



Electronic structure, magnetism and optical properties of orthorhombic GdFeO_3 from first principles

Xu-Hui Zhu,^{ab} Xiang-Bo Xiao,^a Xiang-Rong Chen^b and Bang-Gui Liu^{*ac}

Orthorhombic GdFeO_3 has attracted considerable attention in recent years because its magnetic structure is similar to that of the well-known BiFeO_3 material. Here, we investigate the electronic structure, magnetism and optical properties of orthorhombic GdFeO_3 in terms of density-functional-theory calculations. The modified Becke-Johnson (mBJ) exchange potential is adopted to improve on the description of the electronic structure. Our calculations show that the G-type antiferromagnetic (G-AFM ordering of Fe spins) phase of orthorhombic GdFeO_3 is stable compared to other magnetic phases. The semiconductor gap calculated with mBJ, substantially larger than that with GGA, is in good agreement with recent experimental values. Additionally, we also investigate the effects of spin-orbit coupling on the electronic structure, and calculate the complex dielectric functions and other optical functions of photon energy. The magnetic exchange interactions are also investigated, which gives a Néel temperature close to experimental observation. For confirming these mBJ results, we also study the electronic structure of rhombohedral ($R\bar{3}c$) BiFeO_3 with mBJ, obtaining good consistency with experiment. These lead to a satisfactory theoretical understanding of the electronic structure, magnetism and optical properties of orthorhombic GdFeO_3 and can help elucidate the electronic structures and optical properties of other similar materials.

Received 14th October 2016
Accepted 2nd December 2016

DOI: 10.1039/c6ra25259a
www.rsc.org/advances

I. Introduction

Bismuth ferrite (BiFeO_3) is a representative of single-phase multiferroic materials, which displays antiferromagnetic order below $T_N \sim 643$ K and possesses a relatively high spontaneous electric polarization of $59.4 \mu\text{C cm}^{-2}$ up to $T_c \sim 1100$ K.¹⁻³ Besides, BiFeO_3 is a perovskite oxide whose most stable phase is a rhombohedral distorted structure with space group $R\bar{3}c$. As a magnetic material similar to BiFeO_3 , orthorhombic distorted GdFeO_3 , which has a Néel temperature 661 K and belongs to the perovskite rare-earth orthoferrites, has sparked substantial curiosity and stimulated relatively deep research.⁴⁻⁷

The GdFeO_3 compound has a complex $H-T$ phase diagram and undergoes a plurality of magnetic phase transitions, accompanying the dramatic changes in electrical properties.⁸ The orthorhombic distorted GdFeO_3 compound ($Pbnm$), with Gd^{3+} ions at the center and Fe^{3+} ions at the corners surrounded by oxygen octahedra, possesses weak ferromagnetism and ferroelectricity.^{4,5,7} Under electric and magnetic fields, the

ferroelectric polarization and magnetization of GdFeO_3 has been successfully brought under control for a variety of applications.⁹ A spontaneous polarization of about $0.12 \mu\text{C cm}^{-2}$ was obtained at 2 K,⁷ which is basically identical with the measured value in the perpendicular magnetic system.¹⁰ According to the Bertaut notation, the spin structure of Fe^{3+} is $\text{G}_x\text{A}_y\text{F}_z$.¹¹ Below $T_N^{\text{Gd}} = 2.5$ K, the magnetic order of Gd^{3+} is antiferromagnetic along the a -axis, showing G_x antiferromagnetic order and ferroelectric polarization characteristics.^{4,12} The interaction between adjacent Fe^{3+} and Gd^{3+} layers induces ferroelectric polarization along the c -axis.¹² It was reported that the Fe spins in BiFeO_3 form a G-type antiferromagnetic (G-AFM) ordering, with the spins on the Fe^{3+} ions being aligned antiferromagnetically along the [111] axis.¹³ Interestingly, GdFeO_3 can show a ferromagnetism below 5 K,^{4,14,15} and at $T_N^{\text{Fe}} = 661$ K, Fe^{3+} also forms G-AFM ordering in GdFeO_3 and shows a weak ferromagnetism due to the Dzyaloshinskii-Moriya interaction.¹⁶⁻¹⁸ On the other hand, first-principles calculations suggested that the antiferromagnetic phase of the orthorhombic GdFeO_3 is more stable than the ferromagnetic phase,¹⁹ and it was also pointed out that the transition from antiferromagnetic to paramagnetic ordering could occur at 670 K.^{6,7,9}

Despite a large number of experimental studies concerning the electromagnetic behavior of orthorhombic distorted GdFeO_3 , the theoretical reports are extremely meagre and it is necessary to adopt suitable theoretical approaches to perform

^aBeijing National Laboratory for Condensed Matter Physics, Institute of Physics, Chinese Academy of Sciences, Beijing 100190, China. E-mail: bgliu@iphy.ac.cn

^bInstitute of Atomic and Molecular Physics, College of Physical Science and Technology, Sichuan University, Chengdu 610065, China

^cSchool of Physical Sciences, University of Chinese Academy of Sciences, Beijing 100190, China



further study. Here, we investigate the electronic structure and magnetic and optical properties of orthorhombic GdFeO_3 through density functional theory (DFT) calculations. In order to better understand the electronic properties of GdFeO_3 , we also investigate the electronic structure of BiFeO_3 for comparison. The remainder of this paper is organized as follows: computational details are described in the second section, main calculated results and analysis are presented in the third section, and finally, conclusions are presented in the fourth section.

