Design, synthesis, and biological evaluation of oxazolidone derivatives as highly potent N-acylethanolamine acid amidase (NAAA) inhibitors†

Jie Ren,a Yuhang Li,*ab Hongwei Ke,c Yanting Li,a Longhe Yang,d Helin Yu,a Rui Huang,a Canzhong Lu and Yan Qiu*a

N-Acylethanolamine-hydrolyzing acid amidase (NAAA) is a lysosomal enzyme that catalyzes the hydrolysis of endogenous fatty acid ethanolamides (FAEs), such as N-palmitoylethanolamide (PEA). PEA exhibits anti-inflammatory and analgesic activities by engaging peroxisome proliferator-activated receptor α (PPAR-α). Preventing PEA degradation by inhibition of NAAA has been proposed as a novel strategy for the treatment of inflammation and pain. In the present study, we reported the discovery of the oxazolidone derivative as a novel scaffold for NAAA inhibitors, and studied the structure–activity relationship (SAR) by modification of the side chain and terminal lipophilic substituents. The results showed that the link chain length of C5, straight and saturated linkages were the preferred shape patterns for NAAA inhibition. Several nanomolar NAAA inhibitors were described, including 2f, 3h, 3i and 3j with IC_{50} values of 270 nM, 150 nM, 100 nM and 190 nM, respectively. Enzymatic degradation studies suggested that 2f inhibited NAAA in a selective, noncompetitive and reversible pattern. Moreover, 2f showed high anti-inflammatory and analgesic activities after systemic and oral administration.

1 Introduction

N-Acylethanolamine-hydrolyzing acid amidase (NAAA) is a lysosomal cysteine hydrolase widely expressed in many tissues, especially those associated with immune responses, such as the lungs, spleen, and small intestine. NAAA participates in the control of inflammatory and pain via involving in the hydrolysis of endogenous palmitoylethanolamide (PEA). PEA is a multifunctional lipid mediator that has been shown to inhibit peripheral inflammation and cell degranulation, and exhibits anti-inflammatory and nociceptive activities in rat and mouse models of acute and chronic pain. Moreover, PEA also could alleviate skin inflammation and neuropathic pain in humans. Those effects are mainly attributed to the ability of PEA to engage nuclear receptor peroxisome proliferator-activated receptor-alpha (PPAR-α). In recent years, PEA has attracted increasing attention because of its potential therapeutic effects in inflammation and pain. Preventing PEA degradation by inhibition of NAAA may provide a novel strategy for the treatment of inflammation and pain. For this reason, NAAA has been pursued as a potential drug target.

Despite the increasing interest in the design of NAAA inhibitors as anti-inflammatory and analgesic agents, only few potent NAAA inhibitors have been identified so far. There are two main categories of NAAA inhibitors reported, irreversible covalent inhibitors, such as β-lactone derivatives and non-covalent reversible inhibitors, such as benzothiazole-piperazine derivatives. Although covalent inhibitors, such as 1b (ref. 20), ARN077 and ARN726 have demonstrated promising anti-inflammatory effects in several animal models, their application is limited due to the poor chemical and plasma stability. These compounds degraded in phosphate buffered saline (PBS) within 120 min and were rapidly cleaved in plasma. Non-covalent reversible inhibitors, such as, benzothiazole-piperazine derivatives (Fig. 1) are selective, reversible, and competitive NAAA inhibitors. However, only few potent reversible NAAA inhibitors with drug-like properties have been identified so far. Highly potent, stable and reversible NAAA inhibitors are still highly desired for the treatment of inflammation and pain.

