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# Fluorescent silver nanoparticle based highly sensitive immunoassay for early detection of HIV infection†

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For the first time, we have engineered streptavidin labeled fluorescent silver nanoparticles for their application in immunosensing of biomolecules which will significantly increase sensitivity without compromising the specificity. A computational perspective for understanding the efficiency of the conjugation process is provided as a proof of concept. The fluorescence based sandwich immunoassay was demonstrated to detect HIV-1 p24 antigen in clinical specimens with improved sensitivity and specificity. The detection range of the fluorescent silver nanoparticle-based immunoassay (FSNIA) was found to be between 10 and 1000 pg mL<sup>-1</sup> in a linear dose dependent manner. False positives were not observed with plasma samples from healthy adults (HIV<sup>-ve</sup>), hepatitis B (HBV<sup>+ve</sup>) and hepatitis C (HCV<sup>+ve</sup>). Plasma samples that were HIV<sup>-ve</sup> showed no interference with detection of HIV-1 p24 antigen. This technology can be used in resource limited settings and easily adopted for the detection of other pathogen antigens.

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## 1 Introduction

Developing countries are witnessing a rise in infectious diseases which are the leading cause of loss of life.<sup>1,2</sup> Although different diseases such as malaria, tuberculosis, hepatitis, pulmonary infections, to mention a few, are prevalent worldwide, the disease which has caught the attention of the globe due to its severity is Acquired Immune Deficiency Syndrome (AIDS) caused by the Human Immunodeficiency Virus (HIV). Every year nearly 2.2 million people are infected and 1.3 million die from AIDS-related illness.<sup>3</sup> At the end of 2015, around 40 million people were living with HIV/AIDS.<sup>4</sup> AIDS and its related illnesses

can affect large part of populations if not diagnosed and treated at the earliest stage.

The viral agent responsible for AIDS (Acquired Immune Deficiency Syndrome), human immunodeficiency virus (HIV), destroys cells of the immune system and progressively impairs the body's ability to defend against other infections and certain cancers.<sup>5</sup> Antiretrovirals (ARVs) and their use in combination "cocktails" have drastically reduced mortality and morbidity, prolonged life expectancy and improved the quality of lives of infected people, although they have not been able to cure patients of HIV/AIDS.<sup>6</sup> However, 95% of people infected with HIV live in developing countries, where the availability and access to these medicines remains unacceptably inadequate and the costs unreasonably exorbitant. Hence, rapid and accurate diagnosis of HIV is pivotal in the management and control of this global outbreak. The diagnostic tools currently available for early detection of HIV often require advanced and elaborate facilities which restrict their application in resource limited settings. Currently available diagnostic assays need further improvement for early detection and this can be achieved using nanoparticles for signal amplification coupled with sensitive and specific antibodies. Thus, the current need is rapid, sensitive and specific diagnostics at point of care which are capable of rapid identification and characterization of the disease allowing patients to get immediate treatment for effective clinical outcomes.

When it comes to detection of HIV, there are a wide variety and range of biomarkers related to HIV infection, which can

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even predict the progress of HIV infection to AIDS.<sup>7</sup> Although HIV antibody tests are the most appropriate for identifying the infection, they are insufficient due to the time the antibodies take to manifest in the plasma.<sup>8</sup> Hence, alternate efficient technologies are required for accurate diagnosis for monitoring response to therapy, and to effectively predict disease outcome. Virus isolation through viral culture, nucleic acid detection techniques to detect viral RNA,<sup>9</sup> and tests to detect p24 antigen<sup>10</sup> are commonly used clinical methods to identify virus or viral components in blood early, and confirm an infection. While, these methods are highly specific, each has limitations and their use is restricted to highly specific testing situation depending on the resources available. Although tests for viral RNA detection have been introduced recently, their requirements of complex instrumentation and well-trained personnel makes them unsuitable for resource limited settings.<sup>11</sup>

The p24 antigen tests stand out amongst all other tests as it is one of the earliest protein biomarkers to appear in blood after infection. The HIV-1 capsid (core) p24 is detected much earlier than antibodies during an acute HIV infection.<sup>12</sup> A major advantage that p24 testing offers is that it is highly specific for HIV infection. Testing for p24 can be of value in: early detection of HIV infection,<sup>13</sup> screening blood donor samples,<sup>14,15</sup> diagnosing infection in the newborn,<sup>16,17</sup> and monitoring antiviral therapy.<sup>18</sup>

There are many sensitive detection platforms for HIV p24 detection but user friendly point-of-care rapid test platforms are an unmet need. The most widely used test format is the enzyme-linked immunosorbent assays (ELISAs)<sup>19</sup> which requires separate wells in the assay for each analyte to be detected and use of an enzyme which can be unstable making it less convenient. Since it is frequently necessary or helpful to detect several biomarkers simultaneously, conventional ELISAs make the process time consuming.<sup>20</sup> Current ELISA methods are further limited by problems common to the use of enzymes, contamination of the substrate solution and the need to read the micro well shortly after completion of the enzyme/substrate reaction.<sup>21</sup>

The use of fluorescence-linked immunoassays gives advantages over ELISAs such as increased sensitivity, a larger dynamic range and greater stability of reagents.<sup>22,23</sup> Fluorescence methods may also be multiplexed, thereby using lower volumes of reagents and decreasing the assay time.<sup>24</sup> The ability to spectroscopically detect multiple analytes in a single well assay is not without its challenges. These include cross reactivity of antibodies,<sup>25</sup> the need for discrete spectral bands which do not overlap too closely,<sup>26</sup> and the possibility of interaction between the fluorophores. The use of fluorescent nanoparticles removes some of these challenges which has prompted the application of fluorescent nanoparticles as immunoassay labels.

Noble metal nanoparticles which are generally clusters of 5–100 atoms shielded by capping agents,<sup>27</sup> are of particular interest for their exceptional properties like high tunability of their optical spectra and high surface to volume ratios allowing a higher surface area for specific bio-functionalization.<sup>28–30</sup> In the field of protein detection, noble metal nanoparticles have played a key role in the development of highly sensitive

detection platforms for rapid identification of disease biomarkers.<sup>31,32</sup>

Silver nanoparticles (AgNPs), have been extensively studied for their application in biosensing based on their photoluminescence, surface plasmon resonance, optical scattering, and labelling features.<sup>33,34</sup> These nanoparticles are being explored for other sensing applications in composites and assembled nanostructures for their synergistic properties.<sup>35,36</sup> AgNPs have also been exploited for the development of protocols involving metal enhanced fluorescence based detection, surface-enhanced Raman scattering (SERS) and in combination with gold to form bimetallic NPs which can either be alloys or core shell nanoparticles.<sup>37,38</sup>

Fluorescent silver nanoparticles (FSNPs) are emerging functional materials which have generated significant interest in the field of bioimaging and biosensing.<sup>39,40</sup> These particles can interact with light as they have discreet energy levels due to quantum confinement,<sup>41</sup> allowing intense light absorption and emission.<sup>42,43</sup> These optical properties along with superior biocompatibility, brightness and photostability have led to their application as fluorescent biolabels.<sup>44,45</sup> Additionally, these FSNPs can have application as sensitive probes, as their fluorescence emission is a function of their interaction with the organic scaffold.<sup>46–48</sup>

Combining the efficiency of FSNPs in biosensing with the current need for sensitive detection of biomarkers, we report the development of a versatile fluorescent silver nanoparticle immunoassay (FSNIA). The FSNPs bioconjugated to streptavidin can be used with biotin labeled antibodies or antigens for detection of target antigen or antibodies. The conjugation and functioning of FSNPs was interpreted using computational modeling. A mechanistic interpretation of the bio-conjugation was sought for immunosensing applications for which *in silico* studies were performed using various computational methods to explain the thermodynamics of these processes. In this regard, the energetics of individual steps was evaluated to elucidate the process of the conjugation observed. FSNIA was then employed for the detection of HIV. HIV-1 p24 was chosen as the model analyte as it is the biomarker for early detection of HIV. The results in this study can help in development of a multiplexed screening protocol based on FSNPs which could be potentially beneficial in clinical settings and in the control and prevention of the spread of AIDS.

