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# Replica exchange molecular dynamics study of the amyloid beta (11–40) trimer penetrating a membrane†

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Alzheimer's disease is characterized by the interaction of neurotoxic A $\beta$  oligomers with cellular membranes, which disturbs ion homeostasis. To determine the putative structures of the transmembrane 3A $\beta_{11-40}$  oligomer, temperature replica exchange molecular dynamics (REMD) simulations with an explicit solvent have been employed to monitor the structural changes when interaction of the oligomer with the membrane DPPC lipid bilayer is induced. Although the initial conformation of the 3A $\beta_{11-40}$  transmembrane was fibril-like, the obtained results are in good agreement with previous experiments, in which the  $\beta$ -structure of the A $\beta$  oligomer represents  $\sim 40\%$  of the structure in the average of all considered snapshots. The statistical coil structure, which is located near and interacts with the membrane headgroups, amounts to almost 60% of the structure. The transmembrane A $\beta$  oligomer helix structure basically disappears during the REMD simulations. Instead of the Asp23–Lys28 salt bridge, the polar contact between Asp23 and Asn27 has been found to be a factor stabilizing the structure of the A $\beta$  oligomer. Although numerous polar contacts between lipid headgroups and the peptide have been found, free energy perturbation calculations indicated that van der Waals interactions are the key factor determining the binding between the A $\beta$  trimer and the membrane. It may be argued that the A $\beta_{11-40}$  trimer can be easily inserted into the membrane because the binding free energy between the trimer and the membrane reaches  $-70$  kcal mol $^{-1}$ . The collision cross section of the optimized structures of  $1341 \pm 23$  Å $^2$  agrees well with the experimental values for the solvated A $\beta$  trimer.

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## Introduction

Alzheimer's disease (AD) is known to be one of the most common neurodegenerative disorders<sup>1–5</sup> and is frequently observed among elderly people. In fact, about a third of seniors are affected by AD or other dementias; as yet, there is no effective treatment for AD.<sup>1–6</sup> Different mechanisms of AD have been proposed, including the cholinergic, tau and amyloid hypotheses.<sup>7–9</sup>

Numerous previous studies have indicated that aggregation of amyloid beta (A $\beta$ ) peptides in the extracellular region of brain

tissue is the main cause of AD.<sup>4,10–12</sup> Furthermore, A $\beta$  oligomers have recently been found to be more neurotoxic than the fibril forms.<sup>4,13,14</sup> However, the mechanism by which A $\beta$  oligomers damage neurons remains uncertain. Scientists currently suggest that these oligomers insert into lipid bilayers and establish ion channel-like structures.<sup>15</sup> Consequently, Ca<sup>2+</sup> dication homeostasis is perturbed, ultimately leading to cytotoxicity.<sup>16–18</sup>

In general, although A $\beta$  peptides tend to favor interactions with the phosphate headgroups of zwitterionic lipids through electrostatic interactions,<sup>19–21</sup> the interaction of A $\beta$  with positively charged lipids is equivalent to that with negatively charged lipids.<sup>21,22</sup> In addition, the oligomers appear to insert into the membrane more easily than their corresponding monomers.<sup>19</sup> Several experiments investigating the effects of a membrane on A $\beta$  peptides were reported under various conditions. The structure of an A $\beta$  peptide is transformed into a helix–kink–helix structure when its C-terminus incompletely penetrates the membrane.<sup>23–25</sup> It has been established that acidic phospholipids motivate A $\beta$  peptides to adopt  $\beta$ -form coil structures.<sup>24,26</sup> The kinetics and thermodynamics of the transformation of the random coil structure into the  $\beta$ -structure on the surface of the anionic membrane were studied.<sup>27</sup> These

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† Electronic supplementary information (ESI) available: Include the list of additional figures consist of the mean exchange rates between neighbouring replicas, the distribution of potential energy of all replicas, the diffusion of temperature space of 1<sup>st</sup> and 48<sup>th</sup> replicas, the RMSF of the trimer, and the RMSD of the solvate A $\beta_{11-40}$  trimer. See DOI: 10.1039/c6ra26461a

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experimental results were analyzed and supported by numerous subsequent theoretical studies.<sup>28–33</sup>

Due to the fact that the A $\beta$  trimer is one of the most toxic forms of low weight oligomers,<sup>34</sup> current studies are focused on screening potential inhibitors to prevent the trimer from forming.<sup>35</sup> However, the equilibrated A $\beta$  trimers exist in mixed environments consisting of monomers, dimers, higher order oligomers and mature fibrils; thus, experimental studies about them are few.<sup>36,37</sup> Thus, it is difficult to design effective A $\beta$  trimer inhibitors to treat AD. Moreover, the neurotoxicity and aggregate conformations of A $\beta$  peptides rely upon the sequences and lengths of the peptides.<sup>38</sup> Although several previous investigations indicated that the hydrophilic region of the A $\beta$  peptide N-terminal alters peptide deposition,<sup>39–41</sup> the effects of the hydrophobic core on the deposition of A $\beta$  peptides are greater than the effects of the hydrophilic core. Consequently, previous studies have mainly focused on evaluating the structures of truncated A $\beta$  peptides.<sup>42–44</sup>

In this context, a theoretical study of the transmembrane A $\beta$  trimer and its interactions is thus of great interest. We thus determined to investigate the structures of the A $\beta_{40}$  oligomers when they fully penetrate lipid bilayers. For this purpose, we considered the dipalmitoyl phosphatidylcholine (DPPC) lipid bilayer and inserted the 3A $\beta_{11–40}$  oligomer into it. Moreover, it is likely impossible to study the A $\beta$  trimer starting from the random coil form due to the high CPU time demand,<sup>45</sup> which is in any case beyond our actual computational resources. Therefore, we used a fibril-like structure as the initial conformation of the 3A $\beta_{11–40}$  transmembrane. To study the resulting solvated transmembrane A $\beta$  peptide system, we used replica exchange molecular dynamics (REMD) simulations with an explicit solvent. This method allows the structural changes in the transmembrane 3A $\beta_{11–40}$  peptide to be recorded during the simulation time and allows the mutual interactions between the membrane and the peptide to be probed.

Thus, the most stable structure of the transmembrane 3A $\beta_{11–40}$  peptide corresponds to the energy global minimum of this peptide; this is explored through free energy landscape analyses. In accord with previous experiments and computations, the secondary structure of the oligomer has been predicted using the Define Secondary Structure of Proteins (DSSP) tool;  $\alpha$ -content was lacking during our simulations, while the  $\beta$ -content and statistical coil structures comprised about 40% and 60% of the structure, respectively.<sup>24,26–33,46</sup> The coil domains are located within the membrane surface and strongly interact with the phosphorus atoms of the lipid headgroups.<sup>24,26</sup> The Asp23–Asn27 salt bridge was found to play an important role in stabilizing the structure of the A $\beta$  peptide, instead of the Asp23–Lys28 salt bridge, as previously reported.<sup>30</sup> Moreover, the binding free energy of the trimer to the membrane was determined using free energy perturbation calculations. The size of the trimer was determined through collision cross section prediction. The obtained results may enhance the search for an AD therapeutic agent and establish references for transmembrane mutant A $\beta$  trimer studies.

## Materials and methods

### Replica exchange molecular dynamics (REMD) simulations

The 3A $\beta_{11–40}$  oligomer<sup>44</sup> was presented through the united atom GROMOS 53a6 force field.<sup>47</sup> In this work, this A $\beta$  peptide fully penetrated the membrane DPPC lipid bilayer.<sup>48</sup> The transmembrane system was placed in a periodic boundary conditions box and was then solvated using the simple point charge water model.<sup>49</sup> Three Na<sup>+</sup> ions were added to maintain neutral conditions in the system. The crystal structure of the initial conformation employed is shown in Fig. 1.

The solvated transmembrane oligomer system was used as the initial conformation of the computations using GROMACS version 5.0.7.<sup>50</sup> The starting structure included the A $\beta$  oligomer, 125 DPPC molecules, 3923 water molecules, and 3 Na<sup>+</sup> ions. The steepest descent, conjugate gradient, and low-memory Broyden–Fletcher–Goldfarb–Shanno (L-BFGS) methods<sup>51</sup> were employed to minimize the solvated transmembrane oligomer. The system reached an energy minimum when the maximum force recorded was smaller than 10<sup>−6</sup> kJ (mol<sup>−1</sup> nm). Consequently, the system was simulated over a time of 500 ps in the canonical (NVT) ensemble at 324 K with all atoms restrained using a weak harmonic force. The last snapshot of the NVT simulation was then used as the initial structure of the 500 ps isothermal–isobaric (NPT) simulation at 324 K.

Subsequently, temperature REMD simulations were performed in which the input conformation was the last snapshot of the NPT simulations. The number of replicas was 48, ranging from 290 to 417 K (details are shown in the (ESI†) file). The temperatures of these replicas were determined using the temperature generator for the REMD simulations webserver.<sup>52</sup> The acceptance ratio was chosen to be sufficiently larger than 20%. The exchanges between neighboring replicas were checked every 1 ps, which is sufficient to compare the coupling times of the heat bath. There were 350 000 replica exchange cycles during the computations. The data were collected every 10 ps.

