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Enhanced visible light response of a WO₃ photoelectrode with an immobilized fibrous gold nanoparticle assembly using an amyloid- β peptide†

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A WO₃ photoelectrode immobilizing a fibrous gold nanoparticle (AuNP) assembly using an amyloid- β (A β) peptide was constructed and its ability for photocurrent generation upon visible light irradiation was investigated. AuNPs attached using A β peptide (A β -AuNP) assemble with a fibrous structure at pH 4.5, which disperses at pH 11. The modified A β -AuNPs are immobilized on the surface of WO₃ fabricated on a fluorine-doped tin oxide (FTO) electrode (A β -AuNP_{ass}@WO₃/FTO or A β -AuNP_{dis}@WO₃/FTO, respectively). The photocurrent generation response of the A β -AuNP_{ass}@WO₃/FTO electrode is clearly increased relative to that of A β -AuNP_{dis}@WO₃/FTO upon visible light irradiation (λ = 420–750 nm), indicating that the fibrous AuNP assembly enhances the visible light response of the WO₃ photoelectrode.

Received 17th November 2016
Accepted 28th November 2016

DOI: 10.1039/c6ra26916h

www.rsc.org/advances

Metal nanoparticle–semiconductor (MNP–SC) composites have emerged as a novel class of photocatalysts for a wide range of chemical reactions and photoelectrochemical conversions.^{1–8} Although a variety of semiconductor photocatalysts are known to function under UV light, strategies for improving the visible light response of photocatalysts are required for the efficient use of solar energy. WO₃ is an attractive n-type semiconductor photocatalyst providing a positive conduction band edge lower than the water oxidation potential. However, the band gap energy of WO₃ (~2.6 eV) is sufficiently large to absorb visible light. To improve the visible light response, we investigate a WO₃ photocatalyst capable of immobilizing noble metal nanoparticles that have strong surface plasmon resonance (SPR).^{9–13}

An interfacial structure between semiconductor and nanoparticles in hybrid composites has been identified as crucial feature that improves reactivities and photochemical properties of MNP–SCs.^{14–16} The properties of immobilized MNPs including SPR absorption, refractive index, and light scattering, which are affected by MNP composition, size, and shape, are important parameters to explore in development of new

features in MNP–SC materials. We have therefore focused on developing a method to immobilize MNP assemblies on the surface of SCs. Recently, we reported a TiO₂ composite which immobilizes AuNP assemblies *via* the biotin–streptavidin interaction. This enables efficient electron transfer from TiO₂ to immobilized AuNPs as indicated by photocatalytic activity.¹⁷ An amyloid- β (A β) peptide, which accumulates as the major constituent of the extracellular deposits in Alzheimer's disease, forms fibrous aggregates.¹⁸ It is known that the A β peptide is useful in assembling the metal nanoparticles when linked to the surface of the metal,^{19,20} thereby the A β peptide is advantageous to provide tight interaction between AuNPs. Here, we report a new method for preparing AuNP–WO₃ composites with fibrous AuNP assemblies using amyloid formation promoted by the A β peptide (Fig. 1). The unique AuNP assembly on the WO₃ surface further improves the photocurrent generation upon visible light irradiation.

The soluble A β peptide fragment (consisting of residues 1–16 of amyloid- β)¹⁹ whose aggregation depends on pH was linked to a GGGC fragment at the C-terminus for immobilization on the AuNP surface. This component was synthesized by standard solid-phase peptide synthesis. The crude peptide was purified by HPLC and characterized by ESI-TOF MS (Fig. S1, ESI†). The folding and assembling properties of the A β peptide with the additional fragment were investigated by obtaining CD measurements (Fig. S2, ESI†). The A β peptide exhibits a strong negative minimum at 205 nm both at pH 11 and 7, suggesting prevalent random coil structure. In contrast, A β has a positive maximum at 220 nm in a buffer adjusted to pH 4.5. This feature is consistent with the reported result.¹⁹ The result indicates that the modified A β peptide retains the inherent ability to assemble as β -sheet.

