Direct reuse of two deep-dewatered sludge cakes without a solidifying agent as landfill cover: geotechnical properties and heavy metal leaching characteristics†

Jiakuan Yang,*ab Shinan Zhang, a Yafei Shi,*ac Chao Li, a Wenbo Yu, a Ruonan Guan, a Jun Xiao, a Sha Liang, a Jingping Hu, a Huijie Hou a and Jiukun Hud

Two types of deep-dewatered sewage sludge cakes were produced from pilot-scale experiments by using two composite conditioners: FeCl3 + quick lime (Fe−Lime) and Fenton’s reagent + red mud (Fenton−RM). The feasibility of direct reuse without any solidifying agents of these two deep-dewatered cakes, with water content of about 60 wt%, as landfill cover materials was investigated. Geotechnical properties of these two sludge cakes were found appropriate for reuse as landfill covers. Their plasticity index values increased significantly from 11.6 (raw sludge) to 23.8 (Fe−Lime) and 35.4 (Fenton−RM). The unconfined compressive strength and direct shear strength of the two deep-dewatered sludge cakes could meet or exceed the requirement of landfill cover materials after a certain curing time. Microstructural analyses of scanning electron microscopy (SEM) and X-ray diffraction (XRD) showed that their microstructures were more porous than that of raw sludge since the skeleton builders played a role in building the rigorous framework. There was negligible leaching of Cu, Zn, Pb, Cd and Cr from the deep-dewatered sludge cake from the toxicity characteristic leaching procedure and column leaching test. Both deep-dewatered sludge cakes could be reused as effective landfill cover materials with a suitable curing time.

1. Introduction

A tremendous amount of sewage sludge is being produced worldwide from wastewater treatment plants (WWTPs). In China alone, the annual sludge production has reached up to 24 million tons of dry sludge.1 Its treatment and disposal is costly and difficult. At present, sanitary landfilling is still the major disposal method for sewage sludge in China; while incineration and land application have a few applications.2

Increasing sludge production creates great needs for cost-effective alternatives, in which the most favorable one is beneficial reuse. Reuse of sewage sludge as daily or final cover materials for landfill has attracted substantial attention, and it has been recommended by U.S. EPA.3,4 Commonly, the water content of mechanically dewatered sewage sludge is typically about 75–85 wt% with chemical conditioning with polymers (e.g., polyacrylamide (PAM)). To reduce the high water content of dewatered sludge so that it can be beneficially reused as a landfill cover material, three approaches are often taken. The first approach is sludge drying,5 which is extremely expensive and energy-consuming with a risk of emissions of unpleasant odors. The second approach is blending the dewatered sludge cake with soil. However, the proper type of soil is often not readily available. The third approach is to add solidifying agents into the dewatered sludge cake. With addition of solidifying agents such as Portland cement,6 fluidized bed combustion ash,7 converter slag,8 fly ash and lime,9 calcined aluminum salts,10 or magnesium oxychloride cement,11 the solidified sewage sludge specimens were found effective as landfill cover materials. Unfortunately, the mass ratio of the soil or solidifying agents to the dry solid of the dewatered sludge often needs to be 100–200%. Consequently, the increased volume would take away lots of valuable landfill capacity. Direct reuse of dewatered sewage sludge as landfill cover materials without any soil or solidifying agents is seldom reported in literature.11

For landfill disposal, water content of sludge should be low; and the maximum water content limit is normally 60 wt% in China.12 Many attempts have been made to improve sludge dewatering by employing pretreatment. The pretreatment methods include physical and thermal processes,13,14 chemical
processes, and biological processes. Deep dewatering of sewage sludge using new types of chemical conditioners instead of the traditional polymer conditioners, have been extensively investigated and considered as a cost-effective alternative. In deep dewatering, water content of the dewatered sludge cake could be less than 60 wt%, which makes it more suitable for subsequent reuses or disposal. FeCl₃ combined with quick lime could be less than 60 wt%, which makes it more suitable for most of the heavy metals in sewage sludge were retained in the content of 52.8 and 47.7 wt%, respectively. It was found that deep-dewatered sewage sludge cakes were obtained with water content of the treated stabilization stage for the dewatered sewage sludge. By using either of these two systems, the water content of the treated sludge can be reduced to less than 60 wt%.

