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Co-delivery of hydrophilic and hydrophobic anticancer drugs using biocompatible pH-sensitive lipid-based nano-carriers for multidrug-resistant cancers

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For decades, multi-drug resistance (MDR) to chemotherapeutic drugs has been a serious challenge for researchers and has limited the use of anticancer drugs in malignancy treatment. Combination therapy has been considered as one of the most promising methods to address this problem. In the current study, we optimized niosome nanoparticles containing chemotherapeutic agent doxorubicin and chemosensitizer curcumin in term of surfactant content. Then, a new biocompatible structure (LipoNiosome, combination of niosome and liposome) containing Tween 60: cholesterol: DPPC (at 55 : 30 : 15 : 3) with 3% DSPE-mPEG was designed and developed to serve as a model for selective co-delivery of hydrophilic and hydrophobic drugs to cancerous cells. The proposed formulation provided potential benefits, including pH-sensitive sustained release, smooth globular surface morphology, high entrapment efficiency (~80% for both therapeutic agents) and small diameter (42 nm). Exposure of cancer cells to LipoNiosome-doxorubicin–curcumin has shown an excellent performance of specific cellular internalization and synergistic toxic effect (>40%; as compared to free drugs and >23% when compared to single doxorubicin delivery) against Saos-2, MG-63 and KG-1 cell lines. A new cationic formulation (zeta potential: +35.26 mV; diameter: 52.2 nm) was also designed for co-delivery of above-mentioned drugs and gene as well. Finally, we suggested a kinetic model (Korsmeyer–Peppas with $R^2 = 93\%$ near cancer cells) for *in vitro* drug release of the co-delivery system. The presently formulated nano-based systems would provide researchers with a more obvious understanding of new LipoNiosome formulation as a successful lipid-based nano-carriers for co-delivery of doxorubicin, curcumin and other anticancer agents.

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1. Introduction

Cancer is basically uncontrolled cell proliferation that aggressively invades other parts of the body. According to the International Agency for Research on Cancer (IARC), 14.1 million new cases of cancer were estimated to occur in 2012, with almost 8.2 million mortality cases. Based on IARC estimates, cancer is the 2nd most common cause of death in both economically developing and developed countries. About 13% of global cancer cases are estimated to have occurred in southwestern of Asia.¹ The main cancer treatment modality, chemotherapy, has limitations that including various side effects. To overcome the previously-mentioned impediments of current cancer treatments, nanotechnology can pave the way.

Multidrug resistance (MDR) in cancers, defined as resistance of tumor to the cytotoxic effects of several drugs, has been considered as a major obstacle to the clinical cancer treatment. Resistance against anticancer drug has been attributed to increased drug efflux, decreased drug uptake, activation of DNA repair process along with the activation of detoxifying systems.² In order to treat drug-resistant tumors, combinations of multiple anticancer agents, including chemosensitizers such as, small interfering RNAs (siRNAs) and herbal medicine like curcumin (CUR) and classic antineoplastic drugs have been applied. For example, combination of antileukemic drug and chemosensitizers has been demonstrated to effectively modulate multiple signaling pathways by inactivation of MDR-related mRNAs.^{3–5} In our previous works, formulation of siRNA and doxorubicin (DOX) delivery system were optimized and a new YSA-targeted liposomal DOX-siRNA was developed to exert a synergistic effect to overcome MDR in osteosarcoma.^{6,7} Another challenge for researchers is that single-drug therapy reinforces alternative molecular pathways in cancer cells due to drug-resistance mutations. Co-delivery of multiple anticancer

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agents *via* a nanocarrier has been considered as an approach to overcome MDR.⁸

Anthracycline (antibiotics) are the most commonly used agents, which block the topoisomerase function, thereby inhibiting cell proliferation. Doxorubicin (Adriamycin®), an anthracycline antibiotic, is a potent FDA-approved chemotherapy drug.^{9,10} For many years, it has been acknowledged that DOX could treat several cancer types, such as breast cancer, bone marrow cancer, and osteosarcoma. However, DOX's side effects and unpredictable toxicity to normal cells adversely affects the immune system, which makes patient more susceptible to infections. It also causes cardiotoxicity.¹¹ To address this issue, numerous nanoparticle drug delivery vehicles have emerged during the past two decades.^{2,4}

Compared to free DOX, nano-shielded DOX has less toxicity to normal cells however due to sustained release properties and higher cytotoxicity against cancer cells in short term and overcomes MDR in long term.¹²

Multidrug resistance raised by up-regulation of BCL-2 protein and inactivation DOX by accumulation into acidic cytoplasmic vesicles resulting in induce MDR gene expression, reduced sensitivity to DOX and decreases therapeutic efficiency of DOX for osteosarcoma, bone marrow cancers and *etc.*^{13–17}

Curcumin (CUR), a chemosensitizers, was used to fight cancer governed by blocking NFκB signaling pathway, down-regulation overexpression of P-glycoprotein and Bcl-2.¹⁸ Therefore, CUR is currently being co-administered with DOX and other cytotoxic drugs to reduce the efflux of drugs through upregulated efflux transporters in resistant hepatocellular carcinoma in mice.¹⁹

CUR has shown a number of benefits for patients with colorectal and pancreatic cancer. In addition, neither human nor animal studies have found any side effect or toxicity of CUR.^{20,21} Despite many beneficial pharmacological activities, such as anti-oxidant, antiviral and anti-inflammatory properties and the safety of CUR, its retention time in the body is restricted due to low serum bioavailability, hepatic elimination and poor absorption resulted by hydrophobicity.²² Tsai *et al.* (2011) have reported that CUR may prevent cancer in the colon, skin and stomach after oral administration while no dose-limiting toxicity was observed. However, they showed that it has avid metabolism, resulting in mean plasma concentration of 31.5 mg mL⁻¹ after 8 g ingestion per day.²³ Great efforts have been made by researchers to enhance CUR solubility and protection of CUR against inactivation by hydrolysis.²⁴ Nanobiotechnology has been introduced as a promising strategy in the development of drug delivery systems for hydrophobic drugs such as CUR. Xi Yang *et al.* designed and characterized curcumin-encapsulated polymeric micelles to improve the stability of CUR. They also studied the anti-cancer efficiency of the prepared nano-stabilized aqueous curcumin for colon cancer treatment.²⁵

During the past decade, niosomes and liposomes have been extensively applied due to their stability and biocompatibility, respectively. The performance of single encapsulation of DOX or CUR into the liposome or niosome has already been well investigated.^{17–19,26–28} However, little is known about stability

and biocompatibility of their combined form in liposome and niosome.

The purpose of this research was to investigate the hypothesis whether or not co-delivery of DOX–CUR could show synergistic effects in different cancer cell lines and could overcome against DOX resistance. Moreover, it was hypothesized that this co-delivery could diminish drugs' side effects in normal cells. Various co-formulations were formulated to achieve high entrapment efficiency, small size, and controlled release behavior. For better safety, the doses of both drugs were decreased and the cationic LipoNio formulations were developed. Mathematical model was developed to estimate drug release behavior. Furthermore, co-delivery of DOX and CUR was carried out to estimate cytotoxicity against MG-63, KG-1 and SaOs-2 cell lines.

