


 CrossMark
 click for updates
Cite this: *RSC Adv.*, 2017, 7, 5679

Asymmetric synthesis of C–F quaternary α -fluoro- β -amino-indolin-2-ones *via* Mannich addition reactions; facets of reactivity, structural generality and stereochemical outcome†

Chen Xie,^a Wanxing Sha,^a Yi Zhu,^a Jianlin Han,^{*a} Vadim A. Soloshonok^{bc} and Yi Pan^aReceived 3rd December 2016
Accepted 27th December 2016

DOI: 10.1039/c6ra27710a

www.rsc.org/advances

Reported herein is a new approach for the preparation of enantiomerically pure α -fluoro- β -amino-indolin-2-ones possessing tetrasubstituted fluorinated stereogenic centers. This method includes the detrifluoroacetylative *in situ* generation of tertiary enolates followed by Mannich reaction with (*S*₂)-sulfinylamines. The operationally convenient conditions coupled with perfect diastereoselectivity and functional substituent compatibility bode well for widespread application of this new synthetic method.

Fluoro-organic chemistry plays an essential role in furthering numerous technological advancements, most notably in the energy,¹ food² and healthcare industries.³ Consequently, the current interest in the synthesis of fluorine-containing compounds is at an all-time high, addressing the emerging scientific and practical needs.⁴ Thus, many challenging areas of fluorine methodology have seen important, but somewhat uneven progress.⁵ In particular, despite the impressive recent advances in the asymmetric synthesis of fluorinated amines/amino acids⁶ and derivatives containing quaternary C–F stereogenic carbon,⁷ and compounds possessing both of these, consecutively located, their structural features are still not well studied.⁸

One of the recent advances in fluoro-organic methodology is the development of detrifluoroacetylative, *in situ* generation of unprotected fluoro-enolates **1** (Scheme 1, eqn (1)).⁹ The initial study into the reactivity of enolates **1** using aldol^{9,10} and Mannich^{10,11} reactions has revealed their generally promising synthetic potential along with some problematic aspects awaiting more focused research. For example, enolates **1** easily reacts with aldehydes^{9,10} and highly electrophilic, CF₃-containing ketones¹² and imines,¹¹ while the reactions of **1** with unactivated C=O and C=N compounds still need more innovative

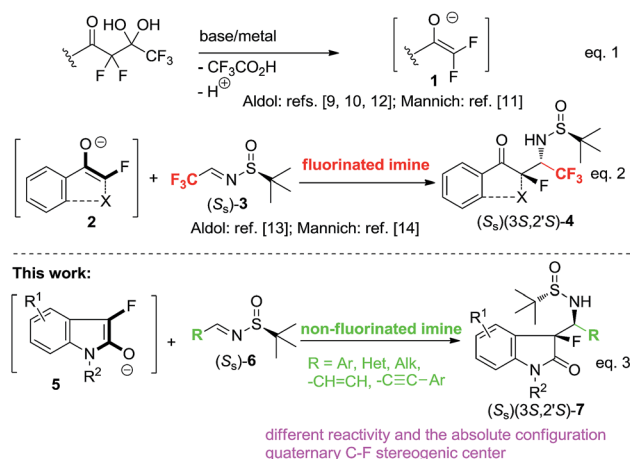
solutions. Over the last several years, our group has contributed to this research area by the development of cyclic enolates **2** (eqn (2)) allowing preparation of structurally complex products possessing tetrasubstituted fluorinated stereogenic centers.¹³ The preliminary data on reactivity of tertiary enolates **2** with imines¹⁴ revealed their moderate reactivity towards highly electrophilic CF₃-imines **3** affording derivatives **4** featuring the structurally challenging quaternary C–F stereogenic center next to the amino functionality. While the preparation of compounds **4** was a noticeable methodological success, one might agree that rather excessive number of fluorine atoms and relatively high configurational liability of the CH stereogenic carbon, would limit the pharmaceutical application of compounds of type **4**. Therefore, to realize the synthetic potential of this methodology we have to develop the corresponding Mannich addition reactions using common, non-