In our previous study, we designed pyrrolidine-based derivatives 1a and 1b (Fig. 1) as NAAA inhibitors. Substitution of the saturated alkyl chain of 1a with an aryl-containing lipophilic moiety led to a potent NAAA inhibitor 1b (IC_{50} = 2.1 μM). In vivo pharmacological studies indicated that 1b reduced the mRNA expression levels of inducible nitric oxide synthase and...
Compounds 1a, 1j−1o, and 1q−1t were prepared by the amidation of palmitoyl chloride with the corresponding amine in the presence of triethylamine (Scheme 1).21 1l and 1m were prepared by the methylation of 1q and 1r with iodomethane in the presence of sodium hydride, respectively (Scheme 2).23 1p was prepared by Dess–Martin periodinane oxidation of 1j or 1k (Scheme 3).24 1e and 1d were prepared by the acidic deprotection of pyrrolidine derivatives 1c−1i and 1d−1l, which were obtained by the amidation of (R)- or (S)-3-(tert-butylcarbonylaminoperyl)pyrrolidine with palmitoyl chloride, respectively (Scheme 4).25,26 1e and 1f were synthesized by a two-step reaction starting from 1q and 1r, which were converted to iodide derivatives by I2 and PPh3, respectively, followed by hydrogenation with H2 (Scheme 5).27,28 1g, 2a−2b, 3a−3d and 3k−3s were synthesized by the imidate reaction, employing the corresponding acyl chloride with 2-pyrrolidone, 2-imidazolidone, or 2-oxazolidone in the presence of n-BuLi, respectively (Scheme 6).29 3e−3g were synthesized by the amidation of 2-oxooxazolidine-3-carbonyl chloride, generated by reaction of 2-oxazolidone and bis[chloromethyl] carbonate, with the corresponding substituted piperazine or piperidine (Scheme 7).30 HPLC analysis indicated >99% purity for compound 1g, 1e−1i, 2b, 2d, 2f−2h, 3c, 3e, and 3g; >98% purity for 1c−1d, 1f, 2a, 2e, 3a, 3b and 3g−3i, and >97% purity for 2c and 3d (Table S3†).

### 2 Results and discussion

#### 2.1 Synthesis of oxazolidone derivatives

First, we synthesized a series of oxazolidone derivatives with different side chains and terminal lipophilic substituents.

![Scheme 1](image1.png)

**Scheme 1** The synthesis of compounds 1a, 1j, 1k, 1n, 1o, and 1q−1t. Reagents and conditions: (a) Et3N, CH2Cl2, room temperature.

![Scheme 2](image2.png)

**Scheme 2** The synthesis of compounds 1l and 1m. Reagents and conditions: (a) NaH, MeI, tetrahydrofuran (THF), 0 °C.

interleukin-6, as well as increased the intracellular PEA levels in mouse macrophages with lipopolysaccharide-induced inflammation.21 Moreover, 1b exhibited excellent chemical stability and plasma stability ($t_1 > 16$ h).

In the present study, we further modified 1a to obtain more potent and biologically stable NAAA inhibitors. A nanomolar NAAA inhibitor 2f was discovered ($IC_{50} = 270$ nM), which exerted profound anti-inflammatory and anti-nociceptive effects in rats.22 We then carried out structure–activity relationship (SAR) studies on 2f by modification of the pyrrolidine ring and the lipophilic chain, which led to the identification of several more potent nanomolar NAAA inhibitors, such as 3h, 3i and 3j (Fig. 2). Compound 2f was then selected as a model molecule to study the NAAA inhibition mechanisms and in vivo anti-inflammatory and analgesic activities of oxazolidone derivatives. The results showed that 2f inhibited NAAA in a selective and highly reversible pattern, and efficiently reduced inflammatory and pain after systemic and oral administration. Oral administration of 2f (30 mg kg$^{-1}$) exhibited the same efficacy with commercial nonsteroidal anti-inflammatory drug ibuprofen at a 6.7-fold higher dose (200 mg kg$^{-1}$).

### 2.2 NAAA inhibition by oxazolidone derivatives

The NAAA inhibition capability of these newly synthesized compounds were tested using heptadecenoylethanolamide as a substrate. Since many NAAA inhibitors also block fatty acid amide hydrolase (FAAH), the FAAH inhibition of these compounds also was examined to obtain NAAA selective inhibitors. (S)-OOPP and well known FAAH inhibitor URB597 were used as standard compounds in the enzymatic assay. The results showed that substitution at 3-position of pyrrolidine with an amino group (1c, 1d) slightly increased NAAA and FAAH...
inhibition, and substitutions at the 2-position with (S)-methyl (1f) or carbonyl (1g) slightly or moderately enhanced inhibition towards NAAA without changing inhibition against FAAH. However, (R)-methyl (1e), hydroxy (1j, 1k), methoxy (1l, 1m), cyano (1n), formamide (1o), hydroxymethyl (1q, 1r) and methoxymethyl (1s, 1t) substituents diminished the NAAA and FAAH inhibitory effects (Table S1†). 1g exhibited low-micromolar inhibition on NAAA activity (IC50 = 4.6 ± 0.48 μM), and it was over 5-fold more potent towards NAAA than FAAH (Table 1). To further optimize the hydrophilic moieties, a series of 1g isosteres were synthesized and tested (1h, 1i) (Table 2). Enzymatic assay results showed that 1i with oxazolidone ring had similar effects on NAAA inhibition as 1g, but 1i exhibited dramatically enhanced selectivity towards NAAA. Less than 10% FAAH activity was inhibited by 1i even at a concentration of 100 μM. Therefore, 2-oxazolidone ring was selected as the template for further optimization.