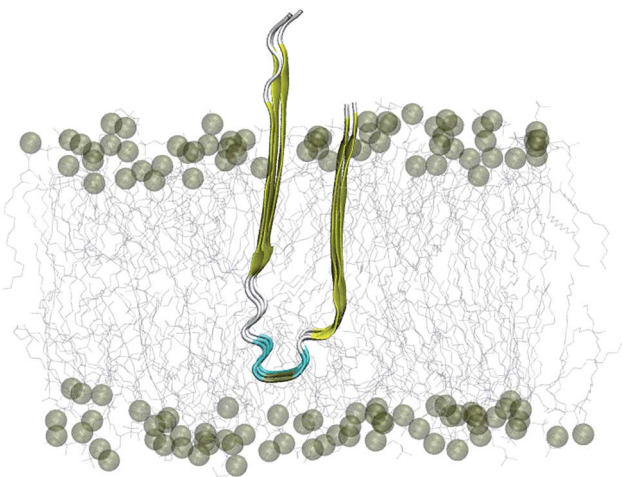


Fig. 1 Starting conformation of the truncated transmembrane 3A $\beta_{11–40}$  peptide inserted into the membrane DPPC lipid bilayer. Water and ion molecules are hidden in this figure.



The MD simulations were integrated using the accurate leap-frog stochastic dynamics integrator.<sup>53</sup> A relaxation time of 0.1 ps was chosen; the system pressure was 1.0 atm and was controlled by the Parrinello–Rahman method.<sup>54</sup> All bonds were constrained through the LINCS<sup>55</sup> with an order of 4. In accord with previous computational studies on the folding/misfolding of A $\beta$  peptides,<sup>39,56</sup> the time step was selected as 2 fs. The non-bonded interaction pair list was renewed every 10 fs, with a cutoff of 1.0 nm. Electrostatic interactions were determined utilizing the fast smooth Particle-Mesh Ewald electrostatics method, with a cutoff of 1.0 nm.<sup>57</sup> The Lennard–Jones interactions were computed with a cutoff equal to the cutoff of non-bonded interactions.

### Free energy perturbation (FEP) method

The interactions between the protein and the membrane were determined through a free energy perturbation method.<sup>58</sup> In this method, the obtained difference of free energy between two bound and unbound states is calculated through MD simulations; during these computations, the system changes from the bound Hamiltonian to the unbound Hamiltonian using  $\lambda$  intervals. The determination is performed when the system exists in the equilibrium state. The bound and unbound states correspond with the coupling parameters  $\lambda = 0$  and 1, respectively. The Bennet's acceptance ratio (BAR) method<sup>59</sup> was used to determine the change of free energy  $\Delta G_{\lambda_i \Rightarrow \lambda_{i+1}}$  between

neighbouring states,  $\lambda_i$  and  $\lambda_{i+1}$ . Consequently, the difference of free energy between the bound and unbound states is a summation over these values (eqn (1)):

$$\Delta G = \sum_{\lambda=0}^{\lambda=1} \Delta G_{\lambda_i \Rightarrow \lambda_{i+1}} \quad (1)$$

In the present work, we changed the coupling parameter  $\lambda$  from 0 to 1 to annihilate the trimer from the transmembrane and solvated systems by altering the non-bonded interactions, as shown in the thermodynamics diagram in Fig. 2. In particular, we used a total of 15 values of  $\lambda$  to reduce the non-bonded interactions from the full-interaction state to the non-interaction state over 15 independent MD simulations with lengths of 5 ns each. The independent MD simulations had the same initial crystal structures and starting velocities; however, they had different coupling parameters  $\lambda$ . In the latter, the Coulomb interaction was decreased through 6 values of the coupling parameter  $\lambda$ , including 0.00, 0.35, 0.55, 0.73, 0.88, and 1.00. Additionally, the van der Waals interactions were modified using 10 values of  $\lambda$ : 0.00, 0.10, 0.20, 0.25, 0.30, 0.40, 0.55, 0.70, 0.85, and 1.00. During these processes, the trimer was annihilated two times in different systems; thus, this method is called the double-annihilation binding free energy method (*cf.* Fig. 2).<sup>60–63</sup> Overall, the binding free energy of the trimer to the membrane lipid bilayers was estimated through expression (2):

$$\Delta G_{\text{bind}} = \Delta G_1 - \Delta G_2 \quad (2)$$

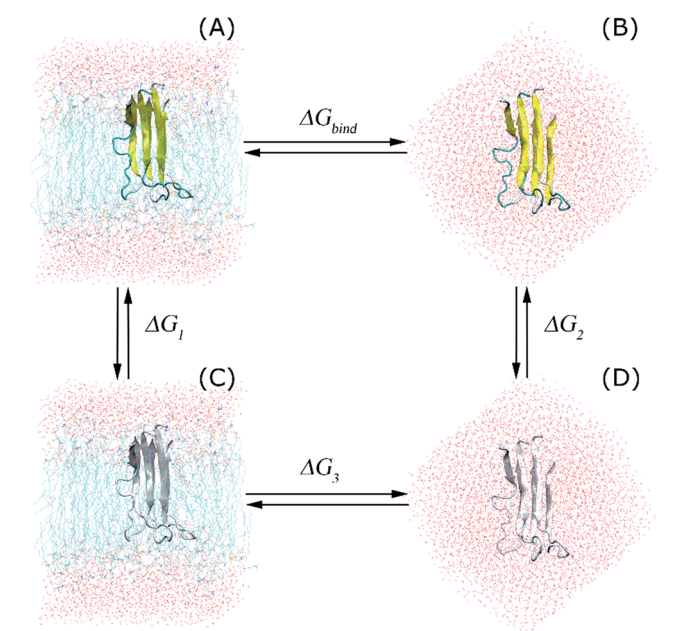


Fig. 2 Thermodynamics diagram of the double-annihilation binding free energy method that was applied to determine the binding free energy of the trimer to the membrane DPPC lipid bilayer. In this diagram, (A) represents the full-interaction state of the A $\beta$  peptide with the solvated membrane system. (B) presents the full-interaction state of the protein with the solution. (C) is a dummy protein penetrating the membrane, and (D) is the dummy protein in solution. The dummy protein represents the protein without any interaction with surrounding molecules.

### Secondary structures

The secondary structures of the transmembrane 3A $\beta_{11-40}$  peptides were estimated using DSSP.<sup>64,65</sup>

### Free energy landscape (FEL)

The “gmsham”<sup>66,67</sup> is a tool in GROMACS that can be employed to determine the FEL of the trimer with two reaction coordinates, including the radius of gyration ( $R_g$ ) and the root mean square deviation (RMSD). The clustering method<sup>68</sup> was employed to find putative conformations of a protein which remained at a global minimum with a tolerance of 3.5 nm  $C_\alpha$  RMSD.

### Collision cross section (CCS)

CCS is a high impact parameter that can be determined by the ion mobility projection approximation calculation tool (IMPACT).<sup>69</sup>

### Contact

The intermolecular contact between the heavy atoms of the residues and the phosphorus atoms of lipid headgroups was investigated through evaluation of the minimum distance of the corresponding atoms with a cutoff of 0.45 nm. Additionally, the distances between the sidechains of the neighbouring chains of the truncated peptide were considered. Thus, the sidechain





contacts were counted when this distance was smaller than 0.45 nm.

### The order of the lipid bilayers

The lipid order parameters measure the oriented mobility of the carbon (C) and deuterium (D) bonds through the parameter  $S_{CD} = \frac{1}{2} \overline{3 \cos^2 \theta - 1}$ , where  $\theta$  is the angle between the molecular axis given by the  $C_{i-1} - C_{i+1}$  vector and the bilayer normal, which is investigated over computational time. The results were averaged over the membrane during the simulation times.

## Results and discussion

### Temperature REMD simulation of the transmembrane 3A $\beta_{11-40}$ peptide

At the atomic computational scale, REMD simulation has become an essential computational method for treating biomolecular systems in general and A $\beta$  aggregation problems in particular; it has been proved to be one of the most powerful enhanced sampling methods.<sup>28,70–72</sup> This method has been validated for investigation of the structures of A $\beta$  peptides in many previous studies.<sup>45,73–76</sup> In contrast, this approach may become less productive compared to normal MD simulations if the maximum temperature chosen is too high,<sup>77</sup> especially if the membrane becomes unstable at high temperature.<sup>78</sup> This may lead to unstable A $\beta_{11-40}$  trimer structures during simulations; however, here, the maximum temperature chosen of 417 K is not too high. However, in this case, although the  $\beta$ -structure of the A $\beta$  trimer was never broken, it fluctuated within a large range, from ~15% to ~55% (Fig. 4C). This may be caused by the use of the GROMOS force field, which is known to favor formation of the  $\beta$ -structure.<sup>79</sup>

As the aggregation process of A $\beta$  peptides is very slow, requiring up to several days, it is very difficult to obtain native

structures of A $\beta$  oligomers from random initial structures through extensive simulations. For example, in previous studies, the computations were proved to reach equilibrium even though the computed  $\beta$ -content was found to be ~18% for the A $\beta_{40}$  dimer;<sup>41</sup> experiments indicated that the  $\beta$ -structure content of the dimer was ~39%.<sup>80</sup> These inconsistent data

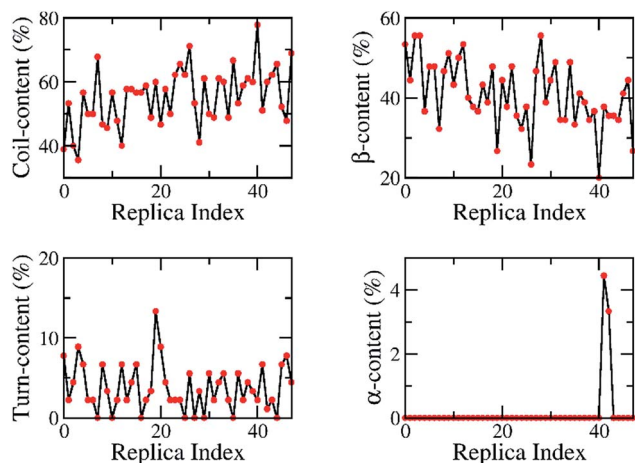


Fig. 3 The secondary structures of 48 replicas of the truncated transmembrane A $\beta_{11-40}$  trimer over 200 ns of REMD simulations, predicted using DSSP tools. The individual metrics are diffused over the entire wide range, suggesting that the computations did not focus on any special conformation.

