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Pb(Lu_{1/2}Nb_{1/2})O₃ single crystals and their superlattice structure

Domain and antiferroelectric properties of

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In order to obtain antiferroelectric single crystals for understanding their structure and the physics and chemistry of the antiferroelectric phenomenon, antiferroelectric single crystals of Pb(Lu_{1/2}Nb_{1/2})O₃ (PLN) were successfully grown. Two sets of superlattice reflections including the B-site atom ordering type reflection and the lead-ion antiparallel ordering type reflection were confirmed and distinguished from the X-ray diffraction patterns and transmission electron microscopy. Micro-domains and macro-domains were observed under the microscopes. The investigations were focused on the extinction angle, the angle of the domain wall and temperature dependence. The Curie temperature $T_{\rm C}$ was about 240 °C and 254 °C for [111]-oriented and [001]-oriented crystal plates respectively. The value of the energy density was calculated to be 3.65 J cm⁻³ for the [111]-oriented sample at 200 °C based on the double hysteresis loops with an electric field $E_{\rm APP}=170~{\rm kV~cm^{-1}}$ applied. All the above features, especially the high Curie temperature and high energy storage density, suggested that the PLN single crystals should be useful for large size energy storage devices that require a wider operation temperature range.

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1. Introduction

An antiferroelectric state is defined as one in which lines of ions in the crystal are spontaneously polarized, but with neighboring lines polarized in antiparallel directions. Among the numerous antiferroelectrics with perovskite structure, a typical and one of the most widely studied antiferroelectric compositions is PbZrO₃.^{2,3} Pb(Zr_{1-x}Ti_x)O₃ (PZT) with low Ti-content (x < 0.05) is an antiferroelectric material at room temperature and has drawn much attention among physics and material science researchers owing to its technological importance and complex phase diagram.3,4 Recently, a number of investigations based on La³⁺ and Sn⁴⁺ modified PZT ceramics and thin films were reported.5-8 On the other hand, compounds of ABO3 including $A(B'_{1/2}B''_{1/2})O_3$ and $(A'_{1/2}A''_{1/2})BO_3$, as an another class of antiferroelectric materials, have recently attracted much research interest.9-12 Especially, a group of complex lead-based perovskite oxides with chemical formula $Pb(B'_xB''_{1-x})O_3$ have considerable interest due to their variety of physical properties and wide range of applications such as high-energy density storage capacitors. Pb(Mg_{1/2}W_{1/2})O₃, Pb(Co_{1/2}W_{1/2})O₃ and Pb(Yb_{1/2}Nb_{1/2})O₃ are the well-known members of antiferroelectrics which belong to a subgroup of B-site atom ordering and a long coherence length of long-range ordering. 10,11 Besides, the A-site complex perovskite compounds, such as $(Na_{1/2}Bi_{1/2})TiO_3 (NBT)^{12}$ and $(K_{1/2}Na_{1/2})NbO_3 (KNN)^{13}$ based solid solution were also studied.

As a member of $Pb(B'_{1/2}B''_{1/2})O_3$ family, $Pb(Lu_{1/2}Nb_{1/2})O_3$ (PLN) have been received some attention in the past. The PLN ceramics were obtained with long range order parameter Ω close to unity, and a sharp antiferroelectric–paraelectric phase transition at about 258 °C in the absence of dielectric dispersion. 10,14 Antonova M. *et al.* also reported the $Pb(Lu_{1/2}Nb_{1/2})O_3$ – $PbTiO_3$ (PLN–PT) ceramics system prepared by a hot-pressed routine method, a pure PLN was presented with long-range order in the B sublattice and an orthorhombic cell. 15 Highly oriented PLN thin film heterostructures with bottom $La_{1/2}Sr_{1/2}CoO_3$ (LSCO) and top Pt electrodes were deposited by pulsed laser deposition. 16 At room temperature, Au/PLN/LSCO thin film heterostructures exhibit ferroelectric behaviors in contrast to typical antiferroelectric features of PLN ceramic. 16 The lower degree of chemical ordering is considered to be the main reason.