## 2 Experimental

### 2.1 Chemicals and reagents

Glutathione (Himedia), silver nitrate (Sigma), hydrazine hydrate (Thermo Scientific), 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide, (Thermo Scientific, USA), *N*-hydroxysuccinimide (Thermo Scientific, USA), streptavidin (Scripps Lab, USA), casein block buffer (Thermo Scientific, USA), anti-HIV-1 p24 antibody (Prospec, USA), HIV-1 p24 antigen (Virogen, USA), biotinylated HIV-1 p24 detector antibody (Perkin-Elmer, USA). Phosphate buffer saline (100 mM, pH-7.2) and carbonate bicarbonate buffer (500 mM, pH-9.5) were prepared by following established procedures. Centrifugal spin columns



(10, 30, 100 kDa) were purchased from GE Health Care, UK. Double distilled water was used for performing all the experiments. Plasma samples from patients and healthy individuals were obtained from the Sri Sathya Sai Institute of Higher Medical Sciences, Prasanthigram, India, (study id-SSSIHL/IEC/PSN/BS/2012/01). The details of the samples were coded and tested in a blinded manner. Samples were confirmed to be positive or negative using fourth generation Microlisa HIV Ag + Ab Elisa kit (J. Mitra & Co. Pvt. Ltd.).

## 2.2 Synthesis of fluorescent silver nanoparticles

Fluorescent silver nanoparticles (FSNPs) were prepared following the process of chemical reduction using hydrazine hydrate as the reducing agent.<sup>49</sup> 5 mL aqueous solution of 2 mM glutathione was mixed (with vigorous stirring) with 5 mL aqueous solution of 1 mM silver nitrate till a milky white acidic solution was obtained. The acidic gel was brought to neutral pH by adding 400  $\mu$ L of sodium hydroxide (1 M) until a clear solution was formed. Following the formation of a clear solution, the rate of stirring was reduced and after 5 minutes, 500  $\mu$ L of hydrazine hydrate was added. The solution was now stirred for 10 hours till it turned pale yellow. The resulting solution was filtered to remove any large particulates using a nitrocellulose membrane. The filtered solution was lyophilized and stored at 4 °C for further characterization and immunoassay experiments.

## 2.3 Characterization of the FSNPs

The morphology and size of the FSNP dispersion was characterized using a Jeol 1400 TEM at 80 kV and LIBRA 200 High Resolution Transmission Electron Microscope operated at 200 kV, respectively. The FSNPs were drop casted on the copper grid and vacuum dried before imaging. Total 514 particles were analyzed using Image J to determine the size distribution of particles. The optical characterization experiments and measurements were carried out using Shimadzu 2450 UV-Vis spectrophotometer and Spectramax M5 microplate reader. To obtain the absorbance and fluorescence spectra, the samples were accordingly diluted. FTIR spectra were obtained using Thermo Scientific Nicolet 10 FT-IR spectrometer on the potassium bromide (KBr) pellet.

## 2.4 Bioconjugation of FSNPs with streptavidin

The bioconjugation of the above purified FSNPs was carried out using the EDC-NHS coupling protocol.<sup>50</sup> 10 mg of nanoparticles were dispersed in 10 mM phosphate buffer (pH = 7.4) and were washed in centrifugal spin columns. Nanoparticles were dispersed in Phosphate Buffered Saline (PBS) and the solution was sonicated with a tip sonicator for 30 s. The activation of carboxyl groups was carried out by mixing the FSNPs with 10 mmol L<sup>-1</sup> of EDC and 10 mmol L<sup>-1</sup> NHS in PBS buffer for 30 min. These activated particles were washed once with wash buffer. To this solution, 50  $\mu$ L solution of 1 mg mL<sup>-1</sup> streptavidin dissolved in carbonate bicarbonate buffer (pH = 9.0) was added. The above prepared solution was incubated at room temperature for 3 h following which the streptavidin conjugated nanoparticles were washed 5 times with the wash buffer.

In order to gain a deeper mechanistic understanding of the bio-conjugation for immunosensing applications, *in silico* studies were performed using various computational methods to rationalize the thermodynamics of these processes. In this regard, the energetics of the individual steps was evaluated to elucidate the process of the conjugation observed.

**2.4.1 Energetics of the conjugation step.** The molecules involved in the studies were constructed in ChemSketch and were optimized in MOE (Molecular Operating Environment) using the MMFX99 force field.<sup>51</sup> In the bio-conjugation protocol, the individual reactants and products were optimized and their net energies for every individual stage of the process were calculated. Further, to enunciate the role of silver nanoparticles in this particular conjugation, the glutathione molecule (GSH) was modified with the thiol group replaced by S-Ag (1 Ag[GSH]), S-6 Ag(6 Ag[GSH]), and lastly S-31 Ag (31 Ag[GSH]).

**2.4.2 Conjugation of the nanoparticle to the protein.** Given that the bio-conjugation is primarily the interaction of the GSH-NHS intermediate with streptavidin, we evaluated the role of Ag in this interaction. In this regard, the molecular docking studies were carried out using Hex 8.0.0 where the interactions between the different Ag-GSH ligands and streptavidin were generated as HEX scores that are an indicator for the binding affinity.<sup>52</sup>

**2.4.3 Interaction of biotin with the conjugated protein.** The deterministic step that allows the sensing of p24 by the fluorescent probe is based on the interaction between the FSNP conjugated streptavidin and biotinylated detector antibody. Hence, we assessed the effect of conjugation on consequent binding to biotin. The 1, 6 and 31 Ag[GSH] systems conjugated to streptavidin were consequently studied for their binding affinity unto biotin using HEX 8.0.0. This was eventually compared to the control, which is reaction in the absence of glutathione conjugation.

## 2.5 Fluorescence polarization

The fluorescence polarization experiments were carried out in order to ascertain whether the process of bioconjugation has produced conjugated nanoparticles. The difference in the fluorescence polarization value is the indicator of size of the bioconjugated molecule, which increases after conjugation.<sup>53</sup> The excitation and emission wavelengths were chosen as 401 nm and 716 nm respectively. The fluorescence polarization value was calculated from the formula given by eqn (1),

$$FP = \frac{I_{\parallel} - G \times I_{\perp}}{I_{\parallel} + G \times I_{\perp}} \quad (1)$$

where, FP is fluorescence polarization,  $I_{\parallel}$  is the parallel polarized emission,  $I_{\perp}$  is the perpendicularly polarized emission,  $G$  is the correction factor which is evaluated for the sensitivity of the instrument.

The  $I_{\parallel}$  and  $I_{\perp}$  were measured in the fluorescence polarization mode by Spectramax M5 plate reader and substituted in the above formula to obtain the FP ratio. Conjugation is confirmed by the higher magnitude of fluorescence polarization for the conjugated FSNPs compared to the unconjugated FSNPs.