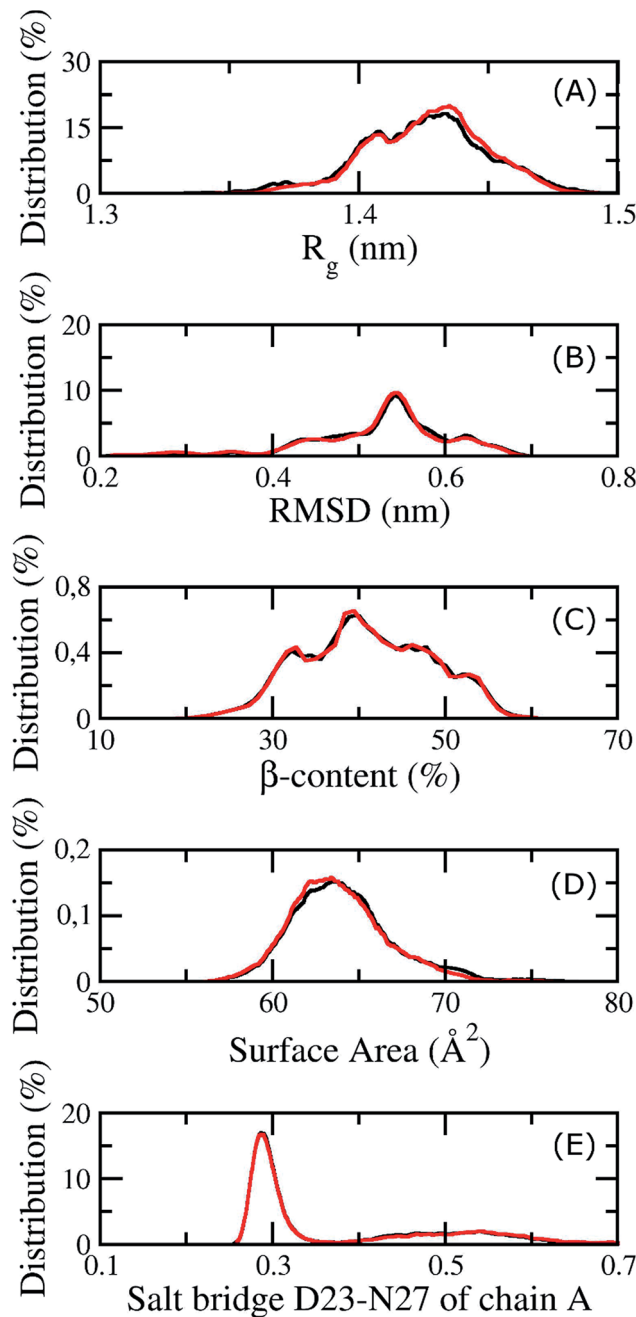


Fig. 4 The convergence of the REMD simulations at 324 K. The black and red lines correspond to the values at different simulation intervals, 200 to 270 ns and 290 to 350 ns. (A) is the distribution of the radius of gyration of the transmembrane A $\beta_{11-40}$  trimer, (B) is the population of the RMSD of the A $\beta$  peptide, (C) is the distribution of the  $\beta$ -content of the trimer, (D) is the population of the surface accessible area of the A $\beta$  peptide, and (E) is the distribution of the salt bridge D23–N27 of chain A.



suggest that the initial conformations of A $\beta$  peptides are very important and that there are several pathways and local minima in the A $\beta$  aggregation process. We assumed that the conformations of the trimer oligomer were close to the those of the mature fibrils;<sup>45</sup> thus, the initial conformation of the solvated 3A $\beta$ <sub>11–40</sub> peptide inserted into the membrane DPPC lipid bilayer was taken from the two-fold fibril form of the 12A $\beta$ <sub>11–40</sub> peptide,<sup>44</sup> as shown in Fig. 1. Although the present computational study may be biased because it was performed with the initial conformation of a fibril-like structure, the metastable structures of the solvated A $\beta$ <sub>11–40</sub> trimer were obtained from the same initial structure.<sup>45</sup>

The solvated system was simulated using explicit solvent temperature REMD simulations involving 48 replicas in the range between 290 and 417 K. The temperature generator for the REMD simulations webserver<sup>52</sup> was employed to choose temperatures for our simulation with the following parameters: exchange probability of 20%, tolerance of  $10^{-4}$ , fully flexible water molecules, constrained hydrogen bonds in proteins, and full hydrogen bonds in proteins. Details of all the replica temperatures are described in the ESI file.<sup>†</sup> 350 ns MD simulations were performed for each replica, amounting to a total of 16 800 ns (16.8  $\mu$ s) of completed MD simulations. The exchanges were attempted every 1 ps; the REMD simulation included 350 000 replica exchange times. To avoid any initial bias, the first 200 ns of the REMD simulation was dismissed from the analyses. The parameters of compatibility of the computational simulations were subsequently evaluated. The results are averaged over individual snapshots.

The exchange rates between two neighbouring replicas were investigated; these are good values because they range from 26% to 38% (Fig. S1 of ESI<sup>†</sup>). Furthermore, the overlap of the sampling potential energies ensures the capability of exchange between the neighbouring replicas.<sup>81</sup> The energetic overlap between different replicas is shown in Fig. S2 (ESI<sup>†</sup>), which indicates that the solvated transmembrane systems have good exchange probabilities according to the Metropolis Criterion. Consequently, the diffusion of the temperature space of the replica temperature index was monitored and is shown in Fig. S3 (ESI<sup>†</sup>),<sup>†</sup> indicating the wide sampling of trajectories over the whole temperature space. Each replica moved through the entire temperature range, and no obstacles were present. Moreover, the secondary structures of the oligomer were estimated at 200 ns over all of the replicas, which are shown in Fig. 3. In particular, these metrics ranged from 36% to 78% of random coil, 20–56% of  $\beta$ -content, 0–13% turn structure, and 0–4% of  $\alpha$ -content. These analyses demonstrate that our simulations were not biased by any distinct conformations.

Because membrane DPPC lipid bilayers have a phase transition at about 315 K, the thermodynamic properties of the 3A $\beta$ <sub>11–40</sub> peptide and the membrane were investigated at 324 K, including the structural changes of the peptide and the mutual interactions between the membrane and A $\beta$ . Our computations were equilibrated at 324 K after 200 ns of REMD simulations because all values considered remained unchanged over two interval simulation times, 200 to 270 ns and 290 to 350 ns, including the gyration radius, RMSD,  $\beta$ -content, surface area,

and salt bridge of the peptide. The corresponding metrics in different time intervals are shown in Fig. 4. In the latter figure, the red lines correspond to the measured metrics at a simulated time interval of 200 to 270 ns, while the black lines highlight these values at the interval of 290 to 350 ns. In particular, the averages of  $R_g$  and RMSD are  $0.47 \pm 0.07$  and  $1.42 \pm 0.02$  nm, respectively. The  $\beta$ -content of the truncated trimer is  $40 \pm 7\%$ , while the surface area of the peptide is estimated to be  $64.73 \pm 3.07$  Å<sup>2</sup>. The distance between the charged groups of D23 and N27 of chain A is  $0.40 \pm 0.17$  nm.

The root mean square fluctuation (RMSF) of the transmembrane 3A $\beta$ <sub>11–40</sub> peptide at 324 K was thus evaluated over the last 150 ns of the REMD simulations. The results presented in Fig. S4 (ESI<sup>†</sup>) are approximately arranged in two regions. The first region, including the sequences 11–13, 21–30, and 38–40, exhibits a high deviation over computational time, almost higher than 0.4 Å. The second region, consisting of the sequences 14–20 and 31–37, is a low deviation domain during the REMD simulations; all the deviations are smaller than 0.4 Å. These regions correspond to the domains where the secondary structure of the peptide regularly forms random coils and  $\beta$ -structures, respectively. The high deviation domain is caused by interactions within the surface regions of the membrane DPPC lipid bilayer, which will be described in the following subsection about the intermolecular interactions between the A $\beta$  oligomer and the membrane DPPC lipid bilayer. The high fluctuation regions emerge as the main contributors to the structural changes of the transmembrane oligomer.