It has been reported that substitutions of the saturated alkyl chain with aryl-containing lipophilic moieties could enhance the inhibitory potency on NAAA and improved the drug-like properties.14-21 Therefore, a series of oxazolidone derivatives incorporating a phenyl ring at the terminus were synthesized and tested. As shown in Table 3, 2a with the chain length of C0 exhibited low-micromolar inhibitory potency toward NAAA (IC50 = 3.4 ± 0.42 μM), while 2f with the chain length of C5 exhibited excellent potency (IC50 = 0.27 ± 0.04 μM), 3-fold more potent than (S)-OOPP tested under the same experimental conditions. Interestingly, sharply reduced or even abolished activity was observed as the chain length decreased from C5 to C1 or increased from C5 to C7. We are not quite sure why their potency dramatically changed when we varied the link chain length, probably because these oxazolidone imides with different chain length are activated to different extent when contacting the NAAA.

To further optimize the NAAA inhibition activity of 2f, we replaced the rotatable bond of the acyl chain of 2f with an additional phenyl group. As shown in Table 4 and S2,† a series of analogs corresponding to biphenyl (3a, 3k–3m), phenoxyphenyl...
(3b, 3n–3p), and benzyloxyphenyl (3c, 3q–3s) were obtained and examined. Compared to 2a, 1,4-substituted phenoxypthyl (3b, IC50 = 0.68 ± 0.11 μM) and benzyloxyphenyl (3c, IC50 = 2.3 ± 0.36 μM) analogs were approximately 10-fold and 2-fold more potent. However, 1,4-substituted biphenyl analog 3a (IC50 = 9.4 ± 1.2 μM) was 2-fold less potent (Table 4), and 1,3-substituted isomers (3k, 3l, 3n, 3o, 3q, 3r) were completely inert for NAAA inhibition (Table S2†). The results indicated that straight structures are preferred shape for the hydrophobic channel of NAAA proximal to the carbonyl moieties. Therefore, incorporation of an extra methylene group between the biaryl and carbonyl groups (3m, 3p and 3s) significantly decreased NAAA inhibition.

It has been reported that NAAA prefers saturated substrates e.g., N-myristoylthanolamine and PEA, rather than unsaturated substrates, e.g., AEA. Therefore, we studied whether the replacement of a phenyl ring with a 6-member ring can improve the NAAA inhibition. A series of additional derivatives of 2f were then synthesized and examined (Table 5). The results showed that introducing hexane rings further enhanced the NAAA inhibitory potency. We found two highly potent NAAA inhibitors 3h and 3i with a IC50 value of 150 nM and 100 nM, respectively, around 2-fold more potent than 2f. Moreover, saturated linkages were found to be more beneficial for the activity than unsaturated benzene rings (3h, 3i vs. 3a–3c; 3j vs. 2f). However, nitrogen atoms substitution of the saturated hexane significantly reduced the activity (3e–3g).

2.3 NAAA inhibition pattern

To further understand the NAAA inhibition pattern of these oxazolidone derivatives, compound 2f was selected as a model for further study. It has been reported that NAAA prefers saturated substrates e.g., N-myristoylthanolamine and PEA, rather than unsaturated substrates, e.g., AEA. Therefore, we studied whether the replacement of a phenyl ring with a 6-member ring can improve the NAAA inhibition. A series of additional derivatives of 2f were then synthesized and examined (Table 5). The results showed that introducing hexane rings further enhanced the NAAA inhibitory potency. We found two highly potent NAAA inhibitors 3h and 3i with a IC50 value of 150 nM and 100 nM, respectively, around 2-fold more potent than 2f. Moreover, saturated linkages were found to be more beneficial for the activity than unsaturated benzene rings (3h, 3i vs. 3a–3c; 3j vs. 2f). However, nitrogen atoms substitution of the saturated hexane significantly reduced the activity (3e–3g).

