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## Antiproliferative activities of alkaloid-like compounds†

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Tricyclic alkaloid-like compounds were synthesised in a few steps, *via* the bridging Ritter reaction. The compounds were evaluated for their antiproliferative activity against the MCF-7 and the aggressive MDA-MB-231 breast cancer cells. The anti-cancer activities of **2c** were found to be selective towards the aggressive and more challenging to treat triple negative (MDA-MB-231) cell line while exhibiting no antiproliferative activities towards the MCF-7 cells at the highest concentration tested (50  $\mu$ M). The  $IC_{50}$  of compound **2c** was determined to be 7.9  $\mu$ M for the MDA-MB-231 cell line. Furthermore, **2c** arrested cell cycle at the G<sub>2</sub>/M phase and induced apoptosis in a dose-dependent manner. Besides in-house anti-cancer screening, compound **3** was selected for anti-cancer screening by the National Cancer Institute and was found to have broad anti-cancer activity with selectivity against particular leukaemia, colon, melanoma, and breast cancer cell lines. Cytotoxicities of compounds **2c** and **3** were also tested against noncancerous mammalian cells (VERO cell line), and found to be selective towards cancerous cells. The facile synthetic route, unique chemical structures and the biological data make these alkaloid-like compounds worthwhile lead compounds for further anti-cancer drug development.

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## Introduction

Alkaloids have played a significant part in drug development. However, research in this area has since started to decline, and the early 1980s saw a move towards derivatisation of natural alkaloids, as well as the synthesis of non-natural alkaloid-like compounds using well known natural product scaffolds.<sup>1–4</sup>

Alkaloids and their derivatives are to date the most successful anti-cancer drugs on the market. Despite tremendous research efforts made in anti-cancer drug development, cancer remains one of the leading causes of death in the modern world with an alarming rising mortality rate.

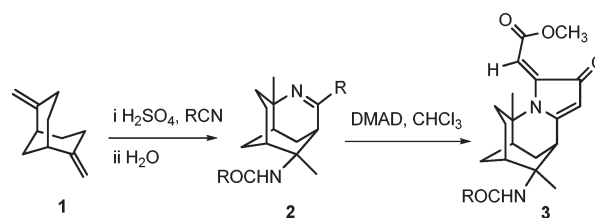
Multidrug resistance is one of the major contributing factors in the battle against cancer. Cancer cells can grow resistance either due to the nature and genetic background of the cancer cell or as a result of genetic changes following chemotherapy treatments. All of these factors make the pursuit of better anti-cancer agents a challenging task.<sup>5–7</sup>

The intramolecular variation of the standard Ritter reaction offers a diverse variety of heterocyclic products with an imine functional group.<sup>8–10</sup> Also, there are a small number of cases involving the formation of a bridged bicyclic structure, known as the bridging Ritter product. This bridging cyclisation has been used previously by our research group to synthesise more complex alkaloid-like molecules.<sup>11,12</sup>

Our research team further explored the reactivity of the tricyclic imine scaffolds with dimethyl acetylenedicarboxylate (DMAD) to form a multicyclic molecule **3** that is highly reminiscent of the natural alkaloids, as shown in Scheme 1.

The unique chemical scaffold and drug-like properties of 3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene **2**, would make it the ideal building block for drug discovery. Therefore, the imine **2** has been developed into a small library of novel and relatively complex alkaloid-like compounds.

Breast cancer is the most common cancer amongst women; it comprises distinct molecular subtypes that affect



**Scheme 1** Two-step synthesis of **3** from 2,6-dimethylenecyclo[3.3.1]nonane **1**.

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the patient's response to different pathway-targeted therapies. Breast cancers that express estrogen receptor (ER) and/or progesterone receptor (PR) can be treated with hormonal therapy.<sup>13</sup> Triple-negative breast cancers lack these receptors, as well as the human epidermal growth factor receptor 2 (HER2), and are therefore unresponsive to targeted breast cancer drugs. Moreover, triple negative breast cancers are more aggressive and metastatic. Thus, in this paper, we report on the synthesis of alkaloid-like compounds and assess their cytotoxicities in-house against the MDA-MB-231 triple negative breast cancer cell line, as well as against a broad panel of cancers using the NCI screen.<sup>14</sup>

## Results and discussion

### Chemistry

As part of our ongoing research into alkaloid-like compounds as anti-cancer agents,<sup>15,16</sup> we have synthesised a library of tricyclic compounds using the bridging Ritter reaction as the key reaction. Various nitriles were used, and subsequent reactions such as reduction, reductive alkylation, and DMAD cycloaddition were used to derivatise these small molecules to heterocyclic alkaloid-like compounds.<sup>17</sup>

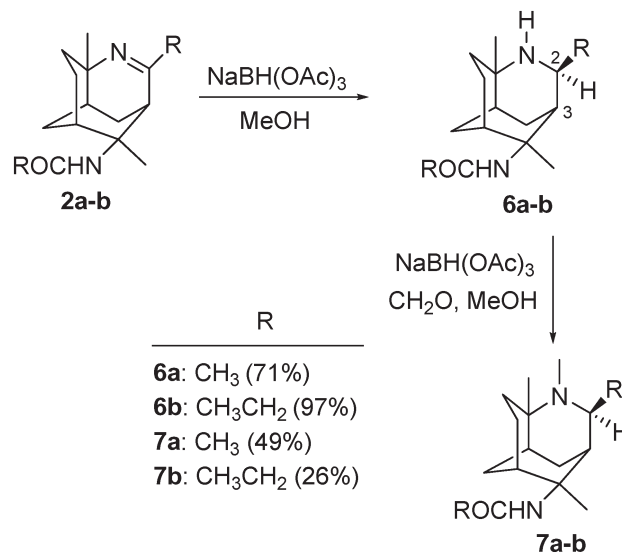
### Bridged Ritter reaction with various nitriles

Our research group previously prepared the cyclic diene **1** as the precursor for generating the tricyclic imine scaffold.<sup>11,12,18,19</sup> Tricyclic imines **2a–c** were synthesised *via* the bridging Ritter reaction in benzene under strongly acidic conditions (Scheme 2). The nitriles were chosen to provide a variety of substituents at the C-2 position that alters the reactivity of the respective imine group. The reaction mechanism for the formation of the tricyclic imines was described in our previous publications.<sup>18,19</sup>

The bulky and electron poor nature of ClCH<sub>2</sub>CN has resulted in the decrease in the percentage yield of **2c** in comparison to **2a** and **2b**. This phenomenon has also been observed in our earlier work, where outcomes of the bridging products were significantly influenced by the nitriles.<sup>15</sup>

### Reduction and reductive alkylation of the 3-azatricyclo-[5.3.1.0<sup>4,9</sup>]undec-2-ene system

Reduction of **2a–b** was carried out under mild reduction conditions with NaBH(OAc)<sub>3</sub> in dry methanol (Scheme 3). The hydride reduces the imine exclusively from the less hindered



Scheme 3 Reduction and reductive alkylation of tricyclic imines.

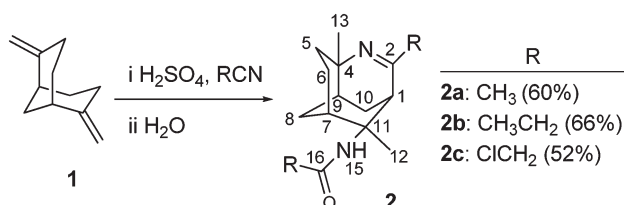
face of the tricyclic ring structure to yield diastereoisomers **6a** and **6b** in 71% and 97% yields, respectively.

