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Facile access to novel 1,2,4-oxadiazinan-5-ones *via* [3 + 3] cycloaddition of *in situ* generated azaoxyallyl cations with nitrones†

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(1) Previous works

(a) Jeffrey's work²

In the presence of Na_2CO_3 , azaoxyallyl cations in situ generated from α -halohydroxamates with nitrones readily underwent [3 + 3] cycloaddition, and gave rise to 1,2,4-oxadiazinan-5-one derivatives in 56–99% chemical yields. The chemical structure of the title compounds was unambiguously identified by X-ray single crystal structure analysis.

1. Introduction

Azaoxyallyl cations represent a family of versatile and powerful synthetic synthons, which are generally in situ generated from α-halohydroxamates in the presence of organic or inorganic bases. Owing to the unique structural features and reactivities of azaoxyallyl cations, some various efforts have been made to enrich the synthetic methodology of azaoxyallyl cations (Scheme 1, 1). In 2011, Jeffrey and co-workers pioneeringly reported the [4 + 3] cycloaddition between azaoxyallyl cations and cyclic dienes (Scheme 1, 1a).2 Since then, the research groups of Jeffrey, Wu and Liao independently devised similar [3 + 2] cycloadditions of azaoxyallyl cations with differently substituted indoles for the preparation of pyrroloindolines (Scheme 1, 1b).3 Moreover, Chen's group discovered the [3 + 1] and [3 + 2] cycloadditions of azaoxyallyl cations with sulfur ylides delivering β- and γ-lactams (Scheme 1, 1c). In 2016, Lin and Jeffrey's groups individually successfully applied the [3 + 2] cycloaddition of azaoxyallyl cations with aldehydes in the synthesis of oxazolidin-4-ones (Scheme 1, 1d).5 Additionally, in the same year, Wu and co-workers established the [3 + 3] cycloaddition of isoquinoline N-oxides as cyclic nitrones with azaoxyallyl cations (Scheme 1, 1e).6 Even though the important and elegant advances in the synthetic methodology of azaoxyallyl cations, it remains highly demanded to develop novel and efficient synthetic methodologies of azaoxyallyl cations for the synthesis of structurally diverse heterocycles.

odology of azaoxyallyl cations, we designed the novel [3+3] cycloaddition of the azaoxyallyl cations *in situ* generated from α -halohydroxamates with acyclic nitrones as 1,3-dipoles with a purpose to prepare potentially bioactive 1,2,4-oxadiazine-5-ones (Scheme 1, 2).⁷ Pleasantly, the [3+3] cycloaddition between azaoxyallyl cations and acyclic nitrones proceeded readily under mild reaction conditions, and gave the title target molecules in the desirable chemical yields. To the best of our knowledge, no such a work has been reported in the literature to date.

Encouraged by the previous works on the synthetic meth-

(e) Wu's work⁶

R1

N-O-[3+3]

"cyclic nitrone"

(2) This work

Scheme 1 Representative cycloadditions involving azaoxyallyl cations.

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(d) Lin and Jeffrey's works

2. Results and discussion

Initially, we screened the solvent effects on the [3+3] cycloaddition of α -halohydroxamate 1a with nitrone 2a as outlined in Table 1. Noticeably, the use of the different solvents significantly affected the chemical yield of the [3+3] cycloaddition. When EtOH was tested as polar protonic solvent, product 3aa was produced in a trace amount in 48 h (entry 6). Compared with the former case, the use of TFE, toluene and DCM as solvents differently increased the chemical yield of the [3+3] cycloaddition (entries 2 and 4-5 νs . 6). Moreover, the significant increase in the chemical yield of product 3aa was observed by using CH₃CN as a polar aprotonic solvent (entry 3). Finally, the [3+3] cycloaddition underwent more efficiently in HFIP as a polar fluorinated solvent, and provided product 3aa in the highest chemical yield (entry 1).

Then, we examined a variety of bases bearing the various basic strength to clarify their effects on the [3+3] cycloaddition of α -halohydroxamate 1a with nitrone 2a using HFIP as solvent as summarized in Table 2. Remarkably, the chemical yield of the [3+3] cycloaddition highly depended on the used bases. As for NaHCO3 as base, it provided 3aa in 13% chemical yield (entry 6). By comparison, the use of K_2CO_3 , Cs_2CO_3 and MeONa bases enhanced the chemical yield of the [3+3] cycloaddition differently (entries 2-3 & 7 νs . 6). As far as other examined bases such as Na2CO3, Et3N, KOH and DBU were concerned, they could promote the [3+3] cycloaddition efficiently, and delivered product 3aa in excellent chemical yields (entries 1, 4-5 & 8). Accordingly, in the presence of by Na2CO3 as an inorganic base, the [3+3] cycloaddition proceeded most efficiently, and produced product 3aa in the highest chemical yield (98%, entry 1).

Meanwhile, we also investigated the effect of the equivalent ratio of $1a/2a/Na_2CO_3$ on the [3+3] cycloaddition of α -halohydroxamate 1a with nitrone 2a as shown in Table 3. Apparently, the used equivalent ratio of $1a/2a/Na_2CO_3$ dramatically influenced the chemical yield of the [3+3] cycloaddition. The application of the ratio of 1:1:1 in the [3+3] cycloaddition formed product 3aa in 59% chemical yield (entry 1). In regard to

Table 1 Screening of solvents^a

$$\begin{array}{c}
 \text{Me} \\
 \text{Me} \\
 \text{H}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{Ph} \\
 \text{H}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{Solv.} \\
 \text{r.t.}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph} \\
 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
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$$\begin{array}{c}
 \text{N} \\
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 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph}
\end{array}$$

$$\begin{array}{c}
 \text{N} \\
 \text{N} \\
 \text{N} \\
 \text{Ph}
\end{array}$$

Entry	Solvent	Time (h)	Yield ^b (%)
1	HFIP	1	98
2	TFE	1.5	40
3	CH_3CN	4	90
4	Toluene	48	30
5	DCM	48	26
6	EtOH	48	Trace

 $[^]a$ Reactions were carried out with 0.2 mmol of **1a** (54.2 mg) and 0.1 mmol of **2a** (19.7 mg) in the presence of 0.2 mmol of Na₂CO₃ (21.2 mg) in 0.5 mL of the indicated solvents at room temperature. b Isolated yield.