II. Computational details

The full-potential linearized augmented plane wave method within the density-functional theory (DFT),^{20,21} as implemented in the package Wien2k,²² is utilized in our calculations. First, the popular generalized gradient approximation (GGA-PBE)²³ is adopted to optimize crystal structures and investigate electronic structures and magnetism. Because the standard semi-local GGA usually underestimates energy band gaps,²⁴ we use the mBJ approximation²⁵ for the exchange potential, taking the local density approximation (LDA)²⁶ to treat the correlation potential, as usual for improved description of electronic structures and optical properties. For the calculations of electronic structures, mBJ has been shown to significantly improve and accurately produce semiconductor gaps for sp semiconductors, wide-band-gap semiconductors, and transition-metal oxide semiconductors and insulators.^{25,27-30} The electronic correlation has been effectively taken into account in this powerful mBJ approach.²⁵ Because the theoretical semiconductor gaps are improved, much better computational results can also be obtained for the optical properties. The full relativistic effects are calculated with Dirac equations for core states, and the scalar relativistic approximation is used for valence states.^{31,32} We also take the spin-orbit coupling (SOC) into consideration. The cut-off energy is set to -6 Ry to separate core states from valence states. The k -mesh size in the first Brillouin zone is $11 \times 10 \times 7$ for GdFeO_3 , and $10 \times 10 \times 10$ for BiFeO_3 . We construct harmonic expansion up to $l_{\text{max}} = 10$, set $R_{\text{mt}} \times K_{\text{max}} = 7$, and use magnitude of the largest vector $G_{\text{max}} = 12$ in charge density Fourier expansions. The radii of Gd, Bi, Fe and O atomic spheres are set to 2.27, 2.12, 1.99 and 1.77 bohr, respectively. The self-consistent calculations are considered to be converged only when the integration of absolute charge-density difference per formula unit between the successive loops is less than $0.0001|e|$, where e is the electron charge.

III. Results and discussion

A. Crystal structure

The orthorhombic GdFeO_3 system investigated here has space group $Pbnm$ (no. 62)^{4,19,33,34} at low temperatures. The experimental lattice constants are $a = 5.349$ Å, $b = 5.611$ Å and $c = 7.669$ Å.³³ The crystal structure of orthorhombic GdFeO_3 is shown in Fig. 1(a). At first, we optimized the lattice parameters and ionic positions with GGA and LDA and the optimized lattice parameters are summarized in Table 1. Existing experimental

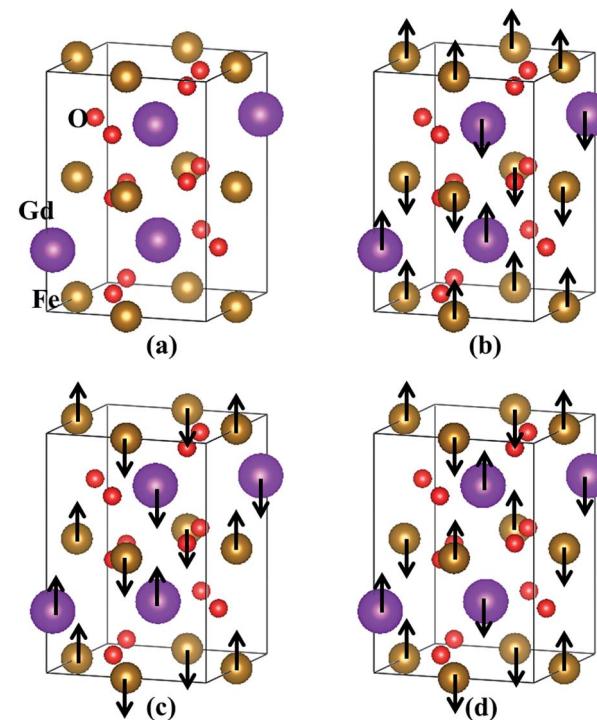


Fig. 1 The crystal structure with all the magnetic moments of Gd and Fe being in the same direction (a) and three antiferromagnetic ordering configurations (b–d) in orthorhombic GdFeO_3 : (b) A-AFM, (c) C-AFM and (d) G-AFM. The arrows indicate magnetic moment orientations on Gd and Fe atoms.

results^{33,34} are also presented for comparison. The GGA optimized volume V is 2.0% larger than the experimental volume,³³ but the LDA optimized volume V is 5.1% smaller. It can be clearly seen that the lattice constants and volume calculated with GGA are closer to the experimental data^{33,34} than those with LDA. Therefore, the other properties are investigated on the basis of the GGA optimized results. The GGA-optimized atomic positions are summarized in Table 2. After internal structure optimization, the Gd atom occupies the (0.9911, 0.0639, 0.25) site, the Fe atom the (0, 0.5, 0) site, the O1 atom the (0.7024, 0.3121, 0.0488) site, and the O2 atom the (0.0927, 0.4726, 0.25) site in Wyckoff coordinates. These are consistent with the experimental orthorhombic structure.^{33,34} Furthermore, the three optimized Gd–O bond lengths are 2.307, 2.347 and 2.399 Å, which are slightly larger than the Fe–O bond lengths of 1.968, 1.974 and 2.124 Å, respectively. These are in line with the relation of ionic radii, $\text{Gd}^{3+} > \text{Fe}^{3+}$, and in accordance with the

Table 1 The lattice parameters optimized with LDA and GGA and experimental data for GdFeO_3

	a (Å)	b (Å)	c (Å)	V (Å 3)	$\alpha = \beta = \gamma$ (°)
GGA	5.399	5.714	7.612	234.83	90.0
LDA	5.222	5.620	7.446	218.51	90.0
Exp. ³³	5.349	5.611	7.669	230.17	90.0
Exp. ³⁴	5.351	5.612	7.671	230.38	90.0



Table 2 The atomic positions (x , y , z) optimized with GGA for GdFeO_3 in comparison with experimental values

Atom	Site	x (exp. ^{33,34})	y (exp. ^{33,34})	z (exp. ^{33,34})
Gd	4c	0.9911 (0.9844, 0.9846)	0.0639 (0.0628, 0.0626)	0.2500 (0.2500, 0.2500)
Fe	4b	0.0000 (0.0000, 0.0000)	0.5000 (0.5000, 0.5000)	0.0000 (0.0000, 0.0000)
O1	8d	0.7024 (0.6957, 0.6966)	0.3121 (0.3016, 0.3011)	0.0488 (0.0506, 0.0518)
O2	4c	0.0927 (0.1005, 0.1009)	0.4726 (0.4672, 0.4669)	0.2500 (0.2500, 0.2500)

previously reported values.³⁴ The distances between Gd^{3+} and Fe^{3+} ions are 3.136, 3.284 and 3.362 Å. The bond angles of $\text{Fe}-\text{O}-\text{Gd}$ are 85.38° and 87.34°, deviating from the ideal values of 90°. These imply that the orthorhombic GdFeO_3 has undergone substantial structural distortion.