## Second structure of transmembrane 3A $\beta$ <sub>11–40</sub> peptide

The average of the secondary structures of the protein which emerged at 324 K during the last 150 ns of the REMD simulation was predicted using the DSSP method.<sup>64,65</sup> The averages of the random coil, beta, turn and helix structures are shown in Fig. 5. During our simulations, on average, the  $\alpha$ -structure was seldom observed, only comprising  $\sim 0.2\%$  over the simulations. The obtained data confirmed that the  $\alpha$ -structure is an intermediate step of the A $\beta$  aggregation process.<sup>37,73,74,82,83</sup> The turn structure was found to be  $\sim 3\%$ . The random coil form and  $\beta$ -content were dominant, with amounts of  $\sim 57\%$  and  $\sim 40\%$ , respectively. This finding is in good agreement with transmembrane A $\beta$  oligomer<sup>24,46</sup> experiments and the trimer of the A $\beta$ <sub>40</sub> peptide in solution.<sup>80</sup> It should be noted that in experiments, the  $\beta$ -contents of transmembrane A $\beta$  oligomers and the solvated A $\beta$  trimer are almost the same, approximately 39–40%.<sup>24,46,80</sup>

In particular, chain C has fewer  $\beta$ -structures compared to both chains A and B. This is internally consistent with the salt bridge analysis given below because chain C was found to make a salt bridge between D23 and K28. This structural change leads to the appearance of the  $\alpha$ -structure in the middle region of chain C (Fig. 5), although the  $\alpha$ -structure appears less frequently.

The secondary structure patterns can roughly be divided into five domains: namely, sequences 11–13, 20–30, and 38–40 are mostly random coil structures, and sequences 14–19 and 31–37 are rigid  $\beta$ -structures in which most of the secondary structures



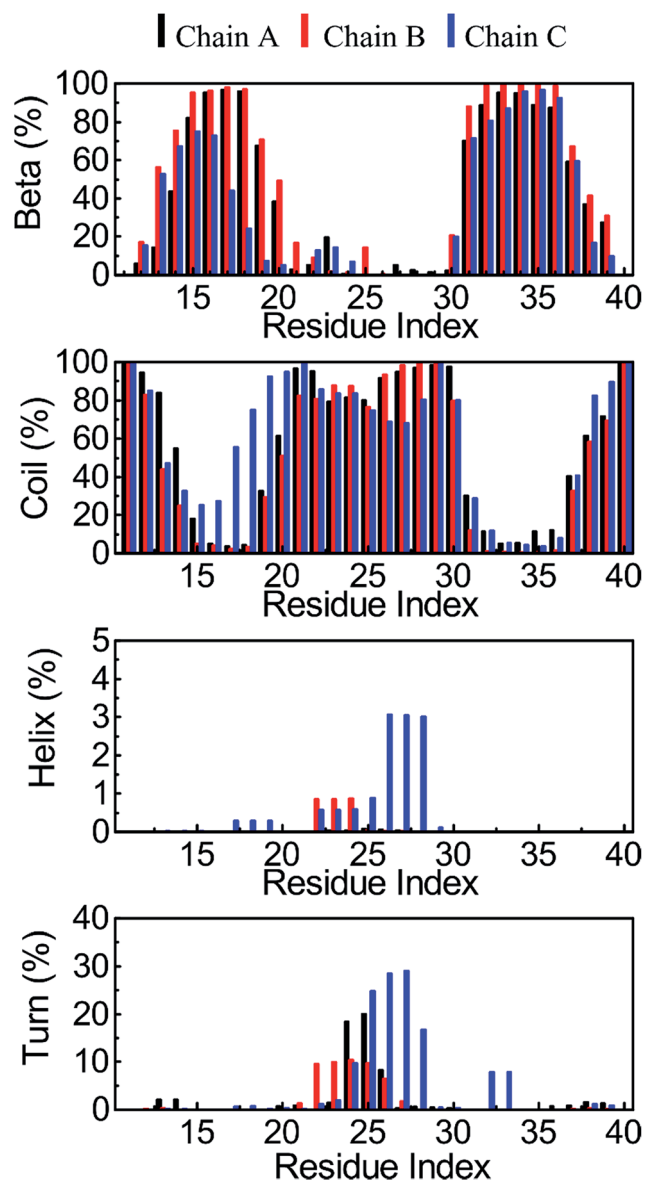


Fig. 5 The secondary structures of the transmembrane  $3A\beta_{11-40}$  peptide, averaged from the last 150 ns of the REMD simulations at 324 K. The secondary structures were predicted using DSSP tools.

have  $\beta$ -content. Overall, in analysing snapshots, the simulated structures of the transmembrane  $3A\beta_{11-40}$  peptide appear to consist of two  $\beta$ -structure domains that are separate from the random coil regions. The  $\beta$ -structure regions are essentially penetrated by the DPPC lipid bilayer. Consequently, the random coil domains are regularly located and interact with the lipid headgroups. These results are in agreement with previous experiments which found that the  $A\beta$  peptide can adopt  $\beta$  conformations when penetrating into the membrane<sup>24,27,46</sup> and can adopt coil conformations when located at the surface of the membrane.<sup>24,26</sup> Both  $\beta$ -sheet domains are characterized by strong intermolecular interactions with each other through both fibril and hydrogen bond contacts. Moreover, the random coil domains have fewer contacts between the two neighbouring chains.

### Mutual interaction contacts of the transmembrane $3A\beta_{11-40}$ peptide

The fibril and hydrogen bond contact maps between two adjacent chains of the  $3A\beta_{11-40}$  peptide were obtained from an analysis of the computational time from 200 to 350 ns of the REMD simulations with explicit solvent molecules at 324 K. The details of the fibril and hydrogen bond contact maps of the  $3A\beta_{11-40}$  oligomer are delineated in Fig. 6. The regions where its secondary structures are rigid  $\beta$ -sheets have frequent contact with each other. Particularly, the rigorous  $\beta$ -structure region of the C-terminal of chain B forms numerous intensive fibril and hydrogen bond contacts with the corresponding C-terminal domains of chains A and C (Fig. 6). Although a comparable N-terminal domain of chain B continued to produce a firm intermolecular interaction with the  $\beta$ -sheet region of the N-terminal of chain A, the fibril and hydrogen bond contacts between the corresponding domains of chains B and C decreased. This is consistent with the increasingly random coil structure of the N-terminal of chain C (Fig. 5). Overall, the obtained data indicate that the central hydrophobic cores of both the N- and C-terminals are essential to maintaining the stability of the  $A\beta$  oligomers.<sup>84</sup>

In addition, the D23–K28 salt bridge has been shown in several previous studies<sup>39,42,43,85</sup> to play an important role in stabilizing the structures of the monomers and the fibril structures of  $A\beta$  peptides in solution; the distribution of the intramolecular D23–K28 salt bridge of the transmembrane  $3A\beta_{11-40}$  peptide was further analysed. This distribution is shown in Fig. 7. At the starting point, no salt bridge appeared. After simulation over a long period of time using REMD, chain C was found to form a salt bridge between Asp23 and Lys28 with a very high probability. However, the Asp23–Lys28 salt bridges of chains A and B were rarely observed. This is due to the fact that

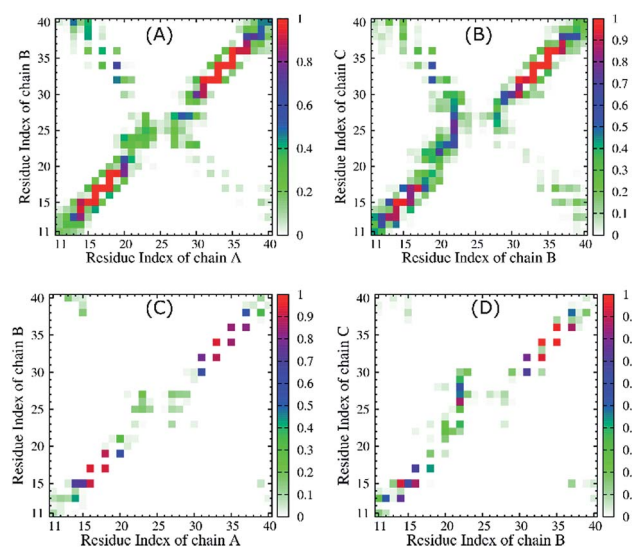


Fig. 6 (A) + (B) and (C) + (D) show the fibril and hydrogen bond contact maps between the neighbouring chains of the transmembrane truncated  $3A\beta_{11-40}$  peptide at 324 K over equilibrium snapshots of the REMD simulations, respectively.