The structures of **6a** and **6b** were confirmed by NMR analysis. The <sup>1</sup>H NMR spectrum showed the newly formed H-2 peaks at 3.50 and 3.12 ppm for **6a** and **6b** respectively. The rest of the spectral features remained similar to those of their corresponding imines, with slight upfield shifts of H-3. The stereochemistry of H-2 was evident in the 1D NOESY NMR studies, where cross peaks were observed for H-2 and H-3. This stereoselectivity results from the Re face of the cyclic ring system being hindered by the bicyclic cyclohexane rings.

Compounds **6a** and **6b** were then subjected to reductive alkylation with 28% formaldehyde to yield compounds **7a** and **7b** in 49% and 26% yield, respectively. The experimental yields of **7a** and **7b** highlight the significance of inaccessibility of N-3 due to the steric hindrance created by the substituents at the C-2 and C-4 positions.

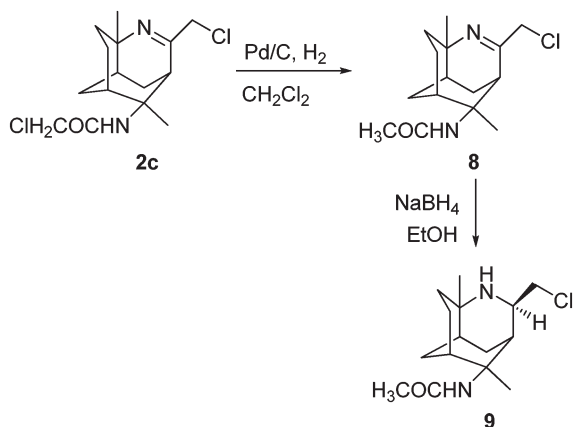
Structures of **7a** and **7b** were confirmed by NMR analysis. The <sup>1</sup>H NMR spectrum showed the newly formed –NCH<sub>3</sub> peaks at 2.29 and 2.42 ppm, respectively, while the <sup>13</sup>C NMR spectrum confirmed the formation of two –NCH<sub>3</sub> peaks at 35.97 and 39.30 ppm corresponding to the structures of **7a** and **7b** respectively.

The reduction of **2c** using either NaBH(OAc)<sub>3</sub> or NaBH<sub>4</sub> provided an unidentified decomposed material as indicated by <sup>1</sup>H NMR analysis of the crude mixtures. In an attempt to reduce **2c** by hydrogenation in the presence of 5% Pd/C, we recovered the unexpected product **8** in 79% yield (Scheme 4). The formation of **8** can be explained *via* the palladium catalysed dehalogenation of α-haloamides.<sup>20,21</sup> Compound **8** was then subjected to reduction using NaBH<sub>4</sub> to give **9** in 72% yield. The structure of **9** was confirmed by NMR analysis; the <sup>1</sup>H NMR spectrum showed the newly formed –CH peak at 3.61 ppm as a quintet. The splitting of this –CH peak further confirms that the imine has been reduced. The rest



Scheme 2 Synthesis of tricyclic imines (**2a–c**).





Scheme 4 Hydrogenation and reduction of compound 2c.

of the spectrum remained similar to that of the other reduced imines.

### Cycloaddition of DMAD to 3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene and reduction of 3

Cyclic amide 3 was synthesised from a one-pot reflux in chloroform with acetylenedicarboxylate (DMAD) and tricyclic imine 2a in 40% yield, according to the method that was previously reported.<sup>18,22</sup> It was then selectively reduced with NaBH<sub>4</sub> in the presence of NiCl<sub>2</sub>·6H<sub>2</sub>O to yield novel compound 10 in 70% yield (Scheme 5).

The <sup>1</sup>H NMR spectrum of 10 showed the disappearance of the C-21 alkene proton peak of compound 3 at 5.86 ppm. A newly formed doublet of doublets at 2.91 and 2.81 ppm corresponded to the structure of 3 being reduced at the C-12 and C-21 positions. The <sup>13</sup>C NMR spectrum further confirmed the presence of HC-12 and H<sub>2</sub>C-21 at 61.6 and 40.9 ppm, respectively. 2D NOESY measurements were carried out on 10 to ascertain the relative stereochemistry at C-12. There were significant cross-peaks between H-12 & H-15 and between H-13 & H-21. The lowest-energy conformation of 10 was calculated using Spartan 10 molecular modelling software (WAVEFUNCTION, INC). The calculation was carried out using the setup menu in Spartan, with the specification of equilibrium conformation, at ground states, semi-empirical model at AM1 level. The molecular modelling of compound 10 (Fig. 1) predicted the distances between H-12 & H-15, and between H-13 & H-21 to be 2.61 and 2.37 Å, re-

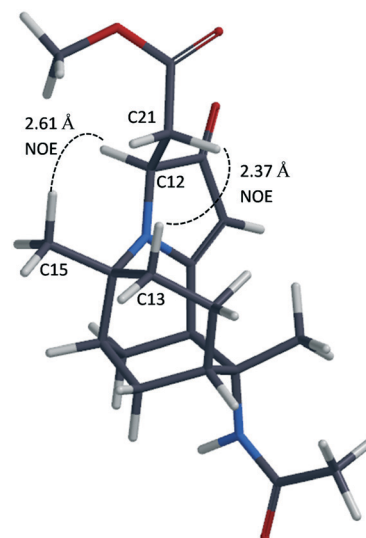


Fig. 1 Molecular model of compound 10 (Spartan 10, generated structure), showing NOE correlations between H-12 and H-15, and H-13 and H-21.

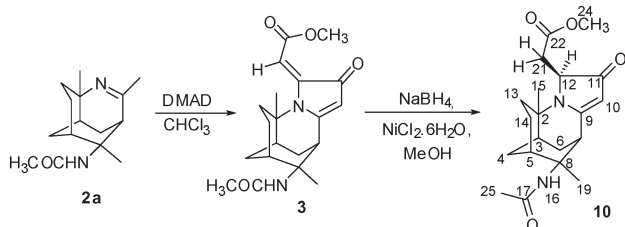
spectively. This observation supports the (*S*)-configuration at C-12 which further indicates that the hydride attacked the C-12 position from the less hindered *Si* face of the heterocyclic ring.

### Biological evaluation

**Broad screening for Antiproliferative activity of compound 3.** Compound 3 was selected by the US National Cancer Institute (NCI) for their standard anti-cancer screening program.<sup>11</sup> Compound 3 was screened for antiproliferative activity *in vitro* against 48 cell lines covering nine cancer types (leukaemia, non-small cell lung cancer, colon cancer, CNS cancer, melanoma, ovarian cancer, renal cancer, prostate cancer and breast cancer). Significant cytotoxicities with IC<sub>50</sub> in the μM range were determined for most of cancer cell lines employed in the screening program; selective results are summarised in Table 1. The testing results of 3 against of 48 cancer cell lines tested are provided in the ESI.†

The dose response curves from the screening of compound 3 against leukaemia, breast cancer, colon and melanoma cells lines determined by NCI are presented in Fig. 2.