In addition, we have calculated the GGA total energies of four different magnetic ordering configurations: the ferromagnetic and three antiferromagnetic (AFM) ones. The three AFM structures are shown in Fig. 1(b-d) and are denoted by A-AFM, C-AFM and G-AFM, respectively. Taking the total energy of the lowest G-AFM structure as a reference, the other three energies are higher, which is consistent with the experimental results.⁴⁸ The ground-state magnetic structure is similar to the magnetic ordering of rhombohedral BiFeO_3 where the Fe spins form a G-AFM structure.³⁵ These results are also consistent with LSDA + U calculation.¹⁹

B. Electronic structures

With the optimized crystal structure, we calculate with both GGA and mBJ potentials the spin-dependent energy band structures and the densities of states (DOSs) of orthorhombic GdFeO_3 between -6 eV and 4 eV. The two band structures are shown in Fig. 2. The conduction band bottom and the valence band top are located at the same S point in both of the band structures and indicate a direct gap for orthorhombic GdFeO_3 . It can be seen that the GGA semiconductor gap is 0.61 eV [Fig. 2(a)], a little larger than the earlier first-principles result of 0.54 eV,³⁶ but the mBJ-calculated semiconductor gap, 2.49 eV [Fig. 2(b)], is apparently larger than the GGA value and is in

accordance with the experimental results (2.1–2.53 eV).^{7,9,10,37,38} Very interestingly, there are many similar features between GdFeO_3 and BiFeO_3 as orthoferrite ABO_3 materials. As a typical multiferroic orthoferrite compound, however, BiFeO_3 shows the characteristics of an indirect band gap. For BiFeO_3 , our GGA band gap of 0.965 eV is slightly lower than previous DFT values of 1.06 eV³⁸ and 1.04 eV,³⁹ but they are all too small to accurately describe the experimental values of 2.4 eV,⁴⁰ 2.5 eV,⁴¹ and 2.74 eV.⁴² Our mBJ calculation produces a semiconductor gap of 2.354 eV for BiFeO_3 and it is in good agreement with the experimental values. In contrast, a band gap of 2.8 eV, obtained with a screened exchange potential,³⁵ is too large in comparison with the experimental values. These results show that our mBJ gap of 2.49 eV is reasonable and should be accurate for orthorhombic GdFeO_3 .

Fig. 3 shows the spin-resolved densities of states (DOSs) of orthorhombic GdFeO_3 calculated with both GGA and mBJ. Through comparing GGA DOS [see Fig. 3(a)] and mBJ DOS [see Fig. 3(b)], we can see that the wide valence bands between -6 eV and 0 eV are originated from O 2p and Fe 3p states with a mixture of some Gd 5p6s, and the conduction bands are mainly from Fe 3d and Gd 4f states. Our analysis shows that the filled O 2p states are located between -6 and 0 eV. The filled Gd 4f states are between -3.4 and -1.2 eV, and the empty ones are between 2.8 and 3.4 eV. The filled Fe e_g states are between -0.85 and 0.0 eV, and the empty ones between 2.49 and 2.93 eV. The empty Fe t_{2g} states are between 3.37 and 3.88 eV. In order to understand the electronic properties of orthorhombic GdFeO_3 , the total and partial density of states of BiFeO_3 are also investigated with mBJ (not presented here). The top of the valence bands consist mainly of O 2p states and some Fe 3d and Bi 6p states, and the bottom of the conduction bands are originated from Fe 3d states and O 2p states. Our DOS calculated with mBJ is significantly more accurate than the previous theoretical work with GGA,^{38,39} and is close to the sX potential result.³⁵ It is interesting that the two materials share main features in the density of states.

C. Effects of spin-orbits coupling

The spin-orbit coupling (SOC) is important for electronic materials containing heavy atoms such as Gd and can lead to magnetocrystalline anisotropy. With the GGA + SOC method, we can calculate the total energy of orthorhombic GdFeO_3 by taking the SOC into account. Setting the magnetization in the [100], [010], [001], [110], [101], [011] and [111] directions, we obtain total energies of 2.6, 3.8, 0, 3.2, 8.7, 1.3 and 1.6 μeV , respectively. It is obvious that the lowest energy is along the

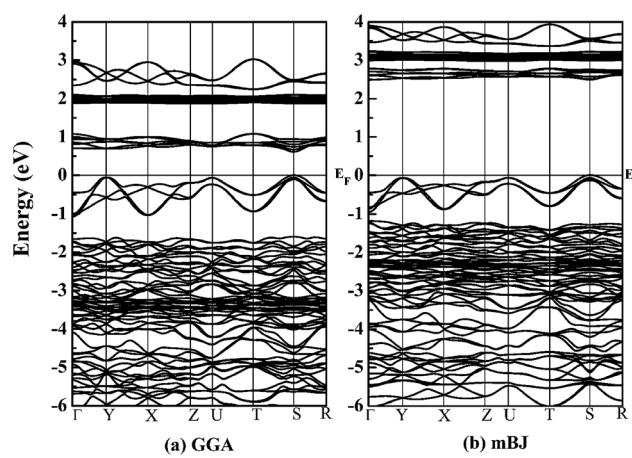


Fig. 2 The spin-resolved energy bands of orthorhombic GdFeO_3 with (a) GGA and (b) mBJ.