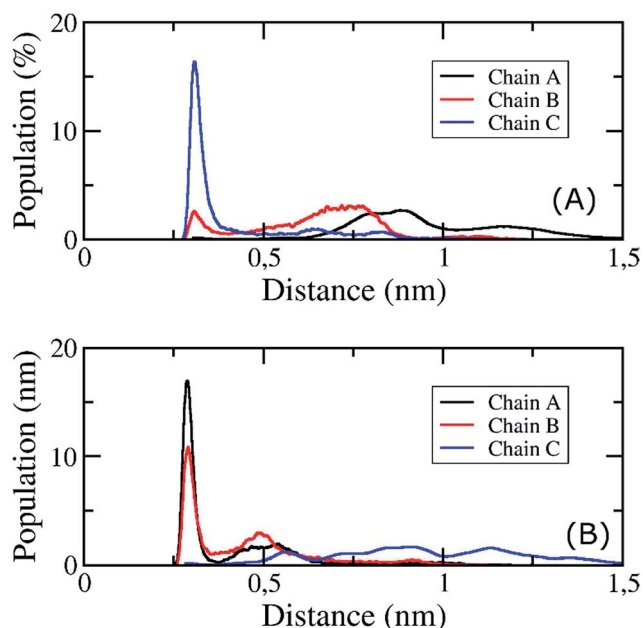


Fig. 7 The distribution of the Asp23–Lys28 salt bridge (A) and the Asp23–Asn27 salt bridge (B) of the transmembrane  $3A\beta_{11-40}$  oligomer at 324 K during the simulation time interval of 200 to 350 ns of the REMD simulations.

both Lys28A and Lys28B form hydrogen bonds with lipid headgroups located within this region. These interactions are the main element altering the secondary structures of the peptides at the loop region. This result is consistent with available computational studies about the effects of lipid bilayers on the structure of the  $A\beta_{1-40}$  monomer.<sup>86,87</sup> Instead, in our present simulation, the polar contact between Asp23 and Asn27 of both chains A and B was found to be a stabilizing contact (Fig. 7B). This contact replaces the salt bridge Asp23–Lys28 in order to secure the turn region, in accord with a previous computational study showing that the membrane alters this salt bridge of the Amyloid precursor protein.<sup>88</sup>

### The optimized structure of the transmembrane $A\beta_{11-40}$ trimer

In an attempt to identify the putative structures of the  $3A\beta_{11-40}$  peptide penetrating the membrane DPPC lipid bilayer, the two-dimensional free energy landscape was constructed using the “gmsham” tool. The RMSD of the protein was chosen as the

first reaction coordinate. The radius of gyration  $R_g = \frac{\sum_i m_i d_i}{\sum_i m_i}$

is defined as the average of the mass-weighted squared distances of all atoms to the center of mass; this was chosen as the second reaction coordinate. The clustering method was then used to search for the stable structures of the peptide which remain at the minima.

Fig. 8 shows the FEL of the truncated  $3A\beta_{11-40}$  peptide at 324 K. In total, there are five representative structures, which are noted as M1 to M5. Four of these structures (*i.e.* M1 to M4) have the same magnitudes of  $R_g$  and RMSD, approximately 1.43 and

0.55 nm, respectively; they have the lowest free energy value of  $-13.9 \text{ kJ mol}^{-1}$ . M5 is located in a different free energy hole, with a free energy of  $-11.59 \text{ kJ mol}^{-1}$ . This conformation has  $R_g$  and RMSD values of  $\sim 1.40$  and  $\sim 0.63$  nm, respectively. The details of the secondary structures of these conformations are shown in Table 1. On average, the coil structure occupies 54% and the  $\beta$ -content occupies 44%. The turn structure represents approximately 2%, and the helix structure completely disappears. This finding may be due to the fact that the putative structures tend to be  $A\beta$  fibril formations because the helix structure is the intermediate step of  $A\beta$  aggregation.<sup>37,73,74,82,83</sup> These observations are in accord with previous experimental findings that lipids alter the secondary structure of  $A\beta$  to approximately 40–60%  $\beta$ -content,<sup>24,46</sup> whereas the random coil structure is found near the lipid headgroups.<sup>24,26</sup>

In particular, the optimized conformation M2 adopts the highest possibility of beta structure, with an amount of 53%. The other metrics, including the alternate random coil and turn structures, occupy 45% and 2%, respectively. It is interesting to note that M2 is the most common conformation of the trimer in the lowest free energy state, with a population of  $\sim 29\%$  of the total snapshots, located in the lowest free energy minimum; this was predicted through the clustering method with a cutoff of 0.35 nm. Therefore, we chose M2 as the initial conformation to estimate the annihilation free energy of the transmembrane  $A\beta_{11-40}$  trimer.

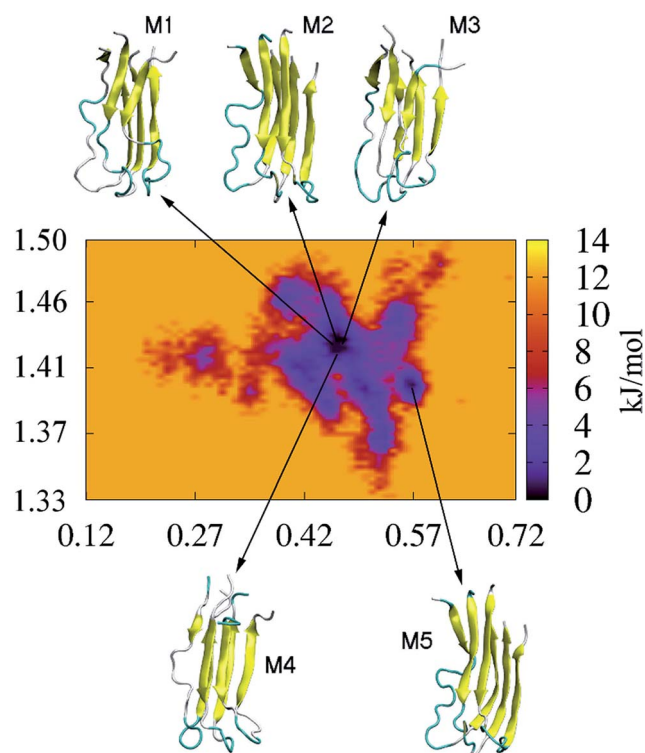


Fig. 8 Free energy landscape of the transmembrane  $3A\beta_{11-40}$  peptide as a function of the first two principal components,  $R_g$  and RMSD. M1 to M5 are minima of the transmembrane truncated  $3A\beta_{11-40}$  peptide inserted into the DPPC lipid bilayer; these were obtained from FEL analysis during the last 150 ns of the REMD simulations at 324 K.





**Table 1** Details of secondary structures and CCS of five globally optimized structures of the A $\beta$  oligomer, predicted through DSSP and MOBCAL tools

Minima	Coil content (%)	Beta content (%)	Turn content (%)	Helix content (%)	CCS ( $\text{\AA}^2$ )
M1	58	40	2	0	1361
M2	45	53	2	0	1311
M3	62	38	0	0	1371
M4	56	40	4	0	1317
M5	51	49	0	0	1343

In this free energy hole, the populations of M1, M3, and M4 amount to 9%, 21% and 13%, respectively. Meanwhile, M5 is located in a different free energy hole, with a population of 38% of all snapshots at this state. The rest are other clusters whose populations are quite low. The putative structure M3 has the lowest probability of  $\beta$ -content, at 38%, and the highest probability of random coil structure, with an amount of 62%. There is no turn structure in the M3 minimum.

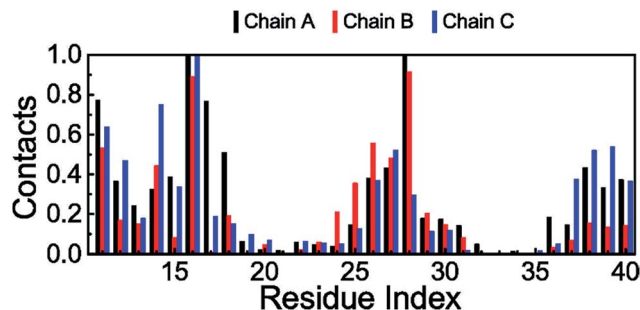
Additionally, both the M1 and M4 conformations form the same amounts of  $\beta$ -structures, at 40%. However, M4 has more turn structure at 4%, whereas M1 only forms 2%. Thus, the random coil structures of M1 and M4 attain 58% and 56%, respectively, due to the lack of helix structures. The optimized structure M5 contains 49% beta structure and 51% random coil. The other metrics,  $\alpha$ - and turn-content, have zero percentages.

Although several previous studies are available on the screening of potential inhibitors for AD, including candidates to inhibit A $\beta$  peptides,<sup>56,89–93</sup> an effective drug for the treatment of AD remains elusive. This may be because the structures of solvated A $\beta$  peptides are employed as targets in most screening of potential inhibitors for A $\beta$  peptides.<sup>56,89–93</sup> However, as discussed above, the major mechanism of A $\beta$  neurotoxicity arises from the membrane penetration of A $\beta$  oligomers. Strong inhibitors of A $\beta$  peptides in solution appear not to effectively inhibit these peptides when they penetrate the membrane. Therefore, the putative structures of the transmembrane trimer A $\beta_{11-40}$  peptide that are determined using REMD simulations can be used as targets for computer-aided drug design that may lead to the discovery of an effective inhibitor of A $\beta$  oligomers.

It is known that the CCS is an important parameter for characterizing proteins. This metric can be determined not only in experiments, such as ion mobility mass spectrometry, but also by computation. The IMPACT method has successfully estimated the CCS of proteins with high correlation to experiments.<sup>69,94</sup> Thus, the CCS of the optimized structure of the transmembrane A $\beta$  trimer was investigated using IMPACT. The obtained results are shown in Table 1; the mean of the metrics is  $1341 \pm 23 \text{ \AA}^2$ . As far as we know, the experimental CCS of the transmembrane A $\beta$  trimer is still unavailable. However, the CCS of the A $\beta$  trimer in solution was determined in previous experiments, with values of  $1265 \pm 16$  and  $1386 \pm 145 \text{ \AA}^2$ .<sup>95,96</sup> Overall, our calculated CCS result is in good agreement with these experiments.