## 2.6 Fluorescent silver nanoparticle immunoassay (FSNIA) for detection of HIV-1 p24 antigen

**2.6.1 Immunoassay protocol.** A capture antibody concentration of  $1 \mu\text{g mL}^{-1}$  was prepared by diluting the stock solution with carbonate-bicarbonate buffer (100 mM, pH 9.6). 55  $\mu\text{L}$  of this capture antibody solution was coated on to the Nunc maxisorp fluorescence microplates and the plates were incubated for 24 hours at  $4^\circ\text{C}$ . After the incubation period, the plates were washed with wash buffer 5 times followed by washing the wells with 300  $\mu\text{L}$  of casein blocking buffer and incubated for 30 minutes at  $37^\circ\text{C}$ . The casein block buffer will ensure that all the non-specific absorption sites are blocked and it will also reduce the background signal. In the next step, different concentrations of HIV-1 p24 antigen diluted in blocking buffer were added to each well and incubated at  $37^\circ\text{C}$  with shaking for 1 hour. Following the incubation step, all the wells were washed 5 times with wash buffer. In the subsequent step, 100  $\mu\text{L}$  of biotinylated detector antibody was added to each well and incubated at  $37^\circ\text{C}$  for 30 minutes. This step completes the formation of this antibody-antigen-antibody sandwich complex to which 100  $\mu\text{L}$  of streptavidin conjugated FSNPs ( $1.5 \text{ mg mL}^{-1}$ ) were added and left to incubate again for 30 minutes at  $37^\circ\text{C}$  with shaking. A final washing step was performed 5 times with PBST buffer to avoid non-specific interactions and reduce the background noise. Finally, the microplate was placed in the Spectramax microplate reader and the fluorescent signal intensities from the sandwich immunocomplexes were measured in a time resolved mode with excitation at 401 nm and emission at 716 nm. All experiments were performed in triplicate. The fluorescence cut off value was derived using the mean of the fluorescence signal intensities of six negative controls plus 2 times their standard deviation (SD). Samples with signal-to-cut off (S/CO) ratios equal to or greater than 2 were considered positive for HIV-1 p24. A schematic representation of the detection of p24 antigen by FSNIA is shown in the Scheme 1.

**2.6.2 Control experiment.** To study the effects of interference of proteins, a control experiment was performed wherein Bovine Serum Albumin (BSA) was taken as an interfering protein and FSNIA was examined for interference. Five different concentrations of BSA were taken:  $0.1 \mu\text{g mL}^{-1}$ ,  $0.5 \mu\text{g mL}^{-1}$ ,  $1.0 \mu\text{g mL}^{-1}$ ,  $5.0 \mu\text{g mL}^{-1}$  and  $10.0 \mu\text{g mL}^{-1}$ , and its effects on the fluorescence signal were observed for a fixed p24 concentration of  $500 \text{ pg mL}^{-1}$ .

In addition to interference of a protein, studies were also carried out to see the interference effect of coexisting viruses. For our study, hepatitis C virus (HCV) was chosen as an interfering virus as it is a common co-infection in HIV patients. To study the effect of the presence of HCV as an interfering agent in the immunoassay, the HCV positive plasma sample was diluted 10, 20, 50 and 100 times.  $500 \text{ pg mL}^{-1}$  of p24 antigen was spiked in all the solution including the undiluted plasma sample. The immunoassay was carried out on the spiked samples.

## 2.7 HIV detection in blood plasma samples

**2.7.1 Preparing plasma samples.** The plasma samples were tested with commercial kits for HIV antigen and antibody.  $2 \mu\text{L}$

of tested plasma was diluted in 188  $\mu\text{L}$  casein block buffer and 10  $\mu\text{L}$  of 10% aqueous solution of Triton X-100 was added.

**2.7.2 HIV-1 p24 spiked plasma samples.** To check the efficacy of our detection system, the first step involved testing HIV negative plasma samples spiked with known concentrations of p24 antigen. These spiked plasma samples were serially diluted with PBS solution to obtain the desired concentrations. The samples were then subjected to FSNIA and the percent recovery for each sample was calculated by the formula as described by eqn (2).

$$\text{Recovery}\% = \frac{\text{measured concentration}}{\text{actual concentration}} \times 100 \quad (2)$$

It is necessary to test the performance of the assay with clinical samples in order to evaluate its applicability to clinical studies. To validate the assay, FSNIA was evaluated using 50 HIV positive and 50 HIV negative clinical samples. These samples were prepared using the procedure described before. The cross reactivity was also evaluated with 10 HBV positive/HIV negative and 10 HCV positive/HIV negative plasma samples.

## 3 Results and discussion

### 3.1 Morphological characterization of fluorescent silver nanoparticles

The synthesis of FSNPs is discussed in Section S1† in ESM. The size and morphology of the silver nanoparticles was confirmed by TEM imaging. The TEM images as seen in Fig. 1 show the particle size to be  $15.4 \pm 3.7 \text{ nm}$  of spherical morphology. The HRTEM confirmed the interplanar spacing to be 0.237 nm which is in accordance with the literature.<sup>54</sup> The nanoparticles were found to be polycrystalline which also is a reported observation.<sup>55</sup>

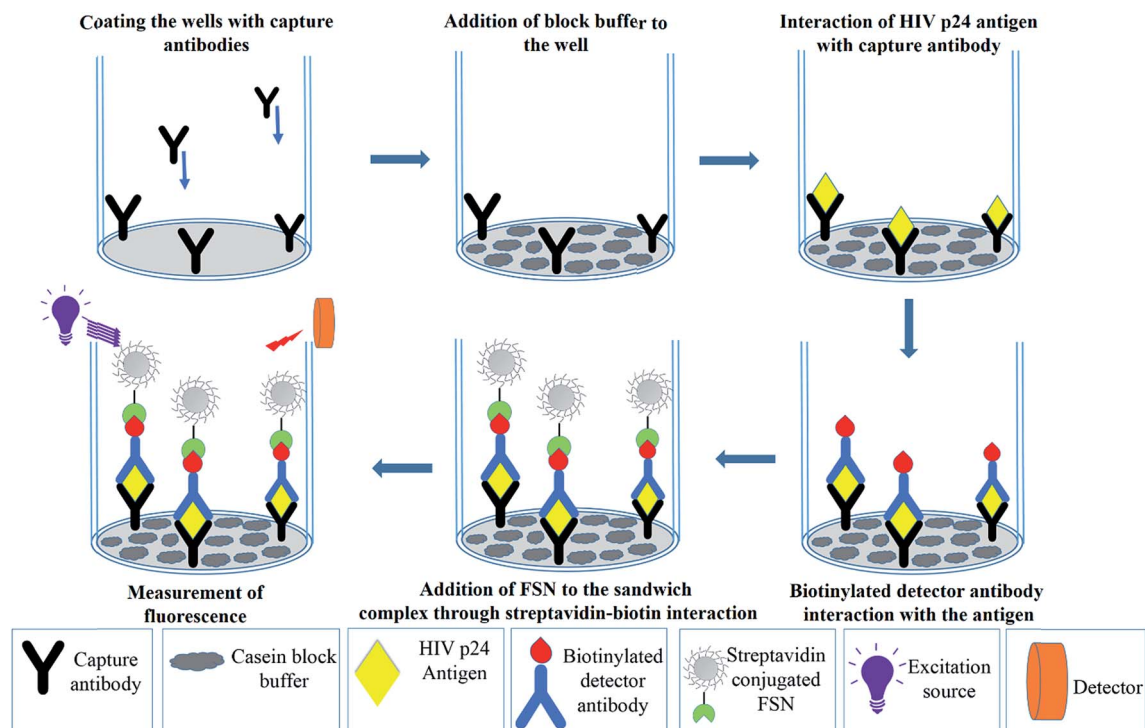
### 3.2 Optical spectral characteristics of fluorescent silver nanoparticles

UV-Vis spectroscopy and photoluminescence spectroscopy studies were used to study the optical properties of the FSNPs. As illustrated in Fig. 2, the absorption spectra revealed a broad peak in the range of 370–450 nm with the peak maximum at 401 nm. On being subjected to UV light (365 nm), a red emission was observed which is visible to naked eyes. A broad emission centered at 716 nm was observed in the emission spectrum with an excitation wavelength peak at 360 nm. This emission can be attributed to hybrid states which are formed due to the interaction of glutathione and silver at their interface.<sup>56</sup>

In order to understand the surface functionalization on the FSNPs, FTIR studies were carried out (Fig. 3a). The most important feature of the FTIR spectra obtained was the missing peak at  $2510 \text{ cm}^{-1}$  which is the characteristic absorption peak for S–H indicating that the S–H group is not freely available in the FSNPs. This suggests that glutathione caps the silver nanoparticles through the thiol from cysteine.<sup>57</sup> The stretching vibration band which is the characteristic peak of carboxylic group was found to have shifted slightly to a lower wavenumber of  $1602 \text{ cm}^{-1}$  which can be attributed to weak interaction







Scheme 1 Schematic illustration of steps involved in FSNI.