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† Electronic supplementary information (ESI) available: Information on instrumentation, synthesis of A β peptide, preparation of electrodes, and additional data on photocurrent generation experiments. See DOI: 10.1039/c6ra26916h

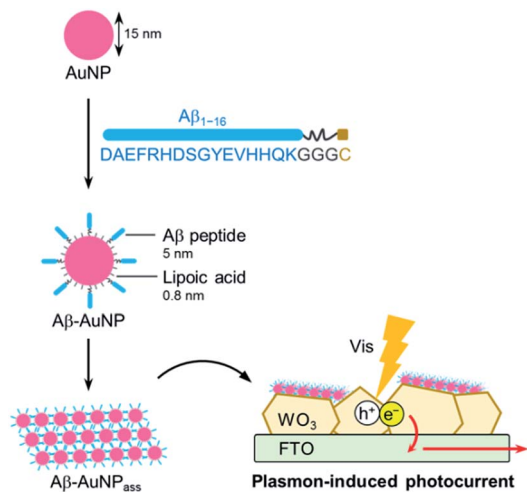


Fig. 1 Preparation of a WO_3 composite with fibrous AuNP assembly using amyloid- β .

The $\text{A}\beta$ peptide was immobilized on the surface of the gold nanoparticle through an Au-S bond with the C-terminal Cys residue. AuNP modified with a lipoic acid was treated with the $\text{A}\beta$ peptide. The mixed solution of AuNP and the $\text{A}\beta$ peptide was incubated for 24 h at 25 °C. Immobilization of the peptide *via* a Cys moiety on the AuNP surface was confirmed by agarose gel electrophoresis (Fig. 2a). The negatively-charged AuNP covered with lipoic acid runs smoothly towards the anode. Interestingly, $\text{A}\beta$ -AuNP has clearly shifted bands due to its increased size and

decreased negative charge (Fig. 2a). In addition, titration experiments with gel electrophoresis indicate that approximately 300 strands of the $\text{A}\beta$ peptide are connected to each AuNP particle. This result clearly shows that the $\text{A}\beta$ peptides modify the surface of the AuNP particle *via* the Cys residue.

The assembly properties of $\text{A}\beta$ -AuNP were analyzed by SPR absorption at different pH values including pH 11, 7, and 4.5 (Fig. 2b). The SPR peak of the AuNP was observed at 523 nm at pH 11 and 7. The red color of $\text{A}\beta$ -AuNP changes to purple at pH 4.5 and an intense absorption peak arises at 675 nm. The red-shifted absorption indicates that AuNPs are assembled and that the average of the interparticle distance is shorter under acidic condition. The assembled structure of $\text{A}\beta$ -AuNP was analyzed by atomic force microscopy (AFM) (Fig. 2c-e). The sample solution was drop-casted on a glass surface for the measurement. When we observed $\text{A}\beta$ -AuNP at pH 11 and 7, the AuNPs were found to be uniformly connected and immobilized on the surface. In contrast, we found that the $\text{A}\beta$ -AuNPs are assembled as a fibrous structure at pH 4.5. The AuNP tightly assembled as a nanofiber provides a significantly shifted SPR absorption.

The assembling feature of $\text{A}\beta$ -AuNP was further analyzed by small angle X-ray scattering (SAXS) (Fig. 3). AuNPs modified only with lipoic acid (LA-AuNP) are compared as a reference (Fig. S3†). Interestingly, the SAXS profiles obtained from the dropcast sample on a polyimide film show peaks in a low q region ($q < 0.5 \text{ nm}^{-1}$), which was not found in the profiles in solution. These new peaks indicate the formation of the AuNP assembly. The d -spacing of $\text{A}\beta$ -AuNP (17.2 nm at pH 11, 16.1 nm