2. Experimental

2.1. Materials

Doxorubicin HCl (DOX) and curcumin (purity ≥ 95%) were obtained from Ebewe Pharma (Austria) and Sigma-Aldrich (St Louis, MO, USA), respectively. Surfactants, Tween 60 and Span 60, were purchased from DaeJung Chemicals & Metals (South Korea). Tween 80 and Tween 20 were obtained from Merck (Darmstadt, Germany). The distearoyl phosphoethanolamine, polyethylene glycol (Lipoid PE 18 : 0/18 : 0 – PEG2000, DSPE-mPEG 2000), DPPC (1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine phospholipid) and SPC80 (soybean phospholipids with 75% phosphatidylcholine) were obtained from Lipoid GmbH (Ludwigshafen, Germany). Cholesterol and DOTAP (1,2-dioleoyl-3-trimethylammonium-propane) were supplied by Sigma-Aldrich (St. Louis, MO, USA) and Avanti Polar Lipids (AL, USA), respectively. Ammonium sulfate salt and NaHCO₃ were obtained from Merck (Darmstadt, Germany). PBS tablets, dialysis bag (MW = 12 kDa), DMSO (dimethyl sulfoxide), HEPES buffer, MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) and paraformaldehyde solution were obtained from Sigma-Aldrich (St. Louis, MO, USA). DAPI (4',6-diamidino-2-phenylindole) was supplied from Thermo Fisher Scientific (Massachusetts, USA). All other chemicals, solvents and salts were of the analytical grade and used without further purification unless specified.

2.2. Cell lines and preparation of biological samples

Human primary (short-term culture) osteoblasts, osteosarcoma cell line MG-63 and SaOs-2, bone marrow acute myeloblastic leukemia (AML) cell line KG-1 and human bone marrow fibroblast-like HBMF-SPH cell line were supplied from the Pasteur Institute of Iran (Tehran, Iran). Human primary osteoblasts SaOs-2 and MG-63 were cultured in DMEM medium (Gibco, Grand Island, USA). HBMF-SPH and KG-1 cells were cultured in RPMI-1640 medium (Gibco, Grand Island, USA). All cells were supplemented with 10% FBS (fetal bovine serum) (Gibco, Grand Island, USA) with penicillin-streptomycin (Gibco, Grand Island, USA) under standard condition (37 °C and 5% CO₂ in a humidified incubator).



2.3. Preparation of drug-loaded nanoparticles

DOX- and CUR-loaded nanoparticles were synthesized and screened for particle size, controlled release and high entrapment efficiency parameters. In order to optimize the results and determine the optimal conditions following experiments were performed:

- Evaluation of the influence of different types of surfactants with various hydrophobic tail and hydrophilic heads and in different molar ratios (0, 15% and 35%) was carried out on dual-drug delivery system.
- The preparation of LipoNiosome was investigated by addition of 15% synthetic and natural phospholipids (SPC80, DPPC).
- Suggested hydration methods for preparing drug-carrier vesicles, including thin film and pH gradient were evaluated.

- After having optimized synthetic conditions, cationic formulations for dual drug/gene delivery were prepared.

Nanoparticles, containing hydrophilic and hydrophobic drugs, were prepared by thin film and pH-gradient technique as fully described elsewhere.¹² Fig. 1 shows the schematic view of research steps. Briefly, the surfactants in these formulations were 70% (w/w) in all, and as can be seen in Table 1, various kinds of surfactants with different concentration ranges were dissolved in chloroform as an organic phase in the presence of 3% DSPE-mPEG and 30% cholesterol with or without phospholipids (Table 2). CUR stock solution was dissolved in methanol (1 mg mL⁻¹) and added to the mixture of surfactant and lipids. The thin films were dried to remove chloroform and methanol, using rotary evaporator (Heidolph, Germany) at 45 °C under reduced pressure. Hydration of dry lipid film was carried

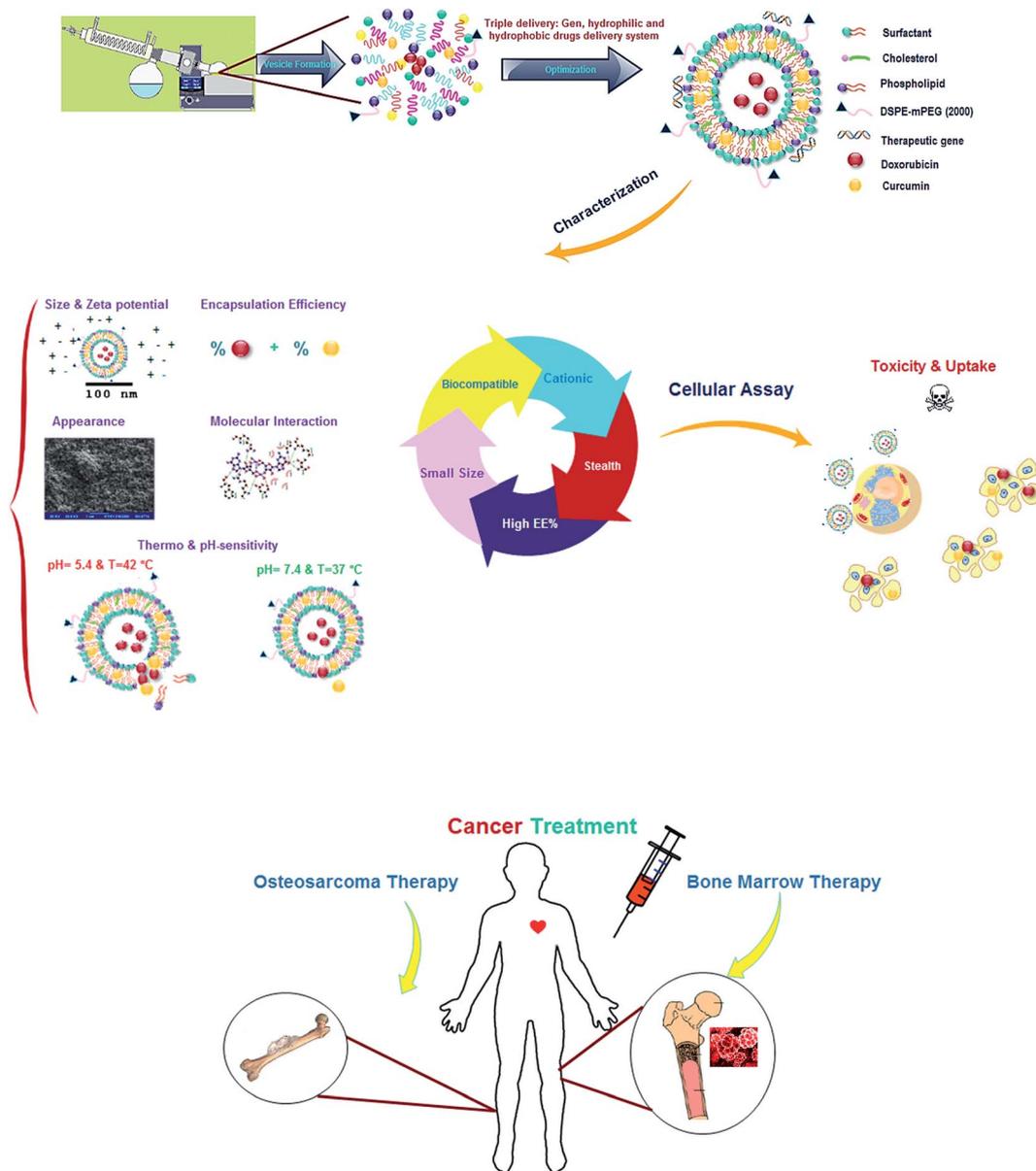


Fig. 1 The schematic representation experimental design of the study.



