activity of BiVO₄. It points that low-cost and effective mixed metal oxide loading on other photocatalysts may be a promising choice in photocatalytic desulfurization.

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Notes and references