In the previous paper, pilot-scale sewage sludge dewatering experiments were conducted using both composite conditioners: Fe–Lime and Fenton–RM. Consequently, two types of deep-dewatered sewage sludge cakes were obtained with water content of 52.8 and 47.7 wt%, respectively. It was found that most of the heavy metals in sewage sludge were retained in the dewatered sludge cake. However, feasibility of direct reuse of those two deep-dewatered sewage sludge cakes as landfill cover materials has not been fully investigated.

The objective of this study was to evaluate geotechnical properties and heavy metal leaching characteristics of these two deep-dewatered sludge cakes. Two issues would be further investigated in this study: (1) how the skeleton builders would behave to improve the geotechnical properties of sludge cakes; (2) what are the leaching behaviors of heavy metals retained in the sludge cakes. Finally, feasibility of direct reuse of them as landfill cover materials would also be evaluated. The schematic of this study is depicted in Fig. 1.

2. Materials and methods

2.1 Materials

The raw sludge (RS) used in this study was a mixture of primary and secondary sludge after a gravity-thickening process from Tangxunhu WWTP in Wuhan, China. Table 1 summarizes some main characteristics of the RS sample. Two batches of RS were investigated in this study. RS-I was sampled in September of 2013; RS-II was sampled in April of 2014.

As presented in Table 1, the volatile suspended solid/total suspended solid (VSS/TSS) of RS-II is higher than that of RS-I since different influent load of sewage mixing with rainwater in various seasons. In the following experiments, RS-I was used in geotechnical tests, and both RS-I and RS-II were used in the leaching properties tests. Both RS samples had a pH close to neutral, and the concentrations of heavy metals (Cu, Zn, Pb, Cd, and Cr) are in the order of Zn > Cu > Cr > Pb > Cd.

FeCl₃, FeSO₄, H₂O₂ and H₂O₂ (27.5% v/v) of industrial grade were obtained from a commercial company in China. H₂SO₄ was used to adjust the initial pH of the RS samples to the optimal value of 5.0 before the addition of Fenton’s reagent. All the acids (including H₂SO₄, HNO₃, HF and HClO₄) used for sludge sample digestion were of analytical grade.

Quick lime and RM, which were used as skeleton builders, were dried, milled and sieved to less than 1 mm in particle size before use. The quick lime was obtained from a local factory. The RM was supplied by an alumina plant employing the Bayer process in Zhengzhou, China. Their chemical compositions are presented in Table S1 as ESL†.

2.2 Dewatering process and dewatered sludge cake

Two composite conditioning systems (Fe–Lime and Fenton–RM) were used in this study. The sludge conditioning and dewatering process is presented in Fig. S1,† studied in our previous study. The scheme of sludge conditioning has been optimized in our previous work and is presented in Table S2. After conditioning, the sludge was dewatered by a pilot-scale diaphragm filter press in the Tangxunhu WWTP (Fig. 2). For each batch, 500–800 kg of RS was pumped into a conditioning tank. A more detailed description of the dewatering process was depicted elsewhere.

Main characteristics of two types of deep-dewatered sludge cakes are presented in Table 2. After the deep-dewatering process, the water contents of the dewatered sludge cakes
Table 1 Some basic characteristics and the concentrations of main heavy metals of the raw sludge samples

<table>
<thead>
<tr>
<th>Batch</th>
<th>pH</th>
<th>Water content (wt%)</th>
<th>VSS/TSS (%)</th>
<th>Cu (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Zn (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Pb (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Cd (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Cr (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
</tr>
</thead>
<tbody>
<tr>
<td>RS-I</td>
<td>7.2</td>
<td>96.0 ± 0.1</td>
<td>41.3 ± 0.3</td>
<td>132.1 ± 2.9</td>
<td>279.6 ± 12.9</td>
<td>39.8 ± 0.5</td>
<td>4.5 ± 0.2</td>
<td>73.3 ± 0.9</td>
</tr>
<tr>
<td>RS-II</td>
<td>6.9</td>
<td>96.9 ± 0.1</td>
<td>59.6 ± 0.2</td>
<td>192.0 ± 2.0</td>
<td>1139.3 ± 10.3</td>
<td>23.3 ± 0.3</td>
<td>2.9 ± 0.1</td>
<td>66.7 ± 0.6</td>
</tr>
</tbody>
</table>

* On the dry solid basis. * Milligrams per kilogram of dry solid (mg kg<sup>-1</sup>).