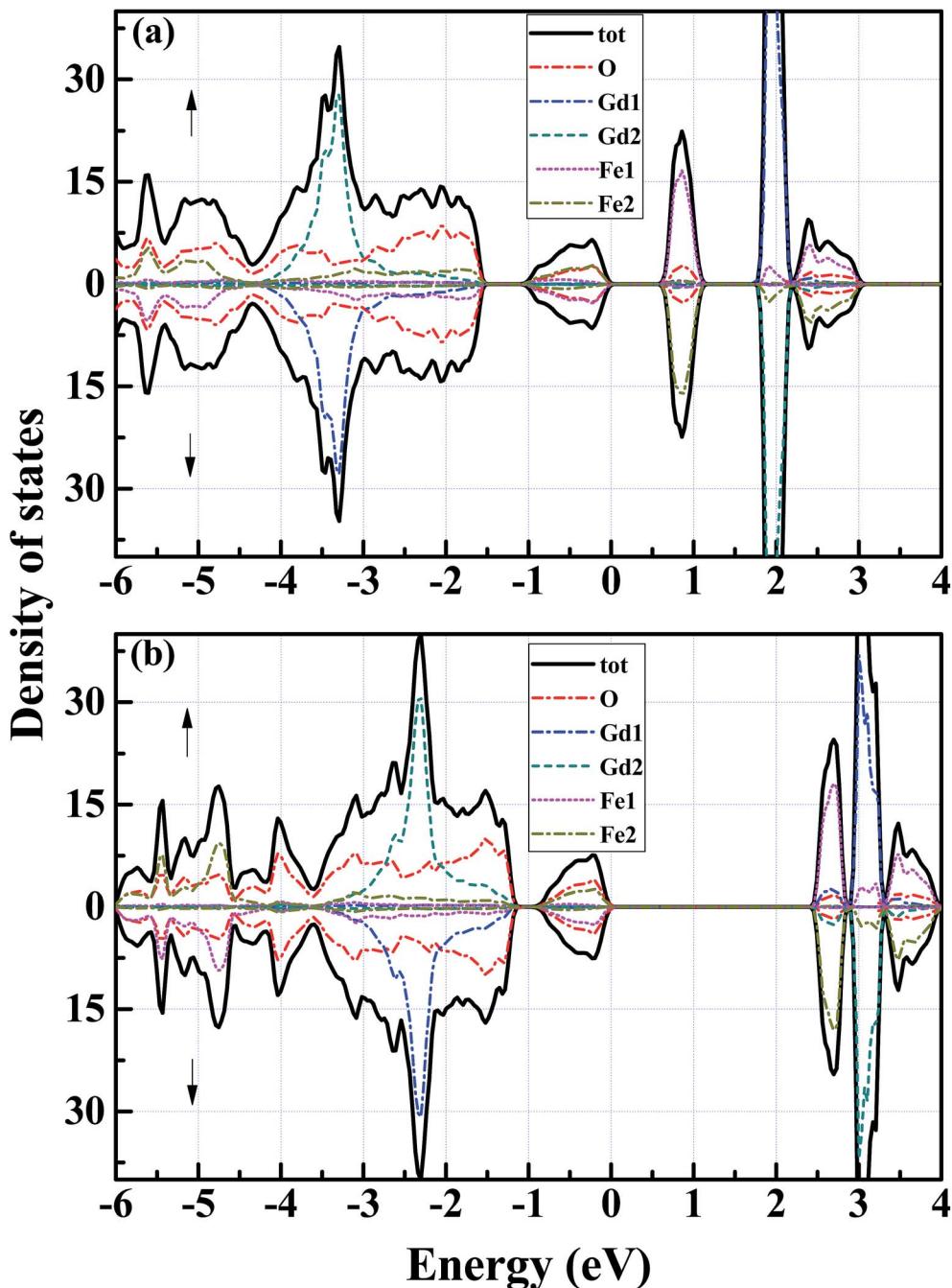


Fig. 3 Spin-resolved total density of states of orthorhombic GdFeO_3 with (a) GGA and (b) mBJ. The upper part is for the majority-spin channel and the lower for the minority-spin one.

[001] direction. These results indicate that the easy magnetization axis (the most stable magnetic orientation) of orthorhombic GdFeO_3 is along the [001] axis.

In the easy axis, the total spin moment is precisely equivalent to $0 \mu_B$ per formula unit without SOC for orthorhombic GdFeO_3 . According to Hund's rule, the cations Gd^{3+} and Fe^{3+} possess high spin values of $s = 7/2$ and $s = 5/2$, respectively, and the antiferromagnetic coupling makes the total spin moment equal to $0 \mu_B$ per formula unit. Since part of the spin moments are located in the interstitial region, the spin moments of the

individual Gd^{3+} and Fe^{3+} are $6.855 \mu_B$ and $4.082 \mu_B$, smaller than the theoretical $7 \mu_B$ and $5 \mu_B$, respectively. When taking the SOC into account, the spin moments of Gd^{3+} and Fe^{3+} are reduced to $6.811 \mu_B$ and $4.080 \mu_B$, respectively. The orbital moment of Fe 3d is $0.183 \mu_B$, which has the same sign as the spin moment, and the orbital moment of Gd^{3+} ion is $0.088 \mu_B$.

The semiconductor gap E_g is also investigated by using mBJ. The semiconductor gap becomes smaller (2.40 eV) when the SOC is taken into account. Fig. 4(a) shows the density of states of orthorhombic GdFeO_3 obtained with mBJ + SOC. Looking