Table 1 Effect of surfactant type with various molar ratios, on entrapment efficiency (EE%), size, long-term and short-term release (R_t^a)

Code	Type of surfactant	Mole of surfactants (%)	EE%	Size (nm)	% R (8 h)	% R (24 h)	% R (48 h)	% R (96 h)
F1	Span 60	70	EE% DOX = 25.01 EE% CUR = 50.21	185.9	% R DOX = 25.42 % R CUR = 20.16	% R DOX = 33.94 % R CUR = 26.88	% R DOX = 39.28 % R CUR = 29.75	% R DOX = 45.6 % R CUR = 34.64
F2	Tween 60	70	EE% DOX = 85.12 EE% CUR = 95.11	85.5	% R DOX = 38.42 % R CUR = 27.16	% R DOX = 46.38 % R CUR = 31.96	% R DOX = 51.67 % R CUR = 37.14	% R DOX = 61.76 % R CUR = 43.78
F3	Tween 80	70	EE% DOX = 62.02 EE% CUR = 87.70	60.9	% R DOX = 50.63 % R CUR = 38.1	% R DOX = 61.66 % R CUR = 43.82	% R DOX = 66.9 % R CUR = 46.39	% R DOX = 77.69 % R CUR = 54.67
F4	Span 60 : Tween 20	55 : 15	EE% DOX = 66.51 EE% CUR = 62.13	54.2	% R DOX = 35.28 % R CUR = 30.96	% R DOX = 45.06 % R CUR = 37.98	% R DOX = 51.13 % R CUR = 41.81	% R DOX = 61.05 % R CUR = 52.06
F5	Span 60 : Tween 60	55 : 15	EE% DOX = 42.02 EE% CUR = 68.05	127.4	% R DOX = 42.48 % R CUR = 25.05	% R DOX = 47.97 % R CUR = 30.06	% R DOX = 53.79 % R CUR = 34.55	% R DOX = 61.26 % R CUR = 37.07
F6	Span 60 : Tween 80	55 : 15	EE% DOX = 48.03 EE% CUR = 71.11	95.9	% R DOX = 39.02 % R CUR = 34.89	% R DOX = 49.50 % R CUR = 40.64	% R DOX = 53.43 % R CUR = 44.99	% R DOX = 59.46 % R CUR = 55.24
F7	Span 60 : Tween 20	35 : 35	EE% DOX = 76.21 EE% CUR = 91.03	48	% R DOX = 41.39 % R CUR = 30.38	% R DOX = 50.81 % R CUR = 37.73	% R DOX = 62.84 % R CUR = 41.83	% R DOX = 63.59 % R CUR = 53.59
F8	Span 60 : Tween 60	35 : 35	EE% DOX = 81.32 EE% CUR = 81.51	97.1	% R DOX = 41.07 % R CUR = 22.95	% R DOX = 48.53 % R CUR = 31.28	% R DOX = 52.81 % R CUR = 34.56	% R DOX = 62.62 % R CUR = 43.49
F9	Span 60 : Tween 80	35 : 35	EE% DOX = 70.43 EE% CUR = 78.91	64.5	% R DOX = 41.65 % R CUR = 32.25	% R DOX = 51.70 % R CUR = 40.91	% R DOX = 55.45 % R CUR = 45.09	% R DOX = 62.84 % R CUR = 61.4

^a EE: entrapment efficiency.

out by adding 1300 μL diluted DOX solution at 63 °C for 60 min. After completion of hydration, ultra-sonication was applied for 45 min (15 seconds on and 10 seconds off, amplitude of 70 at 100 watts) to minimize particle aggregation using ultrasonic homogenizer (model UP200St, Hielscher Ultrasonics GmbH, Germany). For pH-gradient method, the dried films of CUR, surfactants and lipids were hydrated with 1300 μL of ammonium sulfate (pH = 4) at 63 °C for 47 min. Then, nanoparticles were sonicated over an ice bath to produce small vesicles. Afterwards, ammonium sulfate was removed by replacing the medium with fresh PBS (pH = 7.4) using a dialysis cellulose membrane for 2 h at 25 °C. Finally, the diluted DOX with sterile water for injection, was added at 54 °C. The dose of both drugs was 0.5 mg mL⁻¹ for all of the formulations and the L/D ratios were kept at 20 and 10 for thin film and pH-gradient methods, respectively.

2.4. Characterization of DOX/CUR-NPs

2.4.1. Physical characterization of DOX/CUR-NPs. Zeta potential was determined by using ZetaSizer Nano ZS (Malvern, Worcestershire, UK), to measure the surface charge of the synthesized nanocarriers. Dynamic light scattering (DLS) was also applied to determine the hydrodynamic size diameter, polydispersity index (PDI) and distribution size of the obtained nanocarrier. The structure and surface morphology of nano-lipoNio were analyzed using scanning electron microscope (SEM) (model KYKY-EM3200-30 kV, KYKY Technology Development Ltd., Beijing, China) operated at accelerating voltage of 20 kV. To prepare the sample used in SEM an extremely little amount of the lipoNio suspension dispersed in water was placed on the mesh copper grid 400. Then, the grid was put in an evacuated desiccator to evaporate the solvent followed by sputter-coating the sample with gold before being introduced into the microscope.

2.4.2. Entrapment efficiency and drug release study. In order to evaluate the entrapment efficiency and the drug release profile over time, spectroscopic measurements were carried out. The amounts of DOX and CUR loaded into nanoparticle during preparation were estimated by UV-spectrophotometry method at 481 and 427 nm (λ_{max}).²⁶ To estimate encapsulation efficiency and drug release, the separation of nanoparticles from the un-encapsulated drug by the dialysis membrane bag (MW = 12–14 kDa) was required beforehand. Finally, nanoparticles were mixed with isopropyl alcohol in the volume ratios of 1 : 20, 1 : 100 and 1 : 75 (v/v) to lyse the membranes and rapid shed of entrapped drugs.

Dialysis method is one of the methods to determine the concentration of DOX/CUR in PBS solution and subsequently, the drug release behavior of nanoparticles. At the first step, the sample of nanoparticles DOX-CUR was transferred into a dialysis tube and the release of both drugs was monitored in 7 mL of PBS solution (at 37 and 42 °C, pH = 7.4, 6.5 and 5.4) in shaking water bath at 75 rpm for 96 h. Then, in order to determine drug release rate by applying UV/visible spectroscopy (model T80+, PG Instruments, United Kingdom), 1200 μL of the sample was collected at specific time intervals and then substituted with an equal volume of fresh PBS to maintain the

Table 2 Effect of various types of phospholipids on entrapment efficiency (EE%), size, long-term and short-term release

Code	Type of phospholipid	EE%	Size (nm)	% Release (8 h)	Zeta potential (mV)	PDI
F10	SPC80	EE% DOX = 93.02 EE% CUR = 98.01	65.1	% R DOX = 71.07 % R CUR = 73.38	-34	0.283
F11	DPPC	EE% DOX = 88.23 EE% CUR = 77.11	42	% R DOX = 45.29 % R CUR = 30.63	-37	0.324
F12	DOTAP	EE% DOX = 85.02 EE% CUR = 94.03	52.2	—	+35.26	0.302

condition and ensure a constant initial volume. Finally, the amount of drug released was measured and the drug release (R) behavior was described by a semi-empirical mathematical model named Korsmeyer–Peppas.²⁹

2.4.3. Functional group characterization. Fourier transform infrared (FTIR) spectroscopy (Model 8300, Shimadzu Corporation, Tokyo, Japan) was applied to analyze molecular interaction between drugs and carrier for pure DOX, pure CUR, blank lipoNio, LipoNiosomal-DOX, LipoNiosomal-CUR and LipoNiosomal-DOX-CUR. Samples were lyophilized and prepared as dry powder and mixed separately with potassium bromide (KBr), and the pellets were formed by placing samples in a hydraulic press.