Antiferroelectric crystals are promising candidates for dielectrics which are applied in high energy density electrical capacitors, especially for large size capacitor devices, due to the reversible electric field-induced antiferroelectric–ferroelectric phase transition.^{17–19} In addition, antiferroelectric crystals can be used for making some transparent block device which is applied to the optical system. However, until now most of study on antiferroelectric materials only focus on the form of ceramics or thin film due to the huge challenge of the growth of

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the antiferroelectric crystals. Even though much attention has been focused on modified PZ and PZT based single crystals, for example Pb(Zr,Ti)O3,4 (Pb,La)(Zr,Sn,Ti)O3 single crystals20-22 and so on. But the relevant reports on PLN-based AFE single crystal is still quite less until now and these intrigued us to grow the PLN AFE single crystals.²³ Actually, compared with the ceramics or thin film, single crystal is an ideal object to investigate on the antiferroelectric phenomenon. As for the PLN materials, the inherent antiferroelectric properties are unclear due to deficiency of single crystal although PLN ceramics received much attention. Therefore, the investigation on PLN single crystals can help to understand not only the physics and chemistry of the antiferroelectric phenomenon, but also the potential application of PLN single crystals. In this paper, the investigation on the superlattice structure, domain and antiferroelectric properties of PLN single crystals was reported.

2. Experimental section

2.1 Crystal growth

Due to the high melting point of Lu₂O₃, a high-temperature solution technique24 was chosen to grow PLN single crystals by means of adding flux to decrease the growing temperature as low as possible. The starting chemicals, PbO (99.9%), Lu₂O₃ (99.99%), Nb₂O₅ (99.9%) and H₃BO₃ (99.9%), were weighed according to the different ratios of flux to solute in order to realize the optimization of PLN crystal growth process. The proper flux/PLN molar ratio was 6.5: 1 with a molar ratio of 8:1 for PbO/H₃BO₃ after many trials. Mixtures were placed into Pt crucibles, then the crucibles were carried out in a muffle furnace equipped with an automatic temperature controller. The thermal program used for the spontaneous nucleation growth of PLN crystals were controlled by various thermal profiles, namely: (i) heating from room temperature to 1080 °C at a ramp rate of 100 °C h⁻¹ and dwelling for 24 h; (ii) slow cooling from 1080 °C to 1050 °C at 5 °C h⁻¹, from 1050 °C to 1030 °C at 0.1 °C h⁻¹, from 1030 °C to 1000 °C at 0.2 °C h⁻¹; (iii) cooling from 1000 °C down to room temperature at 100 °C h⁻¹. The residual flux was dissolved using 30% hot HNO₃ solutions, and the grown crystals were separated. Crystal seeds (of 2 \times 2 \times 5 mm³ in size) obtained from spontaneous nucleation growth were used for the top seeded solid solution growth (TSSG)

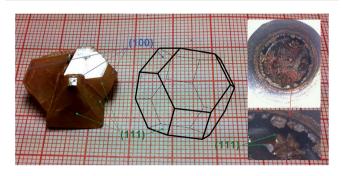


Fig. 1 As-grown PLN single crystals and sketch map for a tetrakaidecahedron combination.

method. Compared with other methods, the TSSG method also offers more advantages in growing single crystals of good quality and high compositional homogeneity due to its slow growth process. The detailed growth process is similar to the one described by C. He et al. 25 and the saturation temperature T_s of the solution were determined to be about 1040 °C by repeated seeding trials. At the stage of growth, the slow cooling rate was set at 0.2 °C h⁻¹. Finally, PLN single crystals were obtained as shown in Fig. 1.