^aSchool of Chemistry and Chemical Engineering, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing, 210093, China. E-mail: hanjl@nju.edu.cn

^bDepartment of Organic Chemistry I, Faculty of Chemistry, University of the Basque Country UPV/EHU, Paseo Manuel Lardizabal 3, 20018 San Sebastián, Spain. E-mail: vadym.soloshonok@ehu.es

^cIKERBASQUE, Basque Foundation for Science Department, Alameda Urquijo 36-5, Plaza Bizkaia, 48011 Bilbao, Spain

† Electronic supplementary information (ESI) available: Detailed procedure and full characterization data, NMR spectra of product **11**, **13**, **14** and X-ray analysis data for **11b**. CCDC 1457060. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c6ra27710a



Scheme 1 Detrifluoroacetylative reactions with CF₃-imines **3** and non-fluorinated imines **6**.



fluorinated imines. In this work we report the very successful implementation of this goal, disclosing the Mannich reactions between 3-fluoroindolin-2-one derived enolates **5** (eqn (3)) and non-fluorinated aldimines **6**. We demonstrate that these addition reactions occur with nearly perfect (>98:2) diastereoselectivity allowing preparation of the target quaternary C–F amino products in enantiomerically pure form.

Drawing from our experience in asymmetric Mannich additions¹⁵ and, in particular, dealing with unactivated imines,¹⁶ we selected non-enolizable and reasonably reactive imine **10a**; also, bearing a trifluoromethylthio group – quite useful as the ¹⁹F-NMR reference for monitoring the reaction outcome (Table 1). As was anticipated, the reaction of imine **10a** conducted under the conditions (DIPEA, 2-Me-THF, –40 °C) previously reported for the additions of CF₃-sulfinylimine **3** (Scheme 1, eqn (2)), did not proceed at all. Lower reactivity of **10a** was counteracted by the elevated reaction temperature. Thus, the addition was rather successful at 0 °C, affording product **11aa** with moderate yield, but with quite respected diastereoselectivity (entry 1). Building on this apparent progress, we proceeded with a thorough optimization of all reaction parameters starting with the base. A selection of examples presented in entries 2–7 convincingly demonstrates that the base's basicity is not of prime importance, while its steric bulk plays rather significant role. This observation can be rationalized considering the sterically congested nature of tertiary enolate and therefore, its sensitivity towards the steric bulk of the base used. Accordingly, for the further optimization study, we selected triethylamine (entry 2) which gave the finest combination of chemical yield and diastereoselectivity.

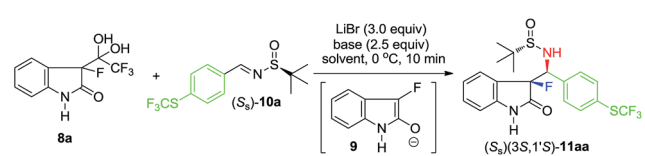
In the second round of the optimization experiments we focused on the properties of reaction solvent and its effect on

the addition reaction under study; the key representative results are presented in entries 8–10. For example, the application of 2-Me-THF gave noticeably better yield (entry 8) as compared with that of the reaction conducted in THF. However, the diastereoselectivity in this reaction was still incomplete. In sharp contrast, the reactions conducted in CH₃CN (entry 9) and DCM (entry 10) proceeded with nearly perfect stereochemical outcome. It should be mentioned that the used in this work notation (the diastereoselectivity of >98:2) is a very conservative estimation, indicating that the NMR signals of other diastereomeric products were not observed in the reaction mixtures. Comparing the results obtained in CH₃CN (entry 9) and DCM (entry 10), one may agree, that the former, providing higher chemical yield, was selected as the optimized solvent.

Next, we proceeded to study the substrate generality of this method. First, we decided to explore the reactivity and substituent effect of sulfinylimines (S₂)-**10a-r** (Scheme 2) derived from aldehydes representing various structural types and classes.