Table 1 Inhibitory effects of compounds 1a–1g, (S)-OOPP and URBS97 on NAAA and FAAH activities

<table>
<thead>
<tr>
<th>Compound</th>
<th>R2</th>
<th>R3</th>
<th>IC50 of NAAA (μM)</th>
<th>IC50 of FAAH (μM)</th>
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<tr>
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<td>25 ± 5.7</td>
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<tr>
<td>1c</td>
<td>–H</td>
<td>(R)-NH2</td>
<td>10 ± 0.95</td>
<td>3.5 ± 0.65</td>
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<tr>
<td>1d</td>
<td>–H</td>
<td>(S)-NH2</td>
<td>11 ± 1.2</td>
<td>4.0 ± 0.81</td>
</tr>
<tr>
<td>1e</td>
<td>(R)-Me</td>
<td>–H</td>
<td>&gt;100</td>
<td>28 ± 2.7</td>
</tr>
<tr>
<td>1f</td>
<td>(S)-Me</td>
<td>–H</td>
<td>15 ± 2.3</td>
<td>18 ± 3.2</td>
</tr>
<tr>
<td>1g</td>
<td>==O</td>
<td>–H</td>
<td>4.6 ± 0.48</td>
<td>25 ± 5.3</td>
</tr>
<tr>
<td>(S)-OOPP</td>
<td></td>
<td></td>
<td>0.83 ± 0.1</td>
<td>&gt;20</td>
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<tr>
<td>URBS97</td>
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<td>0.026 ± 0.005</td>
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Table 2 Inhibitory effects of compounds 1h–1j on NAAA and FAAH activities

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<th>IC50 of FAAH (μM)</th>
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<td>1g</td>
<td>C</td>
<td>4.6 ± 0.48</td>
<td>25 ± 5.3</td>
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<tr>
<td>1h</td>
<td>NH</td>
<td>&gt;100</td>
<td>27 ± 4.9</td>
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<td>1i</td>
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Table 3 Inhibitory effects of compounds 2a–2h on NAAA and FAAH activities

<table>
<thead>
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<th>Compound</th>
<th>n</th>
<th>IC50 of NAAA (μM)</th>
<th>IC50 of FAAH (μM)</th>
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<tr>
<td>1a</td>
<td>0</td>
<td>3.4 ± 0.42</td>
<td>&gt;20</td>
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<tr>
<td>2b</td>
<td>1</td>
<td>&gt;20</td>
<td>&gt;20</td>
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<tr>
<td>2c</td>
<td>2</td>
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<td>&gt;20</td>
</tr>
<tr>
<td>2d</td>
<td>3</td>
<td>12.4 ± 2.0</td>
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</tr>
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<td>2e</td>
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<tr>
<td>2f</td>
<td>5</td>
<td>0.27 ± 0.04</td>
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</tr>
<tr>
<td>2g</td>
<td>6</td>
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</tr>
<tr>
<td>2h</td>
<td>7</td>
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<td>&gt;20</td>
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Table 4 Inhibitory effects of compounds 3a–3d on NAAA and FAAH activities

<table>
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<tr>
<th>Compound</th>
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<th>IC50 of NAAA (μM)</th>
<th>IC50 of FAAH (μM)</th>
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<td>4-Ph</td>
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</tr>
<tr>
<td>3b</td>
<td>0</td>
<td>4-OPh</td>
<td>0.68 ± 0.11</td>
<td>&gt;20</td>
</tr>
<tr>
<td>3c</td>
<td>0</td>
<td>4-OBn</td>
<td>2.4 ± 0.36</td>
<td>&gt;20</td>
</tr>
<tr>
<td>3d</td>
<td>0</td>
<td>4-OCH2CH2Ph</td>
<td>&gt;20</td>
<td>&gt;20</td>
</tr>
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</table>
molecule and its interaction mode with NAAA protein in cells was studied. The results showed that 2f displayed a profound inhibition toward recombinant human NAAA (IC\textsubscript{50} = 0.27 ± 0.04 μM, Fig. S1A†) and recombinant rat NAAA in NAAA-transfected HEK293 cells (IC\textsubscript{50} = 0.42 ± 0.04 μM, Fig. S1B†), which were similar to the data obtained with the NAAA protein extract. Dialysis (Fig. 3B) and rapid dilution (Fig. 3C) of the 2f–NAAA interaction complex almost completely restored the NAAA activity. Moreover, Michaelis–Menten kinetic analysis revealed that 2f induced changes in the maximal catalytic velocity (V\textsubscript{max}) of NAAA (V\textsubscript{max} in nmol per min per mg protein, vehicle: 6.0 ± 0.51; and 2f, 0.3 μM: 1.9 ± 0.15, Fig. 3D) but did not affect Michaelis–Menten constant K\textsubscript{m} (K\textsubscript{m} in μM, vehicle: 174 ± 26; and 2f, 0.3 μM: 212 ± 24). These results indicated that 2f is a reversible and noncompetitive NAAA inhibitor.