### Intermolecular interaction between the DPPC lipid bilayer and the A $\beta$ peptide

Intermolecular interactions between the peptide and the membrane were also computed through evaluation of the minimum distance between the heavy atoms of the A $\beta$  residues and the phosphorus atoms of the headgroups. A contact is recorded whenever the distance becomes smaller than 0.45 nm. The probability of existence of these contacts is demonstrated in Fig. 9. Consistent with the investigation of the RMSF and secondary structures, the residues that have high deviations and solid coil structures frequently form non-bonded contacts with the membrane surface. Additionally, the  $\beta$ -structure domains of the N-terminals of the peptides are commonly found to interact with lipid headgroups. Moreover, both Lys16 and Lys28 have the most regular contact with lipid phosphate headgroups, which is in good agreement with previous studies on the interaction between a membrane and the A $\beta_{1-40}$  monomer.<sup>86,87</sup> Although the observed results imply that electrostatic interactions are the main factor in the binding process between the A $\beta$  trimer and the membrane DPPC lipid bilayer, this qualitative finding is inconsistent with the quantitative analysis of binding free energy calculations, which is mentioned below, indicating that van der Waals interactions emerge as the key element of interaction between the trimer and the membrane. These inconsistent results thus suggest that a study on the interactions of A $\beta$  peptides and membranes using evaluation of the distance between the different heavy atoms of the different molecules could enable artificial observation.



**Fig. 9** Probability of intermolecular contacts between the phosphorus atoms of bilayer DPPC lipids and the heavy atoms of the truncated 3A $\beta_{11-40}$  peptide. The results were investigated at 324 K throughout the last 150 ns of the REMD simulations.





### Free energy difference of binding between the DPPC lipid bilayer and the A $\beta$ trimer

In order to investigate the nature of binding between the trimer and the lipid bilayer, the double-annihilation binding free energy method<sup>61,63</sup> was used to estimate for the first time to estimate the binding free energy of the A $\beta$  trimer to the DPPC lipid bilayer, because it is one of the most accurate methods currently known for this purpose.<sup>97–99</sup> In this scheme, the protein is annihilated from both the transmembrane and solvated systems. To determine the binding free energy, in the first step, the putative structure of the transmembrane A $\beta$  trimer in M2 is chosen as the initial conformation of the first annihilation simulation; it is annihilated from the complex, and subsequently  $\Delta G_1$  is determined. In the second step, the trimer is solvated and then equilibrated in 30 ns of MD simulations. The RMSD analysis indicates that the system reaches equilibrium after 0.2 ns (*cf.* Fig. S5†). The last snapshot of the MD is chosen as the initial conformation for the free energy calculation using the FEP method. In this process, the protein is annihilated from solution; then,  $\Delta G_2$  is determined. The free energy of this process was determined using the BAR method<sup>59</sup> in every ps, and the results are shown in Fig. 10. The free energies reached equilibrium after 2 ns; then, the binding free energy  $\Delta G_{\text{bind}}$  of the oligomer to the DPPC lipid bilayer was determined to be  $-70 \pm 18 \text{ kcal mol}^{-1}$ , which is averaged from the differences of the two annihilation free energies. In particular, the contribution of electrostatic energy ( $\Delta G_{\text{cou}}$ ) to the binding free energy amounts to  $-114 \pm 18 \text{ kcal mol}^{-1}$ , and the difference in free energy of the van der Waals interactions

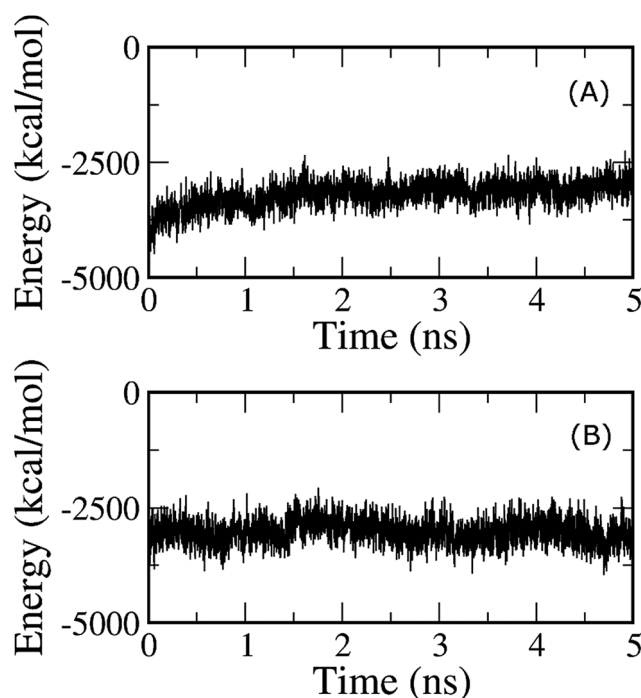


Fig. 10 (A) Annihilation free energy of the A $\beta$  trimer from the solvated membrane system, and (B) annihilation free energy of the A $\beta$  trimer from the solution. The free energy was determined using the BAR method at every ps.

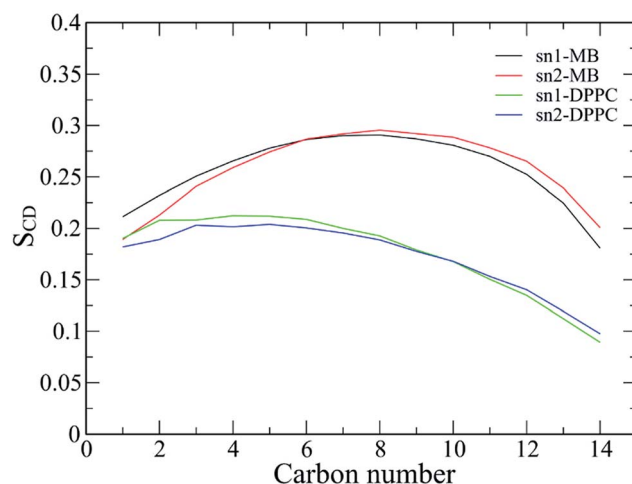


Fig. 11 The lipid order parameters for both carbon atoms of acyl chains sn-1 and sn-2. The results are averaged over the last 150 ns of the temperature REMD simulations.

( $\Delta G_{\text{vdw}}$ ) is  $-184 \pm 3 \text{ kcal mol}^{-1}$ . These results indicate that although intermolecular polar contacts between the A $\beta$  peptide and the membrane are very important, according to the analysis given above and also from previous studies,<sup>21,28</sup> the natures of the binding differ from each other. The van der Waals interactions emerge as dominant over the Coulomb interactions in the binding process of the A $\beta$  peptide to the membrane DPPC lipid bilayer.

### Stabilization of the membrane over simulation time

In order to further detect the effects of the A $\beta$  oligomer on the oriented mobility of the membrane DPPC lipid bilayer, the averages of the lipid order parameters were computed for both carbon atoms of the acyl chains sn-1 and sn-2, as shown in Fig. 11. The membrane DPPC lipid bilayer without the induced oligomer is simulated in solution. The lipid order parameters of the solvated DPPC lipid bilayer are indicated by the green and blue lines in Fig. 11. MD simulation of the pure lipid bilayer was carried out for a length of 200 ns. Our pure solvated DPPC lipid bilayer observations are in good agreement with both previous experimental and computational results.<sup>100–102</sup> In addition, the lipid order parameters of the membrane with the oligomer induced are also investigated, as illustrated by the black and red lines in Fig. 11. Generally, the lipid order parameter curves of the two saturated DPPC chains are quite different from each other for both systems with and without the oligomer. The results obtained here are consistent with the view that the interaction between the A $\beta$  peptides and the lipid headgroups leads to fluctuation of the lipid order parameters.<sup>26,103,104</sup> However, the obtained results also indicate that the membrane DPPC lipid bilayer is stabilized upon protein insertion.

## Conclusions

In the present theoretical study, the high efficiency sampling method of REMD simulations with explicit solvent was applied



to investigate the structural changes of the  $3A\beta_{11-40}$  oligomer inserted in the membrane of a DPPC lipid bilayer.

We firstly generated the stable structures of the  $A\beta_{11-40}$  trimer when it is inserted into the membrane DPPC lipid bilayer. On average, the percentages of the putative trimer secondary structure were estimated to be 44%  $\beta$ -structure, 54% random coil structure, 2% turn structure, and absolutely 0%  $\alpha$ -structure. There was a small shift compared with the averaged metrics of all snapshots, which were 40%  $\beta$ -content, 57% random coil, 3% turn structure and 0%  $\alpha$ -structure. Consequently, the  $\beta$ -content is slightly higher than the  $\beta$ -content of the solvated  $A\beta_{11-40}$  trimer.<sup>45</sup> However, this difference may be caused by the tendency of the GROMOS force field to favor  $\beta$  formation.<sup>79</sup> Moreover, we found that the transmembrane  $A\beta$  trimer structure is arranged in two domains, including the favored  $\beta$ -structure region that is fully inserted into the membrane, involving the sequences 14–19 and 31–37. The mostly random coil region was found to be located close to the surface of the membrane and consists of the sequences 11–13, 20–30 and 38–40. These two domains correspond to the stable and fluctuating regions of the oligomer, respectively. This finding is in agreement with both previous experimental<sup>24,26,27,46</sup> and computational<sup>28–33</sup> studies. We hope that new knowledge of these structures can enhance the search for an  $A\beta$  therapy.