Compound 3 potently inhibited the proliferation of leukaemia, colon, renal and melanoma cancer cell lines with IC<sub>50</sub> values below 10 μM. Amongst the breast cancer cell lines, some variation in activity was observed. The proliferation of most of these cell lines was potently inhibited by 3. However, aggressive and highly metastatic MDA-MB-231 cell line was the least active (IC<sub>50</sub> = 38.9 μM). This suggests that compound 3 is more active towards acute lymphoblastic leukaemia both T-cell and B-cell, as well as acute promyelocytic leukaemia, and less active towards chronic myelogenous leukaemia which observed a threefold increase in the IC<sub>50</sub> values



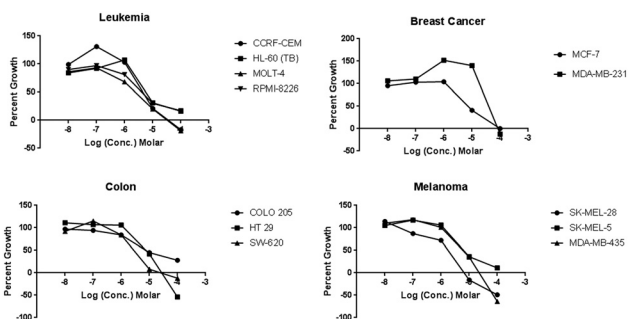
Scheme 5 Derivatisation of tricyclic imine scaffold 2a.



**Table 1** Selected cytotoxicity (IC<sub>50</sub>)<sup>a</sup> of compound 3

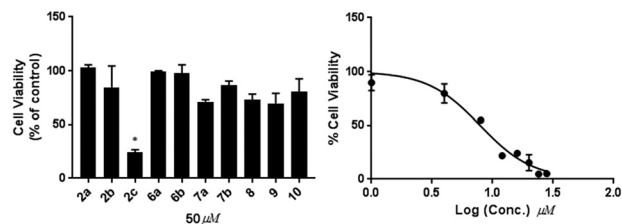
Tumour type	Cell line	IC <sub>50</sub> (μM)
Leukemia	CCRF-CEM	4.4
	HL-60 (TB)	5.5
	MOLT-4	2.4
	RPMI-8226	4.1
Non-small cell lung	NCI-H522	2.0
	EKVX	12.2
Colon cancer	COLO-205	7.8
	HT29	7.5
	SW-620	2.8
CNS cancer	SF-539	3.3
	Melanoma	
Melanoma	SK-MEL-28	1.8
	SK-MEL-5	6.3
Ovarian cancer	MDA-MB-435	5.9
	OVCAR-3	11.8
Renal cancer	CAKI-1	5.3
	RXF 393	6.7
Prostate cancer	DU-145	16.9
	PC-3	15.7
Breast cancer	DU-145	16.9
	MCF-7	7.1
	MDA-MB-231/ATCC	38.9
	HS 578T	10.9
	T-47D	22.7

<sup>a</sup> Concentration of compound 3 required inhibiting cell growth by 50%.

**Fig. 2** Dose response curves of compound 3 against leukaemia, breast cancer, colon, and melanoma cell lines.

comparing to that of lymphoblastic and promyelocytic leukaemia (Table 1). The selectivity of compound 3 towards the lymphoblastic and promyelocytic leukaemia cell lines could be due to the similarities present in the immunohistochemistry of the CCRF-CEM, HL60, MOLT4, and RPMI-8226 cell lines such as the CD antigen expression patterns, and the expression of both TCR- $\alpha$  and TdT genes.<sup>23–25</sup> Future investigation is needed to determine the connection in the observed activity across these cell lines from the broad screening.

Diversification of compound structures was achieved by selectively reducing the  $\alpha,\beta$ -unsaturated system in compound 3 to yield compound 10. However, the removal of the double bond at C-21 resulted in decreased the cytotoxicities of compound 10 (IC<sub>50</sub> > 50  $\mu$ M) against MDA-MB-231 cells as shown in Fig. 3.

**Fig. 3** Effects of 2a–c, 6a–b, 7a–7b, 8, 9 and 10 (50  $\mu$ M, 48 hours) on the proliferation of MDA-MB-231 breast cancer cells (left). \* indicate  $p = 0.001$ . Dose response curve of 2c is shown on the right.

### Antiproliferative activity in breast cancer (MDA-MB-231).

The anti-proliferative activity of the remaining alkaloid-like compounds 2a–c, 6a–b, 7a–b, 8, 9 and 10 was assessed against MDA-MB-231 breast cancer cells using the 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulphophenyl)-2H-tetrazolium (MTS) assay (Fig. 3, left panel). Treatment of MDA-MB-231 cells with compound 2c markedly decreased the cell viability to 24% after 48 hours ( $p < 0.001$ ) at the maximum concentration tested 50  $\mu$ M, while other compounds did not display activity under these conditions. Dose–response curves were constructed for compound 2c, and an IC<sub>50</sub> of  $7.9 \pm 1.5$   $\mu$ M was determined (Fig. 3, right panel). The lack of activity observed for compounds 2a–b, 6a–b, 7a–b, 8, 9 and 10 suggests that antiproliferative activity increases dramatically in the presence of chlorine at the C-2 and C-18 positions of the tricyclic scaffold.

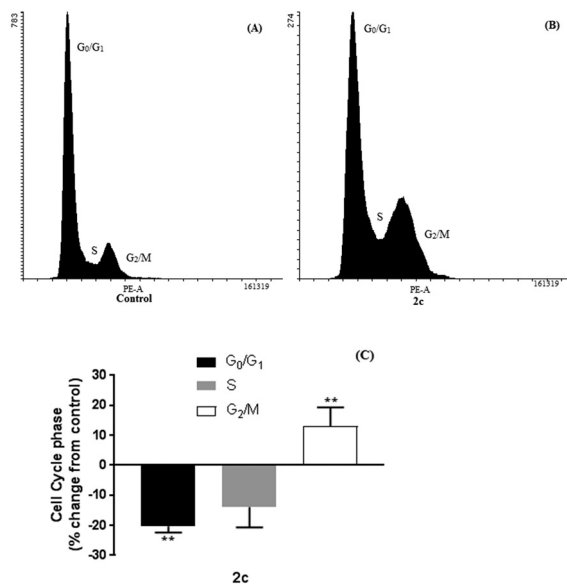
### Cell cycle analysis

Further investigation into the mechanism of action of these compounds was determined to explain for the observed selectivity. To investigate the primary mechanism of action for active compound 2c, we assessed the ability of 2c to induce cell cycle arrest and apoptosis in MDA-MB-231 cells using flow cytometry. Cell cycle studies revealed that compound 2c (20  $\mu$ M, 24 hours) induced cell cycle arrest and altered the distribution of cells throughout the cell cycle (Fig. 4). Thus, treatment with 2c significantly decreased the population of MDA-MB-231 cells in G<sub>0</sub>/G<sub>1</sub> by 20% ( $p < 0.0001$ ), and increased the proportion of cells in the G<sub>2</sub>/M phase by 15% ( $p < 0.0001$ ) comparing to that of control. These findings suggest that cell cycle arrest at the G<sub>2</sub>/M phase contributes to the anti-proliferative activity of 2c.