### Table 5

Inhibitory effects of compounds 3e–3j on NAAA and FAAH activities

<table>
<thead>
<tr>
<th>Compound</th>
<th>( R )</th>
<th>IC\textsubscript{50} of NAAA (μM)</th>
<th>IC\textsubscript{50} of FAAH (μM)</th>
</tr>
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<tbody>
<tr>
<td>3e</td>
<td>-</td>
<td>&gt;20</td>
<td>&gt;20</td>
</tr>
<tr>
<td>3f</td>
<td>-</td>
<td>&gt;20</td>
<td>15.3 ± 2.4</td>
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<tr>
<td>3g</td>
<td>-</td>
<td>&gt;20</td>
<td>21.2 ± 4.2</td>
</tr>
<tr>
<td>3h</td>
<td>-</td>
<td>0.15 ± 0.03</td>
<td>&gt;20</td>
</tr>
<tr>
<td>3i</td>
<td>-</td>
<td>0.10 ± 0.03</td>
<td>&gt;20</td>
</tr>
<tr>
<td>3j</td>
<td>-</td>
<td>0.19 ± 0.02</td>
<td>&gt;20</td>
</tr>
</tbody>
</table>

2.4 *In vivo* anti-inflammatory and analgesic activities of 2f

Finally, the *in vivo* anti-inflammatory and analgesic activities of 2f was studied in different pain and inflammation mice models. Experimental mice were treated with 2f via intraperitoneal injection or intragastrical administration. The behavioural response of mice to formalin stimulation was divided to the first-phase of acute pain (0–10 min) and the second-phase of chronic pain (15–35 min). Single treatment of 2f (10 mg kg\textsuperscript{-1}, i.p.) decreased second-phase but not first-phase (Fig. 4), suggesting a central mechanism of action of 2f. The reduced licking time by 2f at the second-phase was reversed by PPAR-\( \alpha \) antagonist MK886 (2 mg kg\textsuperscript{-1}, i.p.), which indicated that 2f relieve pain in a PPAR-\( \alpha \) dependent pathway.

The anti-inflammatory activity of 2f was tested in 12-O-tetradecanoylphorbol-13-acetate (TPA) induced mice ear edema model. A commercial nonsteroidal anti-inflammatory drug, ibuprofen (200 mg kg\textsuperscript{-1}) was intragastrically administered as a positive control. The results showed that intragastrical administration of 2f (30 mg kg\textsuperscript{-1}) significantly reduce ear edema. Oral administration of compound 2f (30 mg kg\textsuperscript{-1}) presented the same efficacy with ibuprofen (200 mg kg\textsuperscript{-1}). Moreover, 2f significantly prevent the production of proinflammatory cytokines, including TNF-\( \alpha \), IL-6, COX-2 and NF-\( \kappa \)B in mRNA level (Fig. 5).

For many nonsteroidal anti-inflammatory drugs, such as COX-2 inhibitors, adverse gastrointestinal effects and increased cardiovascular risks are two main limitations for their clinical application. We tested the toxicity of 2f by monitoring its influence on food intake, body weight and ether-a-go-go-related-gene (hERG) channel activity. Long-term studies suggested that intragastrical administration of 2f (30 mg kg\textsuperscript{-1} for 30 days, 100 mg kg\textsuperscript{-1} for 10 days) had no significant effects on food intake and body weight in mice (Fig. S2†). Mice stomach anatomy indicated that 2f did not cause gastrointestinal hemorrhages, while indomethacin (20 mg kg\textsuperscript{-1} for 3 days, ig) resulted in obvious hemorrhages (Fig. S3†). The low gastrointestinal side effects of 2f may be because it induces analgesia and anti-inflammatory via enhancement of PEA rather than reducing prostaglandins (PGs) in the gut. Moreover, the activity of hERG channels in CHO cells was only inhibited by 38.3% by
3 Conclusion