Table 2 Screening of bases^a

Entry	Base	Time (h)	Yield ^b (%)
1	Na ₂ CO ₃	1	98
2	K_2CO_3	1	78
3	Cs_2CO_3	1	82
4	Et ₃ N	1	98
5	KOH	1	92
6	$NaHCO_3$	1	13
7	MeONa	1	48
8	DBU	1	98

^a Reactions were carried out with 0.2 mmol of **1a** (54.2 mg) and 0.1 mmol of **2a** (19.7 mg) in the presence of 0.2 mmol of the indicated bases in 0.5 mL of HFIP at room temperature. ^b Isolated yield.

the ratios such as 1.5:1:1.5, 1:1.5:1 and 1:2:1, they provided product 3aa in the increased chemical yields (entries 1 vs. 2 & 4-5). Moreover, it was found that product 3aa was obtained in excellent chemical yields with the use of ratios of 2:1:2 and 2:1:1 in the [3+3] cycloaddition (entries 3 & 6). Noticeably, among all the screened ratios of $1a/2a/\text{Na}_2\text{CO}_3$, the ratio of 2:1:2 should be the most optimal for the [3+3] cycloaddition, and furnished product 3aa in 98% chemical yield (entry 3).

Finally, we broadened the reaction scope of the [3 + 3] cycloaddition under the optimal reaction conditions by employing structurally different α -halohydroxamates **1** and nitrones **2** as summarized in Table 4. Obviously, the variations of R_1 – R_4 groups of substrates **1** and **2** significantly affected the chemical yield of the [3 + 3] cycloaddition. Nitrones **2a–2l** reacted easily with **1a** bearing two methyl groups at the α -position, and gave products **3aa–3al** in 76–99% chemical yields

Table 3 Screening of ratios of 1a/2a/Na₂CO₃^a

$$\begin{array}{c} O \\ Me \\ Me \\ H \end{array} + \begin{array}{c} O \\ N \\ H \end{array} + \begin{array}{c} O \\ N \\ Ph \\ H \end{array} + \begin{array}{c} O \\ N \\ HFIP \\ r.t. \end{array} + \begin{array}{c} O \\ N \\ Ph \\ Ph \end{array} + \begin{array}{c} O \\ N \\ Ph \\ Ph \end{array}$$

Entry	Equivalentratio (1a/2a/Na ₂ CO ₃)	Time (h)	$Yield^{b}$ (%)	
1	1:1:1	1	59	
2	1.5:1:1.5	1	85	
3	2:1:2	1	98	
4	1:1.5:1	1	68	
5	1:2:1	1	78	
6	2:1:1	1	92	

^a Reactions were carried out with 1a and 2a in the presence of Na₂CO₃ in 0.5 mL of HFIP at the indicated equivalent ratios of 1a/2a/Na₂CO₃ at room temperature. ^b Isolated yield.

Table 4 Extension of reaction scope^a

Entry	1 (R ₁ , R ₂ , X)	$2(R_3, R_4)$	3	Time (h)	Yield ^b (%)
1	1a (Me, Me, Br)	2a (Ph, Ph)	3aa	1	98
2	1a (Me, Me, Br)	$2\mathbf{b}$ (4-MeOC ₆ H ₄ , Ph)	3ab	1	96
3	1a (Me, Me, Br)	$2c (4-MeC_6H_4, Ph)$	3ac	1	93
4	1a (Me, Me, Br)	2d (4-ClC ₆ H ₄ , Ph)	3ad	1	99
5	1a (Me, Me, Br)	$2e (3-MeOC_6H_4, Ph)$	3ae	1	89
6	1a (Me, Me, Br)	2f (3-ClC ₆ H ₄ , Ph)	3af	1	99
7	1a (Me, Me, Br)	2g (2-MeOC ₆ H ₄ , Ph)	3ag	1	76
8	1a (Me, Me, Br)	2h (2-ClC ₆ H ₄ , Ph)	3ah	1	99
9	1a (Me, Me, Br)	2i (4-BrC ₆ H ₄ , Ph)	3ai	1	97
10	1a (Me, Me, Br)	2j (4-FC ₆ H ₄ , Ph)	Зај	1	87
11	1a (Me, Me, Br)	2k (4-CNC ₆ H ₄ , Ph)	3ak	1	99
12	1a (Me, Me, Br)	2l $(4-NO_2C_6H_4, Ph)$	3al	1	97
13	1a (Me, Me, Br)	2m (Ph, Bn)	3am	1	81
14	1a (Me, Me, Br)	2n (Et, Bn)	3an	1	77
15	1a (Me, Me, Br)	20 (Ph, Me)	3ao	1	93
16	1a (Me, Me, Br)	2p (2-naphthyl, Ph)	Зар	1	92
17	1a (Me, Me, Br)	2q (2-furyl, Ph)	3aq	1	99
18	1b (H, Et, Br)	2a (Ph, Ph)	3ba	1	nr^c
19	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	2a (Ph, Ph)	3ca	1	92
20	1d (H, Cl, Cl)	2a (Ph, Ph)	3da	12	nr^c
21	1e (H, phenyl, Cl)	2a (Ph, Ph)	3ea	1	56
	1f 0				
22	1f O Me Me N Br	2a (Ph, Ph)	3fa	12	nr^c
23	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	2m (Ph, Bn)	3cm	1	78
24	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	$2\mathbf{g}$ (2-MeOC ₆ H ₄ , Ph)	3cg	1	67
25	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	$2h (2-ClC_6H_4, Ph)$	3ch	1	86
26	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	2p (2-naphthyl, Ph)	3ср	1	90
27	1c $(R_1, R_2 = -CH_2(CH_2)_3CH_2-, X = Br)$	2q (2-furyl, Ph)	3cq	1	61