Table 2 Main characteristics of the dewatered sludge cakes

<table>
<thead>
<tr>
<th>Conditioning system</th>
<th>Water content (wt%)</th>
<th>VSS/TSS (wt%)</th>
<th>Cu (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Zn (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Pb (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Cd (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
<th>Cr (mg kg&lt;sup&gt;-1&lt;/sup&gt;)</th>
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</thead>
<tbody>
<tr>
<td>Fe–Lime-I</td>
<td>52.8</td>
<td>35.2</td>
<td>91.7 ± 0.9</td>
<td>251.3 ± 7.0</td>
<td>30.3 ± 1.3</td>
<td>2.4 ± 0.4</td>
<td>59.9 ± 3.4</td>
</tr>
<tr>
<td>Fenton–RM-I</td>
<td>47.7</td>
<td>37.8</td>
<td>101.6 ± 1.6</td>
<td>253.6 ± 5.6</td>
<td>36.0 ± 0.0</td>
<td>4.8 ± 0.1</td>
<td>117.4 ± 0.9</td>
</tr>
<tr>
<td>Fenton–RM-II</td>
<td>53.7</td>
<td>45.7</td>
<td>93.7 ± 0.5</td>
<td>797.2 ± 4.3</td>
<td>30.5 ± 0.5</td>
<td>3.0 ± 0.1</td>
<td>77.7 ± 0.3</td>
</tr>
</tbody>
</table>

* On the dry solid basis. * Milligrams per kilogram of dry solid (mg kg<sup>-1</sup>).
According to TCLP, 50 g air-dried sludge specimens (Fe–Lime-I and RS-I) was grinded and sieved to achieve particle sizes smaller than 9.5 mm and then placed in a polyethylene bottle, followed by an addition of one liter acetic acid solution (5.7 mL acetic acid in 1 L of distilled water, pH = 2.88). The bottle was placed in a rotary extractor and rotated for 18 h at 30 ± 2 rpm. After rotation, the sample was filtered through a 0.45 μm membrane filter, and then the filtrate was collected and analyzed for metal concentrations by using an atomic absorption spectrometer (Analytik Jena AG NovAA 400, Germany).

The schematic of the CLT is presented in Fig. S2†. As a dynamic test, CLT reportedly has a better leaching effect. Eighty grams of air-dried sludge specimens (Fenton–RM-II and RS-II) were grinded and sieved to achieve particle sizes smaller than 4 mm, and placed in a glass column (5 cm inside diameter and 35 cm height) and a perforated plate on both ends. The dewatered sludge was leached from the bottom up by a continuous flow of the acetic acid solution (pH = 2.88) using a peristaltic pump. A flowing rate of 5 mL h⁻¹, commonly used in literature, was adopted here; and a rate of 10 mL h⁻¹ was also applied for a comparison.

The leachate was sampled at different liquid-to-solid ratios (L/S, mL g⁻¹). L represents the volume of the acetic acid solution passed through the column, in mL; and S represents the mass of the sludge sample in the column, in g. Heavy metals in the leachate were analyzed. The extraction percentage of each heavy metal is the mass ratio of heavy metal leached out to the total mass of heavy metal in the original sludge sample.

2.5 Microstructural analysis

Microstructural characteristics of the dewatered sludge samples were investigated by X-ray diffraction (XRD) and scanning electron microscopy (SEM). The specimens were immersed in ethanol to end the hydration reaction and then dried in an oven at 40 °C for 24 h. XRD analyses were carried out using Cu Kα radiation, operated at 40 kV, with a scan speed of 0.2785° s⁻¹ and in the 2θ range from 5° to 75°. Morphology studies were carried out using a Sirion 200 scanning microscope after the samples were coated with gold.