2.2 Characterization

The phase and structure of the grown crystal were investigated by X-ray diffraction (XRD) analysis using Cu-Kα radiation (Ultima-IV, Rigaku, Japan) at room temperature. Powders X-ray diffraction was studied using ground and sieved powders sample. The scan step of 2θ is 0.02° , with an angular range of 10-80° and a rotating speed of 5° min⁻¹. Moreover, fine crystal powder for the electron microscopy study was polished mechanically, which was dispersed in alcohol to form a dilute suspension. One droplet of the suspension was dropped onto a copper grid with carbon support film. All the investigations were performed on a transmission electron microscope (TEM) (Tecnai F20 FEG, FEI, America). A polarized light microscope (PLM) (LV100POL, Nikon, Japan) equipped with hot plate (THMS 600 E, Linkam, U.K.) was employed to determine the symmetry, domain structure and its evolution as the temperature of the as-grown crystal increases. The optical crystal plates were cut from as grown crystal along the [001] and [111] directions, which were then grinded and polished to 50 µm in thickness. For electric characterization, different thicknesses of plates were sliced from the as-grown crystals along the [001] and [111] directions. The samples were then polished and coated with silver paste as electrodes. Measurements of the dielectric constant (ε') was carried out on samples by computer-controlled Alpha-broadband dielectric/impedance spectrometer (Novocontrol GmbH, Novocontrol, Germany) with an AC signal of 0.5 V (peak to peak) applied. Differential Scanning Calorimeter (DSC) was performed on differential thermal analysis apparatus (DTA 404 PC, NETZSCH, Germany), from room temperature to 1000 °C. Polarization hysteresis loops were displayed by standard ferroelectric test system (TF Analyzer 2000E, aix-ACCT, Germany) combined with a high-voltage supply amplifier/ controller (Model 610E, Trek, America) and an environmental test chamber (DELTA 9023, DELTA, America) controlled from room temperature to 240 °C.

Results and discussion 3.

3.1 Crystals

The photographs of the obtained PLN crystals are shown in Fig. 1. Compared with the size (2-6 mm) of the PLN crystals grown by spontaneous nucleation, the dimensions of the as-grown crystal obtained by TSSG method reached as large as $20 \times 20 \times 12$ mm³. The chemical bonding theory of single crystal growth^{26,27} indicates that the micro structure of lattice planes corresponds to the anisotropic distribution of constituent bonding atoms governing **RSC Advances**

crystal growth with particular rates. Compared with the configuration of PLN crystals, it was found that the (111) plane grew preferentially at the initial stage of crystal growth, and the (001) plane was generated gradually later as a result of the increase of the crystal size. External macro structure of the crystal was a tetrakaidecahedron combination with pseudocubic (111) and (001) planes in habit, and color of dark yellow as shown in Fig. 1.

3.2 Structure analysis

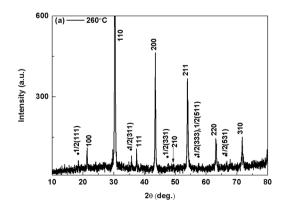
Powder X-ray diffraction patterns for PLN was measured at 260 °C and room temperature, showing perovskite structure as shown in Fig. 2. In addition, some superlattice reflections were found. Superlattice reflections might originate from three sources: long range ordering of B-site cations, antiparallel cation displacements and tilting of the BO₆ octahedra. Due to the high structural stability of perovskite structured compounds,28 the two atom species are allowed to occupy at the B sites. Pure PLN has a pronounced long-range order in the B sublattice that arises from long range order of Lu3+ and Nb3+ cations. The structure adopts a doubled perovskite unit cell due to the $A(B'_{1/2}B''_{1/2})O_3$ type '1:1' ordering with a face-centered arrangement of two different (B' and B") sublattices.29 Just above the transition temperature as shown in Fig. 2(a), the superlattice reflections as indicated by full circles only originate from the long-range ordering of Lu3+ and Nb3+ cations on the octahedral site of the perovskite lattice in addition to the fundamental reflections. These peaks corresponding to d-spacing values twice as much as those of (111), (311), (331), (333), and (531) major ones can be indexed as 1/2(111), 1/2(311), 1/2(331), 1/2(333), and 1/2(531), respectively, showing good order feature of a single crystal. The presence of superlattice reflection due to B-site atom ordering suggests that the prototypic crystal structure is an ordered perovskite with an effective unit cell $(2a_c \times 2a_c \times 2a_c)$ (c for cubic) doubled in size in all crystallographic axis directions of a cubic, simple perovskite cell $(a_c \times a_c \times a_c)$, in terms of crystallography, it means that B3+ and B5+ ions follow each other at