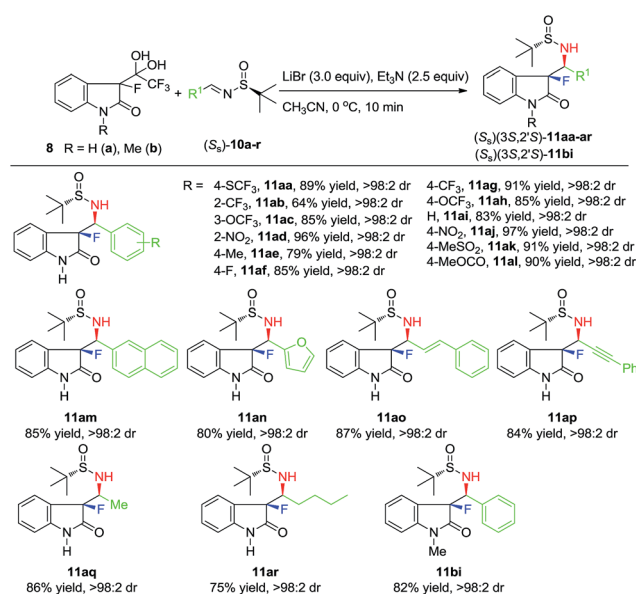
Considering the 19 examples listed in Scheme 2, one might agree that the stereochemical outcome of the Mannich additions between di-ketone hydrates **8** and sulfinylimines (S₂)-**10** is rather exceptional. In fact we failed to find an imine substrate reacting with less than >98:2 diastereoselectivity. In terms of the chemical yields, the lowest value of 64% was obtained for product **11ab** and, most likely, can be attributed to a slow reaction rate due to the steric shielding of the C=N group in **10b** by the *ortho*-trifluoromethyl group. The highest, 97% yield, was registered for product **11aj**, possessing *para*-nitro group. In general, for the compounds bearing electron-withdrawing substituents, the chemical yields were normally above 85%, while unsubstituted **11ai** and *para*-methyl products **11ae** gave a bit lower (83% and 79%) yields. The observed tendency is apparently a function of the electrophilicity/reactivity of the

Table 1 Optimization of the reaction conditions for the detrifluoroacetylative Mannich addition reaction between **8a** and imine (S)-**10a**^a



Entry	Base	Solvent	Yield ^b (%)	Dr ^c
1	DIPEA	THF	54	97:3
2	Et ₃ N	THF	88	97:3
3	<i>n</i> Pr ₃ N	THF	<10	n.d.
4	<i>n</i> Bu ₃ N	THF	<10	n.d.
5	DABCO	THF	92	96:4
6	DBU	THF	87	96:4
7	TMEDA	THF	76	96:4
8	Et ₃ N	2-Me-THF	94	97:3
9	Et ₃ N	CH ₃ CN	89	>98:2
10	Et ₃ N	DCM	80	>98:2

^a Reaction condition: di-ketone hydrates **1a** (0.6 mmol), sulfinylimine **10a** (0.5 mmol), LiBr (156.3 mg, 1.8 mmol, 3.0 equiv.), base (1.5 mmol, 2.5 equiv.) in 5 mL solvent at 0 °C for 10 min. ^b Isolated yields of the major diastereomers. ^c Determined by the ¹⁹F-NMR analysis of the crude products.



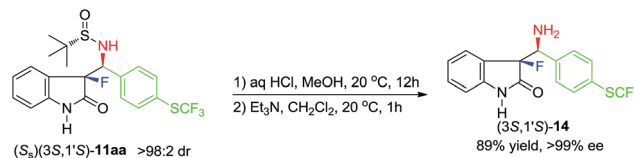
Scheme 2 Substrate generality study of di-ketone hydrates **8** with (S₂)-**10**.



C=N bond in sulfinylimines (S_S)-**10**. Generality of this reaction was further exemplified by the preparation of products **11am** and **11an**, bearing naphthyl and hetero-aromatic rings. Of particular interest were derivatives **11ao** and **11ap** featuring unsaturated moieties, which can be used for synthetic elaborations. Quite remarkably, this method was found to be readily applicable for enolizable alkyl-aldehyde derived sulfinylimines (S_S)-**10q,r**. Thus, products **11q,r**, bearing methyl and *n*-butyl groups, were isolated with the same excellent (>98:2) diastereoselectivity and preparatively attractive chemical yields. Finally, *N*-Me β -keto-amide-hydrate **8b** was reacted with benzaldehyde derived imine **10i**. The addition proceeded without any observable problems giving rise to diastereomerically pure product **11bi** with isolated 82% yield.