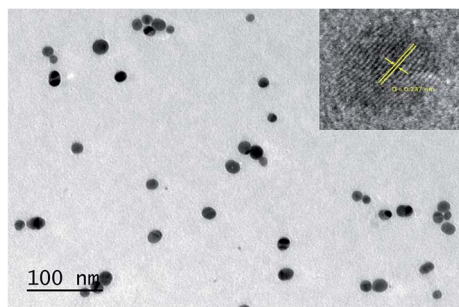


Fig. 1 TEM image of FSNPs which indicates the size of the FSNPs to be around  $15.4 \pm 3.7$  nm. The inset shows the HRTEM image of FSNPs which indicate a  $d$  spacing of 0.237 nm.

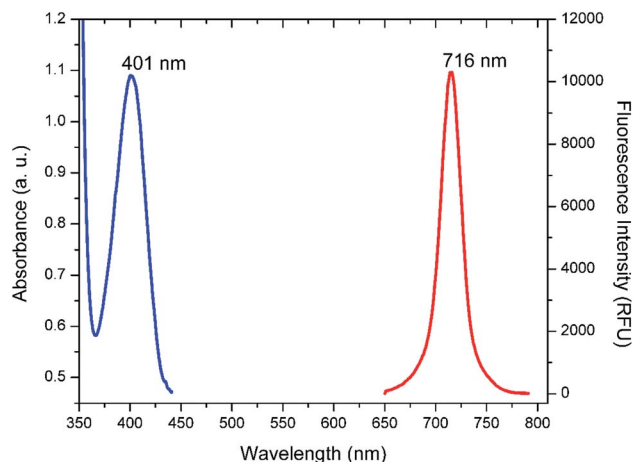


Fig. 2 UV-Vis absorbance and fluorescence emission spectra of FSNPs. The blue peak is the absorbance spectra while the red one is fluorescence spectra of FSNPs.

between the negatively charged  $\text{COO}^-$  and electropositive silver core as depicted in Fig. 3b.<sup>58</sup> The peak at  $1356\text{ cm}^{-1}$  is an indicator of deformation and vibration features of C–N linkage which is predominant in this tripeptide. Meanwhile, the  $1093\text{ cm}^{-1}$  peak is due to the vibration band of C–O. The peaks at  $2974\text{ cm}^{-1}$  and  $1464\text{ cm}^{-1}$  can be ascribed to the stretching and bending vibrations of aliphatic C–H group in glutathione.<sup>59</sup> The 1010–880 and 850–740 bands are the out of plane vibration bands of C–H and N–H group which are typically formed by interaction of the silver nanoparticle with GSH.<sup>60</sup>

### 3.3 Bioconjugation of fluorescent silver nanoparticles to streptavidin

Streptavidin conjugation to FSNPs was performed to make them specific for biomolecular recognition of biotin labeled antibodies.

Since it is difficult to directly conjugate a protein to a nanoparticle with high efficiency, an intermediary coupling protocol was chosen. Now, the most well established and versatile methods that exist for conjugating or crosslinking a moiety to carboxylic acid are based on carbodiimide compounds. Since the medium is aqueous, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC) was chosen. *N*-hydroxysuccinimide (NHS) is an essential part of EDC based coupling protocols, whose purpose is to enhance the coupling efficiency and create amine-reactive intermediates which are stable. EDC helps in coupling of NHS to carboxyls forming an NHS ester which is more stable than the *O*-acylisourea



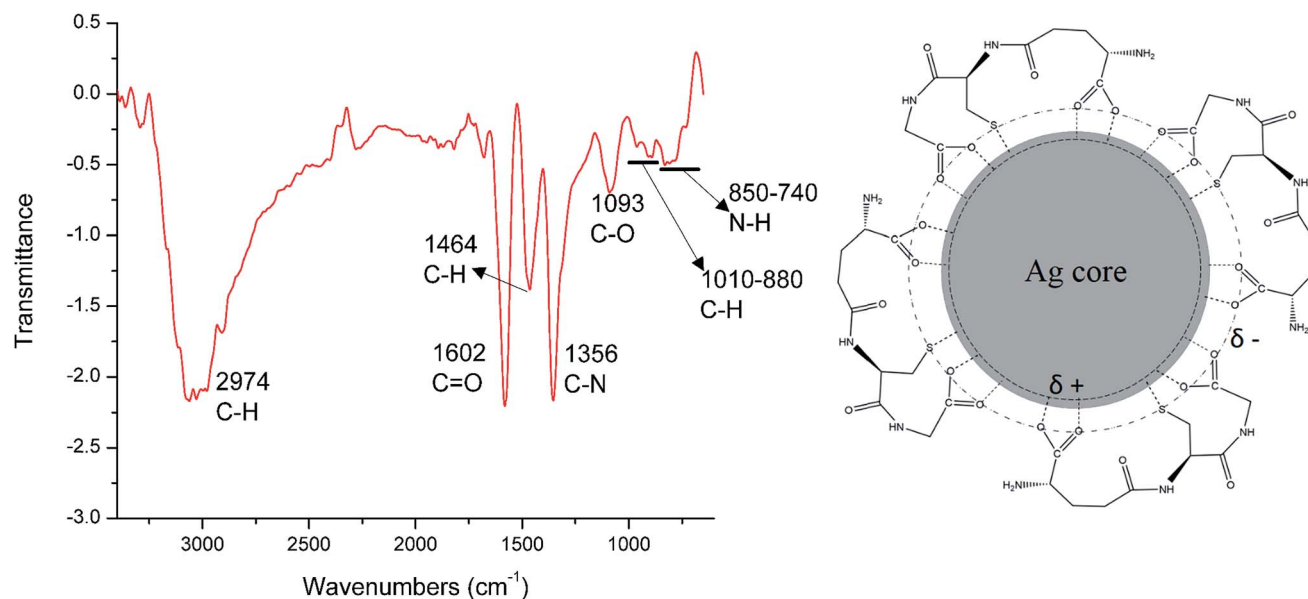
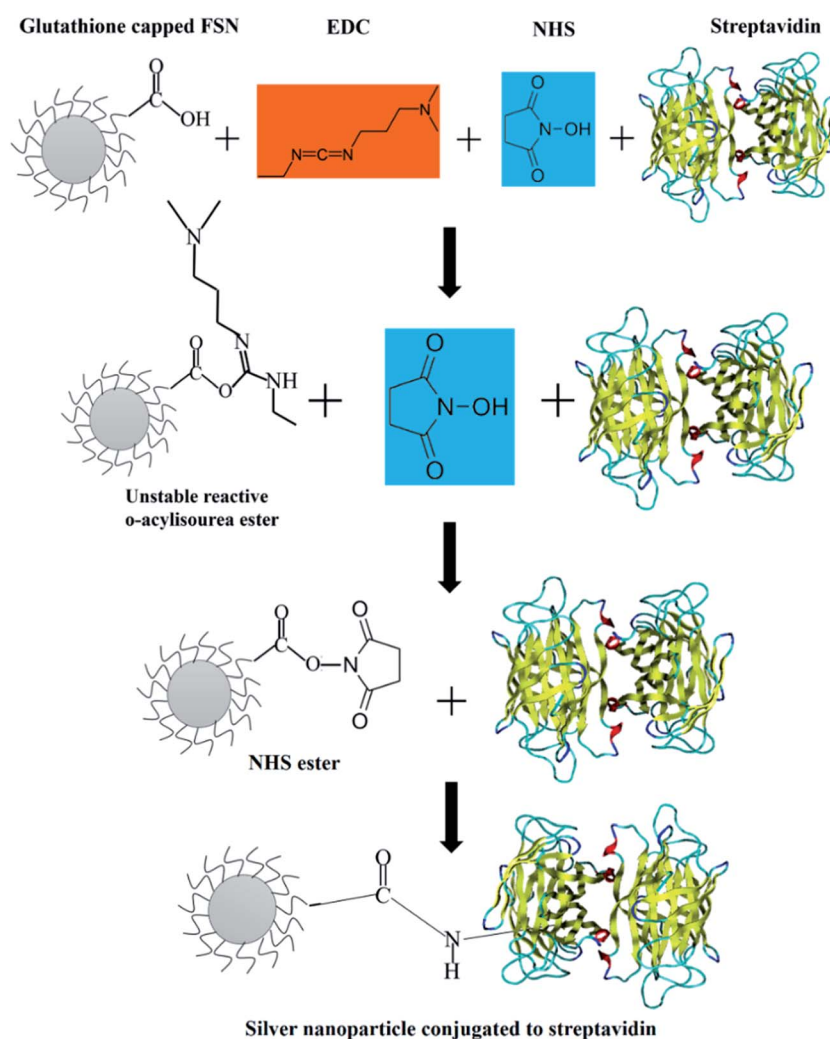


Fig. 3 (a) FTIR spectrum of FSNPs (b) interaction between the silver core and glutathione molecules.