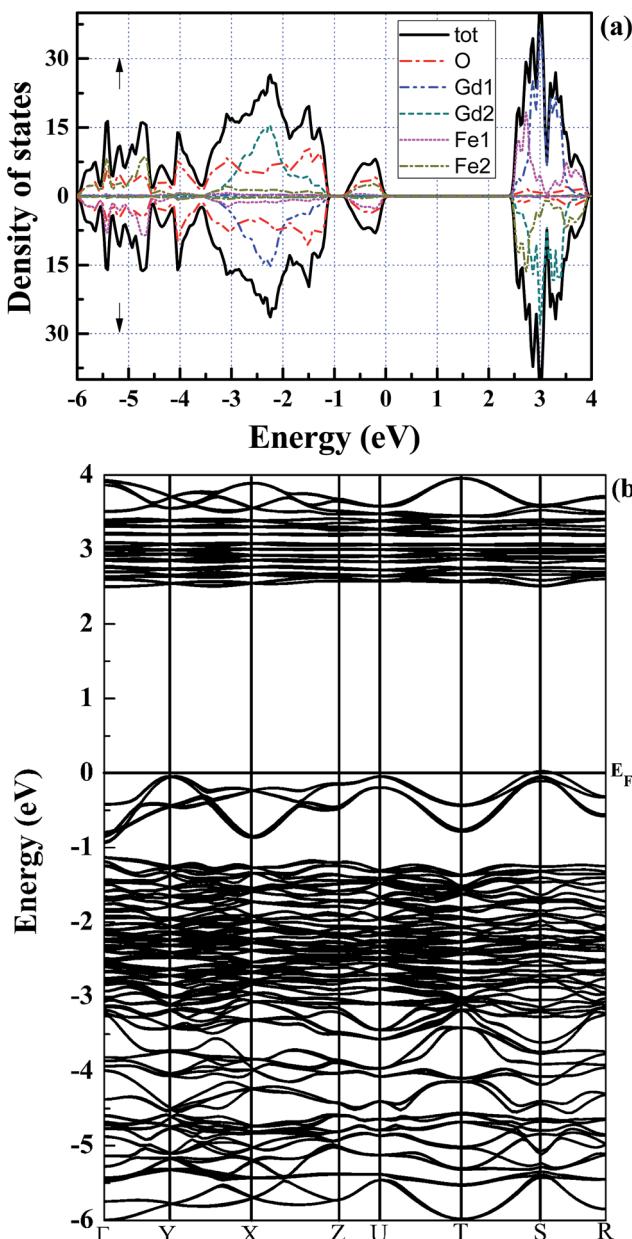


Fig. 4 The spin-resolved (a) total density of states and (b) energy bands of orthorhombic GdFeO_3 with mBJ + SOC.

closely at Fig. 4(a), the semiconductor gap is slightly smaller than that without SOC. This should be because the Fe 3d and Gd 4f bands become wider due to SOC. Fig. 4(b) explicitly demonstrates the energy bands with mBJ + SOC. The band structures and density of states show that the energy bands, especially the conduction bands, in both of the spin channels hybridize with each other.

D. Exchange interactions

The magnetic exchange interactions on Gd^{3+} and Fe^{3+} can be investigated in terms of total energy calculations. We consider four different magnetic configurations, namely antiferromagnetic (ground state) and ferromagnetic ordering, and two other

magnetic orderings constructed by changing the Fe and Gd spins of the antiferromagnetic configuration to the ferromagnetic ordering, respectively. With the first principles total energies of the different magnetic configurations, we can determine the coupling constants of the effective Heisenberg model (eqn. (1)): ⁴³

$$H = \sum_{ij} J_{ij} S_i \times S_j, \quad (1)$$

where S_i is the spin operator at site i . Here, the summation is over spin pairs, and the spin exchange parameter J_{ij} is limited to the nearest (Fe-Gd) and the next nearest (Fe-Fe and Gd-Gd) spin pairs. Although the absolute values of the magnetic moments in the spheres of Gd and Fe are $6.855 \mu_B$ and $4.082 \mu_B$, the Gd^{3+} and Fe^{3+} cations should theoretically contribute $7 \mu_B$ and $5 \mu_B$, respectively. We can assign spin values $s = 7/2$ and $s = 5/2$ to the Gd and Fe spins, respectively. Accordingly, there exists a relation between the magnetic energies e_{ij} and exchange parameter J_{ij} , $e_{ij} = J_{ij} s_i s_j$, where s_i takes values of either $7/2$ or $5/2$. Taking the G-AFM ground state as a reference, the calculated total energies of the other three states are 159.9, 158.6 and 3.5 meV per formula unit, respectively. The total energy can be split into $E_0 + \sum e_{ij}$, where E_0 is defined to be independent of spin configuration. We obtain the following equations from the four magnetic structures (eqn (2)).

$$\begin{cases} 0 = E_0 - 3e_{\text{FeFe}} - 3e_{\text{GdGd}} \\ 159.9 = E_0 + 8e_{\text{FeGd}} + 3e_{\text{FeFe}} + 3e_{\text{GdGd}} \\ 158.6 = E_0 + 3e_{\text{FeFe}} - 3e_{\text{GdGd}} \\ 3.5 = E_0 - 3e_{\text{FeFe}} + 3e_{\text{GdGd}} \end{cases} \quad (2)$$

From the above equations, we can calculate the e_{ij} parameters, and then obtain exchange coupling parameters J_{ij} . The Gd-Gd and Fe-Gd spin exchange energies are much smaller than the Fe-Fe value (26.43 meV). As a result, the spin exchange parameters J_{ij} are 0.03 meV between the nearest Fe-Gd pair, 4.23 meV between Fe-Fe, and 0.05 meV between Gd-Gd. It is clear that the Fe-Fe spin coupling is dominant over the other two.

With the exchange constants J_{ij} given, we can calculate the magnetic phase transition temperature, T_N . For this purpose, the quantum Green-function method is a good approach because it can produce accurate transition temperatures and reliable temperature dependence of magnetization or sublattice magnetization for a Heisenberg spin model. ⁴⁴⁻⁴⁷ For our spin model, we aim to use this method to estimate the Néel temperature. If neglecting the much smaller Gd-Gd and Gd-Fe interaction energies, we can estimate the Néel temperature, $T_N = 605$ K, in terms of an analytical approach. ⁴⁷ Considering that we have not taken the SOC effect into account, this Néel temperature is very satisfactory compared to experimental values: 670 K⁷ and 661 K. ^{16,17}

E. Optical properties

Optical spectroscopy analysis is a powerful tool to determine the energy band structure of a solid material. ^{48,49} The complex dielectric function is directly related to the energy band



structure of solids. For orthorhombic GdFeO_3 , Fig. 5 shows the mBJ calculated curves of the complex dielectric function (the real and imaginary parts), absorption coefficient, reflectivity, energy loss function, refractive index, extinction coefficient, and optical conductivity as functions of the photon energy in the

range of 0–9 eV. All the three polarization directions ($E//x, y, z$) are considered.