^a Reactions were carried out with 0.2 mmol of **1** and 0.1 mmol of **2** in the presence of 0.2 mmol of Na₂CO₃ (21.2 mg) in 0.5 mL of HFIP at room temperature. ^b Isolated yield. ^c No reaction.

(entries 1–12). Basically, with respect to the [3 + 3] cycloaddition with 1a, the nitrones 2 could well tolerate the existence of electron-poor or electron-rich phenyl rings as R₃ group, and furnished products 3 in excellent chemical yields (entries 2–4, 8 and 9, 11 and 12). Moreover, the nitrones 2b, 2e and 2g, involving a 4-, 3- or 2-MeO-substituted phenyl ring as R₃ group respectively, afforded products 3ab, 3ae and 3ag in the quite different chemical yields in [3 + 3] the cycloaddition with 1a (entries 2, 5 & 7); in contrast, the nitrones 2d, 2f and 2h, including a 4-, 3- or 2-Cl-substituted phenyl ring as R₃ group individually, provide products 3ad, 3af and 3ah in the same chemical yields (entries 4, 6 & 8).

Simultaneously, the nitrones **2** well endured the varying bulky size of R_3 and R_4 groups in the cycloaddition with **1a**, and gave the corresponding products **3** in 77–99% chemical yields (entries 13–17). By comparison, in the cycloaddition with **2a**, the α -halohydroxamates **1** could not widely tolerate the structural change of R_1 and R_2 groups. For example, the [3 + 3] cycloaddition of substrates **1b** or **1d** with **2a** did not take place at all (entries 18 & 20); however, substrates **1c** and **1e** could well react

with **2a**, and furnished products **3ca** and **3ea** in 92% and 56% chemical yields, respectively (entries 19 & 21). In addition, we performed the [3+3] cycloaddition of α -halohydroxamate **1c** containing a cyclohexyl subunit with the nitrones **2** with different R_3 or R_4 groups, and discovered that all these [3+3] cycloadditions proceeded ready to form products **3** in the reasonable chemical yields (entries 23–27). Finally, the chemical structure of **3ad** was unambiguously determined by single crystal X-ray analysis as depicted in Fig. 1.8

Finally, we proposed the reaction mechanism for the formation of 3ad (Scheme 2). In the presence of Na_2CO_3 , the elimination reaction of 1a takes place to give azaoxyallyl cation 4. Then, two possible transition states TS1 and TS2 will be produced for the [3+3] cycloaddition between 4 and 2d. With the aid of the molecular model, it was found that in TS2 phenyl group at nitrogen atom of 2d sterically repulses α -methyl group of 4 severely; whereas, this strong destabilizing interaction does not exist in TS1 at all. Therefore, the transition state TS1 is more stable than the transition state TS2, and mainly accounts for the formation of the desired cycloadduct 3ad.

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Fig. 1 X-ray single crystal structure of **3ad** (with thermal ellipsoils shown at the 50% probability level).

Scheme 2 Proposed mechanism for the formation of 3ad.

3. Conclusions

In conclusion, the [3 + 3] cycloadditions of *in situ* generated azaoxyallyl cations with nitrones underwent efficiently, and provided an easy access to the novel potentially bioactive 1,2,4-oxadiazinan-5-ones in reasonable chemical yields. Furthermore, the exploration of other novel cycloadditions of azaoxyallyl cations with various 1,3-, 1,4- and 1,5-dipoles is ongoing in our organic lab, and will be reported in due course.

4. Experimental section

4.1 General information

4.2 Procedure for the synthesis of products 3

A mixture of α -halohydroxamate 1 (2.0 equiv., 0.2 mmol), nitrone 2 (1.0 equiv., 0.1 mmol) and Na₂CO₃ (2.0 equiv., 0.2 mmol) in 0.5 mL of HFIP was stirred at room temperature for 1 h. After the reaction was completed as indicated by TLC plate, the solvent was removed by evaporation and the crude product was purified by flash column chromatography on silica gel (petroleum ether/ethyl acetate = 15:1) to afford the pure products 3 as white solid (56–99% yields).

4-(Benzyloxy)-6,6-dimethyl-2,3-diphenyl-1,2,4-oxadiazinan-5-one (3aa). White solid, yield: 38.0 mg, 98%; mp = 110.8–111.2 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.39–7.35 (m, 3H), 7.32–7.30 (m, 3H), 7.29–7.23 (m, 4H), 7.20–7.16 (m, 2H), 6.97 (t, J = 7.2 Hz, 1H), 6.81 (d, J = 10.0 Hz, 2H), 5.64 (s, 1H), 5.17 (d, J = 10.0 Hz, 1H), 4.63 (d, J = 10.0 Hz, 1H), 1.73 (s, 3H), 1.67 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 145.6, 134.9, 133.6, 130.0, 129.3, 129.0, 128.9, 128.6, 128.5, 128.1, 123.8, 118.3, 83.9, 83.5, 77.4, 24.0, 23.5 ppm; HRMS (ESI) calculated for C₂₄H₂₅N₂O₃ [M + H][†]: 389.18597, found 389.18533.