Scheme 2 Schematic illustration of conjugation of FSNPs to streptavidin through EDC-NHS.



intermediate.<sup>61</sup> It also permits bioconjugation to primary amines at physiological pH. The coupling reaction was carried out in PBS whose pH is 7.4, which is compatible with the bioconjugation chemistry.<sup>62</sup>

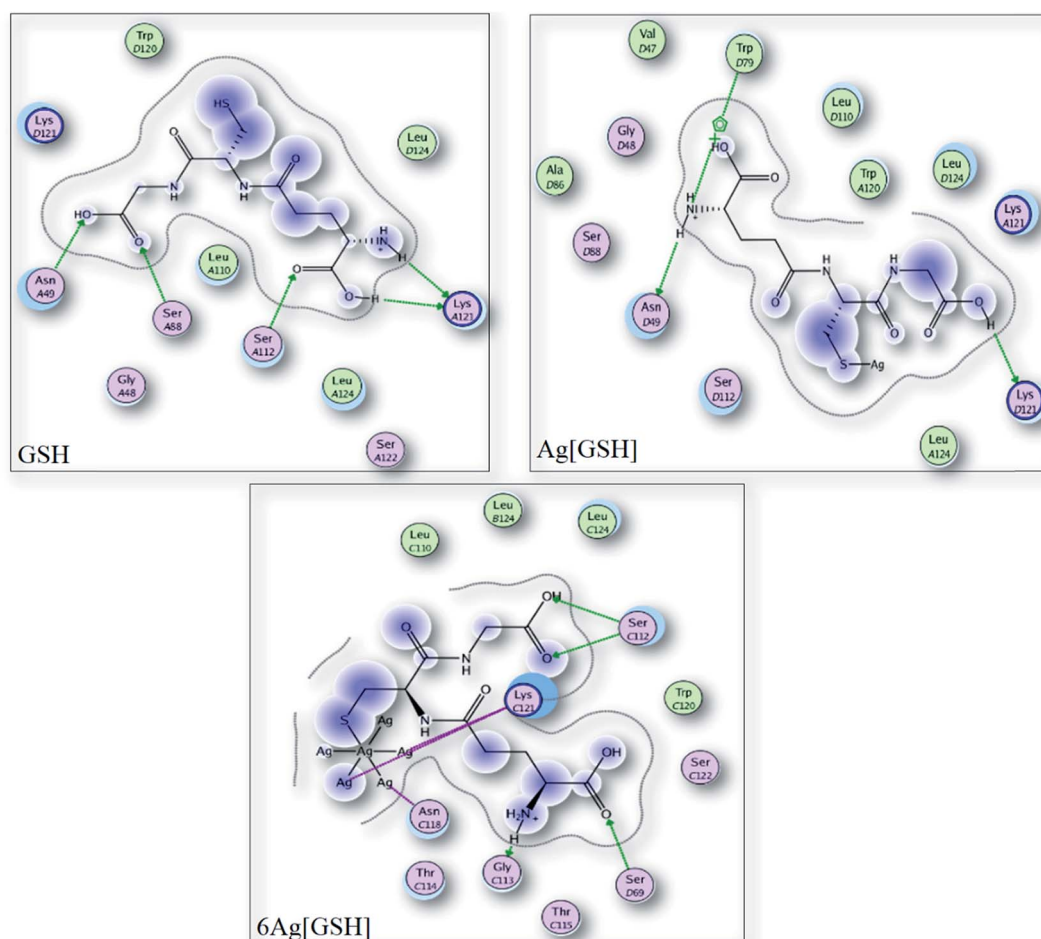
A schematic representation of the conjugation protocol is presented in Scheme 2. The first step of this reaction is the interaction between the glutathione from the FSNPS with EDC. It results in the formation of an unstable *o*-acylisourea intermediate. This intermediate reacts with NHS and forms an amine reactive NHS ester which is considerably more stable than the *o*-acylisourea intermediate. The final addition to this reaction mixture is the protein, streptavidin. The NH<sub>2</sub> from the amino acids in streptavidin reacts with the amine reactive NHS ester and results in formation of an amide bond between

carboxylic group of glutathione and amine group from streptavidin.

**3.3.1 Computational modelling of interaction between FSNPs with streptavidin.** The preliminary calculations on efficiency of conjugation are presented as Section S2† in ESM. Computational studies on bio-conjugation of the EDC–NHS coupled molecules to streptavidin in HEX has given significant understanding as to the nature of conjugation of all silver glutathione systems to the protein. All the ligand interactions were found to have a negative HEX score which indicated the favorable nature of binding for all ligands. From Table 1, it is clear that out of all the ligands, 31 Ag[GSH] has the best binding affinity. Interestingly, the binding affinity for 1 Ag[GSH] was found to be lower than glutathione which could be understood from the interaction profile which indicates that the number of interactions which the Ag[GSH] has is lesser than GSH interacting with the protein (Fig. 4). These interactions stabilize the ligands docking with the protein and thus we obtain a more negative Hex score. Similarly the interactions for 6 Ag[GSH] and 31 Ag[GSH] are even higher which clearly explains the more negative hex score and higher binding affinity of 31 Ag[GSH]. Thus, we see a trend of increasing conjugation efficiency with increasing number of particles.

**Table 1** Hex scores for different ligand systems binding to streptavidin

Molecule	Hex 8.0.0 score
GSH	−262.1
Ag[GSH]	−260
6 Ag[GSH]	−324.42
31 Ag[GSH]	−380.03



**Fig. 4** Interactions of (a) glutathione (b) 1 Ag-glutathione and (c) 6 Ag-glutathione with the streptavidin as determined by HEX docking.



**3.3.2 Computational modelling of conjugated streptavidin-biotin interaction.** The preliminary calculations have been presented in Section S3† in ESM. The detection step, which involves the streptavidin-biotin interaction was studied for all the Ag-GSH systems as to determine the effect of bio-conjugation on the interaction. Based on the biotin docking studies, the calculated binding affinities are presented in Table 2. While the most stable configuration is 1 Ag[GSH], the second most stable system is GSH followed by the 31 Ag[GSH] and next comes the 6 Ag[GSH]. The minimal difference between the HEX score indicates that the 31 atoms conjugated system can also interact with biotin with the same affinity as the other systems.

The slight destabilization of protein explains the difference in binding affinity. The tradeoff between the stabilization of the Ag-GSH (by interaction between Ag and the amino acids) with destabilization of protein due to the steric hindrance plays crucial role in this protein biotin interaction. While 31 atom system may sterically destabilize the protein but the silver amino acid interactions counteract the destabilization allowing a successful interaction of the conjugated protein with biotin. Thus, the highlight of this study is that the bioconjugation process still allows the protein to interact with biotin with significant binding affinity which is important for immunoassay applications.

### 3.4 Fluorescence polarization studies

The confirmation of conjugation was obtained through the fluorescence polarization studies which were performed on the Spectramax M5 plate reader in the fluorescence polarization mode. The correction factor ( $G$ ) of the plate reader was calculated to be 1. The results of the FP are depicted in the Table 3.