As evidenced from the split in the fundamental lines around $2\theta = 44^{\circ}$ of the low-temperature phase, shown in the insets of Fig. 2(b), the lattice distortion is found to be orthorhombic in nature.30 The unit cell of the room-temperature phase can be constructed from a primitive orthorhombic with lattice parameters $a_0 = 4.156 \text{ Å}$, $b_0 = 4.134 \text{ Å}$ and $c_0 = 4.090 \text{ Å}$, where the subscript o refers to the orthorhombic lattice. The roomtemperature XRD reveals complicated extra reflections as indicated by the full rhombus in Fig. 2(b), which appear only below the transition temperature. These extra reflects are most likely to originate from the antiparallel displacements of Pb²⁺ ions. Its crystal symmetry is similar to that of lead zirconate (PbZrO₃) where the crystal structure is characterized by the antiparallel Pb2+ displacement.31

The superlattice reflections were further identified by transmission electron microscopy. Fig. 3(a) shows the electron diffraction pattern obtained in the [1-11] zone-axes, exhibiting the fundamental spots of the prototype and the extra superlattice spots. The superlattice spots contain two sets of superlattice spots (B-site atom ordering superlattice reflections and Pb²⁺ antiparallel displacements superlattice reflections). The superlattice spots of Pb²⁺ antiparallel displacements are confirmed by the spots around $2\theta = 16^{\circ}$, 27° and others. The electron diffraction pattern [Fig. 3(b)] obtained in the [01-1]zone is composed of fundamental spots and the B-site atom ordering superlattice reflections. The superlattice spot is only the B-cation ordering observed in this zone in contrast to that shown in Fig. 3(a). It is also confirmed that the extra superlattice spots in Fig. 3(a) can be observed only along the orthorhombic a_0 direction. The agreement between the present XRD and highresolution TEM investigations reveals the existence of a longrange superlattice resulted from orthorhombic symmetry. Therefore, the two sets of superlattice can be distinguished from structural analysis.

3.3 Domain

Antiferroelectric micro-domains configurations of PLN crystal were examined by TEM. Although low accelerating voltages and low ion currents were used to prevent the loss of PbO, the domain configurations can be easily modified,32 and the domain wall are expanded by the electron beam heating and the



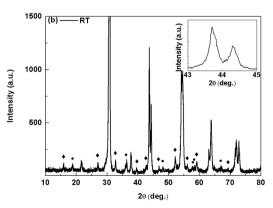


Fig. 2 X-ray powder diffraction patterns of the crashed PLN crystal sample at (a) 260 °C and (b) room temperature; (♦) superlattice reflections due to antiparallel displacements of cation; (\bullet) superlattice reflections due to B-site atom ordering. The split in the fundamental lines around 2θ = 44° inset in (b) clearly show the orthorhombic phase.

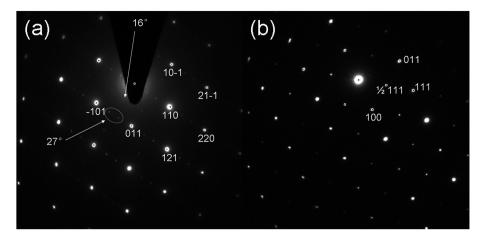


Fig. 3 Electron diffraction patterns of antiferroelectric phase obtained (a) in the [1–11] zone and (b) in the [01–1] zone.

stability of the domains are changed. The change happens so fast that there is no time to take pictures. TEM study of the disorder–order perovskite $Pb(In_{I/2}Nb_{I/2})O_3$ has shown that it is a relaxor ferroelectric in disordered PIN whereas an antiferroelectric behavior is observed in ordered PIN.³³ In ordered PLN, an antiferroelectric micro-domains with 1–2 nm in size was also found, as shown in Fig. 4. In general, the spontaneous polarization (P_s) in the paraelectric perovskite cubic structure below T_C appears in [110] directions in orthorhombic (O) perovskites. The vibration directions of all the domains in the (001) plate are

practically parallel or perpendicular to each other and the extinction angle θ is defined as the smaller angle between the domain's vibration direction and the [100] crystallographic axis is practically the same for all domains.³⁴ Viewed under crossed polarizers, the domains appear to be in extinction when their slow and fast vibration directions are along the vibration directions of polarizer (P) and analyzer (A), namely, $\theta = 45^{\circ}$ or $\theta = 0^{\circ}$ in the O phase.³⁵ Fig. 5 shows the micrograph of the [001]-oriented plate of the PLN crystal under crossed polarizer at room temperature. The extinction of domains appear when $\theta = 0^{\circ}$

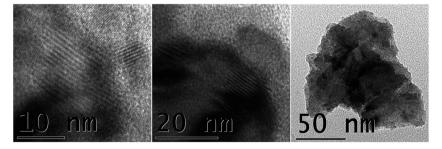


Fig. 4 Micro-domains configurations under TEM in the [1–11] zone.