### Induction of apoptosis

The translocation of phospholipid phosphatidylserine (PS) to the outer leaflet of the cell plasma membrane is one of the hallmarks of apoptosis.<sup>26</sup> This is followed by a loss of membrane integrity and eventual cell death. The ability of compound 2c to induce apoptosis in MDA-MB-231 cells was also assessed using flow cytometry. MDA-MB-231 cells were treated with 20 and 40  $\mu$ M of 2c for 48 h and then stained with Annexin V-FITC (which binds to exposed PS)

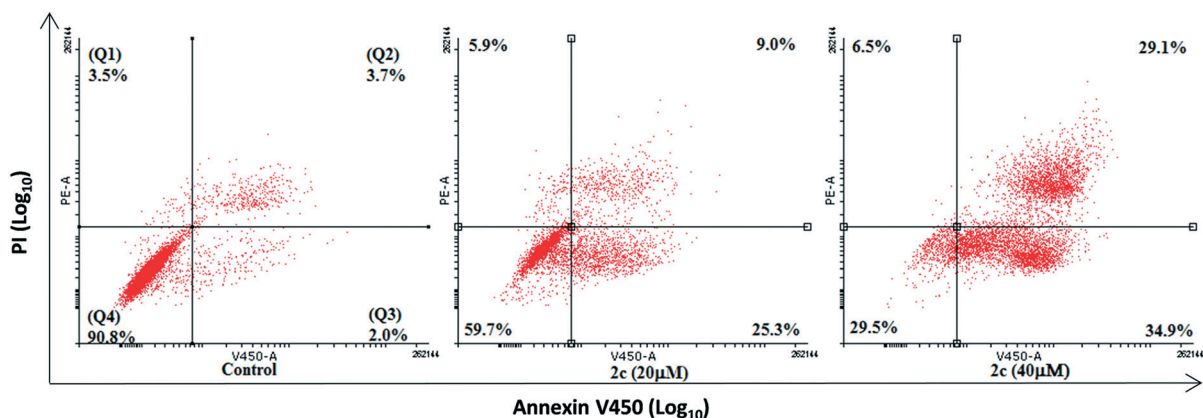




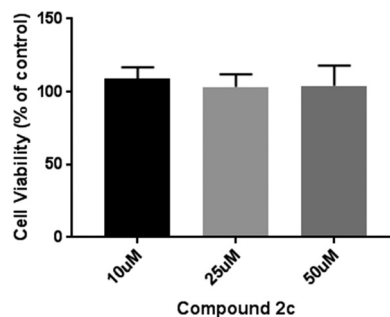
**Fig. 4** Effect of **2c** on the cell cycle progression of MDA-MB-231 cells; cells were treated for 24 h with (A) DMSO (control), (B) **2c** (20  $\mu\text{M}$ ), (C) percentage change from control. Data represent the mean  $\pm$  SEM of at least 3 independent experiments. \*\* indicates  $p < 0.0001$ .

and PI. This allows for the determination of the proportion of cells of undergoing apoptosis at either early apoptosis (Annexin-positive, PI-negative), late apoptosis (Annexin-positive, PI-positive) or total apoptosis (early + late apoptosis) (Fig. 5).

MDA-MB-231 cells tested positive for Annexin V-FITC after treatment with **2c** (20 and 40  $\mu\text{M}$ , 48 hours)  $p = 0.0223$  and  $p = 0.0047$  respectively, which indicates the cells are in the early stage of apoptosis after treatment of the drug (Fig. 5). The population of cells that underwent early apoptosis after treatment with **2c** (20 and 40  $\mu\text{M}$ ) was increased by 21% and 31% respectively, which suggests a dose-dependent increase in the apoptotic population.



**Fig. 5** Effect of compound **2c** on the induction of apoptosis in MDA-MB-231 cells; cells were treated with 20 and 40  $\mu\text{M}$  concentrations of **2c** for 48 h, stained with Annexin V-FITC and PI, and the apoptotic effect was assessed by flow cytometry. Representative results are shown, with quadrants indicating the proportion of cells that are necrotic: Q1, late apoptotic: Q2, early apoptotic: Q3, and viable: Q4. Images are representative of 3 independent experiments.



**Fig. 6** Effects of **2c** (10–50  $\mu\text{M}$ , 48 hours) on the proliferation of MCF-7 breast cancer cells.

#### Cytotoxicity of active of compound **2c** against MCF-7

The selectivity of compound **2c** was investigated by performing MTS cell proliferation assay in varying concentrations (10–50  $\mu\text{M}$ ) on the ER+ (MCF-7) breast cancer cell line Fig. 6. The effects of compounds **2c** on the proliferation and viability of MCF-7 cells after 48 h treatment was determined. Compound **2c** exhibited no antiproliferative activities towards the MCF-7 cells at the highest concentration tested (50  $\mu\text{M}$ ).

#### Cytotoxicity of compounds **2c** and **3** against VERO cells

To assess the selectivity of compounds **2c** and **3** towards cancer cells, the cytotoxicities of these compounds were assessed against the noncancerous VERO cell line at 50  $\mu\text{g mL}^{-1}$  (or 158  $\mu\text{M}$  for **2c**, and 139  $\mu\text{M}$  for **3**). In these assays, **2c** and **3** reduced VERO cell viability by 84% and 77% of control, respectively. In contrast **2c** reduced MDA-MB-231 cell viability to 90% of control at 20  $\mu\text{M}$ , and in the case of **3** viability was reduced to 80% at 75  $\mu\text{M}$ . These data clearly demonstrate the selectivity of **2c** and **3** towards cancer cells over noncancerous cell lines.



## Conclusions

A small library of alkaloid-like compounds was synthesised and their anticancer activities were evaluated. Promising anti-cancer activity was established for compound **3** across multiple cancer types, and for compound **2c** against the MDA-MB-231 breast cancer cell line, while showed no activity against MCF-7 at 50  $\mu$ M. Furthermore, cell cycle and apoptosis studies revealed compound **2c** promotes cell cycle arrest in the G<sub>2</sub>/M phase, as well as inducing apoptosis after 48 hours of treatment. The observed selectivity of **2c** against MDA-MB-231 over noncancerous VERO cells warrants further development in these alkaloid-like compounds *via* bridging Ritter reaction.

## Experimental

### General information

Reagents and analytical grade solvents were purchased from commercial sources. Compounds were purified by column chromatography using flash silica gel (60–80  $\mu$ m). The purity of compounds was determined by <sup>1</sup>H NMR and GC/MS. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on an Agilent 500 MHz spectrometer (500 MHz <sup>1</sup>H, 125 MHz <sup>13</sup>C) in deuterated chloroform (CDCl<sub>3</sub>) unless otherwise specified. NMR assignments were based on COSY, HSQC, NOESY and DEPT experiments. <sup>1</sup>H and <sup>13</sup>C NMR assignments are based on the numbering system used in the systematic name. Low-resolution mass spectra were obtained on an Agilent 6890GC fitted with 5% polysilphenylene, 95% polydimethylsiloxane column, and an Agilent 5973n MS (EI) spectrometer. High-resolution mass spectra were obtained on an Agilent 6510 Accurate-Mass Q-TOF mass spectrometer, equipped with an ESI source.

### General procedure for the synthesis of imines *via* bridging Ritter reaction

A mixture of 98% H<sub>2</sub>SO<sub>4</sub> (2 mL) and nitrile (41.60 mmol, 8 mole equiv.) in 100 mL round bottomed flask fitted with a condenser and drying tube, was stirred in an ice bath. A solution of 2,6-dimethylenebicyclo[3.3.1]nonane **1** (0.77 g, 5.20 mmol) in benzene (5 mL) was added to the mixture dropwise but rapidly. The reaction mixture was stirred at 0 °C for 30 min, then at room temperature overnight. Water was added to the mixture and stirring was continued for a further 30 min. The mixture was extracted with diethyl ether. The aqueous layer was washed with 1 M NaOH solution (18 mL) and extracted with chloroform. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and solvent removed under reduced pressure. The crude product was purified by column chromatography ethyl acetate : hexane (4 : 6).