The higher value of FP of the streptavidin conjugated FSNPs in comparison to unconjugated FSNPs confirms the bio-conjugation of FSNPs to streptavidin. To understand this, we need to have a look at the correlation between fluorescence polarization and size of the fluorophores. The smaller size of unconjugated particles leads to their movement in solutions that results in depolarization of the incident plane polarized

excitation energy thus leading to very low polarization value.<sup>63,64</sup> However, the conjugation of FSNPs to streptavidin reduces the motion of the particle due to an increase in the size and volume of the particles. This reduces the rotation of particles which causes the excitation energy and the transmitted energy to be in the same state of polarization expressed as a higher magnitude of polarization value.<sup>65</sup> This manifests as a higher value of FP for the conjugated particle than that of the unconjugated particle.

### 3.5 Results of fluorescent silver nanoparticle immunoassay (FSNIA) for the detection of HIV-1 p24

HIV-1 p24 antigen is the target viral protein in this study because of its abundance in the early stages of HIV infection when antibody concentrations are low and not detectable. This in turn helps in early detection of the disease which can allow better management of the disease and antiretroviral treatment. In this study, we adapted p24 antigen to assess the performance of FSNIA. In order to test the sensitivity of the FSNIA, we decided to test its limits in detecting p24 antigen. The immunoassay involved the formation of antibody-antigen-antibody sandwich complex where in the detection step involved interaction between the biotinylated detector antibody and streptavidin conjugated FSNPs. The detection process makes use of the strong non-covalent chemistry between biotin and streptavidin which immobilizes the fluorophore to the sandwich complex. When illuminated by excitation energy, the fluorophore emits and the signal intensity is proportional to the amount of streptavidin conjugated FSNPs. The amount of streptavidin conjugated FSNPs are in turn proportional to amount of p24 present. Thus, the signal intensity is directly proportional to the concentration of p24 present in the sample. The fluorescence signal intensity is then measured using a SpectraMax M5 microplate reader. A calibration curve for the purified HIV-1 p24 antigen is plotted with the measured data.

**3.5.1 Calibration curve.** The FSNIA was found to work efficiently in detecting p24 in the linear working range of 10 pg mL<sup>-1</sup> to 1000 pg mL<sup>-1</sup> as observed in Fig. 5a and b. This dynamic range observed here is very useful for detecting p24 antigen concentrations within days after exposure upto 3 weeks, after which, due to seroconversion, the antigen becomes undetectable.<sup>66</sup> Thus, this range is very significant in the early detection of the infection. A linear relation between the concentration of p24 and signal intensity was observed for the whole range tested. A very clear difference was observed between the measured signal intensity and the blank intensity until 10 pg mL<sup>-1</sup> of the p24, lower concentration than this could not be distinguished from background. Thus, our FSNIA could detect p24 antigen reproducibly up to 10 pg mL<sup>-1</sup>.

**3.5.2 Analytical performance of FSNIA.** Statistically, for FSNIA the fluorescence cut-off value was obtained using the sum of the means of the signal intensity of negative controls plus 2 times their standard deviation (SD) and was found to be 33.283 RFUs.<sup>67</sup> The limit of detection was found out to be 8.2 pg mL<sup>-1</sup> for the dynamic range of 10–1000 pg mL<sup>-1</sup> for a signal to cutoff value of 2 : 1, which was chosen as the measure for

**Table 2** Hex scores for interaction of conjugated protein with biotin for different ligand systems

Binding moieties	HEX score
GSH-streptavidin-biotin	−224.5
Ag[GSH]-streptavidin-biotin	−233.3
6 Ag[GSH]-streptavidin-biotin	−221.1
31 Ag[GSH]-streptavidin-biotin	−223.6

**Table 3** Comparison of fluorescence polarization value of the unconjugated FSNPs with the conjugated FSNPs

Sample	$I_{  }$	$I_{\perp}$	$FP = \frac{I_{  } - G \times I_{\perp}}{I_{  } + G \times I_{\perp}}$
Unconjugated FSNPs	93	76	0.105
Streptavidin conjugated FSNPs	138	104	0.140





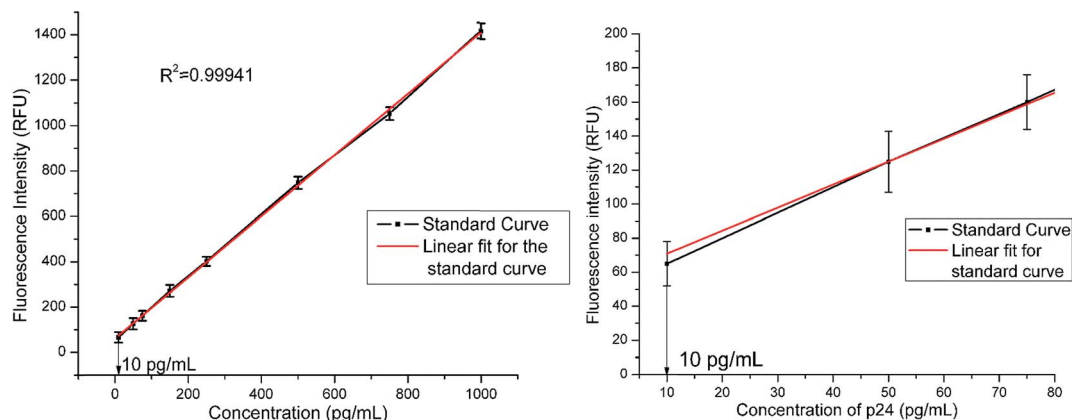


Fig. 5 (a) Plot of the calibration curve for FSNIA. The red line indicates the fitting curve obtained by linear regression. (b) Standard plot of FSNIA with resolved axis that points to the lower detection limit of FSNIA.

determining positive samples.<sup>68</sup> The equation of linear correlation was determined to be

$$FI = 1.34 \times C + 55.38 \quad (3)$$

where FI is fluorescence intensity,  $C$  is the concentration of p24.

We obtained very good linear correlation in eqn (3) between the concentrations of p24 and the fluorescence signal intensity in FSNIA with a high correlation coefficient of  $R^2 = 0.99941$ . This also suggests that FSNIA is a linear dose response assay. In this paper, we report that this FSNIA could achieve an analytical sensitivity at the picogram level which is comparable to the sensitivity levels achieved with conventional colorimetric ELISA (Table 4).

### 3.6 Control experiment

**3.6.1 Effect of presence of an interfering protein.** The presence of various proteins and other biomolecules necessitated the study of interference from other proteins on the immunoassay.

With BSA as a model protein, it was observed that the BSA has no effect on the signal intensities in the immunoassay. Fig. 6 clearly illustrates that there is barely any alteration in the signal strength with increasing BSA concentration. This further reinforces the stability of the immunoassay protocol.

**3.6.2 Effect of coexistent viruses.** To further analyze the extent of specificity and cross reactivity of FSNIA, HCV positive plasma samples were spiked with HIV-1 p24 antigen, a fixed concentration of  $500 \text{ pg mL}^{-1}$ . The results as depicted in Fig. 7 indicate that there is no perceivable effect on the fluorescence signal from the HCV virions suggesting that FSNIA is an exceptionally specific and stable testing protocol.

### 3.7 Optimization of FSNIA

The optimization of immunoassay was performed by optimizing two critical parameters in the immunoassay: capture antibody concentration and FSN-SA concentration.