2.5. Cytotoxicity study

To study anti-proliferative activity of free drugs (DOX & CUR), blank niosome, blank LipoNio (biocompatible niosome), LipoNio-DOX (5, 12.5 and 25 $\mu\text{g mL}^{-1}$), LipoNio-CUR (12.5, 25 and 50 $\mu\text{g mL}^{-1}$), as well as co-administration of LipoNio-DOX-LipoNio-CUR (5, 12.5 and 25 $\mu\text{g mL}^{-1}$), and co-delivery of LipoNio-DOX-CUR (5, 12.5 and 25 $\mu\text{g mL}^{-1}$), were incubated 72 h with 10^4 cells in a 96-well plate prior to assessment with the colorimetric MTT assay. After 72 h after cell seeding in medium, the control wells and samples were removed and washed with PBS and then incubated with 20 μL of 5 mg mL^{-1} MTT in PBS for 3 h. The resultant formazan crystals were dissolved in DMSO. The absorbance of resulting samples was measured using EPOCH microplate spectrophotometer (synergy HTX, Bio-Tek, USA) at 570 nm.

2.6. Nano-LipoNiosomal DOX-CUR localization assay

To determine distribution of the DOX and CUR in the nucleus, the DOX and CUR fluorescence intensity was detected. In short, MG-63, KG-1 and SaOs-2 cells (5×10^5 per well) were seeded in 30 mm dishes, then treated with LipoNio-DOX-CUR, free CUR and free DOX at the concentration of 10 $\mu\text{g mL}^{-1}$. After 3 and 8 h of incubation, cells were washed three times with PBS (pH 7.4). All floating cells were collected using centrifuge (at 1200 rpm for 10 minutes) before washing with PBS. The nucleus of cells was counterstained with DAPI (0.125 $\mu\text{g mL}^{-1}$) for 15 min. Images were obtained by fluorescence microscopy (Olympus, Japan).³⁰

2.7. Statistical analysis

Statistical data analyses were performed *via* Student *t*-test to compare the differences between groups and $P < 5\%$ was

considered significant. Experiments were done in triplicate and average values were reported. The quality of fitting was evaluated by R^2 (ref. 31) and non-linear regression analysis was performed using MATLAB software (version 7.8). The relative standard deviation was calculated to show the precision.

To determine the amount of drug entrapment and drug release, standard solutions were prepared over a concentration range, and analyzed by UV spectroscopy at the characteristic value of λ_{max} with support by the standard curve of drugs measured in a similar condition.

3. Results and discussion

3.1. The effect of surfactant on DOX/CUR niosome formula

In order to determine the optimal formulation for achieving high entrapment efficiency, controlled-release (at 37 °C and pH = 7) and small vesicle size, various niosomal CUR/DOX formulations were evaluated (Table 1). Comparing surfactants used in formulation (F1 vs. F2), including Span 60 and Tween 60, indicated that the hydrophobic chain of Tween 60 is longer than Span 60 resulting in higher CUR entrapment. Moreover, the results indicated that hydrophilic end of Tween 60 played a more predominant role in higher DOX entrapment over that of Span 60. In addition, acyl chain of Tween 20 (F4 vs. F1) had little effect on the enhancement of CUR release due to weaker interaction between CUR and the hydrophobic acyl chain. Although the hydrophilic and hydrophobic parts of Tween 60 and Tween 80 were similar to each other, but the presence of a double bond in the alkyl chain of Tween 80, formation of momentary polar functional groups and the presence of electron cloud have made it more unsaturated and mobile. This phenomenon increased drug leakage during preparation step and subsequently caused a reduction in final entrapment efficiency (F2 vs. F3 and F8 vs. F9). In addition, it increased drug release in Tween 80 formulation. The higher phase transition temperature (T_m) of Span 60 (F1 vs. F2 and F3) resulted in sustained release pattern and less drug entrapment in hydration step of preparation, as well as slow drug release, and subsequently brought the more stable system.

Also, with the addition of 15% Tween (20, 60 and 80) (F4, F5 and F6) to Span 60 formulations, DOX and CUR entrapment into nanoparticles were constantly increased due to increasing aqueous and non-aqueous space of vesicles provided by the long chain of Tween. Since in F5 and F6, the dominant content of formulations is contained span 60, a little addition of Tween 60/80 (15%) increases entrapment efficiency and drug release



independent to type of Tween. Since the chains of Tween became entangled in those of Span 60, it led to improved entrapment. However, the effective role of Tween 20 (F7 and F4) in improving entrapment of DOX into niosome can be attributed to the dominance of hydrophilic part compared to the total volume of the molecule. Thanks to flexible chains, Tween 80 molecules (F6) could compact themselves among Span 60 chains and became smaller in size as compared to Tween 60 (F5). Formulation prepared with Tween 20 (F4) was also smaller in size compared to F5 due to smaller size of alkyl length. This implies that the enhancement of entrapment efficiency and reduced diameter of Tween formulation can also be explained by the presence of PEG molecule tagged to Tween molecule in addition to DSPE-mPEG in all formulations. The Presence of DSPE-mPEG (2000) in formulations has made niosome smaller, less aggregated and stable *in vivo* and *in vitro*. PEGylation also improved drug entrapments.^{7,32,33}

Examining the entrapment efficiency in different drug systems indicated better performance of dual-drug system compared to single drug one. These findings suggest that delivering multiple drugs simultaneously (a hydrophobic and a hydrophilic drug) restores balance followed by synergistic co-entrapment effect in the system, and also provides stability.

In order to improve drug penetration into vesicles, controlled drug release and make LipoNio formulation more stable and biocompatible for *in vivo* and *in vitro* application, 30% cholesterol was applied in all of our formulations.^{12,34} Zeta potential and PDI of all formulations ranged from -27 to -58 and 0.06 to 0.21 , respectively. Based on sustained drug release, small diameter, high entrapment efficiency and simplicity of the formulation F2 with Tween 60 as surfactant has chosen as the formulation for further studies.

3.2. Effect of phospholipid on DOX/CUR LipoNio formulations

For induction biocompatibility to niosome, various LipoNiosomal formulations were compared based on the type of phospholipids in terms of entrapment efficiency, mean diameter, zeta potential and percentage of release during 8, 24, 48 and 96 hours (Table 2). F10 and F11 were composed of natural and synthetic phospholipids, respectively. According to the results, the LipoNio formula containing SPC80 showed higher drug entrapment and larger diameter compared to F11. However, the percentage of drug release was higher for F10 compared to F11. These results were consistent with those of

our previous work.¹² The hydrophobic alkyl chains were approximately equal in length for both SPC80 and DPPC, but unsaturation of SPC80 has made it more mobile and flexible. Thus, the flexibility of binding was improved by the addition of SPC80; however, it made the drug release fairly rapid (burst drug release, contrary to the purpose of sustained-drug release) and also resulted in nanoparticles with larger diameter. It can also be attributed to the fact that T_m of SPC80, unlike DPPC, was lower than body temperature, which resulted in its instability.³⁵ Another explanation would be that the higher entrapment efficiency of F10 which led to increase of release rate due to concentration gradient between both sides of the niosomal membrane.