4-(Benzyloxy)-3-(4-methoxyphenyl)-6,6-dimethyl-2-phenyl-1,2,4-oxadiazinan-5-one (3ab). White solid, yield: 40.2 mg, 96%; mp = 105.2–105.8 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.38–7.28 (m, 5H), 7.20–7.16 (m, 4H), 6.98–6.95 (m, 1H), 6.83–6.77 (m, 4H), 5.60 (s, 1H), 5.14 (d, J = 10.0 Hz, 1H), 4.63 (d, J = 10.0 Hz, 1H), 3.78 (s, 3H), 1.72 (s, 3H), 1.66 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 145.2, 135.6, 134.7, 134.1, 130.0, 129.4, 129.3, 129.1, 128.9, 128.8, 128.6, 127.0, 124.0, 118.1, 83.7, 83.2, 77.4, 24.0, 23.4 ppm; HRMS (ESI) calculated for C₂₅H₂₇N₂O₄ [M + H]⁺: 419.19653, found 419.19543.

4-(Benzyloxy)-6,6-dimethyl-2-phenyl-3-(p-tolyl)-1,2,4-oxadiazinan-5-one (3ac). White solid, yield: 37.4 mg, 93%; mp = 131.3–132.0 °C; ¹H NMR (400 MHz,CDCl₃): δ 7.38–7.28 (m, 5H), 7.20–7.15 (m, 2H), 7.12 (d, J = 8.0 Hz, 2H), 7.06 (d, J = 8.0 Hz, 2H), 6.98–6.94 (m, 1H), 6.81 (d, J = 8.0 Hz, 2H), 5.61 (s, 1H), 5.15 (d, J = 10.0 Hz, 1H), 4.64 (d, J = 10.0 Hz, 1H), 2.31 (s, 3H), 1.71 (s, 3H), 1.66 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.2, 145.6, 139.1, 134.9, 130.6, 130.0, 128.9, 128.8, 128.7, 128.6, 128.5, 123.6, 118.2, 83.4, 83.2, 77.3, 24.0, 23.5, 21.2 ppm; HRMS (ESI) calculated for C₂₅H₂₇N₂O₃ [M + H]⁺: 403.20162, found 403.20087.

4-(Benzyloxy)-3-(4-chlorophenyl)-6,6-dimethyl-2-phenyl-1,2,4-oxadiazinan-5-one (3ad). White solid, yield: 41.8 mg, 99%; mp = 131.6–132.4 °C; ¹H NMR (400 MHz,CDCl₃): δ 7.39–7.36 (m, 3H), 7.33–7.28 (m, 2H), 7.22–7.16 (m, 4H), 7.11 (d, J = 8.0 Hz, 2H), 7.00–6.96 (m, 1H), 6.77 (d, J = 4.0 Hz, 2H), 5.60 (s, 1H), 5.15 (d, J = 10.0 Hz, 1H), 4.71 (d, J = 10.0 Hz, 1H), 1.70 (s, 3H), 1.66 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 145.3, 135.2, 134.8, 132.1, 130.1, 130.0, 129.0, 128.8, 128.6, 128.3, 124.0, 118.1, 83.7, 77.4, 24.0, 23.4 ppm; HRMS (ESI) calculated for C₂₄H₂₄ClN₂O₃ [M + H]⁺: 423.14700, found 423.14612.

4-(Benzyloxy)-3-(3-methoxyphenyl)-6,6-dimethyl-2-phenyl-1,2,4-oxadiazinan-5-one (3ae). White solid, yield: 37.1 mg, 89%; mp = 98.7–99.2 °C;

¹H NMR (400 MHz, CDCl₃): δ 7.37 (d, J = 4.0 Hz, 3H), 7.33 (d, J = 4.0 Hz, 2H), 7.28–7.16 (m, 3H), 6.98–6.95 (m, 1H), 6.85–6.78 (m, 5H), 5.61 (s, 1H), 5.17 (d, J = 10.0 Hz, 1H), 4.67 (d, J = 10.0 Hz, 1H), 3.70 (s, 3H), 1.72 (s, 3H), 1.67 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.2, 159.2, 145.6,

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135.1, 134.9, 130.0, 129.0, 128.9, 128.6, 128.5, 123.6, 121.3, 118.1, 114.9, 114.4, 83.6, 83.5, 77.4, 55.1, 24.0, 23.4 ppm; HRMS (ESI) calculated for $C_{25}H_{27}N_2O_4$ [M + H]⁺: 419.19653, found 419.19495.

4-(Benzyloxy)-3-(3-chlorophenyl)-6,6-dimethyl-2-phenyl-1,2,4oxadiazinan-5-one (3af). White solid, yield: 41.8 mg, 99%; mp = 99.9–100.4 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.40–7.38 (m, 3H), 7.34 (s, 2H), 7.26-7.14 (m, 5H), 7.04 (d, J = 7.6 Hz, 1H), 7.00-6.97(m, 1H), 6.78 (d, J = 7.6 Hz, 2H), 5.54 (s, 1H), 5.16 (d, J = 10.0 Hz,1H), 4.70 (d, J = 10.0 Hz, 1H), 1.71 (s, 3H), 1.66 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.2, 160.2, 145.7, 135.0, 130.2, 129.9, 128.9, 128.6, 128.5, 125.7, 123.6, 118.4, 113.5, 83.4, 77.3, 55.2, 24.0, 23.5 ppm; HRMS (ESI) calculated for $C_{24}H_{24}ClN_2O_3$ $[M + H]^+$: 423.14700, found 423.14612.