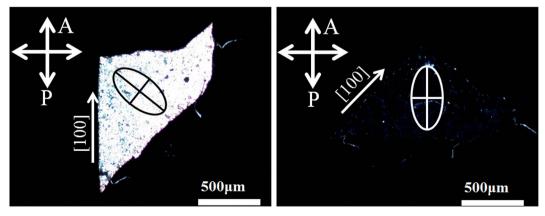


Fig. 5 Micrographs and its extinction angle of [001]-oriented plates of PLN crystal under crossed polarizer at room temperature.

45°, while the field manifests full bright when the objective table was rotated 45°, further confirming the orthorhombic symmetry of PLN crystal at room temperature. The domain structure and its extinction features in (001) PLN crystal plates as a function of temperature are shown in Fig. 6. The evolution of domain pattern was studied in the full bright field under crossed polarizers by heated from room temperature to 258 °C,

where few change until 240 $^{\circ}$ C was observed. However, by continuous heating, some parts of the domains appear to be in extinction and the field slowly turns into all dark state at 258 $^{\circ}$ C, which demonstrates the phase transition from orthorhombic to cubic phase between 240–258 $^{\circ}$ C.

For [111]-oriented platelet of O phase, two or more domains with nonparallel vibration directions appear overlapped. The

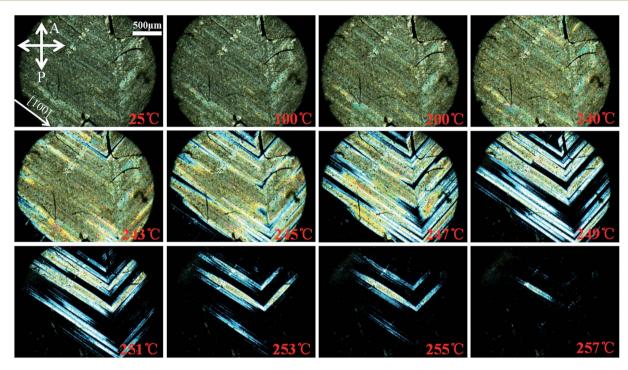


Fig. 6 Micrographs and its temperature dependence of [100]-oriented plates of PLN crystal under crossed polarizer.

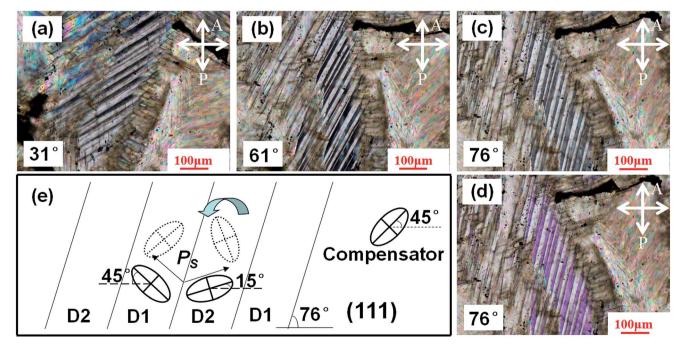


Fig. 7 Micrographs, domain size and domain wall angle of [111]-oriented plates PLN crystal under crossed polarizer at room temperature.

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macro-domains are consisting of numerous micro-domains. Therefore, not all macro-domains show extinction simultaneously. It is found that the presence of macro-domains are about 10-20 μm in size within two sets of domains, as shown in Fig. 7. One set of the domains (D1) appears to be in extinction when the objective table is in the angle of 31° (the angle defined as the angle between polarizer and the strip of domain) but the other set of the domains (D2) appears to be in extinction when the objective table is in the angle of 61°, as shown in Fig. 7(a) and (b), respectively. When the angle is set as 76°, D1 manifests full bright, as shown in Fig. 7(c). In order to confirm the orientation of the spontaneous polarization, a 530 nm optical path compensator plate was inserted, the color of D1 remains first order white-grey (200-300 nm) but the color of D2 turns from first order black-grey to first order pink (500-550 nm), as shown in Fig. 7(d). Based on crystal optical principle, 34 the $P_{\rm s}$ of the adjacent domains was confirmed as shown in Fig. 7(e). The domain walls form an angle of 120° within one of the (111) plate, which is 60° domain walls in space, namely.