**Exo-11-propanamido-4,endo-11-dimethyl-2-ethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene 2b.** As yellow waxy solid (1.22 g, 66%); *R*<sub>f</sub> (9 : 1 ethyl acetate/methanol) 0.75; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.22 (s, 1H, NH-15), 3.36 (s, 1H, H-1), 2.36 (q, *J* = 7.5 Hz, 2H, CH<sub>2</sub>-18), 2.21–2.15 (m, 2H, CH<sub>2</sub>-14), 1.82 (d, *J* = 10.5 Hz, 1H, H-9), 1.77–1.68 (m, 3H, H-7, H-10a,

H-5a), 1.64–1.48 (m, 4H, H-8a, H-10b, H-5b, H-6a), 1.46 (s, 3H, CH<sub>3</sub>-13), 1.27 (s, 3H, CH<sub>3</sub>-12), 1.32–1.22 (m, 2H, H-6b, H-8b), 1.14 (t, *J* = 7.5 Hz, 6H, CH<sub>3</sub>-19, CH<sub>3</sub>-20); <sup>13</sup>C NMR  $\delta$  174.8 (C=O), 173.1 (C-2), 62.1 (C-11/4), 61.0 (C-11/4), 40.7 (CH-1), 40.1 (CH-9), 37.7 (CH<sub>2</sub>-18), 35.7 (CH<sub>2</sub>-5), 34.9 (CH-7), 34.4 (CH<sub>3</sub>-12), 33.6 (CH<sub>2</sub>-14), 30.6 (CH<sub>2</sub>-8), 30.3 (CH<sub>2</sub>-10), 25.4 (CH<sub>3</sub>-13), 22.1 (CH<sub>2</sub>-6), 14.3 (CH<sub>3</sub>-19/20), 12.6 (CH<sub>3</sub>-19/20); FT-IR  $\nu_{\max}$  (neat) 3440, 3249, 3190, 2925, 2110, 1650, 1554, 1449, 1371, 1354, 1112, 959, 941, 744 cm<sup>-1</sup>; HRMS (ESI): [M + H]<sup>+</sup>, found 277.2273, C<sub>17</sub>H<sub>29</sub>N<sub>2</sub>O required 277.2280.

**Exo-11-chloroacetamido-4,endo-11-dimethyl-2-chloromethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene 2c.** As yellow waxy solid (1.10 g, 52%); *R*<sub>f</sub> (9 : 1 ethyl acetate/methanol) 0.76; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.14 (br s, 1H, NH-15), 4.19 (d, *J* = 11.50 Hz, 1H, H-18a), 4.12 (d, *J* = 11.5 Hz, 1H, H-18b), 4.00 (s, 2H, CH<sub>2</sub>-14), 3.48 (br s, 1H, H-1), 2.05 (d, *J* = 10.5 Hz, 1H, H-9), 1.85 (br s, 1H, H-7), 1.84–1.73 (m, 2H, H-10a, H-5a), 1.69 (d, *J* = 10.5 Hz, 1H, H-8a), 1.64–1.52 (m, 3H, H-6a, H-10b, H-5b), 1.50 (s, 3H, CH<sub>3</sub>-12), 1.38–1.36 (m, 1H, H-8b), 1.33 (s, 3H, CH<sub>3</sub>-13); <sup>13</sup>C NMR  $\delta$  169.1 (C=O), 165.0 (C=N), 60.4 (C-11/4), 60.1 (C-11/4), 49.7 (CH<sub>2</sub>-18), 43.5 (CH<sub>2</sub>-14), 37.0 (CH-1), 36.6 (CH-9), 33.2 (CH<sub>2</sub>-10), 32.5 (CH-7), 31.6 (CH<sub>3</sub>-13), 28.3 (CH<sub>2</sub>-8), 27.8 (CH<sub>2</sub>-10), 22.7 (CH<sub>3</sub>-12), 19.6 (CH<sub>2</sub>-6); FT-IR  $\nu_{\max}$  (neat) 3530, 3455, 3267, 3210, 2954, 2146, 1660, 1554, 1450, 1391, 1367, 1355, 1120, 944, 850, 722 cm<sup>-1</sup>; HRMS (ESI): [M + H]<sup>+</sup>, found 317.1195, C<sub>15</sub>H<sub>23</sub>Cl<sub>2</sub>N<sub>2</sub>O required 317.1187.

### General procedure for hydride reduction of tricyclic imines

Imines **2a** (1.81 mmol) was dissolved in dry MeOH (15 mL) and placed in a round bottom flask fitted with a drying tube, NaBH(OAc)<sub>3</sub> (2.30 g, 10.86 mmol, 6 mole equiv.) was slowly added. The mixture was stirred overnight at room temperature. The reaction was quenched with a saturated solution of NaHCO<sub>3</sub> (20 mL), extracted with CHCl<sub>3</sub> (20 mL  $\times$  2), and washed with a saturated solution of NaCl (20 mL). The organic layer was dried with Na<sub>2</sub>CO<sub>3</sub> and solvent was removed under reduced pressure to yield a viscous liquid which solidified upon standing over a day. The product was purified by column chromatography ethyl acetate : methanol (8 : 2) to afford the reduced compound **6a**.

**Exo-11-acetamido-2,4,endo-11-trimethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undecane 6a.** As yellow waxy solid (0.355 g, 71%); *R*<sub>f</sub> (8 : 2 ethyl acetate/methanol) 0.60; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.31 (s, 1H, NH-15), 3.50 (q, *J* = 7.5 Hz, 1H, H-2), 2.44 (br s, 1H, H-1), 2.09–2.06 (m, 1H, H-9), 1.95 (s, 3H, CH<sub>3</sub>-12), 1.83–1.59 (m, 7H, H-6a, H-6b, H-8a, H-8b, H-7, H-10a, H-5a), 1.58 (s, 3H, CH<sub>3</sub>-18), 1.50 (br d, *J* = 13.0 Hz, 1H, H-10b), 1.34 (d, *J* = 7.5 Hz, 3H, CH<sub>3</sub>-14), 1.17–1.12 (m, 1H, H-5b), 1.10 (s, 3H, CH<sub>3</sub>-13); <sup>13</sup>C NMR  $\delta$  171.7 (C=O), 63.9 (C-11/4), 56.8 (CH-2), 55.2 (C-11/4), 41.6 (CH-1), 38.8 (CH<sub>2</sub>-5), 37.5 (CH<sub>3</sub>-18), 37.1 (CH-9), 36.1 (CH-7), 34.3 (CH<sub>2</sub>-8), 30.5 (CH<sub>2</sub>-10), 28.1 (CH<sub>3</sub>-12), 27.8 (CH<sub>3</sub>-14), 26.3 (CH<sub>3</sub>-13), 22.8 (CH<sub>2</sub>-6); FT-IR  $\nu_{\max}$  (neat) 3256, 3191, 2956, 2925, 2873, 1648, 1550, 1457, 1372, 1353, 1234, 1112, 1076, 940, 721 cm<sup>-1</sup>; HRMS (ESI): [M + H]<sup>+</sup>, found 251.2111, C<sub>15</sub>H<sub>27</sub>N<sub>2</sub>O required 251.2123.