4-(Benzyloxy)-3-(2-methoxyphenyl)-6,6-dimethyl-2-phenyl-1,2,4-oxadiazinan-5-one (3ag). White solid, yield: 31.8 mg, 76%; mp = 101.2–102.0 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.66 (d, J = 7.2 Hz, 1H, 7.34 (d, J = 3.6 Hz, 3H), 7.31 (m, 2H), 7.25 (d, J = 3.6 Hz, 3H)8.0 Hz, 1H), 7.17–7.13 (m, 2H), 6.95 (d, J = 8.0 Hz, 2H), 6.90 (d, J =10.0 Hz, 2H), 6.73 (d, J = 10.0 Hz, 1H), 6.47 (s, 1H), 5.15 (d, J =8.0 Hz, 1H), 4.67 (d, J = 8.0 Hz, 1H), 3.52 (s, 3H), 1.74 (s, 3H), 1.66 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.3, 157.9, 145.7, 134.9, 130.3, 129.7, 128.7, 128.3, 128.1, 123.4, 122.4, 120.5, 118.4, 110.6, 83.3, 77.2, 55.2, 24.0, 23.5 ppm; HRMS (ESI) calculated for $C_{25}H_{27}N_2O_4[M+H]^+$: 419.19653, found 419.19522.

4-(Benzyloxy)-3-(2-chlorophenyl)-6,6-dimethyl-2-phenyl-1,2,4oxadiazinan-5-one (3ah). White solid, yield: 41.7 mg, 99%; mp = 92.6–93.2 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.81 (d, J = 6.8.2 Hz, 1H), 7.32 (d, J = 6.4 Hz, 5H), 7.21-7.17 (m, 3H), 7.15-7.11 (m, 2H), 6.98-6.95 (m, 1H), 6.90 (d, J = 6.0 Hz, 2H), 6.42 (s, 1H), 5.09(d, J = 9.6 Hz, 1H), 4.62 (d, J = 10.0 Hz, 1H), 1.73 (s, 3H), 1.62 (s, 3H), 1.62 (s, 3H), 1.63 (s, 3H), 1.64 (s, 33H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.3, 145.0, 134.9, 134.5, 131.8, 130.5, 130.4, 129.9, 129.8, 129.4, 129.0, 128.6, 128.5, 127.0, 124.6, 118.8, 83.5, 77.5, 24.1, 23.5 ppm; HRMS (ESI) calculated for $C_{24}H_{24}ClN_2O_3$ [M + H]⁺: 423.14700, found 423.14548.

4-(Benzyloxy)-3-(4-bromophenyl)-6,6-dimethyl-2-phenyl-1,2,4oxadiazinan-5-one (3ai). White solid, yield: 45.2 mg, 97%; mp = 109.8–110.2 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.39–7.35 (m, 5H), 7.32 (d, J = 6.8 Hz, 2H), 7.18 (t, J = 8.0 Hz, 2H), 7.04 (d, J = 8.0 Hz, 2H)2H), 6.98 (t, J = 7.2 Hz, 1H), 6.77 (d, J = 8.0 Hz, 2H), 5.55 (s, 1H), 5.15 (d, J = 10.0 Hz, 1H), 4.71 (d, J = 10.0 Hz, 1H), 1.70 (s, 3H), 1.66 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.5, 145.3, 134.8, 132.6, 131.3, 130.3, 129.9, 129.0, 128.8, 128.6, 123.5, 118.2, 83.7, 77.4, 23.9, 23.4 ppm; HRMS (ESI) calculated for $C_{24}H_{24}BrN_2O_3 [M + H]^+$: 467.09648, found 467.09604.

4-(Benzyloxy)-3-(4-fluorophenyl)-6,6-dimethyl-2-phenyl-1,2,4oxadiazinan-5-one (3aj). White solid, yield: 45.2 mg, 87%; mp = 98.0–98.8 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.38 (d, J = 6.0 Hz, 3H), 7.32 (d, J = 5.6 Hz, 2H), 7.20-7.16 (m, 4H), 7.00-6.91 (m, 3H), 6.79 (d, J = 7.6 Hz, 2H), 5.59 (s, 1H), 5.16 (d, J = 10.4 Hz, 1H), 4.68 (d, J = 10.0 Hz, 1H), 1.71 (s, 3H), 1.67 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 164.4, 161.9, 145.4, 134.8, 130.6, 130.5, 129.9, 129.4, 129.0, 128.7, 128.6, 123.9, 118.3, 115.2, 115.0, 83.6, 77.4, 24.0, 23.4 ppm; HRMS (ESI) calculated for $C_{24}H_{24}FN_2O_3[M+H]^+$: 407.17655, found 407.17587.

4-(4-(Benzyloxy)-6,6-dimethyl-5-oxo-2-phenyl-1,2,4-oxadiazinan-3-yl)benzonitrile (3ak). White solid, yield: 40.8 mg, 99%; mp = 123.1–123.7 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.49 (d, I = 8.4 Hz, 2H), 7.44-7.36 (m, 3H), 7.33-7.28 (m, 2H), 7.21-7.16 (m, 4H), 6.99 (t, J = 8.0 Hz, 1H), 6.73 (d, J = 8.0 Hz, 2H), 5.56 (s, 1H), 5.17 (d, I = 10.4 Hz, 1H), 4.78 (d, I = 10.4 Hz, 1H), 1.69 (s, 3H), 1.67 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 145.0, 138.5, 134.7, 131.7, 129.9, 129.4, 129.3, 129.2, 129.0, 128.9, 128.7, 124.2, 118.3, 117.9, 113.1, 83.9, 83.4, 77.4, 23.9, 23.4 ppm; HRMS (ESI) calculated for $C_{25}H_{24}N_3O_3$ [M + H]⁺: 414.18122, found 414.18063.