The entrapment efficiency of CUR for F11 is lower than to F2, due to loss of free space in biliary of vesicle, between Tween 60 chains, after filling with phospholipid chains.

To prepare formulation for co-drugs-gene delivery, F12 was synthesized with the cationic phospholipid (DOTAP). As can be seen, zeta potential has made particles extremely positively charged with the addition of 15% DOTAP with the entrapment efficiency slightly increased. The PDI (~ 0.3 for all) indicated that no agglomeration occurred which can be attributed to mutually repellent force between the particles with the same-charge in the suspension.

3.3. Effect of preparation method

The LipoNiosomal formulation was prepared using pH gradient and thin film methods. Table 3 shows the effect of preparation methods on characteristics of vesicles. It was found out that the LipoNiosomal formulation prepared by the pH gradient method led to formation of particles with larger diameter and with a little higher encapsulation efficiency of CUR. It also showed a $+10$ mV change in zeta potential that resulted in prevention of agglomeration due to repulsive electrostatic forces. Although it confirms our previous findings for the single delivery of DOX, there is some concern about the formation of a complex between PBS solution and CUR during buffer replacement step. As described in literature, CUR was decomposed after 30 min of incubation in PBS.³⁶ In fact, the hydration step was performed in thin film method by injection of acidic water (sterile water for injection) with pH value of 5, while for pH gradient method a PBS solution (at pH = 7.4) was used for buffer replacement step. So, the stability of CUR depends on the pH, *i.e.*, improved with decreased pH. These results can be attributed to the fact that the proton was removed from the phenolic group in CUR, leading to decomposition of CUR.

3.4. Characterization of optimized formulation

3.4.1. Morphological characterization. SEM photographs of the carrier before and after CUR and DOX encapsulation are illustrated in Fig. 2. Comparison of blank LipoNiosome and dual drug-loaded LipoNiosome indicated that despite of no significant change in the size of nanoparticles, drug-loaded nanoparticles tend to agglomerate. All LipoNio particles show smooth surface, round shape and separated rigid boundaries, with homogeneous distribution. The nanoparticles had

Table 3 Characterization of formulations prepared by various types of preparation methods

Code	Preparation method	EE%	Size (nm)	Zeta potential (mV)	PDI
F11	Thin-film	EE% DOX = 88.23 EE% CUR = 77.11	42	-37	0.324
F13	pH-Gradient	EE% DOX = 88.01 EE% CUR = 95.1	158.4	-23	0.15



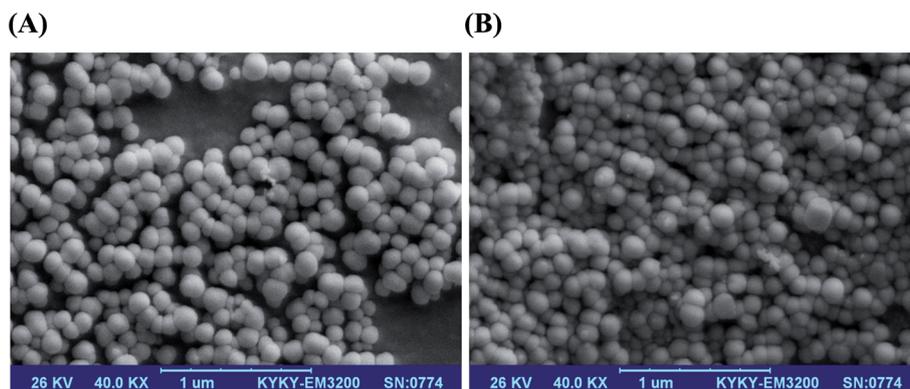


Fig. 2 Scanning electron microscopy (SEM) of (A) blank LipoNiosome; and (B) dual drug-loaded LipoNiosome (DOX-CUR-LipoNiosome).

diameter of less than 100 nm, also confirmed by DLS, which enabling it to pass through blood barrier and accumulating in bone marrow.

3.4.2. Drug-excipients interaction evaluation. To investigate the presence of chemical interactions between dual drugs carrier, a single drug carrier, DOX, CUR and blank carrier, the

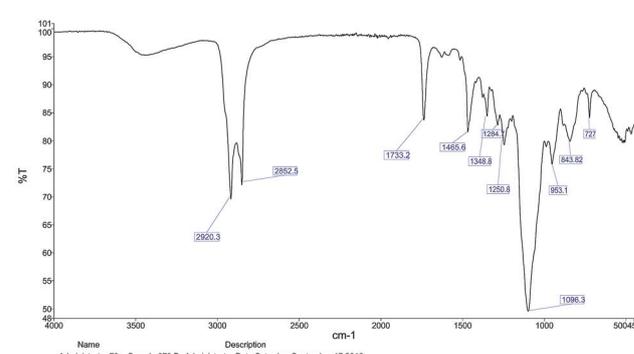
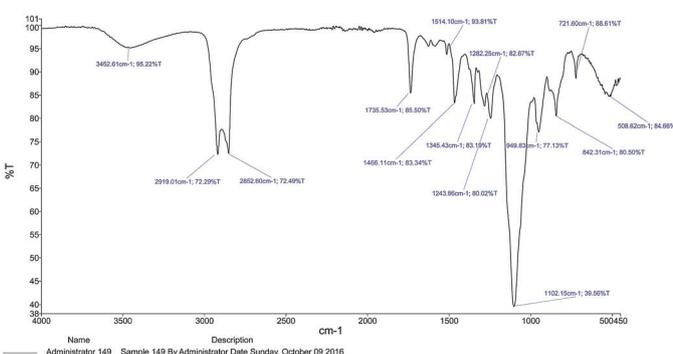
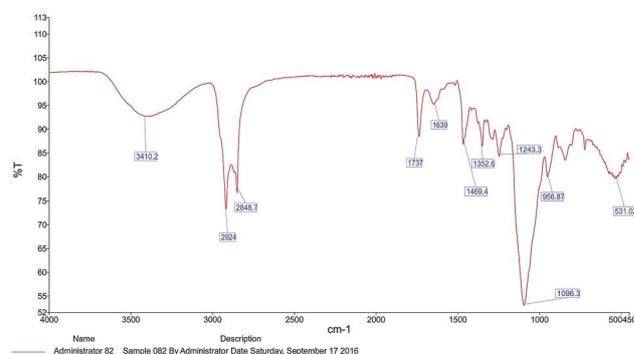
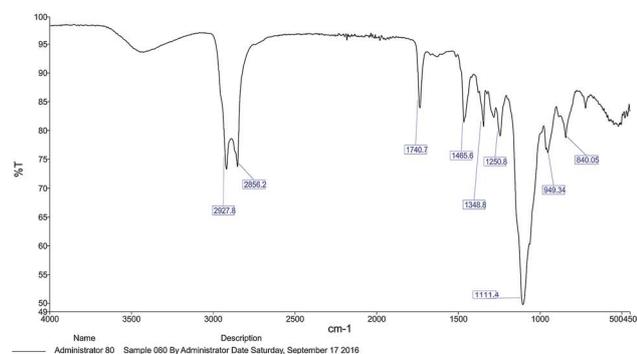
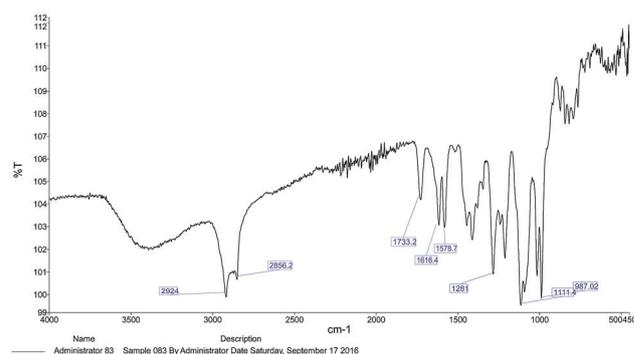
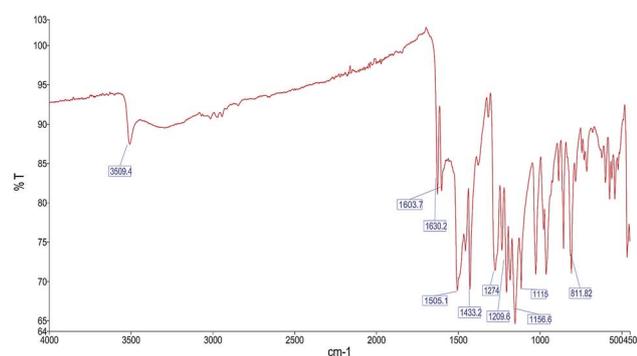