3.4 Phase transition

Fig. 8 shows the temperature dependence of the dielectric constant (ε') at various frequencies for unpoled [111]-oriented and [001]-oriented PLN crystal plate. It shows a sharp peak at $T=240~^{\circ}\mathrm{C}$ and 254 $^{\circ}\mathrm{C}$ for [111]-oriented and [001]-oriented crystal plates respectively, corresponding to the $T_{\rm max}$, which is the temperature of the maximum dielectric constant. Generally, the $T_{\rm max}$ is related to the measurement frequency. So the value of $T_{\rm max}$ at 1k Hz is considered to be Curie temperature $T_{\rm C}$ habitually. No obvious frequency dispersion is found near T_{max} indicating the normal antiferroelectric behavior. The Curie temperature of PLN single crystals is lower than that of PLN ceramics which also undergoes a sharp antiferroelectric-paraelectric phase transition at about 258 °C without dielectric dispersion. 10,14 The difference temperature is due the component differences between single crystals and ceramics. Like most antiferroelectric materials, the value of ε' of PLN crystal (about 162, 90 for [111]-oriented and [001]-oriented crystal at room temperature respectively) is not

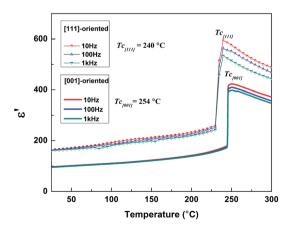


Fig. 8 Temperature dependence of the dielectric constant (ϵ') at various frequencies for unpoled [111]-oriented and [001]-oriented plates of PLN crystal.

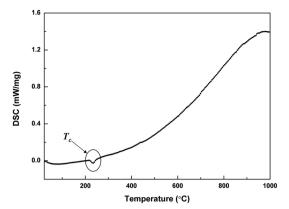


Fig. 9 Differential Scanning Calorimeter of the crashed PLN crystal

very high. The difference of dielectric properties can be explained by the anisotropic behavior. It is known that the possible polarization directions for orthorhombic perovskite symmetry is along the [110] direction. The polarization rotation mechanism induced by external electric fields along certain paths is proposed to explain the high dielectric constant for [111]-oriented samples due to the easy rotation and incline of domain and polar vectors.36 In addition, the orthorhombic to cubic transition temperature was confirmed by the DSC curves of the PLN crystal, exhibiting at 234 °C as shown in Fig. 9, in agreement with the data obtained from dielectric data.

3.5 Antiferroelectric properties

When high-occupancy order exists between the B-site cations, nanoscale chemical gradient appears with long-range normal antiferroelectric behavior. With an electric field $E_{APP} = 170 \text{ kV}$ cm⁻¹ applied at room temperature (f = 10 Hz), the polarizationelectric field (P-E) hysteresis loops of [111]-oriented PLN crystal platelets show that the P-E relationship is rather linear. It should be pointed out that it is difficult to obtained saturated double hysteresis loops at low temperature because the samples are easily broken down due to high AFE/FE transition electric field and low breakdown strength. In order to get the saturated double hysteresis loops, the AFE/FE hysteresis loops for PLN crystals were measured at high temperature which close to $T_{\rm C}$. Fig. 10 shows the polarization-electric field (P-E) hysteresis loops of [111]-oriented PLN crystal platelets with an electric field $E_{\rm APP} = 170 \; \rm kV \; cm^{-1}$ ($f = 10 \; \rm Hz$) applied at 200 °C. It can be seen that there is a double hysteresis loop with small remnant polarization ($\sim 1 \mu C \text{ cm}^{-2}$) and a high induced polarization $(P_{\rm max} \sim 43.2~\mu{\rm C~cm}^{-2})$ in the *P-E* curves. When electric field is increased larger than 136 kV cm⁻¹, the antiferroelectric phase was transformed into a ferroelectric phase, namely, the antiferroelectric phase to ferroelectric phase transition field E_{AFE-FE} $= 136 \text{ kV cm}^{-1}$. Whereas the electric field was reduced less than 75 kV cm⁻¹, the metastable ferroelectric phase returned to antiferroelectric phase and macro-polarization disappeared. Large electric energy is stored and released in the transition process. The smallest value for an electric field that can prevent