4-(Benzyloxy)-6,6-dimethyl-3-(4-nitrophenyl)-2-phenyl-1,2,4oxadiazinan-5-one (3al). White solid, yield: 45.2 mg, 97%; mp = 137.2-138.0 °C; ¹H NMR (400 MHz, CDCl₃): δ 8.04 (d, J = 8.4 Hz, 2H), 7.41–7.35 (m, 3H), 7.32 (d, J = 6.8 Hz, 2H), 7.26 (d, J = 6.8 Hz, 2H = 8.4 Hz, 2H, 7.20-7.16 (m, 2H), 7.01-6.97 (m, 1H), 6.74 (d, J =8.0 Hz, 2H), 5.62 (s, 1H), 5.17 (d, J = 10.4 Hz, 1H), 4.80 (d, J = 10.10.4 Hz, 1H), 1.70 (s, 3H), 1.68 (s, 3H) ppm; ¹³C NMR (100 MHz, $CDCl_3$): δ 170.6, 148.3, 144.9, 140.4, 134.7, 130.0, 129.5, 129.2, 128.9, 128.7, 124.3, 123.1, 118.0, 84.0, 83.1, 77.4, 24.0, 23.4 ppm; HRMS (ESI) calculated for $C_{24}H_{24}N_3O_5$ [M + H]⁺: 434.17105, found 434.17041.

2-Benzyl-4-(benzyloxy)-6,6-dimethyl-3-phenyl-1,2,4-oxadiazinan-5-one (3am). White solid, yield: 32.4 mg, 81%; mp = 98.1-98.7 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.47 (s, 5H), 7.33–7.29 (m, 4H), 7.27 (d, J = 5.2 Hz, 2H), 7.21-7.19 (m, 2H), 7.14 (d, J = 6.0 Hz, 2H), 5.11(s, 1H), 5.02 (d, J = 9.2 Hz, 1H), 4.41 (d, J = 9.2 Hz, 1H), 3.63 (d, J =13.6 Hz, 1H), 3.47 (d, J = 13.6 Hz, 1H), 1.47 (s, 3H), 1.41 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.6, 136.1, 134.9, 134.2, 129.8, 129.7, 129.4, 129.2, 128.8, 128.6, 128.3, 128.0, 127.4, 81.9, 77.2, 57.2, 24.0, 23.7 ppm; HRMS (ESI) calculated for $C_{25}H_{27}N_2O_3$ $[M + H]^+$: 403.20162, found 403.20114.

2-Benzyl-4-(benzyloxy)-3-ethyl-6,6-dimethyl-1,2,4-oxadiazinan-5-one (3an). White solid, yield: 27.3 mg, 77%; mp = 92.3-93.2 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.47–7.45 (m, 2H), 7.40–7.38 (m, 3H), 7.33-7.30 (m, 5H), 5.00 (s, 2H), 4.23-4.21 (m, 1H), 3.99 (d, J = 13.2 Hz, 1H), 3.60 (d, J = 13.2 Hz, 1H), 2.06-2.09 (m, 1H), 1.80-1.75 (m, 1H), 1.29 (s, 3H), 1.27 (s, 3H), 1.06–1.02 (m, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 136.2, 135.1, 129.8, 129.4, 128.9, 128.5, 128.1, 127.5, 81.9, 81.4, 76.4, 57.4, 23.9, 23.5, 21.5 ppm; HRMS (ESI) calculated for $C_{21}H_{27}N_2O_3$ [M + H]⁺: 355.20162, found 355.20074.

4-(Benzyloxy)-2,6,6-trimethyl-3-phenyl-1,2,4-oxadiazinan-5one (3ao). White solid, yield: 30.3 mg, 93%; mp = 67.7-68.5 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.44–7.40 (m, 5H), 7.31– 7.28 (m, 3H), 7.10 (d, J = 6.0 Hz, 2H), 5.00 (d, J = 9.6 Hz, 1H), 4.90 (s, 1H), 4.39 (d, J = 10.0 Hz, 1H), 2.35 (s, 3H), 1.71 (s, 3H),1.51 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.4, 134.8, 134.1, 129.8, 129.7, 129.1, 128.7, 128.6, 128.3, 87.6, 81.9, 77.1, 41.3, 24.1, 23.9 ppm; HRMS (ESI) calculated for C₁₉H₂₃N₂O₃ $[M + H]^+$: 327.17032, found 327.16949.

4-(Benzyloxy)-6,6-dimethyl-3-(naphthalen-1-yl)-2-phenyl-1,2,4oxadiazinan-5-one (3ap). White solid, yield: 40.3 mg, 92%; mp = 143.8–144.6 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.83–7.80 (m, 1H), 7.78–7.74 (m, 2H), 7.65 (s, 1H), 7.52–7.49 (m, 2H), 7.45 (d, J =8.4 Hz, 1H), 7.37–7.31 (m, 3H), 7.27 (d, J = 6.4 Hz, 2H), 7.18–7.14

(m, 2H), 6.96–6.93 (m, 1H), 6.87 (d, J=8.0 Hz, 2H), 5.83 (s, 1H), 5.17 (d, J=10.0 Hz, 1H), 4.63 (d, J=10.0 Hz, 1H), 1.80 (s, 3H), 1.72 (s, 3H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.5, 145.6, 134.9, 133.7, 132.6, 131.2, 130.0, 129.1, 128.9, 128.7, 128.5, 128.2, 128.0, 127.7, 126.7, 126.3, 125.7, 123.8, 118.5, 83.6, 77.4, 24.1, 23.5 ppm; HRMS (ESI) calculated for $C_{28}H_{27}N_2O_3$ [M + H] † : 439.20162, found 439.20071.