Fig. 3 FTIR spectra of CUR FTIR, DOX FTIR, blank LipoNio FTIR, LipoNio-DOX FTIR, LipoNio-CUR FTIR, LipoNio-DOX-CUR FTIR.



FTIR spectral data were obtained, as shown in Fig. 3. The FTIR pattern for LipoNiosomal DOX-CUR shows various characteristic peaks of DPPC, Tween 60, cholesterol and DSPE-mPEG in the range of 3400–1096.30 cm^{-1} which are representative of the hydroxyl bands vibrating in stretching and bending motion (broad band at 3400 cm^{-1}), $-\text{CH}_3$ asymmetric and symmetric stretching (2920.3 cm^{-1}), and symmetric vibration of ethylene ($-\text{CH}_2$) group (2852.5 cm^{-1}). The peaks at 1733.2 cm^{-1} and 1465.6 cm^{-1} substantiated the existence of $\text{C}=\text{O}$ stretching of the ester group and $-\text{CH}_2$ bending in lipids and surfactant, respectively. The characteristic peaks centered at 1348.8 cm^{-1} and 1250.8 cm^{-1} belonged to alkane $\text{C}-\text{H}$ rock and alkyl $\text{C}-\text{N}$ stretch (for DPPC and DSPE-mPEG), respectively. The band which appeared at 1096.3 cm^{-1} was attributed to $\text{C}-\text{O}$ stretch in ether and ester groups (Tween 60 and DSPE-mPEG). All peaks were repeated in the FTIR spectrum of blank LipoNios, LipoNio-DOX and LipoNio-CUR, which clearly confirmed that there were no additional peaks and no chemical interaction between drugs and other components in the formulation. The FTIR spectrum of DOX and CUR shows several characteristic peaks centered at 1274 cm^{-1} for CUR and 1616.4 cm^{-1} which was characteristic of DOX.

Some peaks got broader or were slightly shifted to another wave number; resulted by intermolecular forces, *e.g.*, hydrogen bonding in Tween 60. The results also confirmed that both drugs encapsulated by the LipoNio were stable during formulation. All peaks were also observed in the spectrum of the single drug encapsulation of DOX or CUR; it shows that co-encapsulation of drugs has kept the integrity of drugs and carrier formulation.

3.4.3. Cytotoxicity study. AML is a disease of infants and older adults with an estimated 13 000 new cases per year in USA.³⁷ Osteosarcoma is the most common malignant bone tumor among children. Classic treatment of osteosarcoma needs the children to endure painful chemotherapy.³⁸ As shown in Fig. 4A, the cell viability efficacy was improved when the cell lines were exposed to blank LipoNiosome instead of blank niosome ($\sim 14.75\% \uparrow$). In order to verify the enhanced anti-proliferative activity of the LipoNio-DOX-CUR compared to free DOX/CUR and LipoNio-DOX-CUR, the cytotoxicity of our dual-drugs delivery system was tested against MG-63, KG-1, primary bone and HBMF-SPH cells. Free DOX, free CUR, co-administration of LipoNio-DOX + LipoNio-CUR as well as single delivery of LipoNio-DOX or CUR were used as the controls (Fig. 4B and C). The cytotoxic capacity of LipoNio-DOX-CUR in all concentrations was higher than other under study group (Fig. 4B-E). The proliferation inhibition by drugs was strongly dose-dependent, and it was more effective for entrapment drugs due to slow internalization of drugs.

Free CUR was less toxic compared to free DOX in normal cells. Furthermore, the highest concentrations of free CUR and DOX decreased the % survival in other cell lines treated with DOX up to 45% and up to 50–60% when the cells were treated with CUR.

Our findings indicate that the anti-proliferative effect of CUR entrapped inside LipoNiosome was enhanced which can be due to increased contact surface and solubility. Because of toxic