N-Methyl derivative **11bi** was a logical connection to the subsequent second series of substrate generality experiments where we explored the effects of the substituents positioned on the starting β -keto-amide-hydrates **12** (Scheme 3). Imine (S_S)-**10a**, derived from 4-(trifluoromethylthio)benzaldehyde, was selected as standard electrophile and all reactions were conducted under the same conditions. The twelve examples shown in Scheme 3 were selected to represent the two-pronged substituent positioning on the starting di-ketone hydrates **12**. The major conclusion that one can draw from these data, is the synthetically attractive generality of this method, as the nearly perfect diastereoselectivity (>98:2) and respected chemical yields of products **13a-l** were maintained in all cases studied. Preparation of compounds bearing hydrolytically liable groups (**13c**), di-substitution (**13e**) and electronically conjugated *N*-allyl (**13k**), *N*-phenyl (**13l**) moieties, deserve some particular mentioning.

As the last synthetic goal of this study we decided to demonstrate the deprotection procedure affording the target compounds with free amino functionality. We selected derivative **11aa**, possessing unsubstituted NH group on the indolin-2-one moiety (Scheme 4). The deprotection reaction was



Scheme 4 Deprotection of product (S_S)(3*S*,1'*S*)-**11aa**.

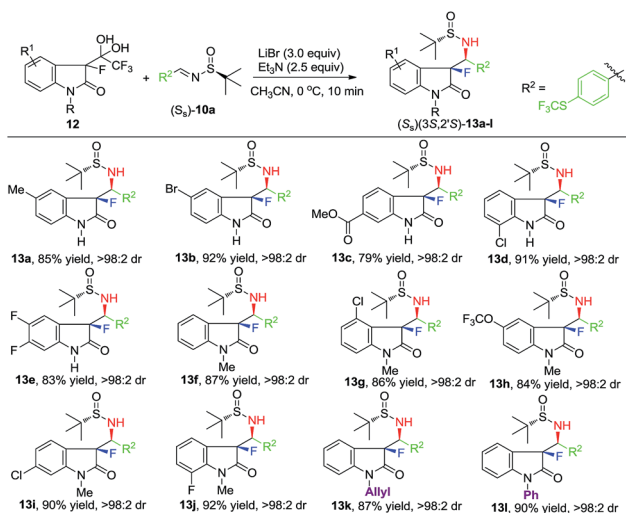
conducted under typical mild acidic conditions¹⁷ giving α -fluoro- β -amino product ($3S,1'S$)-**14**, which was isolated with 89% yield. Furthermore, product **14** was found to be of >99% ee, strongly suggesting that compounds **11** and **13**, listed in Schemes 2 and 3, are virtually diastereomerically pure.

The final objective of this research was the assignment of the absolute configuration of products **11** and **13**. To this end we performed a crystallographic analysis of *N*-Me derivative **11bi** (Scheme 2), which gave rather unexpected result. Thus, according to the X-ray data (see ESI[†]), the configuration of **11bi** was found to be (S_S)(3*S*,2'*S*), that is in sharp contrast to the (S_S)(3*S*,2'*S*)^{18,19} configuration reported for the reactions of CF₃-sulfinylimine (S_S)-**3** (Scheme 1).^{10,14}