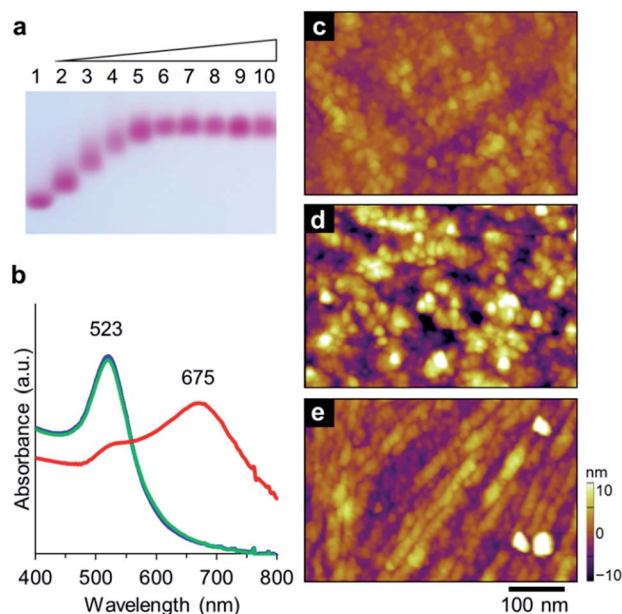


Fig. 2 (a) Agarose gel electrophoresis of $\text{A}\beta$ -modified AuNPs. The concentration of $\text{A}\beta$ increases from lane 1 to 10 (0, 80, 160, 240, 320, 390, 470, 550, 630, and 710 eq.) vs. AuNP (per particle). (b) UV/vis absorption spectra of $\text{A}\beta$ -AuNP at pH 11 (blue), pH 7.0 (green), and pH 4.5 (red) in 10 mM KPi or NaOAc buffer. AFM images of $\text{A}\beta$ -AuNP at (c) pH 11, (d) pH 7.0, and (e) pH 4.5. The sample solution at an adjusted pH was casted onto a glass plate.

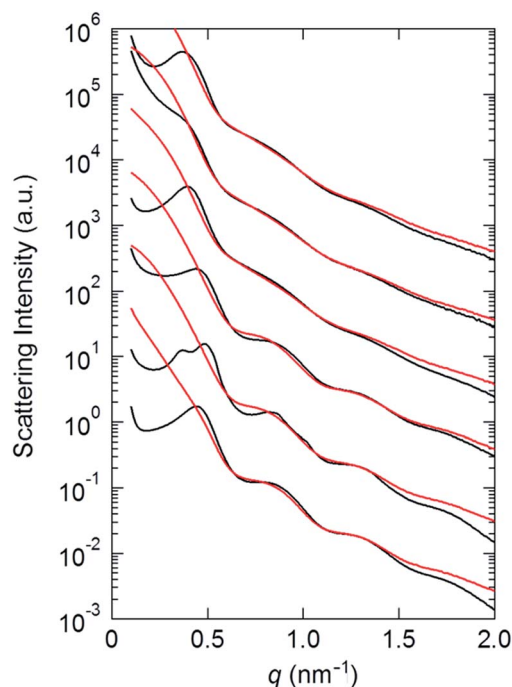


Fig. 3 SAXS profiles plotted as scattering intensity $I(q)$ vs. magnitude of scattering vector (q). The SAXS profile of the sample in cast film (black line) and in solution (red). From top; the profiles of $\text{A}\beta$ -AuNP at pH 11, 7, 4.5, and LA-AuNP without $\text{A}\beta$ peptide at pH 11, 7, 4.5.



at pH 7 and pH 4.5) increases relative to the values in LA-AuNP (14.3 nm at pH 11, 16.1 and 15.9 nm at pH 7, 14.3 nm in pH 4.5). This result suggests that the modification of the AuNP with the A β peptide provides a unique assembly mode. In addition, the observed *d* value of 17.2 nm for A β -AuNP at pH 11 decreases to 15.9 nm at pH 4.5. This appears to correspond with the fibrous morphology that we observed in AFM imaging.

The AuNP assembly using the A β peptide was immobilized on WO₃ fabricated on fluorine doped tin oxide (FTO) electrode. The crystal of WO₃ was grinded in a solution of 0.5% diacetyl cellulose using a ball mill. The resulting solution of WO₃ was drop-casted onto the FTO electrode and the electrode was heated to 450 °C and incubated for 2 h. The WO₃/FTO electrode was then immersed in the solution of A β -AuNP at pH 4.5 or pH 11 to immobilize AuNPs, and the two electrodes, A β -AuNP_{ass}@WO₃/FTO and A β -AuNP_{dis}@WO₃/FTO, were prepared. The electrode immobilizing AuNP without the A β peptide was also prepared (LA-AuNP@WO₃/FTO). SEM analysis indicates that the AuNPs with the size of *ca.* 15 nm are immobilized on the surface of A β -AuNP_{ass}@WO₃/FTO (Fig. 4a and b). The elemental mapping by EDX also supports the existence of Au and S atoms of AuNPs on the surface (Fig. S4, ESI†). The AuNP assembly was observed in large area of the WO₃ surface in A β -AuNP_{ass}@WO₃/FTO. A similar large area of AuNP was not found in A β -AuNP_{dis}@WO₃/FTO or LA-AuNP@WO₃/FTO immobilizing AuNP without the A β peptide (Fig. S5, ESI†).