3. Results and discussion

3.1 Geotechnical properties of solidified sludge

The geotechnical properties of the clay, RS-I and the dewatered sludge cakes are presented in Table 3. As shown, when compared to RS-I, the PL of the Fe–Lime-I decreased significantly from 58.1 wt% to 40.9 wt%, while the PI increased from 11.6 wt% to 23.8 wt%. As for the Fenton–RM-I, its PL and PI were 33.1 wt% and 35.4 wt%, respectively. Moreover, the PL and PI of RS-PAM were close to those of RS. The changes can be seen more readily in Fig. 3. The LL values of the dewatered sludge cakes were nearly the same as that of the RS, but PL values of the dewatered sludge cakes were significantly less than that of the RS. Thus the PL values increased obviously from RS to the sludge cakes since PL equals to LL minus PL, as presented in Table 3. PI value was in the order of RS < Fe–Lime < Fenton–RM. According to the classification of ASTM, the RS belongs to high-plasticity silt while the sludge cake of Fenton–RM is on the boundary of silt and clay. When compared to deep-dewatered sludge cakes (Fe–Lime and Fenton–RM), the standard clay has a smaller plasticity index values.

In addition, RS-I showed a high OMC of 37.5 wt% and an MDD of 1.581 g cm⁻³. While for both deep-dewatered sludge cakes, the values of OMC and MDD decreased considerably. It implied that deep-dewatered sludge cakes have lower water-absorption capacity, when compared with RS and RS-PAM.

It can also be seen from Table 3 that the deep-dewatered sludge cakes had higher permeability coefficient values (>10⁻² cm s⁻¹) than that of RS (6.45 × 10⁻⁸ cm s⁻¹). These results validate the roles of skeleton builders in the conditioning stage that increased the porosity of dewatered sludge.

Table 3 Geotechnical properties of the sludge and other specimens

<table>
<thead>
<tr>
<th>Specimens</th>
<th>LL (wt%)</th>
<th>PL (wt%)</th>
<th>PI (wt%)</th>
<th>OMC (wt%)</th>
<th>MDD (g cm⁻³)</th>
<th>Coefficient of permeability (cm s⁻¹)</th>
</tr>
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<tbody>
<tr>
<td>Clay</td>
<td>40.6</td>
<td>20.0</td>
<td>20.6</td>
<td>12.4</td>
<td>1.890</td>
<td>3.31 × 10⁻⁸</td>
</tr>
<tr>
<td>RS-I</td>
<td>69.7</td>
<td>58.1</td>
<td>11.6</td>
<td>37.5</td>
<td>1.581</td>
<td>6.45 × 10⁻⁸</td>
</tr>
<tr>
<td>RS-PAM</td>
<td>69.3</td>
<td>59.8</td>
<td>9.5</td>
<td>36.9</td>
<td>1.561</td>
<td>6.71 × 10⁻⁸</td>
</tr>
<tr>
<td>Fe–Lime-I</td>
<td>64.7</td>
<td>40.9</td>
<td>23.8</td>
<td>25.5</td>
<td>1.421</td>
<td>5.90 × 10⁻⁷</td>
</tr>
<tr>
<td>Fenton–RM-I</td>
<td>68.5</td>
<td>33.1</td>
<td>35.4</td>
<td>23.0</td>
<td>1.435</td>
<td>7.20 × 10⁻⁸</td>
</tr>
</tbody>
</table>

Fig. 3 Casagrande plasticity chart for clay. RS and the dewatered sludge cake (CL: low-plasticity clay; CH: high-plasticity clay; ML: low-plasticity silt; OL: low-plasticity organic; MH: high-plasticity silt; OH: high-plasticity organic. According to ASTM D2487-93, 1997).
No existing Chinese standards define acceptable parameters with regards to the permeability coefficient of landfill cover. In USA, the permeability coefficients of lime-stabilized sludge as landfill cover range from $1 \times 10^{-3}$ to $10^{-6}$ cm s$^{-1}$. In general, the permeability coefficient values of the deep-dewatered sludge and the RS are less than the reported permeability coefficient values of lime-stabilized sludge. Both deep-dewatered sludge cakes had a permeability coefficient value (between $10^{-6}$ and $10^{-8}$ cm s$^{-1}$), which is smaller to the reported permeability coefficient of the lime-stabilized sludge.

### 3.2 Mechanical strength properties

The relationship between the UCS and the curing age is shown in Fig. 4(a). As shown, the UCS values of the dewatered sludge cakes increase with the curing age. In general, a minimum UCS of 50 kPa is the permissible limit in Chinese standard for an effective landfill cover material. It took 14 days and 28 days for the dewatered sludge of Fe-Lime and the Fenton–RM systems to go above this minimum limit, respectively. As shown in Fig. 4(b), the DSS values of the dewatered sludge cakes increase.