To rationalize this unexpected stereochemical outcome, we can propose the transition state (TS-A) presented in Fig. 1. First of all, based on the literature data,¹⁷ the observed ($1'S$) absolute configuration in products (S_S)(3*S*,2'*S*)-**11/13**, strongly suggest that sulfinylimines (S_S)-**10** reacts in the corresponding *s-trans* geometric configuration. Furthermore, the ($3S$) stereochemistry of the quaternary carbon indicates that sulfinylimines (S_S)-**10** approach the re-face of the corresponding tertiary enolates. Taking these experimental facts together, and complying with the principle of minimum charge separation,²⁰ we constructed plausible TS-A leading to the products of the observed (S_S) ($3S,2'S$) absolute configuration. Noteworthy, the close proximity of enolate O-Li moiety and the imine nitrogen in TS-A, allows not only for the concerted oxygen-nitrogen charge transfer, but also for the additional stabilization *via* Li-coordination to the S-O oxygen. This feature is never observed in the case of CF₃-sulfinylimines **3** (ref. 5a and 10) and may be the underlying reason of the different stereochemical outcome observed in the present study. Moreover, in TS-A the substituent R on the imine, the aromatic ring of the tertiary enolate and *N*-R group are perfectly located to avoid unfavorable overlaps or any other steric interactions. Accordingly, one may agree that TS-A can be quite credibly used to elucidate the remarkable substrate generality observed in these Mannich addition reactions.



Fig. 1 Plausible transition state TS-A accounting for the observed stereochemical outcome and *s-trans* geometry of sulfinylimines (S_S)-**10**.



Scheme 3 Substrate generality study of di-ketone hydrates **12** with (S_S)-**10a**.



Conclusions

In summary, we have developed highly stereoselective dehydroacetylation Mannich reactions between the *in situ* generated tertiary enolates and sulfinyl-imines. The reactions feature nearly complete stereochemical outcome and remarkable generality accommodating several type of functional substitution on both enolates and imines. These reaction characteristics constitute a method of high synthetic value for preparation of enantiomerically pure α -fluoro- β -amino-indolin-2-ones with tetrasubstituted fluorinated stereogenic center.

Acknowledgements

We gratefully acknowledge the financial support from the National Natural Science Foundation of China (No. 21472082).