**Exo-11-propanamido-4,endo-11-dimethyl-2-ethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undecane 6b.** As waxy solid (0.488 g, 97%);  $R_f$  (8:2 ethyl acetate/methanol) 0.62;  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  5.29 (s, 1H, NH-15), 3.12 (quint,  $J = 5.5$  Hz, 1H, H-2), 2.58 (br s, 1H, H-1), 2.16 (dq,  $J = 1.0, 7.5$  Hz, 2H,  $\text{CH}_2$ -14), 1.96 (dd,  $J = 1.0, 11.0$  Hz, 1H, H-9), 1.98–1.82 (m, 1H, H-6a), 1.82–1.76 (m, 2H, H-5a, H-7), 1.72–1.65 (m, 2H, H-6b, H-10a), 1.58 (s, 3H,  $\text{CH}_3$ -12), 1.60–1.55 (m, 3H, H-10b,  $\text{CH}_2$ -18), 1.48 (t,  $J = 1.5$  Hz, 1H, H-10b), 1.46 (br s, 1H, H-8a), 1.27 (s, 1H, H-8b), 1.14 (t,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -20), 1.10 (s, 3H,  $\text{CH}_3$ -13), 0.97 (t,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -19);  $^{13}\text{C NMR}$   $\delta$  172.63 (C=O), 61.9 (CH-2), 61.0 (C-11/4), 37.3 (CH-1), 36.6 ( $\text{CH}_2$ -18), 35.5 (CH-9), 34.6 ( $\text{CH}_3$ -12), 34.3 (CH-7), 32.3 ( $\text{CH}_2$ -5), 31.8 ( $\text{CH}_2$ -14), 31.2 ( $\text{CH}_2$ -8), 28.2 ( $\text{CH}_2$ -10), 23.3 ( $\text{CH}_3$ -13), 20.4 ( $\text{CH}_2$ -6), 12.7 ( $\text{CH}_3$ -19), 10.1 ( $\text{CH}_3$ -20); FT-IR  $\nu_{\text{max}}$  (neat) 3286, 3260, 2985, 2883, 2673, 1659, 1549, 1432, 1377, 1320, 1247, 1102, 1060, 989, 756  $\text{cm}^{-1}$ ; HRMS (ESI):  $[\text{M} + \text{H}]^+$ , found 279.2436,  $\text{C}_{17}\text{H}_{31}\text{N}_2\text{O}$  required 279.2436.

**Exo-11-acetamido-4,endo-11-dimethyl-2-chloromethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene 8.** Imine 2c (50 mg, 0.158 mmol) was dissolved in dry  $\text{CH}_2\text{Cl}_2$  (10 mL) and placed in a round bottom flask fitted with a hydrogen balloon, 5% Pd/C (25 mg, 0.012 mmol, 0.074 mole equiv.) was added to the mixture. The mixture was stirred under  $\text{H}_2$  atmosphere overnight at room temperature. Upon the completion of reaction, Pd/C was filtered and the solvent was removed under reduced pressure. The product was purified by column chromatography to yield compound 8 as a viscous liquid (0.035 g, 79%);  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  6.51 (br s, 1H, NH-15), 3.99 (s, 2H, H-14), 3.14 (s, 1H, H-1), 2.15 (s, 3H,  $\text{CH}_3$ -18), 1.93 (d,  $J = 11.0$  Hz, 1H, H-9), 1.84–1.77 (m, 1H, H-7), 1.84–1.53 (m, 6H, H-8a, H-8b, H-6a, H-5a, H-5b, H-10a), 1.51 (s, 3H,  $\text{CH}_3$ -12), 1.38–1.32 (m, 2H, H-10b, H-6b), 1.29 (s, 3H,  $\text{CH}_3$ -13);  $^{13}\text{C NMR}$   $\delta$  167.3 (C=O), 165.7 (C-2), 62.6 (C-11/4), 61.4 (C-11/4), 45.9 ( $\text{CH}_2$ -14), 42.8 (CH-1), 39.5 (CH-9), 35.6 ( $\text{CH}_2$ -5), 34.4 (CH-7), 34.3 ( $\text{CH}_3$ -13), 31.9 ( $\text{CH}_3$ -18), 30.2 ( $\text{CH}_2$ -8), 30.0 ( $\text{CH}_2$ -10), 25.2 ( $\text{CH}_3$ -12), 22.0 ( $\text{CH}_2$ -6); FT-IR  $\nu_{\text{max}}$  (neat) 3554, 3420, 3228, 3210, 2983, 2150, 1665, 1595, 1450, 1351, 1331, 1311, 1125, 957, 880, 778  $\text{cm}^{-1}$ ; HRMS (ESI):  $[\text{M} + \text{H}]^+$ , found 283.1581,  $\text{C}_{15}\text{H}_{24}\text{ClN}_2\text{O}$  required 283.1579.

**Exo-11-acetamido-4,endo-11-dimethyl-2-chloromethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undecane 9.** Compound 8 (0.050 g, 0.158 mmol) was dissolved in dry ethanol (10 mL) and placed in a round bottom flask fitted with a drying tube.  $\text{NaBH}_4$  (0.067 g, 1.58 mmol, 10 mole equiv.) was added to the mixture. The mixture was stirred overnight at room temperature. The solvent was removed from under reduced pressure. The product was redissolved in  $\text{CH}_2\text{Cl}_2$  and washed with saturated  $\text{NaHCO}_3$ . The organic layer was dried over  $\text{Na}_2\text{CO}_3$  and solvent was removed under reduced pressure. The product was purified by column chromatography, ethyl acetate:methanol (9.5:0.5) to yield amine 9 as an off white viscous liquid (0.035 g, 72%),  $R_f$  (9.5:0.5 ethyl acetate/methanol) 0.64;  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  6.58 (br s, 1H, NH-15), 3.98 (s, 2H, H-14), 3.61 (dt,  $J = 7.0, 14.0$  Hz, 1H, H-2), 2.45 (br s, 1H, CH-1), 2.18–2.12 (m, 1H, H-9), 1.86–1.76 (m, 4H, H-8a, H-7, H-6a,

H-6b), 1.74–1.65 (m, 2H, H-8b, H-10a), 1.64–1.62 (m, 1H, H-5a), 1.60 (s, 3H,  $\text{CH}_3$ -18), 1.56–1.50 (m, 1H, H-10b), 1.39 (d,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -13), 1.34–1.27 (m, 1H, H-5b), 1.17 (s, 3H,  $\text{CH}_3$ -12);  $^{13}\text{C NMR}$   $\delta$  166.8 (C=O), 64.2 (C-11/4), 56.6 (CH-2), 56.0 (C-11/4), 46.0 ( $\text{CH}_2$ -14), 41.3 (CH-1), 37.9 ( $\text{CH}_2$ -5), 36.9 (CH-9), 36.9 ( $\text{CH}_3$ -12), 35.8 (CH-7), 33.7 ( $\text{CH}_2$ -8), 30.1 ( $\text{CH}_2$ -10), 27.5 ( $\text{CH}_3$ -13), 26.1 ( $\text{CH}_3$ -18), 22.7 ( $\text{CH}_2$ -6); FT-IR  $\nu_{\text{max}}$  (neat) 3279, 3265, 2952, 2800, 2651, 1660, 1589, 1450, 1395, 1388, 1200, 1120, 1080, 989, 877, 756  $\text{cm}^{-1}$ ; HRMS (ESI):  $[\text{M} + \text{H}]^+$ , found 285.1715,  $\text{C}_{15}\text{H}_{26}\text{ClN}_2\text{O}$  required 285.1734.