The electronic polarizability of a material can be understood from the real part of the dielectric function $\varepsilon_1(\omega)$ [Fig. 5(a)]. The static dielectric constant $\varepsilon_1(0)$ along the three crystallographic directions is found to be 4.28 for $E//x$, 4.38 for $E//y$, and 4.20 for

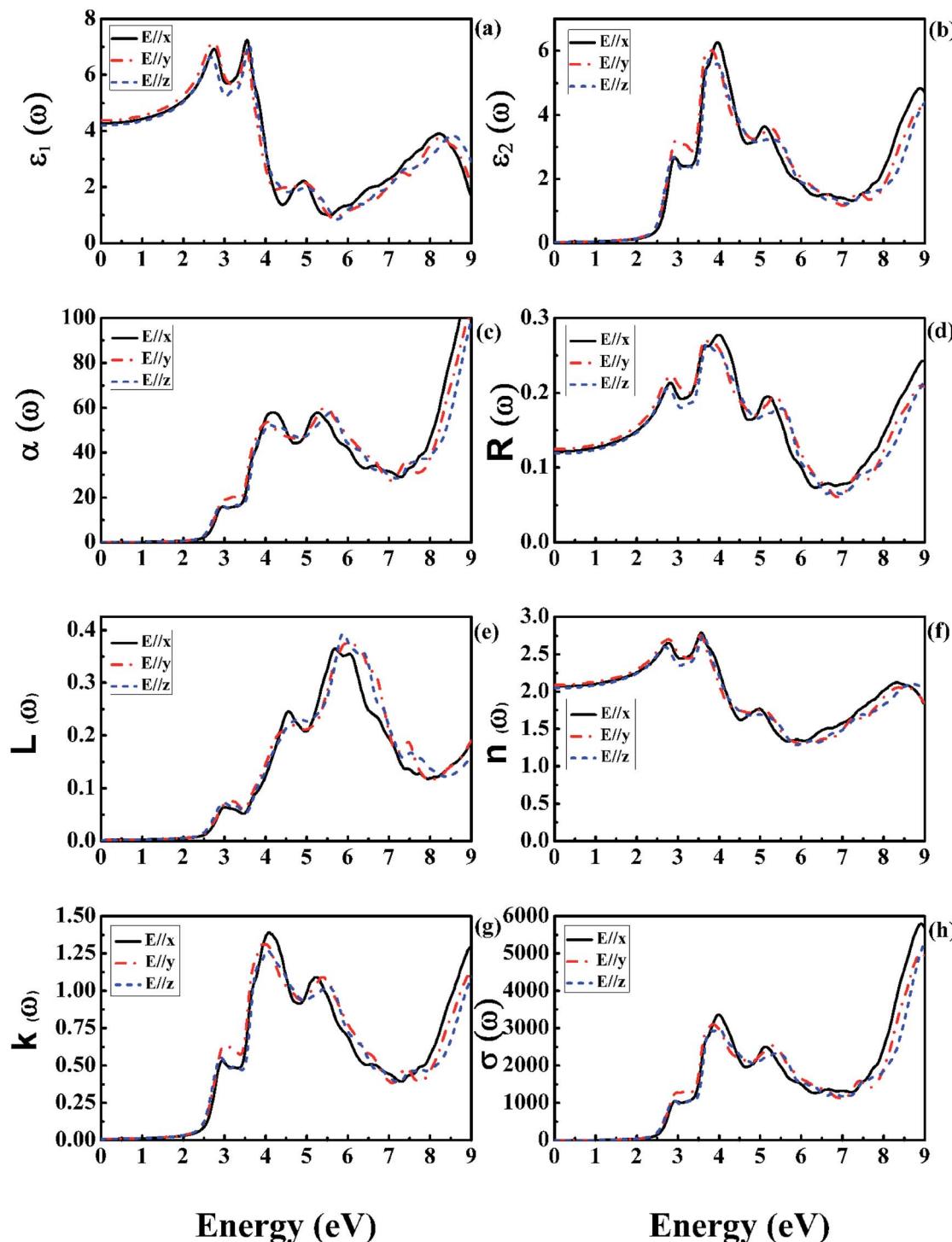


Fig. 5 Optical spectra as a function of photon energy for orthorhombic GdFeO_3 calculated with mBJ: (a) real $\varepsilon_1(\omega)$, (b) imaginary $\varepsilon_2(\omega)$ parts of dielectric function, (c) absorption coefficient $\alpha(\omega)$, (d) reflectivity $R(\omega)$, (e) energy loss function $L(\omega)$, (f) refractive index $n(\omega)$, (g) extinction coefficient $k(\omega)$, and (h) optical conductivity $\sigma(\omega)$.



$E//z$, respectively. The average value of the zero frequency dielectric constant $\varepsilon_1(0)$ is 4.29. However, there is no experimental polarized zero frequency dielectric constant available for comparison. These results clearly indicate anisotropy in the optical properties of orthorhombic GdFeO_3 . The ratio $\varepsilon_1^{yy}(0)/\varepsilon_1^{zz}(0)$ is equal to 1.043 for estimating the degree of anisotropy. From the zero frequency limit, they start increasing and reach the maximum value of 6.92 at 2.76 eV for $E//x$, 7.16 at 2.73 eV for $E//y$, and 6.63 at 2.65 eV for $E//z$, respectively. The imaginary part $\varepsilon_2(\omega)$ [Fig. 5(b)] gives information of the absorption behavior of GdFeO_3 . The threshold energy of the dielectric function is at $E_0 = 2.49$ eV, in good accordance with the fundamental gap. The obtained optical gap once again proves that mBJ can provide an accurate band gap for magnetic semiconductors. The imaginary part $\varepsilon_2(\omega)$ [Fig. 5(b)] indicates that GdFeO_3 is anisotropic and its maximum absorption peak values are around 3.96, 3.82 and 3.77 eV for $E//x$, $E//y$ and $E//z$, respectively. From Fig. 4(a), for the imaginary part $\varepsilon_2(\omega)$, it is clear that there are strong absorption peaks in the energy range of 2.5–9 eV. Because $\varepsilon_2(\omega)$ is related to the DOS, these peaks reflect some transitions between different orbitals. Compared with Fig. 3, it can be recognized that the peaks around 3.5–4.5 eV are mainly due to transitions from Gd-4f valence bands to O-2p conduction bands.