4-(Benzyloxy)-3-(furan-2-yl)-6,6-dimethyl-2-phenyl-1,2,4-oxadiazinan-5-one (3aq). White solid, yield: 37.3 mg, 99%; mp = 120.8-121.5 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.40–7.28 (m, 6H), 7.24 (t, J = 8.0 Hz, 2H), 7.00 (t, J = 7.2 Hz, 1H), 6.87 (d, J = 8.0 Hz, 2H), 6.30 (s, 2H), 5.75 (s, 1H), 5.18 (d, J = 10.0 Hz, 1H), 4.74 (d, J = 10.0 Hz, 1H), 1.67 (s, 3H), 1.66 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 169.9, 147.8, 145.5, 143.0, 134.8, 129.8, 129.0, 128.7, 128.6, 123.5, 117.0, 111.0, 110.5, 83.5, 77.6, 23.8, 23.4 ppm; HRMS (ESI) calculated for $C_{22}H_{23}N_2O_4$ [M + H]*: 379.16523, found 379.16406.

4-(Benzyloxy)-2,3-diphenyl-1-oxa-2,4-diazaspiro[5.5]**undecan-5-one** (3ca). White solid, yield: 39.4 mg, 92%; mp = 106.7–107.5 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.37–7.35 (m, 3H), 7.31–7.29 (m, 2H), 7.25 (s, 1H), 7.23–7.20 (m, 2H), 7.17 (s, 2H), 7.15–7.13 (m, 2H), 6.94–6.90 (m, 1H), 6.79 (d, J = 8.0 Hz, 2H), 5.56 (s, 1H), 5.14 (d, J = 10.0 Hz, 1H), 4.63 (d, J = 10.0 Hz, 1H), 2.36–2.33 (m, 1H), 2.18–2.16 (m, 2H), 1.87–1.86 (m, 1H), 1.70–1.68 (m, 2H), 1.60–1.56 (m, 2H), 1.47–1.42 (m, 1H), 1.40–1.32 (m, 1H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 146.0, 135.0, 133.6, 129.9, 129.2, 128.9, 128.8, 128.6, 128.5, 127.9, 123.3, 84.7, 83.9, 77.3, 32.7, 32.0, 30.0, 29.7, 25.0, 24.5, 22.0, 21.4, 21.2 ppm; HRMS (ESI) calculated for C₂₇H₂₉N₂O₃ [M + H]⁺: 429.21727, found 429.21619.

4-(Benzyloxy)-2,3,6-triphenyl-1,2,4-oxadiazinan-5-one (3ea). Colorless oil, yield: 24.4 mg, 56%; 1 H NMR (400 MHz, CDCl₃): δ 7.54–7.52 (m, 2H), 7.43–7.41 (m, 2H), 7.36–7.34 (m, 5H), 7.32–7.31 (m, 4H), 7.26–7.24 (m, 1H), 7.20–7.16 (m, 2H), 7.12–7.09 (m, 1H), 7.01–6.98 (m, 1H), 6.83–6.79 (m, 2H), 5.84 (d, J=54.8 Hz, 1H), 5.60 (d, J=41.6 Hz, 1H), 5.19–5.14 (m, 1H), 4.72 (d, J=9.6 Hz, 1H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 166.1, 165.5, 145.3, 144.9, 135.3, 134.9, 134.8, 134.0, 132.9, 130.0, 129.8, 129.6, 129.4, 129.2, 129.1, 129.0, 128.9, 128.8, 128.7, 128.6, 128.5, 128.4, 128.3, 128.2, 128.1, 125.3, 124.3, 120.5, 118.3, 85.2, 83.2, 83.1, 77.7 ppm; HRMS (ESI) calculated for $C_{28}H_{25}N_2O_3$ [M + H] $^+$: 437.18597, found 437.18463.

2-Benzyl-4-(benzyloxy)-3-phenyl-1-oxa-2,4-diazaspiro[5.5]-undecan-5-one (3cm). White solid, yield: 34.5 mg, 78%; mp = 76.4–77.2 °C; 1 H NMR (400 MHz, CDCl₃): δ 7.45 (s, 5H), 7.29–7.25 (m, 6H), 7.20–7.18 (m, 2H), 7.11 (d, J = 5.2 Hz, 2H), 5.06 (s, 1H), 4.98 (d, J = 10.0 Hz, 1H), 4.36 (d, J = 4.4 Hz, 1H), 3.58 (d, J = 11.6 Hz, 1H), 3.45 (d, J = 13.2 Hz, 1H), 2.20–2.16 (m, 1H), 1.89 (s, 2H), 1.73–1.71 (m, 1H), 1.42–1.35 (m, 3H), 1.26 (s, 1H), 1.12–1.09 (m, 2H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.6, 136.5, 135.0, 134.4, 129.9, 129.8, 129.7, 129.4, 129.2, 128.8, 128.7, 128.6, 128.5, 128.3, 128.0, 127.5, 83.1, 82.8, 77.4, 77.0, 76.7, 57.5, 32.7, 29.7, 24.9, 24.5, 22.1, 21.5, 19.9 ppm; HRMS (ESI) calculated for $C_{28}H_{31}N_2O_3$ [M + H] $^+$: 443.23292, found 443.23169.