In conclusion, we report the discovery of oxazolidone derivatives as a novel scaffold of NAAA inhibitors. The NAAA inhibitory activity of oxazolidone derivatives was modulated by the side chain and terminal lipophilic substituents. The results showed that the link chain length of C5, straight and saturated linkages were the preferred shape patterns. Several nanomolar NAAA inhibitors (2f, 3b, 3h, 3i and 3j) were developed. Compound 2f exhibited highly potent and selective inhibitory activity towards NAAA via a competitive and reversible pattern. Moreover, in vivo study showed that 2f had significant anti-inflammatory and analgesic effects with low side effects. Our study demonstrated that oxazolidone derivatives is a class of potent NAAA inhibitors holds great promise for clinical translation. We anticipate that these oxazolidone derivatives will help better understand the function of NAAA and facilitate the development of improved therapies for inflammation and pain.

4 Materials and methods

4.1 Materials and instruments

All reagents used in the present study were purchased from Sigma-Aldrich (Shanghai, China) unless otherwise indicated. Tetrahydrofuran was distilled prior to use from sodium benzenophenone ketyl. Dichloromethane (CH2Cl2) was distilled from phosphorus pentoxide. Dimethylformamide (DMF) was distilled from calcium hydride. Silica gel (300–400 mesh) from Yantai Athy Chemical Technology Co. Ltd. (Zhifu, China) was used for column chromatography, and compounds were eluted with an ethyl acetate/petroleum ether (PE) (60–90 °C) mixture (unless otherwise stated).

The 1H-NMR and 13C-NMR spectra were recorded on a Bruker 400 spectrometer (400 MHz for 1H, 100 MHz for 13C), using CDCl3, d6-DMSO, or CD3OD as solvent and Me4Si as internal substance. IR spectra were recorded on a Nicolet Avatar 360 RT-IR spectrophotometer. Mass spectra were recorded on an Applied Bio systems MDS SCIEX 3200Q-TRAP mass spectrometry (MS) system with electro spray ionization (ESIϯ) and direct injection. The HRFABMS spectra were recorded on a Bruker APEX-FTMS apparatus. Elemental analysis was performed using a Vario RL analyzer. Melting points were determined on a Yanaco MP-500 melting point apparatus and are uncorrected. HPLC analysis were run on Agilent-1200 series HPLC system equipped with a photodiode array detector, using a Hypersil Gold C18 column (dimensions 250 × 4.6 mm, particle size 5 μm). Photodiode array (PDA) detector range was set as 210–600 nm.

4.2 Synthesis of oxazolidone derivatives

4.2.1 Synthesis of amides 1a, 1j, 1k, 1n, 1o and 1q–t.

Palmitoyl chloride (83 mg, 0.3 mmol) in CH2Cl2 (2 mL) was added dropwise to a mixture of amine (0.3 mmol) and Et3N (0.08 mL, 0.6 mmol) in anhydrous CH2Cl2 (2 mL) under a nitrogen atmosphere. The resulting solution was stirred at room temperature for 4 h, then quenched with saturated aqueous

Fig. 4 2f attenuated formalin-induced pain behaviors. (A) 2f (10 mg kg−1, i.p., open squares), injected 45 min prior to formalin treatment, produced time-dependent changes in composite pain score relative to vehicle (5% PEG/5% Tween-80 in saline, 10 mL kg−1, i.p., open cycles), or a combination of 2f and PPAR-α antagonist MK886 (2 mg kg−1, i.p., close squares). *P < 0.1 vs. vehicle; (B) 2f (close bar) reduced the licking time during phase 2 of the formalin response, **P < 0.1 vs. vehicle (open bar). MK886 (2 mg kg−1, i.p., close bar) prevented the anti-nociceptive effects of 2f. V represents vehicle (5% PEG/5% Tween-80 in saline), and MK represents PPAR-α antagonist MK886.

Fig. 5 Compound 2f alleviated TPA-induced inflammation. Effects of oral administration of 2f and ibuprofen on mice ears edema (A) and mRNA expression levels of TNF-α (B), IL-6 (C), COX-2 (D) and NF-κB (E) in mice ear treated with vehicle (open bars) or TPA (closed bars). Vehicle, 0.1% DMSO; TPA, 0.03% in acetone. ###P < 0.01 vs. vehicle; (B) **P < 0.01 vs. vehicle (open bar). MK886 (2 mg kg−1, i.p., close bar) prevented the anti-nociceptive effects of 2f. V represents vehicle (5% PEG/5% Tween-80 in saline), and MK represents PPAR-α antagonist MK886.