Fig. 5(c–h) show the calculated results of the photon energy dependence of absorption coefficient $\alpha(\omega)$, reflectivity coefficient $R(\omega)$, energy loss function $L(\omega)$, refractive index $n(\omega)$, extinction coefficient $k(\omega)$ and optical conductivity $\sigma(\omega)$ of orthorhombic GdFeO_3 . The absorption coefficient $\alpha(\omega)$ [Fig. 5(c)] shows a very intense absorption up to 9 eV; it begins to increase sharply from 2.49 eV, corresponding to the band gap value. The reflectivity coefficient $R(\omega)$ is shown in Fig. 5(d), the zero-frequency reflectivity values are 12.1% for $E//x$, 12.5% for $E//y$, and 11.9% for $E//z$, respectively. The maximum reflectivity values are about 19.5, 19.5 and 17.9%, which occurs at 5.18 eV for $E//x$, 5.35 eV for $E//y$, and 5.51 eV for $E//z$, respectively. Interestingly, the strong reflectivity maximum between 2.49 and 9 eV originates from the interband transitions. The energy loss function $L(\omega)$ [Fig. 5(e)] is related to the energy loss of a fast electron in the material and is usually large at the plasmon energy.⁵⁰ The most prominent peak in the $L(\omega)$ spectra represents the characteristic associated with the plasmon resonance and is situated at 5.86 eV for $E//z$ polarization. The refractive index values $n(\omega)$ are shown in Fig. 5(f). The static refractive index $n(0)$ is found to have the value 2.07 for $E//x$, 2.09 for $E//y$, and 2.05 for $E//z$, respectively. The average value of $n(0)$ is equal to 2.07. The value of the static refractive index is obtained from the real part of dielectric function to be $n(0) = \varepsilon_1(0)^{1/2} = \sqrt{4.29} = 2.07$, which is the same as that obtained from Fig. 5(f). A similar trend is observed from the behaviour of the imaginary part of the dielectric function $\varepsilon_2(\omega)$ [Fig. 5(b)] and the extinction coefficient $k(\omega)$ [Fig. 5(g)]. The extinction coefficient $k(\omega)$ reflects the maximum absorption in the medium at 4.01 eV for $E//x$, 3.90 eV for $E//y$, and 4.04 eV for $E//z$, respectively. The optical conductivity $\sigma(\omega)$ is shown in Fig. 5(h). It starts from 2.49 eV and has similar features with the absorption coefficient $\alpha(\omega)$ in Fig. 5(c).

IV. Conclusion

We have used the FP-LAPW method to investigate the structural, electronic, magnetic and optical properties of orthorhombic GdFeO_3 . The GGA approach has confirmed that the G-type AFM ordering of Fe spins is the ground-state phase, consistent with the experimental results. The mBJ exchange potential is used for improving on the description of the electronic structures of GdFeO_3 . Our calculated results show that mBJ exchange greatly improves the accuracy of the band gap value. The mBJ result accords well with the experimental value and overcomes the GGA underestimation of the band gap. Besides, the spin-orbit coupling is taken into account to determine the easy magnetic axis and investigate its effects on the electronic structure. We also calculate magnetic exchange constants and thereby achieve a good Néel temperature close to the experimental value. Finally, the optical properties also are investigated with mBJ. In addition, we also calculated the electronic structure of the well-known BiFeO_3 to support our mBJ-calculated results in the case of GdFeO_3 . The magnetic similarity between these two perovskite oxide materials are very interesting. These calculated results should be useful to obtain further insight for GdFeO_3 and similar materials.

Acknowledgements

This work is supported by the Nature Science Foundation of China (No. 11574366), by the Strategic Priority Research Program of the Chinese Academy of Sciences (Grant No. XDB07000000), and by the Department of Science and Technology of China (Grant No. 2016YFA0300701).

References

- 1 R. S. Fishman, J. T. Haraldsen, N. Furukawa and S. Miyahara, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **87**, 134416.
- 2 J. Buhot, C. Toulouse, Y. Gallais, A. Sacuto, R. de Sousa, D. Wang, L. Bellaiche, M. Bibes, A. Barthlmy, A. Forget, D. Colson, M. Cazayous and M.-A. Measson, *Phys. Rev. Lett.*, 2015, **115**, 267204.
- 3 D. Sando, Y. Yang, E. Bousquet, C. Carrero, V. Garcia, S. Fusil, D. Dolfi, A. Barthlmy, P. Ghosez, L. Bellaiche and M. Bibes, *Nat. Commun.*, 2016, **7**, 10718.
- 4 Y. Tokunaga, N. Rurukawa, H. Sakai, Y. Taguchi, T. Arima and Y. Tokura, *Nat. Mater.*, 2009, **8**, 558.
- 5 M. W. Lufaso and P. M. Woodward, *Acta Crystallogr., Sect. B: Struct. Sci.*, 2001, **57**, 725.
- 6 A. Wu, H. Shen, J. Xu, Z. Wang, L. Jiang, L. Luo, S. Yuan, S. Cao and H. Zhang, *Bull. Mater. Sci.*, 2012, **35**, 259.
- 7 P. R. Babu, I. Bhaumik, S. Ganesamoorthy, S. Kalainathan, R. Bhatt, A. K. Karnal and P. K. Gupta, *J. Alloys Compd.*, 2015, **631**, 232.
- 8 Z. Y. Zhao, X. M. Wang, C. Fan, W. Tao, X. G. Liu, W. P. Ke, F. B. Zhang, X. Zhao and X. F. Sun, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 014414.
- 9 A. Wu, Z. L. Wang, B. Wang, X. L. Ban, L. W. Jiang, J. Xu, S. J. Yuan and S. X. Cao, *Solid State Commun.*, 2014, **185**, 14.