Immobilization of A β -AuNP on the surface of the WO₃/FTO electrode was confirmed by diffuse reflection spectra (Fig. 4c). WO₃ absorbs UV-light ($\lambda < 400$ nm), and, A β -AuNP_{dis}@WO₃/FTO has SPR absorption at 523 nm. In contrast, the WO₃ electrode

immobilizing the A β -AuNP assembly (A β -AuNP_{ass}@WO₃/FTO) has the peak maxima shifted to 590 and 680 nm. The SPR redshift is consistent with that observed in solution, suggesting that the A β -AuNP forms a fibrous assembly as visualized in AFM experiments.

Photocurrent measurements of A β -AuNP_{ass}@WO₃/FTO and A β -AuNP_{dis}@WO₃/FTO electrodes were performed to investigate sensitization of fibril-like assemblies of AuNP. Photocurrent response patterns of the AuNP-immobilized electrodes were measured during on/off cycles of visible light (420–750 nm) in the presence of 2-propanol as an electron donor (Fig. 5a). We observed clear rise and fall of the anodic current under the bias potential of +0.3 mV (*vs.* Ag|AgCl). The A β -AuNP_{ass}@WO₃/FTO electrode shows photocurrent density of *ca.* 54 $\mu\text{A cm}^{-2}$ in the fifth irradiation, which is significantly increased relative to that found in WO₃/FTO electrode (*ca.* 24 $\mu\text{A cm}^{-2}$) and LA-AuNP@WO₃/FTO (*ca.* 25 $\mu\text{A cm}^{-2}$). Interestingly, the A β -AuNP@WO₃/FTO electrode prepared at pH 11 has considerably lower photocurrent density of *ca.* 32 $\mu\text{A cm}^{-2}$. The results

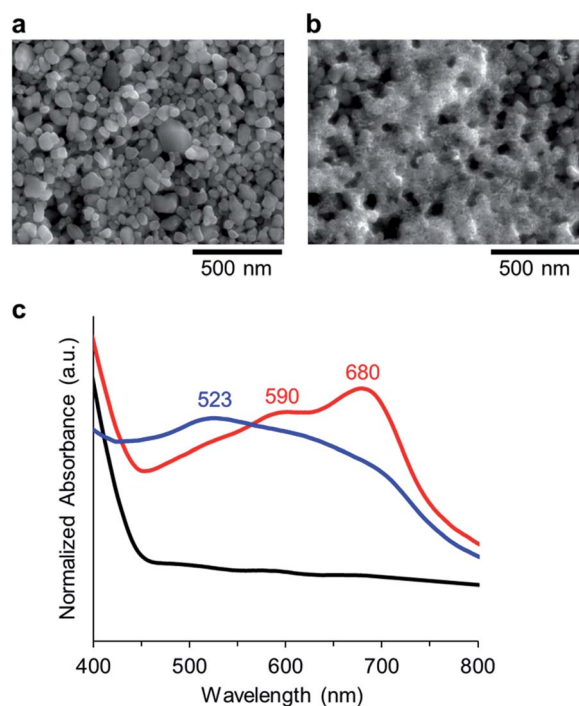


Fig. 4 SEM images of (a) WO₃/FTO, and (b) A β -AuNP_{ass}@WO₃/FTO. (c) Diffuse reflection spectra of WO₃ (black), A β -AuNP_{dis}@WO₃/FTO (blue), and A β -AuNP_{ass}@WO₃/FTO (red).