#### General conditions for reductive alkylation of 3-azatricyclo[5.3.1.0<sup>4,9</sup>]undec-2-ene

Amine 6 (0.92 mmol) was dissolved in MeCN (20 mL), to which  $\text{NaBH}_3\text{CN}$  (1.15 g, 18.4 mmol, 20 mole equiv.), 40% formaldehyde solution (2 mL) and a few drops of glacial acetic acid were added. The solution was stirred for 3 hours at room temperature. The solvent was removed under reduced pressure. The crude was dissolved in  $\text{CHCl}_3$  and washed with saturated  $\text{NaHCO}_3$ , saturated NaCl, and dried with  $\text{Na}_2\text{SO}_4$  to yield a pale red solid 7a.

**Exo-11-acetamido-2,3,4,endo-11-tetramethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undecane 7a.** As pale red viscous solid (0.117 g, 49%);  $R_f$  (9:1 ethyl acetate/methanol) 0.26;  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  5.17 (s, 1H, NH-15), 3.05–3.01 (m, 1H, H-2), 2.47 (br 1H, H-1), 2.29 (s, 3H, NCH<sub>3</sub>), 2.01–1.98 (m, 1H, H-9), 1.93 (s, 3H,  $\text{CH}_3$ -12), 1.90–1.64 (m, 7H, H-6a, H-6b, H-8a, H-8b, H-7, H-10a, H-5a), 1.58 (s, 3H,  $\text{CH}_3$ -18), 1.49 (d,  $J = 4.0$  Hz, 1H, H-10b), 1.30 (d,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -14), 1.22–1.14 (m, 1H, H-5b), 1.11 (s, 3H,  $\text{CH}_3$ -13);  $^{13}\text{C NMR}$   $\delta$  171.7 (C=O), 63.9 (C-11/4), 56.8 (CH-2), 55.2 (C-11/4), 41.6 (CH-1), 38.8 ( $\text{CH}_2$ -5), 37.5 ( $\text{CH}_3$ -18), 37.1 (CH-9), 36.1 (CH-7), 36.0 (NCH<sub>3</sub>), 34.3 ( $\text{CH}_2$ -8), 30.5 ( $\text{CH}_2$ -10), 28.1 ( $\text{CH}_3$ -12), 27.8 ( $\text{CH}_3$ -14), 26.3 ( $\text{CH}_3$ -13), 22.8 ( $\text{CH}_2$ -6); FT-IR  $\nu_{\text{max}}$  (neat) 3360, 3225, 2976, 2803, 2670, 1660, 1548, 1432, 1377, 1320, 1247, 1102, 1020, 982, 750  $\text{cm}^{-1}$ ; HRMS (ESI):  $[\text{M} + \text{H}]^+$ , found 265.2256,  $\text{C}_{16}\text{H}_{29}\text{N}_2\text{O}$  required 265.2280.

**Exo-11-propanamido-3,4,endo-11-trimethyl-2-ethyl-3-azatricyclo[5.3.1.0<sup>4,9</sup>]undecane 7b.** As red viscous liquid (0.055 g, 26%),  $R_f$  (9:1 ethyl acetate/methanol) 0.30;  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  5.27 (s, 1H, NH-15), 2.57 (br s, 1H, H-1), 2.38 (br s, 3H, NCH<sub>3</sub>), 2.15 (ddd,  $J = 3.0, 7.5, 10.5$  Hz, 2H,  $\text{CH}_2$ -14), 1.98–1.90 (m, 2H, H-6a, H-9), 1.84–1.76 (m, 3H, H-5a, H-7, H-8a), 1.74–1.68 (m, 2H, H-5b, H-10a), 1.66–1.60 (m, 4H, H-6b,  $\text{CH}_2$ -18, H-8b), 1.59 (s, 3H,  $\text{CH}_3$ -12), 1.48–1.42 (m, 1H, H-10b), 1.15 (br s, 3H,  $\text{CH}_3$ -13), 1.13 (t,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -20), 1.07 (t,  $J = 7.5$  Hz, 3H,  $\text{CH}_3$ -19);  $^{13}\text{C NMR}$   $\delta$  175.9 (C=O), 63.7 (C-11/4), 42.0 (CH-2), 40.0 (CH-1), 39.3 (NCH<sub>3</sub>), 39.3 (CH-9), 39.2 (CH-7), 34.1 ( $\text{CH}_2$ -14), 33.8 ( $\text{CH}_2$ -8), 31.6 ( $\text{CH}_2$ -18), 30.3 ( $\text{CH}_3$ -13), 29.7 ( $\text{CH}_2$ -10), 25.1 ( $\text{CH}_3$ -12), 23.9 ( $\text{CH}_2$ -8), 16.7 ( $\text{CH}_2$ -6), 15.3 ( $\text{CH}_3$ -19), 12.7 ( $\text{CH}_3$ -20); FT-IR  $\nu_{\text{max}}$  (neat) 3360, 3200, 2966, 2873, 2652, 1648, 1540, 1467, 1377, 1335, 1240, 1174, 1033, 960, 763  $\text{cm}^{-1}$ ; HRMS (ESI):  $[\text{M} + \text{H}]^+$ , found 293.2576,  $\text{C}_{18}\text{H}_{32}\text{N}_2\text{O}$  required 293.2593.



**Methyl(*E*)-[exo-11-acetylamino-9 $\alpha$ ,endo-11-dimethyl-2-oxo-1,2,4,5,5 $\alpha$ ,6,7,8,9,9 $\alpha$ -decahydro-4,7-methanopyrrolo[1,2-*a*]-quinolin-1-ylidane]ethanoate 10.** Compound 3 (ref. 15) (0.050 g, 0.140 mmol) was dissolved in dry MeOH (10 mL) and placed in a round bottom flask fitted with a drying tube NiCl<sub>2</sub>·6H<sub>2</sub>O (0.100 g, 0.43 mmol) was dissolved in dry MeOH (5 mL) and added to the lactam solution. NaBH<sub>4</sub> (0.200 g, 5.60 mmol, 40 mole equiv.) was added slowly to the mixture while stirring. The mixture was stirred overnight at room temperature. The reaction was quenched with saturated NaHCO<sub>3</sub> (20 mL), extracted with CHCl<sub>3</sub> (20 mL × 2), and washed with saturated NaCl (20 mL). The organic layer was dried over Na<sub>2</sub>CO<sub>3</sub> and solvent was removed under reduced pressure to yield a viscous liquid which upon standing solidified over a day. The product was purified *via* column chromatography to afford the reduced compound 10 as a yellow viscous liquid (0.035 g, 70%); *R*<sub>f</sub> (9:1 ethyl acetate/methanol) 0.53; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  5.48 (br s, 1H, NH-16), 5.09 (s, 1H, H-10), 3.95 (d, *J* = 2.0 Hz, H-7), 3.44 (s, 3H, OCH<sub>3</sub>), 2.91 (dd, *J* = 15.5, 3.5 Hz, 1H, H-21a), 2.81 (dd, *J* = 15.5, 6.5 Hz, 1H, H-21b), 2.14 (s, 1H, H-12), 2.07 (br s, 1H, H-3), 1.70–1.60 (m, 3H, H-6a, H-14a, H-4a), 1.98 (m, 1H, H-5), 1.96 (s, 3H, COCH<sub>3</sub>), 1.95–1.85 (m, 3H, H-13a, H-4b, H-6b), 1.54–1.30 (2H, H-14b, H-13b), 1.50 (s, 3H, CH<sub>3</sub>-15), 1.34 (s, 3H, CH<sub>3</sub>-19); <sup>13</sup>C NMR  $\delta$  201.9 (C-11), 187.1 (C-9), 173.4 (C-17), 172.5 (C-22), 103.5 (CH-10), 62.9 (C-2/8), 62.6 (C-2/8), 61.6 (CH-12), 54.5 (OCH<sub>3</sub>), 40.9 (CH<sub>2</sub>-21), 39.6 (CH-3), 39.0 (CH-7), 38.0 (CH-5), 33.9 (CH<sub>3</sub>-15), 32.8 (CH<sub>2</sub>-13), 32.3 (CH<sub>2</sub>-4), 28.5 (CH<sub>2</sub>-6), 27.4 (CH<sub>3</sub>-25), 24.5 (CH<sub>3</sub>-19), 21.8 (CH<sub>2</sub>-14). HRMS (ESI): [M + H]<sup>+</sup>, found 361.2107, C<sub>20</sub>H<sub>29</sub>N<sub>2</sub>O<sub>4</sub> required 361.2127.