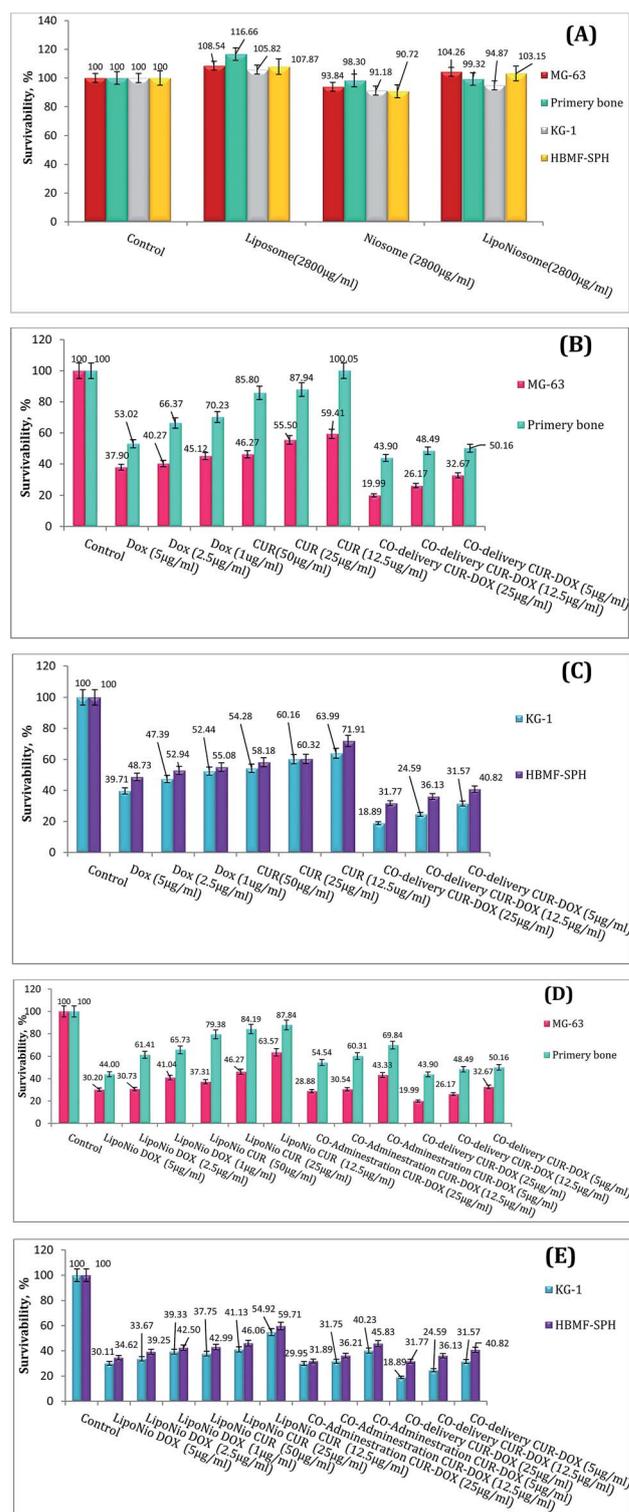


Fig. 4 Survival analysis of MG-63, primary bone cell, KG-1 and HBMF-SPH in various drug concentrations (1, 2.5, 5, 12.5, 25 and 50 $\mu\text{g mL}^{-1}$); biocompatibility comparison of liposome, niosome and LipoNio (A); comparison between toxicity of free drugs and LipoNio-DOX-CUR in various concentrations for OS (B); for AML (C); comparison between cytotoxicity of single drug carrier, co-administration of both single carriers (LipoNio-CUR + LipoNio-DOX), and co-delivery system (LipoNio-DOX-CUR) for OS (D) and for AML (E).



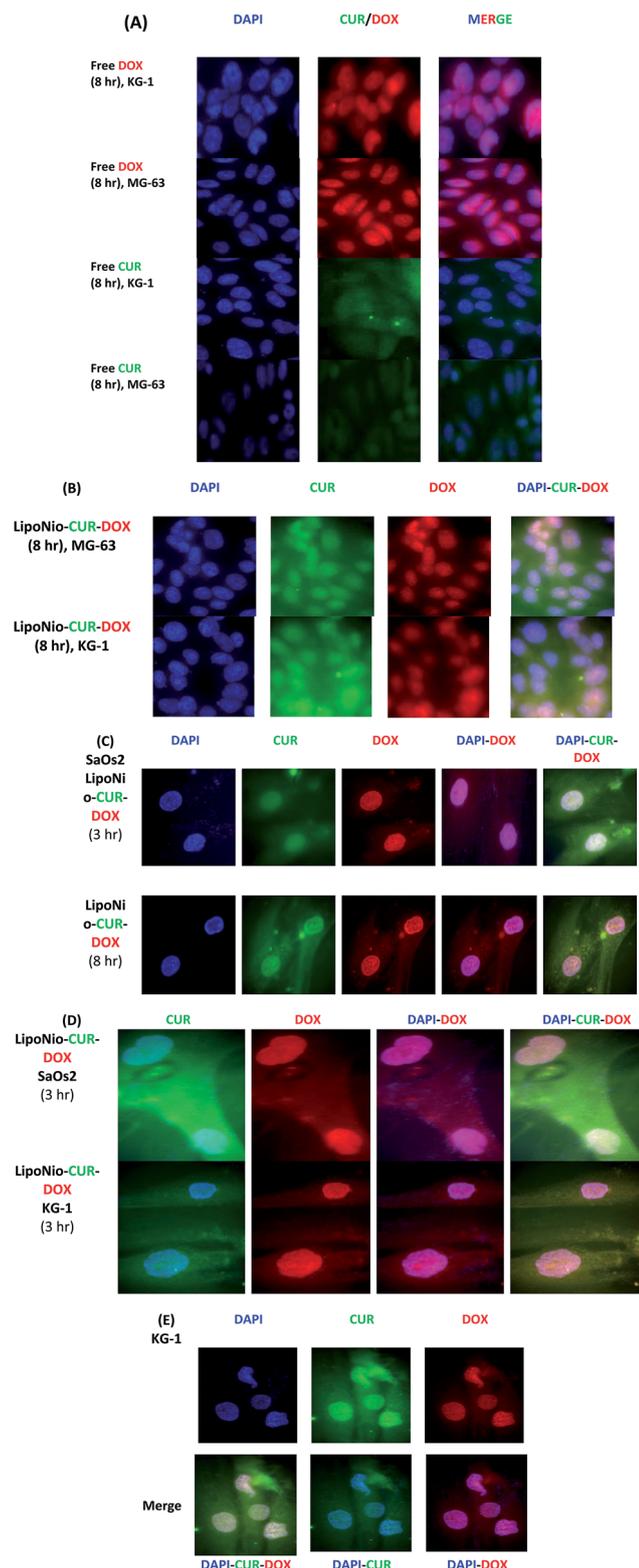


Fig. 5 Cellular uptake images of MG-63, KG-1 and SaOs2 cells, treated with free DOX, free CUR and drugs entrapped into LipoNio vesicles. DAPI was used for labeling the nuclei. DOX and CUR have intrinsic red and green fluorescence, respectively. (A) Comparison between the cellular uptake of free drugs after 8 h-treatment in KG-1 and MG-63 cell line; (B) cellular uptake of MG-63 and KG-1 cells treated with LipoNio-DOX-CUR; (C) cellular uptake of SaOs2 cells treated with LipoNio-DOX-CUR for 3 and 8 h; (D) cellular uptake of SaOs2 and KG-1 cells treated with LipoNio-DOX-CUR for 3 h; (E) typical details of merging.

nature of DOX and concerns about increased cytotoxicity of the current formulations resulted from sustained drug release, it is possible that the formulation could also result in increased cytotoxicity to primary bone cells. LipoNio-DOX-CUR showed significant higher cytotoxicity ($p > 0.1$) compared to the primary bone cells and HBMF-SPH cells.

3.4.4. Nano-LipoNiosomal DOX-CUR localization assay. The cellular uptake of MG-63 and KG-1 cells, treated with free drugs (DOX and CUR) and drug-loaded LipoNio, was studied by fluorescence microscopy, as illustrated in Fig. 5A-E. MG-63 and KG-1 cells were selected as models of sensitive adherent and non-adherent cancerous cells to study the capability of our new formulation. As shown in Fig. 5A and C, the cells, treated with entrapped drugs showed greater purple and turquoise blue color intensity compared to cells treated with free drugs. It is well-known that entrapped drugs (at nano scale) could penetrate the cells by endocytosis, whereas the free drug molecules (at Angstrom scale) were moved by diffusion mechanism.

The MG-63 cell line successfully absorbed the entrapped DOX. The prepared LipoNio-DOX-CUR formulation could effectively enter the cancerous cells, mostly into the nucleus whereas free DOX was predominately distributed in the cytoplasmic region. The accumulation of DOX in nucleus could induce apoptosis and inhibit DNA replication.³⁹ Furthermore, CUR with its antioxidant activity can neutralize free radicals and increase activity of the topoisomerase II, DNA cleavage enzyme.^{28,40,41} Concentrating CUR within cancerous cells *via* carrier effectively enhanced anticancer activity of DOX. The prepared formulation served as a sustained-release carrier for both drugs so that most of the purple and turquoise blue colors were resulted by drugs transfected *via* carrier or previously released drug in acidic compartments. The release of the drug was increased in the cells due to the low pH value of the lysosome, leading to loosening the carrier membrane and enhanced drug release.