4-(Benzyloxy)-3-(2-methoxyphenyl)-2-phenyl-1-oxa-2,4-diaza-spiro[5.5]undecan-5-one (3cg). White solid, yield: 30.7 mg,

67%; mp = 87.1–87.7 °C; 1 H NMR (400 MHz, CDCl₃): δ 7.61–7.59 (m, 1H), 7.33–7.30 (m, 5H), 7.23–7.19 (m, 1H), 7.14–7.10 (m, 2H), 6.92–6.88 (m, 4H), 6.76 (d, J = 10.0 Hz, 1H), 6.40 (s, 1H), 5.13 (d, J = 10.0 Hz, 1H), 4.65 (d, J = 10.0 Hz, 1H), 3.47 (s, 3H), 2.20–2.16 (m, 2H), 1.83–1.74 (m, 1H), 1.70–1.67 (m, 3H), 1.63–1.54 (m, 2H), 1.49–1.46 (m, 1H), 1.42 (s, 1H) ppm; 13 C NMR (100 MHz, CDCl₃): δ 170.3, 157.8, 146.0, 135.0, 130.2, 129.8, 129.7, 128.7, 128.3, 128.0, 123.0, 122.4, 120.3, 117.6, 110.5, 84.5, 77.2, 55.1, 32.1, 29.9, 29.7, 27.0, 25.1, 21.4, 21.3 ppm; HRMS (ESI) calculated for $C_{28}H_{31}N_2O_4$ [M + H] † : 459.22783, found 459.22702.

4-(Benzyloxy)-3-(2-chlorophenyl)-2-phenyl-1-oxa-2,4-diazaspiro-[5.5]**undecan-5-one** (3**ch**). White solid, yield: 39.7 mg, 86%; mp = 96.9–97.5 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.81 (d, J = 7.2 Hz, 1H), 7.32 (s, 5H), 7.23–7.17 (m, 3H), 7.13 (d, J = 8.0 Hz, 2H), 6.93 (d, J = 7.6 Hz, 3H), 6.39 (s, 1H), 5.10 (d, J = 10.0 Hz, 1H), 4.62 (d, J = 10.0 Hz, 1H), 2.18–2.16 (m, 2H), 1.87 (s, 1H), 1.72–1.68 (m, 3H), 1.63–1.57 (m, 2H), 1.50–1.42 (m, 1H), 1.40–1.33 (m, 1H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 145.4, 134.9, 134.6, 131.9, 130.5, 130.3, 129.9, 129.3, 129.2, 128.8, 128.6, 128.5, 128.4, 126.8, 84.7, 77.4, 77.2, 77.0, 76.7, 32.7, 32.0, 30.1, 29.7, 25.0, 24.5, 22.1, 21.3, 21.2 ppm; HRMS (ESI) calculated for C₂₇H₂₈ClN₂O₃ [M + H][†]: 463.17830, found 463.17780.

4-(Benzyloxy)-3-(naphthalen-1-yl)-2-phenyl-1-oxa-2,4-diaza-spiro[5.5]undecan-5-one (3cp). White solid, yield: 43.0 mg, 90%; mp = 123.5–124.0 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.77–7.75 (m, 1H), 7.73–7.71 (m, 1H), 7.69–7.67 (m, 1H), 7.56 (s, 1H), 7.45–7.43 (m, 2H), 7.36–7.30 (m, 4H), 7.26–7.24 (m, 2H), 7.12 (t, J = 8.4 Hz, 2H), 6.89 (t, J = 7.2 Hz, 1H), 6.83 (d, J = 8.0 Hz, 2H), 5.73 (s, 1H), 5.15 (d, J = 10.4 Hz, 1H), 4.61 (d, J = 10.4 Hz, 1H), 2.39–2.36 (m, 1H), 2.22 (s, 2H), 1.90–1.89 (m, 1H), 1.71–1.68 (m, 3H), 1.61–1.57 (m, 2H), 1.49–1.45 (m, 1H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 145.9, 134.9, 134.8, 133.6, 132.6, 131.2, 130.0, 129.3, 129.0, 128.9, 128.7, 128.6, 128.5, 128.4, 128.2, 127.8, 127.6, 126.6, 126.2, 125.7, 123.8, 117.5, 84.8, 77.4, 32.7, 32.1, 30.0, 27.0, 25.0, 24.5, 22.1, 21.4, 21.3 ppm; HRMS (ESI) calculated for C₃₁H₃₁N₂O₃ [M + H]⁺: 479.23292, found 479.23166.

4-(Benzyloxy)-3-(furan-2-yl)-2-phenyl-1-oxa-2,4-diazaspiro[5.5]-**undecan-5-one** (**3cq**). White solid, yield: 25.5 mg, 61%; mp = 77.5–78.2 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.38 (s, 6H), 7.26–7.20 (m, 2H), 6.97 (t, J = 7.2 Hz, 1H), 6.87–6.85 (m, 2H), 6.24 (d, J = 1.2 Hz, 2H), 5.65 (s, 1H), 5.16 (d, J = 10.4 Hz, 1H), 4.42 (d, J = 10.4 Hz, 1H), 2.35–2.32 (m, 1H), 2.17–2.12 (m, 1H), 2.07–2.04 (m, 1H), 1.83–1.81 (m, 1H), 1.72–1.65 (m, 3H), 1.64–1.57 (m, 2H), 1.51–1.47 (m, 1H) ppm; ¹³C NMR (100 MHz, CDCl₃): δ 169.9, 147.9, 145.9, 142.9, 134.8, 129.9, 129.3, 128.9, 128.7, 128.6, 128.5, 123.3, 116.7, 110.8, 110.4, 84.6, 77.6, 32.6, 31.9, 29.7, 25.0, 24.5, 22.0, 21.4, 21.2 ppm; HRMS (ESI) calculated for C₂₅H₂₇N₂O₄ [M + H][†]: 419.19653, found 419.19577.

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