The membrane of the DPPC lipid bilayer was in addition found to be stable during the REMD simulations, and the intermolecular interaction contacts between the oligomer and the DPPC lipid bilayer indicated that the random coil domain favors interaction with the membrane headgroups. Moreover, the van der Waals interaction energy is predominant over the electrostatic energy in the binding of the  $A\beta$  trimer to the membrane, based on double-annihilation binding free energy calculations. Consequently, the trimer experiences a strong attraction with the membrane, with a large binding free energy of  $-70 \pm 18 \text{ kcal mol}^{-1}$ .

The size of the transmembrane  $A\beta_{11-40}$  trimer is approximately equal to the size of the  $A\beta$  trimer in solution, with a cross section (CCS) of  $1341 \pm 23 \text{ \AA}^2$ ; this compares well with the experimental values of the solvated trimer of  $1265 \pm 16$  and  $1386 \pm 145 \text{ \AA}^2$ .<sup>95,96</sup>

The membrane DPPC lipid bilayer induces a strong effect on chain C, which leads to a smaller number of non-bonded contacts with chain B compared to the contacts between chains A and B. Although the membrane of the DPPC lipid bilayer was stable during all the MD computations, the lipid order parameters fluctuated when the oligomer was induced.<sup>26,103,104</sup>

The polar contact Asp23–Asn27 was also found to be the stabilizing factor of the oligomer structure, rather than the Asp23–Lys28 salt bridge as reported in a previous study by Miyashita *et al.*<sup>30</sup> on the transmembrane amyloid precursor protein.

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## References

- 1 A. S. Henderson and A. F. Jorm, *Dementia*, John Wiley & Sons Ltd, 2002.
- 2 D. J. Selkoe, *Neuron*, 1991, **6**, 487–498.
- 3 J. L. Cummings, *N. Engl. J. Med.*, 2004, **351**, 56–67.
- 4 H. W. Querfurth and F. M. LaFerla, *N. Engl. J. Med.*, 2010, **362**, 329–344.
- 5 J. Nasica-Labouze, P. H. Nguyen, F. Sterpone, O. Berthoumieu, N.-V. Buchete, S. Coté, A. De Simone, A. J. Doig, P. Faller, A. Garcia, A. Laio, M. S. Li, S. Melchionna, N. Mousseau, Y. Mu, A. Paravastu, S. Pasquali, D. J. Rosenman, B. Strodel, B. Tarus, J. H. Viles, T. Zhang, C. Wang and P. Derreumaux, *Chem. Rev.*, 2015, **115**, 3518–3563.
- 6 Alzheimer's association, Journal, 2015.
- 7 D. J. Selkoe, *Science*, 2002, **298**, 789–791.
- 8 I. Khlistunova, J. Biernat, Y. Wang, M. Pickhardt, M. von Bergen, Z. Gazova, E. Mandelkow and E.-M. Mandelkow, *J. Biol. Chem.*, 2006, **281**, 1205–1214.
- 9 K. SantaCruz, J. Lewis, T. Spires, J. Paulson, L. Kotilinek, M. Ingelsson, A. Guimaraes, M. DeTure, M. Ramsden, E. McGowan, C. Forster, M. Yue, J. Orne, C. Janus, A. Mariash, M. Kuskowski, B. Hyman, M. Hutton and K. H. Ashe, *Science*, 2005, **309**, 476–481.
- 10 J. Hardy and D. J. Selkoe, *Science*, 2002, **297**, 353–356.
- 11 M. Citron, *Nat. Rev. Neurosci.*, 2004, **5**, 677–685.
- 12 A. Aguzzi and T. O'Connor, *Nat. Rev. Drug Discovery*, 2010, **9**, 237–248.
- 13 D. M. Walsh and D. J. Selkoe, *J. Neurochem.*, 2007, **101**, 1172–1184.
- 14 M. Bucciantini, E. Giannoni, F. Chiti, F. Baroni, L. Formigli, J. Zurdo, N. Taddei, G. Ramponi, C. M. Dobson and M. Stefani, *Nature*, 2002, **416**, 507–511.
- 15 A. Quist, I. Doudevski, H. Lin, R. Azimova, D. Ng, B. Frangione, B. Kagan, J. Ghiso and R. Lal, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 10427–10432.
- 16 T. L. Williams and L. C. Serpell, *FEBS J.*, 2011, **278**, 3905–3917.
- 17 Y. Nakazawa, Y. Suzuki, M. P. Williamson, H. Saitô and T. Asakura, *Chem. Phys. Lipids*, 2009, **158**, 54–60.
- 18 H. Jang, J. Zheng and R. Nussinov, *Biophys. J.*, 2007, **93**, 1938–1949.
- 19 C. Ege and K. Y. C. Lee, *Biophys. J.*, 2004, **87**, 1732–1740.
- 20 H. Zhao, E. K. J. Tuominen and P. K. J. Kinnunen, *Biochemistry*, 2004, **43**, 10302–10307.
- 21 C. Poojari, A. Kukol and B. Strodel, *Biochim. Biophys. Acta, Biomembr.*, 2013, **1828**, 327–339.
- 22 J. J. Kremer, D. J. Sklansky and R. M. Murphy, *Biochemistry*, 2001, **40**, 8563–8571.
- 23 H. Shao, S.-c. Jao, K. Ma and M. G. Zagorski, *J. Mol. Biol.*, 1999, **285**, 755–773.
- 24 E. Terzi, G. Hölzemann and J. Seelig, *J. Mol. Biol.*, 1995, **252**, 633–642.