![Fig. 4](image)

Fig. 4 (a) Unconfined compressive strength and (b) direct shear strength of the dewatered sludge cakes versus curing time. Dotted lines denote the minimum limits of Chinese standards for an effective landfill cover.

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![Fig. 5](image)

Fig. 5 SEM images of (a) RS-I, and the (b) dewatered sludge cakes of Fe–Lime-I, (c) Fenton–RM-I; and (d) the XRD patterns of these samples.

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![Fig. 6](image)

Fig. 6 Concentrations of heavy metal in the leachates from the TCLP tests: (a) RS-I, and Fe–Lime-I; (b) RS-II, and Fenton–RM-II.
Fig. 7  Leached concentrations and extraction percentages of heavy metals from the column leaching tests: (a) Cu, at a leaching rate of 5 mL h\(^{-1}\) and (b) 10 mL h\(^{-1}\); (c) Zn, at a leaching rate of 5 mL h\(^{-1}\) and (d) 10 mL h\(^{-1}\); (e) Pb, at a leaching rate of 5 mL h\(^{-1}\) and (f) 10 mL h\(^{-1}\); (g) Cd, at a leaching rate of 5 mL h\(^{-1}\) and (h) 10 mL h\(^{-1}\); (i) Cr, at a leaching rate of 5 mL h\(^{-1}\) and (j) 10 mL h\(^{-1}\).
with the curing age, which is consistent with the case for the UCS. In general, the required minimum limit of DSS for an effective landfill cover material is 25 kPa. It took 28 and 90 days for the dewatered sludge of Fe–Lime and Fenton–RM systems to go above this minimum limit, respectively. The results imply that lime could enhance the mechanical strengths of the dewatered sludge in Fe–Lime system more effectively than red mud (a skeleton builder in Fenton–RM system), and it took a shorter curing time for Fe–Lime system to attain the required mechanical strength.

3.3 Microstructure characteristics

Fig. 5 shows the SEM images and XRD patterns of the RS-I and the dewatered sludge cakes (Fe–Lime-I, and Fenton–RM-I). As shown in Fig. 5(a), the morphological structure of the RS is flake-like, dense and compact, since more organic matters in the RS filled most of voids in the dried RS specimen. The XRD pattern indicates that the RS consists mainly of quartz and mica. Microstructure characteristics of the RS are consistent with those found in the previous studies.20,21

The dewatered sludge cake of Fe–Lime-I consists of smaller particulates and it is more porous with an irregular shape (Fig. 5(b)). It indicates that the raw sludge with a high organic content and poor dewaterability can be improved with skeleton builders to generate a porous, but relatively incompressible structure. The XRD pattern indicates the Fe–Lime-I consists mainly of quartz, calcium hydroxide and calcite.

Some irregularly shaped crystals embedded among the sludge particles were found in Fenton–RM-I (Fig. 5(c)). This dewatered sludge cake appears to be more porous. The XRD patterns indicate that it consists mainly of quartz, mica, gypsum, anatase, and gibbsite.

3.4 Leaching tests for heavy metals

3.4.1 TCLP tests. Fig. 6 shows the heavy metals concentrations in the leachates of the TCLP tests for the RS and the deep-dewatered sludge cakes.

As shown in Fig. 6(a) and (b), leached concentrations of Cu, Zn and Pb from the two dewatered sludge cakes of Fe–Lime-I are significantly lower than those of the RS-I; while Cr concentration in the dewatered sludge of Fenton–RM-II are the highest among all of RS and dewatered sludge specimens (Fig. 6(b)). The higher concentrations of Cr might come from the red mud which was added as the conditioner. In general, the sludge cake from the Fe–Lime system has a good retention for heavy metals.

Typical acceptable TCLP concentrations of heavy metals effluent discharge standards of municipal solid waste landfills are 40, 100, 0.25, 0.15 and 4.5 mg L\(^{-1}\) of Cu, Zn, Pb, Cd, and Cr (GB16889-2008), respectively.20 The results show that Pb in Fe–Lime-I exceeded the acceptable limit, while the retention capability of Fenton–RM-II on Pb and Cd needs to be improved. In order to facilitate technology adaptability, two improvements could be used to promote the solidification effects of the dewatered sludge cakes. Firstly, during the conditioning and dewatering process, a little more dosage of conditioner (lime or cement) could be added to further improve the dewatering effects and the heavy metal solidification behavior. Besides, to avoid the risk of heavy metal release in the landfill process, small amount of cement could be mixed with the dewatered sludge cakes to strengthen the heavy metal solidification effects, given the good cooperation of alkali cement and red mud.