Notes and references

- (a) Y. Liang, D. Feng, Y. Wu, S.-T. Tsai, G. Li, C. Ray and L. Yu, *J. Am. Chem. Soc.*, 2009, **131**, 7792; (b) R. Berger, G. Resnati, P. Metrangolo, E. Weber and J. Hulliger, *Chem. Soc. Rev.*, 2011, **40**, 3496.
- (a) P. Jeschke, *ChemBioChem*, 2004, **5**, 570; (b) T. Fujiwara and D. O'Hagan, *J. Fluorine Chem.*, 2014, **167**, 16.
- (a) Y. Zhou, J. Wang, Z. Gu, S. Wang, W. Zhu, J. L. Aceña, V. A. Soloshonok, K. Izawa and H. Liu, *Chem. Rev.*, 2016, **116**, 422; (b) W. Zhu, J. Wang, S. Wang, Z. Gu, J. L. Aceña, K. Izawa, H. Liu and V. A. Soloshonok, *J. Fluorine Chem.*, 2014, **167**, 37; (c) J. Wang, M. Sánchez-Roselló, J. L. Aceña, C. del Pozo, A. E. Sorochinsky, S. Fustero, V. A. Soloshonok and H. Liu, *Chem. Rev.*, 2014, **114**, 2432; (d) K. Izawa, J. L. Aceña, J. Wang, V. A. Soloshonok and H. Liu, *Eur. J. Org. Chem.*, 2016, 8–16; (e) E. P. Gillis, K. J. Eastman, M. D. Hill, D. J. Donnelly and N. A. Meanwell, *J. Med. Chem.*, 2015, **58**, 8315.
- (a) J. Han, A. E. Sorochinsky, T. Ono and V. A. Soloshonok, *Curr. Org. Synth.*, 2011, **8**, 281; (b) K. Mikami, S. Fustero, M. Sánchez-Roselló, J. L. Aceña, V. A. Soloshonok and A. E. Sorochinsky, *Synthesis*, 2011, 3045; (c) N. Shibata, T. Ishimaru, S. Nakamura and T. Toru, *J. Fluorine Chem.*, 2007, **128**, 469; (d) N. Shibata, S. Mizuta and H. Kawai, *Tetrahedron: Asymmetry*, 2008, **19**, 2633; (e) X. H. Xu and F. L. Qiang, *Curr. Org. Chem.*, 2015, **19**, 1566.
- (a) H. B. Mei, C. Xie, J. L. Han and V. A. Soloshonok, *Eur. J. Org. Chem.*, 2016, 5917; (b) X. H. Xu, K. Matsuzaki and N. Shibata, *Chem. Rev.*, 2015, **115**, 731; (c) S. Lectard, Y. Hamashima and M. Sodeoka, *Adv. Synth. Catal.*, 2010, **352**, 2708; (d) C. Ni, M. Hu and J. Hu, *Chem. Rev.*, 2015, **115**, 765; (e) X. Yang, T. Wu, R. J. Phipps and F. D. Toste, *Chem. Rev.*, 2015, **115**, 826–870; (f) J. Charpentier, N. Früh and A. Togni, *Chem. Rev.*, 2015, **115**, 650; (g) S. Zhang, L. Li, Y. Hu, Y. Li, Y. Yang, Z. Zha and Z. Wang, *Org. Lett.*, 2015, **17**, 5036.
- (a) J. L. Aceña, A. E. Sorochinsky and V. A. Soloshonok, *Synthesis*, 2012, 1591; (b) K. V. Turcheniuk, V. P. Kukhar, G.-V. Roeschenthaler, J. Luis Acena, V. A. Soloshonok and A. E. Sorochinsky, *RSC Adv.*, 2013, **3**, 6693; (c) X.-L. Qiu and F.-L. Qing, *Eur. J. Org. Chem.*, 2011, 3261.
- (a) X. Yang, R. J. Phipps and F. D. Toste, *J. Am. Chem. Soc.*, 2014, **136**, 5225; (b) N. Mankad and F. D. Toste, *Chem. Sci.*, 2012, **3**, 72; (c) R. J. Phipps and F. D. Toste, *J. Am. Chem. Soc.*, 2013, **135**, 1268; (d) N. Shibata, J. Kohno, K. Takai, T. Ishimaru, S. Nakamura, T. Toru and S. Kanemasa, *Angew. Chem., Int. Ed.*, 2005, **44**, 4204; (e) I. Saidalimu, S. Suzuki, E. Tokunaga, E. Tokunaga and N. Shibata, *Chem. Sci.*, 2016, **7**, 2106; (f) W. Yu, X. H. Xu and F. L. Qing, *Adv. Synth. Catal.*, 2015, **357**, 2039; (g) Y. Hamashima, K. Yagi, H. Takano, L. Tamas and M. Sodeoka, *J. Am. Chem. Soc.*, 2002, **124**, 14530; (h) L. Hintermann and A. Togni, *Angew. Chem., Int. Ed.*, 2000, **39**, 4359; (i) H. Ibrahim and A. Togni, *Chem. Commun.*, 2004, 1147.