40 μM 2f (Fig. S4†), a concentration far above its threshold for NAAA inhibition. However, cisapride, a COX-2 inhibitor, inhibited 50% hERG at a concentration of 0.1 μM. These results suggested that with lower gastrointestinal and cardiovascular side effects, oxazolidone derivatives are promising anti-inflammatory drug candidates in clinical translation.
ammonium chloride, and extracted with CH₂Cl₂ (3 × 10 mL). The combined organic phases were washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel and 1a, 1j, 1k, 1n, 1o and 1q–1t were then obtained.  

4.2.2 Synthesis of amides 1l and 1m. A solution of 1-[(3-hydroxypropyl)pyrrolidin-1-yl]hexadecan-1-one (19 mg, 0.06 mmol) in THF (1 mL) was added slowly to a suspension of NaH (60% in mineral oil, 5 mg) in anhydrous THF (5 mL), under a nitrogen atmosphere at 0 °C. After stirring at the same temperature for 5 min, a solution of CH₄I (0.15 mmol) in THF (0.5 mL) was slowly added. The reaction mixture was stirred at 0 °C for 3 h before quenched with saturated aqueous ammonium chloride (5 mL) and extracted with EtOAc (3 × 5 mL). The combined organic layers were treated using the same procedure as 2.2.1 to afford 1l and 1m.  

4.2.3 Synthesis of amides 1c and 1d. A solution of 4 M HCl (1 mmol, 0.25 mL dioxane) was added slowly to 10 mL tert-buty1 1-palmitoylpyrrolidin-3-ylcarbamate (60.0 mg, 0.14 mmol) was dissolved in acetone (15 mL) and treated with K₂CO₃ (552 mg, 4 mmol) and alkyl bromide (2.5 mmol). The reaction mixture was filtered and concentrated under reduced pressure after being refuxed for 8 h. The residue was dissolved in a 1 : 1 mixture of aqueous 2 M NaOH and MeOH (16 mL), stirred at room temperature for 4 h, then treated with 2 M HCl, and extracted with EtOAc (3 × 5 mL). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel to afford 1c and 1d.  

4.2.4 Synthesis of amides 1e and 1f. I₂ solution (76 mg, 0.3 mmol) in THF (2 mL) was added slowly to a mixture of imidazole (27 mg, 0.4 mmol) and PPH₃ (79 mg, 0.3 mmol) in THF (5 mL), under a nitrogen atmosphere at 0 °C. After stirring at the same temperature for 1 h, a THF solution (2 mL) of 1-[(2-(hydroxymethyl) pyrrolidin-1-yl)hexadecan-1-one (71 mg, 0.2 mmol) was slowly added. The reaction mixture was stirred at 0 °C for 0.5 h and allowed to warm slowly to room temperature over 2 h. The reaction was quenched with a saturated solution of Na₂CO₃, and extracted with EtOAc (3 × 5 mL). The combined organic phases were washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (eluent : MeOH/CH₂Cl₂ = 1 : 15) to afford 1e and 1f.  

4.2.5 Synthesis of oxazolidone imides 1i, 2a–2h, 3a–3d and 3h–3s. A CH₂Cl₂ (5 mL) solution of oxalyl dichloride (0.06 mL, 0.66 mmol) was added slowly to a solution of the suitable acid (0.55 mmol) and dimethylformamide (0.05 mL) under a nitrogen atmosphere at 0 °C. The reaction mixture was stirred at 0 °C for 1 h and evaporated under reduced pressure. The acid chloride obtained was used as such for the next reaction due to its high instability. A hexane solution (0.2 mL) of 2.5 Mn-BuLi (0.55 mmol, hexane) was added slowly to a THF solution (5 mL) of 2-oxazolidone (44 mg, 0.55 mmol), under a nitrogen atmosphere at −78 °C. After stirring at the same temperature for 10 min, a THF solution (2 mL) of acid chloride (0.55 mmol) was slowly added. The reaction mixture was stirred at −78 °C for 0.5 h and allowed to warm slowly to room temperature over 5 h. The reaction was quenched with saturated aqueous ammonium chloride (1.0 mL) and extracted with EtOAc (3 × 5 mL). The combined organic layers were treated using the same procedure as 2.2.1 to afford 1i, 2a–2h, 3a–3d and 3h–3s.  