#### 3.7.1 Capture antibody

**3.7.1.1 Optimization of the capture antibody concentration.** A very important parameter that affects the sensitivity of the immunoassay is the concentration of capture antibodies. There is an optimal concentration of capture antibody at

Table 4 Comparison of sensitivities of detection techniques based on application of silver nanoparticles used for sensing various analytes

Method	Analyte	Concentration ( $\text{pg mL}^{-1}$ )	Ref.
Conventional ELISA	p24 antigen	10–15	70
EIS	Arsenic	13	71
CLMI	Cysteine	25	72
Multiaarray-LSPR	Proteins	100	73
Colorimetric	Cancer biomarker cyclin A <sub>2</sub>	30	74
SPRI	IgG	7	75
RLS	Alpha fetoprotein	40	76
Lab on chip	Staphylococcal enterotoxin B	100	77
Lab on disc	Anti-hepatitis B virus	510	78
SERS	DNA	50	79
FSNIA	HIV-1 p24	10	Present study

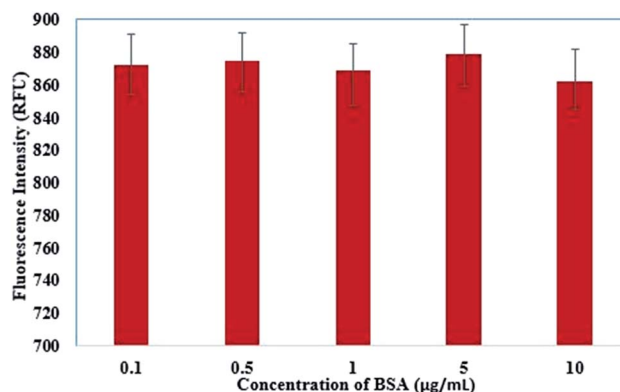


Fig. 6 Plot of fluorescence intensities for  $500 \text{ pg mL}^{-1}$  p24 antigen in samples with varying concentration of BSA.



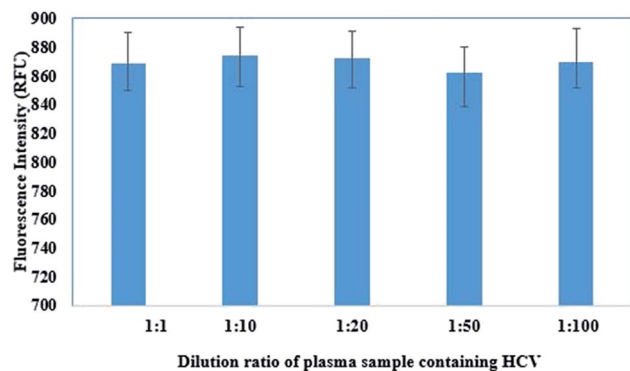


Fig. 7 Effect of dilution of HIV-1 p24 spiked HCV sample on fluorescence intensity from FSNIA.

which the signal will be highest; at higher concentrations the background signal can also be higher and concentrations below this, the signal will be weaker.<sup>69</sup> Optimization of concentration of capture antibody required for the assay was done by testing different concentrations of  $0.5 \mu\text{g mL}^{-1}$ ,  $0.8 \mu\text{g mL}^{-1}$ ,  $1 \mu\text{g mL}^{-1}$  and  $2 \mu\text{g mL}^{-1}$ . As shown in Fig. 8, for  $0.5 \mu\text{g mL}^{-1}$  and  $0.8 \mu\text{g mL}^{-1}$  the signal response was poor and there was no good correlation between the observed signal and antibody concentration. However, at  $1 \mu\text{g mL}^{-1}$ , there was a strong response observed. For concentration higher than  $1 \mu\text{g mL}^{-1}$ , there was no change in the response which can be attributed to the saturation of the detection area with capture antibodies at  $1 \mu\text{g mL}^{-1}$ . As a result,  $1 \mu\text{g mL}^{-1}$  was chosen as the optimal concentration of capture antibody for further experimentation.

**3.7.2 FSN-SA.** The streptavidin conjugated FSNPs are the fluorescent particles which are used in this detection process. The streptavidin-biotin interaction allows FSNPs to indirectly interact with the detector antibody and when the FSNPs are excited with the excitation signal, the intensity of the output

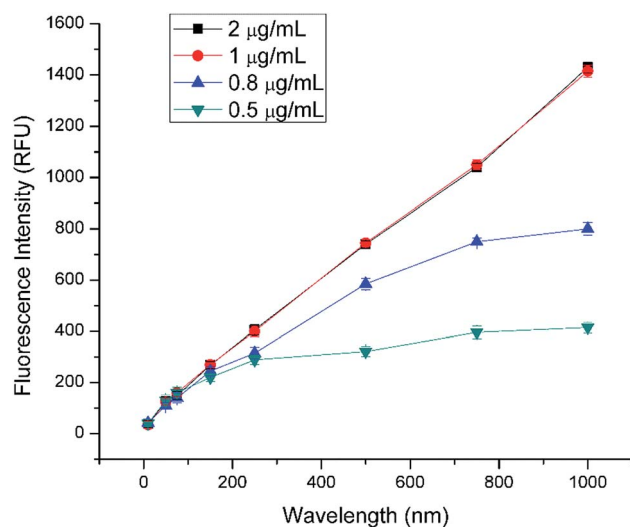


Fig. 8 Effect of varying concentration of capture antibodies on fluorescence intensities in FSNIA.

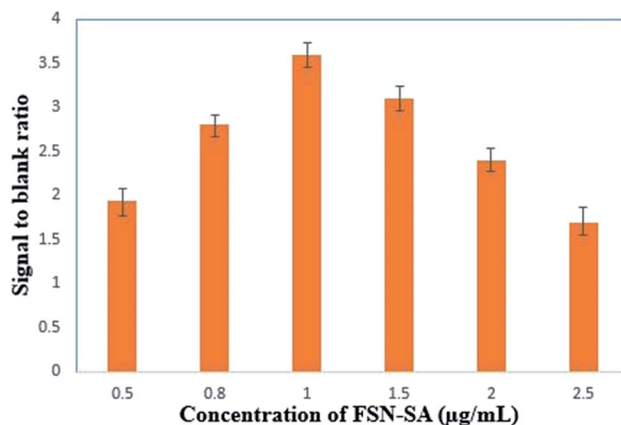


Fig. 9 Effect of varying concentration of FSN-SA on signal to blank ratio of fixed concentration of  $250 \text{ pg mL}^{-1}$  of p24 in FSNIA.

(fluorescence) signal is an indicator of the amount of antigen present. The signal was read with a spectrophotometer. The linear correlation between the fluorescence intensity and the antigen concentration allows us to plot the calibration curve.

**3.7.2.1 Optimization of the streptavidin conjugated FSNPs concentration.** Concentration of FSN-SA greatly affects the performance of the assay. Six concentrations of FSN-SA:  $0.5 \text{ mg mL}^{-1}$ ,  $0.8 \text{ mg mL}^{-1}$ ,  $1 \text{ mg mL}^{-1}$ ,  $1.5 \text{ mg mL}^{-1}$ ,  $2 \text{ mg mL}^{-1}$ ,  $2.5 \text{ mg mL}^{-1}$ , were taken to study its effect on the signal to blank ratio (S/B ratio) in FSNIA. The blank measurements were obtained by performing the immunoassay in absence of the antigen. The optimization of FSN-SA was carried out by measuring and calculating the S/B ratio for concentration of  $250 \text{ pg mL}^{-1}$  of p24 antigen. It is clear from Fig. 9 that no linear relation exists between the FSN-SA concentration and the S/B ratio. At a concentration  $1.5 \text{ mg mL}^{-1}$  of FSN-SA the highest S/B ratio was observed. With an increase in FSN-SA concentration beyond the optimal concentration the blank intensity was also observed to have increased rapidly when compared to the signal intensity. This increase of blank intensity could be attributed to the non-specific absorption of FSN-SA.