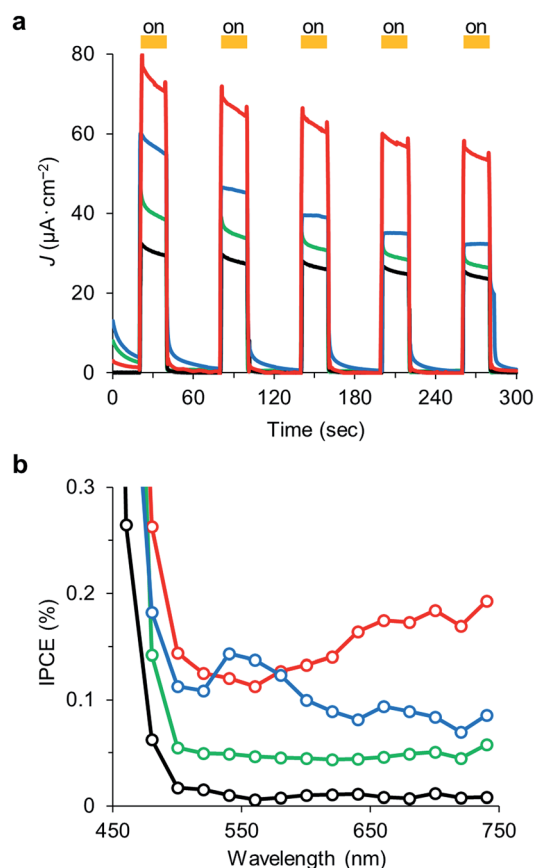


Fig. 5 (a) Photocurrent response of WO₃/FTO (black), LA-AuNP@WO₃/FTO (green), A β -AuNP_{dis}@WO₃/FTO (blue), and A β -AuNP_{ass}@WO₃/FTO (red) during on–off cycles of visible light irradiations ($\lambda = 420$ –750 nm). Photoanodes under constant bias voltage at +300 mV (*vs.* Ag|AgCl) in 1.0 M KOH with 0.5 M 2-propanol. The photocurrent response of A β -AuNP_{ass}@WO₃/FTO decreases to the similar level of WO₃/FTO after a 10 min irradiation. (b) Action spectra of WO₃/FTO (black), AuNP@WO₃/FTO (green), A β -AuNP_{dis}@WO₃/FTO (blue), and A β -AuNP_{ass}@WO₃/FTO (red). IPCE values are averaged through three independent experiments.



indicate that the morphology of AuNP on the WO_3 surface significantly improves the visible light response in photocurrent generation. In addition, the stability of response over the longer cycling period in $\text{A}\beta\text{-AuNP}_{\text{ass}}@\text{WO}_3/\text{FTO}$ is improved relative to that in $\text{A}\beta\text{-AuNP}_{\text{dis}}@\text{WO}_3/\text{FTO}$.

The action spectrum of the $\text{A}\beta\text{-AuNP}_{\text{ass}}@\text{WO}_3/\text{FTO}$ electrode exhibits a gradual increase in current efficiency above 550 nm corresponding to its absorption spectrum (Fig. 5b). In contrast, the spectrum of $\text{A}\beta\text{-AuNP}_{\text{dis}}@\text{WO}_3/\text{FTO}$ has a peak at 550 nm. The WO_3/FTO and $\text{LA-AuNP}@\text{WO}_3/\text{FTO}$ electrodes do not show peak maxima above 500 nm. The results also support the evidence that shifted SPR promotes the photocurrent generation upon the visible light irradiation.

The tight assembly of plasmonic AuNPs immobilized on the WO_3 surface would enhance light absorbance around the absorption edge.¹⁰ In addition, clear but not intense IPCE maximum that corresponds to the plasmon absorption peak suggests the effective generation of hot electrons, leading to photocurrent increase (Fig. 6).¹¹

In conclusion, we prepared an AuNP- WO_3 composite immobilizing the AuNP assembly which is connected using an $\text{A}\beta$ peptide and investigated the photocurrent generation activity of the composites with visible light-irradiation. The AuNPs modified with $\text{A}\beta$ peptides were confirmed to be assembled as a fibrous nanostructure, enabling the absorption of visible light at 650 nm. In addition, the $\text{A}\beta\text{-AuNP}$ assembly fabrication on a FTO electrode is more effective than the composites such as $\text{A}\beta\text{-AuNP}_{\text{dis}}@\text{WO}_3$ or WO_3 without AuNP with respect to photocurrent generation upon visible light irradiation. The results indicate that the fibrous AuNP assembly on the WO_3 surface greatly improves SPR-assisted light absorption of the AuNP- WO_3 composite photoelectrode. This work demonstrates that fabricated noble metal nanoparticle assemblies on semiconductor materials can be precisely designed as visible light responsive photocatalysts. It is expected that further studies of such metal nanoparticle-semiconductor composites using $\text{A}\beta$ peptides will significant insights into our understanding of the photocatalytic behavior of semiconductor nanomaterials.