4.2.6 Synthesis of acids 3q-1 and 3t-1. Thionyl dichloride (0.2 mL, 3 mmol) was slowly added to 10 mL CH₂OH solution of the corresponding carboxylic acid (2.0 mmol) under a nitrogen atmosphere at 0 °C. The reaction mixture was stirred at 0 °C for 2 h and evaporated under reduced pressure. The product was dissolved in acetone (15 mL) and treated with K₂CO₃ (552 mg, 4 mmol) and alkyl bromide (2.5 mmol). The reaction mixture was filtered and concentrated under reduced pressure after being refuxed for 8 h. The residue was dissolved in a 1 : 1 mixture of aqueous 2 M NaOH and MeOH (16 mL), stirred at room temperature for 4 h, then treated with 2 M HCl, and extracted with EtOAc (3 × 5 mL). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel to afford 3q-1 and 3t-1.  

4.2.7 Synthesis of carboxamides 3e–3g. A CH₂Cl₂ solution (2 mL) of bis(trichloromethyl) carbonate (98 mg, 0.33 mmol) was added dropwise to a solution of 2-oxazolidone (44 mg, 0.55 mmol) and Et₃N (0.16 mL, 1.2 mmol) in CH₂Cl₂ (5 mL) under a nitrogen atmosphere. After stirring at 0 °C for 1 h and room temperature for 2 h, the amine (0.4 mmol) in CH₂Cl₂ (1 mL) was added dropwise to the mixture under a nitrogen atmosphere. The resulting mixture was stirred at room temperature for 5 h, quenched with saturated aqueous ammonium chloride, and extracted with CH₂Cl₂ (3 × 10 mL). The combined organic phases were washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel to afford 3e–3g.  

4.3 In vitro biological evaluation of oxazolidone derivatives  

4.3.1 HPLC-MS/MS analysis. A HPLC-MS/MS method was developed and validated for free fatty acids (FFAs) quantification. An ABI 3200 Q-Trap mass spectrometer equipped with Agilent 1200-HPLC system was used. FFAs were eluted at 1.0 mL min⁻¹ at 40 °C with MeOH : water : 95 : 5, v/v, each containing 0.25% acetic acid and 5 mmol L⁻¹ ammonium acetate, pH = 7.4. The molecular ions were monitored by ESI⁺ negative mode at m/z = 267 for heptadecanoic acid, m/z 303 for arachidonic acid, and m/z 269 for C17:0 FFAs.  

4.3.2 Protein preparation and enzymatic assay. HEK293 cells overexpressing rat NAAA (HEK293-rNAAA) or rat FAAH (HEK293-rFAAH) were maintained in Dulbecco’s modified Eagle medium (DMEM, Hyclone, Beijing, China) supplemented with 10% fetal bovine serum (FBS, Gibco®, Shanghai, China) containing 0.3 mg mL⁻¹ geneticin G418. HEK293-rNAAA or HEK293-rFAAH cells were harvested, washed with PBS, sonicated in 20 mM Tris–HCl (pH 7.5) containing 0.32 M sucrose, and centrifuged at 800 × g for 15 min at 4 °C. The supernatants were collected, and the protein concentrations were measured.
Compound 2f (30, 100 mg kg\(^{-1}\)) was then administered by intragastric gavage 30 min before TPA application. TPA (0.03% in acetone, 10 \(\mu\)L) was applied evenly to both sides of the left ear of each mouse to cause edema and inflammation, while the same amount of acetone alone was smeared to the right ear. Animals were sacrificed 3 h after treatment. Both ears were excised immediately and punched in the same position, with a diameter of 9 mm. The ear swelling was determined by weighing pieces of the ears obtained. The mRNA expression levels of TNF-\(\alpha\), IL-6, COX-2 and NF-\(\kappa\)B in mice ear were determined following our previously reported method.\(^{21}\)

4.4.3 Long term toxicity of compound 2f. Mice were treat with compound 2f (100 mg kg\(^{-1}\) for 10 d, 30 mg kg\(^{-1}\) for 30 d) or indomethacin (20 mg kg\(^{-1}\) for 3 d) via intragastral administration with normal diet. Mice body weight and food consumption were measured and recorded every day. The mice were sacrificed 6 h after the last administration and stomachs were harvested, rinsed with distilled water and photographed.

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