### Cell biology

Compound 3 was screened under the standard conditions of the NCI anticancer screening program, against 48 cell lines covering nine cancer types.<sup>14</sup>

### General reagents for cell culture

Penicillin and streptomycin, trypsin and fetal bovine serum were purchased from Thermo Fisher Scientific (Scoresby, Vic, Australia). Dulbecco's modified Eagle's medium (DMEM), phosphate-buffered saline (PBS) and DMSO were purchased from Sigma-Aldrich (Castle Hill, NSW, Australia). The CellTiter 96® Aqueous One Solution Reagent was purchased from Promega (Alexandria, NSW, Australia).

### Cell culture and viability assays

Human MDA-MB-231 breast cancer cells were obtained as a gift from Prof. Michael Murray (University of Sydney) and grown at 37 °C in a humidified atmosphere of 5% CO<sub>2</sub> in DMEM supplemented with 10% fetal bovine serum and 1% penicillin/streptomycin. Confluent cells (80–90%) were harvested using trypsin/EDTA after washing with PBS.

For the MTS assay cells were seeded in 96-well flat-bottom plates at a density of 7 × 10<sup>3</sup> cells per well. Serum was re-

moved after 24 hours, after which cells were treated with various concentrations of 2a–c, 6a–b and 10 (1–50 μM) in DMSO (maximum concentration 0.1%) for 48 hours; control cells received solvent alone. Cell viability was determined using the CellTiter 96® Aqueous One Solution Cell Proliferation Assay (Promega, USA) as per the manufacturer's recommendation. MTS reagent was added to each well, and the absorbance of all wells was determined by measuring optical density at 550 nm (Infinite M1000Pro, Tecan) after 4 hours incubation at 37 °C. Blank-subtracted absorbance values were normalised to the DMSO control, which was arbitrarily assigned as 100%. IC<sub>50</sub> values were defined as the drug concentration that prevented cell growth of more than 50% relative to the DMSO control and were determined using nonlinear regression analysis with Prism 7.0 (GraphPad Software, CA, USA). Each compound was tested in triplicates. Note: the MTS assay was repeated with human MCF-7 breast cancer cells obtained as a gift from Asso. Prof. Mary Bebawy (University of Technology Sydney).

### Cell cycle analysis

Cell cycle distribution was evaluated in MDA-MB-231 cells that were seeded in 6-well flat bottom plates at a density of 5 × 10<sup>5</sup> cells per well. Serum was removed after 24 hours, after which the cells were treated with various concentrations of drugs (20–40 μM) in DMSO (maximum concentration 0.1%) for 24 hours; control wells only contained DMSO at 0.1%. Treated cells were then harvested and fixed overnight at 0 °C in 80% ethanol, centrifuged at 500 g for 10 minutes at 4 °C. Fixed cells were then washed twice with PBS and resuspended in 0.1 M PBS containing 0.1% (v/v) Triton X-100 and 0.1 mg mL<sup>-1</sup> RNase A. After staining with PI (50 μg mL<sup>-1</sup>), cells were incubated for 30 minutes in the dark at 37 °C and subjected to flow cytometry (BD LSRFortessa X-20), with 1 × 10<sup>4</sup> events, per sample acquired. Cell cycle distribution was calculated using Flowing Software version 2.5.1.

### Apoptosis/necrosis

Annexin V-FITC/PI staining was used to evaluate the extent of apoptosis in MDA-MB-231 cells as per manufacturer's recommendation (BD Bioscience, Sydney, NSW, Australia). Cells were seeded in 6-well plates at a density of 1 × 10<sup>5</sup> cells per well, 24 hours after serum removal the cells were treated with 2c (20–40 μM) for 48 hours. Control wells were treated with DMSO only at 0.1% (v/v). Treated cells were trypsinised and washed twice with cold PBS. The cells were then resuspended in binding buffer (0.1 mL), 5 μL of Annexin V-FITC reagent and PI (50 μg mL<sup>-1</sup>). The cells were incubated for 20 minutes in the dark at 0 °C and subjected to flow cytometry as described above. The percentages of viable (Annexin V-/PI-), early apoptotic (Annexin V+/PI-), late apoptotic (Annexin V+/PI+), and necrotic (Annexin V-/PI+) cells were calculated using Flowing Software version 2.5.1.





## Cytotoxicity against VERO cells

Cytotoxicity of compounds **2c** and **3** against VERO cells (African Green Monkey kidney cell lines) was determined externally by Bioassay Laboratory, BIOTEC, Thailand,<sup>27</sup> using the following procedure.<sup>28</sup>

GFP-expressing Vero cell line was generated in-house by stably transfecting the African green monkey kidney cell line (VERO, ATCC CCL-81) with plasmid carried *gfp* gene (pEGFP-N1, Clontech). The cell line was maintained in minimal essential medium (MEM) supplemented with 10% heat-inactivated fetal bovine serum, 1 mM sodium pyruvate, 2.2 g L<sup>-1</sup> NaHCO<sub>3</sub> and 0.8 mg mL<sup>-1</sup> geneticin and incubated at 37 °C in a humidity incubator with 5% CO<sub>2</sub>. Before performing the assay, cells at a logarithmic growth were harvested and diluted to 3.3 × 10<sup>4</sup> cells per mL suspension in complete medium without geneticin.

Compounds **2c** and **3** were assayed in 384-well plate in a quadruplicate. Briefly, test wells were added with 5 µL of samples (or positive or negative control agents) and followed with 45 µL of cell suspension addition. Positive and negative control agents were ellipticine, and 0.5% DMSO, respectively. The plate was then incubated at 37 °C in humidity incubator with 5% CO<sub>2</sub> for 4 days. Fluorescence was measured in a bottom-reading mode with excitation and emission wavelengths of 485 and 535 nm. The fluorescent signal of day-4 was subtracted with that of day-0 before calculation. The percentage of cytotoxicity was calculated by the following equation:

$$\% \text{ Cytotoxicity} = [1 - (\text{FU}_T/\text{FU}_C)] \times 100$$

where FU<sub>T</sub> and FU<sub>C</sub> are the mean fluorescent unit from cells treated with test compound and that treated with 0.5% DMSO respectively.

## Statistical analysis

All measurements were performed at least in triplicates unless otherwise stated. Data from multiple treatments were analysed by one-way ANOVA in combination with Fisher's protected least significant difference test.

## Conflicts of interest

There are no conflicts to declare.

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