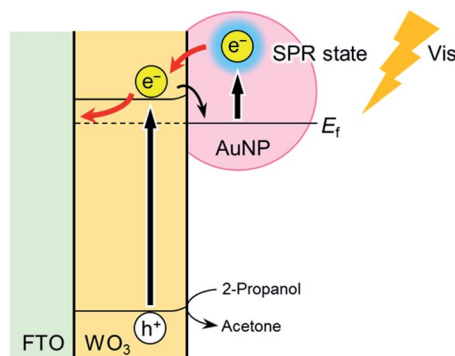


Fig. 6 Schematic representation of surface plasmon-induced photocurrent generation.

Acknowledgements

This work was supported by JSPS KAKENHI Grant Number JP25102527 and JP15H00746 in Innovative Areas “New Polymeric Materials Based on Element Block” to A. O., JSPS KAKENHI Grant Number JP15H05804 in Innovative Areas “Precisely Designed Catalysts with Customized Scaffolding” to T. H. This program was supported by the Strategic International Collaborative Research Program (SICORP), JST. H. H. acknowledges support from the Program for Leading Graduate Schools for Osaka University: Interdisciplinary Program for Biomedical Sciences (IPBS).

Notes and references

- 1 N. Zhou, V. Lopez-Puente, Q. Wang, L. Polavarapu, I. Pastoriza-Santos and Q.-H. Xu, *RSC Adv.*, 2015, **5**, 29076–29097.
- 2 A. Primo, A. Corma and H. Garcia, *Phys. Chem. Chem. Phys.*, 2011, **13**, 886–910.
- 3 S. G. Kumar and L. G. Devi, *J. Phys. Chem. A*, 2011, **115**, 13211–13241.
- 4 P. V. Kamat, *J. Phys. Chem. Lett.*, 2012, **3**, 663–672.
- 5 H. Tada, T. Kiyonaga and S. I. Naya, *Chem. Soc. Rev.*, 2009, **38**, 1849–1858.
- 6 S. Linic, P. Christopher and D. B. Ingram, *Nat. Mater.*, 2011, **10**, 911–921.
- 7 J. S. DuChene, B. C. Sweeny, A. C. Johnston-Peck, D. Su, E. A. Stach and W. D. Wei, *Angew. Chem., Int. Ed.*, 2014, **53**, 7887–7891.
- 8 D. B. Ingram, P. Christopher, J. L. Bauer and S. Linic, *ACS Catal.*, 2011, **1**, 1441–1447.
- 9 D. Hu, P. Diao, D. Xu and Q. Wu, *Nano Res.*, 2016, **9**, 1735–1751.
- 10 R. Solarska, K. Bienkowski, S. Zoladek, A. Majcher, T. Stefaniuk, P. J. Kulesza and J. Augustynski, *Angew. Chem., Int. Ed.*, 2014, **53**, 14196–14200.
- 11 F. Xu, Y. Yao, D. Bai, R. Xu, J. Mei, D. Wu, Z. Gao and K. Jiang, *RSC Adv.*, 2015, **5**, 60339–60344.
- 12 A. Tanaka, K. Hashimoto and H. Kominami, *J. Am. Chem. Soc.*, 2013, **136**, 586–589.
- 13 T. He, Y. Ma, Y. Cao, W. Yang and J. Yao, *Phys. Chem. Chem. Phys.*, 2002, **4**, 1637–1639.
- 14 S. Zou and G. C. Schatz, *Chem. Phys. Lett.*, 2005, **403**, 62–67.
- 15 Z. Liu, W. Hou, P. Pavaskar, M. Aykol and S. B. Cronin, *Nano Lett.*, 2011, **11**, 1111–1116.
- 16 P. Spinelli, M. Hebbink, R. de Waele, L. Black, F. Lenzmann and A. Polman, *Nano Lett.*, 2011, **11**, 1760–1765.
- 17 H. Harada, A. Onoda, T. Uematsu, S. Kuwabata and T. Hayashi, *Langmuir*, 2016, **32**, 6459–6467.
- 18 J. Hardy and D. J. Selkoe, *Science*, 2002, **297**, 353–356.
- 19 K. Yokoyama and D. R. Welchons, *Nanotechnology*, 2007, **18**, 105101.
- 20 A. Majzik, L. Fülöp, E. Csapó, F. Bogár, T. Martinek, B. Penke, G. Bíró and I. Dékány, *Colloids Surf., B*, 2010, **81**, 235–241.