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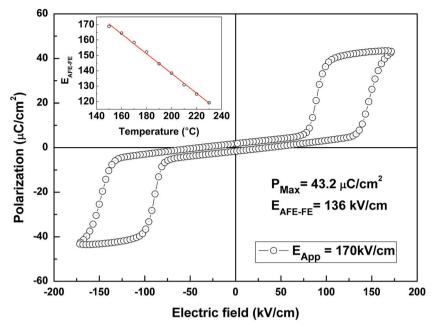


Fig. 10 Polarization vs. electric field (P-E) loops for the [111]-oriented plates PLN crystal at 200 °C (f=10 Hz). Temperature dependence of antiferroelectric parameter of E_{AFE-FE} for PLN crystal below T_C (inset), showing an ideal linear relationship between temperature and E_{AFE-FE} .

the transition from metastable ferroelectric phase into antiferroelectric phase is $E_{\text{FE-AFE}} = 75 \text{ kV cm}^{-1}$. Antiferroelectric to ferroelectric transition leads to high energy storage for the DC field, because of the large increase in polarization that accompanies the transition. Energy storage capabilities are determined experimentally by simple electrical circuit theory.³⁷ The value of energy density was calculated, which was approximately 3.65 J cm⁻³ at 200 °C with 170 kV cm⁻¹ electric field applied. From the dependence of energy density on field strength and temperature, it can be predicted that the attained saturation energy density is higher at room temperature. Such materials show promise for energy storage applications. Compared with the energy density of PLN ceramics which is 1.95 J cm⁻³ measured at 200 °C with f = 50 Hz, PLN crystals has obvious advantages.17 Meanwhile, the temperature dependence of antiferroelectric parameter of $E_{\rm AFE-FE}$ for PLN crystal below $T_{\rm C}$ is also displayed in the inset of Fig. 10. It is obvious that the $E_{\rm AFE-FE}$ shows a linear reduction as temperature increased from 140 °C to 230 °C, indicating coincidence antiferroelectric characterization: with the increasing temperature, the antiparallel dipole would be easily reversed under E_{APP} . Therefore, the $E_{\text{AFE-FE}}$ at room temperature can be extrapolated to be 250 kV cm⁻¹, which is much higher than the breakdown field. Hence a double hysteresis loop at room temperature cannot be obtained.

4. Conclusions

The antiferroelectric single crystals of PLN with a tetrakaidecahedron combination were successfully grown by high-temperature solution technique using PbO-H₃BO₃ flux. The crystal structure was found to be orthorhombic in nature, as revealed by the analysis of the XRD date. Both XRD and TEM show two sets of

superlattice reflections originating respectively from the B-site atom ordering and the lead-atom antiparallel ordering. High degree of order of B-sites occupancy was demonstrated by the superlattice reflections and the chemical order of the as grown PLN crystals. An antiferroelectric micro-domains with 1-2 nm in size has been found on (111) surfaces of PLN single crystal using TEM. The macro-domains consisting of numerous micro-domains have also been found by PLM. The temperature dependence of domains, extinction angle and domain wall angle has been studied under PLM. The normal dielectric behavior is somewhat different from the previous reports in PLN ceramics. The Curie temperature $T_{\rm C}$ was detected, to be near 240 °C. The polarization-electric field (P-E) hysteresis loops of PLN crystal platelets along [111]-oriented were displayed, with $E_{AFE-FE} = 136 \text{ kV cm}^{-1}$ and $E_{FE-AFE} = 75 \text{ kV cm}^{-1}$ under $E_{APP} = 170 \text{ kV cm}^{-1}$. The value of energy density was calculated, which was approximately 3.65 J cm $^{-3}$ at 200 °C. It is suggested that the PLN single crystals can constitute a new resource of high Curie temperature, high energy storage capabilities antiferroelectric-crystals potentially useful for high energy density electrical capacitors especially for large size capacitor devices that require a wider operation temperature range.

Acknowledgements

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