3.4.2 Column leaching test (CLT). Plots of leached concentrations and extraction percentages of various heavy metals versus the L/S ratios from the CLTs are shown in Fig. 7. In general, the leached concentrations of all heavy metals declined and the extraction percentages increased with an increase in the L/S ratio (i.e., age of landfill cover). In addition, at the same L/S ratio, more heavy metals were expectedly leached out at a low leaching rate of 5 mL h\(^{-1}\) than at 10 mL h\(^{-1}\), because of the longer contact time. The extraction percentage of Cu from the Fe–Lime-I was lower than that of the RS-I; while the extraction percentage of Cu from Fenton–RM-II is the lowest (<20 wt%). The results of CLT are consistent with those of the TCLP experiments. The leaching characteristics of Zn are essentially the same as those of Cu in all four samples. However, the leached concentration of Zn from Fenton–RM-II increased significantly with time. The reason might be that the concentration of Zn in RS-II was much higher than that of RS-I. The leached concentrations of Pb and Cd from Fe–Lime-I are essentially the same as those of the RS-I; while addition of RM appears to have a better retention of Pb and Cd as their extraction percentages from Fenton–RM-II were less than 30 wt%. For the case of Cr, nearly 30 wt% of Cr was leached out from the RM, the extraction percentages of Cr in all the dewatered sludge samples were at a low level (<2 wt%).

The results of this CLT study indicate that the deep-dewatered sludge cakes conditioned with skeleton builders are relative stable for a long period of time, with regards to heavy metal leaching, when they are used as daily or final landfill cover.

4. Conclusion

In this study, feasibility of reusing deep-dewatered sludge cakes without addition of solidifying agents as landfill cover materials was evaluated. Results of geotechnical tests indicate that deep-dewatered sludge cakes, conditioned with skeleton builders, improved the geotechnical properties of the dewatered sludge, when compared with RS and RS-PAM. The deep-dewatered sludge cakes can attain the required USC and DSS needed for landfill covers within a short curing time. The XRD patterns and SEM images validated the framework effect of the skeleton builders and showed that these cakes are less compressible. Results of leaching tests using both TCLP and CLT revealed that both deep-dewatered sludge cakes could retain heavy metals for a long period of time.

In summary, the skeleton builders introduced in the conditioning stage during deep-dewatering process improve the geotechnical properties and heavy metal leaching characteristics of the dewatered sludge cakes which make them appropriate for re-use as landfill cover materials.
## Abbreviation

<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>Description</th>
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<tbody>
<tr>
<td>CLT</td>
<td>Column leaching test</td>
</tr>
<tr>
<td>DS</td>
<td>Dry solid</td>
</tr>
<tr>
<td>DSS</td>
<td>Direct shear strength</td>
</tr>
<tr>
<td>LL</td>
<td>Liquid limit</td>
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<tr>
<td>L/S</td>
<td>Liquid-to-solid ratio</td>
</tr>
<tr>
<td>MDD</td>
<td>Maximum dry density</td>
</tr>
<tr>
<td>OMC</td>
<td>Optimum moisture content</td>
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<td>PAM</td>
<td>Polyacrylamide</td>
</tr>
<tr>
<td>PI</td>
<td>Plasticity index</td>
</tr>
<tr>
<td>PL</td>
<td>Plastic limit</td>
</tr>
<tr>
<td>RM</td>
<td>Red mud</td>
</tr>
<tr>
<td>RS</td>
<td>Raw sludge</td>
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<td>SEM</td>
<td>Scanning electron microscopy</td>
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<td>TCLP</td>
<td>Toxicity characteristic leaching procedure</td>
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<tr>
<td>TSS</td>
<td>Total suspended solids</td>
</tr>
<tr>
<td>UCS</td>
<td>Unconfined compressive strength</td>
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<td>Volatilized suspended solids</td>
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<td>Wastewater treatment plant</td>
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<td>XRD</td>
<td>X-ray diffraction</td>
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## References