### 3.8 Precision and recovery rate of the assay

Our next objective was to study the applicability of FSNIA in real world scenario by testing human plasma samples. The first step

Table 5 The recovery percentages calculated for each of the concentration

S. no	p24 spiked ( $\text{pg mL}^{-1}$ )	p24 detected ( $\text{pg mL}^{-1}$ )	Recovery (%)
1	10	$9.2 \pm 0.5$	$92.0 \pm 5.0$
2	50	$47.6 \pm 2.1$	$95.2 \pm 4.2$
3	75	$71.3 \pm 4.5$	$95.0 \pm 4.3$
4	150	$145.1 \pm 5.8$	$96.7 \pm 3.9$
5	250	$242.7 \pm 6.5$	$97.1 \pm 2.7$
6	500	$482.3 \pm 15.9$	$96.4 \pm 3.3$
7	750	$740.0 \pm 19.6$	$98.6 \pm 2.8$
8	1000	$989.2 \pm 39.4$	$98.9 \pm 3.9$



was to determine the sensitivity of the assay which was done by testing 50 plasma samples of healthy individuals. All the samples tested negative and no false positives were observed. The next step was to further evaluate the analytical sensitivity of the assay and this was done by taking known concentrations of p24 antigen and spiking the HIV negative plasma samples collected from healthy individuals (HIV negative sample). The plasma samples were diluted 100 times which in turn reduced the sample amount required for analysis. To the diluted plasma sample, purified p24 antigen concentrations of 10, 50, 150, 250 and 500  $\text{pg mL}^{-1}$  were added. All experiments were conducted in triplicate and the concentration was calculated by taking the average signal response and comparing it with the standard curve (Fig. 5a). It is clear from Table 5 that the concentrations of

p24 antigen calculated from the results of FSNIA are comparable to the actual amounts spiked into sample. Thus FSNIA can quantitatively determine the concentration of HIV-1 p24 antigen in clinical plasma samples.

### 3.9 Clinical serum sample analysis

To further establish the effectiveness of FSNIA, we tested the specificity of FSNIA using samples from infected individuals (HIV positive samples). Plasma samples were diluted 100 times prior to the analysis. We tested 50 HIV positive samples (which were confirmed by 3rd and 4th generation commercial ELISA kits) using FSNIA and no false negatives were observed. To test FSNIA for cross reactivity we also evaluated the assay with 10 HBV positive/HIV negative and 10 HCV positive/HIV negative

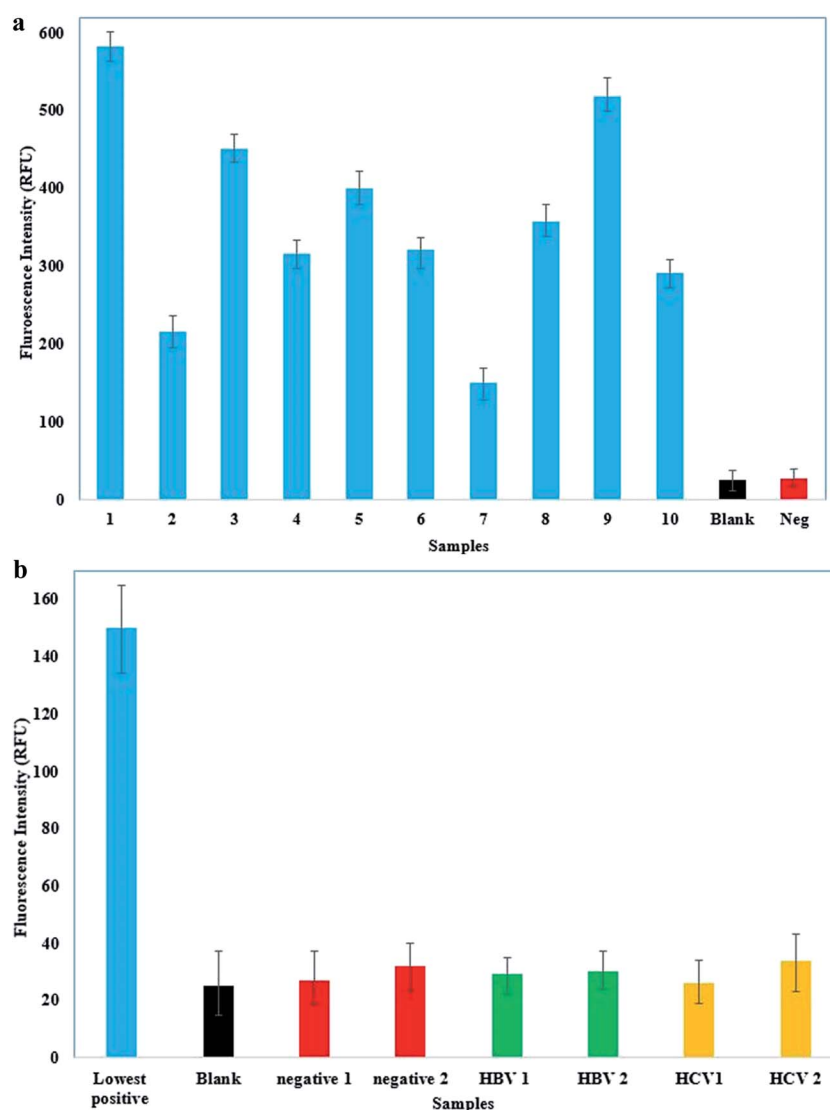


Fig. 10 (a) Results of FSNIA for 10 HIV positive samples chosen at random compared with the results from blank and the intensity of a HIV-negative sample. The blue, black and red bars indicate the intensity of the positive, blank and the highest measured negative sample, respectively. (b) Results of FSNIA for 2 randomly chosen HIV negative, HBV positive/HIV negative and HCV positive/HIV negative samples respectively in comparison with the intensity of the lowest positive tested sample and blank. The blue and black bars indicate the intensity of the HIV positive sample and the blank, while the red, green and yellow indicate HIV negative, HBV positive/HIV negative and HCV positive/HIV negative samples respectively.

**Table 6** Results of clinical serum analysis which show the high specificity of FSNIA

S. no	Sample type	No. of samples	No. of samples tested to be HIV positive	No. of samples tested to be HIV negative
1	HIV positive	50	50	NIL
2	HIV negative	50	NIL	50
3	HIV negative/HBV positive	10	NIL	10
4	HIV negative/HCV positive	10	NIL	10

plasma samples for cross reactivity. We observed that the signal intensity from these samples were as low as the negative control. The comparison of the signals from different plasma samples is presented in Fig. 10a and b. Thus, we could confirm that the assay is very specific to p24 antigen and is not prone to cross reactivity effects from other biomolecules. As clearly evident from the Table 6 the FSNIA has very high specificity and low cross reactivity.

Thus, we have successfully demonstrated the detection of picogram level concentration of HIV-1 p24 antigen using FSNIA. Our findings confirm that fluorescent nanoparticle based immunoassays could dramatically increase detection sensitivity and have applications in improved diagnosis and point-of-care use. Due to high specificity of streptavidin conjugated silver nanoparticles which have been confirmed by the *in silico* studies, FSNIA could be developed into a universal labeling technology by replacing capture and detection antibodies for detection of various other antigens. There is no need for specific training of the technicians to handle FSNIA as it is similar to traditional ELISA which is widely practiced across various laboratories and clinics. Upon further optimization and simplification, this could be developed into a rapid and ultrasensitive testing platform for clinical diagnosis and laboratory research in resource limited settings.

## 4 Conclusion

Our work has shown the use of fluorescent silver nanoparticles in infectious disease assay development can result in enhanced assay sensitivity. Although there are similar assays, this assay is capable of detecting HIV p24 antigen in the  $\text{pg mL}^{-1}$  range which will be useful in detecting early HIV infection. The findings described in this paper are important for those developing highly sensitive assays for the estimation of HIV incidence and detection of acute infection cases as part of the strategy towards an AIDS-free generation. This highly sensitive p24 assay can also be adapted to microfluidic platforms and be made available as point of care technology which can help improve blood safety by reducing the antibody negative window period in blood donors in resource limited settings where nucleic acid testing is not practical or feasible.

## Disclaimer

The findings and conclusions in this report are those of the authors and do not necessarily represent the views of the Food

and Drug Administration, U.S. Department of Health and Human Services.

## Statement of informed consent

The authors declare that all the experiments that were performed are in compliance with the guidelines of the Institutional Ethics Committee (IEC) of Sri Sathya Sai Institute of Higher Learning, Prasanthi Nilayam. The study was approved by the IEC under the study id SSSIHL/IEC/PSN/BS/2012/01. The plasma samples used in the study were obtained from the patients with their informed consent.

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