Comparing MG-63 and KG-1 cell lines indicated that more intense and widespread purple and turquoise blue fluorescence was observed for KG-1, and the fluorescence intensity of drug within cells was significantly stronger than that in osteosarcoma cell line.

Interestingly, cell viability results also confirmed that KG-1 cell line was more sensitive for both entrapped and free DOX. Zhang *et al.* recently studied the localization of DOX and CUR entrapped in PEG nanoparticles after 2, 4 and 8 h exposure of HepG2 cells and found successful drug internalization after 8 h.²⁸ On the contrary, our results showed high drug accumulation after 3 h treatment. This may be due to smaller size of nanoparticles in our study that led to an increase in their diffusion. Also, our new nano LipoNiosomal formulation was more sensitive to changes in pH than other studies,²⁶ *i.e.*, it resulted in significant improvement in nuclear transfection of DOX in cancerous cells (increased release of DOX in lysosome with pH = 4). The negatively charged surface of LipoNio-DOX-CUR contradicted the assumption that the interaction between nanoparticles and a negatively-charged cell membrane resulted in the improvement of charge-dependent cellular transfection. Thus, it is hypothesized that electrostatic force could not have



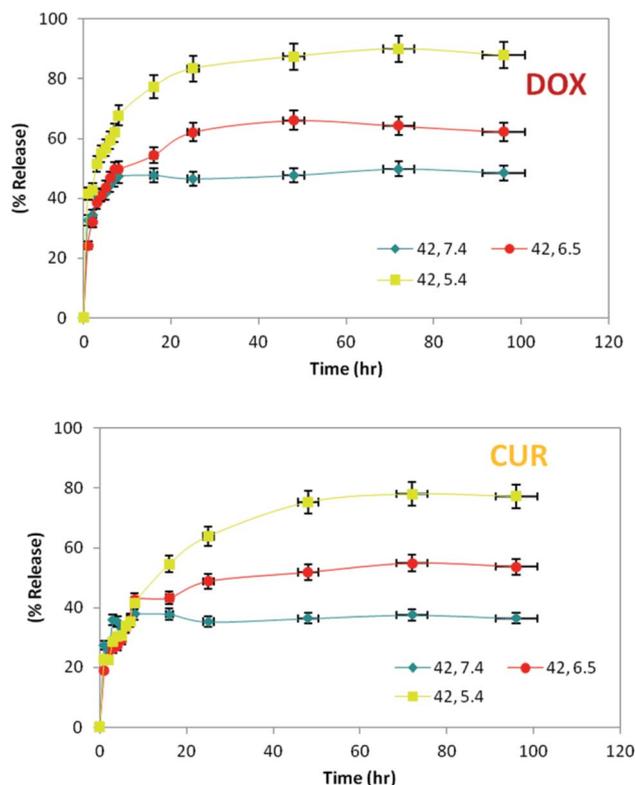


Fig. 6 Drug release profile in various pH.

Table 4 Kinetic model's constant and statistical parameters

	K	n	R^2	Drug
Adjacent healthy cells	32.88	0.0959	0.4599	DOX
	20.4	0.1227	0.7448	CUR
Adjacent cancerous cells	44.12	0.1699	0.9300	DOX
	21.14	0.3053	0.9555	CUR

played specific role in the enhancement of cellular transfection of nanoparticles. In addition, electrostatic force reduced the side effects resulted by undesirable uptake of DOX into normal cells. After 8 h of incubation, it should be noted that the fluorescence intensity of LipoNio-DOX-CUR was significantly enhanced. We also checked the cellular uptake of SaOs2 cells treated with LipoNio-DOX-CUR as an external model validation.

3.4.5. Drug release profile. The results of 96 h release profile of DOX and CUR from LipoNio particles in PBS (pH 7.4, 6.5 and 5.4) at 42 °C are presented in Fig. 6. After 96 h drug release triphasic pattern showed with an initial burst release, secondary linear phase release followed by an apparent zero-order release phase. Approximately 48.49% and 36.45% of loaded drugs were released at pH = 7.4 and 42 °C for DOX and CUR, respectively.

Since DOX and CUR are small molecules and the permeability cut-off of the dialysis tube was 12–14 kDa, released DOX and CUR could pour easily out of the tube. Therefore, the release of drug was not limited by the dialysis tube or the size of drugs; however, the amount of drugs released was influenced by

the temperature, pH of surrounding buffer and also the structure of LipoNio membrane. Three chosen pH values which represented the typical levels that nanoparticle are encountered in the physiological condition, tumor tissue and cancerous cells were 7.4, 6.5 and 5.4, respectively. Cancerous cells and tumor tissue usually have hypoxia that decreases the pH value. We used this phenomenon to make our formulation pH-sensitive to increase the toxic effect against malignant cells and subsequently reduce side effects against normal cells.

The most rapid drug release took place at 42 °C and pH = 5.4–6.5. Thus, our study indicated timing of sustained release of drugs over 4 days from the LipoNiosome. Cytotoxicity assay confirmed the pH-sensitive nature of our formulation that resulted in higher toxic effect of LipoNio-DOX-CUR on cancerous cells compared to normal cells. The release of drugs from inside the LipNio membrane to external fluid is mainly through solubility, diffusion and convection, as fully described in literature.⁷ Rapid release of drugs at low pH values or high temperature can be attributed to the enhancement of solubilization of aggregated drugs into vesicles, loosening of the LipoNio membrane and slightly improving temperature dependence of diffusion. Considering drug release, cell viability, and cellular uptake results, the total amount of released drug was increased over time and reached to approximately 87.86% for DOX and 77.2% for CUR (pH = 5.4, $T = 42$ °C) after 72 h. In other words, about 80% of LipoNio-DOX-CUR and 100% of free drugs killed 37% and 32% of the cells, respectively. It can be concluded, a lower dose of the drug created more cytotoxicity due to the fact that the LipoNio-drug particles were disassembled after uptake by the cells, and meanwhile, DOX and CUR continuously accumulated within nucleus and cytoplasm, and then played anti-cancer effect. The regression analysis was performed with the corresponding nonlinear regression correlation coefficient value (R^2) equal to 0.91. The estimated model parameters of Korsmeyer–Peppas are given at Table 4. According to the results, Korsmeyer–Peppas model could sufficiently estimate the drug release behavior near the cancerous cells. The exponent value of Peppas's model was increased when LipoNio-DOX-CUR was made in contact with cancerous cells instead of normal cells, and reached about 0.5. Thus, the drug release was mainly controlled by Fickian diffusion. In fact, with decreased drug release rate, it became less controlled by Fick's law. These results are in consistent with those of release experiments carried out at various pH and temperature.

4. Conclusions

Our results successfully suggested a new model called LipoNiosome for hydrophobic and hydrophilic drug delivery *via* single nanocarrier. Phospholipid combined with surfactant was suggested as a new strategy to improve biocompatibility and characteristics of our formulations. With the addition of surfactant to nanoparticles, the drug loading capacity of carriers and the stability of the formulation were improved and particle surface tension reduced. The anti-proliferative performance of dual-drug delivery system with four different cell lines showed



uptake and effectiveness of these drugs. Our findings indicated that nanocarrier-based approach adopted for delivery of CUR/DOX combinations was effective in combating cancer cells *in vitro*. Our long-term objective is to provide the proof-of-principle for the comprehensive model of targeted dual-drug delivery system for multidrug-resistant cancer treatment *via* biodegradable, biocompatible and stable carrier for simultaneous delivery of high amounts of siRNA and hydrophobic and hydrophilic drugs.

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