- (a) K. Hiramatsu, T. Honjo, V. Rauniyar and F. D. Toste, *ACS Catal.*, 2016, **6**, 151; (b) R. J. Phipps, K. Hiramatsu and F. D. Toste, *J. Am. Chem. Soc.*, 2012, **134**, 8376; (c) H. P. Shunatona, N. Fruh, Y. M. Wang, V. Rauniyar and F. D. Toste, *Angew. Chem., Int. Ed.*, 2013, **52**, 7724; (d) Y. Hamashima, T. Suzukia, H. Takanoa, Y. Shimura and M. Sodeoka, *J. Am. Chem. Soc.*, 2005, **127**, 10164; (e) B. M. Trost, T. Saget, A. Lerchen and C. I. Hung, *Angew. Chem., Int. Ed.*, 2016, **55**, 781.
- (a) C. Han, E. H. Kim and D. A. Colby, *J. Am. Chem. Soc.*, 2011, **133**, 5802; (b) P. Zhang and C. Wolf, *Angew. Chem., Int. Ed.*, 2013, **52**, 7869; (c) I. Saidalimu, X. Fang, X.-P. He, J. Liang, X. Yang and F. Wu, *Angew. Chem., Int. Ed.*, 2013, **52**, 5566.
- H. Mei, C. Xie, J. L. Aceña, V. A. Soloshonok, G.-V. Roeschenthaler and J. Han, *Eur. J. Org. Chem.*, 2015, 6401.
- (a) C. Xie, L. Wu, H. Mei, V. A. Soloshonok, J. Han and Y. Pan, *Org. Biomol. Chem.*, 2014, **12**, 7836; (b) C. Xie, L. Wu, H. Mei, V. A. Soloshonok, J. Han and Y. Pan, *Tetrahedron Lett.*, 2014, **55**, 5908; (c) C. Xie, L. Wu, J. Zhou, H. Mei, V. A. Soloshonok, J. Han and Y. Pan, *J. Fluorine Chem.*, 2015, **172**, 13.
- P. Zhang and C. Wolf, *J. Org. Chem.*, 2012, **77**, 8840.
- (a) C. Xie, L. Wu, J. Han, V. A. Soloshonok and Y. Pan, *Angew. Chem., Int. Ed.*, 2015, **54**, 6019; (b) L. Zhang, C. Xie, Y. Dai, H. Mei, V. A. Soloshonok, Y. Pan and J. Han, *J. Fluorine Chem.*, 2016, **184**, 28; (c) W. Sha, L. Zhang, W. Zhang, H. Mei, V. A. Soloshonok, J. Han and Y. Pan, *Org. Biomol. Chem.*, 2016, **14**, 7295.
- (a) C. Xie, Y. Dai, H. Mei, J. Han, V. A. Soloshonok and Y. Pan, *Chem. Commun.*, 2015, **51**, 9149; (b) C. Xie, L. Zhang, W. Sha, V. A. Soloshonok, J. Han and Y. Pan, *Org. Lett.*, 2016, **18**, 3270.
- (a) H. Mei, C. Xie, L. Wu, V. A. Soloshonok, J. Han and Y. Pan, *Org. Biomol. Chem.*, 2013, **11**, 8018; (b) C. Xie, H. Mei, L. Wu, V. A. Soloshonok, J. Han and Y. Pan, *RSC Adv.*, 2014, **4**, 4763; (c) H. Mei, Y. Xiong, C. Xie, V. A. Soloshonok, J. Han and Y. Pan, *Org. Biomol. Chem.*, 2014, **12**, 2108.
- (a) L. M. Wu, C. Xie, H. B. Mei, V. A. Soloshonok, J. L. Han and Y. Pan, *Org. Biomol. Chem.*, 2014, **12**, 4620; (b) Y. L. Dai, C. Xie, L. M. Wu, H. B. Mei, V. A. Soloshonok, J. L. Han and Y. Pan, *RSC Adv.*, 2015, **5**, 3491.



- 17 (a) C. Xie, L. Zhang, H. Mei, R. Pajkert, M. Ponomarenko, Y. Pan, G.-V. Rösenthaller, V. A. Soloshonok and J. Han, *Chem.-Eur. J.*, 2016, **22**, 7036; (b) M. T. Robak, M. A. Herbage and J. A. Ellman, *Chem. Rev.*, 2010, **110**, 3600.
- 18 To avoid a confusion, it should be noted that while the descriptors of the absolute configuration for products **11**, **13** and those derived from CF₃-imine **3** (Scheme 1) are the same, these compounds are stereochemically different due to the Cahn-Ingold-Prelog priority rules ref. 19.
- 19 R. S. Cahn, C. K. Ingold and V. Prelog, *Angew. Chem., Int. Ed.*, 1966, **5**, 385.
- 20 (a) V. A. Soloshonok, C. Cai, V. J. Hruby, L. V. Meervelt and T. Yamazaki, *J. Org. Chem.*, 2000, **65**, 6688; (b) V. A. Soloshonok, H. Ohkura, A. Sorochinsky, N. Voloshin, A. Markovsky, M. Belik and T. Yamazaki, *Tetrahedron Lett.*, 2002, **43**, 5445.