10 Y. Tokunaga, S. Iguchi, T. Arima and Y. Tokura, *Phys. Rev. Lett.*, 2008, **101**, 097205.

11 G. T. Rado and H. Suhl, *Magnetism: A treatise on modern theory and materials*, Academic Press, New York, 1963, vol. III.

12 E. F. Bertaut, *Magnetism*, Academic Press, New York, 1963.

13 L. Bi, A. R. Taussig, H.-S. Kim, L. Wang, G. F. Dionne, D. Bono, K. Persson, G. Ceder and C. A. Ross, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **78**, 104106.

14 J. Shah and R. Kotnala, *Scr. Mater.*, 2012, **67**, 316.

15 K. C. Nowack, F. H. L. Koppens, Y. V. Nazarov and L. M. K. Vandersypen, *Science*, 2007, **318**, 1430.

16 D. Treves, *J. Appl. Phys.*, 1965, **36**, 1033.

17 R. M. Bozorth, H. J. Williams and D. E. Walsh, *Phys. Rev.*, 1956, **103**, 572.

18 I. Dzyaloshinskii, *J. Phys. Chem. Solids*, 1958, **4**, 241; T. Moriya, *Phys. Rev.*, 1960, **120**, 91.

19 N. Singh and J. Y. Rhee, *J. Korean Phys. Soc.*, 2008, **53**, 806.

20 P. Hohenberg and W. Kohn, *Phys. Rev.*, 1964, **136**, B864.

21 W. Kohn and L. J. Sham, *Phys. Rev.*, 1965, **140**, A1133.

22 P. Blaha, K. Schwarz, G. K. H. Madsen, D. Kvasnicka and J. Luitz, *WIEN2k, An Augmented Plane Wave + Local Orbitals Program for Calculating Crystal Properties*, ed. K. Schwarz, Techn. Universitat Wien, Austria, 2001, ISBN 3-9501031-1-2.

23 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.

24 H. Fan, A. S. Barnard and M. Zacharias, *Appl. Phys. Lett.*, 2007, **90**, 143116.

25 F. Tran and P. Blaha, *Phys. Rev. Lett.*, 2009, **102**, 226401.

26 J. P. Perdew and Y. Wang, *Phys. Rev. B: Condens. Matter*, 1992, **45**, 13244.

27 D. J. Singh, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **82**, 205102.

28 D. Koller, F. Tran and P. Blaha, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 155109.

29 S. W. Fan, L. J. Ding, Z. L. Wang and K. L. Yao, *Appl. Phys. Lett.*, 2013, **102**, 022404.

30 A. Ghosh, R. Thangavel and M. Rajagopalan, *J. Mater. Sci.*, 2015, **50**, 1710–1717.

31 A. H. MacDonald, W. E. Pickett and D. D. Koelling, *J. Phys. C: Solid State Phys.*, 1980, **13**, 2675.

32 J. Kunes, P. Novak, R. Schmid, P. Blaha and K. Schwarz, *Phys. Rev. B: Condens. Matter*, 2001, **64**, 153102.

33 M. Marezio, J. P. Remeika and P. D. Dernier, The crystal chemistry of the rare earth orthoferrites, *Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem.*, 1970, **26**, 2008–2022.

34 N. L. Ross, J. Zhao, J. B. Burt and T. D. Chaplin, *J. Phys.: Condens. Matter*, 2004, **16**, 5721–5730.

35 S. J. Clark and J. Robertson, *Appl. Phys. Lett.*, 2007, **90**, 132903.

36 B. L. Xing, L. Wu, G. P. Qin, Y. Li, Y. F. Zhang and J. Q. Li, *Acta Chim. Sinica*, 2007, **65**, 17.

37 P. Tang, Y. Y. Hu, T. T. Lin, Z. Jiang and C. W. Tang, *Integr. Ferroelectr.*, 2014, **153**, 1.

38 H. Wang, Y. Zheng, M. Q. Cai, H. Huang and H. L. W. Chan, *Solid State Commun.*, 2009, **149**, 641–644.

39 Q. Xu, M. Sobhan, F. Anariba, J. W. C. Ho, Z. Chen and P. Wu, *Phys. Chem. Chem. Phys.*, 2014, **16**, 23089.

40 T. P. Gujar, V. R. Shinde and C. D. Lokhande, *Mater. Chem. Phys.*, 2007, **103**, 142.

41 F. Gao, Y. Yuan, X. Y. Chen, F. Chen, J. M. Liu and F. Z. Ren, *Appl. Phys. Lett.*, 2006, **89**, 102506.

42 J. F. Ihlefeld, N. J. Podraza, Z. K. Liu, R. C. Rai, X. Xu, T. Heeg, Y. B. Chen, J. Li, R. W. Collins, J. L. Musfeldt, X. Q. Pan, J. Schubert, R. Ramesh and D. G. Schlom, *Appl. Phys. Lett.*, 2008, **92**, 142908.

43 S. Gong, P. Chen and B. G. Liu, *J. Magn. Magn. Mater.*, 2014, **349**, 74–79.

44 S. V. Tyablikov, *Ukrainian Mathematical Journal*, 1959, **11**, 287.

45 R. A. Tahir-Kheli, *Phys. Rev.*, 1963, **132**, 689.

46 H. B. Callen, *Phys. Rev.*, 1963, **130**, 890.

47 R. H. Swendsen, *Phys. Rev. B: Solid State*, 1975, **11**, 1935.

48 B. Amin, R. Khenata, A. Bouhemadou, I. Ahmad and M. Maqbool, *Phys. B*, 2012, **407**, 2588–2592.

49 M. Maqbool, B. Amin and I. Ahmad, *J. Opt. Soc. Am. B*, 2009, **26**, 2181–2184.

50 P. Nozieres, *Phys. Rev. Lett.*, 1959, **8**, 1.