- 25 O. Crescenzi, S. Tomaselli, R. Guerrini, S. Salvadori, A. M. D'Ursi, P. A. Temussi and D. Picone, *Eur. J. Biochem.*, 2002, **269**, 5642–5648.
- 26 J. McLaurin and A. Chakrabartty, *Eur. J. Biochem.*, 1997, **245**, 355–363.
- 27 M. Meier and J. Seelig, *J. Mol. Biol.*, 2007, **369**, 277–289.
- 28 B. Strodel, J. W. L. Lee, C. S. Whittleston and D. J. Wales, *J. Am. Chem. Soc.*, 2010, **132**, 13300–13312.
- 29 C. Lockhart and D. K. Klimov, *J. Phys. Chem. B*, 2014, **118**, 2638–2648.
- 30 N. Miyashita, J. E. Straub and D. Thirumalai, *J. Am. Chem. Soc.*, 2009, **131**, 17843–17852.
- 31 Q. Wang, J. Zhao, X. Yu, C. Zhao, L. Li and J. Zheng, *Langmuir*, 2010, **26**, 12722–12732.
- 32 Z. Chang, Y. Luo, Y. Zhang and G. Wei, *J. Phys. Chem. B*, 2011, **115**, 1165–1174.
- 33 H. Jang, J. Zheng, R. Lal and R. Nussinov, *Trends Biochem. Sci.*, 2008, **33**, 91–100.
- 34 M. K. Jana, R. Cappai, C. L. L. Pham and G. D. Cicciotosto, *J. Neurochem.*, 2016, **136**, 594–608.
- 35 Y. Sun, W. Xi and G. Wei, *J. Phys. Chem. B*, 2015, **119**, 2786–2794.
- 36 G. Bitan, M. D. Kirkitadze, A. Lomakin, S. S. Vollers, G. B. Benedek and D. B. Teplow, *Proc. Natl. Acad. Sci. U. S. A.*, 2003, **100**, 330–335.
- 37 M. D. Kirkitadze, M. M. Condrón and D. B. Teplow, *J. Mol. Biol.*, 2001, **312**, 1103–1119.
- 38 L. K. Thompson, *Proc. Natl. Acad. Sci. U. S. A.*, 2003, **100**, 383–385.
- 39 M. H. Viet, P. H. Nguyen, S. T. Ngo, M. S. Li and P. Derreumaux, *ACS Chem. Neurosci.*, 2013, **4**, 1446–1457.
- 40 B. Murray, M. Sorci, J. Rosenthal, J. Lippens, D. Isaacson, P. Das, D. Fabris, S. Li and G. Belfort, *Proteins: Struct., Funct., Bioinf.*, 2016, **84**, 488–500.
- 41 P. H. Nguyen, F. Sterpone, J. M. Campanera, J. Nasica-Labouze and P. Derreumaux, *ACS Chem. Neurosci.*, 2016, **7**, 823–832.
- 42 A. T. Petkova, Y. Ishii, J. J. Balbach, O. N. Antzutkin, R. D. Leapman, F. Delaglio and R. Tycko, *Proc. Natl. Acad. Sci. U. S. A.*, 2002, **99**, 16742–16747.
- 43 A. T. Petkova, W. M. Yau and R. Tycko, *Biochemistry*, 2006, **45**, 498–512.
- 44 I. Bertini, L. Gonnelli, C. Luchinat, J. Mao and A. Nesi, *J. Am. Chem. Soc.*, 2011, **133**, 16013–16022.
- 45 S. T. Ngo, H. M. Hung, D. T. Truong and M. T. Nguyen, *Phys. Chem. Chem. Phys.*, 2017, DOI: 10.1039/c6cp05511g.
- 46 E. Terzi, G. Hoelzemann and J. Seelig, *Biochemistry*, 1994, **33**, 7434–7441.
- 47 C. Oostenbrink, A. Villa, A. E. Mark and W. F. Van Gunsteren, *J. Comput. Chem.*, 2004, **25**, 1656–1676.
- 48 J. F. Nagle, *Biophys. J.*, 1993, **64**, 1476–1481.
- 49 H. J. C. Berendsen, J. P. M. Postma, W. F. van Gunsteren and A. J. Hermans, *Intermolecular Forces*, Reidel, Dordrecht, Jerusalem, Israel, 1981.
- 50 M. J. Abraham, T. Murtola, R. Schulz, S. Páll, J. C. Smith, B. Hess and E. Lindahl, *SoftwareX*, 2015, **1–2**, 19–25.
- 51 D. F. Shanno, *Math. Comput.*, 1970, **24**, 647–656.
- 52 A. Patriksson and D. van der Spoel, *Phys. Chem. Chem. Phys.*, 2008, **10**, 2073–2077.
- 53 W. F. Van Gunsteren and H. J. C. Berendsen, *Mol. Simul.*, 1988, **1**, 173–185.
- 54 M. Parrinello and A. Rahman, *J. Appl. Phys.*, 1981, **52**, 7182–7190.
- 55 B. Hess, H. Bekker, H. J. C. Berendsen and J. G. E. M. Fraaije, *J. Comput. Chem.*, 1997, **18**, 1463–1472.
- 56 M. H. Viet, S. T. Ngo, N. S. Lam and M. S. Li, *J. Phys. Chem. B*, 2011, **115**, 7433–7446.
- 57 T. Darden, D. York and L. Pedersen, *J. Chem. Phys.*, 1993, **98**, 10089–10092.
- 58 R. W. Zwanzig, *J. Chem. Phys.*, 1954, **22**, 1420–1426.
- 59 C. H. Bennett, *J. Comput. Phys.*, 1976, **22**, 245–268.
- 60 W. L. Jorgensen, J. K. Buckner, S. Boudon and J. Tirado-Rives, *J. Chem. Phys.*, 1988, **89**, 3742–3746.
- 61 G. Jayachandran, M. R. Shirts, S. Park and V. S. Pande, *J. Chem. Phys.*, 2006, **125**, 084901.
- 62 H. Fujitani, Y. Tanida, M. Ito, G. Jayachandran and C. D. Snow, *J. Chem. Phys.*, 2005, **123**, 084108.
- 63 S. T. Ngo, B. K. Mai, D. M. Hiep and M. S. Li, *Chem. Biol. Drug Des.*, 2015, **86**, 546–558.
- 64 W. G. Touw, C. Baakman, J. Black, T. A. H. te Beek, E. Krieger, R. P. Joosten and G. Vriend, *Nucleic Acids Res.*, 2015, **43**, D364–D368.
- 65 R. P. Joosten, T. A. H. te Beek, E. Krieger, M. L. Hekkelman, R. W. W. Hooft, R. Schneider, C. Sander and G. Vriend, *Nucleic Acids Res.*, 2011, **39**, D411–D419.
- 66 Y. Mu, P. H. Nguyen and G. Stock, *Proteins: Struct., Funct., Bioinf.*, 2005, **58**, 45–52.
- 67 E. Papaleo, P. Mereghetti, P. Fantucci, R. Grandori and L. De Gioia, *J. Mol. Graphics Modell.*, 2009, **27**, 889–899.
- 68 X. Daura, K. Gademann, B. Jaun, D. Seebach, W. F. van Gunsteren and A. Mark, *Angew. Chem., Int. Ed.*, 1999, **38**, 236–240.
- 69 E. G. Marklund, M. T. Degiacomi, C. V. Robinson, A. J. Baldwin and J. L. P. Benesch, *Structure*, 2015, **23**, 791–799.
- 70 A. Morriss-Andrews and J. E. Shea, in *Annual Review of Physical Chemistry*, ed. M. A. Johnson and T. J. Martinez, 2015, vol. 66, pp. 643–666.
- 71 Y. Sugita and Y. Okamoto, *Chem. Phys. Lett.*, 1999, **314**, 141–151.
- 72 C. H. Davis and M. L. Berkowitz, *J. Phys. Chem. B*, 2009, **113**, 14480–14486.
- 73 S. Jang and S. Shin, *J. Phys. Chem. B*, 2006, **110**, 1955–1958.
- 74 S. Jang and S. Shin, *J. Phys. Chem. B*, 2008, **112**, 3479–3484.
- 75 A. Okamoto, A. Yano, K. Nomura, S. i. Higai and N. Kurita, *Chem. Phys. Lett.*, 2013, **577**, 131–137.
- 76 A. Yano, A. Okamoto, K. Nomura, S. i. Higai and N. Kurita, *Chem. Phys. Lett.*, 2014, **595–596**, 242–249.
- 77 H. Nymeyer, *J. Chem. Theory Comput.*, 2008, **4**, 626–636.
- 78 T. Mori, J. Jung and Y. Sugita, *J. Chem. Theory Comput.*, 2013, **9**, 5629–5640.
- 79 A. K. Somavarapu and K. P. Kepp, *ChemPhysChem*, 2015, **16**, 3278–3289.
- 80 K. Ono, M. M. Condrón and D. B. Teplow, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 14745–14750.





- 81 M. Meli and G. Colombo, *Int. J. Mol. Sci.*, 2013, **14**, 12157–12169.
- 82 Y. Fezoui and D. B. Teplow, *J. Biol. Chem.*, 2002, **277**, 36948–36954.
- 83 D. K. Klimov and D. Thirumalai, *Structure*, 2003, **11**, 295–307.
- 84 W. Han and K. Schulten, *J. Am. Chem. Soc.*, 2014, **136**, 12450–12460.
- 85 T. Luhers, C. Ritter, M. Adrian, D. Riek-Loher, B. Bohrmann, H. Doeli, D. Schubert and R. Riek, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 17342–17347.
- 86 J. A. Lemkul and D. R. Bevan, *Arch. Biochem. Biophys.*, 2008, **470**, 54–63.
- 87 J. A. Lemkul and D. R. Bevan, *Protein Sci.*, 2011, **20**, 1530–1545.
- 88 N. Miyashita, J. E. Straub, D. Thirumalai and Y. Sugita, *J. Am. Chem. Soc.*, 2009, **131**, 3438–3439.
- 89 S. T. Ngo and M. S. Li, *J. Phys. Chem. B*, 2012, **116**, 10165–10175.
- 90 P. D. Huy, Y. C. Yu, S. T. Ngo, T. V. Thao, C. P. Chen, M. S. Li and Y. C. Chen, *Biochim. Biophys. Acta*, 2013, **1830**, 2960–2969.
- 91 S. T. Ngo and M. S. Li, *Mol. Simul.*, 2013, **39**, 279–291.
- 92 S.-H. Wang, F.-F. Liu, X.-Y. Dong and Y. Sun, *J. Phys. Chem. B*, 2010, **114**, 11576–11583.
- 93 W.-J. Du, J.-J. Guo, M.-T. Gao, S.-Q. Hu, X.-Y. Dong, Y.-F. Han, F.-F. Liu, S. Jiang and Y. Sun, *Sci. Rep.*, 2015, **5**, 7992.
- 94 Y. Sun, S. Vahidi, M. A. Sowole and L. Konermann, *J. Am. Soc. Mass Spectrom.*, 2016, **27**, 31–40.
- 95 E. Sitkiewicz, J. Olędzki, J. Poznański and M. Dadlez, *PLoS One*, 2014, **9**, e100200.
- 96 M. Kłoniecki, A. Jabłonowska, J. Poznański, J. Langridge, C. Hughes, I. Campuzano, K. Giles and M. Dadlez, *J. Mol. Biol.*, 2011, **407**, 110–124.
- 97 J. Aqvist, C. Medina and J.-E. Samuelsson, *Protein Eng.*, 1994, **7**, 385–391.
- 98 J. Srinivasan, T. E. Cheatham, P. Cieplak, P. A. Kollman and D. A. Case, *J. Am. Chem. Soc.*, 1998, **120**, 9401–9409.
- 99 S. T. Ngo, H. M. Hung and M. T. Nguyen, *J. Comput. Chem.*, 2016, **37**, 2734–2742.
- 100 H. I. Petrache, S. W. Dodd and M. F. Brown, *Biophys. J.*, 2000, **79**, 3172–3192.
- 101 D. P. Tieleman, S. J. Marrink and H. J. C. Berendsen, *BBA, Biochim. Biophys. Acta, Rev. Biomembr.*, 1997, **1331**, 235–270.
- 102 H. M. Hung, V. P. Nguyen, S. T. Ngo and M. T. Nguyen, *Biophys. Chem.*, 2016, **217**, 1–7.
- 103 C. E. Dempsey, *BBA, Biochim. Biophys. Acta, Rev. Biomembr.*, 1990, **1031**, 143–161.
- 104 S. H. White, W. C. Wimley and M. E. Selsted, *Curr. Opin. Struct. Biol.*, 1995, **5**, 521–527.

